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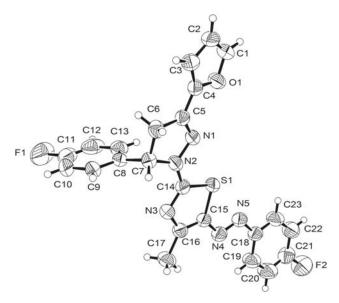
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Crystal structure of (*E*)-2-(5-(4-fluorophenyl)-3-(furan-2-yl)-4,5-dihydro-1*H*-pyrazol-1-yl)-5-((4fluorophenyl)diazenyl)-4-methylthiazole, $C_{23}H_{17}F_2N_5OS$



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Abstract

C₂₃H₁₇F₂N₅OS, monoclinic, P2₁/c (no. 14), a = 5.2272(4) Å, b = 26.7398(15) Å, c = 15.2645(10) Å, $\beta = 97.726(7)^{\circ}$, V = 2114.2(2) Å³, Z = 4, $R_{gt}(F) = 0.0547$, $wR_{ref}(F^2) = 0.1371$, T = 296(2) K.

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Amany S. Hegazy and Benson M. Kariuki: School of Chemistry, Cardiff University, Main Building, Park Place, Cardiff CF10 3AT, UK Table 1: Data collection and handling.

Crystal:	Orange needle		
Size:	0.82 imes 0.27 imes 0.06 mm		
Wavelength:	Mo <i>Kα</i> radiation (0.71073 Å)		
μ:	2.0 cm ⁻¹		
Diffractometer, scan mode:	SuperNova, ω-scans		
2 θ_{max} , completeness:	59.4°, >84% (99% to 50.4 2 <i>0</i>)		
N(hkl) _{measured} , N(hkl) _{unique} , R _{int} :	10634, 5088, 0.031		
Criterion for I _{obs} , N(hkl) _{gt} :	$I_{ m obs}$ $>$ 2 $\sigma(I_{ m obs})$, 3027		
N(param) _{refined} :	290		
Programs:	CrysAlis ^{PRO} [14], SHELX [15], PLATON [16]		

The asymmetric unit of the title crystal structure is shown in the figure. Tables 1 and 2 contain details on crystal structure and measurement conditions and a list of the atoms including atomic coordinates and displacement parameters.

Source of material

The title compound was synthesized from reaction of a mixture of 1:1 molar ratios of 5-(4-fluorophenyl)-3-(furan-2-yl)-4,5-dihydro-1*H*-pyrazole-1-carbothioamide and *N'*-(4-fluorophenyl)-2-oxopropanehydrazonoyl chloride in ethanol under reflux condition for 2 h. The solid obtained on cooling was recrystallized from dimethylformamide to give the title compound as orange crystals in 64% yield, Mp. 225–226 °C [1].

Experimental details

All hydrogen atoms were placed in calculated positions and refined using a riding model. Methyl, methylene and methine C—H bonds were fixed at 0.96 Å, 0.97 Å and 0.98 Å respectively. Displacement parameters were 1.5 times $U_{eq}(C)$ for methyl groups and 1.2 times $U_{eq}(C)$ for methylene and methine hydrogens. Methyl groups were allowed to spin about the C—C bond. Aromatic C—H distances were set to 0.93 Å and their U_{iso} set to 1.2 times $U_{eq}(C)$.

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Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

Atom	X	у	Z	U _{iso} */U _{eq}
C1	0.6495(5)	0.04738(10)	0.59780(18)	0.0655(7)
H1	0.7744	0.0224	0.6059	0.079*
C2	0.6361(5)	0.08585(10)	0.65183(18)	0.0649(7)
H2	0.7473	0.0926	0.7033	0.078*
С3	0.4207(5)	0.11438(10)	0.61618(17)	0.0615(7)
H3	0.3617	0.1436	0.6396	0.074*
C4	0.3163(5)	0.09110(8)	0.54141(15)	0.0500(6)
C5	0.0979(5)	0.10124(8)	0.47659(15)	0.0484(5)
C6	-0.0673(5)	0.14694(9)	0.47670(16)	0.0561(6)
H6A	0.0359	0.1772	0.4797	0.067*
H6B	-0.1676	0.1465	0.5257	0.067*
C7	-0.2426(5)	0.14272(8)	0.38738(15)	0.0511(6)
H7	-0.4242	0.1455	0.3963	0.061*
C8	-0.1805(4)	0.18103(8)	0.32111(15)	0.0474(5)
C9	-0.3144(5)	0.22552(9)	0.31469(19)	0.0663(7)
H9	-0.4475	0.2306	0.3485	0.080*
C10	-0.2526(8)	0.26247(12)	0.2587(3)	0.0966(13)
H10	-0.3431	0.2925	0.2542	0.116*
C11	-0.0596(9)	0.25460(16)	0.2104(2)	0.1000(14)
C12	0.0778(7)	0.21132(16)	0.21420(19)	0.0889(11)
H12	0.2106	0.2070	0.1800	0.107*
C13	0.0149(5)	0.17366(10)	0.27052(16)	0.0610(7)
H13	0.1049	0.1436	0.2739	0.073*
C14	-0.3075(5)	0.06521(8)	0.29387(15)	0.0510(6)
C15	-0.4537(5)	0.00457(9)	0.18424(16)	0.0521(6)
C16	-0.5903(5)	0.04815(9)	0.17850(16)	0.0534(6)
C17	-0.8163(5)	0.05989(11)	0.11044(18)	0.0706(8)
H17A	-0.9659	0.0658	0.1391	0.106*
H17B	-0.8485	0.0322	0.0705	0.106*
H17C	-0.7791	0.0322	0.0780	0.106*
C18	-0.3962(5)	-0.11526(9)	0.09743(16)	0.0549(6)
C18 C19	-0.6020(6)	-0.11320(9) -0.12125(11)	0.0320(2)	0.0349(0)
H19	-0.7214	-0.0956	0.0320(2)	0.0775(9)
C20	-0.6318(7)	-0.16490(13)	-0.0153(2)	0.093
H20	-0.7712	-0.16490(13) -0.1691	-0.0594	0.0914(10)
C21	-0.4549(6)	-0.20178(11)	0.00311(19)	0.0728(8)
C21	-0.4549(6) -0.2510(6)	-0.20178(11) -0.19752(11)	0.0663(2)	
	-0.2510(6)		0.0663(2)	0.0793(9)
H22		-0.2235		0.095*
C23	-0.2219(6)	-0.15362(11)	0.1137(2)	0.0751(8)
H23	-0.0816	-0.1499	0.1576	0.090*
N1	0.0261(4)	0.07057(7)	0.41315(13)	0.0538(5)
N2	-0.1851(4)	0.09153(7)	0.36206(13)	0.0564(5)
N3	-0.5072(4)	0.08338(7)	0.24102(13)	0.0548(5)
N4	-0.5024(4)	-0.03614(8)	0.13060(13)	0.0567(5)
N5	-0.3458(4)	-0.07204(8)	0.15071(13)	0.0579(5)
01	0.4563(3)	0.04935(6)	0.52908(11)	0.0641(5)
F1	0.0032(6)	0.29140(10)	0.15608(16)	0.1654(13)
F2	-0.4863(4)	-0.24579(7)	-0.04261(13)	0.1111(7)
S1	-0.20373(13)	0.00572(2)	0.27253(4)	0.0553(2)

Discussion

Many pyrazolylthiazoles have been synthesized using different procedures and showed antinociceptive, anti-inflammatory and antimicrobial activities [2–10]. The X-ray crystal structures for related compounds have been published recently [11, 12].

The asymmetric unit consists of one molecule. In the molecule, the furan(A)-pyrazole(B)-thiazole(C)-fluorophenyl(D) ring system is almost planar. Thus the largest deviation from the least-squares plane through the four rings is 0.22(1) Å (by O1). The greatest difference between the planes through adjacent rings (A and B) is 7.1(2)°. The second fluorophenyl group (E) is almost perpendicular (85.0(5)°) to the A—B—C—D system. In the crystal, pairs of molecules related by an inversion centre interact through two edge-to-face interactions involving rings D and E with centroid-to-centroid distances of 5.3 Å. A short intermolecular O···O contact (2.84 Å) occurs between furan moieties of pairs of molecules related by inversion symmetry. Such contact is not unique, as shown by a search of the CSD [13] which gave 78 hits for contacts within the sum of van-der-Waals radii for furan oxygens.

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