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Streamlined Synthesis of C(sp³)-Rich N-Heterospirocycles Enabled by Visible-Light-Mediated Photocatalysis

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* Supporting Information

ABSTRACT: We report a general visible-light-mediated strategy that enables the construction of complex C(sp³)-rich N-heterospirocycles from feedstock aliphatic ketones and aldehydes with a broad selection of alkene-containing secondary amines. Key to the success of this approach was the utilization of a highly reducing Ir-photocatalyst and orchestration of the intrinsic reactivities of 1,4-cyclohexadiene and Hantzsch ester. This methodology provides streamlined access to complex C(sp³)-rich N-heterospirocycles displaying structural and functional features relevant to fragment-based lead identification programs.

Growing evidence suggests that an increasing number of aromatic rings in lead compounds can result in greater attrition rates among pharmaceutical candidates due to poor solubility, bioavailability, and pharmacokinetics.¹ However, libraries of lead-like structures often comprise compounds with predominantly C(sp²)-rich scaffolds. Consequently, the assessment of structurally distinct libraries of C(sp³)-rich small molecules displaying diverse polar functionality could help to identify new lead candidates in fragment-based screening approaches that may exhibit enhanced physical and biological properties.² In considering the design of novel C(sp³)-rich small molecules for fragment-based approaches,³ it is noticeable that conformationally restricted scaffolds show highly reproducible results in in silico screening programs; the well-defined spatial orientation of molecular features in a candidate compound often leads to increased binding affinities. One approach toward limiting structural flexibility within C(sp³)-rich small molecules is through the introduction of a spirocycle.⁴ In comparison to aromatic scaffolds, fine-tuning the structural and functional features in these frameworks offers an alternative means through which to probe interactions with target binding sites by varying ring size, adjusting electronic properties, and manipulating substituent effects.⁵ As a result, spirocyclic motifs displaying polar functionality are emerging in pharmaceutical agents and lead compounds (Figure 1a).⁴

Despite the attractive properties offered by rigid saturated polar spirocyclic scaffolds, access to diversely functionalized variants mainly relies on multistep alkyl- and acylative methods.⁵ Notable advances on these approaches include dearomatization-based strategies, using Pd, Ir, or Fe catalysts,^{6a-c} which harness the intrinsic reactivity of tethered (hetero)arene building blocks for the generation of a selection of spirocyclic N-heterocycles

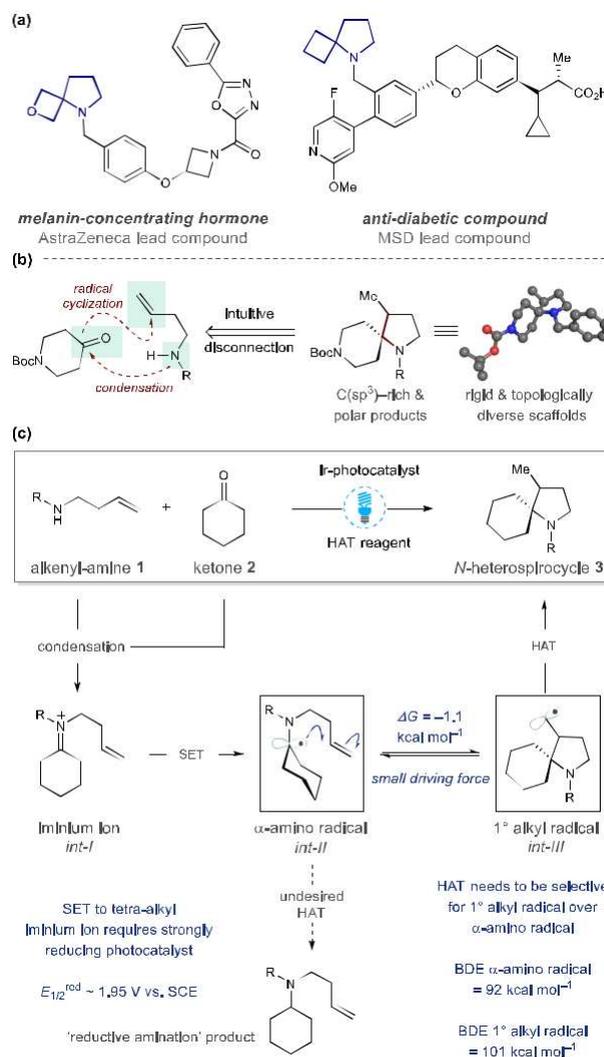


Figure 1. Evolution of the cyclization strategy. Gibbs free energy calculated by DFT methods (UPW6B95-D3BJ/def2-QZVP). Hydrogens removed for clarity.

with different ring sizes. Bode and co-workers produced a range of spirocyclic N-heterocycles using their SnAP technology;^{6d,e}

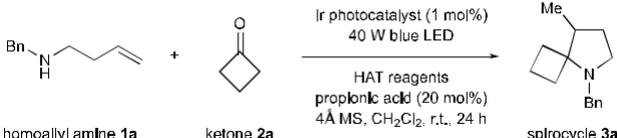
iminium ion formation between amino-stannane reagents and cyclic ketones enables a stoichiometric copper(II) cyclization to access free(NH)-secondary amino-spirocyclic products. To complement these approaches, we reasoned that a photocatalytic strategy⁷ based on an intuitive retrosynthetic disconnection of tertiary amine-based N-heterospirocycles directly to feedstock ketones and readily available alkenyl-derived secondary amines could be realized through the intermediacy of an α -amino radical (Figure 1b).⁸ The structurally distinct C(sp³)-rich polar N-heterospirocycles generated by such a catalytic transformation could be of interest to synthetic and medicinal chemists.

We envisioned that visible-light-mediated photocatalytic single-electron reduction of an alkyl-iminium ion (int-I, Figure 1c), formed from the condensation of a saturated cyclic ketone with a secondary alkylamine, would form a cyclic-tertiary α -amino radical (int-II). Addition of this α -amino radical to an appended alkene on the amine component would give rise to a substituted pyrrolidine-based spirocyclic scaffold. Recent work from our group established a photocatalytic platform for the addition of nucleophilic α -amino radicals (formed from the corresponding iminium ions) to electron-deficient olefins.⁹ Critically, however, a catalytic manifold for the addition of all-alkyl-substituted α -amino radicals to unactivated alkenes remains an unsolved problem, principally due to the mismatched electronics and low thermodynamic driving force for such a reaction. The lone pair on the nitrogen atom stabilizes the α -amino radical but also renders it nucleophilic. Therefore, its addition to an unactivated alkene is polarity-mismatched and produces an alkyl radical (int-III) with no stabilizing substituents that, in turn, can undergo the reverse reaction to re-form the α -amino radical via C–C bond β -scission. Indeed, the use of SmI₂, a strong stoichiometric reductant, was required to trap related alkyl radicals via organosamarium intermediates following addition to unactivated alkenes.¹⁰ However, the metal waste produced in these reactions and frequent need for highly dilute conditions hinder the wider application of this type of process.¹¹ We, therefore, questioned whether an alkyl radical could be selectively intercepted in a milder fashion by way of a hydrogen-atom-transfer (HAT) reaction and in preference to HAT to the α -amino radical.

Herein, we detail the successful realization of these ideas through the development of a streamlined strategy for the synthesis of complex C(sp³)-rich N-heterospirocycles from readily available secondary amines and ketones enabled by visible-light photocatalysis. Keys to the success of this strategy were utilizing a highly reducing Ir-photocatalyst and orchestrating the intrinsic reactivities of 1,4-cyclohexadiene (1,4-CHD) and Hantzsch ester, which facilitated controlled generation and the fate of the radical intermediates present in the reaction. Under optimized conditions, the process combines a range of feedstock aliphatic ketones and aldehydes with alkene-containing secondary amines to forge complex C(sp³)-rich N-heterospirocycles displaying structural, functional, and physical features that are likely to be attractive in fragment-based lead identification programs.

We began our investigations by testing reaction conditions based on our previously established photocatalytic process.⁹ Using amine 1a and cyclobutanone (2a) as substrates, 1 mol% of Ir(ppy)₃ and 2.0 equiv of Hantzsch ester, we were encouraged by the formation of spirocycle 3a in 30% yield and only trace amounts of the corresponding reductive amination product (Table 1, entry 1). Following extensive optimization, we found that the use of 1,4-CHD as an additive (entry 2) and a more

Table 1. Selected Optimization Data^a



entry	catalyst	Hantzsch ester (equiv)	1,4-CHD (equiv)	yield (%)
1	Ir(ppy) ₃	2.0	–	30
2	Ir(ppy) ₃	2.0	7.5	45
3	Ir(dMeppy) ₃	2.0	7.5	97
4	Ir(dMeppy) ₃	2.0	–	24
5	Ir(dMeppy) ₃	–	7.5	43
6	–	2.0	7.5	trace
7	Ir(dMeppy) ₃	–	–	trace

^aReactions employed a 40 W Blue Kessil lamp, and yields were calculated by GC assay with dodecane as internal standard.

reducing photocatalyst, Ir(dMeppy)₃ ($[E_{1/2}^{\text{red}}(\text{Ir}^{\text{IV}}/\text{Ir}^{\text{III}})] = -1.86 \text{ V vs SCE in CH}_2\text{Cl}_2$) (entry 3), improved the yield to 97%.¹² We reasoned that the use of 1,4-CHD as a HAT-donor in high concentration would restrict the role of the Hantzsch ester to turning over the Ir(IV) species formed after single-electron reduction of the iminium ion, while the more reducing photocatalyst would provide a stronger driving force for the reduction step. A series of control experiments showed that the yield of the reaction dropped to 24% in the absence of 1,4-CHD (entry 4); to 43% without the Hantzsch ester (entry 5); and to trace amounts without photocatalyst or light or in the absence of both 1,4-CHD and Hantzsch ester (entries 6 and 7).

Having evaluated the reaction conditions, a range of commercially available ketones (Table 2) were reacted to afford N-heterospirocycles 3a–3i. Aldehydes also reacted to afford the corresponding pyrrolidines 3j–3m, with the syn-isomer of 3k being the major diastereomer. The reaction was amenable to heterocyclic ketones, with a series of saturated N-heterocycles readily transformed into rigid, C(sp³)-rich, and polar bis-N-heterospirocyclic scaffolds with orthogonal protecting groups on the two nitrogen atoms (3n–3t), facilitating further derivatization. The reaction to form azetidine-derived spirocycle 3p could be performed on a gram scale, even with a reduced catalyst loading of 0.5 mol%, forming the product in a 63% yield; the N-groups in 3p could also be selectively deprotected for further derivatization.¹³ Such derivatives are frequently employed as β -turn mimetics for GPCR protein binding, where the orthogonality of the two nitrogen substituents is key to achieve the desired turn-characteristics.¹⁴ While the Ir(dMeppy)₃ photocatalyst had proven optimal for reactions using electron-rich ketones (to form 3a–3e, 3i, 3l), higher yields were observed for electron-deficient ketones (to form 3f–3h, 3j, 3k, and 3m–3t) using the less-reducing [Ir(dMeppy)₂{dtbbpy}]PF₆ photocatalyst.

We next assessed N-substituted alkenylamines 3u–3an, where useful functional groups such as sulfonamide, N-Boc-piperidine, heteroaryl motifs (3u–3ab), and a less nucleophilic aniline (3ac) could be built into the spirocycle products. Substitution on the alkene was tolerated (3ad–3ah), introducing valuable functionalities such as the ester and gem-difluoro in useful yields. While no cyclized product was observed from attempts at reacting N-benzyl allylamine (to afford azetidine-based spiro-cycles), a piperidine scaffold 3ai could be assembled via a 6-exo cyclization, further expanding scope of the process. Homoallylic amines containing α -substituents were effective in the

nection for difficult-to-access C(sp³)-rich N-heterospirocyclic scaffolds that may be of interest to practitioners of both synthetic and medicinal chemistry.

ASSOCIATED CONTENT

* Supporting Information

Experimental procedures and spectral data (PDF)

X-ray crystallographic data for 6 (CIF)

X-ray crystallographic data for 4a (CIF)

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Notes

The authors declare no competing financial interest.

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