# Supplementary Information for

## 2 Molecular organization of fibroin heavy chain and mechanism of fibre 3 formation in *Bombyx mori*.

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#### 7 Supplementary Note

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#### 1- <u>NOESY</u>

9 Within the entire sequence of FibH, Y is only found next to A a total of 13 times, out of the 275 Y residues present. However, the <sup>1</sup>H- <sup>1</sup>H NOESY experiments showed consistent NOE signals between the A βH and Y 10 δ,εH (aromatic). It is known that NOE signal depends inversely on the interatomic distance to the sixth power; 11 12 or alternatively, the distance is directly proportional to the intensity to the -1/6 power. Thus, generally for NOE 13 signal to appear, the two atoms need to be within about 6 Å. In fact, after using the internal Y interatomic distances as mean of calibration, it was possible to estimate average distances between A  $\beta$ H and Y  $\delta$ H to be 14 about 3.4 Å and between A and Y EH 3.3 Å. Although these might be underestimated. There are 221 Y residues 15 within the repetitive domains in total, with an estimated of 74 motifs that would place Y next to A in an inter-16 strand motif as described in the main text (Figure 2), thus after this adjustment, the estimated distance would 17 be about 2.2-2.3 Å between the aromatic protons of Y and A βH, in an almost perfect agreement with the model 18 19 (ca. 2.5 Å). Similar calculations can be made for the observed and expected NOEs from Y to S or V, where Y to S are expected to be similarly across strand distances at least 19 times, and Y to V intra-strand distances 20 (motifs VGY) a total of 27 times, the expected distances are of about 2.4-2.8 Å and about 3.2-3.7 Å, while the 21 22 corrected measured distances are 2.1 Å and 2.3 Å, respectively.

#### 23 2- <u>Polymer viscosity</u>

For polymers beyond a critical entanglement concentration, the viscosity follows a power law proportionalityto the molecular weight as shown in Equation 1.

$$\eta = kM^a$$
 Equation 1

26 Where  $\eta$  is the shear viscosity, k is a polymer dependent constant, M is the molecular weight, and a is the 27 proportionality exponent, which for most polymers assumes the value of 3.5 with very low deviation for a 28 myriad of polymers. So much so that the proportionality is often directly expressed with this value. 29 Consequently,  $\eta \propto M^{3.5}$  and thus, if one considers the relative molecular weight reduction observed in the 30 SDS-PAGE, it is about half of the original MW. This reduction in MW would be translated in a proportional 31 reduction of  $\eta$  by  $(0.5)^{3.5}$ , or 0.088.

Furthermore, viscometry analysis of the samples, NLSF and RSF, was conducted in the LiBr solutions at 20 wt% of concentration. The standing assumption is that in the LiBr solution, the protein is in its statistical coil conformation as a polymer in solution. As it would be expected, under these conditions, the value of the zero shear viscosity ( $\eta_0$ ) for NLSF and RSF are 58.1 ± 0.2 and 2.13 ± 0.02 Pa.s, respectively. One order of magnitude difference was predicted from the previous power law relationship. The fitting of the experimental curves was done using the Cross method (see equation contained within Figure S19), where viscosity as a function of shear rate ( $\dot{\gamma}$ ), is fitted using the infinite shear viscosity ( $\eta_\infty$ ),  $\eta_0$ , a relaxation time ( $\tau$ ) and a power coefficient (*m*).



41 Figure S1.

42 2D X-ray diffraction in Fibre diffraction obtained from Silk-I film obtained from NLSF at different illumination

43 angles relative to the film plane. Diffraction pattern obtained at  $0^{\circ}$  (a), between 0 and  $90^{\circ}$  (b) and  $90^{\circ}$ 44 illumination (c).



47 Figure S2.

AF2 simulation of NTD and first repetive domain. (a) Multisequence alignment (MSA) results from a jackhammer database search showing +16000 found sequences aligned to the sequence of the first 650 residues of FibH (NTD+ first repetitive domain). (b) Ribbon representation of one of the five obtained models showing β-solenoid conformation coloured blue to red in an N to C direction. 2D plots of the predicted alignment error (PAE, c), predicted contacts (PC, d) and predicted distogram (PD, e) for the shown model. In the 2D plots, sequence residue position is plotted both on the horizontal and vertical axis, and the predicted value is in the out-of-plane axis, with the range and coloured coding being shown on the right-hand side of each plot.



- 56 Figure S3.
- 57 Conservation of the  $\beta$ -solenoid topology for different generated models. Schematic representation of the five
- 58 obtained models for the first repetitive domain, aligned at the apparent first rigid sub-domain and superimposed
- 59 (centre) to illustrate possible flexibility of the solenoidal structure.



#### 61 Figure S4.

62 Analysis and interpretation of TEM micrograph and FibH structure. TEM image obtained from lightly stained

63 NLSF sample at pH 6 (a), a drawing showing interpretation of the observed profile (b) and a surface

64 representation of a segment of the obtained model from the first repetitive domain showing similar curvature to

65 that observed in the image (c). Scale bar 40 nm.



- 66 67 Figure S5.
- 68 Structure and topology of full FibH and FibL. At the top, a simplified cartoon of FibH plus FibL showing the
- 69 multidomain structure at the primary level, with a legend of domains and colouring shown below. At the bottom
- is the proposed secondary level structure with the domains coloured following the previous cartoon.





72 Figure S6.

One dimensional proton of fibroin. Simple proton spectra of partially deuterated NLSF with chemical shift
 assignment ran using a Bruker Advance III 700 MHz instrument at pH 6 and referenced with DSS.



### Figure S7.

77 Chemical shift assignment of proton. Homonuclear <sup>1</sup>H- <sup>1</sup>H TOCSY experiment conducted using a Bruker

Advance III 700 MHz instrument at pH 6 and referenced with DSS of NLSF (5 mg/mL). Chemical shift assignment is shown regarding the amino acid and coupling connections shown in coloured boxes/lines.

Boxes/lines are coloured by amino acid, with G, A, S, Y, and V is shown with black, red, blue, purple, and green

81 lines, respectively).



82

83 Figure S8.

84 NMR chemical shift residue assignment for NLSF. Heteronuclear <sup>1</sup>H- <sup>15</sup>N HSQC (a, blue) and <sup>1</sup>H- <sup>13</sup>C HSQC

85 (b, green) with their respective shift assignment.



87 Figure S9.

Effect of pH on chemcial shifts of NLSF. Superimposed <sup>1</sup>H- <sup>13</sup>C HSQC spectra obtained for NLSF samples
 buffered at pH 6 and 8, showing no change in the chemical environment of the detectable residues.





91 Figure S10.

- 92 Silk I to Silk II transformation, dihedral angles perspective. Proposed stretching induced transformation of the
- 93 proposed Silk-I to the theoretical Silk-II (DOI: 10.5452/ma-cs24y) as observed from dihedral angles.



95 Figure S11.

96 Analysis and interpretation of NOESY data from NLSF. Distance against intensity <sup>-1/6</sup> for the assigned shifts

and sticks representation of a Y residue showing estimated intra-residue distances used as internal calibrationand estimated experimental values according to the internal calibration.





100 Figure S12.

101 Constant frquency rheology characterisation of NLSF. Shear rheology oscillatory experiment where the 102 complex strain was varied between 0.1 and 100 % at a constant frequency (1 Hz) for the samples of NLSF 103 buffered at different pH (indicated within the graphs). Curves are shown as averages of 5 replicates, with the 104 standard deviation shown as the shadowed area around the solid line. Experiments were intended to determine 105 the linear viscoelastic limit of the materials.



107 Figure S13.

108 Constant strain rheology characterisation of NLSF. Shear rheology oscillatory experiment where the frequency 109 was varied between 0.1 and 10 Hz at a constant strain (2 %) for the samples of NLSF buffered at different pH 110 (indicated within the graphs). Curves are shown as averages of 5 replicates, with the standard deviation shown 111 as the shadowed area around the solid line. Experiment shows the clear sol-gel transition suffered by the material

as the pH is lowered below 7.







115 Shear viscosity against shear rate for liquid samples of NLSF at different pH. Samples at pH (7, 8 and 9) and

116 (a) and first normal force difference vs shear rate of same samples (b). Curves showed averages (dashed black

117 lines) and standard deviation (shadowed area around average), technical replicates N = 5.



118

119 Figure S15.

120 Circular dichroism analysis of NLSF samples. Overlapped normalized Ellipticity values of the buffered samples

121 obtained by far-UV CD spectroscopy (a). Bestsel deconvolution of the curves showing estimated structural

122 contributions to the ellipticity signal (b). Series temperature scans of the buffered samples in the far-UV region,

where the vertical axis is the temperature, horizontal axis the wavelength, and the far-right the out-of-plane scale
with red being the maximum range and blue minimum (c).



125

- 126 Figure S16.
- 127 Normalized  $\beta$ -sheet content as a function of temperature for all buffered samples (NLSF). The  $\beta$ -sheet content
- 128 was estimated as the ratio between the ellipticity at 218 and 195 nm, normalized against the maximum obtained
- value for the data set.







132 Dynamic light scattering, DLS, analysis. DLS size distribution of NLSF obtained at different pH values plotted

as the Numbers (%) against the size in a lineal vs log scale (a). Values of the Z-average size for the buffered

134 NLSF samples as a function of temperature (b). Values presented for technical replicates (N: 3).



135

136 Figure S18.

137 Comparison of RSF and NLSF materials. Uncropped SDS-PAGE gel of RSF and NLSF at different protein

138 concentrations with reference ladder next to them (a). profiles extracted from the gels image from the selected

139 bands in rectangles (b). Photos of samples loaded onto a rheometer geometry of unbuffered RSF and NLSF

140 indicated (c).





142 Figure S19.

143 Viscometry analysis of silk ion LiBr solutions. Flow curves obtained for NLSF and RSF in their LiBr solutions 144 plotted as shear viscosity against shear rate in a log-log scale. Solid lines represent averaged experimental data,

and dotted or dashed lines represent fits to a Cross fluid model, with constitutive equation shown on the right,

and the table below it shows the fitting parameters.



148 Figure S20.

Results from the proteomics analysis. Cartoons representing the sequence coverage of FibL (a) P25 (b)) and FibH (c). Plot representing the frequency distribution of peptides against residue position of the covered peptides from FibH (d). Plot representing the confidence of found peptides against their residue position (e). Series of example mass spectrograms obtained for the labelled peptides in FibH (f-i). Within the cartoons, the magenta section represents low confidence assignments, and green and black represent high confidence assignments.



154

155 Figure S21.

Regeneration induced hydrolysis on fibroin. Carton showing the multidomain architecture of the primary structure of FibH (top), followed by an aligned cartoon representing the total sequence coverage from the hydrolysate peptides recovered from RSF. Below, different plots show several indicated residues' positions along the sequence of FibH (a). Scheme representing the recovered peptides from RSF, with logo representations of the observed residues at terminal positions of the detected peptides, and those expected from the retained chains (b)



Figure S22. 

TEM analysis of morphological changes in silk fibroin (NLSF). Negatively stained TEM images of globular units observed at pH 8 (a), and elongated fibrillar structures observed at pH 6 (b). Scale bars represent 100 nm 

in both cases.



167

- 168 Figure S23.
- 169 TEM analysis of supramolecualr assemblies obbtained for NLSF. Negatively stained TEM images of intact 170 supramolecular brush-like fibres (a, b) and illustrative images showing possible lateral interactions (c) and
- 171 breakage of the structure (d), with arrows added as guides. Scale bars are 500 nm in (a-d).



172 173 Figure S24.

N-terminal domain oligomerisation. (a) Ribbons representation of two stacked tetrameric units showing the 174 175 manually curated model (left) and the equivalent relaxed and docked minimal energy model (right) as observed from the top (top). (b) Estimated rosseta binding free energies of docking simulations against pH for the n4/n4 176 177 interface. (c) Estimated rosetta binding free energies of docking simulations against residue median standard 178 deviation (RMSD) of the different simulations with the manually curated and relaxed model as reference for the n4/n4 interface. (d) Estimated rosseta binding free energies of docking simulations against pH for the n2/n2 179 180 interface. (e) Estimated rosetta binding free energies of docking simulations against residue median standard 181 deviation (RMSD) of the different simulations with the manually curated and relaxed model as reference for the n2/n2 interface. 182



Dimer solenoid regular YY-subdomain



Dimer solenoid regular nY-subdomain



#### 184

185 Figure S25.

The role of topology and Ala residues in driving lateral interactions. Complete proposed FibH surface model coloured to highlight Ala (black), Y (magenta) and Val (black) from all other residues (green or light pink) shown at the top. At the centre of the composite image, the detail of the first and sixth repetitive domain is shown, highlighting the detected subdomains and the difference in the spatial distribution of Y residues. At the bottom, two laterally docked models of the two different subdomains, showing binding at Ala interfaces with avoidance of Y residues; colour wheat or green represent all other residue but Ala and Y.



193 Figure S26.

Proposed NTD-driven self-assembly process: FibH, as a multidomain solenoidal molecule exists, takes a 194 globular appearance with a diameter of about 20 nm at pH values above 7. At pH 7, the linkers are partially 195 196 protonated, and the protein can extend and start participating in lower-order oligomer species. As pH continues 197 to drop, assembly continues into the supramolecular bottlebrush fibres. PDB models represent only the NTD with the first repetitive domain, whereas drawings represent, schematically, the complete molecule. The 198 199 background image's colour and shape allude to the pH gradient (from 8 to 6) and the anterior section of the silk 200 gland (ASG), respectively. Negatively stained TEM images shown below are intended to show snapshots of the 201 assembly process, and scale bars represent 500 nm.



204 Figure S27.

Microscopy analysis of natural silk fibre. SED-SEM image of a sheared degummed fibre (a). SEM detail off
the same fibre showing nanofibrillar texture and void formation during shearing (b). TEM image of a
microtomed section of an embedded raw fibre showing twin fibroin fibres coated by a less electron-dense sericin
layer (c). SEM image showing further detail of the nanofibrillar texture on a tilted specimen (D). Scale bars are
µm in (a) and 500 nm (b-d).



- 212 Figure S28.
- Fibre X-ray diffraction of natural silk fibre. 2D X-ray diffraction pattern of a raw silk fibroin fibre with assigned
- 214 distances.



215216 Figure S29.

Proposed network formation is driven by lateral interaction of solenoid units: Structure of a tetrameric unit within the supramolecular complex, showing the direction towards c-termini where the first repetitive domain would be located and the possible initiation intra-unit lateral interactions (a). Network of supramolecular fibres stabilized by lateral interactions of solenoids (b). Breakage of the NTD interactions and transformation to a flow-aligned solenoid network formation (c). Cryo-TEM image obtained from an apparent proto fibre at low magnification from NLSF (d), and at high magnification (e), showing what is believed to be evidence of the pre-extended solenoid network (NLSF) (e). Scale bar 1 µm in (d) and 100 nm in (e).



224

225 Figure S30.

Evidence of two yield stresses. Shear yield stress experiments conducted on a pH 7 viscoelastic sample: Shear viscosity and shear stress against time in a log-lineal scale, showing two yield points shown as maxima in

viscosity and shear stress against time in a log mean searce, showing two yield points showin as maxima in viscosity (a). Shear stress and its derivative against accumulated shear strain for two independent samples

showing a first sharp and a second broad yield (b).

230 Table S1. Different literature unit cell parameters for Silk I structures derived from X-ray diffraction.

		Unit co	en para	imet	ers				
ĺ	а	b	С	α	β	γ	reference	Year	Molecular Feature
	4.59	7.20	9.08	90	90	90	Konishi & Kurokawa <sup>43</sup>	1968	Helical folded strands
	4.72	14.4	9.4	90	90	90	Lotz & Keith <sup>30</sup>	1971	Crankshaft model*
	4.65	14.24	8.88	90	90	90	Okuyama et al. <sup>31</sup>	1988	Crankshaft-like*
	4.79	4.79	22.2	90	90	120	Muñoz-Guerra et al.44	1988	Almost extended strand
	8.94	6.46	11.26	90	90	90	Fossey et al.45	1991	Almost extended strand
	22.66	5.70	20.82	90	90	90	He et al. <sup>28</sup>	1999	Almost extended strand

231 \*Model used to derive the type-II  $\beta$ -turn model.

## Table S2.

233	Reflections	obtained fro	m the	oriented	film	diffraction	pattern	show ty	vpical	Silk-Is	pacing	3
233	remeenions	obtained no	in une	onicited	111111	annaction	pattern	SHOW U	picui	DIR I D	pacing	- ۲

Experimental results									
Equator	ial	Meridional							
d-Spacings (Å)	Intensity	d-Spacings (Å)	Intensity						
7.35	226	5.21	169						
5.12	132	3.50	111						
3.70	170	3.15	104						
2.56	62	2.04	42						
2.06	33								

- 235 Table S3.
- 236 NMR chemical shifts. Assigned chemical shifts for simplified motifs compared to those found in the literature 237  $^{22}$ .



<u>h</u>	<u>k</u>	Ī	d-spac	ing															
-5	1	0	1.754	-2	0	-1	2.562	0	-2	1	2.064	2	-3	-1	1.721	4	-2	1	1.824
-5	2	0	1.788	-2	0	0	3.897	0	-1	-1	2.845	2	-3	0	1.996	4	-1	-1	1.863
-4	0	-1	1.691	-2	0	1	2.562	0	-1	0	5.196	2	-3	1	1.721	4	-1	0	2.227
-4	0	0	1.949	-2	1	-1	2.673	0	-1	1	2.845	2	-2	-1	2.226	4	-1	1	1.863
-4	0	1	1.691	-2	1	0	4.323	0	0	-2	1.7	2	-2	0	2.946	4	0	-1	1.691
-4	1	-1	1.863	-2	1	1	2.673	0	0	-1	3.4	2	-2	1	2.226	4	0	0	1.949
-4	1	0	2.227	-2	2	-1	2.226	0	0	1	3.4	2	-1	-1	2.673	4	0	1	1.691
-4	1	1	1.863	-2	2	0	2.946	0	0	2	1.7	2	-1	0	4.323	5	-2	0	1.788
-4	2	-1	1.824	-2	2	1	2.226	0	1	-1	2.845	2	0	1	2.562	5	-1	0	1.754
-4	2	0	2.162	-2	3	-1	1.721	0	1	0	5.196	2	1	-1	2.046				
-4	2	1	1.824	-2	3	0	1.996	0	1	1	2.845	2	1	0	2.563				
-4	3	0	1.824	-2	3	1	1.721	0	2	-1	2.064	2	1	1	2.046				
-3	-1	-1	1.701	-1	-2	-1	1.824	0	2	0	2.598	2	2	0	1.788				
-3	-1	0	1.964	-1	-2	0	2.162	0	2	1	2.064	3	-3	-1	1.701				
-3	-1	1	1.701	-1	-2	1	1.824	0	3	0	1.732	3	-3	0	1.964				
-3	0	-1	2.064	-1	-1	-1	2.464	1	-3	0	1.904	3	-3	1	1.701				
-3	0	0	2.598	-1	-1	0	3.576	1	-2	-1	2.226	3	-2	-1	2.064				
-3	0	1	2.064	-1	-1	1	2.464	1	-2	0	2.946	3	-2	0	2.598				
-3	1	-1	2.25	-1	0	-1	3.116	1	-2	1	2.226	3	-2	1	2.064				
-3	1	0	3	-1	0	0	7.794	1	-1	-1	2.945	3	-1	-1	2.25				
-3	1	1	2.25	-1	0	1	3.116	1	-1	0	5.892	3	-1	0	3				
-3	2	-1	2.064	-1	1	-1	2.945	1	-1	1	2.945	3	-1	1	2.25				
-3	2	0	2.598	-1	1	0	5.892	1	0	-1	3.116	3	0	-1	2.064				
-3	2	1	2.064	-1	1	1	2.945	1	0	0	7.794	3	0	0	2.598				
-3	3	-1	1.701	-1	2	-1	2.226	1	0	1	3.116	3	0	1	2.064				
-3	3	0	1.964	-1	2	0	2.946	1	1	-1	2.464	3	1	-1	1.701				
-3	3	1	1.701	-1	2	1	2.226	1	1	0	3.576	3	1	0	1.964				
-2	-2	0	1.788	-1	3	0	1.904	1	1	1	2.464	3	1	1	1.701				
-2	-1	-1	2.046	0	-3	0	1.732	1	2	-1	1.824	4	-3	0	1.824				
-2	-1	0	2.563	0	-2	-1	2.064	1	2	0	2.162	4	-2	-1	1.824				
-2	-1	1	2.046	0	-2	0	2.598	1	2	1	1.824	4	-2	0	2.162				

239	Table S4.
240	NMR chemical shifts. Assigned chemical shifts for simplified motifs compared to those found in the literature
241	<u>22</u> .

Chemical shift (ppm, measured/literature*)															
		HN	HA	HB	HD	HE F	IG	HD F	HE	CA	СВ	CG	CD	CE	Ν
	G 1	8.4/8.46*	4.1/3.89*						2	44.8/ 44.7*					108.15/ 107.9*
	A 2	8.25/8.25*	4.25/4.26*	1.3/ 1.33*					5	52.7/ 52.2*	19.1/ 18.6 <sup>°</sup>	/ *			123.6/1 23.5*
<	G 3	8.48/8.49*	3.94/3.93*						2	44.8/ 44.7*					108.17/ 108.1*
	s 4	8.3/8.3*	4.46/4.4*	3.86;3. 91/ 3.81;3. 87*					5	58.5/ 58*	63.9, 63.4	/ *			115.5/ 115.3*
	G 5	8.54/8.55*	3.96/3.92*						2	44.8/ 44.7*					110.9/ 110.8*
	A 6	8.25/8.25*	4.25/4.26*	1.3/ 1.33*					5	52.7/ 52.2*	19.1, 18.6 <sup>°</sup>	/ *			123.6/ 123.5*
	G 1	8.41/8.36*	3.89/3.76;3 .84*						2	44.8/ 44.7*					107.9/ 107.9*
	A 2	8.18/8.17*	4.30/4.22*	1.35/ 1.3*					5	52.7/ 52.2*	19.0, 18.6 <sup>°</sup>	/ *			123.6/ 123.4*
<	G 2 3	8.41/8.39*	3.76/3.84							44. 44.	78/ .7*				108.14/ 107.9*
	Эрү 4	8.1/8.1*	4.51/4.46*	2.9;3.0 4/ 2.85;2. 98*	7.0 7/ 6.9 5*	6.8/ 6.71	/ *	7.06	6.7	78 57 57	8.2/ 7.8*	38.7/ 38.3*	133 .45	118.2 6	119.86/ 119.8*
	G 5	8.41/8.41*	3.94/3.76;3 .85*						2	44.8/ 44.7*					110.68/ 110.6*
	A 6	8.18/8.18*	4.3/4.23*	1.3/ 1.3*					5	52.7/ 52.2*	19.1, 18.6	/ *			123.6/ 123.4*
	G 1	8.47/8.39*	3.89/3.76;3 .84*						2	44.8/ 44.7*					108.15/ 107.9*
	A 2	8.19/8.18*	4.29/4.23*	1.3/ 1.3*					5	52.7/ 52.2*	19.1, 18.6	/ *			123.56/ 123.4*
<	G 3	8.47/8.37*	3.89/3.83*						2	44.8/ 44.7*					108.15/ 107.5*
	V 4	7.98/7.97*	4.11/4.04*	2.06/ 2*			).8 7/ ).8 3*		e	62.36 / 62*	32.63 / 32.1 <sup>3</sup>	3 20.36 / * 20.1*	5 *		119.18/ 118.5*
	G 5	8.52/8.51*	3.9/3.86*						2	44.8/ 44.7*					110.7/ 110.8*
	A 6	8.19/8.18*	4.29/4.23*	1.3/ 1.3*					5	52.7/ 52.2*	19.1 18.6	/ *			123.6/ 123.4*

243 Table S5.

Predicted chemical shifts for simplified motifs within the sixth repetitive domain. Chemical shifts were
 predicted using the webserver from SHIFTX2.

		ΗN	HA	HB	HD	HE	CA	СВ	CD	CE	Ν
	G1	8.20	4.10				44.50				108.00
4	A2	8.50	4.50	1.40			51.40	18.70			123.00
SG/	G3	8.00	4.20				43.80				108.20
DAG	<b>S</b> 4	8.40	4.70	3.75/3.79			57.60	63.00			115.40
U	G5	7.90	4.10				44.70				109.30
	A6	8.40	4.50	1.40			51.60	18.20			122.90
	G1	8.5	3.96				45.24				108.7
4	A2	8.25	4.3	1.4			52.7	19.1			123.6
λG/	G3	8.5	3.96				45.24				108.7
DAG	Y4	8.1	4.5	2.91/3.03	7.06	6.78	58.16	38.77	133.45	118.26	119.88
U	G5	8.5	3.96				45.24				108.7
	A6	8.25	4.3	1.4			52.7	19.1			123.6

247 Table S6.

248 Table with the logarithmic shifts used to calculate the master curves

infinite shifts used to calculate the master eurves											
	Sol master c	Gel master curve									
pН	Log(a <sub>pH</sub> ), Freq.	*Log(a <sub>pH</sub> ), G	pH	Log(a <sub>pH</sub> ), Freq.							
7	0	1.83	7	0							
8	-0.86	0	6.5	1.23							
			6	3.14							

\*Given the suspected lower concentration of the sol fraction at this pH, and knowing that like viscosity, the

moduli follow a power-law relationship with concentration, the moduli were scaled to match that of pH 8,however leaving the frequency domain unaffected.