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1 Heavy $\delta^{57}\text{Fe}$ in ocean island basalts: a non-unique signature of
2 processes and source lithologies in the mantle

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18 **Abstract**

19 Lithological heterogeneity is a widely accepted feature of the Earth's mantle, with recycled
20 crustal material accounting for a significant part of heterogeneity in ocean island basalt
21 (OIB) geochemistry. Fe isotopes have been used to link geochemical heterogeneity in OIB
22 sources to distinct mantle lithologies due to their mineral-specific equilibrium fractionation
23 effects, or their composition, such as incorporation of kinetically-fractionated core liquids
24 entrained from the core-mantle boundary.

25 Here we present Fe isotope data for Samoan shield, and Azores volcanoes, together with a
26 combined phase-equilibria and equilibrium mineral-melt isotope fractionation model. These
27 OIB lavas allow us to study the roles of core-derived and recycled mantle components in

28 generating heavy $\delta^{57}\text{Fe}$ melts. Heavy $\delta^{57}\text{Fe}$ correlates with radiogenic isotope signatures of
29 enrichment by a crustal component and not with Fe/Mn or any indicator of core involve-
30 ment. However, single-stage melting of a MORB-like eclogitic pyroxenite cannot generate
31 the heavy $\delta^{57}\text{Fe}$ seen in Pitcairn, Azores, and rejuvenated Samoa lavas. Melts of a reaction-
32 zone pyroxenite (commonly suggested to form part of the OIB source), derived from eclogite
33 melts hybridised with peridotite, also fail to generate the heaviest Fe isotopic compositions
34 seen in OIB. Instead, the generation of heavy $\delta^{57}\text{Fe}$ melts in OIB requires: (1) processes
35 that make subducted eclogite isotopically heavier than its pristine precursor MORB (e.g.,
36 hydrothermal alteration, metamorphism, sediment input); (2) lithospheric processing, such
37 as remobilisation of previously frozen small-degree melts, or a contribution from lithospheric
38 material metasomatised by silicate melts; and/or (3) melting conditions that limit the di-
39 lution of melts with heavy $\delta^{57}\text{Fe}$ by ambient lower $\delta^{57}\text{Fe}$ materials. No single process we
40 consider can generate the heavy $\delta^{57}\text{Fe}$ seen in the Azores, Pitcairn, and rejuvenated Samoan
41 lavas.

42 Therefore, it cannot be assumed that a pyroxenite lithology derived from recycled crustal
43 material is the sole producer of heavy $\delta^{57}\text{Fe}$ melts in OIB, nor can these signatures be
44 related to contributions from the Earth's core. Instead, the observation of heavy $\delta^{57}\text{Fe}$
45 OIB melts cannot be ascribed to a unique source or process. This ambiguity reflects the
46 multitude of processes operating from the generation of recycled lithologies through to their
47 mantle melting: from MORB generation, its low temperature alteration, through mantle
48 heterogeneity development and lithospheric processing, to eruption at ocean islands.

49 1 Introduction

50 Numerous isotopic and trace element studies of ocean island basalts (OIB) suggest that the
51 Earth's mantle is heterogeneous on scales from hundreds of kilometres, across entire man-
52 tle plume systems, to less than a kilometre or metres, as recorded by melt inclusions (e.g.,
53 Zindler and Hart, 1986; Weaver, 1991; Hofmann, 1997; Stracke et al., 2005; MacLennan, 2008).
54 The variance in long-lived radiogenic isotopic composition (Sr-Nd-Pb) in most OIB can be ex-
55 plained by mixing between five primary components: depleted mantle, the prevalent mantle
56 (PREMA)/focal zone (FOZO) component, two enriched mantle (EM1/EM2) components and a
57 HIMU component (Stracke, 2012; Fig. 1). Radiogenic isotopes have also been used to argue for
58 enriched (and depleted) components in the mid-ocean ridge basalt source (e.g., Hirschmann and

59 Stolper, 1996; Salters and Dick, 2002; Liu et al., 2008; Byerly and Lassiter, 2014). In addition to
60 these components, osmium isotopes (Walker et al., 1995; Brandon et al., 1998), combined noble
61 gas and short-lived radiogenic isotopic systematics (e.g., $^3\text{He}/^4\text{He}$ and $^{182}\text{W}/^{184}\text{W}$ anomalies;
62 Mundl et al., 2017; Mundl-Petermeier et al., 2020) and transition element ratios (e.g., Fe/Mn;
63 Humayun et al., 2004) have been used to support the presence of a small amount of a primordial,
64 lower mantle or core-derived component entrained in some OIB.

65 Linking the heterogeneity in long-lived radiogenic isotopes to mantle lithology has been achieved
66 through combining radiogenic and stable isotopes (e.g., Day et al., 2009, 2010), major element
67 compositions of basalts (Jackson and Dasgupta, 2008; Shorttle and MacLennan, 2011) and trace
68 elements in olivine (e.g., Sobolev et al., 2005, 2007; Herzberg, 2011; Neave et al., 2018). These
69 techniques have led to the widely accepted view that recycled oceanic crust generates a signif-
70 icant amount of the variation seen in OIB geochemistry, along with small contributions from
71 sediments and continental crust (e.g., Cohen and O’Nions, 1982; Hofmann and White, 1982;
72 Weaver, 1991; Chauvel et al., 1992; see also Stracke, 2012). The recycled basaltic compo-
73 nent is thought to be present as discrete eclogite or, more commonly, pyroxenite (olivine-poor,
74 pyroxene-rich) lithologies, the latter possibly formed by solid-state or melt reaction of eclogite
75 with ambient peridotite (Sobolev et al., 2007; Herzberg, 2011). However, the success of using
76 major elements to identify pyroxenite in OIB sources may depend on the type of enriched com-
77 ponent invoked (Lambart et al., 2013); and due to uncertainties over the melting conditions and
78 role of crustal processes (magma recharge events, mixing, diffusional reequilibration; Matzen
79 et al., 2017; Hole, 2018; Gleeson and Gibson, 2019) it is unclear whether source compositional
80 differences unambiguously control the trace element concentrations in olivine phenocrysts.

81 Non-traditional stable isotope systems, such as Fe, provide an important, alternative method
82 for identifying lithological heterogeneity (of both enriched and depleted components) in the OIB
83 source. Unlike incompatible trace elements and radiogenic isotopes of incompatible elements,
84 both of which are likely to be disproportionately affected by contributions from enriched mantle
85 components (e.g., Burton et al., 2012), the Fe abundances of melts derived from pyroxenitic
86 and peridotitic source mineralogies are similar (e.g., Sobolev et al., 2005). Therefore, neither
87 pyroxenite nor peridotite lithologies should dominate the Fe isotopic composition of erupted
88 melts, and the Fe isotopic composition of the lavas should reflect the relative contributions

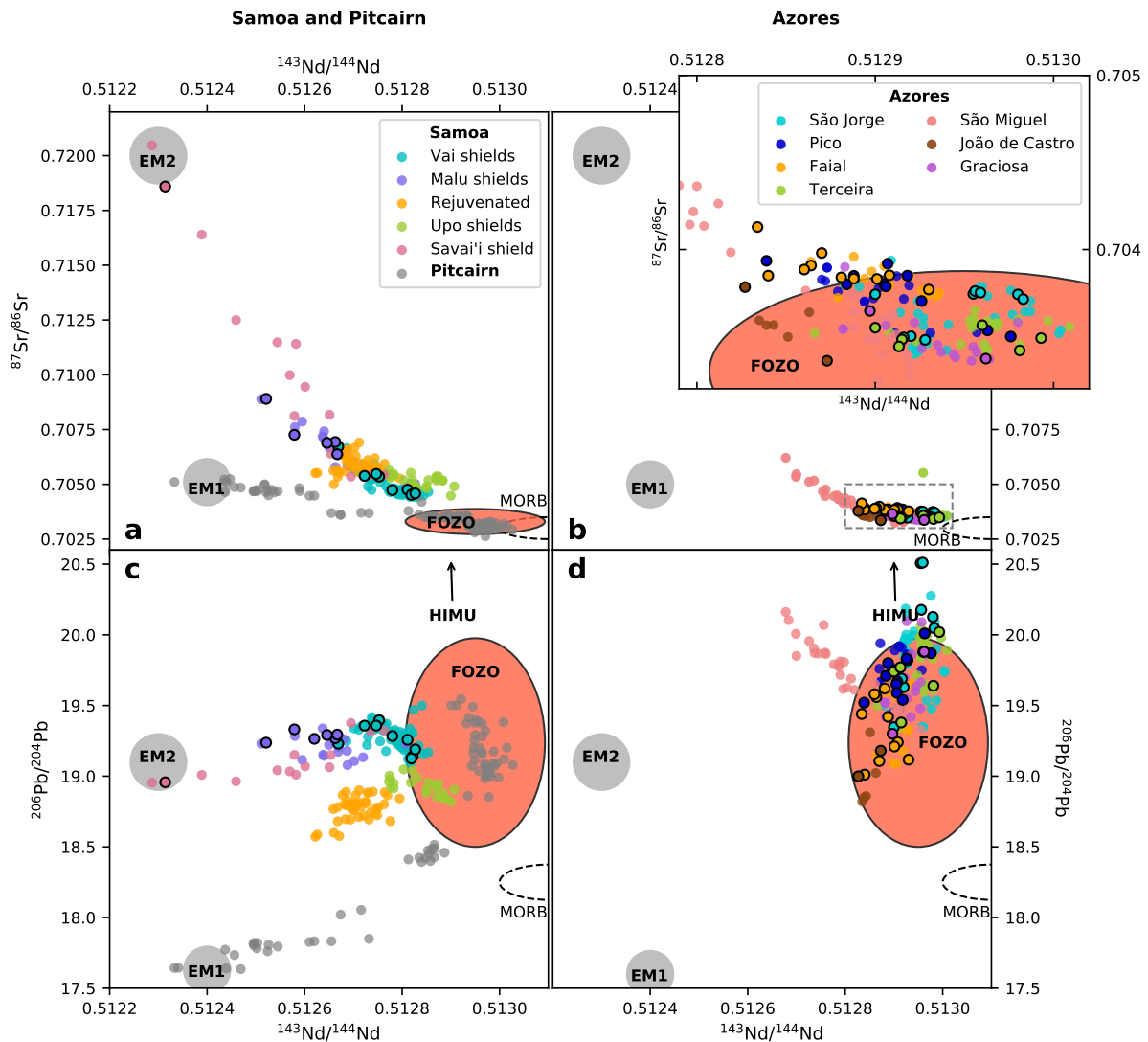


Figure 1: Compilation of Samoa, Pitcairn and Azores radiogenic isotope data showing their different geochemical trends, and comparing them to average MORB (black dashed region; Stracke et al., 2005) and global mantle endmembers (red and grey ovals: EM2 from Jackson et al., 2007b; EM1, HIMU and FOZO – FOZO shown with 2σ uncertainty – from Konter et al., 2008). Samples in this study outlined in black. Pitcairn lavas show a compositional range between FOZO and EM1, Samoan shield lavas between FOZO and EM2 with Samoan rejuvenated lavas showing a weak EM1 component, and Azores lavas between FOZO and HIMU with eastern São Miguel not trending to any global endmember. Samoan data (panels a, c) from Palacz and Saunders (1986); Wright and White (1987); Farley et al. (1992); Hauri and Hart (1993, 1997); Workman et al. (2004); Jackson et al. (2007a,b, 2010, 2014); Hart and Jackson (2014); Pitcairn data (panels a, c) from GEOROC; Azores data (panels b, d) from Turner et al. (1997); Moreira et al. (1999); Beier et al. (2007); Elliott et al. (2007); Beier et al. (2008); Millet et al. (2009); Watanabe (2010); Madureira et al. (2011); Yu (2011); Beier et al. (2012); Hildenbrand et al. (2014); Larrea et al. (2014); Genske et al. (2016); Béguelin et al. (2017); Waters et al. (2020).

89 from these different lithologies (Williams and Bizimis, 2014). Since Fe isotopes show mineral-
 90 specific fractionation effects, they provide a unique opportunity to explore the petrological and
 91 mineralogical characteristics of mantle heterogeneity, and could help unravel common melting
 92 lithologies and processes in the mantle (Williams and Bizimis, 2014; Konter et al., 2016; Nebel
 93 et al., 2018, 2019; Gleeson et al., 2020).

94 Equilibrium inter-mineral Fe isotope fractionations are driven by the different bonding envi-
 95 ronments of Fe in mineral structures (Macris et al., 2015; Young et al., 2015). Strong, short
 96 Fe–O bonds (due to low coordination numbers and/or the presence of oxidised Fe³⁺) concen-
 97 trate heavy Fe, i.e., more ⁵⁷Fe favoured over ⁵⁴Fe (Sossi and O’Neill, 2017), which is reported
 98 as higher δ⁵⁷Fe, where

$$\delta^{57}\text{Fe} = \left(\frac{(^{57}\text{Fe}/^{54}\text{Fe})_{\text{sample}}}{(^{57}\text{Fe}/^{54}\text{Fe})_{\text{IRMM-014}}} - 1 \right) \times 1000 \quad (1)$$

99 This equilibrium fractionation effect suggests that an olivine-dominated peridotite will be iso-
 100 topically lighter than a more pyroxene-rich (pyroxenite) recycled component (Macris et al., 2015;
 101 Sossi and O’Neill, 2017) due to the presence of a small amount of Fe³⁺ in pyroxene, and hence
 102 stronger Fe–O bonds, relative to olivine. Similarly, pure iron metal (Fe⁰) is expected to be
 103 isotopically lighter than lower mantle silicate (Fe^{2+,3+}) due to the difference in Fe valence state
 104 between the phases (Shahar et al., 2016), although this effect is predicted to be small at pres-
 105 sures and temperatures relevant to core equilibration (Shahar et al., 2016; Liu et al., 2017) and
 106 unlikely to be relevant to OIB. Kinetic fractionation effects may also contribute to Fe isotopic
 107 variation in OIB melt sources, with thermodiffusion (Soret diffusion) in material diffusing from
 108 the outer core into the lowermost mantle recently proposed to generate heavy δ⁵⁷Fe in entrained
 109 plume material (Leshner et al., 2020).

110 Qualitatively consistent with an isotopically heavy recycled source component, OIB show vari-
 111 able δ⁵⁷Fe relative to average N- and T-MORB (the latter two dominated by peridotite melting),
 112 generally extending to heavier compositions, such as over 0.25‰ in Samoa and Pitcairn. These
 113 heavy Fe isotopic compositions are suggested to relate to pyroxenitic mantle components in
 114 the OIB source. There are some N-MORB with δ⁵⁷Fe as heavy as 0.2‰, which could also be
 115 consistent with indications of small amounts of enriched pyroxenite or eclogite in the MORB

116 source (c.f., Hirschmann and Stolper, 1996). The radiogenic isotope systematics of hotspots
117 (plumes) that display heavy $\delta^{57}\text{Fe}$ signatures show mixing between a common peridotitic man-
118 tle component and recycled crustal endmembers (Konter et al., 2016; Nebel et al., 2019; Fig.
119 1), and in several cases Fe isotopes correlate positively with indices of recycling (e.g., Sr-Nd-Pb
120 isotopes; Nebel et al., 2019). However, it is unclear whether mixing between different mantle
121 components is represented to the same extent in Fe isotopes as in long-lived radiogenic isotopes,
122 and whether Fe and radiogenic isotopic systems can be linked to identify the mineralogy (e.g.,
123 pyroxene enrichment) of different mantle components identified in Fig. 1. In using heavy Fe
124 isotopic compositions to better understand mantle heterogeneity, both the sources and processes
125 generating heavy $\delta^{57}\text{Fe}$ melts in OIB need to be considered.

126 The Fe isotopic compositions recorded in OIB are heavier (and more variable) than the expected
127 equilibrium isotopic compositions of melts from crustal endmembers contributing to mantle
128 heterogeneity. Isotopic composition estimates for the mantle are $\delta^{57}\text{Fe} = 0.03 \pm 0.03\text{‰}$ (DM;
129 Craddock et al., 2013) to $0.05 \pm 0.01\text{‰}$ (BSE; Sossi et al., 2016); for average oceanic crust,
130 represented by MORB, are $\delta^{57}\text{Fe} = 0.15\text{‰}$ (Teng et al., 2013; Sossi et al., 2016); and bulk
131 continental crust is indistinguishable from, or lighter than, oceanic crust ($\delta^{57}\text{Fe} = 0.08\text{--}0.16\text{‰}$;
132 Johnson et al., 2020). Highly differentiated ($\text{SiO}_2 > 70 \text{ wt}\%$) crust, which could contribute to
133 continentally-derived sediment, records $\delta^{57}\text{Fe} < 0.9\text{‰}$ (Du et al., 2017) although the average
134 $\delta^{57}\text{Fe}$ of rocks with $\text{SiO}_2 > 60 \text{ wt}\%$ is around 0.3‰ (Johnson et al., 2020). Experimental and
135 theoretical estimates of fractionation during partial mantle melting are known to be small (e.g.,
136 Dauphas et al., 2009; Sossi and O'Neill, 2017; Gleeson et al., 2020), likely $\Delta^{57}\text{Fe}_{\text{melt-mantle}}$ (=
137 $\delta^{57}\text{Fe}_{\text{melt}} - \delta^{57}\text{Fe}_{\text{mantle}}$) $< 0.1\text{‰}$ depending on Fe^{3+} buffering in the mantle (Dauphas et al.,
138 2014). However, data from Pitcairn and Galapagos Spreading Centre lavas require a mantle
139 component with $\delta^{57}\text{Fe} = 0.30\text{‰}$ (Nebel et al., 2019; Gleeson et al., 2020). The heaviest isotopic
140 values in OIB from Samoa have previously been explained by combining source heterogeneity,
141 partial melting, and fractional crystallisation effects (Konter et al., 2016). However, even in this
142 multi-process scenario, equilibrium fractionation factors for these processes are required to be
143 high (crystallisation with $\Delta^{57}\text{Fe}_{\text{olivine-melt}} = -0.45\text{‰}$, and mantle melting with $\Delta^{57}\text{Fe}_{\text{melt-mantle}}$
144 $> 0.15\text{‰}$; Konter et al., 2016), and possibly unrealistically high (Gleeson et al., 2020). It
145 therefore remains unclear if the heaviest Fe isotopic values in OIB ($\delta^{57}\text{Fe} > 0.25\text{‰}$) can be
146 explained by simple melting processes of recycled crustal components embedded in ambient

147 plume mantle.

148 Samoan rejuvenated lavas (a later, volumetrically less significant stage than the main shield
149 lavas, erupted far from the plume; Natland, 1980) record the heaviest Fe isotopic compositions
150 in the global OIB dataset (Konter et al., 2016). Samoan shield samples also show correlated
151 $^3\text{He}/^4\text{He}-\mu^{182}\text{W}$, with $^3\text{He}/^4\text{He} \leq 33.8 \text{ R}/\text{R}_a$ (Jackson et al., 2007a) and $\mu^{182}\text{W} \geq -17.3$ (Mundl
152 et al., 2017), proposed to relate to core-equilibrated material (Mundl-Petermeier et al., 2020).
153 Thus, Samoan lavas were selected for further Fe isotopic characterisation. We have expanded
154 the shield lava dataset allowing us to study pyroxenite versus core contributions to heavy $\delta^{57}\text{Fe}$
155 liquids, as the two contributions could be associated with other distinct geochemical signa-
156 tures. The Azores was chosen as a second OIB locality because it shows radiogenic isotopic
157 mixing between a common Azores mantle component and recycled components, so could allow
158 identification of recycled mantle pyroxenite. There may also be a minor lower mantle or core
159 component associated with the Azores plume, identified by raised $^3\text{He}/^4\text{He}$ relative to MORB
160 ($\leq 18.4 \text{ R}/\text{R}_a$; Moreira et al., 2012) and small negative $\mu^{182}\text{W}$ anomalies (≥ -9.9 ; Mundl-
161 Petermeier et al., 2020). Unlike Samoa, the Azores is also (1) a cooler plume ($> 1400^\circ\text{C}$; Beier
162 et al., 2012, compared to $> 1600^\circ\text{C}$ in Samoa; Putirka et al., 2018) meaning melts of enriched
163 (and possibly heavy $\delta^{57}\text{Fe}$) components will be minimally obscured by contemporaneous melting
164 of ambient, relatively depleted mantle; and (2) shows a well spatially-resolved distribution of
165 melts from different components sampled by volcanoes (Béguelin et al., 2017; Beier et al., 2018;
166 Fig. 1). Therefore, Samoa and the Azores offer different perspectives on the links between
167 distinct mantle components and the source of heavy $\delta^{57}\text{Fe}$ lavas.

168 **2 Samples and methods**

169 **2.1 Samoa**

170 The Samoan islands show an age-progressive volcanic track (Koppers et al., 2011), with a shield
171 building stage influenced by multiple mantle components (Jackson et al., 2014; Fig. 1). These
172 components include the global EM2 endmember, which is distinguished by high $^{87}\text{Sr}/^{86}\text{Sr}$ rel-
173 ative to FOZO (but FOZO-like $^{206}\text{Pb}/^{204}\text{Pb}$) and proposed to relate to recycled continental

174 sediment (White and Hofmann, 1982; Jackson et al., 2007b); and a high $^3\text{He}/^4\text{He}$ -negative
175 $\mu^{182}\text{W}$ common plume component (Jackson et al., 2007a; Mundl et al., 2017), proposed to re-
176 late to a contribution from the outer core (Mundl-Petermeier et al., 2020). Recent shield-stage
177 volcanism is split into two sub-parallel and geochemically distinct volcanic chains (Fig. 1), sim-
178 ilar to Hawai'i's Loa and Kea trends (Workman et al., 2004; Huang et al., 2011; Koppers et al.,
179 2011): the Vai and Malu trends. Subsequently, rejuvenated lavas have erupted on the older
180 islands of Savai'i and Upolu (Konter and Jackson, 2012). Rejuvenated lavas record heavy Fe
181 isotopic signatures of $\delta^{57}\text{Fe} \geq 0.3\text{‰}$ (Konter et al., 2016), and an EM1-like component (Work-
182 man et al., 2004; Konter and Jackson, 2012; Reinhard et al., 2019; Fig. 1c). Metasomatism of
183 the lithosphere by silicate melts and/or mineralogical heterogeneity (a pyroxenite component)
184 are proposed as the most likely origin of the heavy $\delta^{57}\text{Fe}$ in Samoa (Konter et al., 2016).

185 The 14 new Samoan samples we studied are subaerial and submarine shield lavas, covering six
186 volcanic centres (Savai'i, Tutuila, Malumalu, Ofu, Ta'u and Vailulu'u). These samples cover
187 the range of the long-lived radiogenic isotope space spanned by Samoan shield lavas (Fig. 1),
188 and include high $^3\text{He}/^4\text{He}$, negative $\mu^{182}\text{W}$ samples (Ofu; Mundl et al., 2017). Major and trace
189 element, and radiogenic isotopic data has been published previously (Vailulu'u and Malumalu
190 seamounts and subaerial Ta'u lavas from Workman et al., 2004; Ofu from Jackson et al., 2007a,
191 2010; Hart and Jackson, 2014). Only samples with greater than 5 wt% MgO were selected, to
192 limit the effects of fractional crystallisation, particularly magnetite fractionation (Williams et al.,
193 2018). Three high MgO samples (> 18 wt%) were chosen to explore the $\delta^{57}\text{Fe}$ variability that
194 could be introduced to lavas by olivine accumulation during melt storage and transport. The
195 Savai'i sample (ALIA115-18) is a dredged lava with heavy $\delta^{57}\text{Fe}$ (0.36‰) previously measured
196 by Konter et al. (2016), which was suggested to have undergone seawater alteration based on
197 trace element data (combined high LOI of 4.4 wt% and Ba/Rb < 6.7 ; Konter et al., 2016).
198 This sample was the only sample available where alteration is documented, and therefore was
199 measured to get an idea of the Fe isotopic variability that could be induced by alteration. The
200 other samples do not show a trace element signature of alteration (Fig. S1).

201 **2.2 Azores**

202 The Azores consist of nine volcanic islands, on the west and east of the Mid-Atlantic Ridge, and
203 associated with a triple junction. Many of the islands are still active, with most subaerial lavas
204 less than 1 Myr old (Larrea et al., 2018). The Azores volcanoes sample a number of recycled
205 components identified in radiogenic isotopic space, spanning much of the global variation of
206 OIB (Beier et al., 2018), although the eastern São Miguel component is not a globally-defined
207 endmember (Fig. 1). São Jorge lavas have the most radiogenic $^{206}\text{Pb}/^{204}\text{Pb}$ from the Azores,
208 a signature attributed to the presence of a HIMU-like recycled oceanic crustal component (Mil-
209 let et al., 2009). These recycled components are mixed with a common Azores component,
210 but appear to undergo limited mixing with each other (Beier et al., 2018). Limited mixing is
211 hypothesised to occur in part because the Terceira rift drains melts from fusible mantle compo-
212 nents to the surface, allowing only short scale horizontal migration and mixing of melts between
213 volcanoes towards the Mid-Atlantic Ridge (Béguelin et al., 2017). However, the presence of
214 a recycled oceanic crustal component in Azores lavas is debated, with uranium series disequi-
215 libria (Prytulak and Elliott, 2009) and transition element ratios (Beier et al., 2013) consistent
216 with a garnet peridotite source and not requiring pyroxenite/eclogite. It is also suggested that
217 carbonated peridotitic mantle may be present in the Azores source region (Beier et al., 2013),
218 explaining melting despite a relatively small thermal anomaly (Beier et al., 2012). Additionally,
219 a negative $\mu^{182}\text{W}$ signature has been recorded in the Azores, suggesting a lower mantle or core
220 component entrained in the Azores plume (Rizo et al., 2019; Mundl-Petermeier et al., 2020).

221 Here we present Fe isotope data for São Jorge, Terceira, Pico, Faial, João de Castro and Graciosa.
222 Major and trace element, and radiogenic isotopic data are from Turner et al. (1997); Beier et al.
223 (2008); Millet et al. (2009); Watanabe (2010); Yu (2011); Beier et al. (2010, 2012); Béguelin
224 et al. (2017); Waters et al. (2020). Only samples with MgO > 5 wt% were selected, to limit the
225 effects of fractional crystallisation.

226 **2.3 Methods**

227 Iron isotope analyses were carried out following established procedures (Williams et al., 2005,
228 2009; Cheng et al., 2014) in the laboratories identified in Table 1. In brief, ~ 20 mg of whole-rock

229 powders were dissolved in a mixture of HF-HNO₃ on a hotplate at 120°C for up to 72 hours,
230 evaporated to dryness and subsequently refluxed in either 6M HCl (Cambridge/Durham), or
231 concentrated HNO₃ and H₂O₂ (Monash University) to remove fluoride bonds. Samples were
232 taken up in 6M HCl (Cambridge/Durham) or 9M HCl (Monash) for column chemistry. Iron
233 purification was achieved through progressive elution in anion exchange resins (AG1-X4 or AG-
234 MP-1), first of matrix followed by Fe, with purified Fe redissolved in 0.1M HNO₃ for isotopic
235 analysis. Full details of methods are given in the Appendix.

236 Analyses were performed on a MC-ICP-MS NeptunePlus in Cambridge, Durham and Monash
237 in wet plasma and medium resolution, with sample standard bracketing to the IRMM-014 (or
238 isotopically identical IRMM-524a) standard. The international geological reference materials
239 BHVO-2, BCR-2 and BIR-1 were used to evaluate column chemistry procedures, and give values
240 in agreement with published values (Craddock and Dauphas, 2010; Table S1 in Appendix). The
241 in-house FeCl₃ standard was used to assess mass dependence, reproducibility and accuracy in
242 Cambridge and Durham, giving $\delta^{56}\text{Fe} = -0.73 \pm 0.03\text{‰}$ (2 S.D.), $\delta^{57}\text{Fe} = -1.08 \pm 0.05\text{‰}$ in
243 Cambridge and $\delta^{56}\text{Fe} = -0.69 \pm 0.03\text{‰}$, $\delta^{57}\text{Fe} = -1.03 \pm 0.05\text{‰}$ in Durham. The long term 2
244 S.D. reproducibility of $\delta^{57}\text{Fe}$ analyses is therefore taken as $\pm 0.05\text{‰}$.

245 **3 Results**

246 The measured Fe isotope compositions for Samoa and the Azores are given in Table 1. We
247 discuss the Fe-isotope systematics of the two localities separately below.

248 **3.1 Samoa**

249 The measured Fe isotope composition of Samoan shield lavas ranges from $\delta^{57}\text{Fe} = 0.07\text{--}0.21\text{‰}$
250 ($\pm 0.02\text{‰}$, 2 S.E.), with an average of $\delta^{57}\text{Fe} = 0.15\text{‰}$, coinciding with average MORB (Teng
251 et al., 2013). There is no systematic variation between Vai ($\delta^{57}\text{Fe} = 0.07\text{--}0.20\text{‰} \pm 0.02\text{‰}$ 2
252 S.E.) and Malu ($\delta^{57}\text{Fe} = 0.15\text{--}0.21\text{‰} \pm 0.02\text{‰}$ 2 S.E.) trend lavas. Duplicate dissolutions of
253 samples ALIA115-18 and AVON-3-71-22 reproduce within 2 S.D. of each other (Table 1). The
254 shield lava dataset measured in this study is comparable with previously published Samoan

Table 1: Iron isotope data for samples and standards measured in this study. Samples measured at [†]University of Cambridge, [‡]Durham University, [§]Monash University. *d* represents duplicate dissolution of sample. Data from Konter et al. (2016) for the shared sample with this study is shown for reference, with $\delta^{57}\text{Fe}$ and 2 S.E. calculated assuming mass dependence from published $\delta^{56}\text{Fe}$ value. S.E. calculated as S.D./ \sqrt{n} ; n = number of repeat measurements. Major element data sources for Samoa: Workman et al., 2004; Jackson et al., 2007a, 2010; Hart and Jackson, 2014, the Azores: Turner et al., 1997; Beier et al., 2008; Millet et al., 2009; Beier et al., 2010; Yu, 2011; Beier et al., 2012; Waters et al., 2020.

Volcano	Sample	MgO (wt%)	FeO _T (wt%)	$\delta^{56}\text{Fe}$ (‰)	2 S.D.	2 S.E.	$\delta^{57}\text{Fe}$ (‰)	2 S.D.	2 S.E.	n
Samoa[†]										
Savai'i	ALIA115-18	5.09	7.09	0.12	0.05	0.02	0.19	0.05	0.02	7
	ALIA115-18 <i>d</i> <i>Konter, 115-18</i>			0.12 <i>0.24</i>	0.04 <i>0.04</i>	0.02 <i>0.04</i>	0.20 <i>(calc) 0.36</i>	0.01 <i>0.06</i>	0.01 <i>0.06</i>	3
Malumalu	AVON-3-77-1	7.62	12.17	0.13	0.03	0.02	0.21	0.05	0.02	4
Malumalu	AVON-3-77-9	8.45	11.08	0.13	0.06	0.03	0.18	0.08	0.03	5
Malumalu	ALIA106-03	6.56	12.97	0.13	0.03	0.01	0.21	0.04	0.01	8
Malumalu	AVON-3-78-1	19.68	11.89	0.10	0.03	0.01	0.15	0.05	0.02	9
Malumalu	AVON-3-76-8	6.27	11.93	0.12	0.04	0.02	0.19	0.07	0.03	5
Tutuila	ALIA112-02	6.68	11.11	0.10	0.04	0.02	0.15	0.04	0.02	6
Ofu	OFU-04-14	16.02	12.70	0.07	0.04	0.01	0.12	0.02	0.01	8
Ofu	OFU-05-18	9.81	12.78	0.07	0.02	0.01	0.12	0.04	0.01	11
Vailulu'u	AVON-3-73-1	7.52	11.26	0.11	0.01	0.01	0.14	0.03	0.01	5
Vailulu'u	AVON-3-70-9	6.47	11.65	0.10	0.04	0.02	0.16	0.05	0.02	6
Vailulu'u	AVON-3-63-2	23.87	10.39	0.08	0.03	0.01	0.13	0.04	0.02	5
Vailulu'u	AVON-3-71-22	21.79	10.77	0.05	0.06	0.02	0.07	0.07	0.02	11
Vailulu'u	AVON-3-71-22 <i>d</i>			0.08	0.01	0.01	0.13	0.04	0.02	3
Ta'u	T33	12.40	11.61	0.12	0.03	0.01	0.21	0.03	0.02	4
Azores										
São Jorge	SJ01 [‡]	8.24	10.43	0.20	0.02	0.01	0.34	0.05	0.02	5
São Jorge	SJ05 [‡]	8.95	12.05	0.13	0.06	0.03	0.24	0.05	0.03	4
São Jorge	SJ07 [‡]	8.45	11.56	0.14	0.02	0.01	0.24	0.08	0.03	6
São Jorge	SJ50 [‡]	8.84	10.70	0.14	0.04	0.02	0.22	0.06	0.03	3
São Jorge	SJ51 [‡]	11.99	11.18	0.18	0.04	0.02	0.23	0.03	0.02	5
São Jorge	SJ52 [‡]	8.20	11.59	0.14	0.02	0.01	0.23	0.03	0.01	6
São Jorge	SJ30 [‡]	9.70	11.32	0.17	0.05	0.03	0.28	0.07	0.04	3
São Jorge	SJ31b [‡]	8.91	12.40	0.25	0.09	0.05	0.28	0.04	0.02	3
São Jorge	WASJ18 [‡]	9.72	10.35	0.20	0.06	0.02	0.28	0.06	0.02	6
Terceira	T2 [‡]	5.54	10.53	0.12	0.03	0.02	0.14	0.03	0.02	3
Terceira	T6 [‡]	8.70	11.03	0.14	0.05	0.03	0.16	0.04	0.02	3
Terceira	T18 [‡]	7.94	10.48	0.16	0.04	0.02	0.22	0.02	0.01	5
Terceira	WAT3 [‡]	11.80	8.68	0.15	0.02	0.01	0.20	0.02	0.01	4
Terceira	AZT-03-12 [§]	5.45	12.44	0.02	0.00	0.00	0.05	0.03	0.02	3
Terceira	AZT-03-016 [§]	5.67	10.88	0.07	0.04	0.02	0.13	0.03	0.02	3
Pico	P4 [‡]	10.36	9.94	0.14	0.04	0.03	0.18	0.03	0.02	2
Pico	P5 [‡]	9.63	9.80	0.20	0.06	0.04	0.24	0.02	0.02	2
Pico	P25 [‡]	8.24	9.99	0.18	0.05	0.02	0.22	0.05	0.02	5
Pico	P26 [‡]	10.83	10.28	0.16	0.03	0.01	0.24	0.05	0.02	6
Pico	P29 [‡]	8.16	9.31	0.22	0.08	0.04	0.28	0.05	0.02	4
Pico	WAP9 [‡]	13.21	9.14	0.15	0.03	0.02	0.17	0.06	0.04	2
Pico	AZP-03-10 [§]	6.60	10.82	0.16	0.03	0.02	0.24	0.03	0.02	3
Pico	AZP-03-23 [§]	5.79	11.41	0.17	0.04	0.02	0.25	0.05	0.03	3
Pico	AZP-03-50 [§]	6.00	10.53	0.17	0.02	0.01	0.27	0.00	0.00	3
Faial	F/FG-28 [‡]	8.13	9.75	0.18	0.06	0.03	0.26	0.03	0.01	5
Faial	F/CP-18 [‡]	7.83	9.32	0.20	0.03	0.01	0.27	0.02	0.01	6
Faial	F/CA-6 [‡]	9.73	9.66	0.20	0.03	0.01	0.26	0.04	0.02	5
Faial	F/CA-24 [‡]	13.99	10.21	0.15	0.07	0.04	0.19	0.03	0.02	4
Faial	WAF1a [‡]	13.51	9.77	0.14	0.08	0.04	0.26	0.05	0.03	4
Faial	AZF-03-11 [§]	5.47	9.73	0.23	0.01	0.01	0.34	0.03	0.02	3
João de Castro	525DS2 [§]	15.78	10.74	0.14	0.02	0.01	0.27	0.00	0.00	3
João de Castro	523DS2 [§]	6.25	9.34	0.15	0.01	0.01	0.21	0.05	0.03	2
Graciosa	AZG-03-07 [§]	8.17	10.36	0.26	0.04	0.02	0.37	0.02	0.01	2
Graciosa	AZG-03-28 [§]	7.02	9.52	0.18	0.02	0.01	0.27	0.08	0.04	3

255 shield data (Konter et al., 2016). The rejuvenated lavas from Konter et al. (2016) are much
256 heavier ($\delta^{57}\text{Fe} > 0.3\text{‰}$) than any of the combined dataset of shield lavas. Our measurement of
257 sample ALIA115-18, the common sample between the two studies, is $\Delta^{57}\text{Fe} = 0.18\text{‰}$ lighter
258 than that reported by Konter et al. (2016).

259 **3.2 Azores**

260 The Fe isotope composition of Azores samples shows considerable variation, ranging from $\delta^{57}\text{Fe}$
261 $= 0.05\text{--}0.37\text{‰}$ ($\pm 0.02\text{‰}$ 2 S.E.), with all but three samples displaying $\delta^{57}\text{Fe} > 0.15\text{‰}$ (average
262 MORB, Teng et al., 2013) even at the same MgO content as MORB (Fig. 2). Two of these are
263 Terceira samples (T2, AZT-03-016) with slightly low TiO_2 for their MgO (Fig. S4), which may
264 indicate they have experienced early magnetite saturation (Williams et al., 2018) or formed as
265 a result of mixing between primitive melts unsaturated in magnetite and magnetite-saturated
266 evolved melts, and therefore are not considered further. No samples have $\text{Ba/Rb} < 9$, and there
267 is no correlation between $\delta^{57}\text{Fe}$ and Ba/Rb or LOI (Fig. S2). There is no apparent difference
268 in $\delta^{57}\text{Fe}$ between different islands.

269 **4 What processes could generate heavy $\delta^{57}\text{Fe}$ liquids?**

270 Several processes can modify the whole rock $\delta^{57}\text{Fe}$ from the primary liquid $\delta^{57}\text{Fe}$, and the primary
271 liquid $\delta^{57}\text{Fe}$ from the isotopic composition of ambient peridotite (0.05‰). In this section, we
272 consider how: (1) post melt emplacement processing (fractional crystallisation, olivine accumu-
273 lation); (2) partial melting; and (3) pre-melt emplacement considerations of mantle lithological
274 heterogeneity, may contribute to the heavy $\delta^{57}\text{Fe}$ ($\gtrsim 0.20\text{‰}$) seen in Samoa and the Azores.

275 **4.1 Olivine accumulation and fractional crystallisation**

276 Olivine accumulation is known to produce light whole-rock $\delta^{57}\text{Fe}$ (Teng et al., 2008; McCoy-West
277 et al., 2018). Olivine accumulation is likely for three Samoan samples measured here, as they
278 have high (> 18 wt%) MgO content (Fig. 2). It is possible to correct for olivine accumulation
279 by modelling the removal of olivine (e.g., McCoy-West et al., 2018), however it is unlikely that

280 the melt has accumulated olivine that is in equilibrium with the final bulk composition which
 281 makes the correction hard to use with reliability. Since there is no statistical $\delta^{57}\text{Fe}$ difference
 282 between the high and low MgO shield lavas, we include the (uncorrected) high MgO samples in
 283 following plots.

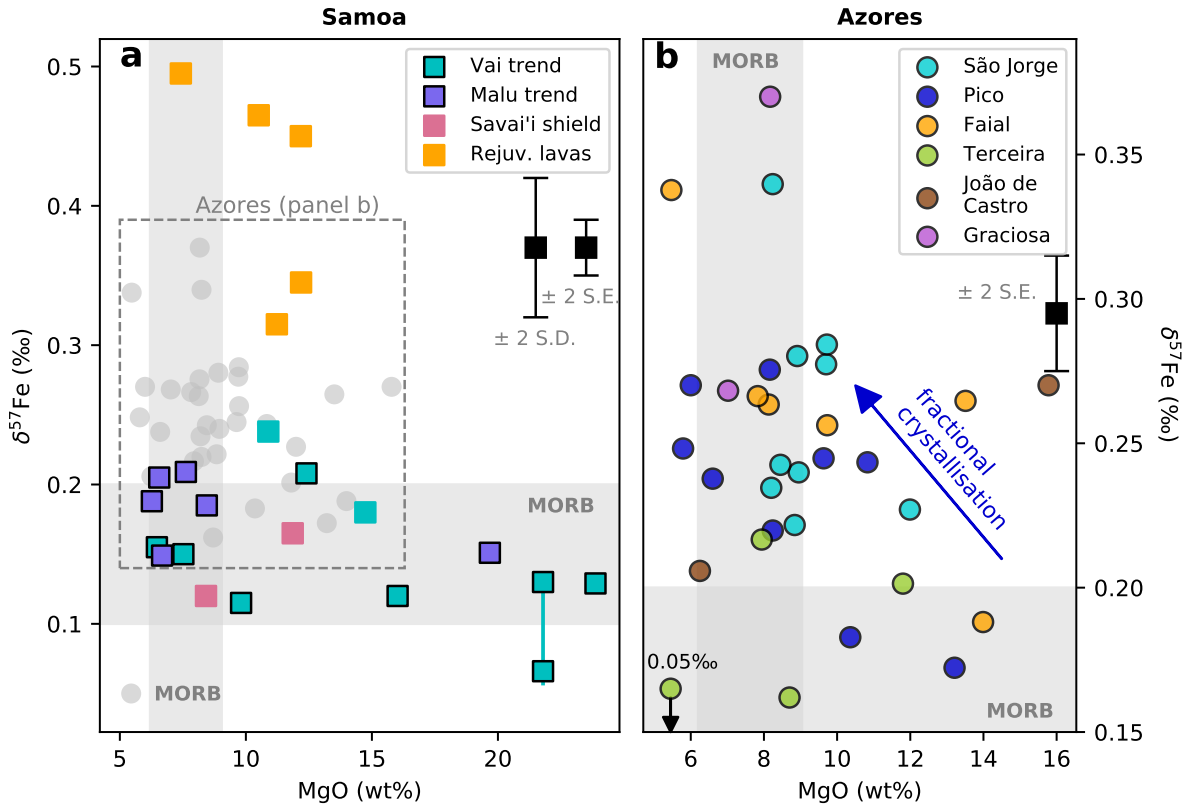


Figure 2: Measured Fe isotope data vs whole-rock MgO: black outlines, data from this study; no outline, data from Konter et al. (2016). At the same MgO content as MORB, the Azores lavas and Samoan rejuvenated lavas are relatively isotopically heavy. Samoan samples with MgO > 18 wt% that have accumulated olivine show no resolvable Fe isotopic difference from lower MgO samples. Some Azores volcanoes (Pico, Faial) show evidence for fractional crystallisation trends. Shaded region is average MORB ($\delta^{57}\text{Fe}$ from Teng et al., 2013; MgO from range of ‘ALL MORB’ from Gale et al., 2013). Errorbars show average long term 2 S.D. and average 2 S.E. on repeat measurements; duplicate dissolutions of the same sample linked by a vertical line.

284 The removal of light $\delta^{57}\text{Fe}$ mineral phases (e.g., olivine and pyroxene, both isotopically light
 285 relative to melt) by fractional crystallisation is an important control on driving the $\delta^{57}\text{Fe}$ of
 286 residual melts to heavier values (e.g., Teng et al., 2008). This isotopic fractionation is thought
 287 to be caused by the low Fe^{3+} content of olivine and pyroxene (Dauphas et al., 2014) and is
 288 consistent with the MgO– $\delta^{57}\text{Fe}$ trend seen in Pico lavas (Fig. 2b). Previous studies have
 289 shown that fractional crystallisation corrections can be large ($\Delta^{57}\text{Fe}$ up to -0.15‰ for Pitcairn
 290 samples; Nebel et al., 2019, with the biggest corrections for the lowest wt% MgO samples),
 291 therefore we perform the same correction to samples with < 18 wt% MgO in this dataset. For

292 each sample, we incrementally add equilibrium olivine back into our melt composition until the
293 liquid is in equilibrium with Fo₉₀ olivine. Given the MgO range of Samoa and Azores samples,
294 pyroxene likely crystallised from these melts (e.g., Beier et al., 2012). However, pyroxene–melt
295 fractionation is expected to be smaller than olivine–melt fractionation (Macris et al., 2015) due
296 to the differing Fe bonding environment in olivine and pyroxene, although the fractionation is
297 poorly constrained. Therefore, we only correct for olivine crystallisation, which provides an
298 upper limit on the fractional crystallisation correction.

299 The olivine–melt fractionation factor ($\Delta^{57}\text{Fe}_{\text{ol-melt}}$) generates significant uncertainty in the frac-
300 tional crystallisation correction used by Sossi et al. (2016) and Nebel et al. (2019). Fig. 3 shows
301 how the magnitude of the correction varies with $\Delta^{57}\text{Fe}_{\text{ol-melt}} = -0.1 \times 10^6/T^2$ (calculated us-
302 ing force constants following Sossi and O’Neill, 2017; see Appendix) to $\Delta^{57}\text{Fe}_{\text{ol-melt}} = -0.4 \times$
303 $10^6/T^2$ (following Nebel et al., 2019). Since we are interested in the heavy Fe isotope signatures,
304 we use the largest fractional crystallisation correction, even though this may be overcorrecting
305 the data. The correction therefore represents an upper limit on $\Delta^{57}\text{Fe}$ during fractional crys-
306 tallisation, which will be conservative from the perspective of diminishing as much heavy $\delta^{57}\text{Fe}$
307 enrichment that is observed as possible.

308 This calculation has been applied to all the MORB and OIB data shown in Fig. 4a. After
309 accounting for the processes of olivine accumulation and fractional crystallisation, the Samoan
310 and Azores datasets still show heavy Fe isotopic compositions relative to MORB (Figs. 3, 4),
311 which must then be a mantle source-derived signal.

312 **4.2 Heterogeneity derived from the mantle**

313 Variability in the Fe isotopic composition of primary mantle melts could be the result of partial
314 melting and/or heterogeneous mantle domains. Mantle heterogeneity is well-studied in the
315 Samoa, Pitcairn and Azores plumes, with different mantle components well-characterised in
316 radiogenic isotope space (Eisele et al., 2002; Jackson et al., 2014; Beier et al., 2018; Waters
317 et al., 2020; Fig. 1). Differences in mantle mineralogy associated with these components may
318 account for variations in Fe isotope composition (Williams and Bizimis, 2014; Konter et al.,
319 2016), as suggested for Pitcairn, where a negative Fe-Pb correlation is interpreted as mixing

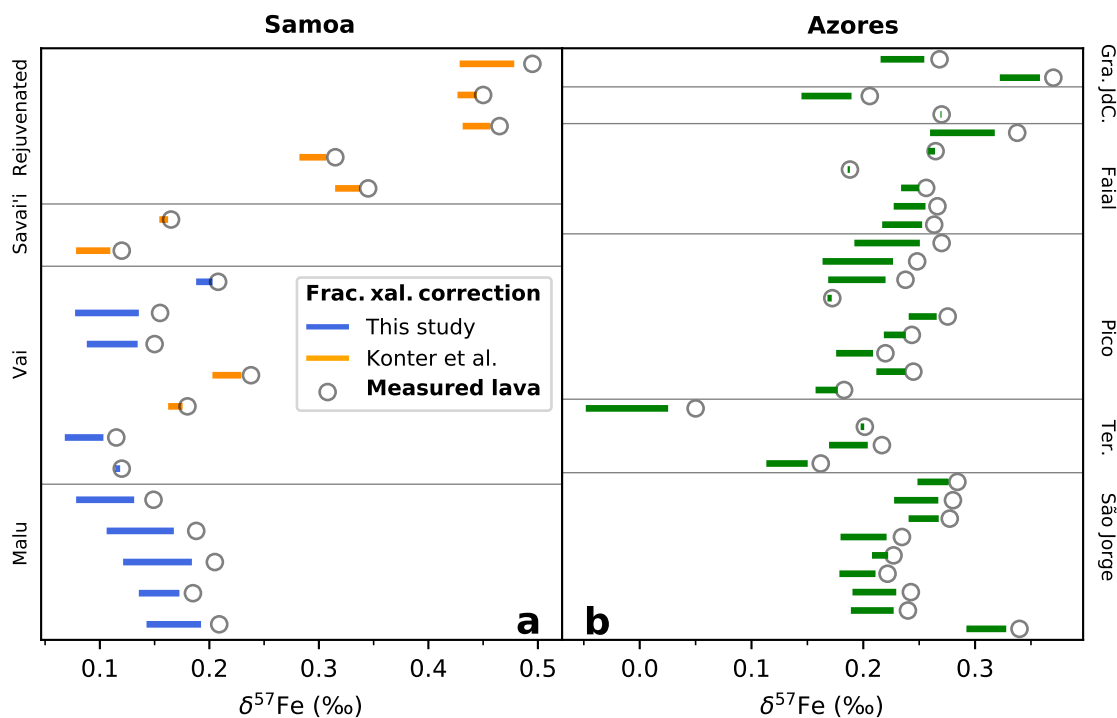


Figure 3: Fractional crystallisation correction following Nebel et al. (2019) and Sossi et al. (2016). The coloured bars show the magnitude of the correction for values of $\Delta^{57}\text{Fe}_{\text{ol-melt}} = -0.1 \times 10^6/T^2 - 0.4 \times 10^6/T^2$. Using the smaller correction gives corrected values that are unresolvable from the internal precision of the data collected in this study. Uncorrected data shown in outline for comparison. Samples with $\text{MgO} > 18 \text{ wt\%}$ were not corrected and are not shown. Abbreviations: ‘Gra.’: Graciosa; ‘JdC.’: João de Castro; ‘Ter.’: Terceira.

320 between a FOZO-like peridotite and an EM1, heavy $\delta^{57}\text{Fe}$ pyroxenite (Nebel et al., 2019; Fig.
 321 5). In Samoa, the distinct populations of shield and rejuvenated lavas also form a negative $\delta^{57}\text{Fe}$ -
 322 $^{206}\text{Pb}/^{204}\text{Pb}$ array (Fig. 5); the heavy $\delta^{57}\text{Fe}$ endmember is associated with the less radiogenic
 323 Pb component. Although the Azores data as a whole shows no Fe-Pb correlation, despite Pb
 324 isotopic variability, the island of São Jorge does display a statistically significant correlation
 325 distinguishable with external 2 S.D. precision on each sample. However, while we note that
 326 the São Jorge samples with the more radiogenic Pb isotopic signature do not show the heaviest
 327 $\delta^{57}\text{Fe}$ (Fig. 5), the spread of $\delta^{57}\text{Fe}$ in São Jorge is small, so a link between the Pb isotopes and
 328 high $\delta^{57}\text{Fe}$ may be largely masked by analytical precision in this case.

329 All three plumes produce melts that have heavier $\delta^{57}\text{Fe}$ than MORB, and show Pb isotopic
 330 variability reflecting different components in the mantle source. The likely dominant endmember
 331 lithologies responsible for mantle heterogeneity are peridotite (anhydrous or volatile-rich) and
 332 pyroxenite derived from recycled oceanic crust (Section 1). Therefore, we first examine whether
 333 melting of these endmember lithologies can generate the heavy $\delta^{57}\text{Fe}$ seen in Samoa, Pitcairn

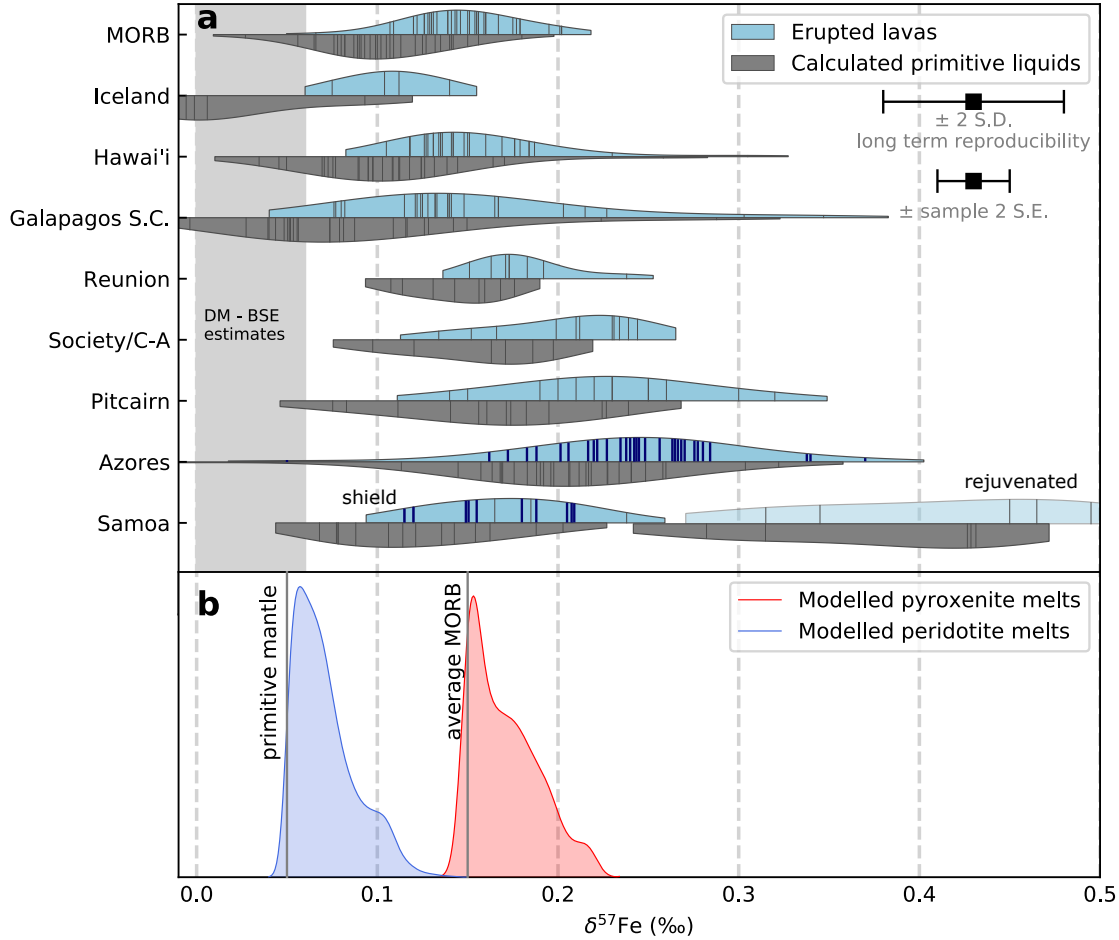


Figure 4: **a)** Compilation of kernel density plots of OIB and MORB Fe isotope data, with each vertical line a sample (bold lines are samples measured in this study). OIB data filtered to only include samples with 5–16 wt% MgO; top errorbar shows long term reproducibility 2 S.D. on $\delta^{57}\text{Fe}$ of standards run at the University of Cambridge. Samoan rejuvenated lavas record the heaviest $\delta^{57}\text{Fe}$ in the global dataset, whereas the Samoan shield dataset is indistinguishable from MORB. The Azores show the isotopically heaviest lavas after Samoan rejuvenated lavas. Only normal (N-MORB) and transitional (T-MORB) class MORB data is shown, following the classification by Teng et al., 2013. Data sources: MORB, Teng et al. (2013); Pitcairn, Nebel et al. (2019); Iceland, Schuessler et al. (2009); La Réunion, Peters et al. (2019); Society and Cook-Austral, Teng et al. (2013); Galapagos Spreading Centre, Gleeson et al. (2020); Hawai'i, Teng et al. (2008, 2013); Samoa, Konter et al. (2016), this study; Azores, this study. Fractional crystallisation correction shows liquids calculated to be in equilibrium with Fo₉₀ olivine, described in section 4.1. Shaded region shows range of estimates, including quoted errors, for depleted mantle, DM (Craddock et al., 2013; Johnson et al., 2020) and Bulk Silicate Earth, BSE (Sossi et al., 2016). **b)** Model results of compositions of melts of a peridotite (KLB1) and pyroxenite (G2) lithology, discussed in section 4.2.1. Primitive mantle estimate of $\delta^{57}\text{Fe} = 0.05\text{‰}$ (Sossi et al., 2016) is shown.

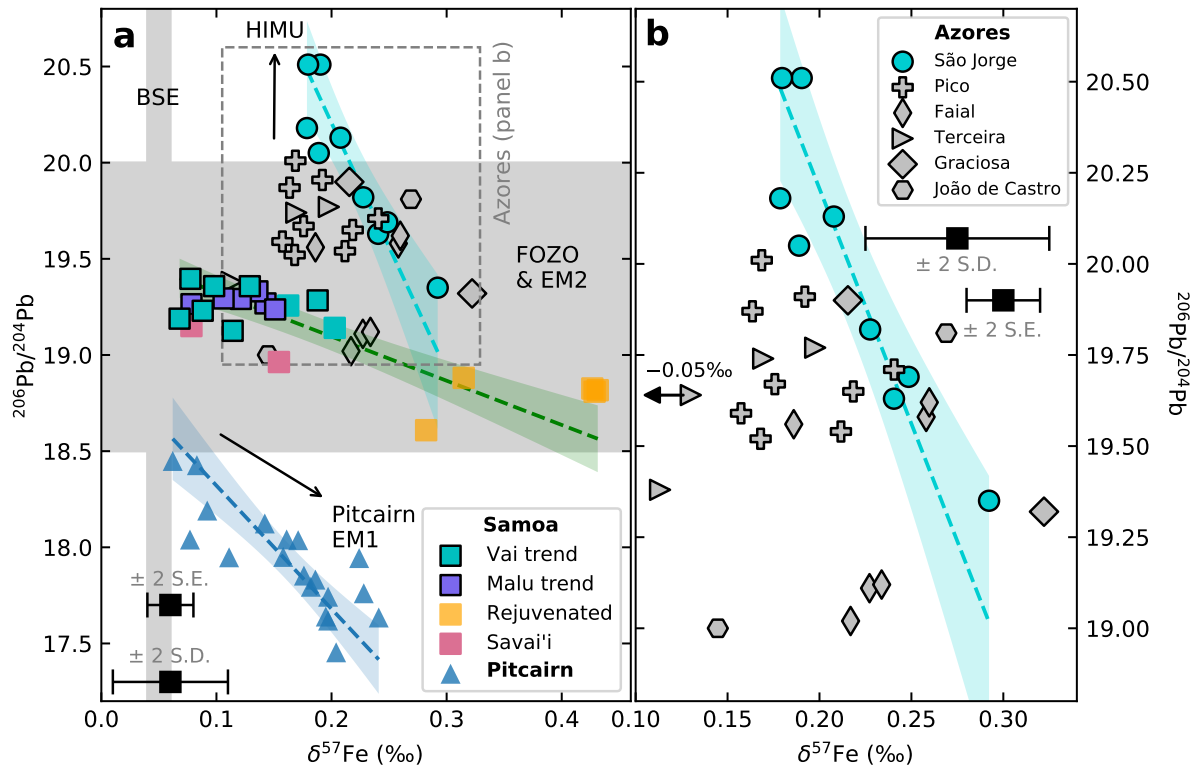


Figure 5: Lead and Fe isotopic variability in the Samoan, Azores and Pitcairn plumes. Data shown have been corrected for fractional crystallisation (Section 4.1). Black outlines indicate iron isotope measurement in this study; no outline indicates published iron isotope data (Samoa, Konter et al., 2016; Pitcairn, Nebel et al., 2019). The plumes show negative arrays with heavy $\delta^{57}\text{Fe}$ components also having less radiogenic Pb isotopic compositions within each array: orthogonal distance regression lines and 2σ confidence bands for Pitcairn, Samoa, and São Jorge are shown, calculated by weighting each data point with the sample 2 S.D. if known, or the long term 2 S.D. whichever is greater, as errors on $\delta^{57}\text{Fe}$.

335 4.2.1 Iron isotope fractionation by peridotite melting

336 First, we consider the contribution of anhydrous peridotite melting to the generation of heavy Fe
 337 isotopic signatures in OIB. To estimate the heaviest Fe isotopic composition of melts generated
 338 from single-stage melting of mantle peridotite, we calculated the modal mineralogy of KLB1
 339 peridotite (a commonly used experimental composition used as an analogue for the upper mantle,
 340 based on a Kilborne Hole spinel lherzolite xenolith; see Davis et al., 2009 and Table S2) over
 341 P-T space (5–40 kbar, 1100–1930°C) using the dataset of Holland et al. (2018), implemented in
 342 THERMOCALC (Powell et al., 1998). Taking the calculated modal mineralogies, the Fe isotope
 343 compositions of the phases present (including melts) were calculated using force constants (see
 344 Appendix) following Sossi and O’Neill (2017).

345 The KLB1 pseudosection, showing equilibrium phase assemblages, and melt $\delta^{57}\text{Fe}$ fractionation
 346 results (reported as $\Delta^{57}\text{Fe} = \delta^{57}\text{Fe}_{\text{melt}} - \delta^{57}\text{Fe}_{\text{bulk}}$) are shown in Fig. 6a & c. The maximum
 347 partial melting fractionation is small ($\Delta^{57}\text{Fe} = 0.07\text{‰}$), consistent with estimates from previous
 348 studies of $\Delta^{57}\text{Fe}$ (0.07–0.11‰; Dauphas et al., 2009; Williams and Bizimis, 2014; Sossi et al.,
 349 2016; Gleeson et al., 2020). Superimposing this melting fractionation onto the bulk BSE $\delta^{57}\text{Fe}$
 350 (0.05‰; Sossi et al., 2016) generates insufficiently heavy melts to explain many of the fractional-
 351 crystallisation-corrected OIB data. Some MORB also show heavier $\delta^{57}\text{Fe}$ than would be expected
 352 for melting of homogeneous peridotite.

353 *Role of carbonated peridotite melting:* Small-degree (carbonatite) melts from a carbonated peri-
 354 dotite mantle have been proposed to explain geochemical variability in Azores lavas (Beier
 355 et al., 2013), and carbonatite metasomatism has been proposed to explain trace element data
 356 in peridotite xenoliths from Savai'i (Hauri et al., 1993). It is important to consider the role
 357 of carbonatite melts in generating heavy $\delta^{57}\text{Fe}$ lavas because, although carbonatites in equi-
 358 librium with silicate mantle are predicted to have $\delta^{57}\text{Fe} < -0.45\text{‰}$, some measured intrusive
 359 carbonatites extend up to $\delta^{57}\text{Fe} = 1.2\text{‰}$ (Johnson et al., 2010). Some of these heavy isotopic
 360 compositions are attributed to the addition of a heavy $\delta^{57}\text{Fe}$ (Fe^{3+} rich) fluid at the edges
 361 of a carbonatite intrusion in the upper crust (generating compositions of $\delta^{57}\text{Fe} \lesssim 0.6\text{‰}$), and
 362 other disequilibrium processes including phenocryst mixing, although the complete process of
 363 generation of the measured heavy $\delta^{57}\text{Fe}$ in carbonatites relevant to OIB lavas is unclear.

364 Carbonated peridotite melts can be identified by their fractionation of trace and major element
 365 ratios, including Nb/La, Ti/Sm and K/La (Cottrell and Kelley, 2013; Beier et al., 2013) relative
 366 to anhydrous peridotite melts. However, the Azores samples in this study show no relationship
 367 between $\delta^{57}\text{Fe}$ and these ratios (Figs. S6, S7). Beier et al. (2013) suggest that the low Ti/Sm
 368 and K/La signature of carbonatite will be obscured by the contribution from peridotitic melts if
 369 the contribution of carbonatite in the final melt is $< 20\%$, or the anhydrous ambient peridotite
 370 undergoes $> 4\%$ partial melting. By mass balance, with (1) 20% carbonatite; (2) equal pro-
 371 portions of Fe in carbonated peridotite melts and anhydrous peridotite melts (Dasgupta et al.,
 372 2009); and (3) volatile-free peridotite $\delta^{57}\text{Fe} = 0.05\text{‰}$ (Sossi et al., 2016), producing an aver-
 373 age fractional-crystallisation-corrected $\delta^{57}\text{Fe}$ for Azores lavas of 0.20‰ requires $\delta^{57}\text{Fe}_{\text{carbonatite}}$
 374 $= 0.8\text{‰}$. If the carbonatite contribution is 10%, the required $\delta^{57}\text{Fe}_{\text{carbonatite}} = 1.55\text{‰}$. While

375 the lightest of these hypothetical carbonatite compositions are consistent with some measured
376 intrusive carbonatites, in general carbonatite lavas show significantly lighter $\delta^{57}\text{Fe}$ compositions
377 (Johnson et al., 2010). Therefore, it seems unlikely that carbonatite melts are contributing
378 to the heavy $\delta^{57}\text{Fe}$ in Azores lavas, despite the possible presence of a carbonated peridotite
379 lithology in their source region (Beier et al., 2013).

380 **4.2.2 Iron isotope fractionation by pyroxenite melting**

381 Given that pure peridotite melts cannot account for heavy $\delta^{57}\text{Fe}$ in Samoa, Pitcairn and the
382 Azores, we now consider the effect of using a pyroxenite lithology on $\Delta^{57}\text{Fe}$ during partial melt-
383 ing. We follow Lambart et al. (2016) in using the term pyroxenite to refer to pyroxene-rich rocks
384 with insufficient olivine ($< 40\%$) to be considered peridotites (Le Maitre et al., 2005). There
385 are a number of recorded natural pyroxenite compositions, but it is unclear which pyroxenite
386 types may best explain global Fe isotopic variability. Natural pyroxenites can be broadly sepa-
387 rated into two types: silica-deficient (SD) and silica-excess (SE); producing distinct partial melt
388 compositions (Kogiso et al., 2004; Lambart et al., 2016). Most previously studied pyroxenites
389 are SD type; those that are MgO-rich are thought to be cumulates produced by the fractional
390 crystallisation of peridotite partial melts (Gonzaga et al., 2010; Lambart et al., 2016). Following
391 Lambart et al. (2016), eclogites (from recycled oceanic crust or sediment) and pyroxenites pro-
392 duced from hybridisation between silica-rich eclogite melts and peridotite (Sobolev et al., 2005)
393 are SE type. Previous studies have proposed that the pyroxenitic component in the OIB mantle
394 source may be formed by this hybridisation process (e.g., Sobolev et al., 2005, 2007; Nebel et al.,
395 2019), therefore we consider the melting of SE eclogitic pyroxenite first. Lithospheric pyrox-
396 enites with a cumulate origin, possibly through multi-stage processes, are discussed in Section
397 5.4.

398 The G2 composition (Table S2), an important MORB-like bulk composition in melting experi-
399 ments and models from Pertermann and Hirschmann (2003*a,b*) and Lambart et al. (2016), is a
400 low MgO quartz eclogite and a SE pyroxenite endmember (Lambart et al., 2016). This eclogitic
401 pyroxenite composition should have a heavier bulk $\delta^{57}\text{Fe}$ composition than peridotite, at least
402 as heavy as mean MORB (Section 4.2.3). The G2 equilibrium phase assemblage was modelled
403 in the same way as for KLB1 (Fig. 6b) to give the partial melting $\delta^{57}\text{Fe}$ fractionation over P-T

404 space (Fig. 6d).

405 The maximum calculated partial melting fractionation for G2 is small ($\Delta^{57}\text{Fe} = 0.07\text{‰}$) and
406 similar compared to that of KLB1, although the P-T distribution of melt compositions is different
407 owing to the different melting behaviours of the two lithologies. Iron isotope fractionation during
408 partial melting of individual endmember lithologies alone cannot explain the range in $\delta^{57}\text{Fe}$
409 observed for OIB. Therefore, we consider the role of bulk $\delta^{57}\text{Fe}$ enriched mantle compositions
410 in producing the observed $\delta^{57}\text{Fe}$ of melts from a two lithology mantle.

411 **4.2.3 Effect of bulk $\delta^{57}\text{Fe}$ compositions of peridotite and pyroxenite**

412 The pyroxenite bulk $\delta^{57}\text{Fe}$ is not expected to be the same as that of peridotite (e.g., Williams and
413 Bizimis, 2014; Konter et al., 2016). For eclogitic G2 pyroxenite, which was chosen to be similar in
414 major element chemistry to average modern oceanic crust (Pertermann and Hirschmann, 2003b),
415 our initial assumption of starting Fe isotopic composition is that of average MORB ($\delta^{57}\text{Fe} =$
416 0.15‰ ; Teng et al., 2013; Williams and Bizimis, 2014), although the effect of subduction and
417 eclogite formation during high P, low T metamorphism is poorly understood (Section 5.2.1).

418 The model results (Fig. 4) show that, based on this assumed G2 starting composition and
419 single-stage melting of each lithology, mixed peridotite and eclogitic pyroxenite melts span the
420 Fe isotope range of Samoan shield and some Azores melts. However, the heaviest of the Samoan
421 and Azores melts ($\delta^{57}\text{Fe} > 0.2\text{‰}$) require that they are almost pure pyroxenite melts. Although
422 melts derived from a dominantly or exclusively pyroxenitic source have been proposed for Hawai'i
423 based on olivine compositions (Sobolev et al., 2005; Herzberg, 2011), a pure pyroxenite source is
424 generally incompatible with other global OIB (e.g., Dasgupta et al., 2010; Herzberg, 2011) and
425 is not supported by the major element compositions of the measured Samoan and Azores lavas
426 relative to the model liquids (Fig. 7). The heaviest modelled pyroxenite melts are produced
427 at low-degrees of melting, 13–14 kbar (~ 50 km depth) and $\sim 1200^\circ\text{C}$ (Fig. 6c & d). Plume
428 geotherms are unlikely to intersect this part of the melting region, e.g., Azores melts are thought
429 to form at $> 1400^\circ\text{C}$ and > 25 kbar (Beier et al., 2012); Samoa melts at $> 1600^\circ\text{C}$ and > 30
430 kbar (Putirka et al., 2018). Moreover, we expect mixing between pyroxenite and peridotite melts
431 upon ascent in the mantle, even if pyroxenite melts first (Lambart et al., 2016). Therefore, pure,

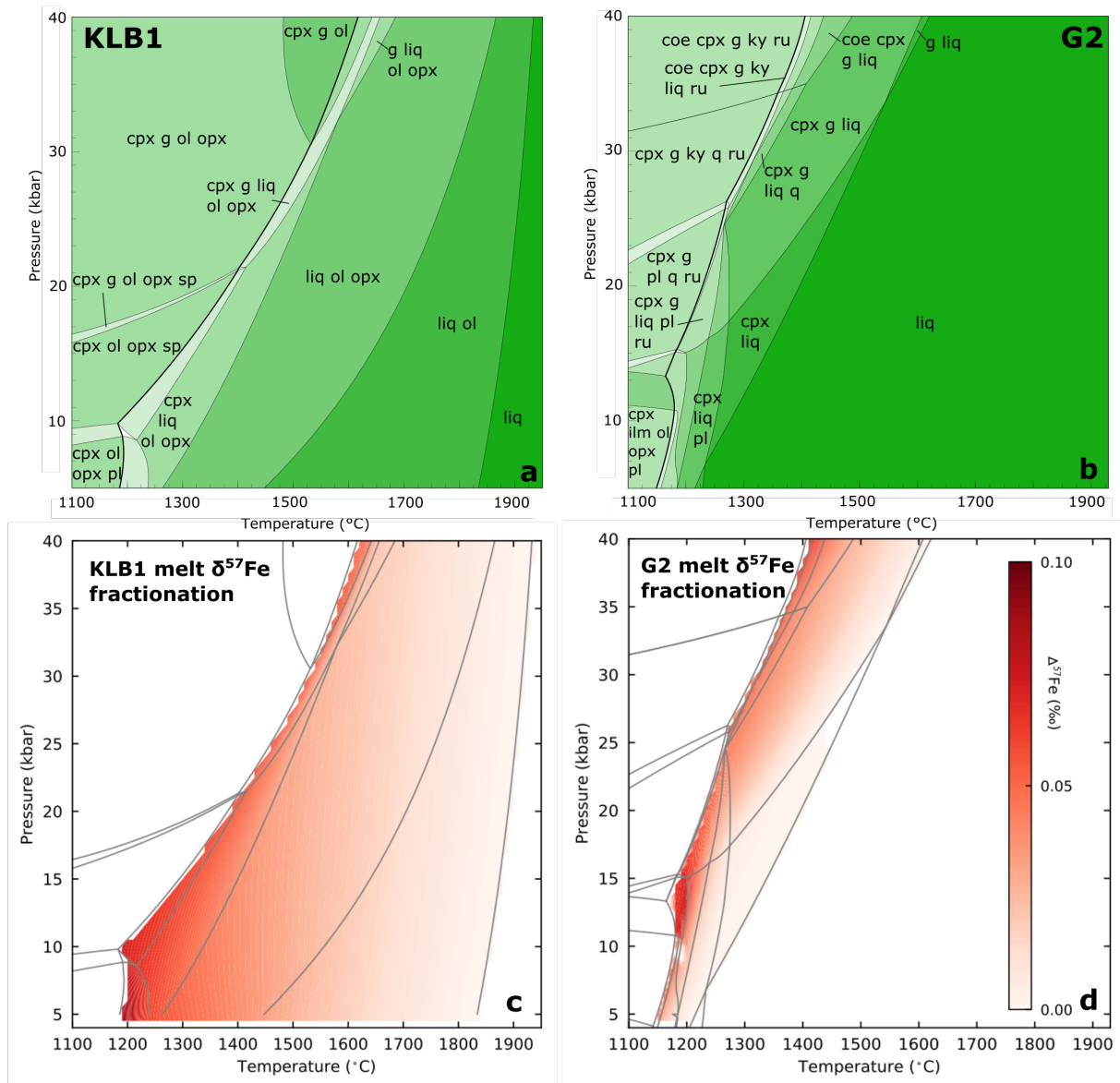


Figure 6: Results of the combined phase equilibria and isotope fractionation model for KLB1 and G2 lithologies. a, b show the stable mineral phases calculated using THERMOCALC, with the solidus highlighted in bold. c, d show the calculated partial melting fractionation from the pseudosections in a and b. The magnitude of the maximum partial melting Fe isotopic fractionation, $\Delta^{57}\text{Fe}$, is small and independent of the bulk composition used.

432 heavy- $\delta^{57}\text{Fe}$ eclogitic pyroxenite melts are unlikely.

433 We also note that a contribution to a peridotite-dominated melt from an eclogitic pyroxenite
434 melt could explain why average MORB, after correction for fractional crystallisation, sits at the
435 heavier end of predicted KLB1 melt $\delta^{57}\text{Fe}$ compositions. An enriched, fertile lithology has been
436 proposed previously to be a small, ubiquitously distributed component of the MORB source,
437 to balance depleted components seen in abyssal peridotites (Salters and Dick, 2002; Byerly and
438 Lassiter, 2014) and the garnet signature seen in trace elements in MORB (Hirschmann and
439 Stolper, 1996). The Fe isotope data would be consistent with this enriched component being a
440 refertilised peridotite (i.e., still an olivine-bearing lithology, but with more clinopyroxene than
441 depleted mantle and a slightly elevated bulk $\delta^{57}\text{Fe}$), or even a pyroxenite to match the heaviest
442 MORB $\delta^{57}\text{Fe}$ compositions. Sun et al. (2020) record MORB glass samples from EPR seamounts
443 with $\delta^{56}\text{Fe} \leq 0.36\text{‰}$ ($\delta^{57}\text{Fe} \leq 0.54\text{‰}$), where the high Fe isotope variability relative to average
444 MORB is suggested to relate to limited melt mixing with ambient mantle in the seamounts
445 compared to the ridge axis. The heavy $\delta^{57}\text{Fe}$ compositions are attributed to garnet pyroxenite
446 veins in recycled, metasomatised oceanic lithosphere, processes which are discussed further in
447 Section 5.4.

448 The modelled major element chemistry could be made more consistent with the Samoa and
449 Azores lavas by changing the type of pyroxenite: SD pyroxenites, such as MIX1G (a composi-
450 tion resembling average natural SD pyroxenites, produced by mixing MORB with peridotite),
451 produce melts with Al_2O_3 that overlap more OIB compositions than G2 (Hirschmann et al.,
452 2003; Kogiso et al., 2003). In agreement with this, mixing the modelled KLB1 and G2 melts in
453 Fig. 7 could produce melts that cover more of the OIB lava compositions than each individual
454 lithology. However, pyroxenites generated by mixing MORB with peridotite would produce a
455 bulk $\delta^{57}\text{Fe}$ lighter than G2 which, given the same $\Delta^{57}\text{Fe}_{\text{melt-solid}}$, would not be able to generate
456 heavy $\delta^{57}\text{Fe}$ in OIB.

457 The isotopically heaviest Samoan shield and Azores melts, and all rejuvenated Samoan melts,
458 are isotopically heavier than those produced from plausible conditions of a single-stage mixed
459 peridotite–pyroxenite melting model.

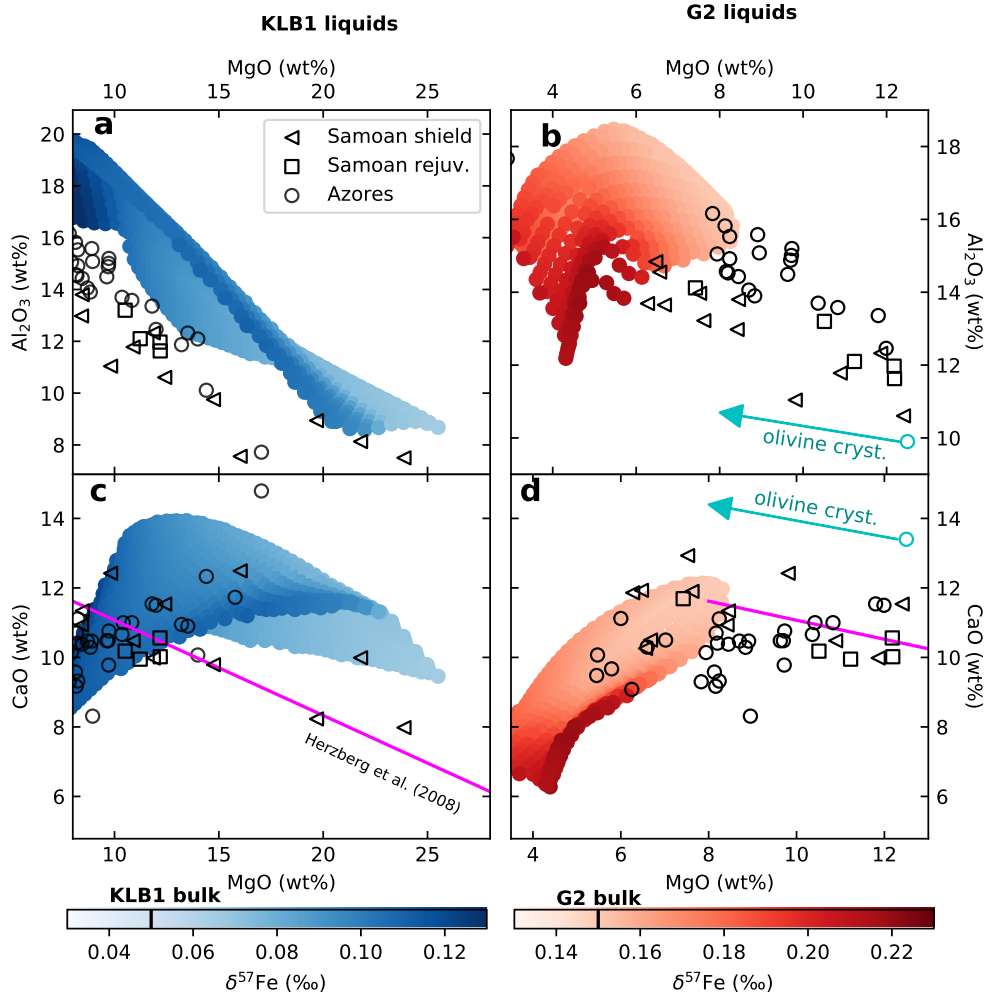


Figure 7: Coloured circles show the compositions of KLB1 (left) and G2 (right) melts produced in the model, coloured for $\delta^{57}\text{Fe}$ (KLB1 melts filtered for melt fraction $< 30\%$ to give a realistic MgO range). Samoa and Azores lavas are superimposed. Although pure pyroxenite melts are implied by the Fe isotopic composition, observed lava compositions are not consistent with pure pyroxenite melts. Pink line shows the separation of accumulated fractional melts from pyroxenite and peridotite source melts in CaO-MgO space from Herzberg and Asimow (2008) — peridotite source melts mostly plot above the line, pyroxenite melts mostly below. An example vector of melts related by olivine fractional crystallisation is shown.

460 5 How to generate heavy $\delta^{57}\text{Fe}$ mantle components?

461 The inability of a simple peridotite–eclogitic pyroxenite melting model (Section 4.2.3) to generate
 462 the heavy $\delta^{57}\text{Fe}$ of Samoa, Pitcairn and Azores melts means that pre-final melt emplacement
 463 processes (i.e., processes affecting the source $\delta^{57}\text{Fe}$ composition) must be considered. After
 464 core contributions, we sequentially consider processes operating from a mid-ocean ridge setting,
 465 through subduction, to upwelling and melting in a mantle plume (summarised in Fig. 8).

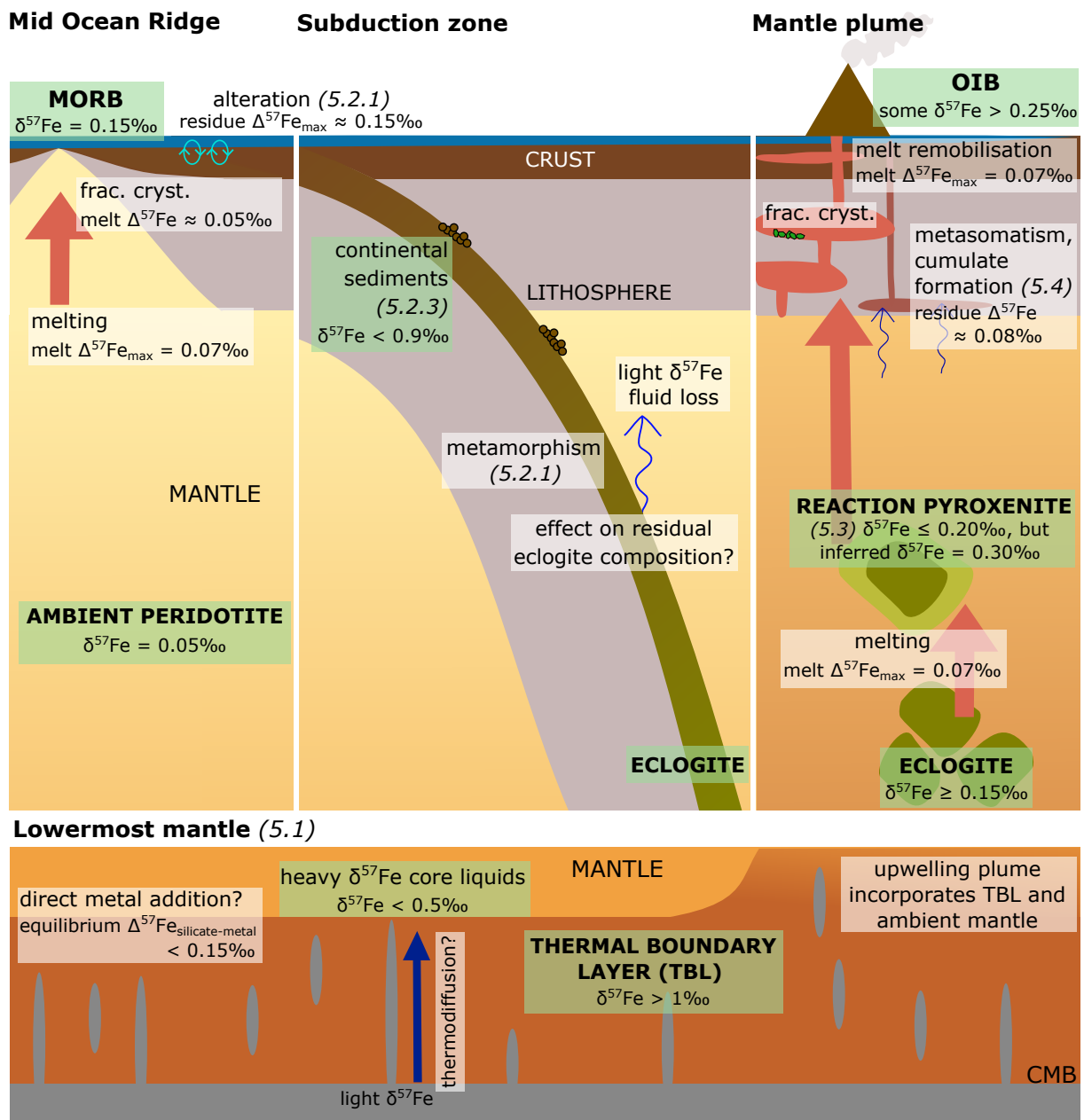


Figure 8: Schematic showing processes that may contribute to OIB liquids with heavier $\delta^{57}\text{Fe}$ than those predicted by single-stage melting of eclogite (Section 4.2.3). Processes including hydrothermal alteration (El Korh et al., 2017) and metamorphism (Debret et al., 2016) may increase subducted eclogite $\delta^{57}\text{Fe}$ ($> 0.15\text{‰}$), and the addition of continental sediments could contribute to recycled crustal domains with $\delta^{57}\text{Fe} > 0.15\text{‰}$. Reaction pyroxenites have inferred $\delta^{57}\text{Fe} \approx 0.30\text{‰}$ from erupted lavas (Nebel et al., 2019; Gleeson et al., 2020), however generation of this component requires eclogite to have $\delta^{57}\text{Fe}$ significantly greater than 0.15‰ (Section 5.3). Metasomatism (Konter et al., 2016), fractional crystallisation and secondary melting may further contribute to heavy $\delta^{57}\text{Fe}$ ($> 0.25\text{‰}$) liquids measured in OIB. Contributions from core liquids have also been proposed to generate heavy $\delta^{57}\text{Fe}$ in OIB (Leshner et al., 2020). Numbers in italics refer to the sections each process is discussed in.

466 5.1 Is heavy $\delta^{57}\text{Fe}$ a core component?

467 Many OIB are suggested to entrain lower mantle material. Short-lived radiogenic isotope systems
 468 and noble gases support that some OIB have incorporated small amounts of primordial, less-

469 degassed mantle material from the first 0.5 Ga of Earth's history, and/or core-equilibrated liq-
 470 uids, possibly stored near the CMB (e.g., Mukhopadhyay, 2012; Mundl et al., 2017; Peters et al.,
 471 2018; Rizo et al., 2019; Mundl-Petermeier et al., 2019, 2020). Calculated equilibrium silicate-
 472 metal fractionation factors at high pressures and temperatures are small ($\Delta^{57}\text{Fe}_{\text{silicate-metal}} \lesssim$
 473 0.15% ; Polyakov, 2009; Shahar et al., 2016; Liu et al., 2017). Given that W and He isotope
 474 data support $< 0.3\%$ core-equilibrated material in OIB (Mundl-Petermeier et al., 2020), this
 475 small equilibrium fractionation is unlikely to contribute to heavy $\delta^{57}\text{Fe}$ in OIB. However, it
 476 has been proposed (Leshner et al., 2020) that heavy $\delta^{57}\text{Fe}$ in OIB could be generated by the
 477 incorporation of a kinetically-fractionated heavy $\delta^{57}\text{Fe}$ liquid iron and silicate layer from the
 478 lowermost mantle into an upwelling plume (Fig. 9). In this model, isotopic fractionation occurs
 479 by thermodiffusion (Soret diffusion), due to the relative motion of heavy and light isotopes in a
 480 liquid in response to a temperature gradient. In the case of liquid iron infiltrating the lowermost
 481 mantle near the CMB, heavy $\delta^{57}\text{Fe}$ migrates towards lower temperatures further from the CMB,
 482 forming a thin, heavy $\delta^{57}\text{Fe}$ ($< 0.5\%$, average $\approx 0.1\%$) boundary layer just above the CMB,
 483 which can be entrained by a rising mantle plume. As the plume rises, the core liquids freeze to
 484 preserve the isotopic anomaly, and the resultant plume is a mixture of this core component and
 485 ambient silicate mantle. This process produces a plume with average $\delta^{57}\text{Fe} > 0.05\%$ heavier
 486 than chondritic mantle depending on the core contribution (Leshner et al., 2020). The proposed
 487 average $\delta^{57}\text{Fe}$ offset of the plume relative to ambient mantle could explain part of the range of
 488 heavy $\delta^{57}\text{Fe}$ in OIB relative to MORB, so the process is discussed below, but the addition of the
 489 core liquids modelled by Leshner et al. (2020) still cannot explain the heaviest ($\delta^{57}\text{Fe} \gtrsim 0.25\%$)
 490 OIB lavas.

491 Heavy $\delta^{57}\text{Fe}$ in OIB generated from the core-contribution model should be linked to other
 492 geochemical signatures of core material: specifically, low $^{182}\text{W}/^{184}\text{W}$ (negative $\mu^{182}\text{W}$) and high
 493 $^3\text{He}/^4\text{He}$ (Leshner et al., 2020), as well as high $^{186,187}\text{Os}/^{188}\text{Os}$, strong HSE enrichment (Brandon
 494 and Walker, 2005), and possibly high Fe/Mn (Humayun et al., 2004; Rizo et al., 2019). However,
 495 HSE enrichment in OIB that show negative $\mu^{182}\text{W}$ signatures is generally not observed (Mundl
 496 et al., 2017; Mundl-Petermeier et al., 2019), which could be attributed to the presence of a core-
 497 equilibrated reservoir rather than direct incorporation of core material providing the $\mu^{182}\text{W}$
 498 signatures (Mundl-Petermeier et al., 2020).

499 The global OIB $\delta^{57}\text{Fe}$ dataset does not show a relationship between heavy $\delta^{57}\text{Fe}$ and volcanoes
 500 or plumes that show negative $\mu^{182}\text{W}$ and high $^3\text{He}/^4\text{He}$, or high Fe/Mn (Fig. S5). The Samoan
 501 volcano of Ofu records $^3\text{He}/^4\text{He} \leq 33.8 \text{ R/R}_a$ (Jackson et al., 2007a) and negative $\mu^{182}\text{W}$ (\geq
 502 -17.3 ; Mundl et al., 2017; Mundl-Petermeier et al., 2020). However, the same Ofu samples
 503 measured here have $\delta^{57}\text{Fe}$ indistinguishable from MORB. This is inconsistent with the presence
 504 of frozen core liquid contributing to isotopically heavy $\delta^{57}\text{Fe}$ melts in the upwelling plume. There
 505 is a small negative $\mu^{182}\text{W}$ anomaly (and slightly raised Fe/Mn) recorded in the Azores plume
 506 as a whole (Rizo et al., 2019; Mundl-Petermeier et al., 2020), but Fe/Mn ratios for Pico (with
 507 a negative $\mu^{182}\text{W}$; Rizo et al., 2019) show the least evidence of high-Fe in the Azores mantle,
 508 being similar to Icelandic lavas (Humayun et al., 2004; Qin and Humayun, 2008). The highest
 509 Fe/Mn ratio were measured in lavas from São Jorge (and São Miguel) that have not all been
 510 analysed for Fe or W isotopes (see Appendix, Table S4), but where samples have both Fe/Mn
 511 and $\delta^{57}\text{Fe}$ data there is no relationship between the two (Fig. S5). Other high $^3\text{He}/^4\text{He}$ lavas
 512 (e.g., Baffin) also show no heavy $\delta^{57}\text{Fe}$ anomaly relative to average MORB (McCoy-West et al.,
 513 2018), although this could be related to the high mantle potential temperatures associated with
 514 the Icelandic plume and plume head (Matthews et al., 2016; Spice et al., 2016) and therefore high
 515 degrees of melting of ambient peridotite. Overall, current data do not support a link between
 516 heavy $\delta^{57}\text{Fe}$ and the presence of core liquids entrained in an upwelling plume, and we instead
 517 consider modifications to the model of subducted, recycled material.

518 5.2 The Fe isotopic composition of subducted material

519 Pyroxenite melts that are isotopically heavier than the model melts shown in Fig. 4b, and
 520 therefore more consistent with measured OIB melts, could be generated if subducted material
 521 had $\delta^{57}\text{Fe} > 0.15\%$. This heavy $\delta^{57}\text{Fe}$ could be caused by: (i) alteration of oceanic crust either
 522 hydrothermally at the seafloor, or during devolatilisation and metamorphism associated with
 523 subduction; (ii) ancient MORB being inherently isotopically heavier than modern MORB; or
 524 (iii) contributions from isotopically heavy subducted sediments.

525 5.2.1 Hydrothermal alteration and subduction processes

526 Studies of subduction zone metabasites suggest that hydrothermal alteration of basalt at shallow
527 crustal levels could make subducted basalt isotopically heavier than MORB. For example, in Île
528 de Groix, France (a HP-LT metamorphic belt), hydrothermal alteration makes subducted basalt
529 isotopically heavier by $\Delta^{56}\text{Fe} = 0.06\text{--}0.10\text{‰}$ (equivalent to $\Delta^{57}\text{Fe} = 0.09\text{--}0.15\text{‰}$), contributing
530 to the heavy Fe compositions of eclogites ($\delta^{57}\text{Fe} = 0.25\text{--}0.28\text{‰}$; El Korh et al., 2017). Taking
531 the maximum amount of Fe isotope variability attributed to hydrothermal alteration by El Korh
532 et al. (2017), partial melts of hydrothermally altered crust could be as heavy as $\delta^{57}\text{Fe} = 0.37\text{‰}$.
533 Melting of this material, if present in mantle plumes, could explain the entire range of Samoan
534 shield and Azores melts, and some of the rejuvenated Samoan liquids. Further geochemical
535 studies, such as oxygen isotopes (Day et al., 2010), could help test the role of crustal alteration
536 on heavy $\delta^{57}\text{Fe}$. Alteration may also introduce sample heterogeneity, which could explain the
537 different $\delta^{57}\text{Fe}$ found between this study and Konter et al. (2016) for sample ALIA115-18, and
538 highlights that samples chosen for Fe isotope analyses should be carefully picked (due to this
539 heterogeneity, the sample is not included in Figs. 2, 3, 4 & 5). However, Rouxel et al. (2003)
540 showed that on a scale larger than a few metres, bulk hydrothermally altered oceanic crust is
541 indistinguishable in $\delta^{57}\text{Fe}$ from MORB.

542 Therefore, the ability of hydrothermal alteration to generate large scale heavy $\delta^{57}\text{Fe}$ recycled
543 crustal domains in the mantle remains uncertain, but is likely small. Any contribution from
544 hydrothermally altered MORB protolith to OIB melts may not be distinguishable in major
545 element compositions, as large major element variability in altered basalts is only recorded
546 where a large proportion of alteration (e.g., $> 30\%$ Fe^{2+} loss) has taken place (Rouxel et al.,
547 2003).

548 Isotopically heavy mantle components could also be generated by metamorphic and/or de-
549 volatilisation reactions, driving metabasites to heavier $\delta^{57}\text{Fe}$ values than their protolith. For
550 example, Île de Groix blueschists have heavy $\delta^{57}\text{Fe}$ ($0.26\text{--}0.51\text{‰}$; El Korh et al., 2017), which
551 could be partly explained by the loss of isotopically light fluids during prograde metamor-
552 phism: progressive serpentinite devolatilisation in the Western Alps records an increase of
553 $\Delta^{56}\text{Fe}_{\text{blueschist-serpentinite}} = 0.12\text{‰}$ ($\Delta^{57}\text{Fe} = 0.18\text{‰}$) from abyssal serpentinites to blueschists

554 (Debret et al., 2016). Increasing $\delta^{57}\text{Fe}$ with devolatilisation is ascribed to loss of Fe(II)-Cl or
555 Fe(II)-S complexes in serpentinite-derived fluids (Debret et al., 2016), which is consistent with
556 light $\delta^{57}\text{Fe}$ in secondary olivines precipitated from serpentinite fluids (Debret et al., 2018). In-
557 filtration by isotopically light slab-derived fluids have also been proposed to explain light ($\delta^{57}\text{Fe}$
558 $< 0\%$) peridotite fragments observed above the Philippine arc (Turner et al., 2018).

559 However, the origin of heavy $\delta^{57}\text{Fe}$ in Île de Groix blueschists is unclear, with El Korh et al.
560 (2017) suggesting that much of the variability is likely to have been inherited from the original
561 protolith, rather than generated during subduction. Studies have also shown that Fe is not
562 fractionated in the downgoing oceanic crust during prograde metamorphism: orogenic eclogites
563 preserve a MORB-like Fe isotope composition ($\delta^{57}\text{Fe} = 0.15\%$; Williams et al., 2009; Li et al.,
564 2016); Western Alpine eclogites also record no Fe isotope fractionation from MORB $\delta^{57}\text{Fe}$, de-
565 spite evidence from associated metagabbros of infiltration by a light $\delta^{57}\text{Fe}$ fluid (Inglis et al.,
566 2017). Preserved eclogite MORB-like $\delta^{57}\text{Fe}$ could be the result of mass balance between the
567 slab and fluid, leaving no resolvable Fe isotope heterogeneity after fluid loss in the subducted,
568 dehydrated eclogitic material. The role of the loss of slab-derived fluids on the Fe isotopic
569 composition of subducted material is further complicated by sources of fluids other than ser-
570 pentinites, such as fluids derived from the altered oceanic crust, which could have heavy $\delta^{57}\text{Fe}$
571 as heavy $\delta^{57}\text{Fe}$ hydrous minerals (e.g., amphibole, epidote) break down (Huang et al., 2020).

572 **5.2.2 The composition of ancient MORB**

573 While the Fe isotopic composition of modern oceanic crust is well-studied, the material that
574 now contributes to mantle heterogeneity may have been subducted more than a billion years
575 ago (Montanini and Tribuzio, 2015). If ancient oceanic crust was isotopically heavier than the
576 modern equivalent, it may have contributed to the formation of heavy $\delta^{57}\text{Fe}$ mantle domains
577 sampled by OIB. Factors introducing temporal variability in the bulk isotopic composition of
578 oceanic crust could be: (i) isotopic fractionation during partial melting and fractional crystalli-
579 sation; and (ii) the amount and/or isotopic effect of hydrothermal alteration.

580 *Magnitude of isotopic fractionations:* Isotopic fractionation during ancient mantle partial melt-
581 ing, and/or fractional crystallisation of the subsequent melts, would need to be greater than

582 modern values to generate oceanic crust with $\delta^{57}\text{Fe} > 0.15\text{‰}$ from a silicate mantle with $\delta^{57}\text{Fe}$
583 similar to the present-day value. However, hotter average upper mantle temperatures in the
584 past (Herzberg et al., 2010) argue against larger $\Delta^{57}\text{Fe}$, both because equilibrium isotopic frac-
585 tionation factors decrease with temperature (Urey, 1947) and partial melting fractionation is
586 smaller at higher degrees of melting (Fig. 6). This relationship is consistent with komatiites,
587 generally ascribed to high melting temperatures, recording an average $\delta^{57}\text{Fe}$ more similar to
588 BSE than average terrestrial basalts (Hibbert et al., 2012; Johnson et al., 2020). Mantle $f\text{O}_2$
589 may also affect the composition of mantle melts due to varying $\text{Fe}^{3+}/\text{Fe}_T$ over time (Williams
590 et al., 2004), hence the crustal $\delta^{57}\text{Fe}$ composition. However, Hibbert et al. (2012) records no
591 difference in $\delta^{57}\text{Fe}$ from komatiites from the late Archean to the Tertiary associated with a
592 mantle $f\text{O}_2$ effect, which is in agreement with other work suggesting that the mantle oxidation
593 state has remained approximately fixed since the Archean (e.g., Berry et al., 2008).

594 *Ancient hydrothermal alteration:* Different processes operating in ancient hydrothermal systems
595 compared to modern MORB could have contributed to $\delta^{57}\text{Fe}$ variability in subducted crust.
596 Whereas modern hydrothermal systems generally precipitate more light $\delta^{57}\text{Fe}$ sulphides (e.g.,
597 Rouxel et al., 2004) than heavy $\delta^{57}\text{Fe}$ oxides, Precambrian hydrothermal systems precipitated
598 fewer sulphides due to higher Fe/S and Fe^{2+} content in the fluids (Kump and Seyfried Jr, 2005).
599 Therefore, Precambrian vent fluids may have precipitated dominantly heavy $\delta^{57}\text{Fe}$ minerals in
600 the crust, and had lighter $\delta^{57}\text{Fe}$ than the present-day fluids (Johnson et al., 2020). However, even
601 for modern hydrothermal systems, their influence on bulk altered crust $\delta^{57}\text{Fe}$ remains uncertain
602 (Section 5.2.1), and so it is not possible to discern whether or not ancient hydrothermal systems
603 could have generated heavier $\delta^{57}\text{Fe}$ in altered MORB than today.

604 **5.2.3 Contributions from sediments**

605 Subducted slab material includes oceanic lithosphere, crust, and sediments. Downgoing sedi-
606 ments are suggested to contribute to mantle heterogeneity (White and Hofmann, 1982; Plank
607 and Langmuir, 1998), with small amounts of continental sediments traced in enriched mantle
608 reservoirs ($\leq 10\%$; e.g., Jackson et al., 2007b; Rapp et al., 2008). Continental sediments could
609 contribute to heavy $\delta^{57}\text{Fe}$ mantle domains, since differentiated igneous rocks record variable
610 $\delta^{57}\text{Fe}$ ($\leq 0.9\text{‰}$; Du et al., 2017). However, bulk upper continental crust (the main component of

611 recycled sediment in the mantle; Stracke, 2012), is thought to be isotopically indistinguishable
612 from MORB (Johnson et al., 2020), and direct measurements of arc system sediments show
613 $\delta^{57}\text{Fe} = 0.17\text{‰}$ (Nebel et al., 2015). Moreover, large-scale sediment input into the mantle is
614 not expected because most sediment is scraped off in the forearc and never subducted; much of
615 the sediment that does subduct may melt early and be returned to the surface in arc volcanism
616 (Jackson et al., 2007b).

617 The Vai (weak HIMU) and Malu (EM2; Jackson et al., 2014) trends in Samoa, as distinguished
618 in radiogenic isotope space (Fig. 1), are proposed to reflect the sampling of different mantle
619 heterogeneities, including sediments. These heterogeneities may originate from different recy-
620 cled material, explaining their radiogenic isotopic differences, hence might be expected to show
621 $\delta^{57}\text{Fe}$ variability related to the recycled material. However, the melts of any recycled mantle
622 component (e.g., sediments, oceanic crust) are likely to be mixed with ambient peridotite melts
623 during shield stage volcanism. Since the contribution of Fe from non-peridotitic components will
624 not dominate the Fe content of melts (unlike incompatible radiogenic isotopes), mass balance
625 would argue against small fractions of non-peridotitic components in the mantle source being
626 resolved using $\delta^{57}\text{Fe}$ where melt fraction is sufficiently high for peridotite melts to dominate.
627 The Fe isotopic composition of melts is dependent on both mantle processes and source lithology
628 and, in this case, processing rather than source composition may dominate melt $\delta^{57}\text{Fe}$ — small
629 variations in the $\delta^{57}\text{Fe}$ of isotopically heavy melts from dilute HIMU and EM2 components may
630 be eliminated by mixing with isotopically light peridotite melts.

631 **5.3 Generating pyroxenites by peridotite hybridisation**

632 In existing studies, the proposed formation of a solid pyroxenite component in the mantle that
633 contributes to melts in OIB (e.g., the Pitcairn EM1 mantle component; Nebel et al., 2019) is a
634 two-stage process, rather than only eclogite subduction and recycling. Reaction-zone pyroxenite
635 is proposed to form by the reaction of high pressure (> 30 kbar) Si-rich melts from recycled
636 oceanic crust with ambient peridotite, producing an olivine-free hybrid pyroxenite lithology
637 (Sobolev et al., 2005, 2007; Rosenthal et al., 2014; Nebel et al., 2019; Fig. 9), which is suggested
638 to have $\delta^{57}\text{Fe} = 0.30\text{‰}$ based on measured OIB data (Nebel et al., 2019; Gleeson et al., 2020).
639 Here we examine whether our modelled eclogite melts can generate this heavy pyroxenite mantle

640 component.

641 Based on a single-stage melting process (Section 4.2.1), the heaviest melts that can be formed
642 from subducted oceanic crust (with bulk $\delta^{57}\text{Fe} = 0.15\text{‰}$) have $\delta^{57}\text{Fe} = 0.22\text{‰}$. Mixing of
643 these melts with peridotite ($\delta^{57}\text{Fe} = 0.05\text{‰}$) would produce a lithology of $\delta^{57}\text{Fe} = 0.05\text{--}0.22\text{‰}$,
644 assuming equilibrium fractionation and depending on the proportion of eclogite melt involved.
645 Remelting of this component could then produce liquids heavy enough to explain the isotopic
646 compositions of Samoan shield and some of the Azores melts, but would require: (i) the solid
647 reaction-zone pyroxenite to have an isotopic composition close to that of the Si-rich eclogite
648 melt (maximum $\delta^{57}\text{Fe} = 0.22\text{‰}$), and (ii) that pyroxenite melt (with $\delta^{57}\text{Fe}$ up to 0.29‰ , taking
649 the upper limit of melting fractionation from Section 4.2.1) was extracted without mixing with
650 ambient peridotite melts (Fig. 9). It is also likely that the subsequent melts would have major
651 element compositions similar to G2 melts and therefore not match the OIB major element
652 geochemistry. Overall, this process cannot generate a sufficiently isotopically heavy solid mantle
653 component to match previous estimates of pyroxenite composition ($\delta^{57}\text{Fe} \geq 0.30\text{‰}$; Nebel et al.,
654 2019), nor subsequently produce the heaviest observed OIB melt compositions. Instead, forming
655 pyroxenite by hybridisation produces a range of subsequent melt compositions likely dominated
656 by $\delta^{57}\text{Fe}$ similar to MORB, but could produce some heavier $\delta^{57}\text{Fe}$ melts that contribute to the
657 heavy $\delta^{57}\text{Fe}$ seen in OIB.

658 5.4 Lithospheric processing and remobilisation

659 Remobilisation of small-degree pyroxenite melts, frozen in the lithosphere during the passage of
660 the main plume head, could generate liquids isotopically heavier than those from a single-stage
661 mantle melting process. Melt remobilisation superimposes a further partial melting fractionation
662 of $\Delta^{57}\text{Fe} \leq 0.07\text{‰}$ onto the resultant melts, which are likely to be small-degree melts. Therefore,
663 partial melting fractionations may be at the upper limit of the melting fractionation range; and
664 minimal melting of, and thus minimal dilution by, isotopically light peridotite is expected. If the
665 heaviest possible reaction-zone pyroxenite melts ($\delta^{57}\text{Fe} = 0.29\text{‰}$; Fig. 9) could become trapped
666 in the lithosphere and subsequently remobilised, it would be possible to generate small-degree
667 melts with $\delta^{57}\text{Fe} \lesssim 0.35\text{‰}$. These melts could explain some of the rejuvenated Samoan lavas,
668 since the Samoan lithosphere has previously passed over at least two hotspots (Jackson et al.,

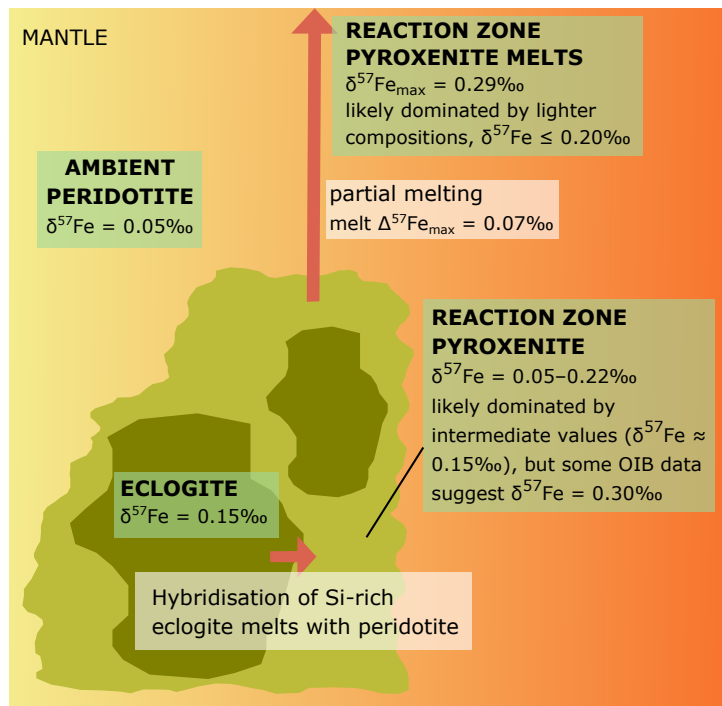


Figure 9: Schematic showing formation of a reaction zone pyroxenite, following Sobolev et al. (2005); Nebel et al. (2019), and the subsequent generation of heavy $\delta^{57}\text{Fe}$ melts. Eclogite melts hybridise with ambient peridotite, generating a pyroxenite mantle component with $\delta^{57}\text{Fe} \leq 0.22\text{‰}$, assuming a subducted eclogite composition of $\delta^{57}\text{Fe} = 0.15\text{‰}$.

669 2010) so is likely to contain trapped veins of melt. The rejuvenated lavas have been proposed to
 670 be small-degree melts remobilised by decompression melting due to plate flexure near the Tonga
 671 trench, consistent with their low SiO_2 , alkalic nature relative to shield samples, and may be an
 672 expression of ‘petit spot’ volcanism (Hawkins Jr and Natland, 1975; Natland, 1980; Konter and
 673 Jackson, 2012; Reinhard et al., 2019). However, erupted melts with compositions close to $\delta^{57}\text{Fe}$
 674 $= 0.35\text{‰}$ would require that: (i) reaction-zone pyroxenite melts were extracted through two
 675 partial melting events without mixing with ambient peridotite melts, and (ii) at each melting
 676 stage, Fe isotopic fractionation was at a maximum ($\Delta^{57}\text{Fe} = 0.07\text{‰}$), despite the limited P-T
 677 conditions where this $\Delta^{57}\text{Fe}$ is possible (Fig. 6). In addition, to generate any large volume of
 678 small-degree, maximum $\Delta^{57}\text{Fe}$ melts requires a residual source that is much larger in mass than
 679 the melts for both melting events. This volume problem would make it difficult to generate
 680 appreciable quantities of eruptable heavy $\delta^{57}\text{Fe}$ melt, but it is possible that remobilisation of
 681 melts with less extreme melting $\Delta^{57}\text{Fe}$ (hence a smaller volume problem) could provide a small
 682 contribution to heavy $\delta^{57}\text{Fe}$.

683 Metasomatic reactions of source peridotite with a melt or fluid, generating isotopic and chem-

684 ical disequilibrium and forming pyroxene-rich domains, may also be able to contribute to
685 heavy isotopic signatures in OIB. Where metasomatism involves enrichment by a Fe-rich, Si-
686 undersaturated melt (e.g., a silicate liquid that has evolved in mantle magma chambers and
687 veins; Weyer and Ionov, 2007), formation of secondary clinopyroxene (as seen in some Samoan
688 xenoliths; Finlayson et al., 2015) and generation of wehrlites (Ionov et al., 2005) could generate
689 a heavy $\delta^{57}\text{Fe}$ metasomatised source (Weyer and Ionov, 2007; Konter et al., 2016). The precip-
690 itation of garnet and clinopyroxene at the interface between subsolidus lherzolite and eclogite-
691 derived melts has also been shown experimentally and has occurred within the lithosphere of the
692 North China Craton (Wang et al., 2020). Sun et al. (2020) proposes that metasomatic garnet
693 pyroxenite veins in recycled oceanic lithosphere could melt and react with ambient peridotite
694 to produce secondary pyroxenites with $\delta^{57}\text{Fe} > 0.3\text{‰}$ although, as discussed in Section 5.3, it is
695 unclear how these heavy compositions will be generated. Metasomatic phlogopite has also been
696 recorded with heavy $\delta^{57}\text{Fe}$ in cratonic mantle xenoliths ($> 0.30\text{‰}$; Zhao et al., 2012), although
697 Beier et al. (2012) argue against phlogopite in the Azores source on the basis of trace elements.

698 The suggested magnitude of the metasomatic effect, based on xenolith data from Samoa (Fin-
699 layson et al., 2015), is $\Delta^{56}\text{Fe} = 0.05\text{‰}$ ($\Delta^{57}\text{Fe} = 0.08\text{‰}$; Konter et al., 2016), which cannot
700 generate the full $\delta^{57}\text{Fe}$ range of liquids observed in Samoa, but could contribute to producing
701 heavy $\delta^{57}\text{Fe}$. Some mantle samples affected by metasomatism record a strong negative correla-
702 tion between $\delta^{56}\text{Fe}$ and $\delta^{26}\text{Mg}$, which is consistent with an inter-diffusion disequilibrium origin
703 to the isotopic signatures (Zhao et al., 2012; Su et al., 2015), and likely due to coupled diffusion
704 into/out of mantle minerals. Therefore, Mg isotopic measurements might be able to identify
705 heavy Fe isotopic signatures of a metasomatic origin, which could be particularly applicable for
706 rejuvenated Samoan lavas (Konter et al., 2016).

707 Some SD pyroxenites are suggested to be lithospheric cumulates from low-degree mantle melts,
708 e.g., some Hawaiian pyroxenite xenoliths, which have $\delta^{57}\text{Fe} \leq 0.27\text{‰}$ (Bizimis et al., 2013;
709 Williams and Bizimis, 2014). These compositions have been explained by the progressive frac-
710 tional crystallisation in the lithosphere of an initial melt with $\delta^{57}\text{Fe} = 0.15\text{‰}$. However, the
711 heaviest $\delta^{57}\text{Fe}$ xenolith is required to be a cumulate fraction in equilibrium with a highly evolved
712 ($> 90\%$ fractionally crystallised) melt (Williams and Bizimis, 2014). For major elements, cumu-
713 late pyroxenite compositions can produce primary magmas that have higher CaO than eclogite-

714 derived melts, plotting above the dividing line in Fig. 7 (Herzberg and Asimow, 2008), although
715 this CaO enriched composition would be inconsistent with the major element chemistry of most
716 of the Azores and Samoan lavas shown here. Therefore, while xenoliths record the presence
717 of heavy $\delta^{57}\text{Fe}$ ($> 0.25\text{‰}$) lithospheric components, their mode of formation and relevance to
718 heavy $\delta^{57}\text{Fe}$ in OIB, particularly large-volume melts, remains uncertain.

719 The pre-final melting processes that have been discussed are summarised in Table 2, highlighting
720 key points about the possibility of each consideration in explaining $\delta^{57}\text{Fe}$ of melts.

Table 2: Summary of processes that are discussed to contribute to heavy $\delta^{57}\text{Fe}$ melts in OIB. We consider the potential effects on melt $\delta^{57}\text{Fe}$, in addition to major elements; and potential melt volume generated from this contribution to mantle heterogeneity, to identify plausible processes that are traced by $\delta^{57}\text{Fe}$ in OIB melts.

Process	Contribution to:			Section
	$\delta^{57}\text{Fe}$	Major elements	Melt volume	
Core contribution	$> 0.05\text{‰}$ heavier than silicate mantle but not consistent with other isotopes	No difference	Could be mixed into whole plume to some extent	5.1
Make eclogite $\delta^{57}\text{Fe} > 0.15\text{‰}$ by:				
(i) Hydrothermal alteration of oceanic crust	Bulk effect suggested to be small	Large variability only from highly altered protolith	Could be large component	5.2.1
(ii) Metamorphism and/or dehydration of subducted basalt	Bulk effect uncertain, could be small due to mass balance in slab	Likely small due to mass balance in slab	Could be large component	5.2.1
(iii) Composition of ancient MORB	Poorly constrained, but likely negligible	Poorly constrained	Could be large component	5.2.2
(iv) Sediment input to the mantle	Sediments $\leq 0.9\text{‰}$, but bulk effect could be small	Bulk effect likely small	Likely small	5.2.3
Generate reaction-zone pyroxenites	Melts $\leq 0.29\text{‰}$, likely closer to 0.15‰	Mix between peridotite and pyroxenite melt compositions	Could be large component	5.3
Lithospheric processing				
(i) Melt remobilisation	Superimposes a further $\Delta^{57}\text{Fe}_{\text{max}} = 0.07\text{‰}$	Small-degree OIB melts likely alkalic	Likely small	5.4
(ii) Metasomatism	Could increase melt $\delta^{57}\text{Fe}$ due to pyroxene enrichment	Measurable e.g., CaO enrichment	Likely small	5.4

721 6 Preservation of heavy $\delta^{57}\text{Fe}$: Importance of plume variables

722 Not all OIB record heavy $\delta^{57}\text{Fe}$ (e.g., Hawai'i, Samoan shield, Réunion), which suggests that the
723 processes controlling the generation and preservation of heavy $\delta^{57}\text{Fe}$ in erupted lavas may rely

724 on factors that differ between plumes, e.g., melt fraction and potential temperature, in addition
725 to the $\delta^{57}\text{Fe}$ of source material.

726 The Azores records heavy $\delta^{57}\text{Fe}$ and is a relatively cool plume (Putirka, 2008; Beier et al., 2012),
727 and the heavy $\delta^{57}\text{Fe}$ rejuvenated Samoan lavas are small-degree melts (Konter and Jackson,
728 2012). Therefore, heavy $\delta^{57}\text{Fe}$ melts may be more likely to be preserved in plumes where they
729 are not diluted by extensive degrees of melting and/or mixing with ambient peridotite mantle
730 melts, regardless of how a heavy $\delta^{57}\text{Fe}$ source component is generated. Hawai'i and Samoa have
731 notably high excess temperatures, with a potential temperature anomaly of $> 200^\circ\text{C}$ (Putirka,
732 2005) compared to only 35°C in the Azores (Beier et al., 2012), and record no heavy $\delta^{57}\text{Fe}$ in
733 erupted shield stage melts (Teng et al., 2008, 2013; Konter et al., 2016). However, a link between
734 plume excess temperature and $\delta^{57}\text{Fe}$ is likely to be complicated by other factors affecting melt
735 fraction, such as lithospheric thickness and mantle source components, and we note that some
736 plumes (for example Réunion) with a similar excess temperature to Pitcairn (Putirka, 2008)
737 have no measured heavy $\delta^{57}\text{Fe}$ lavas.

738 **7 Summary**

739 Stable Fe isotopes have increasingly been used as a tracer of mineralogical heterogeneity in the
740 mantle, but there are multiple processes that could generate heavy $\delta^{57}\text{Fe}$ mantle melts. We
741 show that the existing dataset of $\delta^{57}\text{Fe}$, $\mu^{182}\text{W}$ and $^3\text{He}/^4\text{He}$ of OIB are inconsistent with a
742 contribution from heavy $\delta^{57}\text{Fe}$ core liquids. In agreement with previous work, we also calculate
743 that the magnitude of partial melting fractionation of peridotite is small. We show that the
744 partial melting fractionation of an eclogitic pyroxenite is similar to that of peridotite, and less
745 than that required to explain the heaviest $\delta^{57}\text{Fe}$ seen in basalts — therefore, heterogeneity in
746 melt $\delta^{57}\text{Fe}$ derives from the bulk $\delta^{57}\text{Fe}$ of source lithologies. Mixed melts of KLB1 peridotite
747 and G2 eclogitic pyroxenite (with bulk compositions of $\delta^{57}\text{Fe} = 0.05$ and MORB-like 0.15‰
748 respectively) produce liquids that match some of the Samoan shield and Azores melts, but
749 would need to comprise nearly pure pyroxenite melts to explain the heaviest $\delta^{57}\text{Fe}$, and cannot
750 explain any of the rejuvenated Samoan samples.

751 We show that to produce a mantle source capable of generating melts that match the OIB $\delta^{57}\text{Fe}$

752 range requires consideration of processes operating from MORB generation, through mantle
753 heterogeneity development and lithospheric processing, to eruption at ocean islands. The dom-
754 inant contributor to geochemical heterogeneity recorded in OIB is recycled oceanic crust, which
755 could generate heavy $\delta^{57}\text{Fe}$ ($\gtrsim 0.2\%$) melts if recycled crust was heavier in $\delta^{57}\text{Fe}$ than modern
756 MORB ($> 0.15\%$). Heavy $\delta^{57}\text{Fe}$ in recycled material could be created by hydrothermal alter-
757 ation, metamorphic reactions during subduction, or Fe isotope contributions from sediments,
758 but we suggest no single, unique factor can explain the heaviest OIB $\delta^{57}\text{Fe}$ ($> 0.25\%$) lavas.
759 Reaction-zone pyroxenites in the mantle, formed by the hybridisation of modelled equilibrium
760 eclogitic melts and peridotite, will not be sufficiently isotopically heavy to match the proposed
761 heavy $\delta^{57}\text{Fe}$ mantle component beneath Pitcairn and the Galapagos (Nebel et al., 2019; Gleeson
762 et al., 2020), unless eclogite in the mantle has $\delta^{57}\text{Fe} > 0.15\%$, i.e., has been modified from
763 MORB values. Further processes contributing to heavy $\delta^{57}\text{Fe}$ melts could include the forma-
764 tion of isotopically heavy metasomatic pyroxene-rich domains and cumulates in the lithosphere,
765 which may be remelted during late stages of plume activity (such as small-scale rejuvenated
766 volcanism).

767 Any heavy $\delta^{57}\text{Fe}$ mantle or lithospheric component may most likely be observed in erupted lavas
768 where overall melt volume and/or degree of melting is small, e.g., plumes with relatively low
769 potential temperature. The measurement of MORB-like $\delta^{57}\text{Fe}$ signatures in an ocean island
770 system therefore does not rule out the presence of a heavy $\delta^{57}\text{Fe}$ component in the mantle, as
771 melts derived from this component may be too dilute in the erupted liquid to be distinguished
772 by Fe isotopes.

773 Heavy $\delta^{57}\text{Fe}$ in OIB may have an important role to play in identifying mantle source lithologies,
774 but the presence or absence of a heavy $\delta^{57}\text{Fe}$ signature in melts is unlikely to simply relate to a
775 uniform recycled component in the mantle. Instead, Fe isotopes in OIB reflect a variety of often
776 poorly-understood, difficult-to-distinguish and plume-specific processes that have contributed
777 to variable $\delta^{57}\text{Fe}$ in the mantle and erupted melts.

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