# Design and Synthesis of Imidazole and Triazole Pyrazoles as Mycobacterium Tuberculosis CYP121A1 Inhibitors 

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The emergence of untreatable drug-resistant strains of Mycobacterium tuberculosis is a major public health problem worldwide, and the identification of new efficient treatments is urgently needed. Mycobacterium tuberculosis cytochrome P450 CYP121A1 is a promising drug target for the treatment of tuberculosis owing to its essential role in mycobacterial growth. Using a rational approach, which includes molecular modelling studies, three series of azole pyrazole derivatives were designed through two synthetic pathways. The synthesized compounds were biologically evaluated for their inhibitory activity towards M. tuberculosis and their protein binding affinity ( $K_{\mathrm{D}}$ ). Series 3 biarylpyrazole imidazole derivatives were the most effective


#### Abstract

with the isobutyl (10f) and tert-butyl (10g) compounds displaying optimal activity (MIC $1.562 \mu \mathrm{~g} / \mathrm{mL}, K_{\mathrm{D}} 0.22 \mu \mathrm{M}$ (10f) and $4.81 \mu \mathrm{M}(10 \mathrm{~g})$ ). The spectroscopic data showed that all the synthesised compounds produced a type II red shift of the heme Soret band indicating either direct binding to heme iron or (where less extensive Soret shifts are observed) putative indirect binding via an interstitial water molecule. Evaluation of biological and physicochemical properties identified the following as requirements for activity: $\log \mathrm{P}>4, \mathrm{H}$-bond acceptors/Hbond donors $4 / 0$, number of rotatable bonds $5-6$, molecular volume $>340 \AA^{3}$, topological polar surface area $<40 \AA^{2}$.


## 1. Introduction

Tuberculosis (TB) is the ninth leading cause of death worldwide and the leading cause from a single infectious agent, ranking above HIV/AIDS. ${ }^{[1]}$ In 2016, 6.3 million new cases of TB were reported; of these there were an estimated 1.3 million deaths among HIV-negative people and an estimated 374,000 deaths among HIV-positive people. ${ }^{[1]}$ Drug-resistant TB is a continuing challenge, with an estimated $4.1 \%$ of new cases and $19 \%$ of previously treated cases presenting with multidrug resistant (MDR)/rifampicin resistant (RR) TB in 2016. ${ }^{[1]}$ China, India and the Russian Federation have the largest number of MDR/RR-TB cases accounting for $47 \%$ of the global total. ${ }^{[1]}$

Drugs currently used to treat TB target cell wall synthesis (e.g. isoniazid, ethambutol, cycloserine), protein synthesis (e.g. capreomycin, kanamycin), RNA synthesis (e.g. rifampicin), DNA gyrase (e.g. fluoroquinolones) and ATP synthase (e.g. bedaquiline and delamanid) in Mycobacterium tuberculosis (Mtb), the causative agent of TB. ${ }^{[2]}$ The first line TB drug regimen includes a combination of isoniazid, ethambutol, rifampicin and pyrazinamide over a 6-9 month period and, whilst effective, this therapy is associated with drug intolerance and toxicity. ${ }^{[3]}$ Treatment of MDR-TB and extensively drug resistant (XDR) TB is more complex and also associated with increased side effects. ${ }^{[2,4]}$ The emergence of MDR-TB, XDR-TB and totally drug resistant (TDR) Mtb strains has led to intensified research to identify new anti-TB drugs over the past decade.

Mtb encodes twenty cytochrome P450 enzymes (CYPs or P450s), three of these (CYP121A1, CYP125 A1 and CYP128 A1) were shown to be essential for mycobacterial growth or survival in the host. ${ }^{[5]}$ CYP121A1 is of considerable interest from both a biochemical and drug design perspective. CYP121A1 catalyses a




Lead compounds
(1) $\mathrm{R}=\mathrm{H}, \mathrm{X}=\mathrm{CH}: \mathrm{K}_{\mathrm{D}} 2.63 \pm 0.19 \mu \mathrm{M}$, MIC $12.5 \mu \mathrm{~g} / \mathrm{mL}$
(2) $\mathrm{R}=\mathrm{H}, \mathrm{X}=\mathrm{N}: \mathrm{K}_{\mathrm{D}} 73.5 \pm 0.7 \mu \mathrm{M}$, MIC $50 \mu \mathrm{~g} / \mathrm{mL}$
(3) $\mathrm{R}=\mathrm{I}, \mathrm{X}=\mathrm{CH}: \mathrm{K}_{\mathrm{D}} 13.6 \pm 0.2 \mu \mathrm{M}$, MIC $6.25 \mu \mathrm{~g} / \mathrm{mL}$

Figure 1. Lead compounds (1-3), the natural substrate cYY and CYP121A1 inhibitor fluconazole.
unique C-C bond formation between carbon atoms in the ortho-positions of two tyrosine groups in the cyclodipeptide dicyclotyrosine (cYY, Figure 1), the natural substrate, to form the metabolite mycocyclosin. ${ }^{[6]}$ The role of mycocylosin in Mtb remains unclear, although cyclodipeptides have important biological effects, such as inhibition of production of bacterial virulence factors and biofilm formation. ${ }^{[7,8]}$ Drug design for CYP121A1 inhibitors has followed several approaches: (1) a fragment-based approach has been very effective in developing compounds with low nM enzyme binding affinity, however this has not translated to antimycobacterial activity $;^{[9,10]}$ (2) a recent biophysical screening of a library of described P450 inhibitors has led to a promising lead compound with both good binding affinity and antimycobacterial activity; ${ }^{[11]}$ (3) our own approach has focused on developing a series of different compound scaffolds as mimics of cYY, resulting in both good binding affinity and promising antimycobacterial activity. ${ }^{[12,13]}$ Our previous research identified derivatives of (1,3-diphenyl)-4-((1,3-imidazol-1-yl)methyl)-1H-pyrazole and 1-((1,3-diphenyl-1H-pyra-zol-4-yl)methyl)-1 $--_{-1,2,4]}^{[1]}$ triazole as inhibitors of CYP121A1 (e.g. 1,2 and 3, Figure 1 ), ${ }^{[13]}$ representing lead compounds for further development. Diphenyl pyrazole derivatives were found to interact with the active site of CYP121A1 in a manner similar to fluconazole and the natural substrate cYY (Figure 1), interacting indirectly with the heme via an interstitial water molecule (Figure 2).

Introduction of a N atom in one of the phenyl rings, i.e. a pyridine ring, can potentially result in additional binding interactions, which may improve binding affinity. Likewise, extension of the chain between the pyrazole and the heme prosthetic group may allow the molecule to interact directly with the heme: this is the basis for Series 1 (Figure 3). The requirement for two phenyl rings was also explored in Series 2 by preparation of a small series of imidazole/triazole derivatives, with one of the phenyl rings replaced with a methyl group


Figure 2. (A) 3D and (B) 2D images of the binding interactions of lead compound 1 in the CYP121A1 active site. Binding with the heme is through an interstitial water molecule with Ser237, Gln385, Arg386 and Thr77 as key binding amino acids.
(Figure 3). Finally, our previous research showed a direct correlation between lipophilicity and MIC. ${ }^{[13]}$ Therefore in Series 3 one of the phenyl rings is further substituted with a range of alkyl substituted phenyl and biaryl groups to determine structure-activity relationships (Figure 3).

## 2. Results and Discussion

### 2.1. Chemistry

Series 1 pyridylpyrazole imidazole compounds ( 10 and 11) were obtained via a five-step synthetic route (Scheme 1) beginning with the preparation of the imines $(6)^{[14]}$ on reaction of the appropriate 3 - or 4 -acetylpyridine reagent (4) with phenylhydrazine (5) under acidic conditions. ${ }^{[15,16]}$ The yields obtained for the imines (6) were 98 and $97 \%$ respectively for the 4 pyridyl (6a) and 3-pyridyl (6b) imines. The aldehydes (7) were prepared as previously described ${ }^{[14]}$ by a Vilsmeier-Haack reaction of the imines (6).

Reduction of the aldehydes (7) with $\mathrm{NaBH}_{4}$ resulted in high yields of the corresponding alcohols (8). Thionyl chloride was the chlorinating reagent used to convert the alcohols to the


Figure 3. Modifications of the lead compounds, generating Series 1, 2 and 3
chlorides (9), different equivalents of thionyl chloride were used to optimize the reaction conditions ( 1,4 and 10 equivalents), and the maximum yield was obtained when 10 equivalents of the reagent were used. Reaction of the chlorides (9) with either the potassium salt of imidazole or triazole, prepared in situ by treatment of imidazole or triazole with potassium carbonate in acetonitrile at $45^{\circ} \mathrm{C}$ for $1 \mathrm{~h},^{[12,13]}$ and overnight reflux at $70^{\circ} \mathrm{C}$ gave the required final imidazole (10a and 10b) and triazole ( 11 a and 11 b) pyridyl-pyrazole derivatives (Scheme 1) in yields of $67,54,48$ and $55 \%$ respectively.

To prepare the extended compounds, the aldehydes (7) were oxidized with potassium permanganate to give the carboxylic acids (12). ${ }^{[14]}$ The carboxylic acids were then coupled with either 1-(3-aminopropyl)imidazole or 4-(aminomethyl) pyridine, after activation of the carboxylic acid with carbon-


Scheme 1. Reagents and Conditions: (i) $\mathrm{AcOH}, \mathrm{EtOH}, 80^{\circ} \mathrm{C}, 1 \mathrm{~h}$ (ii) $\mathrm{POCl}_{3}, \mathrm{DMF}$, $60^{\circ} \mathrm{C}, 4 \mathrm{~h}$ (iii) $\mathrm{NaBH}_{4}, \mathrm{EtOH}$, r.t. 1 h (iv) $\mathrm{SOCl}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, r.t. overnight (v) imidazole, $\mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{CH}_{3} \mathrm{CN}, 45^{\circ} \mathrm{C}, 1 \mathrm{~h}$ then $70^{\circ} \mathrm{C}$, overnight (vi) triazole, $\mathrm{K}_{2} \mathrm{CO}_{3}$, $\mathrm{CH}_{3} \mathrm{CN}, 45^{\circ} \mathrm{C}, 1 \mathrm{~h}$ then $70^{\circ} \mathrm{C}$, overnight (vii) $\mathrm{KMnO}_{4}$, acetone $/ \mathrm{H}_{2} \mathrm{O}$, r.t. 1.5 h (viii) (a) CDI, DMF, r.t. 2 h (b) 1-(3-aminopropyl)imidazole, $70^{\circ} \mathrm{C}, 48 \mathrm{~h}$ (ix) (a) CDI, DMF, r.t. 2 h (b) 4 -(aminomethyl)pyridine, $70^{\circ} \mathrm{C}, 48 \mathrm{~h}$. [a, $\mathrm{Ar}=4$-pyridyl; b, $\mathrm{Ar}=3$-pyridyl]
yldiimidazole, to give the N -( 1 H -imidazolyl)propyl)-1-phenyl-3-(pyridyl)-1H-pyrazole-4-carboxamides (13a and 13 b ) and 1-phenyl-3-(pyridyl)- N -(pyridin-4-ylmethyl)-1H-pyrazole-4-carboxamides (14a and $14 b$ ) (Scheme 1).

In series 2, one of the phenyl rings was replaced with a methyl group to determine whether the second phenyl ring was significant for enzyme binding and inhibitory activity.

For the remaining phenyl ring the 4-methoxy phenyl group was chosen to more closely mimic the phenol ring of cYY (Figure 1). The $3 / 5$-(( 1 H -imidazol-1-yl)methyl)-1-(4-methoxy-phenyl)-5/3-methyl-1H-pyrazoles (23 and 25) and 1-((1-(4-methoxyphenyl)-5/3-methyl-1H-pyrazol-3/5-yl)methyl)-1H-1,2,4triazoles (24 and 26) were prepared in four steps (Scheme 2). Formation of the pyrazole ethyl esters (17 and 18) was achieved by the reaction of ethyl-2,4-dioxovalerate (15) with 4-methoxyphenyl hydrazine (16) by a Knorr pyrazole synthesis mechanism ${ }^{[17]}$ as previously described. ${ }^{[18]}$ The two regioisomers, ethyl 1-(4-methoxyphenyl)-5-methyl-1H-pyrazole-3-carboxylate (17) and ethyl 1-(4-methoxyphenyl)-3-methyl-1H-pyrazole-5carboxylate (18) were obtained in a ratio of 1.3:1 and separated by column chromatography. The ethyl esters (17 and 18) were reduced to the alcohols 19 and 20 and then converted to the chlorides using thionyl chloride (21 and 22) followed by reaction with the potassium salt of imidazole or triazole, as previously described, to give the required final imidazole pyrazole derivatives ( 23 and 25) and triazole pyrazole derivatives (24 and 26) (Scheme 2).

The Series 3 biarylpyrazole imidazole ( $10 \mathrm{c}-\mathrm{j}$ ) compounds were obtained via the five-step synthetic route described in Scheme 1 and were prepared to explore the effect of different aryl and biaryl groups on binding and their 'fit' within the active site, and also to evaluate the correlation between MIC and lipophilicity previously observed. Of note, although the yields obtained for the imines ( $6 \mathrm{c}-\mathrm{j}$ ) varied between $84-99 \%$, they were generally found to be highly sensitive to light and moisture, especially those with low melting points, so were used immediately in the next step of the reaction scheme. Also, in the chlorination step of the indole derivative $8 \mathbf{j}$ a C-2 chlorination of the indole ring occurred in addition to the conversion of the alcohol to chloride.


Scheme 2. Reagents and Conditions: (i) $\mathrm{EtOH}, 80^{\circ} \mathrm{C}$, overnight (ii) $\mathrm{LiAlH}_{4}$, THF , r.t. overnight (iii) $\mathrm{SOCl}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, r.t. overnight (iv) imidazole, $\mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{CH}_{3} \mathrm{CN}$, $45^{\circ} \mathrm{C}, 1 \mathrm{~h}$ then $70^{\circ} \mathrm{C}$, overnight (v) triazole, $\mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{CH}_{3} \mathrm{CN}, 45^{\circ} \mathrm{C}, 1 \mathrm{~h}$ then $70^{\circ} \mathrm{C}$, overnight.

### 2.2. CYP121A1 Ligand Binding Affinity

The CYP121A1 binding affinity $\left(K_{D}\right)$ of the various compounds was determined by UV-vis optical titration. For CYP121A1, the typical low-spin ligand free spectrum shows five characteristic peaks; a protein peak from the aromatic amino acid residues at 280 nm , a Soret peak at 416.5 nm , the $\beta$ band at 538 nm , the $\alpha$ band at 565 nm , and small charge transfer band at $\sim 650 \mathrm{~nm}$ owing to a residual amount of high-spin heme iron. A heme iron low-spin (LS) to high-spin (HS) type I shift occurs as cYY binds to CYP121A1, with the Soret band shifting from 416.5 to 393 nm (the cYY $K_{\mathrm{D}}$ is $5.82 \pm 0.16 \mu \mathrm{M}^{[13]}$ ). Other features of note in the spectral titration are apparent isosbestic points at approximately 420 nm and 492 nm , small red shifts in the alpha and beta band maxima at $\sim 572$ and 547 nm , and loss of the small high-spin signal at $\sim 650 \mathrm{~nm}$. Coordination binding of azoles to the CYP121A1 heme iron results in type II (red) spectral shifts of the Soret peak to wavelengths typically between $\sim 418-424 \mathrm{~nm}$ (according to the specific azole and whether it binds directly to the heme iron or via a retained water ligand to the heme iron), accompanied by spectral changes in both the $\alpha$ and $\beta$ bands (e.g. 10f, Figure 4). ${ }^{[19]}$

In Series 1, replacing the phenyl ring of the lead imidazole compound (1) or triazole compound (2) with a pyridyl ring


Figure 4. Compound 10 f binding to CYP121A1 using UV-Vis optical binding titration The left-hand panel shows spectra for compound 10 f titrated with CYP121A1 (ligand-free spectrum is a thick black line, spectra following progressive additions of compound 10 f are thin solid lines, and the final near-saturated protein spectrum is a thick red line). The inset shows overlapped difference spectra generated by subtracting the starting spectrum from each consecutive ligand-bound spectrum. The right-hand panel shows a plot of compound 10 induced-absorbance change calculated as the difference between the peak and trough in the difference spectra in the left-hand panel, using the same wavelength pair ( 429 and 392 nm , respectively). Data was fitted using the Hill equation giving a $K_{\mathrm{D}}$ value of $0.22 \pm 0.01 \mu \mathrm{M}$ for 10 f .
resulted in decreased binding affinity compared with the lead phenyl imidazole (1) and triazole (2) compounds (1 $K_{D}=2.6 \pm$ $0.2 \mu \mathrm{M}$ c.f. $10 \mathrm{a} K_{\mathrm{D}}=60.5 \pm 10.5 \mu \mathrm{M}$ and $10 \mathrm{~b} K_{\mathrm{D}}=53.6 \pm 6.8 \mu \mathrm{M}$; $2 K_{\mathrm{D}}=73.5 \pm 0.7 \mu \mathrm{M}$ c.f. 11 a $K_{\mathrm{D}}=115.3 \pm 27.0 \mu \mathrm{M}$ and $11 \mathrm{~b} K_{\mathrm{D}}=$ $249.7 \pm 51.6 \mu \mathrm{M}$ ) (Table 1).

For the extended pyridyl compounds (13 and 14), low binding affinity was generally observed, with the exception of 14b which displayed moderate binding affinity ( $K_{D}=39.0 \pm$ $8.2 \mu \mathrm{M}$ ). The series 2 imidazole pyrazole regioisomers 23 and 25 displayed moderate binding affinity ( $K_{D}=34.5 \pm 4.8 \mu \mathrm{M}$ and $22.9 \pm 0.2 \mu \mathrm{M}$, respectively). However, the triazole pyrazole

| Compound | $K_{\text {D }}(\mu \mathrm{M})$ | Soret peak <br> shift (nm) | MIC ( $\mu \mathrm{g} / \mathrm{mL}$ ) |
| :---: | :---: | :---: | :---: |
| 10a | $60.5 \pm 10.5$ | 416.5 to 419 | 12.5 |
| 10 b | $53.6 \pm 6.8$ | 416.5 to 421 | 12.5 |
| 10 c | $18.3 \pm 3.4$ | 416.5 to 420 | 1.562 |
| 10d | $53.7 \pm 5.9$ | 416.5 to 420 | 0.781 |
| 10e | $22.5 \pm 3.5$ | 416.5 to 423 | 1.562 |
| 10 f | $0.22 \pm 0.01$ | 416.5 to 423.5 | 1.562 |
| 10 g | $4.81 \pm 0.31$ | 416.5 to 419.5 | 1.562 |
| 10h | $2.91 \pm 0.14$ | 416.5 to 423.5 | 3.125 |
| 10i | $10.1 \pm 0.3$ | 416.5 to 422.5 | 3.125 |
| 10j | $3.31 \pm 0.05$ | 416.5 to 423.5 | 25 |
| 11a | $115.3 \pm 27.0$ | 416.5 to 417.5 | $>100$ |
| 11 b | $249.7 \pm 51.6$ | 416.5 to 417.5 | $>100$ |
| 13a | $83.2 \pm 3.9$ | 416.5 to 423.5 | 100 |
| 13b | $85.6 \pm 12.4$ | 416.5 to 417.5 | $>100$ |
| 14a | $218.6 \pm 13.6$ | 416.5 to 419.5 | $>100$ |
| 14b | $39.0 \pm 8.2$ | 416.5 to 417.5 | $>100$ |
| 23 | $34.5 \pm 4.8$ | 416.5 to 418 | $>100$ |
| 24 | $105 \pm 7$ | 416.5 to 418 | $>100$ |
| 25 | $22.9 \pm 0.2$ | 416.5 to 418.5 | 25 |
| 26 | $98.3 \pm 7.4$ | 416.5 to 416 | 100 |
| 1 | $2.6 \pm 0.2$ | 416.5 to 424 | 12.5 |
| 2 | $73.5 \pm 0.7$ | 416.5 to 422.5 | 50 |
| Fluconazole | $8.6 \pm 0.2$ |  | > 100 |
| Clotrimazole | $0.07 \pm 0.01$ |  | 20 |
| cYY | $5.82 \pm 0.16$ |  | - |

regioisomers 24 and 26 displayed weak binding affinity ( $K_{\mathrm{D}}=$ $105 \pm 7 \mu \mathrm{M}$ and $98.3 \pm 7.4 \mu \mathrm{M}$, respectively) (Table 1).

Series 3 biaryl derivatives with branched butyl substitutions ( 10 f and 10 g ), the biphenyl substitution ( 10 h ) and the indole substitution ( $\mathbf{1 0} \mathbf{j}$ ) gave the highest affinity for CYP121A1, having $K_{D}$ values $<5 \mu \mathrm{M}\left(K_{D}=0.22 \pm 0.01 \mu \mathrm{M}, 4.81 \pm 0.31 \mu \mathrm{M}\right.$, $2.91 \pm 0.14 \mu \mathrm{M}$ and $3.31 \pm 0.05 \mu \mathrm{M}$, respectively) (Table 1). The aryl derivatives with shorter alkyl substitutions, 4-ethyl ( 10 c ), 4propyl (10 d), 4-isopropyl (10 e) and benzo [d][1,3] dioxole (10i) had moderate binding affinity $\left(K_{\mathrm{D}}=18.3 \pm 3.4 \mu \mathrm{M}, 53.7 \pm 5.9 \mu \mathrm{M}\right.$, $22.5 \pm 3.5 \mu \mathrm{M}$, and $10.1 \pm 0.3 \mu \mathrm{M}$, respectively). The values were compared with the $K_{D}$ values of fluconazole $(8.6 \pm 0.2 \mu \mathrm{M})$ and clotrimazole $(0.07 \pm 0.01 \mu \mathrm{M}) .{ }^{[19]}$

Compounds $10 \mathrm{e}, 10 \mathrm{f}, 10 \mathrm{~h}, 10 \mathrm{i}, 10 \mathrm{j}$ and 13 a induce the most extensive Soret absorbance shifts, with the Soret maxima shifting by 6.5 nm (in the case of 10 e ) and by 7 nm (in the case of $10 \mathrm{f}, 10 \mathrm{~h}, 10 \mathrm{j}$ and 13 a ). The shifts observed are comparable with those for azole antifungal drugs (clotrimazole, econazole, fluconazole, ketoconazole and miconazole), which bind tightly to Mtb CYP121A1, inducing a Soret peak shift to between 421$423.5 \mathrm{~nm} .{ }^{[20]}$ Less extensive Soret red shifts were observed for imidazole pyrazole regioisomers (23 and 25), the triazole pyrazole regioisomers (24 and 26), the 4-ethylbenzene (10 c), 4propylbenzene ( 10 d ) and 4-tert-butylbenzene ( 10 g ) and the Series 1 3-pyridyl and 4-pyridyl (10a and 10b) imidazole derivatives on binding to the CYP121A1 heme iron.

### 2.3. MIC Determination Against Mycobacterium Tuberculosis

The derivatives were screened against M. tuberculosis H37Rv by the REMA (Resazurin Microtiter Assay) method. ${ }^{[21]}$ In Series 1 only the imidazole pyridyl derivatives displayed antimicrobial activity ( 10 a and $10 \mathrm{~b}, \mathrm{MIC}=12.5 \mu \mathrm{~g} / \mathrm{mL}$ ), retaining the inhibitory activity observed for the phenyl imidazole lead compound (1).

In Series 2 only the 5 -((1H-imidazol-1-yl)methyl)-1-(4-me-thoxyphenyl)-3-methyl-1H-pyrazole (23) displayed antimycobacterial activity with an MIC of $25 \mu \mathrm{~g} / \mathrm{mL}$ (Table 1). All the compounds in Series 3 displayed antimycobacterial activity, the most promising inhibitory activity ( $\mathrm{MIC}=0.781-1.562 \mu \mathrm{~g} / \mathrm{mL}$ ) was observed for the alkyl substituted aryl derivatives ( $10 \mathrm{c}-$ $\mathbf{1 0 g}$ ). Good inhibitory activity was also obtained for biphenyl ( $10 \mathrm{~h}, \mathrm{MIC}=3.125 \mu \mathrm{~g} / \mathrm{mL}$ ) and benzo[d][1,3]dioxole ( $10 \mathrm{i}, \mathrm{MIC}=$ $3.125 \mu \mathrm{~g} / \mathrm{mL}$ ) derivatives, with the least active in Series 3 being the 2 -chloro- 1 H -indole derivative ( $\mathbf{1 0} \mathbf{j}, \mathrm{MIC}=25 \mu \mathrm{~g} / \mathrm{mL}$ ).

Hansch analysis (Figure 5) showed a clear correlation between the lipophilicity of the compounds and MIC, as previously observed for the pyrazole lead compounds. ${ }^{[13]}$ This increased lipophilicity may facilitate increased drug uptake across the Mtb lipid-rich cell wall resulting in enhanced antimycobacterial activity. The more lipophilic compounds $10 \mathrm{c}-10 \mathrm{~g}$ (cLogP 4.19-4.99) showed the best antimycobacterial activity (MIC $=0.781-1.562 \mu \mathrm{~g} / \mathrm{mL}$ ), while the more polar pyridyl ( 10 a and 10 b ) (cLogP 1.95) and 2 -chloro- 1 H -indole ( 10 j ) (cLogP 3.33) compounds had lower antimycobacterial activity (MIC= $12.5-25 \mu \mathrm{~g} / \mathrm{mL})$.


Figure 5. Hansch analysis illustrating correlation between MIC and calculated $\log \mathrm{P}^{23}$ values for the Series 1 and 3 imidazole compounds (10).

### 2.4. Molecular Modelling

The Molecular Operating Environment (MOE) program ${ }^{[22]}$ was used to perform molecular docking and was found to closely replicate the position and binding interactions of cYY and fluconazole, as observed in the crystal structures PDB 3G5H and PDB 2IJ7, respectively. The 4- and 3-pyridyl derivatives (10/11 a and $10 / 11 \mathrm{~b}$ ) both interacted with the heme iron through an interstitial water molecule bonded with Ser237, and the 4pyridyl (10a) also formed a direct hydrogen bond with Arg386 (e.g. 10a, Figure 6). Both pyridyl imidazoles (10) and triazoles (11) formed additional binding interactions through the pyridine nitrogen, for the 3-pyridyl via a water molecule with Met62 and for the 4-pyridyl via a water molecule with Met62, Met86 and Val83 (Figure 6).

The imidazole derivatives (13) with the 5-atom linker between the imidazole and the pyrazole ring were found to form more interactions with the CYP121A1 active site compared with the 4 -pyridyl derivatives (14) with the 3-atom linker between the 4 -pyridyl and pyrazole ring. However, the


Figure 6. Series 1 pyridyl compound 10a in CYP121A1 active site. The imidazole ring interacts with the heme iron through an interstitial water molecule bonded with Ser237 and a direct hydrogen bond with Arg386. Additional binding interactions through the pyridine nitrogen are observed via a water molecule with Met62, Met86 and Val83.
extended pyridine compounds (13 and 14) were found to form numerous docking conformations owing to the increased flexibility of the extended linker between the heme-bonding group, imidazole for 13 and 4 -pyridyl for 14 . Series 2 imidazole (23 and 25) and triazole (24 and 26) derivatives, with the smaller methyl replacing a phenyl ring, were found to interact with the heme via an interstitial water molecule bound to Ser237, and for imidazole 23 a direct hydrogen bond interaction with Arg386 was formed. Additional hydrophobic interactions were observed with hydrophobic amino acid residues including Thr77, Val78, Val82, Val83 and Met86.

The ethyl ( 10 c ) and propyl ( 10 d ) imidazoles were found to bind in the same manner through interactions with the heme iron via an interstitial water molecule linked to Ser237 and with the phenyl ring interacting with Thr77 via a water molecule. The isopropyl (10e) and isobutyl (10f) imidazoles interacted with the heme in the same manner as described for 10 c and 10 d , but also formed an additional interaction between the pyrazole ring either with Thr77 via a water molecule in the case of 10 e, or with Gln385 and Arg386 via a water molecule for 10 f . The tert-butyl imidazole ( $\mathbf{1 0} \mathbf{g}$ ) and the biphenyl imidazole (10h) both formed a direct hydrogen bond with Arg386 and interacted with the heme through an interstitial water molecule bonded with Ser237. The phenyl ring of the biphenyl imidazole (10h) also formed an interaction with Thr77 via a water molecule. The positioning of the additional binding interactions of the branched alkyl and biphenyl imidazoles (10e-10h) resulted in a better filling of the binding pocket (Figure 7),


Figure 7. Overlapped images of Series 3 ethyl ( $\mathbf{1 0 c}$ ) and isobutyl ( $\mathbf{1 0 f}$ ) derivatives. The ethyl imidazole ( $\mathbf{1 0} \mathbf{c}$, cyan) interacts through the imidazole with the heme indirectly via an interstitial water molecule bonded to Ser237, and through the phenyl ring via an interstitial water molecule bonded to Thr77. The isobutyl imidazole ( $\mathbf{1 0 f}$, magenta) interacts with the heme in a similar manner as observed for 10 c , but forms an additional interaction through the pyrazole ring and occupies the binding site more fully.
especially in the case of 10 f and 10 g , and therefore these compounds would be predicted to be more efficient at blocking binding of the cYY natural substrate.

The heterobiaryl benzo[d][1,3]dioxole (10i) and 2-chloro-1Hindole ( 10 j ) derivatives showed binding of the imidazole with the heme via an interstitial water molecule through Ser237, and
additional H-bond interactions via Thr77, Arg386 and Gln385. The more extended pyrazole derivatives (10f, $10 \mathrm{~g}, 10 \mathrm{~h}, 10 \mathrm{i}$ and 10 j ) either formed additional binding interactions and/or more completely blocked the active site by occupying a similar binding site as the natural substrate cYY and fluconazole, resulting in improved binding affinity.

### 2.5. X-Ray Crystallographic Studies

Two compounds were successful co-crystallized with CYP121A1, 10j and 14a (PDB 6GEO and 6GEQ respectively). Both crystals contained two molecules of sulfate resulting from the crystallography process, one of which is positioned above the heme (Figure 8).

The indole derivative $\mathbf{1 0 j}$ forms an arene-H interaction between the imidazole ring and Gln385, and between the indole benzene ring and Asn85. The indole NH and Cl act as H donors with two water molecules and the imidazole N acts as a H -acceptor with a water molecule. The pyrrole aryl ring is positioned to form a $\pi-\pi$ interaction with Phe168 (Figure 8 A ).


Figure 8. X-ray crystal structures of (A) 10j (PDB 6GEO) and (B) 14 a (PDB 6GEQ) binding to CYP121A1. The compounds are shown in cyan and the haem in orange. Water molecules are illustrated with small red spheres and amino acids are presented either in light grey or, for amino acids that form direct or indirect H -bonding interactions with the inhibitor compound, in magenta. The sulfate molecule from the crystallography process is seen above the heme and forms H -bond interactions with Arg386 and indirectly with Ser237.

The extended 4-pyridyl derivative 14 a makes a H -bond interaction between the carbonyl oxygen and Gln385, and interacts indirectly with Thr77. Both 4-pyridyl groups form indirect H-bond interactions: the amide pyridyl with Thr65 and Arg72, and the pyrrole pyridyl with Val83 and Asn65. The benzene ring forms an arene-H interaction with Trp182 (Figure 8 B ).

Alignment of the crystal structure of 10 j co-crystallized with CYP121A1 (PDB 6GEO) with the crystal structures of CYP121A1 co-crystallized with fluconazole (PDB 2IJ7) and cYY (PDB 3G5H) showed a comparable position of all three structures in the enzyme (Figure 9). All three compounds are positioned to form


Figure 9. Alignment of the crystal structures of CYP121A1 co-crystallized with 10j (cyan) (PDB 6GEO), fluconazole (magenta) (PDB 21J7) and cYY (yellow) (PDB 3G5H) with comparable positioning observed.
a $\pi-\pi$ interaction with Phe168 and interact either directly or indirectly with amino acids Met62, Val83, Asn85. In particular, Hbonding interactions with Gly385 (10j and fluconazole) or Arg386 (cYY) would appear to effectively block access to the active site.

## 3. Conclusions

For optimal binding interactions, inhibitors that effectively fill the CYP121A1 binding site and exhibit strong binding affinity are those that should most effectively inhibit CYP121A1 substrate (cYY) binding. This is observed for the branched alkyl derivatives 10 f and $\mathbf{1 0} \mathrm{g}$, and for the biphenyl 10 h . Compound flexibility also has a notable effect, with the extended pyridine derivatives 13 and 14 displaying very weak binding affinity. Lipophilicity is a major contributor to antimycobacterial activity, as shown by Hansch analysis (Figure 5). The challenge in designing a compound with both good binding affinity and antimycobacterial activity is to combine the most favourable properties for each component. The physicochemical properties of the prepared compounds and reference compounds (cYY, fluconazole and clotrimazole) were calculated, the cLogP was determined using Crippen's fragmentation ${ }^{[23]}$ and the number of H -bond acceptors (nON), H -bond donors (nOHNH) rotatable bonds (nrot), along with the molecular volume (MV) and

| Table 2. Physiocochemical properties of Series 1,2 and 3. |  |  |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- |
| Compound | cLogP | nON/ | nrot | MV $\left(\AA^{3}\right)$ | TPSA $\left(\AA^{2}\right)$ |
|  |  | $n O H N H$ |  |  |  |

[a] Optimal antimycobacterial activity (MIC). [b] Optimal binding affinity $\left(K_{\mathrm{D}}\right) . \mathrm{nON}=\mathrm{H}$-bond acceptor; $\mathrm{nOHNH}=\mathrm{H}$-bond donor; nrot = number of rotatable bonds; $\mathrm{MV}=$ molecular volume; $\mathrm{TPSA}=$ topological polar surface area.
topological polar surface area (TPSA) were calculated using Molinspiration software. ${ }^{[24]}$ Those compounds with cLogP $>4$ and TPSA $<40 \AA^{2}$ had optimal antimycobacterial activity (shaded green, Table 2), while the compounds with 5-6 rotatable bonds, 4-5 H-bond acceptors, no H-bond donors and molecular volume $>340 \AA^{3}$ had optimal binding affinity (shaded red, Table 2).

Therefore, the optimal combined physicochemical properties for antimycobacterial activity and binding affinity (shaded green and red, Table 2) identified Series 3 isobutyl (10f) and tert-butyl derivatives $(10 \mathrm{~g})$ as fulfilling these requirements. These and other high affinity compounds generated in this study will undergo further testing for their ability to kill pathogenic strains of $M$. tuberculosis.

## Experimental Section

## Chemistry

All reagents and solvents were of general purpose or analytical grade and purchased from Sigma-Aldrich Ltd, Fisher Scientific, Fluka and Acros. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were recorded with a Bruker Avance DPX500 spectrometer operating at 500 and 125 MHz , with $\mathrm{Me}_{4} \mathrm{Si}$ as internal standard. Elemental analysis was performed by MEDAC Ltd (Chobham, Surrey, UK). High resolution mass spectra (HRMS) were determined at the EPSRC National Mass Spectrometry Facility at Swansea University and Medac Ltd (Chobham, Surrey, UK), using ESI (Electrospray Ionisation) in positive and negative modes, and a TOF (Time-of-Flight) analyser. Flash column chromatography was performed with silica gel 60 ( $230-400$ mesh) (Merck) and TLC was carried out on precoated silica plates (kiesel gel 60 $\mathrm{F}_{254}, \mathrm{BDH}$ ). Compounds were visualised by illumination under UV light ( 254 nm ) or by the use of potassium permanganate stain followed by heating. Melting points were determined on an
electrothermal instrument and are uncorrected. All solvents were dried prior to use and stored over $4 \AA$ molecular sieves, under nitrogen. All the compounds were $\geq 95 \%$ pure.
The following compounds were prepared as previously described: hydrazines $\mathbf{6 a}$ and $\mathbf{6 b},{ }^{[14]} \mathbf{6 h}{ }^{[23]}$ and $6 \mathbf{j}_{;}^{[25]}$ aldehydes $\mathbf{7 a}$ and $\mathbf{7 b},{ }^{[14]}$ $7 h^{[26]}$ and $7 j_{j}^{[25]}$ alcohol $8 h_{;}^{[26]}$ chloride $9 h ;{ }^{[26]}$ carboxylic acids (12a) and (12b); ${ }^{[14]}$ ethyl 1-(4-methoxyphenyl)-5-methyl-1H-pyrazole-3carboxylate (17) and ethyl 1-(4-methoxyphenyl)-3-methyl-1H-pyrazole-5-carboxylate (18). ${ }^{[18]}$ All compounds were more than $95 \%$ pure.

## General Method for the Preparation of Hydrazine Derivatives

 (6)To a solution of the acetyl reagent (4) ( 5 mmol ) in ethanol ( 20 mL ), was added acetic acid ( 3.3 mmol ) with continuous stirring at room temperature. After 5 min , phenylhydrazine (5) ( 5.5 mmol ) was added dropwise with further stirring for 15 min , then refluxed at $80^{\circ} \mathrm{C}$ for 1 h . The reaction was left to cool, concentrated under reduced pressure, wrapped with aluminium foil and left in the freezer overnight. The precipitate formed was filtered, washed with cold ethanol, dried in vacuo and used immediately in subsequent reactions without further purification.

## 1-(1-(4-Ethylphenyl)ethylidene)-2-phenylhydrazine (6c)

Prepared from 4-ethylacetophenone ( $2 \mathrm{~mL}, 13.39 \mathrm{mmol}$ ). Product obtained as a yellow solid, yield 3.03 g ( $95 \%$ ). M.p. $82-84^{\circ} \mathrm{C}$. TLC ( $1: 1$ petroleum ether/EtOAc), $\mathrm{R} f=0.94$. ${ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 9.19$ (s, 1H, NH), $7.70(\mathrm{~d}, \mathrm{~J}=8.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.23(\mathrm{~m}, 6 \mathrm{H}, \mathrm{Ar}), 6.75(\mathrm{~m}, 1 \mathrm{H}$, para-Ar), 2.61 (q, J=7.6 Hz, 2H, CH 2 ), $2.24\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.19(\mathrm{t}, J=$ $\left.7.6 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO-d $\mathrm{d}_{6}$ ): $\delta 146.7(\mathrm{C}=\mathrm{N}), 143.6(\mathrm{C}, \mathrm{Ar})$, 141.2 (C, Ar), 137.3 (C, Ar), $129.3(2 \times \mathrm{CH}, \mathrm{Ar}), 128.1(2 \times \mathrm{CH}, \mathrm{Ar}), 125.7$ $(2 \times \mathrm{CH}, \mathrm{Ar}), 119.2\left(\mathrm{CH}\right.$, para-Ar), $113.2(2 \times \mathrm{CH}, \mathrm{Ar}), 28.3\left(\mathrm{CH}_{2}\right), 16.0$ $\left(\mathrm{CH}_{3}\right), 13.3\left(\mathrm{CH}_{3}\right)$.

## (1-(1-(4-Propylphenyl)ethylidene)-2-phenylhydrazine (6d)

Prepared from 4-propylacetophenone ( $1 \mathrm{~mL}, 6.04 \mathrm{mmol}$ ). Product obtained as an orange crystalline solid, yield 1.43 g (92\%). M.p. 68$70^{\circ} \mathrm{C}$. TLC (2:1 petroleum ether/EtOAc), $\mathrm{R} f=0.90$. ${ }^{1} \mathrm{H}$ NMR (DMSO$\mathrm{d}_{6}$ ): $\delta 9.19$ (s, 1H, NH), 7.70 (d, J=7.9 Hz, 2H, Ar), 7.23 (m, 6H, Ar), $7.09\left(\mathrm{t}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}\right.$, para-Ar), $2.56\left(\mathrm{t}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.24(\mathrm{~s}$, $\left.3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.60\left(\mathrm{sext}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 0.90\left(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$. ${ }^{13} \mathrm{C}$ NMR (DMSO- $\mathrm{d}_{6}$ ): $\delta 146.7$ (C=N), 142.0 (C, Ar), 141.1 (C, Ar), 137.3 (C, Ar), $128.7(2 \times \mathrm{CH}, \mathrm{Ar}), 127.7(2 \times \mathrm{CH}, \mathrm{Ar}), 125.6(2 \times \mathrm{CH}, \mathrm{Ar}), 119.2$ $\left(1 \times \mathrm{CH}\right.$, para-Ar), $113.2(2 \times \mathrm{CH}, \mathrm{Ar}), 37.4\left(\mathrm{CH}_{2}\right), 24.8\left(\mathrm{CH}_{2}\right), 14.6\left(\mathrm{CH}_{3}\right)$, $13.3\left(\mathrm{CH}_{3}\right)$.

## 1-(1-(4-Isopropylphenyl)ethylidene)-2-phenylhydrazine (6e)

Prepared from 4-isopropylacetophenone ( $2 \mathrm{~mL}, 11.95 \mathrm{mmol}$ ). Product obtained as an orange crystalline solid, yield $2.5 \mathrm{~g}(80 \%)$. M.p. $50-52^{\circ} \mathrm{C}$. TLC ( $1: 1$ petroleum ether/EtOAc), $\mathrm{Rf}=0.93$. ${ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 9.19(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}), 7.70(\mathrm{~d}, \mathrm{~J}=7.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.25$ (d, $\mathrm{J}=$ $7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.21(\mathrm{~m}, 4 \mathrm{H}, \mathrm{Ar}), 6.75(\mathrm{t}, J=6.7 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), 2.90 (m, 1H, CH), $2.24\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.22\left(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 6 \mathrm{H}, 2 \times \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO- $d_{6}$ ): $\delta 148.3(\mathrm{C}=\mathrm{N}), 146.7$ (C, Ar), 141.1 (C, Ar), 137.5 (C, Ar), $129.3(2 \times \mathrm{CH}, \mathrm{Ar}), 126.6(2 \times \mathrm{CH}, \mathrm{Ar}), 125.7(2 \times \mathrm{CH}, \mathrm{Ar}), 119.1$ $\left(\mathrm{CH}\right.$, para-Ar), $113.2(2 \times \mathrm{CH}, \mathrm{Ar}), 33.6(\mathrm{CH}), 24.3\left(2 \times \mathrm{CH}_{3}\right), 13.3\left(\mathrm{CH}_{3}\right)$.

## 1-(1-(4-Isobutylphenyl)ethylidene)-2-phenylhydrazine (6f)

Prepared from 4-isobutylacetophenone ( $5 \mathrm{~mL}, 27.00 \mathrm{mmol}$ ). Product obtained as an orange solid, yield 6.99 g ( $97 \%$ ). M.p. $58-60^{\circ} \mathrm{C}$. TLC ( $3: 1$ petroleum ether/EtOAc), $\mathrm{R} f=0.71 .{ }^{1} \mathrm{H}$ NMR (DMSO- $\mathrm{d}_{6}$ ): $\delta 9.18$ (s, 1H, NH), 7.69 (d, J=7.7 Hz, 2H, Ar), 7.22 (m, 4H, Ar), 7.16 (d, $J=$ $7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 6.75(\mathrm{t}, J=7.3 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $2.44(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}$, $\left.\mathrm{CH}_{2}\right), 2.23\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.83(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}), 0.86(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 6 \mathrm{H}, 2 \times$ $\mathrm{CH}_{3}$ ). ${ }^{13} \mathrm{C}$ NMR (DMSO-d $)$ : $\delta 146.6(\mathrm{C}=\mathrm{N}), 141.2(\mathrm{C}, \mathrm{Ar}), 141.1$ (C, Ar), 137.3 (C, Ar), $129.4(2 \times \mathrm{CH}, \mathrm{Ar}), 129.3(2 \times \mathrm{CH}, \mathrm{Ar}), 125.4(2 \times \mathrm{CH}, \mathrm{Ar})$, $119.2\left(\mathrm{CH}\right.$, para-Ar), $113.2(2 \times \mathrm{CH}, \mathrm{Ar}), 44.8\left(\mathrm{CH}_{2}\right), 30.0\left(\mathrm{CH}_{3}\right), 22.6$ $(\mathrm{CH}), 13.4\left(2 \times \mathrm{CH}_{3}\right)$.

## 1-(1-(4-(Tert-butyl)phenyl)ethylidene)-2-phenylhydrazine (6 g)

Prepared from 4-(tert-butyl)acetophenone ( $3 \mathrm{~mL}, 16.4 \mathrm{mmol}$ ). Product obtained as an orange crystalline solid, yield 4.2 g ( $97 \%$ ). M.p. $66-68^{\circ} \mathrm{C}$. TLC ( $1: 1$ petroleum ether/EtOAc), $\mathrm{R} f=0.92$. ${ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 9.25(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}), 7.75$ (d, $\left.J=8.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}\right), 7.40$ (d, $J=$ $8.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.31(\mathrm{~m}, 4 \mathrm{H}, \mathrm{Ar}), 7.13(\mathrm{t}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), 2.27 $\left(\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.31\left(\mathrm{~s}, 9 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO-d $): \delta 150.4(\mathrm{C}=\mathrm{N})$, 146.7 (C, Ar), 141.1 (C, Ar), 137.1 (C, Ar), $129.3(2 \times \mathrm{CH}, \mathrm{Ar}), 129.0(2 \times$ $\mathrm{CH}, \mathrm{Ar}), 125.4(2 \times \mathrm{CH}, \mathrm{Ar}), 119.2(\mathrm{CH}$, para-Ar), $113.3(2 \times \mathrm{CH}, \mathrm{Ar})$, $34.7\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 31.5\left(3 \times \mathrm{CH}_{3}\right), 13.3\left(\mathrm{CH}_{3}\right)$.

## 1-(1-(Benzo[d][1,3]dioxol-5-yl)ethylidene)-2-phenylhydrazine (6i)

Prepared from 3',4'-(methylenedioxy)acetophenone ( 3.00 g , $18.27 \mathrm{mmol})$. Product obtained as a yellow solid, yield 3.90 g ( $84 \%$ ). M.p. $66-68{ }^{\circ} \mathrm{C}$. TLC ( $1: 1$ petroleum ether/EtOAc), R $f=0.88 .{ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 9.14(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}), 7.40(\mathrm{~s}, 1 \mathrm{H}$, benzo[d][1,3]dioxole), 7.23 (d, $J=1.6 \mathrm{~Hz}, 1 \mathrm{H}$, benzo[d][1,3]dioxole), $7.21(\mathrm{~m}, 4 \mathrm{H}, \mathrm{Ar}), 6.92(\mathrm{~d}, J=$ $8.6 \mathrm{~Hz}, 1 \mathrm{H}$, benzo[d][1,3]dioxole), $6.74(\mathrm{t}, J=4.1 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), 6.04 (s, 2H, benzo[d][1,3]dioxole), 2.21 (s, 3H, CH3). ${ }^{13} \mathrm{C}$ NMR (DMSO$\left.\mathrm{d}_{6}\right): \delta 148.0(\mathrm{C}=\mathrm{N}), 147.5$ (C, benzo[d][1,3]dioxole), 146.7 (C, benzo[d] [1,3]dioxole), 140.9 (C, Ar), 134.3 (C, Ar), 134.3 (C, Ar), 129.3 ( $2 \times \mathrm{CH}$, $\mathrm{Ar}), 119.7(\mathrm{CH}$, para-Ar), $119.1(\mathrm{CH}$, benzo[d][1,3]dioxole), $113.2(\mathrm{CH}$, benzo[d][1,3]dioxole), $108.3(2 \times \mathrm{CH}, \mathrm{Ar}), 105.6$ (C, benzo[d][1,3] dioxole), $101.5\left(\mathrm{CH}_{2}\right.$, benzo[d][1,3]dioxole), $13.5\left(\mathrm{CH}_{3}\right)$.

## General Method for the Preparation of Aldehyde Derivatives (7)

Phosphorus (V) oxychloride ( 15 mmol ) was added dropwise to an ice-cooled solution of hydrazine (6) ( 5 mmol ) in dry DMF ( 50 mL ). The reaction mixture was allowed to reach room temperature, and then heated at $60^{\circ} \mathrm{C}$ for 4 h with continuous stirring. The resulting mixture was poured onto crushed ice ( 50 mL ), brought to pH 8.0 using $10 \%$ aqueous $\mathrm{NaOH}(\approx 50 \mathrm{~mL})$, and left overnight in the freezer. The resulting precipitate was extracted with EtOAc ( 100 mL ) and the organic layer was washed with brine ( $5 \times 50 \mathrm{~mL}$ ), while the aqueous phase was re-extracted with EtOAc $(3 \times 50 \mathrm{~mL})$. The combined organic fractions were washed with brine $(3 \times 50 \mathrm{~mL})$, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated in vacuo to obtain the crude product, which was purified by gradient column chromatography.

## 3-(4-Ethylphenyl)-1-phenyl-1H-pyrazole-4-carbaldehyde (7c)

Prepared from 1-(1-(4-ethylphenyl)ethylidene)-2-phenylhydrazine $(6 \mathrm{c})(2.5 \mathrm{~g}, 10.48 \mathrm{mmol})$. The product was eluted with petroleum ether - EtOAc 85:15 v/v to give the product as a buff solid, yield 1.76 g ( $61 \%$ ). M.p. $112-114^{\circ} \mathrm{C}$. TLC ( $1: 1$ petroleum ether/EtOAc), $\mathrm{Rf}=0.88 .{ }^{1} \mathrm{H}$ NMR (DMSO-d $\mathrm{d}_{6}$ ): $\delta 9.99(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CHO}), 9.31(\mathrm{~s}, 1 \mathrm{H}$, pyrazole), 8.00 (d, J=7.6 Hz, 2H, Ar), 7.86 (d, J= $=8.1 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}$ ), 7.58
(t, J=7.5 Hz, 2H, Ar), 7.43 (t, J=7.5 Hz, 1H, para-Ar), 7.36 (d, J= $8.3 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 2.69\left(\mathrm{q}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 1.24(\mathrm{t}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}$, $\left.\mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 185.0$ (CHO), 153.4 (C, pyrazole), 145.4 (C, Ar), 139.1 (C, Ar), 135.1 (CH, pyrazole), 130.1 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 129.1 $(2 \times \mathrm{CH}, \mathrm{Ar}), 128.5(2 \times \mathrm{CH}, \mathrm{Ar}), 128.3(\mathrm{CH}$, para-Ar), $122.7(\mathrm{C}$, pyrazole), $122.6(\mathrm{C}, \mathrm{Ar}), 119.8(2 \times \mathrm{CH}, \mathrm{Ar}), 28.4\left(\mathrm{CH}_{2}\right), 16.0\left(\mathrm{CH}_{3}\right)$. Anal. Calcd for $\mathrm{C}_{18} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}$ (276.3372): C, 78.24; H, 5.84; N, 10.13. Found: C, 78.11; H, 5.84; N, 10.13.

## 1-Phenyl-3-(4-propylphenyl)-1H-pyrazole-4-carbaldehyde (7 d)

Prepared from 1-(1-(4-propylphenyl)ethylidene)-2-phenylhydrazine ( 6 d) $(1.20 \mathrm{~g}, 4.75 \mathrm{mmol})$. The product was eluted with petroleum ether - EtOAc $85: 15 \mathrm{v} / \mathrm{v}$ to give the product as a white solid, yield $0.91 \mathrm{~g}(66 \%)$. M.p. $86-88^{\circ} \mathrm{C}$. TLC ( $3: 1$ petroleum ether/EtOAc), $\mathrm{Rf}=$ $0.68 .{ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }^{2}$ ): $\delta 9.99$ ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{CHO}$ ), 9.31 ( $\mathrm{s}, 1 \mathrm{H}$, pyrazole), 8.00 ( d, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.85$ (d, $J=7.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.58(\mathrm{t}, J=$ $7.5 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.43(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $7.34(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}$, Ar), 2.63 (t, J=7.4 Hz, 2H, CH 2 ), 1.65 (sext, J=7.3 Hz, 2H, CH $)_{2}$ ), 0.93 ( $\mathrm{t}, \mathrm{J}=7.4 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}$ ). ${ }^{13} \mathrm{C}$ NMR (DMSO-d $\mathrm{d}_{6}$ ): $\delta 185.0$ (CHO), 153.2 ( C , pyrazole), 143.9 (C, Ar), 139.1 (C, Ar), 135.2 (CH, pyrazole), 130.2 ( $2 \times$ CH, Ar), 130.1 (C, Ar), 129.2 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 129.1 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 128.2 (CH, para-Ar), 122.6 (C, pyrazole), $119.7(2 \times \mathrm{CH}, \mathrm{Ar}), 37.5\left(\mathrm{CH}_{2}\right), 24.5\left(\mathrm{CH}_{2}\right)$, $14.1\left(\mathrm{CH}_{3}\right)$. Anal. Calcd for $\mathrm{C}_{19} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}$ (290.364): C, $78.59 ; \mathrm{H}, 6.25 ; \mathrm{N}$, 9.64. Found: C, 78.37; H, 6.36; N, 9.68.

## 3-(4-(Isopropylphenyl)-1-phenyl-1H-pyrazole-4-carbaldehyde (7e)

Prepared from 1-(1-(4-isopropylphenyl)ethylidene)-2-phenylhydrazine ( 6 e) $(2.30 \mathrm{~g}, 9.24 \mathrm{mmol})$. The product was eluted with petroleum ether - EtOAc $85: 15 \mathrm{v} / \mathrm{v}$ to give the product as a white solid, yield $2.30 \mathrm{~g}(78 \%)$. M.p. $114-116^{\circ} \mathrm{C}$. TLC ( $1: 1$ petroleum ether/EtOAc), Rf = 0.83. ${ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }^{2}$ ): $\delta 9.99(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CHO}), 9.31$ (s, 1H, pyrazole), 8.01 ( $d, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}$ ), 7.86 (d, $J=8.1 \mathrm{~Hz}, 2 \mathrm{H}$, Ar), 7.58 ( $\mathrm{t}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}$ ), $7.43(\mathrm{t}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), 7.39 (d, $J=8.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}$ ), 2.97 (sept, $J=6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CH}$ ), 1.26 (d, $J=$ $6.9 \mathrm{~Hz}, 6 \mathrm{H}, 2 \times \mathrm{CH}_{3}$ ). ${ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 185.2$ (CHO), 153.2 (C, pyrazole), 150.1 (C, Ar), 139.1 (C, Ar), 135.2 (CH, pyrazole), 130.2 ( $2 \times$ CH, Ar), 129.3 (C, Ar), 129.2 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 128.2 (CH, para-Ar), 127.0 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 122.6 ( 1 C, pyrazole), $119.7(2 \times \mathrm{CH}, \mathrm{Ar}), 33.8(\mathrm{CH}), 24.3$ $\left(2 \times \mathrm{CH}_{3}\right)$. Anal. Calcd for $\mathrm{C}_{19} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}(290.364)$ : C, $78.59 ; \mathrm{H}, 6.25 ; \mathrm{N}$, 9.64. Found: C, 78.42; H, 6.15; N, 9.58.

## 3-(4-Isobutylphenyl)-1-phenyl-1H-pyrazole-4-carbaldehyde (7f)

Prepared from 1-(1-(4-isobutylphenyl)ethylidene)-2-phenylhydrazine ( 6 f ) $(6.00 \mathrm{~g}, 22.52 \mathrm{mmol})$. The product was eluted with petroleum ether - EtOAc $4: 1 \mathrm{v} / \mathrm{v}$ to give the product as a yellow oil, yield $4.82 \mathrm{~g}(70 \%)$. TLC ( $3: 1$ petroleum ether/EtOAc), $\mathrm{R} f=0.38 .{ }^{1} \mathrm{H}$ NMR (DMSO-d $\mathrm{d}_{6}$ ) $\delta 9.98$ ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{CHO}$ ), 9.27 ( $\mathrm{s}, 1 \mathrm{H}$, pyrazole), 7.97 ( d , $J=7.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.84$ (d, J=7.7 Hz, 2H, Ar), 7.57 (t, J=7.7 Hz, 2H, Ar), 7.42 (t, J=7.6 Hz, 1H, para-Ar), 7.28 (d, J=7.8 Hz, $2 \mathrm{H}, \mathrm{Ar}$ ), 2.51 (d, J=6.8 Hz, 2H, CH 2 ) $, 1.88(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}), 0.88(\mathrm{~d}, J=6.7 \mathrm{~Hz}, 6 \mathrm{H}, 2 \times$ $\mathrm{CH}_{3}$ ). ${ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 185.0$ (CHO), 153.1 (C, Ar), 142.8 (C, pyrazole), 139.1 (C, Ar), 135.4 (CH, pyrazole), 130.2 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 129.6 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 129.2 (C, Ar), 128.9 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 128.2 (CH, para-Ar), 122.6 (C, pyrazole), $119.7(2 \times \mathrm{CH}, \mathrm{Ar}), 44.8\left(\mathrm{CH}_{2}\right), 30.1(\mathrm{CH}), 22.6(2 \times$ $\mathrm{CH}_{3}$ ). [ESI-HRMS] calculated for $\mathrm{C}_{20} \mathrm{H}_{21} \mathrm{~N}_{2} \mathrm{O}: 305.1648[\mathrm{M}+\mathrm{H}]^{+}$. Found: $305.1651[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(4-(Tert-butyl)phenyl)-1-phenyl-1H-pyrazole-4-carbaldehyde

 (7 g)Prepared from 1-(1-(4-(tert-butyl)phenyl)ethylidene)-2-phenylhydrazine $(6 \mathrm{~g})(3.50 \mathrm{~g}, 13.13 \mathrm{mmol})$. The product was eluted with petroleum ether - EtOAc $4: 1 \mathrm{v} / \mathrm{v}$ to give the product as a yellow solid, yield $3.24 \mathrm{~g}(81 \%)$. M.p. $96-98^{\circ} \mathrm{C}$. TLC ( $3: 1$ petroleum ether/ EtOAc), Rf $=0.18 .{ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }^{2}$ ): $\delta 10.01$ ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{CHO}$ ), 9.30 ( s , 1 H, pyrazole), 8.00 (d, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.87$ (d, $J=8.3 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}$ ), $7.56(\mathrm{t}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.51(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.35(\mathrm{t}, \mathrm{J}=$ $7.6 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), 1.31 (s, $\left.9 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO-d $\mathrm{d}_{6}$ ): ठ 185.1 (CHO), 153.2 (C, Ar), 152.2 (C, pyrazole), 139.1 (C, Ar), 135.1 (CH, pyrazole), 130.2 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 129.0 (C, pyrazole), $128.9(2 \times \mathrm{CH}$, Ar), $128.1(\mathrm{CH}$, para-Ar), $125.8(2 \times \mathrm{CH}, \mathrm{Ar}), 122.6(\mathrm{C}$, pyrazole $), 119.7$ $(2 \times \mathrm{CH}, \mathrm{Ar}), 34.9\left(\underline{\mathrm{C}}\left(\mathrm{CH}_{3}\right)_{3}\right), 31.5\left(3 \times \mathrm{CH}_{3}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{20} \mathrm{H}_{21} \mathrm{~N}_{2} \mathrm{O}: 305.1648[\mathrm{M}+\mathrm{H}]^{+}$. Found: $305.1652[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(Benzo[d][1,3] <br> dioxol-5-yl)-1-phenyl-1H-pyrazole-4-carbaldehyde (7i)

Prepared from1-(1-(benzo[d][1,3]dioxol-5-yl)ethylidene)-2-phenylhydrazine ( 6 i ) $(1.62 \mathrm{~g}, 6.37 \mathrm{mmol})$. The product was eluted with petroleum ether - EtOAc $7: 1 \mathrm{v} / \mathrm{v}$ to give the product as a yellow solid, yield 1.50 g ( $81 \%$ ). M.p. $167-169^{\circ} \mathrm{C}$. TLC ( $1: 1$ petroleum ether/EtOAc), $\mathrm{Rf}=0.94 .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{DMSO}-\mathrm{d}_{6}$ ): $\delta 9.96$ ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{CHO}$ ), 9.31 (s, 1H, pyrazole), 7.99 (d, J=7.7 Hz, 2H, Ar), 7.57 (t, J=7.6 Hz, 2 H , Ar), 7.53 ( $\mathrm{s}, 1 \mathrm{H}$, benzo[d][1,3]dioxole), 7.51 ( $\mathrm{d}, J=1.6 \mathrm{~Hz}, 1 \mathrm{H}$, benzo [d][1,3]dioxole), $7.43(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $7.05(\mathrm{~d}, J=8.6 \mathrm{~Hz}$, 1 H , benzo[d][1,3]dioxole), 6.11 ( $\mathrm{s}, 2 \mathrm{H}$, benzo[d][1,3]dioxole). ${ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 185.0$ (CHO), 152.7 (C, pyrazole), 148.6 (C, benzo[d] [1,3]dioxole), 147.9 (C, benzo[d][1,3] dioxole), 139.0 (C, Ar), 135.8 (CH, pyrazole), $130.2(2 \times \mathrm{CH}, \mathrm{Ar}), 128.1$ (CH, para-Ar), 125.6 (C, benzo[d][1,3]dioxole), 123.4 ( CH , benzo[d][1,3]dioxole), 122.4 (C, pyrazole), $119.7(2 \times \mathrm{CH}, \mathrm{Ar}), 109.2(\mathrm{CH}$, benzo[d][1,3]dioxole), 108.9 (CH, benzo[d][1,3]dioxole), $101.8\left(\mathrm{CH}_{2}\right.$, benzo[d][1,3]dioxole). [ESIHRMS] calculated for $\mathrm{C}_{17} \mathrm{H}_{13} \mathrm{~N}_{2} \mathrm{O}_{3}$ : $293.0924[\mathrm{M}+\mathrm{H}]^{+}$. Found: $293.0921[\mathrm{M}+\mathrm{H}]^{+}$.

## General Method for the Preparation of Alcohol Derivatives (8)

To an ice-cooled solution of carbaldehyde (7) ( 1 mmol ) in EtOH $(10 \mathrm{~mL})$ was added $\mathrm{NaBH}_{4}(1 \mathrm{mmol})$ in portions, then the reaction was then stirred at room temperature for 1 h . The solvent was evaporated and $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$ was added slowly and the reaction stirred for 30 min . The reaction mixture was extracted with EtOAc $(2 \times 25 \mathrm{~mL})$, then the combined organic layers washed with $\mathrm{H}_{2} \mathrm{O}(3 \times$ $25 \mathrm{~mL})$, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give the crude alcohol (17), which was was further purified by gradient column chromatography.

## (1-Phenyl-3-(pyridin-4-yl)-1H-pyrazol-4-yl)methanol (8a)

Prepared from 1-phenyl-3-(pyridin-4-yl)-1H-pyrazole-4-carbaldehyde $(7 \mathrm{a})^{[14]}(0.40 \mathrm{~g}, 1.6 \mathrm{mmol})$. The product was obtained as a light yellow solid pure enough for use in the following reaction. Yield $0.33 \mathrm{~g}(82 \%)$. M.p. $120-122^{\circ} \mathrm{C}$. TLC ( $1: 4$ petroleum ether/EtOAc), $R f=0.22 .{ }^{1} H$ NMR (DMSO- $\mathrm{d}_{6}$ ): $\delta 8.66$ ( $\mathrm{d}, \mathrm{J}=6.0 \mathrm{~Hz}, 2 \mathrm{H}$, pyridine), 8.58 (s, 1H, pyrazole), 7.90 (d, $J=7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.85$ (d, $J=6.0 \mathrm{~Hz}, 2 \mathrm{H}$, pyridine), $7.54(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.36(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $5.34(\mathrm{t}, J=5.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{OH} \mathrm{ex}), 4.63\left(\mathrm{~d}, J=5.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO-d $)_{6}$ : $\delta 150.5(2 \times \mathrm{CH}$, pyridine), 148.2 (C, Ar), 140.5 (C, Ar), 139.8 (C, Ar), 130.1 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 129.8 (CH, Ar), 127.1 (CH, Ar), 123.5 (C, Ar), $122.0(2 \times \mathrm{CH}, \mathrm{Ar}), 118.9(2 \times \mathrm{CH}, \mathrm{Ar}), 54.3\left(\mathrm{CH}_{2}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{3} \mathrm{O}: 252.1137[\mathrm{M}+\mathrm{H}]^{+}$. Found: $252.1136[\mathrm{M}+$ $\mathrm{H}]^{+}$.

## (1-Phenyl-3-(pyridin-3-yl)-1H-pyrazol-4-yl)methanol (8b)

Prepared from 1-phenyl-3-(pyridin-3-yl)-1H-pyrazole-4-carbaldehyde $(7 \mathrm{~b})^{[14]}(1.00 \mathrm{~g}, 4.03 \mathrm{mmol})$. The product was obtained as a light yellow solid pure enough for use in the following reaction. Yield $0.85 \mathrm{~g}(85 \%)$. M.p. $111-113^{\circ} \mathrm{C}$. TLC ( $1: 4$ petroleum ether/EtOAc), $\mathrm{Rf}=0.23 .{ }^{1} \mathrm{H}$ NMR (DMSO- $\mathrm{d}_{6}$ ): $\delta 9.09$ ( $\mathrm{s}, 1 \mathrm{H}$, pyridine), 8.60 ( $\mathrm{d}, \mathrm{J}=$ $6.4 \mathrm{~Hz}, 1 \mathrm{H}$, pyridine $), 8.58(\mathrm{~s}, 1 \mathrm{H}$, pyrazole $), 8.28(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}$, pyridine), 7.92 (d, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.53(\mathrm{~m}, 3 \mathrm{H}, 1 \times$ pyridine, $2 \times \mathrm{Ar}$ ), $7.34(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $5.27(\mathrm{t}, J=5.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{OH} \mathrm{ex}), 4.58(\mathrm{~d}$, $J=5.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}$ ). ${ }^{13} \mathrm{C}$ NMR (DMSO-d $\mathrm{d}_{6}$ ): $\delta 149.0$ (CH, pyridine), 148.2 (CH, pyridine), 174.6 (C, Ar), 139.3 (C, pyrazole), 135.1 (CH, pyridine), $128.9(2 \times \mathrm{CH}, \mathrm{Ar}), 128.2$ (C, pyrazole), 128.0 (CH, pyridine), $126.5(2 \times \mathrm{CH}, \mathrm{Ar}), 124.0(\mathrm{CH}$, para-Ar), $119.0(\mathrm{CH}$, pyrazole), 116.7 (C, pyridine), $54.3\left(\mathrm{CH}_{2}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{~N}_{3} \mathrm{O}: 252.1137$ $[\mathrm{M}+\mathrm{H}]^{+}$. Found: $252.1134[\mathrm{M}+\mathrm{H}]^{+}$.

## (3-(4-Ethylphenyl)-1-phenyl-1H-pyrazol-4-yl)methanol (8c)

Prepared from 3-(4-ethylphenyl)-1-phenyl-1H-pyrazole-4-carbaldehyde ( 7 c ) $(1.68 \mathrm{~g}, 6.07 \mathrm{mmol})$. The product was eluted with petroleum ether - EtOAc $7: 3 \mathrm{v} / \mathrm{v}$ to give the product as a white solid, yield $0.86 \mathrm{~g}(51 \%)$. M.p. $54-55^{\circ} \mathrm{C}$. TLC ( $1: 1$ petroleum ether/ $\mathrm{EtOAc}), \mathrm{Rf}=0.71 .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{DMSO}_{6}$ ): $\delta 8.48$ ( $\mathrm{s}, 1 \mathrm{H}$, pyrazole), 7.89 (d, $J=7.5 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.81(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.51(\mathrm{t}, J=7.5 \mathrm{~Hz}$, 1 H, para-Ar), $7.30(\mathrm{~m}, 4 \mathrm{H}, \mathrm{Ar}), 5.22(\mathrm{t}, J=5.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{OH}$ ex), 4.56 (d, $\left.J=4.9 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.65\left(\mathrm{q}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 1.22(\mathrm{t}, J=7.5 \mathrm{~Hz}$, $3 \mathrm{H}, \mathrm{CH}_{3}$ ). ${ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 150.9$ (C, pyrazole), 144.0 (C, Ar), 140.0 (C, Ar), 130.9 (C, Ar), $130.0(2 \times \mathrm{CH}, \mathrm{Ar}), 129.1$ (CH, para-Ar), $128.4(2 \times \mathrm{CH}, \mathrm{Ar}), 127.8(2 \times \mathrm{CH}, \mathrm{Ar}), 126.5(\mathrm{CH}$, pyrazole $), 125.8(\mathrm{C}$, pyrazole), $118.5(2 \times \mathrm{CH}, \mathrm{Ar}), 54.6\left(\mathrm{CH}_{2}\right), 28.4\left(\mathrm{CH}_{2}\right), 16.0\left(\mathrm{CH}_{3}\right)$. [ESIHRMS] calculated for $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{~N}_{2} \mathrm{O}: 279.1497[\mathrm{M}+\mathrm{H}]^{+}$. Found: $279.1503[\mathrm{M}+\mathrm{H}]^{+}$.

## (1-Phenyl-3-(4-propylphenyl)-1H-pyrazol-4-yl)methanol (8d)

Prepared from 1-phenyl-3-(4-propylphenyl)-1H-pyrazole-4-carbaldehyde $(7 \mathrm{~d}) \quad(0.80 \mathrm{~g}, 2.75 \mathrm{mmol})$. The product was eluted with petroleum ether - EtOAc 7:3 v/v to give the product as a yellow solid, yield 0.74 g ( $92 \%$ ). M.p. $86-88^{\circ} \mathrm{C}$. TLC ( $2: 1$ petroleum ether/ EtOAc), $\mathrm{Rf}=0.48 .{ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 8.49$ ( $\mathrm{s}, 1 \mathrm{H}$, pyrazole), 7.89 (d, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.80(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.51(\mathrm{t}, J=7.1 \mathrm{~Hz}$, 1 H, para-Ar), 7.31 (d, $J=7.3 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.29$ (d, $J=8.1 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}$ ), $5.18(\mathrm{t}, J=5.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{OH} \mathrm{ex}), 4.55\left(\mathrm{~d}, J=4.9 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.61(\mathrm{t}, J=$ $7.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}$ ), 1.64 (sext, $\left.J=7.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 0.93$ (t, J=7.3 Hz, $3 \mathrm{H}, \mathrm{CH}_{3}$ ). ${ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 150.8$ (C, pyrazole), 142.3 (C, Ar), 140.0 (C, Ar), 130.9 (C, Ar), $130.0(2 \times \mathrm{CH}, \mathrm{Ar}), 129.1(2 \times \mathrm{CH}, \mathrm{Ar}), 129.0$ (CH, para-Ar), $127.7(2 \times \mathrm{CH}, \mathrm{Ar}), 126.5(\mathrm{CH}$, pyrazole), 122.3 (C, pyrazole), $118.5(2 \times \mathrm{CH}, \mathrm{Ar}), 54.6\left(\mathrm{CH}_{2}\right), 37.5\left(\mathrm{CH}_{2}\right), 24.5\left(\mathrm{CH}_{2}\right), 14.1$ $\left(\mathrm{CH}_{3}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{19} \mathrm{H}_{21} \mathrm{~N}_{2} \mathrm{O}: 293.1654[\mathrm{M}+\mathrm{H}]^{+}$. Found: $293.1654[\mathrm{M}+\mathrm{H}]^{+}$.

## (1-Phenyl-3-(4-isopropylphenyl)-1H-pyrazol-4-yl)methanol (8e)

Prepared from 1-phenyl-3-(4-isopropylphenyl)-1H-pyrazole-4-carbaldehyde ( 7 e ) $(0.90 \mathrm{~g}, 3.09 \mathrm{mmol})$. The product was eluted with petroleum ether - EtOAc 7:3 v/v to give the product as a colourless oil, yield 0.76 g ( $84 \%$ ). TLC ( $1: 1$ petroleum ether/EtOAc), $\mathrm{Rf}=0.67$. ${ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 8.49$ (s, 1H, pyrazole), 7.90 (d, J=7.6 Hz, 2H, Ar), 7.82 (d, $J=8.1 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.51(\mathrm{t}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.35(\mathrm{~d}, J=$ $8.1 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.31(\mathrm{~m}, 1 \mathrm{H}$, para-Ar), $5.18(\mathrm{t}, J=4.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{OH} \mathrm{ex})$, 4.57 (d, $J=4.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}$ ), 2.94 (sept, $J=6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CH}$ ), 1.25 (d, $\left.J=6.9 \mathrm{~Hz}, 6 \mathrm{H}, 2 \times \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO-d $\mathrm{d}_{6}$ ): $\delta 150.8$ (C, pyrazole), 148.5 (C, Ar), 140.0 (C, Ar), 131.1 (C, Ar), 130.0 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 129.1 (CH,
para-Ar), $127.8(2 \times \mathrm{CH}, \mathrm{Ar}), 126.9(2 \times \mathrm{CH}, \mathrm{Ar}), 126.4$ (CH, pyrazole), 122.4 (C, pyrazole), $118.5(2 \times \mathrm{CH}, \mathrm{Ar}), 54.6\left(\mathrm{CH}_{2}\right), 33.7(\mathrm{CH}), 24.3(2 \times$ $\mathrm{CH}_{3}$ ). Anal. Calcd for $\mathrm{C}_{19} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}$ (292.3798): C, 78.92; H, 6.62; N, 9.20. Found: C, 78.63; H, 6.68; N, 9.13.

## (3-(4-Isobutylphenyl)-1-phenyl-1H-pyrazol-4-yl)methanol (8f)

Prepared from 1-phenyl-3-(4-isobutylphenyl)-1H-pyrazole-4-carbaldehyde ( $\mathbf{7}$ f) ( $4.00 \mathrm{~g}, 13.13 \mathrm{mmol}$ ). The product was obtained as a greenish brown oil pure enough for use in the following reaction. Yield 3.67 g ( $91 \%$ ). TLC ( $3: 1$ petroleum ether/EtOAc), $\mathrm{Rf}=0.18 .{ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }^{2}$ ): $\delta 8.32$ ( $\mathrm{s}, 1 \mathrm{H}$, pyrazole), 7.82 (d, $J=7.9 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}$ ), $7.64(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.55(\mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.43(\mathrm{t}, J=$ $7.7 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), 7.26 (d, $J=7.9 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 5.16(\mathrm{t}, J=5.0 \mathrm{~Hz}, 1 \mathrm{H}$, OH ex), $4.56\left(\mathrm{~d}, J=4.9 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.49\left(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 1.90$ ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{CH}$ ), $0.89\left(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 6 \mathrm{H}, 2 \times \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO- $\mathrm{d}_{6}$ ): $\delta 150.7$ (C, pyrazole), 141.5 (C, Ar), 139.9 (C, Ar), 130.9 (C, Ar), 130.0 $(2 \times \mathrm{CH}, \mathrm{Ar}), 129.6(2 \times \mathrm{CH}, \mathrm{Ar}), 129.0(\mathrm{CH}$, para-Ar), $127.6(2 \times \mathrm{CH}, \mathrm{Ar})$, $126.5(\mathrm{CH}$, pyrazole $), 122.3$ ( C, pyrazole), $118.5(2 \times \mathrm{CH}, \mathrm{Ar}), 54.6$ $\left(\mathrm{CH}_{2}\right), 44.8\left(\mathrm{CH}_{2}\right), 30.1(\mathrm{CH}), 22.6\left(2 \times \mathrm{CH}_{3}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{20} \mathrm{H}_{23} \mathrm{~N}_{2} \mathrm{O}: 307.1809[\mathrm{M}+\mathrm{H}]^{+}$. Found: $307.1805[\mathrm{M}+\mathrm{H}]^{+}$.

## (1-Phenyl-3-(4-tert-butylphenyl)-1H-pyrazol-4-yl)methanol ( 8 g )

Prepared from 1-phenyl-3-(4-(tert-butylphenyl)-1H-pyrazole-4-carbaldehyde ( 7 g ) ( $2.3 \mathrm{~g}, 7.55 \mathrm{mmol}$ ). The product was eluted with petroleum ether - EtOAc 3:1 v/v to give the product as a yellow solid, yield 2.0 g ( $87 \%$ ). M.p. $92-94^{\circ} \mathrm{C}$. TLC ( $1: 1$ petroleum ether/ EtOAc), $\mathrm{R} f=0.73 .{ }^{1} \mathrm{H}$ NMR (DMSO-d $\mathrm{d}_{6}$ ): $\delta 8.49$ ( $\mathrm{s}, 1 \mathrm{H}$, pyrazole), 7.89 (d, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.82$ (d, $J=8.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.50(\mathrm{~m}, 4 \mathrm{H}, \mathrm{Ar})$, $7.31(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $5.18(\mathrm{t}, J=4.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{OH}$ ex), $4.57(\mathrm{~d}$, $\left.J=4.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 1.33\left(\mathrm{~s}, 9 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO-d $\left.\mathrm{d}_{6}\right): \delta 150.8$ (C, pyrazole), 140.04 ( $2 \times \mathrm{C}, \mathrm{Ar}$ ), 130.7 (C, Ar), $130.0(2 \times \mathrm{CH}, \mathrm{Ar}), 129.1$ $(2 \times \mathrm{CH}, \mathrm{Ar}), 127.6(\mathrm{CH}$, para-Ar), $126.5(2 \times \mathrm{CH}, \mathrm{Ar}), 125.8(\mathrm{CH}$, pyrazole), 122.4 (C, pyrazole), $118.5(2 \times \mathrm{CH}, \mathrm{Ar}), 54.6\left(\mathrm{CH}_{2}\right), 34.8$ ( $\underline{C}$ $\left.\left(\mathrm{CH}_{3}\right)_{3}\right)$, $31.6\left(3 \times \mathrm{CH}_{3}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{20} \mathrm{H}_{23} \mathrm{~N}_{2} \mathrm{O}: 307.1805$ $[\mathrm{M}+\mathrm{H}]^{+}$. Found: $307.1797[\mathrm{M}+\mathrm{H}]^{+}$.

## (3-(Benzo[d][1,3]dioxol-5-yl)-1-phenyl-1H-pyrazol-4-yl) methanol (8i)

Prepared from 3-(benzo[d][1,3]dioxol-5-yl)-1-phenyl-1H-pyrazole-4carbaldehyde ( 7 i ) $(1.40 \mathrm{~g}, 4.78 \mathrm{mmol})$. The product was eluted with petroleum ether - EtOAc $65: 35 \mathrm{v} / \mathrm{v}$ to give the product as a yellow solid, yield $1.06 \mathrm{~g}(75 \%)$. M.p. $96-98^{\circ} \mathrm{C}$. TLC ( $1: 1$ petroleum ether/ EtOAc), $\mathrm{R} f=0.38 .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{DMSO}-\mathrm{d}_{6}$ ): $\delta 8.48$ ( $\mathrm{s}, 1 \mathrm{H}$, pyrazole), 7.88 (d, J=7.7 Hz, 2H, Ar), $7.51(\mathrm{t}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.47(\mathrm{~s}, 1 \mathrm{H}$, benzo[d] [1,3]dioxole), 7.41 (d, $J=1.6 \mathrm{~Hz}, 1 \mathrm{H}$, benzo[d][1,3]dioxole), $7.30(\mathrm{t}$, $J=7.2 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $7.02(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}$, benzo[d][1,3]dioxole), 6.08 (s, 2H, benzo[d][1,3]dioxole), 5.21 (br s, 1H, OH ex), 4.53 (d, 2H, $J=4.0 \mathrm{~Hz}, \mathrm{CH}_{2}$ ). ${ }^{13} \mathrm{C}$ NMR (DMSO- $\mathrm{d}_{6}$ ): $\delta 150.7$ (C, pyrazole), 148.0 (C, benzo[d][1,3]dioxole), 147.5 (C, benzo[d][1,3]dioxole), 140.0 (C, Ar), $130.0(2 \times \mathrm{CH}, \mathrm{Ar}), 129.2(\mathrm{CH}$, para-Ar), 127.5 (C, benzo[d][1,3] dioxole), 126.5 (CH, pyrazole), 121.9 (C, pyrazole), 121.7 (CH, benzo [d][1,3]dioxole), $118.8(2 \times \mathrm{CH}, \mathrm{Ar}), 108.9(\mathrm{CH}$, benzo[d][1,3]dioxole), $108.1\left(\mathrm{CH}\right.$, benzo[d][1,3]dioxole), $101.6\left(\mathrm{CH}_{2}\right.$, benzo[d][1,3]dioxole), $54.5\left(\mathrm{CH}_{2}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{~N}_{2} \mathrm{O}_{3}: 295.1077[\mathrm{M}+\mathrm{H}]^{+}$. Found: $295.1077[\mathrm{M}+\mathrm{H}]^{+}$.

## (3-(1H-Indol-3-yl)-1-phenyl-1H-pyrazol-4-yl)methanol (8j)

Prepared from 3-(1H-indol-3-yl)-1-phenyl-1H-pyrazole-4-carbaldehyde $(7 \mathbf{j})^{[25]}(8.00 \mathrm{~g}, 27.84 \mathrm{mmol})$. The product was eluted with
petroleum ether - EtOAc $45: 55 \mathrm{v} / \mathrm{v}$ to give the product as a brown waxy solid, yield 5.42 g ( $67 \%$ ). TLC ( $1: 2$ petroleum ether/EtOAc), $\mathrm{Rf}=0.47 .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{DMSO}_{6}$ ): $\delta 11.38$ ( NH , indole), 8.48 ( $\mathrm{s}, 1 \mathrm{H}$, indole), 8.43 (d, $J=7.2 \mathrm{~Hz}, 1 \mathrm{H}$, indole), 7.96 (d, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}$ ), $7.90(\mathrm{~s}, 1 \mathrm{H}$, pyrazole), $7.54(\mathrm{t}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.49(\mathrm{~d}, J=7.3 \mathrm{~Hz}$, 1 H , indole), $7.28(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $7.19(\mathrm{~m}, 2 \mathrm{H}$, indole), 5.19 ( $\mathrm{t}, J=5.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{OH}$ ex), $4.64\left(\mathrm{~d}, J=4.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right.$ ). ${ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 148.1$ (C, Ar), 140.3 (C, indole), 136.8 (C, pyrazole), $130.0(2 \times \mathrm{CH}, \mathrm{Ar}), 127.6(\mathrm{CH}$, indole), $126.0(\mathrm{C}$, indole), $125.8(\mathrm{CH}$, para-Ar), 122.2 (CH, pyrazole), 122.1 (CH, indole), 121.8 (C, pyrazole), 120.1 (CH, indole), $117.9(2 \times \mathrm{CH}, \mathrm{Ar}), 112.0(2 \times \mathrm{CH}$, indole), 108.6 (C, indole), $55.0\left(\mathrm{CH}_{2}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{18} \mathrm{H}_{16} \mathrm{~N}_{3} \mathrm{O}: 290.1293$ $[\mathrm{M}+\mathrm{H}]^{+}$. Found: $290.1295[\mathrm{M}+\mathrm{H}]^{+}$.

## General Method for the Preparation of Alcohol Derivatives (19) and (20)

To an ice-cooled solution of ethyl carboxylate (17 or 18$)^{[18]}$ ( 5 mmol ) in dry THF ( 15 mL ) was added $\mathrm{LiAlH}_{4}$ ( 1 M in THF, 7.5 mmol ) dropwise over 25 min . The reaction was then stirred at room temperature overnight then cooled in an ice-bath and carefully quenched with $\mathrm{H}_{2} \mathrm{O}$ until cessation of effervescence. The reaction mixture was extracted with EtOAc $(2 \times 50 \mathrm{~mL})$, then the combined organic layers washed with $\mathrm{H}_{2} \mathrm{O}(3 \times 50 \mathrm{~mL})$, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give the crude alcohol (5) or (6) which was was further purified by recrystallisation from $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ hexane 1:1 v/v.

## 1-(4-Methoxyphenyl)-5-methyl-1H-pyrazol-3-yl)methanol (19)

Prepared from ethyl 1-(4-methoxyphenyl)-5-methyl-1H-pyrazole-3carboxylate ${ }^{[18]}$ (17) $(1.90 \mathrm{~g}, 7.18 \mathrm{mmol})$. Product obtained as a brown solid, yield 1.02 g ( $64 \%$ ). M.p. $115-118^{\circ} \mathrm{C}$. TLC ( $1: 1$ hexane/ EtOAc), $\mathrm{Rf}=0.21 .{ }^{1} \mathrm{H}$ NMR (DMSO-d $\mathrm{d}_{6}$ ): $\delta 7.33$ (d, J=8.9 Hz, 2H, Ar), $6.87(\mathrm{~d}, J=8.9 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 6.06(\mathrm{~s}, 1 \mathrm{H}$, pyrazole), $4.99(\mathrm{t}, J=5.8 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{ex}), 4.39$ (d, J=5.7 Hz, 2H, CH 2 ), 3.78 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{OCH}_{3}$ ), $2.24(\mathrm{~s}, 3 \mathrm{H}$, $\mathrm{CH}_{3}$ ). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}-\mathrm{d}_{6}\right): \delta 158.0(\mathrm{C}, \mathrm{Ar}), 147.2$ (C, pyrazole), 143.6 (C, pyrazole), 133.8 (C, Ar), $123.5(2 \times \mathrm{CH}, \mathrm{Ar}), 114.3(2 \times \mathrm{CH}, \mathrm{Ar}), 106.3$ ( CH , pyrazole), $55.7\left(\mathrm{OCH}_{3}\right), 53.0\left(\mathrm{CH}_{2}\right), 12.1\left(\mathrm{CH}_{3}\right)$. Anal. Calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot 1 \mathrm{H}_{2} \mathrm{O}$ (220.0563): C, 65.49; H, 6.50; N, 12.73. Found: C, 65.54; H, 6.37; N, 12.83.

## 1-(4-Methoxyphenyl)-3-methyl-1H-pyrazol-5-yl)methanol (20)

Prepared from ethyl 1-(4-methoxyphenyl)-3-methyl-1H-pyrazole-5carboxylate ${ }^{[18]}$ (18) $(1.50 \mathrm{~g}, 5.67 \mathrm{mmol})$. Product obtained as a brown solid, yield 0.82 g ( $66 \%$ ). M.p. $103-105^{\circ} \mathrm{C}$. TLC ( $1: 1$ hexane/ $\mathrm{EtOAc}), \mathrm{Rf}=0.43 .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 7.39(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.05$ (d, J=8.9 Hz, 2H, Ar), 6.19 (s, 1H, pyrazole), 5.02 (t, J=5.8 Hz, 1H, ex), $4.41\left(\mathrm{~d}, \mathrm{~J}=5.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 3.82\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OCH}_{3}\right), 2.26\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$. ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{CDCl}_{3}\right): \delta 158.6(\mathrm{C}, \mathrm{Ar}), 147.8$ (C, pyrazole), 144.0 (C, pyrazole), $133.8(\mathrm{C}, \mathrm{Ar}), 125.5(2 \times \mathrm{CH}, \mathrm{Ar}), 114.6(2 \times \mathrm{CH}, \mathrm{Ar}), 107.3$ $(\mathrm{CH}$, pyrazole $), 55.9\left(\mathrm{OCH}_{3}\right), 54.6\left(\mathrm{CH}_{2}\right), 12.2\left(\mathrm{CH}_{3}\right)$. Anal. Calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{2}$ (218.2548): C, 66.04; H, 6.47; N, 12.83. Found: C, 65.80; H, 6.41; N, 12.68.

General Method for the Preparation of Chlorides (9), (21) and (22)

To an ice-cooled solution of alcohol (8 or 19 or 20) ( 1 mmol ) in dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ was added thionyl chloride ( 10 mmol ) dropwise over 25 min . The reaction was stirred at room temperature overnight and then cooled in an ice-bath and carefully quenched with saturated aqueous $\mathrm{NaHCO}_{3}$ in portions until slightly basic ( pH 8.0 ).

The organic layer was separated, washed with brine $(3 \times 10 \mathrm{~mL})$, $\mathrm{H}_{2} \mathrm{O}(2 \times 10 \mathrm{~mL})$, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give the crude chloride, which was was purified by petroleum ether - EtOAc gradient column chromatography.

## 4-(4-(Chloromethyl)-1-phenyl-1H-pyrazol-3-yl)Pyridine (9 a)

Prepared from (1-phenyl-3-(pyridine-4-yl)-1H-pyrazol-4-yl)methanol ( 8 a ) $(0.19 \mathrm{~g}, 0.75 \mathrm{mmol})$. The product was obtained as a brown solid that was used immediately in the next step without further purification owing to instability. Yield 0.19 g ( $75 \%$ ). M.p. $78-80^{\circ} \mathrm{C}$. TLC (1:4 petroleum ether/EtOAc), $\mathrm{R} f=0.56 .{ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 8.81$ (s, 1H, pyrazole), 8.72 (d, $J=7.3 \mathrm{~Hz}, 2 \mathrm{H}$, pyridine), 7.91 (d, $J=$ $7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.85(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}$, pyridine), $7.56(\mathrm{~d}, J=7.6 \mathrm{~Hz}$, $2 \mathrm{H}, \mathrm{Ar}), 7.39\left(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}\right.$, para-Ar), $5.00\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO-d $\mathrm{d}_{6}$ ): $\delta 150.5(2 \times \mathrm{CH}$, pyridine), 147.4 (C, Ar), 144.5 (C, Ar), 138.9 (C, Ar), $130.2(2 \times \mathrm{CH}, \mathrm{Ar}), 128.5(\mathrm{CH}, \mathrm{Ar}), 124.5(\mathrm{CH}, \mathrm{Ar}), 122.4$ ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), $119.6(2 \times \mathrm{CH}, \mathrm{Ar}), 116.2(\mathrm{C}, \mathrm{Ar}), 37.9\left(\mathrm{CH}_{2}\right)$.

## 3-(4-(Chloromethyl)-1-phenyl-1H-pyrazol-3-yl)Pyridine (9b)

Prepared from (1-phenyl-3-(pyridine-3-yl)-1H-pyrazol-4-yl)methanol ( 8 b) $(0.87 \mathrm{~g}, 3.49 \mathrm{mmol})$. The product was obtained as a pale orange solid that was used immediately in the next step without further purification owing to instability. Yield $0.8 \mathrm{~g}(85 \%)$. M.p. $52^{\circ} \mathrm{C}$ (decomp.). TLC ( $1: 4$ petroleum ether/EtOAc), $\mathrm{R} f=0.76$. ${ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 9.14$ ( $\mathrm{s}, 1 \mathrm{H}$, pyridine), 8.75 ( $\mathrm{d}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}$, pyridine), $8.67(\mathrm{~s}, 1 \mathrm{H}$, pyrazole), 8.52 (d, $J=8.1 \mathrm{~Hz}, 1 \mathrm{H}$, pyridine), 8.01 (d, $J=$ $7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.72(\mathrm{~m}, 3 \mathrm{H}, 1 \times$ pyridine, $2 \times \mathrm{Ar}), 7.40(\mathrm{t}, J=7.4 \mathrm{~Hz}$, 1 H , para-Ar), $4.98\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO-d $)_{6}$ : $\delta 149.1(\mathrm{CH}$, pyridine), 148.4 (CH, pyridine), 148.0 (C, Ar), 139.5 (C, pyrazole), 135.4 (CH, pyridine), $129.0(2 \times \mathrm{CH}, \mathrm{Ar}), 128.2$ (C, pyrazole), 128.2 (CH, pyridine), $126.9(2 \times \mathrm{CH}, \mathrm{Ar}), 124.1(\mathrm{CH}$, para-Ar), $119.2(\mathrm{CH}$, pyrazole), $116.8(\mathrm{C}$, pyridine $), 37.9\left(\mathrm{CH}_{2}\right)$.

## 4-(Chloromethyl)-3-(4-ethylphenyl)-1-phenyl-1H-pyrazole (9 c)

Prepared from (3-(4-ethylphenyl)-1-phenyl-1H-pyrazol-4-yl)methanol $(8 \mathrm{c}) \quad(0.80 \mathrm{~g}, 2.87 \mathrm{mmol})$. The product was eluted with petroleum ether - EtOAc 9:1 v/v to give the product as a yellow solid, yield $0.51 \mathrm{~g}(82 \%)$. M.p. $96-98^{\circ} \mathrm{C}$. TLC ( $2: 1$ petroleum ether/ EtOAc), $\mathrm{Rf}=0.94$. $^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 8.74$ (s, 1H, pyrazole), 7.89 (d, $J=7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.77(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.54(\mathrm{t}, J=7.8 \mathrm{~Hz}$, 1 H, para-Ar), 7.36 (d, $J=8.5 \mathrm{~Hz}, 4 \mathrm{H}, \mathrm{Ar}), 4.91\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}_{2}\right), 2.68(\mathrm{q}, \mathrm{J}=$ $\left.7.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 1.23\left(\mathrm{t}, \mathrm{J}=7.4 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO-d $\mathrm{d}_{6}$ ): $\delta 151.0$ (C, pyrazole), 144.5 (C, Ar), 139.6 (C, Ar), 130.7 (C, pyrazole), 130.1 (C, Ar), $130.1(2 \times \mathrm{CH}, \mathrm{Ar}), 128.7$ (CH, para-Ar, $), 127.8(2 \times \mathrm{CH}$, $\mathrm{Ar}), 127.1(2 \times \mathrm{CH}, \mathrm{Ar}), 118.8(\mathrm{CH}$, pyrazole $), 117.9(2 \times \mathrm{CH}, \mathrm{Ar}), 38.3$ $\left(\mathrm{CH}_{2}\right), 28.43\left(\mathrm{CH}_{2}\right), 16.0\left(\mathrm{CH}_{3}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{18} \mathrm{H}_{18} \mathrm{ClN}_{2}$ : $297.6373[\mathrm{M}+\mathrm{H}]^{+}$. Found: $297.6362[\mathrm{M}+\mathrm{H}]^{+}$.

## 4-(Chloromethyl)-1-phenyl-3-(4-propylphenyl)-1H-pyrazole (9d)

Prepared from (1-phenyl-3-(4-propylphenyl)-1H-pyrazol-4-yl)methanol $(8 \mathbf{d}) \quad(0.70 \mathrm{~g}, 2.39 \mathrm{mmol})$. The product was eluted with petroleum ether - EtOAc $85: 15 \mathrm{v} / \mathrm{v}$ to give the product as a brown solid, yield $0.33 \mathrm{~g}(82 \%)$. M.p. $82-84^{\circ} \mathrm{C}$. TLC ( $2: 1$ petroleum ether/ EtOAc), $\mathrm{Rf}=0.88 .{ }^{1} \mathrm{H}$ NMR (DMSO-d $\mathrm{d}_{6}$ : $\delta 8.09$ (s, 1H, pyrazole), 7.77 ( $\mathrm{m}, 4 \mathrm{H}, \mathrm{Ar}$ ), $7.49(\mathrm{t}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $7.33(\mathrm{~m}, 4 \mathrm{H}, \mathrm{Ar}), 4.76(\mathrm{~s}$, $\left.2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.67\left(\mathrm{t}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 1.72\left(\right.$ sext, $\left.J=7.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right)$, $1.00\left(\mathrm{t}, \quad J=7.3 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO-d $\left.\mathrm{d}_{6}\right): \delta 151.8(\mathrm{C}$, pyrazole), 143.1 (C, Ar), 139.8 (C, Ar), 129.8 (C, Ar), $129.5(2 \times \mathrm{CH}, \mathrm{Ar})$, $128.9(2 \times \mathrm{CH}, \mathrm{Ar}), 128.5(\mathrm{CH}$, para-Ar), $127.9(2 \times \mathrm{CH}, \mathrm{Ar}), 126.7(\mathrm{CH}$,
pyrazole), 119.1 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 117.7 (C, pyrazole), $37.9\left(\mathrm{CH}_{2}\right), 37.3$ $\left(\mathrm{CH}_{2}\right), 24.5\left(\mathrm{CH}_{2}\right), 13.8\left(\mathrm{CH}_{3}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{19} \mathrm{H}_{20} \mathrm{CIN}_{2}$ : $311.1312[\mathrm{M}+\mathrm{H}]^{+}$. Found: $311.1310[\mathrm{M}+\mathrm{H}]^{+}$.

4-(Chloromethyl)-3-(4-isopropylphenyl)-1-phenyl-1H-pyrazole (9e)
Prepared from (1-phenyl-3-(4-isopropylphenyl)-1H-pyrazol-4-yl) methanol (8e) ( $0.62 \mathrm{~g}, 2.11 \mathrm{mmol})$. The product was eluted with petroleum ether - EtOAc $85: 15 \mathrm{v} / \mathrm{v}$ to give the product as a brown solid, yield $0.53 \mathrm{~g}(80 \%)$. M.p. $114-116^{\circ} \mathrm{C}$. TLC ( $2: 1$ petroleum ether/EtOAc), $\mathrm{Rf}=0.85 .{ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 8.74$ (s, 1H, pyrazole), 7.89 ( $\mathrm{d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}$ ), 7.78 (d, $J=8.1 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.53(\mathrm{t}, J=$ $7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.39$ (d, J=8.1 Hz, 2H, Ar), 7.34 (t, J=7.3 Hz, 1H, para-Ar), 4.90 (s, 2H, CH2 $), 2.95$ (sept, $J=6.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CH}), 1.25(\mathrm{~d}, J=$ $6.9 \mathrm{~Hz}, 6 \mathrm{H}, 2 \times \mathrm{CH}_{3}$ ). ${ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 151.0$ (C, pyrazole), 149.1 (C, Ar), 139.7 (C, Ar), 130.7 (CH, para-Ar), $130.3(\mathrm{C}, \mathrm{Ar}), 130.1(2 \times \mathrm{CH}$, Ar), $127.9(2 \times \mathrm{CH}, \mathrm{Ar}), 127.2(2 \times \mathrm{CH}, \mathrm{Ar}), 127.1(\mathrm{CH}$, pyrazole $), 118.8$ $(2 \times \mathrm{CH}, \mathrm{Ar}), 117.9(\mathrm{C}$, pyrazole $), 38.3\left(\mathrm{CH}_{2}\right), 33.7(\mathrm{CH}), 24.3\left(2 \times \mathrm{CH}_{3}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{19} \mathrm{H}_{20} \mathrm{ClN}_{2}: 311.1312[\mathrm{M}+\mathrm{H}]^{+}$. Found: $311.1313[\mathrm{M}+\mathrm{H}]^{+}$.

## 4-(Chloromethyl)-3-(4-isobutylphenyl)-1-phenyl-1H-pyrazole (9f)

Prepared from (1-phenyl-3-(4-isobutylphenyl)-1H-pyrazol-4-yl)methanol ( 8 f ) ( $2.20 \mathrm{~g}, 7.17 \mathrm{mmol}$ ). The product was obtained as a greenish brown solid pure enough for use in the next reaction. Yield 1.90 g ( $82 \%$ ). M.p. $100-102^{\circ} \mathrm{C}$. TLC ( $3: 1$ petroleum ether/ EtOAc), $\mathrm{Rf}=0.62 .{ }^{1} \mathrm{H}$ NMR (DMSO-d $\mathrm{d}_{6}$ ) $\delta 8.46$ ( $\mathrm{s}, 1 \mathrm{H}$, pyrazole), 7.88 (d, J=7.9 Hz, 2H, Ar), 7.79 (d, J=7.7 Hz, 2H, Ar), 7.51 (t, J=7.5 Hz, $2 \mathrm{H}, \mathrm{Ar}), 7.30(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $7.25(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar})$, $4.95\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.49\left(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 1.87(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}), 0.89$ (d, J=6.5 Hz, 6H, $2 \times \mathrm{CH}_{3}$ ). ${ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 149.9$ (C, pyrazole), 142.2 (C, Ar), 139.9 (C, Ar), $131.5(2 \times \mathrm{CH}, \mathrm{Ar}), 129.9(\mathrm{C}, \mathrm{Ar}), 129.5(2 \times$ $\mathrm{CH}, \mathrm{Ar}), 126.5(\mathrm{CH}$, para-Ar), $124.8(2 \times \mathrm{CH}, \mathrm{Ar}), 123.0(\mathrm{CH}$, pyrazole), $120.3(2 \times \mathrm{CH}, \mathrm{Ar}), 119.2(\mathrm{C}$, pyrazole $), 38.3\left(\mathrm{CH}_{2}\right), 36.0\left(\mathrm{CH}_{2}\right), 29.0$ $(\mathrm{CH}), 22.7\left(2 \times \mathrm{CH}_{3}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{ClN}_{2}: 325.1472$ $[\mathrm{M}+\mathrm{H}]^{+}$. Found: $325.1478[\mathrm{M}+\mathrm{H}]^{+}$.

## 4-(Chloromethyl)-3-(4-tert-butylphenyl)-1-phenyl-1H-Pyrazole ( 9 g )

Prepared from (1-phenyl-3-(4-tert-butylphenyl)-1H-pyrazol-4-yl) methanol $(8 \mathrm{~g})(2.00 \mathrm{~g}, 6.52 \mathrm{mmol})$. The product was obtained as a pale yellow solid pure enough for use in the next reaction. Yield 1.46 g ( $69 \%$ ). M.p. $132-134^{\circ} \mathrm{C}$. TLC ( $1: 1$ petroleum ether/EtOAc), $\mathrm{Rf}=0.94 .{ }^{1} \mathrm{H}$ NMR (DMSO-d $\left.{ }_{6}\right) \delta 8.09$ ( $\mathrm{s}, 1 \mathrm{H}$, pyrazole), 7.89 (d. $J=$ $7.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.79$ (d, J=8.4 Hz, 2H, Ar), 7.53 (m, 4H, Ar), 7.35 (t, $J=6.5 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $4.91\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 1.33\left(\mathrm{~s}, 9 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 151.3$ (C, Ar), 150.9 (C, pyrazole), 139.7 (C, Ar), $130.7(\mathrm{CH}$, para-Ar), $130.1(2 \times \mathrm{CH}, \mathrm{Ar}), 129.9(\mathrm{C}, \mathrm{Ar}), 127.6(2 \times \mathrm{CH}$, Ar), $127.0(\mathrm{CH}$, pyrazole $), 126.0(2 \times \mathrm{CH}, \mathrm{Ar}), 118.8(2 \times \mathrm{CH}, \mathrm{Ar}), 118.0$ (C, pyrazole), $38.3\left(\mathrm{CH}_{2}\right), 34.9\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 31.5\left(3 \times \mathrm{CH}_{3}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{20} \mathrm{H}_{22} \mathrm{ClN}_{2}: 325.1466[\mathrm{M}+\mathrm{H}]^{+}$. Found: $325.1461[\mathrm{M}+$ $\mathrm{H}]^{+}$.

## 3-(Benzo[d][1,3] <br> dioxol-5-yl)-4-(chloromethyl)-1-phenyl-1H-pyrazole (9i)

Prepared from (3-(benzo[d][1,3]dioxol-5-yl)-1-phenyl-1H-pyrazol-4$\mathrm{yl}) \mathrm{methanol}(8 \mathrm{i})(0.90 \mathrm{~g}, 3.068 \mathrm{mmol})$. The product was eluted with petroleum ether - EtOAc $85: 15 \mathrm{v} / \mathrm{v}$ to give the product as a yellow solid, yield 0.75 g ( $76 \%$ ). M.p. $158-159^{\circ} \mathrm{C}$. TLC ( $1: 1$ petroleum
ether/EtOAc), $\mathrm{Rf}=0.81 .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{DMSO}-\mathrm{d}_{6}$ ): $\delta 8.54$ ( $\mathrm{s}, 1 \mathrm{H}$, pyrazole), $7.81(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.54(\mathrm{t}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.42(\mathrm{t}, J=$ $7.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ar}), 7.37$ (s, 1H, benzo[d][1,3]dioxole), 7.28 (d, $J=1.6 \mathrm{~Hz}$, 1 H , benzo[d][1,3]dioxole), 6.96 (d, $J=8.1 \mathrm{~Hz}, 1 \mathrm{H}$, benzo[d][1,3] dioxole), 6.07 (s, 2H, benzo[d][1,3]dioxole), $4.92\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 151.4$ (C, pyrazole), 148.0 (C, Ar), 147.9 (C, benzo[d] [1,3]dioxole), 139.7 (C, benzo[d][1,3]dioxole), 129.5 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 128.6 (CH, para-Ar), 126.7 (CH, pyrazole), 126.4 (C, benzo[d][1,3] dioxole), 121.9 (CH, benzo[d][1,3]dioxole), $119.1(2 \times \mathrm{CH}, \mathrm{Ar}), 117.5$ (C, pyrazole), 108.6 (CH, benzo[d][1,3]dioxole), 108.5 (CH, benzo[d] [1,3]dioxole), $101.2\left(\mathrm{CH}_{2}\right.$, benzo[d][1,3]dioxole), $37.2\left(\mathrm{CH}_{2}\right)$. [ESIHRMS] calculated for $\mathrm{C}_{17} \mathrm{H}_{14} \mathrm{ClN}_{2} \mathrm{O}_{2}: 313.0931[\mathrm{M}+\mathrm{H}]^{+}$. Found: $313.0919[\mathrm{M}+\mathrm{H}]^{+}$.

## 2-Chloro-3-(4-(chlorometh- <br> yl)-1-phenyl-1H-pyrazol-3-yl)-1H-indole (9j)

Prepared from (3-(1H-indol-3-yl)-1-phenyl-1H-pyrazol-4-yl)methanol $(8 \mathrm{j})(1.50 \mathrm{~g}, 5.18 \mathrm{mmol})$. The product was eluted with petroleum ether - EtOAc $85: 15 \mathrm{v} / \mathrm{v}$ to give the product as a brown waxy solid, yield 0.90 g ( $51 \%$ ). TLC ( $1: 1$ petroleum ether/EtOAc), $\mathrm{R} f=0.91$. ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ): $\delta 11.07(\mathrm{NH}$, indole), $8.27(\mathrm{~s}, 1 \mathrm{H}$, pyrazole), $7.88(\mathrm{~d}, \mathrm{~J}=$ $7.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.59(\mathrm{t}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}, ~ \mathrm{Ar}), 7.50(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 1 \mathrm{H}$, indole), 7.43 (t, $J=7.5 \mathrm{~Hz}, 1 \mathrm{H}$, indole), 7.34 ( $\mathrm{d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}$, indole), 7.25 (t, $J=7.3 \mathrm{~Hz}, 1 \mathrm{H}$, indole), 7.16 (t, $J=7.5 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), 4.69 ( s , $\left.2 \mathrm{H}, \mathrm{CH}_{2}\right) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 145.8$ (C, Ar), 139.8 (C, indole), 136.7 (C, pyrazole), 134.5 (C, indole), 129.8 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 129.5 (CH, para-Ar), 127.5 (CH, indole), 127.4 (C, pyrazole), 127.3 (C, indole), 123.6 (CH, pyrazole), 120.8 (CH, indole), 120.7 (C, indole), 119.9 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), $119.5(\mathrm{CH}$, indole $), 111.7\left(\mathrm{CH}\right.$, indole), $36.8\left(\mathrm{CH}_{2}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{18} \mathrm{H}_{14} \mathrm{Cl}_{2} \mathrm{~N}_{3}: 342.0565[\mathrm{M}+\mathrm{H}]^{+}$. Found: $242.0550[\mathrm{M}+$ $\mathrm{H}]^{+}$.

## 3-(Chloromethyl)-1-(4-methoxyphenyl)-5-methyl-1H-pyrazole (21)

Prepared from 1-(4-methoxyphenyl)-5-methyl-1H-pyrazol-3-yl)methanol (19) ( $2.0 \mathrm{~g}, 9.16 \mathrm{mmol}$ ). The product was eluted with petroleum ether - EtOAc $40: 60 \mathrm{v} / \mathrm{v}$ to give the product as a brown oil, yield 1.36 g ( $63 \%$ ). TLC ( $1: 1$ hexane/EtOAc), $\mathrm{R} f=0.85$. ${ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ): $\delta \delta 7.52(d, J=9.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.13(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar})$, $6.44\left(\mathrm{~s}, 1 \mathrm{H}\right.$, pyrazole), $4.87\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 3.89\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OCH}_{3}\right), 2.29(\mathrm{~s}, 3 \mathrm{H}$, $\mathrm{CH}_{3}$ ). ${ }^{13} \mathrm{C}-$ NMR (DMSO-d $\mathrm{d}_{6}$ ): $\delta 159.0(\mathrm{C}, \mathrm{Ar}), 148.5(\mathrm{C}$, pyrazole), 140.6 (C, pyrazole), $132.7(\mathrm{C}, \mathrm{Ar}), 126.6(2 \times \mathrm{CH}, \mathrm{Ar}), 114.7(2 \times \mathrm{CH}, \mathrm{Ar}), 106.7$ $(\mathrm{CH}$, pyrazole $), 55.9\left(\mathrm{OCH}_{3}\right), 39.5\left(\mathrm{CH}_{2}\right), 12.4\left(\mathrm{CH}_{3}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{ClN}_{2} \mathrm{O}$ : $237.0795[\mathrm{M}+\mathrm{H}]^{+}$. Found: $237.0799[\mathrm{M}$ $+\mathrm{H}]^{+}$.

## 5-(Chloromethyl)-1-(4-methoxyphenyl)-3-methyl-1H-pyrazole (22)

Prepared from 1-(4-methoxyphenyl)-3-methyl-1H-pyrazol-5-yl)methanol (20) ( $2.0 \mathrm{~g}, 9.16 \mathrm{mmol}$ ). The product was eluted with petroleum ether - EtOAc 40:60 v/v to give the product as a yellow solid, yield 1.43 g ( $66 \%$ ). M.p. $48-50^{\circ} \mathrm{C}$. TLC ( $1: 1$ hexane/EtOAc), $\mathrm{Rf}=0.71$. ${ }^{1} \mathrm{H}$ NMR (DMSO-d $\mathrm{d}_{6}$ : 7.49 (d, $\left.J=9.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}\right), 7.08(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 2 \mathrm{H}$, $\mathrm{Ar}), 6.39(\mathrm{~s}, 1 \mathrm{H}$, pyrazole $), 4.76\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 3.83\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OCH}_{3}\right), 2.22(\mathrm{~s}$, $3 \mathrm{H}, \mathrm{CH}_{3}$ ). ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(\mathrm{DMSO}_{6}\right): \delta 159.5$ (C, Ar), 148.7 (C, pyrazole), 140.9 (C, pyrazole), 133.0 (C, Ar), $126.8(2 \times \mathrm{CH}, \mathrm{Ar}), 115.0(2 \times \mathrm{CH}, \mathrm{Ar})$, $106.9(\mathrm{CH}$, pyrazole $), 56.2\left(\mathrm{OCH}_{3}\right), 39.8\left(\mathrm{CH}_{2}\right), 12.7\left(\mathrm{CH}_{3}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{ClN}_{2} \mathrm{O}: 237.0795[\mathrm{M}+\mathrm{H}]^{+}$. Found: $237.0799[\mathrm{M}$ $+\mathrm{H}]^{+}$.

General Method for the Preparation of Imidazole Derivatives (10), (23) and (25) and Triazole Derivatives (11), (24) and (26)

To a stirred suspension of $\mathrm{K}_{2} \mathrm{CO}_{3}(4 \mathrm{mmol})$ in dry acetonitrile $(10 \mathrm{~mL})$ was added imidazole ( 4 mmol ) or triazole ( 4 mmol ). The reaction mixture was refluxed at $45^{\circ} \mathrm{C}$ for 1 h . After cooling to room temperature the chloromethyl pyrazole compound (9, 21 or 22) ( 1 mmol ) was added and the reaction mixture refluxed at $70^{\circ} \mathrm{C}$ overnight. The solvent was evaporated under reduced pressure and the resulting mixture diluted with EtOAc ( 50 mL ) and washed with brine $(3 \times 20 \mathrm{~mL})$ and $\mathrm{H}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$. The organic layer was dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under reduced pressure to give the crude imidazole or triazole, which was purified by gradient column chromatography.

## 4-(4-((1H-Imidazol-1-yl)methyl)-1-phenyl-1H-pyrazol-3-yl) pyridine (10a)

Prepared from 4-(4-(chloromethyl)-1-phenyl-1H-pyrazol-3-yl) pyridine ( 9 a ) $(0.20 \mathrm{~g}, 0.74 \mathrm{mmol})$. The product was eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{Et}_{3} \mathrm{~N} 91.5: 7.5: 1 \mathrm{v} / \mathrm{v} / \mathrm{v}$ to give the product as a yellow solid, yield 0.15 g ( $67 \%$ ). M.p. $80-82^{\circ} \mathrm{C}$. TLC ( $9: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}$ ), $R f=0.42 .{ }^{1} \mathrm{H}$ NMR (DMSO- $\mathrm{d}_{6}$ ): $\delta 8.65(\mathrm{~d}, \mathrm{~J}=5.7 \mathrm{~Hz}, 2 \mathrm{H}$, pyridine), 8.59 (s, 1H, pyrazole), 7.90 (d, J=7.0 Hz, 2H, Ar), 7.71 (s, 1H, imidazole), 7.66 ( $\mathrm{d}, \mathrm{J}=5.3 \mathrm{~Hz}, 2 \mathrm{H}$, pyridine), $7.55(\mathrm{t}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.38(\mathrm{t}$, $J=7.2 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), 7.16 ( $\mathrm{s}, 1 \mathrm{H}$, imidazole), 6.89 ( $\mathrm{s}, 1 \mathrm{H}$, imidazole), $5.40\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO-d $\mathrm{d}_{6}$ ): $\delta 150.6(2 \times \mathrm{CH}$, pyridine), 147.9 (C, pyridine), 140.1 (C, Ar), 139.5 (C, pyrazole), 135.6 (CH, imidazole), 130.7 (CH, imidazole), 130.1 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 129.2 (CH, pyrazole), 127.5 (CH, para-Ar), $122.1(2 \times \mathrm{CH}$, pyridine), 119.7 (CH, imidazole), 119.1 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 118.4 (C, pyrazole), $40.8\left(\mathrm{CH}_{2}\right)$. [ESIHRMS] calculated for $\mathrm{C}_{18} \mathrm{H}_{16} \mathrm{~N}_{5}$ : $302.1406[\mathrm{M}+\mathrm{H}]^{+}$. Found: 302.1392 $[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(4-((1H-Imidazol-1-yl)methyl)-1-phenyl-1H-pyrazol-3-yl) pyridine (10b)

Prepared from 3-(4-(chloromethyl)-1-phenyl-1H-pyrazol-3-yl) pyridine ( 9 b ) ( $0.50 \mathrm{~g}, 1.85 \mathrm{mmol}$ ). The product was eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{Et}_{3} \mathrm{~N}$ 89:10:1 v/v/v to give the product as a yellow oil, yield $0.30 \mathrm{~g}(54 \%)$. TLC ( $9: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}$ ), $\mathrm{R} f=0.46 .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 8.86(\mathrm{~s}, 1 \mathrm{H}$, pyridine), $8.65(\mathrm{~d}, \mathrm{~J}=4.9 \mathrm{~Hz}, 1 \mathrm{H}$, pyridine), 7.89 (d, $J=7.9 \mathrm{~Hz}, 1 \mathrm{H}$, pyridine), $7.86(\mathrm{~s}, 1 \mathrm{H}$, pyrazole), $7.65(\mathrm{~d}, J=7.6 \mathrm{~Hz}$, $2 \mathrm{H}, \mathrm{Ar}), 7.69$ (s, 1H, imidazole), 7.56 (s, 1H, imidazole), 7.49 (t, $J=$ $8.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.39(\mathrm{~m}, 1 \mathrm{H}$, pyridine $), 7.35(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), 7.97 (s, 1H, imidazole), 5.25 (s, 2H, CH2). ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ): $\delta 149.6$ ( CH , pyridine), 148.6 ( CH , pyridine), 148.2 (C, Ar), 139.4 (C, pyrazole), 135.1 (CH, pyridine), 129.9 (CH, imidazole), $129.6(2 \times \mathrm{CH}, \mathrm{Ar}), 128.5$ (C, pyrazole), $128.0(\mathrm{CH}$, pyridine), $127.3(2 \times \mathrm{CH}, \mathrm{Ar}), 123.7(\mathrm{CH}$, paraAr), 123.7 (CH, imidazole), 119.2 (CH, pyrazole), 119.1 (CH, imidazole), 116.7 (C, pyridine), $41.7\left(\mathrm{CH}_{2}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{18} \mathrm{H}_{16} \mathrm{~N}_{5}: 302.1400[\mathrm{M}+\mathrm{H}]^{+}$. Found: $302.1399[\mathrm{M}+\mathrm{H}]^{+}$.

## 4-((1H-Imidazol-1-yl) <br> methyl)-3-(4-ethylphenyl)-1-phenyl-1H-pyrazole (10c)

Prepared from 4-(chloromethyl)-3-(4-ethylphenyl)-1-phenyl-1Hpyrazole ( 9 c ) $(0.50 \mathrm{~g}, 1.68 \mathrm{mmol})$. The product was eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH} 95: 5 \mathrm{v} / \mathrm{v}$ to give the product as a yellow solid, yield 0.32 g ( $58 \%$ ). M.p. $74-76^{\circ} \mathrm{C}$. TLC ( $9: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}$ ), $\mathrm{Rf}=0.83 .{ }^{1} \mathrm{H}$ NMR (DMSO-d $)_{6}$ : $\delta 7.81$ (s, 1H, pyrazole), 7.73 (d, $J=9 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}$ ), 7.65 (s, 1H, imidazole), 7.52 (d, $J=9 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.47(\mathrm{t}, J=8.3 \mathrm{~Hz}$, 1H, para-Ar), 7.32 ( $\mathrm{m}, 4 \mathrm{H}, \mathrm{Ar}$ ), 7.12 ( $\mathrm{s}, 1 \mathrm{H}$, imidazole), 6.97 ( $\mathrm{s}, 1 \mathrm{H}$, imidazole), 5.24 (s, 2H, CH2 $), 2.72$ (q, J=7.6 Hz, 2H, CH $)_{2}$ ), 1.29 (t, $J=$ $\left.7.5 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 151.5(\mathrm{C}$, pyrazole), $144.8(\mathrm{C}$,

Ar), 139.7 (C, Ar), 129.7 (C, pyrazole), 129.7 (CH, imidazole), 129.5 (C, $\mathrm{Ar}), 128.4(2 \times \mathrm{CH}, \mathrm{Ar}), 127.8(\mathrm{CH}$, para-Ar), $127.6(2 \times \mathrm{CH}, \mathrm{Ar}), 127.4$ ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), $126.7(\mathrm{CH}$, imidazole), $119.0(\mathrm{CH}$, imidazole), $117.7(\mathrm{CH}$, pyrazole), $116.3(2 \times \mathrm{CH}, \mathrm{Ar}), 41.8\left(\mathrm{CH}_{2}\right), 28.7\left(\mathrm{CH}_{2}\right), 15.5\left(\mathrm{CH}_{3}\right)$. Anal. Calcd for $\mathrm{C}_{21} \mathrm{H}_{20} \mathrm{~N}_{4}$ (328.4158): C, 76.80; H, 6.14; N, 17.05. Found: C, 76.68; H, 6.02; N, 17.18.

## 4-((1H-Imidazol-1-yl) methyl)-1-phenyl-3-(4-propylphenyl)-1 H -pyrazole ( 10 d )

Prepared from 4-(chloromethyl)-3-(4-propylphenyl)-1-phenyl-1Hpyrazole ( 9 d ) ( $0.50 \mathrm{~g}, 1.61 \mathrm{mmol}$ ). The product was eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH} 95: 5 \mathrm{v} / \mathrm{v}$ to give the product as a brown oil, yield $0.36 \mathrm{~g}(65 \%)$. TLC ( $\left.9: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}\right), \mathrm{Rf}=0.87$. ${ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : $\delta 9.03$ (s, 1H, pyrazole), 8.37 (s, 1 H , imidazole), 7.78 (d, $J=8.2 \mathrm{~Hz}$, $2 \mathrm{H}, \mathrm{Ar}), 7.48(\mathrm{~m}, 4 \mathrm{H}, \mathrm{Ar}), 7.35(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), 7.30 (d, J= $8.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.28$ ( $\mathrm{s}, 1 \mathrm{H}$, imidazole), 6.93 ( $\mathrm{s}, 1 \mathrm{H}$, imidazole), 5.54 ( s , $\left.2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.67\left(\mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 1.70\left(\mathrm{sext}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right)$, $0.98\left(\mathrm{t}, \mathrm{J}=7.3 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 151.5$ (C, pyrazole), 143.2 (C, Ar), 139.7 (C, Ar), 137.0 (CH, imidazole), 129.7 (C, Ar), 129.6 (CH, imidazole), $129.5(2 \times \mathrm{CH}, \mathrm{Ar}), 129.0(2 \times \mathrm{CH}, \mathrm{Ar}), 127.7(\mathrm{CH}$, imidazole), $127.6(2 \times \mathrm{CH}, \mathrm{Ar}), 126.7(\mathrm{CH}$, para-Ar), $125.2(\mathrm{CH}$, pyrazole), $117.0(2 \times \mathrm{CH}, \mathrm{Ar}), 116.3$ (C, pyrazole), $41.8\left(\mathrm{CH}_{2}\right), 37.8$ $\left(\mathrm{CH}_{2}\right), 24.5\left(\mathrm{CH}_{2}\right), 13.8\left(\mathrm{CH}_{3}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{22} \mathrm{H}_{23} \mathrm{~N}_{4}$ : $343.1917[\mathrm{M}+\mathrm{H}]^{+}$. Found: $343.1915[\mathrm{M}+\mathrm{H}]^{+}$.

## 4-((1H-Imidazol-1-yl) <br> methyl)-3-(4-isopropylphenyl)-1-phenyl-1H-pyrazole (10e)

Prepared from 4-(chloromethyl)-3-(4-isopropylphenyl)-1-phenyl-1Hpyrazole ( 9 e ) $(0.50 \mathrm{~g}, 1.61 \mathrm{mmol})$. The product was eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ 97.5:2.5 v/v to give the product as a brown oil, yield $0.29 \mathrm{~g}(52 \%)$. TLC (9:1 CH2 $\left.\mathrm{Cl}_{2} / \mathrm{MeOH}\right), \mathrm{Rf}=0.74 .{ }^{1} \mathrm{H}$ NMR (DMSO-d $\mathrm{d}_{6}$ ): $\delta 8.51$ (s, 1H, pyrazole), 7.87 (d, J= $8.1 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}$ ), 7.67 (s, 1H, imidazole), 7.56 (d, $J=8.1 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.53(\mathrm{t}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}$, paraAr), 7.35 (d, $J=8.1 \mathrm{~Hz}, 4 \mathrm{H}, \mathrm{Ar}$ ), 7.15 ( $\mathrm{s}, 1 \mathrm{H}$, imidazole), 6.90 ( $\mathrm{s}, 1 \mathrm{H}$, imidazole), $5.30\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.95$ (sept, $J=6.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CH}$ ), 1.25 (d, $J=6.9 \mathrm{~Hz}, 6 \mathrm{H}, 2 \times \mathrm{CH}_{3}$ ). ${ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 150.8$ (C, pyrazole), 149.0 (C, Ar), 139.8 (C, Ar), 130.4 (C, Ar), 130.1 (CH, imidazole), 130.0 $(2 \times \mathrm{CH}, \mathrm{Ar}), 129.1(\mathrm{CH}$, imidazole $), 127.9(2 \times \mathrm{CH}, \mathrm{Ar}), 127.1(2 \times \mathrm{CH}$, $\mathrm{Ar}), 126.9(\mathrm{CH}$, para-Ar), $123.1(\mathrm{CH}$, pyrazole $), 119.7$ (CH, imidazole), $118.8(2 \times \mathrm{CH}, \mathrm{Ar}), 117.2(\mathrm{C}$, pyrazole $), 41.0\left(\mathrm{CH}_{2}\right), 33.7(\mathrm{CH}), 24.3(2 \times$ $\mathrm{CH}_{3}$ ). [ESI-HRMS] calculated for $\mathrm{C}_{22} \mathrm{H}_{23} \mathrm{~N}_{4}: 343.1917[\mathrm{M}+\mathrm{H}]^{+}$. Found: $343.1917[\mathrm{M}+\mathrm{H}]^{+}$.

## 4-((1H-Imidazol-1-yl) methyl)-3-(4-isobutylphenyl)-1-phenyl-1H-pyrazole (10f)

Prepared from 3-(4-isobutylphenyl)-4-(chloromethyl)-1-phenyl-1 Hpyrazole ( 9 f ) $(0.85 \mathrm{~g}, 2.61 \mathrm{mmol})$. The product was eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH} 97.5: 2.5 \mathrm{v} / \mathrm{v}$ to give the product as a yellow oil, yield $0.58 \mathrm{~g}(62 \%)$. TLC ( $\left.95: 5 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}\right), \mathrm{Rf}=0.51$. ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3}$ ): $\delta 7.77$ (s, 1H, pyrazole), 7.70 (d, J= $8.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}$ ), 7.54 (s, 1 H , imidazole), 7.48 (d, $J=7.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.42(\mathrm{t}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.27$ (t, J=7.5 Hz, 1H, para-Ar), $7.22(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.07(\mathrm{~s}, 1 \mathrm{H}$, imidazole), 6.91 ( $\mathrm{s}, 1 \mathrm{H}$, imidazole), 5.16 ( $\mathrm{s}, 2 \mathrm{H}, \mathrm{CH}_{2}$ ), 2.52 (d, J= $\left.7.1 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 1.90(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}), 0.93\left(\mathrm{~d}, \mathrm{~J}=6.6 \mathrm{~Hz}, 6 \mathrm{H}, 2 \times \mathrm{CH}_{3}\right){ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ): $\delta 151.5$ (C, pyrazole), 142.3 (C, Ar), 139.6 (C, Ar), 136.9 (CH, imidazole), 129.7 (C, Ar), $129.6(2 \times \mathrm{CH}, \mathrm{Ar}), 129.5(\mathrm{CH}$, imidazole), $129.5(2 \times \mathrm{CH}, \mathrm{Ar}), 127.6$ ( CH, imidazole), $127.6(2 \times \mathrm{CH}$, Ar), 126.9 ( CH , pyrazole), $126.7(\mathrm{CH}$, para-Ar), $118.9(2 \times \mathrm{CH}, \mathrm{Ar})$, 116.2 (C, pyrazole), $45.2\left(\mathrm{CH}_{2}\right), 41.7\left(\mathrm{CH}_{2}\right), 30.2(\mathrm{CH}), 22.4\left(2 \times \mathrm{CH}_{3}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{23} \mathrm{H}_{25} \mathrm{~N}_{4}$ : $357.2079[\mathrm{M}+\mathrm{H}]^{+}$. Found: $357.2086[\mathrm{M}+\mathrm{H}]^{+}$.

## 4-((1H-Imidazol-1-yl)methyl)-3-(4-(tert-butyl) phenyl)-1-phenyl-1H-pyrazole ( 10 g )

Prepared from 3-(4-(tert-butyl)phenyl)-4-(chloromethyl)-1-phenyl1 H -pyrazole $(9 \mathrm{~g})(0.50 \mathrm{~g}, 1.53 \mathrm{mmol})$. The product was eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH} 97.5: 2.5 \mathrm{v} / \mathrm{v}$ to give the product as a yellow solid, yield 0.31 g ( $57 \%$ ). M.p. $116-117^{\circ} \mathrm{C}$. TLC ( $95: 5 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}$ ), $\mathrm{Rf}=$ 0.72. ${ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 8.52$ (s, 1 H, pyrazole), 7.87 (d, $J=7.8 \mathrm{~Hz}$, $2 \mathrm{H}, \mathrm{Ar}), 7.68$ (s, 1H, imidazole), 7.57 (d, J=8.3 Hz, 2H, Ar), 7.53 (t, J= $7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.49(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.34(\mathrm{t}, J=7.3 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), 7.16 (s, 1H, imidazole), 6.91 (s, 1H, imidazole), 5.30 (s, 2H, $\left.\mathrm{CH}_{2}\right), 1.32\left(\mathrm{~s}, 9 \mathrm{H}, 3 \times \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 151.2(\mathrm{C}, \mathrm{Ar}), 150.7$ (C, pyrazole), 139.8 (C, Ar), 139.8 (C, Ar), 137.5 (CH, imidazole), 130.1 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 130.0 ( CH , imidazole), $129.1(\mathrm{CH}$, para-Ar), $127.6(2 \times \mathrm{CH}$, $\mathrm{Ar}), 126.9$ ( CH, pyrazole), $126.0(2 \times \mathrm{CH}, \mathrm{Ar}), 119.7(\mathrm{CH}$, imidazole), $118.8(2 \times \mathrm{CH}, \mathrm{Ar}), 117.2(\mathrm{C}$, pyrazole $), 41.0\left(\mathrm{CH}_{2}\right), 34.9\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 31.5$ $\left(3 \times \mathrm{CH}_{3}\right)$. Anal. Calcd for $\mathrm{C}_{23} \mathrm{H}_{24} \mathrm{~N}_{4}$ (356.4694): C, 77.50; H, 6.79; N , 15.71. Found: C, 77.92; H, 7.04; N, 15.88.

## 4-((1H-Imidazol-1-yl) <br> methyl)-3-([1,1'-biphenyl]-4-yl)-1-phenyl-1H-pyrazole (10h)

Prepared from 3-([1,1'-biphenyl]-4-yl)-4-(chloromethyl)-1-phenyl-1Hpyrazole $(9 \mathrm{~h})^{[26]}(0.40 \mathrm{~g}, 1.16 \mathrm{mmol})$. The product was eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH} 98: 2 \mathrm{v} / \mathrm{v}$ to give the product as a yellow solid, yield 0.20 g ( $46 \%$ ). M.p. $114-115^{\circ} \mathrm{C}$. TLC ( $95: 5 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}$ ), $\mathrm{Rf}=0.65$. ${ }^{1} \mathrm{H}$ NMR (DMSO-d $\mathrm{d}_{6}$ ): $\delta 8.56$ ( $\mathrm{s}, 1 \mathrm{H}$, pyrazole), 7.90 (d, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}$, Ar), 7.77 (m, 6H, Ar), 7.71 (s, 1H, imidazole), 7.54 (t, $J=7.9 \mathrm{~Hz}, 2 \mathrm{H}$, Ar), $7.50(\mathrm{t}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.40(\mathrm{t}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $7.36(\mathrm{t}$, $J=7.7 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), 7.18 (s, 1H, imidazole), 6.91 (s, 1H, imidazole), 5.37 ( $\mathrm{s}, 2 \mathrm{H}, \mathrm{CH}_{2}$ ). ${ }^{13} \mathrm{C}$ NMR (DMSO-d $\mathrm{d}_{6}$ ): $\delta 150.3$ (C, pyrazole), 140.3 (C, Ar), 140.0 (C, Ar), 139.7 (C, Ar), 132.0 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 130.2 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 130.1 ( CH, imidazole), $129.5(2 \times \mathrm{CH}, \mathrm{Ar}), 129.1$ (CH, imidazole), $128.4(2 \times \mathrm{CH}, \mathrm{Ar}), 128.2(2 \times \mathrm{CH}, \mathrm{Ar}), 127.4(2 \times \mathrm{CH}$, Ar), $127.1(2 \times \mathrm{CH}, \mathrm{Ar}), 127.0(\mathrm{CH}$, para-Ar), 119.7 (CH, imidazole), $118.9\left(\mathrm{CH}\right.$, para-Ar), 117.5 ( C, pyrazole), $41.0\left(\mathrm{CH}_{2}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{25} \mathrm{H}_{21} \mathrm{~N}_{4}: 377.1766[\mathrm{M}+\mathrm{H}]^{+}$. Found: $377.1770[\mathrm{M}+$ $\mathrm{H}]^{+}$.

## 4-((1H-Imidazol-1-yl)methyl)-3-(benzo[d][1,3] dioxol-5-yl)-1-phenyl-1H-pyrazole (10i)

Prepared from 3-(benzo[d][1,3]dioxol-5-yl)-4-(chloromethyl)-1-phenyl-1 H -pyrazole ( 9 i ) $(0.60 \mathrm{~g}, 1.92 \mathrm{mmol})$. The product was eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH} 98: 2 \mathrm{v} / \mathrm{v}$ to give the product as a yellow solid, yield $0.32 \mathrm{~g}(49 \%)$. M.p. $62-64^{\circ} \mathrm{C}$. TLC (95:5 $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}$ ), $\mathrm{Rf}=0.47$. ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 7.72(\mathrm{~s}, 1 \mathrm{H}$, pyrazole), $7.65(\mathrm{~d}, J=7.9 \mathrm{~Hz}$, $2 \mathrm{H}, \mathrm{Ar}), 7.50(\mathrm{~s}, 1 \mathrm{H}$, imidazole), 7.39 (t, J=7.6 Hz, 2H, Ar), 7.24 (t, $J=$ $7.4 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $7.09(\mathrm{~s}, 1 \mathrm{H}$, imidazole), $7.06(\mathrm{~s}, 1 \mathrm{H}$, benzo[d][1,3] dioxole), 6.94 (d, $J=1.6 \mathrm{~Hz}, 1 \mathrm{H}$, benzo[d][1,3]dioxole), 6.89 (s, 1 H , imidazole), 6.82 (d, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}$, benzo[d][1,3]dioxole), 5.93 (s, 2H, benzo[d][1,3]dioxole), $5.10\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 151.0(\mathrm{C}$, pyrazole), 148.1 (C, benzo[d][1,3]dioxole), 147.9 ( $C$, benzo[d][1,3] dioxole), 139.5 (C, Ar), 137.0 (CH, imidazole), 129.7 (CH, imidazole), 129.5 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 127.6 (CH, para-Ar), 126.7 (CH, pyrazole), 121.5 (CH, benzo[d][1,3]dioxole), $118.8(2 \times \mathrm{CH}, \mathrm{Ar}), 118.6$ (CH, imidazole), 108.5 (CH, benzo[d][1,3]dioxole), 108.3 (CH, benzo[d][1,3]dioxole), $101.3\left(\mathrm{CH}_{2}\right.$, benzo[d][1,3]dioxole), $41.9\left(\mathrm{CH}_{2}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{20} \mathrm{H}_{17} \mathrm{~N}_{4} \mathrm{O}_{2}: 345.1352[\mathrm{M}+\mathrm{H}]^{+}$. Found: $345.1366[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(4-((1H-Imidazol-1-yl) <br> methyl)-1-phenyl-1H-pyrazol-3-yl)-2-chloro-1H-indole (10j)

Prepared from 2-chloro-3-(4-(chloromethyl)-1-phenyl-1H-pyrazol-3$\mathrm{yl})$ - 1 H -indole ( 9 j ) $(0.80 \mathrm{~g}, 2.33 \mathrm{mmol})$. The product was eluted with
$\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH} 97: 3 \mathrm{v} / \mathrm{v}$ to give the product as a brown oil, yield $0.49 \mathrm{~g}(56 \%)$. TLC ( $\left.95: 5 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}\right), \mathrm{Rf}=0.72 .{ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : $\delta 10.75(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH}), 7.78$ (s, 1H, imidazole), 7.63 (d, $J=7.7 \mathrm{~Hz}, 2 \mathrm{H}$, Ar), 7.48 (d, $J=7.4 \mathrm{~Hz}, 1 \mathrm{H}$, indole), 7.35 (d, $J=7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}$ ), 7.32 (s, 1 H , pyrazole), 7.19 (d, $J=7.5 \mathrm{~Hz}, 1 \mathrm{H}$, indole), $7.16(\mathrm{t}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), 7.05 ( $\mathrm{m}, 2 \mathrm{H}$, indole), 6.94 ( $\mathrm{s}, 1 \mathrm{H}$, imidazole), $6.75(\mathrm{~s}, 1 \mathrm{H}$, imidazole), $4.99\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right) .{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ): $\delta 144.6$ (C, Ar), 139.8 (C, indole), 134.9 (C, pyrazole), $129.5(2 \times \mathrm{CH}, \mathrm{Ar}), 129.0(\mathrm{CH}$, pyrazole), 127.6 (C, indole), 126.8 ( CH , imidazole), 122.9 (C, indole), 120.8 (CH, imidazole), 119.3 ( CH , imidazole), 119.1 ( $2 \times \mathrm{CH}$, indole), $119.0(3 \times \mathrm{CH}, \mathrm{Ar}), 118.7$ (C, indole), $111.0(2 \times \mathrm{CH}$, indole), 104.4 (C, pyrazole), $41.9\left(\mathrm{CH}_{2}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{21} \mathrm{H}_{17} \mathrm{ClN}_{5}: 374.1172$ $[\mathrm{M}+\mathrm{H}]^{+}$. Found: $374.1187[\mathrm{M}+\mathrm{H}]^{+}$.

## 4-(4-((1H-1,2,4-Triazol-1-yl)methyl)-1-phenyl-1H-pyrazol-3-yl) pyridine (11 a)

Prepared from 4-(4-(chloromethyl)-1-phenyl-1H-pyrazol-3-yl) pyridine ( 9 a ) ( $0.50 \mathrm{~g}, 1.82 \mathrm{mmol})$. The product was eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{Et}_{3} \mathrm{~N} 92.5: 7.5: 1 \mathrm{v} / \mathrm{v} / \mathrm{v}$ to give the product as a white solid, yield 0.27 g ( $48 \%$ ). M.p. $98-100^{\circ} \mathrm{C}$. TLC ( $9: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}$ ), $\mathrm{Rf}=0.69{ }^{1} \mathrm{H}$ NMR (DMSO- $\mathrm{d}_{6}$ ): $\delta 8.67$ ( $\mathrm{m}, 3 \mathrm{H}, 2 \times$ pyridine, $1 \times$ triazole), $8.60(\mathrm{~s}, 1 \mathrm{H}$, pyrazole), 7.98 (s, 1H, triazole), $7.90(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar})$, 7.78 (d, $J=5.8 \mathrm{~Hz}, 2 \mathrm{H}$, pyridine), $7.55(\mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.38(\mathrm{t}$, $J=7.2 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $5.60\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 152.1$ ( CH , triazole), 150.6 ( 2 x CH, pyridine), 148.3 (C, Ar), 140.0 (C, Ar), 139.5 (C, Ar), 131.0 (CH, Ar), 130.2 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 127.5 (CH, Ar), 122.3 $(2 \times \mathrm{CH}, \mathrm{Ar}), 122.1(\mathrm{CH}$, triazole), $119.2(2 \times \mathrm{CH}, \mathrm{Ar}), 117.0(\mathrm{C}, \mathrm{Ar}), 43.5$ $\left(\mathrm{CH}_{2}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{~N}_{6}: 303.1358[\mathrm{M}+\mathrm{H}]^{+}$. Found: $303.1364[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-(4-((1H-1,2,4-Triazol-1-yl)methyl)-1-phenyl-1H-pyrazol-3-yl) pyridine (11 b)

Prepared from 3-(4-(chloromethyl)-1-phenyl-1H-pyrazol-3-yl) pyridine ( 9 b ) ( $0.50 \mathrm{~g}, 1.82 \mathrm{mmol}$ ). The product was eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}-\mathrm{Et}_{3} \mathrm{~N} 92.5: 7.5: 1 \mathrm{v} / \mathrm{v} / \mathrm{v}$ to give the product as a yellow solid, yield 0.31 g ( $55 \%$ ). M.p. $110-112^{\circ} \mathrm{C}$. TLC (9:1 $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}$ ), $\mathrm{R} f=0.68 .{ }^{1} \mathrm{H}$ NMR (DMSO-d $\mathrm{d}_{6}$ ): $\delta 8.67$ (m, $3 \mathrm{H}, 2 \times$ pyridine, $1 \times$ triazole), $8.60(\mathrm{~s}, 1 \mathrm{H}$, pyrazole), 7.98 (s, 1H, triazole), 7.90 (d, $J=7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar})$, 7.78 (d, $J=5.8 \mathrm{~Hz}, 2 \mathrm{H}$, pyridine), $7.55(\mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.38(\mathrm{t}$, $J=7.2 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $5.60\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 152.1$ (CH, triazole), 150.6 ( $2 \times \mathrm{CH}$, pyridine), 148.3 (C, Ar), 140.0 (C, Ar), 139.5 (C, Ar), 131.0 (CH, Ar), 130.2 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 127.5 (CH, Ar), 122.3 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 122.1 (CH, triazole), 119.2 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 117.0 (C, Ar), 43.5 $\left(\mathrm{CH}_{2}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{~N}_{6}: 303.1358[\mathrm{M}+\mathrm{H}]^{+}$. Found: $303.1364[\mathrm{M}+\mathrm{H}]^{+} .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 8.96(\mathrm{~s}, 1 \mathrm{H}$, pyrazole), $8.68(\mathrm{~s}$, 1 H , triazole), $8.08(\mathrm{~s}, 1 \mathrm{H}$, pyridine), $8.05(\mathrm{~s}, 1 \mathrm{H}$, triazole), $8.02(\mathrm{~m}, 2 \mathrm{H}$, pyridine), 7.75 (d, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.51$ (t, $J=7.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.43$ $(\mathrm{t}, J=6.2 \mathrm{~Hz}, 1 \mathrm{H}$, pyridine), $7.37(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $5.48(\mathrm{~s}$, $\left.2 \mathrm{H}, \mathrm{CH}_{2}\right) .{ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ): $\delta 152.5(\mathrm{CH}$, triazole), $149.4(\mathrm{CH}$, pyridine), 148.5 (CH, pyridine), 148.4 (C, Ar), 139.4 (C, pyrazole), 135.5 (CH, pyridine), $129.6(2 \times \mathrm{CH}, \mathrm{Ar}), 128.8$ (CH, triazole), 128.5 (C, pyrazole), $127.3(\mathrm{CH}$, para-Ar), $123.8(\mathrm{CH}$, pyridine $), 119.2(2 \times \mathrm{CH}, \mathrm{Ar}), 119.1$ (CH, pyrazole), 115.0 (C, pyridine), $44.2\left(\mathrm{CH}_{2}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{~N}_{6}: 303.1353[\mathrm{M}+\mathrm{H}]^{+}$. Found: $303.1352[\mathrm{M}+\mathrm{H}]^{+}$.

## 3-((1H-Imidazol-1-yl) methyl)-1-(4-methoxyphenyl)-5-methyl-1 H -pyrazole (23)

Prepared from 3-(chloromethyl)-1-(4-methoxyphenyl)-5-methyl-1Hpyrazole (21) $(1 \mathrm{~g}, 4.22 \mathrm{mmol})$. The product was eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH} 97.5: 2.5 \mathrm{v} / \mathrm{v}$ to give the product as a brown oil, yield $0.65 \mathrm{~g}(58 \%)$. TLC ( $\left.9: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}\right), \mathrm{Rf}=0.47 .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right)$ : $\delta 7.43$ (s, 1H, imidazole), 7.16 (d, $J=8.9 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.08(\mathrm{~s}, 1 \mathrm{H}$,
imidazole), 6.96 (d, J=8.8 Hz, 2H, Ar), 6.82 ( $\mathrm{s}, 1 \mathrm{H}$, imidazole), 6.18 ( s , 1 H , pyrazole), $5.08\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 3.87\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OCH}_{3}\right), 2.34\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$. ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ): $\delta 159.3$ (C, Ar), 147.6 (C, pyrazole), 140.7 (C, pyrazole), 137.2 (CH, imidazole), 132.4 (C, Ar), 129.4 (CH, imidazole), $126.5(2 \times \mathrm{CH}, \mathrm{Ar}), 119.3(\mathrm{CH}$, imidazole $), 114.6(2 \times \mathrm{CH}, \mathrm{Ar}), 105.7$ $(\mathrm{CH}$, pyrazole $), 55.6\left(\mathrm{OCH}_{3}\right), 44.8\left(\mathrm{CH}_{2}\right), 12.2\left(\mathrm{CH}_{3}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{~N}_{4} \mathrm{O}: 269.1397[\mathrm{M}+\mathrm{H}]^{+}$. Found: $269.1395[\mathrm{M}+$ $\mathrm{H}]^{+}$.

## 5-((1H-Imidazol-1-yl) <br> methyl)-1-(4-methoxyphenyl)-3-methyl-1 H -pyrazole (25)

Prepared from 5-(chloromethyl)-1-(4-methoxyphenyl)-3-methyl-1Hpyrazole (22) ( $1 \mathrm{~g}, 4.22 \mathrm{mmol}$ ). The product was eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH} 97.5: 2.5 \mathrm{v} / \mathrm{v}$ to give the product as a brown oil, yield $0.71 \mathrm{~g}(63 \%)$. TLC ( $\left.9: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}\right), \mathrm{Rf}=0.42 .{ }^{1} \mathrm{H} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right)$ : $\delta 7.94$ (s, 1H, imidazole), 7.34 (d, $J=8.9 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.19$ (s, 1 H , imidazole), 7.12 ( $\mathrm{s}, 1 \mathrm{H}$, imidazole), 7.01 ( $\mathrm{d}, J=8.9,2 \mathrm{H}, \mathrm{Ar}$ ), 6.08 (s, 1 H , pyrazole), $5.20\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 3.88\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OCH}_{3}\right), 2.27\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$. ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{CDCl}_{3}$ ): $\delta 159.6$ (C, Ar), 147.8 (C, pyrazole), 140.9 (C, pyrazole), 137.5 (CH, imidazole), 132.6 (C, Ar), 129.7 (CH, imidazole), $126.8(2 \times \mathrm{CH}, \mathrm{Ar}), 119.6(\mathrm{CH}$, imidazole $), 114.7(2 \times \mathrm{CH}, \mathrm{Ar}), 106.0$ ( CH , pyrazole), $55.8\left(\mathrm{OCH}_{3}\right), 45.0\left(\mathrm{CH}_{2}\right), 12.5\left(\mathrm{CH}_{3}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{~N}_{4} \mathrm{O}: 269.1397[\mathrm{M}+\mathrm{H}]^{+}$. Found: $269.1396[\mathrm{M}+$ $\mathrm{H}]^{+}$.

## 1-((1-(4-Methoxyphenyl)-5-methyl-1H-pyrazol-3-yl) methyl)-1H-1,2,4-triazole (24)

Prepared from 3-(chloromethyl)-1-(4-methoxyphenyl)-5-methyl-1Hpyrazole (21) ( $0.5 \mathrm{~g}, 2.11 \mathrm{mmol}$ ). The product was eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH} 97.5: 2.5 \mathrm{v} / \mathrm{v}$ to give the product as a brown oil, yield $0.49 \mathrm{~g}(86 \%)$. TLC ( $\left.9: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}\right), \mathrm{Rf}=0.45 .{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right):$ $\delta 7.95$ (s, 1H, triazole), 7.84 (s, 1H, triazole), 7.24 (m, 2H, Ar), 6.97 (m, $2 \mathrm{H}, \mathrm{Ar}), 6.24\left(\mathrm{~s}, 1 \mathrm{H}\right.$, pyrazole), $5.31\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 3.86\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OCH}_{3}\right)$, $2.33\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 159.4(\mathrm{C}, \mathrm{Ar}), 151.8(\mathrm{CH}$, triazole), 146.0 (C, pyrazole), 140.7 (C, pyrazole), 132.5 (C, Ar), 126.7 (CH, triazole), $126.4(2 \times \mathrm{CH}, \mathrm{Ar}), 114.3(2 \times \mathrm{CH}, \mathrm{Ar}), 105.8$ (CH, para-Ar), $55.6\left(\mathrm{OCH}_{3}\right), 47.4\left(\mathrm{CH}_{2}\right), 12.3\left(\mathrm{CH}_{3}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{14} \mathrm{H}_{16} \mathrm{~N}_{5} \mathrm{O}: 270.1349[\mathrm{M}+\mathrm{H}]^{+}$. Found: $270.1347[\mathrm{M}+\mathrm{H}]^{+}$.

## 1-((1-(4-Methoxyphenyl)-3-methyl-1H-pyrazol-5-yl) methyl)-1H-1,2,4-triazole (26)

Prepared from 5-(chloromethyl)-1-(4-methoxyphenyl)-3-methyl-1Hpyrazole (22) ( $1.0 \mathrm{~g}, 4.22 \mathrm{mmol}$ ). The product was eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH} 97.5: 2.5 \mathrm{v} / \mathrm{v}$ to give the product as a brown crystalline solid, yield 1.02 g ( $90 \%$ ). M.p. $47-48^{\circ} \mathrm{C}$. TLC (97.5:2.5 $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}$ ), $\mathrm{R} f=0.43 .{ }^{1} \mathrm{H}$ NMR ( $\mathrm{DMSO}-\mathrm{d}_{6}$ ): $\delta 8.20$ (s, 1 H , triazole), 7.98 (s, 1H, triazole), $7.33(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 6.99(\mathrm{~d}, J=8.5 \mathrm{~Hz}$, $2 \mathrm{H}, \mathrm{Ar}), 6.16(\mathrm{~s}, 1 \mathrm{H}$, pyrazole $), 5.48\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 3.86\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OCH}_{3}\right)$, $2.26\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right) .{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 159.6(\mathrm{C}, \mathrm{Ar}), 152.0(\mathrm{CH}$, triazole), 146.2 (C, pyrazole), 140.9 (C, pyrazole), 132.7 (C, Ar), 126.9 (CH, triazole), $126.7(2 \times \mathrm{CH}, \mathrm{Ar}), 114.4(2 \times \mathrm{CH}, \mathrm{Ar}), 106.0(\mathrm{CH}$, para-Ar), $55.7\left(\mathrm{OCH}_{3}\right), 47.6\left(\mathrm{CH}_{2}\right), 12.6\left(\mathrm{CH}_{3}\right)$. Anal. Calcd for $\mathrm{C}_{14} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}$ (269.3054): C, 62.44; H, 5.61; N, 25.99. Found: C, 62.32; H, 5.39; N, 26.07.

General Method for the Preparation of N -(3-(1H-Imidazol-1-yl) propyl)-1-phenyl-3-(pyridin-3/4-yl)-1H-pyrazole-4-carboxamides (13) and

## 1-phenyl-3-(pyridin-3/4-yl)-N-(pyridin-4-ylmeth-yl)-1H-pyrazole-4-carboxamides (14)

To a stirred solution of 1-phenyl-3-(pyridin-3/4-yl)-1H-pyrazole-4carboxylic acid (12) ${ }^{[14]}(1.5 \mathrm{mmol})$ in dry DMF ( 15 mL ), was added CDI ( 2.2 mmol ) at room temperature with continuous stirring for 2 h . The reaction was chilled in an ice-cooled water bath and for the preparation of 13 1-(3-aminopropyl)imidazole ( 1.55 mmol ) was added, or for the preparation of 14 4-(aminomethyl)pyridine ( 1.7 mmol ) was added. The mixture was refluxed at $70^{\circ} \mathrm{C}$ for 48 h , then DMF was evaporated under vacuum, and the residue was extracted with EtOAc ( 100 mL ). The organic layer was washed with water $(3 \times 50 \mathrm{~mL})$, dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated under vacuum. The product was purified by gradient column chromatography and eluted with $5 \% \mathrm{MeOH}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$.

## N-(3-(1H-Imidazol-1-yl) <br> propyl)-1-phenyl-3-(pyridin-4-yl)-1H-pyrazole-4-carboxamide (13a)

Prepared from 1-phenyl-3-(pyridin-4-yl)-1H-pyrazole-4-carboxylic acid $(12 \mathrm{a})^{[14]}(0.43 \mathrm{~g}, 1.62 \mathrm{mmol})$ and 1-(3-aminopropyl)imidazole $(0.2 \mathrm{~mL}, 1.67 \mathrm{mmol})$. The product was obtained as a white solid, yield $0.2 \mathrm{~g}(33 \%)$. M.p. $104-106^{\circ} \mathrm{C}$. TLC ( $9: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}$ ), $\mathrm{Rf}=$ 0.29. ${ }^{1} \mathrm{H}$ NMR (DMSO-d $\mathrm{d}_{6}$ ): $\delta 9.01$ (s, 1H, pyrazole), $8.64(\mathrm{~d}, J=6.0 \mathrm{~Hz}$, 2 H , pyridine), $8.46(\mathrm{t}, J=5.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{NH}), 7.93$ (d, $J=8.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar})$, 7.87 (d, $J=6.1 \mathrm{~Hz}, 2 \mathrm{H}$, pyridine), 7.83 (s, 1 H , imidazole), 7.77 (s, 1 H , imidazole), $7.59(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.43(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $7.26\left(\mathrm{~s}, 1 \mathrm{H}\right.$, imidazole), $4.07\left(\mathrm{t}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 3.08(\mathrm{q}, J=7.3 \mathrm{~Hz}$, $2 \mathrm{H}, \mathrm{CH}_{2}$ ), 1.98 (quint, $J=6.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}$ ). ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right): \delta 163.0$ ( $\mathrm{C}=\mathrm{O}$ ), 150.2 ( $2 \times \mathrm{CH}$, pyridine), 148.6 (C, pyridine), 140.0 (C, Ar), 139.1 (C, pyrazole), 130.2 (CH, imidazole), 130.2 (CH, para-Ar), 129.7 $(4 \times \mathrm{CH}, \mathrm{Ar}), 127.8(\mathrm{CH}$, pyrazole $), 123.2(2 \times \mathrm{CH}$, imidazole), $119.4(2 \times$ CH , pyridine), $118.2\left(\mathrm{C}\right.$, pyrazole), $45.0\left(\mathrm{CH}_{2}\right), 37.3\left(\mathrm{CH}_{2}\right), 31.0\left(\mathrm{CH}_{2}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{21} \mathrm{H}_{21} \mathrm{~N}_{6} \mathrm{O}: 373.1777[\mathrm{M}+\mathrm{H}]^{+}$. Found: $373.1789[\mathrm{M}+\mathrm{H}]^{+}$.

## N-(3-(1H-Imidazol-1-yl) <br> propyl)-1-phenyl-3-(pyridin-3-yl)-1H-pyrazole-4-carboxamide (13 b)

Prepared from 1-phenyl-3-(pyridin-3-yl)-1H-pyrazole-4-carboxylic acid $(12 \mathrm{~b})^{[14]}(0.43 \mathrm{~g}, 1.62 \mathrm{mmol})$ and 1 -(3-aminopropyl)imidazole $(0.2 \mathrm{~mL}, 1.67 \mathrm{mmol})$. The product was obtained as a white solid, yield $0.15 \mathrm{~g}(25 \%)$. M.p. $80-82^{\circ} \mathrm{C}$. TLC ( $\left.9: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}\right), \mathrm{Rf}=0.27$. ${ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 9.00$ ( $\mathrm{s}, 1 \mathrm{H}$, pyridine), 8.98 ( $\mathrm{s}, 1 \mathrm{H}$, pyrazole), 8.59 (d, J=3.7 Hz, 1H, pyridine), $8.33(\mathrm{t}, J=5.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{NH}), 8.23$ (d, $J=7.8 \mathrm{~Hz}, 1 \mathrm{H}$, pyridine), 7.92 (d, $J=7.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.66(\mathrm{~s}, 1 \mathrm{H}$, imidazole), 7.59 (t, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.47(\mathrm{~m}, 1 \mathrm{H}$, pyridine), $7.42(\mathrm{t}$, $J=7.3 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), $7.21(\mathrm{~s}, 1 \mathrm{H}$, imidazole), $6.90(\mathrm{~s}, 1 \mathrm{H}$, imidazole), $4.05\left(\mathrm{t}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 3.21\left(\mathrm{q}, J=6.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right)$, 1.95 (quint, $J=6.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}$ ). ${ }^{13} \mathrm{C}$ NMR (DMSO- $\mathrm{d}_{6}$ ): $\delta 163.0(\mathrm{C}=\mathrm{O})$, $150.0(2 \times \mathrm{CH}$, pyridine), $149.6(\mathrm{CH}$, pyridine $), 149.6(\mathrm{CH}$, pyridine $)$, 149.3 (C, pyrazole), 148.8 (C, pyridine), 139.4 (C, Ar), 136.4 (CH, pyridine), $131.0(\mathrm{CH}$, pyridine), $130.3(2 \times \mathrm{CH}, \mathrm{Ar}), 128.8$ (C, pyridine), $127.8(2 \times \mathrm{CH}$, pyridine), $123.6(\mathrm{CH}$, pyrazole $), 122.7(2 \times \mathrm{CH}, \mathrm{Ar})$, 120.2 (C, pyrazole), 119.3 (CH, para-Ar), $41.9\left(\mathrm{CH}_{2}\right)$. Anal. Calcd for $\mathrm{C}_{21} \mathrm{H}_{20} \mathrm{~N}_{6} \mathrm{O}$ (372.4286): C, 67.08; H, 5.47; N, 22.35. Found: C, 66.93; H, 5.57; N, 22.44.

## 1-Phenyl-3-(pyridin-4-yl)-N-(pyridin-4-ylmeth-

 yl)-1H-pyrazole-4-carboxamide (14a)Prepared from 1-phenyl-3-(pyridin-4-yl)-1H-pyrazole-4-carboxylic acid (12 a) ( $0.46 \mathrm{~g}, 1.73 \mathrm{mmol}$ ) and 4-(aminomethyl)pyridine $(0.2 \mathrm{~mL}, 1.97 \mathrm{mmol})$. The product was obtained as a white solid, yield $0.25 \mathrm{~g}(41 \%)$. M.p. $206-208^{\circ} \mathrm{C}$. TLC ( $95: 5 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}$ ), $\mathrm{Rf}=$ 0.25. ${ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 9.09$ ( $\mathrm{s}, 1 \mathrm{H}$, pyrazole), $8.97(\mathrm{t}, J=5.8 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{NH}), 8.63$ (d, $J=6 \mathrm{~Hz}, 2 \mathrm{H}$, pyridine), $8.54(\mathrm{~d}, J=5.9 \mathrm{~Hz}, 2 \mathrm{H}$, pyridine), 7.94 (d, $J=7.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.86$ (d, $J=6.0 \mathrm{~Hz}, 2 \mathrm{H}$, pyridine), $7.60(\mathrm{t}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.43(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), 7.37 (d, $J=$ $5.8 \mathrm{~Hz}, 2 \mathrm{H}$, pyridine), $4.50\left(\mathrm{~d}, \mathrm{~J}=5.9 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2}\right) .{ }^{13} \mathrm{C}$ NMR (DMSO$\left.\mathrm{d}_{6}\right): \delta 163.0(\mathrm{C}=\mathrm{O})$, $150.1(4 \times \mathrm{CH}$, pyridine), 149.0 (C, pyrazole), 148.7 (C, pyridine), 140.0 ( C , pyridine), 139.3 (C, Ar), 131.4 (CH, pyrazole), $130.3(2 \times \mathrm{CH}, \mathrm{Ar}), 128.0(\mathrm{CH}$, para-Ar), $123.1(2 \times \mathrm{CH}$, pyridine $), 122.7$ $\left(2 \times \mathrm{CH}\right.$, pyridine), $119.4(2 \times \mathrm{CH}, \mathrm{Ar}), 118.4(\mathrm{C}$, pyrazole $), 42.0\left(\mathrm{CH}_{2}\right)$. [ESI-HRMS] calculated for $\mathrm{C}_{21} \mathrm{H}_{18} \mathrm{~N}_{5} \mathrm{O}: 356.1511[\mathrm{M}+\mathrm{H}]^{+}$. Found: $356.1523[\mathrm{M}+\mathrm{H}]^{+}$.

## 1-Phenyl-3-(pyridin-3-yl)-N-(pyridin-4-ylmeth-yl)-1H-pyrazole-4-carboxamide (14b)

Prepared from 1-phenyl-3-(pyridin-3-yl)-1H-pyrazole-4-carboxylic acid (12a) ( $0.51 \mathrm{~g}, 1.91 \mathrm{mmol}$ ) and 4-(aminomethyl)pyridine ( $0.22 \mathrm{~mL}, 2.17 \mathrm{mmol}$ ). The product was obtained as a white solid, yield $0.32 \mathrm{~g}(48 \%)$. M.p. $206-208^{\circ} \mathrm{C}$. TLC ( $\left.9: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH}\right), \mathrm{Rf}=$ 0.88. ${ }^{1} \mathrm{H}$ NMR (DMSO-d ${ }_{6}$ ): $\delta 9.10(\mathrm{~s}, 1 \mathrm{H}$, pyridine), $9.00(\mathrm{~s}, 1 \mathrm{H}, \mathrm{NH})$, 8.91 ( $\mathrm{t}, J=5.6 \mathrm{~Hz}, 1 \mathrm{H}$, pyridine), 8.58 ( $\mathrm{d}, J=3.8 \mathrm{~Hz}, 1 \mathrm{H}$, pyridine), 8.53 (d, J=4.9 Hz, 2H, pyridine), 8.21 (d, J=7.8 Hz, 1H, pyridine), 7.92 (d, J=7.8 Hz, 2H, Ar), 7.59 (t, J=7.7 Hz, 2H, Ar), 7.45 (m, 2H, pyridine), 7.36 ( $\mathrm{t}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}$, para-Ar), 4.49 (d, $J=5.8 \mathrm{~Hz}, 2 \mathrm{H}$, $\mathrm{CH}_{2}$ ). ${ }^{13} \mathrm{C}$ NMR ( $\mathrm{DMSO}-\mathrm{d}_{6}$ ): $\delta 163.0(\mathrm{C}=\mathrm{O}), 150.0(2 \times \mathrm{CH}$, pyridine), 149.6 (CH, pyridine), 149.6 (CH, pyridine), 149.3 (C, pyrazole), 148.8 (C, pyridine), 139.4 (C, Ar), 136.4 ( CH, pyridine), 131.0 (CH, pyridine), $130.3(2 \times \mathrm{CH}, \mathrm{Ar}), 128.8$ (C, pyridine), $127.8(2 \times \mathrm{CH}$, pyridine), 123.6 ( CH, pyrazole), 122.7 ( $2 \times \mathrm{CH}, \mathrm{Ar}$ ), 120.2 (C, pyrazole), 119.3 (CH, para-Ar), $41.9\left(\mathrm{CH}_{2}\right)$. Anal. Calcd for $\mathrm{C}_{21} \mathrm{H}_{17} \mathrm{~N}_{5} \mathrm{O}$ (355.3982): C, 70.97; H, 4.82; N, 19.70. Found: C, 71.05; H, 4.79; N, 19.55.

## Biochemistry

CYP121A1 Spectral binding assays for ligand $K_{D}$ determination: CYP121A1 protein was expressed and purified as described previously. ${ }^{[20]}$ Ligand binding assays were performed by spectrophotometric titration using a Cary 60 UV-visible scanning spectrophotometer (Agilent, UK) and a 1 cm path length quartz cuvette, recording spectra between 250 and 800 nm . Titrations were typically done with $3-5 \mu \mathrm{M}$ CYP121A1 at $25^{\circ} \mathrm{C}$ in 100 mM potassium phosphate ( KPi ) buffer, $200 \mathrm{mM} \mathrm{KCl}, \mathrm{pH} 7.85$ with $0.004 \%$ Triton X-100. Ligand stock solutions were prepared in dimethylsulfoxide (DMSO). Ligands were added in small volumes (typically 0.05-0.2 $\mu \mathrm{L}$ aliquots) from concentrated stock solutions to the protein in a 1 mL final volume. Spectral measurements were taken before ligand addition, and following addition of each aliquot of ligand until no further spectral change occurred. Difference spectra at each stage in the titration were obtained by subtraction of the initial ligand-free enzyme spectrum from subsequent spectra collected after each addition of ligand. From the difference spectra, a pair of wavelengths were identified and defined as the absorbance maximum ( $A_{\text {peak }}$ ) and minimum ( $A_{\text {trough }}$ ). The overall absorbance change ( $\Delta \mathrm{A}_{\max }$ ) was calculated by subtracting the $A_{\text {trough }}$ value from the $A_{\text {peak }}$ value for each spectrum collected after a ligand addition. Graphs of $\Delta$ Amax against [ligand] were plotted for each titration. Titrations were done in triplicate and the final $K_{D}$ value presented was determined as the average value across the three sets. The $K_{\mathrm{D}}$ values were determined by fitting the data using either
a standard hyberbolic function (Equation 1) or the Hill equation (Equation 2) using Origin software (OriginLab, Northampton, MA).

$$
\begin{equation*}
\mathrm{A}_{\mathrm{obs}}=\left(\mathrm{A}_{\max }^{*} \mathrm{~L} /\left(K_{\mathrm{d}}+\mathrm{L}\right)\right) \tag{1}
\end{equation*}
$$

In Equation 1 (the standard hyperbolic function, the MichaelisMenten function adapted for ligand binding), $\mathrm{A}_{\text {obs }}$ is the observed absorbance change at ligand concentration $L, A_{\max }$ is the maximal absorbance change observed at apparent ligand saturation, and $K_{\mathrm{d}}$ is the dissociation constant for the binding of the ligand (the substrate concentration at which $\left.A_{\text {obs }}=0.5 \times A_{\text {max }}\right)$.

$$
\begin{equation*}
\mathrm{A}_{\text {obs }}=\left(\mathrm{A}_{\max } \times \mathrm{L}^{\mathrm{n}}\right) /\left(K^{\mathrm{n}}+\mathrm{L}^{\mathrm{n}}\right) \tag{2}
\end{equation*}
$$

In Equation 2 (the sigmoidal Hill equation), $A_{\text {obs }}$ is the observed absorbance change at ligand concentration $L, A_{\max }$ is the absorbance change at apparent ligand saturation, $K$ is the apparent dissociation constant, and $n$ is the Hill coefficient, a value describing the apparent extent of cooperativity observed in ligand binding.

Antimycobacterial Activity Assay: M. tuberculosis $\mathrm{H}_{37} \mathrm{Rv}$ was grown in 7H9 liquid medium with 10\% Middlebrook OADC Growth Supplement enrichment (BBL/Becton-Dickinson, Sparks, MD, USA). Cells were cultured at $37^{\circ} \mathrm{C}$ until mid-log phase was reached $\left(\mathrm{OD}_{600 \mathrm{~nm}}=0.4-0.6\right)$. After cells reached mid-log phase, bacterial suspensions were prepared as described below and REMA assays were performed. The anti-M. tuberculosis activities of the compounds were determined by the REMA (Resazurin Microtiter Assay) method. ${ }^{[21]}$ Stock solutions of the tested compounds ( $10 \mathrm{mg} / \mathrm{mL}$ ) were prepared in DMSO and diluted in Middlebrook 7H9 broth supplemented with $10 \%$ OADC. The microdilution of the compounds was performed in 96-well plates to obtain final compound concentration ranges of $0.39-100 \mu \mathrm{~g} / \mathrm{mL}$. Rifampicin in the concentration range between $0.004-1 \mu \mathrm{~g} / \mathrm{mL}$ was added as control. Bacterial suspensions were prepared and their turbidity adjusted to match the optical density of McFarland no. 1 standard. After a further dilution of $1: 20$ in Middlebrook 7 H 9 broth supplemented with OADC, $100 \mu \mathrm{~L}$ of the inoculum were added to each well of the 96 -well plate. Cultures were incubated for 7 days at $37^{\circ} \mathrm{C}$, and $30 \mu \mathrm{~L}$ of $0.01 \%$ resazurin was added. Wells were read after 24 h for colour change and measured as the fluorescence (excitation/ emission of 530/590 nm filters, respectively) in a microfluorimeter. The MIC was defined as the lowest concentration resulting in $90 \%$ inhibition of $M$. tuberculosis growth. The presented results are representative from two independent experiments.

## Molecular Modeling and Docking

Docking studies were performed using the $\mathrm{MOE}^{[22]}$ program and Mtb CYP121A1 co-crystallized with fluconazole (PDB 2IJ7). All minimisations were performed with MOE until a RMSD gradient of $0.01 \mathrm{Kcal} / \mathrm{mol} / \mathrm{A}$ with the MMFF94 forcefield and partial charges were automatically calculated. The charge of the heme iron at physiological pH was set to $3^{+}$(geometry d2sp3) through the atom manager in MOE. The Alpha Triangle placement, which derives poses by random superposition of ligand atom triplets through alpha sphere dummies in the receptor site, was chosen to determine the poses. The London $\Delta G$ scoring function estimates the free energy of binding of the ligand from a given pose. Refinement of the results was done using the MMFF94 forcefield, and rescoring of the refined results using the London $\Delta \mathrm{G}$ scoring function was applied. The output database dock file was created with different poses for each ligand and arranged according to the final score function (S), which is the score of the last stage that was not set to zero.

## Crystallography Studies on CYP121A1

Untagged CYP121A1 protein and crystals were prepared as previously reported, with the following adaptations. ${ }^{[20]}$ Crystals were prepared using a Mosquito pipetting robot (Molecular Dimensions, Newmarket, UK) in 800 nL drops with protein-tomother liquor at a ratio of $1: 1$, by vapour diffusion in $1.5-2.1 \mathrm{M}$ ammonium sulfate and 0.1 M sodium MES, or Cacodylate from pH 5.5-6.15. Co-crystals were prepared following incubation with 2 mM ligand prepared in DMSO. Protein solutions were centrifuged at $14,000 \mathrm{rpm}$ for 20 mins at $4^{\circ} \mathrm{C}$ immediately before crystallogenesis. Ligand soaks were also carried out either by directly dissolving solid ligand to saturation, or by the addition of a $2-5 \mathrm{mM}$ ligand solution in DMSO to the mother liquor, and soaking was carried out for a minimum period of 24 h . Crystals were immersed in mother liquor supplemented with 10-30\% oil as cryoprotectant, and cryoprotected and flash-cooled in liquid nitrogen. Data were collected on beamline i02 (wavelength 0.9795 Å) at the Diamond Light Source Facility (Harwell, UK). The diffraction data were reduced, scaled and merged using XDS or Xia2. ${ }^{[27,28]}$ Structures were refined using PHENIX ${ }^{[29]}$ with the native CYP121A1 structure (PDB $1 \mathrm{~N} 40)^{[20]}$ as the starting model. Structural rebuilding and validation were performed with COOT, ${ }^{[30]}$ Molprobity ${ }^{[31]}$ and PDB REDO. ${ }^{[32]}$

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## Conflict of Interest

The authors declare no conflict of interest.

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