# CARDIFF UNIVERSITY 

## PRIFYSGOL CAERDYB

# THE EVOLUTION OF THE MAGMATIC PLUMBING SYSTEM ASSOCIATED WITH THE FRANKLIN LARGE IGNEOUS PROVINCE (VICTORIA ISLAND, ARCTIC CANADA) 

Ben Hayes

Submitted in partial fulfilment of the requirements for the degree of PhD .

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#### Abstract

This thesis is an investigation of the magmatic processes that operated in the plumbing system of the Franklin Large Igneous Province (LIP). The Franklin LIP (~723-716 Ma) extends $>2500 \mathrm{~km}$ across northern Canada and is superbly exposed in the Minto Inlier of Victoria Island. The exposed plumbing system is dominated by sills, with subordinate fault-guided feeder dykes. Locally, sills are characterised by olivine-enriched bases, with overlying doleritic gabbros. One example is the Lower Pyramid Sill. This sill was investigated in detail using petrography, whole-rock chemistry and in-situ mineral chemistry. Petrological and geochemical observations, combined with inverse melt modelling show that the olivine-enriched layer formed by olivine slurry replenishment into a resident gabbroic mush. In contrast, the overlying gabbros formed by in-situ differentiation. Intervening clinopyroxene-rich cumulates formed by a metasomatic reaction between different pore melts. The study emphasises that both slurry emplacement and in-situ differentiation played a role in the formation of cumulate layering in this sill.

Sections examined through the same sill along strike, collectively termed the Fort Collinson Sill Complex, reveal that the olivine slurry propagated $>50 \mathrm{~km}$ downstream. Radiogenic and stable isotope data support this interpretation. The geochemical consequences of cumulate remobilisation and mixing in the sill complex are demonstrated, as well as quantified, using $\mathrm{Fe}=\mathrm{Mg}$ modelling. Additionally, the effects of dolostone assimilation are considered, which can potentially affect phase equilibria.

This thesis also investigates the Ni-Cu-PGE potential of the Franklin LIP by studying the Southern Feeder Dyke Complex. This dyke complex was fault-guided and brecciated dykes contain microdoleritic microxenoliths that are impregnated by sulphides. Immiscible sulphide liquids formed when the magma became sulphur saturated due to the ingestion of local sulphur-rich wallrock. This study demonstrates the importance of external sulphur addition to magma in triggering sulphide mineralisation and highlights the economic potential of the Franklin LIP.


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## 1. Introduction

### 1.1 Project rationale \& aims

The principal subject of this thesis is an investigation of the magmatic processes that operated in the magma plumbing system of the Franklin Large Igneous Province (LIP). The study has three aims. The first is to understand the broader scale processes of how LIPs differentiate. The Franklin LIP as exposed on Victoria Island (Arctic Canada), comprises a sill-dominated magmatic plumbing system overlain by flood basalts. Sills act as an intermediary stage for magma traversing the crust, from mantle source to the Earth's surface, and are the location of igneous differentiation. Hence, by studying the Franklin sills, I aim to understand the processes of emplacement, melt transport and crystallisation during the evolution of a LIP plumbing system.

The second aim is to investigate smaller scale differentiation mechanisms that operate in sills. The importance of sills is emphasised by their presence in many igneous terranes; they are common in Proterozoic and Phanerozoic continental flood basalt provinces (Baragar, 1976, Francis \& Walker, 1986, Hawkesworth et al., 1995, Chevallier \& Woodford, 1999, Bédard et al., 2007); ophiolites and oceanic crust (Bédard, 1991a, Lissenberg et al., 2004, Canales et al., 2009); oceanic plateau complexes (Kerr et al., 1998); and Archean tholeiitic lava plains (Bédard et al., 2009). Therefore, the importance of studying sills cannot be understated, and the Franklin sills on Victoria Island represent an excellent target to study magmatic processes operating in sills. Extensive work has been undertaken and our understanding of magmatic processes has advanced considerably over the past $\sim 80$ years, however, many unresolved issues remain, particularly with regard to how cumulate layering forms. The Franklin sills offer the opportunity to investigate the differentiation mechanisms that operated in a basaltic magma chamber. Such investigations have typically focused on large layered intrusions such as the Skaergaard Intrusion and Bushveld Complex. However, the slow cooling times of large layered intrusions allows extensive post-cumulus diffusive exchange, which overprints the primary magmatic signature. The Franklin sills are thin and cooled relatively rapidly in comparison to large layered intrusions. Faster cooling rates of magma allows the preservation of the primary magmatic signature, which provides insights into the differentiation mechanisms that
operated in the sills. Additionally, some of the Franklin sills are characterised by well developed cumulate layering, similar to that observed in large layered intrusions.

The final aim relates to the economic geology of the Franklin LIP. The Franklin LIP is considered to be prospective for $\mathrm{Ni}, \mathrm{Cu}$ and PGE mineralisation (Jefferson et al., 1985, Jefferson et al., 1994, Jowitt \& Ernst, 2013). The Franklin sills have been compared to the Noril'sk-Talnakh intrusions in Siberia (Jefferson et al., 1994). There are known native copper occurrences in the Natkusiak flood basalts, the extrusive part of the Franklin LIP (Jefferson et al., 1985). This thesis aims to constrain and further understand the $\mathrm{Ni}-\mathrm{Cu}-$ PGE economic potential of the Franklin LIP on Victoria Island.

### 1.2 Fieldwork

Fieldwork for this project took place during the summer months of 2010 and 2011 on Victoria Island. The summer months and short lengths ( $\sim 7$ weeks each) of the field season were necessary because of the high latitude $\left(\sim 72^{\circ} \mathrm{N}\right)$. The Geological Survey of Canada organised and sponsored the project fieldwork, which was one component of a mapping initiative called GEM (Geomapping for Energy \& Minerals Project). The principal aim of GEM is to open the Canadian Arctic to geological mapping and research in order to improve understanding of its economic potential. The field base camp was located in different locations for each field season. In 2010, the field base camp was located $\sim 15 \mathrm{~km}$ NE of the easternmost end of Minto Inlet, and the 2011 base camp was located $\sim 10 \mathrm{~km}$ NW of Ulukhaktok (both are labelled on Fig. 2.2). Helicopter support provided transport throughout both field seasons.

Dr Robert Rainbird and Dr Jean Bédard, both of the Geological Survey of Canada, were co-leaders of the mapping expedition to Victoria Island. The expedition consisted of $\sim 15$ students, all of whom were undertaking either an undergraduate, masters or Ph.D. project on the rocks. Other senior researchers also contributed during fieldwork. I was involved in both field seasons, as both a mapping geologist and a researcher for this thesis. The majority of the 2010 field season involved mapping in order to understand the geology, particularly the architecture and composition of the Franklin igneous rocks. The Lower Pyramid Sill (chapter 4) was sampled during the 2010 field season, as were the majority of
the rocks comprising the Fort Collinson Sill Complex (chapter 5). The Uhuk Massif was identified during the 2010 field season and I was involved in the detailed mapping of this important locality. The field relationships of the Uhuk Massif were the focus of an article published in Geological Society of America Bulletin in May 2012, which I co-authored (appendix A). More specific mapping and sampling was targeted during the 2011 field season, following mapping in 2010. I mapped and sampled the Southern Feeder Dyke Complex (chapter 6) during the 2011 field season. Both the Uhuk Massif and Southern Feeder Dyke Complex localities were included in a contact metamorphism article published in Journal of Metamorphic Geology in November 2012, on which I am a coauthor (appendix B).

### 1.3 Thesis outline

Chapter 2 provides a summary of the geological setting of the Franklin LIP, specifically its manifestation in the Minto Inlier of Victoria Island. Chapter 3 provides a summary of the methods used by the author and co-authors. The aim of chapter 4 was to investigate the differentiation of a layered mafic sill called the 'Lower Pyramid Sill'. I carried out a detailed petrological, textural and in-situ mineral chemical investigation of the Lower Pyramid Sill. This study has important implications for how igneous layering forms. Chapter 5 expands on the Lower Pyramid Sill by examining sections through the same sill at other localities > 50 km along strike. Collectively this sill is called the 'Fort Collinson Sill Complex', because of its occurrence just above the prominent Fort Collinson Formation. Chapter 5 provides insights into how the magmatic plumbing system of the Franklin LIP differentiated. Chapter 5 also has implications for how other LIPs differentiate on a regional scale, as well as having implications for the origin of lateral compositional variability that is observed in igneous intrusions. Chapter 6 is an investigation of a fault-guided feeder dyke system known as the 'Southern Feeder Dyke Complex'. Chapter 6 is a field-based study, which includes small-scale geological maps that are central to the chapter. Chapter 6 has important implications for how Ni-Cu-PGE sulphide deposits form, specifically the role crustal sulphur assimilation in the generation of immiscible sulphide liquids. Chapter 6 also considers the regional economic potential of the Franklin LIP. Chapter 7 provides a short conclusion of the overall study.

### 1.4 Author contributions

### 1.4.1 Chapter 4

Chapter 4 has been accepted for publication in Journal of Petrology.

Ben Hayes: I sampled the Lower Pyramid Sill during the 2010 field season. I carried out all of the rock preparation (for whole-rock chemistry and thin sectioning), analytical and data collection work on the Lower Pyramid Sill samples. I conducted all of the geochemical modelling within the chapter. I wrote the chapter with guidance from both supervisors, which helped improve the flow, organisation and scientific impact of the study presented. All interpretations are my own and have not changed during editing from my supervisors.

Jean Bédard: Jean provided guidance during the 2010 field season and through many discussions, provided direction in terms of the modelling carried out in chapter 4. He edited earlier versions of the manuscript to help clarify the arguments that are presented.

Johan Lissenberg: Johan, as my supervisor based at Cardiff University, was available for discussions throughout the course of analytical work, data interpretation and modelling. Johan edited earlier versions of the manuscript and helped clarify data presentation and the scientific impact of the interpretations.

### 1.4.2 Chapter 5

Chapter 5 is currently under review for publication in Contributions to Mineralogy \& Petrology.

Ben Hayes: Chapter 5 builds on chapter 4, and therefore includes a large part of the data and interpretations presented in chapter 4, which are all my own work. Jean Bédard and myself, as well as Dick Naslund and Kathryn Steigerwaldt of Binghamton University sampled the sills of the Fort Collinson Sill Complex, during the 2010 field season. I carried
out all of the modelling and interpretations (except the radiogenic and stable isotopes) that are presented in chapter 5 . I wrote the chapter with guidance from the co-authors.

Johan Lissenberg: Johan provided guidance during the geochemical modelling that is included in the chapter. Johan edited earlier versions of the chapter that helped improve the clarity of the study.

Jean Bédard: Jean helped sample the sills of the Fort Collinson Sill Complex studied in chapter 5. Jean provided direction and guidance during the $\mathrm{Fe}=\mathrm{Mg}$ and trace element inversion modelling. Jean edited earlier versions of the chapter, which improved the flow and organisation of the chapter.

Charlie Beard: Charlie was part of the Victoria Island mapping team in 2011, specifically to carry out a radiogenic isotope investigation of the Franklin sills and Natkusiak flood basalts, principally to understand melt source evolution in the Franklin LIP. Charlie worked on samples from the Fort Collinson Sill Complex, and provided radiogenic isotope methods (chapter 3; section 3.4.1), data and modelling to the chapter. Charlie edited earlier versions of the chapter, helping to improve the clarity of the study.

### 1.4.3 Chapter 6

Chapter 6 has been submitted to Economic Geology for publication.

Ben Hayes: I spent a week mapping and sampling the Southern Feeder Dyke Complex in detail during the 2011 field season. I prepared the samples for whole-rock (including PGE) chemistry and thin sectioning. I undertook all of the petrographic work, data collection and the interpretations of the data. I wrote the chapter, with guidance from the co-authors.

Jean Bédard: Jean assisted in the field in 2011 by helping to plan the strategy for mapping and sampling the Southern Feeder Dyke Complex. Jean provided guidance during the writing of the chapter, which helped improve the flow and organisation of the study.

Matt Hryciuk: Matt assisted me with sampling during fieldwork on the Southern Feeder Dyke Complex. Matt undertook a sulphur isotope based thesis on rocks from the Minto Inlier at McGill University in Montréal. Matt provided details of the sulphur isotope methods (chapter 3; section 3.4.2). Matt provided sulphur isotope interpretations and he edited earlier versions of the chapter, which helped improve the clarity of the material presented, particularly the economic implications of the study.

William MacDonald: Bill spent 3 days sampling at the Southern Feeder Dyke Complex in 2011. Bill processed the samples for magnetic fabrics in his laboratory at Binghamton University, New York state. Bill provided details of the method he used (chapter 3; section 3.4.3) and interpreted the magnetic fabric data that is included in the chapter.

Johan Lissenberg: Johan edited earlier versions of the chapter, which greatly improved the flow, organisation and scientific impact of the study.

## 2. Franklin Large Igneous Province

### 2.1 Extent and age

The Franklin Large Igneous Province (LIP) extends for > 2500 km from the west coast of Greenland as far as the Great Bear Lake and Victoria Island in northern Canada (Fig. 2.1). The Franklin LIP is thought to have formed during Neoproterozoic break-up of Rodinia as Siberia separated from northern Laurentia (Heaman et al., 1992, Rainbird, 1993, Pehrsson \& Buchan, 1999, Ernst et al., 2008, Denyszyn et al., 2009). Geochemically correlative intrusions are thought to be present in Siberia (Ariskin et al., 2009). The emplacement of the Franklin LIP has been dated at between $\sim 723-716 \mathrm{Ma}$, using concordant $\mathrm{U} / \mathrm{Pb}$ ages on baddeleyite/zircon from the Coronation sills and Franklin sills (Heaman et al., 1992, Macdonald et al., 2010). The emplacement of the Franklin LIP has been correlated with the early stages of the Neoproterozoic Sturtian glaciation (Macdonald et al., 2010).


Fig. 2.1: The extent of the Franklin LIP across northern Canada. The Franklin LIP has a radiating dyke swarm, which has been used to infer the location of the melt source (red start located NW of Victoria Island: see Ernst et al. (2008) for details). Franklin sills, dykes and volcanics (Natkusiak) outcrop in the NW region of Victoria Island (western Canadian Arctic). The Coronation sills, on mainland Canada, are also thought to be related to the Franklin LIP (Shellnutt et al., 2004).

### 2.2 Minto Inlier

The Minto Inlier is located on Victoria Island (Fig. 2.2). The Minto Inlier is a NE-SW trending syncline of Neoproterozoic Shaler Supergroup sedimentary rocks (Rainbird, 1993). The Shaler Supergroup comprise intra-cratonic clastic and carbonate sediments that were deposited in the Amundsen Basin (Rainbird et al., 1996). The basin formed during break-up of Rodinia (Young, 1981). The Amundsen Basin is thought to represent an embayment along a larger epeiric basin within the Rodinia supercontinent (Young, 1981, Rainbird et al., 1996). This epeiric sea was episodically connected to open ocean via seaways (Williams \& Stelck, 1975).

The Shaler Supergroup was first defined by Thorsteinsson and Tozer (1962), after their investigations of the Minto Inlier. The Neoproterozoic Shaler Supergroup unconformably overlies Paleoproterozoic ( $\sim 1.9 \mathrm{Ga}$ ) sedimentary rocks of the Goulburn Supergroup, which themselves overlie granite of suspected late Archean age, on the northeast side of Minto Inlier (cf. Rainbird et al. 1996). The Shaler Supergroup is up to $\sim 5 \mathrm{~km}$ thick. It includes, in ascending stratigraphic order, the Rae Group, Reynolds Point Group, Minto Inlet Formation, Wynniatt Formation, Kilian Formation, and Kuujjua Formation (Thorsteinsson \& Tozer, 1962, Jefferson, 1985, Rainbird et al., 1996, Jones et al., 2010, van Acken et al., 2013, Thomson et al., 2014). These Formations generally consist of carbonate rocks with subordinate clastics and sulphate evaporites.

At the top of the Shaler Supergroup are the Natkusiak flood basalts, a sequence of $\sim 1 \mathrm{~km}$ thick lavas that are coeval to the Franklin LIP. These rocks were gently folded about ENEtrending axes, eroded, and then unconformably overlain by a Palaeozoic clastic and carbonate sequence (Thorsteinsson \& Tozer, 1962, Mathieu et al., 2013), itself dissected by a prominent ENE-trending normal fault system (Bédard et al., 2012, Dewing et al., 2013).

Franklin sills and feeder dykes intrude the Shaler Supergroup within the Minto Inlier. There are $\sim 20$ Franklin sills, which range in thickness from a few meters to $\sim 150 \mathrm{~m}$. The Franklin sills are mostly conformable to bedding (Fig. 2.3) and, where exposures permit, are seen to be laterally continuous for tens of kilometres along strike with little change in
thickness (Bédard et al., 2012). There are narrow ( $\sim 2-10 \mathrm{~m}$ ) contact metamorphic haloes around Franklin sills (Nabelek et al., 2013) with wider haloes enveloping some dykes (Hryciuk et al. Submitted).


Fig. 2.2: A simplified geological map of the Minto Inlier on Victoria Island, modified after Thorsteinsson and Tozer (1962), following fieldwork in 2010 and 2011. The localities of the rocks studied in this thesis are labelled. The Lower Pyramid Sill (LPS) is situated near the western end of Minto Inlet and is studied in detail in chapter 4. The LPS is part of the Fort Collinson Sill Complex (FCSC), which is a series of olivine-rich sills that are emplaced just above the Fort Collinson Formation marker unit (within the Reynolds Point Group). The FCSC includes the West Uhuk Sill (WUS) and P-sill (PS). The Uhuk Massif (UM) is a major fault-guided dyke feeder system (appendix A) that is thought to have fed the FCSC. The FCSC magmatic plumbing system is discussed in chapter 5. The Southern Feeder Dyke Complex (SFDC), studied in chapter 6 , is situated at the Wynniatt/Kilian Formation boundary, south of the Minto Inlet and $\sim 1 \mathrm{~km}$ stratigraphically beneath the Natkusiak Formation flood basalts. Other locations mentioned in this thesis are also labelled (SCD $=$ Sulphide City Dyke \& GS = Gossan Sill).

A new aeromagnetic map reveals prominent NW-trending magnetic lineaments (Kiss \& Oneschuk, 2010) that correspond to major topographic lows. These topographic lows locally contain breccias and cataclasites and have been interpreted to be faults (Bédard et al., 2012). Some of these magnetic lineaments correspond to sharp breaks in stratigraphy indicating east-side-down throws and exhibit associated structures indicating normal
motion synchronous with emplacement of magmas with Franklin affinities, and are therefore interpreted to be Neoproterozoic syn-magmatic normal faults. These field relationships are described in detail in Bédard et al. (2012) for the Uhuk Massif (appendix A). The Southern Feeder Dyke Complex (SFDC: Fig 2.2) exhibits very similar field relationships (described in chapter 6), with brecciated country rocks that are intruded by basaltic Franklin dykes, themselves brecciated due to Neoproterozoic syn-magmatic faulting.


Fig. 2.3: Field photo looking towards the SE showing a typical section view within the Minto Inlier of Victoria Island. Sills are brown/black in colour with orange lichen staining at their bases and mostly show poorly developed columnar jointing. Photo by Charlie Beard.

### 2.3 Natkusiak Flood Basalts

The Natkusiak flood basalts (Fig. 2.2), at the top of the Shaler Supergroup, are a $\sim 1 \mathrm{~km}$ thick sequence of basaltic volcanics that are coeval to Franklin sills (Baragar, 1976, Jefferson et al., 1985, Williamson et al., 2013). The Natkusiak flood basalts are preserved as two erosional remnants in the Holman Island syncline (Fig. 2.2). The Shaler Supergroup sedimentary rocks beneath the Natkusiak flood basalts were subjected to pre-magmatic uplift just prior to eruption, thought to be due to thermal doming above a mantle plume (Rainbird, 1993). The lowermost extrusive unit ( $\sim 50-100 \mathrm{~m}$ thick) is a primitive basalt ( $\sim 7-$
$11 \mathrm{wt} \% \mathrm{MgO}$ ) characterised by LREE (light-rare earth element)-LILE (large ion lithophile element)-enrichment, also referred to as Type-1 magmas (Bédard et al., 2013). Overlying this thin unit are laterally extensive basaltic $(\sim 6-10 \mathrm{wt} \% \mathrm{MgO})$ sheet flows with less strongly enriched L/HREE signatures, also referred to as Type-2 magmas (Bédard et al., 2013).


Fig. 2.4: A trace element discrimination plot showing $\mathrm{Ce} / \mathrm{Yb}$ versus Yb (in ppm ). The plot includes all liquid (i.e. sill and dyke chills and Natkusiak lavas) and quasi-liquid (i.e. dolerite and ferrogabbro) compositions that were sampled across the Minto Inlier during fieldwork in 2010 and 2011. These data reveal two populations of magma fed the Franklin LIP. Type-1 magmas are LREE-enriched (high $\mathrm{Ce} / \mathrm{Yb}$ ) while Type-2 magmas have low L/HREE ratios. Based on crosscutting relationships preserved in drill core (sampled by GNME in 2008), Type-1 magmas are older than Type-2 magmas.

The two populations (Type-1 and Type-2) of magma recognised in the Natkusiak volcanics can be correlated with those in the underlying sills. Both magma types form gabbroic/doleritic sills that exhibit systematic inward enrichment of $\mathrm{FeO}, \mathrm{TiO}_{2}$ and incompatible elements as a result of in-situ fractional crystallisation (Naslund et al., 2013). Type-1 sills (including the Lower Pyramid Sill and other sills of the Fort Collinson Sill Complex: chapter $4 \& 5$ ) mostly occur in the lower part of the Shaler Supergroup and are commonly characterised by olivine-rich bases and gabbroic upper parts. They correspond
geochemically to the lower primitive Natkusiak lavas. Intrusions formed from the younger, Type-2 magma population lack olivine-rich layers, but may be prominently plagioclaseclinopyroxene phyric or sub-ophitic in texture (including the majority of the Southern Feeder Dyke Complex: chapter 6).

## 3. Methods

### 3.1 Introduction

This chapter aims to give an overview of the methods used for the data that are presented in this thesis. Section 3.2 will explain the whole-rock chemistry methods utilised at Cardiff University, specifically for major, trace and platinum group elements (PGEs). A selection of samples that are presented in chapter 5 were analysed for whole-rock chemistry in laboratories in Canada, and a brief summary is provided. Section 3.3 will summarise the in-situ mineral chemistry methods used at Laval University in Québec City. Section 3.4 will summarise the methods used by co-authors for other data that form part of this thesis.

### 3.2 Whole-rock chemistry

Rock samples $(\mathrm{N}=81)$ were prepared for whole-rock analyses using the rock preparation facilities at Cardiff University. Initially, samples were trimmed using a diamond bladed mechanical rock saw in order to remove weathered edges. The rock samples were then dried before the next stage of the process. Chunks of each sample were crushed to coarse grit with a steel jaw crusher. Each sample was then reduced to a fine powder in a agate planetary ball mill. Approximately 2 g of each sample was ignited for two hours in a furnace at $\sim 900^{\circ}$ in order to drive off volatiles. This allowed the loss on ignition (LOI) values to be calculated.

Samples were then prepared as solutions for inductively coupled plasma-atomic emission spectrometry or ICP-OES (Jobin Yvon Horiba Ultima 2) and inductively coupled plasmamass spectrometry or ICP-MS (Thermo Elemental) analysis, following the lithium metaborate fusion method described in McDonald and Viljoen (2006). Approximately 0.1 +/- 0.001 g of each sample was mixed with $0.6 \mathrm{~g}+/-0.004 \mathrm{~g}$ of lithium metaborate flux in a platinum crucible. Lithium iodide wetting agent was added to each mixture, which was then fused with a Claisse Fluxy automated fusion system. Each mixture was then dissolved in a 50 ml solution of 20 ml of $10 \% \mathrm{HNO}_{3}$ and 30 ml of 18.2 deionised water obtained
from a Milli-Q purification system. Once the mixture had fully dissolved, 1 ml of 100 ppm Rh spike was added to the solution, which was then filled with 100 ml of 18.2 deionised water. The samples ( $\sim 20 \mathrm{ml}$ of each) were then run on the ICP-OES in order to obtain major element and some trace element ( $\mathrm{Sc}, \mathrm{V}, \mathrm{Cr}, \mathrm{Co}, \mathrm{Ni}, \mathrm{Cu}, \mathrm{Zn}, \mathrm{Sr}, \mathrm{Y}, \mathrm{Zr}, \mathrm{Ba}$ ) abundances. An aliquot of 1 ml of each solution was added to 1 ml of In and Tl and 8 ml of $2 \% \mathrm{HNO}_{3}$ and run on the ICP-MS to obtain trace element abundances (V, $\mathrm{Cr}, \mathrm{Co}, \mathrm{Ni}, \mathrm{Cu}$, Zn. Ga, Rb, Sr, Y, Zr, Nb, Sn, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu, Hf, $\mathrm{Ta}, \mathrm{Pb}, \mathrm{Th}, \mathrm{U}$ ). Dr. Iain McDonald and Dr. Ley Wooley operated the ICP-OES and ICPMS. In order to assess the accuracy and precision of the whole-rock data, external and internal standards were analysed. The external standards were JB-1a and NIM-G (included in appendix E1). The percentage errors of the measured values compared with the certified values are included in appendix E1.

A suite of samples $(\mathrm{N}=51)$ presented in chapter 5 were analysed in IRNS-ETE laboratories in Québec City, Canada. Rock samples were initially trimmed and then reduced to a coarse grit with a steel jaw crusher before being powdered with a planetary agate mortar. Sample fusion was operated on a Claisse fluxer using $\mathrm{LiBO}_{2}$ and LiBr . The $\mathrm{HNO}_{3}-\mathrm{HCl}$ sinter technique was employed to circumvent potential problems associated with high field-strength elements (HFSE) and rare-earth elements (REE) in refractory minerals. The major elements and $\mathrm{Ba}, \mathrm{Sr}, \mathrm{Sc}, \mathrm{V}, \mathrm{Co}, \mathrm{Cr}, \mathrm{Cu}, \mathrm{Ni}$ and Zn compositions were determined by ICP-OES. Trace elements were determined by ICP-MS using a similar method to that described by Varfalvy et al. (1997). Full details of the detection limits, and the accuracy and precision of the data are summarised in Leclerc et al. (2011).

A suite of samples $(\mathrm{N}=16)$ that are presented in chapter 6, were analysed for platinum group elements (PGE) at Cardiff University. PGE concentrations for each sample were obtained by Ni sulphide fire assay, followed by ICP-MS following the method of Huber et al. (2001) and McDonald and Viljoen (2006).

### 3.3 Mineral chemistry

In-situ mineral chemical analyses were carried out using a Cameca SX-100 Electron Microprobe at Laval University in Québec City, Canada. The probe was fitted with 5 wavelength-dispersive spectrometers. All phases were analysed on polished thin sections using a focused beam ( $1 \mu \mathrm{~m}$ diameter), 20 nA current and a 15.0 kV accelerating potential. Counting times were typically 30 seconds on peak and background levels. Core to rim profiles were measured on the major silicate phases. Where grain size was very fine or only a representative core composition was required, single spot analyses were obtained. The dataset for each phase analysed and the analytical conditions for each mineral species are supplied in appendix C. Preparation before undertaking the data collection using the Electron Microprobe was carried out using the Scanning Electron Microscope (Cambridge 360 instrument) at Cardiff University.

### 3.4 Other methods used by co-authors

### 3.4.1 Radiogenic lead isotopes

A suite of samples $(\mathrm{N}=6)$ that are presented in chapter 5 were analysed for radiogenic Pb isotopes at the Pacific Centre for Isotopic and Geochemical Research (PCIGR), University of British Columbia, Vancouver, by chapter 5 co-author, Charlie Beard. Rock samples were carefully prepared before being crushed to gravel and then milled to a fine powder. 100 g of gravel for each sample was milled in agate jars containing agate marbles. Doleritic samples were dissolved using the hot plate method as described in Weis et al. (2006), whilst olivine-rich rocks were digested in high-pressure bombs to produce dissolution of resistant accessory phases such as Zircon (Pretorius et al., 2006, Weis et al., 2006). Elements were separated and purified before isotopic analyses using ion exchange chromatography. Pb was isolated by a two-pass ion exchange chromatography separation method, described by Nobre Silva et al. (2009). Pb-isotopic ratios were corrected for instrumental drift using a Ln-Ln trend function, determined by bracketed analyses of NBS 981 Pb standard solution. External reproducibility was ensured by measurement of USGS reference materials BCR-2 Columbia River Basalt and G-2 Granite, normalised
values for which were within the $2 \sigma$ range of historical values within PCIGR. Internal reproducibility was tested by multiple replicate and complete procedural duplicate analyses. Pb-isotopic compositions were age-corrected to initial ratios at 723 Ma . Full details of the method used are available in Beard (2012) and a more comprehensive description of laboratory methods can be found in Dell'Oro et al. (2012).

### 3.4.2 Sulphur isotopes

A suite of samples $(\mathrm{N}=15)$ presented in chapter 5 and a suite of samples $(\mathrm{N}=32)$ presented in chapter 6 were analysed for sulphur isotopes in the laboratories at McGill University, Montréal. Sulphur isotope powders were ground using case hardened steel and tungsten carbide mills. Sulphate evaporites were extracted using a boiling Thode reduction solution (Forrest \& Newman, 1977). All other samples were extracted using a Cr-reducible solution (Canfield et al., 1986). $\mathrm{H}_{2} \mathrm{~S}$ was reacted with zinc acetate and $\mathrm{AgNO}_{3}$ to produce an $\mathrm{Ag}_{2} \mathrm{~S}$ precipitate. This precipitate was weighed into packets, which were fluorinated overnight in nickel bombs to produce $\mathrm{SF}_{6}$. Analysis was completed with a Thermo Finnigan MAT 253 mass spectrometer using a dual inlet method monitoring ion beams at a mass/charge ratio of 127-131. Analytical reproducibility $(2 \sigma)$ for the full measurement procedure from rock powders to the final mass spectrometer data for $\delta^{34} \mathrm{~S}$ is estimated to be better than $0.2 \%$. All data is reported relative to the Vienna Canyon Diablo Troilite standard (VCDT). Full details of the method used are available in Hryciuk et al. (Submitted).

### 3.4.3 Anisotropy of magnetic susceptibility fabrics

Oriented blocks were taken at 17 sites along the Southern Feeder Dyke Complex (chapter 6) using magnetic and solar compasses. Eleven sites were in dykes, and 6 were in adjacent sills. Typically 5 cores were drilled from each block, producing 85 oriented cores. The anisotropy of magnetic susceptibility (AMS) of each core was measured using the AGICO KLY-3 Kappabridge susceptibility metre. Prof William MacDonald (co-author on chapter 6) carried out this work in his laboratory at Binghamton University, New York state.

# 4. Olivine slurry replenishment and the development of igneous layering in a Franklin sill 

### 4.1 Abstract

The Franklin sills and dykes on Victoria Island in the Canadian Arctic represent the subvolcanic plumbing system to the Natkusiak flood basalts, which are associated with late Neoproterozoic (ca. 723-716 Ma) break-up of Rodinia. The Lower Pyramid Sill (LPS) is the distal end of a sill complex that may be rooted in the Uhuk Massif, a major faultguided feeder system. The LPS is unusual for a thin ( $c a .21 \mathrm{~m}$ ), shallow, tholeiitic intrusion because it displays well-developed cumulate layering similar to that seen in large layered intrusions. The LPS has an aphanitic, olivine-phyric (ca. 5\%) Lower Chilled Margin (LCM); a ( $<1 \mathrm{~m}$ thick) dendritic, olivine-phyric Lower Border Zone (LBZ); an (ca. 7 m thick) olivine-dominated (up to $c a .55 \%$ ) melagabbro/feldspathic-peridotite zone (OZ); a thin ( $c a .1 \mathrm{~m}$ ) clinopyroxene-rich cumulate gabbro (CPZ) containing sector-zoned euhedral clinopyroxene; a (ca. 10 m thick) doleritic gabbro zone (DZ); an (<1 m thick) aphyric, dendritic Upper Border Zone (UBZ); and an aphanitic, olivine-phyric (ca. 5\%) Upper Chilled Margin (UCM). Distinct compositional groups recognized in olivines from the OZ can be associated with specific crystal morphologies, some showing significant reverse zoning. Melt compositions were calculated through application of the olivine-melt $\mathrm{Fe}=\mathrm{Mg}$ exchange coefficient. The calculations suggest that phenocrystic and primocrystic olivine ( $\mathrm{Fo}_{88-82}$ ) in the LCM/LBZ and lower OZ formed from melts with $c a .13-10 \mathrm{wt} \% \mathrm{MgO}$. Modeling implies that reversely zoned olivine primocrysts and chadacrysts have rims in equilibrium with melts of $c a .10-8 \mathrm{wt} \% \mathrm{MgO}$ that were saturated only in olivine (+ minor chromite), whereas some olivine cores formed from melts as evolved as ca. 6-5 $\mathrm{wt} \% \mathrm{MgO}$ that would have coexisted with a gabbroic assemblage. The presence of multiple olivine populations in the OZ (some reverse zoned) indicates that the LPS did not crystallize from a single pulse of melt that evolved by closed-system fractional crystallization. We propose that the reverse zoning pattern records incorporation of evolved crystals, most derived from the mushy gabbroic host, when an olivine-charged replenishment under/intraplated
the partly crystallized basaltic magma, now preserved as the DZ. The intervening CPZ may also owe its origin to the emplacement of the olivine slurry; possibly as a result of porescale melt mixing at this interface. The DZ shows inward differentiation trends that can be explained by in-situ differentiation. The data implies that late emplacement of olivine-rich crystal-slurries and in-situ differentiation both played a role in the development of the layered LPS.

### 4.2 Introduction

There is considerable petrological and geochemical evidence from active volcanoes for periodic replenishment of high-level magma chambers by the influx of new magma from depth (Murphy et al., 2000, Tepley et al., 2000, Garcia et al., 2003, Humphreys et al., 2006, Morgan et al., 2006, Ginibre \& Wörner, 2007). Replenishment is also well documented in layered intrusions (Wager \& Brown, 1968, Huppert \& Sparks, 1980, Ballhaus \& Glikson, 1989, Cawthorn, 1996, Wallace \& Bergantz, 2002, Namur et al., 2010). Influxes of replenishing primitive magma can: (1) prolong the life of a magma chamber (Usselman \& Hodge, 1978, Annen et al., 2014); (2) modify chemical fractionation trends (O'Hara, 1977, Dungan \& Davidson, 2004, Reubi \& Blundy, 2008); (3) affect the cumulate stratigraphy in layered intrusions (Wager \& Brown, 1968, Cawthorn, 1996, Gibb \& Henderson, 2006); and (4) trigger eruption (Brown, 1956, Sparks et al., 1977, Field et al., 2013). It is widely believed that primitive replenishments play a role in the formation of cyclic/macrorhythmic layering in mafic-ultramafic intrusions (Brown, 1956, Irvine \& Smith, 1967, Wager \& Brown, 1968, Irvine, 1977, Huppert \& Sparks, 1980, Smewing, 1981).

Huppert and Sparks (1980) developed a model to explain cyclic olivine-rich/plagioclaserich macrorhythmic layers by ponding of dense, hot, primitive magma recharges beneath a resident, more evolved basaltic magma. In their model (as applied to the Rum intrusion), the contrast in composition and temperature between the two liquid layers drove vigorous convection, holding olivine in suspension in the lower layer until its eventual en masse deposition formed an olivine-rich macro layer. Continuous re-equilibration with the convecting melt prior to deposition would cause olivine compositional variations to approach end-member equilibrium crystallisation paths (Tait, 1985), and explain a lack of
cryptic olivine compositional variation in the Rum olivine-rich macro layers. More recent reviews on the mechanisms of magma influx during replenishment, and the development of igneous layering, can be found in Gorring and Naslund (1995), Cawthorn (1996), Puffer et al. (2009) and Zieg and Marsh (2012).

Recent discussion about how large-scale cyclic/macrorhythmic olivine-rich/plagioclaserich layers form has been strongly polarised. One school of thought advocates the late emplacement of crystal-laden slurries into pre-existing, partly crystallised intrusions (Marsh, 2004, Marsh, 2013). Another favours the emplacement of largely aphyric magmas that evolve by fractional crystallisation to produce in-situ differentiation trends (Latypov, 2009). In this chapter, we present data that bear on this debate, showing that olivineenriched basal layers in the Franklin sills of Victoria Island formed as a result of slurry injection, much as proposed by Marsh $(2004,2013)$, but that the overlying gabbroic rocks are internally differentiated, as advocated by Latypov (2009) and others (Shirley, 1985, Parsons, 1986, Naslund, 1989).

Slow cooling of large intrusions allows significant post-cumulus textural (Hunter, 1996, Holness et al., 2007) and compositional re-equilibration (Barnes, 1986) that may obscure primary magmatic signatures. This is especially true for olivine due to the rapid exchange of Fe and Mg between crystal and melt at magmatic temperatures (Chakraborty, 1997), whereas 'primary magmatic' zoning patterns of clinopyroxene (Müller et al., 2013) and especially plagioclase (Grove et al., 1984, Morse, 1984) seem to better survive the postcumulus stage. In contrast, thin (< 100 m ) intrusions would cool quickly (Carslaw \& Jaeger, 1959) and might better preserve the textural and mineral-chemical evidence of the magmatic processes by which they formed.

The sills constituting the Neoproterozoic Franklin sub-volcanic plumbing system on Victoria Island are examples of such thin, quickly cooled systems. These sills are well exposed, unmetamorphosed and the suite has a capping, cogenetic flood basalt sequence (Natkusiak flood basalts). The excellent preservation and availability of liquid compositions allow the physical and chemical linkages between sills, feeder/transfer dykes and the capping Natkusiak flood basalts to be established. In this chapter, we reconstruct the differentiation history of the Lower Pyramid Sill (LPS), a thin ( $\sim 21 \mathrm{~m}$ ) sill that is prominently layered, with an olivine-enriched lower layer and an olivine-poor gabbroic
upper layer. Our textural and mineral-chemical observations from the LPS are best explained by mixing between a resident gabbroic mush with a late invading olivine slurry, and we propose that this is how the first-order olivine-rich/plagioclase-rich macro layering formed.

### 4.3 Lower Pyramid Sill

The LPS (Fig. 4.1a) is located near Boot Inlet at the western end of the Minto Inlier (Fig. 2.2). It formed from the LREE-LILE-enriched magma population (Type-1 magma). The LPS was emplaced at a depth of $\sim 3 \mathrm{~km}$ based on its stratigraphic position. It is one of many olivine-rich sills that were emplaced just above the Fort Collinson Formation quartzarenites within a broad corridor along the Collingwood hills. Collectively these olivineenriched sills are known as the Fort Collinson Sill Complex (discussed in chapter 5). The thin ( $\sim 50-100 \mathrm{~m}$ ) Fort Collinson marker unit is embedded in much thicker sequences of dolostone and limestone, and this mechanical discontinuity appears to have greatly facilitated intrusion of the early Type-1 magma pulse in the Franklin province. The western, sill-like part ( $\sim 40 \mathrm{~m}$ thick) of the Uhuk Massif (UM), located $\sim 50 \mathrm{~km}$ to the east (Fig. 2.2: cf Bédard et al., 2012), is another example of an olivine-enriched sill emplaced at this stratigraphic position. The LPS has a thin (ca. 5-6 m) metamorphic halo surrounding it, similar to other Franklin sills (Nabelek et al., 2013).

The LPS is unusual for such a thin hypabyssal intrusion in having a well-defined basal olivine-rich cumulate layer, which is overlain by a sub-ophitic doleritic gabbro (Fig. 4.1b). The LPS has chilled contacts against a thin septum of fissile Jago Bay Formation carbonates, only a few meters above the top of the Fort Collinson quartz-arenites (Fig. 4.1a). The LPS (Fig. 4.1b) has an aphanitic Lower Chilled Margin (LCM); a thin (< 1 m ) dendrite-textured Lower Border Zone (LBZ); a $\sim 7 \mathrm{~m}$ thick sequence of olivine cumulate melagabbro to feldspathic-peridotite with $\sim 40-55 \%$ modal olivine (OZ); a $\sim 1 \mathrm{~m}$ thick layer of clinopyroxene-rich gabbro (CPZ); a $\sim 10 \mathrm{~m}$ thick layer of sub-ophitic doleritic gabbro (DZ); a thin (< 1 m ), dendrite-textured Upper Border Zone (UBZ); and a thin ( $<1 \mathrm{~m}$ ) aphanitic Upper Chilled Margin (UCM). We have taken 26 samples from a single LPS profile with sample locations constrained by a global positioning system, and height above the lower contact measured in the field (data in appendix E1). We analysed all samples for
their whole-rock major element concentrations (Fig. 4.1b) and 22 for mineral compositions.

(b)


Fig. 4.1: (a) Field photo of the LPS, with a lower contact just above the Fort Collinson Formation, which is a prominent marker unit. The upper contact is abutted against the Jago Bay Formation. The Upper Pyramid Sill (UPS) is a younger, porphyritic Type-2 sill (unenriched L/HREE magma). In (b), the stratigraphy of the LPS is shown, along with MgO and $\mathrm{TiO}_{2}$ whole-rock compositions (in wt $\%$ ). MgO is highest (up to $\sim 25 \mathrm{wt} \%$ ) in the OZ , which is dominated by olivine. The DZ has lower MgO contents, similar to the chill ( $\sim 10 \mathrm{wt} \% \mathrm{MgO}$ ) and higher $\mathrm{TiO}_{2}$ contents in comparison to the LCM/OZ. The red dashed lines show the mean sill composition. All whole-rock data is provided in appendix E1.

### 4.4 Results

### 4.4.1 Petrography

### 4.4.1.1 Lower Chilled Margin (LCM)

The LCM (sampled at $\sim 5 \mathrm{~cm}$ above the contact) has $\leq 5 \%$ euhedral olivine phenocrysts $(\sim 0.5-2 \mathrm{~mm})$, which may contain tiny chromite microphenocrysts as inclusions. The olivine phenocrysts are surrounded by a very-fine grained ( $\sim 0.1 \mathrm{~mm}$ ) groundmass (Fig. 4.2a) that grades from massive aphanitic material to a felted mass of dendritic plagioclase and clinopyroxene.

### 4.4.1.2 Lower Border Zone (LBZ)

In the LBZ, the proportion of olivine phenocrysts ( $\sim 1-2 \mathrm{~mm}$, also with minor Cr -spinel inclusions) increases to $\sim 10-15 \%$ (at $\sim 20 \mathrm{~cm}$ above the lower contact). The dendritic intergrowths of plagioclase and clinopyroxene reach $\sim 1-2 \mathrm{~mm}$ grain sizes with dendrites commonly nucleating on the surface of olivine phenocrysts (Fig. 4.2b). The rims of some olivine crystals are intergrown with the plagioclase-clinopyroxene dendrites suggesting that the 3-phases cosaturated during crystallisation of the LBZ (Fig. 4.2b). The textures, with the presence of cumulus olivine, are consistent with a weakly olivine-cumulative composition for the LBZ.

### 4.4.1.3 Olivine Zone (OZ)

There is a noticeable textural change $\sim 1 \mathrm{~m}$ above the lower contact of the LPS. At this height, modal olivine increases over a few cm to $\sim 40 \%$ and the groundmass texture changes from dendritic to medium-grained poikilitic. We define this modal and textural transition as the boundary between the LBZ and the OZ. The OZ is $\sim 7 \mathrm{~m}$ thick and is composed of cumulate olivine melagabbro to feldspathic-peridotite. Cr-spinel microphenocrysts are present throughout the OZ, typically as inclusions in olivine primocrysts and as grains embedded in interstitial clinopyroxene and plagioclase.


Fig. 4.2: Photomicrographs of the textures observed in the LCM and LBZ. (a) At the lower contact, euhedral olivine (Ol) phenocrysts reside in a fine-grained quenched groundmass (Gm) (image in plane-light). (b) ~20 cm above the lower contact within the LBZ, the textures grade into coarser dendrites of plagioclase (Plag) and clinopyroxene ( Cpx ) that cement olivine phenocrysts (Olph). Olivine ( Ol ) is also intergrown with the dendrites suggesting the melt may have reached the 3-phase cotectic early in its crystallisation history (image in crossed-nicols).


Fig. 4.3: Backscattered electron (BSE) images showing LOZ olivine morphologies. (a) A clinopyroxene oikocryst (Pcpx) encloses olivine chadacrysts (Olch), and is surrounded by anhedral, interstitial plagioclase ( Pl ), which encloses groundmass olivines (Olpl). (b) Euhedral and clustered groundmass olivines (Olpl) enclosed by anhedral, interstitial plagioclase ( Pl ). The edge of a large, normally zoned olivine primocryst (Olpr) is also visible.

Olivine within the OZ can be broadly assigned to four different textural groups: (1) euhedral olivine primocrysts ( $c a .0 .5-2 \mathrm{~mm}$ ), which commonly contain euhedral Cr-spinel microphenocrysts (Fig. 4.3b \& 4.4a); (2) chadacrystic fine-grained (ca. 0.2-0.5 mm) euhedral-subhedral olivine crystals (some bearing Cr-spinel inclusions), which are enclosed by clinopyroxene oikocrysts (Fig. 4.3a, 4.4a \& 4.5f); (3) fine-grained (ca. 0.2-0.5 mm ) groundmass olivine crystals (some bearing Cr -spinel inclusions) surrounded by interstitial plagioclase (Fig. 4.3b, 4.4a \& 4.5f); (4) hopper olivines (ca. 0.2-2 mm), which are hopper in habit in cross-section and orthorhombic in basal sections where it appears to enclose clinopyroxene and plagioclase (Fig. 4.4d, e \& f). Hopper olivines also have anhedral margins indicating co-crystallization with plagioclase (Fig. 4.4d \& e). We subdivide the OZ based on the spatial distribution of these olivine textures into: a basal pyroxene-poikilitic olivine-melagabbro, which mainly consists of euhedral olivine primocrysts, olivine chadacrysts and groundmass olivines (LOZ; Fig. 4.3a, 4.4a \& 6), with only minor ( $c a .5 \%$ ) proportions of fine-grained ( $<0.5 \mathrm{~mm}$ ) hopper olivine; a middle subzone dominated by hopper olivine crystals that are typically intergrown with clinopyroxene, with subordinate primocrystic, chadacrystic, groundmass and hopper olivine (MOZ; Fig. 4.4b \& c); and an upper subzone (UOZ) dominated by hopper olivine morphologies with subordinate euhedral primocrystic and groundmass olivines that lacks olivine chadacrysts in poikilitic clinopyroxene (Fig. 4.4d, e \& f).

Clinopyroxene ( $\sim 1-2 \mathrm{~mm}$ ) habits range from euhedral to subhedral in the LOZ and it commonly encloses olivine chadacrysts (Fig. 4.3a). Clinopyroxene becomes more subhedral to anhedral in the MOZ, where intergrowths with olivine are common (Fig. 4.4b). In the UOZ, euhedral cumulus clinopyroxene crystals appear ( $\sim 1-2 \mathrm{~mm}$; Fig. 4.4d, f), substituting for now rare clinopyroxene oikocrysts that typify the LOZ/MOZ. Subordinate inter-granular, interstitial clinopyroxene occurs throughout the OZ.

Plagioclase is anhedral throughout the OZ and is interstitial to cumulus olivine and mostly interstitial to clinopyroxene, except for local co-crystallisation textures (Fig. 4.4a, c, d). There are minor accessory amphibole and biotite, and there are traces of groundmass $\mathrm{Fe}-\mathrm{Ti}$ oxides and $\mathrm{Fe}-\mathrm{Ni}-\mathrm{Cu}$ sulphides.


Fig. 4.4: Photomicrographs of OZ textures (all in crossed-nicols). (a) An image from the LOZ showing an olivine primocryst (Olpr) and poikilitic clinopyroxene (Pcpx), which encloses smaller olivine chadacrysts (Olch). Finer-grained olivine crystals (Olpl) can also be seen, surrounded by interstitial plagioclase (Pl). (b) An image from the MOZ showing a hopper olivine (Hol), which is overgrown by clinopyroxene (Cpx). (c) Image showing the abundant olivine of the MOZ, with olivine primocrysts (Olpr) and hopper olivines (Hol). (d) Hopper olivines (Hol) in the UOZ. (e) UOZ hopper olivines (Hol) that are intergrown with interstitial plagioclase ( Pl ). (f) UOZ hopper olivine ( Hol ) that appears to enclose plagioclase and clinopyroxene $(\mathrm{Pl}+\mathrm{cpx})$. Euhedral clinopyroxene crystals $(\mathrm{Cpx})$ are also present as well as interstitial plagioclase $(\mathrm{Pl})$.


Fig. 4.5: Back scattered electron (BSE) image and element maps of a clinopyroxene oikocryst from the LOZ with olivine chadacrysts $\left(\mathrm{Fo}_{83-81}\right)$ : (a) BSE image showing analysis spot numbers (core composition is $\mathrm{Mg} \# 85$ ), (b) Fe , (c) Cr , (d) Ti , (e) Al, (f) Mg. The clinopyroxene oikocryst appears to display sector-zoning, visible in the Mg and Al element maps. The clinopyroxene oikocryst also has a narrow $\mathrm{Fe}-\mathrm{Ti}$ enriched rim (white arrows in d). Interstitial plagioclase is visible in (e). The Fe-rich groundmass olivines are hosted by interstitial plagioclase. $\mathrm{N}=$ north; $\mathrm{S}=$ south $=\mathrm{E}=$ east. $\mathrm{W}=$ west. Additional element maps are included in appendix E7-1.


Fig. 4.6: BSE and element maps of sector-zoned clinopyroxenes from the CPZ: (a) BSE, (b) Ti , (c) Cr , (d) Ca , (e) Mg , (f) Al. Zones characterise the hourglass style sector-zoning with 2-3 times more enrichment in both Cr and Al . Additional element maps are included in appendix E7-1.

### 4.4.1.4 Clinopyroxene Zone (CPZ)

The CPZ is $\sim 1 \mathrm{~m}$ thick. The diffuse base of the CPZ is characterised by a drop in the mode of olivine (to $\sim 5 \%$ ), and a corresponding increase in the proportion of euhedral cumulus clinopyroxene $(\sim 0.5-1 \mathrm{~mm})$ to $\sim 30-40 \%$. The euhedral clinopyroxene crystals are often sector-zoned (Fig. 4.6) and are distributed as trains of grains that may record its formation in some type of channel (Fig. 4.7). Between these trains of sector-zoned clinopyroxene is a gabbroic matrix consisting of serpentinised olivine ( $\sim 5 \%$ ), euhedral-subhedral clinopyroxene and interstitial plagioclase with subordinate interstitial clinopyroxene. There is minor ( $<5 \%$ ) magnetite and ilmenite, as well as accessory amphibole and biotite.


Fig. 4.7: A digitised version of a thin section from the CPZ ( $\sim 8 \mathrm{~m}$ ), which shows how the sector-zoned clinopyroxene (Sz-cpx) may form an interconnected network. The lighter grey is the gabbroic matrix (G) consisting of cumulus clinopyroxene (+ low-Ca pigeonite) and interstitial plagioclase, and minor (<5\%) FeTi oxides. The gabbroic matrix also includes minor ( $\sim 5 \%$ ) olivine ( Ol ), which has been completely serpentinised. The darker grey surrounding the sector-zoned clinopyroxene outlines the morphology of these 'trains'.

The DZ is characterised by the absence of olivine and modal increases in plagioclase (from $\sim 25 \%$ in the OZ to $\sim 40 \%$ in the DZ) and clinopyroxene (from $\sim 30 \%$ in the OZ to $\sim 40 \%$ in the DZ ). The DZ is also marked by a change in the morphology of plagioclase, from subhedral-anhedral interstitial textures to coarser ( $\sim 2-3 \mathrm{~mm}$ ) lath-shaped crystals that mostly form doleritic intergrowths with clinopyroxene. Around 5-10\% of the plagioclase/clinopyroxene mode is comprised of cumulus crystals that do not display intergrowth textures. Full thin section scans show domains of sub-ophitic dolerite with varying grain sizes (varitextured), with coarser pegmatitic patches ( $\sim 5 \mathrm{~mm}$ plagioclase \& $\sim 3-4 \mathrm{~mm}$ clinopyroxene) and finer-grained domains ( $\sim 2-3 \mathrm{~mm}$ plagioclase \& $\sim 1-2 \mathrm{~mm}$ clinopyroxene) present in the DZ.

Clinopyroxene is typically euhedral-subhedral and intergrown with plagioclase, but both augite and low-Ca pigeonite also occur as separate cumulus crystals. Low-temperature alteration is more common in the DZ compared with other zones of the LPS, with plagioclase and clinopyroxene being partly altered to fine-grained albite and saussurite, or chlorite and actinolite (respectively). This is interpreted to represent deuteric alteration caused by volatile exsolution during the final stages of LPS crystallisation. Olivine was not observed in the DZ, but sparse orthopyroxene grains ( $\sim 5-10 \%$ ) in the lower DZ may be pseudomorphous after olivine. Interstitial ilmenite and magnetite is present $(\sim 5-10 \%)$ as are interstitial sulphides ( $\sim 5 \%$ ), with associations of pyrrhotite-pyrite in the lower DZ grading into associations of pyrite-chalcopyrite in the upper DZ. Minor accessory amphibole and biotite occur throughout the DZ, and may be associated with interstitial granophyre in the uppermost DZ ( $\sim 19-20 \mathrm{~m}$ height). Also, in the uppermost DZ, long ( $\sim 5-$ 10 mm ), blade-like clinopyroxene crystals are present with preferred growth directions roughly perpendicular to the upper contact of the LPS (element maps are included in the appendix E7-1).

### 4.4.1.6 Upper Border Zone (UBZ)

The UBZ (from $\sim 20 \mathrm{~m}$ to $\sim 21 \mathrm{~m}$ height) consists largely of fine-grained ( $<1 \mathrm{~mm}$ ) plagioclase and clinopyroxene dendrites. There are also coarser subhedral-anhedral
clinopyroxene grains ( $\sim 1-2 \mathrm{~mm}$ ), which shows both sector-zoning and complex concentric zoning, locally with sieve-textures (element maps included in appendix E7-1). No olivine phenocrysts (nor its pseudomorphs) were observed in the UBZ.

### 4.4.1.7 Upper Chilled Margin (UCM)

The LPS at its upper contact with the Jago Bay Formation carbonate is similar to the LCM in that it is a very fine grained, almost glassy, chilled contact. It contains $\sim 5 \%$ olivine phenocrysts ( $\sim 1-2 \mathrm{~mm}$ ) that have been completely pseudomorphed by secondary serpentine. The groundmass appears to consist of very fine-grained dendrites, but it is more altered in comparison to the LCM. A few amygdules (<5\%; $\sim 2 \mathrm{~mm}$ ) are filled with biotite and chlorite.

### 4.4.2 Mineral chemistry

The main silicate phases present in each layer in the LPS were analysed using the electron microprobe. Chemical zonations were mapped to constrain the crystallisation history of the LPS. We use 'core' when referring to the central region of a crystal; 'mantle', for the intermediate part between the core and rim (often with weak compositional gradients); and 'rim' for the strongly zoned edges of a crystal. We use 'primocryst' when we refer to euhedral cumulus crystals, which are typically $\geq 1 \mathrm{~mm}$ in size, and 'phenocryst' for euhedral crystals in the chills and border zones.

### 4.4.2.1 Olivine

Compositional variations of olivine in the LCM, LBZ and OZ are summarised in Fig. 4.8. Variations in Fo (molar Fo $=100 \times \mathrm{Mg} /\left[\mathrm{Mg}+\mathrm{Fe}^{2+}\right]$ ) are typically correlated with olivine NiO ( $\mathrm{wt} \%$ ) contents.

Olivine phenocrysts in the LCM are normally zoned with near-constant core and mantle compositions and narrow, evolved rims. The average core and rim compositions are $\mathrm{Fo}_{87}$ and $\mathrm{Fo}_{82}$, respectively. The cores and mantles of olivine phenocrysts in the LCM have an average NiO content of $0.32 \mathrm{wt} \%$, whilst the rims have a lower NiO content of $0.28 \mathrm{wt} \%$.

Some euhedral olivine phenocrysts show subtle core to mantle increases in Fo-content ( $\sim 0.3 \& 0.4 \mathrm{~mol} \%$ increases) followed by decreasing Fo at the rim. A small olivine phenocryst in the LCM displays unusually strong zoning from $\mathrm{Fo}_{86}(0.28 \mathrm{wt} \% \mathrm{NiO})$ in the core to $\mathrm{Fo}_{75}(0.24 \mathrm{wt} \% \mathrm{NiO})$ at its rim.


Fig. 4.8: Stratigraphic column showing olivine compositional variations of molar forsterite (Fo) content, through the $\mathrm{LCM}, \mathrm{LBZ}$ and OZ , which is further subdivided into the lower $\mathrm{OZ}(\mathrm{LOZ})$; middle $\mathrm{OZ}(\mathrm{MOZ})$; and upper OZ (UOZ). The grey box covers the compositional range of olivines that are enclosed by clinopyroxene, which is tightly clustered around $\mathrm{Fo}_{83-81}$. 'This melt' refers to the lower chill composition. $\mathrm{NiO}(\mathrm{wt} \%)$ olivine compositions are closely coupled with olivine Fo-compositions. NiO ( $\mathrm{wt} \%$ ) contents are coupled with Fo, in that they show the same zoning patterns. NiO data is provided in appendix xx .

In the LBZ, the average core composition of olivine phenocrysts is $\mathrm{Fo}_{85}(0.29 \mathrm{wt} \% \mathrm{NiO})$. Many olivine phenocrysts in the LBZ show weak reverse zoning from core to mantle ( $\sim 1$ $\mathrm{mol} \%$ ), but with little corresponding change in the NiO content (average $\sim 0.32 \mathrm{wt} \%$ ). Olivine commonly shows strong normal zoning with an average rim composition of $\mathrm{Fo}_{75}$ ( $0.2 \mathrm{wt} \% \mathrm{NiO}$ ).

In the LOZ, olivine primocrysts have core compositions ranging between $\mathrm{Fo}_{88}$ and $\mathrm{Fo}_{82}$ (0.4-0.25 wt\% NiO). Smaller olivine chadacrysts (enclosed by clinopyroxene oikocrysts) have systematically lower Fo-contents and a restricted range of compositions with an average core composition of $\mathrm{Fo}_{83}(0.25 \mathrm{wt} \% \mathrm{NiO})$ and an average rim composition of $\mathrm{Fo}_{81}$ ( $0.2 \mathrm{wt} \% \mathrm{NiO}$ ). Groundmass olivine surrounded by plagioclase has still more evolved compositions with average core and rim compositions of $\mathrm{Fo}_{81}(0.21 \mathrm{wt} \% \mathrm{NiO})$ and $\mathrm{Fo}_{70}$ ( $0.1 \mathrm{wt} \% \mathrm{NiO}$ ), respectively. Most olivines in the LOZ are zoned. The larger olivine primocrysts commonly exhibit normal zoning with near-constant core and mantle compositions (Fig. 4.8), and narrow Fe-rich rims as evolved as $\mathrm{Fo}_{77-75}$. Chadacrystic olivines enclosed by clinopyroxene oikocrysts are mostly normally zoned (typically a $\sim 1-$ $2-\mathrm{mol} \%$ change from core to rim) except for a weakly reverse-zoned olivine grain, which has a core composition of $\mathrm{Fo}_{82}(0.2 \mathrm{wt} \% \mathrm{NiO})$ that increases to $\mathrm{Fo}_{83}(0.24 \mathrm{wt} \% \mathrm{NiO})$ at its rim. Groundmass olivines that are surrounded by plagioclase are generally normally zoned and reach very low Fo compositions at the rims ( $\sim \mathrm{Fo}_{77-70}$ ), but may also show weak reverse zoning with core to rim increases of $\sim 0.5 \mathrm{~mol} \%$ (with NiO contents $<0.2 \mathrm{wt} \%$ for both core and rim).

In the MOZ and UOZ, the relative proportion of olivine primocrysts decreases, whereas the total olivine mode increases (up to $\sim 55 \%$ ). Olivines enclosed by clinopyroxene oikocrysts become less common, but where present they have a restricted compositional range ( $\mathrm{Fo}_{83-81}$ ). In the MOZ, the average Fo composition of groundmass olivines is generally more primitive ( $\sim \mathrm{Fo}_{80}$ ) than the average groundmass olivines from the LOZ $\left(\sim \mathrm{Fo}_{75}\right)$. Subtle ( $\sim 1 \mathrm{~mol} \%$ ) reverse zoning is observed in both groundmass and hopper olivines in the MOZ.

In the UOZ, common hopper olivines have core compositions as low as $\mathrm{Fo}_{73}$ and commonly display reverse zoning with rims between $\mathrm{Fo}_{80-77}$ (Fig. 4.8). An olivine primocryst in the UOZ displays reverse zoning (Fig. 4.9) with a core to rim increase from
$\mathrm{Fo}_{77}(0.16 \mathrm{wt} \% \mathrm{NiO})$ to $\mathrm{Fo}_{80}(0.23 \mathrm{wt} \% \mathrm{NiO})$. Groundmass olivine in the UOZ shows reverse zoning for Fo , with cores of $\sim \mathrm{Fo}_{77}$ and rim compositions ranging between $\mathrm{Fo}_{80-78}$ (with NiO content for core and rim typically < $0.22 \mathrm{wt} \%$ ).


Fig. 4.9: A BSE image and corresponding microprobe traverse of an olivine primocryst from the UOZ. Spot analysis numbers are plotted on the image and correspond to the microprobe traverse. The spot analysis numbers for the Cr-spinel inclusions are also plotted (data in appendix E5). The olivine is reversely zoned, with increasing Fo towards the rim (increasing from $\mathrm{Fo}_{77}$ to $\mathrm{Fo}_{80}$ ), as well as increasing NiO ( $\mathrm{wt} \%$ ) and decreasing $\mathrm{CaO}(\mathrm{wt} \%)$ at the rim.

### 4.4.2.2 Cr-spinel

Cr-spinel microphenocrysts enclosed by phenocrystic olivine in the LCM have compositions that depend on their location in the host olivine phenocryst. For example, a Cr-spinel in the core of an olivine phenocryst has a $\mathrm{Cr} \#(=100 \times \mathrm{Cr} /[\mathrm{Cr}+\mathrm{Al}])$ of 50 , while a Cr-spinel included in the rim of the same olivine has a higher Cr\# of 58. Neither Fe3\# $\left(=100 \times \mathrm{Fe}^{3+} /\left[\mathrm{Fe}^{3+}+\mathrm{Cr}+\mathrm{Al}\right]\right)$ nor $\mathrm{Fe} 2 \#\left(=100 \times \mathrm{Fe}^{2+} /\left[\mathrm{Mg}+\mathrm{Fe}^{2+}\right]\right)$ for these LCM Cr-spinel microphenocrysts show any variation, however.

In the LOZ, Cr-spinels inclusions have very similar compositions (Cr\# 73-46; Fe3\# 18-10; Fe2\# 84-41) to those of the LCM/LBZ, whereas they show more significant compositional variations in the UOZ. For example, a Cr-spinel in the core of a reversely zoned UOZ olivine primocryst ( $\mathrm{Fo}_{77}$; Fig. 4.9) has a $\mathrm{Cr} \#$ of $67, \mathrm{Fe} 3 \#$ of 67 and $\mathrm{Fe} 2 \#$ of 84, whereas the Cr-spinel located in the olivine rim ( $\mathrm{Fo}_{80}$; Fig. 4.9) has a higher $\mathrm{Cr} \#$ of 71 , and markedly lower $\mathrm{Fe} 3 \#$ of 59 and a lower $\mathrm{Fe} 2 \#$ of 80.

### 4.4.2.3 Clinopyroxene

The Mg\# (molar Mg\# $=100 \times \mathrm{Mg} /\left[\mathrm{Mg}+\mathrm{Fe}^{2+}+\mathrm{Mn}\right]$ ) of clinopyroxene versus stratigraphic height is summarised in Fig. 4.10. The Mg\# of groundmass clinopyroxene in the LBZ is relatively low ( $\mathrm{Mg}_{77-70}$ ) compared with clinopyroxenes of the overlying OZ. Most LBZ clinopyroxene crystals are normally zoned (decreasing Mg\# towards the rims), but some are reversely zoned, with Mg\# increasing by up to $\sim 3 \mathrm{~mol} \%$ towards the rim (Fig. 4.11b).

In the OZ , clinopyroxene core compositions show little variation with stratigraphic position (Figs 4.10, 4.11a \& Fig. 4.12) with $\mathrm{Mg}_{86-83}$ (average $\mathrm{Cr}_{2} \mathrm{O}_{3} \sim 0.8 \mathrm{wt} \%$ and $\mathrm{TiO}_{2}$ $\sim 0.4 \mathrm{wt} \%$ ). Clinopyroxene oikocrysts have near-constant core-mantle compositions, but may have narrow $\mathrm{Fe}-\mathrm{Ti}-$ rich rims ( $\sim \mathrm{Mg} \#_{75}$; Fig. 4.11a; also see Fig. 4.5b, d). The evolved rim compositions of oikocrystic and cumulus clinopyroxene in the OZ overlap with the compositions of interstitial OZ clinopyroxene, and with those of LBZ groundmass clinopyroxene (Fig. 4.10). Some LOZ clinopyroxene oikocrysts show subtle sector-zoning, visible on element maps of Cr , Al and Mg (Fig. 4.5c, e, f). Cr and Al (Fig. 4.5c, e) appear
to show similar behaviour and are slightly enriched in a N-S sector, whilst Mg (Fig. 4.5f) is more enriched in the E-W sector of the clinopyroxene.


Fig. 4.10: Stratigraphic column showing clinopyroxene compositional variations in molar magnesium content $\left(\mathrm{Mg} \#=100 \times \mathrm{Mg} /\left[\mathrm{Mg}+\mathrm{Fe}^{2+}+\mathrm{Mn}\right]\right)$ through the entire LPS. The grey box in the OZ covers the range of poikilitic and cumulus clinopyroxene core compositions in the OZ (between $\mathrm{Mg} \# 85-82$ ), illustrating their restricted range in composition.


Fig. 4.11: (a) Poikilitic clinopyroxene crystal ( $\sim 1.5 \mathrm{~mm}$ ) from the LOZ (plane-light photomicrograph), with the corresponding microprobe traverse. The profile starting at 127 traverses the crystal in a N-S direction and avoids the olivine chadacrysts. The clinopyroxene is normally zoned with narrows rims. (b) A subhedral (interstitial?) clinopyroxene from the LBZ, which is adjacent to an olivine phenocryst (BSE image). The corresponding microprobe traverse exhibits increasing $\mathrm{Mg} \#$ towards its rim (i.e. reverse zoning).

In the CPZ, euhedral cumulus clinopyroxenes, which are often sector-zoned, have core compositions similar to those of OZ clinopyroxene (Mg\#83: Fig. 4.10), whilst rims extend to compositions that are more Fe -rich ( $\mathrm{Mg} \# 70-30$ ) than clinopyroxene rims in the underlying OZ. The sector-zoning of clinopyroxene is apparent on element maps (Fig. 4.6), with different zones showing weak contrasts in Cr and Al abundance (Fig. 4.6c, f), whereas Ca and Mg show no clear inter-zone differences (Fig. 4.6d, e). $\mathrm{Cr}_{2} \mathrm{O}_{3}$ typically decreases from $\sim 0.6 \mathrm{wt} \%$ in the cores to $\sim 0.2 \mathrm{wt} \%$ in the rims, whilst $\mathrm{Al}_{2} \mathrm{O}_{3}$ increases from $\sim 1.8 \mathrm{wt} \%$ in the cores to $\sim 2 \mathrm{wt} \%$ at the rims. $\mathrm{TiO}_{2}$ contents also increase slightly from the cores ( $\sim 0.4 \mathrm{wt} \%$ ) to the rims ( $\sim 0.5 \mathrm{wt} \%$ ); which contrasts with the much greater increase in
$\mathrm{TiO}_{2}$ (to $\sim 1 \mathrm{wt} \%$ ) seen in OZ clinopyroxene rims. We suspect the deficit in $\mathrm{TiO}_{2}$ of CPZ clinopyroxene rims may be due to sequestration of $\mathrm{TiO}_{2}$ into interstitial ilmenite in the case of the CPZ. The CPZ grades upward over $\sim 0.8 \mathrm{~m}$ into slightly more evolved clinopyroxene core compositions (Mg\#80), which have reversely zoned mantles (up to $\mathrm{Mg} \# 83$ ), suggesting interaction with a more MgO -rich melt. The presence of wide, strongly evolved rims on all forms of clinopyroxene in the CPZ suggests that all have interacted with abundant, late, evolved pore melts during the final stages of crystallisation, possibly suggesting higher trapped melt fractions in these rocks.


Fig. 4.12: A microprobe traverse across a subhedral clinopyroxene crystal (37), which is surrounded by interstitial plagioclase (the corresponding photomicrograph is in plane-light). The clinopyroxene is normally zoned with narrow Fe -enriched rims. $\mathrm{Cr}_{2} \mathrm{O}_{3}$ (wt \%) shows increases towards one rim, but decreases towards the other. A hopper olivine ( 48 to 49 ) surrounded by plagioclase shows reverse zoning (core $\mathrm{Fo}_{77.6}$ to rim $\mathrm{Fo}_{79.5}$ : data in appendix E2).


Fig. 4.13: (a) Plane-light photomicrograph of a euhedral clinopyroxene crystal from the DZ, with the corresponding microprobe traverse (23 to 26). The clinopyroxene has increasing $\mathrm{Mg} \#$ and $\mathrm{Cr}_{2} \mathrm{O}_{3}$, within its mantle before developing normal zoning at its rim, suggesting it may have mixed with more primitive melt during crystallisation. (b) A BSE image of a clinopyroxene crystal from the UBZ that has a sieve-textured core/mantle. The microprobe traverse ( 10 to 20) shows complex zoning with a high-Mg\# core, followed by low-Mg\# mantles and outwardly decreasing $\mathrm{Mg} \#$ at the rims, a pattern that is resembled by $\mathrm{Al}_{2} \mathrm{O}_{3}$.

In contrast to the muted variations of clinopyroxene compositions in the OZ, clinopyroxene in the DZ shows systematic up-section cryptic variations (Fig. 4.10). At the base of the DZ, clinopyroxene compositions overlap with those of clinopyroxene from the OZ and CPZ ( $\sim \mathrm{Mg} \# 84$ ). A cumulus-textured clinopyroxene at the base of the DZ (Fig.
4.13a) shows complex zoning, with a core composition of Mg\#84 that increases outward gradually to Mg\#85 and then shows a sudden decrease out to Mg\#63 in the rim. This pattern of Mg-enrichment in the mantle of cumulus clinopyroxene occurs sporadically in the DZ (Fig. 4.10). More generally, clinopyroxene displays strong normal core to rim zoning throughout the DZ with changes in Mg\# ranging between $\sim 10-20 \mathrm{~mol} \%$. Rims of euhedral clinopyroxene overlap with the compositions of evolved interstitial clinopyroxene in the DZ . Interstitial clinopyroxene from the DZ is more evolved than interstitial clinopyroxene from the OZ . In the upper DZ , clinopyroxene (including bladed clinopyroxene crystals) have core compositions of Mg\#75 and strong normal zoning at the
rims (to Mg\#30-20). The peak in clinopyroxene Fe-enrichment, just below the UBZ, may represent a type of 'sandwich' horizon.

The UBZ contains rare, euhedral ( $\sim 1 \mathrm{~mm}$ ), sector-zoned clinopyroxene with compositions of Mg\#81 in the crystal core (Fig. 4.13b), with a narrow Fe-Ti-enriched rim (element maps of this crystal are included in appendix E7-1). This specific crystal is sieve-textured with mantle depletions in Mg\# (Mg\#70-67; Fig. 4.13b) relative to the rims, that may indicate a thermal spike, or a reaction with more primitive melt. Interstitial clinopyroxene is intergrown with plagioclase and has evolved compositions ( $\sim \mathrm{Mg} \# 67$ ).

### 4.4.2.4 Plagioclase

Plagioclase anorthite (An; molar An $=100 \times \mathrm{Ca} /[\mathrm{Ca}+\mathrm{Na}+\mathrm{K}]$ ) content versus stratigraphic height is shown in Fig. 4.14. Plagioclase dendrites within the LCM groundmass have relatively low An values, averaging $\mathrm{An}_{64}$. In the LBZ , the coarser plagioclase dendrites become slightly more primitive with compositions of $\sim \mathrm{An}_{71}$.

Plagioclase has core compositions of $\mathrm{An}_{73-68}$ throughout the OZ. Most grains show normal zoning with sporadic occurrences of reverse zoning. Plagioclase rim compositions extend to An contents as low as $\mathrm{An}_{58}$, with the lowest An values adjacent to olivine crystals. Core compositions of interstitial plagioclase in the CPZ (Fig. 4.15a) are between $\mathrm{An}_{72-70}$, whereas rims are strongly evolved ( $\sim \mathrm{An}_{40}$ ). CPZ plagioclase core compositions overlap the composition of OZ plagioclase. In the CPZ , minor euhedral cumulus plagioclase is more primitive with core and mantle compositions up to $\mathrm{An}_{76}$.

Plagioclase core and mantles are typically $\sim \mathrm{An}_{70-60}$ in the DZ , with strongly evolved rims as low as $\mathrm{An}_{45}$. Rare, cumulus-textured plagioclase in the lower DZ has some of the most primitive An compositions ( $\mathrm{An}_{77-76}$; Fig. 4.15b) in the LPS. In some plagioclase crystals in the middle DZ, the cores are mantled by higher An plagioclase suggesting interaction with a more primitive melt prior to the initiation of in-situ fractional crystallisation. The core and rim compositions with the lowest An values occur in the upper DZ (Fig. 4.14) where clinopyroxene has the lowest Mg\# (Fig. 4.10), corresponding to the inferred position of the sandwich horizon.


Fig. 4.14: Stratigraphic column showing plagioclase compositional variations in molar anorthite ( $\mathrm{An}=100 \mathrm{x}$ $\mathrm{Ca} /[\mathrm{Ca}+\mathrm{Na}+\mathrm{K}])$ content through the entire LPS. The vertical grey box covers the typical range of core compositions throughout the LPS $\left(\mathrm{An}_{72-68}\right)$.


Fig. 4.15: (a) Plane-light photomicrograph of an interstitial plagioclase crystal from the CPZ with the corresponding microprobe traverse. A $\sim 1 \mathrm{~mm}$ sized euhedral sector-zoned clinopyroxene can also be seen (22 to 23). The plagioclase is strongly normally zoned with An values of $<50 \mathrm{~mol} \%$ at the rims. (b) A planelight photomicrograph of a euhedral plagioclase lath (possibly cumulus?) in the DZ , with the corresponding microprobe traverse (61). This plagioclase has the highest core $\mathrm{An}\left(\mathrm{An}_{77}\right)$ content recorded in the LPS and is normally zoned towards its rim.

A fine-grained ( $\sim 0.2 \mathrm{~mm}$ ) plagioclase dendrite in the UBZ is reversely zoned at its rim (Fig. 4.16) and, as for clinopyroxene (Fig. 4.13b), may record a temperature increase and/or a primitive replenishment event in the UBZ.

### 4.4.3 Calculated melt chemistry

The $\mathrm{Fe} / \mathrm{Mg}$ ratio of a crystal can be used to calculate the $\mathrm{Fe} / \mathrm{Mg}$ of its equilibrium melt if the $\mathrm{Fe}=\mathrm{Mg}$ exchange coefficient is known (Roeder \& Emslie, 1970). Data from Franklin sill chilled margins and the coeval Natkusiak flood basalts were used to parameterise covariations of melt $\mathrm{Fe} / \mathrm{Mg}$ ratio with melt MgO , allowing melt MgO in equilibrium with both olivine and clinopyroxene to be calculated (model anchors are included in appendix D).


Fig. 4.16: A plane-light photomicrograph and microprobe traverse (7 to 9) of a plagioclase crystal in the UBZ. This plagioclase is reversely zoned with increasing An towards its rim.

### 4.4.3.1 Olivine

An exchange coefficient $(\mathrm{Kd})$ of 0.34 as experimentally determined by Matzen et al. (2011) was used to calculate the exchange of molar FeO and molar MgO between olivine and melt. The cores and mantles of euhedral olivine phenocrysts in the LCM yield model melts with up to $\sim 13 \mathrm{wt} \% \mathrm{MgO}$. The narrow evolved rims on these grains yield model melts of $\sim 11-10 \mathrm{wt} \% \mathrm{MgO}$, closer to the LPS chill composition. The most evolved olivine phenocryst rims yield model melts of $\sim 6.5 \mathrm{wt} \% \mathrm{MgO}$. In the LBZ, phenocrystic olivines have cores equivalent to model melts of $\sim 11 \mathrm{wt} \% \mathrm{MgO}$, and have narrow, Fe-rich rims equivalent to model melts of $\sim 5 \mathrm{wt} \% \mathrm{MgO}$.

Olivine primocryst cores from the LOZ yield model melts between $\sim 13.5-10 \mathrm{wt} \% \mathrm{MgO}$, while their rims yield a melt range of $\sim 8-6 \mathrm{wt} \% \mathrm{MgO}$. Olivine chadacrysts enclosed by clinopyroxene oikocrysts yield magnesian model melts between $\sim 10-8.5 \mathrm{wt} \% \mathrm{MgO}$. A reversely zoned olivine chadacryst has a core in equilibrium with a model melt of $9.8 \mathrm{wt} \%$ and a rim in equilibrium with a model melt of $10.2 \mathrm{wt} \% \mathrm{MgO}$. Groundmass olivines in the LOZ are in equilibrium with model melts ranging between $\sim 7-5 \mathrm{wt} \% \mathrm{MgO}$. In the UOZ , primocrystic olivine and hopper olivine cores yield evolved model melts ranging between $\sim 8-5 \mathrm{wt} \% \mathrm{MgO}$. A reversely zoned olivine primocryst (Fig. 4.9) in the UOZ has a core in equilibrium with a model melt of $\sim 6.3 \mathrm{wt} \% \mathrm{MgO}$, whereas its rim yields a model melt of $\sim 8.5 \mathrm{wt} \% \mathrm{MgO}$. Groundmass olivines in the MOZ and UOZ yield model melts of $\sim 8 \mathrm{wt} \%$ MgO .

### 4.4.3.2 Clinopyroxene

To calculate the MgO and FeO of the melt in equilibrium with clinopyroxene we used the clinopyroxene $\mathrm{Fe} / \mathrm{Mg}$ ratio and a clinopyroxene-melt Kd calculated using equation 14 from Bédard (2010). Interstitial LBZ clinopyroxene yields evolved model melts of $\sim 5-3.5 \mathrm{wt} \%$ MgO . In the LOZ and MOZ, clinopyroxene oikocryst cores yield primitive model melts between $\sim 9-7 \mathrm{wt} \% \mathrm{MgO}$, as do euhedral cumulus clinopyroxene cores in the UOZ. All clinopyroxene rims in the OZ yield Fe -rich compositions with model melts < $7 \mathrm{wt} \% \mathrm{MgO}$. The cores of cumulus clinopyroxenes from the CPZ yield compositions ( $\sim 9-6 \mathrm{wt} \% \mathrm{MgO}$ ), similar to the range of model melts recorded in OZ clinopyroxene, whereas their evolved rims and interstitial clinopyroxene in the CPZ yield very evolved model melts with < 5 $\mathrm{wt} \% \mathrm{MgO}$. Model melts calculated from the DZ clinopyroxene cores show cryptic upsection variation with melt MgO broadly ranging between $\sim 8-1 \mathrm{wt} \%$. In the UBZ, dendritic clinopyroxenes correspond to model melts ranging between $\sim 7-1 \mathrm{wt} \% \mathrm{MgO}$. A coarser, sector-zoned clinopyroxene in the UBZ yields model melts of $\sim 6.5 \mathrm{wt} \% \mathrm{MgO}$, with values of $\sim 2 \mathrm{wt} \% \mathrm{MgO}$ in the sieve-textured mantle and more primitive melts of $\sim 5 \mathrm{wt} \%$ at its rims.

### 4.5 Discussion

### 4.5.1 Constraints provided by the mineral-chemical data and inverse melt compositions

Olivine and clinopyroxene in the LPS yield a broad range of model melt MgO compositions. When coupled with the textural observations, these allow us to track the chemical evolution of melts within the LPS. The cores of euhedral olivine phenocrysts and primocrysts within the LCM, LBZ and LOZ yield the most primitive model melt MgO compositions ( $\sim 13.5-10 \mathrm{wt} \% \mathrm{MgO}$ ), equal or higher than the MgO content of their host chills ( $10.3 \mathrm{wt} \%$ ), but similar to the most primitive chilled margins from other Type-1 Franklin sills (Bédard et al., 2013). The majority of olivine phenocryst rims from the LCM as well as olivine phenocryst cores from the LBZ are, however, in approximate equilibrium with the LCM composition. The model melt compositions in equilibrium with olivine chadacrysts included in clinopyroxene oikocrysts have MgO values between $\sim 10-8 \mathrm{wt} \%$. The melt values calculated to be in equilibrium with the most primitive poikilitic clinopyroxene ( $\sim 9-8 \mathrm{wt} \% \mathrm{MgO}$ ) fall within this range of melt compositions. Since these clinopyroxene oikocrysts contain both normally and reversely zoned olivines, the clinopyroxene oikocrysts in the OZ must have formed after some of their enclosed olivines were exposed to a higher- MgO melt.

The primitive melt values calculated from the cores of olivine and clinopyroxene are in stark contrast to the melts calculated from olivine cores in the UOZ. Melt values in equilibrium with UOZ olivine cores are as low as $\sim 5 \mathrm{wt} \% \mathrm{MgO}$, whilst the rims extend to melt values of $\sim 8 \mathrm{wt} \% \mathrm{MgO}$. These reversely zoned trends are seen in hopper olivines, groundmass olivines and rare primocrystic olivines in the UOZ. Reversely zoned primocrystic olivines contain Cr-spinels that mirror the Fo trend (Fig. 4.9). Cr-spinel hosted within the low-Fo olivine core has lower Cr\# and higher $\mathrm{Fe} 2 \#$ and $\mathrm{Fe} 3 \#$ than the Cr -spinel hosted by the more primitive olivine rim. This pattern suggests that the Cr -spinel in the primocryst rim formed from a less evolved melt than the one in the core, similar to the host olivine primocryst zoning.

In addition to the common occurrence of reversely zoned olivines in the UOZ, groundmass olivines become more primitive upward in the OZ. In the LOZ, groundmass olivines typically have compositions of $\mathrm{Fo}_{75}$ ( $\sim 5 \mathrm{wt} \% \mathrm{MgO}$ model melt). This increases to $\mathrm{Fo}_{80}$ ( $\sim 8$ $\mathrm{wt} \% \mathrm{MgO}$ model melt) in the MOZ. MOZ and UOZ groundmass olivines are therefore in equilibrium with the rims of reversely zoned UOZ olivines, suggesting they may have crystallised from the same melt. The shift to more primitive groundmass olivine compositions upward through the OZ may indicate a change in the crystallisation conditions or differing proportions of trapped melt in this region of the OZ .

Recorded clinopyroxene model melt variations at the base of the DZ are in equilibrium with OZ clinopyroxenes ( $\sim 9-7 \mathrm{wt} \% \mathrm{MgO}$ ), before becoming more evolved up-section to reach minimum MgO contents ( $<4 \mathrm{wt} \% \mathrm{MgO}$ ) in the upper DZ (sandwich horizon). Rims to euhedral clinopyroxenes in the DZ are in equilibrium with DZ interstitial clinopyroxenes ( $\sim 4-1 \mathrm{wt} \%$ ). Throughout the LPS, clinopyroxene rims locally record extreme Fe -Ti-enrichment, presumably as pore melts evolved extensively through in-situ differentiation (cf. Humphreys (2009)). If the interstitial clinopyroxenes crystallised from late, evolved pore melt then this implies that the larger clinopyroxenes may also have crystallised or equilibrated with this melt. The upward evolution in clinopyroxene rim compositions (eventually to values of $\sim \mathrm{Mg} \# 20$ ) is matched by the plagioclase rim/interstitial plagioclase compositions $\left(\sim \mathrm{An}_{40}\right)$ suggesting the presence of a sandwich horizon, possibly as a result of the migration of evolved pore melt (Boudreau \& Philpotts, 2002) towards the last part of the sill to solidify, i.e. the upper DZ. A coarse UBZ clinopyroxene crystal has a broader range of model melts ( $\sim 7-4 \mathrm{wt} \% \mathrm{MgO}$ ) compared to dendrites in the $\mathrm{LBZ}(\sim 4-3 \mathrm{wt} \% \mathrm{MgO}$ ) and may therefore be a 'xenocrystic' crystal that was carried in with the first pulse of LPS magma.

### 4.5.2 PELE modelling

To help constrain the crystallisation history of the sill, we ran the PELE program (Boudreau, 1999) using a chill with $10.3 \mathrm{wt} \% \mathrm{MgO}$ (defined as the average chill composition of the LPS and related intrusions; clarified in chapter 5) as a starting composition. A fractional crystallisation model was run at a pressure of $\sim 0.8 \mathrm{kbar}$, corresponding to the estimated height of the overlying stratigraphic column. Melt $\mathrm{CO}_{2}$ was
set at $0.2 \mathrm{wt} \%, \mathrm{~S}$ at $0.12 \mathrm{wt} \%$ and $\mathrm{H}_{2} \mathrm{O}$ at $0.2 \mathrm{wt} \%$, and chromite was excluded from the assemblage (justifications are provided in the appendix E6-2). PELE predicts a liquidus at $1245^{\circ} \mathrm{C}$ where $\mathrm{Fo}_{88}$ olivine crystallises, values similar to compositions of olivine phenocrysts in the LCM. After $\sim 8 \%$ olivine was extracted, PELE predicts that the melt cosaturates in clinopyroxene ( $\mathrm{Mg} \# 87$ ) and plagioclase $\left(\mathrm{An}_{80}\right)$ at $\sim 7.5 \mathrm{wt} \% \mathrm{MgO}\left(1175^{\circ} \mathrm{C}\right)$, at which juncture the composition of olivine is $\mathrm{Fo}_{84}$. PELE predicts that olivine reaches compositions of $\mathrm{Fo}_{75}$ (similar to primocryst rims in the LOZ and the most evolved primocryst cores from the UOZ) at $\sim 1115^{\circ} \mathrm{C}$, at which point the sill would be $40 \%$ solidified, comprising $11 \%$ olivine $+16 \%$ plagioclase $+13 \%$ clinopyroxene $+60 \%$ pore melt with an MgO of $6 \mathrm{wt} \%$. In conclusion, the crystallisation pathway predicted by PELE is broadly compatible with the observed compositions and textures of olivine in the LPS OZ. However, the most primitive model melts calculated to be in equilibrium with clinopyroxene ( $\sim 9 \mathrm{wt} \%$ ) exceed the melt MgO composition at which clinopyroxene becomes cosaturated according to the PELE model ( $\sim 7.5 \mathrm{wt} \%$ ). Conversely, the model melts calculated to be in equilibrium with most DZ clinopyroxene are in accord with the PELE results. We now discuss the role that fractional crystallisation may have played in the differentiation of the LPS.

### 4.5.3 The role of closed-system fractional crystallisation in the LPS

Mineral compositional trends in a related suite of igneous rocks are commonly explained by fractional or equilibrium crystallisation (Bowen, 1928). The classic example is the Skaergaard intrusion, which exhibits unambiguous mineralogical and cryptic mineralchemical trends, interpreted to reflect closed-system fractional crystallisation of a single pulse of magma (Wager \& Deer, 1939, McBirney, 1996, Tegner, 1997). In the LPS we see similar signs of compositional evolution that could be interpreted as being due to in-situ fractional crystallisation. Such an interpretation would imply that the sill was initially filled with an olivine + chromite phyric magma; that all of the olivine settled out to yield the OZ; that after the olivine settled out, cosaturation of clinopyroxene yielded the CPZ, followed almost immediately by plagioclase cosaturation and reaction of olivine to low-Ca pyroxene to produce the basal DZ. The diffuse inward trends seen in the DZ mineral chemistry of both clinopyroxene (Fig. 4.10) and plagioclase (Fig. 4.14) suggest that the DZ solidified mainly at the floor and reached its furthest chemical evolution in a sandwich horizon just
below the UBZ. In such a model, the systematic decrease in Fo from olivine primocrysts, to chadacrysts, to groundmass in a single thin section would record in-situ evolution of trapped pore melt that partly reacted with its cumulate matrix. While many aspects of such a scenario are plausible, the petrology of the LPS suggests a more complex differentiation history.

None of the main mineral species within the OZ exhibits trends suggesting systematic upsection fractional crystallisation by sequential crystallisation against the floor (Fig. 4.8, 4.10 \& 4.14). Given the extreme compatibility of nickel in olivine (Arndt, 1977, Hart \& Davis, 1978), nickel in accumulated olivine should show a marked and rapid up-section decrease, a pattern that is not observed in the LPS OZ. Alternatively, the uniform compositions of the various olivine morphologies in the LPZ OZ could be explained if these olivines remained in suspension and re-equilibrated with melt (equilibrium crystallisation path) prior to deposition, as in the model proposed by Huppert and Sparks (1980) and Tait (1985). This scenario also fails to explain the observations, because in such a model the deposited olivines should be essentially unzoned, or at least show a consistent zoning pattern, whereas olivines in the LPS OZ show marked and diverse zoning patterns. The common preservation of hopper olivine habits in the LPS OZ also seems inconsistent with this scenario, because hopper olivines would have recrystallized during settling to become more euhedral (Welsch et al., 2012).

Nor are the modal abundances of olivine in the LPS compatible with the amounts of olivine predicted to crystallize from the magma using the PELE modeling. The olivine present in the OZ (when redistributed throughout the LPS) represents $c a .15-20 \%$ of the volume of the LPS (assuming constant thickness). This is far in excess of the $c a .8 \%$ olivine-only crystallization predicted by PELE. This suggests that the amount of olivine in the LPS OZ was not extracted by in-situ crystallization of a magma similar to the LPS chill. PELE models with a starting composition corresponding to some of the most primitive Franklin chills ( $c a .13 .5 \mathrm{wt} \% \mathrm{MgO}$ ) crystallize slightly more olivine ( $c a .10-$ $15 \%$ ) prior to cosaturation in plagioclase and clinopyroxene and are better fits to the observed modal abundances of olivine in the LPS OZ. It might be possible to explain the observed olivine modal abundances if the LPS sill was filled with a more primitive magma (in comparison with the observed chilled margin composition), or if the initial melt pulse carried a significant cargo of olivine crystals (ca. 20\%). However, neither of these
alternative models is compatible with the variety of olivine zoning patterns and their organized distribution within the OZ.

In conclusion, we contend that the common occurrence of reversely zoned olivines in the UOZ and the occasional presence of evolved Cr-spinel inclusions in the low-Fo olivine cores of such grains (Fig. 4.9) cannot be explained by in-situ fractional crystallisation of a single melt pulse and require the operation of an additional mechanism during sill formation and differentiation.

### 4.5.4 The significance of reversely zoned olivine

The survival of strongly zoned olivine in the LPS, particularly the reversely zoned olivines, and the heterogeneous olivine compositions from single thin sections, imply preservation as a result of rapid cooling. The presence of heterogeneous olivine compositions within single thin sections is reminiscent of the diversity of olivine Fo compositions observed in Hawaiian picrites (Garcia et al., 2003). Preliminary olivine $\mathrm{Fe}=\mathrm{Mg}$ diffusion calculations following the method of Costa and Dungan (2005) suggest that the residence times of olivine at magmatic temperatures (with a closure temperature of $\sim 1150^{\circ} \mathrm{C}$ ), were between ca. 2-5 years. Following the method of Carslaw and Jaeger (1959) for 'simple' conductive heat loss of a sill to wallrock at $75^{\circ} \mathrm{C}$, between 3-6 years are needed to cool magma to its solidus, similar to that calculated above. Using a method similar to that described by Cawthorn and Walraven (1998) yields a very similar cooling rate. These calculations indicate that the LPS cooled rapidly and imply that opportunities for re-equilibration were limited. Consequently, we conclude that the mineral-chemical signatures recorded in the LPS are primary igneous features.

Reversely zoned olivines are present throughout the OZ but they are particularly common in the UOZ. In the LOZ, olivine chadacrysts enclosed by clinopyroxene preserve subtle reverse zoning, indicating reaction with primitive melts containing $c a .10 \mathrm{wt} \% \mathrm{MgO}$. According to the PELE modeling, such melts would only be saturated in olivine + chromite. One of the most extreme examples of a reversely zoned olivine in the UOZ has an evolved core $\left(\mathrm{Fo}_{77}\right)$ that formed from a low-MgO ( $c a .6 \mathrm{wt} \%$ ) melt that would have been in equilibrium with an olivine + clinopyroxene + plagioclase assemblage. The core of
this olivine also contains an evolved (low-Cr\#, high-Fe3\#, high Fe2\#) Cr-spinel that must have formed from a melt that had a low Mg\# and high concentrations of ferric iron. In contrast, the $\mathrm{Fo}_{80}$ rim of this reversely zoned olivine indicates formation from a more primitive melt with $c a .8 \mathrm{wt} \% \mathrm{MgO}$, and it contains Cr -spinel inclusions with compositions (high Cr\#, low $\mathrm{Fe} 3 \#$, low $\mathrm{Fe} 2 \#$ ) that also imply an outward shift to more primitive melt compositions. Reverse zoning (rimward $\mathrm{Mg} \mathrm{\#}$ and An increases) are also seen in both clinopyroxene and plagioclase (Fig. $4.12 \& 4.16$ ) throughout the LPS. We hypothesize that these reverse mineral-chemical zoning trends in clinopyroxene and plagioclase may possibly be related to the same mechanism that produced the reversely zoned olivines in the OZ.

The different textural types of olivine, and their distinct compositions, when considered together with the presence of reverse zoning (in both olivine +Cr -spinel inclusions) in the OZ strongly suggest that the OZ rocks are hybrids, representing a mixture of multiple generations of crystallisation products and liquids. Similar textural and phase relationships have been observed in the Lower, Critical and Main Zones of the Bushveld Complex, which have been linked to replenishment (Eales et al., 1991, Mitchell et al., 1998). Other intrusions and volcanic products also bear imprints of variable crystal cargoes, as shown by in-situ isotopic modelling (Davidson et al., 2007, Font et al., 2008, Martin et al., 2010). We will now attempt to constrain the different components involved the differentiation of the LPS using the mineral and melt chemistry in combination with the PELE modelling results.

### 4.5.5 The origin of the $O Z / D Z$ duality

### 4.5.5.1 The source of the high-MgO melts - an olivine slurry?

The presence of reversely zoned olivines in the OZ excludes an origin by simple fractional or equilibrium crystallisation of a single pulse of melt. We propose that this reverse zonation is the result of an intra-sill melt replenishment event, and that the primitive rims record the arrival of a replenishing magma. The range of primitive melt compositions calculated from the reversely zoned rims of olivine in the OZ is between $\sim 10-8 \mathrm{wt} \% \mathrm{MgO}$. These melts resemble the chilled margin composition and would have had about the same
composition as the melts from which most of the primitive olivine primocrysts in the LOZ formed. Some of the olivine primocrysts are too primitive to have crystallised from a LPS melt similar to the chilled margin, and may be exotic to the LPS. The highest model melt MgO contents (up to $\sim 13.5 \mathrm{wt} \%$ ) recorded in the OZ may represent the replenishing melt, but it is also possible that the high-Fo primocrysts that yielded these high melt MgO contents are un-re-equilibrated antecrysts derived from a less-evolved magma that was actively fractionating olivine somewhere upstream as it flowed towards the LPS.

The melt-crystal systematics lead us to infer that that a magma containing olivine primocrysts was emplaced into the LPS as a replenishing olivine slurry. In this scenario, the primitive carrier melt is responsible for the reverse zoning of OZ olivines, with the low-Fo cores of these grains being relicts of the crystal mush that was present in the sill before the slurry arrived. The late arrival of a crystal-liquid slurry carrying a cargo of primitive olivine primocrysts would account for the excess modal olivine in the LPS as a whole compared to the PELE fractional crystallization model. On the other hand, the diverse zoning patterns observed in the OZ, particularly the reverse zoning, appears inconsistent with models invoking simple redistribution of entrained olivine carried in by an initial magma. In this context, the low-Fo hopper olivines in the MOZ and UOZ might also reflect contamination of the primitive slurry by host DZ rocks. Radiogenic and $\delta^{34} \mathrm{~S}$ isotope data (chapter 5) show that the OZ and DZ are not in isotopic equilibrium and represent separate intrusive events, strongly supporting a multiple intrusion model.

We propose that a slurry of olivine primocrysts was emplaced immediately above the dendritic LBZ (at $c a .1 \mathrm{~m}$ above the lower contact) and beneath the DZ, to form the LPS OZ. We speculate that the boundary between the (solidified) dendritic LBZ and the relatively buoyant feldspar-rich resident mush above was an important mechanical discontinuity, facilitating emplacement of a slightly denser olivine slurry. The primitive melts in the replenishing magma would have partly mixed with the resident, evolved, gabbroic mush. Some of the evolved olivines in the OZ may be true relicts, whilst others may record more extensive, earlier hybridization steps. Since the primitive olivine primocrysts with normal zoning patterns typify the LOZ, we suggest that this sub-zone is the least-hybridized facies of the OZ and mostly contains olivines carried in by the replenishing slurry. Reversely zoned mantles to some olivine phenocrysts from the LBZ (Fig. 4.10) that we infer are derived from the first magma pulse emplaced into the LPS,
may also have developed as a result of olivine slurry emplacement, through heating and impregnation of a porous floor. However, reversely zoned mantles are also observed in LCM olivine phenocrysts, so we cannot rule out the possibility that the reversed zoning signatures were created further upstream in the plumbing system prior to emplacement in the LPS. It is also possible that the melt directly in contact with host dolostone may have increased its $\mathrm{Mg} / \mathrm{Fe}$ by assimilating some of the wallrock dolostone (cf. Gaeta et al. (2009), Di Rocco et al. (2012)).

Other examples of reversely zoned olivines and chromite microphenocrysts have been observed in many of the olivine-cumulate sills in the Fort Collinson Sill Complex, suggesting that late olivine slurry emplacement may be ubiquitous in these older Franklin sills (see chapter 5).

### 4.5.5.2 The $D Z$ - a relict of the resident mush?

Thin hypabyssal sills and dikes (like the LPS) are typically considered to have been emplaced near instantaneously and to have cooled rapidly. As a result, they should show limited internal differentiation. They commonly have doleritic textures, which would have hindered crystal-liquid separation, so giving them only a weak cumulate component (Gunn, 1966, Philpotts et al., 1998). The compositions of model melts in equilibrium with DZ clinopyroxene cores are typical of cotectic melt compositions ( $c a .7 .5 \mathrm{wt} \% \mathrm{MgO}$ ). The compositions, textures, scarcity of cumulus minerals ( $<10 \%$ ) and abundance of $\mathrm{Fe}-\mathrm{Ti}$ oxides of the LPS DZ are therefore typical of relatively quickly-crystallized hypabyssal sills.

If the evolved cores of reversely zoned olivines in the UOZ are relicts of the earlier material filling the LPS, as argued above, then we can use their composition to constrain the DZ composition prior to the replenishment. The most evolved olivine core $\left(\sim \mathrm{Fo}_{75}\right)$ in the UOZ that has a reversely zoned rim that is in equilibrium with a model melt composition of $\sim 6 \mathrm{wt} \% \mathrm{MgO}$. The PELE and $\mathrm{Fe}=\mathrm{Mg}$ modelling indicate that plagioclase and clinopyroxene joined olivine on the DZ liquidus at $\sim 7.5 \mathrm{wt} \% \mathrm{MgO}$. This implies that the low-Fo $\left(\mathrm{Fo}_{75}\right)$ cores in the OZ should have been part of a 3-phase cotectic gabbroic assemblage. If the DZ crystallised from a melt similar to the LPS LCM, then the PELE modelling suggests that at the moment when the replenishing event occurred, the DZ
would have comprised $60 \%$ pore melt with about $11 \%$ olivine, $16 \%$ plagioclase, and $13 \%$ clinopyroxene. We will refer to this assemblage as the resident mush. The hopper olivines of the UOZ that enclose clinopyroxene and plagioclase (Fig. 4.4b, c, f) are also inferred to record ingestion of the DZ mush by the olivine slurry.

One disparity between this model and the observations is that no olivine has been observed in the DZ, whereas PELE predicts $\sim 10 \%$ to have been present in the LPS. However, there is commonly a thin olivine-gabbro subzone at the base of the DZ in other Franklin sills of this type (e.g. Bédard et al. 2012). On this basis we infer that the sparse ( $\sim 5-10 \%$ ) orthopyroxene from the basal LPS DZ is pseudomorphous after olivine. This implies that the interstitial melt in the LPS DZ became sufficiently evolved to have reached the olivineorthopyroxene peritectic, which is consistent with the highly evolved compositions of crystal rims in the DZ (e.g. clinopyroxene Mg\#20).

The primitive plagioclase $\left(\mathrm{An}_{77}\right)$ and clinopyroxene (Mg\#85) crystals in the DZ may be relicts of early DZ crystallisation or possibly antecrystic phases carried in the first pulse of LPS magma that are embedded in a mass of later crystallisation products. It is possible that localised reverse zoning in cumulus plagioclase in the DZ may record the arrival of the OZ replenishing magma, but this remains conjectural since plagioclase An-content is sensitive to many other potential effects (e.g. build-up of water content). Reversely zoned mantles in cumulus-textured clinopyroxene in the CPZ and DZ may also be a record of olivine slurry emplacement. The cumulus crystals bearing this signal may have then been entrained and emplaced into the DZ followed mixing with the olivine slurry.

Following olivine slurry emplacement, we suggest that the DZ continued to evolve, reaching its most extreme mineral compositions (Fig. $4.12 \& 4.16$ ) and highest proportions of accessory phases (Fe-Ti oxides, amphibole, biotite) at a high-level sandwich horizon, a pattern suggestive of progressive 'in-situ' fractional crystallization (cf. Boudreau \& Philpotts, 2002). Although we cannot unequivocally determine whether the DZ sandwich horizon developed prior to, or after olivine slurry emplacement, we will argue below that it post-dates it.
4.5.5.3 Mixing and hybridisation between the resident DZ and replenishing slurry

As argued above, we propose that a partly crystalline DZ characterised by a high melt proportion $(\sim 60 \%)$ was resident in the LPS when a replenishing olivine-charged magma was emplaced. The replenishing magma must have contained a high melt proportion ( $\sim 50 \%$ ), since otherwise flow would have been impeded (Paterson, 2009). Mixing between the slurry and resident magma led to the formation of a hybrid melt with an composition of $\sim 10-8 \mathrm{wt} \% \mathrm{MgO}$, from which the most primitive reversely zoned olivine rims and overgrowing clinopyroxene oikocrysts crystallised.

Clinopyroxene oikocryst cores have similar Mg\# throughout the entire OZ, indicating that they all formed from pore melts with $\sim 9-8 \mathrm{wt} \% \mathrm{MgO}$. Conversely, the rims of reversely zoned olivines become less forsteritic up-section, from $\mathrm{Fo}_{83}$ in the LOZ to $\mathrm{Fo}_{80}$ in the UOZ, where olivine rims are calculated to have been in equilibrium with $\sim 8 \mathrm{wt} \% \mathrm{MgO}$ melts. The lack of differentiation in OZ clinopyroxene compositions suggests they cosaturated when local pore melt reached the same point in its compositional evolution. The model melts calculated to be in equilibrium with OZ clinopyroxene exceed their cotectic melt MgO content (according to PELE), suggesting that these clinopyroxenes did not form through fractional crystallisation from a LCM melt composition. Instead, OZ clinopyroxene saturation appears to have been triggered at the point of magma hybridisation when initial hybrid magmas were primitive ( $\sim 9 \mathrm{wt} \% \mathrm{MgO}$ ).

The compositions of groundmass olivines in the MOZ are more primitive than LOZ groundmass olivines and are in equilibrium with reversely zoned rims in the UOZ. The transition to more primitive groundmass olivine in the MOZ/UOZ suggests these groundmass olivines crystallized from hybrid magma percolating up through this mass as the OZ cumulates compacted. As such, the UOZ rim compositions were buffered by throughflow from hybrid pore melt expelled from below. However, compaction of the OZ may have been limited by the relatively rapid cooling rate of the LPS (<6 years). We suggest that the UOZ is a strongly hybridized facies of the OZ that retains undissolved traces of the resident mush, now represented by plagioclase and clinopyroxene inclusions in hopper olivines (Fig. 4.5d \& f), evolved olivine cores and evolved Cr-spinel inclusions (Fig. 4.9). The less common MOZ hopper olivines (Fig. 4.5b) may also have crystallised from hybrid magma, possibly suggesting a weaker contamination signature in the MOZ.

### 4.5.6 Evidence for rapid crystallisation in the centre of the LPS

### 4.5.6.1 Hopper olivine morphologies in the UOZ

The UOZ is characterized by abundant hopper olivine morphologies (Fig. 4.4d, e \& f), some of which are reversely zoned. The hopper olivines of the UOZ that enclose clinopyroxene and plagioclase (Fig. $4.4 \mathrm{~d} \& \mathrm{f}$ ) are also inferred to record ingestion of the DZ mush by the olivine slurry. Experiments carried out by Donaldson (1976) show that olivine crystal morphology is related to the cooling rate and melt MgO composition, and that hopper olivine morphologies form at high growth rates. Huppert and Sparks (1980) suggested that hopper and elongate olivines could be produced by rapid cooling when a hot, primitive basaltic magma quenches beneath a cooler basaltic melt. A very similar mechanism was proposed more recently for the origin of cyclic harristic layering in the Rum intrusion (O'Driscoll et al., 2007). The hopper olivines in the UOZ of the LPS seem to indicate rapid cooling, even though the UOZ is near the middle of the LPS (at $c a .8 \mathrm{~m}$ height) where cooling rates should be the slowest. We suggest that the hopper olivine in the LPS OZ formed when the hotter, primitive olivine slurry intruded a cooler resident gabbroic mush. In this scenario, the 'supercooled' hybrid melt crystallized olivine (hoppertype) around the xenocrystic cores sourced from the resident gabbroic mush.

### 4.5.6.2 Sector-zoned clinopyroxene in the CPZ

Above the OZ there is a striking modal and textural change, as olivine cumulates grade rapidly into a thin layer dominated by euhedral, sector-zoned clinopyroxene (Fig. 4.6). The passage from olivine-dominated cumulates to a clinopyroxene-rich cumulate assemblage (CPZ) could be interpreted as being due to fractional crystallization with cosaturation in clinopyroxene briefly preceding cosaturation with plagioclase. However, as discussed previously, our preferred interpretation is that the OZ formed as a replenishment into a resident gabbroic mush (DZ). This suggests that the CPZ may be related to this event, rather than representing an intermediate fractional crystallization step.

The development of sector-zoning in clinopyroxene has been attributed to rapid growth brought about by rapid cooling, or quenching by magma decompression and volatile exsolution (Nakamura, 1973, Brophy et al., 1999). However, as emphasized in the
discussion about the origin of hopper olivines, the CPZ is roughly in the middle of the sill (at ca. 8-9 m height), where cooling rates should be slowest. Also, clinopyroxene was not a liquidus phase upon emplacement of the initial LPS melt (assumed to have had a composition corresponding to the LCM), so decompression and volatile exsolution are not plausible causes. In the context of the OZ emplacement model, we suggest that the rapid growth textures and abundant clinopyroxene modes of the CPZ record compositional super-saturation induced by pore scale melt mixing. If the DZ was underplated by an olivine slurry, then the OZ/DZ interface may have been overprinted by the migration of pore melts from one system into the other. We speculate that the buoyant, hybrid melts from the OZ percolated upwards, possibly driven by compaction of the OZ below, and infiltrated the base of the DZ, where pore melts would have had cotectic compositions. The phase topology (Fig. 4.17) suggests that mixtures of two such melts could result in a clinopyroxene-only saturated melt (Onuma \& Tohara, 1983, Bédard, 1993). Clinopyroxenes that form from such compositionally super-saturated melts may have crystallized rapidly, much as we suggested for the hopper olivine of the UOZ. This interpretation for the origin of the CPZ is consistent with the distribution of sector-zoned clinopyroxene into 'channel-like' trains (Fig. 4.7). It is also supported by the compositions of the sector-zoned clinopyroxene, which have cores compositions ( $\mathrm{Mg} \#_{85-83}$ ) that overlap with core compositions of clinopyroxene oikocrysts in the OZ below (Fig. 4.10). Since the latter are interpreted to have crystallised from the hybrid melt of the OZ, this compositional similarity is consistent with derivation of the sector-zoned clinopyroxene from pore melt expelled from the OZ.


Fig. 4.17: A Di-An-Fo phase diagram modified after Onuma and Tohara (1983) and Bedard (1993), indicating how mixtures of melt that has crystallised on a pathway between olivine ( Fo ) and clinopyroxene (Di), with melt that has crystallised plagioclase (An), fall in the clinopyroxene-only field (labelled with a star). Melt A would originally be olivine slurry derived carrier melt that became hybridised and crystallised olivine and clinopyroxene (Melt B). Melt C would be derived from the DZ, which had crystallised plagioclase. Melt C was most likely Fe -enriched pore melt that percolated down through the DZ mush.

### 4.5.7 Model - the development of igneous layering in the LPS

Based on the textural and mineral-chemical evidence presented above, we propose that the LPS records three main magmatic stages (Fig. 4.18): (1) initial magma (a dilute crystal slurry) emplacement followed by fractional crystallization; (2) replenishment and partial hybridization with an olivine slurry; (3) late-stage pore melt migration, and limited reequilibration with cooling, trapped pore melt.

### 4.5.7.1 Stage 1 - initial magma emplacement and fractional crystallisation

The presence of $c a .5 \%$ modal olivine phenocrysts ( $c a .1-2 \mathrm{~mm}$ ) in the LPS chills suggests that the initial magma pulse probably carried small amounts of olivine, though we cannot exclude the possibility that higher proportions of olivine were concentrated in the centre of the flow (Bhattacharji, 1967, Simkin, 1967). It is also possible that the initial magma carried a cargo of clinopyroxene and plagioclase, as evidenced by the presence of rare cumulus crystals in the DZ. Model melts in equilibrium with the most primitive olivine phenocryst in the LCM are higher (at $c a .13 \mathrm{wt} \%$ ) than the LPS chill ( $10.3 \mathrm{wt} \% \mathrm{MgO}$ ). As such, some of the olivine carried in with the first pulse may be antecrystic relicts (or products of an earlier crystallization step) carried downstream as the magma fractionated. After emplacement, this basaltic magma crystallized from both margins, producing the LBZ and UBZ, where high cooling rates formed plagioclase and clinopyroxene with dendritic habits that overgrew and cemented olivine phenocrysts. The olivine phenocrysts of the LBZ have core compositions that are in equilibrium with the LCM melt composition suggesting they are derived from this magma. The lack of any olivine crystals in the UBZ suggests that any phenocrysts that were present settled fast enough to evade entrapment by the downward solidification front.

The intergrowth of olivine with clinopyroxene and plagioclase dendrites in the LBZ suggests that the initial magma rapidly reached the 3-phase cotectic after emplacement. If the LCM accurately reflects the melt composition, then only $c a .8 \%$ olivine extraction is needed to reach 3-phase saturation according to the PELE model. However, if the olivine phenocrysts in the chill are 'cumulate' phases, then the LCM composition is slightly biased towards higher MgO contents and <8\% crystallization may be needed to reach 3-phase saturation. If we deduct the average olivine core composition of LBZ olivines from the LCM (considering a $10 \%$ olivine mode for the LBZ), then the residual melt would contain $c a .6 \mathrm{wt} \% \mathrm{MgO}$, similar to our estimate of the resident mush composition using the $\mathrm{Fe}=\mathrm{Mg}$ modeling. The olivine that crystallized from this initial magma pulse into the LPS may have settled to form a first-stage olivine cumulate, but this cannot be determined with certainty. If such a basal olivine-rich layer developed at this stage, then there may have been a weak S-shaped profile with the most evolved material concentrated in the upper third of the sill, as seen in other intrusions (Shirley, 1985, Meurer \& Boudreau, 1998) and in thick lava flows (Bédard, 1987, Boudreau \& Philpotts, 2002). The sandwich horizon
defined by the clinopyroxene and plagioclase data from the LPS (Fig. 4.10 \& 4.14) may have developed early, or might have developed after emplacement of the olivine slurry (see below).

### 4.5.7.2 Stage 2 - the emplacement of an olivine slurry and hybridisation with the basal DZ

The mineral zonation patterns and compositions from the LPS OZ suggest that a crystalcharged olivine slurry was injected near the base of the LPS, beneath most of the buoyant DZ and above the more consolidated, dendritic-textured LBZ. This olivine slurry could have scoured out and mixed with weakly consolidated (olivine-enriched?) host material resting on the sill floor, with only the better-consolidated, largely solidified LBZ adjacent to the cooling surface surviving. The relative coarseness of the LBZ dendrites (Fig. 4.2b) suggests relatively slow cooling rates, possibly due to the emplacement of the hot olivine slurry above. If this is correct, it would imply that the olivine slurry arrived not long after the initial magma was emplaced into the LPS.

Fig. 4.18 (page 58): Cartoons that illustrate the differentiation and subsequent formation of the OZ/DZ duality within the LPS. Stage 1 depicts the emplacement of the first magma in the sill. The initial magma was olivine + chromite phyric (Ol 1). Evidence of this early cargo of normally zoned olivine phenocrysts was quenched in, in both lower (A) and upper (A') chilled margins. As the melt began to crystallise inwardly from its lower and upper margins, it crystallised plagioclase and clinopyroxene dendrites that cemented LBZ olivine phenocrysts (B). No olivine phenocrysts were preserved in the dendritic UBZ due to rapid olivine settling. As the melt crystallised, the sill inflated, crystallising olivine $\left(\mathrm{Fo}_{75}\right)+$ clinopyroxene $(\mathrm{Cpx} 1)+$ plagioclase (Plag) mush with $\sim 60 \%$ pore melt (with $6 \mathrm{wt} \% \mathrm{MgO}$ : Melt 1) (C). In stage 2, the olivine slurry is emplaced from the east (D), carrying a suspended cargo of primitive olivine primocrysts $\left(\mathrm{Fo}_{88-82}\right.$ : Ol 2$)$ and its equilibrium carrier melt ( $13-10 \mathrm{wt} \% \mathrm{MgO}$ : Melt 2 ). As the olivine slurry was emplaced, it partly mixed with and ingested the resident gabbroic mush in the sill (E). The main bulk of the buoyant resident gabbroic mush would have been underplated by the denser olivine slurry ( F ). As the olivine slurry under/intraplates the resident mush, the DZ, as we see it was emplaced and continued to crystallise (G). In stage 3 , the layering becomes enhanced. The majority of the primitive olivine primocrysts delivered into the sill settle in the LOZ $(\mathrm{H})$ whilst the two melts (Melt $1+2$ ) hybridise (Melt 3). This hybridised melt percolates through the OZ, crystallising olivine and poikilitic clinopyroxene ( Cpx 2 ) in the MOZ, as well as the reversely zoned rims of olivine in the UOZ (I). The hybridised melt eventually encounters DZ derived Fe-rich residual liquids, saturating the melt in sector-zoned clinopyroxene (J). The DZ continued to crystallise, with evolved clinopyroxene and plagioclase rims in the upper DZ representing a sandwich horizon where late-stage, evolved pore melt may have pooled (K).

Stage 1


Stage 2


Stage 3


The olivine slurry would have been denser than a partly-solidified, feldspar-rich dolerite and so it would probably have underplated the host DZ, much as was proposed previously for the replenishment of magma chambers (Huppert \& Sparks, 1980, Tegner et al., 1993). On the basis of major element geochemical systematics, Hayes et al. (Submitted) calculated that the olivine slurry contained $c a .20-25 \%$ olivine, making it slightly denser than a resident mush that was $c a .40 \%$ solidified. Also, the $40 \%$ solidified resident mush may have been mechanically strong due to formation of feldspar chains (Philpotts et al., 1998), and so the weakest part of the LPS may have been the interface between the dendritic LBZ and the overlying partly-crystalline resident mush. We infer that the emplacement of the olivine slurry at the base of the LPS created the S-type basal MgO bulge (Fig. 4.1b). During emplacement, we suggest that the incoming olivine slurry mixed to some extent with the resident mush (corresponding to the missing basal DZ) and created a hybrid melt with $c a .10-8 \mathrm{wt} \% \mathrm{MgO}$. The evolved olivine cores (and evolved Cr -spinel inclusions) and possibly the gabbroic inclusions in hopper olivines are interpreted to be partially digested relicts of this basal DZ. The high-An cumulus plagioclase and high-Mg\# cumulus clinopyroxene in the DZ may be remnants of the primitive base of the DZ, much of which may have mixed with and dissolved into the feldspar- and pyroxeneundersaturated olivine slurry. However, the DZ may have had a complex history prior to OZ emplacement, so we cannot exclude the possibility that the uncommon primitive plagioclase and clinopyroxene from the DZ are unrelated to OZ formation. Olivines in the MOZ and especially the UOZ commonly have hopper-type morphologies (Fig. 4.4b-f). These may record the chilling effect of extensive hybridization, coupled with heat loss to the cooler DZ host, similar to models for the harristic layering of the Rum intrusion (Huppert \& Sparks, 1980, O'Driscoll et al., 2007).

### 4.5.7.3 Stage 3 - late-stage processes

After emplacement of the olivine slurry and hybridization with the basal DZ host, the hybrid melt in the resulting OZ had a composition of $c a .10-8 \mathrm{wt} \% \mathrm{MgO}$. This range may represent heterogeneities in the amount of admixed DZ material. Alternatively, the variability of the MgO content may reflect post-hybridization fractional crystallization of the hybrid melt as the OZ solidified. During this consolidation and cooling phase, interstitial and poikilitic clinopyroxene and then plagioclase began to form. The cores of the most primitive clinopyroxenes that envelop the reverse zoned olivines are in
equilibrium with a maximum melt MgO content of $c a .9 \mathrm{wt} \%$, recording clinopyroxene cosaturation from the hybrid melt. We do not yet have a clear explanation for why clinopyroxene appears earlier on the OZ crystallization path in comparison with the PELE model results.

It is plausible to infer that pore melt from the OZ was expelled by compaction to mix with the overlying DZ , and we suggest that this is the origin of the sector-zoned clinopyroxene in the CPZ. At this juncture, the melt being expelled from the OZ would have had compositions roughly corresponding to the compositions of the reversely zoned olivine rims and clinopyroxene in the $\mathrm{UOZ}(c a .8 \mathrm{wt} \% \mathrm{MgO}$ ). When it penetrated the overlying DZ, this OZ-derived pore melt would have mixed with more evolved pore melts, possibly triggering rapid growth of sector-zoned clinopyroxene (Fig. 4.6). We suspect that mixing occurred in dissolution channels in what is now the CPZ and that clinopyroxene formed by this reaction overgrew and replaced the olivine-bearing gabbroic matrix of the CPZ (Fig. 4.7).

When permeability dropped to the point where interstitial melts could no longer move, the trapped pore melts began to evolve via in-situ fractional crystallization (Humphreys, 2009, Holness et al., 2011). Permeability drops may be linked with inwardly migrating solidification fronts (Marsh, 1996). In OZ rocks this generated narrow Fe-rich rims on olivine primocrysts, evolved groundmass olivines, Fe-Ti enriched rims on clinopyroxene, sodic rims on plagioclase, and a cortège of minor phases ( $\mathrm{Fe}-\mathrm{Ti}$ oxides, mica, amphibole, and sulphides). The effects of in-situ differentiation are most prominent just beneath the UBZ in the DZ, where clinopyroxene (Fig. 4.10) and plagioclase (Fig. 4.14) cores reach their most evolved compositions at a sandwich horizon. We tentatively suggest that this sandwich horizon developed late in the crystallization history of the LPS and post-dates the emplacement of the olivine slurry. This is based on the constraints provided from our data, which suggests that the LPS was only $c a .40 \%$ solidified at the time of olivine slurry emplacement, rather than $>70 \%$ solidification required to produce the strongly evolved/fractionated mineral compositions seen in the sandwich horizon.

### 4.5.8 Implications for the differentiation and formation of cyclic/macrorhythmic layering in large layered intrusions

The LPS preserves first order igneous layering that is similar to cyclic/macrorhythmic cumulate layering observed in large layered intrusions (Eales \& Cawthorn, 1996, Wilson, 2012). The rapid cooling rate of the thin LPS preserves fine details of mineral textures and zoning that provide insights into the differentiation and layer-forming mechanisms. Similar mechanisms may also have operated in larger, more slowly cooled, magma chambers. We have presented evidence that suggests that the LPS OZ was emplaced as an olivine slurry near the base of a pre-existing gabbroic mush. The prominent layering seen in the LPS is in marked contrast with most dolerite-textured gabbroic sills. For example, the 150 m thick Beacon sill of the Ferrar suite in Antarctica (Zieg \& Marsh, 2012) shows only limited internal differentiation, and does not develop significant modal layering, even though it is seven times thicker than the $c a .21 \mathrm{~m}$ LPS. Our data imply that the first order cumulate layering structure (OZ/DZ duality) of the LPS was not formed by in-situ closed system differentiation of a single magma pulse, and that the LPS is a composite intrusion that records multiple intrusive pulses, a conclusion that is supported by our unpublished isotopic data (chapter 5: Beard et al. In Preparation). Similar textural and phase relationships have been observed in the Lower, Critical and Main Zones of the Bushveld Complex, which have been linked to replenishment (Eales et al., 1991, Mitchell et al., 1998). Other intrusions and volcanic products also bear imprints of variable crystal cargoes, as shown by in-situ isotopic modeling (Davidson et al., 2007, Font et al., 2008, Martin et al., 2010).

Our results bear on a debate that has created very polarised views in the literature about how magma chambers form and differentiate. The DZ of the LPS (and most other sills of the Franklin suite) shows a D-shaped profile, with the development of an Fe-Ti enriched zone (sandwich horizon) as a result of inward crystallisation, much as argued by Shirley (1985), and others (Latypov, 2009). However, the OZ/DZ duality and S-shaped profile of the LPS as a whole appears to have formed through multiple injections, as advocated by Marsh (2004, 2013). We emphasise that the demonstration that some types of layering (in this instance, a basal olivine-rich layer) formed from a replenishing crystal-charged slurry does not imply that magma cannot differentiate by in-situ fractional crystallisation.

The late emplacement of an olivine slurry in the LPS has other implications for magma chamber evolution. It seems axiomatic that such an event could only occur if the incoming slurry was driven by some type of magmatic overpressure. As the olivine-charged slurry is injected into a previously existing sill, space needs to be created to accommodate it. Either the sill inflates by floor subsidence and/or roof uplift, or an equivalent volume of preexisting resident mush needs to be expelled. This remobilised gabbroic mush could be forced onwards to more distal zones as the overpressure allows the sill to expand laterally, or it could be expelled into parasitic dykes, or could be injected up-section to erupt at the surface if favourable structures are present to facilitate this. A multiple emplacement process similar to what we have documented in the LPS could also have operated in other basaltic provinces and possibly other settings, explaining common glomerocrystic and disequilibrium crystal-melt assemblages (Larrea et al., 2012, Passmore et al., 2012, Leuthold et al., 2014).

We also suggest the formation of a clinopyroxene-enriched layer at the interface between the two systems as a result of post-emplacement migration of interstitial melt. The possible development of the CPZ by mixing between OZ and DZ pore melts is similar to the mechanism of formation proposed for gabbros in the Rum intrusion (Bédard et al., 1988), pyroxenites in the Bay of Islands complex (Bédard, 1991a) and is similar to that proposed for the formation of high-Mg\# clinopyroxene in actively spreading mid-ocean crust (Lissenberg \& Dick, 2008); emphasising the need to better understand the effects of igneous metasomatic processes during the solidification of igneous bodies.

### 4.6 Conclusions

The Lower Pyramid Sill (LPS) forms part of the sill-dominated Franklin magmatic plumbing system that is well exposed in the Minto Inlier of Victoria Island, Arctic Canada. For a thin sill ( $\sim 21 \mathrm{~m}$ ), the LPS is remarkably well layered, characterised by a $\sim 7 \mathrm{~m}$ thick layer of olivine-cumulate melagabbro/feldspathic peridotite $(\mathrm{OZ})$ that is capped by a thin ( $\sim 1 \mathrm{~m}$ ) layer of sector-zoned clinopyroxene-rich cumulate gabbro (CPZ) and a $\sim 10 \mathrm{~m}$ thick layer of sub-ophitic doleritic gabbro (DZ). The OZ is sub-divided based on textures into a basal pyroxene-poikilitic OZ (LOZ); a middle OZ (MOZ); and an upper OZ (UOZ). The
absence of systematic cryptic mineral compositional variations in the LPS OZ; the high proportions of modal olivine; the presence of reversely zoned olivines; and the variety of olivine morphologies within the OZ, when considered together, appear to preclude formation by in-situ fractional crystallisation of a single pulse of magma. Data from the LPS OZ are best explained by an intra-sill mixing event between an invading olivine slurry and a resident gabbroic mush that was $\sim 40 \%$ solidified. The olivine slurry was emplaced just above the LBZ. During its emplacement, the primitive, MgO-rich melts (~13-10 wt\%) mixed with the more evolved resident mush (with $\sim 6 \mathrm{wt} \% \mathrm{MgO}$ pore melt) that was already present in the LPS. Mixing between these two magmas produced a hybrid magma, with higher proportions of DZ contaminant in the UOZ. As the OZ solidified, interstitial melt percolated up through the compacting olivine cumulate, forming reversed rims on olivine, as well as clinopyroxene oikocrysts. The intervening CPZ is dominated by enigmatic sector-zoned clinopyroxene that may have formed when this hybrid melt expelled from the OZ reacted with more Fe -rich residual melts in the DZ . The DZ shows inward fractional crystallisation trends culminating in a sandwich horizon just beneath the UBZ. Our differentiation model for the LPS illustrates that both crystal-slurries and in-situ differentiation were crucial to the development of the OZ/DZ duality seen in the LPS. Such a style of magma emplacement and differentiation may also be prominent in the construction of cyclic/macrorhythmic layering of larger layered intrusions.

# 5. The Geochemical Effects of Olivine Slurry Replenishment and Dolostone Assimilation in the Plumbing System of the Franklin Large Igneous Province 

### 5.1 Abstract

The Neoproterozoic (~723-716 Ma) Franklin Large Igneous Province exposed on Victoria Island (Arctic Canada) is comprised of a sill-dominated magma plumbing system overlain by coeval lavas. We have investigated three sections, separated by a total of $>50 \mathrm{~km}$ of distance, of a sill emplaced just above a prominent sedimentary marker unit. The sill is characterised by a basal olivine-enriched layer (OZ: up to $55 \%$ olivine) overlain by gabbroic rocks. Olivine compositional variations in the OZ can be linked to crystal habit and paragenesis. The diversity of olivine compositions observed in the OZ implies that bulk-rock arrays on an $\mathrm{MgO} v s \mathrm{FeO}$ diagram reflect accumulation of a heterogeneous olivine crystal cargo. We suggest that the OZ represents a late olivine slurry replenishment in a gabbroic sill. This is consistent with radiogenic isotope data, which show that at least three geochemically distinct magmas were emplaced into the sill. The OZ MgO vs FeO arrays exhibit a gradual lateral westward shift towards more Fe-rich bulk compositions. We rule out progressive fractionation or varying amounts of compaction as a cause of this shift. Instead, downstream mixing between the replenishing olivine slurry and a resident gabbroic mush can best explain the westward shift towards more Fe-rich bulk compositions in the OZ . Pb isotopic signatures suggest that magmas near the inferred conduit feeder assimilated small amounts (<10\%) of dolostone (possibly local host rock), which may have locally buffered olivine compositions to high-Fo contents by increasing initial melt MgO contents.

### 5.2 Introduction

To understand the geochemical evolution of basalts we need to constrain the processes active in the associated magmatic plumbing system. Sill-dominated plumbing systems are prominent in Proterozoic and Phanerozoic continental flood basalt provinces (Baragar, 1976, Francis \& Walker, 1986, Hawkesworth et al., 1995, Chevallier \& Woodford, 1999, Bédard et al., 2007), oceanic crust formed by seafloor-spreading (Bédard, 1991b, Lissenberg et al., 2004), oceanic plateau crust (Kerr et al., 1998) and Archaean greenstones (Bédard et al., 2009). Documenting lateral changes in bulk-rock and mineral chemistry in the constituent sills of such basaltic plumbing systems would shed light on mechanisms of magma emplacement in the crust, of magma interaction with host rocks, on igneous differentiation processes, and may enable reconstruction of magma flow directions (Marsh, 2004). In addition to constraining fundamental properties of magmatic systems, this type of information is an important element in the search for Noril'sk type Ni-Cu-PGE sulphide deposits (Naldrett, 1992).

Separation of crystals from melt in magma chambers is frequently assumed to be responsible for the observed geochemical diversity of lavas (Bowen, 1928) and also plays a role in the development of layering in fossil magma chambers (Wager \& Deer, 1939, Irvine, 1980, Irvine, 1987, Naslund \& McBirney, 1996). Recent discussions have emphasised the importance of remobilisation of previously deposited crystal cumulates or crystal slurries, and associated consequences for the magmatic evolution of plutonic systems (Marsh, 2013, Neave et al., 2013, Leuthold et al., 2014). It has also been proposed that the abundant phenocrysts seen in some basalts are remobilised cumulates that were flushed out of underlying feeder systems (Larrea et al., 2012, Passmore et al., 2012). Examples of crystal slurry emplacement have been documented in the Ferrar sills of Antarctica (Bédard et al., 2007); the Shiant Isles sills in NW Scotland (Gibb \& Henderson, 2006); and the Franklin sills on Victoria Island (chapter 4). The emplacement of crystal slurries is also pertinent for the construction of large layered intrusions such as the Bushveld Complex (Mondal \& Mathez, 2007, Roelofse \& Ashwal, 2012, Wilson, 2012). Slurry remobilisation is also implicated in the genesis of the anomalous concentrations of immiscible sulphide liquid observed in some Ni-Cu-PGE deposits (Naldrett, 1992).

An additional factor that can affect melt and crystal evolution is the assimilation of (and/or reaction with) country rocks. It is widely accepted that mantle-derived melts may undergo modification of their isotopic signatures as a result of crustal assimilation e.g. Hildreth and Moorbath (1988). Assimilation of carbonate rocks in particular may impact on the liquid line of descent (LLD) of magma by buffering melt $\mathrm{Mg} / \mathrm{Fe}$ ratios and preferentially stabilising clinopyroxene on the liquidus (Di Rocco et al., 2012, Mollo \& Vona, 2014). The impacts of carbonate assimilation have been demonstrated for Merapi Volcano (Chadwick et al., 2007, Troll et al., 2013) and the Colli Albani volcanic district (Gaeta et al., 2009, Di Rocco et al., 2012), but are not commonly described in tholeiitic systems. Assimilation of carbonate rocks by magma can potentially release significant volumes of $\mathrm{CO}_{2}$ (Iacono Marziano et al., 2007), which may perturb global climate. $\mathrm{CO}_{2}$ excursions in the geological record have been correlated with the formation of Large Igneous Provinces (Wignall, 2001).

The well-preserved and exposed Neoproterozoic Franklin Large Igneous Province (LIP) on Victoria Island provides a unique natural laboratory to address many of these issues. The Franklin LIP has a sill-dominated magmatic plumbing system with local fault-mediated shifts in emplacement level that facilitated magma transgression (Bédard et al., 2012). Franklin sills were emplaced primarily into limestones and dolostones, and calc-silicate skarns that formed by magma-carbonate interaction are locally prominent (Nabelek et al., 2013). Chapter 4 documented the development of first-order modal layering in a composite Franklin sill as a result of two intrusive pulses. In chapter 4, we proposed that the basal olivine-enriched layer of the $\sim 21 \mathrm{~m}$ thick Lower Pyramid Sill (LPS) represents a late olivine slurry replenishment into a resident gabbroic mush. Here we present data from two other localities situated north of the Minto Inlet on Victoria Island (Fig. 2.2), that we believe represent sections through coeval and possibly correlative sills. The P-sill (PS) is located $\sim 20 \mathrm{~km}$ to the ENE of the LPS, whilst the West Uhuk sill (WUS) is another $\sim 30$ km to the ENE (Fig. 2.2). Both have olivine-enriched bases and were emplaced at the same stratigraphic level. The WUS is adjacent to a prominent fault-guided magma upflow zone known as the Uhuk Massif (UM: described in Bédard et al. (2012)), which preserves calcsilicate facies at some intrusive contacts (Nabelek et al., 2013). These three olivineenriched sills (WUS, PS and LPS) are collectively termed the Fort Collinson Sill Complex (FCSC). We will compare sections through them and document how bulk-rock compositions, mineral chemistry and isotopic compositions vary along strike. We will
suggest that the FCSC records a regional scale (> 50 km ) olivine slurry replenishment event in a previously emplaced sill. The along-strike changes in composition recorded in the FCSC provide constraints on the geochemical and textural impacts of cumulate remobilisation and slurry emplacement. We also discuss how small amounts of dolostone assimilation could have increased the $\mathrm{Mg} / \mathrm{Fe}$ ratio of the magma and buffered the Focontent of equilibrium olivine in some parts of the FCSC.

### 5.3 Fort Collinson Sill Complex

The Fort Collinson Sill Complex (FCSC), comprising the West Uhuk Sill (WUS), the Psill (PS) and the Lower Pyramid Sill (LPS), outcrops in the Collingwood Hills region of the Minto Inlier, just north of the Minto Inlet (Fig. 2.2). The FCSC was emplaced $\sim 3 \mathrm{~km}$ stratigraphically beneath the coeval Natkusiak flood basalts. Previous work has documented the field relationships at the Uhuk Massif (Bédard et al., 2012), showing that the Uhuk Massif (UM) is a magma upflow zone that extends towards the west as a conformable sill-like body, the WUS (Fig. 5.1). The WUS was emplaced into the dolostone-dominated Jago Bay Formation, only $\sim 10 \mathrm{~m}$ above the contact with the underlying Fort Collinson Formation, which is dominated by quartz-arenites. The WUS is $\sim 40 \mathrm{~m}$ thick, has sharp chilled upper and lower contacts, and contains a $\sim 12 \mathrm{~m}$ thick olivine-enriched layer in its lower section that we refer to as the Olivine Zone (OZ). About 30 km to the WSW, we sampled a section through another olivine-rich sill (PS) that is located at exactly the same stratigraphic level as the WUS. No chilled margins are exposed at the PS, which has a slightly thinner OZ ( $\sim 10 \mathrm{~m}$ thick) in comparison to the WUS. About 20 km to the WSW of the PS is the $\sim 21 \mathrm{~m}$ thick LPS (chapter 4), which, again, is at the same stratigraphic position as the WUS and PS. Both chilled margins are exposed at the LPS, which has an OZ that is only $\sim 7 \mathrm{~m}$ thick. The petrology and mineral chemistry of the LPS was described in detail by in chapter 4, where we argued on the basis of textural and mineral-chemical data that the LPS OZ is a hybrid rock that formed by mixing between a late olivine slurry and a resident gabbroic mush.


Fig 5.1: Field photo of the UM feeder system (looking west). The olivine-enriched East Uhuk Sill (EUS) jogs up-section into the WUS, with minor olivine-enriched ( Og ) sills between the two main sills. The WUS extends towards the west, towards to the PS and LPS.

### 5.4 Results

### 5.4.1 Petrography

The sills constituting the FCSC exhibit near-identical internal stratigraphies, albeit with differences in layer thicknesses. From bottom to top they have: an olivine-phyric ( $\sim 5 \%$ ) Lower Chilled Margin (LCM); an olivine-phyric ( $\sim 10 \%$ ) Lower Border Zone (LBZ); an Olivine Zone (OZ) composed of olivine-melagabbro to feldspathic-peridotite with olivine modes up to $\sim 55 \%$; a clinopyroxene-rich cumulate gabbro zone (CPZ), which is most prominent at the LPS; a doleritic gabbro zone (DZ) comprised of sub-ophitic clinopyroxene and plagioclase with a minor (< $10 \%$ ) cumulate component; an aphyric Upper Border Zone (UBZ); an olivine-phyric (~5\%) Upper Chilled Margin (UCM). The key features of each layer are summarised in an idealised FCSC stratigraphic log (Fig. 5.2). We will now describe the key features of each layer of the FCSC following the nomenclature of Irvine (1982), and then discuss along-strike sill-to-sill variations.


Fig. 5.2: A schematic log through the FCSC, showing key features of each layer. Both Lower and Upper Chilled Margins (LCM and UCM respectively) contain olivine phenocrysts (a; $\sim 5 \%$ ). The chilled margins grade into Lower and Upper Border Zones (LBZ and UBZ respectively) characterised by plagioclase and clinopyroxene dendrites (b; with local nucleation on the surfaces of olivine phenocrysts). The Olivine Zone (OZ) is characterised by high olivine modes (up to $55 \%$ ) with primitive olivine primocrysts, pyroxenepoikilitic textures (c) and hopper olivine-clinopyroxene textures. The OZ also contains hopper olivines (d). The Clinopyroxene Zone (CPZ) is characterised by abundant (40-50\%) sector-zoned clinopyroxene in an olivine-poor ( $5-10 \%$ ) gabbroic matrix (e). The Doleritic Zone (DZ) consists of sub-ophitic plagioclase and clinopyroxene (f) with minor olivine, sometimes pseudomorphed by orthopyroxene. The top part of the DZ is characterised by acicular pyroxenes (g) and an increase in late-stage accessory phases (quartz, alkali-feldspar, hornblende and biotite).

Where it is visible, the thin ( $\sim 5 \mathrm{~cm}$ ) LCM contains $5 \%$ euhedral ( $0.5-2 \mathrm{~mm}$ ) olivine phenocrysts in a very fine-grained groundmass $(0.1 \mathrm{~mm})$ of plagioclase and clinopyroxene dendrites. Small chromite microphenocrysts occur as inclusions in olivine and in the groundmass. In the WUS LCM, hopper olivine is also present (Fig. 5.3a). The UCM is exposed at both the LPS and WUS and also contains 5\% olivine phenocrysts (or its
pseudomorphs) in a very-fine grained groundmass ( 0.1 mm ). Amygdules ( $1-2 \mathrm{~mm}$ ) filled with secondary chlorite, biotite and quartz, are present in the UCM.

The LPS LCM grades up into a $\sim 1$ m-thick LBZ, where euhedral (1-2 mm) olivine phenocrysts are embedded in a matrix composed of plagioclase and clinopyroxene dendrites ( $1-2 \mathrm{~mm}$ ). The LBZ at the WUS and PS is generally similar, although the slightly thicker ( $\sim 5 \mathrm{~m}$ ) WUS LBZ also contains olivine with hopper morphologies (Fig. 5.3b). At the PS and WUS, the dendrite-textured basal LBZ grades up into an upper LBZ facies with a gabbroic-textured groundmass (Fig. 5.3c). Some olivine primocrysts appear to have been fractured and veined by groundmass plagioclase. Clinopyroxene oikocrysts contain inclusions of olivine, chromite and euhedral plagioclase laths. Minor groundmass orthopyroxene is also observed.

The LPS UCM grades down to an UBZ that is free of olivine phenocrysts, and which is composed of plagioclase and clinopyroxene dendrites. In chapter 4, we suggested that the absence of phenocrysts was due to crystal settling of olivine, such that the upper crystallisation front was no longer able to capture fast-settling phenocrysts. The WUS has a very thin UBZ, whereas the uppermost facies are missing from the PS as a result of erosion. The UBZ at the WUS grades down rapidly into varitextured DZ gabbros, some markedly pegmatitic with acicular pyroxene morphologies.

The LPS LBZ grades up into a $\sim 7 \mathrm{~m}$ thick OZ composed of olivine orthocumulate melagabbro to feldspathic peridotite with $40-55 \%$ modal olivine. The OZ is slightly thicker $(\sim 9-10 \mathrm{~m})$ at the PS, and thicker again $(\sim 12 \mathrm{~m})$ at the WUS. In chapter 4, we subdivided the LPS OZ based on olivine morphologies. The LPS Lower OZ (LOZ) contains euhedral (1-2 mm) olivine primocrysts and finer-grained ( $0.5-1 \mathrm{~mm}$ ) olivine that are enveloped by clinopyroxene oikocrysts and interstitial plagioclase. The LPS Middle OZ (MOZ) has fewer olivine primocrysts, and is dominated by hopper olivine that is intergrown with clinopyroxene (Fig. 5.3d). The LPS Upper OZ (UOZ) mostly contains hopper olivine (Fig. 5.3 e ), with subordinate primocrystic and groundmass olivine. The UOZ clinopyroxene is commonly euhedral and does not contain chadacrystic olivine inclusions.


Fig. 5.3 (page 71): Photomicrographs (all in crossed-nicols) of olivine morphologies and textural relationships in the FCSC. (a) The olivine-phyric WUS LCM with the lower contact with the Jago Bay formation visible. Olivine phenocrysts (Olph) and hopper olivines (Sol) are shown, residing in a very-fine grained groundmass (Gm). (b) The WUS LBZ contains both phenocrystic and hopper olivine can be seen residing in a fine-grained matrix of clinopyroxene ( Cpx ) and plagioclase dendrites (Pld). (c) The PS LBZ showing a euhedral olivine phenocryst residing in a dendritic matrix. (d) The LPS OZ (specifically the MOZ) showing both olivine primocrysts and hopper olivines that are overgrown by clinopyroxene. (e) The LPS OZ (specifically the UOZ) showing hopper olivines that are intergrown with interstitial plagioclase (Plg). (f) The WUS OZ showing hopper (resorbed?) olivine (Aol), olivine chadacrysts (Olc) enclosed in poikilitic clinopyroxene (Pcpx) and groundmass olivines (Olg) residing in interstitial plagioclase. (g) The WUS OZ showing euhedral olivine primocrysts (Olpr) and a coarse poikilitic clinopyroxene enclosing abundant olivine chadacrysts. (h) The PS OZ showing a hopper olivine that is overgrown by clinopyroxene.

In the PS and WUS, the divisions into LOZ, MOZ and UOZ are less clear than at the LPS. In the WUS OZ, hopper olivine can be seen at various levels rather than being restricted to the UOZ (Fig. 5.3f). Near the base of the PS and WUS OZ, clinopyroxene occurs both as oikocrysts (Fig. 5.3 g ) and as euhedral 'cumulus' grains, some with internal 'fairy-rings' of chromite microphenocrysts. Higher in the PS and WUS OZ, subordinate orthopyroxene occurs as oikocrysts, as euhedral prisms or is intergrown with clinopyroxene. Some olivine inclusions in clinopyroxene oikocrysts have hopper shapes and contain clinopyroxene inclusions (Fig. 5.3h). Plagioclase occurs as subhedral inclusions in clinopyroxene oikocrysts, as interstitial grains and as fine groundmass laths.

Overlying the LPS OZ is a thin ( $\sim 1 \mathrm{~m}$ ) CPZ, which is dominated by euhedral sector-zoned clinopyroxene (40-50\%). Equivalents of the CPZ were observed at the PS interlayered with olivine-rich cumulates. CPZ-type rocks were not identified in our section through the WUS. In the PS, the euhedral sector-zoned clinopyroxene commonly has sieved internal zones, and may have alternating normal and reversed zones (in terms of $\mathrm{Mg} \#$ ). The matrix of the PS CPZ preserves hopper olivine that is cemented by plagioclase.

Above the LPS CPZ is a $\sim 9-10 \mathrm{~m}$ thick DZ, which mostly consists of sub-ophitic clinopyroxene and plagioclase. Orthopyroxene ( $\sim 10 \%$ ) in the LPS DZ may be pseudomorphous after olivine. The habit of clinopyroxene changes gradually from more blocky prismatic crystals at the base of the LPS DZ to more acicular elongated prisms in the upper LPS DZ. The PS DZ and WUS DZ show a general similarity to the LPS DZ, but with some differences. At the WUS, the DZ is markedly thicker $(\sim 27 \mathrm{~m})$ than at the LPS. The basal DZ at the WUS and PS contains $10 \%$ euhedral olivine ( $1-3 \mathrm{~mm}$ ) in a gabbroic matrix ( $1-2 \mathrm{~mm}$ ) with only minor Fe -Ti oxides ( $1-2 \%$ ) and granophyric patches. Some
clinopyroxene grains are equant to euhedral, with optically distinct cores reminiscent of CPZ clinopyroxene. Plagioclase is generally anhedral, with a few coarser (1-2 mm) euhedral (cumulus?) laths that may have spongy internal zones. In the PS, this basal olivine-bearing DZ facies is thin and grades up rapidly into a varitextured doleritic gabbro that contains abundant Fe -Ti oxides and shows marked variations of grain size through the development of pegmatoidal patches. The basal WUS DZ described above is succeeded by a thick, doleritic-textured gabbro with acicular clinopyroxene habits. Approximately 12 m below the upper contact, the proportion of Fe -Ti-oxides and sulphides in this gabbro increases markedly (to $\sim 10 \%$ ), as does the amount of granophyre (to $\sim 10 \%$ ) and hornblende ( $\sim 5 \%$ ). The uppermost portions of the WUS DZ resemble the upper varitextured gabbroic facies seen at the PS, with abundant pegmatoidal patches and acicular pyroxene habits. We interpreted the increase in these accessory phases in the upper DZ at the LPS to represent an evolved residue (sandwich horizon) that formed by insitu fractional crystallisation (as described for the LPS chapter 4).

### 5.4.2 Bulk-rock chemistry

The chilled margin compositions of the WUS and LPS are very similar (10.6 vs $10.5 \mathrm{wt} \%$ MgO , respectively: appendix F1). Metre-scale dykes emplaced near the contacts of the UM also have similar compositions (appendix F1). We calculated an average FCSC chill composition of $10.3 \mathrm{wt} \% \mathrm{MgO}$ (appendix F1) from 10 samples (chills, thin sills and dykes). This chill composition was used in chapter 4 as a starting composition for PELE (Boudreau 1999) fractional crystallisation modelling, which will be used to anchor some of the models in the chapter 5 discussion section below.

Many of the OZ rocks from the FCSC show MgO -enrichment and $\mathrm{Al}_{2} \mathrm{O}_{3}$-depletion trends indicating that they contain cumulus olivine (appendix F 1 ). The OZ rocks have MgO contents between 18 and $27 \mathrm{wt} \%, \mathrm{Al}_{2} \mathrm{O}_{3}$ between 6 and $10 \mathrm{wt} \%$, and Ni contents up to 720 ppm (appendix F1). The CPZ rocks have high CaO contents, with values between 12 and $14 \mathrm{wt} \%$, and high Cr and Sc (1500 and 52 ppm , respectively), indicating a cumulate clinopyroxene component. The major element chemistry of DZ rocks is very similar in all three sections of the FCSC, with MgO contents ranging between 6 and $10 \mathrm{wt} \%$. FCSC DZ rocks show clear trends of increasing FeO and $\mathrm{TiO}_{2}$ and decreasing $\mathrm{Al}_{2} \mathrm{O}_{3}$ as MgO
decreases that imply fractionation along a multiphase cotectic (Fig. 5.4). Many DZ rocks have high $\mathrm{Al}_{2} \mathrm{O}_{3}$ contents ( 11 to $15 \mathrm{wt} \%$ ) and Sr contents ( 250 ppm ), and some show positive Eu anomalies on MORB-normalised spidergrams (not shown), implying a plagioclase cumulate component, consistent with plagioclase crystal habits.

The similarity of igneous stratigraphy, petrography and major element compositions at the WUS, PS and LPS, indicates very similar differentiation paths of a common (or at least very similar) magma. The most notable difference between the three sills is that the OZ rocks show a systematic trend towards more Fe -rich bulk compositions towards the west (Fig. 5.4 and appendix F1), which will be discussed below.


Fig. 5.4: A bulk-rock FeO* versus MgO diagram. The FCSC OZ arrays extend towards MgO-enrichment as well as rotating towards more $\mathrm{FeO}^{*}$ compositions (shown by the arrow). The FCSC DZs extend towards $\mathrm{FeO}^{*}$ enrichment, with decreasing MgO . The dashed lines are least square regression fits to the WUS, PS and LPS OZ arrays that are forced through the average FCSC chill (appendix F1). FeO * has been recalculated from total Fe , with the proportion of $\mathrm{Fe}^{2+}$ to $\mathrm{Fe}^{3+}$ set at 0.9 .

### 5.4.3 Olivine chemistry

Olivine chemical data (appendix F2) provides in-situ compositional constraints on the crystals and melts involved in the formation of OZ rocks in the FCSC. An array of different olivines can be recognised in the FCSC OZ and we will show that there is a linkage between the morphology, composition and style of zonation. The chemistry and zonation of olivine within the LPS OZ was described in detail in chapter 4 and we will summarise those findings here before comparing them to the PS and WUS. NiO (wt\%) compositions of olivine range between $0.4-0.1 \mathrm{wt} \%$ and are coupled with Fo behaviour.

Five different types of olivine morphologies were observed within the LPS OZ, each with a particular range of forsterite ( molar $\mathrm{Fo}=100 \times \mathrm{Mg} /\left[\mathrm{Mg}+\mathrm{Fe}^{2+}\right]$ ) content and distinctive core-to-rim zoning styles. Euhedral olivine primocrysts in the LPS Lower OZ (LOZ) are the most primitive, with core compositions ranging between $\mathrm{Fo}_{88-82}$. These olivine primocrysts have narrow Fe -rich rims $\left(\mathrm{FO}_{75}\right)$. This group of primitive olivine primocrysts was interpreted to have been carried as suspended 'phenocrysts' in a late olivine slurry replenishment (e.g. Fig. 4.17), with the Fe-rich rims recording in-situ differentiation of the melt after emplacement. Euhedral-subhedral chadacrystic olivines that are enclosed by clinopyroxene oikocrysts are more evolved than the primocrysts and have a restricted range of composition, between $\mathrm{Fo}_{83-81}$. Although many are normally zoned, some olivine chadacrysts show reverse zoning. The crystallisation of clinopyroxene oikocrysts must therefore post-date the origin of this reverse or normal olivine zoning. Groundmass olivines are similar in size and shape to chadacrysts enclosed by clinopyroxene oikocrysts but are more evolved ( $\mathrm{Fo}_{81-72}$ ). Groundmass olivines are typically normally zoned, but some also show reverse zoning. The hopper olivines of the LPS Upper OZ (UOZ) have similar compositional ranges $\left(\mathrm{Fo}_{80}-73\right)$ to groundmass olivines. The hopper olivines commonly exhibit reverse zoning with the most primitive rim compositions reaching $\mathrm{Fo}_{80}$ in the LPS UOZ.

Olivine primocrysts are also common in the WUS OZ and PS OZ, with core compositions up to $\mathrm{Fo}_{90}$ being observed at the PS. Olivine chadacrysts enclosed by clinopyroxene oikocrysts have $\mathrm{Fo}_{83-78}$ compositions, similar to the LPS range. Groundmass olivines at the WUS and PS range to lower Fo contents ( $\mathrm{Fo}_{80-77}$ and $\mathrm{Fo}_{83-68}$ respectively), as for the LPS.

The Fo-contents of hopper olivines at the WUS and PS sites are very similar to those of the LPS OZ, but at the WUS and PS, they are mostly normally zoned.

### 5.4.4 Lead and sulphur isotopes

Discrimination plots of ${ }^{208} \mathrm{~Pb} /{ }^{204} \mathrm{~Pb} v s^{206} \mathrm{~Pb} /{ }^{204} \mathrm{~Pb}$ and ${ }^{207} \mathrm{~Pb} /{ }^{204} \mathrm{~Pb} v s^{206} \mathrm{~Pb} /{ }^{204} \mathrm{~Pb}$ are presented in Fig. 5.5. The WUS and LPS of the FCSC are plotted (we did not analyse PS samples for Pb -isotopes), along with other olivine-enriched sills (Kat's, Dick's and Uwe's) from the Minto Inlier, some of which may be correlative with the FCSC. The WUS OZ and DZ are in approximate isotopic equilibrium in both Pb -isotope plots, consistent with derivation from the same magma. In contrast, the LPS OZ and DZ are isotopically distinct in both Pb -isotope plots, indicating that these layers were derived from discrete magmas. This observation supports the replenishment model developed in chapter 4.


Fig. 5.5: Pb -isotope discrimination plots with OZ and DZ rocks of the FCSC shown, along with other possibly correlative sills (Dick's, Kat's and Uwe's). (a) ${ }^{207} \mathrm{~Pb} /{ }^{204} \mathrm{~Pb} v s^{206} \mathrm{~Pb}$ / ${ }^{204} \mathrm{~Pb}$ and (b) ${ }^{208} \mathrm{~Pb} /{ }^{204} \mathrm{~Pb}$ vs ${ }^{206} \mathrm{~Pb}{ }^{204} \mathrm{~Pb}$. The compositional field of the basal lava unit of the Natkusiak Flood Basalts are plotted (grey field) showing Pb -isotope compositions that overlap with those of the Type-1 olivine-enriched Franklin sills.

Assimilation trajectories for dolostone and shale are shown with the starting composition set as the average Type-1 Franklin magma composition.

Dick's and Uwe's OZ and DZ are also in Pb-isotope disequilibrium, indicating these are also composite sills. Kat's sill (a few kilometres SE of the LPS: Fig. 2.2) is also located along the top of the Fort Collinson Formation contact, which based on field relationships is thought to be an offshoot of the FCSC. Isotopically, Kat's sill is similar to the WUS, with an OZ and DZ that are in approximate isotopic equilibrium.


Fig. 5.6: S-isotope values for the PS and LPS. The stratigraphic log thickness has been normalised in order to compare the data from the two sills.

S-isotopes for the PS and LPS are presented in Fig. 5.6. The PS and LPS exhibit similar $\delta^{34} \mathrm{~S}$ profiles, with contrasting ranges for their OZ and DZ rocks. The LPS exhibits more of a 'step-change' between OZ and DZ . In general OZ rocks have $\delta^{34} \mathrm{~S}$ values ranging between $+3 \%$ ond $+5 \%$, whilst the DZ rocks have a greater range in $\delta^{34} \mathrm{~S}$ values between $+3 \%$ and $+8 \%$. The apparent decoupling between the OZ and DZ shown by the S -isotopes
is consistent with the Pb -isotope disequilibrium observed in the LPS. Values of +3 to $+4 \%$ o are common for the Franklin magmas and are suggestive of a slightly anomalous melt source composition (Hryciuk et al. Submitted); whereas the range to higher $\delta^{34} \mathrm{~S}$ seen in the DZ rocks suggests these facies include a sedimentary crustal component.

### 5.5 Discussion

### 5.5.1 Are the three FCSC sills a single sill?

The WUS, PS and LPS are all exposed at the same stratigraphic horizon, immediately above the upper contact of the Fort Collinson Formation marker unit. They share a similar internal stratigraphy with similar first-order bimodal layering (OZ/DZ), and have similar petrographic features (Fig. 5.3), chilled margin compositions (appendix F1) and olivine compositions (appendix F2). Additionally, the LPS and PS show a roughly similar split between the S-isotope signatures of the OZ versus the DZ (Fig. 5.6). Dykes at the UM branch towards the west (see Fig. 6 in Bédard et al., (2012) - appendix A) suggesting westward sill propagation. The thickness of the OZ decreases systematically from 12 m at the WUS to 7 m at the LPS, as does the total sill thickness (from 40 m at the WUS to 20 m at the LPS). These relationships suggest that the WUS, PS and LPS may be a single, laterally extensive sill (>50 km), that propagated westward from a feeder dyke at the UM. We cannot completely exclude the null hypothesis, however, that these three sills are separate injections of a similar magma that was emplaced at the same stratigraphic level, and that all three evolved by similar differentiation paths to generate similar igneous layering, bulk-rock chemistry, OZ/DZ S-isotope dichotomy (LPS and PS), petrographic features and mineral compositions. Indeed, many olivine-enriched Type-1 sills from the Minto Inlier that were emplaced at other stratigraphic levels develop very similar facies sequences. However, it would be extremely fortuitous for three separate intrusions to not only share so many characteristics, but also show a systematic westward thinning of the OZ , as well as total sill thickness. We will argue that the systematic westward Fe enrichment trend in the FCSC OZ (Fig. 5.4) is consistent with a westward magma propagation hybridisation model, a scenario that is difficult to explain if the WUS, PS and LPS were three completely separate intrusions.

### 5.5.2 Are the sills of the FCSC composite intrusions?

In chapter 4, we presented mineral-chemical and textural data indicating that the LPS OZ is a hybrid rock that was produced when a primitive olivine slurry was emplaced into a resident gabbroic mush. The OZ from the PS and WUS show similar features and we infer from this that they formed in a similar manner. S-isotope data (Fig. 5.6) support the distinct origin of the OZ and DZ and seem to preclude a simple fractional crystallisation relationship between the two layers at the PS and LPS. Our limited database of bulk rock radiogenic isotope compositions (Fig. 5.5) provides additional constraints and suggests an even more convoluted intra-sill differentiation story for the FCSC. 1) It reinforces the Sisotope conclusion that LPS OZ is distinct from its DZ (Fig. 5.6). 2) It suggests that the WUS DZ and OZ are in approximate Pb -isotopic equilibrium. This could suggest that the WUS DZ developed by internal differentiation of the same magma from which the WUS OZ accumulated; yet the two are very different in terms of S -isotopes (unpublished data). 3) It suggests that magmas either experienced open-system processes with an extremely heterogeneous host stratigraphy as they migrated through the crust, or that magmas originated in an extremely heterogeneous source mantle and were not homogenised enroute, or both. Thus, the LPS OZ acquired radiogenic isotope signatures that are distinct from those in the LPS DZ, and which also differ from those in the WUS. We will now examine the $\mathrm{Fe}=\mathrm{Mg}$ systematics of the FCSC , so as to determine the compositions of crystals and melts involved in intra-sill differentiation. We will reconsider the isotopic constraints afterwards.

### 5.5.3 FCSC OZ Fe=Mg systematics

### 5.5.3.1 Crystal-melt constraints - a westward shift towards more evolved magmas?

The $\mathrm{Fe}=\mathrm{Mg}$ relationships help us to constrain the composition of magmas and crystals involved in the formation and differentiation of the sills constituting the FCSC. Fig. 5.7 shows the average FCSC chill composition, together with analyses of DZ and OZ rocks from the FCSC sills we have sampled. We have also plotted the LLD pathway predicted from PELE (Boudreau 1999) for the average FCSC chill composition, and labelled the point where plagioclase and clinopyroxene join $\mathrm{Fo}_{84}$ olivine on the liquidus. This model LLD pathway overlaps the compositions of DZ rocks from the FCSC, supporting the
notion that these are cotectic precipitates derived from fractional crystallisation of a tholeiitic melt similar to the average FCSC chill composition.


Fig. 5.7 (page 82): The $\mathrm{Fe}=\mathrm{Mg}$ systematics of the FCSC with bulk-rock compositional fields as well as the constraints provided by the olivine compositional data and their inverse melt compositions. The histogram attached to the olivine Fo-Fa join shows the distribution of primocrystic, chadacrystic and groundmass olivine compositions observed in the FCSC OZ from all three sills belonging to the FCSC. Also shown on the diagram is a melt loss trajectory with collinear L1, L2 and L3 poles (see text for details).

Although the petrography (Fig. 5.3) and MgO-enrichment trends of the FCSC OZ rocks (Fig. 5.4) imply olivine accumulation, the OZ arrays in the FCSC sills (WUS, PS and LPS) show distinct trends (Fig. 5.4). In principle, each individual OZ bulk rock composition represents a two-component mixture of cumulus olivine and trapped melt, now mostly represented by clinopyroxene and plagioclase. The modal proportion of olivine in these rocks ranges between $20-55 \%$ (Fig. 5.7), of which most is 'cumulus' textured. A least squares regression trend was fitted to each OZ array, defining both the composition of (bulk) olivine that accumulated from the melt (Cawthorn et al., 1992, Wilson, 2012) and the composition of the dominant entrapped basaltic melt (Fig. 5.7). The WUS OZ array intersects the average FCSC chill composition ( $10.3 \mathrm{wt} \% \mathrm{MgO}$ ) and projects towards an olivine composition of $\mathrm{Fo}_{87}$, which we will later show is not in equilibrium with this melt composition. The PS OZ array projects towards a more evolved basaltic melt composition ( $9 \mathrm{wt} \% \mathrm{MgO}$ ) and a more evolved olivine composition of $\mathrm{Fo}_{86}$. The LPS OZ array projects toward a still more evolved basaltic melt composition ( $8 \mathrm{wt} \% \mathrm{MgO}$ ) and a still more evolved olivine composition of $\mathrm{Fo}_{85}$. If we force the regression of the LPS OZ through the average FCSC chill composition (shown in Fig. 5.4), then the bulk-extract olivine compositions that are required to satisfy the bulk-rock mass balance for the PS and LPS OZ rocks are even more Fe -rich $\left(\sim \mathrm{Fo}_{83}\right)$ than the dominant primocrystic olivines $\left(\sim \mathrm{Fo}_{86}\right)$ that these rocks contain. Although the bulk-rock $\mathrm{Fe}=\mathrm{Mg}$ systematics imply that the olivine cumulates in the FCSC were derived from increasingly more Fe-rich melts towards the west, the observed compositions of olivine in these rocks only approximately matches the required crystal 'cumulates', and do not show systematic Fe-enrichment towards the west (Fig. 5.7 and appendix F2). The diversity of olivine compositions (histograms on the Fo-Fa join: Fig. 5.7) observed within each OZ does not fit an idealised equilibrium crystal-extract model (Cawthorn et al., 1992), and implies that the OZ cumulate arrays reflect an average cumulate formed from a heterogeneous olivine crystal cargo.

### 5.5.3.2 Determining OZ porosities using inverse melt modelling

The fossil porosity of each OZ sample from the FCSC is defined from its position on the $\mathrm{Fe}=\mathrm{Mg}$ diagram (Fig. 5.7), with the proportion of olivine in each rock ranging between 20$55 \%$. We tested these porosities using the equilibrium distribution method of Bédard (1994). The equilibrium distribution method uses the bulk-rock trace element contents of plutonic rocks to evaluate the relationships between the compositions of cumulate crystals and the amount and composition of its trapped melt fraction (TMF). It can be used to constrain the TMF if the melt composition is known. To begin, we assumed that all WUS OZ cumulates formed from a melt similar to the average FCSC chill composition (supported by the position of the WUS OZ array least squares regression: Fig. 5.7). The model melts calculated from the WUS OZ rocks resemble the average FCSC chill composition for TMFs ranging between $50-70 \%$, for complementary olivine modes of 30$50 \%$ (results included in appendix F1). The WUS OZ model melt solutions are almost identical to the olivine modes defined on the $\mathrm{Fe}=\mathrm{Mg}$ diagram (Fig. 5.7). We fitted the LPS OZ model melts to typical $8 \mathrm{wt} \% \mathrm{MgO}$ Franklin suite melt (average of Type-1 chills between 7.5 and $8.5 \mathrm{wt} \% \mathrm{MgO}$ ) because the LPS OZ array least squares regression intersects a melt with this more evolved composition (Fig. 5.7). The model melts calculated from the LPS OZ rocks yield close matches to this $8 \mathrm{wt} \% \mathrm{MgO}$ melt with TMFs that are very similar to those defined on the $\mathrm{Fe}=\mathrm{Mg}$ diagram (Fig. 5.7). The LPS OZ rocks range to slightly lower modal TMF (higher olivine modes), in comparison to those calculated for the WUS OZ.

In summary, the porosities calculated using the equilibrium distribution method closely resemble the porosities defined on the $\mathrm{Fe}=\mathrm{Mg}$ diagram, supporting the inference that OZ rocks are principally composed of olivine and trapped melt with compositions that match those defined by regression of OZ rocks on the $\mathrm{Fe}=\mathrm{Mg}$ diagram (Fig. 5.7). Only 2-3\% fractional crystallisation is needed to explain the incompatible trace element variation between melts calculated to be in equilibrium with the WUS OZ and the LPS OZ (Fig. 5.7). PELE fractional crystallisation modelling implies that $\sim 5 \%$ olivine-only fractionation is required to shift melt MgO from 10 to $8 \mathrm{wt} \%$, consistent with what was calculated from the trace elements.

### 5.5.4 Can progressive magma fractionation explain the westward FCSC OZ differentiation trend?

The westward Fe-enrichment trend in the FCSC OZ (Fig. 5.4) could perhaps be interpreted as a result of olivine fractionation during flow. However, the absence of a systematic trend towards more evolved olivine compositions towards the western end of the FCSC OZ is inconsistent with this hypothesis. Instead, the data show that the compositions of olivine primocrysts and chadacrysts are almost identical in the three OZ sections (appendix F2). The $\mathrm{Fe}=\mathrm{Mg}$ systematics imply that the WUS OZ formed from an accumulation of primitive olivine crystals originating from a melt almost identical in composition to the average FCSC chill composition (Fig. 5.7). Such a melt ( $10.3 \mathrm{wt} \% \mathrm{MgO}$ ) would be saturated only in olivine (+ minor chromite). Therefore, the WUS OZ could plausibly be interpreted to have formed by partial separation of olivine from a melt similar in composition to the average FCSC chill (Cawthorn et al., 1992). In contrast, the LPS OZ array implies extraction of less forsteritic olivines from an evolved melt composition (8 $\mathrm{wt} \% \mathrm{MgO}$ ), which would have been very close to cosaturation in pyroxene and plagioclase (Fig. 5.7). Consequently, this melt should not have been able to create a 7 m thick sequence of olivine cumulates in the LPS, suggesting that the LPS OZ contains 'excess' olivine. One might have inferred that the LCM of the LPS (being only $\sim 1 \mathrm{~m}$ below the LPS OZ) should have been the melt from which the LPS OZ crystallised. However, the misfit between the entrapped melt composition predicted from the LPS OZ array (Fig. 5.7) and the LPS LCM (appendix F1) implies that the melts in equilibrium with the LPS OZ had a different and more complex origin. We contend that the westward Fe -enrichment trend in the OZ cannot be explained solely by fractional crystallisation during magma flow in the FCSC.

### 5.5.5 Can variable degrees of compaction explain the lateral Fe-enrichment trend in the FCSC OZ?

Compaction of cumulates is one mechanism by which pore melt can be expelled from a crystal mush (Sparks et al., 1985, Mathez et al., 1997, Meurer \& Boudreau, 1998). The FCSC thins from 40 to 20 m towards the west. One would predict faster cooling and consequently less efficient compaction at the thinner LPS, in comparison to the thicker

WUS. Preferential retention of an Fe-enriched trapped melt fraction (TMF) might explain why LPS OZ bulk-rock Fe-contents are higher than in the WUS OZ. A similar mechanism was proposed to explain the along-strike increase in TMF within gabbronorites below the pyroxenite marker in the eastern Bushveld Complex (Lundgaard et al., 2006).

We can graphically determine the amount of pore melt loss required to shift the LPS OZ array to the lower Fe-contents that are typical of the WUS OZ array (Fig. 5.7). The composition of this hypothetical extracted pore melt can be determined by drawing a line that passes through two OZ rocks, one from the WUS array and one from the LPS array (Fig. 5.7). The composition of this hypothetical extracted pore melt (Gab75 or L1 on Fig. 5.7) corresponds to a melt in equilibrium with $\mathrm{Fo}_{75}$ olivine and is similar to what was calculated to be in equilibrium with groundmass olivine from the LPS OZ in chapter 4. The amount of pore melt loss required to shift the LPS OZ (L2 pole on Fig. 5.7) array to the WUS OZ (L3 pole on Fig. 5.7) array is $\sim 30 \%$ (Fig. 5.7). We tested this hypothesis by subtracting Gab75 pore melt (corresponding to the trace element composition of a DZ rock matching this $\mathrm{Fe} / \mathrm{Mg}$ bulk-rock composition from the Victoria Island database) from the LPS OZ rock, to see if the result matched the target WUS OZ rock (e.g. L2 - L1 = L3: Fig. 5.7). The best-fit trace element profile requires the loss of $\sim 15 \%$ Gab 75 from LPS OZ to reproduce WUS OZ (Fig. 5.8). This is substantially less than the $\sim 30 \%$ melt loss calculated graphically from the $\mathrm{Fe}=\mathrm{Mg}$ diagram (Fig. 5.7), indicating that the systematic westward Fe-enrichment trend in the FCSC is not a function of variable pore melt loss during compaction.


Fig. 5.8 (page 86): Spidergram showing the trace-element composition of the inferred resident mush pore melt (Gab75; with $6 \mathrm{wt} \% \mathrm{MgO}$ and $13 \mathrm{wt} \% \mathrm{FeO}^{*}$ ) taken from the database of FCSC DZ compositions (appendix F1). Also plotted are the LPS OZ and WUS OZ rocks that fall on a collinear array with Gab 75 on Fig. $5.7(\mathrm{~L} 2-\mathrm{L} 1=\mathrm{L} 3)$. The calculated trace-element signatures of the LPS OZ with Gab75 removed from this (in proportions of 10,20 and $30 \%$ ) are also shown. This melt loss trace-element signature best fits the WUS OZ with $\sim 15 \%$ Gab75 removed from the LPS OZ (see text for details). N-MORB normalizing values are from McDonough and Sun (1995).

### 5.5.6 Mixing between an olivine slurry and a resident gabbroic mush?

In chapter 4, we argued that the LPS OZ is a hybrid rock that formed from a mixture of an evolved component similar in composition to the DZ and an injected magma charged with olivine. This inference is supported by the isotopic data (Fig. 5.5 and 5.6) that demonstrate that the LPS OZ and DZ are not strictly consanguineous. We first constrain the compositions of the components involved, and then quantify the proportions of each component in the mixture using the $\mathrm{Fe}=\mathrm{Mg}$ diagram (Fig. 5.9), in order to evaluate whether variable degrees of mixing can explain the lateral differentiation trend in the FCSC OZ.

### 5.5.6.1 Component 1 - resident gabbroic mush

In chapter 4, we argued that the LPS contained a partly crystallised gabbroic mush at the time of olivine slurry replenishment. The composition of the gabbroic mush at this juncture was constrained by comparing the PELE LLD of the average FCSC chill with the melt composition calculated to be in equilibrium with evolved olivine cores located in the LPS OZ. These cores were interpreted to be xenocrystic, being derived from the host gabbro mush. In fact, the composition of evolved xenocrystic olivine cores is very similar across the FCSC, ranging between $\mathrm{Fo}_{77-75}$. In chapter 4, we calculated from $\mathrm{Fe}=\mathrm{Mg}$ olivine-melt equilibria that these olivine xenocrysts were in equilibrium with melts of $\sim 6 \mathrm{wt} \% \mathrm{MgO}$. The LLD for fractional crystallisation of the average FCSC chill is plotted on Fig. 5.7. At $6 \% \mathrm{MgO}$, the PELE model predicts that $40 \%$ of the melt has solidified, containing $11 \%$ olivine, $16 \%$ plagioclase and $13 \%$ clinopyroxene. The location of this resident mushmixing pole is labelled as Gab75, and it lies approximately in the middle of the FCSC DZ compositional field (Fig. 5.9a \& b). Olivine cores as evolved as $\mathrm{Fo}_{73}$ are present in the LPS UOZ. We label this more evolved resident mush as Gab73 (Fig. 5.9c). We also define a third possible resident mush mixing pole (Sandwich Horizon: SdH: Fig. 5.9c) that would correspond to the most evolved possible FCSC DZ compositions.

### 5.5.6.2 Component 2 - olivine slurry

The olivine primocrysts $\left(\mathrm{Fo}_{88-82}\right)$ in the LPS OZ were interpreted to have been derived from an olivine slurry that was injected into the evolved resident gabbroic mush (e.g. Fig. 4.17). The most primitive olivine primocrysts observed in the FCSC OZ have core compositions up to $\mathrm{Fo}_{90}$, but the median Fo-content of olivine primocrysts/phenocrysts is $\mathrm{Fo}_{86}$ (see histograms on Fig. 5.7). The WUS OZ array links the average FCSC chill composition to olivine compositions ( $\mathrm{Fo}_{87}$ ) that are similar to the olivine primocrysts found within these rocks. Therefore, the WUS OZ appears to represent a mixture of a melt similar to the average FCSC chill and forsteritic olivine. In order to be mobile, such an olivine slurry must have contained >50\% melt (Paterson, 2009); so it seems plausible to suggest that the lower-MgO, melt-rich part of the WUS OZ may approximate the olivine slurry mixing component. We use the average WUS OZ composition (OS87a: Fig. 5.9a) as one potential olivine slurry-mixing pole. This composition (OS87a) corresponds to $20 \%$ olivine ( $\mathrm{Fo}_{87}$ ) in a primitive carrier melt ( $13 \mathrm{wt} \% \mathrm{MgO}$ ). The composition of this primitive carrier melt was determined by calculating the melt in equilibrium with $\mathrm{Fo}_{87}$ olivine using the $\mathrm{Fe}=\mathrm{Mg}$ olivine-melt exchange coefficient $(\mathrm{Kd}=0.34)$ of Matzen et al. $(2011)$ with all Fe as FeO . This calculated carrier melt is more primitive than the FCSC chill, indicating that the melts in equilibrium with this olivine were unfractionated and that the WUS OZ array defined in Fig. 5.7 is not in equilibrium with the projected olivine composition. However, given the large spread of olivine primocryst compositions in the FCSC OZ (Fig. 5.7) it is likely that the olivine slurry was compositionally heterogeneous. The median of olivine primocryst core compositions from the FCSC OZ is $\mathrm{Fo}_{86}$ (see histograms on Fig. 5.7). This second potential olivine slurry-mixing pole (OS86a: Fig. 5.9b \& c) would contain $30 \%$ olivine $\left(\mathrm{Fo}_{86}\right)$ and a carrier melt with $12 \mathrm{wt} \% \mathrm{MgO}$, also calculated using the olivine-melt $\mathrm{Fe}=\mathrm{Mg}$ exchange coefficient of Matzen et al. (2011). More olivine-rich slurry poles can also be defined. One, which contains $45 \%$ olivine, is on the WUS array (OS87b) and another with $55 \%$ olivine is on the $\mathrm{Fo}_{86}$ tieline (OS86b: Fig. 5.9c).

### 5.5.6.3 Mixing calculations

Mixing the Gab75 and OS87a components in a proportion of 70:30 creates a hybrid, olivine charged magma that falls on the LPS OZ array (Fig. 5.9a). Mass balance implies
that this hybrid magma would contain $\sim 15 \%$ olivine primocrysts, primarily inherited from the OS87a component, and would have bulk MgO values between $\sim 15-17 \mathrm{wt} \%$. These hybrid magmas could then unmix to generate the LPS OZ array. Such unmixing was argued to play an important role in the development of the LPS OZ, with upward melt expulsion leading to the formation of the clinopyroxene that dominates the CPZ that overlies the OZ (section 4.5.6.2 \& Fig. 4.16). The mixing lines (Fig. 5.9a) also passes through the PS OZ array, indicating that the PS OZ would contain a slightly ( $\sim 5 \%$ ) higher proportion of olivine slurry. This pattern is consistent for the PS OZ in all the mixing calculations presented below. It is possible to produce a more MgO -rich 1 -stage hybrid that could match the more olivine-rich facies of the OZ (19-21 $\mathrm{wt} \% \mathrm{MgO}$ or LOZ). To do so, the olivine slurry would need to contain a higher proportion of olivine (e.g. OS87b). Mixing OS87b (containing 45-50\% olivine), with Gab75 can produce these more MgOrich compositions as one event, without the need for post-hybridisation unmixing (Fig. 5.9a).


Fig. 5.9: Various $\mathrm{Fe}=\mathrm{Mg}$ mixing models using different mixing poles. (a) OS87a containing ~20\% olivine ( $\mathrm{Fo}_{87}$ ) mixed with Gab75 (with $\sim 6 \mathrm{wt} \% \mathrm{MgO}$ ), and OS87b containing $\sim 45 \%$ olivine ( $\mathrm{Fo}_{87}$ ) mixed with Gab75. (b) OS86a containing $\sim 40 \%$ olivine ( $\mathrm{Fo}_{86}$ ) mixed with Gab75, and OS87b with $>50 \%$ olivine ( $\mathrm{Fo}_{86}$ ) mixed with Gab75. (c) OS87a mixed with Gab73 (with $\sim 5 \mathrm{wt} \% \mathrm{MgO}$ ) and OS87b mixed with SdH (with $\sim 4$ $\mathrm{wt} \% \mathrm{MgO}$ ).

Mixing Gab75 with the OS86a mixing pole in a proportion of 75:25 also generates a more MgO -rich hybrid (Fig. 5.9b). Mass balance implies that this hybrid magma would contain 20-25\% olivine primocrysts, primarily inherited from the OS86a component, and would have bulk MgO values between $\sim 19-21 \mathrm{wt} \%$. The most bulk- MgO -rich OZ compositions along the LPS array contain up to $25 \mathrm{wt} \% \mathrm{MgO}$. The more MgO -rich hybrid magmas defined above could have unmixed to produce these higher- MgO bulk-rock compositions. On the other hand, if the high- MgO compositions formed in one event then a more olivinerich slurry would be needed. An 80:20 mix of Gab75 and OS86b (which contains > 50\% olivine) can produce such hybrid magmas. However, this last model is probably unrealistic because the high proportion of olivine in the slurry would probably inhibit flow and mixing (Paterson, 2009). Alternatively, instead of high olivine modes in the slurry, the resident gabbroic mush may have been more evolved than Gab75. Mixing Gab73 with OS87a in a proportion of 80:20 can produce hybrid magmas with $\sim 19-21 \mathrm{wt} \% \mathrm{MgO}$ (Fig. 5.9 c ). This is a $5 \%$ reduction in the proportion of olivine slurry required, compared to models that use the Gab75 mixing pole.

What these mixing models show, is that it is easy to reproduce the more evolved (lowMgO ) part of the LPS OZ array by some variant of this mixing process, but that it is difficult to generate the more primitive parts of the LPS OZ array (e.g. LPS MOZ with $\sim 25$ $\mathrm{wt} \% \mathrm{MgO}$ ) by a 1 -stage mixing process unless a very high proportion (> $50 \%$ ) of olivine is present in the recharging olivine slurry (e.g. OS86b). If we only consider 'reasonable' olivine modes in the olivine slurry (e.g. OS87b) then the more primitive part of the LPS OZ array can only be attained in a single stage if the resident gabbroic mush had a SdH composition, with mixtures of 80:20 (SdH:OS87b). However, it seems implausible that the invading olivine slurry could have ingested large volumes solely of late-stage evolved magmas that only constitute a small ( $\sim 5 \%$ ) proportion of Type-1 sills. We therefore favour a scenario whereby the MgO-rich part of the OZ array is formed by unmixing of the hybrid magma.

### 5.5.7 Post-hybridisation unmixing and the formation of high- $\mathbf{M g O}$ olivine cumulates

The propagation of a primitive olivine slurry and its hybridisation with a resident gabbroic mush can produce hybrid magmas that account for the formation of the FCSC OZ. The
apparent westward decrease in the proportion of the olivine-rich component in the mixture from which the different OZs accumulated, suggests that the olivine slurry progressively hybridised with an evolved resident gabbroic mush as it migrated through the sill. This progressive hybridisation scenario accounts for the downstream bulk Fe-enrichment trend seen in the FCSC OZ. Following mixing, the resultant hybrid magmas would have unmixed to produce the high- MgO cumulate arrays displayed in Fig. 5.9. These are recorded as the primocrystic olivine in the LPS LOZ. In addition, there are texturally different olivines within OZ rocks containing the highest olivine modes (Fig. 5.7). These rocks, preserved in the LPS MOZ, are predominantly composed of hopper olivine that is intergrown with clinopyroxene (Fig. 5.3d, f, h), with few ( $\sim 5 \%$ ) euhedral olivine primocrysts, and have high- MgO contents ( $\sim 25 \mathrm{wt} \% \mathrm{MgO}$ ). Also, the LPS MOZ olivine compositions $\left(\mathrm{Fo}_{83-81}\right)$ are more evolved than the euhedral olivine primocrysts that characterise the olivine slurry. Based on these textural and compositional constraints we interpret the LPS MOZ to have crystallised directly from the hybrid melt that was produced from the mixing process explained above. To test this scenario we ran an assimilation-fractional-crystallisation (AFC) model in PELE, to see whether the mixing process described above can produce 'excess' olivine crystallisation and so account for the higher olivine modes ( $\sim 55 \%$ ) and by extension, the higher $\mathrm{MgO}(\sim 25 \mathrm{wt} \% \mathrm{MgO})$ contents observed in the LPS MOZ. We simulated in PELE an olivine slurry (containing 30\% olivine and $70 \%$ melt) that assimilates a gabbroic mush (corresponding to an average FCSC DZ composition). The output from this model shows that $\sim 5 \%$ more olivine crystallisation is produced prior to the cotectic. Olivine Fo-contents are also buffered at $\mathrm{Fo}_{82}$, compared to closed-system models where Fo decreases rapidly with crystallisation. These olivine compositions are similar to the composition of olivines observed in the LPS MOZ. The AFC model results (appendix F4) support the hypothesis that the LPS MOZ rocks crystallised from the hybrid melt, consistent with the bulk-mixing model presented above (Fig. 5.9).

### 5.5.8 Lead and sulphur isotopic constraints on the petrogenesis of the FCSC

The spread of Pb -isotope compositions (Fig. 5.5) indicates that several different magmas were involved in the generation of the olivine-enriched Franklin sills. The distribution of compositions in ${ }^{208} \mathrm{~Pb} /{ }^{204} \mathrm{~Pb} v s^{206} \mathrm{~Pb} /{ }^{204} \mathrm{~Pb}$ and ${ }^{207} \mathrm{~Pb} /{ }^{204} \mathrm{~Pb} v s{ }^{206} \mathrm{~Pb} /{ }^{204} \mathrm{~Pb}$ plots could in part
be explained by the assimilation of Shaler Supergroup crustal rocks. The problem is that it is unclear which isotopic composition represents the 'uncontaminated' component. Some of the spread towards the WUS OZ/DZ isotopic compositions could be explained by small ( $<10 \%$ ) amounts of dolostone assimilation, since the composition of Shaler dolostone falls along a suitable mixing trajectory (Beard, 2012). Because the WUS OZ/DZ have similar trace element compositions with enriched Pb concentrations and distinctive radiogenic ${ }^{206} \mathrm{~Pb} /{ }^{204} \mathrm{~Pb}$ and ${ }^{207} \mathrm{~Pb} /{ }^{204} \mathrm{~Pb}$ isotope compositions, we suggest they shared a similar melt source composition and contamination history. Conversely, the unradiogenic Pb isotopic composition seen in the LPS DZ might reflect assimilation of Archaean Basement (Beard, 2012). Whether the range of isotopic compositions in the Franklin magmas is best explained by a heterogeneous mantle source, or a common source magma that was variably contaminated by different crustal components will be discussed in Beard et al. (In Prep).

What is pertinent here is that the LPS OZ and LPS DZ have different Pb - and S -isotope signatures (Fig. 5.5 \& 5.6), consistent with the composite intrusion model presented in chapter 4. In Pb -isotope plots, mixing lines between two geochemical reservoirs are straight lines. As the LPS OZ plots above a tie line between the WUS OZ/DZ and the LPS DZ, a simple mixture of these two magmas cannot explain its isotopic signature (Fig. 5.5 b ), in apparent contradiction with the hybridisation model we favour (Fig. 5.9). If the FCSC does indeed represent a single sill system, then the olivine slurry replenishment event may have been internally heterogeneous or there were two isotopically distinct olivine slurry replenishment events in the FCSC. This is discussed further below.

We suggest that the igneous rocks preserved at the WUS represent a vigorous replenishment event, which may have pushed previously emplaced magmas downstream towards the LPS. Pb-isotope compositions suggest that this final pulse of magma assimilated small amounts of dolostone (< $10 \%$ ) relative to the average Type-1 magma (Fig. 5.5). We speculate that early injections of magma, which created the WUS, are preserved only at the hybrid LPS OZ. These assimilated a smaller fraction of dolostone than the final pulse of magma, now preserved as the WUS. The original, pre-hybridisation olivine-rich LPS OZ component would have been isotopically distinct from the WUS. In this context, the gradual shifts in $\mathrm{OZ} \mathrm{Fe} / \mathrm{Mg}$ in the FCSC that we attribute to downstream hybridisation (Fig. 5.9) are superimposed on a secular variation in the
geochemical/isotopic signatures of the incoming primitive magmas. $\mathrm{Fe} / \mathrm{Mg}$ ratios are buffered by fractional crystallisation, however, radiogenic isotope compositions are unaffected by crystallisation. Therefore, the FCSC records the injection of isotopically distinct magmas.

### 5.5.9 The potential effects of dolostone assimilation on the FCSC magma(s)

The Franklin intrusions of the Minto Inlier are hosted by a predominantly carbonate/dolostone sedimentary sequence (Thomson et al., 2014), and calc-silicate reaction products are locally prominent (Nabelek et al., 2013). We also see field and thin section evidence of dolostone assimilation in the UM (Bédard et al., 2012). If we are correct in our inference that the FCSC magmas are rooted in the UM upflow zone, then fault-related brecciation of host dolostones at the UM (Bédard et al., 2012) may have allowed more extensive assimilation as magma migrated up the fault before propagating laterally along the mechanical discontinuity represented by the contact between the Fort Collinson and Jago Bay Formations.

We investigated the potential impacts of dolostone assimilation on melt and mineral chemistry using PELE. We used the average FCSC chill as our starting composition (10.3 $\mathrm{wt} \% \mathrm{MgO}$ ) and added a dolostone contaminant to the magma ( $0.2 \mathrm{~g}, 0.4 \mathrm{~g}, 0.5 \mathrm{~g}$ and 1 g of contaminant added per fractionation step). The composition of the contaminant was taken as the average of 26 dolostone bulk-rock analyses from sedimentary rocks of the Minto Inlier (appendix F5). The addition of this high-MgO ( $\sim 20 \mathrm{wt} \%$ ) and low-FeO (< $1 \mathrm{wt} \%$ ) dolostone contaminant to the magma has major impacts on the LLD, when compared to regular fractional crystallisation. Prior to olivine saturation, the AFC models with the most dolostone contaminant predict an increase in melt MgO content up to $11.7 \mathrm{wt} \%$ and decrease the melt $\mathrm{FeO} / \mathrm{MgO}$ ratio. In this early stage, abundant $\mathrm{CO}_{2}$ gas is released. Primitive olivine (up to $\mathrm{Fo}_{89}$ ) crystallises at temperatures of $\sim 1260^{\circ} \mathrm{C}$, earlier than in closed-system models for this melt composition (Fig. 5.10). Between $1260^{\circ} \mathrm{C}$ and $1200^{\circ} \mathrm{C}$, $\sim 7 \%$ olivine crystallises, $\sim 16 \%$ dolostone is assimilated, and $7 \% \mathrm{CO}_{2}$-rich gas is released. In contrast to fractionation models without dolostone assimilation where olivine Focontents decrease rapidly (Fig. 5.10), assimilation of MgO-rich dolostone buffers olivine Fo-contents at values of $\sim \mathrm{Fo}_{88-87}$ in the $1260-1200^{\circ} \mathrm{C}$ interval (Fig. 5.10). Clinopyroxene
begins to crystallise earlier in the dolostone-AFC models compared to the closed-system model due to the high CaO -contents (24-47 wt\%) of the dolostone contaminant, with up to $\sim 20 \%$ clinopyroxene being extracted before the arrival of plagioclase. In contrast, both clinopyroxene and plagioclase join olivine on the liquidus simultaneously in closed-system models.


Fig. 5.10: Forsterite (Fo) versus temperature diagram showing the LLD of the average FCSC magma during fractional crystallisation (FC) and AFC. Dolostone is the contaminant, which has been added to the magma in varying proportions per fractionation step. Symbols indicate the arrival of a phase on the liquidus (triangles $=$ clinopyroxene; squares $=$ plagioclase; star $=$ both clinopyroxene and plagioclase). Models are included in appendix F6.

The modelled effects of dolostone assimilation on the FCSC magma are very similar to those observed experimentally (Iacono Marziano et al., 2008, Mollo \& Vona, 2014) and resemble products of volcanic and plutonic systems where assimilation of host carbonates occurred (Wenzel et al., 2002, Chadwick et al., 2007, Di Rocco et al., 2012, Troll et al., 2013). Dolostone assimilation in the FCSC magma could help explain the high-Fo compositions of olivine cores in the WUS OZ. Additionally, the effects of volatile and gas budgets within crystal-mushes have been shown to affect their viscosity. Therefore the
production of $\mathrm{CO}_{2}$ caused by dolostone assimilation may have reduced the viscosity of the crystal-rich FCSC olivine slurry during downstream propagation (Pistone et al., 2013) and facilitated hybridisation.

### 5.5.10 Differentiation model for the FCSC plumbing system

The evidence presented is consistent with a model where the FCSC OZ formed from a late injection of olivine-enriched magma that mixed with a resident, partly consolidated gabbroic mush. This olivine slurry was sourced from upstream in the FCSC, possibly from the UM feeder system.

The isotopic data indicate that there are multiple isotopic signatures within the FCSC. One potential explanation for this is that the olivine slurry replenishment was internally isotopic heterogeneous upon emplacement. The olivine slurry may have acquired varying isotopic signatures deeper in the plumbing system when fresh batches of magma were injected into the plumbing system, possibly during remobilisation of the olivine cargo. However, in this scenario it would be highly fortuitous for the WUS OZ and DZ to be in isotopic equilibrium. We therefore favour a model in which the replenishing olivine-rich magma was injected as two pulses, showing secular shifts in radiogenic isotope composition. Magmas may have been variably contaminated with a diverse range of crustal facies, including dolostone. Even though the olivine slurry was isotopically heterogeneous, the FCSC solidified to form similar textural and petrological facies at the WUS, PS and LPS. In this model, the LPS DZ isotopic signature (Fig. 5.5) represents the earliest pulse of magma into the FCSC. This component could record minor amounts of Archaean basement or sedimentary rock contamination (Beard et al. In Prep). Whether isotopically distinct magma pulses represent a diversity of contamination paths imposed on a common magma stem, or whether these are discrete samples of an isotopically heterogeneous mantle, is beyond the scope of this chapter. An olivine-enriched replenishment seems to have propagated westward through the FCSC, mixing en route with a resident gabbroic mush to create the LPS OZ. A second olivine slurry pulse seems to have experienced more extensive dolostone contamination, as recorded by the Pb isotopes (Fig. 5.5). The olivine slurry replenishment events appear to have been widespread, since Type-1 olivine-enriched sills (Kat's, Dick's and Uwe's sills) are common at the stratigraphic level where the FCSC
was emplaced. Kat's sill is similar to the WUS in that it shows isotopic equilibrium between OZ and DZ , and has isotopic signatures that are very similar to those of the WUS.

Kat's sill is located just to the south of the FCSC (Fig. 2.2) and may represent another branch of the WUS-forming pulse. In contrast, Dick's and Uwe's sills are isotopically distinct and would record different, non-unique sources/differentiation paths. As the OZ of Uwe's sill has similar radiogenic isotope and trace element ratios to rocks from the WUS and Kat's sill, olivine with the dolostone assimilation signature probably crystallised prior to emplacement of this pulse of magma.


Fig. 5.11: Schematic diagram illustrating the formation of the FCSC and the basal lava sequence of the Natkusiak flood basalts (NFB). (a) Deep (> 3 km ) holding sills or feeder conduits are fed by fresh batches of primitive melt from isotopically distinct mantle sources; (b) After initial residual magmas escape from the holding system, primitive olivine cumulates with equilibrium melts are left behind before being remobilised (by overpressure from fresh melt deliveries?); (c) The remobilised olivine slurry ascends up syn-magmatic Neoproterozoic faults (arrows with squiggly contacts), where they assimilate dolostone country rock; (d) Olivine slurry is emplaced into the FCSC, which already hosts a partly-crystallised gabbroic mush derived from an earlier magma; (e) Residual melts to the sills may be erupted at the surface, if structures facilitate melt escape. The erupted melt forms the basal NFB lava sequence, and if melts affected by dolostone assimilation were erupted, then significant quantities of $\mathrm{CO}_{2}$ may have been released. $\mathrm{BI}=\mathrm{Boot}$ Inlet Formation; FC = Fort Collinson Formation; JB = Jago Bay Formation.

We present a possible model for the development of the olivine-enriched rocks of the FCSC in Fig. 5.11. The emplacement dynamics of these heterogeneous olivine slurry replenishment events are similar to the 'braided sills' differentiation model of the Torres Del Paine Mafic Complex in Patagonia (Leuthold et al., 2014). Also, our assimilationreplenishment scenario is analogous to the Roelofse and Ashwal (2012) model proposed for the formation of the Main Zone in the Bushveld Complex. In their model, crustal contamination of crystal-mushes occurred in sub-compartments of an underlying feeder system. A similar scenario may have operated beneath the FCSC at > 3 km depth in the Franklin plumbing system. It is also possible that extensive contamination occurred during transport of magma up fault-guided conduits populated by cataclastic breccias (Bédard et al., 2012). Ultimately, our database of lateral variations in the compositions of the olivineenriched sills of the Minto Inlier provides insight into magma/crystal-slurry flow directions. These have important implications for the evolution of magma plumbing systems, intra-sill mixing and in the search of economic deposits.

### 5.6 Conclusion

The Fort Collinson Sill Complex (FCSC) of the Franklin Large Igneous Province, extends for $>50 \mathrm{~km}$ along strike, and its constituent sills are characterised by basal olivineenriched layers (OZ) and capping gabbros (DZ). We investigated three sections through the FCSC. At its eastern end is the West Uhuk Sill (WUS), in the centre is the P-sill (PS) and at its western end is the Lower Pyramid Sill (LPS). The FCSC OZ rocks exhibit a gradual shift towards more Fe -rich bulk compositions towards the west. Olivine cumulate arrays on $\mathrm{Fe}=\mathrm{Mg}$ plots were used to define bulk crystal-extracts and equilibrium melts, which shift to more evolved compositions towards the west. However, the predicted PS and LPS olivine compositions only approximately match the olivine cargo they contain suggesting excess olivine was added, perhaps by injection of a crystal slurry. The primitive olivine primocrysts within the FCSC OZ were derived from a replenishing olivine slurry, which was emplaced into and mixed with a partly crystalline gabbroic mush. Downstream propagation and hybridisation between these two components can explain the westward Fe-enrichment trend in the FCSC OZ. Radiogenic isotope data suggest that there were at least three geochemically distinct pulses of magma in the FCSC. Isotopic signatures in the WUS in particular can be explained by small amounts of dolostone assimilation, which
buffers the melt $\mathrm{Mg} / \mathrm{Fe}$ ratio enabling crystallisation of the higher-Fo olivine that characterises the WUS. In addition, clinopyroxene is preferentially stabilised on the liquidus ahead of plagioclase and significant quantities of $\mathrm{CO}_{2}$ are released, during dolostone assimilation. The similarities between the FCSC and other olivine-enriched Franklin sills suggest that olivine slurry replenishment events are common, and that the olivine-rich slurries were able to propagate $>50 \mathrm{~km}$ along strike.

# 6. Sulphide immiscibility induced by wallrock assimilation in a fault-guided Franklin magmatic feeder system 

### 6.1 Abstract

The Southern Feeder Dyke Complex (SFDC) is part of the Franklin Large Igneous Province (LIP), which is well exposed in the Minto Inlier of Victoria Island (Arctic Canada). Previous work on the Franklin LIP considers its igneous rocks to be prospective for $\mathrm{Ni}-\mathrm{Cu}-\mathrm{PGE}$ sulphide mineralisation. We investigated the SFDC, which comprises a series of NW-SE trending gabbroic outcrops and sedimentary panels with contact zones that correspond to $1^{\text {st }}$ derivative magnetic anomalies. Field and textural relationships show that the SFDC was emplaced contemporaneously with regional Neoproterozoic faulting. Gabbroic dykes intruded brecciated country rock, and microdoleritic microxenoliths in basaltic matrices indicate multiple intrusive/brecciation events. Host rocks are commonly overprinted by greenschist facies assemblages, with calcite + pyrite veins filling open spaces between breccia fragments. Dykes emplaced into these breccias show globular and net-textured sulphides within microdoleritic microxenoliths that appear to impregnate a porous crystal mush suggesting that immiscibility was triggered on a local scale, presumably by assimilation of local wallrock. Heavy $\delta^{34}$ S isotopic signatures in these dykes also indicate assimilation of sulphur-bearing country rocks. Wallrock assimilation would have been facilitated by fault-related brecciated of the country rocks, which would expose extensive xenolith surface areas to fresh magma. Significant gossan showings associated with a sill up-section from the SFDC suggests that immiscible sulphide liquids may have been flushed downstream by composite dyke systems that experienced magma replenishment. Fault-mediated melt ascent along NW-SE faults has been documented elsewhere in the Minto Inlier, providing opportunities for similar wallrock assimilation and sulphide redistribution events. The evidence preserved in the SFDC confirms the Ni-CuPGE potential of the Franklin LIP.

### 6.2 Introduction

Magma conduits such as dikes are dynamic systems that provide a setting for magma-crust interaction (Naldrett, 1992, Li et al., 2000, Maier et al., 2001, Ding et al., 2010, Lightfoot \& Evans-Lamswood, 2015), and are important sites of Ni-Cu-PGE sulfide mineralization (Ripley \& Li, 2011). Economic conduit-type Ni-Cu-PGE sulfide deposits are typically associated with olivine-rich magmas (Li \& Naldrett, 1999); are commonly situated close to major crustal faults (Naldrett, 1992, Begg et al., 2010, Song et al., 2012, Lightfoot \& Evans-Lamswood, 2015); are generally hosted by (or related to) intrusive or extrusive facies that show a depletion in chalcophile elements (Naldrett, 1992); are commonly emplaced into (or traverse) sulfur-bearing country rocks (Naldrett, 1992, Li et al., 2009); and may show field and/or geochemical evidence of interaction between magma and country rock (Ripley et al., 2003, Li et al., 2009, Keays \& Lightfoot, 2010, Ganino et al., 2014). Most economic Ni-Cu-PGE sulfide deposits of this type are thought to have involved addition of external sulfur to a sulfur-undersaturated magma. Other possible mechanisms to induce sulfide immiscibility include changes of melt composition (Irvine, 1975, Czamanske \& Moore, 1977, Wendlandt, 1982), temperature (Haughton et al., 1974) or $\mathrm{fO}_{2}$ (Buchanan \& Nolan, 1979, Liu et al., 2007). Additionally, a hydraulic trap is commonly invoked to concentrate sufficient volumes of immiscible sulfide liquids (Naldrett, 1992); with an appropriate R-factor, a measure of the ratio of silicate magma to the sulfide liquid, being needed to upgrade the metal tenor of immiscible sulfide liquids (Campbell \& Naldrett, 1979). In conduit-type deposits, upgrading would most likely occur in the context of an open magmatic system, where a previously concentrated sulfide melt would see its PGE tenor upgraded by magma throughflow (Naldrett, 1992).

Examples of conduit-type Ni-Cu-PGE sulfide deposits include the Noril'sk-Talnakh intrusions in Siberia (Naldrett, 1992, Arndt, 2011), Voisey's Bay in Labrador (Naldrett et al., 2000, Ripley \& Li, 2011) and the Eagle deposit in northern Michigan (Ding et al., 2010). Each of these magma conduits is associated with craton margins, regional faultsystems and are comprised of differentiated (mafic-ultramafic intrusions). Additionally, there is petrological and geochemical evidence that magma-crust interaction triggered the formation of immiscible sulfide liquids in each of these magma conduits. Data on $\delta^{34} \mathrm{~S}$ isotopes is especially informative about incorporation of crustal sulfur, because crustal
sulfur is typically enriched in comparison to mantle sulfur (Fiorentini et al., 2012, Ripley \& Li, 2013, Black et al., 2014). The heavy $\delta^{34}$ S isotopic values ( 6 to $14 \%$ ) of mafic rocks at Noril'sk have been explained by the ingestion of crustal sulfur into the magma (Grinenko, 1985, Li et al., 2009). However, other Ni-Cu-PGE sulfide deposits such as Jinchuan in China (Ripley et al., 2005, Lehmann et al., 2007) and Nebo-Babel in Australia (Seat et al., 2009) do not show isotopic evidence for the involvement of external sulfur. For deposits where external sulfur is considered an important factor, a key issue is how and where in the magma plumbing system the sulfur was incorporated into the magma (Ripley et al., 2003, Ripley \& Li, 2013).

The metallogenetic setting of the Franklin Large Igneous Province (LIP) in northern Canada (Fig. 1) shares many similarities with those of other economic conduit-type Ni-CuPGE sulfide deposits (Jefferson et al., 1994). The Franklin LIP is situated at the margin of the Slave craton, a typical feature of Ni-Cu-PGE sulfide deposits (Begg et al., 2010). A Geological Survey of Canada mapping project (2010 and 2011) aimed to constrain the Ni-Cu-PGE potential of the Franklin LIP, considered prospective by Jowitt and Ernst (2013). The associated Natkusiak flood basalts have a known native copper potential (Jefferson et al., 1985), not unlike the Keewenawan Basalts, which are associated with sulfide mineralization in the Duluth Complex (Miller et al., 2002). In the Minto Inlier (Fig. 1), the Franklin igneous suite is dominated by mafic and ultramafic-mafic sills with subordinate NNW/NW-trending feeder dikes (Bédard et al., 2012). There are small ( $\sim \mathrm{mm}$ disseminated sulfides) sulfide showings in the Franklin sills and dikes, as well as larger ( $\sim 1$ km diameter) surface gossans of uncertain origin (Peterson et al., 2014). The Southern Feeder Dike Complex (SFDC) within the Minto Inlier is a NNW/NW-trending dike system with associated sulfide showings. The SFDC preserves evidence of magma emplacement that was contemporaneous with regional-scale Neoproterozoic normal faulting. We carried out a field, geochemical (bulk-rock and in-situ), $\delta^{34}$ S-isotope and anisotropy of magnetic susceptibility study of the SFDC in order to constrain how magmas were emplaced, how magmas interacted with host rocks, how the associated sulfides formed, and to better understand the impact of the syn-magmatic faulting on the overall system. Field evidence (Bédard et al., 2012) and results of a companion $\delta^{34} \mathrm{~S}$-isotope study (Hryciuk et al. submitted), suggest that fault-related brecciation commonly facilitated the ingestion of crustal sulfur into Franklin magmas. In the SFDC, some of the small- and medium-scale processes by which this occurred are preserved. Our results bear on the Ni-Cu-PGE sulfide
potential of the Franklin LIP, and provide insights into the processes involved in the generation of magmatic Ni-Cu-PGE sulfides elsewhere, particularly those that operated in conduit-type magmatic plumbing systems.


Fig. 6.1: (A) Geological map of the SFDC with each zone labelled. Regional bedding typically dips a few degrees towards the SE, except for perturbations near to the SFDC. (B) The SFDC coincides with a $1^{\text {st }}$ derivation NNW/NW-trending magnetic anomaly (Kiss \& Oneschuk, 2010). The locations of the mapped Zones are labelled. Circles with labels correspond to outcrop localities or 'stations' that were used to sync the SFDC with the Geological Survey of Canada mapping system. The 'climbing sill' is a separate locality that will not be discussed here.

### 6.3 Southern Feeder Dyke Complex

The SFDC (Fig. 6.1) was briefly discussed in Bédard et al. (2012) as an example of a synfaulting dyke system. The SFDC is situated $\sim 10 \mathrm{~km}$ south of the Minto Inlet (Fig. 2.2), and is located along a NNW/NW-trending magnetic lineament (Fig. 6.2). On the kilometer scale, the area is characterised by a series of NW- or SE-trending gabbroic and sedimentary panels that are separated by prominent linear valleys or scarps, and which are interpreted to be faults and/or intrusive contacts (Fig. 6.1). The gabbroic outcrops comprise sills that form capping mesas, and dyke-like bodies that are locally exposed in valleys that commonly have a thick Quaternary cover. The intrusive rocks were emplaced into southeasterly younging and shallowly SE-dipping bedded carbonates and evaporites of the Wynniatt and Kilian Formations of the Shaler Supergroup at $\sim 1 \mathrm{~km}$ paleodepth. We examined the SFDC in five, well-exposed areas along a $\sim 6 \mathrm{~km}$ transect. We will describe the field relationships of each from NW to SE (i.e., moving up-section). Also labeled on Fig. 2.2 are the locations of the Gossan Sill (GS) and the Sulphide City Dyke (SCD), which both represent gabbroic bodies emplaced at a similar stratigraphic level. These will be discussed in future studies.

### 6.3.1 Zone D

Zone D is at the northwestern end of the SFDC (Fig. 6.1). In this area, a large gabbroic body (SW end of Fig. 6.3) at the ridge top forms a capping mesa (Figs. 6.3 and 6.4A). Preserved contact relationships imply that this body is a sill that extends for tens of kilometers towards the SW. Contacts with the host Wynniatt Formation are poorly exposed as the SFDC is approached, however, probably due to renewed post-intrusive motion on faults. The contact region of this capping sill with the SFDC (labeled 'b' on Fig. 6.3) shows a bench of strongly recrystallised bedded marble (labeled ' $m$ ' on Fig. 6.3) and a disrupted outcrop of interlayered calc-silicate and Fe -oxide skarn rock. These facies closely resemble contact metamorphic rocks seen at the Uhuk Massif where there are faultguided up-section magmatic transgressions (Bédard et al., 2012, Nabelek et al., 2013). About 50 meters NE of this hidden fault contact is an irregular body of gabbro, $\sim 20 \mathrm{~m}$ thick (labeled 'a' on Fig. 6.3) that is embedded in contact-metamorphosed limestone. The
gabbro has chilled margins at both contacts, as well as internal chills. Bedding in the host Wynniatt Formation increases markedly in dip towards the margins of the dyke (from $5^{\circ}$ to $>70^{\circ}$ ), with drag fold polarity indicating NE-side-down motion on a NNW/NW-trending fault (Fig. 6.3). The host Wynniatt Formation adjacent to the dyke is recrystallised as well as heavily fractured and brecciated. We interpret this part of Zone $D$ to be a downdropped hanging wall limestone panel. Magma was injected into parasitic faults within this downdropped panel to form a gabbroic roof apophysis, similar to what was seen at the Uhuk Massif (Bédard et al. (2012).


Fig. 6.3: A detailed geological map of Zone D, showing the moderately dipping NNW/NW-trending dykelike roof apophysis emplaced into moderately dipping Wynniatt Formation country rock. The line of sight for the field photo in Fig. 6.4A is shown.


Fig. 6.4: (A) Field photo of Zone D, showing the dyke-like roof apophysis feeding towards the capping sill (dashed red line shows inferred contact region and dashed white lines show gently dipping Wynniatt Formation rocks adjacent to the sill). (B) Brecciated and Fe-stained Wynniatt Formation country rocks at Zone B (pencil is 5 cm ). (C) Igneous breccia with the dyke propagator tips at Zone B. Microphyric basalt ( FgG ) hosts xenoliths of microdoleritic sulphide-bearing gabbro ( MgG ). The dyke propagator tips have a green colouration, possibly because of greenschist grade alteration. (D) Field photo of Zone A showing the steeply dipping dyke feeding and splaying towards the capping sills (dashed red lines are igneous contacts). The Kilian Formation bedding dips steeply towards the dyke (dashed white lines). The dyke is gossan stained (labelled) and dips towards the SW, with the dark material extending towards the viewer being talus of basalt deposited on limestone. It is unclear from the outcrop, which sill the dyke feeds into.

### 6.3.2 Zone C

Zone C (Fig. 6.5) is situated $\sim 2 \mathrm{~km}$ to the SE of Zone D (Fig. 6.1). Limited exposures reveal a $\sim 20 \mathrm{~m}$ thick NNW-trending gabbroic dyke that has steeply dipping margins ( $\sim 75^{\circ}$ to the SW), and preserves a chilled margin on the western side (Fig. 6.5). The Wynniatt Formation in contact with this chilled margin is heavily fractured and bedding increases in dip towards the dyke (Fig. 6.5: up to $\sim 27^{\circ}$ to the NE). The dip directions on either side of
the gabbroic dyke indicate NE-side-down motion on a fault, as for Zone D. There is a thinner parallel $(\sim 1 \mathrm{~m})$ near-vertically dipping $\left(\sim 85^{\circ}\right)$ dyke that appears to merge into the thicker dyke over a distance of $\sim 15 \mathrm{~m}$ (Fig. 6.5). The gabbroic body at Zone C is interpreted to be an extension of the main gabbroic intrusion (labeled 'b' on Fig. 6.3) of Zone D.


Fig. 6.5: A detailed geological map of Zone C showing the dyke and local Wynniatt Formation bedding perturbations.

### 6.3.3 Zone B

Zone B (Fig. 6.6) is located $\sim 0.5 \mathrm{~km}$ to the SE of Zone C (Fig. 6.1). A rubbly mound ( $\sim 20$ $m$ thick) of dark, NNW/NW-trending gabbroic rock can be seen at the poorly exposed northeastern end of Zone B. No contacts are preserved, but the thickness and orientation of the gabbro mound, as well as its proximity to Zone C , suggests it may be a continuation of the Zone C dyke. At the SW end of Zone B (Fig. 6.6) there are several, thinner ( $\sim 0.5-2 \mathrm{~m}$ thick) N/NNE-trending dykes, which are oblique to the larger NNW/NW-trending dyke. Some of these thinner dykes can be traced for hundreds of meters through the host

Wynniatt Formation and may connect to the larger NNW/NW-trending dyke (Fig. 6.6). The bedding in the Wynniatt Formation limestone hosting the thinner dykes dips (up to $\sim 45^{\circ}$ ) towards the SE, and is commonly brecciated, with veins being filled by calcite, prehnite, quartz, pyrite and sparse chalcopyrite (Fig. 6.4B). The thin N/NNE-trending dykes intrude the brecciated country rocks, and we infer them to be dyke propagator tips that were emplaced into an active fault. In places, the breccias are igneous, with microdoleritic microxenoliths cemented by microphyric basalt (Fig. 6.4C). Many of these breccias have a distinctly green colouration, indicating a strong greenschist-grade hydrothermal overprint (Fig. 6.4C) that may have been driven by heat derived from repeated emplacement of dykes into the active fault. These greenschist facies breccias are cut by brown weathered, weakly altered, and locally brecciated dykes, which appear to represent the last igneous event. These late dykes and breccias commonly contain disseminated sulphides, typically pyrite, pyrrhotite and minor chalcopyrite.


Fig. 6.6: A detailed geological map of Zone B showing the inferred location of the main NNW/NW-trending dyke as well as the N/NNE-trending dyke propagator tips that extend towards the south into brecciated Wynniatt Formation country rock.

### 6.3.4 Zone E

Zone E (Fig. 6.7) is located midway between Zones B and A (Fig. 6.1). We did not create a detailed map of Zone E because of the limited size of the outcrop (Fig. 6.7A). Zone E consists of a $\sim 20 \mathrm{~m}$ thick NNW-trending gabbroic body emplaced into the Kilian Formation. Both margins of the gabbroic body are preserved, dipping very steeply towards the SW $\left(\sim 80^{\circ}\right)$, and both are chilled contacts. This gabbroic body is interpreted to be a dyke, which is parallel to the NNW-trending dykes of Zone D-C-B. The host Kilian Formation, near to the dyke margins, is strongly discoloured (Fig. 6.7B), has steeper dips ( $\sim 40^{\circ}$ to the SW ), and is heavily brecciated, with the fractures containing coarse $(1 \mathrm{~cm}$ sized) pyrite and calcite. Parasitic folds with NW-trending hinge lines are observed up to $\sim 10 \mathrm{~m}$ away from the dyke contact (Fig. 6.7B). The deflection of the Kilian Formation bedding indicates SW-side-down motion on a NNW-trending fault, the opposite sense to Zone D-C.


Fig. 6.7: (A) Field photo of Zone E showing the dyke outcrop (dashed red lines show igneous contacts). The location of photo B is shown at the western margin of the dyke. (B) Folded and discoloured Kilian Formation bedding at the western margin of the dyke. Hinge lines trend NW, parallel to the dyke.

### 6.3.5 Zone A

Zone A (Fig. 6.8) is at the southeastern end of the SFDC, and represents the highest stratigraphic level of the SFDC we have examined. A $\sim 20 \mathrm{~m}$ thick gabbroic dyke that trends NNW/NW has chilled margins against Kilian Formation sulphate evaporates that dip $\sim 80^{\circ}$ to the SW (Fig. 6.4D). The dyke is slightly offset ( $<10 \mathrm{~m}$ ) by ENE-trending postPalaeozoic faults (Fig. 6.8). The dips of the bedding in host rocks increases (up to $\sim 32^{\circ}$ to the SW) at dyke margins, indicating SW-side-down motion on a NNW/NW-trending fault, as for Zone E. We interpret this dyke to be a direct continuation of the Zone E dyke (Fig. 6.1). The dyke at Zone A contains disseminated sulphides at its margin, as well as sulphide staining in its core (Fig. 6.4D). The dyke appears to terminate where it joins a set of conformable gabbroic sills (Figs. 6.1 and 6.4D). Two sills are well developed towards the NE, but only one, poorly exposed sill can be traced towards the SW. The junction of these two sets of sills with the Zone A dyke is rubbly and poorly exposed (Fig. 6.4D), such that we cannot ascertain from field relationships if the dyke fed the sills.


Fig. 6.8: A detailed geological map of Zone A, showing an inferred continuous dyke that splays towards two capping sills. The local Kilian Formation (sulphate-bearing) bedding is significantly perturbed near to where the dyke splays.

### 6.4 Results

### 6.4.1 Petrography and mineral chemistry

There are a variety of igneous and country rock facies preserved in the SFDC, including chilled intrusive contacts, medium-grained doleritic to gabbroic rocks, brecciated host rocks, all of which may show hydrothermal overprints. Chilled contacts typically grade inwards over a few cm to medium-grained doleritic facies. In-situ mineral compositions for olivine $(\mathrm{Fo}=100 \times \mathrm{Mg} /(\mathrm{Mg}+\mathrm{Fe} 2+))$, clinopyroxene $\left(\mathrm{Mg} \#=100 \mathrm{x} \mathrm{Mg} /\left(\mathrm{Mg}+\mathrm{Fe}^{2+}+\mathrm{Mn}\right)\right)$ and plagioclase $(\mathrm{An}=100 \times \mathrm{Ca} /(\mathrm{Ca}+\mathrm{Na}+\mathrm{K}))$ are provided where analysed. 'Core' corresponds to the centre of the crystal, 'mantle' corresponds to the intermediate region and 'rim' corresponds to the edge of the crystal.

At Zone D, the capping sill is composed of sub-ophitic gabbro with euhedral, normally zoned (core $\mathrm{An}_{70}$; rim $\mathrm{An}_{50}$ ) plagioclase chadacrysts hosted by anhedral, normally zoned (core Mg\#80; rim Mg\#45) clinopyroxene oikocrysts (Fig. 6.9A). Minor (< 15\%) olivine and interstitial ilmenite-magnetite are also present. The gabbroic body/roof apophysis (labeled ' $a$ ' on Fig. 6.3) has a chilled margin that is microphyric with phenocrysts of olivine (core $\mathrm{Fo}_{80}$ and $\mathrm{NiO} 0.25 \mathrm{wt} \%$ ) and clinopyroxene (core $\mathrm{Mg} \# 83$; rim $\mathrm{Mg} \# 80$ ) residing in a very fine-grained groundmass. This chilled margin grades into mediumgrained gabbro that is texturally identical to the capping sill gabbro, with intergrown plagioclase (core $\mathrm{An}_{60}$; rim $\mathrm{An}_{50-45}$ ) and clinopyroxene (core $\mathrm{Mg} \# 75$; rim $\mathrm{Mg} \# 40$ ) and isolated olivine crystals (core $\mathrm{Fo}_{75}$ and $\mathrm{NiO} 0.15 \mathrm{wt} \%$ ). The internal chill in the gabbroic roof apophysis is very similar to the outer chilled margin with sparse clinopyroxene (core $\mathrm{Mg} \# 79$; rim $\mathrm{Mg} \# 76$ ) and plagioclase (core $\mathrm{An}_{70}$; rim $\mathrm{An}_{68}$ ) phenocrysts (no olivine was observed in the internal chill).

The Zone C dyke is very similar to the gabbroic roof apophysis at Zone D. A chilled margin grades into a medium-grained gabbro. However, at Zone C the dyke contains interstitial-textured olivine (Fig. 6.9B) with compositions that are more evolved $\left(\sim \mathrm{Fo}_{50}\right.$ and $\mathrm{NiO}<0.1 \mathrm{wt} \%$ ) in comparison to Zone D olivine. Both clinopyroxene (core Mg\#80; rim $\mathrm{Mg} \mathrm{\# 63}$ ) and plagioclase (core $\mathrm{An}_{70}$; rim $\mathrm{An}_{55}$ ) core compositions at Zone C are very similar to those of Zone D, but crystal rims are more evolved at Zone C.


Fig. 6.9 (page 109): Photomicrographs of minerals and textures from the SFDC (FOV = field of view): (A) Clinopyroxene ( Cpx ) oikocryst enclosing plagioclase ( Plg ) chadacrysts (image in crossed-polars and FOV $=$ 4 mm ) at Zone D. (B) Interstitial olivine ( Ol ) intergrown with plagioclase and clinopyroxene in the Zone C dyke (image in crossed-polars and $\mathrm{FOV}=4 \mathrm{~mm}$ ). (C) Plane-light image ( $\mathrm{FOV}=2 \mathrm{~mm}$ ) of igneous sulphide trapped in a wedge-shaped pore in Zone B dyke-propagator tip igneous breccia. (D) Reflected light image ( $\mathrm{FOV}=1 \mathrm{~mm}$ ) of sulphide from image (C), which consists of pyrite (Py) and pyrrhotite (Po). (E) Plane-light image ( $\mathrm{FOV}=1 \mathrm{~mm}$ ) of microdoleritic microxenolith in the Zone B dyke propagator tip igneous breccia. Note that the microdoleritic microxenolith is coarser than the quenched microphyric basalt in which it is embedded, suggesting it represents a wallrock fragment, and that it hosts several globular concentrations of sulphide. ( F ) Plane-light image ( $\mathrm{FOV}=1 \mathrm{~mm}$ ) and ( G ) reflected light image ( $\mathrm{FOV}=1 \mathrm{~mm}$ ) close-ups of the two patches in lower centre of the microdoleritic microxenolith shown in (E). Note that sulphide impregnates the doleritic textured silicates. (H) Another example of sulphide melt impregnation of microphyric dolerite (reflected light with FOV $=1 \mathrm{~mm}$ ). (I) Clinopyroxene phenocryst intergrown with plagioclase in the microphyric basalt of the Zone B dyke propagator tips (plane-light and FOV $=1 \mathrm{~mm}$ ). (J) Clinopyroxene phenocryst with spongy core (red arrows) hosted within groundmass at Zone A (plane-light and FOV =1 mm ).

At Zone B, the Wynniatt Formation limestones surrounding the narrow dykes we interpret to be propagator tips are heavily brecciated (Fig. 6.4B) with fractures/veins containing abundant calcite and hydrothermal pyrite crystals up to 1 cm in size. Veins also contain prehnite, quartz and sparse chalcopyrite. The brecciated limestone is crosscut by the dyke propagator tips, which themselves are brecciated by younger basaltic dykes. These brecciated propagator tips are composed of microdoleritic microxenoliths, cemented by microphyric basalt (Fig. 6.9I), which has reversely zoned clinopyroxene (core Mg\#81; rim $\mathrm{Mg} \# 83$ ) and normally zoned plagioclase (core $\mathrm{An}_{60}$; rim $\mathrm{An}_{50}$ ) microphenocrysts. The microdoleritic microxenoliths have granular clinopyroxene (core Mg \#80; rim Mg\#55) and fine, randomly oriented plagioclase laths (core $\mathrm{An}_{68} ; \operatorname{rim~}_{\mathrm{An}}^{48}$ ). Small ( $\sim 2-3 \mu \mathrm{~m}$ ) euhedral grains of pentlandite, pyrrhotite and pyrite were observed within plagioclase crystals in some of these microdoleritic microxenoliths. Large ( $\sim 200 \mu \mathrm{~m}$ ) grains of pyrite and pyrrhotite were observed infilling voids between microxenoliths in these dykes (Fig. 6.9 CD), suggesting that the sulphide melt was mobile during brecciation. The microdoleritic microxenoliths also contain globular masses of net-textured sulphide (Fig. 6.9 E-H) that appear to have impregnated the porous dolerite.

The Zone E dyke displays very similar patterns to the dyke at Zone D-C, with a chilled margin that grades into medium-grained gabbro. Olivine in the chill is present as isolated, normally zoned phenocrysts (core $\mathrm{Fo}_{80}$; rim $\mathrm{Fo}_{65}$ ) and as more evolved crystals in olivine glomerocrysts (core $\mathrm{Fo}_{72} ;$ rim $\mathrm{Fo}_{70}$ ). Microphenocrysts of clinopyroxene in the chill are normally zoned (core Mg\#80; rim Mg\#55), with more evolved variants in the dyke core (core Mg\#75; rim Mg\#35). The Kilian Formation directly at the dyke margins at Zone E is
brecciated with fractures being infilled with euhedral pyrite and calcite crystals (up to 1 cm in size), similar to those observed in the brecciated Wynniatt Formation limestones at Zone B. Pyrite in these fractures appears to have been deposited in the permeable host rock breccia by hydrothermal fluids.

The Zone A dyke contains very similar textures and compositions to those observed in the Zone E dyke. However, there is an occurrence of a microporphyritic texture in a dolerite from the gossan-stained dyke core (Fig. 6.4D) at Zone A. Here, microphenocrysts of clinopyroxene are normally zoned (core Mg\#83; rim Mg\#80) with sieve-textured cores (Fig. 6.9J) suggesting resorption. Plagioclase phenocrysts are typically normally zoned with primitive cores (core $\mathrm{An}_{85}$; rim $\mathrm{An}_{75}$ ). However, reversely zoned plagioclase phenocrysts are also observed with rims in equilibrium with the cores of normally zoned plagioclase phenocrysts (core $\mathrm{An}_{72} ;$ rim $\mathrm{An}_{85}$ ). Sparse pyrite and chalcopyrite grains (< 1 mm in size) are present in the gossan stained core region of the Zone A dyke (Fig. 6.4D).

### 6.4.2 Bulk rock geochemistry

The majority of gabbroic samples from the SFDC have MgO contents < $10 \mathrm{wt} \%$. Most gabbros show an FeOt-enrichment trend (Fig. 6.10B), suggesting melt evolution was controlled by the olivine + clinopyroxene + plagioclase cotectic. Most show L/HREEratios typical of Type-2 Franklin magmas (Fig. 6.10A). However, there are isolated occurrences of slightly more primitive ( $\sim 10-12 \mathrm{wt} \% \mathrm{MgO}$ ) Type-1 Franklin magma in the SFDC (Fig. 6.10B). These occur in the dyke propagator tips at Zone B and in the Zone D dyke core.

Primitive mantle normalised multi-element spidergrams show a general similarity of magmas emplaced in the SFDC (Fig. 6.11). The large-ion-lithophile-elements (LILEs), including Pb , show erratic behaviour, however, suggesting hydrothermal alteration of the rocks and remobilisation of these elements (Fig. 6.11). There may be subtle differences between dyke core and chilled margin compositions. For example, at Zone D, the dyke core is more enriched in incompatible trace elements than the dyke chill (Fig. 6.11A). A sample from within the Zone D gabbroic roof apophysis shows higher L/HREE-ratios similar to the Type-1 Franklin magma (Fig. 6.11A). The gabbro from the sill-like body in
the SW region of Zone D is the most depleted in incompatible trace elements (Fig. 6.11A). The Zone C dyke has a trace element pattern identical to the Zone D gabbros, while the dyke chill at Zone C is most enriched (Fig. 6.11B). Zone B dyke propagator tip trace element patterns are very similar to Zone D-C gabbros, indicating they are Type-2 Franklin magmas (Fig. 6.11C). LILEs and high-field-strength-elements (HFSEs) are notably more enriched in the Zone B propagator tips compared to Zones D-C (Fig. 6.11C). Dyke rocks from Zone A-E are very similar to one another, indicating they formed from the same, or a very similar magma (Fig. 6.11 D-E).


Fig. 6.10: (A) $\mathrm{Ce} / \mathrm{Yb}$ versus MgO showing SFDC data. (B) FeOt versus MgO showing the SFDC data. The FCSC field, which is composed of Type-1 Franklin magma, is shown for reference on both plots.


Fig. 6.11: Primitive mantle normalised spidergrams showing gabbroic samples from the SFDC: (A) Zone D; including in blue the Type-2 Natkusiak Flood Basalts, which are very similar to Zone D gabbros. (B) Zone C. (C) Zone B. (D) Zone E. (E) Zone A. Normalizing values are from Sun and McDonough (1989).

Chondrite normalised PGE profiles from the SFDC show that the gabbroic samples contain relatively low amounts of PGE, compared to chondrite abundances (Fig. 6.12). The IPGEs are strongly depleted, suggesting either retention of Os-Ir alloys in the source or fractionation of minor chromite/alloys or olivine (Capobianco \& Drake, 1990, Tredoux et al., 1995, Brenan et al., 2003) or in local reduction fronts at chromite grain boundaries (Finnigan et al., 2008). SFDC gabbros show a minor enrichment in Pt and Pd , relative to other PGEs. Pt is particularly enriched in the dykes at Zone E-A (Fig. 6.12A).


Fig. 6.12: Chondrite mantle normalised PGE spidergrams showing data from gabbroic samples from the SFDC: (A) Zones A, E and B. (B) Zones C and D. Normalizing values are from McDonough and Sun (1995). The yellow field is Type-2 Natkusiak Flood Basalts ( $\mathrm{N}=45$ ). PGE data can be found in appendix G1.

### 6.4.3 Sulphur isotopes

The sulphur isotope data $\left(\delta^{34} \mathrm{~S}\right)$ for the SFDC is presented in Fig. 6.13 and appendix G5. Samples are plotted from bottom (NW) to top (SE) in stratigraphic order. Gabbroic samples at Zone D have $\delta^{34} \mathrm{~S}$ values between +2.7 and $+4.5 \%$. This is consistent with many Minto Inlier Franklin magmas, which commonly have $\delta^{34}$ S values between +2 and +4 possibly reflecting a slightly anomalous mantle source (Beard, 2012, Hryciuk et al., 2014). The Wynniatt Formation limestone at Zone D has $\delta^{34}$ S values between +7.4 and $+11.5 \%$, suggesting that only limited (if any) host-rock contamination is recorded in the gabbros of Zone D. In contrast, Zone C gabbroic samples have heavier $\delta^{34} \mathrm{~S}$ values between +9.0 and $+11.0 \%$, suggesting more extensive contamination of the magma. At Zone B, Wynniatt Formation limestone has $\delta^{34}$ S values between +12.7 and $+19.6 \%$. The thin N/NNE-trending dyke propagator tips at Zone B have $\delta^{34}$ S values between +4.8 and $+12.7 \%$, and have higher sulphur contents ( 0.24 to $0.58 \mathrm{wt} \%$ ) than most igneous rocks in the SFDC, implying significant wallrock contamination at Zone B. The Zone E dyke has $\delta^{34} \mathrm{~S}$ values of +9.7 to $10.4 \%$, suggesting contamination by Kilian Formation sulphates, which generally have $\delta^{34} \mathrm{~S}$ values between +17.0 to $+24.0 \%$. A sample traverse from the Kilian Formation to the west of the Zone E dyke shows a pattern of bulk sulphur-depletion towards the dyke contact, with values decreasing from 0.43 to $0.13 \mathrm{wt} \%$ (appendix G5). Kilian Formation carbonates at Zone A have $\delta^{34} \mathrm{~S}$ values between $+19.1 \%$ and $+21.5 \%$, whereas the sulphates have $\delta^{34} \mathrm{~S}$ values between $+31.8 \%$ and $+32.6 \%$. The Zone A dyke has $\delta^{34} \mathrm{~S}$ values between $+5.0 \%$ and $+8.9 \%$, also consistent with wallrock contamination.

In summary, the SFDC shows an unambiguous up-section $\delta^{34}$ S enrichment trend between Zone D and B, with elevated $\delta^{34}$ S at Zone E, that suggests incorporation of heavy sedimentary sulphur by the magma. The most enriched $\delta^{34} \mathrm{~S}$ values occur in the dykes that are hosted by brecciated sulphur-bearing country rocks in Zone B (Fig. 6.13).


Fig. 6.13: $\delta^{34}$ S (relative to VCDT $=$ Vienna Canyon Diablo Troilite) plot showing dyke, sill and local country rock $\delta^{34}$ S compositions across the SFDC, in stratigraphic order. Sulp = Sulphate evaporites.

### 6.4.4 Anisotropy of magnetic susceptibility fabrics

AMS was performed on 17 samples from the SFDC to constrain magma flow directions (appendix G6). Variation of susceptibility in cores of igneous rocks can be represented by a susceptibility ellipsoid, with maximum, intermediate and minimum axes K1, K2 and K3, respectively (Tarling \& Hrouda, 1993). The plane of 'flattening' of the ellipsoid is the K1K2 plane, and the axis of elongation is the K1 axis. Studies of AMS in magnetite-bearing rocks show that the magnetofoliation, for convenience represented by the K3 axis normal to the K1-K2 plane, and the magnetolineation, represented by the K1 axis, correspond to the preferred shape orientation of multi-domain magnetite (Borradaile \& Henry, 1997).

Thus the principal AMS axes, K 3 and K 1 , in unstrained pristine igneous rocks, are used as proxies for planes and lines of magmatic flow, respectively.

For the SFDC, the mean magnetic susceptibility ( Km ) is generally slightly higher and more variable for the sills ( $5.25+/-2.6 \times 10^{-2}$ SI: Fig. 6.14 B-D) than for the dykes (3.38 $+/-1.4 \times 10^{-2}$ SI: Fig. $6.14 \mathrm{~A}-\mathrm{C}$ ). This is probably due to more abundant and larger grain size of magnetite in the sills. The parameter P , for the degree of anisotropy, is low, < 1.066 for both sills and dykes, and is typical for fabrics of intrusive igneous rocks (Hrouda, 1982). The range of $P$ is about the same for the sills as for the dykes (Fig. 6.14). For both sills and dykes, the shape factor $U$ suggests that prolate magnetofabrics $(\mathrm{U}<0)$ are more common than oblate magnetofabrics $(\mathrm{U}>0)$, the ratio being about 1.4:1, respectively (Fig. 6.14 C-D).

In Fig. 6.14 E-F, poles for the AMS axes are shown for dykes and sills, respectively. Squares represent poles of K1 axes and circles represent poles of K3 axes. The AMS axial distributions for the dykes are somewhat scattered, consistent with both variable dyke orientation and variable flow patterns within dykes (Fig. 6.14E). In general, the K1 axes are better grouped than the K3 axes. The mean K3 of combined dyke sites corresponds to a magnetofoliation striking $\mathrm{N} 33^{\circ} \mathrm{W}$ and dipping $73^{\circ} \mathrm{SW}$. The dyke mean magnetolineation K 1 in the plane of the magnetofoliation is steeply inclined to the SW at $72^{\circ}$ towards azimuth $237^{\circ}$. However, these are 'in-situ' orientations in the northern flank of a regional syncline (Fig. 2.2). Nearby to the west, the host Wynniatt Formation bedding strikes $\mathrm{N} 67^{\circ} \mathrm{E}$ and dips $12^{\circ} \mathrm{SE}$. If dyke emplacement preceded tilting of the bedding, as seems likely, then the present attitudes can be corrected for structural tilt, to estimate the original attitudes at the time of dyke emplacement. Using the above estimate for tilt correction, the 'tilt-corrected' orientation of the mean K1 axis for the dykes becomes approximately $65^{\circ}$ inclined towards azimuth $205^{\circ}$. This supports the interpretation that steep upward flow of magma during dyke emplacement was from a southern source.


Fig. 6.14: Summary plots of magnetic susceptibility parameters for sills (A, C) and dykes (B, D) along the SFDC. $P$ is degree of anisotropy, $U$ is shape parameter, and Km is mean sample susceptibility in S.I. units. Stereoprojection of AMS K1 axes (blue squares, maximum susceptibility) and of K3 axes (red circles, minimum susceptibility) for dykes (E) and sills (F) of the SFDC; equal area, lower hemisphere projection. The contours are clustered around the majority of data points for each population shown (K1 and K3).

The AMS axial distributions for the sills (Fig. 6.14F) tend to be more streaked along greatcircle trends and less irregularly scattered than for the dykes. The mean magnetofoliation of combined sill sites, interpreted from the mean K3 axis, strikes $\mathrm{N} 55^{\circ} \mathrm{W}$ and dips $31^{\circ} \mathrm{SW}$, significantly steeper than typical sill-margin contacts. The sills sampled along the SFDC exhibit mainly gently plunging K1 axes of NW/SE trend. This could be consistent with near-horizontal flow, but tectonic influences may also be involved. K1 axes are mainly well clustered at individual sill sites with larger scatter between sites. In deformed rocks, finite strain may affect AMS axial orientations (Borradaile \& Henry, 1997). Combined magmatic-tectonic influences may produce streaked AMS axial patterns of primary and
secondary (i.e. of magmatic and tectonic), origins. Such transitional magnetofabrics (Borradaile \& Henry, 1997) could occur especially in the sill samples of the SFDC.

### 6.5 Discussion

### 6.5.1 The structure and emplacement of the Southern Feeder Dyke Complex

The SFDC is located along a set of prominent regional NNW-trending $1^{\text {st }}$-derivative magnetic lineaments (Fig. 6.2), interpreted to represent syn-magmatic normal faults by Bédard et al. (2012). The SFDC consists of NNW/NW-trending dykes and NE-SW striking, SE-dipping sills (Fig. 6.1). In many cases, host rocks show evidence of brecciation (Fig. 6.4B) and at Zone B, previously brecciated and hydrothermally altered dyke material was reinjected by fresh dykes, indicating syn-magmatic faulting. The rheologically weak country rock carbonates soften markedly during magma emplacement and thus preserve shear sense directions on the faults (Žák et al., 2012). At Zone D-C (Fig. $6.3 \& 6.5$ ), metre-scale drag folds indicate east-side-down motion on a NE-dipping normal fault. The softening of the country rock implies significant heat loss from the dykes, which could have driven hydrothermal cells around the SFDC.

Zone D is the stratigraphically deepest part of the SFDC. It preserves $\mathrm{a} \sim 20 \mathrm{~m}$ thick gabbroic roof apophysis (labeled 'a' on Fig. 6.3) that injected a parasitic fault in the downdropped limestone roof panel. The edge of the main gabbroic mass may also be a dyke (labeled 'b' on Fig. 6.3), the contact subsequently being sharpened by faulting. These syn-faulting dykes probably fed the capping sill, which extends for tens of kilometers towards the SW, but no direct linkage between the sill and the dyke-like facies is observed. However, interlayered calc-silicate and Fe -oxide skarn rocks occur in the inferred contact region. The calc-silicates in the skarn suggest intense contact metamorphism during sill emplacement (Nabelek et al., 2013). The Fe-oxide facies of the skarn was inferred to represent the expulsion of late-Fe-rich magmatic fluids from the sills into the roof panel limestones at the Uhuk Massif (Bédard et al., 2012, Nabelek et al., 2013). If the analogy with the field relationships at the Uhuk Massif holds, then the skarn facies at Zone D would be remnants of a metamorphic aureole that sheathed the conduit-system that fed the

Zone D sill. This is consistent with the idea that the carbonate and discrete dyke ('a' on Fig. 6.3) that composes most of Zone D is a downdropped roof panel, and that the dykelike feeder of the sill located near the gabbro/limestone contact ('b' on Fig. 6.3) was partly faulted out. The proximity of the Zone C dyke to Zone D , together with its similar size and orientation, suggest it is an extension to a higher stratigraphic level of the Zone D feeder dyke. The Zone C dyke contains more evolved mineral compositions in comparison to the Zone D gabbros, suggesting the magma fractionated as it traversed the crust. This is consistent with the AMS data that indicate dominant up-flow in the dykes (Fig. 6.14). At the present erosional level, the SFDC is segmented, but the dykes constituting this system may be connected at depth, as is typical of many dyke systems (Delaney \& Pollard, 1981).

We infer that the NNW/NW-trending rubbly gabbro at the NE-end of Zone B is also an upsection continuation of the Zone D-C feeder dyke. However, Zone B also has numerous thinner ( $\sim 0.5-2 \mathrm{~m}$ ) N/NNE-trending dykes that splayed off from the main fault-guided dyke (Fig. 6.6). The orientation of the thinner N/NNE-trending dykes suggests injection of magma into a conjugate fracture system. The sulphide-bearing igneous breccias in the thin dykes are cemented by microphyric basalt, confirming the syn-faulting nature of magmatism. The overprinting hydrothermal system generated by dyke intrusion appears to have remobilised country rock sulphides, and deposited them in the surrounding, permeable limestone breccia. We interpret the thinner dykes at Zone B to be propagator tips emplaced into an active fault.

Zone E-A consist of a single NNW/NW-trending dyke emplaced into sulphates and carbonates of the Kilian Formation. This is the stratigraphically highest region of the SFDC. No dykes are present above this level (at this location), suggesting that the dykes may have fed the overlying sills (Figs. 6.4D \& 6.8). At Zone E-A (Fig. $6.7 \& 6.8$ ), SW-side-down motion on a normal fault is preserved, the opposite of that seen at Zone D-C. This suggests that the SFDC fault system has a horst-and-graben geometry, and is composed of numerous, synchronously active fault strands, many of which were exploited by the Franklin magma.

### 6.5.2 Evidence for a composite magmatic system

The presence of internal chills within the dyke-like roof apophysis at Zone D, and the presence of igneous breccias (Figs. 6.4C and 6.9 E-F) at Zone B, implies that the SFDC was a composite magmatic system. Sieve-textured clinopyroxene phenocrysts indicate resorption caused by magmatic replenishment in the plumbing system. Geochemically, the SFDC is mostly composed of relatively fractionated Type-2 Franklin magma (Fig. 6.10A). However, there are occurrences of more primitive Type-1 magma (Fig. 6.10A) at Zone DB. The presence of both populations of Franklin magma suggests that the SFDC acted as a conduit system for the $1^{\text {st }}$ (Type-1 magma) stage of Franklin LIP magmatic activity, and that this older conduit system was then reactivated and flooded by Type-2 magma during the main basaltic effusion stage. AMS fabrics in the dykes show steep upflow fabrics indicating that magma ascended from depth along fault-guided dykes (Fig. 6.14). The apparent northerly magma flow direction measured in the SFDC is probably a local effect caused by the channelling of Franklin magmas through a complex underlying plumbing system.

### 6.5.3 Downstream/up-section $\delta^{34}$ S increase in the SFDC caused by crustal assimilation

Various mechanisms that trigger sulphur-saturation in magma have been suggested, such as: 1) fractional crystallisation (Czamanske \& Moore, 1977, Wendlandt, 1982): 2) changing $\mathrm{fO}_{2}$ conditions (Haughton et al., 1974): 3 ) magma mixing (Irvine, 1975): and 4) addition of external crustal sulphur (Ripley et al., 2003, Keays \& Lightfoot, 2010). The assimilation of crustal sulphur into magma is considered a particularly potent way to trigger formation of an immiscible sulphide liquid. The $\delta^{34} \mathrm{~S}$ data imply that magmas in the SFDC interacted with and commonly assimilated their host rocks (Fig. 6.13).

Heavy $\delta^{34} \mathrm{~S}$ values imply ingestion of crustal sulphur (Li et al., 2009), because mantle-like $\delta^{34} \mathrm{~S}$ values are typically $\sim 0 \%$ (Labidi et al., 2014). Franklin magmas have $\delta^{34} \mathrm{~S}$ values typically between +2 and $+4 \%$, however, which we believe reflects derivation from a slightly anomalous source mantle (Beard, 2012, Hryciuk et al., 2014). $\delta^{34} \mathrm{~S}$ values in the
gabbros at Zone D are similar to typical Franklin magmatic values and therefore preserve minimal evidence of host rock contamination. Slightly up-section at Zone C , the $\delta^{34} \mathrm{~S}$ values increase to $\sim 10 \%$, indicating host rock assimilation. The dyke propagator tips at Zone B also have enriched $\delta^{34} \mathrm{~S}$ values between +4.8 and $+12.7 \%$, trending towards the composition of their host country rocks (Fig. 6.13). The Wynniatt and Kilian Formation sedimentary rocks hosting the SFDC have elevated $\delta^{34} \mathrm{~S}$ values typical of sedimentary rocks deposited in a restricted basin (Prince, 2014). This may account for the elevated $\delta^{34} \mathrm{~S}$ values (up to $+32.6 \%$ ) of Kilian Formation sulphate evaporites compared to the lighter values ( +15 to $+17 \%$ ) of the Minto Inlet Formation evaporites. The Zone E dyke has $\delta^{34} \mathrm{~S}$ values of +9.7 to $10.4 \%$, indicating significant contamination by host rocks. The hydrothermally altered and sulphur-depleted country rock that occurs in proximity to the Zone E dyke suggests that sulphur was liberated by metamorphic devolatilization reactions, as suggested for the Northern Feeder Dyke Complex (Hryciuk et al. In Prep). The $\delta^{34} \mathrm{~S}$ values $(+5.0 \%$ and $+8.9 \%$ o) of the Zone A dyke also indicate minor evaporite or carbonate assimilation (Fig. 6.13). The most elevated Zone $\mathrm{B} \delta^{34} \mathrm{~S}$ values and sulphur contents can be explained by a simple mixing model (Ripley \& Li, 2003). Assuming an uncontaminated magma with 620 ppm sulphur and $\delta^{34} \mathrm{~S}$ of $+3.6 \%$, then 0.05 to $0.06 \mathrm{wt} \%$ anhydrite assimilation (with $\delta^{34} \mathrm{~S}$ of $+32 \%$ ) is required to generate $\delta^{34} \mathrm{~S}$ values greater than $+8 \%$.

In summary, the $\delta^{34} \mathrm{~S}$ data indicate that most gabbroic rocks of the SFDC have been contaminated by local country rocks, which may have induced sulphide immiscibility in the magma. We will now discuss in more detail the evidence preserved at Zone B, where sulphide immiscibility caused by wallrock assimilation appears to have been 'caught in the act'.

### 6.5.4 Sulphide immiscibility induced by wallrock assimilation

There is an ongoing debate about how and where magma becomes sulphur-saturated within the magmatic plumbing systems of Ni-Cu-PGE sulphide deposits (Naldrett, 1992, Ripley et al., 2003, Ripley \& Li, 2013). The dyke-fault relationships of the SFDC share many similarities to conduit-style ore deposits such as Noril'sk (Naldrett, 1992, Li et al., 2009), Jinchuan (Lehmann et al., 2007) and the Eagle deposit (Ding et al., 2012), where it has been inferred that faulting associated with rifting facilitated ore genesis. At Noril'sk it
is thought that the ingestion of sulphate-bearing country rocks into a PGE-enriched magma generated the orebodies (Li et al., 2009). The latter is supported by the heavy $\delta^{34} \mathrm{~S}$ composition of the magma, which is higher than mantle-like values and trends towards the $\delta^{34} \mathrm{~S}$ composition of the surrounding sulphate-bearing country rocks (Grinenko, 1985). However, the actual mechanism for ingestion of sulphur-rich country rock into the Noril'sk magma is unclear (Ripley et al., 2003). Naldrett (1999) proposed that thermal erosion of the magma chamber walls during multiple magmatic pulses led to the ingestion of Devonian evaporites and carbon-bearing country rocks. In the SFDC, the country rock is heavily brecciated as a result of Neoproterozoic faulting. Emplacement of magma into fault zones could allow more extensive interaction because of the increased surface area provided by the brecciated country rock. At Zone B in particular, the dyke propagator tips intrude sulphur-bearing brecciated carbonates and older brecciated dyke material (Fig. 6.4 C ). The $\delta^{34} \mathrm{~S}$ values of the Zone B propagator tips are especially high (up to $+12.7 \%$ ), and trend towards $\delta^{34} \mathrm{~S}$ values of the surrounding country rock (Fig. 6.13).

Sulphur contents of $\sim 1400 \mathrm{ppm}$ are typically required to trigger sulphur saturation in a basaltic magma (Jugo et al., 2005) similar to the Type-2 magmas (Fig. 6.10A), which constitute most of the SFDC. However, this value is dependent upon various factors such as temperature, pressure, magma composition and oxygen fugacity. The sulphur contents of the Zone B dyke propagator tips (up to $0.58 \mathrm{wt} \% \mathrm{ppm}$ : appendix G5) are typically greater than the value required to trigger sulphur saturation. The sulphur-bearing local country rocks represent plausible sulphur-sources. The abundance of sulphur-minerals in the hydrothermal breccia-fills suggest that hydrothermal cells generated by magmatism stripped sulphides from host country rocks (and possibly earlier dyke pulses) and reconcentrated them within the breccia zone (Fig. 6.4B), where they would have the greatest likelihood of being assimilated by the next pulse of magma. In turn, such hydrothermal cells would have been made more efficient by local country rock brecciation, which would have greatly enhanced the permeability of the country rocks.

The microdoleritic microxenoliths within the syn-faulting dyke propagator tips at Zone B host globular impregnations of sulphides that provide direct evidence of syn-faulting magmatic sulphide immiscibility (Fig. 6.9 E-H). The textures show that immiscible globules of magmatic sulphide impregnated a porous microdoleritic matrix (Fig. 6.9 C-H). These sulphide-impregnated microdoleritic rocks were then reworked (brecciated) by the
next magma pulse. It is possible that the injecting basaltic melt that now cements the microdoleritic microxenoliths could have scavenged and ingested sulphide hosted within microdoleritic microxenoliths, in addition to incorporating external crustal-derived sulphur directly.

The Zone B dyke propagator tips appear to represent magmatic dead ends where rapid cooling limited the upgrading process. Conversely, this is why textural evidence for immiscible sulphide generation has been preserved there. The processes involved in the early stages of ore-genesis that are preserved in Zone B may be applicable to the entire SFDC, given the elevated $\delta^{34} \mathrm{~S}$ values recorded in other parts of the system (Fig. 6.13). However, in larger, more robust parts of the system (e.g. Zone D: Fig. 6.3), where large amounts of magma flowed through the conduits to inflate a major sill, the contaminated signatures and any immiscible sulphides formed this way may have been flushed downstream ( $c f$ Hryciuk et al. Submitted). Given the availability of structural or hydraulic traps in these sills, it is possible that immiscible sulphides may have been trapped there.

In summary, the SFDC on Victoria Island represents a fossilised magmatic feeder system that was emplaced into an active normal fault system. Our data imply that external sulphur was ingested by the magma, triggering local sulphide saturation and the formation of immiscible sulphide liquids. Brecciation associated with this faulting may have facilitated the assimilation of host sulphur, both by increasing available reactant surface area, and more speculatively, by pre-concentrating sulphur and Cu through forced hydrothermal circulation in high-permeability fault breccias. Similar mechanisms may have been involved in forming Ni-Cu-PGE ore-bodies in other conduit-style magmatic systems; the evidence having been obliterated by repeated magma throughflow.

### 6.5.5 Economic Ni-Cu-PGE potential of the Franklin LIP

The evidence that sulphide immiscibility occurred in Zone B of the SFDC raises the question of where any possible immiscible sulphide liquids were deposited. A hydraulic trap is required in order to produce economically viable quantities of immiscible sulphide liquids (Naldrett, 1992). The most probable targets would be the downstream extensions of sills that were fed by SFDC dykes. Approximately 10 km SW of the capping sill at Zone A
there are sulphide-rich (~5\%) dykes (e.g. Sulphide City Dyke: Fig. 2.2) with small semimassive sulphide concentrations at its contacts, and there are large calcite-pyrite-rich gossans (GS: Fig. 6.15 and Peterson et al. (2014)). Although the sulphide at the GS appears to be mainly hydrothermal (Fig. 6.15B), it is possible that some of it was scavenged from immiscible sulphide previously concentrated in one of the adjoining sills.


Fig. 6.15: (A) Field photo of the Gossan Sill (GS) situated $\sim 10 \mathrm{~km}$ WSW of the SFDC (Fig. 2.2). The dashed red line shows the sill contact with the Kilian Formation country rocks (dashed white lines for bedding). (B) Hydrothermal pyrite globules with oxidised halos in the Kilian Formation just below the contact with the sill.

The syn-faulting scenario for magma emplacement and contamination documented in the SFDC and the Uhuk Massif (Bédard et al., 2012) probably also applies to the entire Franklin event in the Minto Inlier. The NNW/NW-trending magnetic lineaments (Fig. 6.2) are found throughout the Minto Inlier (Kiss \& Oneschuk, 2010), and each may represent a syn-faulting dyke that could have transported significant quantities of magma up-section. In these syn-magmatic fault systems, where cataclasis increases reactant surface areas,
flowing magma could have ingested crustal sulphur, triggering the formation of immiscible sulphide liquids. This is the scenario that appears to be preserved at Zone B of the SFDC. The intersections of these syn-magmatic faults with sulphate-rich sediments of the Kilian and Minto Inlet Formations would appear to be particularly promising exploration targets. Such fault systems could also be favourable sites for the emplacement of subsequent magma pulses, generating composite magma systems with the capacity to upgrade the tenor of early sulphides. Where the magma systems are better developed, as at Zone D, prior evidence of contamination (if it indeed occurred) would have been flushed downstream.

In addition to the dyke-fault systems acting as dynamic environments, the occurrence of Type-1 magma in the SFDC is significant (Fig. 6.10A). Type-1 magmas were the first to be emplaced in the Franklin LIP and are commonly found deeper in the Shaler Supergroup. Type-1 rocks are often primitive (up to $\sim 13 \mathrm{wt} \% \mathrm{MgO}$ ) with high Ni -contents up to 250 ppm (appendix E1). Type-1 Franklin magmas were hotter $\left(\sim 1300^{\circ} \mathrm{C}\right)$ than Type-2 Franklin magmas, and therefore would have had more potential to reactivate crustal rocks. The Uhuk Massif was inferred to be the major melt and crystal-slurry feeder to the Type-1 olivine-enriched Fort Collinson Sill Complex (chapter 5). Pulses of magma and crystal slurries may have waxed and waned during episodic fault-slip, and upstream sill collapse during melt evacuation at the Uhuk Massif. It has been proposed that melt evacuation triggered by sill collapse played a role in the generation of the Voisey's Bay Ni-Cu-PGE sulphide deposit (Cruden et al., 2008). There is also evidence for magmatic replenishment and magma mixing in sills associated with Type-1 magmas (chapter 5), which can act as mechanism for triggering sulphur-saturation (Irvine, 1975). Later Type-2 magma pulses that utilised the SFDC may have ingested earlier sulphides associated with Type-1 magmas. The association of Type-1 magmas in the SFDC as well as other dyke feeder systems in the Minto Inlier therefore enhances the Ni-Cu-PGE economic potential of the Franklin LIP.

### 6.6 Conclusion

The Southern Feeder Dyke Complex (SFDC) on Victoria Island in the Canadian Arctic is part of the Franklin Large Igneous Province (LIP). The Franklin LIP on Victoria Island is
primarily composed of gabbroic sills, with sub-ordinate NNW/NW-trending feeder dykes. Igneous breccias cemented by basalt and drag folds within wallrocks indicate that Franklin magmas were emplaced contemporaneously with regional Neoproterozoic normal faulting. The SFDC dykes acted as dynamic magma conduits that facilitated the ingestion of crustal sulphur into the magma, which we believe triggered the formation of immiscible sulphide liquids. This scenario is supported by heavy $\delta^{34} \mathrm{~S}$ values of the dykes. There are sulphide showings associated with Franklin sills and dykes across the Minto Inlier, including the SFDC area. At the SFDC, globular and locally net-textured sulphides are associated with microdoleritic microxenoliths that were cemented by later basaltic magmas. Evidence that the SFDC was a composite magmatic system suggests that previously formed immiscible sulphides may have been ingested by fresh magma. In the more robust part of the SFDC where dykes and sills are well developed, the evidence for sulphide immiscibility may have been flushed downstream. The SFDC appears to feed laterally extensive sills at different crustal levels. These sills may represent hydraulic traps that could have concentrated immiscible sulphide liquids. There are gossan showings in the sills downstream from the SFDC, which may represent late hydrothermal reworking of such sulphides. Similar processes of wallrock induced sulphide immiscibility may have operated in dyke-feeder systems across the Franklin LIP in the Minto Inlier, suggesting that the Franklin igneous rocks on Victoria Island are indeed prospective for Ni-Cu-PGE. The textures and field relationships documented in the SFDC support models where sulphide immiscibility is triggered by sedimentary wallrock assimilation during fault-guided magmatic activity.

## 7. Conclusions and future work

### 7.1 Introduction

This thesis has investigated the magmatic processes that operated in the Neoproterozoic Franklin plumbing system. The excellent exposure and preservation (lack of metamorphism and deformation) of the Franklin sills, dykes and flood basalts within the Minto Inlier of Victoria Island has enabled a detailed petrological and in-situ mineral chemical investigation of the igneous rocks.

At the beginning of this project, three aims were set out: 1) Investigate the large-scale processes that operated in the magmatic plumbing systems of LIPs. 2) Investigate the small-scale differentiation mechanisms operating in sills. 3) Provide insights into the economic Ni-Cu-PGE potential of the Franklin LIP.

### 7.2 Large-scale processes operating in the Franklin magmatic plumbing system

In chapter 5, I studied the larger-scale differentiation mechanisms of the Franklin magmatic plumbing system. This was achieved by examining the Fort Collinson Sill Complex (FCSC) for > 50 km along strike. The FCSC has a continuous olivine zone with an overlying doleritic gabbro zone, with the former exhibiting an along strike bulk-rock Fe-enrichment trend. $\mathrm{Fe}=\mathrm{Mg}$ systematics and trace element modelling show that this differentiation trend is the result of olivine slurry replenishment into a resident, more evolved gabbroic mush.

The FCSC study provides important insights into the regional differentiation dynamics of the Franklin magmatic plumbing system. Other olivine-enriched sills have been observed at a similar stratigraphic level to the FCSC, and some of them show isotopic disequilibrium between their olivine and dolerite zones. Therefore, a similar olivine slurry replenishment scenario may have operated in these other Franklin sills. If this is the case, then there may
have been a magma chamber system beneath the level of emplacement of the sills, where primitive magmas similar to the parental magmas to the sills, crystallised olivine cumulates. Periodic replenishment into this underlying feeder system could have triggered cumulate remobilisation events with olivine slurries emplaced up-section into sills containing partly crystalline gabbroic mushes. Such an emplacement process may have operated in other LIPs or large layered intrusions elsewhere, and can potentially account for the lateral variability seen in some of these rocks.

### 7.3 Small-scale differentiation mechanisms operating in sills

In chapter 4, I studied a layered mafic Franklin sill, called the Lower Pyramid Sill (LPS), in detail. The LPS is characterised by a basal olivine-enriched layer (OZ) that is overlain by a sequence of doleritic gabbro (DZ). This study utilised field relationships, whole-rock chemistry, petrography and in-situ mineral chemistry, to help understand how its igneous layering formed. The rapid cooling and preservation of the LPS enabled the use of reliable inverse melt modelling, which was used to assess the compositions of crystals and melts involved during differentiation of the sill. I showed that the LPS OZ formed by an olivine slurry replenishment into a resident, more evolved gabbroic mush. The DZ shows upward in-situ fractional crystallisation trends in comparison to the underlying OZ. Intervening clinopyroxene-rich cumulates were shown to have formed by a metasomatic reaction between pore melts derived from the OZ and DZ .

The differentiation model developed for the LPS has implications for the development of igneous layering in layered intrusions elsewhere, particularly with regards to the formation olivine-rich layers. Also, the study fits into a current debate about how magma(s) are emplaced and how igneous layering forms. One school of thought advocates the emplacement of aphyric liquids that evolved by in-situ fractional crystallisation, while another suggests the emplacement of crystal-charged slurries. The model presented for the LPS shows that its cumulate layering formed from both of these mechanisms. In addition, the LPS model also shows the importance of porous melt flow within crystal mushes, which can potentially overprint primary igneous facies. The differentiation mechanisms that operated in the LPS may have operated in the development of igneous layering large layered intrusions. For example, in the Bushveld Complex, large-scale up-section isotopic
shifts have been observed, as well as crystal-scale isotopic disequilibrium within individual layers. The larger-scale isotopic shifts suggest the periodic emplacement of replenishing magmas, whilst the smaller-scale isotopic disequilibrium between constituent crystals of individual layers suggest that slurry emplacement, as shown for the LPS, may have been an important differentiation mechanism in the Bushveld Complex.

### 7.4 Economic Ni-Cu-PGE potential of the Franklin LIP

The Franklin LIP is considered to be prospective for Ni-Cu-PGE mineralisation. In chapter 6, I investigated a fault-guided Franklin feeder dyke network, called the Southern Feeder Dyke Complex (SFDC). The SFDC provided an ideal target to investigate the economic potential of the Franklin LIP on Victoria Island. The exceptionally exposed field relationships, sulphide mineralisation and emplacement into sulphate-bearing country rocks of the SFDC allowed the processes involved in the formation of immiscible sulphide liquids to be investigated.

At the SFDC, magmas were fault-guided and they may have fed the adjacent sills. Dykes at the SFDC are brecciated and contain microdoleritic microxenoliths cemented by basalt. The microdoleritic microxenoliths contain globular and locally net-textured sulphides that impregnated a porous doleritic mush. I inferred, based on the field and textural relationships, that sulphide immiscibility in the SFDC magma was triggered by local wallrock assimilation. This would have been facilitated by fault-related brecciation. Our interpretation of external sulphur addition into the magma is supported by elevated $\delta^{34} \mathrm{~S}$ values of the dykes, which are towards the values of the local sulphur-bearing country rocks.

NW-trending syn-magmatic Neoproterozoic faults are pervasive across the Minto Inlier, and based on the evidence preserved in the dykes at the SFDC, I infer that that similar mechanisms inducing sulphide immiscibility operated throughout the Franklin magmatic plumbing system. However, in composite magmatic systems that experienced renewed magmatism, immiscible sulphide liquids may have been flushed downstream into neighbouring sills. There is evidence in the SFDC for magmatic replenishment and coupled with the slurry replenishment mechanisms preserved in the Franklin sills, indicates
that the Franklin magmatic plumbing system was a dynamic environment. The SFDC ultimately emphasises the importance of external, crustal-derived sulphur in the formation of immiscible sulphide liquids in basaltic magmas. The SFDC allows for a reconstruction of the mechanisms by which this external sulphur is incorporated into the magma, providing insights into $\mathrm{Ni}-\mathrm{Cu}-\mathrm{PGE}$ mineralisation processes.

### 7.5 Future work

The superbly exposed Neoproterozoic Franklin LIP on Victoria Island provides an excellent opportunity to understand the differentiation of LIPs. Further work on the olivine-enriched sills that may be related to the Fort Collinson Sill Complex (FCSC: chapter 5) could help determine if they formed from a similar emplacement process to that preserved in the Lower Pyramid Sill (LPS: chapter 4) and the other sills constituting the FCSC. If indeed Kat's sill is related to the FCSC (as inferred in chapter 5), then there may have been a regional olivine slurry replenishment event. The exposure of the Franklin sills would enable this olivine slurry replenishment event and its geochemical consequences to be characterised in 3D.

The next phase of the LPS study (chapter 4) would be to characterise in detail the nature of the hybrid melt and its porous flow through the proto-OZ mush. Detailed in-situ trace element quantification and inverse melt modelling using partition coefficients parameterised using the Franklin melt database would provide constraints on porous melt flow in the LPS and its role in redistributing incompatible trace elements in the sill. Also, sector-zoned clinopyroxenes that appear to be preserved in a channel-like network in the geometric centre of the LPS can be investigated to understand the impacts of sector-zoned growth on trace element partitioning.

The majority of Franklin sills exposed in the Minto Inlier are geochemically related to the Type-2 magma population. These Type-2 sills are characterised by doleritic and porphyritic textures. The porphyritic sills are comprised of isolated phenocrysts and glomerocrysts that appear to be ripped up gabbroic mushes. The porphyritic sills show Sand D-shaped whole-major element profiles that characterises other layered sills. Considering the 'random' constituents of these sills, it is surprising that they form well-
developed geochemical profiles. One of these porphyritic sills should be examined in detail, much like the LPS in chapter 4, with crystal and melt compositions constrained to see how their geochemical profiles formed and if similar processes operated in these sills to those in the LPS/FCSC. This work would therefore transcend both Franklin magma populations (Type $1 \& 2$ ) and reveal any linkages between differentiation mechanisms and specific magma types.

To further improve our understanding of the Franklin magmatic plumbing system and the processes operating during the remobilisation of cumulates, the Natkusiak flood basalts need to be studied. A $\sim 1 \mathrm{~km}$ thick sequence of Natkusiak lavas are preserved in the Minto Inlier and they have been geochemically correlated to the underlying Franklin intrusives. Therefore, detailed whole-rock and in-situ mineral chemical work on the Natkusiak flood basalts would provide an opportunity to understand the dynamics of the Franklin plumbing system. It would be interesting to see if the processes preserved in the sills (chapter $4 \& 5$ ) are preserved in the coeval flood basalts, or if, in addition, the lavas preserve a record of other processes. Additionally, geothermobarometry studies of the phenocrysts preserved in the Natkusiak flood basalts would provide constraints on the depth distribution of magma chambers in the Franklin LIP.

## 8. References

Annen, C., Paulatto, M., Sparks, R. S. J., Minshull, T. A. \& Kiddley, E. J. (2014). Quantification of the intrusive magma fluxes during magma chamber growth at soufrière hills volcano (montserrat, Lesser Antilles). Journal of Petrology 55, 529-548.

Ariskin, A. A., Konnikov, E. G., Danyushevsky, L. V., Kislov, E. V., Nikolaev, G. S., Orsoev, D. A., Barmina, G. S. \& Bychkov, K. A. (2009). The Dovyren intrusive complex: Problems of petrology and Ni sulfide mineralization. Geochemistry International 47, 425-453.

Arndt, N. T. (1977). Partitioning of nickel between olivine and ultrabasic and basic komatiite liquids. Carnegie Institution of Washington Year Book 76, 553-557.

Arndt, N. T. (2011). Insights into the geologic setting and origin of Ni-Cu-PGE sulfide deposits of the Noril'sk-Talnakh region, Siberia. Reviews in Economic Geology 17, 199-215.

Ballhaus, C. G. \& Glikson, A. Y. (1989). Magma Mixing and Intraplutonic Quenching in the Wingellina Hills Intrusion, Giles Complex, Central Australia. Journal of Petrology 30, 1443-1469.

Baragar, W. R. A. (1976). The Natkusiak basalts, Victoria Island, District of Franklin; in Report of Activities, Part A. Geological Survey of Canada Paper 76-1A, 347352.

Barnes, S. (1986). The effect of trapped liquid crystallization on cumulus mineral compositions in layered intrusions. Contributions to Mineralogy and Petrology 93, 524531.

Beard, C. (2012). Magmatic plumbing and melt source characterisation: A Hf-Pb-Sr-Nd isotopic study of the sill-dominated intrusive complex of the Franklin Large Igneous Province, Victoria Island, Arctic Canada. Unpublished thesis.

Bedard, J. H. (1993). Oceanic crust as a reactive filter: synkinematic intrusion, hybridization, and assimilation in an ophiolitic magma chamber, western Newfoundland. Geology 21, 77-80.

Bédard, J. H. (1987). The development of compositional and textural layering in Archaean komatiites and in Proterozoic basaltic komatiites from Cape Smith, Quebec, Canada. In: I. Parsons (Ed) The origin of layering in igneous rocks Proceedings of a NATO workshop. D. Reidel Publ. Co. Dordrecht, Holland, 399-418.

Bédard, J. H. (1991a). Cumulate recycling and crustal evolution in the Bay of Islands ophiolite. Journal of Geology 99, 225-249.

Bédard, J. H. (1991b). Cumulate recycling and crustal evolution in the Bay of Islands ophiolite. Journal of Geology 99, 225-249.

Bédard, J. H. (1993). Oceanic crust as a reactive filter: Synkinematic intrusion, hybridization, and assimilation in an ophiolitic magma chamber, western Newfoundland. Geology 21, 77-80.

Bédard, J. H. (1994). A procedure for calculating the equilibrium distribution of trace elements among the minerals of cumulate rocks, and the concentration of trace elements in the coexisting liquids. Chemical Geology 118, 143-153.

Bédard, J. H. (2010). Parameterization of the $\mathrm{Fe}=\mathrm{Mg}$ exchange coefficient (Kd) between clinopyroxene and silicate melts. Chemical Geology 274, 169-176.

Bédard, J. H., Hayes, B., Hryciuk, M., Wing, B., Beard, C., Dell'Oro, T. A., Weis, D., Scoates, J. S., Williamson, N., Cousens, B., Naslund, H. R., MacDonald, W. \& Nabelek, P. (2013). The Neoproterozoic Franklin Large Igneous Province on Victoria Island. Geological Asssociation of Canada - Mineralogical Association of Canada annual meeting. Program with Abstracts 35, 82.

Bédard, J. H., Leclerc, F., Harris, L. B. \& Goulet, N. (2009). Intra-sill magmatic evolution in the Cummings Complex, Abitibi greenstone belt: Tholeiitic to calc-alkaline magmatism recorded in an Archaean subvolcanic conduit system. Lithos 111, 47-71.

Bédard, J. H., Naslund, H. R., Nabelek, P., Winpenny, A., Hryciuk, M., Macdonald, W., Hayes, B., Steigerwaldt, K., Hadlari, T., Rainbird, R., Dewing, K. \& Girard, É. (2012). Fault-mediated melt ascent in a Neoproterozoic continental flood basalt province, the Franklin sills, Victoria Island, Canada. Geological Society of America Bulletin 124, 723-736.

Bédard, J. H., Sparks, R. S. J., Renner, R., Cheadle, M. J. \& Hallworth, M. A. (1988). Peridotite sills and metasomatic gabbros in the Eastern Layered Series of the Rhum complex. Journal of the Geological Society 145, 207-224.

Bédard, J. H. J., Marsh, B. D., Hersum, T. G., Naslund, H. R. \& Mukasa, S. B. (2007). Large-scale mechanical redistribution of orthopyroxene and plagioclase in the basement sill, ferrar dolerites, McMurdo Dry Valleys, Antarctica: Petrological, mineralchemical and field evidence for channelized movement of crystals and melt. Journal of Petrology 48, 2289-2326.

Begg, G. C., Hronsky, J. A. M., Arndt, N. T., Griffin, W. L., O’Reilly, S. Y. \& Hayward, N. (2010). Lithospheric, Cratonic, and Geodynamic Setting of Ni-Cu-PGE Sulfide Deposits. Economic Geology 105, 1057-1070.

Bhattacharji, S. (1967). Mechanics of flow differentiation in ultramafic and mafic sills. Journal of Geology 75, 101-112.

Black, B. A., Hauri, E. H., Elkins-Tanton, L. T. \& Brown, S. M. (2014). Sulfur isotopic evidence for sources of volatiles in Siberian Traps magmas. Earth and Planetary Science Letters 394, 58-69.

Borradaile, G. J. \& Henry, B. (1997). Tectonic applications of magnetic susceptibility and its anisotropy. Earth Sciences 42, 49-93.

Boudreau, A. E. (1999). PELE-a version of the MELTS software program for the PC platform. Computers \& Geosciences 25, 201-203.

Boudreau, A. E. \& Philpotts, A. R. (2002). Quantitative modeling of compaction in the Holyoke flood basalt flow Hartford Basin, Connecticut. Contributions to Mineralogy and Petrology 144, 176-184.

Bowen, N. L. (1928). The evolution of igneous rocks. Princeton University, Princeton.

Brenan, J. M., McDonough, W. F. \& Dalpé, C. (2003). Experimental constraints on the partitioning of rhenium and some platinum-group elements between olivine and silicate melt. Earth and Planetary Science Letters 212, 135-150.

Brophy, J. G., Whittington, C. S. \& Young-Rok, P. (1999). Sector-zoned augite megacrysts in Aleutian high alumina basalts: implications for the conditions of basalt crystallization and the generation of calc-alkaline series magmas. Contributions to Mineralogy and Petrology 135, 277-290.

Brown, G. M. (1956). The layered ultrabasic rocks of Rhum Inner Hebrides. Philosophical Transactions of the Royal Society of London B240, 1-53.

Buchanan, D. L. \& Nolan, J. (1979). Solubility of sulfur and sulfide immiscibility in synthetic tholeiitic melts and their relevance to Bushveld-Complex rocks. The Canadian Mineralogist 17, 483-494.

Campbell, I. H. \& Naldrett, A. J. (1979). The influence of silicate:sulfide ratios on the geochemistry of magmatic sulfides. Economic Geology 74, 1503-1506.

Canales, J. P., Nedimovi, M. R., Kent, G. M., Carbotte, S. M. \& Detrick, R. S. (2009). Seismic reflection images of a near-axis melt sill within the lower crust at the juan de fuca ridge. Nature 460, 89-93.

Canfield, D. E., Raiswell, R., Westrich, J. T., Reaves, C. M. \& Berner, R. A. (1986). The use of chromium reduction in the analysis of reduced inorganic sulfur in sediments and shales. Chemical Geology 54, 149-155.

Capobianco, C. J. \& Drake, M. J. (1990). Partitioning of ruthenium, rhodium, and palladium between spinel and silicate melt and implications for platinum group element fractionation trends. Geochimica et Cosmochimica Acta 54, 869-874.

Carslaw, H. S. \& Jaeger, J. C. (1959). Conduction of heat in solids. Oxford University Press, Oxford, 510.

Cawthorn, R. G., Sander, B. K. \& Jones, I. M. (1992). Evidence for the trapped liquid shift effect in the Mount Ayliff Intrusion, South Africa. Contributions to Mineralogy and Petrology 111, 194-202.

Cawthorn, R. G. \& Walraven, F. (1998). Emplacement and Crystallization Time for the Bushveld Complex. Journal of Petrology 39, 1669-1687.

Cawthorn, R. G. e. (1996). Layered Intrusions. Elsevier.
Chadwick, J. P., Troll, V. R., Ginibre, C., Morgan, D., Gertisser, R., Waight, T. E. \& Davidson, J. P. (2007). Carbonate assimilation at Merapi Volcano, Java, Indonesia: Insights from crystal isotope stratigraphy. Journal of Petrology 48, 1793-1812.

Chakraborty, S. (1997). Rates and mechanisms of $\mathrm{Fe}-\mathrm{Mg}$ interdiffusion in olivine at $980^{\circ}-1300^{\circ}$ C. Journal of Geophysical Research: Solid Earth 102, 12317-12331.

Chevallier, L. \& Woodford, A. (1999). Morph-tectonics and mechanism of emplacement of the dolerite rings and sills of the western Karoo, South Africa. South African Journal of Geology 102, 43-54.

Costa, F. \& Dungan, M. (2005). Short time scales of magmatic assimilation from diffusion modeling of multiple elements in olivine. Geology 33, 837-840.

Cruden, A. R., Burrows, D. R. \& Evans-Lamswood, D. (2008). Structure and emplacement of the Voisey's Bay troctolite and associated Ni-Cu-Co mineralisation. GAC-MAC-SEG-SGA program with abstracts, Quebec City, May 26-28 2008, 39.

Czamanske, G. K. \& Moore, J. G. (1977). Composition and phase chemistry of sulfide globules in basalt from the Mid-Atlantic Ridge rift valley near $37^{\circ} \mathrm{N}$ lat. Geological Society of America Bulletin 88, 587-599.

Davidson, J. P., Morgan, D. J., Charlier, B. L. A., Harlou, R. \& Hora, J. M. (2007). Microsampling and isotopic analysis of igneous rocks: Implications for the study of magmatic systems. 273-311.

Delaney, P. T. \& Pollard, D. D. (1981). Deformation of host rocks and flow of magma during growth of minette dikes and breccia-bearing intrusions near Ship Rock, New Mexico. U.S. Geological Survey Professional Paper 1202, 61.

Dell'Oro, T. A., Weis, D., Scoates, J. S. \& Bédard, J. H. (2012). Sr-Nd-Hf-Pb Isotope and Trace Element Geochemistry of the Natkusiak Formation Continental Flood Basalts of the Neoproterozoic Franklin Large Igneous Province, Victoria Island, Canada. Unpublished thesis.

Denyszyn, S. W., Halls, H. C., Davis, D. W. \& Evans, D. A. D. (2009). Paleomagnetism and $\mathrm{U}-\mathrm{Pb}$ geochronology of Franklin dykes in High Arctic Canada and Greenland: a revised age and paleomagnetic pole constraining block rotations in the Nares Strait region. Canadian Journal of Earth Sciences 46, 689-705.

Dewing, K., Pratt, B. R., Hadlari, T., Brent, T., Bédard, J. \& Rainbird, R. H. (2013). Newly identified "Tunnunik" impact structure, Prince Albert Peninsula, northwestern Victoria Island, Arctic Canada. Meteoritics and Planetary Science 48, 211223.

Di Rocco, T., Freda, C., Gaeta, M., Mollo, S. \& Dallai, L. (2012). Magma Chambers Emplaced in Carbonate Substrate: Petrogenesis of Skarn and Cumulate Rocks and Implications for CO2 Degassing in Volcanic Areas. Journal of Petrology 53, 23072332.

Ding, X., Li, C., Ripley, E. M., Rossell, D. \& Kamo, S. (2010). The Eagle and East Eagle sulfide ore-bearing mafic-ultramafic intrusions in the Midcontinent Rift System, upper Michigan: Geochronology and petrologic evolution. Geochemistry, Geophysics, Geosystems 11, Q03003.

Ding, X., Ripley, E. \& Li, C. (2012). PGE geochemistry of the Eagle Ni-Cu-(PGE) deposit, Upper Michigan: constraints on ore genesis in a dynamic magma conduit. Mineralium Deposita 47, 89-104.

Donaldson, C. (1976). An experimental investigation of olivine morphology. Contributions to Mineralogy and Petrology 57, 187-213.

Dungan, M. A. \& Davidson, J. (2004). Partial assimilative recycling of the mafic plutonic roots of arc volcanoes: An example from the Chilean Andes. Geology 32, 773776.

Eales, H. V. \& Cawthorn, R. G. (1996). The Bushveld Complex. In: Richard Grant, C. (ed.) Developments in Petrology: Elsevier, 181-229.

Eales, H. V., Maier, W. D. \& Teigler, B. (1991). Corroded plagioclase feldspar inclusions in orthopyroxene and olivine of the Lower and Critical Zones, Western Bushveld Complex. Mineralogical Magazine 55, 479-486.

Ernst, R. E., Wingate, M. T. D., Buchan, K. L. \& Li, Z. X. (2008). Global record of 1600-700 Ma Large Igneous Provinces (LIPs): Implications for the reconstruction of the proposed Nuna (Columbia) and Rodinia supercontinents. Precambrian Research 160, 159178.

Field, L., Blundy, J., Calvert, A. \& Yirgu, G. (2013). Magmatic history of Dabbahu, a composite volcano in the Afar Rift, Ethiopia. Geological Society of America Bulletin 125, 128-147.

Finnigan, C. S., Brenan, J. M., Mungall, J. E. \& McDonough, W. F. (2008). Experiments and Models Bearing on the Role of Chromite as a Collector of Platinum Group Minerals by Local Reduction. Journal of Petrology 49, 1647-1665.

Fiorentini, M. L., Bekker, A., Rouxel, O., Wing, B. A., Maier, W. \& Rumble, D. (2012). Multiple Sulfur and Iron Isotope Composition of Magmatic Ni-Cu-(PGE) Sulfide Mineralization from Eastern Botswana. Economic Geology 107, 105-116.

Font, L., Davidson, J. P., Pearson, D. G., Nowell, G. M., Jerram, D. A. \& Ottley, C. J. (2008). Sr and Pb isotope micro-analysis of plagioclase crystals from skye lavas: An insight into open-system processes in a flood basalt province. Journal of Petrology 49, 1449-1471.

Forrest, J. \& Newman, L. (1977). Silver-110 microgram sulfate analysis for the short time resolution of ambient levels of sulfur aerosol. Analytical Chemistry 49, 15791584.

Francis, E. H. \& Walker, B. H. (1986). Emplacement of alkali-dolerite sills relative to extrusive volcanism and sedimentary basins in the Carboniferous of Fife, Scotland. Transactions of the Royal Society of Edinburgh 77, 309-323.

Gaeta, M., Di Rocco, T. \& Freda, C. (2009). Carbonate Assimilation in Open Magmatic Systems: the Role of Melt-bearing Skarns and Cumulate-forming Processes. Journal of Petrology 50, 361-385.

Ganino, C., Arndt, N., Chauvel, C. \& Tornos, F. (2014). Metamorphic degassing of carbonates in the contact aureole of the Aguablanca $\mathrm{Cu}-\mathrm{Ni}-\mathrm{PGE}$ deposit, Spain. Contributions to Mineralogy and Petrology 168, 1-21.

Garcia, M. O., Pietruszka, A. J. \& Rhodes, J. M. (2003). A Petrologic Perspective of Kīlauea Volcano's Summit Magma Reservoir. Journal of Petrology 44, 2313-2339.

Gibb, F. G. F. \& Henderson, C. M. B. (2006). Chemistry of the Shiant Isles Main Sill, NW Scotland, and Wider Implications for the Petrogenesis of Mafic Sills. Journal of Petrology 47, 191-230.

Ginibre, C. \& Wörner, G. (2007). Variable parent magmas and recharge regimes of the Parinacota magma system (N. Chile) revealed by $\mathrm{Fe}, \mathrm{Mg}$ and Sr zoning in plagioclase. Lithos 98, 118-140.

Gorring, M. L. \& Naslund, H. R. (1995). Geochemical reversals within the lower 100 m of the Palisades sill, New Jersey. Contributions to Mineralogy and Petrology 119, 263-276.

Grinenko, L. I. (1985). Sources of sulfur of the nickeliferous and barren gabbrodolerite intrusions of the northwest Siberian platform. International Geology Review 27, 695-708.

Grove, T. L., Baker, M. B. \& Kinzler, R. J. (1984). Coupled CaAl-NaSi diffusion in plagioclase feldspar: Experiments and applications to cooling rate speedometry. Geochimica et Cosmochimica Acta 48, 2113-2121.

Gunn, B. M. (1966). Modal and element variation in Antarctic tholeiites. Geochimica et Cosmochimica Acta 30, 881-920.

Hart, S. R. \& Davis, K. E. (1978). Nickel partitioning between olivine and silicate melt. Earth and Planetary Science Letters 40, 203-219.

Haughton, D. R., Roeder, P. L. \& Skinner, B. J. (1974). Solubility of Sulfur in Mafic Magmas. Economic Geology 69, 451-467.

Hawkesworth, C. J., Lightfoot, P. C., Fedorenko, V. A., Blake, S., Naldrett, A. J., Doherty, W. \& Gorbachev, N. S. (1995). Magma differentiation and mineralisation in the Siberian continental flood basalts. Lithos 34.

Heaman, L. M., LeCheminant, A. N. \& Rainbird, R. H. (1992). Nature and timing of Franklin igneous events, Canada: Implications for a Late Proterozoic mantle plume and the break-up of Laurentia. Earth and Planetary Science Letters 109, 117-131.

Hildreth, W. \& Moorbath, S. (1988). Crustal contributions to arc magmatism in the Andes of Central Chile. Contributions to Mineralogy and Petrology 98, 455-489.

Holness, M. B., Stripp, G., Humphreys, M. C. S., Veksler, I. V., Nielsen, T. F. D. \& Tegner, C. (2011). Silicate liquid immiscibility within the crystal mush: late-stage magmatic microstructures in the skaergaard intrusion, east greenland. Journal of Petrology 52, 175-222.

Holness, M. B., Tegner, C., Nielsen, T. F. D., Stripp, G. \& Morse, S. A. (2007). A textural record of solidification and cooling in the skaergaard intrusion, East Greenland. Journal of Petrology 48, 2359-2377.

Hrouda, F. (1982). Magnetic anisotropy of rocks and its application in geology and geophysics. Geophysical surveys 5, 37-82.

Hryciuk, M., Bédard, J. H., Nabelek, P. \& Wing, B. A. (2014). (Submitted) Dynamic dikes and sterile sills: Contamination in the Neoproterozoic Franklin large igneous province, Victoria Island, Canada. Geology.

Huber, H., Koeberl, C., McDonald, I. \& Reimold, W. U. (2001). Geochemistry and petrology of Witwatersrand and Dwyka diamictites from South Africa: search for an extraterrestrial component. Geochimica et Cosmochimica Acta 65, 2007-2016.

Humphreys, M. C. S. (2009). Chemical Evolution of Intercumulus Liquid, as Recorded in Plagioclase Overgrowth Rims from the Skaergaard Intrusion. Journal of Petrology 50, 127-145.

Humphreys, M. C. S., Blundy, J. D. \& Sparks, R. S. J. (2006). Magma evolution and open-system processes at Shiveluch Volcano: Insights from phenocryst zoning. Journal of Petrology 47, 2303-2334.

Hunter, R. H. (1996). Texture Development in Cumulate Rocks. In: Richard Grant, C. (ed.) Developments in Petrology: Elsevier, 77-101.

Huppert, H. E. \& Sparks, R. S. J. (1980). The fluid dynamics of a basaltic magma chamber replenished by influx of hot, dense ultrabasic magma. Contributions to Mineralogy and Petrology 75, 279-289.

Iacono Marziano, G., Gaillard, F. \& Pichavant, M. (2007). Limestone assimilation and the origin of CO2 emissions at the Alban Hills (Central Italy): Constraints from experimental petrology. Journal of Volcanology and Geothermal Research 166, 91-105.

Iacono Marziano, G., Gaillard, F. \& Pichavant, M. (2008). Limestone assimilation by basaltic magmas: an experimental re-assessment and application to Italian volcanoes. Contributions to Mineralogy and Petrology 155, 719-738.

Irvine, T. N. (1975). Crystallization sequences in the Muskox intrusion and other layered intrusions-II. Origin of chromitite layers and similar deposits of other magmatic ores. Geochimica et Cosmochimica Acta 39, 991-1020.

Irvine, T. N. (1977). Origin of chromitite layers in the Muskox intrusion and other stratiform intrusions: A new interpretation. Geology 5, 273-277.

Irvine, T. N. (1980). Magmatic density currents and cumulus processes. American Journal of Science 280A, 1-58.

Irvine, T. N. (1982). Terminology for Layered Intrusions. Journal of Petrology 23, 127-162.

Irvine, T. N. (1987). Layering and related structures in the Duke Island and Skaergaard Intrusions: Similarities, differences, and origins. In. In: Parsons, I. (ed.) Origins of Igneous Layering. NATO ASI Series C. Mathematical and Physical Sciences 196, 185245.

Irvine, T. N. \& Smith, C. H. (1967). The ultramafic rocks of the Muskox intrusion, Northwest Territories, Canada. In: Wyllie, P.J. (ed.) Ultramafic and Related Rocks. New York: John Wiley \& Sons, Inc, 38-49.

Jefferson, C. W. (1985). Uppermost Shaler Group and its contact with the Natkusiak Basalts, Victoria Island, District of Franklin: in Current Research Part A. Geological Survey of Canada Paper 85-1A, 103.

Jefferson, C. W., Hulbert, L., Rainbird, R., H., Hall, G. E. M., Gregoire, D. C. \& Grinenko, L. I. (1994). Mineral resource assessment of the Neoproterozoic Franklin igneous events of Arctic Canada: comparison with the Permo-Triassic Noril'sk-Talnakh Ni-Cu-PGE deposits of Russia. Geological Survey of Canada Open File Report 2789, 48.

Jefferson, C. W., Nelson, W. E., Kirkham, R. V., Reedman, J. H. \& Scoates, R. F. J. (1985). Geology and Copper Occurrences of the Natkusiak Basalts, Victoria Island, District of Frankalin. Current Research, Part A. Geological Survey of Canada Paper 851A, 203-214.

Jones, D. S., Maloof, A. C., Hurtgen, M. T., Rainbird, R. H. \& Schrag, D. P. (2010). Regional and global chemostratigraphic correlation of the early Neoproterozoic Shaler Supergroup, Victoria Island, Northwestern Canada. Precambrian Research 181, 4363.

Jowitt, S. M. \& Ernst, R. E. (2013). Geochemical assessment of the metallogenic potential of Proterozoic LIPs of Canada. Lithos 174, 291-307.

Jugo, P. J., Luth, R. W. \& Richards, J. P. (2005). An Experimental Study of the Sulfur Content in Basaltic Melts Saturated with Immiscible Sulfide or Sulfate Liquids at $1300^{\circ} \mathrm{C}$ and $1 \cdot 0 \mathrm{GPa}$. Journal of Petrology 46, 783-798.

Keays, R. \& Lightfoot, P. (2010). Crustal sulfur is required to form magmatic NiCu sulfide deposits: evidence from chalcophile element signatures of Siberian and Deccan Trap basalts. Mineralium Deposita 45, 241-257.

Kerr, A. C., Tarney, J., Nivia, A., Marriner, G. F. \& Saunders, A. D. (1998). The internal structure of oceanic plateaus: inferences from obducted Cretaceous terranes in western Colombia and the Caribbean. Tectonophysics 292, 173-188.

Kiss, F. \& Oneschuk, D. (2010). First vertical derivative of the magnetic field, Minto Inlier aeromagnetic survey, Victoria Island, NTS 87 G/SE and parts of 87 G/NW, 88B/SE and 88 B/SW, Northwest Territories, . Geological Survey of Canada Open File Map 6705, scale 1:100,000.

Labidi, J., Cartigny, P., Hamelin, C., Moreira, M. \& Dosso, L. (2014). Sulfur isotope budget (32S, 33S, 34S and 36S) in Pacific-Antarctic ridge basalts: A record of mantle source heterogeneity and hydrothermal sulfide assimilation. Geochimica et Cosmochimica Acta 133, 47-67.

Larrea, P., França, Z., Lago, M., Widom, E., Galé, C. \& Ubide, T. (2012). Magmatic Processes and the Role of Antecrysts in the Genesis of Corvo Island (Azores Archipelago, Portugal). Journal of Petrology 54, 769-793.

Latypov, R. (2009). Testing the Validity of the Petrological Hypothesis 'No Phenocrysts, No Post-emplacement Differentiation'. Journal of Petrology.

Leclerc, F., Bédard, J. H., Harris, L. B., McNicoll, V. J., Goulet, N., Roy, P. \& Houle, P. (2011). Tholeiitic to calc-alkaline cyclic volcanism in the Roy Group, Chibougamau area, Abitibi Greenstone Belt - revised stratigraphy and implications for VHMS explorationGeological Survey of Canada Contribution 20100254. Ministère des Ressources naturelles et de la Faune Contribution 8439-2010-2011-17. Canadian Journal of Earth Sciences 48, 661-694.

Lehmann, J., Arndt, N., Windley, B., Zhou, M.-F., Wang, C. Y. \& Harris, C. (2007). Field Relationships and Geochemical Constraints on the Emplacement of the Jinchuan Intrusion and its Ni-Cu-PGE Sulfide Deposit, Gansu, China. Economic Geology 102, 75-94.

Leuthold, J., Müntener, O., Baumgartner, L. P. \& Putlitz, B. (2014). Petrological Constraints on the Recycling of Mafic Crystal Mushes and Intrusion of Braided Sills in the Torres del Paine Mafic Complex (Patagonia). Journal of Petrology 55, 917-949.

Li, C., Lightfoot, P. C., Amelin, Y. \& Naldrett, A. J. (2000). Contrasting Petrological and Geochemical Relationships in the Voisey's Bay and Mushuau Intrusions, Labrador, Canada: Implications for Ore Genesis. Economic Geology 95, 771-799.

Li, C. \& Naldrett, A. J. (1999). Geology and petrology of the Voisey's Bay intrusion: reaction of olivine with sulfide and silicate liquids. Lithos 47, 1-31.

Li, C., Ripley, E. M. \& Naldrett, A. J. (2009). A new genetic model for the giant $\mathrm{Ni}-\mathrm{Cu}-\mathrm{PGE}$ sulfide deposits associated with the Siberian flood basalts. Economic Geology 104, 291-301.

Lightfoot, P. C. \& Evans-Lamswood, D. (2015). Structural controls on the primary distribution of mafic-ultramafic intrusions containing $\mathrm{Ni}-\mathrm{Cu}-\mathrm{Co}-(\mathrm{PGE})$ sulfide mineralization in the roots of large igneous provinces. Ore Geology Reviews 64, 354-386.

Lissenberg, C. J., Bédard, J. H. \& van Staal, C. R. (2004). The structure and geochemistry of the gabbro zone of the Annieopsquotch ophiolite, Newfoundland: implications for lower crustal accretion at spreading ridges. Earth and Planetary Science Letters 229, 105-123.

Lissenberg, C. J. \& Dick, H. J. B. (2008). Melt-rock reaction in the lower oceanic crust and its implications for the genesis of mid-ocean ridge basalt. Earth and Planetary Science Letters 271, 311-325.

Liu, Y., Samaha, N.-T. \& Baker, D. R. (2007). Sulfur concentration at sulfide saturation (SCSS) in magmatic silicate melts. Geochimica et Cosmochimica Acta 71, 17831799.

Lundgaard, K. L., Tegner, C., Cawthorn, R. G., Kruger, F. J. \& Wilson, J. R. (2006). Trapped intercumulus liquid in the Main Zone of the eastern Bushveld Complex, South Africa. Contributions to Mineralogy and Petrology 151, 352-369.

Macdonald, F. A., Schmitz, M. D., Crowley, J. L., Roots, C. F., Jones, D. S., Maloof, A. C., Strauss, J. V., Cohen, P. A., Johnston, D. T. \& Schrag, D. P. (2010). Calibrating the Cryogenian. Science 327, 1241-1243.

Maier, W. D., Li, C. \& De Waal, S. A. (2001). WHY ARE THERE NO MAJOR Ni-Cu SULFIDE DEPOSITS IN LARGE LAYERED MAFIC-ULTRAMAFIC INTRUSIONS? The Canadian Mineralogist 39, 547-556.

Marsh, B. D. (1996). Solidification fronts and magmatic evolution. Mineralogical Magazine 60, 5-40.

Marsh, B. D. (2004). A magmatic mush column rosetta stone: the McMurdo Dry Valleys, Antarctica. EOS Trans Am Geophys Union 85.

Marsh, B. D. (2013). On some fundamentals of igneous petrology. Contributions to Mineralogy and Petrology 166, 665-690.

Martin, V. M., Davidson, J., Morgan, D. \& Jerram, D. A. (2010). Using the Sr isotope compositions of feldspars and glass to distinguish magma system components and dynamics. Geology 38, 539-542.

Mathez, E. A., Hunter, R. H. \& Kinzler, R. (1997). Petrologic evolution of partially molten cumulate: the Atok section of the Bushveld Complex. Contributions to Mineralogy and Petrology 129, 20-34.

Mathieu, J., Kontak, D. J. \& Turner, E. C. (2013). A fluid inclusion study of diagenetic fluids in Proterozoic and Paleozoic carbonate rocks, Victoria Island, NWT. Geofluids 13, 559-578.

Matzen, A. K., Baker, M. B., Beckett, J. R. \& Stolper, E. M. (2011). Fe-Mg Partitioning between Olivine and High-magnesian Melts and the Nature of Hawaiian Parental Liquids. Journal of Petrology 52, 1243-1263.

McBirney, A. R. (1996). The Skaergaard Intrusion. In: Richard Grant, C. (ed.) Developments in Petrology: Elsevier, 147-180.

McDonald, I. \& Viljoen, K. S. (2006). Platinum-group element geochemistry of mantle eclogites: a reconnaissance study of xenoliths from the Orapa kimberlite, Botswana. Applied Earth Science 115, 81-93.

McDonough, W. F. \& Sun, S. s. (1995). The composition of the Earth. Chemical Geology 120, 223-253.

Meurer, W. P. \& Boudreau, A. E. (1998). Compaction of Igneous Cumulates Part I: Geochemical Consequences for Cumulates and Liquid Fractionation Trends. The Journal of Geology 106, 281-292.

Miller, J. D., Green, J. C., Severson, M. J., Chandler, V. W., Hauck, S. A., Peterson, D. M. \& Wahl, T. E. (2002). Geology and mineral potential of the Duluth Complex and related rocks of northeastern Minnesota. Minnesota Geological Survey Report of Investigations 58, 207.

Mitchell, A. A., Eales, H. V. \& Krueger, F. J. (1998). Magma replenishment, and the significance of poikilitic textures, in the Lower Main Zone of the western Bushveld Complex, South Africa. Mineralogical Magazine 62, 435-450.

Mollo, S. \& Vona, A. (2014). The geochemical evolution of clinopyroxene in the Roman Province: A window on decarbonation from wall-rocks to magma. Lithos 192-195, 1-7.

Mondal, S. K. \& Mathez, E. A. (2007). Origin of the UG2 chromitite layer, Bushveld Complex. Journal of Petrology 48, 495-510.

Morgan, D. J., Blake, S., Rogers, N. W., De Vivo, B., Rolandi, G. \& Davidson, J. P. (2006). Magma chamber recharge at Vesuvius in the century prior to the eruption of A.D. 79. Geology 34, 845-848.

Morse, S. A. (1984). Cation Diffusion in Plagioclase Feldspar. Science 225, 504505.

Müller, T., Dohmen, R., Becker, H. W., Heege, J. \& Chakraborty, S. (2013). FeMg interdiffusion rates in clinopyroxene: experimental data and implications for $\mathrm{Fe}-\mathrm{Mg}$ exchange geothermometers. Contributions to Mineralogy and Petrology 166, 1563-1576.

Murphy, M. D., Sparks, R. S. J., Barclay, J., Carroll, M. R. \& Brewer, T. S. (2000). Remobilization of andesite magma by intrusion of mafic magma at the Soufriere Hills volcano, Montserrat, West Indies. Journal of Petrology 41, 21-42.

Nabelek, P. I., Bédard, J. H., Hryciuk, M. \& Hayes, B. (2013). Short-duration contact metamorphism of calcareous sedimentary rocks by Neoproterozoic Franklin gabbro sills and dykes on Victoria Island, Canada. Journal of Metamorphic Geology 31, 205-220.

Nakamura, Y. (1973). Origin of Sector-Zoning of Igneous Clinopyroxenes. American Mineralogist 58, 986-990.

Naldrett, A. J. (1992). A Model for the Ni-Cu-PGE Ores of the Noril'sk Region and Its Application to Other Areas of Flood Basalt. Economic Geology 87, 1945-1962.

Naldrett, A. J. (1999). World-class Ni-Cu-PGE deposits: key factors in their genesis. Mineralium Deposita 34, 227-240.

Naldrett, A. J., Asif, M., Krstic, S. \& Li, C. (2000). The Composition of Mineralization at the Voisey's Bay Ni-Cu Sulfide Deposit, with Special Reference to Platinum-Group Elements. Economic Geology 95, 845-865.

Namur, O., Charlier, B., Toplis, M. J., Higgins, M. D., Liégeois, J.-P. \& Vander Auwera, J. (2010). Crystallization Sequence and Magma Chamber Processes in the Ferrobasaltic Sept Iles Layered Intrusion, Canada. Journal of Petrology 51, 1203-1236.

Naslund, H. R. (1989). Petrology of the Basistoppen Sill, East Greenland: A Calculated Magma Differentiation Trend. Journal of Petrology 30, 299-319.

Naslund, H. R., Bédard, J. H., Steigerwaldt, K. \& Dye, D. (2013). Origin of S and M and reversed S and M shaped profiles in two Proterozoic mafic sills from the Franklin magmatic event, Victoria Island, NWT, Canada. Geophysical Research Abstracts 15.

Naslund, H. R. \& McBirney, A. R. (1996). Mechanisms of Formation of Igneous Layering. In: Richard Grant, C. (ed.) Developments in Petrology: Elsevier, 1-43.

Neave, D. A., Passmore, E., Maclennan, J., Fitton, G. \& Thordarson, T. (2013). Crystal-Melt Relationships and the Record of Deep Mixing and Crystallization in the ad 1783 Laki Eruption, Iceland. Journal of Petrology 54, 1661-1690.

Nobre Silva, I. G., Weis, D., Barling, J. \& Scoates, J. S. (2009). Leaching systematics and matrix elimination for the determination of high-precision Pb isotope compositions of ocean island basalts. Geochemistry, Geophysics, Geosystems 10, Q08012.

O'Driscoll, B., Donaldson, C. H., Troll, V. R., Jerram, D. A. \& Emeleus, C. H. (2007). An Origin for Harrisitic and Granular Olivine in the Rum Layered Suite, NW Scotland: a Crystal Size Distribution Study. Journal of Petrology 48, 253-270.

O'Hara, M. J. (1977). Geochemical evolution during fractional crystallization of a periodically refilled magma chamber. Nature 266, 503-507.

Onuma, K. \& Tohara, T. (1983). Effect of chromium on phase relations in the join forsterite-anorthite-diopside in air at 1 Atm. Contributions to Mineralogy and Petrology 84, 174-181.

Parsons, I. (1986). Origins of Igneous Layering. NATO ASI Series C. Mathematical and Physical Sciences 196, 666.

Passmore, E., Maclennan, J., Fitton, G. \& Thordarson, T. (2012). Mush Disaggregation in Basaltic Magma Chambers: Evidence from the ad 1783 Laki Eruption. Journal of Petrology 53, 2593-2623.

Paterson, S. R. (2009). Magmatic tubes, pipes, troughs, diapirs, and plumes: Latestage convective instabilities resulting in compositional diversity and permeable networks in crystal-rich magmas of the Tuolumne batholith, Sierra Nevada, California. Geosphere 5, 496-527.

Pehrsson, S. J. \& Buchan, K. L. (1999). Borden dykes of Baffin Island, Northwest Territories: a Franklin U-Pb baddeleyite age and a paleomagnetic reinterpretation. Canadian Journal of Earth Sciences 36, 65-73.

Peterson, R. C., Williamson, M.-C. \& Rainbird, R. H. (2014). Gossan Hill, Victoria Island, Northwest Territories: An analogue for mine waste reactions within permafrost and implication for the subsurface mineralogy of Mars. Earth and Planetary Science Letters 400, 88-93.

Philpotts, A. R., Shi, J. \& Brustman, C. (1998). Role of plagioclase crystal chains in the differentiation of partly crystallized basaltic magma. Nature 395, 343-346.

Pistone, M., Caricchi, L., Ulmer, P., Reusser, E. \& Ardia, P. (2013). Rheology of volatile-bearing crystal mushes: Mobilization vs. viscous death. Chemical Geology 345, 16-39.

Pretorius, W., Weis, D., Williams, G., Hanano, D., Kieffer, B. \& Scoates, J. (2006). Complete Trace Elemental Characterisation of Granitoid (USGS G-2, GSP-2) Reference

Materials by High Resolution Inductively Coupled Plasma-Mass Spectrometry. Geostandards and Geoanalytical Research 30, 39-54.

Prince, J. K. G. (2014). Sequence stratigraphic, lithostratigraphic and stable isotope analysis of the Minto Inlet Formation and Kilian Formation, of the Shaler Supergroup, Northwest Territories. Thesis. Carleton University, Ottawa, 127.

Puffer, J. H., Block, K. A. \& Steiner, J. C. (2009). Transmission of Flood Basalts through a Shallow Crustal Sill and the Correlation of Sill Layers with Extrusive Flows : The Palisades Intrusive System and the Basalts of the Newark Basin, New Jersey, U.S.A. Chicago, II, ETATS-UNIS: University of Chicago Press.

Rainbird, R. (1993). The Sedimentary Record of Mantle Plume Uplift Preceding Eruption of the Neoproterozoic Natkusiak Flood Basalt. Journal of Geology 101, 305-318.

Rainbird, R. H., Jefferson, C. W. \& Young, G. M. (1996). The early Neoproterozoic sedimentary Succession B of northwestern Laurentia: Correlations and paleogeographic significance. Geological Society of America Bulletin 108, 454-470.

Reubi, O. \& Blundy, J. (2008). Assimilation of Plutonic Roots, Formation of HighK 'Exotic’ Melt Inclusions and Genesis of Andesitic Magmas at Volcán De Colima, Mexico. Journal of Petrology 49, 2221-2243.

Ripley, E. M. \& Li, C. (2003). Sulfur isotope exchange and metal enrichment in the formation of magmatic Cu-Ni-(PGE) deposits. Economic Geology 98, 635-641.

Ripley, E. M. \& Li, C. (2011). A Review of Conduit-Related Ni-Cu-(PGE) Sulfide Mineralization at the Voiseys Bay Deposit, Labrador, and the Eagle Deposit, Northern Michigan. Reviews in Economic Geology 17.

Ripley, E. M. \& Li, C. (2013). Sulfide Saturation in Mafic Magmas: Is External Sulfur Required for Magmatic Ni-Cu-(PGE) Ore Genesis? Economic Geology 108, 45-58.

Ripley, E. M., Lightfoot, P. C., Li, C. \& Elswick, E. R. (2003). Sulfur isotopic studies of continental flood basalts in the Noril'sk region: implications for the association between lavas and ore-bearing intrusions. Geochimica et Cosmochimica Acta 67, 28052817.

Ripley, E. M., Sarkar, A. \& Li, C. (2005). Mineralogic and Stable Isotope Studies of Hydrothermal Alteration at the Jinchuan Ni-Cu Deposit, China. Economic Geology 100, 1349-1361.

Roeder, P. L. \& Emslie, R. F. (1970). Olivine-liquid equilibrium. Contributions to Mineralogy and Petrology 29, 275-289.

Roelofse, F. \& Ashwal, L. D. (2012). The Lower Main Zone in the Northern Limb of the Bushveld Complex—a>1.3 km Thick Sequence of Intruded and Variably Contaminated Crystal Mushes. Journal of Petrology 53, 1449-1476.

Seat, Z., Beresford, S. W., Grguric, B. A., Gee, M. A. M. \& Grassineau, N. V. (2009). Reevaluation of the Role of External Sulfur Addition in the Genesis of Ni-Cu-PGE Deposits: Evidence from the Nebo-Babel Ni-Cu-PGE Deposit, West Musgrave, Western Australia. Economic Geology 104, 521-538.

Shellnutt, J. G., Dostal, J. \& Keppie, J. D. (2004). Petrogenesis of the 723 Ma Coronation sills, Amundsen basin, Arctic Canada: implications for the break-up of Rodinia. Precambrian Research 129, 309-324.

Shirley, D. N. (1985). Compaction of igneous cumulates. Journal of Geology 94, 795-809.

Simkin, T. (1967). Flow differentiation in the picritic sills of North Skye. In Wyllie, P.J., ed, Ultramafic and Related Rocks, John Wiley, Hoboken, N.J., 64-69.

Smewing, J. D. (1981). Mixing characteristics and compositional differences in mantle-derived melts beneath spreading axes; evidence from cyclically layered rocks in the ophiolite of North Oman. Journal of Geophysical Research 86, 2645-2659.

Song, X.-Y., Danyushevsky, L., Keays, R., Chen, L.-M., Wang, Y.-S., Tian, Y.-L. \& Xiao, J.-F. (2012). Structural, lithological, and geochemical constraints on the dynamic magma plumbing system of the Jinchuan $\mathrm{Ni}-\mathrm{Cu}$ sulfide deposit, NW China. Mineralium Deposita 47, 277-297.

Sparks, R. S. J., Huppert, H. E., Kerr, R. C., McKenzie, D. P. \& Tait, S. R. (1985). Postcumulus processes in layered intrusions. Geological Magazine 122, 555-568.

Sparks, R. S. J., Sigurdsson, H. \& Wilson, L. (1977). Magma mixing: a mechanism of triggering acid explosive eruptions. Nature 267, 315-318.

Sun, S. S. \& McDonough, W. F. (1989). Chemical and isotopic systematics of oceanic basalts: implications for mantle composition and processes. Geological Society, London, Special Publications 42, 313-345.

Tait, S. R. (1985). Fluid dynamic and geochemical evolution of cyclic unit 10, Rhum, Eastern Layered Series. Geological Magazine 122, 469-484.

Tarling, D. H. \& Hrouda, F. (1993). The Magnetic Anisotropy of Rocks. Chapman and Hall, London, 217.

Tegner, C. (1997). Iron in plagioclase as a monitor of the differentiation of the Skaergaard intrusion. Contributions to Mineralogy and Petrology 128, 45-51.

Tegner, C., Wilson, J. R. \& Brooks, C. K. (1993). Intraplutonic Quench Zones in the Kap Edvard Holm Layered Gabbro Complex, East Greenland. Journal of Petrology 34, 681-710.

Tepley, F. J., Davidson, J. P., Tilling, R. I. \& Arth, J. G. (2000). Magma Mixing, Recharge and Eruption Histories Recorded in Plagioclase Phenocrysts from El Chichón Volcano, Mexico. Journal of Petrology 41, 1397-1411.

Thomson, D., Rainbird, R. H. \& Dix, G. (2014). Architecture of a Neoproterozoic intracratonic carbonate ramp succession: Wynniatt Formation, Amundsen Basin, Arctic Canada. Sedimentary Geology 299, 119-138.

Thorsteinsson, R. \& Tozer, E. T. (1962). Banks, Victoria and Stefansson Islands, Arctic Archipelago. Geological Survey of Canada Memoir 330, 85.

Tredoux, M., Lindsay, N. M., Davies, G. \& McDonald, I. (1995). The fractionation of platinum-group elements in magmatic systems, with the suggestion of a novel causal mechanism. South African Journal of Geology 98, 157-167.

Troll, V. R., Deegan, F. M., Jolis, E. M., Harris, C., Chadwick, J. P., Gertisser, R., Schwarzkopf, L. M., Borisova, A. Y., Bindeman, I. N., Sumarti, S. \& Preece, K. (2013). Magmatic differentiation processes at Merapi Volcano: inclusion petrology and oxygen isotopes. Journal of Volcanology and Geothermal Research 261, 38-49.

Usselman, T. M. \& Hodge, D. S. (1978). Thermal control of low-pressure fractionation processes. Journal of Volcanology and Geothermal Research 4, 265-281.
van Acken, D., Thomson, D., Rainbird, R. H. \& Creaser, R. A. (2013).
Constraining the depositional history of the Neoproterozoic Shaler Supergroup, Amundsen Basin, NW Canada: Rhenium-osmium dating of black shales from the Wynniatt and Boot Inlet Formations. Precambrian Research 236, 124-131.

Varfalvy, V., Hébert, R., Bedard, J. H. \& Laflèche, M. R. (1997). Petrology and geochemistry of pyroxenite dykes in upper mantle peridotites of the North Arm Mountain massif, Bay of Islands Ophiolite, Newfoundland: Implications for the genesis of boninitic and related magmas. Canadian Mineralogist 35, 543-570.

Wager, L. R. \& Brown, G. M. (1968). Layered igneous rocks. Olivier Boyd, Edinburgh.

Wager, L. R. \& Deer, W. A. (1939). Geological investigations in East Greenland Part III. The petrology of the Skaergaard Intrusion, Kangerdlugssuaq, East Greenland. Meddelelser om Grønland 105, 1-352.

Wallace, G. S. \& Bergantz, G. W. (2002). Wavelet-based correlation (WBC) of zoned crystal populations and magma mixing. Earth and Planetary Science Letters 202, 133-145.

Weis, D., Kieffer, B., Maerschalk, C., Barling, J., de Jong, J., Williams, G. A., Hanano, D., Pretorius, W., Mattielli, N., Scoates, J. S., Goolaerts, A., Friedman, R. M. \& Mahoney, J. B. (2006). High-precision isotopic characterization of USGS reference materials by TIMS and MC-ICP-MS. Geochemistry, Geophysics, Geosystems 7, Q08006.

Welsch, B., Faure, F., Famin, V., Baronnet, A. \& Bachèlery, P. (2012). Dendritic Crystallization: A Single Process for all the Textures of Olivine in Basalts? Journal of Petrology.

Wendlandt, R. F. (1982). Sulfide saturation of basalt and andesite melts at high pressures and temperatures. American Mineralogist 67, 877-885.

Wenzel, T., Baumgartner, L. P., Brugmann, G. E., Konnikov, E. G. \& Kislov, E. V. (2002). Partial Melting and Assimilation of Dolomitic Xenoliths by Mafic Magma: the Ioko-Dovyren Intrusion (North Baikal Region, Russia). Journal of Petrology 43, 20492074.

Wignall, P. B. (2001). Large igneous provinces and mass extinctions. EarthScience Reviews 53, 1-33.

Williams, G. D. \& Stelck, C. R. (1975). Speculations on the Cretaceous paleogeography of North America. In: Caldwell, W.G.E. (Ed.), The Cretaceous system in the Western Interior of North America. . Geological Association of Canada, Special Paper Number 13, 1-20.

Williamson, N., Bédard, J. H., Ootes, L., Rainbird, R., Cousens, B. \& Zagorevski, A. (2013). Volcanostratigraphy and significance of the southern lobe Natkusiak Formation flood basalts, Victoria Island, Northwest Territories. Geological Survey of Canada, Current Research 2013-16, 13.

Wilson, A. H. (2012). A Chill Sequence to the Bushveld Complex: Insight into the First Stage of Emplacement and Implications for the Parental Magmas. Journal of Petrology 53, 1123-1168.

Young, G. M. (1981). The Amundsen Embayment, Northwest Territories, relevance to the Upper Proterozoic evolution of North America, in Campbell, F.H.A., ed., Proterozoic Basins of Canada. Geological Survey of Canada Paper 81-10, 203-211.

Žák, J., Verner, K., Johnson, K. \& Schwartz, J. J. (2012). Magma emplacement process zone preserved in the roof of a large Cordilleran batholith, Wallowa Mountains, northeastern Oregon. Journal of Volcanology and Geothermal Research 227-228, 61-75.

Zieg, M. J. \& Marsh, B. D. (2012). Multiple Reinjections and Crystal-mush Compaction in the Beacon Sill, McMurdo Dry Valleys, Antarctica. Journal of Petrology 53, 2567-2591.

## 9. Appendices

A. Copy of "Bédard et al. (2012) Fault-mediated melt ascent in a Neoproterozoic continental flood basalt, the Franklin sills, Victoria Island, Canada" published in Geological Society of America Bulletin in 2012
B. Copy of "Nabelek et al. (2013) Short Duration contact metamorphism of calcareous sedimentary rocks by Neoproterozoic Franklin gabbro sills and dykes on Victoria Island, Canada" published in Journal of Metamorphic Geology in 2013
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## Geological Society of America Bulletin

# Fault-mediated melt ascent in a Neoproterozoic continental flood basalt province, the Franklin sills, Victoria Island, Canada 

J.H. Bédard, H.R. Naslund, P. Nabelek, A. Winpenny, M. Hryciuk, W. Macdonald, B. Hayes, K. Steigerwaldt, T. Hadlari, R. Rainbird, K. Dewing and E. Girard

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#### Abstract

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## Notes

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# Fault-mediated melt ascent in a Neoproterozoic continental flood basalt province, the Franklin sills, Victoria Island, Canada 

J.H. Bédard ${ }^{1, \dagger}$, H.R. Naslund ${ }^{2}$, P. Nabelek ${ }^{3}$, A. Winpenny ${ }^{4}$, M. Hryciuk $^{5}$, W. Macdonald ${ }^{2}$, B. Hayes ${ }^{6}$, K. Steigerwaldt ${ }^{2}$, T. Hadlari ${ }^{7}$, R. Rainbird ${ }^{8}$, K. Dewing ${ }^{7}$, and É. Girard ${ }^{9}$<br>${ }^{1}$ Geological Survey of Canada (GSC-Québec), 490 de la Couronne, Québec, Québec G1K 9A9, Canada<br>${ }^{2}$ Department of Earth Sciences, State University of New York, Binghamton, New York 13902, USA<br>${ }^{3}$ Department of Geological Sciences, University of Missouri, Columbia, Missouri 65211, USA<br>${ }^{4}$ Department of Earth Sciences, University of Bristol, Wills Memorial Building, Bristol BS8 IRJ, UK<br>${ }^{5}$ Department of Earth and Planetary Sciences, McGill University, 3450 University Street, Montréal, Québec H3A 2A7, Canada<br>${ }^{6}$ School of Earth, Ocean and Planetary Sciences, Cardiff University, Cardiff CF10 3YE, Wales, UK<br>${ }^{7}$ Geological Survey of Canada (GSC-Calgary), 3303 33rd Street NW, Calgary, Alberta T2L 2A7, Canada<br>${ }^{8}$ Geological Survey of Canada (GSC-Ottawa), 601 Booth Street, Ottawa, Ontario K1A 0E8, Canada<br>${ }^{9}$ Institut National de la Recherche Scientifique - Eau Terre et Environnement, 490 de la Couronne, Québec, Québec G1K 9A9, Canada


#### Abstract

The Neoproterozoic Franklin large igneous province on Victoria Island, Canada, is characterized by continental flood basalts and a sill-dominated feeder system. Field relationships indicate that fault-guided transfer zones allowed magma to jump up-section to form higher-level intrusions. Where sills connect to dikes and magmas moved upsection, roof and wall rocks are characterized by wide and intense contact-metamorphic haloes, consistent with throughflow of magma. The geometric constraints suggest that conduits may have opened episodically and then closed when magma pressure waned. The episodic nature of conduit opening events can explain the pulsed ascent of crystal slurries, and may also play a role in the deposition of Ni -sulfides.


## INTRODUCTION

Most flood basalt provinces are dominated by evolved lava compositions, requiring significant intracrustal differentiation (Cox, 1980; Dostal and Dupuy, 1984, Hooper and Hawkesworth, 1993; Hawkesworth et al., 1995; Elliot and Fleming, 2004; Jourdan et al., 2007; Larsen and Pedersen, 2009). Tholeiitic dike swarms riddle shields and pericratonic areas and are thought to be the root zones of flood basalt events, possibly related to mantle plumes (e.g., Fahrig, 1987; Ernst et al., 1995; Bryan and Ernst, 2008). Where dike swarms can be related to

[^1]well-developed passive-margin sequences with associated volcanic rocks, dike orientations reflect the interplay between preexisting structural anisotropy and the stress field that is developed as continental breakup proceeds (May, 1971; Sykes, 1978; Ernst et al., 1995; Watanabe et al., 1999; McHone et al., 2005). On the other hand, sills dominate many tholeiitic feeder systems beneath Proterozoic and Phanerozoic continental flood basalt provinces (e.g., Baragar, 1976; Francis and Walker, 1986; Hawkesworth et al., 1995; Chevallier and Woodford, 1999; Bédard et al., 2007; Bryan and Ernst, 2008), magmatically active rifted margins (Bédard et al., 1984; Thomson and Hutton, 2004; Rocchi et al., 2007; White et al., 2008), oceanic plateaus (Kerr et al., 1998), ophiolitic and oceanic lower crust (Bédard, 1991; Dick et al., 2000; Lissenberg et al., 2004), and Archean tholeiitic lava plains (Bédard et al., 2009a). In sill-dominated plumbing systems, there is generally a dearth of steeply dipping conduits that could explain ascent of melt to the surface from their mantle sources.
Dikes are thought to change into sills as a consequence of the dike propagator intersecting a discontinuity or stress barrier, such as an anomalously weak or strong unit (e.g., Watanabe et al., 1999; Gudmundsson and Brenner, 2001; Gudmundsson, 2011). In contrast, much less is known about how sills change into dikes. Some basalt provinces are dominated by saucershaped sills (Thomson and Hutton, 2004; Hansen and Cartwright, 2006; Galerne et al., 2008; Polteau et al., 2008), where the up-curved edges of a lower sill fed higher-level sills (Thomson and Hutton, 2004), such that "feeder" dikes are
steep offshoots from the terminations of underlying sills. In saucer-shaped sills from the Karoo (South Africa) and Rockall Trough (west of Great Britain), the data imply shallow, nearly "subvolcanic" emplacement (Thomson and Hutton, 2004; Hansen and Cartwright, 2006; Galerne et al., 2008), and it has been proposed that their upward transgression is a consequence of lateral expansion and interaction with the free surface (Fialko, 2001; Malthe-Sørenssen et al., 2004; Galland et al., 2009). On the basis of a comparison between observed thicknesses and emplacement depth, Polteau et al. (2008) proposed that sills with diameters less than 10 km would correspond to emplacement depths of less than 2 km . While these transgression mechanisms may characterize subvolcanic sill-dominated feeder systems, they may not apply to more deeply emplaced sills, where interactions with the free surface should not be a factor. There seem to be two end-member possibilities to explain upward transgression of deeply emplaced sills.
(1) The pathway by which a sill migrated up-section did not exist prior to sill emplacement and was generated by an interplay among the stress field, magma flow, magma pressure, and host-rock deformation (Pollard and Segall, 1987; Lister and Kerr, 1991; Rubin, 1995; Gerya and Burg, 2007; Gudmundsson, 2011). Because magmatic intrusion rates are much faster than any possible tectonic accommodation rates (Paterson and Tobisch, 1992; Vigneresse et al., 1999), the local stress field in a rapidly inflating and propagating sill will be dominated by magmatic flow. Menand (2008) suggested that during rapid lateral propagation of magma in a
sill, viscously controlled dynamics may lead to upward transgression of the intrusion along an induced reverse fault where the sill propagator has its maximum curvature.
(2) Magma may jog up-section because of the presence of preexisting steep structures such as faults. This possibility is often suggested but rarely demonstrated (but see Hutton and Ingram, 1992; Thomson, 2007). Structurally mediated melt ascent is intuitively attractive, but unambiguous demonstration of a genetic relationship between guiding structures and magma emplacement from field relationships requires exceptional exposures and preservation of fabrics and contacts. Perversely, robust melt channels tend to consume and recrystallize the wall rocks that contain the evidence (e.g., Huppert and Sparks, 1988; Holness and Humphreys, 2003). In addition, well-exposed crustal sections beneath basaltic lava fields are rare, and in seismic-reflection profiles, steeply dipping
up-flow conduits would be difficult to image. Consequently, the poor documentation of structurally controlled magmatic ascent in silldominated basaltic feeder systems may partly be an observational problem. In this paper, we present evidence from the superbly exposed and preserved Neoproterozoic Franklin magmatic suite of Victoria Island, Canada, that strongly supports an ephemeral, fault-mediated, magma transfer mechanism to explain up-section sill transgression.

## VICTORIA ISLAND GEOLOGY

The Neoproterozoic Shaler Supergroup (Rainbird, 1993) is located in the Minto Inlier, central Victoria Island, Canada (Fig. 1; Thorsteinsson and Tozer, 1962). The Shaler Supergroup is dominated by shallow-water carbonate rocks, with subordinate mud rocks, sandstones, and sulfate evaporites, and it represents a shal-
low-water Proterozoic intracontinental basin developed within the supercontinent Rodinia (Young, 1981; Rainbird, 1993). Archean granites and Paleoproterozoic sedimentary rocks underlie the Neoproterozoic strata (Rainbird et al., 1994). The Natkusiak flood basalts (Baragar 1976; Jefferson, 1985; Jefferson et al., 1985; Dostal et al., 1986; Dupuy et al., 1995) are the uppermost formation of the Shaler Supergroup. The Natkusiak Formation has a thin, heterogeneous basal unit (flows, volcaniclastics, and lahars), succeeded by sheet flows defining two geochemical volcanic cycles (each $\sim 500 \mathrm{~m}$ thick; Dostal et al., 1986). In each cycle, basalts range from high-MgO olivine-phyric basal flows (up to $11 \mathrm{wt} \% \mathrm{MgO}$ ), to evolved plagioclase + clinopyroxene + magnetite-phyric upper flows ( $\sim 3-4 \mathrm{wt} \% \mathrm{MgO}$ ). The upper contact is a Cambrian erosional unconformity upon which a Paleozoic clastic and carbonate sequence was deposited.


Figure 1. Simplified regional map of Victoria Island (Red polygon on inset labeled VI for Victoria Island, which shows location in Canada), based on the compilation of Hulbert et al. (2005) and our new observations from 2010 and 2011 mapping. The boxes show the locations of the northern and southern feeder dikes (NFD and SFD), and of Figure 2. Individual sills are not represented.

The Franklin sills occur throughout the Shaler Supergroup, and some of them may represent feeders to the lavas (Baragar, 1976; Rainbird et al., 1994; Jefferson et al., 1994; Hulbert, 1998; Hulbert et al., 2005). Franklin sills range in thickness from less than 1 m to over 100 m , but most are $20-60 \mathrm{~m}$ thick. Individual sills commonly extend for 20 km or more along-strike with little obvious change in thickness. Depending on the specific location, sills constitute anywhere from $10 \%$ to $75 \%$ of the section. Sills of similar type and age also occur in the Coronation Gulf to the south (Rainbird et al., 1996; Shellnutt et al., 2004), and coeval, geochemically very similar magmas correlated with the Franklin event extend from Greenland to Siberia (Heaman et al., 1992; Pehrsson and Buchan, 1999; Denyszyn et al., 2009; Richard Ernst, 2011, personal commun.). Two age populations are present in the sills, ca. 723 and 716 Ma (Heaman et al., 1992; Michael Hamilton, 2009, personal commun.).

Shaler Supergroup rocks are gently folded along NE-trending axes into open upright folds with a wavelength of $\sim 200 \mathrm{~km}$ (Fig. 1). Beds typically dip no more than $5^{\circ}$ or $10^{\circ}$, however, and there is rarely a cleavage or other apparent outcrop-scale fabric related to folding. The total thickness (excluding sills) of the Supergroup is $\sim 4 \mathrm{~km}$, including the $\sim 1.1$-km-thick Natkusiak Formation flood basalts that cap the sequence.
A new aeromagnetic map reveals prominent NNW-trending magnetic lineaments (Kiss and Oneschuk, 2010; Fig. 2B) that correspond to prominent linear topographic lows, locally containing cataclasites, which we interpret to represent faults. Some of these lineaments mark sharp breaks in stratigraphy indicating east-side-down throws (maps in preparation) and exhibit associated structures indicating normal motion synchronous with emplacement of magmas with Franklin affinities (this paper)-and so we interpret them to be Neoproterozoic synmagmatic normal faults.

The Neoproterozoic rocks are unconformably overlain (Fig. 1) by a Paleozoic sequence ranging in age from Cambrian to Devonian, composed of clastic sedimentary rocks at the base grading up into dominant carbonate rocks (Thorsteinsson and Tozer, 1962; Durbano et al., 2010). The feather-edge of the unconformity is very irregular in detail, with development of an oxidized regolith and local paleotopographic effects. All rocks are dissected into horst and graben structures by ENE- to E-W-trending block faults of post-Devonian age (Hadlari and Bédard, 2008; this study), which we call postPaleozoic faults henceforth. The distribution of these faults on Figure 1 represents the limit of the 2010 mapping, and they are more widespread than is shown.

## Franklin Sills

Most Franklin sills contain fairly homogeneous, evolved diabasic rocks and show only modest internal differentiation (e.g., Shellnutt et al., 2004), but there are subordinate occurrences of plagioclase + clinopyroxene + olivineporphyritic sills and of strongly differentiated primitive sills with olivine cumulate bases. Apparent contact-metamorphic haloes are typically less than a few meters wide. Olivine-rich sills have lower chilled margins with up to 12 $\mathrm{wt} \% \mathrm{MgO}$, and olivine up to $\mathrm{Fo}_{87}$ with NiO up to $0.35 \mathrm{wt} \%$. Minor chromite microphenocrysts
are also present $(\mathrm{Cr} /[\mathrm{Cr}+\mathrm{Al}]$ between 0.5 and 0.7 ). Diabasic sills typically contain plagioclase and clinopyroxene ( $\pm$ olivine) microphenocrysts and generally have chilled margins with MgO around $6 \%-8 \%$. Most sills are concordant with bedding in host sedimentary rocks and, where later block faulting has spared the evidence, extend along-strike for many tens of kilometers with no apparent change in thickness or position in the host stratigraphy. In the course of a regional mapping program (Rainbird et al., 2010), several occurrences and types of sill terminations were observed, the most significant of which are described next. In this paper, we focus

Figure 2. (A) Landsat image showing the Uhuk Massif. The purple colors correspond to mafic igneous rocks. A few of the larger faults and units are also shown, as is the location of Figure 4 (box). (B) The first derivative of the magnetic field (Kiss and Oneschuk, 2010) for the same area. Note the prominent NNW-trending lineament that crosses the eastern end of the Uhuk Massif. The igneous rocks from part A are outlined by a white line. Pz-Paleozoic.

on three areas that show evidence for up-section sill migration.
(1) The northern feeder dike (NFD; Figs. 1, 3A, 3B, and 3C) records injection of magma from a sill into a normal fault, intense thermal metamorphism of capping strata, and possible deflation of the lower sill and partial closure of the feeder channel as the capping strata subsided.
(2) The southern feeder dike (SFD; Figs. 1 and 3D) contains blocks of limestone that are tilted over $60^{\circ}$ in a magmatically dominated upflow zone in a synmagmatic normal fault.
(3) The Uhuk Massif (Figs. 1, 2, and 4-9) exposes a complex up-flow zone, where olivinerich magma migrated up-section and then extended to the west for tens of kilometers (Figs. 2 A and 6 A ) as a concordant sill. Capping and floor strata are intensely metamorphosed, consistent with the existence of a magmatic throughflow system. The roof zone is characterized by breccias, branching dikes, and partially detached slivers (Figs. 4-7 and 9), implying synkinematic magmatic emplacement along an east-dipping normal fault. The contact-metamorphosed roof zone carbonates were injected by Fe-rich fluids, leading to the development of oxide-sulfide-rich skarns (Fig. 8).

## Northern Feeder Dike

The thin-bedded limestones and silty limestones of the Boot Inlet Formation ( $\sim 3-4 \mathrm{~km}$ paleodepth) dip $\sim 5^{\circ}$ toward the southeast in this area, and they are intruded by two sills linked by a dike (Fig. 3A). The diabasic upper sill extends at least $2 \times 5 \mathrm{~km}$ and forms a prominent mesa that terminates abruptly on valley walls. Sill-capped mesas on the other sides of wide valleys to the east and west may be extensions of the same sill. In the area pictured in Figure 3A, the upper sill's roof has been eroded away, but exposed portions are up to 50 m thick. Lower chilled margins are well exposed locally, but talus generally obscures the contact. The lower sill is separated from the upper sill by $\sim 25-30 \mathrm{~m}$ (stratigraphic thickness) of thin-bedded limestone, which is exposed on a shallowly south-sloping bench, $\sim 250 \mathrm{~m}$ wide (Fig. 3A). The lower sill is also diabasic and is at least 20 m thick. Its floor is buried in turf, but its roof is well exposed along a southfacing cliff (an ENE-trending post-Paleozoic fault scarp), and as two erosional windows in the sloping limestone bench (Fig. 3A). The lower sill's roof zone does not show a well-developed chill, but it has granular textures, is affected by pervasive deuteric hydrothermal alteration, and is enriched in pyrite and minor chalcopyrite. A prominent NNW-trending diabase dike (which we call the northern feeder dike) has well-
developed chilled margins on both sides and cuts the limestone bench. The dike is $\sim 3 \mathrm{~m}$ wide at the southern end (Fig. 3B) and widens to $\sim 30 \mathrm{~m}$ where it approaches the contact with the upper sill. This widening effect is primarily an alongstrike change, unlike the dikes documented by Gudmundsson et al. (2008). Where it is widest, the feeder dike is composed of medium-grained diabase at its core. Preliminary magnetic susceptibility data suggest a steep flow fabric. Locally, sulfides are concentrated along the hangingwall contact. The upper sill does not show any trace of the northern feeder dike within it, and we now believe that the upper sill is a later intrusion. Dikes with similar orientations, spaced $1-2 \mathrm{~km}$ apart, many plagioclase + clinopyroxene + olivine-phyric, occur along this same limestone bench to the east, indicating that this dike is not an isolated or unique occurrence. A series of postPaleozoic ENE-trending faults cuts the limestone bench, generating small offsets in the northern feeder dike and obscuring the contact zone with the upper sill.

Along the southern cliff exposure, the lower sill merges seamlessly with the dike, which dips $\sim 45^{\circ}$ toward the east (Fig. 3B). Immediately above the junction with the sill on the southern scarp, the dike is only $\sim 3 \mathrm{~m}$ wide and has chilled margins. The limestone above the sill's roof shows textures indicating extensive recrystallization. This probably represents a combination of thermal metamorphism, reaction between invading magma and host carbonates, and fluidmediated metasomatism. Brittle normal offsets of limestone laminae in the hanging wall indicate extension. The limestone on the footwall side of the dike appears to be less metamorphosed, with a brownish halo that is only $\sim 2 \mathrm{~m}$ wide. A few, small, 0.5 -m-wide plagioclase + clinopyroxene + olivine-phyric splay dikes penetrate the footwall. Limestone in contact with the dike locally exhibits drag folds on a $1-3 \mathrm{~m}$ scale (Fig. 3C), indicating that the eastern (hangingwall) block moved down, making this a normal fault. Tiny back bends on bent laminae suggest either fluctuations to the overall motion, or compression of hot, ductile limestone related to elevated magmatic pressure.

## Southern Feeder Dike Complex

In this area, the bedded limestone of the Upper Wynniatt Formation ( $\sim 1-2 \mathrm{~km}$ paleodepth) and evaporites of the Kilian Formation are intruded by large bodies of diabase and gabbro. On the kilometer scale, the outcrop pattern is characterized by a series of NNWtrending gabbroic and limestone panels, separated by prominent linear valleys or scarps that we interpret as faults and/or intrusive contacts.

The area corresponding to the southern feeder dike shows a prominent NNW-trending positive magnetic anomaly similar to the one shown on Figure 2A, suggesting that this may be a magmatic up-flow zone. The action of the faults on the wall rocks and metasedimentary septa is pronounced. A large intrusive apophysis shows a shallowly dipping limestone bench (Fig. 4A) that bends abruptly down into the fault-feeder zone. Limestone of the roof is broken up into large (e.g., 2 m thick $\times 25 \mathrm{~m}$ long) slabs of layered marble that dip $60^{\circ}-70^{\circ}$ to the east (Fig. 4B) near the fault zone. Both the steeply dipping and flat-lying limestones along this structure are intensely recrystallized to white marble over many meters. Pipe-like magmatic structures drill up into thermally softened limestone/ marble and are surrounded by prominent calcsilicate coronas. The steeply dipping metasedimentary block is sheathed in diabase, some of which shows chilled margins.

## Uhuk Massif

A major feeder system is superbly exposed and preserved on the eastern edge of the Uhuk Massif (Fig. 2). The eastern part of the Uhuk Massif is located above a prominent positive NNW-trending magnetic lineament (Fig. 2B) that also corresponds to a major regional-scale lithostratigraphic break, suggesting that the rocks to the east have dropped along a normal fault. The common occurrence of dikes with this orientation (see previous and following sections), commonly associated with breccias and other evidence for faulting, implies that faults with this orientation are synmagmatic faults of Franklin age. The sill-like western part of the Uhuk Massif has a thick olivine-cumulate base and extends toward the west for $\sim 10 \mathrm{~km}$ (Figs. 2 A and 6 A ). We have yet to unambiguously correlate a sill from one fault block to another, and so we cannot be certain how far this sill actually extends. Except for its eastern end (described later herein), the sill appears to have a planar, conformable lower contact against bedded limestones and dolostones of the Jago Bay Formation ( $\sim 2-3 \mathrm{~km}$ paleodepth). Unlike footwall contacts of most diabasic sills, the carbonate rocks that underlie the distal flat-lying parts of the Uhuk sill show a wide contactmetamorphic halo.

Near-vertical, E-W-trending, post-Paleozoic faults break the massif into four main blocks, exposing cross sections, preserving different levels of emplacement, and facilitating interpretation of geological relationships (Figs. 4 and 6A). The post-Paleozoic faults cut and offset N - to NNW-trending faults that are parallel to the prominent magnetic lineament of Figure 2B,


Figure 3. Northern and southern feeder dikes. (A) General view from the air looking northeast of the northern feeder dike (extending from top left to lower right). The dike is over 30 m wide at its northern end, where it nearly joins the upper sill. The dotted white line shows the upper contact of the lower sill. The "B" and "C" arrows show the viewpoints for Figures 4B and 4C. Note the small dark patches labeled "sill windows" where the upper contact of the lowersill peeps through the thinlimestone cap.(B) View looking northwest, nearly along the strike of the feeder dike/fault. Theirregular surface labeled "dike" on the left is a plating of dike on the cliff surface. The dotted line in the foreground is the trace of a post-Paleozoic normal fault (foreground down). (C) View looking south at the upper contact of the northern feeder dike against intensely deformed and metamorphosed limestone. Note that the disrupted bedding dips steeply to the left (east), suggesting that the hanging wall moved down (normal fault). (D) Southern feeder dike. View looking S-SW. In the background, forming the ridge, there is a large dike of diabase and gabbro (Gab). The dotted line shows the approximate trace of the inferred contact. The white lines next to the limestone(Lst) outcrops show the bedding dip. This varies from nearly flat-lying on the right to steeply dipping to the left. The steeply dipping limestone is bounded by fine-grained diabase at its base and beyond. We interpret the outcrop pattern and change in dip as the result of movement along a Proterozoic synmagmatic fault (down to left). The arrow shows the geologist for scale.
and that are associated with cataclastic breccias into which dikes are emplaced (Fig. 7B). These are unambiguously synmagmatic normal faults (down to the east). A third, less-well-developed fault population trends NNE. The NNE-trending faults appear to cut many of the NNW-trending faults, but not all of them, and so their age remains uncertain. We suspect that they may also be Proterozoic, but we cannot be sure. To facili-
tate the presentation, major panels separated by intrusive contacts or Neoproterozoic faults are referred to as domains, with igneous rocks having a number code and metasedimentary rocks having a letter code.
The Southern Block (Figs. 4 and 6A) is a deeply foundered graben where the Uhuk Massif (if present) is hidden below Shaler Supergroup rocks and unconformably over-
lying Cambrian clastic and carbonate rocks. The Uhuk Main Block is $\sim 1.5 \mathrm{~km}$ wide ( $\mathrm{N}-\mathrm{S}$ direction). It preserves the main conformable part of the sill to the west and downdropped panels of sill, roof limestone, and contact zone rocks to the east. The next block to the north is the $500-\mathrm{m}$-wide Horst Block, which exposes the same relationships as the Main Block, but at a different level. The Northern Block is a


Figure 4. Map of the eastern termination of the Uhuk Massif. The geology is shown as a color wash over a rectified air photo base. Contour lines and lakes are also shown. Letters and numbers are domains referred to in the text. Black lines are faults. Interpreted motion on normal faults is shown by the stick-ball symbol (ball-side down). Skarns are shown as bold orange lines. The location of Figure 5 and viewpoint of Figure 6A are shown.

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Figure 5. Detail map of the sill/limestone upper contact zone, showing two incipient limestone roof pendants and the scalloped shape of the sill/limestone contact. Viewpoints of Figures 6B and 6C are shown as arrows. Center of map is at ca. $116^{\circ} 19 . \mathbf{3}^{\prime} \mathrm{W}, 71^{\circ} 39.15^{\prime} \mathrm{N}$.

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Figure 6. (A) General view from the air of the Uhuk Massif, looking west (see Fig. 4 for viewpoint). Some of the domains are marked. Dashed white lines highlight sill contacts. Black dashed lines highlight post-Paleozoic faults. Thick orange lines show the locations of skarns. (B) View looking southeast. Photo was taken from the small lake between the 2 and the $A$ on Figure 6A; and the viewpoint is marked on Figure 5 . The pale gray in the foreground is a partly detached marble roof pendant, dissected by dikes. The black lenses to the left of, and just beyond, the small lake are the intrusive upper contact of the sill (detail in Fig. 6C). The helicopters in the background are on a large limestone roof panel (domain A). (C) Detail of one of the steeply dipping, scalloped upper intrusive contacts from Figure 2B; viewpoint shown on Figure 5. Hammer is 1 m long. (D) View looking ENE at the sill upper contact (ridge in foreground of Figure 6A, domain 4-C contact; photo taken from just below the 3 ). The cliff face extends $\sim 200 \mathrm{~m}$ from left to right ( $\mathbf{N}$ to S ). Note how the irregular contact comes to a point and breaches the limestone surface.
$1.5-\mathrm{km}$-wide graben that preserves the uppermost facies of the sill to the west and a downdropped panel of roof limestone and contact zone rocks to the east.

## Main Block

In the western part of the Uhuk Massif, the sill is conformable (domain 1, Figs. 4 and 5A), and it is divisible into a lower border zone with upwardly increasing proportions of olivine, a peridotite zone dominated by cumulate olivine
(commonly with a poikilitic top), a lower, medium-grained gabbroic zone, and an upper gabbroic zone characterized by more acicular pyroxene and coarser patches. The upper chilled margin is not exposed. Metasedimentary rocks beneath (e.g., domain $H$ ) show a prominent con-tact-metamorphic halo.

In the next sill domain (domain 2) to the east, the intrusive upper contact of the sill against roof limestone (domain A) dips steeply toward the east. The contact is irregular, showing a scalloped pattern that may represent "conchoi-
dal" fracture of massive, annealed roof rocks (Figs. 5, 6B, and 6C). This process was fossilized in action as arcuate dikes that partly detach slivers of roof-contact limestone, each $\sim 5-10 \mathrm{~m}$ wide (Figs. 5 and 6B). The roof limestone is cut by many thin ( 1 cm to 3 m ), NNW-trending dikes that locally flatten out parallel to bedding and branch toward the west (Fig. 7A), suggesting westward fracture propagation and emplacement. Domain A limestone is most intensely metamorphosed where it abuts sill rocks. The upper sill contact is exposed along the NE-fac-


Figure 7. (A) Branching dike emplaced within limestone cap rock of domain $A$. The branching suggests emplacement toward the northwest. Camera case is $\mathbf{\sim 1 2} \mathbf{~ c m}$ long. (B) Close-up of fault breccia injected by microdike, indicating the synkinematic nature of the magmatism. This is from the roof limestone of the ridge shown in Figure 6D (domain C).
ing fault scarp, where it is markedly lower than the upper contact on the western side. Since there does not seem to be a large fault within domain A, this implies an overall dip of $\sim 25^{\circ}$ to the upper intrusive contact. The upper sill contact on the NE-facing scarp is characterized by the development of contact-metamorphic calcsilicate assemblages in the silty dolostone beds (Fig. 8), with more massive marble in the purer carbonates. Garnet porphyroblasts commonly have vesuvianite overgrowths (Fig. 8C). The calc-silicate rocks are replaced by oxide-rich ( $80 \%$ magnetite + minor pyrite and chalcopyrite) skarns (Fig. 8A). The oxide-rich assemblages preferentially extend along-strike in silty layers, and magnetite coronas are developed on garnet porphyroblasts (Fig. 8D). Individual oxide-skarn pods extend along-strike for up to 30 m , can be up to 3-4 m thick, and are developed discontinuously all along this upper silllimestone contact (Fig. 4).
The next domain to the SE (domain 3) is dominated by olivine-rich gabbro near the base and gabbro and diabase near the top, but contact


Figure 8. (A) View of contact-metamorphosed bedded limestone with silty layers of the Jago Bay Formation (domain A looking west). The silty layers contain prominent garnet and vesuvianite porphyroblasts (Figs. 8B and 8C). The black patch just above the sledgehammer ( 1 m long) is massive Fe -oxide skarn. The oxide-rich material preferentially seeps laterally along silty layers and forms coronas on garnet (Fig. 8D). Photo was taken from the skarn immediately to the right of the box showing the location of Figure 5 on the map in Figure 4. (B) Close-up of garnet porphyroblasts in a silty layer. (C) Close-up of vesuvianite coronas on garnet porphyroblasts. Hammer head for scale. (D) Close-up of magnetite coronas on garnet porphyroblasts; this image is located close to the massive skarn zone.
relationships are not preserved. These gabbros represent either a downdropped block or a discordant magma conduit.
The next domain to the NE (domain B) is dominated by flat-bedded dolostone and limestone with silty interbeds. It is at the same altitude as gabbros of domain 3, and it is interpreted to represent a downdropped panel of roof material. Thin dikes are common along the southwestern trailing edges of domain B, and their morphology closely resembles that of the dikes on the SW edge of domain A.

A prominent ridge to the east (domains 4 and C) exposes the sill upper contact on both sides (foreground of Fig. 6A; Fig. 6D). The roof carbonates are brecciated and injected by dikes (Fig. 7B), suggesting that magma exploited mechanical weaknesses. Host rocks above sill contacts are intensely metamorphosed and replaced by calc-silicate and oxide + sulfide-rich skarns over tens of meters. Dike contacts are also decorated by zoned garnet-vesuvianite porphyroblasts. The sill contact breaches the surface of the limestone on the SW face (Fig. 6D),
but it is at least 20 m lower on the east face of this ridge (Fig. 6A), implying westward-upward transgression of the intrusive contact.

## Horst Block

In the high-standing Horst Block (Figs. 4 and 6A), the normal sill stratigraphy is well developed in the west (domain 5), with a concordant lower contact against carbonate rocks of the Jago Bay Formation (domain E). Metamorphosed, impure limestones to the east (domain F) appear to be downdropped along the same NNW-trending contact that separates domains A and 2 in the Main Block. The SW face of this ridge is characterized by lobate, scalloped, steeply eastdipping intrusive contacts between underlying sill (domain 6) and contact-metamorphosed dolostone. A narrow band of calc-silicate rocks with relict gabbroic textures separates limestone from fresher diabase to the east (domain 7). Meter-scale pods of oxide-rich skarn are developed in laminated carbonate rocks a few meters above the calc-silicate rocks. Domain 7 is com-

Figure 9. (A) Ductile fold in contact zone of limestone (domain G). The dominant bedding orientation is parallel to the pen and is nearly vertical here. An isoclinal closure is highlighted by the silty layers in the middle of the photo, implying ductile behavior. The sill contact is $\boldsymbol{\sim 1} \mathbf{m}$ to the left of the image. (B) Intrusive breccia at upper contact of sill, located close to photo in A .

posed of medium-grained gabbroic rocks, and neither the uppermost gabbro nor the olivine cumulate zone is exposed.

## Northern Block

The rocks here are almost identical to the contact between domains 2 and A of the Main Block, and domains 6 and F of the Horst Block, and this is probably the northward extension of the same contact exposed at a different structural level. Rocks on the SW side of the contact (domain 8) are upper zone gabbroic rocks. The subhorizontal upper chilled margin against flat-bedded limestone (top left corner of Fig. 4) is preserved. The rocks on the NE side of the fault are mostly flat-bedded limestones (domain G). Near the domain 8-G contact, limestone is studded with dikelets, magmatic breccias (Fig. 9B), and small oxide-rich skarn zones. In many places, the contact zone is characterized by dikes and partly detached slivers of limestone up to 20 m long. The limestone nearest to the sill contact is intensely metamorphosed and appears to have been ductile, as attested by the presence of vertically dipping bedding and minor drag folds (Fig. 9A). This suggests that the limestone near the contact was sheared synchronously with magmatic emplacement.

## DISCUSSION AND INTERPRETATION

The geological relationships described here suggest that in this part of Victoria Island, melt moved up-section along east-dipping faults,
many of which correspond to prominent regional magnetic anomalies (Fig. 2B), suggesting that the melt-filled conduits may extend to some depth. It is not known if these are deep feeder dikes extending to the mantle, or if they simply represent anomalous thicknesses of basalt corresponding to up-section jogs. The dominant recorded motions and lithological offsets on many of these NNW-trending faults suggest they were east-dipping normal faults ( $\sim 45^{\circ}-$ $70^{\circ}$ ), and the field evidence implies that the faults guided ascent of magma. It could instead be argued that the faulting is a consequence of sill emplacement (e.g., Pollard and Johnson, 1973; Thomson, 2007). However, theoretical arguments and experimental data (Rubin, 1995; Maaløe, 1998; Galland et al., 2008, 2009) imply that the propagation front of a dike or sill should be curved, and so structures resulting from sill propagation and inflation should also be curved. The straightness and lateral extent (up to 50 km ) of these NNW-trending lineaments (Kiss and Oneschuk, 2010), together with the fact that all are parallel, suggest to us that these faults were not caused by sill inflation during horizontal propagation but were related to regional-scale tectonic extension and began to develop prior to the emplacement of the Franklin sills.

The most common emplacement pattern for Franklin sills is lateral propagation of magma parallel to bedding in the host rock. We can trace individual Franklin sills for tens of kilometers, with no obvious changes in thickness or stratigraphic level. Our field data, as discussed
already, imply that sill propagation proceeded until a favorably dipping fault was encountered (Fig. 10A). Thereafter, lateral migration of melt halted because it was probably easier for melt to lift the roof and ascend the fault plane through this opening conduit. It seems inescapable that during inflation, the Franklin sills must either have lifted their roof strata, or depressed the floor strata, or both (Petford et al., 2000; Acocella and Rossetti, 2002). Seismic data (Hansen and Cartwright, 2006) and experiments (Acocella et al., 2004; Galland et al., 2008, 2009) suggest that in the absence of a preexisting structural control, the end points of roof uplift zones will be accommodated by reverse faults into which magmas could intrude (e.g., Fig. 10A). As pointed out already, the uniform and regionally extensive linearity of the NNW-trending normal faults on Victoria Island suggests to us that these faults developed prior to sill emplacement. If a propagating sill intercepts a fault that dips toward it, then sill inflation should lift the roof clear of the footwall (Fig. 10B). Since there is no fault-parallel slip, there would be no shear along the fault to generate observable structures recording reverse motion. With this type of geometry (Fig. 10B), sill inflation would widen the up-flow conduit, facilitating melt ascent to a higher stratigraphic level.

Field observations suggest the contact-metamorphic aureoles in the roof zone near the junctions between sills and dikes, and beneath the Uhuk Massif's westward extension, are wider than those that surround most of the dia-

Figure 10. Cartoon showing possible seenario for ephemeral fault-mediated melt transfer, slurry emplacement, and possible metallogenetic consequences. (A) As magma spreads laterally parallel to bedding (open arrows), and the roof is raised ( $\mathbf{u}$ ), a compressional stress field is developed at the propagator tip (r). This represents a possible mechanism to develop reverse faults along which melt could jump up-section (Menand et al., 2008). Previously entrained or newly formed crystals (open boxes, mostly olivine + minor chromite) and immiscible sulfide (black circles) are deposited, forming a cumulate layer. The thermal aureole (h) is thin here. Eventually, the sill encounters a preexisting normal fault (f). (B) Magma preferentially ascends the fault, rather than continuing onward, although minor splays may develop in the footwall (p). The melt moves up the fault until a favorable horimon is reached $(z)$, at which juncture a new sill begins to form. The newly nucleated sill preferentially transgresses in the footwall of the fault ( $\mathbf{z}$ ), because magmatic overpressure makes the hanging wall unfavorable (c). A wider metamorphic aureole is developed in the corner region (m). Because the transgressive conduit is narrow, cumulus crystals are unlikely to be entrained. (C) A new pulse of magma arrives (dark-gray arrow), remobilizing previously deposited, unconsolidated cumulates and sulfides. The new pulse increases the internal pressure, raising the roof ( $\mathbf{u}$ ). This widens the transgressive conduit, allowing large amounts of magma to ascend and enter the upper sill, and also allows remobilized cumulates and sulfides to be emplaced higher in the conduit system. Sulfides and cumulates would preferentially concentrate at the lower cornee region where flow is disturbed (i) and at the hydraulic jump ( $\mathbf{j}$ ) where the upper sill is fed. Because of the enhanced magmatic flux through the conduit, the metamorphic aureoles in corner regions and around the conduit are widened ( $\mathbf{m}$ ). This is accentuated by dikes ( $\mathbf{y}$ ) in the roof zone that detach blocks (cf. Figs. 5, 6B, 7A, and 9B). Completely detached blocks rotate into subparallelism with the fault plane (a) as in the southern feeder dike (Fig. 3D). (D) When the magmatic overpressure associated with the new magma pulse wanes and decreases to less than the lithostatic load, the roof subsides (s). Roof subsidence would potentially be accommodated by expulsion of magma from the lower to the upper sill, continuing until the roof zone grounds against the footwall of the fault, as seen in the northern feeder dike. Alternatively, if the lower sill has largely crystallized prior to roof subsidence, then small amounts of fluid-enriched residual melt (f) may be expelled, to concentrate in favorable structural or porosity traps to form oxide-sulfide skarns (Uhuk Massif; Fig. 8A).
basic sills in the area. This inference needs to be confirmed with microscopic and other data, and to be compared with thermal models, but if correct, it implies that considerable volumes of magma may have passed through these channels

A


D
$\hat{\Delta}$




## Bédard et al.

Bédard et al., 2007), and there are Nd isotopic data (Hulbert, 1998, unpublished report) suggesting that the olivine-enriched bases of some Franklin sills are not cumulates produced by local "in situ" crystallization, but represent late emplacement of isotopically distinct olivine slurries instead. Emplacement of crystal-charged slurries requires a period of crystallization and crystal concentration in a holding chamber, followed by remobilization of the crystal mush. Remobilization of immiscible Ni -sulfide accumulations is an economically important facet of the same problem. However, there are few constraints on the triggers or controls of slurry mobilization and emplacement. Inflation of a fault-terminated sill, as discussed here, represents a possible mechanism to modulate slurry emplacement (Fig. 10C).

In this model, a set of sills develops and begins to crystallize. If magmas are primitive (like some of the Franklin sills), olivine-rich cumulates would likely develop. Magmatic sulfides may also have formed and accumulated at the base of the largest, most primitive sills. Minute pentlandite-pyrrhotite globules occur in many olivine-rich Franklin sills (Bédard et al., 2009b), indicating that sulfide immiscibility occurred at some stage. We infer that the magma pressure in the Franklin conduit system increased markedly when new batches of magma arrived from the mantle (e.g., Lister and Kerr, 1991; Gerya and Burg, 2007). This could have led to rupture of the roof (e.g., Pollard and Johnson, 1973), or reactivation of preexisting faults, lifting the sill roof and opening the sloping conduit (fault), and allowing penetration of magma to higher levels. In its early stages, the dipping conduit would have been too narrow to allow large volumes of melt to ascend, and only small amounts of crystal-poor melt would have been able to seep out (Fig. 10B). Attempts to evacuate crystalrich melts (or melt enriched in sulfide globules) would probably have clogged the channel, leading to a further buildup of magma pressure. We suggest that it was the sudden widening of ascent channels when the roof lifted in response to increases in magmatic pressure that allowed accumulated slurries to be evacuated in a catastrophic pulse, and that this could potentially also have remobilized early-formed sulfides (Fig. 10C). In terms of Ni-sulfide exploration (Naldrett et al., 1995; Li et al., 2001; Naldrett, 2005), the most favorable locations for deposit formation would be just below the junction with the transfer conduit ("I" in Fig. 10C, an entrapment zone), or where the fast-moving magma entered a more quiescent upper sill (" j " in Fig. 10C, hydraulic jump). Since the wide contactmetamorphic haloes are consistent with considerable magmatic throughflow, accumulations of
sulfide could potentially be exposed to larger volumes of silicate melt as at Noril'sk, thus increasing the amount of melt seen by the deposited sulfides, and boosting platinum group element grades. The prominent magnetic lineaments at Uhuk and the southern feeder dike suggest that the conduit systems may be large, and that such lineaments may be favorable exploration targets.

At some stage, the magmatic pulse would wane, and so the pressure in the conduit system would decrease. When the magmatic pressure decreased to values less than the lithostatic load (1-4 km of metasedimentary rock, $1-2 \mathrm{~km}$ of congealed basaltic sills, and over 1 km of basaltic lava), the roof would have subsided (Fig. 10D) by expelling whatever magma remained in the lower sill. Roof subsidence may also have pressed out evolved, volatile-enriched residual melts enriched in precious and base metals from a near-rigid, crystal framework (filter-pressing). In the Uhuk Massif, the late oxide-sulfide skarns that overprint the contact-metamorphic aureole are developed all along the roof contact zone (Figs. 4 and 8). It seems reasonable to postulate that these skarns formed from late-magmatic or deuteric fluids pressed out from the underlying, partly crystallized, Fe-rich tholeiitic sills. We suggest that the fluids were focused by a sloping upper contact, guided by favorable structures (breccias, faults), and then ponded against a relatively "impermeable" meta-limestone cap. We are testing this hypothesis with geochemical and isotopic data.

As the hanging-wall panel descended, it would tend to close the dipping, dike-like conduit (Fig. 10D). An intermediate stage may be preserved in the northern feeder dike, where the lateral variation in dike thickness may indicate that subsidence of the roof rock was asymmetric, such that the southern edge became "grounded" against the footwall rocks, jamming the dike partly open (Fig. 3A). If the movement had been smoother and more orthogonal, the feeder dike might have been nearly obliterated, obscuring the record of this event. Thus, dikelike feeder systems developed along dipping faults may be ephemeral. They would widen during periods when the magmatic flux peaked and melts were overpressured, and close during waning stages, so as to become geophysically invisible and geologically negligible. Closure is unlikely to be complete, but if outcrop quality is poor, then the thin dikes and chilled margins that might remain may be missed, or be considered insignificant. Furthermore, anastomosing composite fault systems could excise or obliterate relics of a collapsed feeder dike.

This interpretation suggests a possible explanation for the rarity of dikes observed in some
sill-dominated feeder systems. Fluctuations of magmatic pressure should lead to alternating phases of sill inflation and deflation as melt is transferred from lower to upper sills along faultguided conduits. Subsidence of roof rock as magmatic pressure decreases would lead to deflation of the lower sills and close up many faultdike conduits. The ephemeral nature of the steep conduits implies that fault-guided melt transfer zones might also exist elsewhere, but they have not been observed, either because exceptional exposures are required for their recognition, or because closure of the feeder dike-faults may have been virtually complete.

## CONCLUSIONS

The sill-dominated feeder system for the Franklin event is characterized by fault-guided transfer zones that allowed magma to jump up-section to feed higher-level intrusions. The sill-dike junctions appear to be characterized by unusually wide and intense contact-metamorphic haloes, in part due to the development of skarn, possibly coupled with magmatic throughflow. The geometric constraints suggest that the conduits may have opened episodically when new magma pulses arrived from the mantle, generating magmatic overpressure. These events may also have been associated with remobilization of crystal slurries, and may also have played a role in localizing or enriching Ni-sulfide orebodies. In this and other systems, the conduits might be ephemeral, closing when magma pressure wanes. Roof subsidence during the deflation stage may press out evolved metal-rich fluids and play a role in generating mineralization.

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## REFERENCES CITED

Acocella, V., and Rossetti, F., 2002, The role of extensional tectonics at different crustal levels on granite ascent and emplacement: An example from Tuscany (Italy): Tectonophysics, v. 354, p. 71-83, doi:10.1016 /S0040-1951(02)00290-1.
Acocella, V., Funiciello, R., Marotta, E., Orsi, G., and deVita, S., 2004, The role of extensional structures on experimental calderas and resurgence: Journal of Volcanology and Geothermal Research, v. 129, p. 199217, doi:10.1016/S0377-0273(03)00240-3.
Baragar, W.R.A., 1976, The Natkusiak Basalts, Victoria Island, District of Franklin, in Report of Activities: Geological Survey of Canada Paper 76-1A, p. 347-352.
Bédard, J.H., 1991, Cumulate recycling and crustal evolution in the Bay of Islands ophiolite: The Journal of Geology, v. 99, p. 225-249, doi:10.1086/629486.
Bédard, J.H., Francis, D.M., Hynes, A.J., and Nadeau, S., 1984, Fractionation in the feeder system at a Proterozoic rifted margin: Canadian Journal of Earth Sciences, v. 21, p. 489-499, doi:10.1139/e84-052.

Bédard, J.H., Sparks, R.S.J., Renner, R., Cheadle, M.J., and Hallworth, M.A., 1988, Peridotite sills and metasomatic gabbros in the Eastern Layered Series of the Rhum complex: Journal of the Geological Society of London, v. 25, p. 207224.
Bédard, J.H., Marsh, B.D., Hersum, T.G., Naslund, H.R., and Mukasa, S.B., 2007, Large-scale mechanical redistribution of orthopyroxene and plagioclase in the basement sill, Ferrar dolerites, McMurdo Dry Valleys, Antarctica: Petrological, mineral-chemical and field evidence for channelized movement of crystals and melt: Journal of Petrology, v. 48, p. 2289-2326, doi:10.1093/petrology/egm060.
Bédard, J.H., Leclerc, F., Harris, L., and Goulet, N., 2009a, Intra-sill magmatic evolution in the Cummings complex, Abitibi greenstone belt: Tholeiitic to calc-alkaline magmatism recorded in a subvolcanic conduit system: Lithos, v. 111, p. 47-71, doi:10.1016/j.lithos.2009.03.013.

Bédard, J.H., Naslund, H.R., Morgan, D., and Montjoie, R., 2009b, Geochemical and mineral-chemical systematics from a picritic Franklin sill on Victoria Island, N.W.T.; insights on internal fractionation processes [abs.], in Yellowknife Geoscience Forum Technical Program: Northwest Territories Geoscience Office, Canada.
Bryan, S.E., and Ernst, R.E., 2008, Revised definition of large igneous provinces (LIPs): Earth-Science Reviews, v. 86, p. 175-202, doi:10.1016/j.earscirev.2007.08.008.

Chevallier, L., and Woodford, A., 1999, Morph-tectonics and mechanism of emplacement of the dolerite rings and sills of the western Karoo, South Africa: South African Journal of Geology, v. 102, p. 43-54.
Cox, K.G., 1980, A model for flood basalt vulcanism: Journal of Petrology, v. 21, p. 629-650.
Denyszyn, S.W., Davis, D.W., and Halls, H.C., 2009, Paleomagnetism and U-Pb geochronology of the Clarence Head dykes, Arctic Canada: Orthogonal emplacement of mafic dykes in a large igneous province: Canadian Journal of Earth Sciences, v. 46, p. 155-167, doi:10.1139/E09-011.
Dick, H.J.B., Natland, J.H., Alt, J.C., Bach, W., Bideau, D., Gee, J.S., Haggas, S., Hertogen, J.G.H., Hirth, G., Holm, P.M., Ildefonse, B., Iturrino, G.J., John, B.E., Kelley, D.S., Kikawa, E., Kingdon, A., LeRoux, P.J., Maeda, J., Meyer, P.S., Miller, D.J., Naslund, H.R., Niu, Y.L., Robinson, P.T., Snow, J., Stephen, R.A., Trimby, P.W., Worm, H.U., and Yoshinobu, A., 2000, A long in situ section of the lower ocean crust: Results of ODP Leg 176 drilling at the Southwest Indian Ridge: Earth and Planetary Science Letters, v. 179, p. 31-51, doi:10.1016/S0012-821X(00)00102-3.
Dostal, J., and Dupuy, C., 1984, Geochemistry of the North Mountain Basalts (Nova Scotia, Canada): Chemical Geology, v. 45, p. 245-261, doi:10.1016/0009-2541 (84)90040-8.

Dostal, J., Baragar, W.R.A., and Dupuy, C., 1986, Petrogenesis of the Natkusiak continental basalts, Victoria Island, Northwest Territories, Canada: Canadian Journal of Earth Sciences, v. 23, p. 622-632, doi:10.1139/e86-064.
Dupuy, C., Michard, A., Dostal, J., Dautel, D., and Baragar, W.R.A., 1995, Isotope and trace-element geochemistry
of Proterozoic Natkusiak flood basalts from the northwestern Canadian Shield: Chemical Geology, v. 120, p. 15-25, doi:10.1016/0009-2541(94)00128-U.

Durbano, A., Pratt, B.R., Hadlari, T., and Dewing, K., 2010, Sedimentology and stratigraphy of the Lower Clastic Unit of the Cambrian, Northwest Victoria Island, in 38th Annual Yellowknife Geoscience Forum, Abstracts Volume: Northwest Territories Geoscience Office, p. 73-74.

Elliot, D.H., and Fleming, T.H., 2004, Occurrence and dispersal of magmas in the Jurassic Ferrar large igneous province, Antarctica: Gondwana Research, v. 7, p. 223237, doi:10.1016/S1342-937X(05)70322-1.
Ernst, R.E., Buchan, K.L., and Palmer, H.C., 1995, Giant radiating dyke swarms: Characteristics, distribution and geotectonic applications, in Baer, G., and Heimann, A., eds., Physics and Chemistry of Dykes: Rotterdam, Netherlands, Balkema, p. 3-21.
Fahrig, W.F., 1987, The tectonic settings of continental mafic dyke swarms: Failed arm and early passive margin, in Halls, H.C., and Fahrig, W.F., eds., Mafic Dyke Swarms: Geological Association of Canada Special Paper 34, p. 331-348.
Fialko, Y., 2001, On origin of near-axis volcanism and faulting at fast spreading mid-ocean ridges: Earth and Planetary Science Letters, v. 190, p. 31-39, doi:10.1016 /S0012-821X(01)00376-4.
Francis, E.H., and Walker, B.H., 1986, Emplacement of alkali-dolerite sills relative to extrusive volcanism and sedimentary basins in the Carboniferous of Fife, Scotland: Transactions of the Royal Society of Edinburgh, v. 77, p. 309-323.

Galerne, C.Y., Neumann, E.R., and Planke, S., 2008, Emplacement mechanisms of sill complexes: Information from the geochemical architecture of the Golden Valley Sill Complex, South Africa: Journal of Volcanology and Geothermal Research, v. 177, p. 425-440, doi:10.1016/j.jvolgeores.2008.06.004.
Galland, O., Cobbold, P.R., Hallot, E., and d'Ars, J.D., 2008, Magma-controlled tectonics in compressional settings: Insights from geological examples and experimental modelling: Bollettino della Societa Geologica Italiana, v. 127, p. 205-208.

Galland, O., Planke, S., Neumann, E.R., and MaltheSørenssen, A., 2009, Experimental modelling of shallow magma emplacement: Application to saucer-shaped intrusions: Earth and Planetary Science Letters, v. 277, p. 373-383, doi:10.1016/j.epsl.2008.11.003.

Gerya, T.V., and Burg, J.P., 2007, Intrusion of ultramafic magmatic bodies into the continental crust: Numerical simulation: Physics of the Earth and Planetary Interiors, v. 160, p. 124-142, doi:10.1016/j.pepi.2006.10.004.

Gibb, F.G.F., and Henderson, C.M.B., 2006, Chemistry of the Shiant Isles Main Sill, NW Scotland, and wider implications for the petrogenesis of mafic sills: Journal of Petrology, v. 47, p. 191-230, doi:10.1093/petrology legi072.
Gudmundsson, A., 2011, Deflection of dykes into sills at discontinuities and magma-chamber formation: Tectonophysics, v. 500, p. 50-64, doi:10.1016/j.tecto. 2009 .10.015.
Gudmundsson, A., and Brenner, S.L., 2001, How hydrofractures become arrested: Terra Nova, v. 13, p. 456-462, doi:10.1046/j.1365-3121.2001.00380.x.
Gudmundsson, A., Friese, N., Galindo, I., and Philipp, S.L., 2008, Dike-induced reverse faulting in a graben: Geology, v. 36, p. 123-126, doi:10.1130/G24185A.1
Hadlari, T., and Bédard, J.H., 2008, Reconnaissance fieldwork on Cambrian siliciclastic rocks, Victoria island, NT, in Yellowknife Geoscience Forum, Abstract Volume: Northwest Territories Geoscience Office.
Hansen, D.M., and Cartwright, J., 2006, Saucer-shaped sill with lobate morphology revealed by 3D seismic data: Implications for resolving a shallow-level sill emplacement mechanism: Journal of the Geological Society of London, v. 163, p. 509-523, doi:10.1144 /0016-764905-073.
Hawkesworth, C.J., Lightfoot, P.C., Fedorenko, V.A., Blake, S., Naldrett, A.J., Doherty, W., and Gorbachev, N.S., 1995, Magma differentiation and mineralisation in the Siberian continental flood basalts: Lithos, v. 34, p. 61-88.

Heaman, L.M., LeCheminant, A.N., and Rainbird, R.H., 1992, Nature and timing of Franklin igneous events, Canada: Implications for a Late Proterozoic mantle plume and the break-up of Laurentia: Earth and Planetary Science Letters, v. 109, p. 117-131, doi:10.1016 /0012-821X(92)90078-A.
Holness, M.B., and Humphreys, M.C.S., 2003, The Traigh Bhŕan na Sgůrra Sill, Isle of Mull: Flow localization in a major magma conduit: Journal of Petrology, v. 44, p. 1961-1976, doi:10.1093/petrology/egg066.

Hooper, P.R., and Hawkesworth, C.J., 1993, Isotopic and geochemical constraints on the origin and evolution of the Columbia River basalt: Journal of Petrology, v. 34, p. 1203-1246.

Hulbert, L., 1998, Mineralogical and Chemostratigraphic Investigation of Ultramafic-Bearing Sills on Ground Held by Aber Resources in the Wynniatt Bay-Glenelg Bay-Kilian Lake Area, Victoria Island: Northwest Territories Geoscience Office, 41 p .
Hulbert, L.J., Rainbird, R.W., Jefferson, C.W., and Friske, P., 2005, Map of mafic and ultramafic bodies related to the Franklin magmatic event, Minto Inlier, Victoria Island: Geological Survey of Canada Open-File Map 4928 , map sheet scale $1: 1,000,000+$ CD-ROM.
Huppert, H.E., and Sparks, R.S.J., 1988, The generation of granitic magmas by intrusion of basalt into continental crust: Journal of Petrology, v. 29, p. 599-624.
Husch, J.M., 1990, Palisades sill: Origin of the olivine zone by separate magmatic injection rather than gravity settling: Geology, v. 18, p. 699-702, doi:10.1130 /0091-7613(1990)018<0699:PSOOTO>2.3.CO;2.
Hutton, D.H.W., and Ingram, G.M., 1992, The Great Tonalite Sill of southeastern Alaska and British ColumbiaEmplacement into an active contractional high angle reverse shear zone: Transactions of the Royal Society of Edinburgh-Earth Sciences, v. 83, p. 383-386.
Jefferson, C.W., 1985, Uppermost Shaler Group and its contact with the Natkusiak Basalts, Victoria Island, District of Franklin, in Current Research Part A: Geological Survey of Canada Paper 85-1A, p. 103-110.
Jefferson, C.W., Nelson, W.E., Kirkham, R.V., Reedman, J.H., and Scoates, R.F.J., 1985, Geology and copper occurrences of the Natkusiak Basalts, Victoria Island, District of Franklin, in Current Research Part A: Geological Survey of Canada Paper 85-1A, p. 203-214.
Jefferson, C.W., Hulbert, L.J., Rainbird, R.H., Hall, G.E.M., Gregoire, D.C., and Grinenko, L.I., 1994, Mineral Resource Assessment of the Neoproterozoic Franklin Igneous Events of Arctic Canada: Comparison with the Permo-Triassic Noril'sk-Talnakh Ni-Cu-PGE Deposits of Russia: Geological Survey of Canada Open-File Report 2789, 48 p .
Jourdan, F., Bertrand, H., Schärer, U., Blichert-Toft, J., Féraud, G., and Kampunzu, A.B., 2007, Major and trace element and $\mathrm{Sr}, \mathrm{Nd}, \mathrm{Hf}$ and Pb isotope compositions of the Karoo large igneous province, BotswanaZimbabwe: Lithosphere vs. mantle plume contribution: Journal of Petrology, v. 48, p. 1043-1077, doi:10.1093 /petrology/egm010.
Kerr, A.C., Tarney, J., Nivia, A., Marriner, G.F., and Saunders, A.D., 1998, The internal structure of oceanic plateaus: Inferences from obducted Cretaceous terranes in western Colombia and the Caribbean: Tectonophysics, v. 292, p. 173-188, doi:10.1016/S0040-1951 (98)00067-5.

Kiss, F., and Oneschuk, D., 2010, First Vertical Derivative of the Magnetic Field, Minto Inlier Aeromagnetic Survey, Victoria Island, NTS 87 G/SE and Parts of 87 G/NW, 88 B/SE and 88 B/SW, Northwest Territories: Geological Survey of Canada Open-File Map 6705, scale 1:100,000.
Larsen, L.M., and Pedersen, A.K., 2000, Processes in highMg , high- $T$ magmas: Evidence from olivine, chromite and glass in Palaeogene picrites from West Greenland: Journal of Petrology, v. 41, p. 1071-1098, doi:10.1093 /petrology/41.7.1071.
Larsen, L.M., and Pedersen, A.K., 2009, Petrology of the Paleocene picrites and flood basalts on Disko and Nuussuaq, West Greenland: Journal of Petrology, v. 50, p. 1667-1711, doi:10.1093/petrology/egp048.

Li, C., Naldrett, A.J., and Ripley, E.M., 2001, Critical factors for the formation of a nickel-copper deposit in an
evolved magma system: Lessons from a comparison of the Pants Lake and Voisey's Bay sulfide occurrences in Labrador, Canada: Mineralium Deposita, v. 36, p. 8592, doi:10.1007/s001260050288.
Lissenberg, C.J., Bédard, J.H., and van Staal, C.R., 2004, The structure and geochemistry of the gabbro zone of the Annieopsquotch ophiolite, Newfoundland: Implications for lower crustal accretion at spreading ridges: Earth and Planetary Science Letters, v. 229, p. 105123, doi:10.1016/j.eps1.2004.10.029.
Lister, J.R., and Kerr, R.C., 1991, Fluid-mechanical models of crack propagation and their application to magma transport in dykes: Journal of Geophysical Research, v. 96, p. 10,049-10,077, doi:10.1029/91JB00600.

Maaløe, S., 1998, Shape of ascending feeder dikes, and ascent modes of magma: Journal of Volcanology and Geothermal Research, v. 81, p. 207-214, doi:10.1016 /S0377-0273(98)00008-0.
Malthe-Sørenssen, A., Planke, S., Svensen, H., and Jamveit, B., 2004, Formation of saucer-shaped sills, in Breitkreuz, C., and Petford, N., eds., Physical Geology of HighLevel Magmatic Systems: Geological Society of London Special Publication 234, p. 215-277.
Marsh, B.A., 2004, Magmatic mush column Rosetta Stone: The McMurdo Dry Valleys of Antarctica: Eos (Transactions, American Geophysical Union), v. 85, no. 47, p. 497, doi:10.1029/2004EO470001

May, P.R., 1971, Pattern of Triassic-Jurassic diabase dikes around the North Atlantic in the context of pre-drift positions of the continents: Geological Society of America Bulletin, v. 82, p. 1285-1292, doi:10.1130 /0016-7606(1971)82[1285:POTDDA]2.0.CO;2.
McHone, J.G., Anderson, D.L., Beutel, E.K., and Fialko, Y.A., 2005, Giant dikes, rifts, flood basalts, and plate tectonics: A contention of mantle models, in Foulger, G.R., Natland, J.H., and Anderson, D.L., eds., Plates, Plumes and Paradigms: Geological Society of America Special Paper 388, p. 401-420.
Menand, T., 2008, The mechanics and dynamics of sills in layered elastic rocks and their implications for the growth of laccoliths and other igneous complexes: Earth and Planetary Science Letters, v. 267, p. 93-99, doi:10.1016/j.epsl.2007.11.043.
Naldrett, A.J., 2005, A history of our understanding of magmatic Ni-Cu sulfide deposits: Canadian Mineralogist, v. 43, p. 2069-2098, doi:10.2113/gscanmin.43.6.2069.

Naldrett, A.J., Fedorenko, V.A., Lightfoot, P.C., Kunilov, V.I., Gorbachev, N.S., Doherty, W., and Johan, Z., 1995, Ni-Cu-PGE deposits of Noril'sk region, Siberia: Their formation in conduits for flood basalt volcanism: Transactions of the Institution of Mining and Metallurgy, Section B, Applied Earth Science, v. 104, p. B18-B36.
Paterson, S.R., and Tobisch, O.T., 1992, Rates of processes in magmatic arcs-Implications for the timing and
nature of pluton emplacement and wall rock deformation: Journal of Structural Geology, v. 14, p. 291-300, doi:10.1016/0191-8141(92)90087-D.
Pehrsson, S.J., and Buchan, K.L., 1999, Borden dykes of Baffin Island, Northwest Territories: A Franklin U-Pb baddeleyite age and a paleomagnetic reinterpretation: Canadian Journal of Earth Sciences, v. 36, p. 65-73, doi:10.1139/e98-091.
Petford, N., Cruden, A.R., McCaffrey, K.J.W., and Vigneresse, J.L., 2000, Granite magma formation, transport and emplacement in the Earth's crust: Nature, v. 408, p. 669-673, doi:10.1038/35047000.

Pollard, D.D., and Johnson, A.M., 1973, Mechanics of growth of some laccolithic intrusions in the Henry Mountains, Utah: II. Bending and failure of overburden layers and sill formation: Tectonophysics, v. 18, p. 311354, doi:10.1016/0040-1951(73)90051-6.
Pollard, D.D., and Segall, P., 1987, Theoretical displacements and stresses near fractures in rock: With applications to faults, joints, veins, dikes, and solution surfaces, in Atkinson, B.K., ed., Fracture Mechanics of Rock: London, Academic Press, p. 277-349.
Polteau, S., Mazzini, A., Galland, O., Planke, S., and MaltheSørenssen, A., 2008, Saucer-shaped intrusions: Occurrences, emplacement and implications: Earth and Planetary Science Letters, v. 266, p. 195-204, doi:10.1016 /j.epsl.2007.11.015.
Rainbird, R.H., 1993, The sedimentary record of mantle plume uplift preceding eruption of the Neoproterozoic Natkusiak flood basalt: The Journal of Geology, v. 101, p. 305-318, doi:10.1086/648225.

Rainbird, R.H., Hodgson, D.A., Darch, W., and Lustwerk, R., 1994, Bedrock and surficial geology of northeast Minto Inlier, Victoria Island, NTS 76B7: Geological Survey of Canada Open-File Map 2781, scale 1:50,000.
Rainbird, R.H., LeCheminant, A.N., and Lawyer, J.I., 1996, The Duke of York and related Neoproterozoic inliers of southern Victoria Island, District of Franklin, Northwest Territories, in Current Research: Geological Survey of Canada Paper 1996-E, p. 125-134.
Rainbird, R.H., Bédard, J., Dewing, K., Hadlari, T., Kiss, Miles, W. Ootes, L., 2010, Victoria Island GEM Project: Results from 2010 field mapping and thematic studies, in Palmer, E. (compiler), 38th Annual Yellowknife Geoscience Forum, Abstracts Volume, p. 50-51.
Rocchi, S., Mazzotti, A., Marroni, M., Pandolfi, L., Con stantini, P., Giuseppe, B., Biase, D., Federici, F., and Lo, P.G., 2007, Detection of Miocene saucer-shaped sills (offshore Senegal) via integrated interpretation of seismic, magnetic and gravity data: Terra Nova, v. 19, p. 232-239, doi:10.1111/j.1365-3121.2007.00740.x.

Rubin, A.M., 1995, Propagation of magma-filled cracks Annual Review of Earth and Planetary Sciences, v. 23, p. 287-336, doi:10.1146/annurev.ea.23.050195.001443.

Shellnutt, J.G., Dostal, J., and Keppie, J.D., 2004, Petrogenesis of the 723 Ma Coronation sills, Amundsen basin, Arctic Canada: Implications for the break-up of Rodinia: Precambrian Research, v. 129, p. 309-324, doi:10.1016/j.precamres.2003.10.006.
Simkin, T., 1967, Flow differentiation in the picritic sills of North Skye, in Wyllie, P.J., ed., Ultramafic and Related Rocks: Hoboken, New Jersey, John Wiley, p. 64-69.
Sykes, L.R., 1978, Intraplate seismicity, reactivation of preexisting zones of weakness, alkaline magmatism, and other tectonism predating continental fragmentation: Reviews of Geophysics and Space Physics, v. 16, p. 621-688, doi:10.1029/RG016i004p00621.

Thomson, K., 2007, Determining magma flow in sills, dykes and laccoliths and their implications for sill emplacement mechanisms: Bulletin of Volcanology, v. 70, p. 183-201, doi:10.1007/s00445-007-0131-8.

Thomson, K., and Hutton, D., 2004, Geometry and growth of sill complexes: Insights using 3D seismic from the North Rockall Trough: Bulletin of Volcanology, v. 66, p. 364-375, doi:10.1007/s00445-003-0320-z.

Thorsteinsson, R., and Tozer, E.T., 1962, Banks, Victoria and Stefansson Islands, Arctic Archipelago: Geological Survey of Canada Memoir 330, 85 p.
Vigneresse, J.L., Tikoff, B., and Ameglio, L., 1999, Modifi cation of the regional stress field by magma intrusion and formation of tabular granitic plutons: Tectonophysics, v. 302, p. 203-224, doi:10.1016/S0040-1951 (98)00285-6

Watanabe, T., Koyaguchi, T., and Seno, T., 1999, Tectonic stress controls on ascent and emplacement of magmas: Journal of Volcanology and Geothermal Research, v. 91, p. 65-78, doi:10.1016/S0377-0273(99)00054-2.

White, R.S., Smith, L.K., Roberts, A.W., Christie, P.A.F., Kusznir, N.J., and the iSIMM Team, 2008, Lowercrustal intrusion on the North Atlantic continental margin: Nature, v. 452, p. 460-464, doi:10.1038 /nature06687.
Young, G.M., 1981, The Amundson Embayment, Northwest Territories: Relevance to the Upper Proterozoic evolution of North America, in Campbell, F.H.A., ed., Proterozoic Basins of Canada: Geological Survey of Canada Paper 81-10, p. 203-211.

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# Short-duration contact metamorphism of calcareous sedimentary rocks by Neoproterozoic Franklin gabbro sills and dykes on Victoria Island, Canada 

P. I. NABELEK, ${ }^{1}$ J. H. BÉDARD, ${ }^{2}$ M. HRYCIUK ${ }^{3}$ AND B. HAYES ${ }^{4}$<br>${ }^{1}$ Department of Geological Sciences, University of Missouri, Columbia, MO 65211, USA (nabelekp@missouri.edu)<br>${ }^{2}$ Geological Survey of Canada, 490, rue de la Couronne, Quebec City, QC G1K9A9, Canada<br>${ }^{3}$ Department of Earth and Planetary Sciences, McGill University, 3450 University, St. Montreal, QC H3A 2A7, Canada<br>${ }^{4}$ School of Earth and Ocean Sciences, Cardiff University, Cardiff, CF10 3AT Wales, UK


#### Abstract

This contribution addresses contact metamorphism and fluid flow in calcareous rocks of the Neoproterozoic Shaler Supergroup on Victoria Island, Arctic Canada. These processes occurred due to intrusion of gabbroic sills and dykes at $c .720$ Ma during the Franklin magmatic event, which was associated with the break-up of Rodinia. The intrusive sheets (sills and dykes) are a few metres to $\sim 50 \mathrm{~m}$ thick. Metasedimentary rocks were examined in three locations with very good exposures of vertical dykes feeding horizontal sills, the Northern Feeder Dyke (NFD) complex, the Southern Feeder Dyke (SFD) complex and the Uhuk Massif. In the NFD and SFD complexes, protoliths were limestones and dolostones with minor silicates, and at the Uhuk Massif, the protoliths were silty dolostones. At the time of magma emplacement, these locations were at depths of $1-4 \mathrm{~km}$. The widths of contact aureoles are only several decametres wide, commensurate with thicknesses of the dykes and sills. Splays of tremolite mark incipient metamorphism. Highest grade rocks in the NFD and SFD complexes contain the prograde assemblage diopside + phlogopite whereas at Uhuk they contain the assemblage vesuvianite + garnet + diopside. The assemblages are successfully modelled with $T-X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}$ pseudosections that suggest achievement of $\mathrm{CO}_{2}$-rich fluid compositions due to early decarbonation reactions, followed by influx of aqueous fluids after peak metamorphism. Rapid heating of host rocks and short near-peak temperature intervals are demonstrated by the prevalent morphology of diopside as radial splays of acicular crystals that appear to pseudomorph tremolite and by incomplete recrystallization of calcite in marbles. Calcsilicates in the roof of one sill at Uhuk experienced metasomatic influx of Fe that is evidenced by nearly pure andradite rims on grossular garnet. Vesuvianite, which overgrew the grossular portions of garnet, also contains ferric iron. Vesuvianite was partially consumed during retrograde growth of serpentine and andradite. The occurrence of serpentine in high-grade portions of aureoles is consistent with eventual levelling-off of temperatures between 350 and $400^{\circ} \mathrm{C}$, an inference that is supported by modelled conductive heat transfer from the cooling magma sheets. Focused fluid flow near intrusion-wall rock contacts is demonstrated by narrow zones of anomalously low $\delta^{13} \mathrm{C}$ and $\delta^{18} \mathrm{O}$ values of carbonate minerals. Although the up to $5 \%$ decrease of both $\delta^{13} \mathrm{C}$ and $\delta^{18} \mathrm{O}$ values from sedimentary values is much smaller than is typical for calcsilicate aureoles around large plutons, it is greater than what could have been achieved by decarbonation alone. The decrease in $\delta^{13} \mathrm{C}$ is attributed to fluidmediated exchange with organic low- ${ }^{13} \mathrm{C}$ carbon that is dispersed through the unmetamorphosed rocks and the decrease in $\delta^{18} \mathrm{O}$ is attributed to fluid-mediated isotopic exchange with the gabbroic intrusive sheets. This study shows that when gabbroic sills and dykes intrude a sedimentary basin, (i) contact aureoles are likely to be narrow, only on the scale of several decametres; (ii) short high-temperature regimes prevent achievement of equilibrium metamorphic textures; and (iii) $T-X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}$ paths in calcareous contact aureoles are likely to be complex, reflecting a transition from prograde decarbonation reactions to influx of aqueous fluids during cooling.


Key words: contact metamorphism; fluid flow; gabbroic sills; stable isotopes; Victoria Island.

## INTRODUCTION

A consequence of magma intrusion into the upper crust is heat transfer to the surrounding rocks and generation of hydrothermal systems. Most previous
studies of contact metamorphic systems have focused on aureoles that surround large plutons, where hydrothermal systems can operate for tens of thousands of years or much longer, depending on sizes of plutons and durations of magmatic system (Norton \&

Taylor, 1979; Cathles, 1981; Norton, 1984; Fournier, 1989; Cook et al., 1997; Cui et al., 2001, 2002; Nabelek, 2009; Nabelek \& Morgan, 2012). On the other hand, metamorphism by individual, tens of metres thick sheet intrusions (dykes and sills) occurs only on short time scales, with near-peak temperatures lasting only tens to hundreds of years, depending on the distance from an intrusion (Nabelek et al., 2012). On these time scales, disequilibrium metamorphic features, including heterogeneous metamorphic textures, can be expected. However, in a large igneous province where magmatism can last for millions of years, hydrothermal systems surrounding sills and dykes can potentially exist on time scales that are orders of magnitude longer than individual heat pulses from invading sheet intrusions. Therefore, retrograde hydrothermal features can potentially develop and be preserved in metamorphosed rocks.

The purpose of this contribution is to describe shortduration metamorphism that was driven by emplacement of mafic sills and dykes into the sedimentary rocks of the Neoproterozoic Shaler Supergroup, which is exposed within the Minto Inlier in central Victoria Island, Arctic Canada (Fig. 1). The Shaler Supergroup
was deposited into a shallow epeiric sea, referred to as the Amundsen Basin (Rainbird et al., 1996). After lithification, strata were invaded c. 720 Ma ago by gabbro dykes and sills and were overlain by Natkusiak flood basalts during the Franklin magmatic event (Dostal et al., 1986; Heaman et al., 1992). The Franklin magmatic event is widespread across the Arctic region and can be correlated with mafic magmatism in Greenland and Siberia. It is interpreted as being related to the break-up of Rodinia (Heaman et al., 1992). Shallow marine limestones and dolostones, which are interbedded with restricted-basin evaporites in the upper part of the stratigraphic section, dominate the Shaler Supergroup (Rainbird et al., 1994, 1996). Subordinate units include sandstones and organic-rich shales, all of which were invaded by the sills. This contribution focuses on three separate localities within the Minto Inlier where the local magmatic plumbing systems are better understood (Bédard et al., 2012). The localities contain remnant, fault-guided feeder dykes that fed prominent sills. These localities present an opportunity to examine metamorphism caused by invasion of mafic sheet intrusions in a large igneous province.


Fig. 1. Geological map of Victoria Island showing the major units of the Shaler Supergroup and the overlying Palaeozoic rocks (after Hulbert et al., 2005; Bédard et al., 2012). Heavy northwest-trending lines show Proterozoic vertical faults that have guided emplacement of sill-feeding dykes (not shown) into the Shaler Supergroup.

Metamorphism of the calcareous rocks and compositions of attending fluids are evaluated by employing calculated pseudosections and by stable isotope compositions of carbonate minerals. The data provide information on the development of mineral assemblages and the textural equilibration in contact aureoles of the gabbroic intrusive sheets, on the heat flow surrounding the crystallizing melts, and on the duration of related hydrothermal systems.

## GEOLOGICAL SETTING

The Shaler Supergroup developed an open anticlinalsynclinal structure with shallowly plunging axes trending to the east-northeast (Fig. 1). The Supergroup is dominated by shallow-water marine carbonates and siliciclastic rocks with subordinate restricted-basin sulphate evaporites (Rainbird, 1993; Rainbird et al., 1996). In the study area, around Minto Inlet, the oldest unit is the Reynolds Point Group, which is dominated by carbonates, except near the top where it is more silty. Overlying this group is the Minto Inlet Formation, which contains thick, gypsum-dominated evaporites interbedded with carbonates. To the south of and overlying the Minto Inlet Formation are thick exposures of shallow marine carbonate rocks and deeper marine shales of the Wynniatt Formation. Overlying the Wynniatt Formation is the Kilian Formation, which includes interbedded carbonates, siliciclastics and sulphate evaporites. The top of the sedimentary sequence is the quartzarenite-dominated Kuujjua Formation. The sedimentary rocks are overlain conformably to uncomformably by the Natkusiak flood basalts. Cambrian to Devonian platform rocks uncomformably overlay the Natkusiak flood basalts.

The three areas in which contact metamorphism was examined in detail include: (i) the Northern Feeder Dyke (NFD) complex; (ii) the Southern Feeder Dyke (SFD) complex; and (iii) the Uhuk Massif (Fig. 1). Magmatism in all three areas was localized by north-west-trending Proterozoic faults. The NFD complex (Fig. 2a) occurs within the Boot Inlet Formation of the Reynolds Point Group (Rainbird et al., 1994, 1996). The formation is mostly a limestone, but the proportion of clay increases upwards. The complex includes a lower diabase sill ( $>20 \mathrm{~m}$ thick) that terminates to the SW against limestones on the footwall of one of the Proterozoic faults. A subvertical feeder dyke trends from this sill towards a younger, $\sim 50 \mathrm{~m}$ thick sill that caps the mesa at this locality (Bédard et al., 2012). Upslope from the lower sill there are windows with exposures of the apparently quite uneven upper surface of this particular sill. The stratigraphic distance between the two sills is at least 50 m , based on the current elevation difference between the sill windows and the upper sill. Bédard et al. (2012) showed that the exposed fault-guided dyke might have served as a conduit for upflowing magma. Anisotropy of magnetic susceptibility data, indicating a steep palaeodip of the
magnetic lineation, is consistent with this interpretation (MacDonald et al., 2012). The palaeodepth of the NFD complex is estimated to have been 3-4 km based on the stratigraphy. Samples were collected with the aim of determining the 3D structure of metamorphism, extending from the roof of the lower sill at its lower exposure and the sill windows toward the base of the mesa-capping upper sill. Samples were also collected across the feeder dyke, and along several traverses perpendicular to the feeder dyke.

Northwest-trending dykes characterize the SFD complex (Fig. 2). Its palaeodepth during metamorphism is estimated to have been $1-2 \mathrm{~km}$ (Bédard et al., 2012). Metamorphism is discussed here for zones A and D of the complex. Zone D is $\sim 6 \mathrm{~km}$ to the northwest (downsection) of zone A. The dyke-sill system exposed in the intervening zones B and C is less well preserved. Along the trend of these zones is a set of prominent magnetic lineaments that suggests that the feeder dykes represent parts of a fault-related magmatic feeder zone (Bédard et al., 2012). The wall rocks of the feeder dyke(s) in zone A (Fig. 2b) belong to the Kilian Formation and include limestones with variable proportions of detrital silicates and remnants of evaporite beds. The dyke in zone A is $\sim 20 \mathrm{~m}$ wide and dips steeply to the west. To the south, it feeds into a sill that forms a prominent bench. In zone $A$, the shallowly dipping bedding in the host rocks shows a marked downward deflection on the east side and upward deflection on the west side of the dyke, suggesting that these are fault-related drag folds and that dyke ascent was fault guided. Within a metres-wide zone next to the dyke, some limestone beds are brecciated with conchoidal fractures filled by calcite ( $\pm$ pyrite). Samples include a marble from the intersection of the dyke with the capping sill at the highest sampled stratigraphic level, from a lower massive, partially graphitic limestone ledge that is brecciated near the dyke, and from the next lower limestone ledge, which has an evident skarn near the dyke. Calcsilicate samples interbedded with evaporites were collected $\sim 200 \mathrm{~m}$ further to the north (station 63).

Zone SFD-D is within dolostones of the Wynniatt Formation (Fig. 2c). The dolostones were deposited in subtidal and intertidal settings as evidenced by the presence of mudcracks, ooids and stromatolites (Rainbird et al., 1994). The dolostones, particularly in stromatolite horizons, are now fairly graphitic. A feature of the SFD-D zone is an irregular dyke-like mafic body, $\sim 50 \mathrm{~m}$ wide, that trends in the northwesterly direction and that may be related to a larger sill that makes up the mesa to the southwest. On the eastern side of the dyke-like body, the dolostone bedding was rotated up into parallelism with the mafic body and now has an $\sim 45^{\circ}$ dip to the northeast (Bédard et al., 2012). Elsewhere in the SFD-D zone, the beds are near their original, near-horizontal orientation, but their dip increases somewhat towards the western margin of the mafic body, with a geometry that suggests an origin


Fig. 2. Geological maps of (a) the Northern Feeder Dyke complex, (b) zone A of the Southern Feeder Dyke complex, (c) zone D of the Southern Feeder Dyke complex and (d) the eastern edge of the Uhuk Massif (after the more complete map of Bédard et al., 2012). Station symbols denote characteristic minerals or rock types. Shaded symbols show occurrences of serpentine, chlorite or epidote. Contours, where shown, have 20 m interval. UTM coordinates for zone 11 W are shown along margins. Northwest-trending faults are typically Proterozoic, coeval with magma emplacement; northeast-trending faults are typically post-Palaeozoic and dismembered the Proterozoic geology. Locations of exoskarns and skarn veins at the Uhuk Massif are shown.
as drag folds due to downward motion of the northeastern block. Samples were collected from ledges between the mesa-forming sill and the smaller dyke-like mafic body with the rotated sedimentary beds. Multiple samples with different morphologies and/or mineral contents were collected at some locations in the SFD-D zone.

The Uhuk massif contains on its eastern side a series of down-dropped blocks of a large primitive sill and its roof rocks, with normal motion along north-north-west-trending, northeast-dipping faults (Fig. 2d; Bédard et al., 2012). Ascending magma actively intruded and brecciated the roof and then appears to have formed a sill to the west. A prominent northwesttrending magnetic lineament suggests that the Proterozoic faults guided magma ascent (Bédard et al., 2012). Further dismemberment of the massif occurred by post-Palaeozoic faulting. The host rocks to the
intrusions belong to the Jago Bay Formation (uppermost Reynolds Point Group; Rainbird et al., 1994, 1996), which is characterized by yellow-weathering silty dolostone. Exoskarns produced by variable amounts of metasomatism of calcsilicates occur in several places above sill exposures. Samples were collected along two vertical traverses through the calcsilicates above the sill in block III and from the roof of the sill in block I. Station 8 occurs near massive skarns and where small gabbroic dykes have penetrated the overlying calcsilicates.

## METHODS

Mineralogy and mineral chemistry of the metamorphic rocks were determined by petrography, semiquantitative energy-dispersive analysis (EDA) using a Quanta 600 scanning electron microscope (SEM;

University of Missouri, Columbia, MO, USA), and in selected samples were determined by quantitative electron probe microanalysis (EPMA; Washington University, St. Louis, MO, USA). Carbonate minerals were analysed for oxygen and carbon isotope ratios by a dual-inlet Finnigan Delta Plus mass spectrometer with an automated Kiel acid reaction device at the University of Missouri. The carbonate minerals were extracted from billets that were left from thin-section preparation using a fine drill. Values are reported relative to the standard VSMOW for oxygen and the standard VPDB for carbon.

Garnet on Victoria Island is dominantly a grossularandradite mixture, but also has variable concentrations of Ti (Table 1; mineral symbols are from Kretz, 1983). High-Ti garnet has been reported in hornfelses elsewhere (e.g. Labotka, 1995). One substitution mechanism is $2 \mathrm{Ti}^{4+}$ for $2 \mathrm{Si}^{4+}$ in andradite, which leads to the schorlomite (sch) end-member $\left(\mathrm{Ca}_{3} \mathrm{Ti}_{2} \mathrm{Fe}_{2}{ }^{3+}\right.$ $\mathrm{SiO}_{12}$ ), and another mechanism is $\mathrm{Fe}^{2+} \mathrm{Ti}^{4+}$ for $2 \mathrm{Fe}^{3+}$, which leads to the morimotoite (mor) endmember $\left(\mathrm{Ca}_{3} \mathrm{Fe}^{2+} \mathrm{Ti}^{4+} \mathrm{Si}_{3} \mathrm{O}_{12}\right.$; Armbruster et al., 1998). The morimotoite substitution may also involve incorporation of hydrogen into garnet, which, however, we did not characterize in this study. Proportions
of garnet end-members (Table 1) were calculated from EPMA analysis by a least-squares regression. The maximum residual sum of squares for all spot analysis was 0.02 .

## MINERALOGY

## Northern Feeder Dyke complex

Mineralogy of the NFD complex is dominated by calcite that can contain up to $4 \%$ dolomite component. In unmetamorphosed or weakly metamorphosed rocks, detrital quartz occurs in variable proportions, from none to tens of percent. It is generally absent from metamorphosed rocks near the mafic intrusive rocks. In these calcite-dominated rocks, dolomite was not petrographically identified and we did not come across it during EDA or EPMA analysis. It is therefore possible that magnesian calcite is the main Mg-bearing mineral in the unmetamorphosed rocks and was the source of Mg for metamorphic Mg -silicates. Muscovite is the main aluminium-bearing mineral, but its proportion is very small. The calcite ( $\pm$ quartz)-dominated mineralogy in the NFD complex reflects the $\mathrm{CaO}-$ dominated compositions of the rocks (Table 2).

Table 1. Representative analyses of silicate minerals. ${ }^{\text {a }}$

|  | $\begin{gathered} \mathrm{Di} \\ 40-1 \end{gathered}$ | $\begin{gathered} \mathrm{Di} \\ 7-12 \end{gathered}$ | $\begin{gathered} \mathrm{Phl} \\ 40-1 \end{gathered}$ | $\begin{gathered} \text { Srp } \\ 40-1 \end{gathered}$ | $\begin{gathered} \text { Srp } \\ 8-1 \mathrm{~A} \end{gathered}$ | $\begin{aligned} & \text { Epid } \\ & 26-1 \end{aligned}$ | $\begin{gathered} \text { Vsv } \\ 7-12 \end{gathered}$ | $\begin{gathered} \text { Vsv } \\ 8-1 \mathrm{~A} \end{gathered}$ | $\begin{gathered} \text { Grt } \\ 26-1 \end{gathered}$ | $\begin{gathered} \text { Grt } \\ \text { core } 7-12 \end{gathered}$ | $\begin{gathered} \text { Grt rim } \\ 7-12 \end{gathered}$ | $\begin{aligned} & \text { b} \text { Grt spot } \\ & \text { "A" } 8-1 \mathrm{~A} \end{aligned}$ | $\begin{aligned} & { }^{\text {b}} \text { Grt spot } \\ & \text { "B" } 8-1 \mathrm{~A} \end{aligned}$ | ${ }^{\mathrm{b}}$ Grt spot <br> "C" 8-1A | $\begin{aligned} & \text { b} \text { brt spot } \\ & \text { "D" } 8-1 \mathrm{~A} \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| wt.\% |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| $\mathrm{SiO}_{2}$ | 54.42 | 52.93 | 42.11 | 40.48 | 41.18 | 38.64 | 36.33 | 36.88 | 36.52 | 40.18 | 39.37 | 35.78 | 38.07 | 36.27 | 36.32 |
| $\mathrm{TiO}_{2}$ | 0.1 | 0.76 | 0.54 | 0.07 | 0 | 0.03 | 1.01 | 0.68 | 3.56 | 0.27 | 1.69 | 6.25 | 1.7 | 0.13 | 0.09 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 1.2 | 2.54 | 14.7 | 1.86 | 2.94 | 27.72 | 14.95 | 14.55 | 13.81 | 20.55 | 19.22 | 11.55 | 11.03 | 1.76 | 1.62 |
| FeO | 2.71 | 5.44 | 1.24 | 3.07 | 0.44 | 7.62 | 4.6 | 3.86 | 9.82 | 4.35 | 4.02 | 8.96 | 12.32 | 25.51 | 25.45 |
| MnO | 0.08 | 0.11 | 0.04 | 0.08 | 0.1 | 0.03 | 0.09 | 0.05 | 0.16 | 0.2 | 0.14 | 0.1 | 0.04 | 0 | 0.06 |
| MgO | 18.05 | 14.46 | 26.66 | 38.46 | 39.42 | 0 | 3.98 | 4.41 | 0.79 | 0.51 | 0.28 | 1.3 | 0.68 | 0.26 | 0.47 |
| CaO | 23.86 | 24.66 | 0.18 | 0.14 | 0.15 | 23.91 | 35.5 | 36.23 | 34.75 | 35.63 | 36.37 | 35.31 | 35.96 | 33.94 | 34.21 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.07 | 0.18 | 0.39 | 0.03 | 0.06 | 0.02 | 0.1 | 0.05 |  |  |  |  |  |  |  |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.01 | 0.03 | 10.08 | 0.02 | 0.01 | 0.03 | 0.01 | 0 |  |  |  |  |  |  |  |
| F | nd | nd | 1.26 | nd | nd | nd | 0.5 | 0.28 |  |  |  |  |  |  |  |
| Cl | nd | nd | 0.09 | 0.41 | 0.09 | nd | 0.01 | 0.02 |  |  |  |  |  |  |  |
| Sum | 100.51 | 101.11 | 97.3 | 84.63 | 84.39 | 98.01 | 97.07 | 97.02 | 99.42 | 101.68 | 101.08 | 99.25 | 99.81 | 97.88 | 98.22 |
| Atoms/formula |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| Si | 1.97 | 1.93 | 2.92 | 1.86 | 1.86 | 2.99 | 17.7 | 17.87 | 2.86 | 2.99 | 2.97 | 2.82 | 2.98 | 3.01 | 3 |
| Al | 0.05 | 0.11 | 1.2 | 0.1 | 0.16 | 2.53 | 8.59 | 8.31 | 1.27 | 1.81 | 1.71 | 1.07 | 1.02 | 0.17 | 0.16 |
| Ti | 0 | 0.02 | 0.03 | 0 | 0 | 0 | 0.37 | 0.25 | 0.21 | 0.02 | 0.1 | 0.37 | 0.1 | 0.01 | 0.01 |
| $\mathrm{Fe}^{3+}$ | 0 | 0 | 0 | 0 | 0 | 0.43 | 1.38 | 1.56 | 0.59 | 0.18 | 0.17 | 0.54 | 0.81 | 1.77 | 1.76 |
| $\mathrm{Fe}^{2+}$ | 0.08 | 0.17 | 0.07 | 0.12 | 0.02 | 0.07 | 0.49 | 0 | 0.05 | 0.1 | 0.09 | 0.05 | 0 | 0 | 0 |
| Mn | 0 | 0 | 0 | 0 | 0 | 0 | 0.04 | 0.02 | 0.01 | 0.01 | 0.01 | 0.01 | 0 | 0 | 0 |
| Mg | 0.97 | 0.79 | 2.75 | 2.64 | 2.66 | 0 | 2.89 | 3.19 | 0.09 | 0.06 | 0.03 | 0.15 | 0.08 | 0.03 | 0.06 |
| Ca | 0.92 | 0.97 | 0.01 | 0.01 | 0.01 | 1.98 | 18.54 | 18.81 | 2.91 | 2.85 | 2.94 | 2.98 | 3.02 | 3.01 | 3.02 |
| Na | 0 | 0 | 0.05 | 0 | 0.01 | 0 | 0 | 0 |  |  |  |  |  |  |  |
| K | 0 | 0 | 0.89 | 0 | 0 | 0 | 0 | 0 |  |  |  |  |  |  |  |
| F | nd | nd | 0.28 | nd | nd | nd | 0.38 | 0.22 |  |  |  |  |  |  |  |
| Cl | nd | nd | 0.01 | 0.02 | 0 | nd | 0 | 0.01 |  |  |  |  |  |  |  |
| Minera components |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| \% En | 49 | 41 |  |  |  |  |  |  |  |  |  |  |  |  |  |
| \% Fs | 4 | 9 |  |  |  |  |  |  |  |  |  |  |  |  |  |
| \% Wo | 47 | 50 |  |  |  |  |  |  |  |  |  |  |  |  |  |
| \% Prp |  |  |  |  |  |  |  |  | 3 | 3 | 1 | 4 | 2 | 0 | 1 |
| \% Grs |  |  |  |  |  |  |  |  | 61 | 86 | 83 | 53 | 51 | 9 | 8 |
| \% Adr |  |  |  |  |  |  |  |  | 23 | 8 | 7 | 19 | 40 | 89 | 89 |
| \% Mor |  |  |  |  |  |  |  |  | 6 | 3 | 7 | 15 | 6 | 2 | 2 |
| \% Sch |  |  |  |  |  |  |  |  | 7 | 0 | 1 | 11 | 1 | 0 | 0 |

${ }^{\text {a }}$ Normalizations: diopside to 6 oxygen; phlogopite to 11 oxygen; serpetinite to 7 oxygen and 5 cations; epidote to 15 oxygen and 16 cations; garnet to 12 oxygen and 8 cations; vesuvianite to
73.5 oxygen and 50 cations; nd, not detected.
${ }^{\mathrm{b}}$ Spots 'A-D' refer to analysis locations in Fig.4.

Incipient metamorphism is identified by the appearance of radial tremolite (Fig. 3b). Tremolite occurs in rocks as close as 30 m from the feeder dyke and the capping sill and in most samples between the windows to the lower sill (Fig. 2a). Diopside occurs only very close to the gabbroic sheets. In contrast to the typical equant shape of diopside in contact aureoles

Table 2. Representative compositions of rocks used to compute pseudosections.

|  | Northern Feeder Dyke and Southern <br> Feeder Dyke carbonates | Uhuk silty dolostones |
| :--- | :---: | :---: |
| $\mathrm{SiO}_{2}$ | 1.53 | 25.30 |
| $\mathrm{TiO}_{2}$ | 0.02 | 0.25 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 0.44 | 4.49 |
| FeO | 0.37 | 9.50 |
| MnO | 0.01 | 0.06 |
| MgO | 0.91 | 4.19 |
| CaO | 54.63 | 29.04 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.02 | 0.00 |
| $\mathrm{~K}_{2} \mathrm{O}$ | 0.09 | 0.00 |
| $\mathrm{CO}_{2}$ | 43.02 | 24.70 |
| $\mathrm{Sum}^{2}$ | 101.04 | 97.50 |

around large plutons (e.g., Berger et al., 2010; Nabelek \& Morgan, 2012), diopside in these carbonates from Victoria Island typically occurs as radial splays of acicular crystals that resemble the common morphology of tremolite (Fig. 3c). Petrographically it is distinguished from tremolite by its larger extinction angle. Its morphology suggests that it pseudomorphs tremolite and that there was not a sufficient amount of time for it to equilibrate texturally. Phlogopite is common in more aluminous rocks and occurs together with both tremolite and diopside. A feature of many diop-side-bearing rocks is the occurrence of serpentine (Fig. 3d; the specific form of serpentine was not determined).

Only locally, at the intersections of the dyke with the lower sill and near the sill window (stations 26-29 \& 126), is there evidence for metasomatism. Samples from these locations contain variable amounts of garnet $\left(\sim \operatorname{Prp}_{3} \mathrm{Grs}_{61} \mathrm{Adr}_{23} \mathrm{Mor}_{6} \mathrm{Sch}_{7}\right), \mathrm{Fe}^{3+}$-bearing epidote (Table 1), chlorite and titanite, in addition to tremolite or diopside as metamorphic minerals. A sample located $\sim 2 \mathrm{~m}$ above the lower sill (132-1) and one above the sill window (31-1) contain veinlets of


Fig. 3. (a) Variably recrystallized marble at the Southern Feeder Dyke (SFD)-D (field of view (fov) $=2 \mathrm{~mm}$ ); (b) tremolite splay at the SFD-D complex (fov $=2 \mathrm{~mm}$ ); (c) splay of acicular and bladed diopside in the Northern Feeder Dyke (NFD) complex (fov $=0.5 \mathrm{~mm}$ ); (d) serpentine surrounding a recrystallized or newly formed pokioblastic calcite containing phlogopite in the NFD complex (fov $=0.5 \mathrm{~mm}$ ).
anhydrite and the latter sample, unsurprisingly, also contains scapolite.

In high-grade contact aureoles around large plutons, calcite is usually highly recrystallized (annealed), even in rocks that apparently remained closed to external fluids (e.g. Nabelek et al., 1984; Nabelek \& Morgan, 2012). On Victoria Island, the calcite in unmetamorphosed and incipiently metamorphosed rocks is a finegrained micrite $(\sim 5-10 \mu \mathrm{~m})$ that becomes variably coarser (up to $\sim 50 \mu \mathrm{~m}$ ) next to mafic intrusions (Fig. 3a). In skarns and breccia-filling veins, calcite crystals can be on a millimetre scale. Neoformed calcite grains in skarns are typically poikiloblastic, enclosing metamorphic silicate minerals, whereas in calcitedominated rocks, silicate minerals typically occur at calcite triple junctions, with the exception of some tremolite splays that overgrow micritic calcite.

In the NFD complex, the carbonate and silicate minerals are Fe poor (Table 1), except in the local skarns. This indicates that most of the Fe in the rocks is sequestered by pyrite and pyrrhotite. Pyrrhotite occurs in increasing proportions towards the mafic rocks, consistent with increasing temperature (Toulmin \& Barton, 1964).

## Southern Feeder Dyke complex

Metamorphic mineralogy in the SFD complex is similar to that in the NFD complex. In the SFD-A zone, massive limestone ledges below the capping sill are essentially monomineralic with micritic calcite and graphitic material; however, some samples contain detrital quartz. Near the dyke, calcite grain size is variable and triple junctions are more abundant. Marbles are commonly brecciated at contacts with the dyke with fractures filled by coarse calcite veins. Some of the marbles contain interstitial serpentine, showing the presence of Mg and Si in addition to the calcite components. The skarn in the lowest massive limestone ledge next to the dyke contains garnet, epidote and chlorite. At station $63,10 \mathrm{~m}$ from the dyke there is tremolite, calcite and detrital quartz, and 2 m from the dyke, there is diopside enclosed in variably recrystallized calcite ( $\sim 50 \mu \mathrm{~m}$ ) and minor anhydrite. Fifty centimetres from the dyke there is a rare occurrence of wollastonite, due to a large amount of detrital quartz, in addition to diopside and neoformed, recrystallized calcite. Thus, next to the dyke in the SFD-A zone, the high-grade aureole, defined as having diopside, is very narrow.

The dominant outcrops in the SFD-D zone are variably recrystallized marble ledges. Those that have fossil stromatolites are fairly graphitic. Index metamorphic minerals are tremolite and diopside. Tremolite overgrowing a micritic matrix (Fig. 3b) occurs on a ledge approximately half way between the mesacapping sill and the smaller mafic dyke-like body (Fig. 2c). Diopside is confined to close proximities of the mafic bodies, except in the eastern ledge (stations 77-81) where it occurs as much as $\sim 200 \mathrm{~m}$ horizontally
from the current edge of the upper sill, suggesting that a mafic body capped this ledge in the past, or that an unexposed intrusive apophysis may be present beneath.

## Uhuk Massif

The metamorphic mineralogy at the Uhuk Massif is distinguished from that in the NFD and SFD complexes by containing the frequent assemblage of garnet + diopside. Garnet is stable due to elevated Al concentrations in the dolomitic Jago Bay Formation (Table 2). These two minerals plus vesuvianite occur throughout the whole 10 m vertical traverse above the lowermost exposure of the down-dropped sill at station 8 and through at least the first 5 m at station 7 (section III of Fig. 2d). The silty beds are interbedded with variably recrystallized marbles containing conchodial fractures filled by calcite. Diopside is nearly always acicular and together with some small calcite grains, also occurs as inclusions in garnet. In the matrix, calcite typically occurs as large recrystallized grains. Vesuvianite is always poikiloblastic, enclosing all other minerals. This indicates that vesuvianite was the last mineral that grew in the rocks during prograde metamorphism. Andradite and some diopside in the matrix appear to be intimately associated with retrograde serpentine. Mineralogy of calcsilicates in the sill's roof in section I of the Uhuk map (stations 10-17; Fig. 2d) is similar, although vesuvianite was found only in sample 16. Scapolite was found in samples 7-20 and 15-1. Magnetite, pyrite and pyrrhotite occur in various proportions. The sulphides occur as separate grains or as two-phase grains with lamellae of pyrrhotite in pyrite.

## Metasomatism

The compositions of garnet and vesuvianite (Table 1) potentially record variable additions of Fe to the rocks by metasomatism and high oxidation conditions. For example, in sample $7-12$ that came from 1 m above the sill, garnet is fairly homogeneous, dominated by the grossular component, but also contains significant andradite component $\left(\mathrm{Grs}_{86} \mathrm{Adr}_{8} \mathrm{Mor}_{3}-\mathrm{Grs}_{83} \mathrm{Adr}_{7} \mathrm{Mor}_{7}\right)$. Most Fe in vesuvianite in $7-12$ is also apparently ferric. It is possible to grow these minerals from the bulk composition in Table 2 (see below), but there had to be oxidation of Fe in the rocks before or during mineral growth. In several samples at station 8, all Fe in vesuvianite appears to be ferric. The presence of ferric Fe indicates that these minerals grew at a fairly high $f \mathrm{O}_{2}$.

The composition of garnet is particularly interesting in sample 8-1A (Table 1; Fig. 4). Core of a larger garnet is $\mathrm{grs}_{62} \mathrm{adr}_{34}$. Another brighter zone in the larger garnet in the BSE image of Fig. 4 is grs $_{54} \mathrm{adr}_{44}$. This zone is surrounded by vesuvianite, which suggests that there was an influx of some oxidized, Fe-bearing fluid at this stage of metamorphism. The most Fe-enriched fluid


Fig. 4. A back-scatter electron image of a portion of sample $8-1$ A showing an assemblage of garnet, vesuvianite, calcite, diopside and serpentine. The bright rim on the larger garnet and small microgarnet are nearly pure andradite. Note that the andradite rim grew after vesuvianite and the small andradite crystals occur within serpentine-containing fractures in vesuvianite. Spots labelled by letters correspond to microprobe analyses in Table 1.
appears to have arrived after the growth of vesuvianite. It is indicated by a bright grs $_{9} \mathrm{adr}_{89}$ rim on the garnet that occurs only within a patch of serpentine with smallbladed diopside crystals. There are also small, $\sim 10 \mu \mathrm{~m}$ andradite crystals (seen as brightest crystals in the BSE image) within the serpentine and within serpentine-filled cracks inside vesuvianite. The coexistence of these minerals suggests that assemblage andradite + serpentine + diopside was stable at the prevailing oxidizing conditions. Serpentine-filled embayments and cracks within vesuvianite suggest that it was consumed during formation of the three-phase assemblage, with ferric iron from vesuvianite used to stabilize andradite. Fe metasomatism at Uhuk is also clearly evidenced by occurrences of massive magnetite-rich exoskarns at other locations above the sill. At the locations discussed in this article, Fe metasomatism was more subtle but clearly evident in the mineral chemistry.

## PHASE EQUILIBRIA CONSTRAINTS ON METAMORPHIC CONDITIONS

Temperature- $X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}$ pseudosections, showing theoretical mineral assemblages in the calcite-dominated rocks in the NFD and SFD complexes and the calcsilicates at Uhuk, were constructed using the theriak/domino software (De Capitani \& Petrakakis, 2010). An updated version (db55, 2007) of the activity models and thermodynamic database of Holland \& Powell (1998) was used in the calculations. The pseudosections were calculated for 250 bar, appropriate for a hydrostatic pressure regime in the sedimentary rocks. With increasing pressure, the topology of
the diagrams stays essentially intact, but the assemblage boundaries move to higher temperatures because nearly all reactions are volatile producing.

## Northern Feeder Dyke and Southern Feeder Dyke zones

Assemblages in the calcite-dominated rocks (Fig. 5a,b) were modelled using a representative composition (Table 2), but without Fe . A Fe-absent composition is appropriate because almost all Fe in the NFD and SFD complexes is sequestered in sulphides or magnetite, and the silicate minerals contain very little Fe (Table 1). The composition used is one with the highest molar $\mathrm{CO}_{2} /(\mathrm{CaO}+\mathrm{MgO})$ ratio (0.98) of all the analysed rocks from the NFD complex (unpublished data), meaning that it is the least decarbonated. Except for some rocks that had abundant detrital quartz, most reactions were quartz absent because quartz was consumed early by incipient reactions. The pseudosection has a swath of tremolite-bearing assemblages that extends from $\mathrm{CO}_{2}$-absent conditions to $X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}=0.7$. In terms of temperature, the permissible tremolite-present field is fairly narrow. Tremolite was probably formed continuously on the univariant reaction:

$$
\begin{equation*}
5 \mathrm{Dol}+8 \mathrm{Qtz}+\mathrm{H}_{2} \mathrm{O} \rightarrow \mathrm{Tr}+3 \mathrm{Cal}+7 \mathrm{CO}_{2} \tag{1}
\end{equation*}
$$

so that the interstitial fluid became $\mathrm{CO}_{2}$-bearing. However, given the apparent paucity of dolomite in the NFD rocks, it is possible that Reaction 1 involved a dolomite component in calcite instead of dolomite as a separate phase. Initial univariant, closed-system behaviour is predicted for the onset of decarbonation reactions because the locally generated fluid pressure typically exceeds the background hydrostatic pressure, thus preventing influx of an external $\mathrm{H}_{2} \mathrm{O}$ that would make the reaction discontinuous (Nabelek, 2007, 2009). Observed brecciation of rocks near dykes and sills demonstrates that fluid pressure sometimes exceeded the local lithostatic pressure. The pseudosection predicts an extremely small field for talc (Fig. 5b), which explains the apparent absence of talc from the metamorphosed rocks on Victoria Island.
Rocks containing the assemblage diopside + phlogopite occur directly above sills and near dykes. Indeed, these two minerals are predicted to coexist above $400^{\circ} \mathrm{C}$ across most of the range of $X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}$. The typically acicular, radial morphology of diopside on Victoria Island suggests that most of it replaced previously formed tremolite by the reaction:

$$
\begin{equation*}
\mathrm{Tr}+3 \mathrm{Cal} \rightarrow 4 \mathrm{Di}+\mathrm{Dol}_{\mathrm{ss}}+\mathrm{CO}_{2}+\mathrm{H}_{2} \mathrm{O} \tag{2}
\end{equation*}
$$

which has the dolomite component in calcite $\left(\mathrm{Dol}_{\mathrm{ss}}\right)$ as the product. The solubility of dolomite into calcite increases with temperature along the surface of the solvus between the two minerals (Goldsmith \& Newton, 1969). However, if quartz was still present and if $X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}$ reached more than $\sim 0.7$, diopside could have formed by the reaction


Fig. 5. Temperature- $X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}$ pseudosections showing stable mineral assemblages in metamorphosed rocks. The pseudosections were calculated using the theriak/domino computer program (De Capitani \& Petrakakis, 2010). Some narrow, low-variance assemblage fields are not labelled. (a) Pseudosection for rocks in Northern Feeder Dyke and Southern Feeder Dyke complexes showing assemblages across the full range of $X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}$ with the tremolite-present field highlighted. (b) Pseudosection for the same rocks highlighting the extent of serpentine (antigorite) stability in the $X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}$ range from 0 to 0.01 . (c) Pseudosection for rocks at Uhuk calculated assuming the $\mathrm{Ni}-\mathrm{NiO}$ buffer. Thin lines show fraction of andradite in garnet. The vesuvianite-present field is approximate due to a possible incompatibility between the thermodynamic data of Ogorodova et al. (2011) for Fe-vesuvianite and the db55 (2007) version of the thermodynamic database of Holland \& Powell (1998).

$$
\begin{equation*}
\mathrm{Dol}_{s s}+2 \mathrm{Qtz} \rightarrow \mathrm{Di}+\mathrm{CO}_{2} . \tag{3}
\end{equation*}
$$

Reactions 2 and 3 may have occurred at the same time if Reaction 2 was thermally overstepped, which would have effectively cancelled out the involvement of $\mathrm{dol}_{\text {ss }}$. Phlogopite most likely formed by the reaction:

$$
\begin{equation*}
3 \mathrm{Dol}_{\mathrm{ss}}+\mathrm{Mc}+\mathrm{H}_{2} \mathrm{O} \rightarrow \mathrm{Phl}+3 \mathrm{Cal}+3 \mathrm{CO}_{2} \tag{4}
\end{equation*}
$$

The restriction of the diopside + phlogopite assemblage to close proximities of the intrusive sheets, in spite of its large $T-X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}$ stability range, points to steep thermal gradients in the contact aureoles. Reaction 4 involves the addition of $\mathrm{H}_{2} \mathrm{O}$, which indicates that the rocks were invaded by $\mathrm{H}_{2} \mathrm{O}$ after decarbonation reactions that produced diopside ceased. The reaction would have also produced new, probably relatively coarse calcite in the presence of $\mathrm{H}_{2} \mathrm{O}$.

Many rocks that contain diopside, and a few that contain tremolite, have serpentine. Forsterite, which is predicted to occur above $400{ }^{\circ} \mathrm{C}$ in $\mathrm{H}_{2} \mathrm{O}$-rich conditions was not found in the rocks. Serpentine-bearing assemblages were modelled using the thermodynamic data for antigorite $\left(\mathrm{Mg}_{48} \mathrm{Si}_{34} \mathrm{O}_{147} \mathrm{H}_{62}\right)$ in the db55 database. (Using chrysotile instead of antigorite affects the assemblage boundaries only little). The antigorite field extends only to $X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}$ of 0.008 (Fig. 5b), which shows that the $\mathrm{CO}_{2}$ that was generated by prograde decarbonation reactions in the serpentinebearing high-grade rocks must eventually have been flushed away and that magnesian silicates (diopside and perhaps forsterite, if present) reacted to serpentine. The pseudosection suggests the following hydration reactions:

$$
\begin{equation*}
14 \mathrm{Fo}+10 \mathrm{Di}+10 \mathrm{Dol}_{\mathrm{ss}}+31 \mathrm{H}_{2} \mathrm{O} \rightarrow \mathrm{Ant}+20 \mathrm{Cal} \tag{5}
\end{equation*}
$$

or

$$
\begin{equation*}
17 \mathrm{Di}+31 \mathrm{Dol}_{\mathrm{ss}}+31 \mathrm{H}_{2} \mathrm{O} \rightarrow \mathrm{Ant}+48 \mathrm{Cal}+14 \mathrm{CO}_{2} \tag{6}
\end{equation*}
$$

These reactions also produce calcite that is likely to be fairly coarse due to the presence of $\mathrm{H}_{2} \mathrm{O}$.

Scapolite (not analysed and not included in the pseudosection) that occurs in some samples in the NFD complex and at the Uhuk Massif, with or without anhydrite veinlets, probably records influx of a salty fluid:

$$
\begin{array}{r}
\mathrm{An}+\mathrm{Cal}+\mathrm{NaCl}_{\mathrm{aq}}+\mathrm{CaSO}_{4, \mathrm{aq}} \\
\rightarrow(\mathrm{Ca}, \mathrm{Na})_{4}(\mathrm{Si}, \mathrm{Al})_{12} \mathrm{O}_{24}\left(\mathrm{CO}_{3}, \mathrm{SO}_{4}, \mathrm{Cl}\right) . \tag{7}
\end{array}
$$

## Uhuk Massif

The calcsilicate rocks at the Uhuk Massif include Febearing silicate minerals, including diopside, garnet and vesuvianite, that require the incorporation of Fe in the construction of a pseudosection (Fig. 5c). The whole-rock composition chosen has the highest $\mathrm{CO}_{2}$ concentration and among the lowest Fe concentrations of all analysed rocks (unpublished data) from the massif (Table 2). Because all the analysed rocks from the Uhuk Massif were decarbonated to various extents during metamorphism, $\mathrm{CO}_{2}$ was added to the composition to make its molar proportion equal to the proportion of $\mathrm{CaO}+\mathrm{MgO}$, under the assumption that before metamorphism, both of these oxides were contained only in calcite and dolomite. There is no evidence in the protoliths for detrital plagioclase that would have also contained calcium.

Fe-bearing vesuvianite is a frequent mineral in metamorphosed Uhuk rocks. However, the db55 thermodynamic database does not include data for Fe vesuvianite (one with two Fe atoms per formula unit). Therefore, thermodynamic data were taken from measurements on a natural Fe -vesuvianite by Ogorodova et al. (2011). The composition of vesuvianite is very close to $\mathrm{Mg}_{2} \mathrm{Ca}_{19} \mathrm{Fe}_{2} \mathrm{Al}_{9} \mathrm{Si}_{18} \mathrm{O}_{78} \mathrm{H}_{9}$ in which all Fe is ferric and it corresponds closely to the composition of vesuvianite at Uhuk (Table 1). Because the activitycomposition relationship between Fe -vesuvianite and Al-vesuvianite $\left(\mathrm{Mg}_{2} \mathrm{Ca}_{19} \mathrm{Al}_{11} \mathrm{Si}_{18} \mathrm{O}_{78} \mathrm{H}_{9}\right)$ is unknown, the solution was assumed ideal.

The oxygen fugacity was constrained to the $\mathrm{Ni}-\mathrm{NiO}$ buffer so that both ferrous and ferric iron could be present. This constraint agrees with magmatic oxygen fugacities of the mafic dykes and sills on Victoria Island, which were about two orders of magnitude higher than the quartz-fayalite-magnetite (QFM) buffer, as revealed by Fe-Ti-oxide oxybarometry (J. H. Bédard, unpublished data). A pseudosection calculated assuming the QFM buffer does not produce vesuvianite or andradite garnet. Ti in garnet was not considered in calculating the pseudosection.

The pseudosection for the Uhuk rocks (Fig. 5c) reproduces well the observed assemblages. However, because it is for a single composition, it does not address Fe metasomatism that was discussed above. All assemblages in the pseudosection include calcite. In unmetamorphosed rocks, the assemblage includes chlorite as the Fe-bearing silicate mineral. The ubiquitously coexisting diopside + garnet + magnetite in the high-grade rocks at Uhuk spans the full range of $X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}$ above $400^{\circ} \mathrm{C}$, and so provides no constraints on $X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}$. Diopside was most likely produced by a reaction similar to Reaction 2, but with magnetite as the source of Fe and, instead of using the dolomite component in calcite, it used dolomite that is common at Uhuk. Garnet probably initially formed by the reaction:

$$
\begin{array}{r}
0.702 \mathrm{Cpx}\left(\mathrm{Di}_{59} \mathrm{Hd}_{41}\right)+0.779 \mathrm{Cal}\left(\mathrm{Cal}_{96.4} \mathrm{Dol}_{3.6}\right) \\
+1.505 \mathrm{An}+0.104 \mathrm{Mt} \rightarrow \mathrm{Grt}\left(\mathrm{Grs}_{70} \mathrm{Adr}_{30}\right)  \tag{8}\\
+0.807 \mathrm{CO}_{2}
\end{array}
$$

The reaction is based on compositions of the reactants before the onset of the reaction and of garnet at the reaction boundary as predicted by THERIAK output at $X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}=0.5$. Elsewhere on the reaction boundary, the coefficients and compositions of solutions would be slightly different. The predicted garnet composition corresponds closely to compositions of garnet cores at Uhuk (Table 1). As shown by the contours in Fig. 5c, andradite component increases with temperature, particularly in the vesuvianite-present field, which is consistent with the andraditedominated garnet in vesuvianite-present rocks. Equilibrium diopside becomes more magnesian with increasing temperature, reaching nearly pure diopside
in the vesuvianite-present field, in agreement with the composition of diopside in vesuvianite-present rocks (Table 1). Reaction 8 is also consistent with the occurrence of both diopside and calcite as inclusions in garnet at Uhuk.

The vesuvianite-present field in the pseudosection occurs only above $500^{\circ} \mathrm{C}$ and low $X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}$. Due to the poor constraint on the activity relationship between Al and Fe -vesuvianite, the location of the vesuvianite-present field is only approximate. Nevertheless, its confinement to low $X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}$ is consistent with a location of the vesuvianite-in reaction at $X\left(\mathrm{CO}_{2}\right)^{\text {fluid }} \approx 0.003$ in an Fe-absent system (Nabelek \& Morgan, 2012). The model predicts a nearly pure Fe-vesuvianite, in accord with composition of vesuvianite in Uhuk rocks (Table 1). The growth of vesuvianite involved the consumption of garnet + diopside and hydration by the approximate reaction:

$$
\begin{array}{r}
2.85 \mathrm{Grt}_{\left(\mathrm{Grs}_{60} \mathrm{Adr}_{40}\right)+1.935 \mathrm{Cpx}\left(\mathrm{Di}_{93} \mathrm{Hd}_{7}\right)}+5.725 \mathrm{Cal}\left(\mathrm{Cal}_{96.5} \mathrm{Dol}_{3.5}+2.79 \mathrm{An}\right. \\
+4.5 \mathrm{H}_{2} \mathrm{O} \rightarrow \mathrm{Fe}-\mathrm{Ves}+0.138 \mathrm{Mt}+5.926 \mathrm{CO}_{2} \tag{9}
\end{array}
$$

Thus, the presence of vesuvianite in the highest grade calcsilicate rocks demonstrates that the $\mathrm{CO}_{2}$ that was produced earlier by decarbonation reactions was flushed away by $\mathrm{H}_{2} \mathrm{O}$, allowing vesuvianite to grow. A late crystallization of vesuvianite in the paragenetic sequence of minerals is consistent with its occurrence as a poikiloblastic phase that encloses all other phases in the rocks, except for serpentine.

Notable also is the confinement of chlorite and epidote-present fields to $\mathrm{CO}_{2}$-poor conditions (Fig. 5c). This shows that the fluids that were responsible for the formation of epidote and chlorite-bearing skarns at several locations on Victoria Island must have been $\mathrm{H}_{2} \mathrm{O}$-dominated.

## OXYGEN AND CARBON ISOTOPE RATIOS

Figure 6 shows summary plots for covariances of $\delta^{13} \mathrm{C}$ and $\delta^{18} \mathrm{O}$ values for carbonate minerals in metamorphosed rocks on Victoria Island (Table S1). The values fall within ranges of previously published data of Jones et al. (2010) for each of the examined formations. $\delta^{13} \mathrm{C}$ and $\delta^{18} \mathrm{O}$ values in the SFD-D zone are tightly clustered with $\sim 2 \%$ ranges for both. Exceptions to the clustering are samples collected within several metres of the dyke at the locality (highlighted in Fig. 6c). Both $\delta^{18} \mathrm{O}$ and $\delta^{13} \mathrm{C}$ are lower by $\sim 2 \%$ from the cluster in these rocks. Similar isotope ratio systematics are seen in the SFD-A zone. There are apparently two trends for the furthest samples from the dyke, with $\delta^{18} \mathrm{O}$ more elevated in the middle limestone ledge. The difference probably reflects an original variability related to stratigraphy (Jones et al., 2010). There seems to be a slight positive correlation between $\delta^{13} \mathrm{C}$ and $\delta^{18} \mathrm{O}$ in both trends. As in the SFD-D zone, highlighted
samples with lower $\delta^{13} \mathrm{C}$ values in Fig. 6b came from within a few metres of the dyke. The lowest $\delta^{13} \mathrm{C}$ values occur in samples that contain serpentine and/or garnet with epidote and have recrystallized calcite. Only one sample has an anomalously low $\delta^{18} \mathrm{O}$ by $\sim 3 \%$ and it contains a large proportion of chlorite. Overall, $\delta^{18} \mathrm{O}$ values are more elevated in the dolomitic Wynniatt

Formation in the SFD-D zone than in Killian Formation limestones in the SFD-A zone. Dolomitization typically elevates $\delta^{18} \mathrm{O}$ due to the larger oxygen isotope fractionation between dolomite and $\mathrm{H}_{2} \mathrm{O}\left(\Delta^{18} \mathrm{O}_{\text {Dol- }} \mathrm{H}_{2} \mathrm{O}\right)$ than between calcite and $\mathrm{H}_{2} \mathrm{O}\left(\Delta^{18} \mathrm{O}_{\mathrm{Cal}-} \mathrm{H}_{2} \mathrm{O}\right)$.

Isotope ratio shifts are especially well developed at Uhuk (Fig. 6d). All but one sample from the roof of


$\square$

Fig. 6. Plots showing $\delta^{13} \mathrm{C}$ and $\delta^{18} \mathrm{O}$ values of carbonate minerals in metamorphosed rocks. Samples highlighted by dashed lines were collected very near gabbroic dykes and sills. Lines show predicted effect of Rayleigh decarbonation on the $\delta^{13} \mathrm{C}$ and $\delta^{18} \mathrm{O}$ of remaining calcite in the rocks. Tick marks on the lines show the fraction of the original carbonate mineral remaining. The metamorphic rocks on Victoria Island probably did not lose more than $20 \%$ of calcite by calcsilicate reactions.


Fig. 7. Vertical profiles of $\delta^{13} \mathrm{C}$ and $\delta^{18} \mathrm{O}$ above a sill at the Uhuk Massif at stations 7 and 8 .
the sill and the vertical profile at station 7 define a tight, $2 \%$ cluster of $\delta^{13} \mathrm{C}$ and $\delta^{18} \mathrm{O}$. One skarn sample with the assemblage garnet + vesuvianite + diopside has a lower $\delta^{18} \mathrm{O}$ by $\sim 2 \%$ than the cluster. Details of the vertical profiles at stations 7 and 8 are shown in Fig. 7. There is a trend of decreasing $\delta^{13} \mathrm{C}$ toward the sill, particularly at station 8 that has andradite garnet together with vesuvianite and diopside throughout. Even a sample 10 m above the sill has a lower $\delta^{13} \mathrm{C}$ than most other samples at Uhuk. Trends for $\delta^{18} \mathrm{O}$ are analogous, giving a positive correlation between the two isotope ratios at station 8 (Fig. 6d). Station 8 samples within 3 m of the sill have the lowest $\delta^{13} \mathrm{C}$ and $\delta^{18} \mathrm{O}$ values. Such extents of heavy isotope depletions are not seen at station 7 or in the roof samples.

The largest spread of $\delta^{13} \mathrm{C}$ and $\delta^{18} \mathrm{O}$ occurs in the NFD complex with ranges of $\sim 5 \%$ for both (Fig. 6a). It is notable that most rocks with $\delta^{18} \mathrm{O}$ values of $<20 \%$, highlighted in Fig. 6, occur directly above the lower sill and the sill windows on the east side of the dyke (down-thrown side of the dyke-guiding fault), whereas on the west side of the dyke, most samples have $\delta^{18} \mathrm{O}$ values that appear unchanged from the primary $\delta^{18} \mathrm{O}$ values of the Boot Inlet Formation (Fig. 8). Many of the ${ }^{18} \mathrm{O}$-depleted rocks contain serpentine or chlorite, and the lowest $\delta^{18} \mathrm{O}$ values occur in epidote-bearing skarns at the intersection of the sill window and the dyke.

The systematics of the $\delta^{13} \mathrm{C}$ variation (Fig. 6a) are harder to unravel. Part of the range may be related to a primary stratigraphic variation, as suggested by different average values of sample traverses. However, analogously to the other locations, the lowest $\delta^{13} \mathrm{C}$
values occur close to the dyke and below the capping sill, indicating a causal relationship.

## DISCUSSION

## Fluids

The lowest $\delta^{13} \mathrm{C}$ and $\delta^{18} \mathrm{O}$ values in each studied locality occur within a few metres of dykes and sills in each traverse. Although the lowest $\delta^{18} \mathrm{O}$ values occur right next to mafic bodies, they are still significantly higher than the probable primary $\delta^{18} \mathrm{O}$ of the intru-


Fig. 8. Map distribution of $\delta^{18} \mathrm{O}$ in calcite in the Northern Feeder Dyke complex. Dark areas are sills and dykes (see Fig. 2a). Where more than one sample per station was analysed, the lowest value is shown.
sions and flood basalts on Victoria Island, which should be $5-9 \%$ based on analysis of continental basalts elsewhere (Harmon \& Hoefs, 1995). Clearly, there was only a limited oxygen isotope transfer from the mafic intrusions to the wall rocks, which contrasts with the frequent approach of oxygen isotope ratios of wall rocks to those of large granitic intrusions due to fluid expulsion from magmas (Valley, 1986; Nabelek, 1991; Baumgartner \& Valley, 2001). Many metamorphic rocks near the mafic bodies on Victoria Island are
${ }^{18} \mathrm{O}$-depleted beyond what can be explained by decarbonation. Calculated effects of Rayleigh decarbonation on $\delta^{18} \mathrm{O}$ and $\delta^{13} \mathrm{C}$ are superimposed on observed data in Fig. 6. They were calculated using isotopic fractionations between $\mathrm{CO}_{2}$ and calcite at $400{ }^{\circ} \mathrm{C}$, the approximate temperature at which most decarbonation reactions occurred (Fig. 5). The $\Delta^{18} \mathrm{O}_{\left(\mathrm{CO}_{2}-\mathrm{Cal}\right)}$ is $7 \%$ and $\Delta^{13} \mathrm{C}_{\left(\mathrm{CO}_{2}-\mathrm{Cal}\right)}$ is $3.6 \%$ (Chacko et al., 1991). Rayleigh decarbonation implies instantaneous removal of $\mathrm{CO}_{2}$ from the rock and therefore leads to greater isotopic depletion than batch decarbonation in which generated $\mathrm{CO}_{2}$ stays in the rock (Nabelek, 1991). Due to the limited amount of silicates in the examined metamorphic rocks, a maximum of $\sim 20 \%$ carbonate minerals were lost by metamorphic reactions. This implies the maximum of $\sim 2 \%$ decrease in $\delta^{18} \mathrm{O}$ by decarbonation. Thus, another mechanism must account for the excess ${ }^{18} \mathrm{O}$ depletion in the highlighted high-grade rocks in Fig. 6. The most likely mechanism was a fluid-mediated isotopic exchange with the proximal mafic sheets while calcite in the wall rocks was undergoing recrystallization or was produced by metamorphic reactions (e.g. Reactions 4 and 6).

The oxygen isotope systematics in the NFD complex (Fig. 8) present a particularly good illustration for the influence of mafic sheets on oxygen isotope depletion and they speak to the geometry of syn-metamorphic fluid flow. On the west side of the dyke, the largely unshifted $\delta^{18} \mathrm{O}$ values indicate little isotopic exchange between the dyke and wall rocks. This suggests that fluid flow there was mostly vertical, parallel to the dyke. In contrast, the evident ${ }^{18} \mathrm{O}$ depletion in the wall rocks on the east side of the dyke points to a more vigorous isotopic exchange with the mafic sheets, particularly the sills. The sills show a fair amount of chloritization of mafic minerals and the wall rocks have serpentine, chlorite or epidote, all of which point to flow of aqueous fluids through all the rocks on the east side of the fault while the rocks were hot. While the sills were still partially molten, hence impermeable, early fluid flow was probably convective between the sills. Upon solidification, the sills apparently became more permeable due to shrinking, and therefore were invaded by aqueous fluids.

Pore fluids from within the sedimentary basin probably dominated the aqueous fluids near the intrusive sheets, at least during the retrograde stages of metamorphism when serpentine grew. However, during early stages of metamorphism, after fluid pressure
next to the mafic sheets dropped, magmatic water emanating from the crystallizing mafic magma, sills especially, may have invaded the wall rocks. Mafic melts can contain as much as $3 \mathrm{wt} \%$ water at 1 kbar (Dixon et al., 1995), and this amount of exsolved water may have caused some, if not all, of the ${ }^{18} \mathrm{O}$ depletion in the wall rocks. Expulsion of water from sills is consistent with the development of Fe-enriched skarns that typically occur above sills, especially at Uhuk Massif. There is no evidence for influx of meteoric water into the basin during the mafic magmatism. If meteoric waters invaded the basin, the evidence for their presence must have been erased by isotopic exchange with the host rocks. Alternatively, because isotopic exchange of minerals with fluids typically occurs during recrystallization (Nabelek et al., 1992; Roselle et al., 1999), flow of sedimentary and meteoric fluids may not have been registered by unrecrystallized carbonate minerals.

There are three mechanisms by which $\delta^{13} \mathrm{C}$ in carbonate minerals can decrease from primary sedimentary values during metamorphism: (i) decarbonation; (ii) influx of low- ${ }^{13} \mathrm{C}$ carbon from a nearby magma; and (iii) exchange with organic carbon. $\delta^{13} \mathrm{C}$ values in host rocks proximal to the intrusions in the SFD-A and SFD-D zones and at Uhuk Massif are clearly lowered beyond what could have been accomplished by decarbonation (Fig. 6). Although a continental flood basalt magma can contain some $\mathrm{CO}_{2}$, the amount is unlikely to influence $\delta^{13} \mathrm{C}$ of carbon-dominated wall rocks. Assuming $\mathrm{CO}_{2}$ saturation at 1 kbar ( 3 km depth), the basaltic melts would have contained no more than $\sim 500 \mathrm{ppm} \mathrm{CO}_{2}$ (Dixon et al., 1995). Exchange with local low- ${ }^{13} \mathrm{C}$ organic carbon was probably the main mechanism by which $\delta^{13} \mathrm{C}$ values in carbonate minerals were decreased. $\delta^{13} \mathrm{C}$ of organic carbon ranges from -19 to $-28 \%$ (D. Thompson, personal communication). During metamorphic recrystallization of carbonates, $\delta^{13} \mathrm{C}$ of carbonate minerals commonly decreases due to exchange with organic material and can mimic apparent climatic $\delta^{13} \mathrm{C}$ excursions (Bergfeld et al., 1996; Labotka et al., 2000). Because the diffusion of carbon in carbonate minerals is very slow, $\sim 10^{-21} \mathrm{~m}^{2} \mathrm{~s}^{-1}$ over the full range of $X\left(\mathrm{CO}_{2}\right)^{\text {fluid }}$ (Labotka et al., 2011), carbon isotope exchange in Victoria Island metamorphic rocks must have occurred during recrystallization of calcite in the presence of carbon-bearing fluids. Indeed, in samples with anomalously ${ }^{13} \mathrm{C}$-depleted carbon, calcite has been recrystallized and is typically poikiloblastic. Recrystallization to a larger grain size itself implies the presence of an aqueous fluid component.

Overall, the carbon and oxygen isotope data and mineralogy point to focused fluid flow through narrow zones at the interfaces between the sheet intrusions and their wall rocks. The focusing may have been guided by a relative impermeability of the intrusions and by creation of reaction-enhanced permeability in the wall rocks due to a decrease in the
volume of the solid assemblages by reactions (e.g., Rumble et al., 1982; Nabelek et al., 1984; Nabelek, 2009). Moreover, high fluid pressures generated by heating of pore fluids and by early metamorphic decarbonation reactions may have driven fluid flow near the intrusions. Reaction-enhanced permeability and high transient fluid pressures would have been particularly important at Uhuk Massif due to the high proportion of silicates in the protoliths. They probably promoted the evident Fe metasomatism, the development of skarns and the brecciation of the sill's roof at Uhuk. Syn-intrusion faulting and junctions between sills and dykes may also have enhanced fluid flow near the mafic sheets.

## Duration of metamorphism

Rapid heating and a short duration of elevated temperature regimes in contact aureoles of the gabbroic sheets on Victoria Island is implied by the prevalent acicular morphology of diopside and the variable recrystallization of calcite in thick marble horizons (Fig. 3). The morphology differs from the usual equant morphology of diopside and coarse marbles in aureoles of large intrusions (Berger et al., 2010), where elevated temperature regimes can prevail for tens of thousands of years due to large pluton sizes and long durations of pluton growth (Annen, 2011; Nabelek et al., 2012). Around large plutons, rock textures have a greater chance of annealing. On Victoria Island, rapid emplacement and much smaller thicknesses of gabbroic sheet intrusions limited the time that the wall rocks spent at elevated temperatures. However, an extended period of retrograde, $\mathrm{H}_{2} \mathrm{O}$-present metamorphism is evident in the frequent occurrence of hydrous minerals, including serpentine, chlorite and epidote.

Calculated temperature-time $(T-t)$ paths of the aureole rocks support the occurrence of a short-duration elevated temperature regime, followed by an extended period of a retrograde regime in the contact aureoles on Victoria Island (Fig. 9). The profiles were calculated following the method of Nabelek et al. (2012) that employs temperature-dependent thermal diffusivity of crystallizing melts and aureole rocks (Whittington et al., 2009). The necessary liquidus and solidus temperatures, 1150 and $840^{\circ} \mathrm{C}$, respectively, were obtained with the program melts (Ghiorso \& Sack, 1995) for compositions of the Victoria Island sills (Dostal et al., 1980). The initial temperature of the wall rocks was assumed to be $80^{\circ} \mathrm{C}$, which is appropriate for magma emplacement depth of $\sim 3 \mathrm{~km}$. A 50 m thick gabbroic sheet intrusion and heat transfer only by conduction were assumed.

The $T-t$ paths for a rock 5 m from the sheet intrusion show a rapid temperature increase into a temperature regime in which diopside and phlogopite coexist in metamorphic rocks of the NFD and SFD complexes and in which vesuvianite, garnet and diop-


Fig. 9. Calculated conductive temperature-time paths surrounding a 50 m thick intrusive sheet. Liquidus and solidus temperatures, 1150 and $840^{\circ} \mathrm{C}$, respectively, were obtained with the program melts (Ghiorso \& Sack, 1995) for compositions of the Victoria Island sills (Dostal et al., 1980). The initial temperature of the wall rocks was assumed to be $80^{\circ} \mathrm{C}$. Note that 150 years after emplacement of a sheet intrusion, there is a convergence of temperatures onto the $300-400{ }^{\circ} \mathrm{C}$ range, in which retrograde serpentine, chlorite and epidote are predicted to be stable in the metamorphic rocks on Victoria Island (Fig. 5).
side coexist in Uhuk rock compositions (Fig. 5). The rapid temperature increase is consistent with prograde growth of acicular diopside splays. Irrespective how thick a sheet intrusion is, temperature will always increase rapidly in rocks right next to it, although the maximum temperature that will be reached will vary. Rocks 25 m from the model sheet intrusion reach diopside-present fields, but c. 60 years later. Rocks 50 m from the sheet intrusion barely reach the tremolite present field, if at all. Notable is the convergence after $c .150$ years of the three illustrated $T-t$ paths onto the $350-400{ }^{\circ} \mathrm{C}$ temperature range in which serpentine, chlorite and epidote are stable (Fig. 5). The calculations suggest that this temperature regime remained active around individual intrusive sheets for an extended period of time, giving an opportunity for these retrograde hydrous minerals to grow in the presence of convecting aqueous fluids.

The Shaler Supergroup has numerous sills throughout its exposed stratigraphy. There are at least two geochemically distinct populations of dykes, sills and flood basalt that, based on field relationships, may represent different phases of the Franklin magmatic event (Williamson et al., 2012). This would suggest that magmatism in the region occurred over a time period that was several orders of magnitude longer than the time it took to inject one dyke or one sill into the Shaler Supergroup. The narrow contact aureoles around the intrusive sheets are consistent with a long duration of the Franklin magmatic event on Victoria Island, because they point to only a limited heat input into the sedimentary basin by injected mafic magma at any one time.

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## REFERENCES

Annen, C., 2011. Implications of incremental emplacement of magma bodies for magma differentiation, thermal aureole dimensions and plutonism-volcanism relationships. Tectonophysics, 500, 3-10.
Armbruster, T., Birrer, J., Libowitzky, E. \& Beran, A., 1998. Crystal chemistry of Ti-bearing andradites. European Journal of Mineralogy, 10, 907-921.
Baumgartner, L.P. \& Valley, J.W., 2001. Stable isotope transport and contact metamorphic fluid flow. Reviews in Mineralogy and Geochemistry, 43, 415-467.
Bédard, J.H., Naslund, H.R., Nabelek, P. et al., 2012. Faultmediated melt ascent in a Neoproterozoic continental flood basalt province, the Franklin sills, Victoria Island, Canada. Geological Society of America Bulletin, 124, 723-736.
Berger, A., Brodhag, S.H. \& Herwegh, M., 2010. Reaction-induced nucleation and growth v . grain coarsening in contact metamorphic, impure carbonates. Journal of Metamorphic Geology, 28, 809-824.
Bergfeld, D., Nabelek, P.I. \& Labotka, T.C., 1996. Carbon isotope exchange during polymetamorphism in the Panamint Mountains, California. Journal of Metamorphic Geology, 14, 199-212.
Cathles, L.M., 1981. Fluid flow and genesis of hydrothermal ore deposits. Economic Geology, 75th Anniversary Volume, 424-457.
Chacko, T., Mayeda, T.K., Clayton, R.N. \& Goldsmith, J.R., 1991. Oxygen and carbon isotope fractionations between $\mathrm{CO}_{2}$ and calcite. Geochimica et Cosmochimica Acta, 55, 2867-2882.
Cook, S.J., Bowman, J.R. \& Forster, C.B., 1997. Contact metamorphism surrounding the Alta stock: finite element model simulation of heat- and ${ }^{18} \mathrm{O} /{ }^{16} \mathrm{O}$ mass-transport during prograde metamorphism. American Journal of Science, 297, 1-55.
Cui, X., Nabelek, P.I. \& Liu, M., 2001. Heat and fluid flow in contact metamorphic aureoles with layered and transient permeability, with application to the Notch Peak aureole, Utah. Journal of Geophysical Research, 106, 6477-6491.
Cui, X., Nabelek, P.I. \& Liu, M., 2002. Numerical modeling of fluid flow and oxygen isotope exchange in the Notch Peak contact-metamorphic aureole, Utah. Geological Society of America Bulletin, 114, 869-882.

De Capitani, C. \& Petrakakis, K., 2010. The computation of equilibrium assemblage diagrams with Theriak/Domino software. American Mineralogist, 95, 1006-1016.
Dixon, J.E., Stolper, E.M. \& Holloway, J.R., 1995. An experimental study of water and carbon dioxide solubilities in midocean ridge basaltic liquids. Part I: Calibration and solubility models. Journal of Petrology, 36, 1607-1631.
Dostal, J., Dupuy, C. \& Leyreloup, A., 1980. Geochemistry and petrology of meta-igneous granulitic xenoliths in Neogene volcanic rocks of the Massif Central, France - implications for the lower crust. Earth and Planetary Science Letters, 50, 31-40.
Dostal, J., Baragar, W.R.A. \& Dupuy, C., 1986. Petrogenesis of the Natkusiak continental basalts, Victoria Island, Northwest Territories, Canada. Canadian Journal of Earth Sciences, 23, 622-632.
Fournier, R.O., 1989. Geochemistry and dynamics of the Yellostone National Park hydrothermal system. Annual Reviews of Earth and Planetary Sciences, 17, 13-53.
Ghiorso, M.S. \& Sack, R.O., 1995. Chemical mass transfer in magmatic processes. IV. A revised and internally consistent thermodynamic model for the interpolation and extrapolation of liquid-solid equilibria in magmatic systems at elevated temperatures and pressures. Contributions to Mineralogy and Petrology, 119, 197-212.
Goldsmith, J.R. \& Newton, R.C., 1969. P-T-X relations in the system $\mathrm{CaCO}_{3}-\mathrm{MgCO}_{3}$ at high temperatures and pressures. American Journal of Science, 267-A, 160-190.
Harmon, R.S. \& Hoefs, J., 1995. Oxygen isotope heterogeneity of the mantle deduced from global ${ }^{18} \mathrm{O}$ systematics of basalts from different tectonic settings. Contributions to Mineralogy and Petrology, 120, 95-114.
Heaman, L., Lecheminant, A. \& Rainbird, R., 1992. Nature and timing of Franklin igneous events, Canada: implications for a Late Proterozoic mantle plume and the break-up of Laurentia. Earth and Planetary Science Letters, 109, 117-131.
Holland, T.J.B. \& Powell, R., 1998. An internally consistent thermodynamic data set for phases of petrological interest. Journal of Metamorphic Geology, 16, 309-343.
Hulbert, L.J., Rainbird, R.W., Jefferson, C.W. \& Friske, P., 2005. Map of Mafic and Ultramafic Bodies Related to the Franklin Magmatic Event, Minto Inlier. Geological Survey of Canada, Victoria Island.
Jones, D.S., Maloof, A.C., Hurtgen, M.T., Rainbird, R.H. \& Schrag, D.P., 2010. Regional and global chemostratigraphic correlation of the early Neoproterozoic Shaler Supergroup, Victoria Island, Northwestern Canada. Precambrian Research, 181, 43-63.
Kretz, R., 1983. Symbols for rock-forming minerals. American Mineralogist, 68, 277-279.
Labotka, T.C., 1995. Evidence for immiscibility in Ti-rich garnet in a calc-silicate hornfels from northeastern Minnesota. American Mineralogist, 80, 1026-1030.
Labotka, T.C., Bergfeld, D. \& Nabelek, P.I., 2000. Two diamictites, two cap carbonates, two $\delta^{13} \mathrm{C}$ excursions, two rifts: the Neoproterozoic Kingston Peak Formation, Death Valley, California: comment and Reply. Geology, 28, 191-192.
Labotka, T.C., Cole, D.R., Fayek, M.J. \& Chacko, T., 2011. An experimental study of the diffusion of C and O in calcite in mixed $\mathrm{CO}_{2}-\mathrm{H}_{2} \mathrm{O}$ fluid. American Mineralogist, 96, 1262-1269.
MacDonald, W.D., Bédard, J., Hayes, B., Naslund, H.R., Carpenter, J. \& Steigerwaldt, K., 2012. AMS magnetofabrics and emplacement of Franklin Dykes, Victoria Island, Arctic Canada. Geological Association of Canada - Mineralogical Association of Canada annual meeting, Program with Abstracts, 35, 82.
Nabelek, P.I., 1991. Stable isotope monitors. Reviews in Mineralogy, 26, 395-435.
Nabelek, P.I., 2007. Fluid evolution and kinetics of metamorphic reactions in calc-silicate contact aureoles - From $\mathrm{H}_{2} \mathrm{O}$ to $\mathrm{CO}_{2}$ and back. Geology, 35, 927-930.
Nabelek, P.I., 2009. Numerical simulation of kinetically-controlled calc-silicate reactions and fluid flow with transient
permeability around crystallizing plutons. American Journal of Science, 309, 517-548.
Nabelek, P.I. \& Morgan, S.S., 2012. Metamorphism and fluid flow in the contact aureole of the Eureka Valley-Joshua FlatBeer Creek pluton, California. Geological Society of America Bulletin, 124, 228-239.
Nabelek, P.I., Labotka, T.C., O’Neil, J.R. \& Papike, J.J., 1984. Contrasting fluid/rock interaction between the Notch Peak granitic intrusion and argillites and limestones in western Utah: evidence from stable isotopes and phase assemblages. Contributions to Mineralogy and Petrology, 86, 25-34.
Nabelek, P.I., Labotka, T.C. \& Russ-Nabelek, C., 1992. Stable isotope evidence for the role of diffusion, infiltration and local structure on contact metamorphism of calc-silicate rocks at Notch Peak, Utah. Journal of Petrology, 33, 557-583.
Nabelek, P.I., Hofmeister, A.M. \& Whittington, A.G., 2012. The influence of temperature-dependent thermal diffusivity on the conductive cooling rates of plutons and temperature-time paths in contact aureoles. Earth and Planetary Science Letters, 317, 318, 157-164.
Norton, D.L., 1984. Theory of hydrothermal systems. Annual Review of Earth and Planetary Sciences, 12, 155-177.
Norton, D.L. \& Taylor, H.P., 1979. Quantitative simulation of the hydrothermal systems of crystallizing magmas on the basis of transport theory and oxygen isotope data: an analysis of the Skaergaard Intrusion. Journal of Petrology, 20, 421-486.
Ogorodova, L.P., Kiseleva, I.A., Melchakova, L.V. \& Spiridonov, E.M., 2011. Thermodynamic characteristics of minerals of the vesuvianite group. Geochemistry International, 49, 191-195.
Rainbird, R.H., 1993. The sedimentary record of mantle plume uplift preceding eruption of the Neoproterozoic Natkusiak flood basalt. The Journal of Geology, 101, 305-318.
Rainbird, R.H., Jefferson, C.W., Hildebrand, R.S. \& Worth, J.K., 1994. The Shaler Supergroup and revision of Neoproterozoic stratigraphy in Amundsen Basin, Northwest Territories. Current Research 1994-C, Geological Survey of Canada, 61-70.

Rainbird, R., Jefferson, C. \& Young, G., 1996. The early Neoproterozoic sedimentary Succession B of northwestern Laurentia: correlations and paleogeographic significance. Geological Society of America Bulletin, 108, 454-470.
Roselle, G.T., Baumgartner, L.P. \& Valley, J.W., 1999. Stable isotope evidence of heterogeneous fluid infiltration at the Ubehebe Peak contact aureole, Death Valley National Park, California. American Journal of Science, 299, 93-138.
Rumble, D., Ferry, J.M., Hoering, T.C. \& Boucot, A.J., 1982. Fluid flow during metamorphism at the Beaver Brook Fossil Locality, New Hampshire. American Journal of Science, 282, 886-919.
Toulmin, P. \& Barton, P.B., 1964. A thermodynamic study of pyrite and pyrrhotite. Geochmica et Cosmochimica Acta, 28, 641-671.
Valley, J.W., 1986. Stable isotope geochemistry of metamorphic rocks. Review in Mineralogy, 16, 445-489.
Whittington, A.G., Hofmeister, A.M. \& Nabelek, P.I., 2009. Temperature-dependent thermal diffusivity of Earth's crust and implications for magmatism. Nature, 458, 319-321.
Williamson, N., Cousens, B., Ootes, L., Bédard, J., Rainbird, R. \& Dell'Oro, T., 2012. Volcano-stratigraphy and major element geochemistry of the Southern Lobe of the Natkusiak Formation Flood Basalts of Victoria Island: insights into the initiation of the Neoproterozoic Franklin Magmatic Event. Geological Association of Canada - Mineralogical Association of Canada annual meeting, Program with Abstracts, 35, 152.

## SUPPORTING INFORMATION

Additional Supporting Information may be found in the online version of this article at the publisher's web-site:

Table S1. $\delta 13 \mathrm{C}$ and $\delta 18 \mathrm{O}$ values of carbonate minerals in metamorphosed Victoria Island rocks.

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## Appendix C - electron microprobe methods for mineral species analysed

The following methods were used for all mineral chemical data presented in this thesis.

Olivine compositions. Analyses were done at 20 nA at 15 kV . Detection limits are $c a$. $0.01 \mathrm{wt} \%$ for all elements. Background values were counted for $1 / 2$ the time counted at the peak. Spectrometer 1: LIF crystal, $\mathrm{Mn}-\mathrm{K} \alpha, 20$ seconds, standard = busx. $\mathrm{Fe}-\mathrm{K} \alpha$, 20 seconds, standard $=$ haematite. Spectrometer 2: LLIF crystal, $\mathrm{Cr}-\mathrm{K} \alpha, 20$ seconds, standard $=$ chromite. $\mathrm{Na}-\mathrm{K} \alpha, 20$ seconds, standard $=$ nickel metal. Spectrometer 3: LPET crystal, $\mathrm{Ca}-\mathrm{K} \alpha, 20$ seconds, standard $=$ diopside. Ti-K $\alpha, 20$ seconds, standard $=$ rutile. Spectrometer 4: TAP crystal, $\mathrm{Si}-\mathrm{K} \alpha, 20$ seconds, standard = olvx (olivine). Mg $\mathrm{K} \alpha, 20$ seconds, standard $=$ olvx. Spectrometer 5: TAP crystal, Al-K $\alpha, 20$ seconds, standard = plagvz (plagioclase). Not analyzed: O. Number of Oxygen's $=4$. Number of $\mathrm{H} 2 \mathrm{O}=0$.

Pyroxene compositions. Analyses were done at 20 nA at 15 kV . Detection limits are ca. $0.01 \mathrm{wt} \%$ for all elements. Background values were counted for $1 / 2$ the time counted at the peak. Spectrometer 1: LIF crystal, $\mathrm{Fe}-\mathrm{K} \alpha, 15$ seconds, standard $=$ haematite. $\mathrm{Cr}-\mathrm{K} \alpha, 15$ seconds, standard $=$ chromite. $\mathrm{Ti}-\mathrm{K} \alpha, 15$ seconds, standard $=$ rutile. Spectrometer 2: LLIF crystal, $\mathrm{Ni}-\mathrm{K} \alpha, 20$ seconds, standard = nickel metal. Mn$\mathrm{K} \alpha, 20$ seconds, standard = busx. Spectrometer 3: LPET crystal, K-K $\alpha, 20$ seconds, standard $=$ orthoclase $. \mathrm{Ca}-\mathrm{K} \alpha, 20$ seconds, standard $=$ diopside. Spectrometer 4: TAP crystal, $\mathrm{Si}-\mathrm{K} \alpha, 20$ seconds, standard $=$ gzvz. $\mathrm{Mg}-\mathrm{K} \alpha, 20$ seconds, standard $=$ olvx. Spectrometer 5: TAP crystal, $\mathrm{Na}-\mathrm{K} \alpha, 20$ seconds, standard $=$ albite. Al-K $\alpha, 20$ seconds, standard = cor. Not analyzed: O. Number of Oxygen's $=6$. Number of H2O $=0$. Same method used for both clinopyroxene and orthopyroxene.

Plagioclase compositions. Analyses were done at 20 nA at 15 kV . Detection limits are ca. $0.01 \mathrm{wt} \%$ for all elements. Background values were counted for $1 / 2$ the time counted at the peak. Spectrometer 1: LIF crystal, Ti-K $\alpha, 20$ seconds, standard $=$ rutile . $\mathrm{Mn}-\mathrm{K} \alpha, 20$ seconds, standard = busx. Spectrometer 2: LIFF crystal, $\mathrm{Fe}-\mathrm{K} \alpha, 20$
seconds, standard $=$ haematite $. \mathrm{Ba}-\mathrm{K} \alpha, 20$ seconds, standard $=$ barite. Spectrometer 3: LPET crystal, $\mathrm{Ca}-\mathrm{K} \alpha, 20$ seconds, standard $=$ diopside. $\mathrm{K}-\mathrm{K} \alpha, 15$ seconds, standard $=$ orthoclase. $\mathrm{Sr}-\mathrm{L} \alpha, 15$ seconds, standard = celestite. Spectrometer 4: TAP crystal, Si$\mathrm{K} \alpha, 20$ seconds, standard = plagvz. Mg-K $\alpha, 20$ seconds, standard $=$ olvx.

Spectrometer 5: $\mathrm{Na}-\mathrm{K} \alpha, 20$ seconds, standard $=$ albite $. \mathrm{Al}-\mathrm{K} \alpha, 20$ seconds, standard $=$ cor. Not analyzed: O. Number of Oxygen's $=8$. Number of $\mathrm{H} 2 \mathrm{O}=0$.

Spinel compositions. Analyses were done at 20 nA at 15 kV . Detection limits are $c a$. $0.01 \mathrm{wt} \%$ for all elements. Background values were counted for $1 / 2$ the time counted at the peak. Spectrometer 1: LIF crystal, $\mathrm{Co}-\mathrm{K} \alpha, 15$ seconds, standard $=$ ATX-Co. Fe$\mathrm{K} \alpha, 15$ seconds, standard $=$ haematite $\mathrm{Mn}-\mathrm{K} \alpha, 15$ seconds, standard $=$ busx. Spectrometer 2: LLIF crystal, $\mathrm{Zn}-\mathrm{K} \alpha, 15$ seconds, standard $=$ sphalerite. $\mathrm{Ni}-\mathrm{K} \alpha, 15$ seconds, standard $=$ nickel metal. $\mathrm{Cr}-\mathrm{K} \alpha, 15$ seconds, standard $=$ chromite. Spectrometer 3: LPET crystal, $\mathrm{Ca}-\mathrm{K} \alpha, 20$ seconds, standard $=$ diopside. Ti-K $\alpha, 20$ seconds, standard = rutile. Spectrometer 4: TAP crystal, Si-K $\alpha, 20$ seconds, standard $=$ gzvz. $\mathrm{Na}-\mathrm{K} \alpha, 20$ seconds, standard $=$ albite. Spectrometer 5: Al-K $\alpha, 20$ seconds, standard = plagvz. $\mathrm{Mg}-\mathrm{K} \alpha, 20$ seconds, standard = olvx. Not analyzed: H. Not analyzed: O . Number of Oxygen's $=32$. Number of $\mathrm{H} 2 \mathrm{O}=0$.

## Appendix D1 - Franklin liquid line of descent

Jean Bédard, co-supervisor of this PhD thesis, carried out much of the Franklin magma liquid line of descent (LLD) modelling. Jean constructed and organised the Victoria Island geochemical database and modelled the magma suites to find the best-fit liquid line of decent to the two suites of magma (Franklin Type-1 and Type-2). Below is a summary of the geochemical database and justifications for the model parameters.

The major element distributions of the Type-1 versus Type-2 suites were examined using simple Harker plots. The database was culled to remove obvious olivine and clinopyroxene cumulates, as well as obvious hybrids with limestone. The culled database includes samples classed as lavas (v), chilled margins ( + ) and diabasic facies ( x ). It is probable that many of these compositions are perturbed by a) post emplacement alteration, b) metasomatism or exchange with wallrocks and/or c) pre-, syn- or post-emplacement accumulation of phases. Although the most obvious olivine cumulates were excluded, it is probable that some of the higher- MgO analyses from the database (many with no corresponding petrographic data) contain 'cumulus' olivine. This will be tested using the PELE modelling. Some of the diabasic samples may represent 2- or 3-phase cumulates (cpx+plag+olivine, $c p x+$ plag+pigeonite, plag+cpx+magnetite...), principally reflecting loss of pore melt during solidification. This would mainly act to decrease incompatible element contents (TiO2, K2O, P2O5..), shift bulk compositions back along the cotectic Fe enrichment trend, and possible cause enrichment of Al 2 O 3 and CaO above the true LLD.

There are obvious differences in the distributions of the two suites of Franklin magmas (Type-1 versus Type-2), which imply they should be modelled separately. Magmas of the Type-1 suite extend to higher MgO contents than those of the Type-2 suite. Although much of the Type- 1 high- MgO tail is probably an accumulation effect, the almost complete lack of such a tail for the Type- 2 suite implies that these magmas did not experience a long olivine-only crystallisation history after emplacement in the crust. Furthermore, Type-1 sills contain relatively little ferrogabbro with high $\mathrm{FeO} / \mathrm{MgO}$, in contrast with the Type-2 sills, which do. As a result, most sills related to the Type-1 suite show a clear Feenrichment trend, but rarely 'turn the corner' towards an Fe-depletion/SiO2 enrichment trend that indicates abundant extraction of magnetite. This contrasts with sills of Type-2 suite, that show more extended evolution towards low FeO and higher SiO 2 . The overall
trend of geochemical evolution also differs between the two suites. The Type-1 suite shows a trajectory at a slightly lower FeO and TiO 2 , and a slightly higher SiO 2 content for a given MgO value, and starts its $\mathrm{FeO}-\mathrm{TiO} 2$-enrichment trend at a slightly lower MgO content than the Type-2 sills. Thus, the FeO-TiO2-enrichment trends of the two suites are offset one from the other. Additionally, Type-1 magmas show a slightly higher K2O content, but the two suites have similar Na 2 O contents.

Since many of the most detailed studies were done on sills emplaced near the Fort Collinson Formation, pressures for Type-1 magma differentiation were estimated from the height of this overlying stratigraphic column, $\sim 2.5 \mathrm{~km}$, which translates to $\sim 0.8 \mathrm{kbar}$. All simulations were done at this pressure. Preliminary QUILF calculations by Bédard et al (2008) on Type-1 magmas suggest $\mathrm{QFM}+1$ to +2 , so calculations were done at $\mathrm{QFM}+1$ and +2 , bracketing observed fO2 conditions. The gabbroic segment of the PELE models cannot exactly mimic the observed trends, however, because the PELE code does not include pigeonite, which is observed near the base of the gabbroic sequence in several Type-1 sills.

Trace hornblende and biotite occur as rims on higher-temperature silicates in many sills, suggesting that some magmas were not entirely dry, such that extended in-situ fractional crystallization locally caused water contents to build up to the point where these phases were stabilized. An initial water content of $0.2 \%$ was assumed for some models, since $95 \%$ crystallization of such a starting point will yield the $4 \% \mathrm{H} 2 \mathrm{O}$ content needed to stabilize hornblende. Of course, F and Cl will stabilize these phases even at lower H 2 O contents. Limited data show that Hb and biotite in Franklin sills do indeed contain significant Cl and F-contents (up to $1-1.5 \mathrm{wt} \%$ combined $\mathrm{F}+\mathrm{Cl}$ ), but some water may also be present. There is no direct analytical data on H in these phases to constrain this at present. Melt CO2 solubilities are very low at these pressures, and initial staring contents were assumed to be ca $0.2 \%$ also. These values were varied.

Typical S-contents of Franklin magmas are ca $0.04 \%$, with some chilled margins recording higher values (up to ca $0.15 \%$ ). Small Ni-Fe sulphide blebs in chills and olivine cumulates suggest early, small-scale sulphide immiscibility may be common. Even though PELE does not consider the effects of Cu and Ni on sulphide immiscibility, early Fe-sulphide melt exsolution is commonly observed in models that assumed higher initial S contents of
ca $0.12 \%$, and small amounts of sulphide exsolves intermittently throughout the model crystallization history. The amounts are always small, however, in keeping with the small amount of initial S in the melt. More evolved ferrogabbroic rocks typically contain $\mathrm{Cu}-\mathrm{Fe}$ sulphide blebs. Models that assumed low initial $S$ contents of ca $0.04 \%$ only showed immiscibility in the later crystallization history, often after cosaturation of magnetite. This suggests that the sharp decrease in FeO caused by magnetite crystallization often triggers immiscibility by lowering S-solubility in the melt, which is strongly dependent on FeO content (Mavrogenes?). These results and observations imply that the v0 melts were probably very close to sulphide saturation, but that $S$ contents were probably too low to create sulphide orebodies during closed-system crystallization. On the other hand, even minor assimilation of S-rich limestone or sulphates would probably trigger immiscible sulphide separation, with the amount of sulphide produced correlating directly to S-input.


A final 'optimized' model K was run with slightly lower CaO , water ( $0.1 \%$ ) and CO2 ( $0.1 \%$ ), and higher Al 2 O 3 and SiO 2 the initial Uhuk chill model ( C ), and yields an optimal fit to the data. Na 2 O is probably very slightly too high, since mixing lines with plausible olivine compositions do not fit the accumulative high- MgO tail of the data. This presupposes that these compositions are primary, however. This model hits the solidus at about 1240C, initially crystallizing Fo 88 olivine. About 7.5\% olivine is extracted prior to feldspar cosaturation at ca 1175C. At these conditions Fo85 olivine and An79 plagioclase form, compositions very similar to the observed data. Only a short interval of ol+plag crystallization is allowed before cpx joins the assemblage below 1165C at SiO 2 ca 51.7 and $\mathrm{MgO} 7.3 \%$, in accord with the absence of thick wehrlitic cumulate sequences. The 3phase assemblage crystallizes between 1160C and 1060 C , with an average proportion of Ol:Cpx:plag of 16:37:47. Magnetite joins the assemblage at $\mathrm{SiO} 2>54 \%, \mathrm{FeO} *>16.6 \%$, $\mathrm{TiO} 2>2.25 \%$ and $\mathrm{MgO}<3 \%$, with an initial pulse of $3.5 \% / 10 \mathrm{C}$ crystallization at 1060 C , also triggering minor sulphide immiscibility. Thereafter, the proportion decreases to $0.8 \%$ and then to $0.5,0.4$, and 0.3 at 1050, 1040 and 1030C. After 1030C, orthopyroxene joins the assemblage. Ilmenite is absent, however. The subsequent evolutionary path is probably not realistic.

I compared the LLD derived from PELE that best fits the Franklin Type-1 suite to MELTS. Below are simple excel plots showing the LLDs from PELE (green triangles) and MELTS (red squares) run at the same parameters (high $\mathrm{Ce} / \mathrm{Yb}$ family in blue diamonds refers to the Type-1 Franklin magma suite). Fe-enrichment starts earlier in the PELE model. The PELE model has Fe-depletion beginning at $4 \% \mathrm{MgO}$ (after reaching a maximum of $16 \% \mathrm{FeO}$ ). MELTS reaches $13 \% \mathrm{FeO}$ at a slightly higher $\mathrm{MgO}(5-6 \%)$. MELTS has CaO decreasing at $9 \% \mathrm{MgO}$ (cpx in). PELE has CaO decreasing at $7 \% \mathrm{MgO}$. MELTS has Al2O3 increasing at $9 \% \mathrm{MgO}$ (cpx in) and then decreasing at $7.5 \% \mathrm{MgO}$ (plag in). Al2O3 in the PELE model decreases earlier and thus matches the high $\mathrm{Ce} / \mathrm{Yb}$ family liquid line of descent much better.

The plots below show the LLD derived from PELE is a better fit to the Type-1 Franklin magma suite.




Below are other LLDs calculated from MELTS for other parameters. These plots concur with the above plots, showing that the LLD calculated from PELE is the best-fit model to the data.




## Appendix D2 - Model melt parameterisations for mineral chemistry

Below are simple excel plots showing $\mathrm{Fe} / \mathrm{Mg}$ versus MgO and FeO that were used to calculate the model melt compositions in equilibrium with olivine and clinopyroxene. The plots include Type-1 Franklin liquid compositions from the Victoria Island database.


## Appendix E1 - LPS whole-rock chemistry

| Sample | $\begin{gathered} \text { 10JB327 } \\ \text { B1 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A2 } \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A3 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 11BH311 } \\ \text { B2 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A4 } \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A5 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A1 } \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A2 } \\ \hline \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | LCM | LBZ | LBZ | LOZ | LOZ | LOZ | LOZ | MOZ |
| $\mathrm{SiO}_{2}$ | 48.28 | 46.47 | 45.07 | 44.36 | 44.01 | 44.49 | 43.66 | 42.39 |
| $\mathrm{TiO}_{2}$ | 1.02 | 0.92 | 0.8 | 0.79 | 0.72 | 0.73 | 0.7 | 0.58 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 12.76 | 11.23 | 9.49 | 9.65 | 8.95 | 9 | 8.83 | 6.68 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{~T}$ | 11.13 | 13.25 | 13.28 | 13.49 | 13.4 | 13.46 | 13.68 | 14.83 |
| FeOT | 10.01 | 11.91 | 11.94 | 12.12 | 12.05 | 12.1 | 12.3 | 13.33 |
| MnO | 0.15 | 0.19 | 0.19 | 0.2 | 0.19 | 0.2 | 0.18 | 0.2 |
| MgO | 10.74 | 15.32 | 17.84 | 19.64 | 20.69 | 19.95 | 21.42 | 23.52 |
| CaO | 12.25 | 10.16 | 9.65 | 9.33 | 8.92 | 8.81 | 7.97 | 6.47 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 1.53 | 1.33 | 1.31 | 0.54 | 1.01 | 0.97 | 0.96 | 0.31 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.56 | 0.52 | 0.45 | 0.38 | 0.46 | 0.41 | 0.36 | 0.27 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | 0.09 | 0.07 | 0.07 | 0.07 | 0.06 | 0.06 | 0.06 | 0.05 |
| S | 0.11 |  |  |  |  |  |  |  |
| LOI | 1.4 | 1.55 | 1.39 | 1.57 | 1.64 | 1.92 | 2.23 | 5.78 |
| Total | 100.3 | 101.02 | 99.55 | 100 | 100.06 | 100.01 | 100.05 | 101.09 |
| Cs | 1.32 | 0.91 | 0.81 | 0.55 | 0.59 | 0.79 | 0.77 | 1.21 |
| K | 4621.83 | 5214.31 | 4542.21 | 3751.91 | 4581.02 | 4140.43 | 3627.98 | 2745.24 |
| Rb | 4 | 20.4 | 17 | 10.4 | 15.3 | 16.1 | 13.6 | 10.3 |
| Ba | 133 | 96.5 | 58.5 | 95.9 | 131.1 | 68.7 | 53.2 | 53.3 |
| Th | 0.6 | 1.4 | 1.36 | 0.87 | 1.17 | 1.31 | 1.23 | 0.69 |
| U | 0.21 | 0.27 | 0.26 | 0.2 | 0.24 | 0.24 | 0.26 | 0.14 |
| Nb | 4.6 | 3.63 | 3.5 | 2.21 | 2.78 | 2.97 | 3.08 | 1.97 |
| Ta | <0.3 | 0.24 | 0.25 | 0.17 | 0.21 | 0.21 | 0.21 | 0.13 |
| La | 6.8 | 6.61 | 6.39 | 4.53 | 5.66 | 5.57 | 5.71 | 3.48 |
| Ce | 16.3 | 15.34 | 14.3 | 10.45 | 12.96 | 12.63 | 12.93 | 8.01 |
| Pr | 2.5 | 2.09 | 1.93 | 1.41 | 1.76 | 1.76 | 1.72 | 1.13 |
| Pb | 1.51 | 2.83 | 3.09 | 1.71 | 3.27 | 4.38 | 2.98 | 2.19 |
| P | 381.63 | 729.04 | 679.89 | 651.94 | 608.74 | 627.9 | 600.75 | 522.92 |
| Sr | 140 | 143.9 | 106.2 | 104.8 | 110.1 | 101.5 | 98.1 | 60.1 |
| Nd | 10.7 | 9.09 | 8.55 | 6.16 | 7.67 | 7.45 | 7.32 | 4.95 |
| Sm | 2.7 | 2.38 | 2.15 | 1.68 | 1.96 | 2.07 | 1.86 | 1.3 |
| Zr | 72 | 65.8 | 67.7 | 49.6 | 51.4 | 81. | 62.8 | 49.5 |
| Hf | 1.87 | 1.77 | 1.72 | 1.05 | 1.39 | 1.9 | 1.52 | 0.96 |
| Eu | 0.87 | 0.77 | 0.74 | 0.55 | 0.65 | 0.63 | 0.62 | 0.44 |
| Gd | 3.18 | 2.65 | 2.48 | 1.79 | 2.19 | 2.16 | 2.06 | 1.39 |
| Tb | 0.54 | 0.43 | 0.43 | 0.3 | 0.36 | 0.34 | 0.34 | 0.25 |
| Dy | 3.2 | 2.83 | 2.73 | 2.01 | 2.32 | 2.35 | 2.3 | 1.6 |
| Y | 19 | 17.6 | 15.9 | 13.9 | 14.1 | 12.5 | 13.5 | 8.8 |
| Ho | 0.68 | 0.5 | 0.5 | 0.36 | 0.43 | 0.43 | 0.44 | 0.28 |
| Er | 1.84 | 1.55 | 1.45 | 1.03 | 1.26 | 1.27 | 1.27 | 0.83 |
| Tm | 0.26 | 0.24 | 0.24 | 0.17 | 0.21 | 0.2 | 0.19 | 0.14 |
| Yb | 1.75 | 1.53 | 1.46 | 1.05 | 1.25 | 1.22 | 1.25 | 0.87 |
| Lu | 0.26 | 0.24 | 0.23 | 0.17 | 0.2 | 0.19 | 0.19 | 0.14 |
| Co | 45 | 66.5 | 61.8 | 60.1 | 62.8 | 63.2 | 65.8 | 64.5 |
| Cr | 660 | 788.6 | 1030.2 | 1096.2 | 1364.4 | 1345.9 | 1314.8 | 1490 |
| Cu | 133 | 93.5 | 89.7 | 83 | 80.7 | 104 | 91.1 | 64.9 |
| Ni | 219 | 325.4 | 403.4 | 451.2 | 723.2 | 453.8 | 536 | 640.9 |
| Sc | 38.9 | 37.3 | 32.1 | 33 | 31.3 | 30.6 | 29.5 | 25.5 |
| V | 299 | 291.1 | 229.8 | 230.1 | 224.7 | 214.8 | 230.6 | 180.7 |
| Zn | 43 | 71.6 | 76.9 | 76.4 | 77.1 | 82 | 76.3 | 58.9 |


| Sample | $\begin{gathered} \text { 10BH167 } \\ \text { A3 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A4 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 11BH311 } \\ \text { B3 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 11BH311 } \\ \text { B4 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A5 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A6 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A7 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A8 } \\ \hline \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | MOZ | UOZ | UOZ | CPZ | CPZ | DZ | DZ | DZ |
| $\mathrm{SiO}_{2}$ | 41.61 | 42.72 | 43.95 | 49.94 | 48.63 | 47.85 | 47.45 | 48.1 |
| $\mathrm{TiO}_{2}$ | 0.55 | 0.69 | 0.64 | 0.9 | 0.94 | 1.02 | 1.05 | 0.96 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 6.56 | 8.07 | 7.9 | 10.55 | 12.08 | 13.59 | 14.12 | 14.39 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{~T}$ | 14.14 | 13.99 | 14.43 | 10.11 | 11 | 11.96 | 12.03 | 10.71 |
| FeOT | 12.71 | 12.58 | 12.97 | 9.09 | 9.89 | 10.75 | 10.81 | 9.63 |
| MnO | 0.2 | 0.19 | 0.2 | 0.17 | 0.16 | 0.19 | 0.17 | 0.17 |
| MgO | 25.43 | 21.83 | 20.52 | 11.45 | 10.56 | 8.63 | 8.58 | 8.63 |
| CaO | 6.21 | 8.17 | 8.36 | 14.11 | 12.09 | 11.65 | 11.54 | 10.82 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 0.4 | 0.81 | 0.78 | 1.26 | 1.51 | 1.84 | 1.76 | 1.83 |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.27 | 0.31 | 0.26 | 0.85 | 1.02 | 1.04 | 1.35 | 1.98 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | 0.05 | 0.06 | 0.05 | 0.06 | 0.07 | 0.07 | 0.08 | 0.08 |
| S |  |  |  |  |  |  |  |  |
| LOI | 5.53 | 2.82 | 3.08 | 1.9 | 2.01 | 2.7 | 2.2 | 3.31 |
| Total | 100.95 | 99.67 | 100.17 | 101.31 | 100.08 | 100.56 | 100.33 | 100.98 |
| Cs | 1.32 | 0.77 | 0.71 | 0.68 | 0.77 | 0.8 | 0.7 | 0.58 |
| K | 2698.49 | 3061.85 | 2596.98 | 8474.67 | 10277.55 | 10444.53 | 13523.71 | 19766.51 |
| Rb | 11.8 | 12.7 | 9.4 | 23.8 | 28.1 | 31.4 | 37.4 | 70.8 |
| Ba | 30 | 87.4 | 4.6 | 105.1 | 81.9 | 79.5 | 221.9 | 179.5 |
| Th | 0.79 | 0.95 | 0.81 | 1.25 | 1.31 | 1.42 | 1.56 | 1.45 |
| U | 0.16 | 0.2 | 0.17 | 0.24 | 0.28 | 0.28 | 0.34 | 0.3 |
| Nb | 1.78 | 2.64 | 1.81 | 3.26 | 3.67 | 3.86 | 3.89 | 4.06 |
| Ta | 0.14 | 0.17 | 0.15 | 0.22 | 0.25 | 0.27 | 0.29 | 0.27 |
| La | 3.68 | 4.88 | 3.99 | 5.83 | 6.22 | 6.59 | 7.46 | 6.9 |
| Ce | 8.38 | 11.24 | 9.11 | 13.93 | 14.59 | 15.26 | 17.44 | 16.27 |
| Pr | 1.16 | 1.56 | 1.25 | 1.89 | 2.01 | 2.08 | 2.4 | 2.22 |
| Pb | 1.87 | 2.77 | 1.77 | 1.85 | 1.96 | 2.76 | 2.4 | 2.21 |
| P | 514.89 | 639.41 | 516.62 | 640.06 | 671.02 | 718.88 | 841.73 | 758.71 |
| Sr | 58.7 | 91.2 | 80.4 | 166.6 | 206.2 | 234.9 | 305.3 | 318.7 |
| Nd | 5.14 | 6.75 | 5.49 | 8.42 | 8.82 | 9.11 | 10.39 | 9.66 |
| Sm | 1.43 | 1.85 | 1.39 | 2.29 | 2.35 | 2.5 | 2.74 | 2.46 |
| Zr | 42.9 | 54.9 | 46.9 | 68.8 | 79.9 | 76.2 | 82.4 | 80.7 |
| Hf | 1.02 | 1.33 | 0.94 | 1.68 | 1.76 | 2.05 | 2.22 | 2.01 |
| Eu | 0.45 | 0.59 | 0.49 | 0.72 | 0.79 | 0.97 | 0.89 | 0.85 |
| Gd | 1.46 | 1.96 | 1.64 | 2.49 | 2.6 | 2.74 | 3.01 | 2.71 |
| Tb | 0.25 | 0.31 | 0.26 | 0.45 | 0.44 | 0.48 | 0.53 | 0.46 |
| Dy | 1.68 | 2.14 | 1.86 | 2.88 | 2.9 | 3.19 | 3.25 | 3.11 |
| Y | 8.8 | 11.6 | 10.5 | 18.1 | 18.6 | 18.2 | 21 | 18.9 |
| Ho | 0.3 | 0.39 | 0.34 | 0.54 | 0.57 | 0.59 | 0.59 | 0.57 |
| Er | 0.92 | 1.21 | 0.98 | 1.59 | 1.65 | 1.72 | 1.81 | 1.68 |
| Tm | 0.14 | 0.19 | 0.15 | 0.25 | 0.24 | 0.26 | 0.28 | 0.26 |
| Yb | 0.87 | 1.19 | 1.04 | 1.51 | 1.57 | 1.68 | 1.72 | 1.66 |
| Lu | 0.15 | 0.18 | 0.16 | 0.25 | 0.25 | 0.26 | 0.27 | 0.26 |
| Co | 64.6 | 64.6 | 61.5 | 49.6 | 50.9 | 58.9 | 55.1 | 50.4 |
| Cr | 1591.6 | 1244.9 | 1704.8 | 1945.7 | 1023.8 | 324.9 | 269 | 292.9 |
| Cu | 89 | 92.1 | 64 | 87.7 | 95.8 | 104.6 | 108.4 | 91 |
| Ni | 622.8 | 513 | 562.2 | 124.7 | 80.9 | 56.4 | 87.4 | 68.1 |
| Sc | 24.1 | 30.3 | 31.5 | 56.4 | 49.5 | 42.6 | 41.4 | 40.7 |
| V | 167 | 205.7 | 201 | 363.7 | 342.6 | 346.7 | 343.4 | 321.2 |
| Zn | 119.3 | 123.8 | 55.9 | 54 | 80.3 | 61.4 | 75.5 | 43.1 |


| Sample | $\begin{gathered} \text { 10BH167 } \\ \text { A9 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 11BH311 } \\ \text { B6 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A10 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A10 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A11-1 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A12 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A14 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 11BH311 } \\ \text { B7 } \\ \hline \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | DZ | DZ | DZ | DZ | DZ | DZ | UBZ | UCM |
| $\mathrm{SiO}_{2}$ | 46.59 | 47.88 | 46.62 | 46.62 | 47.95 | 47.15 | 48.61 | 48.34 |
| $\mathrm{TiO}_{2}$ | 1.24 | 1.18 | 1.17 | 1.17 | 1.03 | 1.07 | 1.02 | 1.08 |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 13.36 | 14.24 | 13.51 | 13.51 | 12.88 | 13.07 | 12.87 | 13.08 |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{~T}$ | 13.81 | 12.21 | 11.68 | 11.68 | 12.29 | 12.26 | 12.56 | 12.21 |
| FeOT | 12.41 | 10.98 | 10.5 | 10.5 | 11.05 | 11.02 | 11.29 | 10.97 |
| MnO | 0.18 | 0.18 | 0.17 | 0.17 | 0.2 | 0.18 | 0.19 | 0.19 |
| MgO | 8.01 | 8.46 | 7.92 | 7.92 | 8.85 | 8.81 | 9.18 | 8.68 |
| CaO | 9.81 | 11.26 | 11.29 | 11.29 | 10.26 | 11.03 | 12.06 | 12.04 |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2.4 | 1.78 | 2.01 | 2.01 | 1.67 | 1.52 | 1.6 | 1.63 |
| $\mathrm{K}_{2} \mathrm{O}$ | 1.17 | 1.26 | 1.07 | 1.07 | 1.61 | 1.55 | 0.98 | 1.13 |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | 0.09 | 0.09 | 0.09 | 0.09 | 0.08 | 0.09 | 0.09 | 0.09 |
| S |  |  |  |  |  |  |  |  |
| LOI | 4.09 | 2.2 | 3.52 | 3.52 | 3.13 | 2.61 | 2.04 | 2.39 |
| Total | 100.74 | 100.76 | 99.06 | 99.06 | 99.95 | 99.33 | 101.19 | 100.86 |
| Cs | 0.36 | 1.11 | 0.37 | 0.37 | 0.84 | 1.28 | 1.81 | 0.54 |
| K | 11704.8 | 12584.77 | 10.655 .67 | 10655.67 | 16124.63 | 15465.69 | 9834.86 | 11313.22 |
| Rb | 39.8 | 37.2 | 35.4 | 35.4 | 52.5 | 40.8 | 34.3 | 27.2 |
| Ba | 140.9 | 143.1 | 112.7 | 112.7 | 211.5 | 161.3 | 125.3 | 174.3 |
| Th | 1.63 | 1.72 | 1.85 | 1.85 | 1.56 | 1.43 | 1.69 | 1.7 |
| U | 0.35 | 0.35 | 0.37 | 0.37 | 0.31 | 0.32 | 0.37 | 0.34 |
| Nb | 4.44 | 4 | 4.8 | 4.8 | 4.32 | 3.92 | 3.55 | 4.29 |
| Ta | 0.3 | 0.32 | 0.35 | 0.35 | 0.28 | 0.26 | 0.29 | 0.31 |
| La | 7.55 | 7.94 | 7.97 | 7.97 | 7.64 | 7.53 | 7.92 | 8 |
| Ce | 17.34 | 18.34 | 19.1 | 19.1 | 17.6 | 17.15 | 17.84 | 18.53 |
| Pr | 2.34 | 2.51 | 2.59 | 2.59 | 2.42 | 2.33 | 2.41 | 2.54 |
| Pb | 1.56 | 1.74 | 1.72 | 1.72 | 1.61 | 1.3 | 2.63 | 2.43 |
| P | 896.02 | 853.7 | 896.2 | 896.2 | 836.12 | 890.75 | 854.83 | 861.78 |
| Sr | 240.9 | 261.4 | 247.2 | 247.2 | 224.4 | 232.3 | 231.1 | 226.1 |
| Nd | 10.33 | 10.64 | 11.51 | 11.51 | 10.5 | 10.37 | 10.4 | 10.78 |
| Sm | 2.88 | 2.82 | 2.94 | 2.94 | 2.74 | 2.74 | 2.78 | 2.81 |
| Zr | 81.2 | 81.1 | 92.4 | 92.4 | 88.9 | 73.8 | 77.3 | 75.6 |
| Hf | 2.19 | 2.08 | 2.39 | 2.39 | 2.07 | 2.03 | 1.9 | 2.03 |
| Eu | 1.13 | 1.02 | 1.03 | 1.03 | 0.93 | 0.91 | 0.9 | 0.96 |
| Gd | 2.98 | 3.25 | 3.37 | 3.37 | 3.06 | 2.98 | 2.98 | 3.04 |
| Tb | 0.53 | 0.53 | 0.62 | 0.62 | 0.5 | 0.52 | 0.5 | 0.54 |
| Dy | 3.34 | 3.52 | 3.72 | 3.72 | 3.26 | 3.32 | 3.24 | 3.45 |
| Y | 21 | 21.4 | 21.5 | 21.5 | 18.2 | 18.9 | 18.3 | 19.2 |
| Ho | 0.63 | 0.66 | 0.69 | 0.69 | 0.59 | 0.6 | 0.58 | 0.61 |
| Er | 1.87 | 1.9 | 2 | 2 | 1.8 | 1.79 | 1.75 | 1.84 |
| Tm | 0.3 | 0.29 | 0.31 | 0.31 | 0.28 | 0.28 | 0.28 | 0.29 |
| Yb | 1.89 | 1.8 | 2.04 | 2.04 | 1.85 | 1.73 | 1.76 | 1.81 |
| Lu | 0.29 | 0.3 | 0.31 | 0.31 | 0.28 | 0.29 | 0.27 | 0.28 |
| Co | 66.1 | 57.1 | 57.5 | 57.5 | 54.5 | 55.3 | 58.4 | 52.5 |
| Cr | 153.5 | 266.2 | 211.5 | 211.5 | 147.4 | 172.2 | 615.4 | 609.5 |
| Cu | 102.9 | 108.2 | 152.6 | 152.6 | 96 | 104.1 | 128.1 | 122.4 |
| Ni | 80.7 | 61.5 | 75.3 | 75.3 | 66.1 | 62.5 | 91.3 | 81.4 |
| Sc | 40.4 | 41.3 | 39.9 | 39.9 | 41.1 | 41 | 41.2 | 42.1 |
| V | 407.4 | 372.7 | 346.3 | 346.3 | 298.2 | 311.7 | 328.1 | 317.7 |
| Zn | 48 | 116.8 | 44.1 | 44.1 | 39.2 | 54.5 | 76.4 | 28.3 |


| Sample | $\begin{gathered} \text { 10JB328 } \\ \text { B1 } \\ \hline \end{gathered}$ | Standard 1 JB1-a | Certified JB1-a | Standard 2 NIM-g | Certified NIM-g |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | UCM |  |  |  |  |  |
| $\mathrm{SiO}_{2}$ | 48 | 52.14 | 52.16 | 75.5 | 75.7 |  |
| $\mathrm{TiO}_{2}$ | 1.05 | 1.29 | 1.3 | 0.08 | 0.09 |  |
| $\mathrm{Al}_{2} \mathrm{O}_{3}$ | 12.78 | 14.26 | 14.51 | 12.25 | 12.08 |  |
| $\mathrm{Fe}_{2} \mathrm{O}_{3} \mathrm{~T}$ | 11.33 | 9.27 | 9.1 | 2.23 | 2.02 |  |
| FeOT | 10.18 |  |  |  |  |  |
| MnO | 0.15 | 0.15 | 0.15 | 0.02 | 0.02 |  |
| MgO | 10.32 | 7.92 | 7.75 | 0.03 | 0.06 |  |
| CaO | 10.16 | 9.28 | 9.23 | 0.86 | 0.78 |  |
| $\mathrm{Na}_{2} \mathrm{O}$ | 2.45 | 2.7 | 2.74 | 3.32 | 3.36 |  |
| $\mathrm{K}_{2} \mathrm{O}$ | 0.5 | 1.43 | 1.42 | 5.08 | 4.99 |  |
| $\mathrm{P}_{2} \mathrm{O}_{5}$ | 0.09 | 0.26 | 0.27 | 0 | 0.01 |  |
| S | 0.08 |  |  |  |  |  |
| LOI | 2.7 | 0.78 | 0.78 | 0.55 | 0.51 |  |
| Total | 99.8 | 99.47 | 99.39 | 99.93 | 99.62 |  |
|  |  |  |  |  |  |  |
| Cs | 0.6 | 1.10 | 1.2 | 0.76 | 1 |  |
| K | 4149.85 |  |  |  |  |  |
| Rb | 13.3 | 40.6 | 41 | 313.8 | 320 |  |
| Ba | 109 | 502.7 | 497 | 113.6 | 120 |  |
| Th | 0.7 | 8.74 | 8.8 | 48.97 | 50 |  |
| U | 0.2 | 1.61 | 1.6 | 15.66 | 15 |  |
| Nb | 4.58 | 26.12 | 27 | 50.36 | 53 |  |
| Ta | 0.3 | 1.63 | 2 | 4.5 | 4.5 |  |
| La | 6.4 | 36.72 | 38.1 | 105.41 | 109 |  |
| Ce | 16.1 | 64.56 | 66.1 | 189.58 | 195 |  |
| Pr | 2.4 | 7.01 | 7.3 | 19.59 |  |  |
| Pb | 1.06 | 7.33 | 7.2 | 28.24 | 40 |  |
| P | 403.1 |  |  |  |  |  |
| Sr | 252 | 450.6 | 443 | 10.6 | 10 |  |
| Nd | 10.5 | 24.23 | 25.5 | 69.92 | 72 |  |
| Sm | 2.8 | 4.93 | 5.02 | 14.95 | 15.8 |  |
| Zr | 65 | 143.9 | 146 | 311.4 | 300 |  |
| Hf | 1.79 | 3.4 | 3.48 | 12.27 | 12 |  |
| Eu | 0.77 | 1.44 | 1.47 | 0.32 | 0.35 |  |
| Gd | 3.11 | 4.82 | 4.54 | 14.52 | 14 |  |
| Tb | 0.52 | 0.7 | 0.69 | 2.68 | 3 |  |
| Dy | 3.2 | 3.98 | 4.19 | 16.73 | 17 |  |
| Y | 19 | 23.9 | 24 | 154.3 | 143 |  |
| Ho | 0.65 | 0.72 | 0.72 | 3.41 |  |  |
| Er | 1.86 | 2.09 | 2.18 | 11.26 |  |  |
| Tm | 0.26 | 0.32 | 0.31 | 1.95 | 2 |  |
| Yb | 1.74 | 1.99 | 2.10 | 13.68 | 14.2 |  |
| Lu | 0.25 | 0.32 | 0.32 | 1.99 | 2 |  |
| Co | 52 | 42.2 | 39.5 | 3.9 | 4 |  |
| Cr | 689 | 421.9 | 415 | 7.0 | 12 |  |
| Cu | 149 | 58.8 | 55 | 16.2 | 12 |  |
| Ni | 201 | 142.3 | 134 | 2.8 | 8 |  |
| Sc | 39.5 | 29.9 | 27.9 | 0.3 | 1 |  |
| V | 306 | 216.2 | 206 | 4.7 | 2 |  |
| Zn | 52 | 84.7 | 82 | 48.2 | 50 |  |

Appendix E2 - LPS Olivine mineral chemistry \& inverse melt compositions

| Sample | 10BH166 <br> A1 | 10BH166 <br> A1 | 10BH166 <br> A1 | 10 BH 166 <br> A1 | 10BH166 <br> A1 | 10BH166 <br> A1 | 10BH166 <br> A1 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | LCM | LCM | LCM | LCM | LCM | LCM | LCM |
| X Region | 1 Core | 2 Rimw | 3 Rim | 4 Core | 5 Rimw | 6 Rim | 10 Core |
| $\mathrm{SiO2}$ | 40.861 | 40.531 | 39.982 | 40.411 | 40.316 | 39.750 | 39.613 |
| $\mathrm{TiO2}$ | 0 | 0 | 0 | 0 | 0 | 0 | 0 |
| $\mathrm{Al2O3}$ | 0.082 | 0.078 | 0.060 | 0.102 | 0.084 | 0.035 | 0.038 |
| $\mathrm{Cr2O3}$ | 0.081 | 0.037 | 0.051 | 0.057 | 0.068 | 0.052 | 0.024 |
| FeO | 12.136 | 11.798 | 16.214 | 11.951 | 12.051 | 14.627 | 13.099 |
| MnO | 0.168 | 0.136 | 0.198 | 0.142 | 0.168 | 0.150 | 0.202 |
| MgO | 47.411 | 47.048 | 43.920 | 47.320 | 47.272 | 45.013 | 46.862 |
| CaO | 0.289 | 0.280 | 0.229 | 0.274 | 0.270 | 0.290 | 0.264 |
| NiO | 0.333 | 0.365 | 0.307 | 0.338 | 0.379 | 0.321 | 0.302 |
| Total | 101.361 | 100.273 | 100.961 | 100.595 | 100.608 | 100.238 | 100.404 |
| Fo (mole \%) | 87.440 | 87.664 | 82.839 | 87.587 | 87.485 | 84.577 | 86.441 |
| Melt MgO | 12.909 | 13.054 | 10.131 | 13.004 | 12.938 | 11.137 | 12.275 |
| Melt FeO | 9.077 | 9.020 | 10.319 | 9.040 | 9.066 | 9.834 | 9.336 |
|  |  |  |  |  |  |  |  |


| Sample | $\begin{gathered} \text { 10BH166 } \\ \text { A1 } \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A1 } \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A1 } \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A1 } \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A1 } \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A1 } \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A1 } \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | LCM | LCM | LCM | LCM | LCM | LCM | LCM |
| X Region | 11 Rimw | 12 Rim | 14 Core | 15 Rimw | 16 Rim | 18 Core | 19 Rim |
| SiO 2 | 39.577 | 39.483 | 40.024 | 40.191 | 39.157 | 39.846 | 37.780 |
| TiO 2 | 0 | 0 | 0.001 | 0 | 0 | 0 | 0.011 |
| Al2O3 | 0.059 | 0.036 | 0.047 | 0.017 | 0.059 | 0.036 | 0.032 |
| Cr 2 O 3 | 0.056 | 0.019 | 0.058 | 0.046 | 0.037 | 0.056 | 0.023 |
| FeO | 13.317 | 15.284 | 12.404 | 12.062 | 14.313 | 13.597 | 22.654 |
| MnO | 0.199 | 0.144 | 0.208 | 0.209 | 0.147 | 0.173 | 0.278 |
| MgO | 46.819 | 45.491 | 47.045 | 47.281 | 45.460 | 45.983 | 38.931 |
| CaO | 0.276 | 0.250 | 0.191 | 0.194 | 0.301 | 0.238 | 0.209 |
| NiO | 0.355 | 0.262 | 0.348 | 0.352 | 0.292 | 0.278 | 0.236 |
| Total | 100.658 | 100.969 | 100.325 | 100.352 | 99.766 | 100.207 | 100.153 |
| Fo (mole \%) | 86.235 | 84.137 | 87.111 | 87.477 | 84.985 | 85.768 | 75.384 |
| Melt MgO | 12.146 | 10.877 | 12.699 | 12.933 | 11.381 | 11.857 | 6.425 |
| Melt FeO | 9.390 | 9.955 | 9.162 | 9.068 | 9.723 | 9.513 | 12.654 |


| Sample | $\begin{gathered} \text { 10BH166 } \\ \text { A2 } \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A2 } \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A2 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A2 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A2 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A2 } \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A2 } \\ \hline \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | LBZ | LBZ | LBZ | LBZ | LBZ | LBZ | LBZ |
| X Region | 1 Core | 2 Rimw | 3 Rimw | 4 Rimw | 5 Rim | 6 Core | 7 Rimw |
| SiO 2 | 38.628 | 39.362 | 39.252 | 39.350 | 36.954 | 39.758 | 38.949 |
| TiO 2 | 0 | 0 | 0 | 0.005 | 0 | 0 | 0 |
| Al2O3 | 0.406 | 0.045 | 0.035 | 0.060 | 0.033 | 0.058 | 0.037 |
| Cr2O3 | 0.525 | 0.048 | 0.063 | 0.072 | 0.017 | 0.047 | 0.015 |
| FeO | 13.812 | 13.372 | 13.282 | 13.272 | 25.216 | 14.175 | 16.254 |
| MnO | 0.124 | 0.162 | 0.199 | 0.156 | 0.234 | 0.288 | 0.209 |
| MgO | 44.690 | 46.684 | 46.531 | 46.213 | 35.895 | 46.051 | 44.556 |
| CaO | 1.148 | 0.283 | 0.257 | 0.286 | 0.240 | 0.258 | 0.245 |
| NiO | 0.323 | 0.320 | 0.301 | 0.320 | 0.259 | 0.297 | 0.287 |
| Total | 99.656 | 100.276 | 99.920 | 99.734 | 98.848 | 100.932 | 100.552 |
| Fo (mole \%) | 85.220 | 86.152 | 86.193 | 86.121 | 71.725 | 85.271 | 83.007 |
| Melt MgO | 11.523 | 12.094 | 12.120 | 12.075 | 4.963 | 11.553 | 10.227 |
| Melt FeO | 9.660 | 9.411 | 9.401 | 9.420 | 13.977 | 9.646 | 10.271 |
|  |  |  |  |  |  |  |  |


| Sample | 10BH166 | 10BH166 | 10BH166 | 10 BH 166 | 10BH166 | 10BH166 | 10BH166 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | A2 | A2 | A2 | A2 | A3 | A3 | A3 |
| Zone | LBZ | LBZ | LBZ | LBZ | LBZ | LBZ | LBZ |
| X Region | 8 Rim | 24 Core | 25 Rimw | 26 Rim | 9 Core | 10 Rimw | 11 Rim |
| SiO 2 | 37.195 | 39.989 | 39.625 | 39.787 | 37.520 | 37.341 | 37.640 |
| TiO 2 | 0 | 0 | 0 | 0 | 0.011 | 0 | 0.019 |
| $\mathrm{Al2O} 3$ | 0.038 | 0.046 | 0.168 | 0.041 | 0.061 | 0.032 | 0.048 |


| Cr 2 O 3 | 0.023 | 0.051 | 0.220 | 0.053 | 0.012 | 0 | 0 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| FeO | 25.707 | 14.915 | 15.355 | 18.019 | 23.174 | 22.858 | 24.282 |
| MnO | 0.282 | 0.181 | 0.181 | 0.335 | 0.249 | 0.308 | 0.373 |
| MgO | 36.612 | 45.097 | 44.040 | 43.266 | 38.123 | 37.944 | 36.920 |
| CaO | 0.258 | 0.247 | 0.658 | 0.265 | 0.244 | 0.268 | 0.156 |
| NiO | 0.215 | 0.257 | 0.266 | 0.284 | 0.208 | 0.244 | 0.205 |
| Total | 100.330 | 100.783 | 100.513 | 102.050 | 99.602 | 98.995 | 99.643 |
| Fo (mole \%) | 71.735 | 84.346 | 83.636 | 81.056 | 74.565 | 74.735 | 73.042 |
| Melt MgO | 4.967 | 11.000 | 10.586 | 9.156 | 6.077 | 6.149 | 5.463 |
| Melt FeO | 13.974 | 9.898 | 10.094 | 10.838 | 12.939 | 12.879 | 13.486 |


| Sample | 10BH166 <br> A3 | 10BH166 <br> A3 | 10BH166 <br> A3 | 10 BH 166 <br> A3 | 11 BH311 <br> B2 | 11BH311 <br> B2 | 11BH311 <br> B2 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | LBZ | LBZ | LBZ | LBZ | LOZ | LOZ | LOZ |
| X Region | 21 Core | 22 Rim | 23 Core | 24 Rim | 125 Core | 126 Rim | 128 Core |
| SiO 2 | 37.042 | 36.796 | 37.151 | 37.654 | 38.867 | 38.874 | 39.294 |
| TiO 2 | 0 | 0.029 | 0 | 0.013 | 0.002 | 0.006 | 0.006 |
| $\mathrm{Al2O3}$ | 0.055 | 0 | 0.029 | 0.009 | 0.013 | 0.029 | 0.044 |
| $\mathrm{Cr2O3}$ | 0.013 | 0.015 | 0.025 | 0.019 | 0.027 | 0.003 | 0.023 |
| FeO | 22.935 | 24.553 | 22.542 | 22.521 | 18.680 | 24.400 | 19.286 |
| MnO | 0.238 | 0.347 | 0.399 | 0.297 | 0.226 | 0.299 | 0.343 |
| MgO | 38.177 | 36.293 | 38.059 | 38.421 | 42.022 | 37.602 | 41.754 |
| CaO | 0.278 | 0.202 | 0.261 | 0.188 | 0.256 | 0.233 | 0.229 |
| NiO | 0.229 | 0.176 | 0.206 | 0.258 | 0.247 | 0.169 | 0.216 |
| Total | 98.967 | 98.411 | 98.672 | 99.380 | 100.340 | 101.615 | 101.195 |
| Fo (mole \%) | 74.787 | 72.483 | 75.054 | 75.248 | 80.035 | 73.306 | 79.415 |
| Melt MgO | 6.171 | 5.247 | 6.284 | 6.367 | 8.622 | 5.566 | 8.307 |
| Melt FeO | 12.861 | 13.692 | 12.768 | 12.701 | 11.146 | 13.389 | 11.337 |


| Sample | 11BH311 <br> B2 | 11BH311 <br> B2 | 11BH311 <br> B2 | 10 BH 166 <br> A4 | 10 BH 166 <br> A4 | 10BH166 <br> A4 | 10BH166 <br> A4 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | LOZ | LOZ | LOZ | LOZ | LOZ | LOZ | LOZ |
| X Region | 129 Rim | 133 Core | 134 Rim | 7 Core | 8 Rim | 12 Core | 13 Rim |
| SiO 2 | 38.394 | 38.347 | 37.917 | 38.105 | 38.3 | 39.727 | 39.702 |
| TiO 2 | 0 | 0.005 | 0.009 | 0 | 0.006 | 0 | 0 |
| $\mathrm{Al2O} 3$ | 0.117 | 0.042 | 0.034 | 0.043 | 0.043 | 0.047 | 0.028 |
| $\mathrm{Cr2O3}$ | 0.050 | 0.029 | 0.023 | 0.011 | 0.029 | 0.040 | 0.053 |
| FeO | 21.462 | 17.318 | 23.053 | 17.324 | 17.447 | 16.813 | 16.070 |
| MnO | 0.274 | 0.227 | 0.244 | 0.260 | 0.335 | 0.324 | 0.144 |
| MgO | 39.935 | 43.594 | 38.526 | 42.571 | 42.730 | 43.532 | 44.104 |
| CaO | 0.168 | 0.238 | 0.199 | 0.164 | 0.239 | 0.208 | 0.212 |
| NiO | 0.186 | 0.284 | 0.215 | 0.225 | 0.192 | 0.202 | 0.241 |
| Total | 100.586 | 100.084 | 100.220 | 98.703 | 99.321 | 100.893 | 100.554 |
| Fo (mole \%) | 76.829 | 81.771 | 74.862 | 81.409 | 81.358 | 82.187 | 83.024 |
| Melt MgO | 7.067 | 9.540 | 6.202 | 9.344 | 9.317 | 9.768 | 10.236 |
| Melt FeO | 12.166 | 10.628 | 12.835 | 10.734 | 10.749 | 10.507 | 10.267 |


| Sample | $\begin{gathered} \hline \text { 10BH166 } \\ \text { A4 } \end{gathered}$ | $\begin{gathered} \hline \text { 10BH166 } \\ \text { A4 } \end{gathered}$ | $\begin{gathered} \text { 10BH166 } \\ \text { A4 } \end{gathered}$ | $\begin{gathered} \hline \text { 10BH166 } \\ \text { A4 } \end{gathered}$ | $\begin{gathered} \hline \text { 10BH166 } \\ \text { A4 } \end{gathered}$ | $\begin{gathered} \hline \text { 10BH166 } \\ \text { A4 } \end{gathered}$ | $\begin{gathered} \hline \text { 10BH166 } \\ \text { A4 } \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | LOZ | LOZ | LOZ | LOZ | LOZ | LOZ | LOZ |
| X Region | 14 Core | 15 Rim | 19 Core | 20 Rimw | 21 Rim | 24 Core | 25 Rimw |
| SiO 2 | 39.569 | 39.398 | 38.942 | 38.632 | 38.207 | 40.179 | 40.157 |
| TiO 2 | 0 | 0.005 | 0 | 0.008 | 0 | 0 | 0 |
| Al2O3 | 0.060 | 0.042 | 0.042 | 0.042 | 0.061 | 0.059 | 0.049 |
| Cr 2 O 3 | 0.045 | 0.017 | 0.039 | 0.039 | 0.056 | 0.034 | 0.055 |
| FeO | 16.535 | 16.496 | 16.712 | 16.376 | 22.495 | 12.944 | 14.534 |
| MnO | 0.264 | 0.293 | 0.246 | 0.235 | 0.306 | 0.139 | 0.181 |
| MgO | 43.738 | 43.459 | 43.034 | 43.446 | 38.386 | 46.354 | 45.750 |
| CaO | 0.260 | 0.210 | 0.220 | 0.216 | 0.234 | 0.258 | 0.263 |
| NiO | 0.294 | 0.228 | 0.316 | 0.307 | 0.240 | 0.330 | 0.321 |
| Total | 100.765 | 100.148 | 99.551 | 99.301 | 99.985 | 100.297 | 101.310 |
| Fo (mole \%) | 82.498 | 82.440 | 82.107 | 82.541 | 75.253 | 86.453 | 84.870 |
| Melt MgO | 9.941 | 9.908 | 9.724 | 9.964 | 6.369 | 12.282 | 11.312 |
| Melt FeO | 10.417 | 10.434 | 10.530 | 10.405 | 12.699 | 9.333 | 9.754 |


| Sample | 10BH166 <br> A4 | 10BH166 <br> A5 | 10BH166 <br> A5 | 10 BH 166 <br> A5 | 10 BH 166 <br> A5 | 10BH166 <br> A5 | 10BH166 <br> A5 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | LOZ | LOZ | LOZ | LOZ | LOZ | LOZ | LOZ |
| X Region | 26 Rim | 6 Core | 7 Rimw | 8 Rim | 15 Core | 16 Rimw | 17 Rimw |
| $\mathrm{SiO2}$ | 38.471 | 38.269 | 38.451 | 38.270 | 39.837 | 39.770 | 39.391 |
| $\mathrm{TiO2}$ | 0 | 0 | 0 | 0.013 | 0 | 0 | 0 |
| $\mathrm{Al2O3}$ | 0.075 | 0.053 | 0.044 | 0.052 | 0.055 | 0.060 | 0.048 |
| $\mathrm{Cr2O3}$ | 0.070 | 0.019 | 0.061 | 0.048 | 0.038 | 0.074 | 0.078 |
| FeO | 20.387 | 21.408 | 21.632 | 21.137 | 11.500 | 11.636 | 11.290 |
| MnO | 0.322 | 0.315 | 0.301 | 0.307 | 0.185 | 0.283 | 0.177 |
| MgO | 40.140 | 39.211 | 39.637 | 39.623 | 47.678 | 47.432 | 47.638 |
| CaO | 0.198 | 0.279 | 0.288 | 0.208 | 0.302 | 0.286 | 0.251 |
| NiO | 0.293 | 0.215 | 0.194 | 0.184 | 0.376 | 0.405 | 0.358 |
| Total | 99.956 | 99.769 | 100.608 | 99.842 | 99.971 | 99.946 | 99.231 |
| Fo (mole \%) | 77.820 | 76.547 | 76.554 | 76.961 | 88.078 | 87.899 | 88.262 |
| Melt MgO | 7.528 | 6.939 | 6.942 | 7.127 | 13.324 | 13.207 | 13.445 |
| Melt FeO | 11.842 | 12.260 | 12.257 | 12.122 | 8.915 | 8.960 | 8.869 |


| Sample | 10BH166 <br> A5 | 10BH166 <br> A5 | 10BH167 <br> A1 | 10 BH 167 <br> A1 | 10 BH 167 <br> A1 | 10BH167 <br> A1 | 10BH167 <br> A1 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | LOZ | LOZ | LOZ | LOZ | LOZ | LOZ | LOZ |
| X Region | 18 Rimw | 19 Rim | 7 Core | 8 Rim | 16 Core | 17 Rim | 9 Core |
| SiO2 | 38.929 | 37.659 | 39.760 | 38.484 | 39.768 | 38.327 | 39.071 |
| $\mathrm{TiO2}$ | 0.007 | 0 | 0.002 | 0 | 0 | 0.009 | 0 |
| $\mathrm{Al2O3}$ | 0.049 | 0.032 | 0.153 | 0.024 | 0.057 | 0.026 | 0.036 |
| $\mathrm{Cr2O3}$ | 0.046 | 0.006 | 0.221 | 0 | 0.054 | 0.019 | 0.037 |
| FeO | 14.322 | 19.870 | 16.317 | 22.165 | 14.457 | 21.130 | 15.603 |
| MnO | 0.245 | 0.328 | 0.163 | 0.258 | 0.253 | 0.261 | 0.192 |
| MgO | 45.491 | 40.880 | 43.403 | 38.590 | 45.191 | 40.147 | 44.162 |
| CaO | 0.268 | 0.206 | 0.482 | 0.232 | 0.260 | 0.225 | 0.250 |
| NiO | 0.324 | 0.213 | 0.297 | 0.266 | 0.298 | 0.229 | 0.245 |
| Total | 99.681 | 99.194 | 100.798 | 100.019 | 100.338 | 100.373 | 99.596 |
| Fo (mole \%) | 84.985 | 78.569 | 82.579 | 75.625 | 84.780 | 77.199 | 83.454 |
| Melt MgO | 11.381 | 7.888 | 9.985 | 6.529 | 10.481 | 10.483 | 11.258 |
| Melt FeO | 9.723 | 11.602 | 10.394 | 12.571 | 10.145 | 10.144 | 9.779 |


| Sample | $\begin{gathered} \text { 10BH167 } \\ \text { A1 } \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A2 } \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A2 } \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A2 } \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A2 } \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A2 } \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A2 } \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | LOZ | MOZ | MOZ | MOZ | MOZ | MOZ | MOZ |
| X Region | 10 Rim | 1 Core | 2 Rimw | 3 Rim | 12 Core | 13 Rim | 14 Core |
| SiO 2 | 39.642 | 38.633 | 39.057 | 39.102 | 38.424 | 38.946 | 38.273 |
| TiO 2 | 0.003 | 0 | 0 | 0 | 0 | 0 | 0 |
| $\mathrm{Al2O} 3$ | 0.073 | 0.037 | 0.037 | 0.030 | 0.043 | 0.036 | 0.057 |
| Cr 2 O 3 | 0.015 | 0.063 | 0 | 0 | 0.029 | 0.012 | 0.025 |
| FeO | 15.597 | 17.164 | 18.997 | 20.044 | 16.414 | 16.292 | 17.634 |
| MnO | 0.198 | 0.232 | 0.217 | 0.307 | 0.163 | 0.169 | 0.203 |
| MgO | 44.156 | 42.727 | 41.648 | 41.064 | 43.322 | 42.907 | 41.625 |
| CaO | 0.167 | 0.297 | 0.283 | 0.202 | 0.264 | 0.221 | 0.292 |
| NiO | 0.205 | 0.226 | 0.248 | 0.252 | 0.262 | 0.281 | 0.212 |
| Total | 100.056 | 99.379 | 100.487 | 101.001 | 98.921 | 98.864 | 98.321 |
| Fo (mole \%) | 83.457 | 81.604 | 79.620 | 78.498 | 82.466 | 82.435 | 80.793 |
| Melt MgO | 7.237 | 9.449 | 8.410 | 7.854 | 9.923 | 9.905 | 9.016 |
| Melt FeO | 12.044 | 10.677 | 11.274 | 11.625 | 10.426 | 10.435 | 10.917 |


| Sample | 10BH167 <br> A2 | 10BH167 <br> A3 | 10 BH 167 A <br> 3 | 10 BH 167 <br> A 3 | 10 BH 167 <br> A 3 | 10 BH 167 <br> A 3 | 10BH167 <br> A 3 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | MOZ | MOZ | MOZ | MOZ | MOZ | MOZ | MOZ |
| X Region | 15 Rim | 1 Core | 2 Rim | 8 Core | 9 Rim | 18 Core | 19 Rim |
| SiO 2 | 38.899 | 38.860 | 38.636 | 39.167 | 38.994 | 38.182 | 38.351 |
| TiO 2 | 0 | 0.002 | 0.003 | 0 | 0 | 0 | 0 |
| $\mathrm{Al2O} 3$ | 0.046 | 0.033 | 0.029 | 0.044 | 0.035 | 0.036 | 0.027 |
| $\mathrm{Cr2O3}$ | 0.050 | 0.035 | 0.054 | 0.023 | 0.039 | 0.013 | 0.026 |
| FeO | 17.921 | 19.525 | 20.328 | 18.111 | 17.074 | 19.481 | 18.845 |


| MnO | 0.237 | 0.225 | 0.270 | 0.169 | 0.197 | 0.262 | 0.248 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| MgO | 42.645 | 41.049 | 40.454 | 42.396 | 42.600 | 40.425 | 41.748 |
| CaO | 0.293 | 0.237 | 0.199 | 0.297 | 0.278 | 0.258 | 0.244 |
| NiO | 0.235 | 0.226 | 0.226 | 0.244 | 0.269 | 0.185 | 0.188 |
| Total | 100.326 | 100.192 | 100.199 | 100.451 | 99.486 | 98.842 | 99.677 |
| Fo (mole \%) | 80.918 | 78.932 | 78.004 | 80.663 | 81.638 | 78.714 | 79.789 |
| Melt MgO | 9.082 | 8.066 | 7.616 | 8.948 | 9.468 | 7.959 | 8.496 |
| Melt FeO | 10.880 | 11.488 | 11.783 | 10.956 | 10.667 | 11.557 | 11.222 |


| Sample | $\begin{gathered} \text { 10BH167 } \\ \text { A3 } \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A3 } \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A4 } \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A4 } \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A4 } \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A4 } \end{gathered}$ | $\begin{gathered} \text { 10BH167 } \\ \text { A4 } \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | MOZ | MOZ | UOZ | UOZ | UOZ | UOZ | UOZ |
| X Region | 23 Rep | 24 Rep | 4 Core | 5 Rimw | 6 Rim | 7 Rimw | 8 Rim |
| SiO 2 | 38.887 | 38.251 | 38.754 | 37.594 | 38.332 | 38.105 | 37.382 |
| TiO 2 | 0 | 0 | 0 | 0.023 | 0 | 0.002 | 0.004 |
| Al2O3 | 0.048 | 0.032 | 0 | 0.048 | 0.042 | 0.028 | 0.019 |
| Cr 2 O 3 | 0.040 | 0.025 | 0.034 | 0 | 0.026 | 0.025 | 0.004 |
| FeO | 17.318 | 17.877 | 20.306 | 24.617 | 21.751 | 23.537 | 27.056 |
| MnO | 0.232 | 0.240 | 0.247 | 0.316 | 0.317 | 0.342 | 0.362 |
| MgO | 43.492 | 42.671 | 40.352 | 36.728 | 39.444 | 37.548 | 34.962 |
| CaO | 0.225 | 0.230 | 0.285 | 0.151 | 0.244 | 0.251 | 0.133 |
| NiO | 0.249 | 0.235 | 0.197 | 0.165 | 0.211 | 0.134 | 0.167 |
| Total | 100.491 | 99.561 | 100.175 | 99.642 | 100.367 | 99.972 | 100.089 |
| Fo (mole \%) | 81.736 | 80.965 | 77.979 | 72.668 | 76.368 | 73.977 | 69.722 |
| Melt MgO | 9.521 | 9.107 | 7.604 | 5.318 | 6.858 | 5.836 | 4.260 |
| Melt FeO | 10.638 | 10.866 | 11.791 | 13.623 | 12.320 | 13.147 | 14.761 |


| Sample | 10BH167 <br> A4 | 10BH167 <br> A4 | 10 BH 167 <br> A4 | 10 BH 167 <br> A4 | 11 BH 311 <br> B3 | 11BH311 <br> B3 | 11BH311 <br> B3 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | UOZ | UOZ | UOZ | UOZ | UOZ | UOZ | UOZ |
| X Region | 19 Core | 20 Rim | 26 Core | 27 Rim | 13 Rim | 14 Rimw | 15 Rimw |
| SiO2 | 38.600 | 38.423 | 37.243 | 38.144 | 39.378 | 38.573 | 38.540 |
| TiO 2 | 0.001 | 0 | 0.011 | 0.006 | 0 | 0.002 | 0 |
| $\mathrm{Al2O3}$ | 0.051 | 0.032 | 0.030 | 0.037 | 0.037 | 0.026 | 0.015 |
| $\mathrm{Cr2O3}$ | 0.044 | 0.028 | 0.025 | 0.039 | 0.021 | 0.038 | 0.004 |
| FeO | 19.578 | 24.323 | 22.739 | 19.862 | 20.973 | 21.200 | 21.868 |
| MnO | 0.310 | 0.293 | 0.421 | 0.296 | 0.314 | 0.369 | 0.293 |
| MgO | 41.471 | 37.350 | 38.558 | 41.324 | 39.894 | 39.475 | 39.003 |
| CaO | 0.281 | 0.265 | 0.222 | 0.282 | 0.239 | 0.251 | 0.231 |
| NiO | 0.185 | 0.184 | 0.190 | 0.227 | 0.222 | 0.225 | 0.214 |
| Total | 100.521 | 100.898 | 99.439 | 100.217 | 101.078 | 100.159 | 100.168 |
| Fo (mole \%) | 79.056 | 73.236 | 75.135 | 78.757 | 77.219 | 76.842 | 76.067 |
| Melt MgO | 8.128 | 5.539 | 6.318 | 7.980 | 7.246 | 7.073 | 6.724 |
| Melt FeO | 11.449 | 13.415 | 12.740 | 11.543 | 12.037 | 12.162 | 12.421 |


| Sample | $\begin{gathered} \text { 11BH311B } \\ 3 \end{gathered}$ | $\begin{gathered} \text { 11BH311 } \\ \text { B3 } \end{gathered}$ | $\begin{gathered} \text { 11BH311B } \\ 3 \end{gathered}$ | $\begin{gathered} \hline 11 \mathrm{BH} 311 \\ \text { B3 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 11BH311 } \\ \text { B3 } \end{gathered}$ | $\begin{gathered} \hline 11 \mathrm{BH} 311 \\ \text { B3 } \\ \hline \end{gathered}$ | $\begin{gathered} \hline \text { 11BH311 } \\ \text { B3 } \\ \hline \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | UOZ | UOZ | UOZ | UOZ | UOZ | UOZ | UOZ |
| X Region | 16 Core | 17 Rimw | 19 Rim | 20 Rim | 21 Core | 22 Rimw | 23 Rim |
| SiO 2 | 38.012 | 37.842 | 38.028 | 38.343 | 38.179 | 38.242 | 38.970 |
| TiO 2 | 0.012 | 0.002 | 0 | 0.010 | 0 | 0 | 0 |
| Al2O3 | 0.017 | 0.015 | 0.041 | 0.019 | 0.040 | 0.039 | 0.021 |
| Cr 2 O 3 | 0 | 0.001 | 0.019 | 0 | 0.039 | 0.022 | 0.014 |
| FeO | 22.844 | 22.459 | 22.573 | 22.059 | 21.714 | 21.053 | 20.908 |
| MnO | 0.330 | 0.321 | 0.298 | 0.342 | 0.342 | 0.299 | 0.314 |
| MgO | 38.685 | 38.594 | 38.559 | 39.216 | 39.331 | 39.527 | 39.619 |
| CaO | 0.213 | 0.218 | 0.174 | 0.144 | 0.226 | 0.228 | 0.247 |
| NiO | 0.193 | 0.133 | 0.229 | 0.206 | 0.186 | 0.206 | 0.195 |
| Total | 100.306 | 99.585 | 99.921 | 100.339 | 100.057 | 99.616 | 100.288 |
| Fo (mole \%) | 75.110 | 75.383 | 75.272 | 76.008 | 76.347 | 76.989 | 77.152 |
| Melt MgO | 6.308 | 6.425 | 6.377 | 6.698 | 6.849 | 7.140 | 7.215 |
| Melt FeO | 12.748 | 12.654 | 12.692 | 12.441 | 12.327 | 12.113 | 12.059 |


| Sample | $\begin{gathered} \text { 11BH311B } \\ 3 \end{gathered}$ | $\begin{gathered} \text { 11BH311 } \\ \text { B3 } \end{gathered}$ | $\begin{gathered} \text { 11BH311B } \\ 3 \end{gathered}$ | $\begin{gathered} \text { 11BH311 } \\ \text { B3 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 11BH311 } \\ \text { B3 } \\ \hline \end{gathered}$ | $\begin{gathered} \hline \text { 11BH311 } \\ \text { B3 } \\ \hline \end{gathered}$ | $\begin{gathered} \text { 11BH311 } \\ \text { B3 } \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zone | UOZ | UOZ | UOZ | UOZ | UOZ | UOZ | UOZ |
| X Region | 25 Core | 26 Rimw | 27 Rim | 48 Core | 49 Rim | 50 Core | 51 Rim |
| SiO 2 | 39.041 | 38.529 | 38.218 | 39.003 | 39.194 | 39.677 | 39.689 |
| TiO2 | 0 | 0.004 | 0.018 | 0.016 | 0 | 0 | 0 |
| Al2O3 | 0.035 | 0.044 | 0.020 | 0.034 | 1.097 | 0.045 | 0.030 |
| Cr 2 O 3 | 0.009 | 0.035 | 0.048 | 0.034 | 0.015 | 0.062 | 0.046 |
| FeO | 23.508 | 23.240 | 23.614 | 20.830 | 18.951 | 20.215 | 18.997 |
| MnO | 0.341 | 0.374 | 0.295 | 0.311 | 0.271 | 0.276 | 0.344 |
| MgO | 37.933 | 38.231 | 38.350 | 40.556 | 41.327 | 40.522 | 41.601 |
| CaO | 0.215 | 0.223 | 0.219 | 0.249 | 0.244 | 0.263 | 0.239 |
| NiO | 0.161 | 0.174 | 0.209 | 0.197 | 0.215 | 0.211 | 0.184 |
| Total | 101.243 | 100.854 | 100.991 | 101.230 | 101.314 | 101.271 | 101.130 |
| Fo (mole \%) | 74.197 | 74.564 | 74.320 | 77.626 | 79.534 | 78.128 | 79.602 |
| Melt MgO | 5.925 | 6.077 | 5.976 | 7.437 | 8.367 | 7.675 | 8.401 |
| Melt FeO | 13.069 | 12.939 | 13.025 | 11.905 | 11.300 | 11.743 | 11.279 |


| Sample | 11 BH 311 B <br> 3 | 11BH311 <br> B3 | 11BH311B <br> 3 | 11 BH 311 <br> B3 |
| :--- | :---: | :---: | :---: | :---: |
| Zone | UOZ | UOZ | UOZ | UOZ |
| X Region | 57 Core | 58 Rimw | 59 Rimw | 60 Rim |
| SiO 2 | 38.742 | 38.899 | 39.347 | 39.585 |
| TiO 2 | 0 | 0.007 | 0 | 0.012 |
| $\mathrm{Al2O3}$ | 0.052 | 0.042 | 0.033 | 0.021 |
| $\mathrm{Cr2O3}$ | 0 | 0.046 | 0.012 | 0.024 |
| FeO | 20.701 | 19.642 | 19.391 | 18.848 |
| MnO | 0.311 | 0.282 | 0.236 | 0.206 |
| MgO | 39.971 | 41.107 | 41.620 | 41.558 |
| CaO | 0.242 | 0.260 | 0.264 | 0.172 |
| NiO | 0.215 | 0.161 | 0.187 | 0.230 |
| Total | 100.234 | 100.446 | 101.090 | 100.656 |
| Fo (mole \%) | 77.482 | 78.856 | 79.274 | 79.712 |
| Melt MgO | 7.369 | 8.029 | 8.236 | 8.457 |
| Melt FeO | 11.952 | 11.512 | 11.381 | 11.245 |

Appendix E1-A - Point counting results and error


Error - 1 standard deviation

|  |  | 1sig- <br> O1 | 1sig- <br> Plag | 1sig- <br> $\mathbf{C p x}$ | 1sig-Fe- <br> $\mathbf{T i}$ | 1sig- <br> Gm |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| 166A1 | LCM | 1.414 | 0.000 | 0.000 | 1.414 | 2.828 |
| 166A2 | LBZ | 2.828 | 0.000 | 0.000 | 1.414 | 4.243 |
| 166A3 | OZ | 2.828 | 0.707 | 0.707 | 2.828 | 0.000 |


| 166A4 | OZ | 1.414 | 0.707 | 0.707 | 1.414 | 0.000 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| 166A5 | OZ | 0.707 | 2.121 | 0.707 | 0.707 | 0.000 |
| 167A1 | OZ | 2.121 | 2.828 | 0.707 | 1.414 | 0.000 |
| 167A2 | OZ | 2.828 | 0.000 | 1.414 | 1.414 | 0.000 |
| 167A3 | OZ | 2.121 | 4.243 | 3.536 | 1.414 | 0.000 |
| 167A4 | OZ | 2.828 | 0.707 | 3.536 | 0.000 | 0.000 |
| 167A5 | CPZ | 0.707 | 1.414 | 2.121 | 1.414 | 0.000 |
| 167A6 | DZ | 0.000 | 2.828 | 0.707 | 2.121 | 0.000 |
| 167A7 | DZ | 0.000 | 1.414 | 2.828 | 4.243 | 0.000 |
| 167A8 | DZ | 0.000 | 0.707 | 1.414 | 0.707 | 0.000 |
| 167A9 | DZ | 0.000 | 0.707 | 2.121 | 1.414 | 0.000 |
| 167A10 | DZ | 0.000 | 0.707 | 1.414 | 0.707 | 0.000 |
| 167A11 | DZ | 0.000 | 0.000 | 1.414 | 1.414 | 0.000 |
| 167A12 | DZ | 0.000 | 0.707 | 0.000 | 0.707 | 0.000 |
| 167A14 | UBZ | 0.000 | 0.000 | 0.000 | 0.707 | 0.707 |
| 167A15 | UCM | 2.121 | 0.000 | 0.000 | 0.000 | 2.121 |

## Appendix E1-B - LPS whole-rock plots and descriptions

## Major Elements

SiO2: Consistently around $45 \mathrm{wt} \%$ through the sill, with LCM and UCM showing good similarities. There is a slight increase in SiO 2 between the LCM and the LBZ. There is a slight trend of decreasing SiO 2 from the LBZ through the OZ . The highest SiO 2 values preserved in the sill are within the CPZ. There is very little variation in SiO 2 through the DZ .

TiO2: Overall, a Z-style profile. Both LCM and the UCM show a reasonable fit with a slightly higher TiO 2 value in the UCM. There is a decreasing trend in TiO 2 from the LCM, through the LBZ and OZ. The UOZ (Upper OZ) shows an increase in TiO2 leading towards the CPZ. There is a sudden jump in TiO 2 in the CPZ from 0.65 to $0.95 \mathrm{wt} \%$ (values similar to the LCM). Then there appears to be two cycles in TiO 2 in the DZ , with higher values in the upper cycle, which decrease from 1.2 to $1 \mathrm{wt} \%$. The UBZ and UCM have values around 1.05 $\mathrm{wt} \%$ compared within $0.9 \mathrm{wt} \%$ at the LCM.

Al2O3: Overall, a Z-style profile. Values decrease from the LCM up through the LBZ and OZ . They begin to increase in the roof of the OZ (much like TiO2) with a jump up to $8 \mathrm{wt} \%$ from $6.5 \mathrm{wt} \%$ and then another jump to $12 \mathrm{wt} \%$ in the CPZ. The highest values are in the DZ, typically around $14 \mathrm{wt} \%$. They steadily decrease towards the UBZ/UCM ( $\sim 13 \mathrm{wt} \%$ ) but these are slightly higher than at the LCM.

Fe 2 O 3 (total iron expressed as Fe 2 O 3 ): There appears to be two separate trends for Fe 2 O 3. The LCM value is $13 \mathrm{wt} \%$, which increases slightly into the LBZ, through the OZ. Midway through the OZ, the value jumps to $15 \mathrm{wt} \%$ and then back to $14 \mathrm{wt} \%$ for the upper part of the OZ. Then the second trend starts within the CPZ at values of $10 \mathrm{wt} \%$. Values increase back up to $\sim 14 \mathrm{wt} \%$ in the DZ, but are consistently $\sim 12 \mathrm{wt} \%$ through the DZ and the UBZ/UCM (which incidentally are lower than the LCM/LBZ).

MnO : The MnO profile is similar in appearance to the Fe 2 O 3 profile with two trends visible. MnO increases from the $\mathrm{LCM}(0.18 \mathrm{wt} \%)$ to the LBZ ( $0.19 \mathrm{wt} \%$ ) and is constantly around 0.19-0.2 wt \% for the OZ. There is a decrease at the CPZ from $0.19 \mathrm{wt} \%$ to $0.16 \mathrm{wt} \% \mathrm{MnO}$ values are around $0.15-0.17 \mathrm{wt} \%$ for the DZ . The UBZ and LBZ are very similar. The UCM contains slightly less MnO than the LCM.

MgO : Almost shows a 'traditional' S-shaped curve, but there isn't a big enough increase in the UBZ and UCM to create symmetry with the LCM-LBZ. The LCM value (frozen liquid) is $13 \mathrm{wt} \%$, with an increase through the LBZ ( $15 \mathrm{to} 17 \mathrm{wt} \%$ ), and an increase through the OZ to a highest value of $25 \mathrm{wt} \%$. It decreases to around $21 \mathrm{wt} \%$ in the UOZ and then shows a big decrease from 21 to $11 \mathrm{wt} \%$ in the CPZ. The value drops again, to < $10 \mathrm{wt} \%$ and remains consistently so through the DZ. The UBZ replicates this, and the UCM shows a slight decrease to values $>10 \mathrm{wt} \%$ - similar to the LCM.

CaO : Appears to show two broad trends. CaO decreases from $11 \mathrm{wt} \%$ in the LCM, through the LBZ to $6 \mathrm{wt} \%$ at the middle $\mathrm{OZ}(\mathrm{MOZ})$. There is a slight increase in CaO in the UOZ to $8-9 \mathrm{wt} \%$. There is a sudden increase at the CPZ to $14-15 \mathrm{wt} \%$, and then the value is consistently $\sim 11 \mathrm{wt} \%$ in the DZ, similar to the UBZ. The UCM shows good fit with the LCM, with $\sim 11 \mathrm{wt} \% \mathrm{CaO}$.

Na2O: Na2O shows a Z-style curve but with much higher values at the UBZ, and especially the UCM. There is a decrease in Na 2 O from the $\mathrm{LCM}(1.55 \mathrm{wt} \%)$ through the LBZ to the
lower olivine zone (LOZ; values of $0.5 \mathrm{wt} \%$ ). There is an increase in Na 2 O just above the LOZ to values of 1 , followed by a decrease to values of $\sim 0.35 \mathrm{wt} \%$, followed by an increase to $0.75 \mathrm{wt} \%$ in the UOZ. Values increase to $1.4-1.5 \mathrm{wt} \%$ in the CPZ as well as in the lower DZ (1.8-1.9 $\mathrm{wt} \%$ ). There is a decreasing trend in the DZ, including some peaks (up to $\sim 2.5$ $\mathrm{wt} \%$ ). The UBZ ( $1.6 \mathrm{wt} \%$ ) is similar to the LCM, and the UCM shows values ( $2.35 \mathrm{wt} \%$ ) similar to the DZ, as opposed to the LCM.

K2O: Again, two 'cycles' are visible. There is a slight decrease in K2O from the LCM (0.50 $\mathrm{wt} \%$ ) up through the LBZ and OZ (to as low as $\sim 0.25 \mathrm{wt} \%$ ). There is a sudden increase in K 2 O at the boundary of the CPZ to $1.4-1.5 \mathrm{wt} \%$. There is a consistent value ( $\sim 1.4 \mathrm{wt} \%$ ) through the DZ with elevations up to $2 \mathrm{wt} \%$ in one sample. The UBZ and UCM have greater K2O contents compared to the LBZ and LCM ( $\sim 1 \mathrm{wt} \%)$.

P2O5: Roughly a Z-style curve, with a LCM value of $0.08 \mathrm{wt} \%$ decreasing through the LBZ and the OZ (to $\sim 0.05 \mathrm{wt} \%$ ). The UOZ has a slightly higher value and starts the increasing trend up through the CPZ and DZ to values of $\sim 0.09 \mathrm{wt} \%$. The UBZ contains more P2O5 than the LBZ, however, the LCM and UCM are similar.
$\mathrm{FeO}^{*}$ ( FeO * calculated from Fe 2 O 3 with the proportion of $\mathrm{Fe} 2+$ to $\mathrm{Fe} 3+$ set at 0.9 ): Identical to Fe 2 O 3 trend.

LOI: There is a rough trend indicating an increase in LOI values up-section through the LPS. The MOZ has elevated LOI values (up to 5.6-5.7 wt \% compared with $1.3 \mathrm{wt} \%$ for the LCM). The UCM shows a higher LOI value compared with the LCM.

Trace Elements:
ICP-OES
Sc: A 'two-cycle trend', with a LCM value of 38 ppm decreasing through the LBZ and OZ, until the UOZ where it climbs to 33 ppm . Then there is a big jump to 57 ppm within the CPZ, and then the DZ consistently contains $\sim 43 \mathrm{ppm}$, including the UBZ and UCM (both higher than the LBZ and LCM, respectively).

V: Similar to Sc. LCM values are at 320 ppm , which decrease through the LBZ and OZ to the lowest value in the LPS of 165 ppm . There is an increase in the UOZ to 210 ppm , and then a jump to $>350 \mathrm{ppm}$ in the CPZ. Then there appears to be two cycles in the DZ, both decreasing upwards, but generally with high V contents than the lower half of the LPS. The UBZ and UCM are roughly similar to the LCM.

Cr: Cr displays a S-shaped curve, with LCM values of 750 ppm , increasing through the LBZ and OZ. The CPZ at the contact with the OZ contains 2000 ppm , the highest value in the LPS. This decreases in the next CPZ sample, and into the DZ, to a value of $<500 \mathrm{ppm}$, and then stays around the same throughout the DZ. The UBZ and UCM crudely correlate with the LCM (750 ppm).

Co: Weak trends, with a steady increase from the LCM through the LBZ and OZ, followed by a sharp decrease in the CPZ. There are possibly two cycles in the DZ. The UBZ and UCM crudely correlate with the LCM.

Ni: An asymmetric S-shape. The LCM has a value of 230 ppm, followed by a increase through the LBZ and the OZ (highest value of $\sim 720 \mathrm{ppm}$ ). The UOZ begins to show a
decrease in comparison to the MOZ, followed by a sharp decrease at the CPZ to values of $<100 \mathrm{ppm}$. Values are $\sim 90 \mathrm{ppm}$ in the DZ, with a slight increase in the UBZ to $\sim 100 \mathrm{ppm}$. The UCM is roughly similar to the LCM ( 200 ppm ).
$\mathrm{Cu}: \mathrm{LCM}$ values are at 100 ppm , with a decrease through the LBZ and in the LOZ. There is a slight increase in the MOZ before a decrease in the UOZ. Cu is at 90 ppm in the CPZ and stays roughly around 100 ppm throughout the DZ except for one sample where a value of $\sim 160 \mathrm{ppm}$ is observed. The UBZ and UCM have higher Cu values than the LCM/LBZ.

Zn : There may be two cycles in Zn through the LPS. The LCM has a value of $\sim 80 \mathrm{ppm}$, with decreases through the LBZ, and there is a slight increase LOZ, before a decrease in the MOZ and then an increase in the UOZ. The CPZ has a higher value, leading to a decreasing trend in the DZ, except for two values, which show an increase (up to 120 ppm ). The UBZ shows some scatter, and the UCM is similar to the LCM.

Sr: An asymmetric Z-shape curve. The LCM has a value of 150 ppm , which decreases through the LBZ and OZ, until the UOZ where it begins to increase, leading into the CPZ with values of $160-200 \mathrm{ppm}$. Sr increases in the DZ to $\sim 300 \mathrm{ppm}$. The UBZ and UCM are similar, but with higher values compared with the LCM/LBZ, respectively.

Y: A 'two-cycle trend'. LCM values are at 17 ppm , which decrease through the LBZ and OZ, but begin to increase in the UOZ, eventually increasing to 17 ppm in the CPZ. They increase slightly through the DZ before decreasing in the upper DZ. The UBZ/UCM values fit well, and correlate with the LCM.

Zr: A weak 'two-cycle trend'. The LCM has a value of 70 ppm , which decreases through the LBZ (decreasing TMF?). Values increase in the LOZ with a further increase in the MOZ, before decreasing in the UOZ. The CPZ has values similar to the LCM, and they stay constant through the DZ, with the UBZ and UCM showing similar values, that corresponds with the LCM.

Ba: A weak 'two-cycle trend'? The LCM has values of 130 ppm , which decrease through the LBZ. They jump up slightly in the LOZ, but decrease again in the MOZ. Values increase in the UOZ and increase again in the CPZ to LCM-like values. The lower DZ is similar to the CPZ, before values increase to $>200 \mathrm{ppm}$ followed by a decreasing trend upward through the rest of the DZ. Values increase slightly at the very top of the DZ. The UBZ and UCM roughly correlate with the LCM.

ICP-MS:
Co: There is an almost S-shaped curve for Co, with both LCM and UCM values very similar. The LCM has a value of $\sim 60 \mathrm{ppm}$, with increases in the first LBZ sample and then decreasing towards the top of the LBZ and into the LOZ. Co increases in the MOZ ( $\sim 70 \mathrm{ppm}$ ) and then decreases slightly in the UOZ ( $\sim 60 \mathrm{ppm}$ ) and into the base of the CPZ ( $\sim 50 \mathrm{ppm}$ - lowest values in the sill). Values increase slightly in the next CPZ sample ( $\sim 55 \mathrm{ppm}$ ) and then are fairly consistent through the DZ ( $\sim 55 \mathrm{ppm}$ ) apart from one sample which has a value of $\sim 70$ ppm in the mid-DZ. The UBZ value increases slightly from the DZ mean value.

Ga: Smooth asymmetrical Z-shaped trend. The LCM has a value of $\sim 14 \mathrm{ppm}$, increasing in the first LBZ sample and then decreasing in the next LBZ sample and into the LOZ. The Ga content shows a slight decrease before increasing slightly in the UOZ and the CPZ ( $\sim 14-15$ $\mathrm{ppm})$. Ga shows a further increase in the DZ. The value of Ga may increase slightly up
through the DZ, but values are all between 16-17 ppm, with the UBZ and UCM being similar, with both higher than the LCM and LBZ.

Rb : Rb displays almost two separate trends but it is quite a smooth 'weak' Z-shape overall. There is an increase in Rb from $\sim 18$ to 20 ppm in the first LBZ sample but then it decreases slightly in the higher LBZ sample. Similar increases are seen in the LOZ, but then it says consistently the same in the rest of the OZ towards the UOZ. There is a sudden increase to $\sim 28 \mathrm{ppm}$ in the CPZ and it increases again slightly at the bottom of the DZ with an increase up to $\sim 72 \mathrm{ppm}$. It begins to decrease again in the upper DZ, continuing in the UBZ and finally the UCM ( $\sim 15 \mathrm{ppm}$ ), which is lower than the LCM.

P: A good Z-shape curve is produced. The P value is $\sim 800 \mathrm{ppm}$ at the LCM , and then it decreases through the LBZ and the OZ (there may be a slight increase in P in the LOZ). P values start to increase in the UOZ and into the CPZ. It continues to increase through the DZ with $P$ values greater the LCM value. The UBZ is similar to the mean value of the DZ, before decreasing slightly into the UCM, which has almost the same P value as the LCM.

Nb : An asymmetrical Z-shape profile. Values increase from the LCM into the LBZ, and then decrease in the next LBZ sample and into the LOZ. There is a slight increase in the LOZ, before the trend decreases again, before an increase in the UOZ. This is continued into the CPZ, with an increasing trend in the base and middle parts of the DZ. Values fluctuate slightly through the upper DZ, UBZ and then UCM preserves a higher Nb concentration than the LCM.
$\mathrm{Sn}: \mathrm{Sn}$ decreases from the LCM and through the LBZ and into the OZ. One sample shows an increase in the LOZ but then the trend generally decreases through the OZ, before UOZ where an increase is observed. This is continued into the CPZ, with a range of values in the base of the DZ, before a smooth decreasing trend is observed towards the UBZ. The UBZ shows a recovery, with values greater than the LBZ, but the UCM has a value less than the LCM.

Cs: A decreasing trend through the LBZ and into the LOZ. Values increase in the MOZ before decreasing in the UOZ. The CPZ shows a slightly increasing trend, which leads into the base of the DZ, followed by fluctuating values in the middle/upper parts of the DZ. The UBZ shows a range of values, before the UCM, which has a value less than the LCM.

Ba: Ba shows a decreasing trend from the LCM and through the LBZ. There is a slight increase in the LOZ, before values decrease in the MOZ, which is followed by a slight increase in the UOZ, which leads into the CPZ. The mid/upper DZ is then detached from the basal DZ/CPZ by a separate decreasing trend up-sill. The UBZ shows a slight skew in values, before the UCM seems to preserve a value close to that of the LCM.

La: A weak Z-shape is displayed. There is a decreasing trend through the LBZ from the LCM and into the OZ. The basal OZ bulge is seen again, before it decreases again. The roof of the OZ hints at a partial recovery, which steps up into the CPZ, followed by a slight increasing trend in the basal regions of the DZ, which values remain fairly consistently, as does the UBZ, which have higher values than the LBZ. The UCM and LCM are very similar.
$\operatorname{REE}(\mathrm{Ce} / \mathrm{Pr} / \mathrm{Nd} / \mathrm{Sm} / \mathrm{Gd} / \mathrm{Tb} / \mathrm{Dy} / \mathrm{Ho} / \mathrm{Er} / \mathrm{Tm} / \mathrm{Yb} / \mathrm{Lu})$ : All behave, generally, the same as La . However, the difference between the UOZ and CPZ becomes greater towards the HREE end of the REE spectrum.

Eu: Eu behaves similarly to the other REE, however, more complex behaviour is seen through the DZ and UBZ, with two decreasing cycles separated by an increase in Eu values mid way through the DZ.

Hf: Similar to the REE but a more less pronounced jump from the roof of the OZ to the CPZ with good symmetry between the LCM and UCM.

Ta: Ta behaves much like the REE, but with a less pronounced increase between the OZ and CPZ.
$\mathrm{Pb}: \mathrm{Pb}$ shows more complex behaviour in comparison to the REE. There is an increase in Pb from the LCM, through the LBZ; a decrease in the LOZ, followed by a sharp increase to the highest values in the sill (OZ sample 10BH1665A5); followed by a decrease through the MOZ and UOZ. Pb values decrease in the CPZ before increasing in the basal DZ, followed by a smooth decreasing trend all through the DZ. The UBZ has Pb values similar to the LCM, much like the UCM.

Th: The LCM and LBZ values remain fairly constant before a sharp decrease in the LOZ, followed by an increase in sample 10BH166A5. Th values decrease through the MOZ and into the UOZ, before a slight increase in the sample immediately beneath the CPZ. Th values increase in the CPZ, before increasing through the DZ. Th values decrease in the upper DZ. Th values increase in the UCM and UBZ, with both showing higher values than the LCM and LBZ, respectively.

U: U behaves like Th, but with a less pronounced increase in sample 10BH166A5; a less pronounced increase in the CPZ.

LREE enrichment ratios
$\mathrm{La} / \mathrm{Yb}$ : A slight decrease from the LCM to the UOZ (a slight increase in the LOZ - sample 10BH166A5). The $\mathrm{La} / \mathrm{Yb}$ is consistently the same throughout the DZ and is the same in the UBZ. The UBZ and DZ has a $\mathrm{La} / \mathrm{Yb}$ similar to the LCM . The UCM La/Yb ratio is slightly lower than the LCM.
$\mathrm{Ce} / \mathrm{Yb}$ : Behaves the same as $\mathrm{La} / \mathrm{Yb}$.
$\mathrm{La} / \mathrm{Sm}$ : Behaves like $\mathrm{La} / \mathrm{Yb}$ and $\mathrm{Ce} / \mathrm{Yb}$, however, the LOZ increase (10BH166A5) is more pronounced.


Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | Melt <br> MgO | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| LBZ | 10BH166 A2 9 | 50.900 | 3.000 | 9.800 | 0.800 | 0.300 | 16.500 | 0.200 | 18.300 | 0.000 | 0.200 | 0.000 | 100.100 | 39.083 | 42.363 | 18.554 | 0.750 | 4.166 | 14.876 |
|  | 10BH166 A2 10 | 50.000 | 4.000 | 8.800 | 0.700 | 0.400 | 15.800 | 0.300 | 19.100 | 0.000 | 0.300 | 0.000 | 99.400 | 41.616 | 41.386 | 16.997 | 0.762 | 4.849 | 14.097 |
|  | 10BH166 A2 11 | 52.400 | 1.700 | 12.800 | 0.500 | 0.200 | 19.200 | 0.300 | 13.200 | 0.000 | 0.100 | 0.000 | 100.500 | 27.714 | 48.462 | 23.823 | 0.728 | 3.580 | 15.652 |
|  | 10BH166 A2 9-11 av | 51.100 | 2.900 | 10.467 | 0.667 | 0.300 | 17.167 | 0.267 | 16.867 | 0.000 | 0.200 | 0.000 | 100.000 | 36.053 | 44.114 | 19.833 | 0.745 | 4.119 | 14.934 |
|  | 10BH166 A2 21 | 53.000 | 1.300 | 16.500 | 0.200 | 0.200 | 24.400 | 0.400 | 3.700 | 0.000 | 0.100 | 0.000 | 99.800 | 7.763 | 61.547 | 30.690 | 0.725 | 4.809 | 14.140 |
|  | 10BH166 A2 22 | 53.300 | 0.800 | 16.200 | 0.200 | 0.100 | 24.800 | 0.400 | 3.400 | 0.000 | 0.100 | 0.000 | 99.300 | 7.147 | 62.668 | 30.186 | 0.732 | 4.952 | 13.990 |
|  | 10BH166 A2 23 | 52.700 | 1.300 | 14.100 | 0.300 | 0.200 | 23.300 | 0.300 | 7.200 | 0.000 | 0.200 | 0.000 | 99.600 | 15.091 | 58.711 | 26.198 | 0.746 | 5.365 | 13.579 |
|  | $\begin{aligned} & \text { 10BH166 A2 21-23 } \\ & \text { av } \end{aligned}$ | 53.000 | 1.133 | 15.600 | 0.233 | 0.167 | 24.167 | 0.367 | 4.767 | 0.000 | 0.133 | 0.000 | 99.567 | 10.004 | 60.973 | 29.023 | 0.734 | 5.024 | 13.916 |
| LBZ | 10BH166 A3 1 | 50.300 | 4.100 | 5.500 | 0.400 | 0.800 | 16.600 | 0.100 | 20.900 | 0.000 | 0.200 | 0.000 | 99.000 | 45.701 | 43.638 | 10.661 | 0.843 | 9.098 | 10.871 |
|  | 10BH166 A3 2 | 51.500 | 3.200 | 5.800 | 0.400 | 0.600 | 17.700 | 0.100 | 19.900 | 0.000 | 0.200 | 0.000 | 99.400 | 42.962 | 45.938 | 11.100 | 0.845 | 9.004 | 10.925 |
|  | 10BH166 A3 3 | 51.500 | 2.400 | 5.600 | 0.300 | 0.400 | 18.100 | 0.200 | 19.700 | 0.000 | 0.200 | 0.000 | 98.300 | 42.435 | 46.872 | 10.693 | 0.852 | 9.343 | 10.735 |
|  | 10BH166 A3 4 | 50.300 | 4.400 | 6.400 | 0.400 | 0.900 | 16.800 | 0.100 | 20.300 | 0.000 | 0.200 | 0.000 | 99.700 | 43.968 | 43.744 | 12.288 | 0.824 | 8.194 | 11.408 |
|  | 10BH166 A3 5 | 51.100 | 3.600 | 6.600 | 0.500 | 0.500 | 17.300 | 0.100 | 19.500 | 0.000 | 0.200 | 0.000 | 99.500 | 42.255 | 45.067 | 12.678 | 0.824 | 7.828 | 11.643 |
|  | 10BH166 A3 1-5 av | 50.940 | 3.540 | 5.980 | 0.400 | 0.640 | 17.300 | 0.120 | 20.060 | 0.000 | 0.200 | 0.000 | 99.180 | 43.459 | 45.057 | 11.484 | 0.838 | 8.680 | 11.113 |
|  | 10BH166 A3 6 | 52.200 | 2.100 | 15.400 | 0.600 | 0.100 | 26.200 | 0.300 | 2.200 | 0.100 | 0.000 | 0.000 | 99.100 | 4.646 | 66.522 | 28.832 | 0.752 | 6.621 | 12.501 |
|  | 10BH166 A3 7 | 52.500 | 2.000 | 15.500 | 0.600 | 0.100 | 26.100 | 0.300 | 2.100 | 0.100 | 0.000 | 0.000 | 99.300 | 4.448 | 66.452 | 29.100 | 0.750 | 6.423 | 12.656 |
|  | 10BH166 A3 8 | 52.500 | 1.800 | 15.600 | 0.600 | 0.000 | 25.900 | 0.300 | 2.000 | 0.100 | 0.100 | 0.000 | 98.900 | 4.258 | 66.297 | 29.445 | 0.747 | 6.245 | 12.800 |
|  | 10BH166 A3 6-8 av | 52.400 | 1.967 | 15.500 | 0.600 | 0.067 | 26.067 | 0.300 | 2.100 | 0.100 | 0.033 | 0.000 | 99.100 | 4.451 | 66.424 | 29.125 | 0.750 | 6.430 | 12.651 |
|  | 10BH166 A3 12 | 49.900 | 4.600 | 6.000 | 0.600 | 0.800 | 16.700 | 0.100 | 20.400 | 0.000 | 0.200 | 0.000 | 99.300 | 44.546 | 43.840 | 11.614 | 0.832 | 8.773 | 11.058 |
|  | 10BH166 A3 13 | 50.700 | 4.000 | 5.600 | 0.500 | 0.600 | 16.800 | 0.100 | 20.900 | 0.000 | 0.200 | 0.000 | 99.500 | 45.375 | 43.848 | 10.778 | 0.842 | 9.003 | 10.925 |
|  | 10BH166 A3 14 | 50.400 | 3.000 | 7.100 | 0.400 | 0.300 | 16.800 | 0.200 | 19.300 | 0.100 | 0.200 | 0.000 | 97.800 | 42.148 | 44.107 | 13.745 | 0.808 | 6.760 | 12.394 |
|  | $\begin{aligned} & \text { 10BH166 A3 12-14 } \\ & \text { av } \end{aligned}$ | 50.850 | 3.392 | 8.550 | 0.525 | 0.442 | 19.092 | 0.175 | 15.675 | 0.050 | 0.158 | 0.000 | 98.925 | 33.924 | 49.673 | 16.403 | 0.799 | 7.163 | 12.098 |
|  | 10BH166 A3 33 | 50.100 | 4.400 | 5.400 | 0.500 | 0.800 | 16.200 | 0.100 | 21.200 | 0.000 | 0.200 | 0.000 | 99.000 | 46.632 | 42.838 | 10.529 | 0.842 | 9.033 | 10.908 |
|  | 10BH166 A3 34 | 50.700 | 4.600 | 6.000 | 0.400 | 1.000 | 16.700 | 0.100 | 20.100 | 0.000 | 0.200 | 0.000 | 99.900 | 44.180 | 44.129 | 11.691 | 0.832 | 8.502 | 11.219 |
|  | 10BH166 A3 35 | 52.500 | 1.900 | 8.800 | 0.400 | 0.300 | 18.000 | 0.200 | 17.400 | 0.000 | 0.200 | 0.000 | 99.700 | 37.148 | 46.198 | 16.654 | 0.785 | 5.261 | 13.680 |
|  | $\begin{aligned} & \text { 10BH166 A3 33-35 } \\ & \text { av } \end{aligned}$ | 51.100 | 3.633 | 6.733 | 0.433 | 0.700 | 16.967 | 0.133 | 19.567 | 0.000 | 0.200 | 0.000 | 99.533 | 42.599 | 44.406 | 12.995 | 0.818 | 7.396 | 11.933 |
| LOZ | BH311 B2 124 | 50.539 | 2.353 | 9.768 | 1.195 | 0.104 | 16.280 | 0.191 | 19.613 | 0.027 | 0.301 | 0.012 | 100.383 | 40.994 | 40.907 | 18.099 | 0.748 | 4.190 | 14.846 |
|  | BH311 B2 124 | 51.254 | 2.215 | 8.956 | 0.690 | 0.389 | 17.694 | 0.234 | 17.705 | 0.030 | 0.268 | 0.003 | 99.438 | 37.738 | 45.340 | 16.922 | 0.779 | 5.487 | 13.464 |
|  | BH311 B2 124 | 52.484 | 2.044 | 7.410 | 0.371 | 0.678 | 18.515 | 0.171 | 18.314 | 0.042 | 0.221 | 0.011 | 100.261 | 38.849 | 47.217 | 13.934 | 0.817 | 7.205 | 12.068 |
|  | BH311 B2 124 | 51.404 | 2.772 | 6.255 | 0.329 | 0.702 | 18.062 | 0.177 | 18.983 | 0.052 | 0.198 | 0.031 | 98.965 | 41.052 | 46.957 | 11.991 | 0.837 | 8.607 | 11.156 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{MgO} \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | BH311 B2 124 | 51.500 | 2.909 | 6.354 | 0.387 | 0.741 | 18.351 | 0.161 | 18.992 | 0.010 | 0.205 | 0.010 | 99.620 | 40.680 | 47.255 | 12.065 | 0.837 | 8.797 | 11.044 |
|  | BH311 B2 124 | 51.324 | 2.915 | 5.920 | 0.308 | 0.668 | 18.396 | 0.135 | 19.233 | 0.080 | 0.215 | 0.008 | 99.202 | 41.276 | 47.462 | 11.262 | 0.847 | 9.432 | 10.687 |
|  | BH311 B2 124 | 51.421 | 3.136 | 5.997 | 0.265 | 0.667 | 17.910 | 0.138 | 19.710 | 0.050 | 0.215 | 0.003 | 99.512 | 42.335 | 46.246 | 11.419 | 0.842 | 8.961 | 10.950 |
|  | BH311 B2 124 | 51.551 | 2.625 | 5.324 | 0.395 | 0.669 | 17.712 | 0.123 | 20.899 | 0.051 | 0.167 | 0.002 | 99.518 | 44.550 | 45.390 | 10.061 | 0.856 | 9.622 | 10.584 |
|  | BH311 B2 124 | 50.625 | 4.081 | 6.098 | 0.516 | 0.885 | 17.438 | 0.151 | 19.724 | 0.023 | 0.201 | 0.007 | 99.749 | 42.791 | 45.481 | 11.728 | 0.836 | 8.954 | 10.953 |
|  | BH311 B2 124 | 49.804 | 4.111 | 5.970 | 0.602 | 0.781 | 16.801 | 0.116 | 20.798 | 0.019 | 0.212 | 0.010 | 99.224 | 44.932 | 43.635 | 11.433 | 0.834 | 8.876 | 10.998 |
|  | BH311 B2 124 | 50.400 | 4.039 | 5.716 | 0.495 | 0.598 | 16.942 | 0.125 | 20.564 | 0.084 | 0.196 | 0.007 | 99.166 | 44.706 | 44.278 | 11.016 | 0.841 | 9.061 | 10.892 |
|  | BH311 B2 124 | 50.743 | 3.998 | 5.767 | 0.366 | 0.804 | 16.740 | 0.117 | 20.695 | 0.054 | 0.233 | 0.019 | 99.536 | 45.056 | 43.814 | 11.130 | 0.838 | 8.739 | 11.078 |
|  | BH311 B2 124 | 49.939 | 4.059 | 5.754 | 0.530 | 0.803 | 16.993 | 0.133 | 20.406 | 0.066 | 0.198 | 0.007 | 98.888 | 44.423 | 44.473 | 11.104 | 0.840 | 9.222 | 10.802 |
|  | BH311 B2 124 | 50.310 | 3.883 | 5.915 | 0.387 | 0.746 | 16.943 | 0.120 | 20.526 | 0.041 | 0.218 | 0.006 | 99.095 | 44.488 | 44.147 | 11.365 | 0.836 | 8.799 | 11.043 |
|  | BH311 B2 124 | 50.001 | 3.370 | 7.709 | 0.413 | 0.654 | 22.083 | 0.101 | 13.128 | 0.074 | 0.165 | 0.025 | 97.723 | 28.226 | 57.080 | 14.693 | 0.836 | 10.323 | 10.224 |
|  | BH311 B2 124 | 54.102 | 3.350 | 6.595 | 0.429 | 0.688 | 17.048 | 0.189 | 19.664 | 0.032 | 0.197 | 0.007 | 102.301 | 42.743 | 44.549 | 12.708 | 0.822 | 6.664 | 12.468 |
|  | BH311 B2 124 | 37.336 | 1.898 | 7.614 | 0.884 | 0.480 | 18.460 | 0.205 | 20.140 | 0.000 | 0.143 | 0.017 | 87.177 | 41.033 | 45.215 | 13.752 | 0.812 | 11.042 | 9.879 |
|  | BH311 B2 124 | 49.677 | 3.004 | 8.690 | 1.019 | 0.212 | 16.108 | 0.254 | 19.208 | 0.065 | 0.286 | 0.015 | 98.538 | 41.507 | 41.846 | 16.647 | 0.768 | 5.076 | 13.863 |
|  | BH311 B2 124 | 50.611 | 2.361 | 8.885 | 0.799 | 0.017 | 15.560 | 0.235 | 20.358 | 0.050 | 0.376 | 0.004 | 99.256 | 43.370 | 39.851 | 16.779 | 0.757 | 4.119 | 14.934 |
|  | BH311 B2 124 | 50.207 | 2.441 | 8.800 | 0.869 | 0.169 | 16.925 | 0.246 | 18.307 | 0.022 | 0.250 | 0.007 | 98.243 | 39.408 | 43.799 | 16.792 | 0.774 | 5.282 | 13.659 |
|  | BH311 B2 124 av | 50.262 | 3.078 | 6.975 | 0.562 | 0.573 | 17.548 | 0.166 | 19.348 | 0.044 | 0.223 | 0.011 | 98.790 | 41.508 | 45.247 | 13.245 | 0.818 | 7.827 | 11.643 |
|  | BH311 B2 127 | 50.689 | 2.514 | 6.869 | 0.165 | 0.448 | 18.068 | 0.256 | 19.510 | 0.045 | 0.144 | 0.000 | 98.708 | 41.230 | 45.902 | 12.868 | 0.824 | 8.026 | 11.514 |
|  | BH311 B2 127 | 52.906 | 2.433 | 6.027 | 0.380 | 0.694 | 18.152 | 0.155 | 19.999 | 0.015 | 0.202 | 0.001 | 100.964 | 42.403 | 46.269 | 11.328 | 0.843 | 8.576 | 11.175 |
|  | BH311 B2 127 | 52.530 | 2.530 | 6.242 | 0.294 | 0.493 | 18.814 | 0.165 | 19.143 | 0.055 | 0.182 | 0.011 | 100.459 | 40.476 | 47.824 | 11.700 | 0.843 | 8.915 | 10.976 |
|  | BH311 B2 127 | 51.205 | 4.123 | 5.848 | 0.516 | 0.978 | 16.792 | 0.174 | 20.451 | 0.006 | 0.204 | 0.009 | 100.306 | 44.631 | 44.055 | 11.313 | 0.837 | 8.594 | 11.164 |
|  | BH311 B2 127 | 51.762 | 4.125 | 5.951 | 0.502 | 0.816 | 17.070 | 0.125 | 20.647 | 0.060 | 0.222 | 0.008 | 101.288 | 44.456 | 44.185 | 11.359 | 0.836 | 8.599 | 11.161 |
|  | BH311 B2 127 | 51.660 | 4.170 | 5.645 | 0.466 | 0.812 | 17.059 | 0.152 | 20.521 | 0.064 | 0.209 | 0.016 | 100.774 | 44.579 | 44.551 | 10.871 | 0.843 | 8.994 | 10.931 |
|  | BH311 B2 127 | 51.322 | 4.225 | 5.871 | 0.387 | 0.847 | 17.087 | 0.124 | 20.525 | 0.080 | 0.225 | 0.005 | 100.698 | 44.358 | 44.394 | 11.248 | 0.838 | 8.843 | 11.018 |
|  | BH311 B2 127 | 51.346 | 4.192 | 5.943 | 0.566 | 0.825 | 17.007 | 0.145 | 20.568 | 0.031 | 0.224 | 0.008 | 100.855 | 44.441 | 44.176 | 11.383 | 0.836 | 8.686 | 11.109 |
|  | BH311 B2 127 | 51.247 | 4.176 | 6.079 | 0.516 | 0.977 | 17.126 | 0.153 | 20.083 | 0.000 | 0.214 | 0.012 | 100.583 | 43.601 | 44.699 | 11.700 | 0.834 | 8.598 | 11.162 |
|  | BH311 B2 127 | 51.420 | 4.267 | 5.620 | 0.631 | 0.926 | 16.830 | 0.172 | 20.464 | 0.045 | 0.235 | 0.007 | 100.617 | 44.800 | 44.294 | 10.907 | 0.842 | 8.953 | 10.954 |
|  | BH311 B2 127 | 51.717 | 4.244 | 5.696 | 0.445 | 0.983 | 16.870 | 0.129 | 20.764 | 0.040 | 0.211 | 0.011 | 101.110 | 45.047 | 43.999 | 10.954 | 0.841 | 8.798 | 11.044 |
|  | BH311 B2 127 | 51.411 | 4.221 | 5.556 | 0.416 | 0.826 | 16.780 | 0.117 | 20.932 | 0.013 | 0.187 | 0.006 | 100.465 | 45.475 | 43.825 | 10.700 | 0.843 | 8.933 | 10.965 |
|  | BH311 B2 127 | 51.800 | 4.176 | 5.866 | 0.573 | 0.917 | 16.531 | 0.199 | 21.365 | 0.019 | 0.220 | 0.008 | 101.674 | 46.007 | 42.795 | 11.198 | 0.834 | 8.306 | 11.338 |
|  | BH311 B2 127 | 51.068 | 4.176 | 5.417 | 0.438 | 0.778 | 16.548 | 0.131 | 21.319 | 0.070 | 0.198 | 0.006 | 100.149 | 46.331 | 43.233 | 10.436 | 0.845 | 9.047 | 10.900 |
|  | BH311 B2 127 | 51.408 | 4.196 | 5.461 | 0.502 | 0.883 | 16.544 | 0.164 | 21.515 | 0.050 | 0.224 | 0.005 | 100.952 | 46.524 | 43.008 | 10.468 | 0.844 | 8.977 | 10.940 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{MgO} \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | BH311 B2 127 | 51.308 | 4.213 | 5.377 | 0.423 | 1.027 | 16.432 | 0.131 | 21.501 | 0.023 | 0.219 | 0.006 | 100.660 | 46.719 | 42.924 | 10.357 | 0.845 | 9.006 | 10.924 |
|  | BH311 B2 127 | 51.449 | 4.288 | 5.732 | 0.524 | 0.992 | 16.456 | 0.081 | 21.250 | 0.000 | 0.236 | 0.006 | 101.014 | 46.081 | 42.900 | 11.019 | 0.836 | 8.505 | 11.217 |
|  | BH311 B2 127 | 51.528 | 4.286 | 5.707 | 0.516 | 1.040 | 16.711 | 0.119 | 20.959 | 0.023 | 0.220 | 0.001 | 101.110 | 45.457 | 43.571 | 10.972 | 0.839 | 8.740 | 11.077 |
|  | BH311 B2 127 | 50.226 | 4.846 | 6.761 | 0.550 | 0.900 | 16.629 | 0.152 | 19.620 | 0.053 | 0.185 | 0.001 | 99.923 | 43.022 | 43.836 | 13.142 | 0.814 | 7.704 | 11.724 |
|  | BH311 B2 127 | 53.278 | 5.799 | 6.009 | 0.366 | 0.554 | 17.997 | 0.162 | 20.105 | 0.041 | 0.189 | 0.000 | 104.500 | 42.715 | 45.967 | 11.317 | 0.842 | 9.313 | 10.752 |
|  | BH311 B2 127 av | 51.564 | 4.060 | 5.884 | 0.459 | 0.836 | 17.075 | 0.150 | 20.562 | 0.037 | 0.208 | 0.006 | 100.840 | 44.418 | 44.320 | 11.262 | 0.838 | 8.695 | 11.104 |
|  | BH311 B2 135 | 51.711 | 1.937 | 9.216 | 0.697 | 0.285 | 17.802 | 0.259 | 17.409 | 0.005 | 0.239 | 0.005 | 99.565 | 37.056 | 45.554 | 17.390 | 0.775 | 5.119 | 13.819 |
|  | BH311 B2 135 | 50.522 | 2.089 | 8.787 | 0.548 | 0.302 | 20.190 | 0.232 | 15.984 | 0.027 | 0.213 | 0.017 | 98.911 | 33.268 | 50.519 | 16.213 | 0.804 | 7.742 | 11.699 |
|  | BH311 B2 135 | 51.079 | 1.879 | 7.699 | 0.443 | 0.456 | 17.516 | 0.204 | 19.975 | 0.052 | 0.203 | 0.011 | 99.517 | 41.739 | 44.000 | 14.261 | 0.802 | 6.533 | 12.570 |
|  | BH311 B2 135 | 51.752 | 2.647 | 7.451 | 0.464 | 0.434 | 18.893 | 0.208 | 17.118 | 0.055 | 0.208 | 0.022 | 99.252 | 36.864 | 48.912 | 14.224 | 0.819 | 7.718 | 11.715 |
|  | BH311 B2 135 | 52.183 | 1.923 | 6.648 | 0.508 | 0.575 | 18.357 | 0.166 | 18.740 | 0.016 | 0.208 | 0.018 | 99.342 | 40.127 | 47.254 | 12.619 | 0.831 | 7.928 | 11.578 |
|  | BH311 B2 135 | 52.153 | 2.499 | 5.690 | 0.287 | 0.516 | 18.255 | 0.179 | 19.753 | 0.001 | 0.199 | 0.000 | 99.532 | 42.259 | 46.950 | 10.791 | 0.851 | 9.245 | 10.789 |
|  | BH311 B2 135 | 50.652 | 4.361 | 6.322 | 0.579 | 1.080 | 17.168 | 0.118 | 19.338 | 0.053 | 0.229 | 0.010 | 99.910 | 42.425 | 45.280 | 12.295 | 0.829 | 8.507 | 11.216 |
|  | BH311 B2 135 | 51.685 | 3.118 | 6.170 | 0.358 | 0.728 | 18.139 | 0.215 | 18.909 | 0.066 | 0.193 | 0.005 | 99.586 | 40.942 | 47.215 | 11.843 | 0.840 | 8.806 | 11.039 |
|  | BH311 B2 135 | 53.048 | 2.360 | 6.149 | 0.351 | 0.624 | 18.945 | 0.155 | 18.784 | 0.014 | 0.176 | 0.010 | 100.616 | 39.957 | 48.448 | 11.595 | 0.846 | 8.915 | 10.976 |
|  | BH311 B2 135 | 52.467 | 2.217 | 5.810 | 0.423 | 0.616 | 18.808 | 0.160 | 19.025 | 0.025 | 0.185 | 0.006 | 99.742 | 40.664 | 48.328 | 11.008 | 0.852 | 9.379 | 10.716 |
|  | BH311 B2 135 | 50.219 | 4.427 | 5.682 | 0.559 | 1.140 | 16.394 | 0.126 | 20.933 | 0.043 | 0.213 | 0.012 | 99.748 | 45.827 | 43.146 | 11.027 | 0.837 | 8.892 | 10.989 |
|  | BH311 B2 135 | 50.672 | 4.306 | 5.709 | 0.696 | 0.795 | 16.569 | 0.140 | 20.927 | 0.057 | 0.197 | 0.013 | 100.081 | 45.586 | 43.390 | 11.024 | 0.838 | 8.823 | 11.029 |
|  | BH311 B2 135 | 50.802 | 4.312 | 5.764 | 0.531 | 0.830 | 16.677 | 0.168 | 20.699 | 0.020 | 0.210 | 0.012 | 100.025 | 45.138 | 43.720 | 11.142 | 0.838 | 8.775 | 11.057 |
|  | BH311 B2 135 | 48.533 | 4.070 | 6.891 | 0.485 | 0.947 | 17.998 | 0.161 | 18.033 | 0.041 | 0.207 | 0.027 | 97.393 | 39.392 | 47.264 | 13.344 | 0.823 | 8.943 | 10.960 |
|  | BH311 B2 135 | 51.048 | 4.252 | 5.588 | 0.660 | 0.852 | 16.488 | 0.109 | 20.937 | 0.086 | 0.245 | 0.012 | 100.277 | 45.802 | 43.362 | 10.836 | 0.840 | 8.815 | 11.034 |
|  | BH311 B2 135 | 50.332 | 3.971 | 5.569 | 0.688 | 0.838 | 17.074 | 0.119 | 19.607 | 0.064 | 0.254 | 0.031 | 98.547 | 43.504 | 45.543 | 10.954 | 0.845 | 9.369 | 10.721 |
|  | BH311 B2 135 | 48.625 | 4.001 | 6.019 | 0.529 | 0.967 | 17.630 | 0.170 | 19.056 | 0.043 | 0.216 | 0.004 | 97.260 | 41.802 | 46.493 | 11.705 | 0.839 | 9.701 | 10.543 |
|  | BH311 B2 135 | 50.703 | 3.905 | 6.472 | 0.529 | 0.810 | 17.574 | 0.141 | 18.920 | 0.039 | 0.213 | 0.014 | 99.320 | 41.324 | 46.145 | 12.531 | 0.829 | 8.460 | 11.244 |
|  | BH311 B2 135 | 52.056 | 2.681 | 7.912 | 0.378 | 0.590 | 18.204 | 0.212 | 18.052 | 0.033 | 0.267 | 0.007 | 100.392 | 38.449 | 46.612 | 14.939 | 0.804 | 6.747 | 12.404 |
|  | BH311 B2 135 av | 51.839 | 2.437 | 9.272 | 0.569 | 0.406 | 18.941 | 0.237 | 15.948 | 0.022 | 0.211 | 0.014 | 99.896 | 33.977 | 48.512 | 17.511 | 0.784 | 6.025 | 12.984 |
|  | BH311 B2 136 | 51.104 | 3.170 | 6.741 | 0.514 | 0.690 | 17.881 | 0.174 | 18.907 | 0.038 | 0.214 | 0.013 | 99.446 | 40.805 | 46.332 | 12.863 | 0.826 | 8.118 | 11.456 |
|  | BH311 B2 136 | 51.787 | 2.405 | 7.472 | 0.372 | 0.439 | 17.952 | 0.210 | 18.892 | 0.012 | 0.223 | 0.022 | 99.786 | 40.113 | 45.823 | 14.064 | 0.811 | 6.953 | 12.250 |
|  | BH311 B2 136 | 49.102 | 2.884 | 5.915 | 0.351 | 0.737 | 17.846 | 0.170 | 19.928 | 0.000 | 0.206 | 0.013 | 97.152 | 42.740 | 46.014 | 11.246 | 0.843 | 9.637 | 10.577 |
|  | BH311 B2 136 | 53.065 | 2.707 | 5.952 | 0.401 | 0.672 | 18.059 | 0.156 | 19.882 | 0.044 | 0.199 | 0.003 | 101.140 | 42.421 | 46.322 | 11.258 | 0.844 | 8.597 | 11.162 |
|  | BH311 B2 136 | 52.628 | 2.516 | 5.850 | 0.402 | 0.638 | 18.604 | 0.114 | 19.632 | 0.011 | 0.225 | 0.009 | 100.629 | 41.608 | 47.401 | 10.991 | 0.850 | 9.293 | 10.763 |
|  | BH311 B2 136 | 53.075 | 2.395 | 5.673 | 0.416 | 0.730 | 18.355 | 0.182 | 19.696 | 0.052 | 0.179 | 0.000 | 100.753 | 42.094 | 47.159 | 10.748 | 0.852 | 9.120 | 10.859 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | Melt <br> MgO | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | BH311 B2 136 | 49.485 | 2.390 | 6.074 | 0.351 | 0.702 | 18.589 | 0.167 | 19.658 | 0.041 | 0.174 | 0.006 | 97.637 | 41.482 | 47.157 | 11.362 | 0.845 | 9.828 | 10.476 |
|  | BH311 B2 136 | 52.968 | 2.729 | 6.171 | 0.330 | 0.611 | 18.363 | 0.158 | 19.266 | 0.090 | 0.185 | 0.012 | 100.883 | 41.156 | 47.158 | 11.686 | 0.841 | 8.549 | 11.190 |
|  | BH311 B2 136 | 52.625 | 2.753 | 5.828 | 0.308 | 0.716 | 17.962 | 0.173 | 19.979 | 0.034 | 0.177 | 0.020 | 100.575 | 42.746 | 46.200 | 11.054 | 0.846 | 8.795 | 11.045 |
|  | BH311 B2 136 | 52.575 | 3.075 | 6.378 | 0.408 | 0.571 | 17.704 | 0.160 | 20.008 | 0.023 | 0.199 | 0.008 | 101.109 | 42.620 | 45.337 | 12.044 | 0.832 | 7.992 | 11.536 |
|  | BH311 B2 136 | 52.803 | 3.036 | 5.984 | 0.387 | 0.672 | 17.590 | 0.167 | 20.193 | 0.026 | 0.210 | 0.006 | 101.074 | 43.292 | 45.336 | 11.373 | 0.840 | 8.317 | 11.331 |
|  | BH311 B2 136 | 52.463 | 2.619 | 6.809 | 0.308 | 0.653 | 17.483 | 0.218 | 19.940 | 0.037 | 0.235 | 0.023 | 100.788 | 42.431 | 44.725 | 12.844 | 0.821 | 7.218 | 12.058 |
|  | BH311 B2 136 | 51.023 | 3.367 | 7.663 | 0.929 | 0.569 | 16.201 | 0.208 | 20.096 | 0.061 | 0.326 | 0.014 | 100.457 | 43.343 | 42.007 | 14.651 | 0.790 | 5.923 | 13.072 |
|  | BH311 B2 136 | 52.825 | 2.872 | 6.035 | 0.151 | 0.624 | 17.987 | 0.177 | 20.022 | 0.034 | 0.209 | 0.000 | 100.936 | 42.604 | 46.012 | 11.384 | 0.842 | 8.519 | 11.209 |
|  | BH311 B2 136 | 52.768 | 2.531 | 5.792 | 0.394 | 0.751 | 18.211 | 0.174 | 19.799 | 0.024 | 0.222 | 0.004 | 100.670 | 42.282 | 46.753 | 10.965 | 0.849 | 9.003 | 10.925 |
|  | BH311 B2 136 | 53.218 | 2.407 | 5.912 | 0.229 | 0.712 | 18.278 | 0.131 | 19.848 | 0.056 | 0.201 | 0.008 | 101.000 | 42.174 | 46.690 | 11.136 | 0.846 | 8.695 | 11.104 |
|  | BH311 B2 136 | 52.964 | 2.589 | 5.864 | 0.394 | 0.616 | 17.955 | 0.187 | 19.867 | 0.059 | 0.188 | 0.020 | 100.703 | 42.587 | 46.270 | 11.143 | 0.845 | 8.583 | 11.170 |
|  | BH311 B2 136 | 52.781 | 2.768 | 5.628 | 0.309 | 0.677 | 17.768 | 0.178 | 19.882 | 0.079 | 0.205 | 0.018 | 100.293 | 43.005 | 46.203 | 10.791 | 0.849 | 8.818 | 11.032 |
|  | BH311 B2 136 | 52.919 | 2.128 | 6.817 | 0.380 | 0.222 | 18.472 | 0.200 | 19.077 | 0.051 | 0.180 | 0.003 | 100.449 | 40.309 | 46.922 | 12.769 | 0.828 | 7.652 | 11.759 |
|  | BH311 B2 136 | 52.086 | 3.067 | 6.757 | 0.401 | 0.362 | 17.371 | 0.158 | 19.731 | 0.045 | 0.230 | 0.021 | 100.229 | 42.338 | 44.810 | 12.853 | 0.821 | 7.317 | 11.988 |
|  | BH311 B2 136 | 53.214 | 1.780 | 6.596 | 0.358 | 0.283 | 18.418 | 0.196 | 18.947 | 0.039 | 0.188 | 0.007 | 100.026 | 40.368 | 47.174 | 12.458 | 0.833 | 7.617 | 11.782 |
|  | BH311 B2 136 av | 52.319 | 2.651 | 6.259 | 0.379 | 0.598 | 17.958 | 0.174 | 19.717 | 0.041 | 0.208 | 0.011 | 100.314 | 42.086 | 46.074 | 11.841 | 0.836 | 8.298 | 11.343 |
| LOZ | 10BH166 A4 1 | 52.600 | 2.500 | 6.000 | 0.200 | 0.500 | 18.700 | 0.200 | 19.000 | 0.000 | 0.200 | 0.000 | 99.900 | 40.599 | 48.036 | 11.365 | 0.847 | 9.029 | 10.910 |
|  | 10BH166 A4 2 | 51.500 | 2.600 | 6.400 | 0.500 | 0.600 | 17.600 | 0.200 | 19.700 | 0.000 | 0.200 | 0.000 | 99.200 | 42.337 | 45.471 | 12.193 | 0.831 | 8.005 | 11.528 |
|  | 10BH166 A4 3 | 50.200 | 2.900 | 9.000 | 1.100 | 0.000 | 15.600 | 0.200 | 20.100 | 0.000 | 0.300 | 0.000 | 99.400 | 42.919 | 40.045 | 17.036 | 0.755 | 4.284 | 14.733 |
|  | 10BH166 A4 1-3 av | 51.433 | 2.667 | 7.133 | 0.600 | 0.367 | 17.300 | 0.200 | 19.600 | 0.000 | 0.233 | 0.000 | 99.500 | 41.951 | 44.515 | 13.534 | 0.812 | 6.942 | 12.258 |
|  | 10BH166 A4 4 | 49.700 | 4.200 | 5.900 | 0.500 | 1.100 | 17.100 | 0.100 | 19.800 | 0.000 | 0.200 | 0.000 | 98.800 | 43.433 | 45.094 | 11.473 | 0.838 | 9.190 | 10.820 |
|  | 10BH166 A4 5 | 49.600 | 4.200 | 6.000 | 0.600 | 1.100 | 16.800 | 0.200 | 19.900 | 0.000 | 0.200 | 0.000 | 98.600 | 43.818 | 44.471 | 11.711 | 0.833 | 8.849 | 11.014 |
|  | 10BH166 A4 6 | 49.800 | 2.700 | 9.000 | 1.000 | 0.100 | 16.600 | 0.200 | 18.700 | 0.100 | 0.300 | 0.000 | 98.400 | 40.099 | 42.793 | 17.108 | 0.767 | 5.120 | 13.819 |
|  | 10BH166 A4 4-6 av | 49.700 | 3.700 | 6.967 | 0.700 | 0.767 | 16.833 | 0.167 | 19.467 | 0.033 | 0.233 | 0.000 | 98.600 | 42.430 | 44.109 | 13.461 | 0.812 | 7.529 | 11.842 |
|  | 10BH166 A4 9 | 49.900 | 3.700 | 5.600 | 0.400 | 0.900 | 16.900 | 0.100 | 20.500 | 0.100 | 0.200 | 0.000 | 98.200 | 44.778 | 44.378 | 10.843 | 0.843 | 9.246 | 10.789 |
|  | 10BH166 A4 10 | 49.500 | 3.800 | 5.500 | 0.300 | 0.900 | 16.400 | 0.100 | 20.800 | 0.100 | 0.200 | 0.000 | 97.700 | 45.824 | 43.435 | 10.741 | 0.842 | 9.041 | 10.904 |
|  | 10BH166 A4 11 | 51.100 | 2.300 | 6.200 | 0.400 | 0.500 | 17.200 | 0.200 | 20.200 | 0.100 | 0.200 | 0.000 | 98.300 | 43.559 | 44.589 | 11.852 | 0.832 | 7.908 | 11.590 |
|  | 10BH166 A4 9-11 av | 50.167 | 3.267 | 5.767 | 0.367 | 0.767 | 16.833 | 0.133 | 20.500 | 0.100 | 0.200 | 0.000 | 98.067 | 44.712 | 44.138 | 11.150 | 0.839 | 8.727 | 11.085 |
| LOZ | 10BH166 A5 1 | 53.100 | 2.100 | 5.800 | 0.200 | 0.600 | 18.900 | 0.200 | 18.400 | 0.000 | 0.200 | 0.000 | 99.500 | 39.773 | 49.113 | 11.114 | 0.853 | 9.120 | 10.859 |
|  | 10BH166 A5 2 | 50.300 | 4.400 | 5.500 | 0.400 | 1.000 | 16.700 | 0.100 | 19.700 | 0.000 | 0.200 | 0.000 | 98.300 | 44.119 | 44.962 | 10.919 | 0.844 | 9.142 | 10.847 |
|  | 10BH166 A5 3 | 51.000 | 3.300 | 6.600 | 0.600 | 0.800 | 16.600 | 0.200 | 19.300 | 0.000 | 0.200 | 0.000 | 98.600 | 42.787 | 44.242 | 12.971 | 0.818 | 7.103 | 12.141 |
|  | 10BH166 A5 4 | 50.700 | 3.000 | 8.300 | 0.800 | 0.300 | 16.600 | 0.200 | 18.600 | 0.000 | 0.300 | 0.000 | 98.800 | 40.511 | 43.464 | 16.025 | 0.781 | 5.438 | 13.509 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{MgO} \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 10BH166 A5 5 | 50.500 | 2.900 | 8.500 | 0.800 | 0.300 | 16.200 | 0.200 | 18.600 | 0.000 | 0.300 | 0.000 | 98.300 | 40.780 | 42.699 | 16.520 | 0.773 | 4.914 | 14.029 |
|  | 10BH166 A5 1-5 av | 51.120 | 3.140 | 6.940 | 0.560 | 0.600 | 17.000 | 0.180 | 18.920 | 0.000 | 0.240 | 0.000 | 98.700 | 41.574 | 44.908 | 13.518 | 0.814 | 6.998 | 12.217 |
|  | 10BH166 A5 25 | 50.200 | 4.400 | 5.800 | 0.600 | 1.200 | 16.900 | 0.200 | 19.600 | 0.000 | 0.200 | 0.000 | 99.100 | 43.499 | 45.090 | 11.411 | 0.839 | 9.063 | 10.891 |
|  | 10BH166 A5 26 | 49.600 | 4.500 | 6.300 | 0.500 | 0.900 | 16.700 | 0.200 | 19.700 | 0.000 | 0.200 | 0.000 | 98.500 | 43.429 | 44.259 | 12.312 | 0.825 | 8.387 | 11.289 |
|  | 10BH166 A5 27 | 35.700 | 5.000 | 11.100 | 0.400 | 0.400 | 16.400 | 0.300 | 17.300 | 0.000 | 0.100 | 0.100 | 86.800 | 36.922 | 42.078 | 21.000 | 0.725 | 7.305 | 11.997 |
|  | $\begin{aligned} & \text { 10BH166 A5 25-27 } \\ & \text { av } \end{aligned}$ | 45.167 | 4.633 | 7.733 | 0.500 | 0.833 | 16.667 | 0.233 | 18.867 | 0.000 | 0.167 | 0.033 | 94.800 | 41.231 | 43.787 | 14.982 | 0.793 | 7.967 | 11.552 |
|  | 10BH166 A5 28 | 52.800 | 1.500 | 15.000 | 0.600 | 0.100 | 26.600 | 0.300 | 2.000 | 0.100 | 0.000 | 0.000 | 98.900 | 4.231 | 67.642 | 28.127 | 0.760 | 6.767 | 12.389 |
|  | 10BH166 A5 29 | 52.500 | 1.600 | 15.600 | 0.700 | 0.000 | 26.400 | 0.300 | 1.900 | 0.100 | 0.000 | 0.000 | 99.100 | 4.003 | 66.863 | 29.134 | 0.751 | 6.469 | 12.620 |
|  | 10BH166 A5 30 | 53.800 | 1.100 | 15.800 | 0.500 | 0.000 | 26.800 | 0.300 | 1.700 | 0.100 | 0.000 | 0.000 | 100.200 | 3.547 | 67.227 | 29.225 | 0.751 | 6.208 | 12.831 |
|  | $\begin{aligned} & \text { 10BH166 A5 28-30 } \\ & \text { av } \end{aligned}$ | 53.033 | 1.400 | 15.467 | 0.600 | 0.033 | 26.600 | 0.300 | 1.867 | 0.100 | 0.000 | 0.000 | 99.400 | 3.925 | 67.244 | 28.831 | 0.754 | 6.479 | 12.612 |
| LOZ | 10BH167 A1 1 | 51.300 | 3.900 | 5.500 | 0.500 | 0.800 | 16.600 | 0.100 | 20.700 | 0.100 | 0.200 | 0.000 | 99.800 | 45.463 | 43.829 | 10.708 | 0.843 | 8.750 | 11.072 |
|  | 10BH167 A1 2 | 51.300 | 3.900 | 5.500 | 0.400 | 1.000 | 17.100 | 0.200 | 19.900 | 0.100 | 0.200 | 0.000 | 99.500 | 43.897 | 45.347 | 10.755 | 0.847 | 9.184 | 10.823 |
|  | 10BH167 A1 3 | 50.400 | 3.000 | 9.000 | 1.400 | 0.100 | 15.800 | 0.300 | 18.900 | 0.000 | 0.300 | 0.000 | 99.100 | 41.201 | 41.407 | 17.392 | 0.758 | 4.369 | 14.631 |
|  | 10BH167 A1 1-3 av | 51.000 | 3.600 | 6.667 | 0.767 | 0.633 | 16.500 | 0.200 | 19.833 | 0.067 | 0.233 | 0.000 | 99.467 | 43.514 | 43.520 | 12.966 | 0.815 | 7.148 | 12.108 |
|  | 10BH167 A1 11 | 51.000 | 3.800 | 6.200 | 0.400 | 1.000 | 17.900 | 0.200 | 18.600 | 0.000 | 0.200 | 0.000 | 99.300 | 40.776 | 47.175 | 12.049 | 0.837 | 8.937 | 10.963 |
|  | 10BH167 A1 12 | 51.300 | 3.700 | 5.400 | 0.400 | 0.900 | 17.300 | 0.100 | 19.800 | 0.000 | 0.200 | 0.000 | 99.000 | 43.627 | 45.825 | 10.547 | 0.851 | 9.383 | 10.713 |
|  | 10BH167 A1 13 | 51.600 | 3.400 | 6.800 | 0.600 | 0.600 | 17.200 | 0.200 | 19.200 | 0.000 | 0.200 | 0.000 | 99.800 | 41.825 | 45.044 | 13.131 | 0.818 | 7.315 | 11.989 |
|  | $\begin{aligned} & \text { 10BH167 A1 11-13 } \\ & \text { av } \end{aligned}$ | 51.300 | 3.633 | 6.133 | 0.467 | 0.833 | 17.467 | 0.167 | 19.200 | 0.000 | 0.200 | 0.000 | 99.367 | 42.073 | 46.013 | 11.914 | 0.835 | 8.513 | 11.212 |
|  | 10BH167 A1 21 | 53.500 | 1.900 | 15.300 | 0.700 | 0.100 | 26.200 | 0.300 | 2.500 | 0.000 | 0.000 | 0.000 | 100.500 | 5.256 | 66.226 | 28.518 | 0.753 | 6.360 | 12.707 |
|  | 10BH167 A1 22 | 53.200 | 2.000 | 14.400 | 0.600 | 0.200 | 26.500 | 0.300 | 2.500 | 0.100 | 0.100 | 0.000 | 99.800 | 5.305 | 67.606 | 27.089 | 0.766 | 7.108 | 12.137 |
|  | $\begin{aligned} & \text { 10BH167 A1 21-22 } \\ & \text { av } \end{aligned}$ | 53.350 | 1.950 | 14.850 | 0.650 | 0.150 | 26.350 | 0.300 | 2.500 | 0.050 | 0.050 | 0.000 | 100.150 | 5.281 | 66.913 | 27.807 | 0.760 | 6.729 | 12.418 |
| MOZ | 10BH167 A2 4 | 52.300 | 1.900 | 12.300 | 0.500 | 0.300 | 28.000 | 0.300 | 2.400 | 0.100 | 0.100 | 0.000 | 98.200 | 5.110 | 71.673 | 23.217 | 0.802 | 9.284 | 10.768 |
|  | 10BH167 A2 5 | 54.900 | 1.100 | 12.300 | 0.200 | 0.300 | 28.900 | 0.300 | 2.200 | 0.100 | 0.000 | 0.000 | 100.300 | 4.598 | 72.613 | 22.789 | 0.807 | 8.995 | 10.930 |
|  | 10BH167 A2 4-5 av | 53.600 | 1.500 | 12.300 | 0.350 | 0.300 | 28.450 | 0.300 | 2.300 | 0.100 | 0.050 | 0.000 | 99.250 | 4.852 | 72.148 | 23.001 | 0.805 | 9.142 | 10.847 |
|  | 10BH167 A2 9 | 51.100 | 2.400 | 5.400 | 0.500 | 0.700 | 18.100 | 0.100 | 20.500 | 0.000 | 0.200 | 0.000 | 99.000 | 43.574 | 46.251 | 10.175 | 0.857 | 9.933 | 10.422 |
|  | 10BH167 A2 10 | 51.800 | 2.500 | 5.200 | 0.300 | 0.700 | 17.700 | 0.100 | 20.300 | 0.000 | 0.200 | 0.000 | 98.700 | 43.951 | 46.069 | 9.980 | 0.858 | 9.566 | 10.614 |
|  | 10BH167 A2 11 | 51.700 | 3.100 | 6.700 | 0.600 | 0.600 | 17.500 | 0.100 | 19.500 | 0.000 | 0.200 | 0.000 | 100.200 | 41.956 | 45.265 | 12.779 | 0.823 | 7.634 | 11.771 |
|  | 10BH167 A2 9-11 av | 51.533 | 2.667 | 5.767 | 0.467 | 0.667 | 17.767 | 0.100 | 20.100 | 0.000 | 0.200 | 0.000 | 99.300 | 43.160 | 45.863 | 10.977 | 0.846 | 8.999 | 10.928 |
| MOZ | 10BH167 A3 5 | 51.600 | 3.600 | 5.300 | 0.400 | 0.600 | 17.200 | 0.100 | 20.700 | 0.000 | 0.200 | 0.000 | 99.700 | 44.926 | 44.877 | 10.197 | 0.853 | 9.386 | 10.712 |
|  | 10BH167 A3 6 | 51.000 | 4.000 | 5.500 | 0.600 | 1.100 | 17.100 | 0.100 | 20.300 | 0.000 | 0.200 | 0.000 | 100.000 | 44.388 | 44.951 | 10.661 | 0.847 | 9.379 | 10.716 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{MgO} \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 10BH167 A3 7 | 49.500 | 3.800 | 8.200 | 0.600 | 0.500 | 17.900 | 0.300 | 17.400 | 0.000 | 0.200 | 0.000 | 98.500 | 37.672 | 46.590 | 15.738 | 0.795 | 7.127 | 12.123 |
|  | 10BH167 A3 5-7 av | 50.700 | 3.800 | 6.333 | 0.533 | 0.733 | 17.400 | 0.167 | 19.467 | 0.000 | 0.200 | 0.000 | 99.400 | 42.320 | 45.475 | 12.205 | 0.830 | 8.467 | 11.240 |
|  | 10BH167 A3 10 | 52.400 | 2.400 | 5.800 | 0.300 | 0.600 | 18.500 | 0.100 | 19.400 | 0.100 | 0.200 | 0.000 | 99.800 | 41.469 | 47.541 | 10.990 | 0.850 | 9.212 | 10.808 |
|  | 10BH167 A3 11 | 52.100 | 2.900 | 5.800 | 0.500 | 0.700 | 18.300 | 0.100 | 19.100 | 0.000 | 0.200 | 0.000 | 99.800 | 41.305 | 47.576 | 11.119 | 0.849 | 9.248 | 10.788 |
|  | 10BH167 A3 12 | 52.400 | 2.200 | 6.900 | 0.500 | 0.500 | 18.400 | 0.200 | 18.100 | 0.000 | 0.200 | 0.000 | 99.400 | 39.062 | 47.738 | 13.200 | 0.826 | 7.598 | 11.795 |
|  | 10BH167 A3 10-12 <br> av | 52.300 | 2.500 | 6.167 | 0.433 | 0.600 | 18.400 | 0.133 | 18.867 | 0.033 | 0.200 | 0.000 | 99.667 | 40.614 | 47.618 | 11.768 | 0.842 | 8.659 | 11.125 |
|  | 10BH167 A3 20 | 51.000 | 4.300 | 5.600 | 0.400 | 1.200 | 17.200 | 0.100 | 19.800 | 0.100 | 0.200 | 0.000 | 99.800 | 43.572 | 45.503 | 10.924 | 0.846 | 9.359 | 10.726 |
|  | 10BH167 A3 21 | 53.100 | 2.400 | 5.500 | 0.400 | 0.700 | 18.500 | 0.200 | 19.400 | 0.000 | 0.200 | 0.000 | 100.300 | 41.706 | 47.812 | 10.482 | 0.857 | 9.437 | 10.684 |
|  | 10BH167 A3 22 | 52.200 | 2.400 | 7.600 | 1.200 | 0.400 | 17.600 | 0.200 | 18.800 | 0.000 | 0.300 | 0.000 | 100.700 | 40.261 | 45.311 | 14.428 | 0.805 | 6.549 | 12.557 |
|  | $\begin{aligned} & \text { 10BH167 A3 20-22 } \\ & \text { av } \end{aligned}$ | 52.100 | 3.033 | 6.233 | 0.667 | 0.767 | 17.767 | 0.167 | 19.333 | 0.033 | 0.233 | 0.000 | 100.267 | 41.831 | 46.213 | 11.956 | 0.836 | 8.360 | 11.305 |
| UOZ | 10BH167 A4 13 | 51.600 | 3.300 | 5.600 | 0.300 | 1.200 | 17.500 | 0.100 | 20.100 | 0.000 | 0.300 | 0.000 | 100.100 | 43.599 | 45.634 | 10.768 | 0.848 | 9.276 | 10.772 |
|  | 10BH167 A4 14 | 52.300 | 2.500 | 6.200 | 0.300 | 0.700 | 17.900 | 0.200 | 19.300 | 0.000 | 0.200 | 0.000 | 99.600 | 41.671 | 46.462 | 11.867 | 0.837 | 8.186 | 11.413 |
|  | 10BH167 A 415 | 51.900 | 2.500 | 8.900 | 0.900 | 0.100 | 16.800 | 0.200 | 18.600 | 0.000 | 0.300 | 0.000 | 100.300 | 39.840 | 43.260 | 16.899 | 0.771 | 4.696 | 14.262 |
|  | $\begin{aligned} & \text { 10BH167 A4 13-15 } \\ & \text { av } \end{aligned}$ | 51.933 | 2.767 | 6.900 | 0.500 | 0.667 | 17.400 | 0.167 | 19.333 | 0.000 | 0.267 | 0.000 | 100.000 | 41.695 | 45.113 | 13.192 | 0.818 | 7.182 | 12.084 |
|  | 10BH167 A4 16 | 53.600 | 1.900 | 14.500 | 0.700 | 0.100 | 27.100 | 0.300 | 2.100 | 0.000 | 0.100 | 0.000 | 100.400 | 4.418 | 68.540 | 27.042 | 0.769 | 7.295 | 12.004 |
|  | 10BH167 A4 17 | 53.600 | 1.900 | 14.800 | 0.600 | 0.200 | 27.000 | 0.300 | 2.300 | 0.100 | 0.000 | 0.000 | 100.800 | 4.804 | 67.794 | 27.402 | 0.765 | 7.068 | 12.166 |
|  | 10BH167 A4 18 | 53.400 | 1.600 | 15.300 | 0.800 | 0.000 | 26.500 | 0.300 | 1.900 | 0.000 | 0.100 | 0.000 | 100.000 | 4.015 | 67.323 | 28.662 | 0.755 | 6.522 | 12.578 |
|  | $\begin{aligned} & \text { 10BH167 A4 16-18 } \\ & \text { av } \end{aligned}$ | 53.533 | 1.800 | 14.867 | 0.700 | 0.100 | 26.867 | 0.300 | 2.100 | 0.033 | 0.067 | 0.000 | 100.400 | 4.414 | 67.886 | 27.700 | 0.763 | 6.959 | 12.245 |
|  | 10BH167 A4 23 | 52.900 | 2.500 | 5.600 | 0.300 | 0.900 | 18.300 | 0.100 | 19.800 | 0.000 | 0.200 | 0.000 | 100.700 | 42.340 | 47.044 | 10.615 | 0.853 | 9.262 | 10.780 |
|  | 10BH167 A4 24 | 51.900 | 3.300 | 5.900 | 0.300 | 1.100 | 17.700 | 0.100 | 19.500 | 0.000 | 0.200 | 0.000 | 100.000 | 42.383 | 46.249 | 11.368 | 0.842 | 8.812 | 11.035 |
|  | 10BH167 A4 25 | 52.700 | 1.900 | 6.600 | 0.300 | 0.700 | 18.400 | 0.200 | 19.100 | 0.000 | 0.200 | 0.000 | 100.200 | 40.577 | 46.993 | 12.430 | 0.832 | 7.897 | 11.597 |
|  | $\begin{aligned} & \text { 10BH167 A4 23-25 } \\ & \text { av } \end{aligned}$ | 52.500 | 2.567 | 6.033 | 0.300 | 0.900 | 18.133 | 0.133 | 19.467 | 0.000 | 0.200 | 0.000 | 100.300 | 41.761 | 46.766 | 11.474 | 0.843 | 8.640 | 11.136 |
| UOZ | BH311 B3 28 | 51.746 | 2.573 | 9.000 | 1.060 | 0.026 | 15.888 | 0.195 | 20.233 | 0.053 | 0.351 | 0.009 | 100.091 | 42.720 | 40.328 | 16.952 | 0.759 | 4.127 | 14.924 |
|  | BH311 B3 28 | 51.076 | 2.956 | 9.045 | 1.457 | 0.043 | 16.360 | 0.220 | 19.284 | 0.080 | 0.345 | 0.012 | 100.878 | 41.056 | 41.873 | 17.071 | 0.763 | 4.774 | 14.177 |
|  | BH311 B3 28 | 51.296 | 2.923 | 8.453 | 1.008 | 0.352 | 16.771 | 0.238 | 19.002 | 0.060 | 0.272 | 0.003 | 100.378 | 40.727 | 43.213 | 16.060 | 0.780 | 5.410 | 13.536 |
|  | BH311 B3 28 | 51.464 | 2.798 | 7.952 | 0.866 | 0.448 | 16.656 | 0.227 | 19.070 | 0.015 | 0.269 | 0.018 | 99.783 | 41.328 | 43.395 | 15.277 | 0.789 | 5.600 | 13.360 |
|  | BH311 B3 28 | 51.640 | 3.027 | 8.444 | 0.665 | 0.534 | 17.255 | 0.196 | 18.500 | 0.032 | 0.279 | 0.018 | 100.590 | 39.590 | 44.391 | 16.019 | 0.785 | 5.720 | 13.251 |
|  | BH311 B3 28 | 51.973 | 2.932 | 7.874 | 0.637 | 0.626 | 17.149 | 0.248 | 18.705 | 0.039 | 0.269 | 0.000 | 100.452 | 40.398 | 44.526 | 15.075 | 0.795 | 5.998 | 13.007 |
|  | BH311 B3 28 | 51.611 | 2.870 | 7.526 | 0.695 | 0.632 | 17.319 | 0.223 | 19.039 | 0.042 | 0.235 | 0.010 | 100.202 | 40.917 | 44.745 | 14.338 | 0.804 | 6.601 | 12.516 |
|  | BH311 B3 28 | 51.863 | 3.158 | 7.482 | 0.437 | 0.880 | 17.319 | 0.184 | 18.993 | 0.000 | 0.258 | 0.007 | 100.581 | 40.893 | 44.827 | 14.280 | 0.805 | 6.649 | 12.479 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{MgO} \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | BH311 B3 28 | 52.170 | 3.115 | 6.935 | 0.624 | 0.781 | 17.017 | 0.170 | 19.713 | 0.067 | 0.260 | 0.005 | 100.857 | 42.560 | 44.167 | 13.273 | 0.814 | 6.911 | 12.281 |
|  | BH311 B3 28 | 51.816 | 3.308 | 7.114 | 0.552 | 0.924 | 16.915 | 0.166 | 19.879 | 0.049 | 0.234 | 0.007 | 100.964 | 42.732 | 43.712 | 13.556 | 0.809 | 6.807 | 12.359 |
|  | BH311 B3 28 | 51.965 | 3.248 | 6.957 | 0.552 | 0.868 | 16.963 | 0.194 | 19.904 | 0.037 | 0.272 | 0.014 | 100.974 | 42.838 | 43.889 | 13.273 | 0.813 | 6.980 | 12.230 |
|  | BH311 B3 28 | 52.017 | 3.302 | 6.310 | 0.546 | 0.918 | 16.856 | 0.151 | 20.061 | 0.001 | 0.257 | 0.009 | 100.428 | 43.688 | 44.130 | 12.182 | 0.826 | 7.574 | 11.812 |
|  | BH311 B3 28 | 51.228 | 4.146 | 6.933 | 0.544 | 0.898 | 17.213 | 0.122 | 18.098 | 0.001 | 0.399 | 0.119 | 99.701 | 40.274 | 46.049 | 13.677 | 0.816 | 7.582 | 11.806 |
|  | BH311 B3 28 | 51.757 | 3.378 | 6.591 | 0.416 | 1.070 | 17.032 | 0.162 | 20.324 | 0.035 | 0.226 | 0.009 | 101.000 | 43.574 | 43.899 | 12.527 | 0.822 | 7.578 | 11.809 |
|  | BH311 B3 28 | 51.481 | 4.018 | 6.076 | 0.359 | 1.242 | 16.726 | 0.142 | 20.647 | 0.040 | 0.264 | 0.021 | 101.016 | 44.748 | 43.579 | 11.673 | 0.831 | 8.235 | 11.382 |
|  | BH311 B3 28 | 52.439 | 3.244 | 5.489 | 0.310 | 0.978 | 17.122 | 0.156 | 20.684 | 0.021 | 0.247 | 0.011 | 100.701 | 44.835 | 44.618 | 10.547 | 0.848 | 8.801 | 11.042 |
|  | BH311 B3 28 | 52.323 | 3.067 | 5.724 | 0.360 | 0.924 | 17.103 | 0.132 | 20.820 | 0.007 | 0.253 | 0.017 | 100.730 | 44.817 | 44.260 | 10.923 | 0.842 | 8.470 | 11.238 |
|  | BH311 B3 28 | 52.503 | 2.984 | 5.485 | 0.324 | 0.952 | 17.387 | 0.158 | 20.587 | 0.053 | 0.227 | 0.006 | 100.666 | 44.415 | 45.095 | 10.490 | 0.850 | 8.952 | 10.955 |
|  | BH311 B3 28 | 52.541 | 2.969 | 5.744 | 0.360 | 0.845 | 17.459 | 0.166 | 20.239 | 0.038 | 0.244 | 0.002 | 100.607 | 43.694 | 45.313 | 10.993 | 0.844 | 8.607 | 11.156 |
|  | BH311 B3 28 | 52.364 | 3.060 | 5.681 | 0.331 | 0.933 | 17.464 | 0.162 | 20.223 | 0.064 | 0.239 | 0.005 | 100.526 | 43.722 | 45.391 | 10.888 | 0.846 | 8.787 | 11.050 |
|  | BH311 B3 28 | 52.458 | 3.036 | 5.907 | 0.374 | 0.889 | 17.548 | 0.164 | 20.119 | 0.023 | 0.238 | 0.003 | 100.759 | 43.312 | 45.415 | 11.273 | 0.841 | 8.523 | 11.206 |
|  | BH311 B3 28 | 52.777 | 2.869 | 6.212 | 0.316 | 0.835 | 17.489 | 0.167 | 19.881 | 0.051 | 0.256 | 0.017 | 100.870 | 42.835 | 45.300 | 11.865 | 0.834 | 7.908 | 11.590 |
|  | BH311 B3 28 | 52.334 | 2.814 | 6.134 | 0.352 | 0.788 | 17.681 | 0.171 | 19.691 | 0.019 | 0.225 | 0.013 | 100.222 | 42.452 | 45.825 | 11.723 | 0.837 | 8.247 | 11.375 |
|  | BH311 B3 28 | 52.620 | 2.556 | 6.412 | 0.424 | 0.813 | 17.877 | 0.156 | 19.540 | 0.039 | 0.261 | 0.003 | 100.701 | 41.828 | 46.005 | 12.167 | 0.832 | 7.981 | 11.543 |
|  | BH311 B3 28 | 52.965 | 2.477 | 6.440 | 0.395 | 0.822 | 17.895 | 0.183 | 19.244 | 0.043 | 0.248 | 0.015 | 100.727 | 41.415 | 46.299 | 12.286 | 0.832 | 7.786 | 11.670 |
|  | BH311 B3 28 | 52.912 | 2.407 | 7.034 | 0.380 | 0.794 | 17.972 | 0.223 | 18.971 | 0.017 | 0.248 | 0.013 | 100.971 | 40.526 | 46.154 | 13.320 | 0.820 | 7.177 | 12.088 |
|  | BH311 B3 28 | 52.606 | 2.458 | 7.572 | 0.494 | 0.740 | 17.782 | 0.174 | 18.604 | 0.039 | 0.247 | 0.015 | 100.731 | 39.843 | 45.782 | 14.375 | 0.807 | 6.523 | 12.578 |
|  | BH311 B3 28 | 52.386 | 2.459 | 8.423 | 0.686 | 0.482 | 17.612 | 0.228 | 18.343 | 0.000 | 0.266 | 0.008 | 100.893 | 39.042 | 45.065 | 15.893 | 0.788 | 5.665 | 13.300 |
|  | BH311 B3 28 | 52.289 | 2.592 | 7.001 | 0.395 | 0.314 | 17.484 | 0.204 | 18.375 | 0.026 | 0.428 | 0.047 | 99.155 | 40.296 | 46.094 | 13.610 | 0.817 | 6.910 | 12.282 |
|  | BH311 B3 28 | 51.211 | 2.939 | 8.958 | 1.043 | 0.299 | 16.787 | 0.196 | 18.821 | 0.029 | 0.317 | 0.018 | 100.618 | 40.093 | 42.990 | 16.916 | 0.770 | 5.078 | 13.861 |
|  | BH311 B3 28 av | 52.061 | 2.989 | 7.030 | 0.565 | 0.722 | 17.203 | 0.183 | 19.520 | 0.034 | 0.271 | 0.015 | 100.559 | 42.039 | 44.544 | 13.417 | 0.814 | 6.952 | 12.251 |
|  | BH311 B3 29 | 50.752 | 2.728 | 8.403 | 1.051 | 0.269 | 16.582 | 0.225 | 19.330 | 0.000 | 0.317 | 0.008 | 99.665 | 41.380 | 42.674 | 15.946 | 0.779 | 5.418 | 13.529 |
|  | BH311 B3 29 | 51.337 | 2.719 | 8.218 | 0.823 | 0.439 | 16.991 | 0.189 | 19.364 | 0.000 | 0.266 | 0.005 | 100.351 | 41.134 | 43.391 | 15.475 | 0.787 | 5.752 | 13.222 |
|  | BH311 B3 29 | 51.372 | 2.730 | 7.433 | 0.767 | 0.562 | 17.101 | 0.174 | 19.347 | 0.013 | 0.276 | 0.027 | 99.802 | 41.611 | 44.217 | 14.172 | 0.804 | 6.561 | 12.547 |
|  | BH311 B3 29 | 51.289 | 3.427 | 6.512 | 0.294 | 1.066 | 16.974 | 0.145 | 20.275 | 0.059 | 0.232 | 0.012 | 100.285 | 43.646 | 43.927 | 12.427 | 0.823 | 7.732 | 11.706 |
|  | BH311 B3 29 | 51.501 | 3.687 | 5.493 | 0.453 | 1.087 | 16.950 | 0.122 | 20.719 | 0.005 | 0.220 | 0.021 | 100.258 | 45.075 | 44.331 | 10.594 | 0.846 | 9.061 | 10.892 |
|  | BH311 B3 29 | 51.373 | 3.380 | 4.878 | 0.295 | 1.212 | 16.668 | 0.130 | 21.611 | 0.000 | 0.155 | 0.005 | 99.707 | 47.008 | 43.586 | 9.406 | 0.859 | 9.680 | 10.554 |
|  | BH311 B3 29 | 50.629 | 3.699 | 5.318 | 0.338 | 1.193 | 16.917 | 0.153 | 20.710 | 0.031 | 0.234 | 0.013 | 99.235 | 45.256 | 44.442 | 10.302 | 0.850 | 9.542 | 10.627 |
|  | BH311 B3 29 | 52.422 | 2.271 | 4.778 | 0.224 | 1.059 | 16.540 | 0.170 | 22.827 | 0.042 | 0.149 | 0.008 | 100.490 | 48.623 | 42.355 | 9.022 | 0.861 | 9.200 | 10.815 |
|  | BH311 B3 29 | 47.376 | 5.618 | 8.880 | 0.249 | 0.964 | 16.638 | 0.227 | 16.119 | 0.039 | 0.172 | 0.024 | 96.306 | 36.640 | 45.466 | 17.894 | 0.770 | 6.279 | 12.773 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{MgO} \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | BH311 B3 29 | 51.288 | 3.555 | 5.767 | 0.201 | 1.235 | 17.133 | 0.152 | 20.381 | 0.000 | 0.242 | 0.007 | 99.961 | 44.220 | 44.688 | 11.092 | 0.841 | 8.856 | 11.010 |
|  | BH311 B3 29 | 51.010 | 3.519 | 5.622 | 0.359 | 1.183 | 17.349 | 0.170 | 20.394 | 0.024 | 0.203 | 0.012 | 99.845 | 44.110 | 45.111 | 10.779 | 0.846 | 9.333 | 10.741 |
|  | BH311 B3 29 | 51.186 | 3.463 | 5.371 | 0.417 | 1.236 | 17.346 | 0.153 | 20.495 | 0.001 | 0.210 | 0.012 | 99.890 | 44.449 | 45.225 | 10.326 | 0.852 | 9.635 | 10.578 |
|  | BH311 B3 29 | 51.163 | 3.408 | 5.399 | 0.410 | 1.241 | 17.304 | 0.099 | 20.534 | 0.049 | 0.233 | 0.000 | 99.840 | 44.520 | 45.103 | 10.377 | 0.851 | 9.563 | 10.616 |
|  | BH311 B3 29 | 51.885 | 2.805 | 6.073 | 0.366 | 0.893 | 17.953 | 0.164 | 19.719 | 0.000 | 0.234 | 0.009 | 100.101 | 42.238 | 46.230 | 11.532 | 0.840 | 8.756 | 11.068 |
|  | BH311 B3 29 | 51.466 | 2.946 | 5.313 | 0.151 | 1.175 | 17.655 | 0.136 | 20.447 | 0.022 | 0.223 | 0.002 | 99.536 | 44.085 | 45.761 | 10.155 | 0.856 | 9.725 | 10.530 |
|  | BH311 B3 29 | 51.960 | 2.695 | 5.413 | 0.360 | 1.039 | 17.700 | 0.091 | 20.511 | 0.000 | 0.223 | 0.006 | 99.998 | 44.026 | 45.674 | 10.300 | 0.854 | 9.423 | 10.692 |
|  | BH311 B3 29 | 51.763 | 2.789 | 5.475 | 0.367 | 0.987 | 17.666 | 0.109 | 20.412 | 0.066 | 0.226 | 0.010 | 99.870 | 43.894 | 45.669 | 10.437 | 0.852 | 9.388 | 10.711 |
|  | BH311 B3 29 | 52.072 | 2.836 | 5.633 | 0.309 | 0.789 | 17.564 | 0.120 | 20.475 | 0.034 | 0.241 | 0.002 | 100.075 | 43.953 | 45.327 | 10.719 | 0.847 | 8.961 | 10.950 |
|  | BH311 B3 29 | 51.957 | 2.935 | 5.676 | 0.360 | 0.876 | 17.655 | 0.153 | 20.525 | 0.026 | 0.211 | 0.003 | 100.377 | 43.875 | 45.370 | 10.756 | 0.847 | 9.081 | 10.881 |
|  | BH311 B3 29 | 51.463 | 2.887 | 5.519 | 0.374 | 0.798 | 17.356 | 0.138 | 20.501 | 0.034 | 0.240 | 0.009 | 99.319 | 44.318 | 45.105 | 10.576 | 0.849 | 9.119 | 10.860 |
|  | BH311 B3 29 | 51.894 | 2.645 | 5.765 | 0.425 | 0.631 | 17.573 | 0.143 | 20.229 | 0.006 | 0.216 | 0.001 | 99.528 | 43.536 | 45.466 | 10.998 | 0.845 | 8.719 | 11.090 |
|  | BH311 B3 29 | 51.670 | 2.686 | 6.686 | 0.495 | 0.930 | 17.511 | 0.159 | 19.879 | 0.031 | 0.213 | 0.000 | 100.260 | 42.425 | 44.927 | 12.649 | 0.824 | 7.680 | 11.740 |
|  | BH311 B3 29 | 52.079 | 2.427 | 6.697 | 0.330 | 0.690 | 17.715 | 0.181 | 19.304 | 0.000 | 0.226 | 0.011 | 99.660 | 41.481 | 45.763 | 12.757 | 0.825 | 7.507 | 11.857 |
|  | BH311 B3 29 | 47.406 | 2.019 | 8.747 | 5.162 | 0.472 | 15.070 | 0.323 | 19.311 | 0.031 | 0.122 | 0.012 | 98.675 | 42.741 | 40.098 | 17.162 | 0.754 | 5.029 | 13.911 |
|  | BH311 B3 29 | 52.184 | 2.163 | 7.292 | 0.366 | 0.549 | 18.016 | 0.173 | 18.866 | 0.018 | 0.229 | 0.006 | 99.862 | 40.150 | 46.093 | 13.757 | 0.815 | 7.005 | 12.212 |
|  | BH311 B3 29 | 52.136 | 2.144 | 7.507 | 0.552 | 0.462 | 18.273 | 0.226 | 18.413 | 0.000 | 0.238 | 0.007 | 99.958 | 39.147 | 46.704 | 14.148 | 0.813 | 7.029 | 12.194 |
|  | BH311 B3 29 | 48.513 | 3.696 | 9.751 | 1.096 | 0.328 | 16.395 | 0.307 | 17.313 | 0.049 | 0.311 | 0.019 | 97.778 | 37.912 | 43.160 | 18.928 | 0.750 | 4.932 | 14.010 |
|  | BH311 B3 29 | 46.944 | 4.493 | 10.090 | 1.187 | 0.203 | 16.826 | 0.280 | 16.504 | 0.031 | 0.266 | 0.290 | 97.114 | 36.132 | 44.285 | 19.582 | 0.748 | 5.633 | 13.329 |
|  | BH311 B3 29 | 51.229 | 2.435 | 8.629 | 0.951 | 0.295 | 16.985 | 0.221 | 18.894 | 0.084 | 0.287 | 0.009 | 100.019 | 40.232 | 43.480 | 16.288 | 0.778 | 5.341 | 13.602 |
|  | BH311 B3 29 | 51.008 | 2.545 | 8.301 | 1.130 | 0.222 | 16.551 | 0.216 | 19.277 | 0.034 | 0.289 | 0.007 | 99.580 | 41.427 | 42.760 | 15.814 | 0.780 | 5.321 | 13.622 |
|  | BH311 B3 29 av | 51.011 | 3.045 | 6.688 | 0.662 | 0.803 | 17.165 | 0.175 | 19.756 | 0.024 | 0.230 | 0.019 | 99.578 | 42.641 | 44.546 | 12.812 | 0.821 | 7.627 | 11.776 |
|  | BH311 B3 31 | 50.007 | 3.086 | 9.123 | 1.457 | 0.238 | 16.494 | 0.222 | 18.993 | 0.018 | 0.299 | 0.009 | 99.946 | 40.489 | 42.271 | 17.240 | 0.763 | 5.120 | 13.819 |
|  | BH311 B3 32 | 50.927 | 2.449 | 8.128 | 1.131 | 0.222 | 16.361 | 0.211 | 19.834 | 0.063 | 0.329 | 0.013 | 99.668 | 42.464 | 42.110 | 15.426 | 0.782 | 5.383 | 13.562 |
|  | BH311 B3 31-32 av | 50.467 | 2.768 | 8.626 | 1.294 | 0.230 | 16.428 | 0.217 | 19.414 | 0.041 | 0.314 | 0.011 | 99.807 | 41.476 | 42.190 | 16.333 | 0.773 | 5.244 | 13.696 |
|  | BH311 B3 33 | 50.708 | 2.506 | 8.366 | 1.045 | 0.196 | 16.925 | 0.192 | 19.047 | 0.076 | 0.263 | 0.002 | 99.326 | 40.690 | 43.467 | 15.843 | 0.783 | 5.642 | 13.321 |
|  | BH311 B3 33 | 52.028 | 2.046 | 7.601 | 0.480 | 0.318 | 18.383 | 0.216 | 17.892 | 0.000 | 0.221 | 0.004 | 99.189 | 38.288 | 47.293 | 14.419 | 0.812 | 6.908 | 12.283 |
|  | BH311 B3 33 | 52.222 | 2.168 | 7.188 | 0.531 | 0.375 | 18.589 | 0.193 | 18.608 | 0.037 | 0.232 | 0.020 | 100.163 | 39.318 | 47.219 | 13.464 | 0.822 | 7.638 | 11.768 |
|  | BH311 B3 33 | 52.018 | 2.234 | 6.621 | 0.581 | 0.559 | 18.329 | 0.170 | 19.052 | 0.034 | 0.214 | 0.001 | 99.813 | 40.574 | 46.926 | 12.500 | 0.831 | 8.143 | 11.440 |
|  | BH311 B3 33 | 51.962 | 2.318 | 6.164 | 0.230 | 0.704 | 18.050 | 0.170 | 19.198 | 0.000 | 0.214 | 0.000 | 99.010 | 41.409 | 46.805 | 11.786 | 0.839 | 8.399 | 11.282 |
|  | BH311 B3 33 | 51.557 | 2.735 | 6.289 | 0.410 | 0.796 | 17.833 | 0.160 | 19.601 | 0.000 | 0.226 | 0.008 | 99.615 | 42.049 | 45.991 | 11.960 | 0.835 | 8.416 | 11.271 |
|  | BH311 B3 33 | 51.625 | 2.898 | 5.785 | 0.230 | 0.915 | 17.726 | 0.169 | 20.217 | 0.034 | 0.198 | 0.000 | 99.797 | 43.333 | 45.675 | 10.992 | 0.845 | 9.024 | 10.913 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | Melt <br> MgO | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | BH311 B3 33 | 51.653 | 2.570 | 5.545 | 0.273 | 0.973 | 17.465 | 0.156 | 20.350 | 0.022 | 0.202 | 0.009 | 99.218 | 43.989 | 45.386 | 10.625 | 0.849 | 9.002 | 10.926 |
|  | BH311 B3 33 | 51.692 | 2.810 | 5.299 | 0.424 | 1.171 | 17.635 | 0.149 | 20.447 | 0.035 | 0.204 | 0.015 | 99.881 | 44.119 | 45.745 | 10.136 | 0.856 | 9.665 | 10.561 |
|  | BH311 B3 33 | 51.478 | 3.100 | 5.538 | 0.209 | 1.201 | 17.824 | 0.167 | 20.513 | 0.055 | 0.209 | 0.008 | 100.302 | 43.784 | 45.737 | 10.479 | 0.852 | 9.659 | 10.565 |
|  | BH311 B3 33 | 49.243 | 7.202 | 6.749 | 0.688 | 0.637 | 13.973 | 0.181 | 19.290 | 0.003 | 0.745 | 0.027 | 98.738 | 45.851 | 39.928 | 14.221 | 0.787 | 6.178 | 12.856 |
|  | BH311 B3 33 | 51.364 | 3.535 | 5.618 | 0.273 | 1.008 | 17.517 | 0.158 | 19.913 | 0.046 | 0.206 | 0.000 | 99.638 | 43.335 | 45.828 | 10.838 | 0.847 | 9.284 | 10.768 |
|  | BH311 B3 33 | 51.476 | 3.509 | 5.805 | 0.338 | 1.204 | 17.981 | 0.110 | 19.509 | 0.047 | 0.202 | 0.004 | 100.185 | 42.162 | 46.717 | 11.121 | 0.847 | 9.437 | 10.684 |
|  | BH311 B3 33 | 51.190 | 3.322 | 5.656 | 0.338 | 1.165 | 17.888 | 0.126 | 19.647 | 0.055 | 0.217 | 0.014 | 99.618 | 42.558 | 46.582 | 10.861 | 0.849 | 9.596 | 10.598 |
|  | BH311 B3 33 | 51.939 | 2.881 | 5.387 | 0.230 | 1.092 | 18.074 | 0.135 | 19.653 | 0.064 | 0.202 | 0.000 | 99.657 | 42.579 | 47.075 | 10.346 | 0.857 | 9.745 | 10.519 |
|  | BH311 B3 33 | 51.737 | 2.525 | 5.762 | 0.309 | 0.867 | 17.795 | 0.136 | 20.286 | 0.000 | 0.224 | 0.004 | 99.645 | 43.358 | 45.724 | 10.917 | 0.846 | 8.999 | 10.927 |
|  | BH311 B3 33 | 51.826 | 3.045 | 5.692 | 0.295 | 0.854 | 17.630 | 0.152 | 20.168 | 0.029 | 0.220 | 0.004 | 99.915 | 43.458 | 45.670 | 10.873 | 0.847 | 9.039 | 10.905 |
|  | BH311 B3 33 | 51.685 | 2.517 | 7.012 | 0.330 | 0.598 | 17.801 | 0.171 | 19.015 | 0.063 | 0.230 | 0.006 | 99.428 | 40.778 | 45.892 | 13.330 | 0.819 | 7.360 | 11.959 |
|  | BH311 B3 33 | 52.260 | 2.427 | 7.651 | 0.409 | 0.383 | 18.050 | 0.189 | 18.359 | 0.000 | 0.203 | 0.004 | 99.935 | 39.194 | 46.326 | 14.480 | 0.808 | 6.637 | 12.489 |
|  | BH311 B3 33 | 50.850 | 2.532 | 8.316 | 0.866 | 0.296 | 16.723 | 0.172 | 19.470 | 0.000 | 0.284 | 0.008 | 99.517 | 41.473 | 42.824 | 15.703 | 0.782 | 5.497 | 13.454 |
|  | BH311 B3 33 av | 51.526 | 2.944 | 6.402 | 0.424 | 0.766 | 17.610 | 0.164 | 19.512 | 0.030 | 0.246 | 0.007 | 99.630 | 42.115 | 45.640 | 12.245 | 0.831 | 8.131 | 11.448 |
|  | BH311 B3 37 | 51.780 | 3.821 | 5.857 | 0.618 | 1.029 | 16.824 | 0.133 | 20.582 | 0.038 | 0.229 | 0.004 | 100.915 | 44.744 | 43.969 | 11.287 | 0.837 | 8.423 | 11.267 |
|  | BH311 B3 37 | 51.605 | 3.770 | 5.738 | 0.345 | 1.170 | 16.918 | 0.150 | 20.883 | 0.000 | 0.226 | 0.006 | 100.811 | 45.096 | 43.920 | 10.984 | 0.840 | 8.727 | 11.085 |
|  | BH311 B3 37 | 51.610 | 3.391 | 5.637 | 0.381 | 1.035 | 17.091 | 0.126 | 20.928 | 0.050 | 0.231 | 0.000 | 100.480 | 45.034 | 44.213 | 10.753 | 0.844 | 8.914 | 10.976 |
|  | BH311 B3 37 | 51.416 | 3.247 | 5.477 | 0.410 | 1.056 | 17.675 | 0.121 | 20.007 | 0.017 | 0.194 | 0.008 | 99.628 | 43.389 | 46.082 | 10.529 | 0.852 | 9.557 | 10.619 |
|  | BH311 B3 37 | 52.000 | 3.174 | 5.172 | 0.194 | 1.290 | 17.128 | 0.103 | 20.802 | 0.048 | 0.270 | 0.017 | 100.198 | 45.244 | 44.785 | 9.972 | 0.855 | 9.434 | 10.686 |
|  | BH311 B3 37 | 51.658 | 3.489 | 5.548 | 0.209 | 1.275 | 17.280 | 0.157 | 20.255 | 0.030 | 0.257 | 0.004 | 100.162 | 44.084 | 45.212 | 10.704 | 0.847 | 9.172 | 10.830 |
|  | BH311 B3 37 | 51.308 | 2.934 | 5.207 | 0.345 | 1.338 | 17.136 | 0.158 | 21.048 | 0.021 | 0.195 | 0.010 | 99.700 | 45.495 | 44.528 | 9.977 | 0.854 | 9.547 | 10.624 |
|  | BH311 B3 37 | 51.669 | 3.412 | 5.196 | 0.446 | 1.431 | 17.150 | 0.141 | 20.765 | 0.067 | 0.248 | 0.005 | 100.530 | 45.153 | 44.832 | 10.016 | 0.855 | 9.655 | 10.567 |
|  | BH311 B3 37 | 51.479 | 3.458 | 5.239 | 0.338 | 1.416 | 17.431 | 0.156 | 20.378 | 0.011 | 0.228 | 0.008 | 100.142 | 44.322 | 45.577 | 10.101 | 0.856 | 9.836 | 10.472 |
|  | BH311 B3 37 | 52.832 | 1.505 | 6.029 | 0.115 | 1.304 | 15.767 | 0.167 | 22.182 | 0.061 | 0.326 | 0.008 | 100.296 | 47.722 | 40.779 | 11.498 | 0.823 | 6.270 | 12.780 |
|  | BH311 B3 37 | 52.381 | 2.718 | 5.420 | 0.381 | 1.018 | 18.055 | 0.145 | 19.857 | 0.021 | 0.213 | 0.003 | 100.212 | 42.826 | 46.812 | 10.362 | 0.856 | 9.554 | 10.621 |
|  | BH311 B3 37 | 51.531 | 3.143 | 6.434 | 0.323 | 0.786 | 18.707 | 0.198 | 18.223 | 0.025 | 0.229 | 0.034 | 99.633 | 39.260 | 48.452 | 12.288 | 0.838 | 9.025 | 10.913 |
|  | BH311 B3 37 | 50.896 | 3.669 | 6.279 | 0.381 | 0.813 | 16.425 | 0.208 | 20.431 | 0.019 | 0.313 | 0.012 | 99.446 | 44.665 | 43.167 | 12.168 | 0.823 | 7.672 | 11.746 |
|  | BH311 B3 37 | 51.641 | 3.481 | 5.813 | 0.547 | 0.902 | 16.346 | 0.188 | 21.140 | 0.099 | 0.283 | 0.003 | 100.443 | 46.013 | 42.771 | 11.216 | 0.834 | 8.026 | 11.514 |
|  | BH311 B3 37 | 51.390 | 3.104 | 5.748 | 0.295 | 1.160 | 17.790 | 0.158 | 19.664 | 0.045 | 0.258 | 0.022 | 99.634 | 42.612 | 46.346 | 11.042 | 0.847 | 9.293 | 10.763 |
|  | BH311 B3 37 | 52.256 | 2.980 | 5.855 | 0.360 | 1.064 | 18.118 | 0.145 | 19.900 | 0.019 | 0.208 | 0.004 | 100.909 | 42.457 | 46.470 | 11.073 | 0.846 | 9.173 | 10.829 |
|  | BH311 B3 37 | 50.564 | 4.802 | 6.670 | 0.552 | 0.948 | 16.354 | 0.147 | 19.950 | 0.000 | 0.236 | 0.000 | 100.223 | 43.824 | 43.188 | 12.988 | 0.814 | 7.497 | 11.864 |
|  | BH311 B3 37 | 51.972 | 3.286 | 5.591 | 0.302 | 1.210 | 17.502 | 0.141 | 20.231 | 0.030 | 0.252 | 0.006 | 100.523 | 43.764 | 45.515 | 10.721 | 0.848 | 9.181 | 10.825 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{MgO} \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | BH311 B3 37 | 51.526 | 3.417 | 5.855 | 0.431 | 1.243 | 17.544 | 0.125 | 19.920 | 0.071 | 0.244 | 0.004 | 100.380 | 43.116 | 45.650 | 11.234 | 0.842 | 9.031 | 10.910 |
|  | BH311 B3 37 | 51.590 | 3.345 | 5.805 | 0.431 | 1.143 | 17.150 | 0.313 | 20.128 | 0.048 | 0.231 | 0.005 | 100.189 | 43.860 | 44.927 | 11.213 | 0.840 | 8.694 | 11.105 |
|  | BH311 B3 37 | 51.405 | 3.482 | 5.768 | 0.388 | 1.222 | 17.593 | 0.155 | 19.881 | 0.047 | 0.245 | 0.001 | 100.187 | 43.085 | 45.835 | 11.081 | 0.845 | 9.226 | 10.800 |
|  | BH311 B3 37 | 51.542 | 3.567 | 5.940 | 0.302 | 1.274 | 17.487 | 0.166 | 19.873 | 0.006 | 0.236 | 0.012 | 100.405 | 43.051 | 45.542 | 11.407 | 0.840 | 8.874 | 10.999 |
|  | BH311 B3 37 | 52.237 | 3.014 | 5.596 | 0.446 | 0.938 | 17.639 | 0.139 | 20.222 | 0.042 | 0.228 | 0.000 | 100.501 | 43.593 | 45.713 | 10.694 | 0.849 | 9.104 | 10.868 |
|  | BH311 B3 37 | 52.567 | 2.731 | 5.373 | 0.346 | 0.974 | 17.570 | 0.165 | 20.545 | 0.034 | 0.209 | 0.009 | 100.523 | 44.249 | 45.493 | 10.258 | 0.854 | 9.180 | 10.826 |
|  | BH311 B3 37 | 52.202 | 2.596 | 5.682 | 0.389 | 0.780 | 17.784 | 0.131 | 20.208 | 0.000 | 0.208 | 0.010 | 99.990 | 43.342 | 45.855 | 10.803 | 0.848 | 8.948 | 10.957 |
|  | BH311 B3 37 | 52.399 | 2.504 | 6.050 | 0.345 | 0.727 | 18.080 | 0.160 | 19.877 | 0.004 | 0.229 | 0.010 | 100.385 | 42.313 | 46.270 | 11.417 | 0.842 | 8.651 | 11.130 |
|  | BH311 B3 37 | 52.711 | 2.304 | 6.252 | 0.424 | 0.608 | 18.163 | 0.163 | 19.380 | 0.011 | 0.219 | 0.007 | 100.242 | 41.448 | 46.699 | 11.853 | 0.838 | 8.239 | 11.380 |
|  | BH311 B3 37 | 51.581 | 2.302 | 6.994 | 0.186 | 0.580 | 17.933 | 0.207 | 17.167 | 0.051 | 0.271 | 0.010 | 97.282 | 38.212 | 47.988 | 13.800 | 0.820 | 7.177 | 12.088 |
|  | BH311 B3 37 | 52.763 | 2.187 | 6.891 | 0.402 | 0.633 | 18.615 | 0.174 | 18.861 | 0.026 | 0.245 | 0.002 | 100.799 | 39.835 | 47.264 | 12.902 | 0.828 | 7.880 | 11.609 |
|  | BH311 B3 37 | 52.725 | 2.135 | 6.843 | 0.259 | 0.590 | 18.369 | 0.218 | 19.433 | 0.044 | 0.219 | 0.000 | 100.835 | 40.841 | 46.410 | 12.749 | 0.827 | 7.727 | 11.709 |
|  | BH311 B3 37 av | 51.841 | 3.099 | 5.865 | 0.363 | 1.032 | 17.454 | 0.162 | 20.091 | 0.033 | 0.239 | 0.007 | 100.187 | 43.422 | 45.342 | 11.236 | 0.841 | 8.706 | 11.097 |
|  | BH311 B3 45 | 54.036 | 1.802 | 13.535 | 0.532 | 0.216 | 27.431 | 0.318 | 2.123 | 0.052 | 0.048 | 0.007 | 100.100 | 4.508 | 70.017 | 25.475 | 0.783 | 7.827 | 11.643 |
|  | BH311 B3 46 | 51.385 | 2.305 | 14.287 | 0.580 | 0.164 | 26.515 | 0.296 | 1.863 | 0.045 | 0.053 | 0.013 | 97.506 | 4.015 | 68.692 | 27.293 | 0.768 | 7.544 | 11.832 |
|  | BH311 B3 47 | 53.881 | 1.538 | 14.189 | 0.629 | 0.004 | 27.410 | 0.312 | 1.887 | 0.031 | 0.047 | 0.013 | 99.941 | 3.980 | 69.494 | 26.527 | 0.775 | 7.431 | 11.909 |
|  | BH311 B3 45-47 av |  | 1.882 | 14.004 | 0.580 | 0.128 | 27.119 | 0.309 | 1.958 | 0.043 | 0.049 | 0.011 | 99.182 | 4.167 | 69.401 | 26.432 | 0.775 | 7.598 | 11.795 |
| CPZ | BH311 B4 1 | 50.879 | 2.618 | 13.019 | 0.698 | 0.213 | 15.154 | 0.274 | 16.504 | 0.000 | 0.246 | 0.025 | 99.630 | 35.674 | 39.379 | 24.946 | 0.675 | 1.615 | 19.734 |
|  | BH311 B4 1 | 52.109 | 3.347 | 5.636 | 0.408 | 0.973 | 17.753 | 0.138 | 19.675 | 0.056 | 0.230 | 0.010 | 100.335 | 42.759 | 46.383 | 10.858 | 0.849 | 9.226 | 10.800 |
|  | BH311 B4 1 | 52.395 | 2.997 | 5.310 | 0.301 | 0.910 | 17.887 | 0.160 | 20.071 | 0.042 | 0.208 | 0.009 | 100.290 | 43.367 | 46.462 | 10.171 | 0.857 | 9.622 | 10.585 |
|  | BH311 B4 1 | 52.478 | 2.847 | 6.669 | 0.279 | 0.470 | 17.556 | 0.153 | 19.492 | 0.036 | 0.202 | 0.006 | 100.188 | 41.910 | 45.379 | 12.711 | 0.824 | 7.341 | 11.971 |
|  | BH311 B4 1 | 51.603 | 3.443 | 6.271 | 0.437 | 0.745 | 17.757 | 0.160 | 19.468 | 0.012 | 0.197 | 0.000 | 100.093 | 41.980 | 46.032 | 11.987 | 0.835 | 8.514 | 11.212 |
|  | BH311 B4 1 av | 51.893 | 3.050 | 7.381 | 0.425 | 0.662 | 17.221 | 0.177 | 19.042 | 0.029 | 0.217 | 0.010 | 100.107 | 41.138 | 44.727 | 14.135 | 0.808 | 6.503 | 12.593 |
|  | BH311 B4 2 | 52.254 | 2.401 | 8.906 | 0.456 | 0.195 | 16.864 | 0.243 | 18.827 | 0.005 | 0.186 | 0.010 | 100.347 | 40.061 | 43.139 | 16.799 | 0.771 | 4.523 | 14.454 |
|  | BH311 B4 3 | 52.822 | 2.698 | 7.658 | 0.415 | 0.221 | 17.246 | 0.183 | 19.703 | 0.043 | 0.213 | 0.000 | 101.202 | 41.722 | 43.903 | 14.375 | 0.801 | 5.936 | 13.060 |
|  | BH311 B4 4 | 52.727 | 2.788 | 7.277 | 0.508 | 0.209 | 16.957 | 0.191 | 19.184 | 0.000 | 0.241 | 0.005 | 100.087 | 41.686 | 44.297 | 14.017 | 0.806 | 5.978 | 13.025 |
|  | BH311 B4 2-4 av | 52.601 | 2.629 | 7.947 | 0.460 | 0.208 | 17.022 | 0.206 | 19.238 | 0.016 | 0.213 | 0.005 | 100.545 | 41.156 | 43.780 | 15.064 | 0.793 | 5.436 | 13.512 |
|  | BH311 B4 11 | 52.081 | 3.272 | 5.594 | 0.365 | 1.297 | 17.720 | 0.164 | 19.934 | 0.024 | 0.217 | 0.001 | 100.669 | 43.151 | 46.114 | 10.735 | 0.850 | 9.318 | 10.749 |
|  | BH311 B4 12 | 52.571 | 3.233 | 6.648 | 0.321 | 1.188 | 18.976 | 0.156 | 17.640 | 0.030 | 0.222 | 0.003 | 100.988 | 38.062 | 49.223 | 12.716 | 0.836 | 8.750 | 11.072 |
|  | BH311 B4 13 | 52.868 | 2.275 | 6.116 | 0.287 | 0.467 | 17.681 | 0.164 | 20.177 | 0.007 | 0.193 | 0.009 | 100.244 | 43.063 | 45.366 | 11.571 | 0.837 | 7.896 | 11.598 |
|  | BH311 B4 11-13 av | 52.507 | 2.927 | 6.119 | 0.324 | 0.984 | 18.126 | 0.161 | 19.250 | 0.020 | 0.211 | 0.004 | 100.634 | 41.425 | 46.901 | 11.674 | 0.841 | 8.650 | 11.131 |
|  | BH311 B4 14 | 53.839 | 1.933 | 6.127 | 0.229 | 0.654 | 18.923 | 0.162 | 19.437 | 0.023 | 0.198 | 0.007 | 101.532 | 40.819 | 47.774 | 11.406 | 0.846 | 8.661 | 11.124 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{MgO} \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | BH311 B4 15 | 53.830 | 2.240 | 6.264 | 0.193 | 0.754 | 19.173 | 0.173 | 18.651 | 0.005 | 0.208 | 0.012 | 101.503 | 39.470 | 48.779 | 11.751 | 0.845 | 8.741 | 11.077 |
|  | BH311 B4 16 | 49.354 | 2.542 | 12.374 | 0.558 | 0.120 | 14.752 | 0.304 | 18.776 | 0.047 | 0.226 | 0.009 | 99.062 | 39.545 | 37.352 | 23.103 | 0.680 | 1.955 | 18.754 |
|  | BH311 B4 14-16 av | 52.341 | 2.238 | 8.255 | 0.327 | 0.509 | 17.616 | 0.213 | 18.955 | 0.025 | 0.211 | 0.009 | 100.699 | 39.945 | 44.635 | 15.420 | 0.790 | 5.651 | 13.312 |
|  | BH311 B4 17 | 53.276 | 2.090 | 5.946 | 0.258 | 0.782 | 17.636 | 0.147 | 20.608 | 0.014 | 0.216 | 0.004 | 100.977 | 43.772 | 45.033 | 11.196 | 0.841 | 8.022 | 11.517 |
|  | BH311 B4 18 | 52.161 | 3.450 | 5.530 | 0.401 | 1.232 | 17.618 | 0.164 | 20.116 | 0.039 | 0.216 | 0.008 | 100.935 | 43.543 | 45.846 | 10.611 | 0.850 | 9.359 | 10.727 |
|  | BH311 B4 19 | 52.596 | 2.206 | 9.654 | 0.512 | 0.073 | 16.992 | 0.244 | 18.320 | 0.056 | 0.191 | 0.000 | 100.844 | 38.727 | 43.182 | 18.091 | 0.758 | 3.960 | 15.136 |
|  | BH311 B4 17-19 av | 52.678 | 2.582 | 7.043 | 0.390 | 0.696 | 17.415 | 0.185 | 19.681 | 0.036 | 0.208 | 0.004 | 100.919 | 42.014 | 44.687 | 13.299 | 0.816 | 6.775 | 12.383 |
|  | BH311 B4 22 | 53.539 | 2.046 | 5.837 | 0.158 | 0.808 | 19.034 | 0.134 | 18.802 | 0.036 | 0.205 | 0.001 | 100.600 | 40.125 | 48.833 | 11.042 | 0.853 | 9.161 | 10.836 |
|  | BH311 B4 23 | 51.453 | 2.063 | 13.831 | 0.747 | 0.038 | 14.411 | 0.350 | 17.729 | 0.023 | 0.249 | 0.000 | 100.894 | 37.471 | 36.616 | 25.913 | 0.650 | 0.979 | 22.301 |
|  | BH311 B4 22-23 av | 52.496 | 2.055 | 9.834 | 0.453 | 0.423 | 16.723 | 0.242 | 18.266 | 0.030 | 0.227 | 0.001 | 100.747 | 38.798 | 42.724 | 18.478 | 0.752 | 3.612 | 15.607 |
|  | BH311 B4 24 | 53.468 | 1.967 | 7.014 | 0.186 | 0.726 | 20.060 | 0.182 | 16.527 | 0.041 | 0.167 | 0.004 | 100.342 | 35.269 | 51.463 | 13.269 | 0.836 | 8.429 | 11.263 |
|  | BH311 B4 25 | 53.824 | 1.860 | 6.006 | 0.186 | 0.746 | 19.302 | 0.161 | 18.305 | 0.051 | 0.184 | 0.001 | 100.626 | 39.085 | 49.547 | 11.368 | 0.851 | 8.975 | 10.941 |
|  | BH311 B4 26 | 52.353 | 2.931 | 5.971 | 0.380 | 0.524 | 17.101 | 0.155 | 20.124 | 0.012 | 0.200 | 0.008 | 99.759 | 43.771 | 44.716 | 11.513 | 0.836 | 7.857 | 11.623 |
|  | BH311 B4 24-26 av | 53.215 | 2.253 | 6.330 | 0.251 | 0.665 | 18.821 | 0.166 | 18.319 | 0.035 | 0.184 | 0.004 | 100.242 | 39.375 | 48.575 | 12.050 | 0.841 | 8.418 | 11.270 |
|  | BH311 B4 33 | 54.047 | 1.891 | 7.009 | 0.214 | 0.561 | 20.641 | 0.193 | 15.911 | 0.035 | 0.162 | 0.008 | 100.672 | 33.898 | 52.865 | 13.237 | 0.840 | 8.664 | 11.122 |
|  | BH311 B4 34 | 52.895 | 2.486 | 6.529 | 0.344 | 0.414 | 17.727 | 0.175 | 19.915 | 0.025 | 0.187 | 0.002 | 100.699 | 42.360 | 45.329 | 12.311 | 0.829 | 7.490 | 11.869 |
|  | BH311 B4 35 | 50.706 | 0.839 | 27.473 | 0.468 | 0.016 | 14.185 | 0.666 | 5.492 | 0.000 | 0.082 | 0.000 | 99.927 | 11.710 | 36.361 | 51.929 | 0.479 | 0.058 | 36.789 |
|  | BH311 B4 36 | 52.502 | 3.468 | 6.275 | 0.322 | 0.684 | 18.115 | 0.147 | 18.985 | 0.026 | 0.215 | 0.001 | 100.740 | 40.982 | 47.010 | 12.008 | 0.837 | 8.519 | 11.209 |
|  | BH311 B4 37 | 52.927 | 2.633 | 6.920 | 0.458 | 0.279 | 17.827 | 0.189 | 19.317 | 0.024 | 0.204 | 0.006 | 100.784 | 41.203 | 45.713 | 13.085 | 0.821 | 7.129 | 12.122 |
|  | BH311 B4 38 | 52.495 | 2.642 | 7.134 | 0.415 | 0.257 | 17.457 | 0.170 | 19.798 | 0.033 | 0.179 | 0.004 | 100.584 | 42.026 | 44.549 | 13.424 | 0.813 | 6.709 | 12.433 |
|  | BH311 B4 37-38 av | 52.711 | 2.638 | 7.027 | 0.437 | 0.268 | 17.642 | 0.180 | 19.558 | 0.029 | 0.192 | 0.005 | 100.684 | 41.615 | 45.131 | 13.254 | 0.817 | 6.918 | 12.276 |
|  | $\begin{aligned} & \text { BH311 B4 36- } \\ & \text { RIMAV } \\ & \hline \end{aligned}$ | 52.607 | 3.053 | 6.651 | 0.379 | 0.476 | 17.879 | 0.163 | 19.271 | 0.027 | 0.203 | 0.003 | 100.712 | 41.298 | 46.071 | 12.631 | 0.827 | 7.705 | 11.723 |
| CPZ | 10BH167 A5 1 | 51.300 | 3.900 | 6.800 | 0.600 | 0.900 | 17.300 | 0.100 | 19.100 | 0.000 | 0.200 | 0.000 | 100.200 | 41.589 | 45.286 | 13.126 | 0.819 | 7.682 | 11.739 |
|  | 10BH167 A5 2 | 52.400 | 2.900 | 7.000 | 0.200 | 0.600 | 18.300 | 0.200 | 18.700 | 0.000 | 0.200 | 0.000 | 100.500 | 39.868 | 46.903 | 13.229 | 0.823 | 7.674 | 11.744 |
|  | 10BH167 A5 3 | 51.400 | 3.500 | 7.100 | 0.400 | 0.700 | 17.600 | 0.100 | 19.000 | 0.000 | 0.200 | 0.000 | 100.000 | 40.902 | 45.549 | 13.549 | 0.815 | 7.431 | 11.909 |
|  | 10BH167 A5 1-3 av | 51.700 | 3.433 | 6.967 | 0.400 | 0.733 | 17.733 | 0.133 | 18.933 | 0.000 | 0.200 | 0.000 | 100.233 | 40.780 | 45.918 | 13.302 | 0.819 | 7.597 | 11.796 |
|  | 10BH167 A5 8 | 51.500 | 2.000 | 7.200 | 0.200 | 0.600 | 17.500 | 0.200 | 19.300 | 0.000 | 0.200 | 0.000 | 98.600 | 41.309 | 45.030 | 13.661 | 0.812 | 6.743 | 12.407 |
|  | 10BH167 A5 9 | 50.800 | 2.200 | 6.900 | 0.300 | 0.600 | 18.400 | 0.100 | 18.800 | 0.100 | 0.200 | 0.000 | 98.400 | 39.969 | 47.027 | 13.004 | 0.826 | 8.159 | 11.430 |
|  | 10BH167 A5 10 | 51.000 | 2.100 | 9.500 | 0.400 | 0.100 | 17.200 | 0.200 | 18.500 | 0.000 | 0.200 | 0.000 | 99.100 | 38.866 | 43.441 | 17.693 | 0.763 | 4.637 | 14.326 |
|  | 10BH167 A5 8-10 av | 51.100 | 2.100 | 7.867 | 0.300 | 0.433 | 17.700 | 0.167 | 18.867 | 0.033 | 0.200 | 0.000 | 98.700 | 40.041 | 45.159 | 14.800 | 0.800 | 6.402 | 12.673 |
|  | 10BH167 A5 23 | 50.200 | 3.600 | 6.400 | 0.400 | 0.900 | 16.900 | 0.200 | 19.800 | 0.000 | 0.200 | 0.000 | 98.600 | 43.241 | 44.369 | 12.390 | 0.825 | 8.036 | 11.508 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{MgO} \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 10BH167 A5 24 | 50.100 | 3.400 | 6.400 | 0.300 | 0.800 | 17.200 | 0.100 | 19.700 | 0.000 | 0.200 | 0.000 | 98.400 | 42.779 | 44.901 | 12.320 | 0.827 | 8.235 | 11.382 |
|  | 10BH167 A5 25 | 52.000 | 2.400 | 8.600 | 0.400 | 0.100 | 17.300 | 0.200 | 19.100 | 0.000 | 0.200 | 0.000 | 100.400 | 40.192 | 43.765 | 16.043 | 0.782 | 5.242 | 13.698 |
|  | 10BH167 A5 26 | 49.400 | 1.800 | 18.900 | 0.700 | 0.100 | 14.100 | 0.400 | 14.700 | 0.000 | 0.200 | 0.000 | 100.300 | 30.369 | 35.019 | 34.613 | 0.571 | 0.440 | 26.398 |
|  | 10BH167 A5 27 | 48.200 | 1.500 | 22.900 | 0.600 | 0.000 | 11.700 | 0.500 | 14.300 | 0.000 | 0.200 | 0.000 | 99.800 | 29.384 | 28.902 | 41.713 | 0.477 | 0.050 | 37.552 |
|  | $\begin{aligned} & \text { 10BH167 A5 23-27 } \\ & \text { av } \end{aligned}$ | 49.980 | 2.540 | 12.640 | 0.480 | 0.380 | 15.440 | 0.280 | 17.520 | 0.000 | 0.200 | 0.000 | 99.500 | 37.051 | 39.254 | 23.696 | 0.685 | 2.109 | 18.365 |
|  | 10BH167 A5 48 | 50.200 | 3.100 | 10.800 | 0.600 | 0.000 | 15.900 | 0.200 | 18.000 | 0.000 | 0.200 | 0.000 | 99.100 | 38.553 | 40.941 | 20.506 | 0.724 | 3.204 | 16.222 |
|  | 10BH167 A5 49 | 50.100 | 2.000 | 9.100 | 0.400 | 0.100 | 17.900 | 0.200 | 17.900 | 0.000 | 0.200 | 0.000 | 97.800 | 37.695 | 45.317 | 16.988 | 0.778 | 5.692 | 13.276 |
|  | 10BH167 A5 50 | 51.000 | 2.000 | 9.200 | 0.400 | 0.100 | 17.700 | 0.200 | 18.000 | 0.000 | 0.200 | 0.000 | 98.600 | 37.947 | 44.859 | 17.193 | 0.774 | 5.199 | 13.740 |
|  | $\begin{aligned} & \text { 10BH167 A5 48-50 } \\ & \text { av } \end{aligned}$ | 50.433 | 2.367 | 9.700 | 0.467 | 0.067 | 17.167 | 0.200 | 17.967 | 0.000 | 0.200 | 0.000 | 98.500 | 38.063 | 43.721 | 18.217 | 0.759 | 4.640 | 14.324 |
|  | 10BH167 A5 51 | 49.400 | 3.600 | 8.000 | 0.400 | 0.400 | 18.300 | 0.200 | 17.500 | 0.000 | 0.200 | 0.000 | 98.100 | 37.560 | 47.219 | 15.221 | 0.803 | 7.588 | 11.802 |
|  | 10BH167 A5 52 | 47.900 | 2.200 | 8.600 | 0.300 | 0.400 | 15.600 | 0.200 | 20.300 | 0.100 | 0.200 | 0.000 | 95.700 | 43.490 | 40.178 | 16.333 | 0.764 | 4.904 | 14.040 |
|  | 10BH167 A5 53 | 49.600 | 2.300 | 7.700 | 0.300 | 0.300 | 17.300 | 0.200 | 18.500 | 0.000 | 0.200 | 0.000 | 96.300 | 40.110 | 45.091 | 14.799 | 0.800 | 6.562 | 12.546 |
|  | 10BH167 A5 54 | 48.400 | 2.500 | 7.800 | 0.300 | 0.400 | 16.900 | 0.200 | 19.100 | 0.000 | 0.200 | 0.000 | 95.900 | 41.225 | 43.851 | 14.924 | 0.794 | 6.596 | 12.520 |
|  | 10BH167 A5 55 | 49.600 | 2.600 | 7.000 | 0.300 | 0.500 | 17.300 | 0.100 | 19.600 | 0.100 | 0.200 | 0.000 | 97.200 | 42.057 | 44.627 | 13.315 | 0.815 | 7.538 | 11.836 |
|  | 10BH167 A5 56 | 49.200 | 2.200 | 8.600 | 0.400 | 0.200 | 17.400 | 0.200 | 18.800 | 0.000 | 0.200 | 0.000 | 97.100 | 39.711 | 44.185 | 16.103 | 0.783 | 6.045 | 12.968 |
|  | 10BH167 A5 57 | 50.000 | 2.100 | 9.300 | 0.300 | 0.000 | 17.000 | 0.200 | 18.300 | 0.000 | 0.200 | 0.000 | 97.400 | 38.952 | 43.501 | 17.548 | 0.765 | 4.781 | 14.170 |
|  | 10BH167 A5 58 | 50.100 | 2.000 | 10.700 | 0.400 | 0.000 | 17.000 | 0.200 | 17.400 | 0.000 | 0.200 | 0.000 | 98.000 | 36.769 | 43.187 | 20.044 | 0.739 | 3.828 | 15.310 |
|  | 10BH167 A5 59 | 48.200 | 2.300 | 15.500 | 0.600 | 0.000 | 14.400 | 0.300 | 16.700 | 0.000 | 0.200 | 0.000 | 98.300 | 34.973 | 36.253 | 28.774 | 0.623 | 1.078 | 21.804 |
|  | 10BH167 A5 60 | 45.900 | 1.100 | 27.500 | 0.500 | 0.100 | 7.700 | 0.600 | 15.200 | 0.000 | 0.200 | 0.000 | 98.800 | 31.125 | 18.955 | 49.919 | 0.333 | 0.000 | 70.717 |
|  | $\begin{aligned} & \text { 10BH167 A5 51-60 } \\ & \text { av } \end{aligned}$ | 48.830 | 2.290 | 11.070 | 0.380 | 0.230 | 15.890 | 0.240 | 18.140 | 0.020 | 0.200 | 0.000 | 97.280 | 38.550 | 40.596 | 20.854 | 0.719 | 3.233 | 16.176 |
|  | 10BH167 A5 64 | 51.100 | 2.100 | 7.200 | 0.200 | 0.400 | 17.800 | 0.200 | 19.200 | 0.100 | 0.200 | 0.000 | 98.300 | 40.867 | 45.547 | 13.585 | 0.815 | 7.165 | 12.096 |
|  | 10BH167 A5 65 | 50.600 | 2.100 | 8.400 | 0.300 | 0.100 | 17.700 | 0.200 | 18.600 | 0.000 | 0.200 | 0.000 | 98.200 | 39.303 | 44.963 | 15.734 | 0.790 | 5.997 | 13.008 |
|  | 10BH167 A5 66 | 47.500 | 1.800 | 17.800 | 1.000 | 0.100 | 13.400 | 0.400 | 16.300 | 0.000 | 0.200 | 0.000 | 98.500 | 33.826 | 33.430 | 32.745 | 0.573 | 0.519 | 25.551 |
|  | 10BH167 A5 64-66 $\begin{aligned} & \text { av } \\ & \hline \end{aligned}$ | 49.733 | 2.000 | 11.133 | 0.500 | 0.200 | 16.300 | 0.267 | 18.033 | 0.033 | 0.200 | 0.000 | 98.333 | 37.966 | 41.255 | 20.778 | 0.723 | 3.271 | 16.115 |
|  | 10BH167 A5 67 | 48.900 | 3.400 | 6.600 | 0.400 | 0.700 | 17.000 | 0.200 | 19.600 | 0.000 | 0.200 | 0.000 | 97.000 | 42.713 | 44.537 | 12.750 | 0.821 | 8.161 | 11.429 |
|  | 10BH167 A5 68 | 50.100 | 3.400 | 6.700 | 0.400 | 0.800 | 17.400 | 0.100 | 19.500 | 0.000 | 0.200 | 0.000 | 98.600 | 42.065 | 45.123 | 12.812 | 0.822 | 8.090 | 11.474 |
|  | 10BH167 A5 69 | 51.000 | 2.700 | 6.600 | 0.300 | 0.500 | 17.600 | 0.100 | 19.500 | 0.000 | 0.200 | 0.000 | 98.600 | 41.927 | 45.493 | 12.580 | 0.826 | 7.846 | 11.630 |
|  | 10BH167 A5 70 | 50.100 | 2.200 | 12.400 | 0.500 | 0.100 | 15.700 | 0.200 | 17.700 | 0.000 | 0.200 | 0.000 | 99.100 | 37.211 | 39.680 | 23.109 | 0.693 | 2.241 | 18.055 |
|  | $\begin{aligned} & \text { 10BH167 A5 67-70 } \\ & \text { av } \end{aligned}$ | 50.025 | 2.925 | 8.075 | 0.400 | 0.525 | 16.925 | 0.150 | 19.075 | 0.000 | 0.200 | 0.000 | 98.325 | 40.951 | 43.682 | 15.368 | 0.789 | 6.070 | 12.946 |
|  | 10BH167 A5 93 | 49.500 | 1.400 | 25.600 | 0.500 | 0.000 | 15.200 | 0.500 | 7.000 | 0.000 | 0.100 | 0.100 | 99.900 | 14.593 | 38.095 | 47.311 | 0.514 | 0.230 | 29.733 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{MgO} \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 10BH167 A5 94 | 49.500 | 0.900 | 29.700 | 0.400 | 0.000 | 14.800 | 0.600 | 5.000 | 0.000 | 0.100 | 0.000 | 101.100 | 10.179 | 36.222 | 53.599 | 0.470 | 0.103 | 33.833 |
|  | 10BH167 A5 95 | 46.900 | 0.900 | 34.200 | 0.400 | 0.000 | 10.400 | 0.700 | 6.700 | 0.000 | 0.100 | 0.000 | 100.300 | 13.530 | 25.248 | 61.222 | 0.351 | 0.002 | 54.661 |
|  | $\begin{aligned} & \text { 10BH167 A5 93-95 } \\ & \text { av } \end{aligned}$ | 48.633 | 1.067 | 29.833 | 0.433 | 0.000 | 13.467 | 0.600 | 6.233 | 0.000 | 0.100 | 0.033 | 100.433 | 12.755 | 33.128 | 54.117 | 0.446 | 0.050 | 37.538 |
| DZ | 10BH167 A6 1 | 49.900 | 3.300 | 8.700 | 0.300 | 0.600 | 15.400 | 0.200 | 20.200 | 0.100 | 0.200 | 0.000 | 98.800 | 43.510 | 39.878 | 16.612 | 0.759 | 4.412 | 14.582 |
|  | 10BH167 A6 2 | 51.400 | 2.400 | 7.100 | 0.200 | 0.400 | 18.000 | 0.200 | 18.800 | 0.000 | 0.200 | 0.000 | 98.800 | 40.228 | 46.304 | 13.468 | 0.819 | 7.384 | 11.941 |
|  | 10BH167 A6 3 | 51.100 | 2.200 | 6.900 | 0.300 | 0.300 | 17.600 | 0.200 | 18.900 | 0.100 | 0.200 | 0.000 | 97.700 | 40.931 | 45.822 | 13.247 | 0.820 | 7.240 | 12.043 |
|  | 10BH167 A6 4 | 52.600 | 2.200 | 7.600 | 0.200 | 0.300 | 18.300 | 0.200 | 18.600 | 0.100 | 0.200 | 0.000 | 100.300 | 39.293 | 46.475 | 14.232 | 0.811 | 6.761 | 12.394 |
|  | 10BH167 A6 5 | 49.700 | 1.900 | 10.000 | 0.300 | 0.100 | 17.200 | 0.300 | 17.900 | 0.000 | 0.200 | 0.000 | 97.600 | 37.730 | 43.585 | 18.685 | 0.754 | 4.545 | 14.429 |
|  | 10BH167 A6 6 | 49.400 | 1.900 | 11.600 | 0.500 | 0.000 | 16.500 | 0.200 | 17.300 | 0.000 | 0.200 | 0.000 | 97.600 | 36.483 | 41.831 | 21.685 | 0.717 | 3.154 | 16.302 |
|  | 10BH167 A6 1-6 av | 50.683 | 2.317 | 8.650 | 0.300 | 0.283 | 17.167 | 0.217 | 18.617 | 0.050 | 0.200 | 0.000 | 98.467 | 39.676 | 43.982 | 16.342 | 0.780 | 5.379 | 13.566 |
|  | 10BH167 A6 7 | 49.300 | 1.600 | 21.300 | 0.700 | 0.000 | 14.100 | 0.400 | 12.500 | 0.000 | 0.200 | 0.000 | 100.100 | 25.862 | 35.071 | 39.066 | 0.541 | 0.280 | 28.717 |
|  | 10BH167 A6 19 | 45.200 | 7.300 | 13.400 | 1.300 | 0.000 | 15.100 | 0.200 | 11.000 | 0.000 | 2.000 | 0.300 | 95.800 | 26.809 | 44.241 | 28.950 | 0.668 | 3.873 | 15.249 |
|  | 10BH167 A6 20 | 43.800 | 9.400 | 14.500 | 0.800 | 0.000 | 14.400 | 0.300 | 10.500 | 0.000 | 2.100 | 0.300 | 96.200 | 25.821 | 42.571 | 31.609 | 0.639 | 3.487 | 15.788 |
|  | 10BH167 A6 21 | 48.000 | 5.000 | 19.500 | 0.400 | 0.000 | 11.800 | 0.200 | 11.000 | 0.000 | 0.900 | 0.100 | 97.000 | 25.899 | 33.400 | 40.700 | 0.519 | 0.193 | 30.637 |
|  | $\begin{aligned} & \text { 10BH167 A6 19-21 } \\ & \text { av } \end{aligned}$ | 45.667 | 7.233 | 15.800 | 0.833 | 0.000 | 13.767 | 0.233 | 10.833 | 0.000 | 1.667 | 0.233 | 96.333 | 26.174 | 39.986 | 33.840 | 0.608 | 1.906 | 18.885 |
|  | 10BH167 A6 22 | 48.900 | 2.100 | 7.900 | 0.300 | 0.400 | 15.100 | 0.200 | 21.500 | 0.000 | 0.200 | 0.000 | 96.600 | 46.082 | 38.908 | 15.010 | 0.773 | 4.771 | 14.181 |
|  | 10BH167 A6 23 | 49.700 | 2.900 | 5.300 | 0.300 | 0.600 | 15.600 | 0.100 | 22.700 | 0.000 | 0.100 | 0.000 | 97.400 | 49.185 | 40.635 | 10.180 | 0.840 | 8.267 | 11.363 |
|  | 10BH167 A6 24 | 49.600 | 3.700 | 7.000 | 0.500 | 0.700 | 16.900 | 0.200 | 19.400 | 0.000 | 0.200 | 0.000 | 98.100 | 42.245 | 44.242 | 13.513 | 0.811 | 7.506 | 11.857 |
|  | 10BH167 A6 25 | 49.600 | 2.400 | 10.500 | 0.500 | 0.100 | 15.800 | 0.200 | 18.300 | 0.000 | 0.200 | 0.000 | 97.500 | 39.268 | 40.759 | 19.973 | 0.728 | 3.263 | 16.128 |
|  | 10BH167 A6 26 | 47.500 | 2.400 | 15.200 | 0.700 | 0.000 | 14.000 | 0.300 | 17.600 | 0.000 | 0.200 | 0.000 | 98.000 | 36.739 | 35.133 | 28.127 | 0.621 | 1.098 | 21.710 |
|  | $\begin{aligned} & \text { 10BH167 A6 22-26 } \\ & \text { av } \end{aligned}$ | 49.060 | 2.700 | 9.180 | 0.460 | 0.360 | 15.480 | 0.200 | 19.900 | 0.000 | 0.180 | 0.000 | 97.520 | 42.660 | 39.894 | 17.445 | 0.750 | 4.145 | 14.901 |
|  | 10BH167 A6 27 | 50.000 | 1.900 | 8.100 | 0.300 | 0.300 | 14.600 | 0.200 | 21.800 | 0.000 | 0.300 | 0.000 | 97.400 | 46.849 | 37.720 | 15.431 | 0.763 | 3.830 | 15.307 |
|  | 10BH167 A6 28 | 49.600 | 2.000 | 7.700 | 0.300 | 0.400 | 15.300 | 0.200 | 21.300 | 0.100 | 0.200 | 0.000 | 97.100 | 45.788 | 39.539 | 14.673 | 0.780 | 4.890 | 14.054 |
|  | 10BH167 A6 29 | 51.000 | 2.200 | 7.900 | 0.400 | 0.200 | 17.700 | 0.200 | 18.900 | 0.000 | 0.200 | 0.000 | 98.600 | 40.058 | 45.099 | 14.843 | 0.800 | 6.411 | 12.666 |
|  | 10BH167 A6 30 | 49.700 | 2.100 | 11.300 | 0.500 | 0.100 | 16.300 | 0.300 | 17.200 | 0.000 | 0.200 | 0.000 | 97.600 | 36.742 | 41.860 | 21.398 | 0.720 | 3.114 | 16.368 |
|  | 10BH167 A6 31 | 50.000 | 2.400 | 13.900 | 0.600 | 0.000 | 15.200 | 0.300 | 17.100 | 0.000 | 0.200 | 0.000 | 99.700 | 35.853 | 38.312 | 25.835 | 0.661 | 1.554 | 19.931 |
|  | $\begin{aligned} & \text { 10BH167 A6 27-31 } \\ & \text { av } \end{aligned}$ | 50.060 | 2.120 | 9.780 | 0.420 | 0.200 | 15.820 | 0.240 | 19.260 | 0.020 | 0.220 | 0.000 | 98.080 | 41.024 | 40.510 | 18.466 | 0.742 | 3.610 | 15.610 |
| DZ | 10BH167 A8 1 | 51.500 | 2.100 | 9.900 | 0.200 | 0.000 | 17.000 | 0.300 | 17.700 | 0.000 | 0.200 | 0.000 | 98.800 | 37.729 | 43.564 | 18.707 | 0.754 | 3.936 | 15.168 |
|  | 10BH167 A8 2 | 52.700 | 1.900 | 6.800 | 0.300 | 0.200 | 18.200 | 0.200 | 19.300 | 0.000 | 0.200 | 0.000 | 99.700 | 40.883 | 46.348 | 12.769 | 0.827 | 7.404 | 11.928 |
|  | 10BH167 A8 3 | 52.000 | 2.100 | 5.800 | 0.300 | 0.300 | 18.000 | 0.200 | 19.900 | 0.000 | 0.200 | 0.000 | 98.700 | 42.630 | 46.356 | 11.014 | 0.847 | 8.750 | 11.071 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{MgO} \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 10BH167 A8 4 | 51.900 | 2.700 | 8.500 | 0.500 | 0.100 | 17.200 | 0.200 | 18.200 | 0.100 | 0.200 | 0.000 | 99.600 | 39.214 | 44.552 | 16.235 | 0.783 | 5.236 | 13.704 |
|  | 10BH167 A8 1-4 av | 52.025 | 2.200 | 7.750 | 0.325 | 0.150 | 17.600 | 0.225 | 18.775 | 0.025 | 0.200 | 0.000 | 99.200 | 40.114 | 45.207 | 14.679 | 0.802 | 6.102 | 12.919 |
|  | 10BH167 A8 7 | 51.100 | 1.900 | 14.700 | 0.500 | 0.000 | 14.700 | 0.400 | 16.200 | 0.000 | 0.200 | 0.000 | 99.800 | 34.539 | 37.678 | 27.783 | 0.641 | 0.877 | 22.864 |
|  | 10BH167 A8 8 | 52.100 | 2.800 | 7.900 | 0.500 | 0.200 | 16.700 | 0.200 | 19.700 | 0.000 | 0.200 | 0.000 | 100.400 | 42.112 | 42.917 | 14.970 | 0.790 | 5.411 | 13.535 |
|  | 10BH167 A8 9 | 51.900 | 1.900 | 11.500 | 0.500 | 0.000 | 16.300 | 0.300 | 17.200 | 0.000 | 0.200 | 0.000 | 99.700 | 36.604 | 41.702 | 21.695 | 0.716 | 2.478 | 17.539 |
|  | 10BH167 A8 7-9 av | 51.700 | 2.200 | 11.367 | 0.500 | 0.067 | 15.900 | 0.300 | 17.700 | 0.000 | 0.200 | 0.000 | 99.967 | 37.747 | 40.764 | 21.489 | 0.714 | 2.402 | 17.698 |
| DZ | 10BH167 A9 5 | 49.300 | 1.800 | 20.500 | 0.700 | 0.000 | 11.800 | 0.400 | 16.300 | 0.000 | 0.200 | 0.000 | 101.000 | 33.499 | 29.154 | 37.347 | 0.506 | 0.075 | 35.479 |
|  | 10BH167 A9 6 | 49.100 | 2.400 | 14.200 | 0.600 | 0.100 | 14.700 | 0.300 | 17.500 | 0.000 | 0.200 | 0.000 | 99.100 | 36.642 | 37.002 | 26.357 | 0.648 | 1.394 | 20.487 |
|  | 10BH167 A9 7 | 48.900 | 2.000 | 18.700 | 0.700 | 0.000 | 12.800 | 0.400 | 16.500 | 0.000 | 0.200 | 0.000 | 100.200 | 34.045 | 31.751 | 34.204 | 0.550 | 0.253 | 29.238 |
|  | 10BH167 A9 5-7 av | 49.100 | 2.067 | 17.800 | 0.667 | 0.033 | 13.100 | 0.367 | 16.767 | 0.000 | 0.200 | 0.000 | 100.100 | 34.718 | 32.610 | 32.673 | 0.567 | 0.347 | 27.619 |
|  | 10BH167 A9 11 | 46.900 | 2.000 | 13.100 | 0.700 | 0.000 | 13.400 | 0.200 | 19.900 | 0.000 | 0.200 | 0.000 | 96.500 | 41.787 | 33.827 | 24.385 | 0.646 | 1.326 | 20.746 |
|  | 10BH167 A9 12 | 48.000 | 2.800 | 8.900 | 0.300 | 0.100 | 18.100 | 0.200 | 17.700 | 0.000 | 0.200 | 0.000 | 96.300 | 37.382 | 45.955 | 16.663 | 0.784 | 6.776 | 12.382 |
|  | 10BH167 A9 13 | 48.000 | 2.800 | 8.000 | 0.300 | 0.200 | 17.400 | 0.200 | 18.600 | 0.000 | 0.200 | 0.000 | 95.800 | 39.906 | 44.879 | 15.215 | 0.795 | 7.000 | 12.215 |
|  | 10BH167 A9 14 | 48.900 | 2.200 | 9.000 | 0.300 | 0.000 | 14.900 | 0.200 | 21.600 | 0.000 | 0.200 | 0.000 | 97.400 | 45.483 | 37.718 | 16.800 | 0.747 | 3.777 | 15.378 |
|  | $\begin{aligned} & \text { 10BH167 A9 11-14 } \\ & \text { av } \end{aligned}$ | 47.950 | 2.450 | 9.750 | 0.400 | 0.075 | 15.950 | 0.200 | 19.450 | 0.000 | 0.200 | 0.000 | 96.500 | 41.148 | 40.566 | 18.285 | 0.745 | 4.359 | 14.644 |
| DZ | BH311 B6 62 | 51.758 | 2.390 | 9.739 | 0.671 | 0.000 | 16.428 | 0.205 | 18.785 | 0.000 | 0.212 | 0.008 | 100.196 | 39.826 | 41.871 | 18.303 | 0.750 | 3.748 | 15.417 |
|  | BH311 B6 62 | 52.014 | 2.425 | 8.595 | 0.516 | 0.000 | 16.728 | 0.213 | 19.083 | 0.032 | 0.211 | 0.007 | 99.824 | 40.765 | 42.959 | 16.276 | 0.776 | 4.705 | 14.253 |
|  | BH311 B6 62 | 52.147 | 2.415 | 9.626 | 0.443 | 0.017 | 17.788 | 0.257 | 17.123 | 0.030 | 0.194 | 0.011 | 100.051 | 36.400 | 45.460 | 18.140 | 0.767 | 4.697 | 14.261 |
|  | BH311 B6 62 | 52.839 | 1.877 | 9.404 | 0.364 | 0.026 | 18.302 | 0.283 | 17.114 | 0.006 | 0.192 | 0.021 | 100.428 | 36.065 | 46.367 | 17.568 | 0.776 | 4.954 | 13.987 |
|  | BH311 B6 62 | 53.036 | 1.788 | 9.476 | 0.386 | 0.030 | 18.234 | 0.253 | 17.317 | 0.024 | 0.194 | 0.005 | 100.743 | 36.351 | 46.015 | 17.633 | 0.774 | 4.794 | 14.156 |
|  | BH311 B6 62 | 52.841 | 1.732 | 10.039 | 0.399 | 0.043 | 18.061 | 0.268 | 16.905 | 0.010 | 0.206 | 0.000 | 100.504 | 35.577 | 45.695 | 18.729 | 0.762 | 4.292 | 14.723 |
|  | BH311 B6 62 av | 52.439 | 2.105 | 9.480 | 0.463 | 0.019 | 17.590 | 0.247 | 17.721 | 0.017 | 0.202 | 0.009 | 100.291 | 37.497 | 44.728 | 17.775 | 0.768 | 4.523 | 14.455 |
|  | BH311 B6 65 | 50.547 | 1.761 | 15.640 | 0.780 | 0.008 | 13.990 | 0.326 | 16.334 | 0.006 | 0.211 | 0.016 | 99.619 | 34.741 | 35.771 | 29.488 | 0.614 | 0.582 | 24.966 |
|  | BH311 B6 65 | 50.727 | 2.410 | 13.091 | 0.742 | 0.013 | 15.148 | 0.294 | 17.041 | 0.000 | 0.230 | 0.016 | 99.712 | 36.369 | 38.865 | 24.767 | 0.673 | 1.606 | 19.764 |
|  | BH311 B6 65 | 50.747 | 2.398 | 12.670 | 0.708 | 0.000 | 14.631 | 0.320 | 18.323 | 0.010 | 0.243 | 0.007 | 100.057 | 38.866 | 37.310 | 23.824 | 0.673 | 1.497 | 20.123 |
|  | BH311 B6 65 av | 50.674 | 2.190 | 13.800 | 0.743 | 0.007 | 14.590 | 0.313 | 17.233 | 0.005 | 0.228 | 0.013 | 99.796 | 36.659 | 37.315 | 26.026 | 0.654 | 1.142 | 21.509 |
|  | BH311 B6 67 | 49.903 | 2.064 | 15.940 | 0.653 | 0.000 | 14.998 | 0.370 | 14.984 | 0.000 | 0.209 | 0.010 | 99.131 | 31.783 | 38.245 | 29.972 | 0.626 | 0.962 | 22.388 |
|  | BH311 B6 67 | 50.884 | 2.196 | 10.125 | 0.577 | 0.000 | 16.639 | 0.256 | 18.029 | 0.000 | 0.291 | 0.058 | 99.055 | 38.353 | 42.553 | 19.094 | 0.745 | 3.861 | 15.266 |
|  | BH311 B6 67 | 49.274 | 3.096 | 15.788 | 0.350 | 0.080 | 16.030 | 0.293 | 12.049 | 0.057 | 0.173 | 0.052 | 97.242 | 26.589 | 42.526 | 30.885 | 0.644 | 1.498 | 20.121 |
|  | BH311 B6 67 av | 50.020 | 2.452 | 13.951 | 0.527 | 0.027 | 15.889 | 0.306 | 15.021 | 0.019 | 0.224 | 0.040 | 98.476 | 32.242 | 41.108 | 26.650 | 0.672 | 1.802 | 19.172 |
|  | BH311 B6 68 | 49.996 | 1.497 | 16.231 | 0.695 | 0.000 | 13.935 | 0.358 | 16.193 | 0.043 | 0.297 | 0.004 | 99.249 | 34.210 | 35.392 | 30.398 | 0.605 | 0.555 | 25.208 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | Melt <br> MgO | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | BH311 B6 69 | 50.351 | 2.194 | 15.287 | 0.697 | 0.017 | 13.273 | 0.417 | 17.587 | 0.000 | 0.276 | 0.024 | 100.123 | 37.344 | 33.882 | 28.775 | 0.607 | 0.509 | 25.656 |
|  | BH311 B6 69 | 50.264 | 2.525 | 13.157 | 0.849 | 0.000 | 13.655 | 0.316 | 18.527 | 0.006 | 0.327 | 0.016 | 99.642 | 39.752 | 35.222 | 25.025 | 0.649 | 1.020 | 22.092 |
|  | BH311 B6 69 | 50.708 | 1.907 | 17.069 | 0.637 | 0.029 | 14.719 | 0.381 | 14.062 | 0.007 | 0.210 | 0.006 | 99.735 | 29.991 | 37.739 | 32.271 | 0.606 | 0.605 | 24.771 |
|  | BH311 B6 69 av | 50.441 | 2.209 | 15.171 | 0.728 | 0.015 | 13.882 | 0.371 | 16.725 | 0.004 | 0.271 | 0.015 | 99.833 | 35.696 | 35.614 | 28.690 | 0.621 | 0.677 | 24.188 |
|  | BH311 B6 71 | 51.969 | 2.468 | 8.079 | 0.559 | 0.061 | 17.370 | 0.185 | 19.330 | 0.000 | 0.214 | 0.005 | 100.240 | 40.803 | 44.079 | 15.118 | 0.793 | 5.813 | 13.168 |
|  | BH311 B6 71 | 48.095 | 6.130 | 16.229 | 0.295 | 0.475 | 15.909 | 0.193 | 16.286 | 0.071 | 0.176 | 0.027 | 103.886 | 32.704 | 38.406 | 28.890 | 0.636 | 2.279 | 17.969 |
|  | BH311 B6 71 | 53.499 | 1.882 | 6.963 | 0.215 | 0.328 | 20.253 | 0.206 | 16.899 | 0.061 | 0.173 | 0.013 | 100.492 | 35.637 | 51.346 | 13.017 | 0.838 | 8.623 | 11.147 |
|  | BH311 B6 71 | 52.683 | 2.193 | 6.377 | 0.446 | 0.368 | 18.655 | 0.194 | 19.080 | 0.076 | 0.186 | 0.009 | 100.267 | 40.458 | 47.555 | 11.987 | 0.839 | 8.462 | 11.243 |
|  | BH311 B6 71 | 50.635 | 1.953 | 13.216 | 0.800 | 0.000 | 15.009 | 0.318 | 17.339 | 0.038 | 0.237 | 0.003 | 99.548 | 36.815 | 38.311 | 24.875 | 0.669 | 1.470 | 20.217 |
|  | BH311 B6 71 av | 51.376 | 2.925 | 10.173 | 0.463 | 0.246 | 17.439 | 0.219 | 17.787 | 0.049 | 0.197 | 0.011 | 100.887 | 37.284 | 43.939 | 18.777 | 0.755 | 4.521 | 14.457 |
|  | BH311 B6 72 | 50.421 | 0.582 | 21.544 | 0.035 | 0.050 | 7.736 | 0.519 | 19.661 | 0.000 | 0.241 | 0.001 | 100.790 | 40.910 | 19.351 | 39.739 | 0.390 | 0.000 | 83.836 |
|  | BH311 B6 72 | 31.440 | 13.311 | 27.961 | 0.013 | 0.000 | 13.324 | 0.208 | 1.634 | 0.000 | 0.072 | 0.095 | 88.058 | 3.850 | 37.744 | 58.406 | 0.459 | 1.415 | 20.413 |
|  | BH311 B6 72 | 49.413 | 1.272 | 24.134 | 0.566 | 0.000 | 9.785 | 0.560 | 14.373 | 0.000 | 0.204 | 0.009 | 100.316 | 30.240 | 24.749 | 45.011 | 0.419 | 0.002 | 54.996 |
|  | BH311 B6 72 av | 43.758 | 5.055 | 24.546 | 0.205 | 0.017 | 10.282 | 0.429 | 11.889 | 0.000 | 0.172 | 0.035 | 96.388 | 25.000 | 27.281 | 47.719 | 0.423 | 0.060 | 36.595 |
| DZ | 10BH167 A10 7 | 51.700 | 2.100 | 14.000 | 0.500 | 0.000 | 16.600 | 0.300 | 15.000 | 0.000 | 0.200 | 0.000 | 100.600 | 31.668 | 42.131 | 26.201 | 0.679 | 1.841 | 19.062 |
|  | 10BH167 A10 8 | 50.800 | 2.000 | 16.100 | 0.600 | 0.000 | 13.800 | 0.400 | 16.100 | 0.000 | 0.200 | 0.000 | 100.000 | 34.283 | 35.326 | 30.391 | 0.604 | 0.469 | 26.071 |
|  | 10BH167 A10 9 | 50.700 | 1.300 | 15.200 | 0.200 | 0.000 | 10.100 | 0.300 | 20.200 | 0.000 | 0.200 | 0.000 | 98.400 | 44.089 | 26.502 | 29.409 | 0.542 | 0.015 | 43.888 |
|  | 10BH167 A10 10 | 50.900 | 2.100 | 14.100 | 0.500 | 0.000 | 15.600 | 0.300 | 15.500 | 0.000 | 0.200 | 0.000 | 99.400 | 33.153 | 40.113 | 26.734 | 0.663 | 1.409 | 20.434 |
|  | $\begin{aligned} & \text { 10BH167 A10 7-10 } \\ & \text { av } \end{aligned}$ | 51.025 | 1.875 | 14.850 | 0.450 | 0.000 | 14.025 | 0.325 | 16.700 | 0.000 | 0.200 | 0.000 | 99.600 | 35.741 | 36.085 | 28.174 | 0.627 | 0.617 | 24.663 |
|  | 10BH167 A10 11 | 50.600 | 1.700 | 14.500 | 0.600 | 0.000 | 14.600 | 0.300 | 16.800 | 0.000 | 0.200 | 0.000 | 99.400 | 35.589 | 37.182 | 27.229 | 0.642 | 0.930 | 22.563 |
|  | 10BH167 A10 12 | 50.000 | 2.300 | 14.100 | 0.800 | 0.000 | 14.000 | 0.300 | 17.200 | 0.000 | 0.200 | 0.000 | 99.000 | 36.966 | 36.172 | 26.863 | 0.639 | 0.906 | 22.695 |
|  | 10BH167 A10 13 | 50.200 | 1.700 | 16.700 | 0.600 | 0.000 | 11.800 | 0.400 | 18.000 | 0.000 | 0.200 | 0.000 | 99.700 | 38.306 | 30.189 | 31.505 | 0.557 | 0.116 | 33.217 |
|  | $\begin{aligned} & \text { 10BH167 A10 11-13 } \\ & \text { av } \end{aligned}$ | 50.267 | 1.900 | 15.100 | 0.667 | 0.000 | 13.467 | 0.333 | 17.333 | 0.000 | 0.200 | 0.000 | 99.367 | 36.951 | 34.513 | 28.536 | 0.614 | 0.525 | 25.494 |
|  | 10BH167 A10 14 | 48.700 | 2.500 | 18.300 | 0.300 | 0.000 | 9.400 | 0.400 | 17.500 | 0.000 | 0.200 | 0.100 | 97.300 | 38.869 | 25.099 | 36.031 | 0.478 | 0.007 | 48.000 |
|  | 10BH167 A10 15 | 48.700 | 4.300 | 23.000 | 0.200 | 0.000 | 8.400 | 0.300 | 11.700 | 0.000 | 0.400 | 0.100 | 97.200 | 27.734 | 23.937 | 48.329 | 0.394 | 0.000 | 63.483 |
|  | 10BH167 A10 16 | 44.800 | 5.900 | 23.800 | 1.400 | 0.000 | 7.800 | 0.300 | 10.200 | 0.000 | 1.800 | 0.800 | 96.800 | 25.077 | 23.054 | 51.870 | 0.369 | 0.007 | 47.447 |
|  | $\begin{aligned} & \text { 10BH167 A10 14-16 } \\ & \text { av } \end{aligned}$ | 47.400 | 4.233 | 21.700 | 0.633 | 0.000 | 8.533 | 0.333 | 13.133 | 0.000 | 0.800 | 0.333 | 97.100 | 30.809 | 24.065 | 45.126 | 0.412 | 0.003 | 51.752 |
|  | 10BH167 A10 17 | 52.400 | 2.400 | 9.500 | 0.500 | 0.000 | 16.800 | 0.200 | 18.200 | 0.000 | 0.200 | 0.000 | 100.200 | 38.874 | 43.139 | 17.988 | 0.759 | 3.930 | 15.175 |
|  | 10BH167 A10 18 | 52.600 | 2.700 | 9.200 | 0.600 | 0.100 | 17.200 | 0.200 | 18.300 | 0.000 | 0.200 | 0.000 | 101.100 | 38.826 | 43.871 | 17.303 | 0.769 | 4.574 | 14.397 |
|  | 10BH167 A10 19 | 51.800 | 1.600 | 12.200 | 0.600 | 0.000 | 12.900 | 0.300 | 20.600 | 0.000 | 0.200 | 0.000 | 100.400 | 43.902 | 33.050 | 23.048 | 0.653 | 0.581 | 24.977 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | Melt <br> MgO | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\begin{aligned} & \text { 10BH167 A10 17-19 } \\ & \text { av } \end{aligned}$ | 52.267 | 2.233 | 10.300 | 0.567 | 0.033 | 15.633 | 0.233 | 19.033 | 0.000 | 0.200 | 0.000 | 100.567 | 40.533 | 40.023 | 19.444 | 0.730 | 2.635 | 17.223 |
|  | 10BH167 A10 20 | 53.100 | 1.900 | 7.700 | 0.400 | 0.200 | 17.700 | 0.200 | 18.900 | 0.000 | 0.200 | 0.000 | 100.300 | 40.209 | 45.269 | 14.522 | 0.804 | 5.864 | 13.123 |
|  | 10BH167 A10 21 | 53.400 | 1.900 | 7.600 | 0.300 | 0.200 | 18.000 | 0.200 | 18.900 | 0.000 | 0.200 | 0.000 | 100.700 | 39.978 | 45.772 | 14.251 | 0.808 | 6.148 | 12.881 |
|  | 10BH167 A10 22 | 51.800 | 2.400 | 9.300 | 0.600 | 0.000 | 16.300 | 0.200 | 18.900 | 0.000 | 0.200 | 0.000 | 99.700 | 40.437 | 41.925 | 17.638 | 0.757 | 3.859 | 15.268 |
|  | $\begin{aligned} & \text { 10BH167 A10 20-22 } \\ & \text { av } \end{aligned}$ | 52.767 | 2.067 | 8.200 | 0.433 | 0.133 | 17.333 | 0.200 | 18.900 | 0.000 | 0.200 | 0.000 | 100.233 | 40.207 | 44.329 | 15.464 | 0.790 | 5.231 | 13.709 |
|  | 10BH167 A10 31 | 52.600 | 2.600 | 7.100 | 0.400 | 0.200 | 17.300 | 0.200 | 19.000 | 0.000 | 0.200 | 0.000 | 99.500 | 41.222 | 45.123 | 13.655 | 0.813 | 6.393 | 12.681 |
|  | 10BH167 A10 32 | 52.700 | 2.500 | 8.300 | 0.400 | 0.100 | 17.100 | 0.200 | 17.900 | 0.000 | 0.200 | 0.000 | 99.400 | 39.070 | 44.870 | 16.060 | 0.786 | 4.898 | 14.046 |
|  | 10BH167 A10 33 | 51.500 | 2.300 | 10.000 | 0.400 | 0.100 | 16.000 | 0.300 | 17.700 | 0.000 | 0.200 | 0.000 | 98.400 | 38.647 | 41.998 | 19.355 | 0.740 | 3.097 | 16.396 |
|  | 10BH167 A10 34 | 51.400 | 2.300 | 12.500 | 0.600 | 0.100 | 14.900 | 0.300 | 16.700 | 0.000 | 0.200 | 0.000 | 99.000 | 36.548 | 39.202 | 24.250 | 0.680 | 1.397 | 20.476 |
|  | 10BH167 A10 35 | 50.600 | 2.000 | 18.400 | 0.600 | 0.000 | 11.700 | 0.500 | 15.200 | 0.000 | 0.200 | 0.000 | 99.300 | 33.350 | 30.861 | 35.788 | 0.531 | 0.054 | 37.159 |
|  | 10BH167 A10 31-35 av | 51.760 | 2.340 | 11.260 | 0.480 | 0.100 | 15.400 | 0.300 | 17.300 | 0.000 | 0.200 | 0.000 | 99.120 | 37.777 | 40.427 | 21.796 | 0.709 | 2.037 | 18.543 |
| DZ | 10BH167 A11-1 4 | 50.000 | 2.200 | 10.300 | 0.400 | 0.100 | 19.200 | 0.200 | 14.600 | 0.000 | 0.200 | 0.000 | 97.300 | 31.188 | 49.307 | 19.505 | 0.769 | 5.715 | 13.255 |
|  | 10BH167 A11-1 5 | 49.000 | 2.300 | 12.600 | 0.300 | 0.000 | 18.000 | 0.300 | 14.800 | 0.000 | 0.200 | 0.000 | 97.500 | 31.087 | 45.452 | 23.461 | 0.718 | 3.828 | 15.310 |
|  | 10BH167 A11-1 6 | 49.400 | 2.100 | 13.700 | 0.500 | 0.000 | 17.900 | 0.300 | 14.400 | 0.000 | 0.200 | 0.000 | 98.600 | 29.960 | 44.772 | 25.268 | 0.700 | 3.196 | 16.234 |
|  | 10BH167 A11-1 4-6 <br> av | 49.467 | 2.200 | 12.200 | 0.400 | 0.033 | 18.367 | 0.267 | 14.600 | 0.000 | 0.200 | 0.000 | 97.800 | 30.740 | 46.489 | 22.771 | 0.728 | 4.149 | 14.896 |
|  | 10BH167 A11-1 11 | 48.100 | 3.000 | 10.500 | 0.400 | 0.000 | 16.100 | 0.200 | 18.300 | 0.000 | 0.200 | 0.000 | 96.800 | 38.967 | 41.214 | 19.820 | 0.732 | 4.013 | 15.068 |
|  | 10BH167 A11-1 12 | 48.000 | 2.800 | 11.600 | 0.700 | 0.100 | 16.100 | 0.200 | 17.700 | 0.000 | 0.200 | 0.000 | 97.500 | 37.391 | 40.887 | 21.723 | 0.712 | 3.453 | 15.838 |
|  | 10BH167 A11-1 13 | 48.900 | 2.700 | 13.400 | 0.800 | 0.000 | 15.300 | 0.300 | 17.800 | 0.000 | 0.200 | 0.000 | 99.400 | 37.028 | 38.262 | 24.710 | 0.670 | 2.056 | 18.497 |
|  | 10BH167 A11-1 14 | 48.400 | 1.800 | 19.200 | 0.500 | 0.000 | 11.600 | 0.400 | 17.200 | 0.000 | 0.200 | 0.000 | 99.300 | 35.710 | 28.953 | 35.337 | 0.518 | 0.097 | 34.126 |
|  | $\begin{aligned} & \text { 10BH167 A11-1 11- } \\ & 14 \text { av } \end{aligned}$ | 48.350 | 2.575 | 13.675 | 0.600 | 0.025 | 14.775 | 0.275 | 17.750 | 0.000 | 0.200 | 0.000 | 98.250 | 37.263 | 37.288 | 25.449 | 0.658 | 1.701 | 19.466 |
|  | 10BH167 A11-1 20 | 50.700 | 2.800 | 11.600 | 0.600 | 0.000 | 16.200 | 0.200 | 18.600 | 0.000 | 0.200 | 0.000 | 100.900 | 38.463 | 40.273 | 21.264 | 0.713 | 2.998 | 16.562 |
|  | 10BH167 A11-1 21 | 50.100 | 3.000 | 13.500 | 0.700 | 0.000 | 15.700 | 0.300 | 17.300 | 0.000 | 0.300 | 0.000 | 100.700 | 35.936 | 39.206 | 24.859 | 0.675 | 2.120 | 18.338 |
|  | 10BH167 A11-1 22 | 49.100 | 1.700 | 22.000 | 0.500 | 0.100 | 13.900 | 0.500 | 13.000 | 0.000 | 0.200 | 0.000 | 100.900 | 26.416 | 33.955 | 39.629 | 0.530 | 0.253 | 29.245 |
|  | $\begin{aligned} & \text { 10BH167 A11-1 20- } \\ & 22 \text { av } \end{aligned}$ | 49.967 | 2.500 | 15.700 | 0.600 | 0.033 | 15.267 | 0.333 | 16.300 | 0.000 | 0.233 | 0.000 | 100.833 | 33.559 | 37.787 | 28.654 | 0.634 | 1.253 | 21.035 |
|  | 10BH167 A11-1 23 | 48.600 | 2.400 | 11.700 | 0.400 | 0.100 | 14.100 | 0.300 | 19.500 | 0.100 | 0.200 | 0.000 | 97.300 | 41.647 | 36.202 | 22.151 | 0.682 | 1.853 | 19.028 |
|  | 10BH167 A11-1 24 | 49.200 | 2.500 | 13.000 | 0.500 | 0.000 | 16.700 | 0.300 | 15.900 | 0.000 | 0.200 | 0.000 | 98.300 | 33.473 | 42.266 | 24.261 | 0.696 | 2.834 | 16.850 |
|  | 10BH167 A11-1 25 | 47.300 | 1.800 | 21.000 | 0.600 | 0.100 | 13.600 | 0.500 | 13.600 | 0.000 | 0.200 | 0.000 | 98.600 | 28.003 | 33.665 | 38.331 | 0.536 | 0.335 | 27.794 |
|  | $\begin{aligned} & \text { 10BH167 A11-1 23- } \\ & 25 \mathrm{av} \\ & \hline \end{aligned}$ | 48.367 | 2.233 | 15.233 | 0.500 | 0.067 | 14.800 | 0.367 | 16.333 | 0.033 | 0.200 | 0.000 | 98.067 | 34.292 | 37.356 | 28.352 | 0.634 | 1.264 | 20.988 |
|  | 10BH167 A11-1 31 | 49.000 | 1.900 | 21.900 | 0.800 | 0.000 | 13.300 | 0.400 | 13.900 | 0.000 | 0.200 | 0.000 | 101.300 | 28.193 | 32.430 | 39.376 | 0.520 | 0.195 | 30.569 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | Melt <br> MgO | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 10BH167 A11-1 32 | 48.600 | 2.100 | 18.900 | 0.600 | 0.000 | 12.400 | 0.400 | 16.700 | 0.000 | 0.200 | 0.000 | 99.900 | 34.532 | 30.824 | 34.644 | 0.539 | 0.199 | 30.466 |
|  | 10BH167 A11-1 33 | 47.600 | 2.100 | 21.100 | 0.900 | 0.000 | 12.000 | 0.400 | 15.100 | 0.000 | 0.200 | 0.000 | 99.400 | 31.308 | 29.911 | 38.781 | 0.503 | 0.120 | 33.043 |
|  | $\begin{aligned} & \text { 10BH167 A11-1 31- } \\ & 33 \text { av } \end{aligned}$ | 48.400 | 2.033 | 20.633 | 0.767 | 0.000 | 12.567 | 0.400 | 15.233 | 0.000 | 0.200 | 0.000 | 100.200 | 31.324 | 31.065 | 37.611 | 0.520 | 0.168 | 31.347 |
| DZ | 10BH167 A11-2 5 | 49.500 | 2.000 | 16.300 | 0.600 | 0.000 | 14.600 | 0.400 | 14.900 | 0.000 | 0.200 | 0.000 | 98.600 | 31.769 | 37.423 | 30.808 | 0.615 | 0.788 | 23.412 |
|  | 10BH167 A11-2 6 | 50.900 | 2.600 | 11.200 | 0.600 | 0.000 | 16.900 | 0.300 | 16.600 | 0.000 | 0.200 | 0.000 | 99.200 | 35.436 | 43.370 | 21.194 | 0.729 | 3.419 | 15.889 |
|  | 10BH167 A11-2 7 | 50.400 | 2.300 | 13.400 | 0.500 | 0.000 | 15.300 | 0.300 | 16.000 | 0.000 | 0.200 | 0.000 | 98.500 | 34.578 | 39.751 | 25.671 | 0.670 | 1.529 | 20.015 |
|  | 10BH167 A11-2 5-7 av | 50.267 | 2.300 | 13.633 | 0.567 | 0.000 | 15.600 | 0.333 | 15.833 | 0.000 | 0.200 | 0.000 | 98.767 | 33.924 | 40.182 | 25.894 | 0.671 | 1.680 | 19.533 |
| DZ | 10BH167 A12 1 | 49.400 | 1.300 | 24.500 | 0.500 | 0.000 | 11.000 | 0.600 | 10.800 | 0.000 | 0.100 | 0.000 | 98.300 | 23.610 | 28.910 | 47.480 | 0.444 | 0.006 | 48.654 |
|  | 10BH167 A12 2 | 50.800 | 2.400 | 15.900 | 0.600 | 0.000 | 15.100 | 0.300 | 13.600 | 0.000 | 0.300 | 0.000 | 99.000 | 29.663 | 39.594 | 30.743 | 0.629 | 0.884 | 22.825 |
|  | 10BH167 A12 3 | 52.100 | 2.100 | 10.400 | 0.400 | 0.000 | 17.600 | 0.200 | 15.000 | 0.000 | 0.200 | 0.000 | 98.000 | 33.056 | 46.627 | 20.317 | 0.751 | 3.694 | 15.492 |
|  | 10BH167 A12 4 | 50.400 | 2.700 | 8.400 | 0.600 | 0.000 | 15.600 | 0.200 | 19.100 | 0.100 | 0.200 | 0.000 | 97.400 | 42.163 | 41.399 | 16.438 | 0.768 | 4.269 | 14.751 |
|  | 10BH167 A12 5 | 50.100 | 2.300 | 13.900 | 0.600 | 0.000 | 16.800 | 0.300 | 13.400 | 0.000 | 0.200 | 0.000 | 97.700 | 29.182 | 43.984 | 26.834 | 0.683 | 2.190 | 18.172 |
|  | 10BH167 A12 6 | 50.000 | 2.800 | 17.000 | 0.600 | 0.000 | 15.000 | 0.400 | 12.900 | 0.000 | 0.200 | 0.000 | 98.900 | 28.042 | 39.199 | 32.759 | 0.611 | 0.809 | 23.278 |
|  | 10BH167 A12 1-6 av | 50.467 | 2.267 | 15.017 | 0.550 | 0.000 | 15.183 | 0.333 | 14.133 | 0.017 | 0.200 | 0.000 | 98.217 | 30.927 | 39.943 | 29.130 | 0.643 | 1.049 | 21.946 |
|  | 10BH167 A12 11 | 44.600 | 5.900 | 24.900 | 0.200 | 0.000 | 10.700 | 0.400 | 6.200 | 0.000 | 0.400 | 0.100 | 93.300 | 15.072 | 31.270 | 53.658 | 0.434 | 0.051 | 37.415 |
|  | 10BH167 A12 13 | 44.500 | 4.800 | 26.000 | 1.200 | 0.000 | 7.000 | 0.400 | 9.100 | 0.000 | 1.800 | 0.600 | 95.500 | 22.434 | 20.746 | 56.820 | 0.324 | 0.001 | 59.665 |
|  | 10BH167 A12 21 | 43.100 | 6.400 | 28.300 | 0.900 | 0.000 | 4.900 | 0.400 | 10.100 | 0.000 | 1.700 | 0.900 | 96.600 | 24.588 | 14.340 | 61.072 | 0.236 | 0.000 | 88.177 |
|  | 10BH167 A12 22 | 50.600 | 2.300 | 14.600 | 0.500 | 0.000 | 16.500 | 0.300 | 14.300 | 0.100 | 0.200 | 0.000 | 99.400 | 30.375 | 42.134 | 27.491 | 0.668 | 1.828 | 19.099 |
|  | 10BH167 A12 23 | 50.200 | 3.900 | 14.600 | 0.700 | 0.000 | 17.000 | 0.300 | 13.800 | 0.000 | 0.200 | 0.000 | 100.600 | 29.250 | 43.318 | 27.432 | 0.675 | 2.477 | 17.541 |
|  | 10BH167 A12 24 | 42.200 | 6.400 | 29.900 | 1.300 | 0.000 | 4.100 | 0.400 | 9.400 | 0.000 | 1.900 | 0.700 | 96.300 | 23.020 | 12.071 | 64.910 | 0.196 | 0.000 | 107.020 |
|  | $\begin{aligned} & \text { 10BH167 A12 21-24 } \\ & \text { av } \end{aligned}$ | 46.525 | 4.750 | 21.850 | 0.850 | 0.000 | 10.625 | 0.350 | 11.900 | 0.025 | 1.000 | 0.400 | 98.225 | 27.019 | 29.002 | 43.979 | 0.464 | 0.089 | 34.619 |
| UBZ | 10BH167 A14 1 | 52.000 | 4.100 | 6.900 | 0.400 | 0.500 | 17.800 | 0.200 | 17.200 | 0.000 | 0.200 | 0.000 | 99.200 | 38.465 | 47.856 | 13.679 | 0.821 | 7.552 | 11.826 |
|  | 10BH167 A14 2 | 51.800 | 3.900 | 6.600 | 0.400 | 0.600 | 17.500 | 0.200 | 18.500 | 0.100 | 0.200 | 0.000 | 99.800 | 40.759 | 46.351 | 12.890 | 0.825 | 7.806 | 11.657 |
|  | 10BH167 A14 3 | 52.700 | 1.100 | 19.200 | 0.400 | 0.100 | 21.300 | 0.400 | 4.600 | 0.000 | 0.100 | 0.000 | 99.800 | 9.740 | 54.220 | 36.039 | 0.664 | 2.333 | 17.849 |
|  | 10BH167 A14 1-3 av | 52.167 | 3.033 | 10.900 | 0.400 | 0.400 | 18.867 | 0.267 | 13.433 | 0.033 | 0.167 | 0.000 | 99.600 | 29.345 | 49.547 | 21.108 | 0.755 | 4.660 | 14.301 |
|  | 10BH167 A14 10 | 51.900 | 2.900 | 9.900 | 0.500 | 0.300 | 18.000 | 0.200 | 15.200 | 0.000 | 0.200 | 0.000 | 99.200 | 33.322 | 47.439 | 19.239 | 0.764 | 4.714 | 14.243 |
|  | 10BH167 A14 11 | 50.500 | 4.700 | 6.600 | 0.500 | 0.700 | 16.700 | 0.200 | 18.300 | 0.000 | 0.200 | 0.000 | 98.400 | 41.377 | 45.394 | 13.229 | 0.818 | 7.597 | 11.796 |
|  | 10BH167 A14 12 | 50.300 | 2.300 | 13.900 | 0.800 | 0.200 | 14.300 | 0.300 | 16.500 | 0.000 | 0.200 | 0.000 | 98.800 | 35.859 | 37.362 | 26.779 | 0.647 | 1.007 | 22.154 |
|  | 10BH167 A14 13 | 51.300 | 2.300 | 11.600 | 0.500 | 0.000 | 15.300 | 0.300 | 17.300 | 0.000 | 0.200 | 0.000 | 98.900 | 37.628 | 40.006 | 22.366 | 0.702 | 1.935 | 18.808 |
|  | 10BH167 A14 14 | 51.500 | 3.800 | 7.000 | 0.400 | 0.300 | 16.600 | 0.200 | 19.600 | 0.000 | 0.200 | 0.000 | 99.500 | 42.831 | 43.609 | 13.560 | 0.809 | 6.599 | 12.518 |

Appendix E3 - LPS Clinopyroxene mineral chemistry \& inverse melt compositions

| Zone | Sample code | $\mathrm{SiO}_{2}$ | $\mathrm{Al}_{2} \mathrm{O}_{3}$ | FeO* | $\mathrm{TiO}_{2}$ | $\mathrm{Cr}_{2} \mathrm{O}_{3}$ | MgO | MnO | CaO | NiO | $\mathrm{Na}_{2} \mathrm{O}$ | $\mathrm{K}_{2} \mathrm{O}$ | Total | Womol\% | Enmol\% | Fsmol\% | Mgmol\% | Melt <br> MgO | $\begin{aligned} & \hline \text { Melt } \\ & \mathrm{FeO} \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 10BH167 A14 15 | 51.400 | 5.000 | 6.600 | 0.300 | 0.500 | 16.600 | 0.100 | 19.300 | 0.000 | 0.200 | 0.000 | 100.000 | 42.787 | 44.242 | 12.971 | 0.818 | 7.372 | 11.950 |
|  | 10BH167 A14 16 | 50.700 | 3.300 | 12.200 | 0.500 | 0.300 | 15.200 | 0.300 | 16.800 | 0.000 | 0.200 | 0.000 | 99.600 | 36.611 | 39.821 | 23.568 | 0.689 | 2.007 | 18.619 |
|  | 10BH167 A14 17 | 51.000 | 4.300 | 7.000 | 0.600 | 0.700 | 17.000 | 0.200 | 18.100 | 0.000 | 0.200 | 0.000 | 99.100 | 40.454 | 45.677 | 13.869 | 0.812 | 7.205 | 12.068 |
|  | 10BH167 A14 18 | 51.200 | 4.300 | 7.100 | 0.400 | 0.500 | 16.900 | 0.200 | 18.000 | 0.000 | 0.200 | 0.000 | 98.800 | 40.349 | 45.543 | 14.109 | 0.809 | 6.846 | 12.329 |
|  | 10BH167 A14 19 | 51.200 | 4.200 | 6.700 | 0.600 | 0.400 | 16.800 | 0.200 | 18.500 | 0.000 | 0.200 | 0.000 | 98.700 | 41.446 | 45.247 | 13.306 | 0.817 | 7.219 | 12.058 |
|  | 10BH167 A14 20 | 51.400 | 4.100 | 7.300 | 0.600 | 0.500 | 16.800 | 0.200 | 18.000 | 0.100 | 0.200 | 0.000 | 99.100 | 40.297 | 45.215 | 14.487 | 0.804 | 6.488 | 12.605 |
|  | $\begin{aligned} & \text { 10BH167 A14 10-20 } \\ & \text { av } \end{aligned}$ | 51.127 | 3.745 | 8.718 | 0.518 | 0.400 | 16.382 | 0.218 | 17.782 | 0.009 | 0.200 | 0.000 | 99.100 | 39.337 | 43.567 | 17.097 | 0.770 | 4.788 | 14.162 |
|  | 10BH167 A14 27 | 50.900 | 4.700 | 6.400 | 0.400 | 0.600 | 16.900 | 0.100 | 18.700 | 0.000 | 0.200 | 0.000 | 99.000 | 41.844 | 45.462 | 12.695 | 0.825 | 7.910 | 11.589 |
|  | 10BH167 A14 28 | 51.000 | 3.400 | 8.300 | 0.300 | 0.200 | 17.100 | 0.200 | 16.900 | 0.100 | 0.200 | 0.000 | 97.700 | 37.711 | 45.872 | 16.418 | 0.786 | 5.542 | 13.412 |
|  | 10BH167 A14 29 | 49.200 | 4.500 | 16.100 | 0.700 | 0.000 | 13.800 | 0.300 | 10.900 | 0.000 | 0.900 | 0.100 | 96.700 | 26.100 | 39.725 | 34.175 | 0.604 | 0.819 | 23.217 |
|  | 10BH167 A14 27-29 av | 50.367 | 4.200 | 10.267 | 0.467 | 0.267 | 15.933 | 0.200 | 15.500 | 0.033 | 0.433 | 0.033 | 97.800 | 35.424 | 43.776 | 20.800 | 0.734 | 3.606 | 15.616 |
|  | 10BH167 A14 30 | 50.900 | 2.300 | 12.900 | 0.700 | 0.000 | 14.900 | 0.300 | 16.300 | 0.000 | 0.200 | 0.000 | 98.600 | 35.708 | 39.241 | 25.051 | 0.673 | 1.361 | 20.609 |
|  | 10BH167 A14 31 | 50.800 | 2.600 | 12.800 | 0.700 | 0.000 | 15.000 | 0.200 | 16.300 | 0.000 | 0.200 | 0.000 | 98.600 | 35.683 | 39.477 | 24.840 | 0.676 | 1.499 | 20.116 |
|  | 10BH167 A14 32 | 50.300 | 2.200 | 12.800 | 0.600 | 0.100 | 14.400 | 0.300 | 16.800 | 0.000 | 0.200 | 0.000 | 97.700 | 36.957 | 38.082 | 24.961 | 0.667 | 1.216 | 21.190 |
|  | $\begin{aligned} & \text { 10BH167 A14 30-32 } \\ & \text { av } \end{aligned}$ | 50.667 | 2.367 | 12.833 | 0.667 | 0.033 | 14.767 | 0.267 | 16.467 | 0.000 | 0.200 | 0.000 | 98.300 | 36.115 | 38.934 | 24.951 | 0.672 | 1.357 | 20.627 |

Appendix E4 - LPS Plagioclase compositions

| Zone | Sample code | SiO2 | TiO2 | Al2O3 | MgO | CaO | MnO | FeO | Na2O | K2O | BaO | SrO | Total | Anmol\% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| LCM | 10BH166 A1 20 | 51.240 | 0.076 | 29.739 | 0.274 | 13.994 | 0.000 | 1.119 | 3.619 | 0.199 | 0.000 | 0.066 | 100.326 | 67.344 |
|  | 10BH166 A1 21 | 51.767 | 0.027 | 29.624 | 0.117 | 13.144 | 0.018 | 0.965 | 4.188 | 0.287 | 0.053 | 0.028 | 100.218 | 62.343 |
|  | $\begin{aligned} & \text { 10BH166 A1 20-21 } \\ & \text { av } \end{aligned}$ | 51.504 | 0.052 | 29.682 | 0.196 | 13.569 | 0.009 | 1.042 | 3.904 | 0.243 | 0.027 | 0.047 | 100.272 | 64.825 |
| LBZ | 10BH166 A2 12 | 50.807 | 0.049 | 30.277 | 0.214 | 13.945 | 0.018 | 1.061 | 3.768 | 0.189 | 0.000 | 0.052 | 100.380 | 66.441 |
|  | 10BH166 A2 13 | 51.358 | 0.000 | 29.807 | 0.208 | 13.427 | 0.071 | 1.014 | 4.052 | 0.232 | 0.000 | 0.000 | 100.169 | 63.830 |
|  | $\begin{aligned} & \text { 10BH166 A2 12-13 } \\ & \text { av } \end{aligned}$ | 51.083 | 0.025 | 30.042 | 0.211 | 13.686 | 0.045 | 1.038 | 3.910 | 0.211 | 0.000 | 0.026 | 100.275 | 65.134 |
|  | 10BH166 A2 14 | 49.594 | 0.022 | 30.764 | 0.308 | 14.978 | 0.012 | 0.709 | 3.210 | 0.096 | 0.000 | 0.073 | 99.766 | 71.662 |
|  | 10BH166 A2 15 | 50.859 | 0.005 | 30.383 | 0.210 | 14.363 | 0.041 | 0.781 | 3.576 | 0.150 | 0.000 | 0.021 | 100.389 | 68.354 |
|  | 10BH166 A2 16 | 50.990 | 0.092 | 29.723 | 0.130 | 13.543 | 0.000 | 0.982 | 3.863 | 0.239 | 0.000 | 0.105 | 99.667 | 65.055 |
|  | $\begin{aligned} & \text { 10BH166 A2 14-16 } \\ & \text { av } \end{aligned}$ | 50.481 | 0.040 | 30.290 | 0.216 | 14.295 | 0.018 | 0.824 | 3.550 | 0.162 | 0.000 | 0.066 | 99.941 | 68.361 |
|  | 10BH166 A2 17 | 50.011 | 0.000 | 30.531 | 0.453 | 14.454 | 0.000 | 1.150 | 3.335 | 0.135 | 0.007 | 0.056 | 100.132 | 69.988 |
|  | 10BH166 A2 18 | 50.167 | 0.049 | 30.989 | 0.275 | 14.874 | 0.041 | 0.855 | 3.346 | 0.120 | 0.000 | 0.049 | 100.765 | 70.588 |
|  | 10BH166 A2 19 | 52.400 | 0.038 | 29.555 | 0.143 | 13.206 | 0.000 | 1.118 | 4.134 | 0.247 | 0.000 | 0.042 | 100.883 | 62.943 |
|  | $\begin{aligned} & \text { 10BH166 A2 17-19 } \\ & \text { av } \end{aligned}$ | 50.859 | 0.029 | 30.358 | 0.290 | 14.178 | 0.014 | 1.041 | 3.605 | 0.167 | 0.002 | 0.049 | 100.593 | 67.832 |
| LBZ | 10BH166 A3 15 | 49.557 | 0.027 | 30.476 | 0.232 | 14.527 | 0.000 | 0.715 | 3.432 | 0.130 | 0.000 | 0.084 | 99.180 | 69.533 |
|  | 10BH166 A3 16 | 46.214 | 0.069 | 29.926 | 1.075 | 11.648 | 0.195 | 5.650 | 3.179 | 0.134 | 0.020 | 0.074 | 98.184 | 66.304 |
|  | 10BH166 A3 17 | 49.912 | 0.000 | 31.083 | 0.148 | 14.735 | 0.000 | 0.845 | 3.256 | 0.147 | 0.000 | 0.049 | 100.175 | 70.835 |
|  | $\begin{aligned} & \text { 10BH166 A3 15-17 } \\ & \text { av } \end{aligned}$ | 48.561 | 0.032 | 30.495 | 0.485 | 13.637 | 0.065 | 2.403 | 3.289 | 0.137 | 0.007 | 0.069 | 99.180 | 69.033 |
|  | 10BH166 A3 18 | 50.204 | 0.108 | 30.506 | 0.285 | 14.439 | 0.000 | 0.697 | 3.531 | 0.136 | 0.000 | 0.035 | 99.941 | 68.788 |
|  | 10BH166 A3 19 | 49.908 | 0.114 | 30.681 | 0.231 | 14.642 | 0.000 | 0.767 | 3.364 | 0.132 | 0.000 | 0.056 | 99.895 | 70.103 |
|  | 10BH166 A3 20 | 52.337 | 0.027 | 29.218 | 0.071 | 12.246 | 0.000 | 0.778 | 4.640 | 0.253 | 0.000 | 0.063 | 99.633 | 58.471 |
|  | $\begin{aligned} & \text { 10BH166 A3 18-20 } \\ & \text { av } \end{aligned}$ | 50.816 | 0.083 | 30.135 | 0.196 | 13.776 | 0.000 | 0.747 | 3.845 | 0.174 | 0.000 | 0.051 | 99.823 | 65.786 |
|  | 10BH166 A3 27 | 49.680 | 0.032 | 30.650 | 0.202 | 14.521 | 0.000 | 0.787 | 3.513 | 0.142 | 0.000 | 0.115 | 99.642 | 68.993 |
|  | 10BH166 A3 28 | 48.615 | 0.092 | 31.106 | 0.109 | 14.930 | 0.050 | 0.922 | 3.207 | 0.154 | 0.000 | 0.080 | 99.265 | 71.379 |
|  | 10BH166 A3 29 | 49.744 | 0.065 | 30.453 | 0.144 | 14.146 | 0.000 | 0.950 | 3.470 | 0.206 | 0.000 | 0.052 | 99.230 | 68.436 |
|  | 10BH166 A3 30 | 49.346 | 0.063 | 30.736 | 0.152 | 14.532 | 0.017 | 0.886 | 3.397 | 0.167 | 0.000 | 0.082 | 99.379 | 69.606 |
|  | 10BH166 A3 36 | 49.826 | 0.087 | 30.933 | 0.253 | 14.891 | 0.000 | 0.595 | 3.265 | 0.130 | 0.000 | 0.003 | 99.892 | 71.065 |
|  | 10BH166 A3 37 | 48.971 | 0.000 | 30.760 | 0.247 | 14.941 | 0.000 | 0.646 | 3.264 | 0.130 | 0.000 | 0.000 | 98.959 | 71.140 |
|  | 10BH166 A3 38 | 49.355 | 0.000 | 30.599 | 0.276 | 14.608 | 0.000 | 0.669 | 3.464 | 0.153 | 0.000 | 0.084 | 99.208 | 69.369 |

Appendix E4 - LPS Plagioclase compositions

| Zone | Sample code | SiO2 | TiO2 | Al2O3 | MgO | CaO | MnO | FeO | Na2O | K2O | BaO | SrO | Total | Anmol\% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 10BH166 A3 36-38 <br> av | 49.384 | 0.029 | 30.764 | 0.259 | 14.813 | 0.000 | 0.637 | 3.331 | 0.138 | 0.000 | 0.029 | 99.353 | 70.523 |
| LOZ | BH311 B2 130 | 49.219 | 0.103 | 30.421 | 0.346 | 14.850 | 0.000 | 0.761 | 3.206 | 0.116 | 0.000 | 0.000 | 99.022 | 71.430 |
|  | BH311 B2 131 | 49.557 | 0.038 | 30.440 | 0.221 | 14.601 | 0.000 | 0.645 | 3.269 | 0.134 | 0.000 | 0.080 | 98.985 | 70.618 |
|  | BH311 B2 132 | 49.882 | 0.114 | 30.285 | 0.174 | 14.239 | 0.000 | 0.880 | 3.553 | 0.178 | 0.000 | 0.000 | 99.305 | 68.193 |
|  | $\begin{aligned} & \text { BH311 B2 130-132 } \\ & \text { av } \end{aligned}$ | 49.553 | 0.085 | 30.382 | 0.247 | 14.563 | 0.000 | 0.762 | 3.343 | 0.143 | 0.000 | 0.027 | 99.104 | 70.081 |
| LOZ | 10BH166 A4 16 | 50.602 | 0.027 | 30.590 | 0.204 | 14.632 | 0.000 | 0.687 | 3.313 | 0.178 | 0.000 | 0.035 | 100.268 | 70.214 |
|  | 10BH166 A4 17 | 49.675 | 0.000 | 30.726 | 0.273 | 14.742 | 0.000 | 0.607 | 3.171 | 0.112 | 0.009 | 0.024 | 99.339 | 71.505 |
|  | 10BH166 A4 18 | 50.591 | 0.049 | 30.752 | 0.173 | 14.612 | 0.021 | 0.795 | 3.492 | 0.156 | 0.002 | 0.031 | 100.674 | 69.194 |
|  | $\begin{aligned} & \text { 10BH166 A4 16-18 } \\ & \text { av } \end{aligned}$ | 50.289 | 0.025 | 30.689 | 0.217 | 14.662 | 0.007 | 0.696 | 3.325 | 0.149 | 0.004 | 0.030 | 100.094 | 70.295 |
|  | 10BH166 A4 29 | 51.425 | 0.043 | 30.361 | 0.175 | 13.915 | 0.009 | 0.776 | 3.832 | 0.191 | 0.000 | 0.031 | 100.758 | 66.021 |
|  | 10BH166 A4 30 | 50.890 | 0.076 | 30.543 | 0.188 | 14.133 | 0.024 | 0.750 | 3.679 | 0.177 | 0.013 | 0.035 | 100.508 | 67.281 |
|  | 10BH166 A4 31 | 50.615 | 0.097 | 30.440 | 0.137 | 13.608 | 0.000 | 0.917 | 3.874 | 0.212 | 0.000 | 0.035 | 99.935 | 65.202 |
|  | $\begin{aligned} & \text { 10BH166 A4 29-31 } \\ & \text { av } \end{aligned}$ | 50.977 | 0.072 | 30.448 | 0.167 | 13.885 | 0.011 | 0.814 | 3.795 | 0.193 | 0.004 | 0.034 | 100.400 | 66.170 |
| LOZ | 10BH166 A5 9 | 50.044 | 0.054 | 30.492 | 0.211 | 14.369 | 0.006 | 0.868 | 3.522 | 0.151 | 0.002 | 0.105 | 99.824 | 68.676 |
|  | 10BH166 A5 10 | 50.534 | 0.059 | 30.283 | 0.200 | 14.110 | 0.000 | 0.710 | 3.622 | 0.161 | 0.000 | 0.031 | 99.710 | 67.655 |
|  | 10BH166 A5 11 | 50.854 | 0.000 | 30.049 | 0.172 | 13.880 | 0.000 | 0.707 | 3.715 | 0.186 | 0.000 | 0.059 | 99.622 | 66.654 |
|  | $\begin{aligned} & \text { 10BH166 A5 9-11 } \\ & \text { av } \end{aligned}$ | 50.477 | 0.038 | 30.275 | 0.194 | 14.120 | 0.002 | 0.762 | 3.620 | 0.166 | 0.001 | 0.065 | 99.719 | 67.663 |
|  | 10BH166 A5 12 | 47.763 | 0.043 | 29.055 | 0.148 | 15.794 | 0.000 | 0.769 | 3.339 | 0.160 | 0.007 | 0.021 | 97.099 | 71.696 |
|  | 10BH166 A5 13 | 49.438 | 0.108 | 27.718 | 2.462 | 11.771 | 0.035 | 2.516 | 3.884 | 0.540 | 0.005 | 0.011 | 98.488 | 60.538 |
|  | $\begin{aligned} & \text { 10BH166 A5 12-13 } \\ & \text { av } \end{aligned}$ | 48.601 | 0.076 | 28.387 | 1.305 | 13.783 | 0.018 | 1.643 | 3.612 | 0.350 | 0.006 | 0.016 | 97.794 | 66.464 |
|  | 10BH166 A5 22 | 50.286 | 0.016 | 30.654 | 0.209 | 14.311 | 0.009 | 0.631 | 3.505 | 0.149 | 0.000 | 0.056 | 99.826 | 68.701 |
|  | 10BH166 A5 23 | 50.319 | 0.049 | 30.265 | 0.219 | 14.189 | 0.032 | 0.743 | 3.626 | 0.164 | 0.000 | 0.066 | 99.672 | 67.742 |
|  | 10BH166 A5 24 | 50.725 | 0.140 | 30.794 | 0.170 | 13.276 | 0.006 | 0.841 | 4.134 | 0.256 | 0.000 | 0.035 | 100.377 | 63.034 |
|  | $\begin{aligned} & \text { 10BH166 A5 22-24 } \\ & \text { av } \end{aligned}$ | 50.443 | 0.068 | 30.571 | 0.199 | 13.925 | 0.016 | 0.738 | 3.755 | 0.190 | 0.000 | 0.052 | 99.958 | 66.482 |
| LOZ | 10BH167 A1 5 | 51.035 | 0.054 | 30.269 | 0.277 | 14.238 | 0.000 | 0.755 | 3.506 | 0.181 | 0.000 | 0.035 | 100.350 | 68.459 |
|  | 10BH167 A1 6 | 52.840 | 0.086 | 29.272 | 0.219 | 12.502 | 0.026 | 0.936 | 4.453 | 0.342 | 0.000 | 0.195 | 100.871 | 59.627 |
|  | 10BH167 A1 5-6 av | 51.938 | 0.070 | 29.771 | 0.248 | 13.370 | 0.013 | 0.846 | 3.980 | 0.262 | 0.000 | 0.115 | 100.611 | 64.025 |
|  | 10BH167 A1 14 | 50.482 | 0.049 | 30.346 | 0.201 | 14.060 | 0.000 | 0.742 | 3.681 | 0.179 | 0.000 | 0.105 | 99.845 | 67.163 |

Appendix E4 - LPS Plagioclase compositions

| Zone | Sample code | SiO2 | TiO2 | Al2O3 | MgO | CaO | MnO | FeO | Na2O | K2O | BaO | SrO | Total | Anmol\% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 10BH167 A1 15 | 51.149 | 0.016 | 30.324 | 0.173 | 14.118 | 0.000 | 0.798 | 3.673 | 0.206 | 0.000 | 0.014 | 100.471 | 67.197 |
|  | $\begin{aligned} & \text { 10BH167 A1 14-15 } \\ & \text { av } \end{aligned}$ | 50.816 | 0.033 | 30.335 | 0.187 | 14.089 | 0.000 | 0.770 | 3.677 | 0.193 | 0.000 | 0.060 | 100.158 | 67.180 |
|  | 10BH167 A1 18 | 50.441 | 0.038 | 30.771 | 0.229 | 14.639 | 0.015 | 0.691 | 3.461 | 0.155 | 0.000 | 0.070 | 100.510 | 69.424 |
|  | 10BH167 A1 19 | 48.812 | 0.070 | 28.858 | 2.494 | 12.778 | 0.000 | 2.817 | 3.131 | 0.178 | 0.000 | 0.011 | 99.149 | 68.494 |
|  | 10BH167 A1 20 | 51.174 | 0.054 | 29.897 | 0.137 | 13.704 | 0.000 | 0.888 | 3.912 | 0.215 | 0.018 | 0.059 | 100.058 | 65.116 |
|  | $\begin{aligned} & \text { 10BH167 A1 18-20 } \\ & \text { av } \end{aligned}$ | 50.142 | 0.054 | 29.842 | 0.953 | 13.707 | 0.005 | 1.465 | 3.501 | 0.183 | 0.006 | 0.047 | 99.906 | 67.647 |
| MOZ | 10BH167 A2 6 | 49.667 | 0.059 | 30.653 | 0.250 | 14.633 | 0.018 | 0.809 | 3.335 | 0.131 | 0.000 | 0.070 | 99.625 | 70.270 |
|  | 10BH167 A2 7 | 49.833 | 0.130 | 31.264 | 0.221 | 15.141 | 0.009 | 0.578 | 3.175 | 0.126 | 0.000 | 0.084 | 100.561 | 71.975 |
|  | 10BH167 A2 8 | 50.118 | 0.022 | 30.824 | 0.231 | 14.893 | 0.012 | 0.616 | 3.229 | 0.097 | 0.007 | 0.066 | 100.115 | 71.415 |
|  | 10BH167 A2 6-8 av | 49.873 | 0.070 | 30.914 | 0.234 | 14.889 | 0.013 | 0.668 | 3.246 | 0.118 | 0.002 | 0.073 | 100.100 | 71.223 |
|  | 10BH167 A2 16 | 51.160 | 0.113 | 29.877 | 0.122 | 13.289 | 0.000 | 0.769 | 4.255 | 0.241 | 0.027 | 0.024 | 99.877 | 62.432 |
|  | 10BH167 A2 17 | 50.045 | 0.043 | 28.319 | 2.590 | 11.844 | 0.059 | 2.599 | 3.647 | 0.227 | 0.000 | 0.049 | 99.422 | 63.290 |
|  | 10BH167 A2 18 | 51.373 | 0.151 | 29.766 | 0.182 | 13.329 | 0.035 | 0.957 | 4.077 | 0.245 | 0.000 | 0.080 | 100.195 | 63.477 |
|  | $\begin{aligned} & \text { 10BH167 A2 16-18 } \\ & \text { av } \end{aligned}$ | 50.859 | 0.102 | 29.321 | 0.965 | 12.821 | 0.031 | 1.442 | 3.993 | 0.238 | 0.009 | 0.051 | 99.831 | 63.055 |
|  | 10BH167 A2 19 | 50.090 | 0.027 | 30.897 | 0.278 | 14.961 | 0.009 | 0.597 | 3.247 | 0.109 | 0.031 | 0.017 | 100.263 | 71.318 |
|  | 10BH167 A2 20 | 50.296 | 0.000 | 30.860 | 0.299 | 14.925 | 0.000 | 0.608 | 3.309 | 0.102 | 0.000 | 0.056 | 100.455 | 70.956 |
|  | 10BH167 A2 21 | 49.867 | 0.070 | 31.144 | 0.229 | 15.160 | 0.044 | 0.765 | 3.130 | 0.095 | 0.000 | 0.066 | 100.570 | 72.408 |
|  | $\begin{aligned} & \text { 10BH167 A2 19-21 } \\ & \text { av } \end{aligned}$ | 50.084 | 0.032 | 30.967 | 0.269 | 15.015 | 0.018 | 0.657 | 3.229 | 0.102 | 0.010 | 0.046 | 100.429 | 71.559 |
| MOZ | 10BH167 A3 13 | 49.893 | 0.076 | 30.608 | 0.168 | 14.194 | 0.000 | 0.686 | 3.635 | 0.155 | 0.000 | 0.077 | 99.492 | 67.731 |
|  | 10BH167 A3 14 | 50.869 | 0.005 | 30.672 | 0.146 | 14.091 | 0.000 | 0.799 | 3.636 | 0.220 | 0.000 | 0.000 | 100.438 | 67.316 |
|  | $\begin{aligned} & \text { 10BH167 A3 13-14 } \\ & \text { av } \end{aligned}$ | 50.381 | 0.041 | 30.640 | 0.157 | 14.143 | 0.000 | 0.743 | 3.636 | 0.188 | 0.000 | 0.039 | 99.965 | 67.524 |
|  | 10BH167 A3 15 | 49.970 | 0.022 | 30.688 | 0.203 | 14.390 | 0.000 | 0.598 | 3.646 | 0.147 | 0.000 | 0.063 | 99.727 | 67.997 |
|  | 10BH167 A3 16 | 51.467 | 0.081 | 29.787 | 0.181 | 13.354 | 0.000 | 0.681 | 4.128 | 0.230 | 0.033 | 0.063 | 100.005 | 63.260 |
|  | 10BH167 A3 17 | 51.674 | 0.124 | 29.881 | 0.147 | 13.036 | 0.006 | 0.731 | 4.261 | 0.237 | 0.000 | 0.108 | 100.205 | 61.991 |
|  | $\begin{aligned} & \text { 10BH167 A3 15-17 } \\ & \text { av } \end{aligned}$ | 51.037 | 0.076 | 30.119 | 0.177 | 13.593 | 0.002 | 0.670 | 4.012 | 0.205 | 0.011 | 0.078 | 99.979 | 64.422 |
|  | 10BH167 A3 25 | 49.886 | 0.092 | 30.822 | 0.220 | 14.577 | 0.000 | 0.713 | 3.464 | 0.135 | 0.000 | 0.000 | 99.909 | 69.394 |
|  | 10BH167 A3 26 | 49.320 | 0.054 | 30.410 | 0.809 | 14.430 | 0.021 | 1.365 | 3.321 | 0.147 | 0.000 | 0.056 | 99.933 | 69.999 |
|  | $\begin{aligned} & \text { 10BH167 A3 25-26 } \\ & \text { av } \end{aligned}$ | 49.603 | 0.073 | 30.616 | 0.515 | 14.504 | 0.011 | 1.039 | 3.393 | 0.141 | 0.000 | 0.028 | 99.921 | 69.694 |

Appendix E4 - LPS Plagioclase compositions

| Zone | Sample code | SiO2 | TiO2 | Al2O3 | MgO | CaO | MnO | FeO | Na2O | K2O | BaO | SrO | Total | Anmol\% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| UOZ | 10BH167 A4 1 | 49.766 | 0.097 | 30.649 | 0.269 | 14.802 | 0.000 | 0.539 | 3.254 | 0.127 | 0.014 | 0.045 | 99.562 | 71.004 |
|  | 10BH167 A4 2 | 50.794 | 0.000 | 30.804 | 0.218 | 14.647 | 0.018 | 0.562 | 3.456 | 0.131 | 0.000 | 0.073 | 100.703 | 69.559 |
|  | 10BH167 A4 3 | 52.097 | 0.167 | 29.628 | 0.150 | 13.183 | 0.041 | 0.839 | 4.184 | 0.249 | 0.000 | 0.014 | 100.552 | 62.625 |
|  | 10BH167 A4 1-3 av | 50.886 | 0.088 | 30.360 | 0.212 | 14.211 | 0.020 | 0.647 | 3.631 | 0.169 | 0.005 | 0.044 | 100.272 | 67.719 |
|  | 10BH167 A4 28 | 49.596 | 0.005 | 31.509 | 0.215 | 15.312 | 0.032 | 0.704 | 3.017 | 0.094 | 0.005 | 0.073 | 100.562 | 73.315 |
|  | 10BH167 A4 29 | 49.766 | 0.097 | 30.649 | 0.269 | 14.802 | 0.000 | 0.539 | 3.254 | 0.127 | 0.014 | 0.045 | 99.562 | 71.004 |
|  | $\begin{aligned} & \text { 10BH167 A4 28-29 } \\ & \text { av } \end{aligned}$ | 49.681 | 0.051 | 31.079 | 0.242 | 15.057 | 0.016 | 0.622 | 3.136 | 0.111 | 0.010 | 0.059 | 100.062 | 72.161 |
| UOZ | BH311 B3 24 | 50.548 | 0.054 | 29.870 | 0.661 | 13.567 | 0.027 | 1.145 | 3.609 | 0.171 | 0.000 | 0.000 | 99.652 | 66.828 |
|  | BH311 B3 24 | 50.851 | 0.098 | 29.610 | 0.605 | 13.892 | 0.000 | 0.848 | 3.659 | 0.188 | 0.014 | 0.094 | 99.859 | 66.975 |
|  | BH311 B3 24 | 50.646 | 0.049 | 29.962 | 0.233 | 14.167 | 0.024 | 0.667 | 3.703 | 0.163 | 0.000 | 0.010 | 99.624 | 67.264 |
|  | BH311 B3 24 | 49.542 | 0.098 | 29.672 | 1.099 | 13.703 | 0.006 | 1.099 | 3.376 | 0.137 | 0.007 | 0.000 | 98.739 | 68.591 |
|  | BH311 B3 24 | 51.237 | 0.049 | 30.294 | 0.200 | 14.371 | 0.015 | 0.633 | 3.577 | 0.155 | 0.000 | 0.021 | 100.552 | 68.341 |
|  | BH311 B3 24 | 50.531 | 0.043 | 30.150 | 0.244 | 14.230 | 0.024 | 0.671 | 3.582 | 0.149 | 0.000 | 0.042 | 99.666 | 68.121 |
|  | BH311 B3 24 | 50.600 | 0.087 | 30.213 | 0.304 | 14.307 | 0.052 | 0.711 | 3.502 | 0.143 | 0.000 | 0.115 | 100.034 | 68.736 |
|  | BH311 B3 24 | 50.716 | 0.000 | 30.345 | 0.280 | 14.425 | 0.000 | 0.656 | 3.479 | 0.144 | 0.000 | 0.031 | 100.076 | 69.046 |
|  | BH311 B3 24 | 50.861 | 0.060 | 30.427 | 0.219 | 14.377 | 0.000 | 0.632 | 3.485 | 0.147 | 0.000 | 0.031 | 100.239 | 68.927 |
|  | BH311 B3 24 | 51.039 | 0.027 | 30.454 | 0.245 | 14.381 | 0.018 | 0.617 | 3.402 | 0.158 | 0.000 | 0.010 | 100.351 | 69.389 |
|  | BH311 B3 24 | 50.618 | 0.119 | 30.581 | 0.229 | 14.462 | 0.000 | 0.598 | 3.495 | 0.134 | 0.047 | 0.073 | 100.356 | 68.988 |
|  | BH311 B3 24 | 51.173 | 0.000 | 30.477 | 0.208 | 14.622 | 0.027 | 0.573 | 3.420 | 0.150 | 0.000 | 0.052 | 100.702 | 69.664 |
|  | BH311 B3 24 | 50.246 | 0.000 | 30.709 | 0.207 | 14.547 | 0.000 | 0.583 | 3.455 | 0.149 | 0.000 | 0.031 | 99.927 | 69.349 |
|  | BH311 B3 24 | 50.250 | 0.005 | 30.544 | 0.200 | 14.590 | 0.000 | 0.587 | 3.404 | 0.124 | 0.000 | 0.035 | 99.739 | 69.817 |
|  | BH311 B3 24 | 49.828 | 0.065 | 30.637 | 0.241 | 14.561 | 0.000 | 0.658 | 3.314 | 0.131 | 0.000 | 0.098 | 99.533 | 70.296 |
|  | BH311 B3 24 | 51.067 | 0.049 | 30.609 | 0.206 | 14.442 | 0.039 | 0.625 | 3.513 | 0.144 | 0.000 | 0.024 | 100.718 | 68.868 |
|  | BH311 B3 24 | 50.898 | 0.049 | 30.314 | 0.218 | 14.102 | 0.039 | 0.646 | 3.654 | 0.154 | 0.000 | 0.070 | 100.144 | 67.482 |
|  | BH311 B3 24 | 51.176 | 0.043 | 30.015 | 0.204 | 13.989 | 0.000 | 0.720 | 3.695 | 0.171 | 0.004 | 0.052 | 100.069 | 66.996 |
|  | BH311 B3 24 | 50.537 | 0.022 | 28.792 | 1.627 | 12.356 | 0.061 | 2.085 | 3.702 | 0.164 | 0.000 | 0.056 | 99.402 | 64.186 |
|  | BH311 B3 24 | 41.020 | 0.085 | 22.175 | 10.754 | 5.743 | 0.143 | 10.021 | 1.959 | 0.180 | 0.000 | 0.000 | 92.080 | 60.438 |
|  | BH311 B3 24 av | 50.169 | 0.050 | 29.793 | 0.909 | 13.742 | 0.024 | 1.239 | 3.449 | 0.153 | 0.004 | 0.042 | 99.573 | 67.915 |
|  | BH311 B3 34 | 50.215 | 0.060 | 30.126 | 0.199 | 14.264 | 0.058 | 0.586 | 3.498 | 0.120 | 0.051 | 0.028 | 99.205 | 68.724 |
|  | BH311 B3 35 | 50.166 | 0.087 | 30.216 | 0.209 | 14.625 | 0.027 | 0.602 | 3.411 | 0.114 | 0.000 | 0.038 | 99.495 | 69.865 |
|  | BH311 B3 36 | 50.565 | 0.076 | 29.819 | 0.173 | 14.000 | 0.000 | 0.725 | 3.729 | 0.169 | 0.025 | 0.056 | 99.337 | 66.800 |

Appendix E4 - LPS Plagioclase compositions

| Zone | Sample code | SiO2 | TiO2 | Al2O3 | MgO | CaO | MnO | FeO | Na2O | K2O | BaO | SrO | Total | Anmol\% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | BH311 B3 34-36 av | 50.315 | 0.074 | 30.054 | 0.194 | 14.296 | 0.028 | 0.638 | 3.546 | 0.134 | 0.025 | 0.041 | 99.346 | 68.463 |
|  | BH311 B3 41 | 48.452 | 0.027 | 30.565 | 0.209 | 15.202 | 0.000 | 0.655 | 2.961 | 0.108 | 0.005 | 0.000 | 98.184 | 73.473 |
|  | BH311 B3 42 | 49.579 | 0.109 | 30.498 | 0.199 | 14.942 | 0.000 | 0.633 | 3.105 | 0.129 | 0.011 | 0.000 | 99.205 | 72.120 |
|  | BH311 B3 43 | 50.083 | 0.022 | 30.251 | 0.172 | 14.353 | 0.012 | 0.783 | 3.476 | 0.200 | 0.000 | 0.035 | 99.387 | 68.737 |
|  | BH311 B3 41-43 av | 49.371 | 0.053 | 30.438 | 0.193 | 14.832 | 0.004 | 0.690 | 3.181 | 0.146 | 0.005 | 0.012 | 98.925 | 71.443 |
|  | BH311 B3 44 | 50.161 | 0.043 | 30.427 | 0.130 | 14.500 | 0.000 | 0.850 | 3.323 | 0.203 | 0.002 | 0.080 | 99.719 | 69.861 |
|  | BH311 B3 44 | 44.634 | 0.071 | 27.581 | 1.402 | 18.125 | 0.279 | 1.601 | 2.667 | 0.060 | 0.007 | 0.049 | 96.476 | 78.718 |
|  | BH311 B3 44 | 49.957 | 0.000 | 30.551 | 0.238 | 14.662 | 0.082 | 0.599 | 3.257 | 0.131 | 0.020 | 0.080 | 99.577 | 70.766 |
|  | BH311 B3 44 | 50.239 | 0.000 | 30.320 | 0.252 | 14.645 | 0.052 | 0.543 | 3.335 | 0.146 | 0.000 | 0.073 | 99.605 | 70.227 |
|  | BH311 B3 44 | 49.690 | 0.000 | 30.519 | 0.271 | 14.818 | 0.009 | 0.561 | 3.151 | 0.122 | 0.000 | 0.105 | 99.246 | 71.705 |
|  | BH311 B3 44 | 49.080 | 0.038 | 30.490 | 0.402 | 14.850 | 0.000 | 0.682 | 3.133 | 0.114 | 0.000 | 0.045 | 98.834 | 71.895 |
|  | BH311 B3 44 | 44.358 | 0.037 | 26.349 | 2.669 | 10.981 | 0.314 | 9.214 | 2.332 | 0.097 | 0.000 | 0.106 | 96.457 | 71.694 |
|  | BH311 B3 44 | 46.609 | 0.059 | 28.369 | 1.689 | 12.418 | 0.319 | 5.787 | 2.879 | 0.128 | 0.004 | 0.035 | 98.296 | 69.836 |
|  | BH311 B3 44 | 50.280 | 0.000 | 30.664 | 0.324 | 14.972 | 0.000 | 0.633 | 3.131 | 0.124 | 0.000 | 0.010 | 100.138 | 72.031 |
|  | BH311 B3 44 | 46.985 | 0.032 | 28.366 | 4.385 | 12.424 | 0.082 | 2.888 | 2.625 | 0.114 | 0.000 | 0.032 | 97.933 | 71.774 |
|  | BH311 B3 44 | 49.491 | 0.016 | 30.676 | 0.230 | 14.919 | 0.000 | 0.624 | 3.204 | 0.125 | 0.018 | 0.024 | 99.327 | 71.478 |
|  | BH311 B3 44 | 44.942 | 0.124 | 26.448 | 5.831 | 11.232 | 0.386 | 4.755 | 2.191 | 0.137 | 0.011 | 0.021 | 96.078 | 73.107 |
|  | BH311 B3 44 | 48.722 | 0.103 | 30.770 | 0.223 | 14.974 | 0.024 | 0.717 | 3.187 | 0.136 | 0.014 | 0.147 | 99.017 | 71.618 |
|  | BH311 B3 44 | 50.018 | 0.000 | 30.602 | 0.218 | 14.825 | 0.055 | 0.776 | 3.161 | 0.128 | 0.000 | 0.045 | 99.828 | 71.627 |
|  | BH311 B3 44 | 50.114 | 0.049 | 30.536 | 0.191 | 14.633 | 0.000 | 0.649 | 3.315 | 0.148 | 0.000 | 0.035 | 99.670 | 70.324 |
|  | BH311 B3 44 | 49.733 | 0.000 | 29.926 | 1.035 | 14.127 | 0.055 | 0.833 | 3.355 | 0.155 | 0.000 | 0.049 | 99.268 | 69.309 |
|  | BH311 B3 44 | 47.933 | 0.092 | 27.599 | 4.018 | 12.094 | 0.012 | 3.219 | 2.908 | 0.211 | 0.000 | 0.018 | 98.104 | 68.687 |
|  | BH311 B3 44 | 50.157 | 0.027 | 30.308 | 0.211 | 14.238 | 0.000 | 0.726 | 3.514 | 0.161 | 0.033 | 0.035 | 99.410 | 68.450 |
|  | BH311 B3 44 | 50.857 | 0.130 | 29.873 | 0.288 | 14.065 | 0.000 | 0.853 | 3.615 | 0.168 | 0.000 | 0.108 | 99.957 | 67.599 |
|  | BH311 B3 44 | 51.018 | 0.157 | 30.258 | 0.160 | 14.130 | 0.036 | 0.833 | 3.599 | 0.188 | 0.000 | 0.077 | 100.456 | 67.716 |
|  | BH311 B3 44 av | 48.749 | 0.049 | 29.532 | 1.208 | 14.082 | 0.085 | 1.867 | 3.094 | 0.140 | 0.005 | 0.059 | 98.870 | 70.921 |
|  | BH311 B3 53 | 52.352 | 0.103 | 29.940 | 0.172 | 13.494 | 0.015 | 0.868 | 4.035 | 0.218 | 0.000 | 0.122 | 101.319 | 64.089 |
|  | BH311 B3 53 | 52.539 | 0.065 | 29.689 | 0.190 | 13.225 | 0.000 | 0.673 | 4.028 | 0.224 | 0.000 | 0.076 | 100.709 | 63.641 |
|  | BH311 B3 53 | 52.802 | 0.071 | 29.630 | 0.172 | 13.281 | 0.000 | 0.694 | 4.018 | 0.208 | 0.000 | 0.069 | 100.945 | 63.853 |
|  | BH311 B3 53 | 52.524 | 0.060 | 29.892 | 0.171 | 13.644 | 0.015 | 0.671 | 3.868 | 0.192 | 0.033 | 0.035 | 101.105 | 65.332 |
|  | BH311 B3 53 | 51.885 | 0.136 | 29.800 | 0.163 | 13.579 | 0.043 | 0.660 | 3.908 | 0.184 | 0.005 | 0.101 | 100.464 | 65.060 |
|  | BH311 B3 53 | 52.052 | 0.054 | 30.042 | 0.145 | 13.611 | 0.000 | 0.689 | 3.943 | 0.199 | 0.027 | 0.077 | 100.839 | 64.836 |

Appendix E4 - LPS Plagioclase compositions

| Zone | Sample code | SiO2 | TiO2 | Al2O3 | MgO | CaO | MnO | FeO | Na2O | K2O | BaO | SrO | Total | Anmol\% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | BH311 B3 53 | 52.132 | 0.119 | 29.741 | 0.160 | 13.513 | 0.000 | 0.735 | 3.993 | 0.221 | 0.000 | 0.136 | 100.750 | 64.342 |
|  | BH311 B3 53 | 52.226 | 0.038 | 29.500 | 0.138 | 13.147 | 0.000 | 0.737 | 4.155 | 0.212 | 0.000 | 0.101 | 100.254 | 62.850 |
|  | BH311 B3 53 | 52.335 | 0.141 | 29.476 | 0.127 | 12.862 | 0.094 | 0.769 | 4.209 | 0.261 | 0.005 | 0.097 | 100.376 | 61.863 |
|  | BH311 B3 53 | 53.210 | 0.087 | 28.865 | 0.095 | 11.988 | 0.009 | 0.712 | 4.700 | 0.352 | 0.000 | 0.090 | 100.108 | 57.326 |
|  | BH311 B3 53 av | 52.406 | 0.087 | 29.658 | 0.153 | 13.234 | 0.018 | 0.721 | 4.086 | 0.227 | 0.007 | 0.090 | 100.687 | 63.319 |
| CPZ | BH311 B4 6 | 50.365 | 0.081 | 31.131 | 0.188 | 14.650 | 0.000 | 0.815 | 3.459 | 0.148 | 0.000 | 0.042 | 100.879 | 69.479 |
|  | BH311 B4 7 | 50.652 | 0.016 | 30.624 | 0.230 | 14.757 | 0.000 | 0.686 | 3.295 | 0.155 | 0.000 | 0.128 | 100.543 | 70.594 |
|  | BH311 B4 6-7 av | 50.509 | 0.049 | 30.878 | 0.209 | 14.704 | 0.000 | 0.751 | 3.377 | 0.152 | 0.000 | 0.085 | 100.711 | 70.037 |
|  | BH311 B4 8 | 51.277 | 0.043 | 30.701 | 0.169 | 14.683 | 0.054 | 0.758 | 3.286 | 0.141 | 0.016 | 0.083 | 101.211 | 70.581 |
|  | BH311 B4 9 | 49.979 | 0.043 | 31.061 | 0.196 | 15.045 | 0.000 | 0.605 | 3.137 | 0.126 | 0.007 | 0.024 | 100.223 | 72.075 |
|  | BH311 B4 10 | 50.365 | 0.092 | 31.157 | 0.220 | 14.988 | 0.000 | 0.718 | 3.228 | 0.135 | 0.052 | 0.063 | 101.018 | 71.341 |
|  | BH311 B4 8-10 av | 50.540 | 0.059 | 30.973 | 0.195 | 14.905 | 0.018 | 0.694 | 3.217 | 0.134 | 0.025 | 0.057 | 100.817 | 71.332 |
|  | BH311 B4 21 | 51.797 | 0.087 | 29.651 | 0.109 | 13.266 | 0.003 | 0.746 | 4.094 | 0.228 | 0.000 | 0.076 | 100.057 | 63.335 |
|  | BH311 B4 21 | 51.921 | 0.000 | 29.647 | 0.132 | 13.336 | 0.000 | 0.794 | 3.949 | 0.231 | 0.011 | 0.059 | 100.080 | 64.236 |
|  | BH311 B4 21 | 51.956 | 0.060 | 29.705 | 0.133 | 13.388 | 0.003 | 0.809 | 4.090 | 0.221 | 0.000 | 0.007 | 100.372 | 63.594 |
|  | BH311 B4 21 | 57.107 | 0.065 | 27.261 | 0.040 | 9.774 | 0.064 | 0.508 | 5.937 | 0.378 | 0.047 | 0.062 | 101.243 | 46.577 |
|  | BH311 B4 21 av | 53.195 | 0.053 | 29.066 | 0.104 | 12.441 | 0.018 | 0.714 | 4.518 | 0.265 | 0.015 | 0.051 | 100.438 | 59.435 |
|  | BH311 B4 27 | 51.790 | 0.022 | 30.200 | 0.156 | 13.874 | 0.048 | 0.785 | 3.719 | 0.163 | 0.000 | 0.080 | 100.837 | 66.709 |
|  | BH311 B4 28 | 51.932 | 0.016 | 30.441 | 0.155 | 14.139 | 0.000 | 0.735 | 3.602 | 0.176 | 0.002 | 0.000 | 101.198 | 67.757 |
|  | BH311 B4 29 | 54.161 | 0.032 | 28.989 | 0.108 | 12.482 | 0.000 | 0.774 | 4.469 | 0.265 | 0.000 | 0.166 | 101.446 | 59.767 |
|  | BH311 B4 27-29 av | 52.628 | 0.023 | 29.877 | 0.140 | 13.498 | 0.016 | 0.765 | 3.930 | 0.201 | 0.001 | 0.082 | 101.160 | 64.744 |
|  | BH311 B4 30 | 50.537 | 0.060 | 30.986 | 0.186 | 14.930 | 0.042 | 0.675 | 3.269 | 0.129 | 0.013 | 0.090 | 100.917 | 71.082 |
|  | BH311 B4 31 | 51.298 | 0.016 | 29.309 | 0.708 | 13.168 | 0.042 | 1.823 | 3.461 | 0.267 | 0.000 | 0.121 | 100.213 | 66.678 |
|  | BH311 B4 32 | 53.800 | 0.124 | 28.989 | 0.090 | 12.070 | 0.067 | 0.779 | 4.691 | 0.300 | 0.000 | 0.052 | 100.962 | 57.707 |
|  | BH311 B4 30-32 av | 51.878 | 0.067 | 29.761 | 0.328 | 13.389 | 0.050 | 1.092 | 3.807 | 0.232 | 0.004 | 0.088 | 100.697 | 65.156 |
| CPZ | 10BH167 A5 11 | 51.539 | 0.011 | 29.111 | 0.104 | 12.915 | 0.106 | 0.773 | 4.231 | 0.207 | 0.000 | 0.059 | 99.056 | 62.039 |
|  | 10BH167 A5 12 | 50.635 | 0.000 | 29.524 | 0.103 | 13.361 | 0.032 | 0.804 | 3.980 | 0.191 | 0.023 | 0.017 | 98.670 | 64.239 |
|  | 10BH167 A5 13 | 50.249 | 0.000 | 29.700 | 0.120 | 13.313 | 0.000 | 0.921 | 4.029 | 0.193 | 0.014 | 0.073 | 98.612 | 63.886 |
|  | 10BH167 A5 14 | 54.892 | 0.011 | 26.702 | 0.030 | 9.409 | 0.026 | 0.598 | 6.154 | 0.393 | 0.002 | 0.041 | 98.258 | 44.776 |
|  | 10BH167 A5 15 | 98.067 | 0.000 | 0.074 | 0.013 | 0.036 | 0.006 | 0.258 | 0.002 | 0.008 | 0.000 | 0.000 | 98.464 | 73.254 |
|  | $\begin{aligned} & \text { 10BH167 A5 11-15 } \\ & \text { av } \end{aligned}$ | 61.076 | 0.004 | 23.022 | 0.074 | 9.807 | 0.034 | 0.671 | 3.679 | 0.198 | 0.008 | 0.038 | 98.612 | 58.711 |

Appendix E4 - LPS Plagioclase compositions

| Zone | Sample code | SiO2 | TiO2 | Al2O3 | MgO | CaO | MnO | FeO | Na2O | K2O | BaO | SrO | Total | Anmol\% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 10BH167 A5 16 | 48.337 | 0.022 | 30.850 | 0.201 | 14.978 | 0.000 | 0.644 | 3.093 | 0.111 | 0.013 | 0.080 | 98.329 | 72.316 |
|  | 10BH167 A5 17 | 49.522 | 0.097 | 30.910 | 0.196 | 14.751 | 0.032 | 0.662 | 3.316 | 0.125 | 0.000 | 0.000 | 99.611 | 70.578 |
|  | 10BH167 A5 18 | 49.194 | 0.097 | 30.548 | 0.206 | 14.856 | 0.006 | 0.714 | 3.131 | 0.108 | 0.007 | 0.066 | 98.933 | 71.932 |
|  | 10BH167 A5 19 | 49.307 | 0.000 | 30.531 | 0.188 | 14.701 | 0.000 | 0.657 | 3.259 | 0.138 | 0.000 | 0.010 | 98.791 | 70.805 |
|  | 10BH167 A5 20 | 48.551 | 0.081 | 30.323 | 0.178 | 14.398 | 0.047 | 0.708 | 3.445 | 0.135 | 0.000 | 0.024 | 97.890 | 69.246 |
|  | 10BH167 A5 21 | 49.350 | 0.027 | 30.151 | 0.129 | 14.087 | 0.000 | 0.668 | 3.746 | 0.135 | 0.007 | 0.132 | 98.432 | 66.989 |
|  | 10BH167 A5 22 | 50.472 | 0.011 | 30.173 | 0.150 | 14.001 | 0.000 | 0.800 | 3.606 | 0.150 | 0.000 | 0.076 | 99.439 | 67.622 |
|  | $\begin{aligned} & \text { 10BH167 A5 26-22 } \\ & \text { av } \end{aligned}$ | 49.248 | 0.048 | 30.498 | 0.178 | 14.539 | 0.012 | 0.693 | 3.371 | 0.129 | 0.004 | 0.055 | 98.775 | 69.920 |
|  | 10BH167 A5 34 | 48.740 | 0.070 | 30.703 | 0.172 | 14.798 | 0.000 | 0.676 | 3.213 | 0.112 | 0.000 | 0.059 | 98.543 | 71.331 |
|  | 10BH167 A5 35 | 49.099 | 0.059 | 30.512 | 0.153 | 14.507 | 0.015 | 0.719 | 3.317 | 0.127 | 0.000 | 0.042 | 98.550 | 70.216 |
|  | 10BH167 A5 36 | 49.612 | 0.054 | 30.174 | 0.152 | 14.249 | 0.000 | 0.775 | 3.483 | 0.126 | 0.000 | 0.132 | 98.757 | 68.830 |
|  | 10BH167 A5 37 | 50.164 | 0.032 | 29.521 | 0.139 | 13.370 | 0.026 | 0.840 | 3.851 | 0.198 | 0.000 | 0.090 | 98.231 | 64.984 |
|  | 10BH167 A5 38 | 49.630 | 0.005 | 29.533 | 0.116 | 13.203 | 0.000 | 0.901 | 3.996 | 0.203 | 0.000 | 0.083 | 97.670 | 63.858 |
|  | 10BH167 A5 39 | 49.998 | 0.081 | 28.947 | 0.093 | 12.694 | 0.000 | 0.759 | 4.229 | 0.204 | 0.005 | 0.031 | 97.041 | 61.647 |
|  | 10BH167 A5 40 | 50.930 | 0.102 | 28.639 | 0.084 | 12.037 | 0.000 | 0.765 | 4.628 | 0.255 | 0.000 | 0.073 | 97.513 | 58.107 |
|  | 10BH167 A5 41 | 52.928 | 0.059 | 27.960 | 0.048 | 11.120 | 0.000 | 0.888 | 5.120 | 0.288 | 0.000 | 0.042 | 98.453 | 53.648 |
|  | $\begin{aligned} & \text { 10BH167 A5 34-41 } \\ & \text { av } \end{aligned}$ | 50.138 | 0.058 | 29.499 | 0.120 | 13.247 | 0.005 | 0.790 | 3.980 | 0.189 | 0.001 | 0.069 | 98.095 | 64.077 |
|  | 10BH167 A5 71 | 49.039 | 0.070 | 31.381 | 0.183 | 15.094 | 0.000 | 0.743 | 3.004 | 0.134 | 0.000 | 0.070 | 99.718 | 72.955 |
|  | 10BH167 A5 72 | 48.838 | 0.022 | 31.164 | 0.136 | 14.965 | 0.000 | 0.821 | 3.129 | 0.131 | 0.000 | 0.052 | 99.258 | 72.006 |
|  | 10BH167 A5 73 | 51.080 | 0.054 | 29.505 | 0.116 | 13.058 | 0.000 | 0.810 | 4.181 | 0.199 | 0.013 | 0.014 | 99.030 | 62.582 |
|  | 10BH167 A5 74 | 56.958 | 0.081 | 26.337 | 0.015 | 8.632 | 0.044 | 0.562 | 6.630 | 0.438 | 0.023 | 0.172 | 99.892 | 40.795 |
|  | $\begin{aligned} & \text { 10BH167 A5 71-74 } \\ & \text { av } \end{aligned}$ | 51.479 | 0.057 | 29.597 | 0.113 | 12.937 | 0.011 | 0.734 | 4.236 | 0.226 | 0.009 | 0.077 | 99.475 | 61.977 |
|  | 10BH167 A5 84 | 48.324 | 0.022 | 31.788 | 0.208 | 15.958 | 0.000 | 0.587 | 2.670 | 0.096 | 0.000 | 0.056 | 99.709 | 76.340 |
|  | 10BH167 A5 85 | 46.703 | 0.086 | 31.733 | 0.200 | 15.580 | 0.015 | 0.622 | 2.708 | 0.102 | 0.000 | 0.105 | 97.854 | 75.625 |
|  | 10BH167 A5 86 | 47.501 | 0.005 | 31.694 | 0.200 | 15.669 | 0.000 | 0.636 | 2.789 | 0.082 | 0.025 | 0.052 | 98.653 | 75.250 |
|  | 10BH167 A5 87 | 47.179 | 0.059 | 31.190 | 0.161 | 15.156 | 0.000 | 0.752 | 3.019 | 0.113 | 0.000 | 0.042 | 97.671 | 73.028 |
|  | 10BH167 A5 88 | 47.441 | 0.000 | 30.924 | 0.154 | 14.906 | 0.000 | 0.986 | 3.219 | 0.127 | 0.000 | 0.031 | 97.788 | 71.381 |
|  | $\begin{aligned} & \text { 10BH167 A5 84-88 } \\ & \text { av } \end{aligned}$ | 47.430 | 0.034 | 31.466 | 0.185 | 15.454 | 0.003 | 0.717 | 2.881 | 0.104 | 0.005 | 0.057 | 98.335 | 74.323 |
|  | 10BH167 A5 90 | 49.573 | 0.000 | 31.342 | 0.327 | 15.109 | 0.044 | 0.755 | 2.773 | 0.188 | 0.000 | 0.000 | 100.111 | 74.243 |
|  | 10BH167 A5 91 | 50.088 | 0.000 | 30.628 | 0.157 | 14.469 | 0.018 | 0.783 | 3.432 | 0.140 | 0.000 | 0.000 | 99.715 | 69.409 |

Appendix E4 - LPS Plagioclase compositions

| Zone | Sample code | SiO2 | TiO2 | Al2O3 | MgO | CaO | MnO | FeO | Na2O | K2O | BaO | SrO | Total | Anmol\% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 10BH167 A5 92 | 59.567 | 0.000 | 24.508 | 0.009 | 6.448 | 0.000 | 0.321 | 7.801 | 0.585 | 0.070 | 0.058 | 99.367 | 30.292 |
|  | $\begin{aligned} & \text { 10BH167 A5 90-92 } \\ & \text { av } \end{aligned}$ | 53.076 | 0.000 | 28.826 | 0.164 | 12.009 | 0.021 | 0.620 | 4.669 | 0.304 | 0.023 | 0.019 | 99.731 | 57.657 |
| DZ | 10BH167 A6 12 | 46.672 | 0.005 | 31.563 | 0.211 | 15.482 | 0.000 | 0.613 | 2.735 | 0.106 | 0.000 | 0.024 | 97.411 | 75.311 |
|  | 10BH167 A6 13 | 48.142 | 0.000 | 30.901 | 0.272 | 14.456 | 0.000 | 0.732 | 3.024 | 0.173 | 0.023 | 0.000 | 97.723 | 71.769 |
|  | 10BH167 A6 14 | 49.179 | 0.011 | 30.338 | 0.162 | 14.354 | 0.070 | 0.709 | 3.421 | 0.144 | 0.000 | 0.021 | 98.409 | 69.290 |
|  | 10BH167 A6 15 | 49.598 | 0.022 | 30.085 | 0.158 | 13.997 | 0.000 | 0.878 | 3.617 | 0.157 | 0.000 | 0.070 | 98.582 | 67.523 |
|  | $\begin{aligned} & \text { 10BH167 A6 12-15 } \\ & \text { av } \end{aligned}$ | 48.398 | 0.010 | 30.722 | 0.201 | 14.572 | 0.018 | 0.733 | 3.199 | 0.145 | 0.006 | 0.029 | 98.031 | 70.959 |
|  | 10BH167 A6 32 | 49.692 | 0.065 | 30.040 | 0.138 | 13.915 | 0.088 | 0.865 | 3.659 | 0.153 | 0.000 | 0.080 | 98.695 | 67.163 |
|  | 10BH167 A6 33 | 49.701 | 0.000 | 29.810 | 0.122 | 13.856 | 0.000 | 0.836 | 3.650 | 0.162 | 0.000 | 0.014 | 98.151 | 67.087 |
|  | 10BH167 A6 34 | 48.706 | 0.065 | 30.085 | 0.143 | 13.945 | 0.023 | 0.931 | 3.664 | 0.165 | 0.063 | 0.031 | 97.821 | 67.060 |
|  | $\begin{aligned} & \text { 10BH167 A6 32-34 } \\ & \text { av } \end{aligned}$ | 49.366 | 0.043 | 29.978 | 0.134 | 13.905 | 0.037 | 0.877 | 3.658 | 0.160 | 0.021 | 0.042 | 98.222 | 67.103 |
|  | 10BH167 A6 35 | 47.872 | 0.022 | 30.893 | 0.228 | 15.148 | 0.015 | 0.574 | 2.866 | 0.119 | 0.000 | 0.070 | 97.807 | 73.980 |
|  | 10BH167 A6 36 | 48.132 | 0.016 | 31.063 | 0.230 | 15.194 | 0.026 | 0.689 | 2.964 | 0.106 | 0.000 | 0.105 | 98.525 | 73.459 |
|  | 10BH167 A6 37 | 48.149 | 0.049 | 31.150 | 0.205 | 15.074 | 0.000 | 0.735 | 2.957 | 0.115 | 0.000 | 0.000 | 98.434 | 73.311 |
|  | 10BH167 A6 38 | 47.152 | 0.081 | 30.526 | 0.296 | 14.865 | 0.021 | 0.824 | 3.073 | 0.120 | 0.000 | 0.052 | 97.010 | 72.270 |
|  | $\begin{aligned} & \text { 10BH167 A6 35-38 } \\ & \text { av } \end{aligned}$ | 47.826 | 0.042 | 30.908 | 0.240 | 15.070 | 0.016 | 0.706 | 2.965 | 0.115 | 0.000 | 0.057 | 97.944 | 73.254 |
|  | 10BH167 A6 39 | 50.550 | 0.124 | 29.403 | 0.144 | 13.424 | 0.026 | 0.824 | 3.827 | 0.182 | 0.000 | 0.049 | 98.553 | 65.273 |
|  | 10BH167 A6 40 | 50.153 | 0.048 | 29.634 | 0.129 | 13.595 | 0.000 | 0.819 | 3.697 | 0.165 | 0.045 | 0.080 | 98.365 | 66.324 |
|  | $\begin{aligned} & \text { 10BH167 A6 39-40 } \\ & \text { av } \end{aligned}$ | 50.352 | 0.086 | 29.519 | 0.137 | 13.510 | 0.013 | 0.822 | 3.762 | 0.174 | 0.023 | 0.065 | 98.459 | 65.798 |
|  | 10BH167 A6 44 | 53.108 | 0.113 | 28.552 | 0.088 | 12.051 | 0.000 | 0.687 | 4.654 | 0.230 | 0.000 | 0.076 | 99.559 | 58.087 |
|  | 10BH167 A6 45 | 54.593 | 0.038 | 28.167 | 0.070 | 11.311 | 0.009 | 0.603 | 5.016 | 0.285 | 0.000 | 0.176 | 100.268 | 54.571 |
|  | 10BH167 A6 46 | 55.847 | 0.000 | 26.694 | 0.021 | 9.055 | 0.082 | 0.539 | 6.380 | 0.380 | 0.061 | 0.124 | 99.183 | 42.966 |
|  | $\begin{aligned} & \text { 10BH167 A6 44-46 } \\ & \text { av } \end{aligned}$ | 54.516 | 0.050 | 27.804 | 0.060 | 10.806 | 0.030 | 0.610 | 5.350 | 0.298 | 0.020 | 0.125 | 99.670 | 51.827 |
| DZ | 10BH167 A8 10 | 52.432 | 0.086 | 29.548 | 0.140 | 13.260 | 0.000 | 1.024 | 4.085 | 0.190 | 0.005 | 0.062 | 100.832 | 63.505 |
|  | 10BH167 A8 11 | 50.622 | 0.075 | 30.161 | 0.158 | 14.304 | 0.000 | 0.912 | 3.632 | 0.149 | 0.043 | 0.035 | 100.091 | 67.890 |
|  | 10BH167 A8 12 | 53.479 | 0.027 | 28.784 | 0.105 | 11.666 | 0.000 | 0.867 | 4.727 | 0.406 | 0.000 | 0.000 | 100.061 | 56.349 |
|  | 10BH167 A8 13 | 53.638 | 0.091 | 28.522 | 0.069 | 11.774 | 0.018 | 0.870 | 4.934 | 0.244 | 0.000 | 0.080 | 100.240 | 56.086 |
|  | $\begin{aligned} & \text { 10BH167 A8 10-13 } \\ & \text { av } \end{aligned}$ | 52.543 | 0.070 | 29.254 | 0.118 | 12.751 | 0.005 | 0.918 | 4.345 | 0.247 | 0.012 | 0.044 | 100.306 | 60.976 |

Appendix E4 - LPS Plagioclase compositions

| Zone | Sample code | SiO2 | TiO2 | Al2O3 | MgO | CaO | MnO | FeO | Na2O | K2O | BaO | SrO | Total | Anmol\% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| DZ | 10BH167 A9 1 | 51.604 | 0.054 | 29.529 | 0.124 | 13.239 | 0.032 | 0.846 | 4.088 | 0.201 | 0.000 | 0.052 | 99.769 | 63.418 |
|  | 10BH167 A9 2 | 51.060 | 0.016 | 29.538 | 0.124 | 13.364 | 0.000 | 0.817 | 3.934 | 0.205 | 0.002 | 0.000 | 99.060 | 64.474 |
|  | 10BH167 A9 3 | 50.714 | 0.140 | 29.469 | 0.125 | 13.148 | 0.047 | 0.818 | 4.116 | 0.201 | 0.000 | 0.101 | 98.879 | 63.104 |
|  | 10BH167 A9 4 | 55.391 | 0.054 | 27.344 | 0.018 | 10.117 | 0.000 | 0.542 | 5.755 | 0.336 | 0.043 | 0.041 | 99.641 | 48.299 |
|  | 10BH167 A9 1-4 av | 52.192 | 0.066 | 28.970 | 0.098 | 12.467 | 0.020 | 0.756 | 4.473 | 0.236 | 0.011 | 0.049 | 99.337 | 59.804 |
|  | 10BH167 A9 16 | 49.194 | 0.097 | 29.348 | 0.113 | 12.893 | 0.000 | 0.915 | 4.168 | 0.275 | 0.000 | 0.063 | 97.066 | 62.097 |
|  | 10BH167 A9 17 | 50.116 | 0.000 | 29.567 | 0.128 | 13.256 | 0.029 | 0.861 | 4.074 | 0.201 | 0.000 | 0.059 | 98.291 | 63.525 |
|  | $\begin{aligned} & \text { 10BH167 A9 16-17 } \\ & \text { av } \end{aligned}$ | 49.655 | 0.049 | 29.458 | 0.121 | 13.075 | 0.015 | 0.888 | 4.121 | 0.238 | 0.000 | 0.061 | 97.679 | 62.813 |
| DZ | BH311 B6 61 | 48.888 | 0.043 | 31.706 | 0.220 | 16.226 | 0.070 | 0.546 | 2.461 | 0.088 | 0.063 | 0.049 | 100.360 | 77.983 |
|  | BH311 B6 61 | 49.335 | 0.071 | 31.500 | 0.242 | 16.062 | 0.000 | 0.559 | 2.431 | 0.080 | 0.000 | 0.028 | 100.308 | 78.137 |
|  | BH311 B6 61 | 49.270 | 0.038 | 31.345 | 0.234 | 15.791 | 0.009 | 0.554 | 2.605 | 0.098 | 0.000 | 0.035 | 99.979 | 76.575 |
|  | BH311 B6 61 | 49.537 | 0.033 | 31.302 | 0.219 | 15.520 | 0.000 | 0.556 | 2.733 | 0.092 | 0.000 | 0.066 | 100.058 | 75.431 |
|  | BH311 B6 61 | 51.682 | 0.011 | 29.555 | 0.209 | 13.350 | 0.012 | 0.657 | 3.670 | 0.342 | 0.027 | 0.038 | 99.553 | 65.415 |
|  | BH311 B6 61 av | 49.742 | 0.039 | 31.082 | 0.225 | 15.390 | 0.018 | 0.574 | 2.780 | 0.140 | 0.018 | 0.043 | 100.052 | 74.708 |
|  | BH311 B6 70 | 49.980 | 0.000 | 31.386 | 0.171 | 15.349 | 0.033 | 0.753 | 2.888 | 0.116 | 0.000 | 0.118 | 100.794 | 74.103 |
|  | BH311 B6 70 | 51.314 | 0.011 | 30.631 | 0.222 | 14.665 | 0.079 | 0.663 | 3.391 | 0.138 | 0.000 | 0.066 | 101.180 | 69.948 |
|  | BH311 B6 70 | 52.898 | 0.087 | 29.378 | 0.143 | 13.181 | 0.000 | 0.783 | 4.089 | 0.211 | 0.000 | 0.010 | 100.780 | 63.274 |
|  | BH311 B6 70 | 53.849 | 0.103 | 28.866 | 0.122 | 12.312 | 0.058 | 0.855 | 4.605 | 0.292 | 0.000 | 0.097 | 101.159 | 58.649 |
|  | BH311 B6 70 av | 52.010 | 0.050 | 30.065 | 0.165 | 13.877 | 0.043 | 0.764 | 3.743 | 0.189 | 0.000 | 0.073 | 100.978 | 66.494 |
|  | BH311 B6 77 | 50.236 | 0.027 | 29.906 | 0.170 | 14.297 | 0.000 | 0.778 | 3.425 | 0.129 | 0.000 | 0.087 | 99.055 | 69.240 |
|  | BH311 B6 78 | 51.392 | 0.022 | 29.470 | 0.167 | 13.980 | 0.006 | 0.746 | 3.647 | 0.151 | 0.027 | 0.014 | 99.622 | 67.311 |
|  | BH311 B6 77-78 av | 50.814 | 0.025 | 29.688 | 0.169 | 14.139 | 0.003 | 0.762 | 3.536 | 0.140 | 0.014 | 0.051 | 99.339 | 68.276 |
| DZ | 10BH167 A10 1 | 52.773 | 0.011 | 29.263 | 0.107 | 13.221 | 0.026 | 0.843 | 4.145 | 0.204 | 0.000 | 0.035 | 100.628 | 63.064 |
|  | 10BH167 A10 2 | 51.973 | 0.000 | 29.674 | 0.149 | 13.319 | 0.000 | 0.845 | 3.972 | 0.199 | 0.000 | 0.031 | 100.162 | 64.208 |
|  | 10BH167 A10 3 | 57.179 | 0.005 | 26.792 | 0.022 | 9.487 | 0.000 | 0.500 | 6.126 | 0.388 | 0.000 | 0.062 | 100.561 | 45.102 |
|  | $\begin{aligned} & \text { 10BH167 A10 1-3 } \\ & \text { av } \end{aligned}$ | 53.975 | 0.005 | 28.576 | 0.093 | 12.009 | 0.009 | 0.729 | 4.748 | 0.264 | 0.000 | 0.043 | 100.450 | 57.421 |
|  | 10BH167 A10 4 | 51.961 | 0.022 | 29.425 | 0.143 | 13.204 | 0.012 | 0.836 | 4.130 | 0.199 | 0.000 | 0.007 | 99.939 | 63.133 |
|  | 10BH167 A10 5 | 52.048 | 0.011 | 29.794 | 0.138 | 13.630 | 0.050 | 0.795 | 3.896 | 0.175 | 0.000 | 0.000 | 100.537 | 65.252 |
|  | 10BH167 A10 6 | 51.187 | 0.086 | 30.110 | 0.138 | 13.950 | 0.000 | 0.809 | 3.808 | 0.172 | 0.000 | 0.111 | 100.371 | 66.285 |
|  | $\begin{aligned} & \text { 10BH167 A10 4-6 } \\ & \text { av } \end{aligned}$ | 51.732 | 0.040 | 29.776 | 0.140 | 13.595 | 0.021 | 0.813 | 3.945 | 0.182 | 0.000 | 0.039 | 100.282 | 64.893 |

Appendix E4 - LPS Plagioclase compositions

| Zone | Sample code | SiO2 | TiO2 | Al2O3 | MgO | CaO | MnO | FeO | Na2O | K2O | BaO | SrO | Total | Anmol\% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 10BH167 A10 23 | 52.731 | 0.005 | 29.026 | 0.131 | 11.670 | 0.000 | 0.993 | 4.672 | 0.548 | 0.040 | 0.038 | 99.854 | 56.129 |
|  | 10BH167 A10 24 | 52.344 | 0.054 | 29.420 | 0.124 | 13.186 | 0.021 | 0.911 | 4.212 | 0.212 | 0.000 | 0.083 | 100.567 | 62.611 |
|  | 10BH167 A10 25 | 55.616 | 0.000 | 27.732 | 0.032 | 10.342 | 0.044 | 0.819 | 5.860 | 0.318 | 0.034 | 0.128 | 100.925 | 48.470 |
|  | $\begin{aligned} & \text { 10BH167 A10 23- } \\ & 25 \mathrm{av} \end{aligned}$ | 53.564 | 0.020 | 28.726 | 0.096 | 11.733 | 0.022 | 0.908 | 4.915 | 0.359 | 0.025 | 0.083 | 100.449 | 55.703 |
|  | 10BH167 A10 26 | 51.435 | 0.043 | 30.128 | 0.162 | 13.990 | 0.000 | 0.775 | 3.511 | 0.299 | 0.000 | 0.059 | 100.402 | 67.587 |
|  | 10BH167 A10 27 | 51.368 | 0.000 | 30.178 | 0.163 | 13.945 | 0.000 | 0.756 | 3.484 | 0.477 | 0.000 | 0.069 | 100.440 | 66.987 |
|  | 10BH167 A10 28 | 50.534 | 0.043 | 30.603 | 0.155 | 14.556 | 0.029 | 0.772 | 3.316 | 0.132 | 0.049 | 0.038 | 100.227 | 70.212 |
|  | 10BH167 A10 29 | 52.038 | 0.016 | 29.718 | 0.143 | 13.653 | 0.000 | 0.818 | 3.887 | 0.176 | 0.000 | 0.003 | 100.452 | 65.337 |
|  | 10BH167 A10 30 | 53.792 | 0.097 | 28.710 | 0.097 | 12.233 | 0.000 | 0.725 | 4.619 | 0.242 | 0.000 | 0.035 | 100.550 | 58.589 |
|  | $\begin{aligned} & \text { 10BH167 A10 26- } \\ & 30 \text { av } \end{aligned}$ | 51.833 | 0.040 | 29.867 | 0.144 | 13.675 | 0.006 | 0.769 | 3.763 | 0.265 | 0.010 | 0.041 | 100.414 | 65.732 |
| DZ | 10BH167 A11-1 1 | 49.580 | 0.059 | 30.513 | 0.186 | 14.321 | 0.047 | 0.679 | 3.501 | 0.144 | 0.000 | 0.031 | 99.061 | 68.759 |
|  | 10BH167 A11-1 2 | 50.147 | 0.048 | 30.026 | 0.158 | 13.796 | 0.044 | 0.766 | 3.724 | 0.188 | 0.023 | 0.028 | 98.948 | 66.432 |
|  | 10BH167 A11-1 3 | 54.111 | 0.000 | 27.345 | 0.030 | 10.140 | 0.026 | 0.727 | 5.732 | 0.368 | 0.005 | 0.087 | 98.571 | 48.395 |
|  | $\begin{aligned} & \text { 10BH167 A11-1 1- } \\ & 3 \mathrm{av} \end{aligned}$ | 51.279 | 0.036 | 29.295 | 0.125 | 12.752 | 0.039 | 0.724 | 4.319 | 0.233 | 0.009 | 0.049 | 98.860 | 61.165 |
|  | 10BH167 A11-1 15 | 49.778 | 0.070 | 30.090 | 0.147 | 13.583 | 0.000 | 0.752 | 3.882 | 0.172 | 0.000 | 0.021 | 98.495 | 65.264 |
|  | 10BH167 A11-1 16 | 50.439 | 0.059 | 29.772 | 0.132 | 13.307 | 0.012 | 0.783 | 4.009 | 0.215 | 0.000 | 0.017 | 98.745 | 63.922 |
|  | $\begin{aligned} & \text { 10BH167 A11-1 } \\ & 15-16 \text { av } \end{aligned}$ | 50.109 | 0.065 | 29.931 | 0.140 | 13.445 | 0.006 | 0.768 | 3.946 | 0.194 | 0.000 | 0.019 | 98.620 | 64.593 |
|  | 10BH167 A11-1 19 | 95.481 | 0.065 | 0.122 | 0.011 | 0.000 | 0.000 | 0.076 | 0.039 | 0.008 | 0.000 | 0.018 | 95.820 | 0.000 |
|  | 10BH167 A11-1 26 | 66.239 | 0.011 | 20.024 | 0.065 | 0.136 | 0.000 | 0.462 | 11.862 | 0.099 | 0.022 | 0.079 | 98.999 | 0.626 |
| DZ | 10BH167 A11-2 1 | 52.082 | 0.087 | 29.300 | 0.119 | 13.241 | 0.015 | 0.761 | 3.962 | 0.209 | 0.000 | 0.066 | 99.842 | 64.092 |
|  | 10BH167 A11-2 2 | 52.626 | 0.060 | 29.420 | 0.112 | 13.172 | 0.000 | 0.874 | 4.085 | 0.235 | 0.000 | 0.056 | 100.640 | 63.194 |
|  | 10BH167 A11-2 3 | 53.637 | 0.043 | 28.376 | 0.094 | 11.500 | 0.000 | 0.966 | 5.015 | 0.299 | 0.000 | 0.024 | 99.954 | 54.943 |
|  | $\begin{aligned} & \text { 10BH167 A11-2 1- } \\ & 3 \mathrm{av} \end{aligned}$ | 52.782 | 0.063 | 29.032 | 0.108 | 12.638 | 0.005 | 0.867 | 4.354 | 0.248 | 0.000 | 0.049 | 100.145 | 60.725 |
|  | 10BH167 A11-2 9 | 51.974 | 0.043 | 29.713 | 0.127 | 13.204 | 0.000 | 0.887 | 4.112 | 0.215 | 0.033 | 0.119 | 100.427 | 63.138 |
|  | 10BH167 A11-2 10 | 53.016 | 0.000 | 28.519 | 0.093 | 11.916 | 0.000 | 0.668 | 4.798 | 0.279 | 0.016 | 0.035 | 99.340 | 56.915 |
|  | $\begin{aligned} & \text { 10BH167 A11-2 9- } \\ & 10 \text { av } \end{aligned}$ | 52.495 | 0.022 | 29.116 | 0.110 | 12.560 | 0.000 | 0.778 | 4.455 | 0.247 | 0.025 | 0.077 | 99.884 | 60.025 |
|  | 10BH167 A11-2 13 | 80.034 | 0.000 | 10.958 | 0.414 | 3.876 | 0.042 | 0.973 | 3.119 | 0.147 | 0.024 | 0.016 | 99.603 | 39.943 |
|  | 10BH167 A11-2 14 | 53.397 | 0.103 | 28.689 | 0.071 | 11.760 | 0.000 | 0.919 | 4.934 | 0.283 | 0.020 | 0.056 | 100.232 | 55.913 |
| DZ | 10BH167 A12 7 | 57.379 | 0.065 | 26.288 | 0.036 | 8.680 | 0.029 | 0.456 | 6.641 | 0.328 | 0.000 | 0.000 | 99.902 | 41.161 |

Appendix E4 - LPS Plagioclase compositions

| Zone | Sample code | SiO2 | TiO2 | Al2O3 | MgO | CaO | MnO | FeO | Na2O | K2O | BaO | SrO | Total | Anmol\% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 10BH167 A12 8 | 54.994 | 0.113 | 28.220 | 0.066 | 11.193 | 0.000 | 0.591 | 5.155 | 0.356 | 0.000 | 0.145 | 100.833 | 53.440 |
|  | 10BH167 A12 9 | 54.404 | 0.081 | 28.359 | 0.084 | 11.417 | 0.000 | 0.547 | 5.119 | 0.299 | 0.018 | 0.045 | 100.373 | 54.256 |
|  | 10BH167 A12 10 | 53.312 | 0.016 | 28.855 | 0.093 | 12.218 | 0.000 | 0.622 | 4.657 | 0.257 | 0.000 | 0.066 | 100.096 | 58.317 |
|  | $\begin{aligned} & \text { 10BH167 A12 7-10 } \\ & \text { av } \end{aligned}$ | 55.022 | 0.069 | 27.931 | 0.070 | 10.877 | 0.007 | 0.554 | 5.393 | 0.310 | 0.005 | 0.064 | 100.301 | 51.779 |
|  | 10BH167 A12 14 | 49.869 | 0.005 | 30.423 | 0.167 | 14.142 | 0.000 | 0.696 | 3.599 | 0.160 | 0.000 | 0.077 | 99.138 | 67.843 |
|  | 10BH167 A12 15 | 50.333 | 0.065 | 29.881 | 0.136 | 13.718 | 0.035 | 0.784 | 3.836 | 0.206 | 0.022 | 0.077 | 99.093 | 65.596 |
|  | 10BH167 A12 16 | 52.320 | 0.059 | 29.398 | 0.118 | 12.829 | 0.000 | 0.690 | 4.360 | 0.250 | 0.007 | 0.073 | 100.104 | 61.036 |
|  | 10BH167 A12 17 | 56.179 | 0.000 | 27.224 | 0.041 | 9.888 | 0.076 | 0.610 | 5.954 | 0.449 | 0.036 | 0.055 | 100.512 | 46.620 |
|  | $\begin{aligned} & \text { 10BH167 A12 14- } \\ & 17 \text { av } \end{aligned}$ | 52.175 | 0.032 | 29.232 | 0.116 | 12.644 | 0.028 | 0.695 | 4.437 | 0.266 | 0.016 | 0.071 | 99.712 | 60.220 |
|  | 10BH167 A12 18 | 52.462 | 0.022 | 29.201 | 0.119 | 12.813 | 0.003 | 0.731 | 4.356 | 0.242 | 0.000 | 0.052 | 100.001 | 61.062 |
|  | 10BH167 A12 19 | 54.116 | 0.005 | 28.550 | 0.080 | 11.839 | 0.000 | 0.618 | 4.827 | 0.327 | 0.070 | 0.062 | 100.494 | 56.407 |
|  | 10BH167 A12 20 | 56.282 | 0.000 | 27.315 | 0.031 | 10.164 | 0.053 | 0.623 | 5.792 | 0.416 | 0.083 | 0.000 | 100.759 | 48.010 |
|  | $\begin{aligned} & \text { 10BH167 A12 18- } \\ & 20 \mathrm{av} \end{aligned}$ | 54.287 | 0.009 | 28.355 | 0.077 | 11.605 | 0.019 | 0.657 | 4.992 | 0.328 | 0.051 | 0.038 | 100.418 | 55.138 |
| UBZ | 10BH167 A14 4 | 50.772 | 0.043 | 30.420 | 0.136 | 14.363 | 0.000 | 1.029 | 3.464 | 0.237 | 0.000 | 0.070 | 100.534 | 68.678 |
|  | 10BH167 A14 5 | 51.167 | 0.000 | 30.320 | 0.145 | 14.411 | 0.000 | 0.954 | 3.521 | 0.232 | 0.000 | 0.049 | 100.799 | 68.433 |
|  | 10BH167 A14 6 | 50.966 | 0.048 | 30.603 | 0.172 | 14.660 | 0.000 | 1.001 | 3.389 | 0.222 | 0.031 | 0.066 | 101.158 | 69.583 |
|  | $\begin{aligned} & \text { 10BH167 A14 4-6 } \\ & \text { av } \end{aligned}$ | 50.968 | 0.030 | 30.448 | 0.151 | 14.478 | 0.000 | 0.995 | 3.458 | 0.230 | 0.010 | 0.062 | 100.830 | 68.899 |
|  | 10BH167 A14 7 | 51.527 | 0.000 | 30.425 | 0.141 | 14.159 | 0.006 | 1.063 | 3.641 | 0.172 | 0.000 | 0.049 | 101.183 | 67.577 |
|  | 10BH167 A14 8 | 50.776 | 0.097 | 30.441 | 0.125 | 14.332 | 0.021 | 0.978 | 3.516 | 0.163 | 0.000 | 0.000 | 100.449 | 68.612 |
|  | 10BH167 A14 9 | 50.620 | 0.038 | 30.686 | 0.134 | 14.625 | 0.021 | 1.014 | 3.304 | 0.201 | 0.005 | 0.049 | 100.697 | 70.161 |
|  | $\begin{aligned} & \text { 10BH167 A14 7-9 } \\ & \text { av } \end{aligned}$ | 50.974 | 0.045 | 30.517 | 0.133 | 14.372 | 0.016 | 1.018 | 3.487 | 0.179 | 0.002 | 0.033 | 100.776 | 68.781 |
|  | 10BH167 A14 21 | 50.856 | 0.016 | 30.378 | 0.149 | 14.233 | 0.000 | 0.789 | 3.466 | 0.184 | 0.000 | 0.066 | 100.137 | 68.679 |
|  | 10BH167 A14 22 | 50.713 | 0.016 | 30.763 | 0.135 | 14.519 | 0.000 | 0.875 | 3.345 | 0.172 | 0.047 | 0.024 | 100.609 | 69.823 |
|  | 10BH167 A14 23 | 50.262 | 0.043 | 31.191 | 0.102 | 15.024 | 0.000 | 0.970 | 3.056 | 0.176 | 0.000 | 0.000 | 100.824 | 72.358 |
|  | $\begin{aligned} & \text { 10BH167 A14 21- } \\ & 23 \text { av } \end{aligned}$ | 50.610 | 0.025 | 30.777 | 0.129 | 14.592 | 0.000 | 0.878 | 3.289 | 0.177 | 0.016 | 0.030 | 100.523 | 70.287 |
|  | 10BH167 A14 24 | 51.030 | 0.000 | 30.276 | 0.142 | 14.408 | 0.038 | 0.991 | 3.477 | 0.182 | 0.049 | 0.017 | 100.610 | 68.824 |
|  | 10BH167 A14 25 | 51.214 | 0.054 | 30.287 | 0.156 | 14.399 | 0.000 | 0.946 | 3.494 | 0.174 | 0.000 | 0.153 | 100.877 | 68.800 |
|  | 10BH167 A14 26 | 50.781 | 0.011 | 30.288 | 0.117 | 14.318 | 0.000 | 1.039 | 3.518 | 0.165 | 0.000 | 0.024 | 100.261 | 68.571 |
|  | 10BH167 A14 33 | 50.688 | 0.000 | 30.280 | 0.123 | 14.497 | 0.035 | 1.046 | 3.736 | 0.207 | 0.000 | 0.007 | 100.619 | 67.415 |

Appendix E4 - LPS Plagioclase compositions

| Zone | Sample code | SiO 2 | TiO 2 | $\mathrm{Al2O} 3$ | MgO | CaO | MnO | FeO | Na 2 O | K 2 O | BaO | SrO | Total | Anmol\% |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
|  | 10BH167 A14 34 | 50.181 | 0.065 | 31.336 | 0.117 | 15.088 | 0.023 | 0.982 | 3.192 | 0.144 | 0.000 | 0.052 | 101.180 | 71.726 |
|  | 10BH167 A14 35 | 50.030 | 0.000 | 30.762 | 0.128 | 14.529 | 0.000 | 0.939 | 3.442 | 0.171 | 0.000 | 0.028 | 100.029 | 69.314 |
|  | 10BH167 A14 33- <br> 35 av | 50.300 | 0.022 | 30.793 | 0.123 | 14.705 | 0.019 | 0.989 | 3.457 | 0.174 | 0.000 | 0.029 | 100.609 | 69.470 |

Appendix E5 - LPS Cr-spinel compositions

| Zone | Sample code | SiO 2 | TiO2 | Al2O3 | Cr2O3 | Fe2O3 | MgO | CaO | MnO | FeO | CoO | NiO | ZnO | Na2O | Total | Cr\# | Fe3\# | Fe2\# |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| LCM | 166 A1 7 | 0.091 | 0.695 | 19.937 | 41.393 | 7.865 | 8.783 | 0.000 | 0.000 | $\begin{aligned} & 21.6 \\ & 58 \end{aligned}$ | 0.000 | 0.096 | 0.073 | 0.021 | 100.614 | $\begin{aligned} & \hline 58.20 \\ & 8 \end{aligned}$ | 9.525 | 58.041 |
|  | 166 A1 8 | 0.105 | 0.365 | 25.879 | 38.491 | 6.696 | 14.219 | 0.000 | 0.000 | $\begin{aligned} & 13.9 \\ & 33 \\ & \hline \end{aligned}$ | 0.050 | 0.253 | 0.049 | 0.025 | 100.065 | $\begin{aligned} & 50.16 \\ & 1 \end{aligned}$ | 7.607 | 35.471 |
|  | 166 A1 9 | 0.089 | 0.391 | 25.851 | 38.249 | 6.616 | 13.556 | 0.000 | 0.000 | $\begin{aligned} & 14.9 \\ & 65 \\ & \hline \end{aligned}$ | 0.043 | 0.241 | 0.064 | 0.018 | 100.083 | $\begin{aligned} & 49.81 \\ & 4 \end{aligned}$ | 7.579 | 38.244 |
|  | 166 A1 13 | 0.069 | 0.765 | 23.695 | 37.931 | 9.384 | 13.001 | 0.000 | 0.000 | $\begin{aligned} & 15.6 \\ & 86 \end{aligned}$ | 0.071 | 0.294 | 0.024 | 0.000 | 100.919 | $\begin{aligned} & 51.78 \\ & 1 \end{aligned}$ | $\begin{aligned} & 10.86 \\ & 7 \end{aligned}$ | 40.363 |
| LBZ | 166_A3 25 | 0.197 | 2.030 | 10.917 | 28.695 | $\begin{aligned} & 25.88 \\ & 2 \end{aligned}$ | 4.056 | 0.063 | 0.000 | $\begin{aligned} & \hline 26.2 \\ & 71 \\ & \hline \end{aligned}$ | 0.077 | 0.101 | 0.169 | 0.013 | 98.471 | $\begin{aligned} & 63.81 \\ & 2 \end{aligned}$ | $\begin{aligned} & 35.39 \\ & 4 \end{aligned}$ | 78.418 |
|  | 166_A3 26 | 4.422 | 1.704 | 10.962 | 27.678 | $\begin{aligned} & 28.87 \\ & 0 \end{aligned}$ | 12.006 | 0.009 | 0.000 | $\begin{aligned} & 15.2 \\ & 35 \end{aligned}$ | 0.161 | 0.208 | 0.179 | 0.020 | 101.454 | $\begin{aligned} & 62.87 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 38.43 \\ & 1 \\ & \hline \end{aligned}$ | 41.587 |
| LOZ | 166_A4 22 | 0.073 | 1.483 | 20.449 | 33.048 | $\begin{aligned} & 14.32 \\ & 1 \\ & \hline \end{aligned}$ | 8.821 | 0.000 | 0.000 | $\begin{aligned} & 20.9 \\ & 76 \\ & \hline \end{aligned}$ | 0.094 | 0.205 | 0.223 | 0.000 | 99.693 | $\begin{aligned} & 52.01 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & 17.66 \\ & 3 \\ & \hline \end{aligned}$ | 57.152 |
|  | 166_A4 23 | 0.041 | $\begin{aligned} & 11.30 \\ & 8 \end{aligned}$ | 3.753 | 8.610 | $\begin{aligned} & 45.09 \\ & 3 \end{aligned}$ | 2.729 | 0.000 | 0.144 | $\begin{aligned} & 26.8 \\ & 46 \end{aligned}$ | 0.021 | 0.169 | 0.082 | 0.000 | 98.796 | $\begin{aligned} & 60.61 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & 75.13 \\ & 7 \\ & \hline \end{aligned}$ | 84.665 |
|  | 166_A4 27 | 0.077 | 0.540 | 21.024 | 36.647 | $\begin{aligned} & 10.24 \\ & 3 \end{aligned}$ | 8.748 | 0.003 | 0.000 | $\begin{aligned} & 20.9 \\ & 04 \end{aligned}$ | 0.107 | 0.177 | 0.231 | 0.023 | 98.724 | $\begin{aligned} & 53.90 \\ & 3 \end{aligned}$ | $12.53$ | 57.274 |
|  | 166_A4 28 | 0.066 | 0.545 | 21.681 | 38.113 | 8.833 | 9.697 | 0.000 | 0.000 | $\begin{aligned} & 19.8 \\ & 07 \end{aligned}$ | 0.077 | 0.210 | 0.140 | 0.025 | 99.194 | $\begin{aligned} & 54.11 \\ & 3 \end{aligned}$ | $\begin{aligned} & 10.66 \\ & 2 \end{aligned}$ | 53.403 |
| LOZ | 166_A5 14 | 0.066 | 6.317 | 1.235 | 1.586 | $\begin{aligned} & 57.35 \\ & 1 \end{aligned}$ | 0.371 | 0.021 | 0.680 | $\begin{aligned} & 28.5 \\ & 88 \end{aligned}$ | 0.000 | 0.205 | 0.084 | 0.028 | 96.532 | $\begin{aligned} & 46.29 \\ & 2 \end{aligned}$ | $\begin{aligned} & 94.09 \\ & 1 \end{aligned}$ | 97.734 |
|  | 166_A5 20 | 0.054 | 7.544 | 5.110 | 21.458 | $\begin{aligned} & 34.35 \\ & 0 \end{aligned}$ | 3.690 | 0.001 | 0.000 | $\begin{aligned} & \hline 25.7 \\ & 33 \\ & \hline \end{aligned}$ | 0.080 | 0.177 | 0.073 | 0.000 | 98.270 | $\begin{aligned} & 73.80 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 52.92 \\ & 7 \end{aligned}$ | 79.640 |
|  | 166_A5 21 | 0.068 | 1.945 | 10.004 | 41.377 | $\begin{aligned} & 15.52 \\ & 6 \end{aligned}$ | 5.302 | 0.000 | 0.000 | $\begin{aligned} & \hline 24.7 \\ & 68 \\ & \hline \end{aligned}$ | 0.005 | 0.048 | 0.165 | 0.040 | 99.248 | $\begin{aligned} & 73.50 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 20.79 \\ & 1 \end{aligned}$ | 72.384 |
|  | 166_A5 31 | 0.066 | 6.298 | 3.915 | 1.553 | $\begin{aligned} & 51.92 \\ & 1 \end{aligned}$ | 0.000 | 0.047 | 0.295 | $\begin{aligned} & 29.0 \\ & 65 \end{aligned}$ | 0.000 | 0.295 | 0.000 | 0.007 | 93.462 | $\begin{aligned} & 21.02 \\ & 4 \end{aligned}$ | $\begin{aligned} & 86.99 \\ & 3 \end{aligned}$ | $\begin{aligned} & \hline 100.00 \\ & 0 \end{aligned}$ |
| MOZ | 167_A3 3 | 0.049 | 0.391 | 11.133 | 35.522 | $\begin{aligned} & 20.34 \\ & 3 \end{aligned}$ | 5.159 | 0.027 | 0.000 | $\begin{aligned} & \hline 24.4 \\ & 44 \\ & \hline \end{aligned}$ | 0.000 | 0.184 | 0.128 | 0.012 | 97.392 | $\begin{aligned} & 68.15 \\ & 8 \end{aligned}$ | $\begin{aligned} & 27.08 \\ & 9 \end{aligned}$ | 72.666 |
|  | 167_A3 4 | 0.697 | 5.585 | 3.674 | 20.112 | $\begin{aligned} & 37.44 \\ & 7 \end{aligned}$ | 3.147 | 0.021 | 0.000 | $\begin{aligned} & 25.8 \\ & 56 \\ & \hline \end{aligned}$ | 0.000 | 0.218 | 0.113 | 0.030 | 96.900 | $\begin{aligned} & 78.59 \\ & 5 \end{aligned}$ | $\begin{aligned} & 58.20 \\ & 7 \end{aligned}$ | 82.170 |
| UOZ | 167_A4 9 | 0.037 | 5.894 | 6.133 | 27.590 | $\begin{aligned} & 28.71 \\ & 2 \end{aligned}$ | 3.559 | 0.004 | 0.000 | $\begin{aligned} & 26.3 \\ & 50 \end{aligned}$ | 0.006 | 0.140 | 0.107 | 0.008 | 98.540 | $\begin{aligned} & 75.10 \\ & 9 \end{aligned}$ | $\begin{aligned} & 42.65 \\ & 9 \\ & \hline \end{aligned}$ | 80.591 |
|  | 167_A4 10 | 0.059 | 2.932 | 9.578 | 33.344 | $\begin{aligned} & 22.66 \\ & 5 \end{aligned}$ | 5.481 | 0.027 | 0.000 | $\begin{aligned} & 24.0 \\ & 18 \end{aligned}$ | 0.018 | 0.105 | 0.109 | 0.011 | 98.347 | $\begin{aligned} & 70.02 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & 31.17 \\ & 5 \end{aligned}$ | 71.087 |
|  | 167_A4 11 | 0.044 | $\begin{aligned} & 11.14 \\ & 7 \\ & \hline \end{aligned}$ | 3.772 | 7.857 | $\begin{aligned} & 46.47 \\ & 7 \\ & \hline \end{aligned}$ | 3.078 | 0.000 | 0.079 | $\begin{aligned} & \hline 26.4 \\ & 03 \\ & \hline \end{aligned}$ | 0.101 | 0.186 | 0.059 | 0.009 | 99.212 | $\begin{aligned} & 58.28 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 76.64 \\ & 0 \\ & \hline \end{aligned}$ | 82.798 |
| UOZ | BH311 B3 1 | 0.040 | 8.803 | 3.408 | 6.726 | $\begin{aligned} & 48.22 \\ & 3 \end{aligned}$ | 2.277 | 0.008 | 0.102 | $\begin{aligned} & \hline 28.5 \\ & 01 \end{aligned}$ | 0.082 | 0.220 | 0.016 | 0.027 | 102.149 | $\begin{aligned} & 56.97 \\ & 4 \end{aligned}$ | $\begin{aligned} & 79.54 \\ & 2 \end{aligned}$ | 87.538 |
|  | BH311 B3 2 | 0.019 | 8.048 | 4.583 | 18.321 | 35.51 | 3.544 | 0.000 | 0.000 | 28.8 | 0.000 | 0.198 | 0.032 | 0.000 | 101.212 | 72.83 | 57.33 | 82.041 |

Appendix E5 - LPS Cr-spinel compositions

|  |  |  |  |  |  | 0 |  |  |  | 47 |  |  |  |  |  | 5 | 4 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | BH311 B3 4 | 0.042 | 8.436 | 3.653 | 18.648 | $35.20$ | 3.454 | 0.000 | 0.000 | $\begin{aligned} & \hline 28.3 \\ & 71 \end{aligned}$ | 0.039 | 0.150 | 0.092 | 0.031 | 101.086 | $77.40$ <br> 2 | $\begin{aligned} & 58.16 \\ & 7 \\ & \hline \end{aligned}$ | 82.162 |
|  | BH311 B3 6 | 0.050 | 0.950 | 4.995 | 40.857 | 4.148 | 10.511 | 0.178 | 0.000 | $\begin{aligned} & 22.8 \\ & 24 \end{aligned}$ | 0.033 | 0.078 | 0.043 | 0.017 | 100.660 | $60.36$ $7$ | 5.515 | 54.917 |
|  | BH311 B3 7 | 0.050 | 1.033 | 18.701 | 39.676 | 4.812 | 10.623 | 0.161 | 0.000 | $\begin{aligned} & 22.8 \\ & 09 \end{aligned}$ | 0.064 | 0.126 | 0.071 | 0.009 | 101.045 | $\begin{aligned} & 58.73 \\ & 4 \\ & \hline \end{aligned}$ | 6.348 | 54.640 |
|  | BH311 B3 8 | 0.052 | 0.972 | 17.970 | 40.804 | 4.540 | 10.585 | 0.129 | 0.000 | $\begin{aligned} & 22.5 \\ & 94 \end{aligned}$ | 0.045 | 0.116 | 0.142 | 0.023 | 100.616 | $\begin{aligned} & 60.36 \\ & 9 \end{aligned}$ | 6.010 | 54.490 |
|  | BH311 B3 9 | 0.462 | 1.268 | 17.041 | 39.106 | 6.605 | 10.467 | 0.205 | 0.000 | $\begin{aligned} & 22.5 \\ & 87 \end{aligned}$ | 0.011 | 0.107 | 0.037 | 0.039 | 101.441 | $\begin{aligned} & \hline 60.62 \\ & 2 \end{aligned}$ | 8.884 | 54.760 |
|  | BH311 B3 10 | 0.040 | 2.017 | 12.677 | 35.675 | $\begin{aligned} & 13.17 \\ & 3 \end{aligned}$ | 6.495 | 0.191 | 0.000 | $\begin{aligned} & 27.4 \\ & 60 \end{aligned}$ | 0.042 | 0.063 | 0.051 | 0.041 | 100.254 | $\begin{aligned} & 65.37 \\ & 3 \end{aligned}$ | $\begin{aligned} & 18.68 \\ & 3 \end{aligned}$ | 70.349 |
|  | BH311 B3 11 | 0.041 | 3.168 | 9.496 | 33.634 | $\begin{aligned} & 17.09 \\ & 8 \end{aligned}$ | 5.359 | 0.092 | 0.000 | $\begin{aligned} & 28.2 \\ & 19 \end{aligned}$ | 0.046 | 0.134 | 0.150 | 0.004 | 100.948 | $\begin{aligned} & 70.38 \\ & 0 \end{aligned}$ | $\begin{aligned} & 25.40 \\ & 0 \end{aligned}$ | 74.706 |
|  | BH311 B3 12 | 0.010 | $\begin{aligned} & \hline 10.70 \\ & 1 \\ & \hline \end{aligned}$ | 4.337 | 10.623 | $\begin{aligned} & \hline 42.36 \\ & 6 \end{aligned}$ | 4.341 | 0.031 | 0.000 | $\begin{aligned} & \hline 26.3 \\ & 06 \\ & \hline \end{aligned}$ | 0.069 | 0.187 | 0.126 | 0.022 | 101.413 | $\begin{aligned} & \hline 62.16 \\ & 5 \end{aligned}$ | $\begin{aligned} & 70.23 \\ & 7 \end{aligned}$ | 77.274 |
|  | BH311 B3 55 | 0.029 | 9.152 | 4.226 | 12.993 | $\begin{aligned} & 41.00 \\ & 3 \end{aligned}$ | 3.166 | 0.000 | 0.000 | $\begin{aligned} & 28.6 \\ & 99 \end{aligned}$ | 0.000 | 0.222 | 0.076 | 0.004 | 100.686 | $\begin{aligned} & 67.35 \\ & 1 \end{aligned}$ | $\begin{aligned} & 66.91 \\ & 5 \end{aligned}$ | 83.567 |
|  | BH311 B3 56 | 0.033 | 8.518 | 4.732 | 17.134 | $\begin{aligned} & 36.77 \\ & 9 \end{aligned}$ | 3.991 | 0.000 | 0.000 | $\begin{aligned} & 28.1 \\ & 40 \end{aligned}$ | 0.045 | 0.196 | 0.030 | 0.000 | 101.280 | $\begin{aligned} & 70.83 \\ & 3 \end{aligned}$ | $\begin{aligned} & 59.13 \\ & 3 \end{aligned}$ | 79.824 |
| CPZ | 167 A5 98 | 0.037 | $\begin{array}{\|l\|} \hline 50.16 \\ 8 \end{array}$ | 0.049 | 0.028 | $\begin{aligned} & 19.98 \\ & 7 \\ & \hline \end{aligned}$ | 0.018 | 0.006 | 1.744 | $\begin{aligned} & 29.7 \\ & 43 \end{aligned}$ | 0.023 | 0.000 | 0.000 | 0.020 | 101.823 | $\begin{aligned} & 28.57 \\ & 1 \end{aligned}$ | $\begin{aligned} & 99.45 \\ & 9 \\ & \hline \end{aligned}$ | 99.891 |
| DZ | BH311 B6 76 | 0.027 | $\begin{array}{\|l\|} \hline 51.27 \\ 7 \end{array}$ | 0.046 | 0.006 | $\begin{aligned} & 18.50 \\ & 0 \end{aligned}$ | 0.019 | 0.059 | 1.694 | $\begin{aligned} & 29.5 \\ & 95 \end{aligned}$ | 0.002 | 0.000 | 0.000 | 0.014 | 101.292 | 6.667 | $\begin{aligned} & 99.58 \\ & 3 \\ & \hline \end{aligned}$ | 99.890 |

Appendix E6 - PELE model results

|  |  |  |  |  | Liq Comp | Liq Comp | Liq Comp | Liq Comp | Liq Comp | Liq Comp | Liq Comp | Liq Comp | Liq Comp | Liq Comp | Liq Comp | Liq Comp | Liq Comp | Liq Comp | Liq Comp |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Step | $\begin{aligned} & \mathrm{T} \\ & \mathrm{C} \end{aligned}$ | $\begin{aligned} & \text { Logf } \\ & \text { O2 } \end{aligned}$ | $\begin{aligned} & \text { Logf } \\ & \text { S2 } \end{aligned}$ | Liq Mas s | SiO 2 | TiO2 | Al2O3 | Fe2O3 | Cr2O3 | FeO | FeO* | MgO | CaO | Na 2 O | K2O | P2O5 | H2O | CO 2 | S |
| 5 | $\begin{aligned} & 13 \\ & 05 \\ & \hline \end{aligned}$ | -6.43 | -0.21 | $\begin{aligned} & 99.9 \\ & 6 \end{aligned}$ | 50.22 | 0.91 | 13.26 | 2.16 | 0.05 | 8.34 | 10.28 | 10.31 | 11.23 | 1.97 | 1.17 | 0.08 | 0.20 | 0.01 | 0.11 |
| 10 | $\begin{aligned} & 12 \\ & 55 \end{aligned}$ | -6.94 | -0.25 | $\begin{aligned} & 99.8 \\ & 9 \end{aligned}$ | 50.25 | 0.91 | 13.27 | 2.20 | 0.05 | 8.27 | 10.24 | 10.31 | 11.24 | 1.97 | 1.17 | 0.08 | 0.20 | 0.01 | 0.09 |
| 11 | $\begin{aligned} & 12 \\ & 45 \end{aligned}$ | $-7.04$ | -0.26 | $\begin{aligned} & 99.5 \\ & 6 \end{aligned}$ | 50.29 | 0.91 | 13.31 | 2.21 | 0.05 | 8.25 | 10.23 | 10.19 | 11.27 | 1.98 | 1.17 | 0.08 | 0.20 | 0.01 | 0.09 |
| 12 | $\begin{aligned} & 12 \\ & 35 \end{aligned}$ | $-7.13$ | -0.26 | $\begin{aligned} & 98.5 \\ & 9 \end{aligned}$ | 50.39 | 0.92 | 13.44 | 2.23 | 0.05 | 8.19 | 10.20 | 9.84 | 11.39 | 1.99 | 1.18 | 0.08 | 0.21 | 0.01 | 0.08 |
| 13 | $\begin{aligned} & 12 \\ & 25 \end{aligned}$ | $-7.23$ | -0.28 | $\begin{aligned} & 97.6 \\ & 5 \\ & \hline \end{aligned}$ | 50.49 | 0.93 | 13.57 | 2.25 | 0.05 | 8.15 | 10.17 | 9.48 | 11.50 | 2.01 | 1.19 | 0.08 | 0.21 | 0.01 | 0.08 |
| 14 | $\begin{aligned} & 12 \\ & 15 \end{aligned}$ | $-7.34$ | -0.30 | $\begin{aligned} & 96.7 \\ & 3 \\ & \hline \end{aligned}$ | 50.60 | 0.94 | 13.70 | 2.26 | 0.05 | 8.11 | 10.13 | 9.14 | 11.60 | 2.03 | 1.20 | 0.08 | 0.21 | 0.01 | 0.08 |
| 15 | $\begin{aligned} & 12 \\ & 05 \\ & \hline \end{aligned}$ | -7.45 | -0.32 | $\begin{aligned} & 95.8 \\ & 5 \\ & \hline \end{aligned}$ | 50.70 | 0.94 | 13.83 | 2.27 | 0.05 | 8.06 | 10.09 | 8.80 | 11.71 | 2.05 | 1.22 | 0.08 | 0.21 | 0.01 | 0.07 |
| 16 | $\begin{aligned} & 11 \\ & 95 \\ & \hline \end{aligned}$ | $-7.56$ | -0.33 | $\begin{aligned} & 94.9 \\ & 9 \\ & \hline \end{aligned}$ | 50.80 | 0.95 | 13.95 | 2.28 | 0.05 | 8.00 | 10.05 | 8.48 | 11.82 | 2.07 | 1.23 | 0.08 | 0.21 | 0.01 | 0.07 |
| 17 | $\begin{aligned} & 11 \\ & 85 \end{aligned}$ | $-7.68$ | -0.35 | $\begin{aligned} & 94.1 \\ & 7 \\ & \hline \end{aligned}$ | 50.90 | 0.96 | 14.07 | 2.28 | 0.05 | 7.95 | 10.00 | 8.16 | 11.92 | 2.09 | 1.24 | 0.08 | 0.21 | 0.01 | 0.07 |
| 18 | $\begin{aligned} & 11 \\ & 75 \\ & \hline \end{aligned}$ | -7.79 | -0.37 | $\begin{aligned} & 93.3 \\ & 7 \\ & \hline \end{aligned}$ | 51.00 | 0.97 | 14.19 | 2.29 | 0.05 | 7.89 | 9.95 | 7.85 | 12.02 | 2.11 | 1.25 | 0.09 | 0.22 | 0.02 | 0.07 |
| 18b | $\begin{aligned} & 11 \\ & 70 \\ & \hline \end{aligned}$ | $-7.80$ | -0.34 | $\begin{aligned} & 92.9 \\ & 7 \end{aligned}$ | 51.05 | 0.97 | 14.25 | 2.33 | 0.05 | 7.81 | 9.90 | 7.71 | 12.07 | 2.12 | 1.25 | 0.09 | 0.22 | 0.02 | 0.07 |
| 19 | $\begin{aligned} & 11 \\ & 65 \\ & \hline \end{aligned}$ | $-7.88$ | -0.33 | $\begin{aligned} & 88.6 \\ & 6 \\ & \hline \end{aligned}$ | 51.18 | 1.02 | 14.09 | 2.38 | 0.05 | 8.05 | 10.19 | 7.47 | 11.89 | 2.16 | 1.31 | 0.09 | 0.23 | 0.02 | 0.07 |
| 19b | $\begin{aligned} & 11 \\ & 60 \end{aligned}$ | $-7.91$ | -0.26 | $\begin{aligned} & 83.3 \\ & 4 \\ & \hline \end{aligned}$ | 51.28 | 1.09 | 14.02 | 2.49 | 0.06 | 8.35 | 10.59 | 7.17 | 11.51 | 2.23 | 1.40 | 0.10 | 0.24 | 0.02 | 0.07 |
| 20 | $\begin{aligned} & 11 \\ & 55 \end{aligned}$ | -7.96 | -0.22 | $\begin{aligned} & 78.6 \\ & 9 \\ & \hline \end{aligned}$ | 51.38 | 1.15 | 13.94 | 2.57 | 0.06 | 8.67 | 10.97 | 6.87 | 11.15 | 2.29 | 1.48 | 0.10 | 0.26 | 0.02 | 0.07 |
| 21 | $\begin{aligned} & 11 \\ & 45 \\ & \hline \end{aligned}$ | -7.98 | -0.04 | $\begin{aligned} & 71.0 \\ & 4 \end{aligned}$ | 51.53 | 1.27 | 13.78 | 2.81 | 0.07 | 9.19 | 11.72 | 6.35 | 10.46 | 2.40 | 1.64 | 0.11 | 0.28 | 0.02 | 0.08 |
| 22 | $\begin{aligned} & 11 \\ & 35 \\ & \hline \end{aligned}$ | -8.10 | 0.04 | $\begin{aligned} & \hline 64.8 \\ & 6 \\ & \hline \end{aligned}$ | 51.71 | 1.40 | 13.63 | 2.95 | 0.07 | 9.73 | 12.38 | 5.83 | 9.85 | 2.50 | 1.80 | 0.12 | 0.31 | 0.02 | 0.08 |
| 23 | $\begin{aligned} & 11 \\ & 25 \end{aligned}$ | -8.22 | 0.11 | $\begin{aligned} & 59.8 \\ & 2 \\ & \hline \end{aligned}$ | 51.89 | 1.51 | 13.49 | 3.07 | 0.08 | 10.21 | 12.97 | 5.35 | 9.29 | 2.59 | 1.95 | 0.13 | 0.34 | 0.02 | 0.09 |
| 24 | $\begin{aligned} & 11 \\ & 15 \\ & \hline \end{aligned}$ | -8.35 | 0.19 | $\begin{aligned} & 55.6 \\ & 1 \\ & \hline \end{aligned}$ | 52.06 | 1.63 | 13.34 | 3.18 | 0.09 | 10.66 | 13.52 | 4.90 | 8.77 | 2.67 | 2.10 | 0.14 | 0.36 | 0.02 | 0.09 |
| 25 | $\begin{aligned} & 11 \\ & 05 \\ & \hline \end{aligned}$ | -8.47 | 0.25 | $\begin{aligned} & 52.0 \\ & 4 \end{aligned}$ | 52.24 | 1.74 | 13.19 | 3.29 | 0.09 | 11.06 | 14.01 | 4.48 | 8.29 | 2.74 | 2.24 | 0.15 | 0.39 | 0.03 | 0.09 |

## Appendix E6-1 (accompanying chapter 4) - PELE modelling

PELE modeling - justification for environmental conditions: since many of the most detailed studies were done on sills emplaced near the Fort Collinson Fm, pressures for V0 differentiation were estimated from the height of this overlying stratigraphic column, ca. 2.5 km , which translates to ca. 0.8 kb . All simulations were done at this pressure. Preliminary QUILF calculations by Bédard et al 2008 (Yellowknife geoscience forum poster) on v0 sills suggest $\mathrm{QFM}+1 \mathrm{t}+2$, so calculations were done at $\mathrm{QFM}+1$ and +2 , bracketing observed fO 2 conditions. The gabbroic segment of the PELE models cannot exactly mimic the observed trends, however, because the PELE code does not include pigeonite, which is observed near the base of the gabbroic sequence in several V0 sills. Trace hornblende and biotite occur as rims on highertemperature silicates in many sills, suggesting that some magmas were not entirely dry, such that extended in-situ fractional crystallization locally caused water contents to build up to the point where these phases were stabilized. An initial water content of $0.2 \%$ was assumed for some models, since $95 \%$ crystallization of such a starting point will yield the $4 \% \mathrm{H} 2 \mathrm{O}$ content needed to stabilize hornblende. Of course, F and Cl will stabilize these phases even at lower H 2 O contents. Limited data show that Hb and biotite in Franklin sills do indeed contain significant Cl and F -contents (up to 1$1.5 \mathrm{wt} \%$ combined $\mathrm{F}+\mathrm{Cl}$ ), but some water may also be present. There is no direct analytical data on H in these phases to constrain this at present. Melt CO2 solubilities are very low at these pressures, and initial staring contents were assumed to be ca $0.2 \%$ also. These values were varied. Typical S-contents of Franklin magmas are ca $0.04 \%$, with some chilled margins recording higher values (up to ca $0.15 \%$ ). Small Ni-Fe sulphide blebs in chills and olivine cumulates suggest early, small-scale sulphide immiscibility may be common. Even though PELE does not consider the effects of Cu and Ni on sulphide immiscibility, early Fe-sulfide melt exsolution is commonly observed in models that assumed higher initial S contents of ca $0.12 \%$, and small amounts of sulfide exsolves intermittently throughout the model crystallization history. The amounts are always small, however, in keeping with the small amount of initial S in the melt. More evolved ferrogabbroic rocks typically contain $\mathrm{Cu}-\mathrm{Fe}$ sulphide blebs. Models that assumed low initial $S$ contents of ca $0.04 \%$ only showed immiscibility in the later crystallization history, often after cosaturation of magnetite. This suggests that the sharp decrease in FeO caused by magnetite crystallization often
triggers immiscibility by lowering S-solubility in the melt, which is strongly dependent on FeO content (Mavrogenes?). These results and observations imply that the v 0 melts were probably very close to sulfide saturation, but that S contents were probably too low to create sulphide orebodies during closed-system crystallization. On the other hand, even minor assimilation of S-rich limestone or sulfates would probably trigger immiscible sulfide separation, with the amount of sulphide produced correlating directly to S-input.

Optimal PELE model run - A $3^{\text {rd }}$ starting composition (C), attempted to mimic evolution of the Uhuk/P-Sill/LPS system using (initially) the average composition of the chilled margin and roof-dyke apophyses from the Uhuk massif, which have ca. $10.3 \mathrm{wt} \% \mathrm{MgO}$ and 0.12 \% S. In retrospect, the high S is probably the result of minor wallrock assimilation or exchange, and is probably higher than the magmatic value. V0 model D2, Uhuk2 \#7 is used in the manuscript.

## Appendix E7-2 - Element map descriptions and interpretations (accompanying Chapter 4)

1) 10RAT-BH166A03 - LBZ - a rare orthopyroxene (opx) crystal. BSE: euhedral orthopyroxene with co-crystallized plagioclase. Plagioclase is interstitial to olivine. Some olivine may be partially enclosed by the opx. Fe: twins are visible in the orthopyroxene. Enclosed olivines have less Fe than groundmass/interstitial olivine. Magnetite is visible in the groundmass and in fractured olivine. Cr: a Cr-rich band across the centre of the orthopyroxene crystal. Clinopyroxene to the left that is interstitial may also be sector-zoned? Minimal subhedral chromites in the groundmass. Ti: Ti pattern in the orthopyroxene is very similar to Cr. The northern rim of the orthopyroxene is Ti-enriched. Ilmenite is present in the groundmass. Al: Twinning in orthopyroxene is prominent. Interstitial plagioclase around the groundmass is dendritic-looking. Mg: Orthopyroxene crystal doesn't appear to show twinning in Mg. Chadacrystic olivines are more Mg-rich than groundmass olivines. Olivine in lower right of image appears to be partially intergrown with clinopyroxene.
2) 11RAT-BH311B02 - LOZ - zoned clinopyroxene crystal. BSE: Euhedral clinopyroxene crystal with a fine-grained olivine chadacryst in its core. Rim of the clinopyroxene is slightly irregular because of plagioclase impingement. Ti: hint of Tienrichment on clinopyroxene rim. Trace amounts of ilmenite may be visible? Cr: the western half of the clinopyroxene crystal is enriched in Cr. Possibly a broken crystal/xenocryst? A clinopyroxene crystal in the top left of the image shows sectorzoning. Ca : Ca shows the same behaviour as Cr . Some calcite/carbonate cement in the groundmass? Mg: Clinopyroxene crystal shows similar zoning pattern to Cr and Ca , though area rich in Mg is relatively depleted in Cr and Ca . Olivine is prominent, with some showing partial co-crystallization textures with both clinopyroxene and plagioclase. Al : Al behaves like Cr and Ca in the clinopyroxene crystal. Sector-zoned in the top left clinopyroxene is visible. The interstitial texture of plagioclase is prominent.
3) 11RAT-BH311B03 - UOZ - euhedral clinopyroxene crystal. BSE: a euhedral, cumulus-looking clinopyroxene crystal with a euhedral chromite and tabular olivine
inclusion. Ti: hint of Ti-enrichment in the rim of the clinopyroxene. Cr : clinopyroxene core is depleted, with a 'depleted-halo' surrounding the chromite crystal. Trace amounts of chromite visible in olivine and groundmass. Ca: Ca shows similar behaviour to Cr , though may be more convoluted. There appears to be a $\mathrm{Ca}-$ depleted core, a Ca-enriched mantle and Ca -depleted rim in the clinopyroxene crystal. Rim may wider in Ca than it appears in $\mathrm{Cr} . \mathrm{Mg}$ : there are hints of a patchy Mg enriched mantle in the clinopyroxene. The clinopyroxene rim appears depleted in Mg. Tabular olivine inclusion is visible, as well as surrounding skeletal olivines. Al: Al pattern in clinopyroxene is very complex; Al-enriched core with a radiating pattern of Al-enrichment and Al-depletion in the mantle/rim of the clinopyroxene.
4) 11RAT-BH311B03 - UOZ - a skeletal olivine crystal. BSE: a skeletal olivine with magnetite filled fractures, residing in interstitial plagioclase. Ti: local, fine-grained ilmenite crystals are present - usually associated with magnetite-filled fractures in olivine. Possible hint of Ti-enriched rims on clinopyroxene? Cr: euhedral clinopyroxene in the image are enriched in Cr with normal zoning. A clinopyroxene near the top of the image may have a Cr-depleted/xenocrystic core? Ca : Ca shows similar behaviour to Cr , with euhedral (cumulus?) clinopyroxene enriched in $\mathrm{Ca} . \mathrm{Mg}$ : Olivine is prominent in the image. Euhedral clinopyroxene crystals are enriched. Some minor clinopyroxene inclusions in skeletal olivines? Al: The clinopyroxene at the top of the image has Al-behaviour resembling Cr , with prominent zoning (concentric zoning?). Interstitial plagioclase is clearly visible.
5) 11RAT-BH311B04 - CPZ - sector-zoned clinopyroxene crystal. BSE: a euhedral clinopyroxene crystal. Interstitial plagioclase (partly altered to albite?) also visible in the BSE image. Ti: hint of a Ti-enriched rim on the clinopyroxene. Minor ilmenite associated with the rims of clinopyroxene are visible. Cr: prominent sector-zoned is visible in the euhedral clinopyroxene. Sector-zoning is also discernable in other cpx crystals. Mg: faint sector-zoning visible in Mg but more complex? Sometimes Mghigh's correspond to Cr-high's and Cr-low's. Al: Al behaviour is coupled with Cr . Interstitial plagioclase is clearly visible, surrounding the sector-zoned clinopyroxene crystals. 6) 10RAT-BH167A05 - CPZ - subhedral clinopyroxene crystal. BSE: a subhedral clinopyroxene crystal with normal zoning. Fe: clinopyroxene is normally zoned with an Fe -enriched rim. Local alteration at the rim with Fe-enrichment
(magnetite?)? Cr: similar behaviour to Fe. Finer-grained cpx crystals either side of the larger cpx (synneusis?) appear to have Cr-rich rims; broken grains or resorbed? Ti: banded-like Ti-zoning in the large cpx crystal? Rims of Ti-enriched with ilmenite adjacent to these rims. Al: banded-like (or sector-zoned?) zoning resembling Cr . Interstitial (some partially intergrown with cpx ?) plagioclase is visible. Mg : unclear, but possibly banded-like zoning similar to $\mathrm{Cr} / \mathrm{Al}$ but with inverse enrichment.
6) 11RAT-BH311B06 - DZ - acicular clinopyroxene crystal. BSE: euhedral and twinned clinopyroxene crystal. Other more interstitial-looking cpx may be intergrown with interstitial plagioclase. Ti: hint of Ti-enrichment along the northern rim of the cpx crystal; with adjacent ilmenite crystals. Cr: southern rim of the cpx crystal is enriched in Cr (indicating growth direction towards the LPS upper contact). $\mathrm{Ca}: \mathrm{Ca}$ behaves like Cr. Interstitial plagioclase alteration is visible. $\mathrm{Mg}: \mathrm{Mg}$ is depleted at the northern rim of the cpx crystal with similar levels of enrichment in the rest of the cpx crystal. Al: similar behaviour to $\mathrm{Cr} / \mathrm{Ca}$, with Al -enrichment at the southern rim, and Al-depletion at the northern rim of the cpx crystal. Subhedral/interstitial cpx at the top left shows irregular Al-zoning. Interstitial plagioclase is prominent in the image with co-crystallization textures with interstitial cpx.
7) 10RAT-BH167A14 - UBZ - coarse clinopyroxene crystal amongst dendritic border zone groundmass. BSE: euhedral clinopyroxene showing normal zoning and co-plagioclase-crystallization textures at its rims. Fe: cpx is normally zoned with an Fe-enriched rim. Groundmass cpx is also Fe -enriched. Cr : homogeneous $\mathrm{Cr}-$ enrichment in the core of the cpx with a depleted rim (the inverse of Fe ). Groundmass cpx shows no Cr-enrichment. Ti: Ti behaviour is closely related to Fe . Cpx rim shows Ti-enrichment. Groundmass cpx also shows Ti-enrichment. Minor ilmenite is present in the groundmass. Al: Al-behaviour in cpx appears to be very similar to Cr. Plagioclase impingement in the rims of the cpx are visible as are the habits of plagioclase dendrites/cpx dendrites. Mg : cpx is crudely normally zoned in Mg (similar behaviour to $\mathrm{Cr}-\mathrm{Ti}-\mathrm{Al}-\mathrm{Fe}$ ). Groundmass olivine is more depleted in Mg .

## Appendix E7-1 - Element maps (accompanying Chapter 4)

1) 10RAT-BH166A03-LBZ

2) 11RAT-BH311B02 - LOZ

3) 11RAT-BH311B03 - UOZ

4) 11RAT-BH311B03 - UOZ

5) 11RAT-BH311B04-CPZ


Cr

6) 10RAT-BH167A05-CPZ

7) 11RAT-BH311B06 - DZ




## Appendix F1 - FCSC whole-rock data

| INTRUSION | LPS | LPS | LPS | LPS | LPS | LPS | LPS |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| ZONE | LBZ | LBZ | OZ | OZ | OZ | OZ | OZ |
| LITHOLOGY | ol-gabbro | ol-gabbro | ol-gabbro | ol-gabbro | ol-gabbro | ol-gabbro | ol-gabbro |
| HEIGHT (m) | 0.200 | 1.020 | 1.400 | 1.790 | 2.720 | 2.810 | 3.640 |
| SAMPLE I.D. | 10BH166A2 | 10BH166A3 | 11BH311B2 | 10BH166A4 | 10BH166A5 | 10BH167A1 | 10BH167A2 |
| Lab | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff |
|  |  |  |  |  |  |  |  |
| SiO 2 | 46.471 | 45.073 | 44.357 | 44.006 | 44.491 | 43.659 | 42.389 |
| TiO2 | 0.919 | 0.803 | 0.792 | 0.721 | 0.733 | 0.704 | 0.576 |
| Al2O3 | 11.234 | 9.492 | 9.650 | 8.953 | 8.997 | 8.827 | 6.683 |
| Fe2O3T | 13.250 | 13.278 | 13.485 | 13.399 | 13.461 | 13.680 | 14.832 |
| FeO* | 11.912 | 11.937 | 12.123 | 12.046 | 12.101 | 12.299 | 13.334 |
| Fe2O3* | 1.338 | 1.341 | 1.362 | 1.353 | 1.360 | 1.382 | 1.498 |
| MnO | 0.189 | 0.188 | 0.196 | 0.191 | 0.197 | 0.185 | 0.204 |
| MgO | 15.321 | 17.841 | 19.643 | 20.688 | 19.950 | 21.418 | 23.519 |
| CaO | 10.162 | 9.649 | 9.327 | 8.921 | 8.811 | 7.966 | 6.473 |
| Na 2 O | 1.329 | 1.311 | 0.544 | 1.013 | 0.971 | 0.962 | 0.311 |
| K2O | 0.521 | 0.454 | 0.375 | 0.458 | 0.414 | 0.363 | 0.275 |
| P2O5 | 0.073 | 0.068 | 0.065 | 0.061 | 0.063 | 0.060 | 0.052 |
| LOI | 1.546 | 1.391 | 1.565 | 1.644 | 1.925 | 2.230 | 5.776 |
| SUM | 101.016 | 99.547 | 100.000 | 100.056 | 100.011 | 100.054 | 101.090 |
|  |  |  |  |  |  |  |  |
| ICP-OES (ppm) |  |  |  |  |  |  |  |
| Sc | 37.338 | 32.112 | 32.985 | 31.315 | 30.609 | 29.491 | 25.489 |
| V | 301.203 | 229.759 | 230.126 | 248.462 | 214.833 | 230.551 | 180.719 |
| Cr | 801.270 | 1039.579 | 1096.210 | 1364.362 | 1345.936 | 1314.779 | 1489.962 |
| Co | 63.429 | 61.791 | 61.894 | 62.755 | 63.185 | 65.824 | 64.520 |
| Ni | 325.367 | 403.375 | 451.197 | 723.231 | 453.780 | 536.045 | 640.872 |
| Cu | 98.701 | 91.309 | 83.047 | 80.678 | 103.995 | 91.063 | 64.891 |
| Zn | 74.591 | 72.214 | 76.352 | 82.765 | 82.037 | 76.284 | 58.904 |
| Sr | 143.942 | 106.225 | 104.842 | 110.067 | 96.208 | 98.102 | 60.147 |
| Y | 17.628 | 15.873 | 13.869 | 14.129 | 12.480 | 13.466 | 8.832 |
| Zr | 65.767 | 67.654 | 49.649 | 51.378 | 80.964 | 62.820 | 49.540 |
| Ba | 96.519 | 58.493 | 95.924 | 131.089 | 68.693 | 53.199 | 18.348 |
|  |  |  |  |  |  |  |  |
| ICP-MS |  |  |  |  |  |  |  |
| Ti | 0.902 | 0.807 | 0.763 | 0.697 | 0.759 | 0.711 | 0.564 |
| V | 291.100 | 239.400 | 206.000 | 224.700 | 242.800 | 226.000 | 187.300 |
| Cr | 788.600 | 1030.200 | 1056.500 | 1371.500 | 1343.500 | 1278.300 | 1558.300 |
| Co | 66.500 | 59.100 | 60.100 | 58.900 | 63.700 | 65.100 | 67.800 |
| Ni | 287.000 | 390.500 | 430.800 | 717.300 | 442.400 | 526.800 | 626.800 |
| Cu | 93.500 | 89.700 | 74.800 | 72.100 | 101.000 | 80.800 | 66.600 |
| Zn | 71.600 | 76.900 | 60.600 | 77.100 | 86.400 | 67.400 | 75.300 |
| Rb | 20.400 | 17.000 | 10.400 | 15.300 | 16.100 | 13.600 | 10.300 |
| K | 5214.310 | 4542.210 | 3751.910 | 4581.020 | 4140.430 | 3627.980 | 2745.240 |
| Sr | 133.800 | 113.700 | 94.200 | 104.000 | 101.500 | 94.300 | 56.900 |
| P | 729.037 | 679.891 | 651.937 | 608.736 | 627.897 | 600.748 | 522.919 |
| Y | 18.100 | 16.500 | 11.600 | 14.100 | 14.000 | 13.600 | 9.700 |
| Zr | 69.700 | 69.900 | 51.100 | 59.300 | 81.400 | 62.800 | 45.400 |
| Nb | 3.630 | 3.500 | 2.210 | 2.780 | 2.970 | 3.080 | 1.970 |
| Sn | 0.690 | 0.800 | 0.710 | 1.270 | 0.720 | 0.850 | 0.560 |
| Cs | 0.910 | 0.810 | 0.550 | 0.590 | 0.790 | 0.770 | 1.210 |
| Ba | 108.900 | 71.600 | 88.900 | 137.500 | 80.200 | 67.200 | 53.300 |
| La | 6.610 | 6.390 | 4.530 | 5.660 | 5.570 | 5.710 | 3.480 |
| Ce | 15.340 | 14.300 | 10.450 | 12.960 | 12.630 | 12.930 | 8.010 |
| Pr | 2.090 | 1.930 | 1.410 | 1.760 | 1.760 | 1.720 | 1.130 |
| Nd | 9.090 | 8.550 | 6.160 | 7.670 | 7.450 | 7.320 | 4.950 |
| Sm | 2.380 | 2.150 | 1.680 | 1.960 | 2.070 | 1.860 | 1.300 |
| Eu | 0.770 | 0.740 | 0.550 | 0.650 | 0.630 | 0.620 | 0.440 |
| Gd | 2.650 | 2.480 | 1.790 | 2.190 | 2.160 | 2.060 | 1.390 |
| Tb | 0.430 | 0.430 | 0.300 | 0.360 | 0.340 | 0.340 | 0.250 |
| Dy | 2.830 | 2.730 | 2.010 | 2.320 | 2.350 | 2.300 | 1.600 |
| Ho | 0.500 | 0.500 | 0.360 | 0.430 | 0.430 | 0.440 | 0.280 |
| Er | 1.550 | 1.450 | 1.030 | 1.260 | 1.270 | 1.270 | 0.830 |
| Tm | 0.240 | 0.240 | 0.170 | 0.210 | 0.200 | 0.190 | 0.140 |
| Yb | 1.530 | 1.460 | 1.050 | 1.250 | 1.220 | 1.250 | 0.870 |


| Lu | 0.240 | 0.230 | 0.170 | 0.200 | 0.190 | 0.190 | 0.140 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Hf | 1.770 | 1.720 | 1.050 | 1.390 | 1.900 | 1.520 | 0.960 |
| Ta | 0.240 | 0.250 | 0.170 | 0.210 | 0.210 | 0.210 | 0.130 |
| Pb | 2.830 | 3.090 | 1.710 | 3.270 | 4.380 | 2.980 | 2.190 |
| Th | 1.400 | 1.360 | 0.870 | 1.170 | 1.310 | 1.230 | 0.690 |
| U | 0.270 | 0.260 | 0.200 | 0.240 | 0.240 | 0.260 | 0.140 |
| Inversion models |  |  |  |  |  |  |  |
| Fitted to MgO (wt \%) of > | 8.000 | 8.000 | 8.000 | 8.000 | 8.000 | 8.000 | 8.000 |
| Th |  |  | 1.569 | 1.790 | 2.005 | 1.941 | 1.700 |
| Nb |  |  | 4.012 | 4.272 | 4.564 | 4.883 | 4.910 |
| Ta |  |  | 0.300 | 0.317 | 0.317 | 0.327 | 0.309 |
| La |  |  | 8.216 | 8.694 | 8.555 | 9.048 | 8.661 |
| Ce |  |  | 18.938 | 19.896 | 19.389 | 20.476 | 19.906 |
| Pr |  |  | 2.553 | 2.700 | 2.700 | 2.722 | 2.804 |
| P |  |  | 1173.815 | 930.504 | 959.794 | 946.896 | 1284.183 |
| Nd |  |  | 11.109 | 11.737 | 11.400 | 11.551 | 12.192 |
| Sm |  |  | 3.005 | 2.983 | 3.151 | 2.918 | 3.155 |
| Zr |  |  | 91.413 | 90.259 | 123.896 | 98.525 | 110.194 |
| Hf |  |  | 1.878 | 2.116 | 2.892 | 2.385 | 2.330 |
| Ti |  |  | 1.364 | 1.061 | 1.156 | 1.115 | 1.368 |
| Eu |  |  | 0.984 | 0.989 | 0.959 | 0.973 | 1.068 |
| Gd |  |  | 3.177 | 3.316 | 3.270 | 3.213 | 3.325 |
| Tb |  |  | 0.530 | 0.544 | 0.513 | 0.529 | 0.594 |
| Dy |  |  | 3.556 | 3.505 | 3.550 | 3.579 | 3.806 |
| Y |  |  | 20.536 | 21.314 | 21.162 | 21.177 | 23.106 |
| Ho |  |  | 0.635 | 0.649 | 0.649 | 0.684 | 0.663 |
| Er |  |  | 1.813 | 1.898 | 1.913 | 1.970 | 1.958 |
| Tm |  |  | 0.299 | 0.316 | 0.301 | 0.295 | 0.330 |
| Yb |  |  | 1.834 | 1.873 | 1.828 | 1.928 | 2.023 |
| Lu |  |  | 0.297 | 0.300 | 0.285 | 0.293 | 0.326 |
| Cr |  |  | 949.663 | 1261.149 | 1235.402 | 1170.069 | 1355.043 |
| Sc |  |  | 48.471 | 41.670 | 40.730 | 39.999 | 44.405 |
| Rb |  |  | 18.817 | 23.463 | 24.689 | 21.511 | 25.520 |
| Ba |  |  | 160.325 | 210.406 | 122.724 | 106.044 | 131.281 |
| U |  |  | 0.361 | 0.367 | 0.367 | 0.410 | 0.345 |
| Sr |  |  | 170.159 | 159.314 | 155.484 | 148.983 | 140.563 |
|  |  |  |  |  |  |  |  |
| Residual modes |  |  |  |  |  |  |  |
| TMF |  |  | 0.550 | 0.650 | 0.650 | 0.630 | 0.400 |
| Olivine |  |  | 0.450 | 0.350 | 0.350 | 0.370 | 0.600 |


| INTRUSION | LPS | LPS | LPS | LPS | LPS | LPS | LPS | LPS |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| ZONE | OZ | $\begin{aligned} & \text { OZ ROOF- } \\ & 1 \end{aligned}$ | $\begin{aligned} & \text { OZ ROOF- } \\ & 2 \end{aligned}$ | CPZ | CPZ | DZ | DZ | DZ |
| LITHOLOGY | ol-gabbro | ol-gabbro | ol-gabbro | cpx-rich | cpx-rich | diabase | diabase | diabase |
| HEIGHT (m) | 5.450 | 7.280 | 7.280 | 8.000 | 8.970 | 9.570 | 11.110 | 12.070 |
| SAMPLE I.D. | $\begin{aligned} & \text { 10BH167A } \\ & 3 \end{aligned}$ | $\begin{aligned} & \text { 10BH167A } \\ & 4 \end{aligned}$ | $\begin{aligned} & \text { 11BH311B } \\ & 3 \end{aligned}$ | $\begin{aligned} & \text { 11BH311B } \\ & 4 \end{aligned}$ | $\begin{aligned} & \text { 10BH167A } \\ & 5 \end{aligned}$ | $\begin{aligned} & \text { 10BH167A } \\ & 6 \end{aligned}$ | $\begin{aligned} & \text { 10BH167A } \\ & 7 \end{aligned}$ | $\begin{aligned} & \text { 10BH167A } \\ & 8 \end{aligned}$ |
| Lab | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff |
| SiO 2 | 41.605 | 42.723 | 43.952 | 49.942 | 48.634 | 47.851 | 47.446 | 48.105 |
| TiO2 | 0.549 | 0.693 | 0.637 | 0.902 | 0.944 | 1.023 | 1.052 | 0.964 |
| Al2O3 | 6.563 | 8.070 | 7.897 | 10.551 | 12.075 | 13.590 | 14.118 | 14.386 |
| Fe2O3T | 14.137 | 13.992 | 14.431 | 10.109 | 11.002 | 11.962 | 12.025 | 10.712 |
| FeO* | 12.709 | 12.579 | 12.973 | 9.088 | 9.891 | 10.754 | 10.811 | 9.630 |
| Fe2O3* | 1.428 | 1.413 | 1.458 | 1.021 | 1.111 | 1.208 | 1.215 | 1.082 |
| MnO | 0.197 | 0.193 | 0.202 | 0.167 | 0.164 | 0.193 | 0.169 | 0.167 |
| MgO | 25.433 | 21.827 | 20.523 | 11.453 | 10.565 | 8.632 | 8.579 | 8.625 |
| CaO | 6.209 | 8.167 | 8.355 | 14.110 | 12.088 | 11.647 | 11.542 | 10.824 |
| Na 2 O | 0.405 | 0.812 | 0.781 | 1.264 | 1.506 | 1.841 | 1.758 | 1.831 |
| K2O | 0.270 | 0.306 | 0.260 | 0.847 | 1.023 | 1.044 | 1.352 | 1.977 |
| P2O5 | 0.051 | 0.064 | 0.052 | 0.064 | 0.067 | 0.072 | 0.084 | 0.076 |
| LOI | 5.527 | 2.820 | 3.085 | 1.901 | 2.012 | 2.701 | 2.204 | 3.311 |
| SUM | 100.947 | 99.668 | 100.174 | 101.311 | 100.079 | 100.558 | 100.330 | 100.979 |


|  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| ICP-OES (ppm) |  |  |  |  |  |  |  |  |
| Sc | 24.136 | 30.251 | 31.452 | 56.368 | 49.546 | 42.598 | 41.397 | 40.741 |
| V | 166.999 | 205.681 | 200.964 | 363.709 | 342.559 | 346.740 | 343.431 | 321.218 |
| Cr | 1591.642 | 1244.937 | 1704.769 | 1945.740 | 1023.784 | 324.874 | 268.965 | 292.890 |
| Co | 64.578 | 64.637 | 61.531 | 49.618 | 50.940 | 58.931 | 55.089 | 50.374 |
| Ni | 622.833 | 512.994 | 562.236 | 124.712 | 80.923 | 56.405 | 87.370 | 68.101 |
| Cu | 89.048 | 92.100 | 63.983 | 87.659 | 95.804 | 104.588 | 108.387 | 91.037 |
| Zn | 119.319 | 123.834 | 55.870 | 54.004 | 80.284 | 61.370 | 75.458 | 43.058 |
| Sr | 58.729 | 91.154 | 80.446 | 166.629 | 206.172 | 234.867 | 305.276 | 318.651 |
| Y | 8.764 | 11.556 | 10.462 | 18.379 | 18.535 | 18.238 | 21.050 | 18.858 |
| Zr | 42.879 | 54.872 | 46.906 | 68.792 | 79.921 | 76.238 | 82.355 | 80.722 |
| Ba | 30.026 | 87.394 | 4.585 | 105.062 | 81.940 | 79.461 | 221.917 | 179.504 |
|  |  |  |  |  |  |  |  |  |
| ICP-MS |  |  |  |  |  |  |  |  |
| Ti | 0.549 | 0.718 | 0.674 | 0.894 | 0.941 | 1.033 | 1.037 | 0.998 |
| V | 169.400 | 215.700 | 203.000 | 345.300 | 327.700 | 336.100 | 330.700 | 301.200 |
| Cr | 1618.300 | 1233.900 | 1655.900 | 1988.700 | 992.200 | 328.800 | 263.400 | 276.800 |
| Co | 68.400 | 63.600 | 58.500 | 50.300 | 55.900 | 54.100 | 52.200 | 51.400 |
| Ni | 621.300 | 513.600 | 579.300 | 114.300 | 79.000 | 55.600 | 85.400 | 64.500 |
| Cu | 62.200 | 86.900 | 57.800 | 86.500 | 91.600 | 106.400 | 105.300 | 90.600 |
| Zn | 128.500 | 137.300 | 73.600 | 49.600 | 80.300 | 58.800 | 67.200 | 38.800 |
| Rb | 11.800 | 12.700 | 9.400 | 23.800 | 28.100 | 31.400 | 37.400 | 70.800 |
| K | 2698.490 | 3061.850 | 2596.980 | 8474.670 | 10227.550 | 10444.530 | 13523.710 | 19766.510 |
| Sr | 59.300 | 92.100 | 75.200 | 166.700 | 200.100 | 228.400 | 277.700 | 302.300 |
| P | 514.886 | 639.414 | 516.623 | 640.057 | 671.018 | 718.881 | 841.732 | 758.709 |
| Y | 9.900 | 13.300 | 11.200 | 18.100 | 18.600 | 19.000 | 20.200 | 19.200 |
| Zr | 45.300 | 57.900 | 42.200 | 69.000 | 75.900 | 78.800 | 80.000 | 80.100 |
| Nb | 1.780 | 2.640 | 1.810 | 3.260 | 3.670 | 3.860 | 3.890 | 4.060 |
| Sn | 0.350 | 0.910 | 0.220 | 0.930 | 0.790 | 1.080 | 1.840 | 0.550 |
| Cs | 1.320 | 0.770 | 0.710 | 0.680 | 0.770 | 0.800 | 0.700 | 0.580 |
| Ba | 41.700 | 93.900 | 40.600 | 121.000 | 95.500 | 91.000 | 231.100 | 187.400 |
| La | 3.680 | 4.880 | 3.990 | 5.830 | 6.220 | 6.590 | 7.460 | 6.900 |
| Ce | 8.380 | 11.240 | 9.110 | 13.930 | 14.590 | 15.260 | 17.440 | 16.270 |
| Pr | 1.160 | 1.560 | 1.250 | 1.890 | 2.010 | 2.080 | 2.400 | 2.220 |
| Nd | 5.140 | 6.750 | 5.490 | 8.420 | 8.820 | 9.110 | 10.390 | 9.660 |
| Sm | 1.430 | 1.850 | 1.390 | 2.290 | 2.350 | 2.500 | 2.740 | 2.460 |
| Eu | 0.450 | 0.590 | 0.490 | 0.720 | 0.790 | 0.970 | 0.890 | 0.850 |
| Gd | 1.460 | 1.960 | 1.640 | 2.490 | 2.600 | 2.740 | 3.010 | 2.710 |
| Tb | 0.250 | 0.310 | 0.260 | 0.450 | 0.440 | 0.480 | 0.530 | 0.460 |
| Dy | 1.680 | 2.140 | 1.860 | 2.880 | 2.900 | 3.190 | 3.250 | 3.110 |
| Ho | 0.300 | 0.390 | 0.340 | 0.540 | 0.570 | 0.590 | 0.590 | 0.570 |
| Er | 0.920 | 1.210 | 0.980 | 1.590 | 1.650 | 1.720 | 1.810 | 1.680 |
| Tm | 0.140 | 0.190 | 0.150 | 0.250 | 0.240 | 0.260 | 0.280 | 0.260 |
| Yb | 0.870 | 1.190 | 1.040 | 1.510 | 1.570 | 1.680 | 1.720 | 1.660 |
| Lu | 0.150 | 0.180 | 0.160 | 0.250 | 0.250 | 0.260 | 0.270 | 0.260 |
| Hf | 1.020 | 1.330 | 0.940 | 1.680 | 1.760 | 2.050 | 2.220 | 2.010 |
| Ta | 0.140 | 0.170 | 0.150 | 0.220 | 0.250 | 0.270 | 0.290 | 0.270 |
| Pb | 1.870 | 2.770 | 1.770 | 1.850 | 1.960 | 2.760 | 2.400 | 2.210 |
| Th | 0.790 | 0.950 | 0.810 | 1.250 | 1.310 | 1.420 | 1.560 | 1.450 |
| U | 0.160 | 0.200 | 0.170 | 0.240 | 0.280 | 0.280 | 0.340 | 0.300 |
|  |  |  |  |  |  |  |  |  |
| Inversion models |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |
| Fitted to MgO (wt \%) of > | 8.000 | 8.000 | 8.000 |  |  |  |  |  |
| Model melts in ppm |  |  |  |  |  |  |  |  |
| Th | 1.946 | 1.573 | 1.604 |  |  |  |  |  |
| Nb | 4.437 | 4.394 | 3.613 |  |  |  |  |  |
| Ta | 0.333 | 0.277 | 0.290 |  |  |  |  |  |
| La | 9.159 | 8.117 | 7.956 |  |  |  |  |  |
| Ce | 20.825 | 18.684 | 18.147 |  |  |  |  |  |
| Pr | 2.878 | 2.591 | 2.488 |  |  |  |  |  |
| P | 1264.455 | 1057.231 | 1020.994 |  |  |  |  |  |
| Nd | 12.660 | 11.175 | 10.871 |  |  |  |  |  |
| Sm | 3.471 | 3.043 | 2.725 |  |  |  |  |  |
| Zr | 109.951 | 95.230 | 82.745 |  |  |  |  |  |
| Hf | 2.476 | 2.188 | 1.843 |  |  |  |  |  |
| Ti | 1.334 | 1.181 | 1.321 |  |  |  |  |  |


| Eu | 1.092 | 0.970 | 0.961 |  |  |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| Gd | 3.493 | 3.203 | 3.184 |  |  |  |  |  |
| Tb | 0.594 | 0.505 | 0.502 |  |  |  |  |  |
| Dy | 3.996 | 3.488 | 3.598 |  |  |  |  |  |
| Y | 23.583 | 21.689 | 21.684 |  |  |  |  |  |
| Ho | 0.711 | 0.634 | 0.656 |  |  |  |  |  |
| Er | 2.170 | 1.964 | 1.885 |  |  |  |  |  |
| Tm | 0.330 | 0.308 | 0.288 |  |  |  |  |  |
| Yb | 2.023 | 1.919 | 1.981 |  |  |  |  |  |
| Lu | 0.349 | 0.290 | 0.305 |  |  |  |  |  |
| Cr | 1407.217 | 1121.727 | 1471.911 |  |  |  |  |  |
| Sc | 42.048 | 42.250 | 48.762 |  |  |  |  |  |
| Rb | 29.237 | 21.082 | 18.688 |  |  |  |  |  |
| Ba | 102.709 | 155.464 | 80.396 |  |  |  |  |  |
| U | 0.394 | 0.331 | 0.337 |  |  |  |  |  |
| Sr | 146.492 | 152.686 | 149.206 |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |
| Residual modes |  |  |  |  |  |  |  |  |
| TMF | 0.400 | 0.600 | 0.500 |  |  |  |  |  |
| Olivine | 0.600 | 0.400 | 0.500 |  |  |  |  |  |


| $\begin{aligned} & \text { INTRUSIO } \\ & \mathbf{N} \\ & \hline \end{aligned}$ | LPS | LPS | LPS | LPS | LPS | LPS | LPS | LPS |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| ZONE | DZ | DZ | DZ | DZ | DZ | UBZ | UBZ | UCZ-1 |
| $\begin{aligned} & \text { LITHOLOG } \\ & \text { Y } \end{aligned}$ | diabase | diabase | diabase | diabase | diabase | fg diabase | fg diabase | aphanitic |
| $\begin{aligned} & \text { HEIGHT } \\ & (\mathrm{m}) \end{aligned}$ | 12.710 | 14.000 | 15.700 | 16.880 | 19.700 | 20.680 | 20.880 | 21.430 |
| SAMPLE I.D. | $\begin{aligned} & \text { 10BH167A } \\ & 9 \\ & \hline \end{aligned}$ | 11BH311B $6$ | $\begin{aligned} & \text { 10BH167A1 } \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & \text { 10BH167A1 } \\ & 1 \end{aligned}$ | $\begin{aligned} & \text { 10BH167A1 } \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & \text { 10BH167A1 } \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & \text { 11BH311B } \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \text { 10BH167A1 } \\ & 5 \\ & \hline \end{aligned}$ |
| Lab | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff |
| SiO 2 | 46.588 | 47.884 | 46.621 | 47.954 | 47.146 | 48.608 | 48.341 | 45.701 |
| TiO2 | 1.237 | 1.184 | 1.172 | 1.026 | 1.067 | 1.022 | 1.079 | 1.007 |
| Al2O3 | 13.362 | 14.243 | 13.506 | 12.884 | 13.068 | 12.871 | 13.084 | 12.240 |
| Fe2O3T | 13.809 | 12.215 | 11.680 | 12.291 | 12.258 | 12.561 | 12.206 | 12.274 |
| FeO* | 12.414 | 10.981 | 10.501 | 11.049 | 11.020 | 11.292 | 10.973 | 11.034 |
| Fe2O3* | 1.395 | 1.234 | 1.180 | 1.241 | 1.238 | 1.269 | 1.233 | 1.240 |
| MnO | 0.178 | 0.185 | 0.172 | 0.195 | 0.185 | 0.189 | 0.187 | 0.190 |
| MgO | 8.011 | 8.460 | 7.922 | 8.852 | 8.812 | 9.178 | 8.681 | 10.491 |
| CaO | 9.813 | 11.261 | 11.291 | 10.265 | 11.031 | 12.058 | 12.036 | 11.510 |
| Na 2 O | 2.398 | 1.778 | 2.014 | 1.665 | 1.516 | 1.601 | 1.635 | 1.901 |
| K2O | 1.170 | 1.258 | 1.066 | 1.612 | 1.547 | 0.983 | 1.131 | 0.877 |
| P2O5 | 0.090 | 0.085 | 0.090 | 0.084 | 0.089 | 0.085 | 0.086 | 0.083 |
| LOI | 4.087 | 2.203 | 3.525 | 3.125 | 2.613 | 2.036 | 2.395 | 2.721 |
| SUM | 100.742 | 100.756 | 99.060 | 99.953 | 99.332 | 101.193 | 100.861 | 111.270 |
| Sc | 40.415 | 41.333 | 39.927 | 41.081 | 40.976 | 41.233 | 42.139 | 39.586 |
| V | 407.444 | 372.683 | 346.257 | 298.195 | 311.704 | 328.094 | 317.668 | 332.612 |
| Cr | 153.473 | 266.179 | 211.518 | 147.432 | 172.221 | 615.423 | 609.521 | 782.803 |
| Co | 66.124 | 57.052 | 57.519 | 54.510 | 55.329 | 58.376 | 52.496 | 59.896 |
| Ni | 80.701 | 61.527 | 75.320 | 66.057 | 62.455 | 91.307 | 81.361 | 132.918 |
| Cu | 102.888 | 108.183 | 152.625 | 95.982 | 104.134 | 128.052 | 122.438 | 112.439 |
| Zn | 48.001 | 116.808 | 44.137 | 39.167 | 54.472 | 76.399 | 28.256 | 59.533 |
| Sr | 240.897 | 261.357 | 247.229 | 224.427 | 232.326 | 231.139 | 226.091 | 203.720 |
| Y | 20.961 | 21.448 | 21.537 | 18.249 | 18.882 | 18.286 | 19.160 | 19.067 |
| Zr | 81.169 | 81.136 | 92.382 | 88.942 | 73.837 | 77.295 | 75.579 | 87.164 |
| Ba | 140.940 | 143.081 | 112.716 | 211.527 | 161.288 | 125.344 | 174.296 | 109.987 |
|  |  |  |  |  |  |  |  |  |
| Ti | 1.268 | 1.192 | 1.178 | 1.052 | 1.037 | 1.055 | 1.058 | 0.982 |
| V | 390.800 | 350.200 | 333.100 | 309.500 | 318.800 | 347.200 | 315.900 | 319.300 |
| Cr | 157.100 | 249.300 | 211.700 | 148.300 | 162.100 | 590.500 | 586.300 | 779.900 |
| Co | 70.300 | 55.600 | 53.100 | 56.000 | 54.500 | 58.700 | 56.800 | 60.100 |
| Ni | 75.800 | 62.800 | 73.900 | 61.100 | 65.900 | 85.700 | 83.600 | 128.100 |
| Cu | 96.900 | 108.500 | 152.200 | 107.200 | 110.000 | 133.500 | 124.100 | 119.200 |
| Zn | 47.900 | 124.600 | 37.300 | 36.900 | 54.200 | 75.500 | 51.300 | 57.300 |
| Rb | 39.800 | 37.200 | 35.400 | 52.500 | 40.800 | 34.300 | 27.200 | 9.900 |
| K | 11704.800 | 12584.770 | 10655.670 | 16124.630 | 15465.690 | 9834.860 | 11313.220 | 8774.260 |


| Sr | 222.400 | 254.400 | 242.300 | 232.300 | 234.700 | 232.000 | 229.900 | 209.100 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| P | 896.020 | 853.701 | 896.202 | 836.117 | 890.748 | 854.826 | 861.778 | 828.765 |
| Y | 21.000 | 21.900 | 22.700 | 21.200 | 20.200 | 20.000 | 21.300 | 20.400 |
| Zr | 85.300 | 81.300 | 90.000 | 86.500 | 79.300 | 77.600 | 80.500 | 85.900 |
| Nb | 4.440 | 4.000 | 4.800 | 4.320 | 3.920 | 3.550 | 4.290 | 3.960 |
| Sn | 1.010 | 0.990 | 0.940 | 0.830 | 0.630 | 1.210 | 1.070 | 1.230 |
| Cs | 0.360 | 1.110 | 0.370 | 0.840 | 1.280 | 1.810 | 0.540 | 0.370 |
| Ba | 146.500 | 157.000 | 121.600 | 211.800 | 157.400 | 127.900 | 195.000 | 126.200 |
| La | 7.550 | 7.940 | 7.970 | 7.640 | 7.530 | 7.920 | 8.000 | 7.330 |
| Ce | 17.340 | 18.340 | 19.100 | 17.600 | 17.150 | 17.840 | 18.530 | 17.280 |
| Pr | 2.340 | 2.510 | 2.590 | 2.420 | 2.330 | 2.410 | 2.540 | 2.340 |
| Nd | 10.330 | 10.640 | 11.510 | 10.500 | 10.370 | 10.400 | 10.780 | 10.410 |
| Sm | 2.880 | 2.820 | 2.940 | 2.740 | 2.740 | 2.780 | 2.810 | 2.850 |
| Eu | 1.130 | 1.020 | 1.030 | 0.930 | 0.910 | 0.910 | 0.960 | 0.880 |
| Gd | 2.980 | 3.250 | 3.370 | 3.060 | 2.980 | 2.980 | 3.040 | 3.050 |
| Tb | 0.530 | 0.530 | 0.620 | 0.500 | 0.520 | 0.500 | 0.540 | 0.530 |
| Dy | 3.430 | 3.520 | 3.720 | 3.260 | 3.320 | 3.240 | 3.450 | 3.390 |
| Ho | 0.630 | 0.660 | 0.690 | 0.590 | 0.600 | 0.580 | 0.610 | 0.610 |
| Er | 1.870 | 1.900 | 2.000 | 1.800 | 1.790 | 1.750 | 1.840 | 1.780 |
| Tm | 0.300 | 0.290 | 0.310 | 0.280 | 0.280 | 0.280 | 0.290 | 0.280 |
| Yb | 1.890 | 1.800 | 2.040 | 1.850 | 1.730 | 1.760 | 1.810 | 1.750 |
| Lu | 0.290 | 0.300 | 0.310 | 0.280 | 0.290 | 0.270 | 0.280 | 0.270 |
| Hf | 2.190 | 2.080 | 2.390 | 2.070 | 2.030 | 1.900 | 2.030 | 2.180 |
| Ta | 0.300 | 0.320 | 0.350 | 0.280 | 0.260 | 0.290 | 0.310 | 0.290 |
| Pb | 1.560 | 1.740 | 1.720 | 1.610 | 1.300 | 2.630 | 2.430 | 3.350 |
| Th | 1.630 | 1.720 | 1.850 | 1.560 | 1.430 | 1.690 | 1.700 | 1.800 |
| U | 0.350 | 0.350 | 0.370 | 0.310 | 0.320 | 0.370 | 0.340 | 0.360 |
|  |  |  |  |  |  |  |  |  |


| INTRUSION | LPS | LPS | LPS | WUS | WUS | $\begin{aligned} & \hline \text { UHUK } \\ & \text { MASSIF } \end{aligned}$ | $\begin{aligned} & \hline \text { UHUK } \\ & \text { MASSIF } \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| ZONE | UCZ-2 | LCZ | UCZ |  |  |  |  |
| LITHOLOGY | aphanitic | aphanitic | aphanitic | chill | chill | small dike | small dike |
| HEIGHT (m) | 21.430 | 0.000 | 0.000 |  |  |  |  |
| $\begin{aligned} & \text { SAMPLE } \\ & \text { I.D. } \end{aligned}$ | 11BH311B8 | 11JB327B1 | 11JB328B1 | 10NW037C1 | 10NW038A1 | 10JB160B1 | 10JB160B2 |
| Lab | Cardiff | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec |
| SiO 2 | 46.081 | 48.278 | 48.000 | 48.259 | 47.823 | 48.083 | 47.476 |
| TiO2 | 1.020 | 1.023 | 1.048 | 0.989 | 0.931 | 0.989 | 0.972 |
| Al2O3 | 12.529 | 12.763 | 12.776 | 13.144 | 12.956 | 13.219 | 13.078 |
| Fe2O3T | 11.983 | 11.135 | 11.329 | 11.812 | 12.161 | 11.153 | 11.091 |
| FeO* | 10.773 | 10.010 | 10.185 | 10.619 | 10.932 | 10.027 | 9.971 |
| Fe2O3* | 1.210 | 1.125 | 1.144 | 1.193 | 1.228 | 1.126 | 1.120 |
| MnO | 0.151 | 0.147 | 0.152 | 0.158 | 0.169 | 0.149 | 0.145 |
| MgO | 10.883 | 10.743 | 10.323 | 10.465 | 10.515 | 9.413 | 9.413 |
| CaO | 10.263 | 12.255 | 10.159 | 11.193 | 12.067 | 12.287 | 11.707 |
| Na 2 O | 2.569 | 1.526 | 2.451 | 1.806 | 1.647 | 1.921 | 2.029 |
| K2O | 0.603 | 0.557 | 0.500 | 1.097 | 0.632 | 1.059 | 1.548 |
| P2O5 | 0.083 | 0.087 | 0.092 | 0.081 | 0.068 | 0.079 | 0.082 |
| LOI | 2.965 | 1.443 | 2.673 | 1.818 | 1.513 | 1.204 | 2.131 |
| SUM | 111.113 | 100.254 | 99.777 | 101.178 | 100.942 | 99.841 | 100.159 |
|  |  |  |  |  |  |  |  |
| ICP-OES (ppm) |  |  |  |  |  |  |  |
| Sc | 41.232 | 38.906 | 39.497 | 37.374 | 37.969 | 37.621 | 37.191 |
| V | 335.438 | 299.460 | 306.207 |  |  | 279.057 | 278.312 |
| Cr | 734.756 | 660.194 | 688.506 |  |  | 511.276 | 506.349 |
| Co | 55.411 | 45.109 | 52.274 |  |  | 40.414 | 46.505 |
| Ni | 260.662 | 218.871 | 200.659 |  |  | 148.769 | 157.283 |
| Cu | 120.053 | 133.257 | 148.530 |  |  | 78.401 | 82.513 |
| Zn | 98.873 | 42.689 | 52.096 |  |  | 56.134 | 48.260 |
| Sr | 273.861 | 140.284 | 251.900 |  |  | 462.455 | 1024.056 |
| Y | 20.095 | 19.000 | 19.000 |  |  | 20.000 | 18.000 |
| Zr | 82.760 | 72.000 | 65.000 |  |  | 66.200 | 102.000 |
| Ba | 88.756 | 132.823 | 109.494 |  |  | 182.506 | 357.273 |
|  |  |  |  |  |  |  |  |
| ICP-MS |  |  |  |  |  |  |  |


| Ti | 1.032 | \#REF! | \#REF! | \#REF! | \#REF! | \#REF! | \#REF! |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| V | 336.300 | 299.460 | 306.207 | 278.632 | 283.854 |  |  |
| Cr | 723.500 | 660.194 | 688.506 | 672.582 | 696.834 |  |  |
| Co | 60.100 | 45.109 | 52.274 | 43.390 | 52.983 |  |  |
| Ni | 246.800 | 218.871 | 200.659 | 212.890 | 230.816 |  |  |
| Cu | 125.200 | 133.257 | 148.530 | 75.519 | 135.193 |  |  |
| Zn | 95.400 | 42.689 | 52.096 | 55.387 | 85.043 |  |  |
| Rb | 15.100 | 24.000 | 13.300 | 28.000 | 20.700 | 36.600 | 44.200 |
| K | 6034.240 | 4621.826 | 4149.848 | 9110.821 | 5247.524 | 8795.409 | 12850.823 |
| Sr | 252.100 | 140.284 | 251.900 | 477.476 | 311.910 | 462.455 | 1024.056 |
| P | 826.519 | 381.632 | 403.103 | 353.054 | 296.195 | 4916.171 | 4888.903 |
| Y | 19.900 | 19.000 | 19.000 | 17.800 | 16.700 | 20.000 | 18.000 |
| Zr | 82.900 | 72.000 | 65.000 | 60.500 | 52.000 | 66.200 | 102.000 |
| Nb | 4.190 | 4.600 | 4.580 | 4.510 | 4.160 | 5.140 | 5.030 |
| Sn | 0.940 |  |  |  |  |  |  |
| Cs | 0.600 | 1.320 | 0.600 | 1.490 | 1.210 | 2.100 | 8.900 |
| Ba | 103.000 | 132.823 | 109.494 | 290.650 | 130.420 | 182.506 | 357.273 |
| La | 6.800 | 6.800 | 6.400 | 7.200 | 6.700 | 8.400 | 7.200 |
| Ce | 16.330 | 16.300 | 16.100 | 15.500 | 13.900 | 17.200 | 17.100 |
| Pr | 2.250 | 2.500 | 2.400 | 2.300 | 2.000 | 2.600 | 2.500 |
| Nd | 9.990 | 10.700 | 10.500 | 10.400 | 10.200 | 12.200 | 10.600 |
| Sm | 2.750 | 2.700 | 2.800 | 3.000 | 2.500 | 3.000 | 2.800 |
| Eu | 0.740 | 0.870 | 0.770 | 0.820 | 0.780 | 0.970 | 0.790 |
| Gd | 2.950 | 3.180 | 3.110 | 2.810 | 2.510 | 3.050 | 2.980 |
| Tb | 0.510 | 0.540 | 0.520 | 0.530 | 0.470 | 0.560 | 0.530 |
| Dy | 3.260 | 3.200 | 3.200 | 3.030 | 3.130 | 3.520 | 3.100 |
| Ho | 0.590 | 0.680 | 0.650 | 0.600 | 0.560 | 0.700 | 0.630 |
| Er | 1.710 | 1.840 | 1.860 | 1.680 | 1.700 | 1.960 | 1.860 |
| Tm | 0.270 | 0.260 | 0.260 | 0.250 | 0.220 | 0.280 | 0.250 |
| Yb | 1.750 | 1.750 | 1.740 | 1.700 | 1.610 | 1.810 | 1.720 |
| Lu | 0.270 | 0.260 | 0.250 | 0.230 | 0.230 | 0.270 | 0.250 |
| Hf | 2.040 | 1.870 | 1.790 | 1.790 | 1.600 | 1.860 | 2.610 |
| Ta | 0.280 | $<0.3$ | 0.300 | $<0.3$ | $<0.3$ | 0.330 | 0.350 |
| Pb | 1.730 | 1.510 | 1.060 | 2.000 | 3.620 | 3.250 | 4.500 |
| Th | 1.560 | 0.600 | 0.700 | 1.620 | 1.300 | 2.000 | $<0.7$ |
| U | 0.340 | 0.210 | 0.200 | 0.270 | 0.190 | 0.450 | 0.400 |
|  |  |  |  |  |  |  |  |


| INTRUSION | PS | PS | PS | PS | PS | PS | PS |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| ZONE | LBZ | LBZ | LBZ | OZ | OZ | OZ | OZ |
| LITHOLOGY |  |  |  |  |  |  |  |
| HEIGHT (m) |  |  |  |  |  |  |  |
| SAMPLE I.D. | 10JB287A1 | 10JB367A1 | 10JB367B1 | 10JB367C1 | 10JB367E1 | 10JB367G1 | 10JB367H1 |
| Lab | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec |
|  |  |  |  |  |  |  |  |
| SiO 2 | 46.709 | 46.815 | 45.891 | 46.265 | 44.798 | 45.248 | 44.889 |
| TiO2 | 0.761 | 0.861 | 0.776 | 0.781 | 0.677 | 0.699 | 0.656 |
| Al2O3 | 9.974 | 11.111 | 9.780 | 9.629 | 8.673 | 8.664 | 8.528 |
| Fe2O3T | 12.900 | 12.875 | 13.023 | 13.144 | 13.075 | 13.368 | 13.291 |
| FeO* | 11.597 | 11.575 | 11.708 | 11.816 | 11.754 | 12.017 | 11.948 |
| Fe2O3* | 1.303 | 1.300 | 1.315 | 1.327 | 1.321 | 1.350 | 1.342 |
| MnO | 0.189 | 0.188 | 0.187 | 0.188 | 0.185 | 0.189 | 0.190 |
| MgO | 17.827 | 15.637 | 17.911 | 18.460 | 20.759 | 21.581 | 22.083 |
| CaO | 8.488 | 9.746 | 8.748 | 8.410 | 7.455 | 7.661 | 7.646 |
| Na 2 O | 1.110 | 1.120 | 1.001 | 0.997 | 0.872 | 0.809 | 0.750 |
| K2O | 0.480 | 0.470 | 0.448 | 0.422 | 0.368 | 0.283 | 0.280 |
| P2O5 | 0.064 | 0.061 | 0.063 | 0.061 | 0.049 | 0.053 | 0.054 |
| LOI | 1.940 | 1.498 | 2.104 | 2.260 | 2.633 | 2.353 | 2.552 |
| SUM | 100.724 | 100.660 | 100.216 | 100.902 | 99.814 | 101.201 | 101.258 |
|  |  |  |  |  |  |  |  |
| $\begin{aligned} & \text { ICP-OES } \\ & \text { (ppm) } \\ & \hline \end{aligned}$ |  |  |  |  |  |  |  |
| Sc | 30.049 | 33.213 | 30.189 | 29.959 | 26.980 | 27.686 | 27.543 |
| V |  |  |  |  |  |  |  |
| Cr |  |  |  |  |  |  |  |
| Co |  |  |  |  |  |  |  |
| Ni |  |  |  |  |  |  |  |


| Cu |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Zn |  |  |  |  |  |  |  |
| Sr |  |  |  |  |  |  |  |
| Y |  |  |  |  |  |  |  |
| Zr |  |  |  |  |  |  |  |
| Ba |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |
| ICP-MS |  |  |  |  |  |  |  |
| Ti | 0.761 | 0.861 | 0.776 | 0.782 | 0.677 | 0.700 | 0.656 |
| V | 220.529 | 249.049 | 231.223 | 229.306 | 194.702 | 206.674 | 206.186 |
| Cr | 976.403 | 874.433 | 1033.881 | 1019.141 | 888.763 | 1206.526 | 1520.895 |
| Co | 67.746 | 61.931 | 72.595 | 72.723 | 77.503 | 81.584 | 85.347 |
| Ni | 511.649 | 549.269 | 525.578 | 566.507 | 671.256 | 685.003 | 732.561 |
| Cu | 105.870 | 121.532 | 120.388 | 103.780 | 100.425 | 94.949 | 116.600 |
| Zn | 82.583 | 90.330 | 69.791 | 91.370 | 93.100 | 64.989 | 69.587 |
| Rb | 18.500 | 17.700 | 14.700 | 14.600 | 12.400 | 9.400 | 9.500 |
| K | 3988.100 | 3902.579 | 3715.648 | 3501.683 | 3058.421 | 2350.845 | 2324.680 |
| Sr | 115.942 | 126.808 | 112.429 | 106.760 | 94.352 | 90.789 | 87.993 |
| P | 277.716 | 268.193 | 275.032 | 267.844 | 215.876 | 231.351 | 234.740 |
| Y | 15.000 | 14.600 | 14.300 | 15.000 | 11.600 | 12.500 | 12.500 |
| Zr | 51.000 | 52.000 | 58.000 | 62.000 | 47.000 | 43.000 | 42.000 |
| Nb | 3.660 | 3.690 | 3.650 | 4.000 | 3.170 | 2.860 | 2.920 |
| Sn |  |  |  |  |  |  |  |
| Cs | 1.130 | 0.500 | 0.700 | 0.500 | 0.130 | 1.260 | 0.400 |
| Ba | 86.832 | 85.240 | 80.287 | 81.807 | 70.184 | 59.686 | 55.531 |
| La | 7.000 | 6.400 | 6.500 | 7.100 | 5.700 | 5.000 | 4.900 |
| Ce | 14.700 | 13.200 | 14.000 | 15.600 | 11.500 | 10.900 | 10.800 |
| Pr | 2.000 | 2.000 | 1.900 | 2.000 | 1.730 | 1.510 | 1.530 |
| Nd | 9.400 | 9.600 | 8.040 | 8.540 | 8.180 | 7.300 | 6.490 |
| Sm | 2.300 | 2.500 | 2.100 | 2.200 | 2.000 | 1.900 | 1.720 |
| Eu | 0.740 | 0.820 | 0.670 | 0.720 | 0.660 | 0.590 | 0.620 |
| Gd | 2.370 | 2.480 | 2.430 | 2.530 | 1.980 | 1.890 | 2.110 |
| Tb | 0.410 | 0.410 | 0.410 | 0.430 | 0.330 | 0.340 | 0.350 |
| Dy | 2.680 | 2.920 | 2.400 | 2.400 | 2.390 | 2.160 | 2.100 |
| Ho | 0.510 | 0.580 | 0.530 | 0.540 | 0.450 | 0.430 | 0.460 |
| Er | 1.480 | 1.600 | 1.490 | 1.540 | 1.280 | 1.260 | 1.300 |
| Tm | 0.220 | 0.230 | 0.210 | 0.210 | 0.190 | 0.175 | 0.177 |
| Yb | 1.360 | 1.510 | 1.370 | 1.430 | 1.210 | 1.090 | 1.230 |
| Lu | 0.200 | 0.220 | 0.200 | 0.220 | 0.180 | 0.168 | 0.180 |
| Hf | 1.470 | 1.470 | 1.610 | 1.740 | 1.330 | 1.210 | 1.260 |
| Ta | < 0.3 | < 0.3 | <0.3 | <0.3 | < 0.3 | < 0.3 | <0.3 |
| Pb | 3.400 | 2.100 | 1.570 | 1.760 | 1.900 | 1.900 | 1.080 |
| Th | 1.620 | 1.470 | 0.800 | 0.900 | 1.200 | 1.000 | 1.100 |
| U | 0.250 | 0.240 | 0.260 | 0.310 | 0.210 | 0.131 | 0.210 |


| INTRUSIO <br> N | PS | PS | PS | PS | PS | PS | PS | PS |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| ZONE | DZ! | OZ | OZ | OZ | CPZ | DZ | DZ | DZ |
| $\begin{aligned} & \text { LITHOLOG } \\ & \text { Y } \end{aligned}$ |  |  |  |  |  |  |  |  |
| $\begin{aligned} & \text { HEIGHT } \\ & (\mathrm{m}) \end{aligned}$ |  |  |  |  |  |  |  |  |
| $\begin{aligned} & \text { SAMPLE } \\ & \text { I.D. } \end{aligned}$ | 10JB367J1 | 10JB368A1 | 10JB368B1 | 10JB368G1 | 10JB368D1 | 10JB290A1 | 10JB367I1 | 10JB369A1 |
| Lab | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec |
|  |  |  |  |  |  |  |  |  |
| SiO 2 | 49.063 | 43.495 | 43.499 | 43.900 | 49.124 | 49.882 | 43.433 | 48.644 |
| TiO2 | 1.179 | 0.616 | 0.702 | 0.681 | 0.880 | 1.215 | 0.606 | 1.259 |
| Al2O3 | 14.513 | 8.028 | 8.339 | 8.410 | 11.006 | 14.243 | 7.723 | 15.219 |
| Fe2O3T | 12.457 | 13.238 | 13.470 | 13.677 | 10.353 | 13.218 | 13.279 | 13.087 |
| FeO* | 11.199 | 11.901 | 12.110 | 12.295 | 9.307 | 11.883 | 11.938 | 11.765 |
| Fe2O3* | 1.258 | 1.337 | 1.360 | 1.381 | 1.046 | 1.335 | 1.341 | 1.322 |
| MnO | 0.177 | 0.186 | 0.195 | 0.197 | 0.183 | 0.197 | 0.187 | 0.185 |
| MgO | 7.023 | 23.150 | 20.424 | 20.806 | 11.375 | 7.270 | 23.336 | 6.077 |
| CaO | 10.355 | 7.081 | 7.577 | 7.732 | 13.263 | 10.799 | 6.869 | 9.395 |
| Na 2 O | 1.748 | 0.645 | 0.793 | 0.801 | 1.183 | 1.740 | 0.456 | 1.887 |
| K2O | 1.537 | 0.314 | 0.417 | 0.386 | 1.071 | 1.335 | 0.354 | 1.470 |


| P2O5 | 0.097 | 0.045 | 0.058 | 0.055 | 0.062 | 0.104 | 0.042 | 0.109 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| LOI | 2.176 | 3.470 | 4.251 | 3.477 | 1.815 | 1.448 | 4.005 | 2.689 |
| SUM | 100.545 | 100.756 | 100.084 | 100.483 | 100.640 | 101.637 | 100.798 | 100.253 |
| ICP-OES (ppm) |  |  |  |  |  |  |  |  |
| Sc | 37.406 | 25.140 | 28.660 | 28.823 | 50.667 | 38.651 | 25.057 | 31.949 |
| V |  |  |  |  |  |  |  |  |
| Cr |  |  |  |  |  |  |  |  |
| Co |  |  |  |  |  |  |  |  |
| Ni |  |  |  |  |  |  |  |  |
| Cu |  |  |  |  |  |  |  |  |
| $\mathrm{Zn}$ |  |  |  |  |  |  |  |  |
| Sr |  |  |  |  |  |  |  |  |
| Y |  |  |  |  |  |  |  |  |
| Zr |  |  |  |  |  |  |  |  |
| Ba |  |  |  |  |  |  |  |  |
| ICP-MS |  |  |  |  |  |  |  |  |
|  | ICP-MS |  |  |  |  |  |  |  |
| Ti | 1.179 | 0.616 | 0.702 | 0.681 | 0.881 | 1.215 | 0.606 | 1.259 |
| V | 308.835 | 196.651 | 205.289 | 208.894 | 295.776 | 333.911 | 196.506 | 309.452 |
| Cr | 123.766 | 3126.391 | 1965.958 | 1973.188 | 1516.208 | 80.726 | 3203.112 | 55.127 |
| Co | 41.882 | 77.016 | 78.664 | 81.629 | 41.239 | 41.447 | 88.282 | 48.340 |
| Ni | 74.149 | 726.957 | 548.652 | 541.335 | 143.339 | 82.241 | 778.038 | 82.955 |
| Cu | 155.524 | 89.884 | 99.502 | 96.979 | 118.320 | 171.616 | 89.036 | 193.099 |
| Zn | 82.465 | 78.038 | 93.744 | 68.003 | 164.731 | 101.142 | 84.678 | 78.185 |
| Rb | 47.300 | 9.800 | 11.900 | 10.900 | 25.900 | 41.000 | 12.200 | 47.400 |
| K | 12757.895 | 2609.399 | 3458.906 | 3203.613 | 8892.019 | 11083.326 | 2940.067 | 12201.306 |
| Sr | 325.433 | 82.457 | 90.970 | 87.649 | 201.929 | 235.502 | 84.111 | 325.847 |
| P | 424.789 | 197.764 | 255.153 | 240.710 | 269.191 | 452.342 | 185.145 | 477.326 |
| Y | 22.100 | 10.900 | 11.800 | 12.600 | 15.400 | 21.000 | 10.400 | 24.000 |
| Zr | 81.300 | 37.000 | 43.000 | 44.000 | 52.000 | 82.000 | 38.000 | 91.000 |
| Nb | 5.600 | 2.420 | 2.990 | 2.780 | 3.500 | 4.910 | 2.600 | 5.850 |
| Sn |  |  |  |  |  |  |  |  |
| Cs | 0.980 | 0.640 | 0.260 | 0.980 | < 0.12 | 1.540 | 0.600 | 0.910 |
| Ba | 193.346 | 51.210 | 58.603 | 56.110 | 135.817 | 159.652 | 54.158 | 189.377 |
| La | 9.500 | 3.800 | 5.000 | 4.800 | 5.900 | 8.600 | 4.100 | 10.500 |
| Ce | 19.900 | 8.620 | 10.100 | 10.800 | 12.300 | 20.200 | 9.400 | 22.200 |
| Pr | 3.000 | 1.250 | 1.560 | 1.510 | 1.900 | 2.800 | 1.290 | 3.200 |
| Nd | 14.000 | 6.040 | 7.460 | 7.030 | 9.400 | 12.200 | 5.510 | 14.900 |
| Sm | 3.500 | 1.550 | 1.900 | 1.800 | 2.400 | 3.100 | 1.500 | 3.800 |
| Eu | 1.130 | 0.560 | 0.640 | 0.610 | 0.800 | 0.970 | 0.510 | 1.200 |
| Gd | 3.630 | 1.590 | 1.920 | 1.870 | 2.510 | 3.430 | 1.770 | 3.830 |
| Tb | 0.650 | 0.300 | 0.330 | 0.340 | 0.430 | 0.580 | 0.300 | 0.690 |
| Dy | 4.180 | 1.850 | 2.340 | 2.170 | 3.060 | 3.500 | 1.770 | 4.210 |
| Ho | 0.830 | 0.370 | 0.460 | 0.430 | 0.600 | 0.740 | 0.380 | 0.830 |
| Er | 2.300 | 1.050 | 1.290 | 1.220 | 1.720 | 2.060 | 1.020 | 2.390 |
| Tm | 0.340 | 0.154 | 0.180 | 0.178 | 0.250 | 0.290 | 0.141 | 0.340 |
| Yb | 2.170 | 0.950 | 1.180 | 1.160 | 1.540 | 1.970 | 0.970 | 2.240 |
| Lu | 0.310 | 0.140 | 0.174 | 0.164 | 0.230 | 0.280 | 0.157 | 0.330 |
| Hf | 2.260 | 0.950 | 1.240 | 1.210 | 1.490 | 2.220 | 1.120 | 2.470 |
| Ta | 0.300 | < 0.3 | <0.3 | <0.3 | <0.3 | 0.300 | <0.3 | 0.300 |
| Pb | 1.340 | 3.300 | 1.560 | 1.470 | 0.730 | 2.800 | 1.680 | 2.100 |
| Th | 2.100 | 1.000 | 1.100 | 1.000 | 1.100 | <0.7 | <0.7 | 2.100 |
| U | 0.330 | 0.125 | 0.161 | 0.120 | 0.200 | 0.220 | 0.152 | 0.330 |


| INTRUSION | WUS | WUS | WUS | WUS | WUS | WUS | WUS | WUS |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| ZONE | LBZ | LBZ | LBZ | LBZ | LBZ | LBZ | OZ | OZ |
| LITHOLOGY |  |  |  |  |  |  |  |  |
| HEIGHT (m) |  |  |  |  |  |  |  |  |
| SAMPLE I.D. | $\mathbf{0 7 2 3 A 0 0 1 ~}$ | $\mathbf{0 7 2 3 A 0 0 2}$ | $\mathbf{0 7 2 3 A 0 0 2}$ | $\mathbf{0 7 2 6 A 0 0 8}$ <br> C | $\mathbf{0 7 2 6 A 0 0 8}$ <br> D | $\mathbf{0 7 2 6 A 0 0 8}$ <br> E | $\mathbf{0 7 2 3 A 0 0 3}$ | $\mathbf{0 7 2 3 A A 0 0 4}$ |
| Lab | INRS, <br> Quebec | INRS, <br> Quebec | INRS, <br> Quebec | INRS, <br> Quebec | INRS, <br> Quebec | INRS, <br> Quebec | INRS, <br> Quebec | INRS, <br> Quebec |
|  |  |  |  |  |  |  |  |  |
| SiO2 | 45.500 | 45.680 | 45.680 | 46.920 | 45.220 | 46.380 | 44.460 | 42.590 |
| TiO2 | 0.710 | 0.720 | 0.720 | 0.730 | 0.680 | 0.640 | 0.620 | 0.520 |
| Al2O3 | 9.640 | 9.920 | 9.920 | 10.080 | 9.190 | 8.960 | 8.550 | 7.500 |


| Fe2O3T | 12.610 | 12.370 | 12.370 | 12.430 | 12.640 | 13.800 | 12.350 | 12.920 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| FeO* | 11.336 | 11.121 | 11.121 | 11.175 | 11.363 | 12.406 | 11.103 | 11.615 |
| Fe2O3* | 1.274 | 1.249 | 1.249 | 1.255 | 1.277 | 1.394 | 1.247 | 1.305 |
| MnO | 0.180 | 0.170 | 0.170 | 0.170 | 0.170 | 0.180 | 0.170 | 0.180 |
| MgO | 19.720 | 18.430 | 18.430 | 17.210 | 18.620 | 19.190 | 20.710 | 23.070 |
| CaO | 8.250 | 8.240 | 8.240 | 8.310 | 7.920 | 7.880 | 7.130 | 6.210 |
| Na 2 O | 1.090 | 1.100 | 1.100 | 1.130 | 1.190 | 1.070 | 0.920 | 0.800 |
| K2O | 0.310 | 0.470 | 0.470 | 0.320 | 0.270 | 0.250 | 0.420 | 0.260 |
| P2O5 | 0.060 | 0.060 | 0.060 | 0.070 | 0.060 | 0.060 | 0.060 | 0.050 |
| LOI | 1.600 | 2.500 | 2.500 | 2.200 | 3.700 | 1.200 | 4.200 | 5.500 |
| SUM | 100.010 | 99.980 | 99.980 | 99.850 | 100.000 | 100.000 | 99.980 | 100.010 |
| ICP-OES (ppm) |  |  |  |  |  |  |  |  |
| Sc | 27.000 | 28.000 | 28.000 | 30.000 | 26.000 | 29.000 | 25.000 | 22.000 |
| V | 203.000 | 209.000 | 209.000 | 221.000 | 204.000 | 213.000 | 183.000 | 155.000 |
| Cr | 1505.355 | 1382.189 | 1382.189 | 1375.347 | 1649.047 | 2032.229 | 1655.890 | 1751.685 |
| Co | 92.800 | 89.000 | 89.000 | 85.600 | 90.300 | 91.600 | 105.100 | 106.600 |
| Ni | 628.300 | 589.800 | 589.800 | 463.600 | 504.100 | 499.700 | 793.400 | 903.700 |
| Cu | 93.400 | 89.700 | 89.700 | 88.000 | 65.000 | 74.300 | 77.600 | 69.500 |
| Zn | 47.000 | 45.000 | 45.000 | 44.000 | 43.000 | 47.000 | 50.000 | 53.000 |
| Sr | 140.200 | 167.100 | 167.100 | 118.700 | 115.700 | 105.500 | 135.200 | 90.700 |
| Y | 13.500 | 14.500 | 14.500 | 14.300 | 12.200 | 12.100 | 13.000 | 10.100 |
| Zr | 44.700 | 48.900 | 48.900 | 50.700 | 43.100 | 39.900 | 43.000 | 32.900 |
| Ba | 81.800 | 101.300 | 101.300 | 78.900 | 69.900 | 57.200 | 93.000 | 187.200 |
| ICP-MS |  |  |  |  |  |  |  |  |
| Ti | 1.090 | 1.100 | 1.100 | 1.130 | 1.190 | 1.070 | 0.920 | 0.800 |
| V | 203.000 | 209.000 | 209.000 | 221.000 | 204.000 | 213.000 | 183.000 | 155.000 |
| Cr | 1505.355 | 1382.189 | 1382.189 | 1375.347 | 1649.047 | 2032.229 | 1655.890 | 1751.685 |
| Co | 92.800 | 89.000 | 89.000 | 85.600 | 90.300 | 91.600 | 105.100 | 106.600 |
| Ni | 628.300 | 589.800 | 589.800 | 463.600 | 504.100 | 499.700 | 793.400 | 903.700 |
| Cu | 93.400 | 89.700 | 89.700 | 88.000 | 65.000 | 74.300 | 77.600 | 69.500 |
| Zn | 47.000 | 45.000 | 45.000 | 44.000 | 43.000 | 47.000 | 50.000 | 53.000 |
| Rb | 10.800 | 13.700 | 13.700 | 13.100 | 12.100 | 9.100 | 14.100 | 9.500 |
| K | 2573.461 | 3901.699 | 3901.699 | 2656.476 | 2241.401 | 2075.372 | 3486.624 | 2158.386 |
| Sr | 140.200 | 167.100 | 167.100 | 118.700 | 115.700 | 105.500 | 135.200 | 90.700 |
| P | 261.853 | 261.853 | 261.853 | 305.495 | 261.853 | 261.853 | 261.853 | 218.211 |
| Y | 13.500 | 14.500 | 14.500 | 14.300 | 12.200 | 12.100 | 13.000 | 10.100 |
| Zr | 44.700 | 48.900 | 48.900 | 50.700 | 43.100 | 39.900 | 43.000 | 32.900 |
| Nb | 3.300 | 3.200 | 3.200 | 3.400 | 3.300 | 2.600 | 2.700 | 2.300 |
| Sn |  |  |  |  |  |  |  |  |
| Cs | 0.900 | 1.300 | 1.300 | 1.200 | 2.600 | 0.900 | 1.300 | 1.100 |
| Ba | 81.800 | 101.300 | 101.300 | 78.900 | 69.900 | 57.200 | 93.000 | 187.200 |
| La | 6.100 | 6.400 | 6.400 | 6.200 | 5.000 | 4.700 | 5.900 | 4.200 |
| Ce | 14.600 | 15.200 | 15.200 | 14.900 | 11.400 | 11.200 | 13.400 | 10.700 |
| Pr | 1.790 | 1.980 | 1.980 | 1.940 | 1.540 | 1.450 | 1.600 | 1.290 |
| Nd | 7.000 | 8.000 | 8.000 | 8.000 | 6.400 | 6.700 | 7.500 | 5.800 |
| Sm | 2.000 | 2.200 | 2.200 | 2.400 | 2.000 | 1.900 | 1.900 | 1.300 |
| Eu | 0.740 | 0.760 | 0.760 | 0.720 | 0.650 | 0.630 | 0.490 | 0.460 |
| Gd | 2.150 | 2.550 | 2.550 | 2.410 | 2.100 | 1.890 | 1.970 | 1.740 |
| Tb | 0.330 | 0.430 | 0.430 | 0.430 | 0.360 | 0.340 | 0.350 | 0.260 |
| Dy | 2.200 | 2.530 | 2.530 | 2.600 | 2.060 | 2.290 | 2.070 | 1.790 |
| Ho | 0.480 | 0.480 | 0.480 | 0.510 | 0.420 | 0.430 | 0.420 | 0.360 |
| Er | 1.390 | 1.540 | 1.540 | 1.520 | 1.290 | 1.150 | 1.270 | 1.110 |
| Tm | 0.210 | 0.230 | 0.230 | 0.200 | 0.190 | 0.190 | 0.190 | 0.180 |
| Yb | 1.250 | 1.280 | 1.280 | 1.380 | 1.170 | 1.210 | 1.060 | 1.080 |
| Lu | 0.180 | 0.180 | 0.180 | 0.210 | 0.200 | 0.180 | 0.190 | 0.130 |
| Hf | 1.300 | 1.800 | 1.800 | 1.700 | 1.400 | 1.400 | 1.100 | 1.000 |
| Ta | 0.200 | 0.200 | 0.200 | 0.200 | 0.200 | 0.100 | 0.200 | 0.300 |
| Pb | 6.900 | 3.700 | 3.700 | 3.200 | 2.400 | 2.200 | 4.000 | 3.800 |
| Th | 1.500 | 1.900 | 1.900 | 1.000 | 0.900 | 1.000 | 1.300 | 1.100 |
| U | 0.400 | 0.400 | 0.400 | 0.300 | 0.200 | 0.200 | 0.300 | 0.300 |
|  |  |  |  |  |  |  |  |  |
| Inversion models |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |
| Fitted to MgO (wt \%) of > | 10.000 | 10.000 | 10.000 | 10.000 | 10.000 | 10.000 | 10.000 | 10.000 |
| Model melts in ppm |  |  |  |  |  |  |  |  |
| Th | 2.140 | 1.900 | 1.900 | 1.000 | 0.900 | 1.000 | 1.998 | 1.998 |


| Nb | 4.724 | 3.200 | 3.200 | 3.400 | 3.300 | 2.600 | 4.168 | 4.205 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| Ta | 0.282 | 0.200 | 0.200 | 0.200 | 0.200 | 0.100 | 0.303 | 0.534 |
| La | 8.728 | 6.400 | 6.400 | 6.200 | 5.000 | 4.700 | 9.104 | 7.673 |
| Ce | 20.881 | 15.200 | 15.200 | 14.900 | 11.400 | 11.200 | 20.666 | 19.533 |
| Pr | 2.559 | 1.980 | 1.980 | 1.940 | 1.540 | 1.450 | 2.466 | 2.353 |
| P | 373.222 | 261.853 | 261.853 | 305.495 | 261.853 | 261.853 | 402.107 | 395.738 |
| Nd | 9.986 | 8.000 | 8.000 | 8.000 | 6.400 | 6.700 | 11.530 | 10.536 |
| Sm | 2.841 | 2.200 | 2.200 | 2.400 | 2.000 | 1.900 | 2.905 | 2.342 |
| Zr | 63.492 | 48.900 | 48.900 | 50.700 | 43.100 | 39.900 | 65.745 | 59.273 |
| Hf | 1.846 | 1.800 | 1.800 | 1.700 | 1.400 | 1.400 | 1.681 | 1.800 |
| Ti | 1.548 | 1.100 | 1.100 | 1.130 | 1.190 | 1.070 | 1.406 | 1.441 |
| Eu | 1.051 | 0.760 | 0.760 | 0.720 | 0.650 | 0.630 | 0.749 | 0.829 |
| Gd | 3.041 | 2.550 | 2.550 | 2.410 | 2.100 | 1.890 | 2.996 | 3.110 |
| Tb | 0.466 | 0.430 | 0.430 | 0.430 | 0.360 | 0.340 | 0.531 | 0.463 |
| Dy | 3.106 | 2.530 | 2.530 | 2.600 | 2.060 | 2.290 | 3.142 | 3.189 |
| Y | 19.070 | 14.500 | 14.500 | 14.300 | 12.200 | 12.100 | 19.740 | 18.008 |
| Ho | 0.677 | 0.480 | 0.480 | 0.510 | 0.420 | 0.430 | 0.636 | 0.640 |
| Er | 1.958 | 1.540 | 1.540 | 1.520 | 1.290 | 1.150 | 1.921 | 1.968 |
| Tm | 0.296 | 0.230 | 0.230 | 0.200 | 0.190 | 0.190 | 0.287 | 0.319 |
| Yb | 1.753 | 1.280 | 1.280 | 1.380 | 1.170 | 1.210 | 1.595 | 1.900 |
| Lu | 0.252 | 0.180 | 0.180 | 0.210 | 0.200 | 0.180 | 0.286 | 0.229 |
| Cr | 726.172 | 1382.189 | 1382.189 | 1375.347 | 1649.047 | 2032.229 | 640.700 | 563.515 |
| Sc | 34.386 | 28.000 | 28.000 | 30.000 | 26.000 | 29.000 | 33.387 | 32.501 |
|  |  |  |  |  |  |  |  |  |
| Rb |  |  |  |  |  |  |  |  |
| Ba | 15.433 | 13.700 | 13.700 | 13.100 | 12.100 | 9.100 | 21.722 | 17.314 |
| U | 116.690 | 101.300 | 101.300 | 78.900 | 69.900 | 57.200 | 142.967 | 340.053 |
| Sr | 0.571 | 0.400 | 0.400 | 0.300 | 0.200 | 0.200 | 0.461 | 0.545 |
| Residual modes | 200.171 | 167.100 | 167.100 | 118.700 | 115.700 | 105.500 | 208.063 | 165.028 |
| TMF | Olivine | 0.300 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.350 |


| INTRUSION | WUS | WUS | WUS | WUS | WUS | WUS | WUS |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| ZONE | OZ | OZ | OZ | OZ | OZ | OZ | OZ |
| LITHOLOGY |  |  |  |  |  |  |  |
| HEIGHT (m) |  |  |  |  |  |  |  |
| SAMPLE I.D. | 0723A005 | 0723A008 | 0723A014 | 0723A015 | 0726A001 | 0726A002 | 0726A003 |
| Lab | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec |
| SiO 2 | 41.410 | 40.640 | 41.850 | 42.710 | 41.910 | 44.060 | 44.780 |
| TiO2 | 0.550 | 0.380 | 0.470 | 0.500 | 0.450 | 0.530 | 0.620 |
| Al2O3 | 7.440 | 6.100 | 6.610 | 6.950 | 6.640 | 7.620 | 8.440 |
| Fe2O3T | 13.410 | 12.810 | 13.270 | 12.720 | 13.060 | 12.710 | 11.670 |
| FeO* | 12.056 | 11.516 | 11.930 | 11.435 | 11.741 | 11.426 | 10.491 |
| Fe2O3* | 1.354 | 1.294 | 1.340 | 1.285 | 1.319 | 1.284 | 1.179 |
| MnO | 0.190 | 0.170 | 0.170 | 0.170 | 0.180 | 0.170 | 0.150 |
| MgO | 22.640 | 26.010 | 23.090 | 23.490 | 24.080 | 22.840 | 20.180 |
| CaO | 5.880 | 5.200 | 5.540 | 5.750 | 5.580 | 6.300 | 8.030 |
| Na 2 O | 0.880 | 0.510 | 0.690 | 0.670 | 0.570 | 0.700 | 0.910 |
| K2O | 0.290 | 0.270 | 0.350 | 0.260 | 0.310 | 0.250 | 0.320 |
| P2O5 | 0.060 | 0.040 | 0.040 | 0.040 | 0.040 | 0.060 | 0.040 |
| LOI | 6.800 | 7.500 | 7.400 | 6.200 | 6.600 | 4.300 | 4.400 |
| SUM | 100.000 | 100.020 | 99.870 | 99.860 | 99.860 | 99.860 | 99.860 |
| ICP-OES (ppm) |  |  |  |  |  |  |  |
| Sc | 23.000 | 18.000 | 20.000 | 21.000 | 20.000 | 23.000 | 25.000 |
| V | 166.000 | 114.000 | 146.000 | 157.000 | 137.000 | 153.000 | 172.000 |
| Cr | 2244.347 | 1395.874 | 1676.418 | 1655.890 | 2039.071 | 1245.339 | 1300.079 |
| Co | 107.700 | 114.900 | 102.800 | 99.200 | 105.900 | 96.600 | 84.900 |
| Ni | 729.900 | 1029.400 | 805.600 | 907.400 | 889.600 | 821.300 | 676.100 |
| Cu | 71.000 | 47.200 | 66.100 | 53.500 | 53.300 | 60.300 | 52.800 |
| Zn | 59.000 | 59.000 | 62.000 | 52.000 | 56.000 | 52.000 | 42.000 |
| Sr | 87.900 | 79.200 | 116.100 | 117.400 | 87.800 | 94.700 | 116.300 |
| Y | 10.800 | 7.200 | 8.600 | 9.200 | 8.500 | 10.200 | 11.600 |
| Zr | 34.700 | 25.000 | 32.000 | 33.800 | 29.700 | 35.100 | 41.900 |


| Ba | 76.100 | 43.800 | 54.600 | 71.200 | 54.800 | 60.100 | 71.100 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| ICP-MS |  |  |  |  |  |  |  |
| Ti | 0.880 | 0.510 | 0.690 | 0.670 | 0.570 | 0.700 | 0.910 |
| V | 166.000 | 114.000 | 146.000 | 157.000 | 137.000 | 153.000 | 172.000 |
| Cr | 2244.347 | 1395.874 | 1676.418 | 1655.890 | 2039.071 | 1245.339 | 1300.079 |
| Co | 107.700 | 114.900 | 102.800 | 99.200 | 105.900 | 96.600 | 84.900 |
| Ni | 729.900 | 1029.400 | 805.600 | 907.400 | 889.600 | 821.300 | 676.100 |
| Cu | 71.000 | 47.200 | 66.100 | 53.500 | 53.300 | 60.300 | 52.800 |
| Zn | 59.000 | 59.000 | 62.000 | 52.000 | 56.000 | 52.000 | 42.000 |
| Rb | 10.200 | 8.700 | 8.700 | 8.900 | 8.600 | 8.900 | 10.000 |
| K | 2407.431 | 2241.401 | 2905.520 | 2158.386 | 2573.461 | 2075.372 | 2656.476 |
| Sr | 87.900 | 79.200 | 116.100 | 117.400 | 87.800 | 94.700 | 116.300 |
| P | 261.853 | 174.569 | 174.569 | 174.569 | 174.569 | 261.853 | 174.569 |
| Y | 10.800 | 7.200 | 8.600 | 9.200 | 8.500 | 10.200 | 11.600 |
| Zr | 34.700 | 25.000 | 32.000 | 33.800 | 29.700 | 35.100 | 41.900 |
| Nb | 2.500 | 1.800 | 1.600 | 1.600 | 1.200 | 2.400 | 2.200 |
| Sn |  |  |  |  |  |  |  |
| Cs | 2.100 | 0.700 | 1.400 | 1.200 | 0.900 | 1.100 | 1.600 |
| Ba | 76.100 | 43.800 | 54.600 | 71.200 | 54.800 | 60.100 | 71.100 |
| La | 4.400 | 3.200 | 4.000 | 4.200 | 3.700 | 4.700 | 5.100 |
| Ce | 10.700 | 7.400 | 9.300 | 9.600 | 8.900 | 10.600 | 11.600 |
| Pr | 1.330 | 0.980 | 1.180 | 1.250 | 1.220 | 1.390 | 1.490 |
| Nd | 5.400 | 4.100 | 4.800 | 5.500 | 5.400 | 6.100 | 6.500 |
| Sm | 1.500 | 1.100 | 1.200 | 1.400 | 1.300 | 1.500 | 1.800 |
| Eu | 0.560 | 0.390 | 0.450 | 0.490 | 0.390 | 0.530 | 0.600 |
| Gd | 1.650 | 1.130 | 1.450 | 1.680 | 1.560 | 1.720 | 1.880 |
| Tb | 0.340 | 0.180 | 0.280 | 0.310 | 0.240 | 0.320 | 0.320 |
| Dy | 1.920 | 1.240 | 1.650 | 1.620 | 1.550 | 1.760 | 2.060 |
| Ho | 0.370 | 0.280 | 0.330 | 0.340 | 0.320 | 0.390 | 0.430 |
| Er | 1.070 | 0.800 | 0.970 | 0.990 | 0.920 | 1.110 | 1.200 |
| Tm | 0.180 | 0.110 | 0.120 | 0.130 | 0.130 | 0.140 | 0.170 |
| Yb | 1.080 | 0.650 | 0.950 | 1.010 | 0.950 | 1.110 | 1.190 |
| Lu | 0.140 | 0.100 | 0.110 | 0.140 | 0.110 | 0.150 | 0.160 |
| Hf | 1.200 | 1.000 | 1.000 | 1.100 | 0.900 | 1.100 | 1.300 |
| Ta | 0.200 | 0.100 | 0.200 | 0.200 | 0.200 | 0.100 | 0.200 |
| Pb | 4.400 | 2.300 | 3.600 | 4.600 | 5.700 | 4.000 | 2.300 |
| Th | 0.900 | 0.600 | 0.500 | 0.900 | 0.500 | 0.900 | 0.800 |
| U | 0.300 | 0.100 | 0.200 | 0.200 | 0.200 | 0.200 | 0.200 |
|  |  |  |  |  |  |  |  |
| Inversion models |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |
| $\begin{aligned} & \text { Fitted to } \mathrm{MgO}(\mathrm{wt} \%) \\ & \text { of }> \end{aligned}$ | 10.000 | 10.000 | 101.000 | 10.000 | 10.000 | 10.000 | 10.000 |
| Model melts in ppm |  |  |  |  |  |  |  |
| Th | 1.638 | 1.048 | 0.996 | 1.632 | 1.039 | 1.546 | 1.228 |
| Nb | 4.579 | 3.164 | 3.213 | 2.920 | 2.515 | 4.146 | 3.391 |
| Ta | 0.357 | 0.172 | 0.389 | 0.355 | 0.405 | 0.169 | 0.303 |
| La | 8.053 | 5.621 | 8.024 | 7.659 | 7.748 | 8.114 | 7.858 |
| Ce | 19.568 | 12.989 | 18.637 | 17.493 | 18.616 | 18.286 | 17.863 |
| Pr | 2.430 | 1.719 | 2.362 | 2.276 | 2.549 | 2.396 | 2.293 |
| P | 475.747 | 304.571 | 347.053 | 316.017 | 361.993 | 449.115 | 267.661 |
| Nd | 9.827 | 7.164 | 9.562 | 9.973 | 11.222 | 10.477 | 9.977 |
| Sm | 2.708 | 1.908 | 2.367 | 2.518 | 2.673 | 2.558 | 2.748 |
| Zr | 62.627 | 43.355 | 63.111 | 60.786 | 61.053 | 59.854 | 63.967 |
| Hf | 2.164 | 1.734 | 1.971 | 1.977 | 1.848 | 1.875 | 1.984 |
| Ti | 1.587 | 0.884 | 1.360 | 1.205 | 1.171 | 1.194 | 1.389 |
| Eu | 1.011 | 0.676 | 0.888 | 0.881 | 0.802 | 0.904 | 0.916 |
| Gd | 2.954 | 1.945 | 2.832 | 2.997 | 3.173 | 2.912 | 2.855 |
| Tb | 0.606 | 0.309 | 0.544 | 0.551 | 0.486 | 0.540 | 0.485 |
| Dy | 3.427 | 2.128 | 3.210 | 2.881 | 3.140 | 2.972 | 3.122 |
| Y | 19.291 | 12.367 | 16.748 | 16.374 | 17.236 | 17.234 | 17.588 |
| Ho | 0.659 | 0.480 | 0.640 | 0.603 | 0.646 | 0.657 | 0.651 |
| Er | 1.900 | 1.367 | 1.876 | 1.752 | 1.852 | 1.866 | 1.813 |
| Tm | 0.320 | 0.188 | 0.232 | 0.230 | 0.262 | 0.235 | 0.257 |
| Yb | 1.903 | 1.103 | 1.820 | 1.773 | 1.892 | 1.853 | 1.788 |
| Lu | 0.247 | 0.170 | 0.211 | 0.246 | 0.219 | 0.250 | 0.240 |
| Cr | 622.134 | 662.966 | 639.366 | 634.562 | 652.294 | 592.173 | 623.390 |
| Sc | 34.024 | 25.978 | 31.138 | 30.983 | 31.888 | 32.857 | 33.347 |


|  |  |  |  |  |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| Rb | 18.623 | 15.248 | 17.400 | 16.191 | 17.950 | 15.331 | 15.382 |
| Ba | 138.489 | 76.533 | 108.765 | 129.102 | 113.881 | 103.229 | 109.133 |
| U | 0.546 | 0.175 | 0.398 | 0.363 | 0.416 | 0.344 | 0.307 |
| Sr | 160.225 | 138.597 | 231.736 | 213.221 | 182.855 | 162.894 | 178.703 |
|  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |
| Residual modes | 0.545 | 0.568 | 0.497 | 0.547 | 0.476 | 0.578 | 0.648 |
| TMF | 0.450 | 0.430 | 0.500 | 0.450 | 0.520 | 0.420 | 0.350 |
| Olivine |  |  |  |  |  |  |  |


| INTRUSION | WUS | WUS | WUS | WUS | WUS | WUS | WUS |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| ZONE | OZ | OZ | DZ | DZ | DZ | DZ | DZ |
| LITHOLOGY |  |  |  |  |  |  |  |
| HEIGHT (m) |  |  |  |  |  |  |  |
| SAMPLE I.D. | 0726A004 | 10KS095A6 | 10JB156B2 | 10HY062A1 | 10HY062A2 | 10HY062A3 | 10KS095A15 |
| Lab | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec | INRS, Quebec |
|  |  |  |  |  |  |  |  |
| SiO 2 | 45.430 | 40.175 | 48.192 | 50.522 | 50.266 | 50.924 | 47.379 |
| TiO2 | 0.610 | 0.428 | 1.166 | 0.928 | 1.170 | 1.260 | 0.923 |
| Al2O3 | 8.460 | 6.045 | 13.458 | 14.930 | 13.916 | 14.264 | 13.564 |
| Fe2O3T | 12.830 | 12.247 | 12.623 | 10.272 | 12.823 | 12.690 | 10.238 |
| FeO* | 11.534 | 11.010 | 11.348 | 9.235 | 11.528 | 11.409 | 9.204 |
| Fe2O3* | 1.296 | 1.237 | 1.275 | 1.038 | 1.295 | 1.282 | 1.034 |
| MnO | 0.170 | 0.162 | 0.135 | 0.149 | 0.177 | 0.173 | 0.153 |
| MgO | 20.870 | 23.331 | 7.038 | 8.261 | 7.213 | 6.911 | 7.272 |
| CaO | 7.230 | 5.507 | 10.644 | 10.939 | 9.830 | 9.844 | 9.779 |
| Na 2 O | 0.940 | 0.718 | 2.489 | 1.771 | 1.878 | 1.874 | 2.300 |
| K2O | 0.300 | 0.173 | 1.827 | 1.271 | 1.298 | 1.288 | 1.126 |
| P2O5 | 0.060 | 0.035 | 0.109 | 0.059 | 0.071 | 0.094 | 0.080 |
| LOI | 2.600 | 6.558 | 2.501 | 1.409 | 1.559 | 1.588 | 2.447 |
| SUM | 99.850 | 95.630 | 100.492 | 100.572 | 100.283 | 100.994 | 95.367 |
|  |  |  |  |  |  |  |  |
| ICP-OES (ppm) |  |  |  |  |  |  |  |
| Sc | 26.000 | 19.993 | 36.957 | 37.573 | 41.455 | 38.341 | 34.685 |
| V | 182.000 | 124.591 | 328.279 | 271.841 | 364.848 | 331.975 | 247.192 |
| Cr | 1484.827 | 1213.669 | 76.500 | 265.265 | 13.506 | 49.195 | 70.504 |
| Co | 91.700 | 94.596 | 26.156 | 39.669 | 47.881 | 45.662 | 42.151 |
| Ni | 720.700 | 911.732 | 80.325 | 129.038 | 88.139 | 91.190 | 88.802 |
| Cu | 71.300 | 58.365 | 139.789 | 103.929 | 183.807 | 156.483 | 132.777 |
| Zn | 50.000 | 75.901 | < 40 | 83.119 | 91.711 | 103.441 | 61.340 |
| Sr | 98.500 | 100.000 | 746.485 | 237.289 | 234.915 | 220.129 | 359.000 |
| Y | 11.600 | 8.200 | 20.000 | 13.400 | 20.000 | 22.000 | 20.000 |
| Zr | 39.000 | 32.000 | 78.200 | 72.000 | 89.000 | 105.000 | 70.000 |
| Ba | 72.200 | 50.100 | 279.372 | 114.507 | 122.750 | 163.649 | 186.000 |
|  |  |  |  |  |  |  |  |
| ICP-MS |  |  |  |  |  |  |  |
| Ti | 0.940 | 0.718 | 2.490 | 1.771 | 1.878 | 1.874 | 2.301 |
| V | 182.000 | 124.591 | 328.279 | 271.841 | 364.848 | 331.975 | 247.192 |
| Cr | 1484.827 | 1213.669 | 76.500 | 265.265 | 13.506 | 49.195 | 70.504 |
| Co | 91.700 | 94.596 | 26.156 | 39.669 | 47.881 | 45.662 | 42.151 |
| Ni | 720.700 | 911.732 | 80.325 | 129.038 | 88.139 | 91.190 | 88.802 |
| Cu | 71.300 | 58.365 | 139.789 | 103.929 | 183.807 | 156.483 | 132.777 |
| Zn | 50.000 | 75.901 | < 40 | 83.119 | 91.711 | 103.441 | 61.340 |
| Rb | 9.100 | 7.300 | 29.600 | 26.000 | 33.900 | 34.300 | 34.400 |
| K | 2490.446 | 1434.383 | 15166.978 | 10555.051 | 10777.409 | 10690.636 | 9350.536 |
| Sr | 98.500 | 100.000 | 746.485 | 237.289 | 234.915 | 220.129 | 359.000 |
| P | 261.853 | 154.287 | 475.123 | 259.397 | 309.167 | 411.462 | 347.542 |
| Y | 11.600 | 8.200 | 20.000 | 13.400 | 20.000 | 22.000 | 20.000 |
| Zr | 39.000 | 32.000 | 78.200 | 72.000 | 89.000 | 105.000 | 70.000 |
| Nb | 2.700 | 2.040 | 6.210 | 4.230 | 5.380 | 6.390 | 4.820 |
| Sn |  |  |  |  |  |  |  |
| Cs | 0.600 | 1.650 | 1.900 | 0.600 | 0.600 | 1.000 | 0.400 |
| Ba | 72.200 | 50.100 | 279.372 | 114.507 | 122.750 | 163.649 | 186.000 |
| La | 5.200 | 3.500 | 9.200 | 7.300 | 9.200 | 11.200 | 7.800 |
| Ce | 12.500 | 8.010 | 18.600 | 16.800 | 21.400 | 26.000 | 18.500 |
| Pr | 1.650 | 1.020 | 2.800 | 2.100 | 2.700 | 3.300 | 2.300 |


| Nd | 6.900 | 4.550 | 13.000 | 9.800 | 12.300 | 14.500 | 10.500 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Sm | 2.000 | 1.210 | 3.300 | 2.500 | 3.100 | 3.800 | 2.700 |
| Eu | 0.610 | 0.390 | 1.050 | 0.810 | 1.030 | 1.150 | 0.900 |
| Gd | 2.030 | 1.310 | 3.320 | 2.740 | 3.420 | 4.030 | 3.180 |
| Tb | 0.360 | 0.240 | 0.580 | 0.450 | 0.580 | 0.680 | 0.520 |
| Dy | 2.040 | 1.420 | 3.670 | 2.900 | 3.500 | 4.220 | 3.310 |
| Ho | 0.450 | 0.280 | 0.730 | 0.610 | 0.750 | 0.880 | 0.670 |
| Er | 1.250 | 0.848 | 2.050 | 1.760 | 2.150 | 2.500 | 1.920 |
| Tm | 0.190 | 0.121 | 0.300 | 0.260 | 0.310 | 0.370 | 0.290 |
| Yb | 1.120 | 0.769 | 1.830 | 1.640 | 1.950 | 2.260 | 1.770 |
| Lu | 0.190 | 0.118 | 0.280 | 0.240 | 0.300 | 0.350 | 0.270 |
| Hf | 1.200 | 0.796 | 2.120 | 1.930 | 2.290 | 2.700 | 1.900 |
| Ta | 0.300 | 0.130 | 0.350 | 0.300 | 0.350 | 0.400 | 0.310 |
| Pb | 3.500 | 62.300 | 2.300 | 3.000 | 6.100 | 8.650 | 25.400 |
| Th | 1.200 | 0.900 | 1.900 | 1.400 | 1.100 | 2.400 | 2.000 |
| U | 0.300 | 0.156 | 0.650 | 0.310 | 0.400 | 0.470 | 0.360 |
|  |  |  |  |  |  |  |  |
| Inversion models |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |
| $\begin{aligned} & \text { Fitted to } \mathrm{MgO}(\mathrm{wt} \%) \\ & \text { of }> \end{aligned}$ | 10.000 | 10.000 |  |  |  |  |  |
| Model melts in ppm |  |  |  |  |  |  |  |
| Th | 1.842 | 1.980 |  |  |  |  |  |
| Nb | 4.162 | 4.532 |  |  |  |  |  |
| Ta | 0.454 | 0.278 |  |  |  |  |  |
| La | 8.012 | 7.767 |  |  |  |  |  |
| Ce | 19.249 | 17.753 |  |  |  |  |  |
| Pr | 2.539 | 2.258 |  |  |  |  |  |
| P | 401.491 | 338.646 |  |  |  |  |  |
| Nd | 10.591 | 10.011 |  |  |  |  |  |
| Sm | 3.053 | 2.630 |  |  |  |  |  |
| Zr | 59.539 | 69.563 |  |  |  |  |  |
| Hf | 1.831 | 1.730 |  |  |  |  |  |
| Ti | 1.435 | 1.561 |  |  |  |  |  |
| Eu | 0.931 | 0.848 |  |  |  |  |  |
| Gd | 3.083 | 2.814 |  |  |  |  |  |
| Tb | 0.545 | 0.513 |  |  |  |  |  |
| Dy | 3.091 | 3.036 |  |  |  |  |  |
| Y | 17.588 | 17.553 |  |  |  |  |  |
| Ho | 0.681 | 0.597 |  |  |  |  |  |
| Er | 1.888 | 1.800 |  |  |  |  |  |
| Tm | 0.287 | 0.257 |  |  |  |  |  |
| Yb | 1.683 | 1.614 |  |  |  |  |  |
| Lu | 0.285 | 0.248 |  |  |  |  |  |
| Cr | 711.977 | 741.625 |  |  |  |  |  |
| Sc | 34.681 | 32.851 |  |  |  |  |  |
|  |  |  |  |  |  |  |  |
| Rb | 13.998 | 16.140 |  |  |  |  |  |
| Ba | 110.821 | 110.231 |  |  |  |  |  |
| U | 0.460 | 0.343 |  |  |  |  |  |
| Sr | 151.352 | 220.556 |  |  |  |  |  |
|  |  |  |  |  |  |  |  |
| Residual modes |  |  |  |  |  |  |  |
| TMF | 0.648 | 0.449 |  |  |  |  |  |
| Olivine | 0.350 | 0.550 |  |  |  |  |  |


| INTRUSION | WUS | WUS | WUS | WUS | FCSC |
| :--- | :--- | :--- | :--- | :--- | :--- |
| ZONE | DZ | DZ | UCM | lower chill |  |
| LITHOLOGY |  |  |  |  |  |
| HEIGHT (m) |  |  |  |  |  |
| SAMPLE <br> I.D. | 10JB155A1 | 10JB155C1 | 10HY062A4 | 10AW054B1 | av chill |
| Lab | INRS, <br> Quebec | INRS, <br> Quebec | INRS, <br> Quebec | INRS, <br> Quebec |  |
|  | 36.247 | 34.763 | 49.204 | 48.780 | 48.057 |
| $\mathrm{SiO2}$ | 0.401 | 0.467 | 0.976 | 0.970 | 0.990 |
| $\mathrm{TiO2}$ |  |  |  |  |  |


| Al2O3 | 7.160 | 7.531 | 12.691 | 12.937 | 12.918 |
| :---: | :---: | :---: | :---: | :---: | :---: |
| Fe2O3T | 10.087 | 8.357 | 11.206 | 12.202 | 11.665 |
| FeO* | 9.068 | 7.513 | 10.074 | 10.970 | 10.487 |
| Fe2O3* | 1.019 | 0.844 | 1.132 | 1.232 | 1.178 |
| MnO | 0.089 | 0.109 | 0.119 | 0.186 | 0.160 |
| MgO | 4.714 | 5.268 | 10.273 | 10.898 | 10.328 |
| CaO | 31.751 | 36.845 | 9.317 | 11.461 | 11.170 |
| Na 2 O | 0.518 | < 0.07 | 2.426 | 1.490 | 1.891 |
| K2O | 1.052 | < 0.08 | 1.028 | 0.565 | 1.026 |
| P2O5 | 0.163 | 0.229 | 0.063 | 0.085 | 0.080 |
| LOI | 7.988 | 7.129 | 3.723 | 0.612 | 1.639 |
| SUM | 100.833 | 101.033 | 101.078 | 100.457 | 101.561 |
|  |  |  |  |  |  |
| ICP-OES (ppm) |  |  |  |  |  |
| Sc | 6.663 | 7.643 | 35.727 | 37.468 | 38.204 |
| V | 87.217 | 84.817 | 275.439 | 280.116 | 290.731 |
| Cr | 29.759 | 33.366 | 731.167 | 644.611 | 607.268 |
| Co | 32.360 | 9.484 | 42.640 | 47.692 | 47.064 |
| Ni | 25.712 | 25.594 | 226.309 | 211.648 | 184.990 |
| Cu | 239.772 | 16.120 | 28.023 | 132.485 | 106.992 |
| Zn | < 40 | < 40 | 140.232 | 63.087 | 72.822 |
| Sr | 214.201 | 130.842 | 252.722 | 179.948 | 441.285 |
| Y | 20.000 | 18.000 | 15.400 | 18.000 | 18.416 |
| Zr | 93.000 | 106.000 | 74.000 | 69.000 | 72.314 |
| Ba | 219.648 | < 6 | 134.090 | 113.764 | 193.100 |
|  |  |  |  |  |  |
| ICP-MS |  |  |  |  |  |
| Ti | 0.518 | \#VALUE! | 2.427 | 1.490 | 0.000 |
| V | 87.217 | 84.817 | 275.439 | 280.116 | 290.272 |
| Cr | 29.759 | 33.366 | 731.167 | 644.611 | 628.370 |
| Co | 32.360 | 9.484 | 42.640 | 47.692 | 47.961 |
| Ni | 25.712 | 25.594 | 226.309 | 211.648 | 192.426 |
| Cu | 239.772 | 16.120 | 28.023 | 132.485 | 110.848 |
| Zn | < 40 | < 40 | 140.232 | 63.087 | 74.844 |
| Rb | 25.300 | 0.200 | 22.000 | 21.000 | 28.261 |
| K | 8735.618 | -dl | 8536.621 | 4689.708 | 8653.266 |
| Sr | 214.201 | 130.842 | 252.722 | 179.948 | 435.198 |
| P | 713.116 | 997.605 | 273.601 | 368.814 | 417.700 |
| Y | 20.000 | 18.000 | 15.400 | 18.000 | 18.350 |
| Zr | 93.000 | 106.000 | 74.000 | 69.000 | 70.467 |
| Nb | 6.410 | 5.690 | 4.810 | 4.950 | 4.721 |
| Sn |  |  |  |  | 1.085 |
| Cs | 0.620 | <0.13 | < 0.26 | 7.600 | 2.614 |
| Ba | 219.648 | < 6 | 134.090 | 113.764 | 196.729 |
| La | 8.300 | 31.700 | 6.700 | 7.800 | 7.379 |
| Ce | 22.200 | 42.100 | 17.000 | 17.900 | 16.412 |
| Pr | 3.600 | 4.800 | 2.200 | 2.500 | 2.383 |
| Nd | 16.000 | 17.800 | 10.500 | 10.700 | 10.889 |
| Sm | 3.000 | 3.300 | 2.700 | 2.800 | 2.817 |
| Eu | 0.910 | 0.880 | 0.770 | 0.890 | 0.846 |
| Gd | 2.680 | 2.910 | 2.990 | 3.020 | 2.922 |
| Tb | 0.440 | 0.440 | 0.500 | 0.500 | 0.519 |
| Dy | 2.760 | 2.780 | 3.200 | 3.200 | 3.244 |
| Ho | 0.580 | 0.560 | 0.660 | 0.650 | 0.634 |
| Er | 1.790 | 1.690 | 1.860 | 1.850 | 1.821 |
| Tm | 0.280 | 0.260 | 0.290 | 0.250 | 0.261 |
| Yb | 1.790 | 1.560 | 1.700 | 1.680 | 1.712 |
| Lu | 0.260 | 0.230 | 0.260 | 0.250 | 0.250 |
| Hf | 2.230 | 2.840 | 1.960 | 1.940 | 1.917 |
| Ta | 0.400 | 0.400 | 0.300 | <0.3 | 0.323 |
| Pb | 0.650 | 1.150 | 10.000 | 3.900 | 3.002 |
| Th | 4.900 | 3.800 | 1.600 | 0.700 | 1.434 |
| U | 1.620 | 4.990 | 0.340 | 0.300 | 0.303 |

## Appendix F2 - FCSC olivine mineral chemistry

| Sample code | Sill | SiO 2 | TiO2 | Al2O3 | Cr2O3 | FeO | MnO | MgO | CaO | NiO | Total | Fo |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| JB367C 46 | PS | 40.50 | 0.00 | 0.03 | 0.05 | 11.72 | 0.17 | 47.09 | 0.30 | 0.37 | 100.23 | 0.88 |
| JB367C 47 | PS | 40.38 | 0.00 | 0.07 | 0.05 | 11.79 | 0.21 | 46.82 | 0.28 | 0.37 | 99.97 | 0.88 |
| JB367C 52 | PS | 39.74 | 0.00 | 0.03 | 0.02 | 17.72 | 0.24 | 41.95 | 0.11 | 0.29 | 100.07 | 0.81 |
| JB367C 54 | PS | 40.40 | 0.01 | 0.01 | 0.01 | 13.36 | 0.21 | 45.35 | 0.28 | 0.30 | 99.92 | 0.86 |
| JB367C 59 | PS | 40.28 | 0.01 | 0.04 | 0.06 | 12.23 | 0.20 | 46.16 | 0.26 | 0.36 | 99.58 | 0.87 |
| JB367C 60 | PS | 40.16 | 0.00 | 0.03 | 0.03 | 12.72 | 0.14 | 46.59 | 0.26 | 0.30 | 100.23 | 0.87 |
| JB367C 62 | PS | 41.44 | 0.00 | 0.04 | 0.02 | 11.52 | 0.12 | 44.94 | 0.23 | 0.31 | 98.61 | 0.87 |
| av JB367C ol cores | PS | 40.41 | 0.00 | 0.03 | 0.03 | 13.01 | 0.18 | 45.56 | 0.24 | 0.33 | 99.80 | 0.86 |
| JB367C 63 | PS | 38.09 | 0.00 | 0.02 | 0.02 | 25.07 | 0.37 | 35.70 | 0.21 | 0.21 | 99.68 | 0.72 |
| JB367C 61 | PS | 38.78 | 0.00 | 0.00 | 0.03 | 18.51 | 0.26 | 41.37 | 0.22 | 0.24 | 99.41 | 0.80 |
| JB367C 48 | PS | 53.48 | 0.45 | 1.21 | 0.02 | 17.23 | 0.38 | 24.69 | 1.89 | 0.06 | 99.42 | 0.72 |
| JB367C 53 | PS | 38.96 | 0.01 | 0.00 | 0.00 | 21.80 | 0.31 | 37.92 | 0.19 | 0.22 | 99.41 | 0.76 |
| JB367C 55 | PS | 39.26 | 0.00 | 0.02 | 0.03 | 19.02 | 0.28 | 40.84 | 0.21 | 0.28 | 99.93 | 0.79 |
| av JB367C ol rims | PS | 41.71 | 0.09 | 0.25 | 0.02 | 20.33 | 0.32 | 36.10 | 0.54 | 0.20 | 99.57 | 0.76 |
| JB367H 66 | PS | 39.34 | 0.00 | 0.05 | 0.00 | 18.45 | 0.24 | 41.90 | 0.24 | 0.27 | 100.49 | 0.80 |
| JB367H 67 | PS | 39.39 | 0.00 | 0.03 | 0.03 | 16.38 | 0.23 | 42.52 | 0.22 | 0.20 | 98.99 | 0.82 |
| JB367H 75 | PS | 39.06 | 0.00 | 0.04 | 0.04 | 19.46 | 0.33 | 40.74 | 0.21 | 0.25 | 100.14 | 0.79 |
| JB367H 76 | PS | 39.45 | 0.00 | 0.04 | 0.07 | 16.53 | 0.19 | 42.64 | 0.26 | 0.27 | 99.45 | 0.82 |
| JB367H 77 | PS | 39.14 | 0.01 | 0.02 | 0.00 | 19.98 | 0.30 | 40.16 | 0.24 | 0.24 | 100.08 | 0.78 |
| JB367H 78 | PS | 38.83 | 0.03 | 0.00 | 0.03 | 20.94 | 0.28 | 39.57 | 0.16 | 0.28 | 100.11 | 0.77 |
| JB367H 79 | PS | 39.82 | 0.01 | 0.03 | 0.06 | 15.25 | 0.20 | 43.90 | 0.25 | 0.33 | 99.84 | 0.84 |
| JB367H 80 | PS | 39.37 | 0.00 | 0.01 | 0.04 | 18.42 | 0.22 | 41.45 | 0.25 | 0.23 | 99.98 | 0.80 |
| JB367H 68 | PS | 38.93 | 0.00 | 0.03 | 0.04 | 18.90 | 0.24 | 40.16 | 0.25 | 0.25 | 98.80 | 0.79 |
| JB367H 87 | PS | 38.62 | 0.01 | 0.01 | 0.01 | 20.63 | 0.37 | 39.15 | 0.30 | 0.27 | 99.37 | 0.77 |
| JB367H 88 | PS | 38.85 | 0.00 | 0.02 | 0.05 | 20.25 | 0.25 | 39.56 | 0.20 | 0.22 | 99.40 | 0.78 |
| JB367H 74 | PS | 38.82 | 0.00 | 0.03 | 0.02 | 21.51 | 0.25 | 39.27 | 0.26 | 0.22 | 100.38 | 0.76 |
| JB367H 72 | PS | 38.79 | 0.01 | 0.02 | 0.00 | 20.04 | 0.36 | 40.33 | 0.18 | 0.23 | 99.94 | 0.78 |
| JB367H 73 | PS | 38.93 | 0.00 | 0.02 | 0.00 | 20.56 | 0.24 | 39.96 | 0.13 | 0.28 | 100.14 | 0.78 |
| JB367H 85 | PS | 39.07 | 0.00 | 0.03 | 0.06 | 19.46 | 0.28 | 40.27 | 0.19 | 0.21 | 99.56 | 0.79 |
| JB367H 86 | PS | 38.83 | 0.04 | 0.02 | 0.01 | 20.85 | 0.31 | 38.90 | 0.19 | 0.25 | 99.40 | 0.77 |
| av 10JB367H olivines | PS | 39.08 | 0.01 | 0.03 | 0.03 | 19.23 | 0.27 | 40.65 | 0.22 | 0.25 | 99.75 | 0.79 |
| JB368B 107 | PS | 38.70 | 0.00 | 0.00 | 0.00 | 21.35 | 0.25 | 39.05 | 0.24 | 0.20 | 99.78 | 0.77 |
| JB368B 108 | PS | 38.86 | 0.02 | 0.01 | 0.01 | 21.42 | 0.55 | 38.49 | 0.22 | 0.21 | 99.77 | 0.76 |
| JB368B 109 | PS | 38.57 | 0.00 | 0.02 | 0.03 | 20.22 | 0.26 | 39.55 | 0.30 | 0.23 | 99.17 | 0.78 |
| JB368B 110 | PS | 38.82 | 0.00 | 0.01 | 0.00 | 21.97 | 0.34 | 38.73 | 0.22 | 0.21 | 100.30 | 0.76 |
| JB368B 114 | PS | 38.45 | 0.01 | 0.02 | 0.02 | 22.24 | 0.25 | 38.56 | 0.23 | 0.18 | 99.94 | 0.76 |
| JB368B 115 | PS | 39.07 | 0.02 | 0.01 | 0.01 | 19.56 | 0.27 | 40.24 | 0.25 | 0.21 | 99.63 | 0.79 |
| JB368B 116 | PS | 38.38 | 0.00 | 0.01 | 0.04 | 20.63 | 0.38 | 38.86 | 0.18 | 0.19 | 98.68 | 0.77 |
| JB368B 117 | PS | 38.47 | 0.00 | 0.00 | 0.05 | 21.22 | 0.25 | 37.34 | 0.20 | 0.21 | 97.73 | 0.76 |
| JB368B 118 | PS | 38.72 | 0.00 | 0.02 | 0.01 | 20.64 | 0.18 | 39.37 | 0.24 | 0.25 | 99.42 | 0.77 |
| JB368B 119 | PS | 39.02 | 0.01 | 0.02 | 0.03 | 19.92 | 0.25 | 40.51 | 0.19 | 0.16 | 100.12 | 0.78 |
| JB368B 120 | PS | 38.57 | 0.00 | 0.00 | 0.00 | 22.24 | 0.29 | 38.65 | 0.19 | 0.20 | 100.13 | 0.76 |
| JB368B 121 | PS | 38.41 | 0.00 | 0.02 | 0.02 | 22.44 | 0.30 | 38.15 | 0.24 | 0.13 | 99.71 | 0.75 |
| JB368B 122 | PS | 38.34 | 0.02 | 0.02 | 0.03 | 22.37 | 0.29 | 38.09 | 0.20 | 0.18 | 99.54 | 0.75 |
| av 10JB368B oliv |  | 38.64 | 0.01 | 0.01 | 0.02 | 21.25 | 0.30 | 38.89 | 0.22 | 0.20 | 99.53 | 0.77 |
| 10KS95 A7 b11 | WUS | 38.20 | 0.01 | 0.13 | 0.01 | 23.55 | 0.35 | 35.86 | 0.26 | 0.17 | 98.57 | 0.73 |
| 10KS95 A7 b12 | WUS | 37.73 | 0.01 | 0.14 | 0.00 | 24.02 | 0.28 | 36.06 | 0.27 | 0.11 | 98.62 | 0.73 |
| 10KS95 A7 b13 | WUS | 37.92 | 0.00 | 0.12 | 0.00 | 25.69 | 0.32 | 34.79 | 0.20 | 0.06 | 99.13 | 0.71 |
| 10KS95 A7 b24 | WUS | 38.91 | 0.01 | 0.15 | 0.01 | 21.21 | 0.26 | 38.64 | 0.28 | 0.14 | 99.61 | 0.76 |
| 10KS95 A7 b25 | WUS | 39.07 | 0.00 | 0.13 | 0.00 | 20.54 | 0.20 | 38.96 | 0.21 | 0.20 | 99.33 | 0.77 |
| av 10ks95A7 ol | WUS | 38.37 | 0.01 | 0.13 | 0.00 | 23.00 | 0.28 | 36.86 | 0.25 | 0.14 | 99.05 | 0.74 |
| 10KS95 A5 b4 | WUS | 39.27 | 0.00 | 0.15 | 0.00 | 18.38 | 0.26 | 40.84 | 0.21 | 0.26 | 99.40 | 0.80 |
| 10KS95 A5 b5 | WUS | 39.26 | 0.00 | 0.14 | 0.00 | 17.91 | 0.33 | 40.66 | 0.16 | 0.27 | 98.76 | 0.80 |
| av 10ks95A5 olpheno? | WUS | 39.27 | 0.00 | 0.14 | 0.00 | 18.14 | 0.30 | 40.75 | 0.19 | 0.27 | 99.08 | 0.80 |
| 10KS95 A5 a10 | WUS | 39.29 | 0.00 | 0.15 | 0.04 | 17.81 | 0.33 | 40.87 | 0.18 | 0.24 | 98.92 | 0.80 |
| 10KS95 A5 a11 | WUS | 39.40 | 0.01 | 0.14 | 0.01 | 16.98 | 0.24 | 41.80 | 0.13 | 0.27 | 98.97 | 0.81 |


| Sample code | Sill | SiO 2 | TiO2 | Al2O3 | Cr2O3 | FeO | MnO | MgO | CaO | NiO | Total | Fo |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 10KS95 A5 a12 | WUS | 40.83 | 0.00 | 0.12 | 0.03 | 16.10 | 0.24 | 41.97 | 0.17 | 0.23 | 99.70 | 0.82 |
| 10KS95 A5 b1 | WUS | 39.96 | 0.01 | 0.14 | 0.04 | 14.99 | 0.21 | 43.72 | 0.18 | 0.27 | 99.52 | 0.84 |
| av 10ks95A5 ol incl | WUS | 39.87 | 0.01 | 0.14 | 0.03 | 16.47 | 0.25 | 42.09 | 0.16 | 0.25 | 99.28 | 0.82 |
| 10JB367I a5 | PS | 39.57 | 0.00 | 0.13 | 0.05 | 16.36 | 0.23 | 42.48 | 0.23 | 0.26 | 99.32 | 0.82 |
| 10JB367I a7 | PS | 39.52 | 0.01 | 0.13 | 0.11 | 17.84 | 0.25 | 41.53 | 0.15 | 0.27 | 99.81 | 0.81 |
| 10JB367I b11 | PS | 39.36 | 0.01 | 0.15 | 0.02 | 18.22 | 0.21 | 41.45 | 0.23 | 0.27 | 99.92 | 0.80 |
| 10JB367I b12 | PS | 39.44 | 0.00 | 0.15 | 0.02 | 17.59 | 0.24 | 41.67 | 0.19 | 0.22 | 99.53 | 0.81 |
| av 10jb367i ol incl | PS | 39.47 | 0.00 | 0.14 | 0.05 | 17.50 | 0.23 | 41.78 | 0.20 | 0.25 | 99.64 | 0.81 |
| 10JB367J al | PS | 39.31 | 0.00 | 0.14 | 0.03 | 18.53 | 0.33 | 40.71 | 0.21 | 0.24 | 99.50 | 0.80 |
| 10JB367J a2 | PS | 39.49 | 0.01 | 0.14 | 0.04 | 16.85 | 0.24 | 41.87 | 0.27 | 0.23 | 99.16 | 0.82 |
| 10JB367J a4 | PS | 39.02 | 0.00 | 0.12 | 0.04 | 18.24 | 0.35 | 40.37 | 0.25 | 0.20 | 98.61 | 0.80 |
| av 10jb367j ol incl | PS | 39.28 | 0.00 | 0.14 | 0.04 | 17.87 | 0.31 | 40.98 | 0.24 | 0.22 | 99.09 | 0.80 |
| 10JB367J b11 | PS | 38.68 | 0.00 | 0.13 | 0.01 | 21.68 | 0.33 | 38.25 | 0.22 | 0.20 | 99.54 | 0.76 |
| 10JB367J c16 | PS | 38.86 | 0.00 | 0.13 | 0.03 | 20.53 | 0.24 | 38.66 | 0.28 | 0.17 | 98.91 | 0.77 |
| 10JB367J c17 | PS | 38.77 | 0.00 | 0.15 | 0.00 | 20.69 | 0.30 | 39.65 | 0.29 | 0.20 | 100.06 | 0.77 |
| 10JB367J c18 | PS | 38.94 | 0.00 | 0.12 | 0.02 | 19.94 | 0.30 | 39.27 | 0.17 | 0.21 | 98.99 | 0.78 |
| 10JB367J c19 | PS | 38.68 | 0.00 | 0.13 | 0.00 | 20.31 | 0.40 | 38.67 | 0.16 | 0.15 | 98.52 | 0.77 |
| 10JB367J d31 | PS | 38.63 | 0.00 | 0.13 | 0.00 | 19.85 | 0.35 | 39.19 | 0.27 | 0.16 | 98.61 | 0.78 |
| 10JB367J d32 | PS | 38.54 | 0.01 | 0.15 | 0.03 | 21.14 | 0.38 | 38.27 | 0.22 | 0.18 | 98.92 | 0.76 |
| av 10 jb 367 j ol intdendritic | PS | 38.73 | 0.00 | 0.13 | 0.01 | 20.59 | 0.33 | 38.85 | 0.23 | 0.18 | 99.08 | 0.77 |
| 10JB367G b1 | PS | 40.49 | 0.00 | 0.15 | 0.06 | 12.91 | 0.26 | 45.77 | 0.27 | 0.32 | 100.27 | 0.86 |
| 10JB367G a10 | PS | 39.65 | 0.00 | 0.13 | 0.04 | 14.59 | 0.17 | 44.04 | 0.24 | 0.28 | 99.14 | 0.84 |
| 10JB367G a11 | PS | 40.00 | 0.01 | 0.14 | 0.04 | 14.90 | 0.24 | 44.27 | 0.23 | 0.32 | 100.15 | 0.84 |
| 10JB367G a12 | PS | 39.79 | 0.00 | 0.14 | 0.02 | 14.43 | 0.21 | 44.27 | 0.24 | 0.35 | 99.46 | 0.85 |
| av 10 jb 367 g big ol pheno? - rim | PS | 39.98 | 0.00 | 0.14 | 0.04 | 14.21 | 0.22 | 44.59 | 0.24 | 0.32 | 99.75 | 0.85 |
| 10JB367G b2 | PS | 39.69 | 0.02 | 0.12 | 0.04 | 16.85 | 0.24 | 42.68 | 0.22 | 0.33 | 100.22 | 0.82 |
| 10JB367G a16 | PS | 39.07 | 0.03 | 0.13 | 0.00 | 19.76 | 0.40 | 40.26 | 0.19 | 0.25 | 100.09 | 0.78 |
| 10JB367G b5 | PS | 39.02 | 0.01 | 0.14 | 0.05 | 19.76 | 0.28 | 40.04 | 0.23 | 0.29 | 99.81 | 0.78 |
| 10JB367G b6 | PS | 39.13 | 0.00 | 0.13 | 0.01 | 19.61 | 0.31 | 40.41 | 0.18 | 0.26 | 100.04 | 0.79 |
| av 10 jb 367 g ol gm, rims \& in plag | PS | 39.23 | 0.01 | 0.13 | 0.03 | 18.99 | 0.31 | 40.85 | 0.20 | 0.28 | 100.04 | 0.79 |
| 10JB367E c3 | PS | 40.47 | 0.00 | 0.14 | 0.06 | 12.44 | 0.16 | 45.54 | 0.26 | 0.32 | 99.42 | 0.87 |
| 10JB367E c4 | PS | 40.52 | 0.00 | 0.14 | 0.04 | 12.68 | 0.25 | 47.60 | 0.27 | 0.31 | 101.82 | 0.87 |
| 10JB367E b6 | PS | 41.00 | 0.00 | 0.15 | 0.05 | 11.79 | 0.21 | 46.76 | 0.27 | 0.36 | 100.62 | 0.88 |
| 10JB367E b7 | PS | 40.85 | 0.00 | 0.15 | 0.08 | 11.53 | 0.19 | 46.90 | 0.27 | 0.40 | 100.39 | 0.88 |
| 10JB367E a25 | PS | 40.94 | 0.01 | 0.23 | 0.05 | 11.00 | 0.15 | 46.87 | 0.63 | 0.37 | 100.24 | 0.88 |
| av 10 jb 367 e big ol pheno? Minus rim | PS | 40.75 | 0.00 | 0.16 | 0.06 | 11.89 | 0.19 | 46.73 | 0.34 | 0.35 | 100.50 | 0.88 |
| 10JB367E b8 | PS | 39.60 | 0.00 | 0.14 | 0.05 | 17.44 | 0.22 | 42.05 | 0.24 | 0.30 | 100.10 | 0.81 |
| 10JB367E b12 | PS | 38.71 | 0.01 | 0.15 | 0.00 | 20.74 | 0.21 | 39.26 | 0.26 | 0.25 | 99.60 | 0.77 |
| 10JB367E b13 | PS | 38.78 | 0.00 | 0.15 | 0.04 | 21.99 | 0.31 | 38.24 | 0.22 | 0.26 | 99.99 | 0.76 |
| 10JB367E b14 | PS | 39.58 | 0.01 | 0.14 | 0.03 | 19.44 | 0.23 | 40.84 | 0.20 | 0.29 | 100.77 | 0.79 |
| 10JB367E b15 | PS | 39.41 | 0.00 | 0.14 | 0.00 | 17.23 | 0.24 | 41.65 | 0.24 | 0.26 | 99.20 | 0.81 |
| 10JB367E a20 | PS | 38.86 | 0.00 | 0.13 | 0.05 | 20.01 | 0.34 | 39.15 | 0.19 | 0.25 | 98.99 | 0.78 |
| 10JB367E a21 | PS | 38.97 | 0.00 | 0.14 | 0.04 | 19.27 | 0.29 | 39.45 | 0.20 | 0.22 | 98.60 | 0.78 |
| 10JB367E a22 | PS | 39.00 | 0.00 | 0.14 | 0.00 | 20.27 | 0.28 | 39.32 | 0.16 | 0.30 | 99.47 | 0.78 |
| 10JB367E a23 | PS | 39.59 | 0.00 | 0.15 | 0.02 | 16.56 | 0.14 | 42.73 | 0.16 | 0.29 | 99.63 | 0.82 |
| av 10 jb 367 e ragged looking ol \& pheno rim | PS | 39.17 | 0.00 | 0.14 | 0.03 | 19.22 | 0.25 | 40.30 | 0.21 | 0.27 | 99.59 | 0.79 |
| JB367A 1 | PS | 39.50 | 0.00 | 0.02 | 0.04 | 13.54 | 0.22 | 45.67 | 0.29 | 0.33 | 99.62 | 0.86 |
| JB367A 2 | PS | 39.96 | 0.00 | 0.05 | 0.04 | 13.21 | 0.18 | 45.54 | 0.28 | 0.30 | 99.56 | 0.86 |
| 10JB367A a3 | PS | 40.09 | 0.01 | 0.17 | 0.06 | 13.12 | 0.17 | 45.23 | 0.25 | 0.31 | 99.42 | 0.86 |
| 10JB367A a4 | PS | 40.43 | 0.01 | 0.15 | 0.05 | 13.15 | 0.10 | 45.27 | 0.28 | 0.33 | 99.77 | 0.86 |
| JB367A 3 | PS | 39.23 | 0.00 | 0.06 | 0.13 | 13.59 | 0.30 | 45.76 | 0.30 | 0.31 | 99.69 | 0.86 |
| JB367A 4 | PS | 39.06 | 0.00 | 0.04 | 0.05 | 13.22 | 0.26 | 45.06 | 0.25 | 0.34 | 98.28 | 0.86 |
| 10JB367A a5 | PS | 40.44 | 0.02 | 0.14 | 0.05 | 13.23 | 0.09 | 46.56 | 0.28 | 0.30 | 101.12 | 0.86 |
| 10JB367A a6 | PS | 40.49 | 0.01 | 0.14 | 0.09 | 13.76 | 0.11 | 45.19 | 0.32 | 0.30 | 100.41 | 0.85 |


| Sample code | Sill | SiO 2 | TiO2 | $\begin{aligned} & \hline \mathrm{Al2} \\ & \mathrm{O} 3 \\ & \hline \end{aligned}$ | Cr2O3 | FeO | MnO | MgO | CaO | NiO | Total | Fo |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 10 JB 367 A a 1 | PS | 40.50 | 0.00 | 0.14 | 0.04 | 12.98 | 0.13 | 45.89 | 0.28 | 0.29 | 100.27 | 0.86 |
| JB367A 6 | PS | 39.28 | 0.00 | 0.04 | 0.01 | 13.73 | 0.20 | 45.30 | 0.25 | 0.34 | 99.14 | 0.85 |
| JB367A 7 | PS | 39.50 | 0.00 | 0.05 | 0.01 | 13.62 | 0.20 | 45.21 | 0.26 | 0.36 | 99.21 | 0.86 |
| 10JB367A b8 | PS | 40.58 | 0.01 | 0.14 | 0.07 | 13.18 | 0.21 | 45.77 | 0.27 | 0.27 | 100.51 | 0.86 |
| JB367A 18 | PS | 39.13 | 0.01 | 0.07 | 0.08 | 13.64 | 0.16 | 44.98 | 0.33 | 0.30 | 98.68 | 0.85 |
| 10JB367A b10 | PS | 40.71 | 0.00 | 0.14 | 0.02 | 13.49 | 0.18 | 45.97 | 0.24 | 0.34 | 101.09 | 0.86 |
| JB367A 19 | PS | 38.57 | 0.00 | 0.05 | 0.07 | 13.38 | 0.22 | 45.24 | 0.25 | 0.30 | 98.07 | 0.86 |
| JB367A 20 | PS | 39.06 | 0.00 | 0.04 | 0.02 | 13.17 | 0.21 | 45.50 | 0.24 | 0.33 | 98.56 | 0.86 |
| 10JB367A b11 | PS | 40.79 | 0.00 | 0.13 | 0.02 | 13.77 | 0.17 | 45.42 | 0.23 | 0.33 | 100.86 | 0.85 |
| JB367A 21 | PS | 38.34 | 0.00 | 0.07 | 0.07 | 13.40 | 0.19 | 44.76 | 0.39 | 0.30 | 97.51 | 0.86 |
| JB367A 22 | PS | 39.13 | 0.00 | 0.06 | 0.04 | 14.46 | 0.20 | 44.75 | 0.26 | 0.33 | 99.22 | 0.85 |
| 10JB367B c1 | PS | 40.37 | 0.00 | 0.15 | 0.03 | 13.55 | 0.26 | 45.55 | 0.23 | 0.29 | 100.45 | 0.86 |
| 10JB367B c2 | PS | 40.15 | 0.00 | 0.15 | 0.04 | 13.58 | 0.24 | 45.58 | 0.23 | 0.30 | 100.30 | 0.86 |
| 10JB367B c9 | PS | 40.04 | 0.00 | 0.14 | 0.06 | 13.42 | 0.21 | 45.21 | 0.22 | 0.36 | 99.67 | 0.86 |
| 10JB367B b11 | PS | 41.33 | 0.02 | 0.14 | 0.05 | 10.59 | 0.15 | 48.55 | 0.13 | 0.41 | 101.38 | 0.89 |
| 10JB367B b12 | PS | 40.46 | 0.01 | 0.14 | 0.03 | 11.67 | 0.23 | 47.09 | 0.10 | 0.38 | 100.13 | 0.88 |
| 10JB367B a20 | PS | 40.63 | 0.00 | 0.15 | 0.09 | 10.31 | 0.14 | 48.55 | 0.25 | 0.37 | 100.51 | 0.89 |
| 10JB367B a21 | PS | 41.08 | 0.00 | 0.16 | 0.06 | 10.11 | 0.17 | 48.55 | 0.30 | 0.33 | 100.77 | 0.90 |
| 10JB367B a28 | PS | 47.39 | 0.01 | 0.17 | 0.03 | 9.78 | 0.16 | 36.71 | 0.16 | 0.13 | 94.59 | 0.87 |
| 10JB367B a29 | PS | 40.42 | 0.00 | 0.15 | 0.04 | 15.81 | 0.25 | 44.00 | 0.28 | 0.22 | 101.18 | 0.83 |
| 10JB367B c4 | PS | 39.91 | 0.00 | 0.15 | 0.04 | 15.27 | 0.19 | 43.70 | 0.21 | 0.27 | 99.74 | 0.84 |
| 10JB367B c6 | PS | 34.63 | 0.00 | 0.15 | 0.03 | 23.70 | 0.45 | 37.30 | 0.24 | 0.21 | 96.72 | 0.74 |
| 10JB367B b15 | PS | 38.23 | 0.00 | 0.13 | 0.01 | 24.37 | 0.24 | 36.60 | 0.21 | 0.20 | 100.03 | 0.73 |
| 10JB367B b18 | PS | 37.73 | 0.01 | 0.13 | 0.00 | 28.26 | 0.42 | 33.85 | 0.16 | 0.25 | 100.82 | 0.68 |
| 10JB367B a22 | PS | 39.87 | 0.00 | 0.16 | 0.05 | 16.15 | 0.25 | 43.49 | 0.24 | 0.28 | 100.55 | 0.83 |
| 10JB367B a23 | PS | 39.32 | 0.00 | 0.15 | 0.02 | 19.11 | 0.28 | 41.05 | 0.22 | 0.27 | 100.46 | 0.79 |
| 10JB367B c3 | PS | 39.20 | 0.00 | 0.15 | 0.03 | 18.98 | 0.24 | 41.22 | 0.25 | 0.24 | 100.32 | 0.79 |
| 10JB367B a26 | PS | 39.83 | 0.00 | 0.17 | 0.08 | 17.49 | 0.21 | 42.51 | 0.24 | 0.23 | 100.81 | 0.81 |
| 10 KS 95 A 3 a 1 | WUS | 39.77 | 0.01 | 0.14 | 0.00 | 16.38 | 0.22 | 43.63 | 0.15 | 0.25 | 100.57 | 0.83 |
| 10 KS 95 A 3 a 2 | WUS | 39.15 | 0.00 | 0.13 | 0.00 | 15.91 | 0.21 | 43.65 | 0.14 | 0.29 | 99.48 | 0.83 |
| 10KS95 A3 b16 | WUS | 40.28 | 0.00 | 0.13 | 0.03 | 13.10 | 0.25 | 45.59 | 0.22 | 0.34 | 99.94 | 0.86 |
| 10KS95 A3 b11 | WUS | 39.49 | 0.00 | 0.14 | 0.05 | 17.75 | 0.31 | 42.61 | 0.21 | 0.27 | 100.83 | 0.81 |
| 10KS95 A3 d35 | WUS | 40.01 | 0.00 | 0.12 | 0.04 | 17.09 | 0.22 | 42.92 | 0.20 | 0.34 | 100.93 | 0.82 |
| 10KS95 A3 d36 | WUS | 39.43 | 0.00 | 0.14 | 0.04 | 15.97 | 0.23 | 43.56 | 0.17 | 0.29 | 99.82 | 0.83 |
| 10KS95 A3 a6 | WUS | 39.28 | 0.00 | 0.13 | 0.00 | 18.71 | 0.28 | 41.84 | 0.21 | 0.24 | 100.70 | 0.80 |
| 10KS95 A3 a 7 | WUS | 39.34 | 0.00 | 0.14 | 0.01 | 19.25 | 0.24 | 40.99 | 0.19 | 0.24 | 100.41 | 0.79 |
| 10KS95 A3 b17 | WUS | 39.44 | 0.01 | 0.15 | 0.04 | 18.65 | 0.28 | 41.65 | 0.28 | 0.27 | 100.77 | 0.80 |
| 10KS95 A3 b18 | WUS | 39.01 | 0.02 | 0.12 | 0.02 | 19.93 | 0.31 | 40.34 | 0.13 | 0.23 | 100.11 | 0.78 |
| 10KS95 A3 c19 | WUS | 38.73 | 0.00 | 0.12 | 0.02 | 20.75 | 0.31 | 40.19 | 0.16 | 0.24 | 100.54 | 0.78 |
| 10KS95 A3 c26 | WUS | 39.42 | 0.01 | 0.13 | 0.01 | 20.35 | 0.34 | 40.62 | 0.09 | 0.26 | 101.25 | 0.78 |
| 10KS95 A3 c29 | WUS | 39.13 | 0.00 | 0.14 | 0.02 | 18.31 | 0.31 | 41.43 | 0.11 | 0.25 | 99.70 | 0.80 |
| 10KS95 A1 c11 | WUS | 40.65 | 0.00 | 0.14 | 0.03 | 13.59 | 0.21 | 45.43 | 0.20 | 0.31 | 100.57 | 0.86 |
| 10KS95 A1 c12 | WUS | 40.24 | 0.00 | 0.13 | 0.07 | 13.47 | 0.16 | 45.61 | 0.20 | 0.32 | 100.21 | 0.86 |
| 10KS95 A1 c13 | WUS | 39.68 | 0.00 | 0.14 | 0.04 | 12.12 | 0.15 | 45.85 | 0.24 | 0.34 | 98.56 | 0.87 |
| 10KS95 A1 c14 | WUS | 39.95 | 0.02 | 0.14 | 0.03 | 13.72 | 0.08 | 45.50 | 0.23 | 0.29 | 99.96 | 0.86 |
| 10KS95 A1 b20 | WUS | 40.31 | 0.00 | 0.13 | 0.06 | 12.68 | 0.11 | 46.12 | 0.27 | 0.32 | 100.00 | 0.87 |
| 10KS95 A1 a34 | WUS | 40.12 | 0.00 | 0.13 | 0.05 | 13.51 | 0.25 | 45.65 | 0.27 | 0.24 | 100.23 | 0.86 |
| 10KS95 A1 a35 | WUS | 40.27 | 0.00 | 0.13 | 0.03 | 13.95 | 0.24 | 45.49 | 0.24 | 0.24 | 100.62 | 0.85 |


| Sample code | Sill | SiO2 | TiO2 | $\begin{aligned} & \mathrm{Al2} \\ & \mathrm{O} 3 \\ & \hline \end{aligned}$ | Cr 2 O 3 | FeO | MnO | MgO | CaO | NiO | Total | Fo |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 10KS95 A1 a36 | WUS | 40.00 | 0.00 | 0.14 | 0.03 | 15.70 | 0.22 | 43.75 | 0.24 | 0.27 | 100.36 | 0.83 |
| 10KS95 A1 a41 | WUS | 40.26 | 0.00 | 0.14 | 0.03 | 14.62 | 0.19 | 45.08 | 0.18 | 0.22 | 100.74 | 0.85 |
| 10KS95 A1 a42 | WUS | 41.10 | 0.00 | 0.14 | 0.05 | 13.69 | 0.21 | 45.38 | 0.22 | 0.21 | 101.00 | 0.86 |
| av 10 ks 95 A 1 ol pheno | WUS | 40.26 | 0.00 | 0.14 | 0.04 | 13.70 | 0.18 | 45.38 | 0.23 | 0.28 | 100.22 | 0.86 |
| 10KS95 A1 b22 | WUS | 39.90 | 0.01 | 0.15 | 0.03 | 15.42 | 0.30 | 44.47 | 0.14 | 0.33 | 100.75 | 0.84 |
| 10KS95 A1 b23 | WUS | 39.61 | 0.00 | 0.15 | 0.02 | 16.34 | 0.26 | 43.18 | 0.20 | 0.27 | 100.04 | 0.82 |


| 10KS95 A1 b24 | WUS | 39.17 | 0.00 | 0.14 | 0.06 | 16.04 | 0.23 | 42.78 | 0.19 | 0.22 | 98.84 | 0.83 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| av 10 ks 95 A 1 ol incl in skeletal cpx pheno | WUS | 39.56 | 0.00 | 0.14 | 0.04 | 15.93 | 0.26 | 43.48 | 0.18 | 0.28 | 99.88 | 0.83 |
| 10KS95 A1 a43 | WUS | 39.15 | 0.00 | 0.13 | 0.07 | 18.90 | 0.27 | 40.96 | 0.10 | 0.26 | 99.84 | 0.79 |
| 10KS95 A1 b21 | WUS | 38.34 | 0.01 | 0.15 | 0.02 | 21.00 | 0.30 | 39.25 | 0.20 | 0.24 | 99.51 | 0.77 |
| 10KS95 A1 d1 | WUS | 38.79 | 0.00 | 0.14 | 0.02 | 19.65 | 0.28 | 40.53 | 0.25 | 0.20 | 99.85 | 0.79 |
| 10KS95 A1 d2 | WUS | 39.35 | 0.00 | 0.15 | 0.02 | 19.21 | 0.27 | 40.63 | 0.25 | 0.27 | 100.14 | 0.79 |
| 10KS95 A1 d3 | WUS | 38.87 | 0.00 | 0.13 | 0.02 | 21.68 | 0.35 | 39.00 | 0.22 | 0.23 | 100.50 | 0.76 |
| 10KS95 A1 d5 | WUS | 39.44 | 0.00 | 0.14 | 0.03 | 18.43 | 0.24 | 41.81 | 0.14 | 0.28 | 100.51 | 0.80 |
| av 10ks95A1 skeletal ol \& pheno rims | WUS | 38.99 | 0.00 | 0.14 | 0.03 | 19.81 | 0.28 | 40.36 | 0.19 | 0.25 | 100.06 | 0.78 |
| 10JB368A a1 | PS | 39.17 | 0.00 | 0.14 | 0.03 | 17.98 | 0.31 | 41.98 | 0.28 | 0.29 | 100.19 | 0.81 |
| 10JB368A a2 | PS | 39.11 | 0.02 | 0.15 | 0.01 | 18.86 | 0.25 | 41.06 | 0.23 | 0.26 | 99.96 | 0.80 |
| 10JB368A a3 | PS | 39.22 | 0.00 | 0.15 | 0.01 | 19.28 | 0.34 | 40.82 | 0.18 | 0.24 | 100.25 | 0.79 |
| 10JB368A a10 | PS | 38.73 | 0.01 | 0.14 | 0.02 | 19.30 | 0.31 | 40.57 | 0.18 | 0.25 | 99.51 | 0.79 |
| 10JB368A b21 | PS | 39.20 | 0.00 | 0.13 | 0.03 | 17.89 | 0.28 | 41.69 | 0.17 | 0.25 | 99.64 | 0.81 |
| 10JB368A b22 | PS | 38.81 | 0.00 | 0.14 | 0.06 | 18.34 | 0.20 | 41.64 | 0.16 | 0.29 | 99.64 | 0.80 |
| 10JB368A b23 | PS | 39.09 | 0.00 | 0.14 | 0.04 | 19.23 | 0.25 | 40.65 | 0.23 | 0.22 | 99.86 | 0.79 |
| 10JB368A b25 | PS | 39.32 | 0.00 | 0.13 | 0.00 | 19.40 | 0.22 | 40.62 | 0.24 | 0.24 | 100.19 | 0.79 |
| av 10jb368a ol pheno, crown \& gm | PS | 39.08 | 0.00 | 0.14 | 0.02 | 18.78 | 0.27 | 41.13 | 0.21 | 0.26 | 99.90 | 0.80 |
| 10JB368A c27 | PS | 39.89 | 0.00 | 0.14 | 0.02 | 15.52 | 0.18 | 44.09 | 0.28 | 0.24 | 100.35 | 0.84 |
| 10JB368A c28 | PS | 39.78 | 0.00 | 0.15 | 0.05 | 14.73 | 0.12 | 44.28 | 0.21 | 0.23 | 99.57 | 0.84 |
| 10JB368A c35 | PS | 39.35 | 0.00 | 0.15 | 0.04 | 17.18 | 0.32 | 42.37 | 0.21 | 0.27 | 99.87 | 0.81 |
| 10JB368A c36 | PS | 39.21 | 0.02 | 0.15 | 0.07 | 16.53 | 0.10 | 43.03 | 0.16 | 0.28 | 99.55 | 0.82 |
| av 10jb368a ol incl in px oiks | PS | 39.55 | 0.01 | 0.14 | 0.04 | 15.99 | 0.18 | 43.44 | 0.21 | 0.25 | 99.83 | 0.83 |
| NW38A 15 | WUS | 36.76 | 0.02 | 0.01 | 0.00 | 30.99 | 0.43 | 30.81 | 0.18 | 0.17 | 99.38 | 0.64 |
| NW38A 16 | WUS | 36.93 | 0.00 | 0.00 | 0.03 | 31.72 | 0.41 | 30.64 | 0.12 | 0.12 | 99.97 | 0.63 |
| NW38A 17 | WUS | 36.67 | 0.00 | 0.07 | 0.14 | 31.20 | 0.41 | 30.68 | 0.11 | 0.13 | 99.42 | 0.64 |
| NW38A 18 | WUS | 36.81 | 0.03 | 0.00 | 0.01 | 31.67 | 0.43 | 30.14 | 0.12 | 0.11 | 99.31 | 0.63 |
| NW38A 19 | WUS | 36.88 | 0.01 | 0.04 | 0.00 | 31.19 | 0.42 | 30.74 | 0.21 | 0.15 | 99.64 | 0.64 |
| NW38A 20 | WUS | 36.56 | 0.01 | 0.00 | 0.00 | 31.69 | 0.33 | 30.18 | 0.14 | 0.10 | 99.04 | 0.63 |
| NW38A 21 | WUS | 36.60 | 0.00 | 0.03 | 0.00 | 32.04 | 0.46 | 31.10 | 0.17 | 0.11 | 100.50 | 0.63 |
| NW38A 22 | WUS | 36.68 | 0.04 | 0.05 | 0.02 | 32.08 | 0.49 | 30.10 | 0.35 | 0.15 | 99.96 | 0.63 |
| av NW38A ol phenos | WUS | 36.74 | 0.01 | 0.03 | 0.03 | 31.57 | 0.42 | 30.55 | 0.17 | 0.13 | 99.65 | 0.63 |
| AW54B1 23 | WUS | 40.32 | 0.01 | 0.03 | 0.05 | 12.83 | 0.08 | 45.89 | 0.28 | 0.35 | 99.83 | 0.86 |
| AW54B1 25 | WUS | 40.24 | 0.00 | 0.02 | 0.05 | 13.61 | 0.15 | 44.97 | 0.23 | 0.24 | 99.52 | 0.85 |
| AW54B1 26 | WUS | 40.12 | 0.00 | 0.03 | 0.08 | 13.72 | 0.19 | 44.92 | 0.29 | 0.30 | 99.66 | 0.85 |
| AW54B1 28 | WUS | 40.20 | 0.00 | 0.01 | 0.04 | 13.75 | 0.23 | 44.63 | 0.22 | 0.24 | 99.34 | 0.85 |
| AW54B1 30 | WUS | 40.33 | 0.00 | 0.02 | 0.09 | 13.54 | 0.12 | 44.96 | 0.25 | 0.30 | 99.62 | 0.86 |
| AW54B1 31 | WUS | 40.21 | 0.01 | 0.11 | 0.10 | 13.09 | 0.19 | 44.74 | 0.43 | 0.37 | 99.27 | 0.86 |
| AW54B1 32 | WUS | 40.10 | 0.00 | 0.03 | 0.03 | 14.14 | 0.19 | 44.99 | 0.28 | 0.30 | 100.08 | 0.85 |
| AW54B1 34 | WUS | 39.98 | 0.02 | 0.03 | 0.08 | 13.73 | 0.16 | 45.32 | 0.23 | 0.28 | 99.86 | 0.85 |
| AW54B1 37 | WUS | 39.92 | 0.00 | 0.06 | 0.07 | 13.32 | 0.20 | 45.67 | 0.31 | 0.31 | 99.90 | 0.86 |
| AW54B1 39 | WUS | 39.83 | 0.00 | 0.07 | 0.15 | 12.47 | 0.16 | 46.01 | 0.33 | 0.30 | 99.34 | 0.87 |
| av AW54B1 ol cores | WUS | 40.13 | 0.01 | 0.04 | 0.07 | 13.42 | 0.17 | 45.21 | 0.28 | 0.30 | 99.64 | 0.86 |
| AW54B1 24 | WUS | 39.54 | 0.00 | 0.03 | 0.05 | 16.56 | 0.24 | 42.59 | 0.15 | 0.30 | 99.46 | 0.82 |
| AW54B1 29 | WUS | 39.44 | 0.02 | 0.03 | 0.05 | 17.24 | 0.22 | 41.86 | 0.28 | 0.24 | 99.37 | 0.81 |
| AW54B1 33 | WUS | 38.74 | 0.01 | 0.05 | 0.05 | 19.76 | 0.24 | 40.17 | 0.27 | 0.23 | 99.53 | 0.78 |

## Appendix F3 - FCSC sulphur isotope data

| Sulphur isotopes were analysed at McGill University, Montreal - details of methods are included in Hryciuk et al Prep) |  |  |  |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :---: |
|  |  |  |  |  |  |  |
| SILL | SAMPLE ID | LITHOLOGY | HEIGHT FROM BASE (m) | $\mathbf{\delta}^{34} \mathbf{S}$ (VCDT) | S (wt \%) |  |
|  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |
| LPS | 10BH166 A1 | LCZ | 0.00 | 3.69 | 0.03 |  |
| LPS | 10BH166 A5 | OZ | 2.72 | 3.96 | 0.03 |  |
| LPS | 10BH167 A2 | OZ | 3.64 | 4.10 | 0.02 |  |
| LPS | 10BH167 A4 | OZ | 7.28 | 3.72 | 0.03 |  |
| LPS | 10BH167 A5 | CPZ | 8.97 | 4.07 | 0.05 |  |
| LPS | 10BH167 A8 | DZ | 12.07 | 6.66 | 0.17 |  |
| LPS | 10BH167 A11 | DZ | 16.88 | 5.73 | 0.03 |  |
| LPS | 11BH311 B8 | UCZ | 21.43 | 6.94 | 0.13 |  |
| PS | 10RAT-JB287A01 | LCZ | 0.00 | 5.79 | 0.05 |  |
| PS | 10RAT-JB367A01 | LBZ | 3.00 | 3.77 | 0.05 |  |
| PS | 10RAT-JB367E01 | LBZ | 4.00 | 4.84 | 0.03 |  |
| PS | 10RAT-JB367G01 | OZ | 7.00 | 3.27 | 0.03 |  |
| PS | 10RAT-JB368B01 | OZ | 9.00 | 3.57 | 0.03 |  |
| PS | 10RAT-JB368D01 | DZ | 12.00 | 8.02 | 0.05 |  |
| PS | 10RAT-JB368G01 | DZ | 16.00 | 3.53 | 0.04 |  |
| PS | 10RAT-JB369A01 | DZ | 18.00 | 5.12 | 0.08 |  |

## Appendix F5 - Dolostone whole-rock data

| SAMPLE ID | SiO2 | TiO2 | A12O3 | Fe2O3 | FeOt | FeOt | MnO | MgO | CaO | Na2O | K2O | P205 | Cr2O3 | S wt\% | LOI | H2OT | CO2T | TOT |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 2388-2 | 1.18 | 0.02 | 0.39 | 1.87 | 1.68 |  | 0.27 | 8.09 | 46.74 | 0.03 | 0.05 | 0.04 | 0.01 |  | 41.40 |  |  | 100.10 |
| 87RAR-5B-10 | 16.40 | 0.05 | 0.80 | 1.30 | 1.17 | 1.40 | 0.08 | 17.30 | 24.30 | 0.01 | 0.21 | 0.01 |  | 0.01 |  | 0.70 | 38.40 | 99.60 |
| 92RAT-M104 | 20.90 | 0.18 | 2.50 | 2.80 | 2.52 | 2.70 | 0.10 | 13.20 | 23.70 | 0.01 | 0.83 | 0.03 |  |  |  | 0.80 | 34.80 | 99.90 |
| 93RAT-10 | 9.70 | 0.11 | 1.70 | 1.00 | 0.90 | 1.10 | 0.06 | 18.00 | 26.40 | 0.01 | 0.80 | 0.03 |  | <dl |  | 0.60 | 41.30 | 99.70 |
| 0730A011 | 9.76 | 0.04 | 0.57 | 1.24 | 1.11 |  | 0.04 | 17.86 | 28.13 | 0.03 | 0.26 | 0.01 | 0.00 | 0.12 | 42.00 |  |  | 99.94 |
| HDB-2005-100 |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| GNME-06-04/009 | 16.46 | 0.16 | 2.88 | 2.62 | 2.36 |  | 0.03 | 10.37 | 38.07 | 0.25 | 0.09 | 0.06 | 0.00 | 0.05 | 29.00 |  | 7.32 | 99.99 |
| GNME-06-04/010 | 18.41 | 0.17 | 2.98 | 2.45 | 2.20 |  | 0.03 | 10.28 | 37.21 | 0.24 | 0.06 | 0.06 | 0.00 | 0.25 | 28.10 |  | 7.35 | 99.99 |
| GNME-06-04/121 | 14.62 | 0.05 | 0.99 | 1.11 | 1.00 |  | 0.03 | 18.82 | 31.45 | 0.01 | 0.02 | 0.01 | 0.00 | 0.12 | 32.90 |  | 7.27 | 99.99 |
| GNME-06-04/122 | 14.85 | 0.14 | 2.49 | 1.60 | 1.44 |  | 0.04 | 19.74 | 30.56 | 0.01 | 0.24 | 0.02 | 0.00 | 0.77 | 30.30 |  | 6.40 | 99.99 |
| GNME-06-05/005 | 14.83 | 0.12 | 1.92 | 1.04 | 0.93 |  | 0.02 | 10.90 | 38.03 | 0.11 | 0.32 | 0.02 | 0.00 | 0.16 | 32.60 |  | 8.58 | 99.91 |
| GNME-06-05/032 | 14.62 | 0.07 | 1.49 | 3.18 | 2.86 |  | 0.02 | 22.08 | 26.99 | 0.01 | 0.02 | 0.01 | 0.00 | 0.07 | 31.40 |  | 6.19 | 99.87 |
| GNME-06-05/033 | 13.44 | 0.07 | 1.28 | 1.27 | 1.14 |  | 0.03 | 23.03 | 27.45 | 0.01 | 0.02 | 0.03 | 0.00 | 0.01 | 33.30 |  | 6.43 | 99.92 |
| GNME-06-05/129 | 10.35 | 0.06 | 1.00 | 0.70 | 0.63 |  | 0.02 | 19.57 | 32.78 | 0.01 | 0.02 | 0.02 | 0.00 | 0.14 | 35.40 |  | 8.80 | 99.93 |
| GNME-06-05/130 | 8.17 | 0.07 | 1.13 | 0.83 | 0.75 |  | 0.01 | 19.27 | 36.95 | 0.02 | 0.04 | 0.03 | 0.00 | 3.00 | 32.60 |  | 8.35 | 99.12 |
| GNME-06-05/131 | 12.28 | 0.06 | 1.11 | 0.67 | 0.60 |  | 0.03 | 20.66 | 31.16 | 0.01 | 0.09 | 0.04 | 0.00 | 0.13 | 33.80 |  | 7.84 | 99.91 |
| 10RAT-RR197B01 | 2.39 | 0.01 | 0.19 | 0.55 | 0.50 |  | 0.10 | 19.08 | 29.05 | < 0.07 | < 0.08 | < 0.03 |  | 0.07 | 45.70 |  |  | 97.20 |
| 10RAT-HY074A01b | 16.69 | 0.17 | 3.08 | 1.32 | 1.19 |  | 0.04 | 17.15 | 27.04 | 0.14 | 0.67 | 0.04 |  | 0.38 | 33.20 |  |  | 99.90 |
| 10RAT-HY074A02b | 12.92 | 0.16 | 2.80 | 1.53 | 1.37 |  | 0.04 | 18.20 | 29.43 | 0.09 | 0.85 | 0.04 |  | 0.42 | 34.30 |  |  | 100.80 |
| 11-RAT-BH188A1 | 7.10 | 0.02 | 0.50 | 0.38 | 0.34 |  | 0.01 | 19.60 | 31.00 | < 0,05 | 0.09 | 0.01 |  | 0.18 | 39.23 |  |  | 98.00 |
| 11-RAT-BH367A1 | 1.50 | 0.01 | 0.10 | 0.98 | 0.88 |  | 0.03 | 17.40 | 32.00 | <0,05 | <0,007 | <0,01 |  | 0.43 | 44.17 |  |  | 97.00 |
| 11-RAT-BH369A1 | 1.50 | 0.01 | 0.20 | 0.68 | 0.61 |  | 0.03 | 22.00 | 35.60 | 0.14 | <0,007 | 0.01 |  | 0.26 | 38.81 |  |  | 99.00 |
| Dolostone average | 11.34 | 0.08 | 1.43 | 1.39 | 1.25 | 1.73 | 0.05 | 17.27 | 31.62 | 0.06 | 0.26 | 0.03 | 0.00 | 0.36 | 35.46 | 0.70 | 14.54 | 99.51 |

## Appendix G1 - SFDC whole-rock data

| ZONE | D | D | D | D | D | D | D | D |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| LITHOLOGY | dyke core | dyke core | dyke | dyke core | dyke core | dyke chill | capping sill | dyke core |
|  |  |  | internal chill |  |  |  |  |  |
| SAMPLE I.D. | 11MH123B2 | 11BH296A1 | 11BH268A2 | 11BH270A2 | 11BH266A2 | 11MH126A2 | 11MH128A2 | 11BH264A2 |
| Lab | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff |
| SiO2 | 47.85 | 47.21 | 48.75 | 47.92 | 47.80 | 47.06 | 46.91 | 46.99 |
| TiO2 | 1.23 | 1.90 | 1.56 | 1.32 | 1.38 | 1.29 | 1.07 | 1.96 |
| Al2O3 | 13.39 | 11.92 | 12.60 | 12.76 | 13.17 | 13.11 | 13.90 | 10.84 |
| Fe2O3T | 13.88 | 17.72 | 15.46 | 12.40 | 14.52 | 13.90 | 12.57 | 19.36 |
| FeO* | 12.48 | 15.93 | 13.90 | 11.15 | 13.05 | 12.49 | 11.30 | 17.41 |
| Fe2O3* | 1.40 | 1.79 | 1.56 | 1.25 | 1.47 | 1.40 | 1.27 | 1.96 |
| MnO | 0.20 | 0.23 | 0.22 | 0.18 | 0.20 | 0.20 | 0.19 | 0.26 |
| MgO | 8.19 | 5.63 | 7.09 | 7.45 | 7.71 | 8.47 | 8.43 | 5.57 |
| CaO | 11.22 | 9.92 | 11.66 | 11.66 | 12.25 | 11.67 | 13.53 | 9.77 |
| Na 2 O | 2.03 | 1.95 | 2.36 | 2.21 | 2.08 | 2.07 | 1.65 | 2.59 |
| K2O | 0.49 | 1.77 | 0.72 | 1.31 | 0.52 | 0.83 | 0.47 | 0.79 |
| P2O5 | 0.11 | 0.17 | 0.12 | 0.11 | 0.12 | 0.09 | 0.09 | 0.21 |
| LOI | 0.79 | 0.71 | 0.79 | 1.57 | 1.04 | 0.82 | 0.87 | 1.18 |
| TOTAL | 99.37 | 116.85 | 101.34 | 98.89 | 100.78 | 113.41 | 99.66 | 99.53 |
| Sc | 40.48 | 41.80 | 44.98 | 44.45 | 44.22 | 42.11 | 42.62 | 40.68 |
| V | 367.20 | 464.97 | 462.38 | 401.62 | 427.53 | 415.06 | 298.90 | 474.13 |
| Cr | 237.27 | 5.00 | 56.76 | 228.72 | 197.49 | 211.54 | 304.43 | 5.78 |
| Co | 66.48 | 83.35 | 75.74 | 58.67 | 69.21 | 66.35 | 56.16 | 90.57 |
| Ni | 193.65 | 7.45 | 30.08 | 73.22 | 50.38 | 64.61 | 95.52 | 39.53 |
| Cu | 122.08 | 207.33 | 147.31 | 117.08 | 133.78 | 135.47 | 104.30 | 166.74 |
| Zn | 93.99 | 111.66 | 80.52 | 115.53 | 89.23 | 83.14 | 74.79 | 123.49 |
| Sr | 205.85 | 407.41 | 262.77 | 425.98 | 217.09 | 270.42 | 172.07 | 304.64 |
| Y | 22.05 | 32.53 | 27.06 | 23.95 | 25.26 | 23.25 | 18.70 | 35.60 |
| Zr | 58.37 | 133.91 | 97.34 | 65.35 | 96.16 | 86.36 | 63.41 | 140.64 |
| Ba | 184.80 | 244.08 | 107.74 | 155.74 | 100.80 | 201.41 | 81.76 | 149.62 |
| Ti | 1.23 | 1.95 | 1.49 | 1.30 | 1.36 | 1.28 | 1.08 | 1.96 |
| V | 337.80 | 497.70 | 416.00 | 338.00 | 378.60 | 398.20 | 323.40 | 479.50 |
| Cr | 222.30 | 4.00 | 63.10 | 220.80 | 189.60 | 208.30 | 283.80 | 9.70 |
| Co | 65.10 | 79.20 | 75.70 | 54.50 | 64.70 | 65.90 | 55.40 | 87.80 |
| Ni | 192.60 | 11.80 | 34.40 | 77.20 | 50.70 | 60.00 | 81.80 | 41.90 |
| Cu | 119.60 | 227.00 | 143.10 | 111.70 | 131.70 | 129.30 | 107.80 | 177.50 |
| Zn | 87.00 | 117.50 | 77.50 | 108.10 | 84.90 | 80.20 | 78.70 | 123.30 |
| Rb | 11.10 | 50.40 | 21.50 | 37.20 | 12.80 | 22.10 | 13.80 | 22.30 |
| K | 4946.76 | 17668.01 | 7213.11 | 13093.01 | 5191.70 | 8271.14 | 4652.79 | 7942.05 |
| Sr | 194.10 | 411.70 | 261.90 | 419.40 | 206.50 | 3333.00 | 174.20 | 303.30 |
| P | 1056.73 | 1682.99 | 1246.97 | 1078.19 | 1177.46 | 940.65 | 862.70 | 2139.32 |
| Y | 21.40 | 36.30 | 26.90 | 20.60 | 25.40 | 23.40 | 19.90 | 38.00 |
| Zr | 73.00 | 133.50 | 96.70 | 66.80 | 95.00 | 89.40 | 66.60 | 134.70 |
| Nb | 3.39 | 6.29 | 4.82 | 2.95 | 4.69 | 4.01 | 2.80 | 6.52 |
| Sn | 1.72 | 1.19 | 0.99 | 1.95 | 0.82 | 1.00 | 1.44 | 0.83 |
| Cs | 0.55 | 0.92 | 1.51 | 2.02 | 0.81 | 0.95 | 0.95 | 0.62 |
| Ba | 176.40 | 237.20 | 110.70 | 143.40 | 107.40 | 206.00 | 86.00 | 145.50 |
| La | 5.53 | 8.49 | 6.73 | 5.51 | 6.97 | 5.96 | 4.94 | 10.63 |
| Ce | 13.56 | 20.84 | 16.66 | 13.30 | 16.87 | 14.47 | 11.96 | 25.58 |
| Pr | 1.96 | 3.08 | 2.43 | 1.93 | 2.38 | 2.13 | 1.75 | 3.72 |
| Nd | 9.25 | 14.30 | 11.54 | 8.90 | 11.15 | 9.68 | 8.07 | 17.29 |
| Sm | 2.62 | 4.10 | 3.41 | 2.60 | 3.26 | 2.84 | 2.47 | 4.89 |
| Eu | 0.99 | 1.38 | 1.16 | 0.97 | 1.11 | 1.04 | 0.86 | 1.52 |
| Gd | 3.21 | 4.91 | 3.88 | 2.89 | 3.62 | 3.34 | 2.88 | 5.55 |
| Tb | 0.55 | 0.91 | 0.69 | 0.52 | 0.67 | 0.61 | 0.49 | 0.94 |
| Dy | 3.63 | 5.52 | 4.41 | 3.44 | 4.21 | 3.94 | 3.24 | 6.00 |
| Ho | 0.64 | 1.07 | 0.78 | 0.64 | 0.77 | 0.71 | 0.59 | 1.05 |
| Er | 1.89 | 3.06 | 2.32 | 1.85 | 2.24 | 2.07 | 1.75 | 3.12 |
| Tm | 0.30 | 0.47 | 0.37 | 0.29 | 0.36 | 0.32 | 0.27 | 0.51 |
| Yb | 1.81 | 3.03 | 2.32 | 1.83 | 2.18 | 2.03 | 1.75 | 3.17 |
| Lu | 0.29 | 0.47 | 0.35 | 0.29 | 0.34 | 0.31 | 0.26 | 0.46 |
| Hf | 1.82 | 3.52 | 2.50 | 1.65 | 2.50 | 2.24 | 1.63 | 3.45 |
| Ta | 0.26 | 0.41 | 0.32 | 0.22 | 0.31 | 0.27 | 0.20 | 0.45 |


| Pb | 1.92 | 7.50 | 1.58 | 1.44 | 1.67 | 1.90 |  | 2.85 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| Th | 0.76 | 1.39 | 0.98 | 0.76 | 0.97 | 0.82 |  |  |
| U | 0.18 | 0.30 | 0.20 | 0.17 | 0.23 | 0.71 |  |  |
|  |  |  |  |  |  |  |  |  |
| Os | 0.21 | $<0.03$ | 0.07 |  | 0.15 |  |  |  |
| Ir | 0.44 | 0.17 | 0.04 |  |  |  |  |  |
| Ru | 0.83 | 0.30 | 0.44 |  |  |  |  |  |
| Rh | 1.01 | 17.7 | 9.07 | 16.6 |  |  | 0.06 |  |
| Pt | 22.4 | 4.38 |  |  |  | 0.04 |  |  |
| Au |  |  |  |  | 0.37 |  |  |  |


| ZONE | D | D | D | D | C | C | C | C |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\begin{aligned} & \text { LITHOLOG } \\ & \text { Y } \end{aligned}$ | dyke core | dyke core | dyke core | dyke core | dyke core | dyke core | small dyke | dyke chill |
| $\begin{aligned} & \text { SAMPLE } \\ & \text { I.D. } \\ & \hline \end{aligned}$ | $\begin{aligned} & \text { 11MH124B } \\ & 2 \end{aligned}$ | $\begin{aligned} & \text { 11MH271A } \\ & 2 \end{aligned}$ | $\begin{aligned} & \text { 11MH125A } \\ & 2 \end{aligned}$ | $\begin{aligned} & \text { 11BH269A } \\ & 2 \end{aligned}$ | $\begin{aligned} & \text { 11MH129A } \\ & 2 \end{aligned}$ | $\begin{aligned} & \text { 11BH239A } \\ & 2 \end{aligned}$ | $\begin{aligned} & \text { 11MH131A } \\ & 2 \end{aligned}$ | $\begin{aligned} & \text { 11MH130A } \\ & 2 \end{aligned}$ |
| Lab | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff |
| SiO2 | 47.09 | 47.05 | 47.80 | 48.40 | 47.72 | 48.32 | 47.27 | 47.64 |
| TiO2 | 1.26 | 1.34 | 1.27 | 1.42 | 1.16 | 1.81 | 1.21 | 1.38 |
| Al2O3 | 12.81 | 13.05 | 12.83 | 13.10 | 13.25 | 12.96 | 12.80 | 13.12 |
| Fe2O3T | 14.01 | 12.86 | 14.20 | 14.12 | 14.08 | 17.05 | 13.46 | 14.09 |
| FeO* | 12.59 | 11.56 | 12.76 | 12.69 | 12.66 | 15.32 | 12.10 | 12.67 |
| Fe2O3* | 1.41 | 1.30 | 1.43 | 1.43 | 1.42 | 1.72 | 1.36 | 1.42 |
| MnO | 0.21 | 0.18 | 0.22 | 0.21 | 0.21 | 0.24 | 0.20 | 0.21 |
| MgO | 8.09 | 8.21 | 7.93 | 7.58 | 9.00 | 4.96 | 8.06 | 7.20 |
| CaO | 12.04 | 13.55 | 13.20 | 10.48 | 11.85 | 10.04 | 13.72 | 12.35 |
| Na 2 O | 1.64 | 1.99 | 1.77 | 2.34 | 2.04 | 2.43 | 1.66 | 1.88 |
| K2O | 0.59 | 1.06 | 0.44 | 1.49 | 0.43 | 1.37 | 1.32 | 0.57 |
| P2O5 | 0.11 | 0.11 | 0.09 | 0.11 | 0.09 | 0.15 | 0.09 | 0.11 |
| LOI | 1.16 | 1.30 | 0.90 | 1.91 | 0.81 | 0.66 | 1.07 | 0.56 |
| TOTAL | 98.99 | 100.71 | 100.64 | 101.16 | 114.72 | 99.99 | 100.86 | 99.13 |
|  |  |  |  |  |  |  |  |  |
| Sc | 40.44 | 43.27 | 44.86 | 45.56 | 42.41 | 43.04 | 43.23 | 42.09 |
| V | 350.44 | 369.14 | 364.94 | 390.46 | 383.52 | 475.33 | 346.74 | 377.42 |
| Cr | 230.11 | 208.64 | 181.31 | 215.34 | 306.85 | 35.19 | 196.59 | 132.81 |
| Co | 64.15 | 59.82 | 66.27 | 66.58 | 68.68 | 82.45 | 62.58 | 66.89 |
| Ni | 129.41 | 61.42 | 63.92 | 57.12 | 491.27 | 48.80 | 67.33 | 44.49 |
| Cu | 135.10 | 109.78 | 117.83 | 128.49 | 116.96 | 181.65 | 129.50 | 145.96 |
| Zn | 92.27 | 57.62 | 86.66 | 60.36 | 71.15 | 101.22 | 63.16 | 100.42 |
| Sr | 203.44 | 300.27 | 206.74 | 435.94 | 209.83 | 388.06 | 592.23 | 204.63 |
| Y | 21.10 | 21.58 | 19.17 | 22.37 | 21.24 | 34.61 | 19.52 | 20.99 |
| Zr | 76.96 | 82.58 | 71.78 | 92.63 | 68.79 | 119.59 | 73.77 | 85.27 |
| Ba | 97.15 | 271.18 | 420.29 | 191.78 | 128.66 | 243.95 | 344.47 | 143.44 |
|  |  |  |  |  |  |  |  |  |
| Ti | 1.32 | 1.37 | 1.24 | 1.41 | 1.14 | 1.79 | 1.19 | 1.42 |
| V | 357.20 | 374.00 | 372.70 | 401.80 | 378.00 | 471.40 | 353.00 | 380.20 |
| Cr | 217.90 | 198.40 | 171.60 | 209.00 | 306.80 | 43.20 | 187.30 | 132.10 |
| Co | 61.20 | 54.20 | 62.20 | 63.10 | 66.20 | 86.80 | 57.90 | 66.80 |
| Ni | 109.30 | 60.00 | 52.70 | 54.30 | 461.20 | 52.90 | 63.50 | 44.90 |
| Cu | 148.30 | 118.10 | 128.30 | 135.10 | 107.40 | 204.60 | 138.40 | 151.60 |
| Zn | 103.00 | 55.30 | 88.70 | 61.10 | 72.00 | 106.90 | 64.30 | 105.60 |
| Rb | 19.10 | 45.40 | 14.10 | 53.00 | 11.40 | 40.10 | 37.20 | 16.70 |
| K | 5901.90 | 10597.21 | 4373.96 | 14867.85 | 4322.78 | 13655.46 | 13156.67 | 5734.03 |
| Sr | 212.90 | 319.70 | 217.00 | 458.90 | 191.10 | 376.70 | 609.30 | 215.30 |
| P | 1073.12 | 1094.31 | 892.00 | 1124.10 | 928.05 | 1540.67 | 917.28 | 1122.32 |
| Y | 23.70 | 24.10 | 21.70 | 25.50 | 19.60 | 34.70 | 22.60 | 24.90 |
| Zr | 83.40 | 88.20 | 73.10 | 94.10 | 64.60 | 114.30 | 76.90 | 90.90 |
| Nb | 3.90 | 4.47 | 3.36 | 4.25 | 2.76 | 5.40 | 3.45 | 4.55 |
| Sn | 1.36 | 1.44 | 1.52 | 1.15 | 0.89 | 2.23 | 0.84 | 1.66 |
| Cs | 0.97 | 2.66 | 0.73 | 1.32 | 0.56 | 0.56 | 1.89 | 0.98 |
| Ba | 107.40 | 271.30 | 398.00 | 191.90 | 124.80 | 251.70 | 332.00 | 148.70 |
| La | 6.92 | 6.18 | 4.91 | 6.44 | 4.69 | 8.58 | 5.14 | 6.49 |
| Ce | 16.10 | 15.04 | 11.86 | 15.45 | 11.65 | 20.93 | 12.43 | 15.46 |
| Pr | 2.28 | 2.24 | 1.79 | 2.27 | 1.75 | 3.01 | 1.84 | 2.30 |


| Nd | 10.58 | 10.37 | 8.41 | 10.83 | 7.98 | 14.07 | 8.72 | 10.60 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| Sm | 2.95 | 3.00 | 2.52 | 3.10 | 2.25 | 4.05 | 2.61 | 3.08 |
| Eu | 1.09 | 1.10 | 0.98 | 1.19 | 0.89 | 1.39 | 0.92 | 1.13 |
| Gd | 3.52 | 3.38 | 2.94 | 3.64 | 2.80 | 4.69 | 2.95 | 3.58 |
| Tb | 0.66 | 0.64 | 0.54 | 0.67 | 0.52 | 0.87 | 0.56 | 0.66 |
| Dy | 3.93 | 3.91 | 3.57 | 4.05 | 3.31 | 5.18 | 3.47 | 4.04 |
| Ho | 0.74 | 0.72 | 0.65 | 0.77 | 0.62 | 0.95 | 0.65 | 0.74 |
| Er | 2.10 | 2.17 | 1.90 | 2.21 | 1.81 | 2.88 | 1.96 | 2.15 |
| Tm | 0.33 | 0.34 | 0.30 | 0.34 | 0.28 | 0.45 | 0.30 | 0.33 |
| Yb | 2.04 | 2.04 | 1.84 | 2.10 | 1.78 | 2.78 | 1.88 | 2.10 |
| Lu | 0.33 | 0.33 | 0.29 | 0.34 | 0.29 | 0.44 | 0.29 | 0.34 |
| Hf | 2.27 | 2.27 | 1.89 | 2.55 | 1.65 | 3.09 | 1.97 | 2.47 |
| Ta | 0.28 | 0.29 | 0.22 | 0.31 | 0.21 | 0.38 | 0.23 | 0.30 |
| Pb | 2.18 | 1.03 | 2.36 | 1.04 | 1.50 | 2.10 | 3.09 | 3.55 |
| Th | 0.85 | 0.89 | 0.77 | 0.89 | 0.68 | 1.25 | 0.70 | 0.86 |
| U | 0.18 | 0.24 | 0.16 | 0.21 | 0.16 | 0.31 | 0.17 | 0.20 |
|  |  |  |  |  |  |  |  |  |
| Os |  |  |  |  | $<0.03$ |  | 0.10 |  |
| Ir |  |  |  |  | 0.03 |  | 0.03 |  |
| Ru |  |  |  |  | 0.47 |  | 0.30 |  |
| Rh |  |  |  |  |  |  | 0.49 |  |
| Pt |  |  |  |  |  |  | 9.07 |  |
| Pd | Au |  |  |  |  |  |  | 17.0 |


| ZONE | C/D | B | B | B | B | B | B | B | B |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\begin{aligned} & \text { LITHOLO } \\ & \text { GY } \end{aligned}$ | dyke core | dyke | dyke | dyke | connectin <br> g | connectin <br> g | dyke | connectin <br> g | dyke |
|  |  | prop tip | propr tip | prop tip | dyke | dyke | prop tip | dyke | prop tip |
| $\begin{aligned} & \text { SAMPLE } \\ & \text { I.D. } \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline \text { 11BH292 } \\ & \text { A1 } \\ & \hline \end{aligned}$ | $\begin{aligned} & \text { 11BH185 } \\ & \text { A2 } \\ & \hline \end{aligned}$ | $\begin{aligned} & \text { 11BH197 } \\ & \text { A4 } \\ & \hline \end{aligned}$ | $\begin{aligned} & \text { 11BH185 } \\ & \text { A2 } \\ & \hline \end{aligned}$ | $\begin{aligned} & \text { 11BH207 } \\ & \text { A1 } \\ & \hline \end{aligned}$ | $\begin{aligned} & \text { 11BH209 } \\ & \text { A1 } \\ & \hline \end{aligned}$ | $\begin{aligned} & \text { 11BH185 } \\ & \text { A1 } \\ & \hline \end{aligned}$ | $\begin{aligned} & \text { 11BH210 } \\ & \text { A1 } \end{aligned}$ | $\begin{aligned} & \text { 11JB105 } \\ & \text { C2 } \\ & \hline \end{aligned}$ |
| Lab | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | INRS |
| SiO 2 | 47.46 | 45.36 | 42.65 | 45.36 | 49.66 | 49.05 | 48.53 | 49.60 | 47.28 |
| TiO2 | 1.17 | 1.19 | 0.94 | 1.19 | 1.00 | 0.99 | 1.15 | 1.02 | 1.08 |
| Al2O3 | 13.34 | 13.13 | 10.66 | 13.13 | 13.00 | 12.85 | 13.95 | 13.16 | 13.73 |
| Fe2O3T | 13.79 | 11.46 | 5.57 | 11.46 | 9.43 | 9.12 | 10.17 | 7.69 | 11.70 |
| FeO* | 12.40 | 10.30 | 5.01 | 10.30 | 8.48 | 8.20 | 9.14 | 6.91 | 10.52 |
| Fe2O3* | 1.39 | 1.16 | 0.56 | 1.16 | 0.95 | 0.92 | 1.03 | 0.78 | 1.18 |
| MnO | 0.21 | 0.13 | 0.08 | 0.13 | 0.11 | 0.11 | 0.15 | 0.09 | 0.14 |
| MgO | 7.45 | 8.60 | 9.11 | 8.60 | 10.13 | 9.44 | 8.39 | 10.29 | 7.38 |
| CaO | 12.86 | 13.81 | 18.45 | 13.81 | 10.35 | 11.61 | 11.66 | 11.30 | 12.04 |
| Na 2 O | 1.93 | 1.59 | 2.33 | 1.59 | 0.98 | 1.23 | 1.82 | 0.86 | 1.38 |
| K2O | 0.55 | 1.91 | 1.85 | 1.91 | 3.74 | 3.10 | 2.55 | 3.64 | 2.18 |
| P2O5 | 0.08 | 0.08 | 0.05 | 0.08 | 0.10 | 0.09 | 0.10 | 0.10 | 0.08 |
| LOI | 0.50 | 2.44 | 9.36 | 2.44 | 1.74 | 1.46 | 2.99 | 2.16 | 2.88 |
| TOTAL | 99.36 | 99.71 | 106.63 | 111.17 | 100.24 | 99.07 | 101.46 | 99.90 | 100.02 |
| Sc | 46.31 | 43.60 | 33.06 | 43.56 | 39.37 | 37.97 | 44.24 | 38.61 |  |
| V | 380.10 | 376.30 | 285.36 | 376.32 | 317.35 | 308.35 | 337.87 | 275.80 |  |
| Cr | 93.01 | 199.20 | 151.42 | 199.24 | 518.30 | 477.83 | 207.69 | 440.31 |  |
| Co | 68.22 | 55.00 | 24.72 | 55.03 | 42.64 | 43.79 | 46.70 | 34.72 |  |
| Ni | 36.46 | 67.60 | 33.11 | 67.59 | 116.17 | 116.54 | 58.91 | 153.19 |  |
| Cu | 135.35 | 104.30 | 108.99 | 104.31 | 22.44 | 18.14 | 85.87 | 18.30 |  |
| Zn | 94.31 | 38.20 | 89.67 | 38.19 | 14.24 | 20.13 | 36.37 | 17.84 |  |
| Sr | 225.35 | 602.90 | 346.54 | 602.95 | 1267.76 | 970.20 | 707.70 | 1479.68 |  |
| Y | 21.77 | 22.50 | 19.79 | 22.50 | 22.71 | 22.08 | 20.14 | 20.21 |  |
| Zr | 68.17 | 86.00 | 70.43 | 86.02 | 115.44 | 97.87 | 76.93 | 108.39 |  |
| Ba | 182.55 | 325.10 | 800.25 | 325.11 | 1440.85 | 521.80 | 284.50 | 1448.70 |  |
| Ti | 1.16 | 1.15 | 0.93 | 1.15 | 0.98 | 0.95 | 1.17 | 0.99 | 0.00 |
| V | 356.70 | 366.40 | 276.80 | 366.40 | 299.60 | 288.60 | 334.40 | 291.50 | 31.00 |
| Cr | 100.60 | 195.80 | 148.10 | 195.80 | 471.90 | 456.70 | 201.20 | 431.30 | 4.00 |
| Co | 66.90 | 49.90 | 22.20 | 49.90 | 38.60 | 40.10 | 44.20 | 37.30 | 72.00 |
| Ni | 43.90 | 73.90 | 33.50 | 73.90 | 109.70 | 103.60 | 57.00 | 142.70 | 13.00 |
| Cu | 144.70 | 110.80 | 116.40 | 110.80 | 18.50 | 17.60 | 90.60 | 17.80 | < 7 |
| Zn | 100.40 | 31.30 | 98.30 | 31.30 | 12.60 | 18.60 | 35.00 | 13.40 | < 6 |


| Rb | 16.40 | 58.40 | 68.20 | 58.40 | 123.70 | 93.50 | 77.60 | 131.60 | 0.13 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| K | 5497.20 | 19100.00 | 18496.46 | 19111.44 | 37363.39 | 30951.48 | 25547.07 | 36356.64 |  |
| Sr | 215.40 | 591.80 | 330.10 | 591.80 | 1247.30 | 983.00 | 719.40 | 1398.10 | 111.00 |
| P | 812.31 | 800.00 | 537.76 | 799.53 | 988.82 | 933.79 | 959.27 | 953.97 | 203.59 |
| Y | 21.10 | 22.80 | 19.60 | 22.80 | 22.10 | 22.30 | 21.90 | 21.60 | 9.40 |
| Zr | 67.80 | 87.50 | 75.20 | 87.50 | 115.20 | 104.20 | 78.80 | 106.60 | 1.70 |
| Nb | 3.17 | 3.92 | 4.17 | 3.92 | 5.54 | 5.61 | 3.44 | 5.67 | 0.43 |
| Sn | 1.06 | 1.58 | 1.65 | 1.58 | 0.91 | 1.20 | 1.10 | 0.78 | < 0.5 |
| Cs | 0.45 | 0.55 | 0.37 | 0.55 | 1.65 | 2.04 | 0.71 | 2.32 | $<0.1$ |
| Ba | 187.70 | 340.20 | 804.40 | 340.20 | 1386.00 | 507.30 | 272.30 | 1319.40 | 3.60 |
| La | 4.96 | 5.61 | 6.10 | 5.61 | 7.86 | 14.11 | 5.15 | 8.16 | 13.00 |
| Ce | 11.99 | 13.58 | 14.53 | 13.58 | 20.08 | 28.24 | 12.79 | 18.78 | 22.50 |
| Pr | 1.75 | 1.99 | 2.03 | 1.99 | 2.85 | 3.48 | 1.90 | 2.61 | 2.60 |
| Nd | 8.50 | 9.16 | 9.08 | 9.16 | 11.97 | 13.95 | 8.72 | 11.61 | 9.20 |
| Sm | 2.48 | 2.65 | 2.49 | 2.65 | 3.18 | 3.27 | 2.52 | 3.00 | 2.00 |
| Eu | 0.98 | 1.15 | 0.86 | 1.15 | 1.04 | 1.04 | 1.07 | 1.04 | 1.13 |
| Gd | 2.99 | 3.21 | 2.93 | 3.21 | 3.32 | 3.47 | 3.01 | 3.24 | 1.88 |
| Tb | 0.54 | 0.58 | 0.55 | 0.58 | 0.55 | 0.62 | 0.55 | 0.56 | 0.34 |
| Dy | 3.46 | 3.76 | 3.33 | 3.76 | 3.45 | 3.71 | 3.53 | 3.47 | 1.79 |
| Ho | 0.62 | 0.68 | 0.60 | 0.68 | 0.62 | 0.68 | 0.66 | 0.63 | 0.35 |
| Er | 1.87 | 2.03 | 1.74 | 2.03 | 1.94 | 2.03 | 1.86 | 1.88 | 0.80 |
| Tm | 0.30 | 0.30 | 0.28 | 0.30 | 0.31 | 0.31 | 0.31 | 0.32 | 0.11 |
| Yb | 1.79 | 1.92 | 1.72 | 1.92 | 1.97 | 2.04 | 1.80 | 1.92 | 0.59 |
| Lu | 0.28 | 0.32 | 0.27 | 0.32 | 0.30 | 0.32 | 0.29 | 0.32 | 0.06 |
| Hf | 1.75 | 2.21 | 1.83 | 2.21 | 3.05 | 2.63 | 1.86 | 2.63 | 0.02 |
| Ta | 0.22 | 0.25 | 0.28 | 0.25 | 0.40 | 0.43 | 0.23 | 0.40 | <0.024 |
| Pb | 1.43 | 2.60 | 6.86 | 2.60 | 2.56 | 1.90 | 1.43 | 1.50 | 0.29 |
| Th | 0.75 | 0.79 | 1.34 | 0.79 | 3.12 | 3.30 | 0.77 | 2.99 | <0.24 |
| U | 0.16 | 0.27 | 0.68 | 0.27 | 0.74 | 0.75 | 0.20 | 0.70 | 0.23 |
|  |  | <0.03 |  |  |  |  |  |  |  |
| Os |  | <0.03 |  |  |  |  | 0.20 | 0.14 |  |
| Ir |  | 0.26 |  |  |  |  | 0.16 | 0.15 |  |
| Ru |  | 0.44 |  |  |  |  | 0.34 | 0.33 |  |
| Rh |  | 21.8 |  |  |  |  | 0.42 | 0.22 |  |
| Pt |  | 13.6 |  |  |  |  | 13.6 | 9.36 |  |
| Pd |  | 1.77 |  |  |  |  | 11.1 | 5.24 |  |
| Au |  |  |  |  |  |  | 1.77 | 0.50 |  |


| ZONE | B | A | A | A | A | A | E | E |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\begin{aligned} & \text { LITHOLOG } \\ & \mathrm{Y} \end{aligned}$ | dyke | dyke chill | small dyke | dyke core | dyke core | dyke chill | dyke chill | dyke core |
|  | prop tip |  | apophysis |  |  |  |  |  |
| SAMPLE I.D. | $\begin{aligned} & \text { 11JB109B } \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \text { 11BH154A } \\ & 3 \end{aligned}$ | $\begin{aligned} & \text { 11BH154A } \\ & 2 \end{aligned}$ | $\begin{aligned} & \text { 11BH175A } \\ & 1 \end{aligned}$ | $\begin{aligned} & \text { 11BH167A } \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & \text { 11BH167A } \\ & 2 \end{aligned}$ | $\begin{aligned} & \text { 11BH366B } \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \text { 11BH372A } \\ & 1 \end{aligned}$ |
| Lab | INRS | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff | Cardiff |
| SiO 2 | 50.82 | 47.37 | 47.35 | 46.71 | 47.01 | 46.37 | 46.60 | 48.16 |
| TiO2 | 0.98 | 1.29 | 1.43 | 1.09 | 1.27 | 1.19 | 1.18 | 1.50 |
| Al2O3 | 13.48 | 13.79 | 12.79 | 13.42 | 13.03 | 12.91 | 13.71 | 12.27 |
| Fe2O3T | 10.28 | 12.61 | 13.60 | 12.68 | 14.31 | 13.61 | 12.90 | 15.98 |
| FeO* | 9.24 | 11.33 | 12.23 | 11.40 | 12.86 | 12.24 | 11.60 | 14.37 |
| Fe2O3* | 1.04 | 1.27 | 1.37 | 1.28 | 1.45 | 1.38 | 1.30 | 1.61 |
| MnO | 0.16 | 0.17 | 0.19 | 0.19 | 0.22 | 0.20 | 0.19 | 0.24 |
| MgO | 8.51 | 7.90 | 6.80 | 8.62 | 7.49 | 8.06 | 8.12 | 6.70 |
| CaO | 9.79 | 9.39 | 11.37 | 12.57 | 11.22 | 10.33 | 12.08 | 9.74 |
| Na 2 O | 1.26 | 2.35 | 2.34 | 1.11 | 1.76 | 1.00 | 1.63 | 1.80 |
| K2O | 2.58 | 2.83 | 2.29 | 2.99 | 2.32 | 3.24 | 2.26 | 1.74 |
| P2O5 | 0.09 | 0.09 | 0.11 | 0.10 | 0.10 | 0.09 | 0.09 | 0.13 |
| LOI | 1.22 | 2.49 | 1.90 | 1.53 | 1.68 | 1.83 | 1.44 | 1.42 |
| TOTAL | 99.30 | 100.28 | 100.18 | 101.02 | 100.41 | 112.46 | 100.19 | 99.68 |
| Sc |  | 46.90 | 46.12 | 45.99 | 43.81 | 42.54 | 46.5 | 43.7 |
| V |  | 399.83 | 425.50 | 346.17 | 428.11 | 405.52 | 380.8 | 398.7 |
| Cr |  | 196.66 | 59.38 | 248.07 | 146.11 | 192.95 | 217.3 | 27.3 |
| Co |  | 59.71 | 66.62 | 61.23 | 66.96 | 64.65 | 62.2 | 73.7 |
| Ni |  | 76.58 | 35.62 | 90.82 | 50.69 | 56.54 | 74.2 | 44.5 |
| Cu |  | 113.01 | 126.56 | 114.49 | 138.55 | 131.39 | 111.1 | 150.8 |


| Zn |  | 46.35 | 52.41 | 79.39 | 136.83 | 107.65 | 99.5 | 90.7 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Sr |  | 473.09 | 518.36 | 571.81 | 336.42 | 610.24 | 456.0 | 297.3 |
| Y |  | 24.65 | 27.49 | 21.41 | 23.53 | 21.57 | 22.2 | 25.7 |
| Zr |  | 90.32 | 91.33 | 66.95 | 79.75 | 75.88 | 72.3 | 97.7 |
| Ba |  | 1199.84 | 847.44 | 675.50 | 124.84 | 493.43 | 513.7 | 199.9 |
| Ti | 1.08 | 1.25 | 1.43 | 1.10 | 1.27 | 1.17 | 1.2 | 1.5 |
| V | 316.00 | 386.30 | 426.30 | 329.80 | 400.70 | 396.00 | 372.2 | 423.1 |
| Cr | 203.00 | 200.60 | 61.10 | 239.60 | 141.60 | 188.60 | 209.8 | 38.0 |
| Co | 47.00 | 55.80 | 64.00 | 62.10 | 66.00 | 65.20 | 59.7 | 70.2 |
| Ni | 100.00 | 73.20 | 39.30 | 90.70 | 51.00 | 54.50 | 70.3 | 43.7 |
| Cu | 156.00 | 117.80 | 132.20 | 114.80 | 148.50 | 131.40 | 105.4 | 154.6 |
| Zn | 71.00 | 42.20 | 55.00 | 74.60 | 141.90 | 114.60 | 107.8 | 90.2 |
| Rb | 57.20 | 120.50 | 95.60 | 85.50 | 76.40 | 83.10 | 52.7 | 48.0 |
| K | 18067.50 | 28326.50 | 22855.16 | 29883.01 | 23221.78 | 32421.48 | 22601.50 | 17391.98 |
| Sr | 781.00 | 454.70 | 507.90 | 565.10 | 325.60 | 584.10 | 442.60 | 308.20 |
| P | 358.78 | 913.86 | 1121.59 | 973.52 | 1009.35 | 931.94 | 881.69 | 1280.08 |
| Y | 19.00 | 23.70 | 27.10 | 20.60 | 24.40 | 22.10 | 21.1 | 28.1 |
| Zr | 61.00 | 89.00 | 95.50 | 69.20 | 83.30 | 77.90 | 70.1 | 98.5 |
| Nb | 3.32 | 3.86 | 4.42 | 2.92 | 3.88 | 3.49 | 2.82 | 4.44 |
| Sn | 1.10 | 1.51 | 0.99 | 0.80 | 1.14 | 0.79 | 0.94 | 1.60 |
| Cs | 1.09 | 1.46 | 1.02 | 1.09 | 0.73 | 0.72 | 0.80 | 0.90 |
| Ba | 214.00 | 1298.00 | 884.80 | 651.70 | 134.80 | 477.80 | 511.8 | 185.7 |
| La | 4.80 | 5.47 | 6.12 | 4.88 | 6.02 | 5.18 | 4.86 | 6.81 |
| Ce | 11.50 | 13.70 | 15.42 | 12.02 | 14.61 | 12.54 | 12.03 | 16.59 |
| Pr | 1.74 | 2.02 | 2.31 | 1.80 | 2.12 | 1.87 | 1.77 | 2.47 |
| Nd | 8.08 | 9.57 | 10.68 | 8.38 | 9.82 | 8.67 | 8.25 | 11.36 |
| Sm | 2.40 | 2.79 | 3.26 | 2.46 | 3.03 | 2.61 | 2.49 | 3.44 |
| Eu | 0.99 | 1.20 | 1.21 | 0.92 | 1.08 | 1.07 | 0.97 | 1.22 |
| Gd | 2.82 | 3.29 | 3.72 | 2.80 | 3.42 | 3.07 | 2.94 | 4.00 |
| Tb | 0.57 | 0.60 | 0.67 | 0.50 | 0.64 | 0.56 | 0.54 | 0.69 |
| Dy | 3.30 | 3.77 | 4.18 | 3.18 | 4.09 | 3.69 | 3.36 | 4.56 |
| Ho | 0.74 | 0.68 | 0.77 | 0.58 | 0.74 | 0.68 | 0.60 | 0.83 |
| Er | 1.95 | 2.12 | 2.30 | 1.73 | 2.18 | 1.94 | 1.84 | 2.48 |
| Tm | 0.29 | 0.32 | 0.36 | 0.27 | 0.35 | 0.32 | 0.27 | 0.39 |
| Yb | 1.93 | 2.01 | 2.26 | 1.72 | 2.16 | 1.92 | 1.78 | 2.47 |
| Lu | 0.27 | 0.31 | 0.35 | 0.27 | 0.32 | 0.31 | 0.29 | 0.36 |
| Hf | 1.86 | 2.42 | 2.54 | 1.72 | 2.21 | 2.04 | 1.79 | 2.59 |
| Ta | 0.23 | 0.25 | 0.29 | 0.20 | 0.27 | 0.23 | 0.21 | 0.31 |
| Pb | 3.00 | 4.03 | 1.58 | 2.09 | 4.08 | 2.32 | 2.85 | 1.36 |
| Th | 0.80 | 0.80 | 0.94 | 0.67 | 0.87 | 0.76 | 0.70 | 1.02 |
| U | 0.18 | 0.18 | 0.21 | 0.16 | 0.20 | 0.18 | 0.16 | 0.23 |
| Os |  | 0.04 | <0.03 |  | <0.03 |  | <0.03 |  |
| Ir |  | 0.03 | <0.03 |  | <0.03 |  | <0.03 |  |
| Ru |  | 0.24 | 0.17 |  | 0.20 |  | 0.38 |  |
| Rh |  | 0.34 | 0.38 |  | 0.48 |  | 0.73 |  |
| Pt |  | 7.47 | 15.7 |  | 6.85 |  | 38.7 |  |
| Pd |  | 12.9 | 15.7 |  | 21.6 |  | 17.3 |  |
| Au |  | 1.33 | 1.85 |  | 2.45 |  | 3.30 |  |

## Appendix G2 - SFDC olivine mineral chemistry

| ZONE | SAMPLE ID | X Region | SiO2 | TiO2 | Al2O3 | Cr2O3 | FeO | MnO | MgO | CaO | NiO | Total | Fo (mol \%) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| D | MH123 B2 46 | core | 37.72 | 0.00 | 0.05 | 0.02 | 24.33 | 0.30 | 37.63 | 0.30 | 0.19 | 100.55 | 0.73 |
|  | MH123 B2 47 | rimward | 37.66 | 0.00 | 0.03 | 0.00 | 25.76 | 0.33 | 36.08 | 0.25 | 0.15 | 100.26 | 0.71 |
|  | MH123 B2 48 | rimward | 37.15 | 0.00 | 0.01 | 0.03 | 29.13 | 0.42 | 33.17 | 0.29 | 0.11 | 100.30 | 0.67 |
|  | MH123 B2 49 | rim | 34.30 | 0.02 | 0.03 | 0.02 | 35.46 | 0.56 | 27.97 | 0.23 | 0.05 | 98.64 | 0.58 |
|  | MH123 B2 46-49 | average | 36.71 | 0.01 | 0.03 | 0.02 | 28.67 | 0.40 | 33.71 | 0.27 | 0.13 | 99.94 | 0.68 |
|  | MH123 B2 50 | core | 37.30 | 0.00 | 0.04 | 0.01 | 25.92 | 0.34 | 36.20 | 0.29 | 0.16 | 100.26 | 0.71 |
|  | MH123 B2 51 | rim | 35.34 | 0.02 | 0.00 | 0.02 | 35.04 | 0.47 | 28.70 | 0.21 | 0.10 | 99.90 | 0.59 |
|  | MH123 B2 50-51 | average | 36.32 | 0.01 | 0.02 | 0.01 | 30.48 | 0.41 | 32.45 | 0.25 | 0.13 | 100.08 | 0.65 |
|  | MH123 B2 52 | core | 36.93 | 0.00 | 0.05 | 0.00 | 24.73 | 0.37 | 36.75 | 0.30 | 0.16 | 99.28 | 0.73 |
|  | MH123 B2 53 | rim | 37.14 | 0.01 | 0.04 | 0.02 | 23.79 | 0.32 | 37.37 | 0.19 | 0.14 | 99.02 | 0.74 |
|  | MH123 B2 52-53 | average | 37.03 | 0.01 | 0.05 | 0.01 | 24.26 | 0.34 | 37.06 | 0.24 | 0.15 | 99.15 | 0.73 |
|  | MH123 B2 65 | core | 37.68 | 0.01 | 0.03 | 0.04 | 21.32 | 0.25 | 39.22 | 0.33 | 0.16 | 99.03 | 0.77 |
|  | MH123 B2 66 | rimward | 37.13 | 0.00 | 0.04 | 0.00 | 22.26 | 0.33 | 39.04 | 0.32 | 0.12 | 99.24 | 0.76 |
|  | MH123 B2 67 | rimward | 37.19 | 0.00 | 0.05 | 0.00 | 23.64 | 0.24 | 37.98 | 0.30 | 0.17 | 99.58 | 0.74 |
|  | MH123 B2 68 | rimward | 34.99 | 0.01 | 0.04 | 0.00 | 34.75 | 0.59 | 28.33 | 0.24 | 0.13 | 99.05 | 0.59 |
|  | MH123 B2 69 | rim | 33.82 | 0.01 | 0.01 | 0.01 | 40.81 | 0.67 | 22.87 | 0.26 | 0.12 | 98.58 | 0.50 |
|  | MH123 B2 65-69 | average | 36.16 | 0.00 | 0.03 | 0.01 | 28.56 | 0.41 | 33.49 | 0.29 | 0.14 | 99.10 | 0.67 |
|  | MH123 B2 92 | rep | 34.48 | 0.00 | 0.02 | 0.02 | 32.35 | 0.35 | 29.86 | 0.25 | 0.11 | 97.42 | 0.62 |
|  | MH123 B2 93 | core | 33.16 | 0.00 | 0.03 | 0.03 | 37.88 | 0.60 | 25.19 | 0.26 | 0.08 | 97.23 | 0.54 |
|  | MH123 B2 94 | rim | 35.10 | 0.00 | 0.03 | 0.00 | 35.74 | 0.42 | 27.22 | 0.25 | 0.16 | 98.91 | 0.58 |
|  | MH123 B2 93-94 | average | 34.13 | 0.00 | 0.03 | 0.01 | 36.81 | 0.51 | 26.21 | 0.26 | 0.12 | 98.07 | 0.56 |
|  |  |  |  |  |  |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |  |  |  |  |  |
|  | MH126 A2 32 | core | 39.55 | 0.00 | 0.43 | 0.02 | 19.87 | 0.25 | 40.41 | 0.27 | 0.17 | 100.96 | 0.78 |
|  | MH126 A2 33 | rim | 39.92 | 0.01 | 0.10 | 0.02 | 19.27 | 0.37 | 41.01 | 0.31 | 0.24 | 101.26 | 0.79 |
|  | MH126 A2 34 | rim2 | 39.06 | 0.00 | 0.13 | 0.03 | 24.72 | 0.26 | 36.61 | 0.30 | 0.15 | 101.26 | 0.73 |
|  | MH126 A2 33-34 | rim average | 39.51 | 0.00 | 0.22 | 0.02 | 21.29 | 0.30 | 39.34 | 0.29 | 0.19 | 101.16 | 0.77 |
|  | MH126 A2 32-34 | average (incl A384) | 39.53 | 0.00 | 0.33 | 0.02 | 20.58 | 0.28 | 39.87 | 0.28 | 0.18 | 101.06 | 0.78 |
|  | MH126 A2 36 | core | 39.88 | 0.03 | 0.58 | 0.25 | 19.35 | 0.26 | 39.50 | 1.27 | 0.16 | 101.25 | 0.78 |
|  | MH126 A2 37 | rim | 38.89 | 0.00 | 0.05 | 0.00 | 26.21 | 0.30 | 35.22 | 0.28 | 0.16 | 101.12 | 0.71 |
|  | MH126 A2 36-37 | average | 39.38 | 0.02 | 0.31 | 0.13 | 22.78 | 0.28 | 37.36 | 0.77 | 0.16 | 101.18 | 0.74 |
|  | MH126 A2 38 | core | 38.99 | 0.00 | 0.06 | 0.02 | 18.49 | 0.26 | 42.07 | 0.30 | 0.19 | 100.39 | 0.80 |
|  | MH126 A2 39 | rim | 37.64 | 0.01 | 0.07 | 0.02 | 24.57 | 0.39 | 37.27 | 0.29 | 0.20 | 100.44 | 0.73 |
|  | MH126 A2 38-39 | average | 38.32 | 0.00 | 0.06 | 0.02 | 21.53 | 0.33 | 39.67 | 0.29 | 0.20 | 100.41 | 0.77 |
|  | MH126 A2 40 | core | 38.91 | 0.00 | 0.08 | 0.03 | 18.70 | 0.31 | 42.03 | 0.35 | 0.18 | 100.61 | 0.80 |
|  | MH126 A2 41 | rim | 36.25 | 0.01 | 0.00 | 0.00 | 33.05 | 0.45 | 30.13 | 0.12 | 0.13 | 100.14 | 0.62 |
|  | MH126 A2 40-41 | average | 37.58 | 0.00 | 0.04 | 0.01 | 25.88 | 0.38 | 36.08 | 0.24 | 0.16 | 100.38 | 0.71 |
|  | MH126 A2 42 | core | 39.94 | 0.02 | 0.06 | 0.02 | 19.07 | 0.26 | 42.12 | 0.30 | 0.24 | 102.02 | 0.80 |
|  | MH126 A2 43 | rim | 38.23 | 0.00 | 0.05 | 0.03 | 24.06 | 0.30 | 37.22 | 0.27 | 0.22 | 100.38 | 0.73 |
|  | MH126 A2 42-43 | average | 39.09 | 0.01 | 0.05 | 0.02 | 21.57 | 0.28 | 39.67 | 0.28 | 0.23 | 101.20 | 0.77 |
| C | MH129 A2 136 | core | 34.93 | 0.02 | 0.03 | 0.00 | 41.46 | 0.57 | 23.31 | 0.24 | 0.08 | 100.64 | 0.50 |


|  | MH129 A2 137 | rimwards | 34.24 | 0.02 | 0.05 | 0.00 | 41.19 | 0.57 | 23.22 | 0.24 | 0.06 | 99.59 | 0.50 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | MH129 A2 138 | rim | 33.94 | 0.04 | 0.01 | 0.02 | 40.63 | 0.50 | 23.50 | 0.19 | 0.07 | 98.89 | 0.51 |
|  | $\begin{aligned} & \text { MH129 A2 136- } \\ & 138 \end{aligned}$ | average | 34.37 | 0.03 | 0.03 | 0.01 | 41.09 | 0.55 | 23.35 | 0.22 | 0.07 | 99.70 | 0.50 |
|  | MH129 A2 139 | rep | 34.48 | 0.03 | 0.02 | 0.00 | 42.17 | 0.51 | 22.85 | 0.19 | 0.07 | 100.31 | 0.49 |
|  | MH129 A2 140 | core | 33.90 | 0.03 | 0.03 | 0.01 | 41.23 | 0.58 | 22.99 | 0.20 | 0.07 | 99.03 | 0.50 |
|  | MH129 A2 141 | rim | 34.57 | 0.00 | 0.01 | 0.00 | 42.22 | 0.52 | 22.96 | 0.17 | 0.08 | 100.53 | 0.49 |
|  | $\begin{aligned} & \text { MH129 A2 140- } \\ & 141 \end{aligned}$ | average | 34.24 | 0.01 | 0.02 | 0.00 | 41.72 | 0.55 | 22.97 | 0.19 | 0.08 | 99.78 | 0.50 |
|  | MH129 A2 142 | rep | 34.52 | 0.02 | 0.03 | 0.00 | 40.41 | 0.52 | 23.72 | 0.20 | 0.06 | 99.48 | 0.51 |
|  | MH129 A2 143 | rep | 34.33 | 0.03 | 0.02 | 0.02 | 41.19 | 0.62 | 24.02 | 0.17 | 0.03 | 100.43 | 0.51 |
|  | MH129 A2 149 | core | 34.36 | 0.00 | 0.01 | 0.00 | 41.80 | 0.50 | 22.30 | 0.22 | 0.07 | 99.26 | 0.49 |
|  | MH129 A2 150 | rim | 34.11 | 0.00 | 0.09 | 0.00 | 42.44 | 0.71 | 21.83 | 0.21 | 0.05 | 99.43 | 0.48 |
|  | $\begin{aligned} & \text { MH129 A2 149- } \\ & 150 \end{aligned}$ | average | 34.24 | 0.00 | 0.05 | 0.00 | 42.12 | 0.60 | 22.07 | 0.21 | 0.06 | 99.35 | 0.48 |
|  | MH129 A2 173 | rep | 34.50 | 0.00 | 0.02 | 0.00 | 42.29 | 0.58 | 22.56 | 0.21 | 0.05 | 100.21 | 0.49 |
|  | MH129 A2 174 | rep | 34.76 | 0.01 | 0.01 | 0.00 | 41.87 | 0.65 | 22.69 | 0.22 | 0.02 | 100.23 | 0.49 |
| E | BH366 B1 72 | core | 38.65 | 0.00 | 0.09 | 0.02 | 23.65 | 0.30 | 38.36 | 0.16 | 0.16 | 101.39 | 0.50 |
|  | BH366 B1 73 | rimwards | 38.27 | 0.00 | 0.07 | 0.00 | 26.35 | 0.35 | 35.89 | 0.16 | 0.16 | 101.24 | 0.50 |
|  | BH366 B1 74 | rim | 38.44 | 0.00 | 0.07 | 0.00 | 25.81 | 0.32 | 35.71 | 0.14 | 0.14 | 100.64 | 0.51 |
|  | BH366 B1 72-74 | average | 38.45 | 0.00 | 0.08 | 0.01 | 25.27 | 0.32 | 36.65 | 0.15 | 0.15 | 101.09 | 0.50 |
|  | BH366 B1 75 | rep | 38.19 | 0.01 | 0.22 | 0.09 | 26.80 | 0.38 | 34.38 | 0.67 | 0.10 | 100.83 | 0.49 |
|  | BH366 B1 76 | core | 39.54 | 0.00 | 0.03 | 0.00 | 20.93 | 0.29 | 40.26 | 0.20 | 0.15 | 101.39 | 0.50 |
|  | BH366 B1 77 | rim | 37.68 | 0.00 | 0.07 | 0.00 | 28.58 | 0.40 | 33.20 | 0.22 | 0.14 | 100.29 | 0.49 |
|  | BH366 B1 76-77 | average | 38.61 | 0.00 | 0.05 | 0.00 | 24.75 | 0.35 | 36.73 | 0.21 | 0.15 | 100.84 | 0.50 |
|  | BH366 B1 83 | core | 39.98 | 0.00 | 0.03 | 0.03 | 18.20 | 0.29 | 42.06 | 0.29 | 0.21 | 101.07 | 0.51 |
|  | BH366 B1 84 | rimwards | 39.10 | 0.00 | 0.07 | 0.01 | 22.23 | 0.31 | 38.78 | 0.30 | 0.19 | 100.99 | 0.51 |
|  | BH366 B1 85 | rim | 38.38 | 0.01 | 0.03 | 0.00 | 26.02 | 0.41 | 36.05 | 0.11 | 0.12 | 101.13 | 0.49 |
|  | BH366 B1 83-85 | average | 39.15 | 0.00 | 0.04 | 0.01 | 22.15 | 0.33 | 38.96 | 0.23 | 0.17 | 101.06 | 0.48 |
|  | BH366 B1 89 | core | 38.64 | 0.01 | 2.66 | 0.03 | 21.45 | 0.34 | 39.15 | 0.30 | 0.20 | 102.77 | 0.48 |
|  | BH366 B1 90 | rim | 37.27 | 0.03 | 0.03 | 0.02 | 28.61 | 0.38 | 33.81 | 0.13 | 0.14 | 100.42 | 0.49 |
|  | BH366 B1 89-90 | average | 37.95 | 0.02 | 1.35 | 0.03 | 25.03 | 0.36 | 36.48 | 0.22 | 0.17 | 101.60 | 0.49 |
|  | BH366 B1 91 | core | 37.90 | 0.00 | 0.04 | 0.02 | 26.08 | 0.47 | 35.99 | 0.22 | 0.15 | 100.85 | \#VALUE! |
|  | BH366 B1 92 | rim | 37.49 | 0.01 | 0.08 | 0.00 | 27.59 | 0.32 | 34.48 | 0.13 | 0.11 | 100.21 | \#VALUE! |
|  | BH366 B1 91-92 | average | 37.69 | 0.00 | 0.06 | 0.01 | 26.84 | 0.39 | 35.23 | 0.17 | 0.13 | 100.53 | \#VALUE! |
| A | BH154 A2 73 | core | 38.19 | 0.00 | 0.01 | 0.00 | 22.49 | 0.31 | 38.27 | 0.27 | 0.16 | 99.70 | 0.75 |
|  | BH154 A2 74 | rimwards | 38.82 | 0.00 | 0.00 | 0.00 | 22.68 | 0.39 | 38.57 | 0.33 | 0.13 | 100.91 | 0.75 |
|  | BH154 A2 75 | rimwards | 39.43 | 0.01 | 0.03 | 0.00 | 22.86 | 0.31 | 38.65 | 0.34 | 0.15 | 101.76 | 0.75 |
|  | BH154 A2 76 | rimwards | 38.96 | 0.00 | 0.05 | 0.00 | 22.57 | 0.37 | 38.99 | 0.33 | 0.11 | 101.38 | 0.75 |
|  | BH154 A2 77 | rim | 39.13 | 0.01 | 0.03 | 0.00 | 23.25 | 0.37 | 38.58 | 0.35 | 0.15 | 101.87 | 0.75 |
|  | BH154 A2 73-77 | average | 38.91 | 0.00 | 0.02 | 0.00 | 22.77 | 0.35 | 38.61 | 0.32 | 0.14 | 101.12 | 0.75 |
|  |  |  |  |  |  |  |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |  |  |  |  |  |  |
|  | BH167 A4 1 | core | 38.86 | 0.00 | 0.03 | 0.04 | 21.11 | 0.23 | 39.83 | 0.35 | 0.16 | 100.60 | 0.77 |
|  | BH167 A4 2 | rimwards | 38.36 | 0.00 | 0.04 | 0.00 | 21.46 | 0.40 | 39.35 | 0.32 | 0.15 | 100.08 | 0.77 |
|  | BH167 A4 3 | rim | 39.21 | 0.01 | 0.04 | 0.00 | 19.91 | 0.31 | 40.14 | 0.29 | 0.17 | 100.08 | 0.78 |
|  | BH167 A4 1-3 | average | 38.81 | 0.00 | 0.03 | 0.01 | 20.83 | 0.31 | 39.77 | 0.32 | 0.16 | 100.25 | 0.77 |



## Appendix G3 - SFDC clinopyroxene mineral chemistry

| Z | SAMPLE ID | X REGION | SiO2 | Al2O3 | $\begin{aligned} & \hline \mathrm{FeO} \\ & * \end{aligned}$ | TiO2 | $\begin{aligned} & \mathrm{Cr} 2 \mathrm{O} \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & \mathrm{Mg} \\ & \mathrm{O} \\ & \hline \end{aligned}$ | $\begin{aligned} & \mathrm{Mn} \\ & \mathrm{O} \\ & \hline \end{aligned}$ | CaO | $\begin{aligned} & \hline \mathrm{Ni} \\ & \mathrm{O} \\ & \hline \end{aligned}$ | $\begin{aligned} & \mathrm{Na} 2 \\ & \mathrm{O} \\ & \hline \end{aligned}$ | $\begin{aligned} & \mathrm{K} 2 \\ & \mathrm{O} \\ & \hline \end{aligned}$ | Total | mg\# |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| D | BH268 A2 1 | core | 52.59 | 2.10 | 7.62 | 0.51 | 0.20 | $\begin{aligned} & 17.1 \\ & 0 \end{aligned}$ | 0.22 | $\begin{aligned} & \hline 20.0 \\ & 2 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.20 | 0.01 | $\begin{aligned} & 100.5 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 79.9 \\ & 9 \end{aligned}$ |
|  | BH268 A2 2 | rimwards | 51.81 | 3.09 | 7.87 | 0.52 | 0.37 | $\begin{aligned} & 16.4 \\ & 7 \\ & \hline \end{aligned}$ | 0.22 | $\begin{aligned} & 20.2 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 100.8 \\ & 6 \end{aligned}$ | $\begin{aligned} & 78.8 \\ & 5 \\ & \hline \end{aligned}$ |
|  | BH268 A2 3 | rim | 51.49 | 3.14 | 7.81 | 0.66 | 0.27 | $\begin{aligned} & 16.6 \\ & 8 \end{aligned}$ | 0.18 | $\begin{aligned} & 19.5 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 100.0 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & 79.1 \\ & 8 \end{aligned}$ |
|  | BH268 A2 1-3 | average | 51.96 | 2.78 | 7.77 | 0.56 | 0.28 | $\begin{aligned} & 16.7 \\ & 5 \\ & \hline \end{aligned}$ | 0.21 | $19.9$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.4 \\ & 9 \end{aligned}$ | $79.3$ |
|  | BH268 A2 8 | core | 52.30 | 3.06 | 8.86 | 0.41 | 0.16 | $\begin{aligned} & 16.9 \\ & 5 \\ & \hline \end{aligned}$ | 0.23 | $\begin{aligned} & 18.9 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 101.1 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & 77.3 \\ & 3 \\ & \hline \end{aligned}$ |
|  | BH268 A2 9 | rimwards | 53.51 | 1.55 | 9.46 | 0.33 | 0.18 | $\begin{aligned} & 18.9 \\ & 8 \end{aligned}$ | 0.27 | $16.6$ | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | 0.16 | 0.01 | $\begin{aligned} & 101.1 \\ & 3 \end{aligned}$ | $\begin{aligned} & 78.1 \\ & 5 \\ & \hline \end{aligned}$ |
|  | BH268 A2 10 | rim | 53.59 | 1.80 | 8.91 | 0.34 | 0.10 | $\begin{aligned} & 18.1 \\ & 4 \\ & \hline \end{aligned}$ | 0.28 | $\begin{aligned} & 17.7 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.16 | 0.02 | $\begin{aligned} & 101.0 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & 78.3 \\ & 9 \\ & \hline \end{aligned}$ |
|  | BH268 A2 8-10 | average | 53.13 | 2.14 | 9.08 | 0.36 | 0.15 | $\begin{aligned} & 18.0 \\ & 3 \end{aligned}$ | 0.26 | $\begin{aligned} & 17.7 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.18 | 0.01 | $\begin{aligned} & 101.1 \\ & 2 \end{aligned}$ | $\begin{aligned} & 77.9 \\ & 5 \end{aligned}$ |
|  | BH268 A2 15 | ground rep | 49.98 | 2.53 | $\begin{aligned} & 18.1 \\ & 4 \\ & \hline \end{aligned}$ | 1.04 | 0.01 | $12.9$ | 0.37 | $\begin{aligned} & 15.3 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.25 | 0.00 | $\begin{aligned} & 100.6 \\ & 0 \end{aligned}$ | $\begin{aligned} & 55.8 \\ & 9 \end{aligned}$ |
|  | BH268 A2 16 | core | 53.78 | 1.69 | 8.18 | 0.34 | 0.29 | $\begin{aligned} & 18.6 \\ & 0 \end{aligned}$ | 0.23 | $\begin{aligned} & 17.8 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | 0.16 | 0.01 | $\begin{aligned} & 101.1 \\ & 2 \end{aligned}$ | $\begin{aligned} & 80.2 \\ & 1 \\ & \hline \end{aligned}$ |
|  | BH268 A2 17 | rimwards | 51.85 | 3.11 | 9.61 | 0.49 | 0.19 | $\begin{aligned} & 17.1 \\ & 7 \\ & \hline \end{aligned}$ | 0.23 | $17.9$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.8 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 76.1 \\ & 0 \\ & \hline \end{aligned}$ |
|  | BH268 A2 18 | rim | 52.99 | 1.77 | $\begin{aligned} & 10.6 \\ & 8 \end{aligned}$ | 0.35 | 0.10 | $\begin{aligned} & 18.4 \\ & 2 \\ & \hline \end{aligned}$ | 0.30 | $\begin{aligned} & 15.4 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | 0.13 | 0.00 | $\begin{aligned} & 100.2 \\ & 6 \end{aligned}$ | $\begin{aligned} & 75.4 \\ & 6 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH268 A2 16- } \\ & 18 \end{aligned}$ | average | 52.87 | 2.19 | 9.49 | 0.39 | 0.19 | $\begin{aligned} & 18.0 \\ & 6 \end{aligned}$ | 0.25 | $\begin{aligned} & 17.0 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.17 | 0.00 | $\begin{aligned} & 100.7 \\ & 5 \end{aligned}$ | $\begin{aligned} & 77.2 \\ & 6 \\ & \hline \end{aligned}$ |
|  | BH268 A2 21 | core | 52.20 | 3.09 | 6.68 | 0.58 | 0.62 | $\begin{aligned} & 16.4 \\ & 9 \end{aligned}$ | 0.12 | $\begin{aligned} & 20.8 \\ & 3 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.8 \\ & 5 \end{aligned}$ | $\begin{aligned} & 81.4 \\ & 8 \end{aligned}$ |
|  | BH268 A2 22 | rim | 52.17 | 3.08 | 7.89 | 0.64 | 0.25 | $\begin{aligned} & 16.7 \\ & 2 \\ & \hline \end{aligned}$ | 0.21 | $\begin{aligned} & 19.5 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.7 \\ & 4 \end{aligned}$ | $\begin{aligned} & 79.0 \\ & 6 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH268 A2 21- } \\ & 22 \end{aligned}$ | average | 52.19 | 3.08 | 7.28 | 0.61 | 0.44 | $\begin{aligned} & 16.6 \\ & 1 \\ & \hline \end{aligned}$ | 0.16 | $\begin{aligned} & 20.2 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.7 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & 80.2 \\ & 7 \\ & \hline \end{aligned}$ |
|  | BH268 A2 23 | core | 53.13 | 1.70 | 7.95 | 0.36 | 0.19 | $\begin{aligned} & 17.9 \\ & 8 \\ & \hline \end{aligned}$ | 0.23 | $\begin{aligned} & 18.5 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 5 \end{aligned}$ | 0.17 | 0.00 | $\begin{aligned} & 100.2 \\ & 6 \end{aligned}$ | $\begin{aligned} & 80.1 \\ & 1 \\ & \hline \end{aligned}$ |
|  | BH268 A2 24 | rimwards | 53.13 | 1.82 | 7.82 | 0.27 | 0.22 | $\begin{aligned} & 17.4 \\ & 4 \end{aligned}$ | 0.24 | $\begin{aligned} & 19.5 \\ & 6 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.7 \\ & 2 \end{aligned}$ | $\begin{aligned} & \hline 79.8 \\ & 9 \\ & \hline \end{aligned}$ |
|  | BH268 A2 25 | rim | 50.58 | 4.04 | $\begin{aligned} & 10.0 \\ & 8 \end{aligned}$ | 0.86 | 0.05 | $\begin{aligned} & 16.2 \\ & 7 \\ & \hline \end{aligned}$ | 0.24 | $\begin{aligned} & 17.9 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.3 \\ & 3 \end{aligned}$ | $\begin{aligned} & \hline 74.2 \\ & 0 \end{aligned}$ |
|  | $\begin{aligned} & \text { BH268 A2 23- } \\ & 25 \end{aligned}$ | average | 52.28 | 2.52 | 8.62 | 0.50 | 0.15 | $\begin{aligned} & 17.2 \\ & 3 \\ & \hline \end{aligned}$ | 0.24 | $\begin{aligned} & 18.6 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.4 \\ & 4 \end{aligned}$ | $\begin{aligned} & 78.0 \\ & 7 \\ & \hline \end{aligned}$ |
|  | BH268 A2 26 | ground rep | 51.71 | 2.48 | $\begin{aligned} & 22.0 \\ & 5 \end{aligned}$ | 0.18 | 0.00 | $\begin{aligned} & 13.7 \\ & 5 \\ & \hline \end{aligned}$ | 0.22 | 3.12 | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.23 | 0.06 | 93.80 | $\begin{aligned} & 52.6 \\ & 3 \end{aligned}$ |
|  | MH126 A2 29 | core | 53.12 | 2.04 | 6.61 | 0.46 | 0.64 | $\begin{aligned} & 17.7 \\ & 6 \\ & \hline \end{aligned}$ | 0.17 | $\begin{aligned} & 19.7 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 100.7 \\ & 6 \end{aligned}$ | $82.7$ $1$ |
|  | MH126 A2 30 | rimwards | 53.12 | 2.16 | 6.47 | 0.32 | 0.61 | $\begin{aligned} & 17.6 \\ & 2 \end{aligned}$ | 0.15 | $\begin{aligned} & \hline 20.0 \\ & 2 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 100.7 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 82.9 \\ & 1 \\ & \hline \end{aligned}$ |
|  | MH126 A2 31 | rim | 53.62 | 2.05 | 8.29 | 0.50 | 0.49 | $\begin{aligned} & 19.1 \\ & 7 \\ & \hline \end{aligned}$ | 0.25 | $\begin{aligned} & 16.9 \\ & 2 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 101.5 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 80.4 \\ & 7 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { MH126 A2 29- } \\ & 31 \\ & \hline \end{aligned}$ | average | 53.29 | 2.08 | 7.12 | 0.43 | 0.58 | $\begin{aligned} & 18.1 \\ & 8 \\ & \hline \end{aligned}$ | 0.19 | $\begin{aligned} & 18.8 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 101.0 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 82.0 \\ & 3 \end{aligned}$ |
|  | MH126 A2 45 | ground rep | 49.73 | 5.13 | 8.72 | 0.84 | 0.28 | $\begin{aligned} & 15.0 \\ & 0 \end{aligned}$ | 0.18 | $\begin{aligned} & 20.2 \\ & 4 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.3 \\ & 2 \end{aligned}$ | $\begin{aligned} & 75.4 \\ & 1 \\ & \hline \end{aligned}$ |
|  | BH296 A1 19 | core | 51.60 | 1.92 | $\begin{aligned} & 13.1 \\ & 2 \\ & \hline \end{aligned}$ | 0.58 | 0.00 | $\begin{aligned} & 15.0 \\ & 5 \end{aligned}$ | 0.28 | $\begin{aligned} & 17.4 \\ & 0 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.36 | 0.02 | $\begin{aligned} & 100.3 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 67.1 \\ & 4 \\ & \hline \end{aligned}$ |
|  | BH296 A1 19 | rimwards | 51.90 | 2.00 | $\begin{aligned} & 12.9 \\ & 1 \end{aligned}$ | 0.54 | 0.00 | $\begin{aligned} & 15.1 \\ & 9 \\ & \hline \end{aligned}$ | 0.32 | $\begin{aligned} & \hline 17.6 \\ & 0 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.24 | 0.00 | $\begin{aligned} & 100.6 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 67.7 \\ & 0 \\ & \hline \end{aligned}$ |
|  | BH296 A1 19 | rimwards | 51.86 | 2.02 | $12.7$ | 0.64 | 0.00 | $15.0$ | 0.30 | $17.8$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.25 | 0.00 | $\begin{aligned} & 100.6 \\ & 3 \end{aligned}$ | $\begin{aligned} & 67.7 \\ & 6 \\ & \hline \end{aligned}$ |
|  | BH296 A1 19 | rimwards | 51.90 | 1.96 | $\begin{aligned} & 12.7 \\ & 9 \\ & \hline \end{aligned}$ | 0.58 | 0.00 | $\begin{aligned} & 15.1 \\ & 5 \\ & \hline \end{aligned}$ | 0.28 | $\begin{aligned} & 18.1 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | 0.26 | 0.01 | $\begin{aligned} & 101.1 \\ & 2 \end{aligned}$ | $\begin{aligned} & 67.8 \\ & 6 \\ & \hline \end{aligned}$ |
|  | BH296 A1 19 | rimwards | 49.16 | 2.92 | $\begin{aligned} & 15.4 \\ & 6 \\ & \hline \end{aligned}$ | 0.63 | 0.04 | $\begin{aligned} & 14.3 \\ & 0 \\ & \hline \end{aligned}$ | 0.34 | $\begin{aligned} & 15.6 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 0.22 | 0.00 | 98.68 | $\begin{aligned} & 62.2 \\ & 4 \end{aligned}$ |
|  | BH296 A1 19 | rimwards | 51.65 | 1.86 | $\begin{aligned} & 14.8 \\ & 1 \end{aligned}$ | 0.63 | 0.06 | $\begin{aligned} & 15.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.33 | $\begin{aligned} & 16.3 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 100.9 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & 64.3 \\ & 6 \\ & \hline \end{aligned}$ |
|  | BH296 A1 19 | rimwards | 53.03 | 0.99 | $\begin{aligned} & \hline 21.7 \\ & 3 \end{aligned}$ | 0.30 | 0.00 | $\begin{aligned} & 18.6 \\ & 0 \end{aligned}$ | 0.46 | 4.83 | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.12 | 0.01 | $\begin{aligned} & 100.0 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 60.4 \\ & 0 \\ & \hline \end{aligned}$ |
|  | BH296 A1 19 | rimwards | 52.00 | 1.21 | $\begin{aligned} & 21.6 \\ & 4 \\ & \hline \end{aligned}$ | 0.43 | 0.05 | $\begin{aligned} & 18.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.52 | 6.34 | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.09 | 0.00 | $\begin{aligned} & 100.3 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 59.7 \\ & 8 \\ & \hline \end{aligned}$ |
|  | BH296 A1 19 | rimwards | 51.98 | 0.94 | $\begin{aligned} & 23.6 \\ & 9 \\ & \hline \end{aligned}$ | 0.35 | 0.00 | $\begin{aligned} & 17.3 \\ & 1 \end{aligned}$ | 0.54 | 5.29 | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.08 | 0.01 | $\begin{aligned} & 100.1 \\ & 9 \end{aligned}$ | $\begin{aligned} & 56.5 \\ & 7 \\ & \hline \end{aligned}$ |
|  | BH296 A1 19 | rim | 50.93 | 1.10 | 25.7 | 0.45 | 0.00 | 14.6 | 0.62 | 6.70 | 0.0 | 0.13 | 0.00 | 100.3 | 50.3 |


|  |  |  |  | 3 |  |  | 4 |  |  | 0 |  |  | 1 | 5 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| BH296 A1 19 | x average | 51.60 | 1.69 | $\begin{aligned} & 17.4 \\ & 6 \end{aligned}$ | 0.51 | 0.01 | $\begin{aligned} & 15.8 \\ & 3 \\ & \hline \end{aligned}$ | 0.40 | $\begin{aligned} & 12.6 \\ & 1 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.20 | 0.01 | $\begin{aligned} & 100.3 \\ & 3 \end{aligned}$ | $\begin{aligned} & \hline 62.4 \\ & 2 \end{aligned}$ |
| BH296 A1 21 | rim | 51.73 | 1.84 | $\begin{aligned} & 13.3 \\ & 1 \\ & \hline \end{aligned}$ | 0.48 | 0.04 | $\begin{aligned} & 14.9 \\ & 5 \end{aligned}$ | 0.34 | $\begin{aligned} & 17.4 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.3 \\ & 0 \end{aligned}$ | $\begin{aligned} & \hline 66.6 \\ & 8 \end{aligned}$ |
| BH296 A1 21 | rimward | 52.40 | 1.93 | $\begin{aligned} & 11.4 \\ & 8 \\ & \hline \end{aligned}$ | 0.61 | 0.00 | $\begin{aligned} & 15.7 \\ & 5 \\ & \hline \end{aligned}$ | 0.32 | $18.0$ | $0.0$ | 0.23 | 0.00 | $\begin{aligned} & 100.8 \\ & 1 \end{aligned}$ | $\begin{aligned} & 70.9 \\ & 7 \\ & \hline \end{aligned}$ |
| BH296 A1 21 | rimward | 52.23 | 1.92 | $\begin{aligned} & 12.0 \\ & 5 \end{aligned}$ | 0.55 | 0.00 | $\begin{aligned} & 15.4 \\ & 0 \\ & \hline \end{aligned}$ | 0.29 | $\begin{aligned} & 17.9 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 6 \end{aligned}$ | 0.24 | 0.00 | $\begin{aligned} & 100.7 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 69.4 \\ & 9 \end{aligned}$ |
| BH296 A1 21 | rimward | 51.83 | 1.96 | $12.5$ | 0.71 | 0.04 | $15.0$ | 0.31 | $\begin{aligned} & 17.8 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.26 | 0.02 | $\begin{aligned} & 100.5 \\ & 1 \end{aligned}$ | $\begin{aligned} & 68.0 \\ & 9 \\ & \hline \end{aligned}$ |
| BH296 A1 21 | rimward | 51.17 | 2.09 | $\begin{aligned} & 12.7 \\ & 3 \end{aligned}$ | 0.95 | 0.02 | $\begin{aligned} & 14.9 \\ & 7 \\ & \hline \end{aligned}$ | 0.31 | $\begin{aligned} & 17.7 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.25 | 0.01 | $\begin{aligned} & 100.2 \\ & 5 \end{aligned}$ | $\begin{aligned} & 67.7 \\ & 0 \\ & \hline \end{aligned}$ |
| BH296 A1 21 | rimward | 52.10 | 2.00 | $\begin{aligned} & 11.2 \\ & 7 \\ & \hline \end{aligned}$ | 0.60 | 0.07 | $\begin{aligned} & 15.5 \\ & 2 \end{aligned}$ | 0.26 | $\begin{aligned} & 18.4 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.23 | 0.01 | $\begin{aligned} & 100.5 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 71.0 \\ & 4 \\ & \hline \end{aligned}$ |
| BH296 A1 21 | rimward | 51.64 | 2.09 | $10.7$ | 0.68 | 0.00 | $\begin{aligned} & 15.4 \\ & 8 \\ & \hline \end{aligned}$ | 0.27 | $\begin{aligned} & 18.6 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.22 | 0.01 | 99.74 | $\begin{aligned} & 71.9 \\ & 9 \\ & \hline \end{aligned}$ |
| BH296 A1 21 | rimward | 51.45 | 2.33 | $\begin{aligned} & 10.9 \\ & 2 \end{aligned}$ | 0.70 | 0.00 | $\begin{aligned} & 15.4 \\ & 3 \\ & \hline \end{aligned}$ | 0.26 | $\begin{aligned} & 18.6 \\ & 0 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.24 | 0.01 | 99.93 | $\begin{aligned} & 71.5 \\ & 8 \end{aligned}$ |
| BH296 A1 21 | rimward | 51.42 | 2.43 | $\begin{aligned} & 11.2 \\ & 7 \\ & \hline \end{aligned}$ | 0.74 | 0.06 | $\begin{aligned} & 15.4 \\ & 0 \\ & \hline \end{aligned}$ | 0.32 | $\begin{aligned} & 18.5 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.4 \\ & 3 \end{aligned}$ | $\begin{aligned} & 70.8 \\ & 9 \end{aligned}$ |
| BH296 A1 21 | core | 51.58 | 2.33 | $\begin{aligned} & 11.4 \\ & 1 \\ & \hline \end{aligned}$ | 0.63 | 0.00 | $\begin{aligned} & 15.3 \\ & 7 \\ & \hline \end{aligned}$ | 0.25 | $\begin{aligned} & 18.4 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.25 | 0.00 | $\begin{aligned} & 100.3 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 70.5 \\ & 9 \end{aligned}$ |
| BH296 A1 21 | core | 51.59 | 2.36 | $\begin{aligned} & 11.3 \\ & 4 \\ & \hline \end{aligned}$ | 0.69 | 0.02 | $\begin{aligned} & 15.2 \\ & 3 \\ & \hline \end{aligned}$ | 0.29 | $\begin{aligned} & 18.5 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 5 \end{aligned}$ | 0.23 | 0.01 | $\begin{aligned} & 100.3 \\ & 1 \end{aligned}$ | $\begin{aligned} & 70.5 \\ & 4 \end{aligned}$ |
| BH296 A1 21 | rimward | 51.32 | 2.47 | $\begin{aligned} & 11.7 \\ & 5 \\ & \hline \end{aligned}$ | 0.64 | 0.03 | $\begin{aligned} & 15.2 \\ & 2 \\ & \hline \end{aligned}$ | 0.28 | $\begin{aligned} & 18.5 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.24 | 0.00 | $\begin{aligned} & 100.5 \\ & 5 \end{aligned}$ | $\begin{aligned} & 69.7 \\ & 7 \\ & \hline \end{aligned}$ |
| BH296 A1 21 | rimward | 51.00 | 2.47 | $\begin{aligned} & 11.1 \\ & 9 \\ & \hline \end{aligned}$ | 0.82 | 0.05 | $\begin{aligned} & 14.8 \\ & 9 \end{aligned}$ | 0.26 | $\begin{aligned} & 18.4 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.24 | 0.03 | 99.35 | $\begin{aligned} & 70.3 \\ & 4 \end{aligned}$ |
| BH296 A1 21 | rimward | 51.23 | 2.43 | $\begin{aligned} & 11.2 \\ & 8 \\ & \hline \end{aligned}$ | 0.77 | 0.00 | $\begin{aligned} & 15.2 \\ & 9 \end{aligned}$ | 0.26 | $\begin{aligned} & 18.4 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.27 | 0.00 | 99.98 | $\begin{aligned} & 70.7 \\ & 1 \end{aligned}$ |
| BH296 A1 21 | rimward | 51.19 | 2.40 | $\begin{aligned} & 11.7 \\ & 1 \\ & \hline \end{aligned}$ | 0.74 | 0.02 | $\begin{aligned} & 15.2 \\ & 7 \\ & \hline \end{aligned}$ | 0.25 | $\begin{aligned} & 18.1 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.24 | 0.01 | 99.98 | $\begin{aligned} & 69.9 \\ & 1 \\ & \hline \end{aligned}$ |
| BH296 A1 21 | rimward | 51.26 | 2.41 | $\begin{aligned} & 11.7 \\ & 8 \\ & \hline \end{aligned}$ | 0.58 | 0.01 | $\begin{aligned} & 15.4 \\ & 3 \end{aligned}$ | 0.26 | $\begin{aligned} & 18.1 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.25 | 0.01 | $\begin{aligned} & 100.1 \\ & 2 \end{aligned}$ | $\begin{aligned} & 70.0 \\ & 0 \end{aligned}$ |
| BH296 A1 21 | rimward | 50.64 | 3.72 | $\begin{aligned} & 11.7 \\ & 6 \\ & \hline \end{aligned}$ | 0.96 | 0.07 | $\begin{aligned} & 14.5 \\ & 8 \\ & \hline \end{aligned}$ | 0.24 | $\begin{aligned} & 17.1 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.51 | 0.13 | 99.81 | $\begin{aligned} & \hline 68.8 \\ & 5 \\ & \hline \end{aligned}$ |
| BH296 A1 21 | rimward | 52.50 | 2.21 | 9.81 | 0.25 | 0.00 | $\begin{aligned} & 13.9 \\ & 3 \end{aligned}$ | 0.31 | $\begin{aligned} & 21.1 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | 0.38 | 0.00 | $\begin{aligned} & 100.5 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 71.6 \\ & 9 \end{aligned}$ |
| BH296 A1 21 | rimward | 51.22 | 2.48 | $\begin{aligned} & 12.0 \\ & 3 \end{aligned}$ | 0.69 | 0.00 | $\begin{aligned} & 15.5 \\ & 9 \\ & \hline \end{aligned}$ | 0.28 | $\begin{aligned} & 17.0 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | 0.22 | 0.00 | 99.63 | $\begin{aligned} & 69.7 \\ & 8 \end{aligned}$ |
| BH296 A1 21 | rim | 51.97 | 1.96 | $\begin{aligned} & 13.0 \\ & 9 \end{aligned}$ | 0.57 | 0.00 | $\begin{aligned} & 14.7 \\ & 2 \\ & \hline \end{aligned}$ | 0.28 | $\begin{aligned} & 18.0 \\ & 7 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.19 | 0.00 | $\begin{aligned} & 100.8 \\ & 8 \end{aligned}$ | $\begin{aligned} & 66.7 \\ & 0 \end{aligned}$ |
| BH296 A1 21 | x average | 51.57 | 2.29 | $11.6$ | 0.67 | 0.02 | $\begin{aligned} & 15.1 \\ & 7 \\ & \hline \end{aligned}$ | 0.28 | $\begin{aligned} & 18.2 \\ & 8 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.25 | 0.01 | $\begin{aligned} & 100.2 \\ & 4 \end{aligned}$ | $\begin{aligned} & 69.8 \\ & 7 \\ & \hline \end{aligned}$ |
| BH296 A1 31 | core | 51.50 | 1.49 | $\begin{aligned} & 17.3 \\ & 3 \\ & \hline \end{aligned}$ | 0.68 | 0.00 | $\begin{aligned} & 14.1 \\ & 9 \\ & \hline \end{aligned}$ | 0.40 | $\begin{aligned} & 14.3 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.19 | 0.00 | $\begin{aligned} & 100.1 \\ & 8 \end{aligned}$ | $\begin{aligned} & 59.3 \\ & 3 \end{aligned}$ |
| BH296 A1 31 | rimwards | 51.27 | 1.52 | $\begin{aligned} & 16.1 \\ & 5 \\ & \hline \end{aligned}$ | 0.73 | 0.00 | $\begin{aligned} & 13.5 \\ & 3 \\ & \hline \end{aligned}$ | 0.39 | $\begin{aligned} & 15.8 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.20 | 0.00 | 99.64 | $\begin{aligned} & 59.8 \\ & 9 \\ & \hline \end{aligned}$ |
| BH296 A1 31 | rimwards | 51.22 | 1.62 | $\begin{aligned} & 15.8 \\ & 4 \\ & \hline \end{aligned}$ | 0.79 | 0.00 | $\begin{aligned} & 13.2 \\ & 1 \end{aligned}$ | 0.37 | $\begin{aligned} & 17.1 \\ & 9 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.23 | 0.01 | $\begin{aligned} & 100.4 \\ & 9 \end{aligned}$ | $\begin{aligned} & 59.7 \\ & 8 \end{aligned}$ |
| BH296 A1 31 | rimwards | 51.15 | 1.50 | $\begin{aligned} & 16.8 \\ & 3 \end{aligned}$ | 0.66 | 0.04 | $13.5$ | 0.40 | $\begin{aligned} & 15.7 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.21 | 0.02 | $\begin{aligned} & 100.1 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 58.8 \\ & 5 \end{aligned}$ |
| BH296 A1 31 | rimwards | 51.91 | 1.68 | $\begin{aligned} & 17.1 \\ & 4 \\ & \hline \end{aligned}$ | 0.45 | 0.00 | $\begin{aligned} & 12.9 \\ & 5 \\ & \hline \end{aligned}$ | 0.32 | $\begin{aligned} & 13.8 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.30 | 0.01 | 98.57 | $\begin{aligned} & 57.3 \\ & 7 \\ & \hline \end{aligned}$ |
| BH296 A1 31 | rimwards | 51.62 | 0.82 | $\begin{aligned} & 25.6 \\ & 6 \end{aligned}$ | 0.43 | 0.00 | $\begin{aligned} & 16.1 \\ & 2 \\ & \hline \end{aligned}$ | 0.58 | 5.36 | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.08 | 0.00 | $\begin{aligned} & 100.6 \\ & 9 \end{aligned}$ | $\begin{aligned} & 52.8 \\ & 2 \end{aligned}$ |
| BH296 A1 31 | rimwards | 50.98 | 0.97 | $\begin{aligned} & 25.1 \\ & 3 \end{aligned}$ | 0.55 | 0.00 | $\begin{aligned} & 15.8 \\ & 2 \end{aligned}$ | 0.64 | 5.78 | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.09 | 0.00 | 99.95 | $\begin{aligned} & 52.8 \\ & 6 \end{aligned}$ |
| BH296 A1 31 | rim | 48.04 | 2.16 | $26.7$ | 0.43 | 0.00 | $\begin{aligned} & 13.2 \\ & 6 \\ & \hline \end{aligned}$ | 0.59 | 5.96 | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.10 | 0.02 | 97.27 | $\begin{aligned} & 46.9 \\ & 5 \end{aligned}$ |
| BH296 A1 31 | x average | 50.96 | 1.47 | $\begin{aligned} & 20.1 \\ & 0 \\ & \hline \end{aligned}$ | 0.59 | 0.01 | $\begin{aligned} & 14.0 \\ & 7 \\ & \hline \end{aligned}$ | 0.46 | $\begin{aligned} & 11.7 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 0.17 | 0.01 | 99.61 | $\begin{aligned} & 55.9 \\ & 8 \end{aligned}$ |
| BH296 A1 32 | core | 51.16 | 1.60 | $\begin{aligned} & 16.7 \\ & 2 \\ & \hline \end{aligned}$ | 0.64 | 0.00 | $\begin{aligned} & 13.8 \\ & 4 \\ & \hline \end{aligned}$ | 0.38 | $\begin{aligned} & 15.6 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.25 | 0.00 | $\begin{aligned} & 100.2 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & 59 . \\ & 0 \\ & \hline \end{aligned}$ |
| BH296 A1 32 | rimwards | 51.39 | 1.66 | $\begin{aligned} & 15.5 \\ & 5 \\ & \hline \end{aligned}$ | 0.67 | 0.00 | $\begin{aligned} & 13.8 \\ & 0 \\ & \hline \end{aligned}$ | 0.40 | $\begin{aligned} & 16.1 \\ & 7 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.22 | 0.00 | 99.84 | $\begin{aligned} & 61.2 \\ & 6 \end{aligned}$ |
| BH296 A1 32 | rimwards | 51.27 | 1.65 | $\begin{aligned} & 15.9 \\ & 5 \\ & \hline \end{aligned}$ | 0.57 | 0.00 | $\begin{aligned} & 13.8 \\ & 9 \\ & \hline \end{aligned}$ | 0.38 | $\begin{aligned} & 16.2 \\ & 4 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.1 \\ & 6 \end{aligned}$ | $\begin{aligned} & \hline 60.8 \\ & 2 \end{aligned}$ |
| BH296 A1 32 | rimwards | 50.19 | 2.22 | $\begin{aligned} & 17.7 \\ & 0 \end{aligned}$ | 0.64 | 0.02 | $\begin{aligned} & 13.9 \\ & 5 \\ & \hline \end{aligned}$ | 0.38 | $\begin{aligned} & 14.0 \\ & 5 \\ & \hline \end{aligned}$ | $0.0$ | 0.21 | 0.04 | 99.41 | $\begin{aligned} & 58.4 \\ & 2 \end{aligned}$ |
| BH296 A1 32 | rimwards | 50.20 | 1.70 | $\begin{aligned} & 20.2 \\ & 3 \end{aligned}$ | 0.56 | 0.02 | $\begin{aligned} & 12.2 \\ & 3 \end{aligned}$ | 0.47 | $\begin{aligned} & 14.3 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.20 | 0.00 | 99.92 | $\begin{aligned} & 51.8 \\ & 5 \end{aligned}$ |
| BH296 A1 32 | rimwards | 50.33 | 1.03 | $\begin{aligned} & 22.7 \\ & 1 \\ & \hline \end{aligned}$ | 0.62 | 0.01 | $\begin{aligned} & 10.5 \\ & 4 \end{aligned}$ | 0.56 | $\begin{aligned} & 14.3 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.3 \\ & 5 \end{aligned}$ | $\begin{aligned} & 45.2 \\ & 7 \\ & \hline \end{aligned}$ |
| BH296 A1 32 | rimwards | 50.34 | 1.18 | 23.3 | 0.73 | 0.00 | 11.1 | 0.56 | 12.8 | 0.0 | 0.15 | 0.00 | 100.3 | 46.0 |


|  |  |  |  | 5 |  |  | 7 |  | 7 | 0 |  |  | 5 | 2 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| BH296 A1 32 | rim | 50.83 | 1.19 | $\begin{aligned} & \hline 20.0 \\ & 5 \end{aligned}$ | 0.54 | 0.07 | $\begin{aligned} & 11.6 \\ & 2 \\ & \hline \end{aligned}$ | 0.49 | $\begin{aligned} & 15.1 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.17 | 0.00 | $\begin{aligned} & 100.0 \\ & 7 \end{aligned}$ | $\begin{aligned} & 50.8 \\ & 1 \end{aligned}$ |
| BH296 A1 32 | x average | 50.71 | 1.53 | $\begin{aligned} & 19.0 \\ & 3 \end{aligned}$ | 0.62 | 0.01 | $\begin{aligned} & 12.6 \\ & 3 \end{aligned}$ | 0.45 | $\begin{aligned} & 14.8 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.20 | 0.01 | $\begin{aligned} & 100.0 \\ & 4 \end{aligned}$ | $\begin{aligned} & \hline 54.2 \\ & 6 \end{aligned}$ |
| BH296 A1 33 | core | 53.36 | 1.69 | $12.5$ | 0.49 | 0.03 | $\begin{aligned} & 15.2 \\ & 8 \\ & \hline \end{aligned}$ | 0.17 | $\begin{aligned} & 15.2 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.20 | 0.13 | 99.13 | $\begin{aligned} & 68.5 \\ & 0 \\ & \hline \end{aligned}$ |
| BH296 A1 33 | rimwards | 52.81 | 1.67 | 9.91 | 0.43 | 0.03 | $\begin{aligned} & 15.4 \\ & 5 \\ & \hline \end{aligned}$ | 0.22 | $\begin{aligned} & 19.6 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.18 | 0.01 | $\begin{aligned} & 100.4 \\ & 2 \end{aligned}$ | $\begin{aligned} & 73.5 \\ & 2 \end{aligned}$ |
| BH296 A1 33 | rimwards | 52.50 | 1.94 | $\begin{aligned} & 10.1 \\ & 3 \end{aligned}$ | 0.46 | 0.01 | $\begin{aligned} & 15.8 \\ & 8 \end{aligned}$ | 0.28 | $\begin{aligned} & 18.9 \\ & 3 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.19 | 0.01 | $\begin{aligned} & 100.3 \\ & 4 \end{aligned}$ | $\begin{aligned} & 73.6 \\ & 3 \\ & \hline \end{aligned}$ |
| BH296 A1 33 | rimwards | 52.24 | 1.91 | $\begin{aligned} & 10.3 \\ & 6 \\ & \hline \end{aligned}$ | 0.50 | 0.03 | $\begin{aligned} & 15.9 \\ & 5 \end{aligned}$ | 0.29 | $\begin{aligned} & 18.4 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.20 | 0.00 | 99.94 | $\begin{aligned} & 73.3 \\ & 0 \\ & \hline \end{aligned}$ |
| BH296 A1 33 | rimwards | 52.53 | 2.07 | $\begin{aligned} & 11.6 \\ & 6 \\ & \hline \end{aligned}$ | 0.52 | 0.06 | $\begin{aligned} & 15.8 \\ & 9 \\ & \hline \end{aligned}$ | 0.32 | $\begin{aligned} & 17.8 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 101.1 \\ & 7 \\ & \hline \end{aligned}$ | $70.8$ |
| BH296 A1 33 | rimwards | 52.44 | 1.99 | $12.6$ | 0.57 | 0.00 | $\begin{aligned} & 15.1 \\ & 3 \\ & \hline \end{aligned}$ | 0.33 | $\begin{aligned} & 17.6 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 101.0 \\ & 0 \end{aligned}$ | $\begin{aligned} & 68.1 \\ & 5 \\ & \hline \end{aligned}$ |
| BH296 A1 33 | rimwards | 51.34 | 1.62 | $\begin{aligned} & 17.5 \\ & 6 \\ & \hline \end{aligned}$ | 0.63 | 0.06 | $\begin{aligned} & 14.0 \\ & 0 \end{aligned}$ | 0.42 | $\begin{aligned} & 14.9 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.7 \\ & 7 \end{aligned}$ | $\begin{aligned} & 58.6 \\ & 8 \end{aligned}$ |
| BH296 A1 33 | rim | 49.94 | 1.08 | $\begin{aligned} & 26.3 \\ & 1 \\ & \hline \end{aligned}$ | 0.44 | 0.01 | 9.97 | 0.62 | $\begin{aligned} & 11.4 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.0 \\ & 2 \end{aligned}$ | $\begin{aligned} & \hline 40.3 \\ & 1 \\ & \hline \end{aligned}$ |
| BH296 A1 33 | x average | 52.15 | 1.75 | $\begin{aligned} & 13.8 \\ & 8 \\ & \hline \end{aligned}$ | 0.50 | 0.03 | $\begin{aligned} & 14.6 \\ & 9 \end{aligned}$ | 0.33 | $\begin{aligned} & 16.7 \\ & 8 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.20 | 0.02 | $\begin{aligned} & 100.3 \\ & 5 \end{aligned}$ | $\begin{aligned} & \hline 65.8 \\ & 6 \\ & \hline \end{aligned}$ |
| BH296 A1 34 | core | 52.04 | 2.10 | $\begin{aligned} & 11.5 \\ & 9 \\ & \hline \end{aligned}$ | 0.55 | 0.08 | $\begin{aligned} & 15.1 \\ & 3 \\ & \hline \end{aligned}$ | 0.28 | $\begin{aligned} & 18.2 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.21 | 0.01 | $\begin{aligned} & 100.2 \\ & 0 \end{aligned}$ | $\begin{aligned} & 69.9 \\ & 5 \\ & \hline \end{aligned}$ |
| BH296 A1 34 | rimwards | 52.57 | 1.72 | $\begin{aligned} & 11.4 \\ & 7 \\ & \hline \end{aligned}$ | 0.38 | 0.01 | $\begin{aligned} & 15.8 \\ & 7 \\ & \hline \end{aligned}$ | 0.26 | $17.8$ | $0.0$ | 0.22 | 0.01 | $100.3$ | $\begin{aligned} & 71.1 \\ & 5 \\ & \hline \end{aligned}$ |
| BH296 A1 34 | rimwards | 52.47 | 1.87 | $\begin{aligned} & 12.0 \\ & 1 \end{aligned}$ | 0.54 | 0.09 | $\begin{aligned} & 15.3 \\ & 9 \\ & \hline \end{aligned}$ | 0.28 | $\begin{aligned} & 17.9 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 100.8 \\ & 3 \end{aligned}$ | $\begin{aligned} & 69.5 \\ & 5 \\ & \hline \end{aligned}$ |
| BH296 A1 34 | rimwards | 52.15 | 2.32 | $\begin{aligned} & 11.6 \\ & 3 \\ & \hline \end{aligned}$ | 0.84 | 0.00 | $\begin{aligned} & 14.7 \\ & 4 \end{aligned}$ | 0.29 | $\begin{aligned} & 18.6 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 5 \end{aligned}$ | 0.30 | 0.01 | $\begin{aligned} & 100.9 \\ & 3 \end{aligned}$ | $\begin{aligned} & 69.3 \\ & 1 \\ & \hline \end{aligned}$ |
| BH296 A1 34 | rimwards | 52.12 | 1.92 | $\begin{aligned} & 12.0 \\ & 9 \\ & \hline \end{aligned}$ | 0.56 | 0.01 | $\begin{aligned} & 15.2 \\ & 1 \\ & \hline \end{aligned}$ | 0.29 | $\begin{aligned} & 18.2 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.21 | 0.02 | $\begin{aligned} & 100.6 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 69.1 \\ & 5 \\ & \hline \end{aligned}$ |
| BH296 A1 34 | rimwards | 51.61 | 1.82 | $\begin{aligned} & 15.3 \\ & 4 \\ & \hline \end{aligned}$ | 0.65 | 0.06 | $\begin{aligned} & 14.1 \\ & 7 \\ & \hline \end{aligned}$ | 0.39 | $\begin{aligned} & 16.5 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.18 | 0.01 | $\begin{aligned} & 100.8 \\ & 1 \end{aligned}$ | $\begin{aligned} & 62.2 \\ & 1 \\ & \hline \end{aligned}$ |
| BH296 A1 34 | rimwards | 51.21 | 1.87 | $\begin{aligned} & 16.1 \\ & 7 \\ & \hline \end{aligned}$ | 0.86 | 0.06 | $\begin{aligned} & 13.5 \\ & 5 \\ & \hline \end{aligned}$ | 0.36 | $\begin{aligned} & 16.8 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.23 | 0.01 | $101.2$ | $\begin{aligned} & 59.9 \\ & 0 \\ & \hline \end{aligned}$ |
| BH296 A1 34 | rimwards | 51.32 | 1.62 | $\begin{aligned} & 16.0 \\ & 8 \end{aligned}$ | 0.79 | 0.00 | $\begin{aligned} & 13.2 \\ & 6 \\ & \hline \end{aligned}$ | 0.39 | $\begin{aligned} & 16.4 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.24 | 0.02 | $\begin{aligned} & 100.1 \\ & 8 \end{aligned}$ | $\begin{aligned} & 59.5 \\ & 0 \end{aligned}$ |
| BH296 A1 34 | rimwards | 51.11 | 1.47 | $17.0$ | 0.64 | 0.03 | $12.5$ | 0.39 | $\begin{aligned} & 16.5 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.23 | 0.01 | 99.97 | $\begin{aligned} & 56.6 \\ & 7 \\ & \hline \end{aligned}$ |
| BH296 A1 34 | rim | 49.68 | 1.89 | $\begin{aligned} & 19.3 \\ & 3 \end{aligned}$ | 0.61 | 0.00 | $\begin{aligned} & 11.0 \\ & 6 \end{aligned}$ | 0.41 | $\begin{aligned} & 15.9 \\ & 8 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 5 \end{aligned}$ | 0.23 | 0.00 | 99.23 | $\begin{aligned} & 50.5 \\ & 0 \end{aligned}$ |
| BH296 A1 34 | x average | 51.63 | 1.86 | $14.2$ | 0.64 | 0.03 | $\begin{aligned} & 14.0 \\ & 9 \end{aligned}$ | 0.33 | $\begin{aligned} & 17.3 \\ & 3 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.23 | 0.01 | $\begin{aligned} & 100.4 \\ & 3 \end{aligned}$ | $\begin{aligned} & 63.7 \\ & 9 \\ & \hline \end{aligned}$ |
| BH296 A1 35 | core | 52.98 | 1.65 | $11.3$ | 0.39 | 0.03 | $\begin{aligned} & 16.2 \\ & 1 \\ & \hline \end{aligned}$ | 0.29 | $\begin{aligned} & 18.0 \\ & 9 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.18 | 0.01 | $\begin{aligned} & 101.1 \\ & 2 \end{aligned}$ | $\begin{aligned} & 71.8 \\ & 8 \\ & \hline \end{aligned}$ |
| BH296 A1 35 | rimwards | 52.35 | 2.06 | $\begin{aligned} & 11.2 \\ & 6 \\ & \hline \end{aligned}$ | 0.45 | 0.02 | $\begin{aligned} & 15.4 \\ & 5 \\ & \hline \end{aligned}$ | 0.32 | $\begin{aligned} & 17.8 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.20 | 0.01 | 99.98 | $\begin{aligned} & 70.9 \\ & 7 \\ & \hline \end{aligned}$ |
| BH296 A1 35 | rimwards | 52.47 | 1.89 | $\begin{aligned} & 10.4 \\ & 9 \end{aligned}$ | 0.44 | 0.00 | $\begin{aligned} & 15.5 \\ & 8 \\ & \hline \end{aligned}$ | 0.26 | $\begin{aligned} & 19.3 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.7 \\ & 0 \end{aligned}$ | $\begin{aligned} & \hline 72.5 \\ & 8 \\ & \hline \end{aligned}$ |
| BH296 A1 35 | rimwards | 52.37 | 1.91 | $\begin{aligned} & 10.6 \\ & 8 \end{aligned}$ | 0.42 | 0.06 | $\begin{aligned} & 15.9 \\ & 6 \\ & \hline \end{aligned}$ | 0.29 | $18.6$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.22 | 0.01 | $\begin{aligned} & 100.5 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 72.6 \\ & 9 \\ & \hline \end{aligned}$ |
| BH296 A1 35 | rimwards | 52.67 | 1.88 | $\begin{aligned} & 11.4 \\ & 8 \\ & \hline \end{aligned}$ | 0.50 | 0.05 | $\begin{aligned} & 15.7 \\ & 8 \\ & \hline \end{aligned}$ | 0.29 | $\begin{aligned} & 18.2 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 101.1 \\ & 0 \end{aligned}$ | $\begin{aligned} & 71.0 \\ & 1 \\ & \hline \end{aligned}$ |
| BH296 A1 35 | rimwards | 51.94 | 1.92 | $12.8$ | 0.64 | 0.00 | $\begin{aligned} & 15.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.32 | $\begin{aligned} & 17.6 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.5 \\ & 7 \end{aligned}$ | $\begin{aligned} & \hline 67.4 \\ & 8 \end{aligned}$ |
| BH296 A1 35 | rimwards | 51.44 | 1.89 | $\begin{aligned} & 15.5 \\ & 4 \\ & \hline \end{aligned}$ | 0.60 | 0.00 | $\begin{aligned} & 14.2 \\ & 6 \end{aligned}$ | 0.39 | $\begin{aligned} & 16.3 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.6 \\ & 8 \end{aligned}$ | $\begin{aligned} & 62.0 \\ & 5 \\ & \hline \end{aligned}$ |
| BH296 A1 35 | rim | 51.27 | 1.58 | $\begin{aligned} & 16.5 \\ & 6 \end{aligned}$ | 0.51 | 0.00 | $13.2$ | 0.43 | $\begin{aligned} & 16.1 \\ & 3 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.20 | 0.01 | 99.94 | $\begin{aligned} & 58.7 \\ & 6 \\ & \hline \end{aligned}$ |
| BH296 A1 35 | x average | 52.19 | 1.85 | $\begin{aligned} & 12.5 \\ & 3 \end{aligned}$ | 0.49 | 0.02 | $\begin{aligned} & 15.1 \\ & 9 \\ & \hline \end{aligned}$ | 0.32 | $\begin{aligned} & 17.7 \\ & 8 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.20 | 0.01 | $\begin{aligned} & 100.5 \\ & 8 \end{aligned}$ | $\begin{aligned} & 68.4 \\ & 3 \\ & \hline \end{aligned}$ |
| BH296 A1 45 | core | 50.50 | 1.19 | $\begin{aligned} & 22.4 \\ & 8 \\ & \hline \end{aligned}$ | 0.48 | 0.10 | $\begin{aligned} & 10.2 \\ & 0 \\ & \hline \end{aligned}$ | 0.52 | $\begin{aligned} & 14.3 \\ & 5 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.17 | 0.00 | 99.98 | $\begin{aligned} & 44.7 \\ & 2 \\ & \hline \end{aligned}$ |
| BH296 A1 45 | rimwards | 50.20 | 1.15 | $\begin{aligned} & 22.1 \\ & 8 \end{aligned}$ | 0.68 | 0.00 | 9.92 | 0.47 | $\begin{aligned} & 15.2 \\ & 8 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.18 | 0.00 | $\begin{aligned} & 100.0 \\ & 8 \end{aligned}$ | $\begin{aligned} & \hline 44.3 \\ & 6 \\ & \hline \end{aligned}$ |
| BH296 A1 45 | rimwards | 49.51 | 1.45 | $\begin{aligned} & 21.9 \\ & 3 \end{aligned}$ | 0.66 | 0.03 | 9.75 | 0.51 | $\begin{aligned} & 15.1 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.23 | 0.00 | 99.20 | $\begin{aligned} & \hline 44.1 \\ & 9 \\ & \hline \end{aligned}$ |
| BH296 A1 45 | rimwards | 49.80 | 1.18 | $\begin{aligned} & 22.3 \\ & 4 \end{aligned}$ | 0.68 | 0.00 | 9.62 | 0.51 | $\begin{aligned} & 15.6 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.20 | 0.00 | 99.94 | $\begin{aligned} & 43.4 \\ & 2 \\ & \hline \end{aligned}$ |
| BH296 A1 45 | rimwards | 49.74 | 1.16 | $\begin{aligned} & 22.4 \\ & 2 \\ & \hline \end{aligned}$ | 0.69 | 0.03 | 9.53 | 0.53 | $\begin{aligned} & 15.5 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.20 | 0.01 | 99.85 | $\begin{aligned} & 43.0 \\ & 9 \\ & \hline \end{aligned}$ |
| BH296 A1 45 | rim | 49.80 | 1.13 | $\begin{aligned} & 22.9 \\ & 8 \\ & \hline \end{aligned}$ | 0.66 | 0.00 | 8.96 | 0.53 | $\begin{aligned} & 15.8 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.19 | 0.01 | $\begin{aligned} & 100.1 \\ & 4 \end{aligned}$ | $\begin{aligned} & 40.9 \\ & 9 \end{aligned}$ |
| BH296 A1 45 | x average | 49.92 | 1.21 | 22.3 | 0.64 | 0.03 | 9.66 | 0.51 | 15.3 | 0.0 | 0.20 | 0.00 | 99.86 | 43.4 |


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| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| MH123 B2 58 | rep | 50.77 | 2.12 | $\begin{aligned} & 14.7 \\ & 1 \\ & \hline \end{aligned}$ | 1.00 | 0.05 | $\begin{aligned} & 14.2 \\ & 8 \\ & \hline \end{aligned}$ | 0.40 | $16.5$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.18 | 0.00 | $\begin{aligned} & 100.0 \\ & 2 \end{aligned}$ | $\begin{aligned} & 63.3 \\ & 7 \\ & \hline \end{aligned}$ |
| MH123 B2 61 | core | 52.23 | 2.64 | 8.47 | 0.43 | 0.04 | $\begin{aligned} & 16.6 \\ & 6 \\ & \hline \end{aligned}$ | 0.18 | $\begin{aligned} & 19.1 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 5 \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.0 \\ & 7 \end{aligned}$ | $\begin{aligned} & 77.8 \\ & 0 \end{aligned}$ |
| MH123 B2 62 | rim | 52.34 | 2.74 | 8.49 | 0.53 | 0.13 | $16.3$ | 0.23 | $\begin{aligned} & 19.5 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.24 | 0.00 | $\begin{aligned} & 100.5 \\ & 8 \end{aligned}$ | $\begin{aligned} & \hline 77.4 \\ & 4 \\ & \hline \end{aligned}$ |
| $\begin{aligned} & \text { MH123 B2 58- } \\ & 62 \end{aligned}$ | average | 51.78 | 2.50 | $\begin{aligned} & 10.5 \\ & 5 \end{aligned}$ | 0.65 | 0.07 | $\begin{aligned} & 15.7 \\ & 6 \\ & \hline \end{aligned}$ | 0.27 | $\begin{aligned} & 18.4 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.2 \\ & 2 \end{aligned}$ | $\begin{aligned} & 72.8 \\ & 7 \\ & \hline \end{aligned}$ |
| MH123 B2 76 | core | 53.16 | 0.81 | $\begin{aligned} & 20.6 \\ & 7 \\ & \hline \end{aligned}$ | 0.34 | 0.08 | $\begin{aligned} & 20.7 \\ & 6 \\ & \hline \end{aligned}$ | 0.46 | 4.41 | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.07 | 0.01 | $\begin{aligned} & 100.7 \\ & 8 \end{aligned}$ | $\begin{aligned} & 64.1 \\ & 5 \\ & \hline \end{aligned}$ |
| MH123 B2 77 | rimward | 51.29 | 1.85 | $\begin{aligned} & 17.1 \\ & 5 \\ & \hline \end{aligned}$ | 0.82 | 0.01 | $\begin{aligned} & 15.5 \\ & 6 \\ & \hline \end{aligned}$ | 0.42 | $\begin{aligned} & 13.3 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 5 \end{aligned}$ | 0.19 | 0.01 | $\begin{aligned} & 100.6 \\ & 3 \end{aligned}$ | $\begin{aligned} & 61.8 \\ & 0 \\ & \hline \end{aligned}$ |
| MH123 B2 78 | rim | 50.55 | 1.05 | $\begin{aligned} & 26.3 \\ & 6 \\ & \hline \end{aligned}$ | 0.68 | 0.00 | $\begin{aligned} & 12.9 \\ & 5 \end{aligned}$ | 0.59 | 8.31 | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.12 | 0.00 | $\begin{aligned} & 100.6 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & 46.6 \\ & 9 \end{aligned}$ |
| $\begin{aligned} & \text { MH123 B2 76- } \\ & 78 \\ & \hline \end{aligned}$ | average | 51.66 | 1.24 | $\begin{aligned} & 21.3 \\ & 9 \\ & \hline \end{aligned}$ | 0.61 | 0.03 | $16.4$ | 0.49 | 8.67 | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.13 | 0.01 | $\begin{aligned} & 100.6 \\ & 8 \end{aligned}$ | $\begin{aligned} & 57.5 \\ & 4 \end{aligned}$ |
| MH123 B2 81 | rim | 50.40 | 1.12 | $\begin{aligned} & 21.0 \\ & 7 \\ & \hline \end{aligned}$ | 0.40 | 0.01 | $\begin{aligned} & 10.8 \\ & 7 \\ & \hline \end{aligned}$ | 0.46 | $\begin{aligned} & 15.1 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.19 | 0.00 | 99.69 | $\begin{aligned} & 47.9 \\ & 0 \end{aligned}$ |
| MH123 B2 82 | core | 52.47 | 1.53 | $\begin{aligned} & 13.2 \\ & 9 \\ & \hline \end{aligned}$ | 0.45 | 0.00 | $16.8$ | 0.34 | $\begin{aligned} & 15.2 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.3 \\ & 6 \end{aligned}$ | $\begin{aligned} & \hline 69.2 \\ & 7 \\ & \hline \end{aligned}$ |
| MH123 B2 83 | core | 52.87 | 1.56 | $\begin{aligned} & 10.9 \\ & 9 \end{aligned}$ | 0.41 | 0.06 | $\begin{aligned} & 16.7 \\ & 3 \end{aligned}$ | 0.27 | $\begin{aligned} & 17.5 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.17 | 0.01 | $\begin{aligned} & 100.6 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 73.0 \\ & 6 \\ & \hline \end{aligned}$ |
| MH123 B2 84 | rim | 52.36 | 1.62 | $\begin{aligned} & 12.6 \\ & 6 \end{aligned}$ | 0.60 | 0.00 | $\begin{aligned} & 16.4 \\ & 4 \end{aligned}$ | 0.34 | $\begin{aligned} & 16.5 \\ & 5 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | 0.21 | 0.01 | $\begin{aligned} & 100.8 \\ & 3 \end{aligned}$ | $\begin{aligned} & \hline 69.8 \\ & 3 \end{aligned}$ |
| $\begin{aligned} & \text { MH123 B2 81- } \\ & 84 \end{aligned}$ | average | 52.03 | 1.46 | $14.5$ | 0.46 | 0.02 | $\begin{aligned} & 15.2 \\ & 2 \end{aligned}$ | 0.35 | $\begin{aligned} & 16.1 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.19 | 0.01 | $\begin{aligned} & 100.3 \\ & 7 \end{aligned}$ | $\begin{aligned} & 65.0 \\ & 1 \\ & \hline \end{aligned}$ |
| MH123 B2 85 | core | 50.83 | 2.26 | $\begin{aligned} & 13.4 \\ & 9 \end{aligned}$ | 0.85 | 0.06 | $\begin{aligned} & 14.1 \\ & 4 \\ & \hline \end{aligned}$ | 0.32 | $\begin{aligned} & 17.4 \\ & 1 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 5 \end{aligned}$ | 0.26 | 0.02 | 99.69 | $\begin{aligned} & 65.1 \\ & 2 \end{aligned}$ |
| MH123 B2 86 | rimward | 51.10 | 2.35 | $\begin{aligned} & 13.6 \\ & 7 \\ & \hline \end{aligned}$ | 0.83 | 0.01 | $\begin{aligned} & 14.9 \\ & 4 \end{aligned}$ | 0.34 | $\begin{aligned} & 16.6 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.23 | 0.01 | $\begin{aligned} & 100.1 \\ & 4 \end{aligned}$ | $\begin{aligned} & 66.0 \\ & 7 \\ & \hline \end{aligned}$ |
| MH123 B2 87 | rim | 51.36 | 2.37 | $\begin{aligned} & 11.1 \\ & 0 \\ & \hline \end{aligned}$ | 0.65 | 0.02 | $\begin{aligned} & 15.6 \\ & 5 \\ & \hline \end{aligned}$ | 0.28 | $\begin{aligned} & 17.8 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.24 | 0.02 | 99.52 | $\begin{aligned} & 71.5 \\ & 4 \\ & \hline \end{aligned}$ |
| $\begin{aligned} & \text { MH123 B2 85- } \\ & 87 \\ & \hline \end{aligned}$ | average | 51.10 | 2.33 | $\begin{aligned} & 12.7 \\ & 5 \\ & \hline \end{aligned}$ | 0.78 | 0.03 | $\begin{aligned} & 14.9 \\ & 1 \end{aligned}$ | 0.31 | $\begin{aligned} & 17.3 \\ & 0 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.24 | 0.01 | 99.78 | $\begin{aligned} & \hline 67.5 \\ & 8 \\ & \hline \end{aligned}$ |
| MH123 B2 91 | core | 52.09 | 2.09 | $12.9$ | 0.63 | 0.00 | $\begin{aligned} & 16.5 \\ & 8 \\ & \hline \end{aligned}$ | 0.37 | $\begin{aligned} & 15.1 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.18 | 0.01 | $\begin{aligned} & 100.0 \\ & 9 \end{aligned}$ | $\begin{aligned} & \hline 69.4 \\ & 9 \\ & \hline \end{aligned}$ |
| MH123 B2 91 | rimward | 51.99 | 2.10 | $\begin{aligned} & 13.5 \\ & 5 \end{aligned}$ | 0.73 | 0.00 | $\begin{aligned} & 16.6 \\ & 3 \end{aligned}$ | 0.33 | $\begin{aligned} & 15.0 \\ & 2 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | 0.28 | 0.02 | $\begin{aligned} & 100.6 \\ & 8 \end{aligned}$ | $\begin{aligned} & 68.6 \\ & 2 \end{aligned}$ |
| MH123 B2 91 | rimward | 51.82 | 2.09 | $\begin{aligned} & 14.0 \\ & 8 \\ & \hline \end{aligned}$ | 0.66 | 0.06 | $\begin{aligned} & 16.5 \\ & 4 \\ & \hline \end{aligned}$ | 0.37 | $\begin{aligned} & 14.8 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.21 | 0.01 | $100.7$ | $\begin{aligned} & \hline 67.6 \\ & 7 \end{aligned}$ |
| MH123 B2 91 | rimward | 51.59 | 2.04 | $\begin{aligned} & 13.8 \\ & 1 \\ & \hline \end{aligned}$ | 0.72 | 0.05 | $\begin{aligned} & 16.4 \\ & 6 \\ & \hline \end{aligned}$ | 0.40 | $\begin{aligned} & 14.6 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.24 | 0.02 | 99.98 | $\begin{aligned} & \hline 68.0 \\ & 0 \end{aligned}$ |
| MH123 B2 91 | rim | 51.69 | 1.98 | $\begin{aligned} & 14.7 \\ & 0 \end{aligned}$ | 0.52 | 0.00 | $16.4$ | 0.38 | $\begin{aligned} & 14.1 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.20 | 0.01 | $\begin{aligned} & 100.1 \\ & 2 \end{aligned}$ | $\begin{aligned} & 66.6 \\ & 4 \end{aligned}$ |
| MH123 B2 91 | x average | 51.84 | 2.06 | $13.8$ | 0.65 | 0.02 | $\begin{aligned} & 16.5 \\ & 4 \\ & \hline \end{aligned}$ | 0.37 | $\begin{aligned} & 14.7 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.22 | 0.01 | $\begin{aligned} & 100.3 \\ & 2 \end{aligned}$ | $\begin{aligned} & 68.0 \\ & 8 \end{aligned}$ |
| MH123 B2 99 | rim | 51.22 | 1.11 | $\begin{aligned} & 18.8 \\ & 1 \end{aligned}$ | 0.45 | 0.04 | $\begin{aligned} & 12.3 \\ & 2 \end{aligned}$ | 0.44 | $\begin{aligned} & 15.6 \\ & 6 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.2 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 53.8 \\ & 6 \end{aligned}$ |
| MH123 B2 99 | rimward | 51.23 | 1.37 | $\begin{aligned} & 16.3 \\ & 9 \\ & \hline \end{aligned}$ | 0.51 | 0.04 | $\begin{aligned} & 13.8 \\ & 7 \\ & \hline \end{aligned}$ | 0.38 | $\begin{aligned} & 16.0 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.22 | 0.01 | $\begin{aligned} & 100.0 \\ & 4 \end{aligned}$ | $\begin{aligned} & 60.1 \\ & 3 \end{aligned}$ |
| MH123 B2 99 | rimward | 51.55 | 1.48 | $15.5$ | 0.53 | 0.00 | $\begin{aligned} & 15.3 \\ & 1 \\ & \hline \end{aligned}$ | 0.41 | $\begin{aligned} & 14.9 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.22 | 0.02 | $\begin{aligned} & 100.0 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 63.7 \\ & 7 \\ & \hline \end{aligned}$ |
| MH123 B2 99 | rimward | 52.03 | 1.60 | $\begin{aligned} & 12.6 \\ & 1 \\ & \hline \end{aligned}$ | 0.63 | 0.04 | $\begin{aligned} & 15.7 \\ & 2 \\ & \hline \end{aligned}$ | 0.33 | $\begin{aligned} & 16.3 \\ & 8 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.21 | 0.02 | 99.58 | $\begin{aligned} & 68.9 \\ & 5 \\ & \hline \end{aligned}$ |
| MH123 B2 99 | core | 51.37 | 2.51 | $\begin{aligned} & 11.4 \\ & 9 \\ & \hline \end{aligned}$ | 0.68 | 0.02 | $\begin{aligned} & 15.4 \\ & 7 \\ & \hline \end{aligned}$ | 0.26 | $\begin{aligned} & 17.8 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.24 | 0.02 | 99.95 | $\begin{aligned} & 70.5 \\ & 9 \end{aligned}$ |
| MH123 B2 99 | core | 51.45 | 2.84 | $\begin{aligned} & 10.9 \\ & 2 \end{aligned}$ | 0.75 | 0.03 | $\begin{aligned} & 15.4 \\ & 7 \\ & \hline \end{aligned}$ | 0.26 | $\begin{aligned} & 18.1 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 0.26 | 0.04 | $\begin{aligned} & 100.1 \\ & 3 \end{aligned}$ | $\begin{aligned} & 71.6 \\ & 3 \end{aligned}$ |
| MH123 B2 99 | rimward | 51.45 | 2.66 | 9.61 | 0.69 | 0.10 | $15.6$ | 0.21 | $\begin{aligned} & 18.6 \\ & 2 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.24 | 0.03 | 99.21 | $\begin{aligned} & 74.3 \\ & 3 \end{aligned}$ |
| MH123 B2 99 | rimward | 51.92 | 2.75 | 9.35 | 0.52 | 0.01 | $\begin{aligned} & 16.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.22 | $\begin{aligned} & 18.9 \\ & 4 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 0.25 | 0.01 | $\begin{aligned} & 100.0 \\ & 1 \end{aligned}$ | $\begin{aligned} & 75.3 \\ & 6 \end{aligned}$ |
| MH123 B2 99 | rimward | 52.99 | 1.62 | 9.27 | 0.39 | 0.05 | $\begin{aligned} & 16.9 \\ & 2 \\ & \hline \end{aligned}$ | 0.24 | $\begin{aligned} & 18.6 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.22 | 0.02 | $\begin{aligned} & 100.4 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 76.4 \\ & 8 \\ & \hline \end{aligned}$ |
| MH123 B2 99 | rim | 53.11 | 1.65 | 9.16 | 0.29 | 0.06 | $\begin{aligned} & 16.9 \\ & 6 \end{aligned}$ | 0.25 | $\begin{aligned} & 18.5 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.20 | 0.01 | $\begin{aligned} & 100.2 \\ & 8 \end{aligned}$ | $\begin{aligned} & 76.7 \\ & 5 \end{aligned}$ |
| MH123 B2 99 | x average | 51.83 | 1.96 | $\begin{aligned} & 12.3 \\ & 1 \\ & \hline \end{aligned}$ | 0.54 | 0.04 | $\begin{aligned} & 15.3 \\ & 7 \\ & \hline \end{aligned}$ | 0.30 | $\begin{aligned} & \hline 17.3 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.23 | 0.02 | 99.99 | $\begin{aligned} & \hline 69.1 \\ & 8 \end{aligned}$ |
| MH123 B2 101 | core | 53.40 | 1.59 | 8.51 | 0.34 | 0.07 | $\begin{aligned} & 17.1 \\ & 4 \\ & \hline \end{aligned}$ | 0.25 | $\begin{aligned} & 18.9 \\ & 5 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 6 \end{aligned}$ | 0.17 | 0.01 | $\begin{aligned} & 100.4 \\ & 9 \end{aligned}$ | $\begin{aligned} & 78.2 \\ & 0 \end{aligned}$ |
| MH123 B2 102 | rimward | 51.71 | 2.89 | 7.99 | 0.53 | 0.23 | $\begin{aligned} & 16.1 \\ & 6 \\ & \hline \end{aligned}$ | 0.23 | $\begin{aligned} & 19.4 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.22 | 0.00 | 99.41 | $\begin{aligned} & 78.2 \\ & 7 \end{aligned}$ |
| MH123 B2 103 | rim | 51.81 | 2.80 | 8.55 | 0.61 | 0.12 | $\begin{aligned} & 16.5 \\ & 6 \end{aligned}$ | 0.20 | $\begin{aligned} & 19.0 \\ & 8 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.23 | 0.01 | $\begin{aligned} & 100.0 \\ & 1 \end{aligned}$ | $\begin{aligned} & 77.5 \\ & 3 \end{aligned}$ |
| MH123 B2 101- | average | 52.31 | 2.43 | 8.35 | 0.49 | 0.14 | 16.6 | 0.23 | 19.1 | 0.0 | 0.21 | 0.01 | 99.97 | 78.0 |


|  | 103 |  |  |  |  |  |  | 2 |  | 6 | 3 |  |  |  | 0 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | MH123 B2 104 | core | 50.68 | 1.98 | $\begin{aligned} & 14.6 \\ & 0 \\ & \hline \end{aligned}$ | 0.84 | 0.00 | $\begin{aligned} & \hline 14.1 \\ & 9 \\ & \hline \end{aligned}$ | 0.35 | $\begin{aligned} & 16.4 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.24 | 0.00 | 99.38 | $\begin{aligned} & \hline 63.4 \\ & 0 \\ & \hline \end{aligned}$ |
|  | MH123 B2 105 | rim | 44.47 | 7.31 | $\begin{aligned} & \hline 23.2 \\ & 4 \end{aligned}$ | 1.13 | 0.00 | $\begin{aligned} & 10.2 \\ & 3 \\ & \hline \end{aligned}$ | 0.21 | 5.76 | $\begin{aligned} & \hline 0.0 \\ & 5 \end{aligned}$ | 0.22 | 4.46 | 97.08 | $\begin{aligned} & 43.9 \\ & 6 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { MH123 B2 104- } \\ & 105 \\ & \hline \end{aligned}$ | average | 47.58 | 4.64 | $\begin{aligned} & 18.9 \\ & 2 \end{aligned}$ | 0.98 | 0.00 | $\begin{aligned} & 12.2 \\ & 1 \\ & \hline \end{aligned}$ | 0.28 | $\begin{aligned} & 11.1 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.23 | 2.23 | 98.23 | $\begin{aligned} & 53.6 \\ & 8 \\ & \hline \end{aligned}$ |
| C | MH129 A2 125 | core | 52.13 | 2.90 | 7.95 | 0.59 | 0.22 | $\begin{aligned} & 16.6 \\ & 4 \end{aligned}$ | 0.17 | $\begin{aligned} & 19.5 \\ & 9 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.4 \\ & 5 \end{aligned}$ | $\begin{aligned} & 78.8 \\ & 6 \\ & \hline \end{aligned}$ |
|  | MH129 A2 126 | rimward | 52.40 | 2.62 | 7.92 | 0.54 | 0.17 | $\begin{aligned} & 16.6 \\ & 6 \\ & \hline \end{aligned}$ | 0.21 | $\begin{aligned} & 19.7 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.5 \\ & 2 \end{aligned}$ | $\begin{aligned} & 78.9 \\ & 5 \\ & \hline \end{aligned}$ |
|  | MH129 A2 127 | rim | 51.17 | 1.25 | $\begin{aligned} & 17.4 \\ & 8 \\ & \hline \end{aligned}$ | 0.71 | 0.05 | $\begin{aligned} & 12.6 \\ & 6 \\ & \hline \end{aligned}$ | 0.40 | $\begin{aligned} & 16.2 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 5 \end{aligned}$ | 0.24 | 0.01 | $\begin{aligned} & 100.2 \\ & 9 \end{aligned}$ | $\begin{aligned} & 56.3 \\ & 4 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { MH129 A2 } \\ & 125-127 \\ & \hline \end{aligned}$ | average | 51.90 | 2.26 | $\begin{aligned} & 11.1 \\ & 2 \\ & \hline \end{aligned}$ | 0.61 | 0.15 | $15.3$ | 0.26 | $\begin{aligned} & 18.5 \\ & 4 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.4 \\ & 2 \end{aligned}$ | $\begin{aligned} & 71.3 \\ & 9 \\ & \hline \end{aligned}$ |
|  | MH129 A2 132 | core | 51.89 | 0.71 | $23.7$ | 0.44 | 0.00 | $\begin{aligned} & 17.9 \\ & 6 \\ & \hline \end{aligned}$ | 0.53 | 4.75 | $\begin{aligned} & 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.07 | 0.00 | $\begin{aligned} & 100.0 \\ & 9 \end{aligned}$ | $\begin{aligned} & 57.4 \\ & 3 \\ & \hline \end{aligned}$ |
|  | MH129 A2 133 | rim | 51.70 | 0.47 | $\begin{aligned} & 17.3 \\ & 9 \end{aligned}$ | 0.16 | 0.00 | $\begin{aligned} & 10.4 \\ & 7 \end{aligned}$ | 0.35 | $\begin{aligned} & 19.8 \\ & 2 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 0.18 | 0.00 | $\begin{aligned} & 100.5 \\ & 4 \end{aligned}$ | $\begin{aligned} & 51.7 \\ & 7 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { MH129 A2 } \\ & 132-133 \end{aligned}$ | average | 51.80 | 0.59 | $\begin{aligned} & 20.5 \\ & 6 \\ & \hline \end{aligned}$ | 0.30 | 0.00 | $\begin{aligned} & 14.2 \\ & 2 \end{aligned}$ | 0.44 | $\begin{aligned} & 12.2 \\ & 8 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.12 | 0.00 | $\begin{aligned} & 100.3 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & 54.6 \\ & 0 \\ & \hline \end{aligned}$ |
|  | MH129 A2 151 | rep | 51.29 | 1.26 | $\begin{aligned} & 22.4 \\ & 5 \end{aligned}$ | 0.62 | 0.00 | $\begin{aligned} & 17.5 \\ & 1 \\ & \hline \end{aligned}$ | 0.53 | 6.17 | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.10 | 0.00 | 99.95 | $\begin{aligned} & \hline 58.1 \\ & 6 \\ & \hline \end{aligned}$ |
|  | MH129 A2 152 | core | 52.04 | 1.02 | $\begin{aligned} & 22.2 \\ & 7 \\ & \hline \end{aligned}$ | 0.50 | 0.02 | $\begin{aligned} & 18.2 \\ & 8 \\ & \hline \end{aligned}$ | 0.44 | 5.85 | $\begin{aligned} & \hline 0.0 \\ & 6 \end{aligned}$ | 0.08 | 0.00 | $\begin{aligned} & 100.5 \\ & 5 \end{aligned}$ | $\begin{aligned} & 59.4 \\ & 0 \end{aligned}$ |
|  | MH129 A2 152 | rimwards | 52.38 | 1.00 | $\begin{aligned} & 21.4 \\ & 5 \\ & \hline \end{aligned}$ | 0.49 | 0.00 | $\begin{aligned} & 19.1 \\ & 4 \\ & \hline \end{aligned}$ | 0.44 | 4.85 | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.09 | 0.01 | 99.88 | $\begin{aligned} & 61.3 \\ & 9 \\ & \hline \end{aligned}$ |
|  | MH129 A2 152 | rimwards | 52.48 | 1.06 | $\begin{aligned} & 21.2 \\ & 9 \\ & \hline \end{aligned}$ | 0.39 | 0.01 | $\begin{aligned} & 19.6 \\ & 7 \\ & \hline \end{aligned}$ | 0.54 | 4.27 | $\begin{aligned} & 0.0 \\ & 5 \end{aligned}$ | 0.08 | 0.01 | 99.84 | $\begin{aligned} & \hline 62.2 \\ & 2 \end{aligned}$ |
|  | MH129 A2 152 | rimwards | 51.92 | 1.20 | $\begin{aligned} & 22.1 \\ & 9 \end{aligned}$ | 0.37 | 0.00 | $\begin{aligned} & 19.6 \\ & 3 \end{aligned}$ | 0.47 | 3.36 | $\begin{aligned} & 0.0 \\ & 6 \end{aligned}$ | 0.06 | 0.01 | 99.28 | $\begin{aligned} & 61.1 \\ & 9 \\ & \hline \end{aligned}$ |
|  | MH129 A2 152 | rimwards | 52.41 | 1.00 | $\begin{aligned} & 22.9 \\ & 9 \\ & \hline \end{aligned}$ | 0.42 | 0.00 | $\begin{aligned} & 19.1 \\ & 7 \\ & \hline \end{aligned}$ | 0.51 | 4.08 | $\begin{aligned} & \hline 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.09 | 0.00 | $\begin{aligned} & 100.7 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & 59.7 \\ & 8 \\ & \hline \end{aligned}$ |
|  | MH129 A2 152 | rimwards | 52.20 | 1.22 | $\begin{aligned} & 20.8 \\ & 5 \end{aligned}$ | 0.58 | 0.04 | $\begin{aligned} & 18.1 \\ & 1 \\ & \hline \end{aligned}$ | 0.46 | 7.13 | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.10 | 0.01 | $\begin{aligned} & 100.7 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 60.7 \\ & 5 \end{aligned}$ |
|  | MH129 A2 152 | rimwards | 51.40 | 0.94 | $\begin{aligned} & 24.1 \\ & 5 \\ & \hline \end{aligned}$ | 0.36 | 0.00 | $17.2$ | 0.50 | 4.34 | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.10 | 0.01 | 99.12 | $\begin{aligned} & 56.0 \\ & 6 \\ & \hline \end{aligned}$ |
|  | MH129 A2 152 | rimwards | 51.46 | 1.21 | $\begin{aligned} & 14.9 \\ & 8 \end{aligned}$ | 0.75 | 0.00 | $\begin{aligned} & 12.8 \\ & 3 \\ & \hline \end{aligned}$ | 0.36 | $\begin{aligned} & 18.2 \\ & 4 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.25 | 0.00 | $\begin{aligned} & 100.1 \\ & 1 \end{aligned}$ | $\begin{aligned} & 60.4 \\ & 2 \\ & \hline \end{aligned}$ |
|  | MH129 A2 152 | rimwards | 52.04 | 1.84 | $\begin{aligned} & 11.7 \\ & 9 \\ & \hline \end{aligned}$ | 0.58 | 0.00 | $\begin{aligned} & 15.5 \\ & 9 \end{aligned}$ | 0.32 | $\begin{aligned} & 17.5 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.22 | 0.01 | 99.97 | $\begin{aligned} & 70.2 \\ & 2 \\ & \hline \end{aligned}$ |
|  | MH129 A2 152 | rimwards | 51.92 | 1.96 | $\begin{aligned} & 11.9 \\ & 9 \end{aligned}$ | 0.55 | 0.02 | $\begin{aligned} & 15.7 \\ & 3 \\ & \hline \end{aligned}$ | 0.31 | $\begin{aligned} & 17.4 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.23 | 0.01 | $\begin{aligned} & 100.1 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 70.0 \\ & 3 \\ & \hline \end{aligned}$ |
|  | MH129 A2 152 | rimwards | 52.16 | 1.94 | $\begin{aligned} & 11.6 \\ & 8 \end{aligned}$ | 0.52 | 0.03 | $\begin{aligned} & 15.5 \\ & 3 \\ & \hline \end{aligned}$ | 0.34 | $\begin{aligned} & 17.8 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 100.3 \\ & 4 \end{aligned}$ | $\begin{aligned} & 70.3 \\ & 2 \end{aligned}$ |
|  | MH129 A2 152 | rimwards | 51.97 | 2.12 | $\begin{aligned} & 11.8 \\ & 4 \\ & \hline \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 13.1 \\ & 1 \\ & \hline \end{aligned}$ | 0.28 | $\begin{aligned} & 20.0 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 0.35 | 0.04 | $\begin{aligned} & 100.0 \\ & 0 \end{aligned}$ | $\begin{aligned} & 66.3 \\ & 6 \end{aligned}$ |
|  | MH129 A2 152 | rimwards | 52.45 | 2.11 | 9.24 | 0.62 | 0.13 | $\begin{aligned} & 16.1 \\ & 2 \\ & \hline \end{aligned}$ | 0.22 | $\begin{aligned} & 19.4 \\ & 7 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.21 | 0.01 | $\begin{aligned} & 100.5 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & 75.6 \\ & 7 \\ & \hline \end{aligned}$ |
|  | MH129 A2 152 | rimwards | 52.86 | 2.18 | 7.90 | 0.53 | 0.03 | $\begin{aligned} & 16.5 \\ & 3 \\ & \hline \end{aligned}$ | 0.20 | $\begin{aligned} & 20.2 \\ & 2 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.21 | 0.01 | $\begin{aligned} & 100.6 \\ & 8 \end{aligned}$ | $\begin{aligned} & 78.8 \\ & 5 \\ & \hline \end{aligned}$ |
|  | MH129 A2 152 | rim | 52.45 | 2.01 | $\begin{aligned} & 10.4 \\ & 6 \end{aligned}$ | 0.33 | 0.07 | $\begin{aligned} & 15.1 \\ & 5 \\ & \hline \end{aligned}$ | 0.24 | $\begin{aligned} & 19.7 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.21 | 0.02 | $\begin{aligned} & 100.6 \\ & 5 \end{aligned}$ | $\begin{aligned} & 72.0 \\ & 7 \\ & \hline \end{aligned}$ |
|  | MH129 A2 152 | x average | 52.14 | 1.52 | $\begin{aligned} & 16.3 \\ & 4 \\ & \hline \end{aligned}$ | 0.48 | 0.02 | $\begin{aligned} & 16.7 \\ & 9 \\ & \hline \end{aligned}$ | 0.38 | $\begin{aligned} & 12.3 \\ & 0 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.17 | 0.01 | $\begin{aligned} & 100.1 \\ & 7 \end{aligned}$ | $\begin{aligned} & 65.6 \\ & 5 \\ & \hline \end{aligned}$ |
|  | MH129 A2 153 | core | 52.25 | 2.93 | 7.98 | 0.42 | 0.15 | $\begin{aligned} & 16.8 \\ & 9 \\ & \hline \end{aligned}$ | 0.22 | $\begin{aligned} & 19.2 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.3 \\ & 2 \end{aligned}$ | $\begin{aligned} & 79.0 \\ & 4 \end{aligned}$ |
|  | MH129 A2 153 | rimwards | 52.44 | 2.92 | 8.06 | 0.47 | 0.28 | $\begin{aligned} & 17.2 \\ & 4 \end{aligned}$ | 0.20 | $\begin{aligned} & 18.6 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 6 \end{aligned}$ | 0.18 | 0.00 | $\begin{aligned} & 100.4 \\ & 3 \end{aligned}$ | $\begin{aligned} & 79.2 \\ & 2 \end{aligned}$ |
|  | MH129 A2 153 | rimwards | 52.62 | 2.49 | 7.66 | 0.58 | 0.21 | $\begin{aligned} & 16.6 \\ & 9 \\ & \hline \end{aligned}$ | 0.19 | $\begin{aligned} & 19.6 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.19 | 0.01 | $\begin{aligned} & 100.3 \\ & 1 \end{aligned}$ | $\begin{aligned} & 79.5 \\ & 1 \\ & \hline \end{aligned}$ |
|  | MH129 A2 153 | rimwards | 52.24 | 2.18 | 9.28 | 0.49 | 0.06 | $\begin{aligned} & 15.9 \\ & 3 \\ & \hline \end{aligned}$ | 0.22 | $19.5$ | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.23 | 0.01 | $\begin{aligned} & 100.1 \\ & 6 \end{aligned}$ | $\begin{aligned} & 75.3 \\ & 7 \\ & \hline \end{aligned}$ |
|  | MH129 A2 153 | rimwards | 47.75 | 2.26 | $\begin{aligned} & 10.2 \\ & 0 \end{aligned}$ | 0.31 | 0.04 | $\begin{aligned} & 14.9 \\ & 4 \\ & \hline \end{aligned}$ | 0.29 | $\begin{aligned} & 17.0 \\ & 3 \end{aligned}$ | $\begin{aligned} & \hline 0.1 \\ & 1 \end{aligned}$ | 0.45 | 0.08 | 93.44 | $\begin{aligned} & 72.3 \\ & 0 \end{aligned}$ |
|  | MH129 A2 153 | rimwards | 52.07 | 1.01 | $\begin{aligned} & 23.1 \\ & 4 \end{aligned}$ | 0.51 | 0.00 | $\begin{aligned} & 18.2 \\ & 0 \\ & \hline \end{aligned}$ | 0.54 | 4.95 | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.08 | 0.00 | $\begin{aligned} & 100.5 \\ & 1 \end{aligned}$ | $\begin{aligned} & 58.3 \\ & 6 \end{aligned}$ |
|  | MH129 A2 153 | rimwards | 52.56 | 1.29 | $\begin{aligned} & 19.8 \\ & 9 \\ & \hline \end{aligned}$ | 0.51 | 0.00 | $\begin{aligned} & 19.5 \\ & 2 \\ & \hline \end{aligned}$ | 0.44 | 6.59 | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.07 | 0.01 | $\begin{aligned} & 100.8 \\ & 9 \end{aligned}$ | $\begin{aligned} & \hline 63.6 \\ & 2 \\ & \hline \end{aligned}$ |
|  | MH129 A2 153 | rimwards | 52.68 | 1.07 | $\begin{aligned} & 19.7 \\ & 8 \end{aligned}$ | 0.35 | 0.00 | $\begin{aligned} & 20.0 \\ & 2 \end{aligned}$ | 0.45 | 5.93 | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.10 | 0.01 | $\begin{aligned} & 100.4 \\ & 1 \end{aligned}$ | $\begin{aligned} & 64.3 \\ & 3 \end{aligned}$ |
|  | MH129 A2 153 | rimwards | 52.34 | 1.00 | $\begin{aligned} & 23.5 \\ & 1 \\ & \hline \end{aligned}$ | 0.51 | 0.05 | $\begin{aligned} & 19.4 \\ & 4 \end{aligned}$ | 0.52 | 3.36 | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | 0.04 | 0.01 | $\begin{aligned} & 100.8 \\ & 2 \end{aligned}$ | $\begin{aligned} & 59.5 \\ & 7 \end{aligned}$ |
|  | MH129 A2 153 | rim | 52.20 | 1.12 | $\begin{aligned} & 22.9 \\ & 7 \\ & \hline \end{aligned}$ | 0.32 | 0.00 | $\begin{aligned} & 18.9 \\ & 8 \\ & \hline \end{aligned}$ | 0.48 | 4.73 | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.07 | 0.00 | $\begin{aligned} & 100.8 \\ & 8 \end{aligned}$ | $\begin{aligned} & 59.5 \\ & 5 \end{aligned}$ |
|  | MH129 A2 153 | x average | 51.92 | 1.83 | 15.2 | 0.45 | 0.08 | 17.7 | 0.35 | 11.9 | 0.0 | 0.16 | 0.01 | 99.82 | 69.0 |


|  |  |  |  |  | 5 |  |  | 8 |  | 6 | 3 |  |  |  | 9 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | MH129 A2 154 | core | 53.13 | 1.97 | 7.75 | 0.36 | 0.14 | $\begin{aligned} & 17.3 \\ & 7 \\ & \hline \end{aligned}$ | 0.23 | $\begin{aligned} & 19.4 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.20 | 0.01 | $\begin{aligned} & 100.6 \\ & 3 \end{aligned}$ | $\begin{aligned} & 79.9 \\ & 7 \\ & \hline \end{aligned}$ |
|  | MH129 A2 155 | rimwards | 52.87 | 2.05 | 9.96 | 0.66 | 0.00 | $\begin{aligned} & 16.4 \\ & 2 \\ & \hline \end{aligned}$ | 0.24 | $\begin{aligned} & \hline 18.7 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 101.2 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & 74.6 \\ & 0 \end{aligned}$ |
|  | MH129 A2 156 | rim | 52.23 | 1.07 | $\begin{aligned} & 22.3 \\ & 2 \\ & \hline \end{aligned}$ | 0.41 | 0.07 | $\begin{aligned} & 19.1 \\ & 7 \\ & \hline \end{aligned}$ | 0.50 | 5.03 | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.09 | 0.00 | $\begin{aligned} & 100.9 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 60.4 \\ & 8 \end{aligned}$ |
|  | $\begin{aligned} & \text { MH129 A2 } \\ & 154-156 \\ & \hline \end{aligned}$ | average | 52.75 | 1.69 | $\begin{aligned} & 13.3 \\ & 5 \end{aligned}$ | 0.48 | 0.07 | $\begin{aligned} & 17.6 \\ & 5 \\ & \hline \end{aligned}$ | 0.32 | $\begin{aligned} & 14.4 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.17 | 0.00 | $\begin{aligned} & 100.9 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & 71.6 \\ & 8 \\ & \hline \end{aligned}$ |
|  | MH129 A2 163 | core | 52.20 | 2.03 | $\begin{aligned} & 11.9 \\ & 2 \end{aligned}$ | 0.58 | 0.03 | $\begin{aligned} & 15.5 \\ & 4 \\ & \hline \end{aligned}$ | 0.29 | $\begin{aligned} & 18.4 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.23 | 0.01 | $\begin{aligned} & 101.2 \\ & 5 \end{aligned}$ | $\begin{aligned} & \hline 69.9 \\ & 1 \\ & \hline \end{aligned}$ |
|  | MH129 A2 164 | rimward | 52.54 | 2.21 | 8.78 | 0.56 | 0.07 | $\begin{aligned} & 16.5 \\ & 6 \\ & \hline \end{aligned}$ | 0.21 | $\begin{aligned} & 19.9 \\ & 2 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 101.0 \\ & 7 \end{aligned}$ | $\begin{aligned} & 77.0 \\ & 7 \\ & \hline \end{aligned}$ |
|  | MH129 A2 165 | rim | 52.40 | 2.21 | 8.83 | 0.47 | 0.08 | $\begin{aligned} & 16.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.19 | $\begin{aligned} & 19.8 \\ & 1 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.2 \\ & 6 \end{aligned}$ | $\begin{aligned} & \hline 76.4 \\ & 1 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { MH129 A2 } \\ & 163-165 \\ & \hline \end{aligned}$ | average | 52.38 | 2.15 | 9.84 | 0.54 | 0.06 | $\begin{aligned} & 16.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.23 | $\begin{aligned} & 19.3 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.8 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 74.4 \\ & 6 \\ & \hline \end{aligned}$ |
|  | MH129 A2 166 | rep | 51.52 | 2.28 | $\begin{aligned} & 12.9 \\ & 2 \end{aligned}$ | 0.55 | 0.01 | $\begin{aligned} & 14.7 \\ & 1 \\ & \hline \end{aligned}$ | 0.30 | $\begin{aligned} & 18.0 \\ & 0 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.24 | 0.01 | $\begin{aligned} & 100.5 \\ & 6 \end{aligned}$ | $\begin{aligned} & 66.9 \\ & 9 \end{aligned}$ |
|  | MH129 A2 167 | core | 52.14 | 3.12 | 8.01 | 0.52 | 0.17 | $\begin{aligned} & 16.7 \\ & 2 \\ & \hline \end{aligned}$ | 0.19 | $\begin{aligned} & 19.2 \\ & 4 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 7 \\ & \hline \end{aligned}$ | 0.21 | 0.01 | $\begin{aligned} & 100.3 \\ & 9 \end{aligned}$ | $\begin{aligned} & 78.8 \\ & 2 \\ & \hline \end{aligned}$ |
|  | MH129 A2 168 | rim | 52.49 | 2.71 | 7.86 | 0.55 | 0.05 | $\begin{aligned} & 16.8 \\ & 7 \\ & \hline \end{aligned}$ | 0.21 | $\begin{aligned} & 19.5 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 6 \\ & \hline \end{aligned}$ | 0.22 | 0.01 | $\begin{aligned} & 100.5 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 79.2 \\ & 8 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { MH129 A2 } \\ & 167-168 \end{aligned}$ | average | 52.32 | 2.92 | 7.93 | 0.53 | 0.11 | $\begin{aligned} & 16.7 \\ & 9 \\ & \hline \end{aligned}$ | 0.20 | $\begin{aligned} & 19.3 \\ & 8 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 6 \end{aligned}$ | 0.21 | 0.01 | $\begin{aligned} & 100.4 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 79.0 \\ & 5 \\ & \hline \end{aligned}$ |
| B | BH185 A2_139 | core | 52.75 | 2.44 | 7.34 | 0.42 | 0.57 | $17.8$ | 0.19 | $\begin{aligned} & 18.7 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.25 | 0.00 | $\begin{aligned} & 100.6 \\ & 6 \end{aligned}$ | $\begin{aligned} & 81.2 \\ & 9 \\ & \hline \end{aligned}$ |
|  | BH185 A2_140 | rim | 52.36 | 2.87 | 6.09 | 0.34 | 0.73 | $\begin{aligned} & 16.8 \\ & 7 \\ & \hline \end{aligned}$ | 0.15 | $\begin{aligned} & 20.9 \\ & 1 \end{aligned}$ | $0.0$ | 0.24 | 0.00 | $\begin{aligned} & 100.5 \\ & 7 \end{aligned}$ | $\begin{aligned} & 83.1 \\ & 7 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_1 } \\ & 39-40 \end{aligned}$ | average | 52.56 | 2.66 | 6.71 | 0.38 | 0.65 | $\begin{aligned} & 17.3 \\ & 8 \\ & \hline \end{aligned}$ | 0.17 | $\begin{aligned} & 19.8 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.24 | 0.00 | $\begin{aligned} & 100.6 \\ & 1 \end{aligned}$ | $\begin{aligned} & 82.2 \\ & 3 \\ & \hline \end{aligned}$ |
|  | BH185 A2_1 42 | core | 53.74 | 1.76 | 7.44 | 0.26 | 0.43 | $\begin{aligned} & 19.1 \\ & 2 \\ & \hline \end{aligned}$ | 0.21 | $\begin{aligned} & 17.7 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 6 \\ & \hline \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.9 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & 82.0 \\ & 9 \\ & \hline \end{aligned}$ |
|  | BH185 A2_143 | rim | 52.57 | 2.71 | 6.13 | 0.44 | 0.64 | $16.9$ | 0.15 | $\begin{aligned} & 20.9 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 7 \end{aligned}$ | 0.20 | 0.01 | $\begin{aligned} & 100.7 \\ & 8 \end{aligned}$ | $\begin{aligned} & 83.1 \\ & 0 \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_1 } \\ & 42-43 \end{aligned}$ | average | 53.15 | 2.24 | 6.78 | 0.35 | 0.53 | $\begin{aligned} & 18.0 \\ & 2 \end{aligned}$ | 0.18 | $\begin{aligned} & 19.3 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 6 \\ & \hline \end{aligned}$ | 0.20 | 0.01 | $\begin{aligned} & 100.8 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 82.6 \\ & 0 \\ & \hline \end{aligned}$ |
|  | BH185 A2_149 | core | 51.97 | 3.14 | 6.99 | 0.47 | 0.46 | $\begin{aligned} & 16.8 \\ & 3 \\ & \hline \end{aligned}$ | 0.19 | $\begin{aligned} & 20.3 \\ & 5 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.6 \\ & 4 \end{aligned}$ | $\begin{aligned} & \hline 81.1 \\ & 0 \\ & \hline \end{aligned}$ |
|  | BH185 A2_1 50 | rimwards | 51.69 | 2.76 | 9.33 | 0.66 | 0.00 | $\begin{aligned} & 15.8 \\ & 4 \\ & \hline \end{aligned}$ | 0.20 | $\begin{aligned} & 19.7 \\ & 3 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.23 | 0.00 | $100.4$ | $\begin{aligned} & 75.1 \\ & 6 \\ & \hline \end{aligned}$ |
|  | BH185 A2_1 51 | rim | 50.91 | 2.55 | $\begin{aligned} & 12.9 \\ & 7 \end{aligned}$ | 0.86 | 0.00 | $\begin{aligned} & 14.3 \\ & 0 \\ & \hline \end{aligned}$ | 0.30 | $\begin{aligned} & 18.2 \\ & 8 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.28 | 0.01 | $\begin{aligned} & 100.4 \\ & 5 \end{aligned}$ | $\begin{aligned} & 66.2 \\ & 7 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_1 } \\ & 49-51 \end{aligned}$ | average | 51.52 | 2.82 | 9.76 | 0.66 | 0.15 | $\begin{aligned} & 15.6 \\ & 6 \\ & \hline \end{aligned}$ | 0.23 | $\begin{aligned} & 19.4 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.24 | 0.00 | $\begin{aligned} & 100.5 \\ & 1 \end{aligned}$ | $\begin{aligned} & 74.1 \\ & 8 \\ & \hline \end{aligned}$ |
|  | BH185 A2_2 47 | core | 53.24 | 1.63 | 7.51 | 0.46 | 0.12 | $\begin{aligned} & 17.6 \\ & 8 \\ & \hline \end{aligned}$ | 0.22 | $19.3$ | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.4 \\ & 0 \end{aligned}$ | $\begin{aligned} & 80.7 \\ & 4 \end{aligned}$ |
|  | BH185 A2_2 48 | rim | 51.61 | 2.54 | 9.15 | 0.54 | 0.00 | $\begin{aligned} & 16.6 \\ & 9 \end{aligned}$ | 0.23 | $\begin{aligned} & 18.2 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 0.21 | 0.02 | 99.20 | $\begin{aligned} & 76.4 \\ & 7 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_2 } \\ & 47-48 \end{aligned}$ | average | 52.43 | 2.09 | 8.33 | 0.50 | 0.06 | $\begin{aligned} & 17.1 \\ & 0 \end{aligned}$ | 0.22 | $\begin{aligned} & 18.7 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.20 | 0.01 | 99.80 | $\begin{aligned} & 78.6 \\ & 1 \\ & \hline \end{aligned}$ |
|  | BH185 A2_2 49 | core | 52.02 | 2.95 | 8.19 | 0.68 | 0.09 | $\begin{aligned} & 16.5 \\ & 4 \\ & \hline \end{aligned}$ | 0.15 | $\begin{aligned} & 19.7 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.6 \\ & 1 \end{aligned}$ | $\begin{aligned} & 78.2 \\ & 5 \\ & \hline \end{aligned}$ |
|  | BH185 A2_2 50 | rim | 52.23 | 2.62 | 8.60 | 0.58 | 0.07 | $\begin{aligned} & 16.3 \\ & 8 \\ & \hline \end{aligned}$ | 0.27 | $\begin{aligned} & 19.2 \\ & 8 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | 0.24 | 0.01 | $\begin{aligned} & 100.3 \\ & 4 \end{aligned}$ | $\begin{aligned} & 77.2 \\ & 4 \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_2 } \\ & 49-50 \end{aligned}$ | average | 52.13 | 2.79 | 8.40 | 0.63 | 0.08 | $\begin{aligned} & 16.4 \\ & 6 \\ & \hline \end{aligned}$ | 0.21 | $\begin{aligned} & 19.5 \\ & 2 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.22 | 0.01 | $\begin{aligned} & 100.4 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 77.7 \\ & 4 \end{aligned}$ |
|  | BH185 A2_2 52 | core | 52.23 | 2.20 | $\begin{aligned} & 10.1 \\ & 9 \end{aligned}$ | 0.35 | 0.04 | $\begin{aligned} & 17.0 \\ & 1 \end{aligned}$ | 0.20 | $\begin{aligned} & 17.7 \\ & 7 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.20 | 0.02 | $\begin{aligned} & 100.2 \\ & 2 \end{aligned}$ | $\begin{aligned} & 74.8 \\ & 3 \\ & \hline \end{aligned}$ |
|  | BH185 A2_2 53 | rim | 50.83 | 1.72 | $\begin{aligned} & 16.9 \\ & 5 \end{aligned}$ | 0.77 | 0.01 | $\begin{aligned} & 12.9 \\ & 7 \\ & \hline \end{aligned}$ | 0.44 | $\begin{aligned} & 16.3 \\ & 5 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.24 | 0.01 | $\begin{aligned} & 100.3 \\ & 1 \end{aligned}$ | $\begin{aligned} & 57.6 \\ & 8 \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_2 } \\ & 52-53 \end{aligned}$ | average | 51.53 | 1.96 | $\begin{aligned} & 13.5 \\ & 7 \\ & \hline \end{aligned}$ | 0.56 | 0.03 | $\begin{aligned} & 14.9 \\ & 9 \end{aligned}$ | 0.32 | $\begin{aligned} & 17.0 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.22 | 0.01 | $\begin{aligned} & 100.2 \\ & 6 \end{aligned}$ | $\begin{aligned} & 66.2 \\ & 6 \\ & \hline \end{aligned}$ |
|  | BH185 A2_2 56 | core | 51.98 | 2.68 | 9.90 | 0.55 | 0.03 | $\begin{aligned} & 16.1 \\ & 1 \\ & \hline \end{aligned}$ | 0.25 | $\begin{aligned} & 18.9 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.22 | 0.01 | $\begin{aligned} & 100.6 \\ & 1 \end{aligned}$ | $\begin{aligned} & 74.3 \\ & 6 \\ & \hline \end{aligned}$ |
|  | BH185 A2_2 57 | rim | 52.29 | 2.43 | 8.90 | 0.59 | 0.08 | $17.2$ | 0.20 | $\begin{aligned} & 18.3 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.23 | 0.01 | $\begin{aligned} & 100.3 \\ & 2 \end{aligned}$ | $\begin{aligned} & 77.5 \\ & 2 \end{aligned}$ |
|  | $\begin{aligned} & \hline \text { BH185 A2_2 } \\ & 56-57 \end{aligned}$ | average | 52.13 | 2.55 | 9.40 | 0.57 | 0.06 | $\begin{aligned} & 16.6 \\ & 6 \\ & \hline \end{aligned}$ | 0.22 | $\begin{aligned} & 18.6 \\ & 4 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.22 | 0.01 | $\begin{aligned} & 100.4 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 75.9 \\ & 4 \end{aligned}$ |
|  | BH185 A2_2 58 | rim (overgrowt h) | 53.11 | 1.68 | 8.47 | 0.43 | 0.01 | $\begin{aligned} & 16.9 \\ & 7 \end{aligned}$ | 0.22 | $\begin{aligned} & 19.4 \\ & 4 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.19 | 0.01 | $\begin{aligned} & 100.5 \\ & 7 \end{aligned}$ | $\begin{aligned} & 78.1 \\ & 1 \end{aligned}$ |
|  | BH185 A2_2 59 | core | 52.83 | 2.26 | 5.98 | 0.39 | 0.48 | $\begin{aligned} & 17.3 \\ & 1 \\ & \hline \end{aligned}$ | 0.16 | $\begin{aligned} & 20.7 \\ & 0 \\ & \hline \end{aligned}$ | $0.0$ | 0.20 | 0.01 | $\begin{aligned} & 100.3 \\ & 2 \end{aligned}$ | $\begin{aligned} & 83.7 \\ & 5 \\ & \hline \end{aligned}$ |
|  | BH185 A2_2 60 | rimwards | 52.44 | 2.73 | 5.99 | 0.37 | 0.63 | $\begin{aligned} & 16.9 \\ & 5 \end{aligned}$ | 0.11 | $\begin{aligned} & 21.0 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 5 \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.4 \\ & 8 \end{aligned}$ | $\begin{aligned} & 83.4 \\ & 5 \end{aligned}$ |


|  | BH185 A2_2 61 | rim | 53.16 | 2.18 | 5.67 | 0.36 | 0.55 | $\begin{aligned} & 17.3 \\ & 6 \\ & \hline \end{aligned}$ | 0.16 | $\begin{aligned} & \hline 21.0 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.23 | 0.01 | $\begin{aligned} & 100.6 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & 84.5 \\ & 1 \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\begin{aligned} & \text { BH185 A2_2 } \\ & 59-61 \end{aligned}$ | average | 52.81 | 2.39 | 5.88 | 0.37 | 0.55 | $\begin{aligned} & 17.2 \\ & 0 \\ & \hline \end{aligned}$ | 0.14 | $\begin{aligned} & \hline 20.9 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.21 | 0.01 | $\begin{aligned} & 100.5 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 83.9 \\ & 0 \\ & \hline \end{aligned}$ |
|  | BH185 A2_2 62 | ground rep | 53.13 | 2.47 | 5.90 | 0.29 | 0.51 | $\begin{aligned} & 17.0 \\ & 8 \\ & \hline \end{aligned}$ | 0.15 | $\begin{aligned} & 21.1 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.20 | 0.01 | $\begin{aligned} & 100.9 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 83.7 \\ & 7 \\ & \hline \end{aligned}$ |
|  | BH185 A2_2 63 | ground rep | 49.29 | 3.78 | $\begin{aligned} & 10.6 \\ & 2 \\ & \hline \end{aligned}$ | 0.49 | 0.24 | $\begin{aligned} & 13.6 \\ & 9 \\ & \hline \end{aligned}$ | 0.27 | $\begin{aligned} & 19.9 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 6 \\ & \hline \end{aligned}$ | 0.22 | 0.13 | 98.72 | $\begin{aligned} & \hline 69.6 \\ & 7 \\ & \hline \end{aligned}$ |
|  | BH185 A2_2 65 | core | 52.45 | 2.74 | 6.30 | 0.40 | 0.59 | $\begin{aligned} & 16.9 \\ & 4 \end{aligned}$ | 0.13 | $\begin{aligned} & 20.8 \\ & 8 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.6 \\ & 5 \end{aligned}$ | $\begin{aligned} & 82.7 \\ & 5 \\ & \hline \end{aligned}$ |
|  | BH185 A2_2 66 | rim | 52.62 | 2.76 | 6.34 | 0.33 | 0.67 | $\begin{aligned} & 17.0 \\ & 6 \end{aligned}$ | 0.15 | $\begin{aligned} & 20.7 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.8 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & 82.7 \\ & 4 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_2 } \\ & 65-66 \end{aligned}$ | average | 52.53 | 2.75 | 6.32 | 0.37 | 0.63 | $\begin{aligned} & 17.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.14 | $\begin{aligned} & \hline 20.8 \\ & 0 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.7 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 82.7 \\ & 4 \\ & \hline \end{aligned}$ |
|  | BH185 A2_2 67 | ground rep | 51.00 | 4.59 | 8.63 | 0.72 | 0.36 | $\begin{aligned} & 16.8 \\ & 7 \\ & \hline \end{aligned}$ | 0.19 | $\begin{aligned} & 17.6 \\ & 2 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.24 | 0.01 | $\begin{aligned} & 100.2 \\ & 8 \end{aligned}$ | $\begin{aligned} & 77.6 \\ & 9 \end{aligned}$ |
|  | BH185 A2_2 69 | core | 51.24 | 4.09 | 6.49 | 0.39 | 0.88 | $\begin{aligned} & 16.2 \\ & 9 \end{aligned}$ | 0.12 | $\begin{aligned} & 20.6 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | 0.24 | 0.01 | $\begin{aligned} & 100.3 \\ & 8 \end{aligned}$ | $\begin{aligned} & 81.7 \\ & 3 \end{aligned}$ |
|  | BH185 A2_2 70 | rim | 52.44 | 2.60 | 5.93 | 0.37 | 0.72 | $17.1$ | 0.12 | $21.0$ | $\begin{aligned} & 0.0 \\ & 5 \end{aligned}$ | 0.21 | 0.01 | $\begin{aligned} & 100.6 \\ & 3 \end{aligned}$ | $\begin{aligned} & 83.7 \\ & 4 \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_2 } \\ & 69-70 \end{aligned}$ | average | 51.84 | 3.35 | 6.21 | 0.38 | 0.80 | $\begin{aligned} & 16.7 \\ & 0 \end{aligned}$ | 0.12 | $\begin{aligned} & 20.8 \\ & 4 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | 0.22 | 0.01 | $\begin{aligned} & 100.5 \\ & 1 \end{aligned}$ | $\begin{aligned} & 82.7 \\ & 3 \end{aligned}$ |
| E | BH366 B1 78 | core | 51.53 | 3.66 | 7.66 | 0.72 | 0.23 | $\begin{aligned} & 16.5 \\ & 2 \\ & \hline \end{aligned}$ | 0.21 | $\begin{aligned} & 19.2 \\ & 0 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.21 | 0.01 | $\begin{aligned} & 100.0 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 79.3 \\ & \hline 5 \end{aligned}$ |
|  | BH366 B1 79 | rim | 51.83 | 3.20 | 8.14 | 0.51 | 0.09 | $\begin{aligned} & 15.8 \\ & 3 \end{aligned}$ | 0.17 | $\begin{aligned} & 20.1 \\ & 4 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 0.23 | 0.01 | $\begin{aligned} & 100.1 \\ & 6 \end{aligned}$ | $\begin{aligned} & 77.6 \\ & 0 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH366 B1 78- } \\ & 79 \end{aligned}$ | average | 51.68 | 3.43 | 7.90 | 0.61 | 0.16 | $\begin{aligned} & 16.1 \\ & 8 \\ & \hline \end{aligned}$ | 0.19 | $\begin{aligned} & 19.7 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.22 | 0.01 | $\begin{aligned} & 100.1 \\ & 1 \end{aligned}$ | $\begin{aligned} & \hline 78.4 \\ & 8 \\ & \hline \end{aligned}$ |
|  | BH366 B1 80 | core | 51.60 | 3.63 | 8.83 | 0.51 | 0.00 | $\begin{aligned} & 16.2 \\ & 4 \end{aligned}$ | 0.22 | $19.2$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.24 | 0.00 | $\begin{aligned} & 100.5 \\ & 3 \end{aligned}$ | $\begin{aligned} & \hline 76.6 \\ & 3 \\ & \hline \end{aligned}$ |
|  | BH366 B1 81 | rim | 51.01 | 1.29 | $\begin{aligned} & 20.5 \\ & 2 \end{aligned}$ | 0.53 | 0.06 | $\begin{aligned} & 13.6 \\ & 5 \end{aligned}$ | 0.51 | $\begin{aligned} & 11.8 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.16 | 0.00 | 99.59 | $\begin{aligned} & 54.2 \\ & 3 \end{aligned}$ |
|  | $\begin{aligned} & \text { BH366 B1 80- } \\ & 81 \end{aligned}$ | average | 51.31 | 2.46 | $\begin{aligned} & 14.6 \\ & 8 \end{aligned}$ | 0.52 | 0.03 | $\begin{aligned} & 14.9 \\ & 5 \end{aligned}$ | 0.36 | $15.5$ | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.0 \\ & 6 \end{aligned}$ | $\begin{aligned} & 65.4 \\ & 3 \\ & \hline \end{aligned}$ |
|  | BH366 B1 88 | rep | 51.32 | 3.50 | 9.12 | 0.65 | 0.06 | $\begin{aligned} & 15.7 \\ & 7 \\ & \hline \end{aligned}$ | 0.21 | $\begin{aligned} & 19.6 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.22 | 0.01 | $\begin{aligned} & 100.5 \\ & 4 \end{aligned}$ | $\begin{aligned} & 75.5 \\ & 1 \\ & \hline \end{aligned}$ |
|  | BH366 B1 94 | rep | 51.93 | 3.24 | 8.76 | 0.62 | 0.12 | $\begin{aligned} & 16.3 \\ & 7 \\ & \hline \end{aligned}$ | 0.24 | $\begin{aligned} & 19.2 \\ & 0 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.7 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & 76.9 \\ & 0 \\ & \hline \end{aligned}$ |
|  | BH372 A1 106 | core | 51.37 | 2.83 | $\begin{aligned} & 10.2 \\ & 9 \end{aligned}$ | 0.79 | 0.08 | $15.9$ | 0.28 | $\begin{aligned} & 18.1 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.22 | 0.00 | 99.98 | $\begin{aligned} & 73.4 \\ & 0 \\ & \hline \end{aligned}$ |
|  | BH372 A1 106 | rimward | 51.80 | 2.31 | 9.40 | 0.61 | 0.14 | $\begin{aligned} & 16.0 \\ & 9 \end{aligned}$ | 0.23 | $\begin{aligned} & 18.9 \\ & 9 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.21 | 0.00 | 99.83 | $\begin{aligned} & 75.3 \\ & 2 \end{aligned}$ |
|  | BH372 A1 106 | rimward | 52.01 | 2.20 | 9.90 | 0.66 | 0.06 | $\begin{aligned} & 15.7 \\ & 1 \\ & \hline \end{aligned}$ | 0.27 | $\begin{aligned} & 18.9 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.0 \\ & 1 \end{aligned}$ | $\begin{aligned} & \hline 73.8 \\ & 8 \\ & \hline \end{aligned}$ |
|  | BH372 A1 106 | rimward | 51.69 | 2.12 | $\begin{aligned} & 11.6 \\ & 1 \\ & \hline \end{aligned}$ | 0.60 | 0.00 | $\begin{aligned} & 15.1 \\ & 6 \\ & \hline \end{aligned}$ | 0.30 | $\begin{aligned} & 18.1 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.22 | 0.00 | 99.81 | $\begin{aligned} & 69.9 \\ & 4 \end{aligned}$ |
|  | BH372 A1 106 | rim | 51.09 | 2.07 | $\begin{aligned} & 13.6 \\ & 4 \\ & \hline \end{aligned}$ | 0.52 | 0.00 | $\begin{aligned} & 14.5 \\ & 0 \\ & \hline \end{aligned}$ | 0.37 | $\begin{aligned} & 17.1 \\ & 9 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.24 | 0.00 | 99.61 | $\begin{aligned} & \hline 65.4 \\ & 6 \\ & \hline \end{aligned}$ |
|  | BH372 A1 106 | average | 51.59 | 2.30 | $\begin{aligned} & 10.9 \\ & 7 \end{aligned}$ | 0.63 | 0.06 | $15.4$ | 0.29 | $\begin{aligned} & 18.2 \\ & 9 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.22 | 0.00 | 99.85 | $\begin{aligned} & 71.6 \\ & 0 \\ & \hline \end{aligned}$ |
|  | BH372 A1 107 | rim | 50.69 | 1.77 | $\begin{aligned} & 16.5 \\ & 0 \\ & \hline \end{aligned}$ | 0.78 | 0.00 | $\begin{aligned} & 12.7 \\ & 5 \end{aligned}$ | 0.38 | $\begin{aligned} & 16.8 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.24 | 0.00 | 99.93 | $\begin{aligned} & 57.9 \\ & 4 \end{aligned}$ |
|  | BH372 A1 107 | rimward | 51.21 | 1.93 | $\begin{aligned} & 15.5 \\ & 1 \\ & \hline \end{aligned}$ | 0.86 | 0.01 | $13.7$ | 0.37 | $\begin{aligned} & 17.0 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 8 \\ & \hline \end{aligned}$ | 0.25 | 0.00 | $\begin{aligned} & 100.9 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 61.2 \\ & 3 \\ & \hline \end{aligned}$ |
|  | BH372 A1 107 | rimward | 50.84 | 2.27 | $\begin{aligned} & 12.9 \\ & 3 \end{aligned}$ | 0.54 | 0.03 | $\begin{aligned} & 14.5 \\ & 1 \\ & \hline \end{aligned}$ | 0.29 | $\begin{aligned} & 17.2 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.28 | 0.01 | 98.95 | $\begin{aligned} & 66.6 \\ & 6 \\ & \hline \end{aligned}$ |
|  | BH372 A1 107 | rimward | 52.48 | 1.88 | $\begin{aligned} & 11.0 \\ & 8 \\ & \hline \end{aligned}$ | 0.31 | 0.06 | $\begin{aligned} & 15.6 \\ & 7 \\ & \hline \end{aligned}$ | 0.30 | $\begin{aligned} & 18.7 \\ & 0 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.22 | 0.01 | $\begin{aligned} & 100.7 \\ & 2 \end{aligned}$ | $\begin{aligned} & 71.5 \\ & 9 \end{aligned}$ |
|  | BH372 A1 107 | core | 52.49 | 1.97 | $\begin{aligned} & 11.0 \\ & 8 \\ & \hline \end{aligned}$ | 0.61 | 0.00 | $\begin{aligned} & 15.7 \\ & 3 \end{aligned}$ | 0.27 | $\begin{aligned} & 18.6 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.9 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 71.6 \\ & 8 \\ & \hline \end{aligned}$ |
|  | BH372 A1 107 | core | 52.40 | 1.89 | $\begin{aligned} & 10.7 \\ & 1 \\ & \hline \end{aligned}$ | 0.60 | 0.00 | $15.7$ | 0.26 | $\begin{aligned} & 18.5 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.24 | 0.01 | $\begin{aligned} & 100.3 \\ & 9 \\ & \hline \end{aligned}$ | $72.4$ $1$ |
|  | BH372 A1 107 | rimward | 52.33 | 1.98 | $\begin{aligned} & 11.1 \\ & 9 \end{aligned}$ | 0.50 | 0.00 | $\begin{aligned} & 15.7 \\ & 3 \\ & \hline \end{aligned}$ | 0.27 | $\begin{aligned} & 18.5 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.8 \\ & 2 \end{aligned}$ | $\begin{aligned} & \hline 71.4 \\ & 8 \end{aligned}$ |
|  | BH372 A1 107 | rimward | 52.42 | 1.88 | $\begin{aligned} & 10.2 \\ & 6 \end{aligned}$ | 0.41 | 0.04 | $\begin{aligned} & 15.8 \\ & 6 \\ & \hline \end{aligned}$ | 0.29 | $\begin{aligned} & 18.9 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.18 | 0.00 | $\begin{aligned} & 100.3 \\ & 1 \end{aligned}$ | $\begin{aligned} & 73.3 \\ & 6 \end{aligned}$ |
|  | BH372 A1 107 | rimward | 52.55 | 1.34 | 8.88 | 0.57 | 0.00 | $14.3$ | 0.23 | $\begin{aligned} & 22.8 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.07 | 0.00 | $\begin{aligned} & 100.9 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 74.2 \\ & 2 \\ & \hline \end{aligned}$ |
|  | BH372 A1 107 | rim | 51.12 | 1.64 | $\begin{aligned} & 15.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.66 | 0.05 | $\begin{aligned} & 11.9 \\ & 7 \\ & \hline \end{aligned}$ | 0.37 | $\begin{aligned} & 19.5 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.27 | 0.01 | $\begin{aligned} & 100.7 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 58.6 \\ & 9 \\ & \hline \end{aligned}$ |
|  | BH372 A1 107 | x average | 51.85 | 1.85 | $\begin{aligned} & 12.3 \\ & 1 \end{aligned}$ | 0.58 | 0.02 | $\begin{aligned} & 14.6 \\ & 1 \end{aligned}$ | 0.30 | $\begin{aligned} & 18.6 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.4 \\ & 7 \end{aligned}$ | $\begin{aligned} & 67.9 \\ & 3 \\ & \hline \end{aligned}$ |
|  | BH372 A1 108 | rim | 50.86 | 1.60 | $\begin{aligned} & 17.7 \\ & 8 \end{aligned}$ | 0.62 | 0.00 | $\begin{aligned} & 14.1 \\ & 4 \\ & \hline \end{aligned}$ | 0.41 | $\begin{aligned} & 14.5 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 100.2 \\ & 4 \end{aligned}$ | $\begin{aligned} & 58.6 \\ & 3 \end{aligned}$ |
|  | BH372 A1 108 | rimward | 51.29 | 2.00 | $\begin{aligned} & 14.2 \\ & 8 \end{aligned}$ | 0.75 | 0.00 | $\begin{aligned} & 14.9 \\ & 6 \end{aligned}$ | 0.37 | $\begin{aligned} & 16.2 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.0 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & 65.1 \\ & 2 \end{aligned}$ |


| BH372 A1 108 | rimward | 51.69 | 2.10 | $\begin{aligned} & 12.6 \\ & 5 \end{aligned}$ | 0.40 | 0.00 | $\begin{aligned} & 15.5 \\ & 1 \\ & \hline \end{aligned}$ | 0.31 | $\begin{aligned} & 17.7 \\ & 3 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.6 \\ & 1 \end{aligned}$ | $\begin{aligned} & 68.5 \\ & 9 \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| BH372 A1 108 | rimward | 51.51 | 2.08 | $\begin{aligned} & 10.5 \\ & 3 \end{aligned}$ | 0.48 | 0.00 | $\begin{aligned} & 15.8 \\ & 1 \\ & \hline \end{aligned}$ | 0.25 | $\begin{aligned} & 18.7 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.22 | 0.00 | 99.58 | $\begin{aligned} & 72.7 \\ & 9 \\ & \hline \end{aligned}$ |
| BH372 A1 108 | core | 51.85 | 2.02 | $\begin{aligned} & 10.3 \\ & 6 \end{aligned}$ | 0.65 | 0.04 | $\begin{aligned} & 15.7 \\ & 2 \\ & \hline \end{aligned}$ | 0.26 | $\begin{aligned} & 18.7 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.21 | 0.00 | 99.84 | $\begin{aligned} & \hline 72.9 \\ & 9 \\ & \hline \end{aligned}$ |
| BH372 A1 108 | core | 51.47 | 2.05 | $\begin{aligned} & 10.4 \\ & 7 \\ & \hline \end{aligned}$ | 0.43 | 0.00 | $\begin{aligned} & 15.6 \\ & 7 \\ & \hline \end{aligned}$ | 0.27 | $\begin{aligned} & 18.6 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.19 | 0.00 | 99.27 | $\begin{aligned} & 72.7 \\ & 4 \\ & \hline \end{aligned}$ |
| BH372 A1 108 | rimward | 52.00 | 2.06 | $\begin{aligned} & 10.3 \\ & 4 \\ & \hline \end{aligned}$ | 0.55 | 0.00 | $\begin{aligned} & 15.7 \\ & 6 \\ & \hline \end{aligned}$ | 0.27 | $18.7$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.23 | 0.00 | 99.97 | $\begin{aligned} & 73.1 \\ & 0 \\ & \hline \end{aligned}$ |
| BH372 A1 108 | rimward | 51.20 | 2.18 | $\begin{aligned} & 11.8 \\ & 2 \end{aligned}$ | 0.63 | 0.00 | $\begin{aligned} & 15.3 \\ & 5 \\ & \hline \end{aligned}$ | 0.25 | $\begin{aligned} & 18.2 \\ & 5 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.23 | 0.00 | 99.91 | $\begin{aligned} & 69.8 \\ & 4 \\ & \hline \end{aligned}$ |
| BH372 A1 108 | rimward | 51.58 | 2.07 | $\begin{aligned} & 12.5 \\ & 1 \\ & \hline \end{aligned}$ | 0.69 | 0.08 | $\begin{aligned} & 15.3 \\ & 1 \end{aligned}$ | 0.29 | $\begin{aligned} & 17.9 \\ & 2 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.6 \\ & 5 \end{aligned}$ | $\begin{aligned} & \hline 68.5 \\ & 7 \\ & \hline \end{aligned}$ |
| BH372 A1 108 | rim | 51.22 | 2.29 | $11.7$ | 0.44 | 0.01 | $\begin{aligned} & 13.7 \\ & 6 \\ & \hline \end{aligned}$ | 0.33 | $\begin{aligned} & 20.0 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.21 | 0.01 | $\begin{aligned} & 100.0 \\ & 6 \end{aligned}$ | $\begin{aligned} & 67.6 \\ & 8 \end{aligned}$ |
| BH372 A1 108 | x average | 51.47 | 2.05 | $\begin{aligned} & 12.2 \\ & 4 \end{aligned}$ | 0.56 | 0.01 | $\begin{aligned} & 15.2 \\ & 0 \end{aligned}$ | 0.30 | $\begin{aligned} & 17.9 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.0 \\ & 2 \end{aligned}$ | $\begin{aligned} & 69.0 \\ & 0 \\ & \hline \end{aligned}$ |
| BH372 A1 109 | rim | 51.91 | 2.27 | $\begin{aligned} & 13.3 \\ & 1 \\ & \hline \end{aligned}$ | 0.57 | 0.00 | $13.6$ | 0.31 | $\begin{aligned} & 16.9 \\ & 5 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 6 \end{aligned}$ | 0.44 | 0.11 | 99.59 | $\begin{aligned} & 64.6 \\ & 3 \\ & \hline \end{aligned}$ |
| BH372 A1 109 | rimward | 52.20 | 1.92 | $\begin{aligned} & 12.0 \\ & 3 \end{aligned}$ | 0.60 | 0.05 | $\begin{aligned} & 15.0 \\ & 0 \end{aligned}$ | 0.28 | $\begin{aligned} & 18.3 \\ & 3 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.6 \\ & 0 \end{aligned}$ | $\begin{aligned} & 68.9 \\ & 6 \\ & \hline \end{aligned}$ |
| BH372 A1 109 | rimward | 52.07 | 2.04 | $\begin{aligned} & 10.8 \\ & 1 \\ & \hline \end{aligned}$ | 0.48 | 0.03 | $\begin{aligned} & 15.8 \\ & 4 \end{aligned}$ | 0.31 | $\begin{aligned} & 18.3 \\ & 1 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 100.1 \\ & 2 \end{aligned}$ | $\begin{aligned} & 72.3 \\ & 1 \end{aligned}$ |
| BH372 A1 109 | rimward | 52.28 | 2.14 | $\begin{aligned} & 11.0 \\ & 3 \end{aligned}$ | 0.55 | 0.00 | $\begin{aligned} & 15.4 \\ & 6 \\ & \hline \end{aligned}$ | 0.28 | $\begin{aligned} & 18.7 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.7 \\ & 1 \end{aligned}$ | $\begin{aligned} & 71.4 \\ & 1 \\ & \hline \end{aligned}$ |
| BH372 A1 109 | core | 51.86 | 2.22 | $\begin{aligned} & 11.5 \\ & 1 \\ & \hline \end{aligned}$ | 0.63 | 0.02 | $\begin{aligned} & 15.5 \\ & 7 \\ & \hline \end{aligned}$ | 0.27 | $\begin{aligned} & 18.3 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 100.6 \\ & 5 \end{aligned}$ | $\begin{aligned} & 70.6 \\ & 8 \end{aligned}$ |
| BH372 A1 109 | core | 51.51 | 1.98 | $\begin{aligned} & 11.8 \\ & 9 \end{aligned}$ | 0.66 | 0.00 | $\begin{aligned} & 14.6 \\ & 9 \\ & \hline \end{aligned}$ | 0.29 | $\begin{aligned} & 18.7 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 100.0 \\ & 4 \end{aligned}$ | $\begin{aligned} & \hline 68.7 \\ & 8 \\ & \hline \end{aligned}$ |
| BH372 A1 109 | rimward | 51.49 | 2.06 | $\begin{aligned} & 14.1 \\ & 3 \\ & \hline \end{aligned}$ | 0.66 | 0.00 | $\begin{aligned} & 14.1 \\ & 3 \\ & \hline \end{aligned}$ | 0.30 | $\begin{aligned} & 17.6 \\ & 0 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.24 | 0.01 | $\begin{aligned} & 100.6 \\ & 1 \end{aligned}$ | $\begin{aligned} & 64.0 \\ & 6 \\ & \hline \end{aligned}$ |
| BH372 A1 109 | rimward | 50.82 | 1.78 | $\begin{aligned} & 15.6 \\ & 0 \\ & \hline \end{aligned}$ | 0.77 | 0.02 | $\begin{aligned} & 13.1 \\ & 4 \\ & \hline \end{aligned}$ | 0.38 | $17.4$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.24 | 0.00 | $\begin{aligned} & 100.1 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 60.0 \\ & 2 \end{aligned}$ |
| BH372 A1 109 | rimward | 50.54 | 1.59 | $\begin{aligned} & 16.7 \\ & 6 \end{aligned}$ | 0.81 | 0.00 | $12.5$ | 0.38 | $17.0$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.24 | 0.00 | 99.83 | $57.0$ |
| BH372 A1 109 | rim | 50.53 | 1.33 | $\begin{aligned} & 17.6 \\ & 7 \\ & \hline \end{aligned}$ | 0.75 | 0.01 | $\begin{aligned} & 11.5 \\ & 4 \\ & \hline \end{aligned}$ | 0.46 | $\begin{aligned} & 17.0 \\ & 9 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.20 | 0.00 | 99.58 | $\begin{aligned} & 53.7 \\ & 8 \\ & \hline \end{aligned}$ |
| BH372 A1 109 | x average | 51.52 | 1.93 | $\begin{aligned} & 13.4 \\ & 7 \\ & \hline \end{aligned}$ | 0.65 | 0.01 | $\begin{aligned} & 14.1 \\ & 5 \\ & \hline \end{aligned}$ | 0.33 | $\begin{aligned} & 17.8 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.25 | 0.01 | $\begin{aligned} & 100.1 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 65.1 \\ & 7 \\ & \hline \end{aligned}$ |
| BH372 A1 114 | rim | 51.18 | 1.40 | $\begin{aligned} & 16.7 \\ & 1 \end{aligned}$ | 0.51 | 0.00 | $\begin{aligned} & 11.7 \\ & 9 \\ & \hline \end{aligned}$ | 0.40 | $\begin{aligned} & 17.9 \\ & 6 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 0.19 | 0.01 | $\begin{aligned} & 100.1 \\ & 4 \end{aligned}$ | $\begin{aligned} & 55.7 \\ & 1 \\ & \hline \end{aligned}$ |
| BH372 A1 114 | core | 50.91 | 1.51 | $\begin{aligned} & 16.9 \\ & 0 \end{aligned}$ | 0.60 | 0.00 | $\begin{aligned} & 12.5 \\ & 3 \\ & \hline \end{aligned}$ | 0.41 | $\begin{aligned} & 16.8 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.23 | 0.00 | 99.96 | $\begin{aligned} & 56.9 \\ & 1 \end{aligned}$ |
| BH372 A1 114 | core | 50.80 | 1.53 | $\begin{aligned} & 16.8 \\ & 7 \\ & \hline \end{aligned}$ | 0.77 | 0.06 | $\begin{aligned} & 12.7 \\ & 3 \\ & \hline \end{aligned}$ | 0.40 | $\begin{aligned} & 16.7 \\ & 3 \\ & \hline \end{aligned}$ | $0.0$ | 0.23 | 0.00 | $\begin{aligned} & 100.1 \\ & 1 \end{aligned}$ | $\begin{aligned} & 57.3 \\ & 5 \\ & \hline \end{aligned}$ |
| BH372 A1 114 | rim | 49.69 | 0.82 | $\begin{aligned} & 24.1 \\ & 3 \\ & \hline \end{aligned}$ | 0.44 | 0.03 | 6.98 | 0.64 | $\begin{aligned} & 16.9 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.17 | 0.00 | 99.84 | $\begin{aligned} & 34.0 \\ & 3 \\ & \hline \end{aligned}$ |
| BH372 A1 114 | x average | 50.64 | 1.31 | $\begin{aligned} & 18.6 \\ & 5 \end{aligned}$ | 0.58 | 0.02 | $\begin{aligned} & 11.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.46 | $\begin{aligned} & 17.1 \\ & 3 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.0 \\ & 1 \end{aligned}$ | $\begin{aligned} & 51.0 \\ & 0 \\ & \hline \end{aligned}$ |
| BH372 A1 115 | rim | 50.45 | 3.07 | $\begin{aligned} & 11.2 \\ & 4 \\ & \hline \end{aligned}$ | 0.48 | 0.06 | $\begin{aligned} & 14.6 \\ & 9 \\ & \hline \end{aligned}$ | 0.23 | $\begin{aligned} & 18.1 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.25 | 0.01 | 98.70 | $\begin{aligned} & 69.9 \\ & 5 \\ & \hline \end{aligned}$ |
| BH372 A1 115 | rimward | 51.35 | 2.77 | 9.30 | 0.60 | 0.04 | $\begin{aligned} & 15.7 \\ & 6 \\ & \hline \end{aligned}$ | 0.23 | $\begin{aligned} & 19.0 \\ & 6 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.21 | 0.00 | 99.34 | $\begin{aligned} & \hline 75.1 \\ & 3 \\ & \hline \end{aligned}$ |
| BH372 A1 115 | rimward | 52.67 | 1.67 | 9.61 | 0.41 | 0.04 | $16.6$ | 0.25 | $\begin{aligned} & 18.3 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.18 | 0.00 | 99.86 | $\begin{aligned} & 75.5 \\ & 5 \\ & \hline \end{aligned}$ |
| BH372 A1 115 | rimward | 51.67 | 2.59 | 9.62 | 0.61 | 0.05 | $\begin{aligned} & 15.8 \\ & 6 \\ & \hline \end{aligned}$ | 0.22 | $\begin{aligned} & 19.2 \\ & 8 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.1 \\ & 3 \end{aligned}$ | $\begin{aligned} & 74.6 \\ & 0 \\ & \hline \end{aligned}$ |
| BH372 A1 115 | core | 51.50 | 2.36 | 9.84 | 0.53 | 0.06 | $\begin{aligned} & 16.0 \\ & 6 \end{aligned}$ | 0.23 | $\begin{aligned} & 18.2 \\ & 1 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.22 | 0.00 | 99.03 | $\begin{aligned} & 74.4 \\ & 2 \\ & \hline \end{aligned}$ |
| BH372 A1 115 | core | 49.91 | 3.27 | $\begin{aligned} & 11.5 \\ & 5 \\ & \hline \end{aligned}$ | 0.48 | 0.06 | $\begin{aligned} & 16.1 \\ & 0 \\ & \hline \end{aligned}$ | 0.23 | $\begin{aligned} & 16.5 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.19 | 0.03 | 98.34 | $\begin{aligned} & 71.3 \\ & 0 \\ & \hline \end{aligned}$ |
| BH372 A1 115 | rimward | 52.26 | 1.92 | 9.77 | 0.54 | 0.04 | $\begin{aligned} & 15.8 \\ & 4 \\ & \hline \end{aligned}$ | 0.27 | $\begin{aligned} & 19.0 \\ & 4 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.19 | 0.00 | 99.88 | $\begin{aligned} & 74.2 \\ & 8 \\ & \hline \end{aligned}$ |
| BH372 A1 115 | rimward | 51.04 | 2.46 | $\begin{aligned} & 10.1 \\ & 2 \end{aligned}$ | 0.44 | 0.00 | $\begin{aligned} & 15.7 \\ & 7 \\ & \hline \end{aligned}$ | 0.24 | $\begin{aligned} & 18.1 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.24 | 0.01 | 98.46 | $\begin{aligned} & 73.5 \\ & 3 \\ & \hline \end{aligned}$ |
| BH372 A1 115 | rimward | 52.35 | 2.13 | $\begin{aligned} & 10.2 \\ & 5 \end{aligned}$ | 0.61 | 0.05 | $\begin{aligned} & 15.9 \\ & 2 \end{aligned}$ | 0.28 | $\begin{aligned} & 18.8 \\ & 3 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.19 | 0.00 | $\begin{aligned} & 100.6 \\ & 2 \end{aligned}$ | $\begin{aligned} & \hline 73.4 \\ & 7 \\ & \hline \end{aligned}$ |
| BH372 A1 115 | rim | 52.23 | 2.10 | $\begin{aligned} & 10.9 \\ & 3 \end{aligned}$ | 0.53 | 0.04 | $\begin{aligned} & 15.8 \\ & 0 \end{aligned}$ | 0.28 | $\begin{aligned} & 18.4 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 5 \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.6 \\ & 0 \end{aligned}$ | $\begin{aligned} & 72.0 \\ & 3 \\ & \hline \end{aligned}$ |
| BH372 A1 115 | x average | 51.54 | 2.43 | $\begin{aligned} & 10.2 \\ & 2 \end{aligned}$ | 0.52 | 0.04 | $\begin{aligned} & 15.8 \\ & 5 \\ & \hline \end{aligned}$ | 0.25 | $\begin{aligned} & 18.4 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.21 | 0.01 | 99.50 | $\begin{aligned} & \hline 73.4 \\ & 3 \\ & \hline \end{aligned}$ |
| BH372 A1 116 | rim | 51.05 | 1.75 | $\begin{aligned} & 15.6 \\ & 1 \end{aligned}$ | 0.89 | 0.00 | $12.0$ | 0.33 | $\begin{aligned} & 15.3 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.31 | 0.01 | 97.30 | $57.8$ |
| BH372 A1 116 | rimward | 52.10 | 2.03 | $\begin{aligned} & 11.0 \\ & 9 \\ & \hline \end{aligned}$ | 0.44 | 0.00 | $\begin{aligned} & 15.9 \\ & 5 \\ & \hline \end{aligned}$ | 0.29 | $\begin{aligned} & 18.3 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.5 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 71.9 \\ & 3 \end{aligned}$ |


|  | BH372 A1 116 | rimward | 52.59 | 1.88 | $\begin{aligned} & 10.2 \\ & 6 \\ & \hline \end{aligned}$ | 0.58 | 0.05 | $\begin{aligned} & 16.4 \\ & 3 \end{aligned}$ | 0.22 | $\begin{aligned} & 18.4 \\ & 3 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.19 | 0.00 | $\begin{aligned} & 100.6 \\ & 5 \end{aligned}$ | $\begin{aligned} & 74.0 \\ & 5 \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | BH372 A1 116 | rimward | 52.41 | 1.89 | $\begin{aligned} & 10.2 \\ & 4 \\ & \hline \end{aligned}$ | 0.45 | 0.02 | $\begin{aligned} & 16.3 \\ & 0 \end{aligned}$ | 0.25 | $\begin{aligned} & 18.8 \\ & 0 \\ & \hline \end{aligned}$ | $0.0$ | 0.23 | 0.01 | $\begin{aligned} & 100.7 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 74.0 \\ & 4 \end{aligned}$ |
|  | BH372 A1 116 | core | 52.50 | 1.91 | $\begin{aligned} & 10.1 \\ & 8 \\ & \hline \end{aligned}$ | 0.50 | 0.03 | $\begin{aligned} & 16.3 \\ & 5 \\ & \hline \end{aligned}$ | 0.26 | $\begin{aligned} & 18.8 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.8 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 74.1 \\ & 1 \\ & \hline \end{aligned}$ |
|  | BH372 A1 116 | core | 52.24 | 2.10 | $\begin{aligned} & 10.6 \\ & 7 \end{aligned}$ | 0.49 | 0.00 | $16.3$ | 0.25 | $\begin{aligned} & 18.2 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.5 \\ & 4 \end{aligned}$ | $\begin{aligned} & \hline 73.1 \\ & 5 \\ & \hline \end{aligned}$ |
|  | BH372 A1 116 | rimward | 52.25 | 2.04 | $\begin{aligned} & 10.8 \\ & 7 \\ & \hline \end{aligned}$ | 0.57 | 0.00 | $\begin{aligned} & 16.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.28 | $\begin{aligned} & 18.2 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.4 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 72.4 \\ & 7 \\ & \hline \end{aligned}$ |
|  | BH372 A1 116 | rimward | 51.53 | 1.56 | $\begin{aligned} & 13.8 \\ & 9 \\ & \hline \end{aligned}$ | 0.77 | 0.05 | $\begin{aligned} & 13.2 \\ & 5 \\ & \hline \end{aligned}$ | 0.36 | $17.5$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.27 | 0.00 | 99.24 | $\begin{aligned} & 62.9 \\ & 5 \end{aligned}$ |
|  | BH372 A1 116 | rimward | 51.46 | 1.15 | $16.2$ | 0.47 | 0.00 | $\begin{aligned} & 11.1 \\ & 1 \\ & \hline \end{aligned}$ | 0.39 | $\begin{aligned} & 18.3 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.19 | 0.00 | 99.44 | $\begin{aligned} & 54.8 \\ & 7 \\ & \hline \end{aligned}$ |
|  | BH372 A1 116 | rim | 43.50 | 3.90 | $\begin{aligned} & 27.1 \\ & 6 \\ & \hline \end{aligned}$ | 0.29 | 0.02 | 7.13 | 0.48 | $\begin{aligned} & 11.6 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.25 | 0.03 | 94.36 | $\begin{aligned} & 31.8 \\ & 7 \\ & \hline \end{aligned}$ |
|  | BH372 A1 116 | x average | 51.16 | 2.02 | $\begin{aligned} & 13.6 \\ & 3 \\ & \hline \end{aligned}$ | 0.54 | 0.02 | $\begin{aligned} & 14.1 \\ & 0 \\ & \hline \end{aligned}$ | 0.31 | $\begin{aligned} & 17.3 \\ & 8 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 0.23 | 0.01 | 99.40 | $\begin{aligned} & 64.7 \\ & 3 \\ & \hline \end{aligned}$ |
|  | BH372 A1 121 | core | 51.03 | 1.28 | $\begin{aligned} & 19.7 \\ & 2 \\ & \hline \end{aligned}$ | 0.58 | 0.00 | $\begin{aligned} & 13.7 \\ & 9 \\ & \hline \end{aligned}$ | 0.45 | $\begin{aligned} & 13.5 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.17 | 0.00 | $\begin{aligned} & 100.5 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 55.4 \\ & 8 \\ & \hline \end{aligned}$ |
|  | BH372 A1 122 | rimward | 50.55 | 1.28 | $\begin{aligned} & 21.0 \\ & 5 \end{aligned}$ | 0.44 | 0.00 | $\begin{aligned} & 11.5 \\ & 6 \\ & \hline \end{aligned}$ | 0.54 | $\begin{aligned} & 14.1 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.17 | 0.00 | 99.76 | $\begin{aligned} & 49.4 \\ & 6 \\ & \hline \end{aligned}$ |
|  | BH372 A1 123 | rim (very altered) | 45.74 | 4.21 | $\begin{aligned} & 31.7 \\ & 2 \end{aligned}$ | 0.07 | 0.01 | 3.73 | 0.64 | 9.49 | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.87 | 0.72 | 97.20 | $\begin{aligned} & 17.3 \\ & 1 \end{aligned}$ |
|  | $\begin{aligned} & \text { BH372 A1 121- } \\ & 123 \\ & \hline \end{aligned}$ | average | 49.11 | 2.25 | $\begin{aligned} & 24.1 \\ & 7 \\ & \hline \end{aligned}$ | 0.36 | 0.00 | 9.69 | 0.54 | $\begin{aligned} & 12.4 \\ & 0 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.40 | 0.24 | 99.17 | $\begin{aligned} & 40.7 \\ & 5 \end{aligned}$ |
|  | BH372 A1 124 | rim | 40.81 | 9.78 | $\begin{aligned} & 27.3 \\ & 6 \\ & \hline \end{aligned}$ | 0.24 | 0.00 | 8.44 | 0.10 | 0.15 | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.13 | 6.46 | 93.48 | $\begin{aligned} & 35.4 \\ & 8 \end{aligned}$ |
|  | BH372 A1 124 | core | 51.01 | 2.38 | $\begin{aligned} & 17.6 \\ & 5 \\ & \hline \end{aligned}$ | 0.45 | 0.02 | $\begin{aligned} & 10.9 \\ & 0 \\ & \hline \end{aligned}$ | 0.30 | $\begin{aligned} & 15.1 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.26 | 0.06 | 98.21 | $\begin{aligned} & 52.3 \\ & 8 \\ & \hline \end{aligned}$ |
|  | BH372 A1 124 | core | 52.44 | 2.04 | $\begin{aligned} & 10.9 \\ & 7 \end{aligned}$ | 0.56 | 0.00 | $\begin{aligned} & 16.0 \\ & 0 \end{aligned}$ | 0.25 | $\begin{aligned} & 18.2 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.7 \\ & 5 \end{aligned}$ | $\begin{aligned} & \hline 72.2 \\ & 2 \end{aligned}$ |
|  | BH372 A1 124 | core | 52.37 | 1.91 | $10.6$ | 0.39 | 0.02 | $\begin{aligned} & 16.1 \\ & 0 \\ & \hline \end{aligned}$ | 0.24 | $18.3$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.2 \\ & 2 \end{aligned}$ | $\begin{aligned} & 72.9 \\ & 3 \end{aligned}$ |
|  | BH372 A1 124 | rim | 52.01 | 2.09 | $\begin{aligned} & 10.9 \\ & 3 \end{aligned}$ | 0.51 | 0.00 | $\begin{aligned} & 15.6 \\ & 1 \\ & \hline \end{aligned}$ | 0.28 | $\begin{aligned} & 18.5 \\ & 3 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 5 \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 100.2 \\ & 3 \end{aligned}$ | $\begin{aligned} & 71.7 \\ & 9 \end{aligned}$ |
|  | BH372 A1 124 | x average | 49.73 | 3.64 | $\begin{aligned} & 15.5 \\ & 1 \\ & \hline \end{aligned}$ | 0.43 | 0.01 | $\begin{aligned} & 13.4 \\ & 1 \\ & \hline \end{aligned}$ | 0.23 | $\begin{aligned} & 14.0 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.21 | 1.30 | 98.58 | $\begin{aligned} & \hline 60.9 \\ & 6 \\ & \hline \end{aligned}$ |
| A | BH154 A2 65 | core | 53.64 | 1.56 | 8.08 | 0.36 | 0.09 | $\begin{aligned} & 18.1 \\ & 8 \end{aligned}$ | 0.27 | $\begin{aligned} & 17.7 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.19 | 0.00 | $\begin{aligned} & 100.1 \\ & 8 \end{aligned}$ | $\begin{aligned} & 80.0 \\ & 4 \end{aligned}$ |
|  | BH154 A2 66 | rimwards | 52.54 | 2.57 | 7.75 | 0.42 | 0.18 | $\begin{aligned} & 16.5 \\ & 3 \end{aligned}$ | 0.17 | $\begin{aligned} & 20.2 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.6 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & 79.1 \\ & 7 \\ & \hline \end{aligned}$ |
|  | BH154 A2 67 | rim | 52.85 | 2.27 | 7.22 | 0.38 | 0.21 | $\begin{aligned} & 16.5 \\ & 5 \\ & \hline \end{aligned}$ | 0.16 | $20.7$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.6 \\ & 2 \end{aligned}$ | $80.3$ |
|  | $\begin{aligned} & \text { BH154 A2 65- } \\ & 67 \\ & \hline \end{aligned}$ | average | 53.01 | 2.13 | 7.69 | 0.39 | 0.16 | $\begin{aligned} & 17.0 \\ & 9 \\ & \hline \end{aligned}$ | 0.20 | $\begin{aligned} & 19.5 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.4 \\ & 8 \end{aligned}$ | $\begin{aligned} & 79.8 \\ & 4 \end{aligned}$ |
|  | BH154 A2 70 | ground rep | 50.67 | 2.81 | $\begin{aligned} & 15.1 \\ & 3 \\ & \hline \end{aligned}$ | 0.82 | 0.02 | $\begin{aligned} & 13.4 \\ & 9 \\ & \hline \end{aligned}$ | 0.34 | $\begin{aligned} & 16.9 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.29 | 0.01 | $\begin{aligned} & 100.5 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 61.3 \\ & 8 \\ & \hline \end{aligned}$ |
|  | BH154 A2 71 | core | 52.53 | 2.35 | 7.18 | 0.51 | 0.29 | $\begin{aligned} & 16.8 \\ & 8 \end{aligned}$ | 0.19 | $20.3$ | $\begin{aligned} & 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.5 \\ & 0 \end{aligned}$ | $\begin{aligned} & 80.7 \\ & 3 \\ & \hline \end{aligned}$ |
|  | BH154 A2 72 | rim | 51.96 | 2.65 | $\begin{aligned} & 12.0 \\ & 4 \end{aligned}$ | 0.59 | 0.00 | $\begin{aligned} & 16.5 \\ & 5 \\ & \hline \end{aligned}$ | 0.29 | $\begin{aligned} & 16.1 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.24 | 0.01 | $\begin{aligned} & 100.4 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & 71.0 \\ & 2 \end{aligned}$ |
|  | $\begin{aligned} & \text { BH154 A2 71- } \\ & 72 \\ & \hline \end{aligned}$ | average | 52.25 | 2.50 | 9.61 | 0.55 | 0.15 | $\begin{aligned} & 16.7 \\ & 2 \\ & \hline \end{aligned}$ | 0.24 | $\begin{aligned} & 18.2 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.23 | 0.01 | $\begin{aligned} & 100.4 \\ & 9 \end{aligned}$ | $\begin{aligned} & \hline 75.8 \\ & 8 \\ & \hline \end{aligned}$ |
|  | BH154 A2 80 | ground rep | 51.45 | 2.07 | $\begin{aligned} & 16.4 \\ & 4 \\ & \hline \end{aligned}$ | 0.47 | 0.00 | $\begin{aligned} & 14.1 \\ & 2 \\ & \hline \end{aligned}$ | 0.40 | $\begin{aligned} & 15.4 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.26 | 0.01 | $\begin{aligned} & 100.6 \\ & 3 \end{aligned}$ | $\begin{aligned} & 60.4 \\ & 8 \end{aligned}$ |
|  | BH167 A2 57 | core | 51.63 | 3.27 | 9.16 | 0.76 | 0.10 | $\begin{aligned} & 16.3 \\ & 9 \end{aligned}$ | 0.26 | $\begin{aligned} & 18.5 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.23 | 0.00 | $\begin{aligned} & 100.3 \\ & 6 \end{aligned}$ | $\begin{aligned} & 76.1 \\ & 2 \end{aligned}$ |
|  | BH167 A2 58 | rimwards | 51.04 | 2.66 | $\begin{aligned} & 13.2 \\ & 2 \end{aligned}$ | 0.63 | 0.00 | $\begin{aligned} & 15.9 \\ & 9 \end{aligned}$ | 0.31 | $\begin{aligned} & 15.9 \\ & 0 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | 0.22 | 0.02 | $\begin{aligned} & 100.0 \\ & 2 \end{aligned}$ | $\begin{aligned} & 68.3 \\ & 0 \end{aligned}$ |
|  | BH167 A2 59 | rim | 42.80 | 7.11 | $27.5$ | 0.40 | 0.04 | 7.40 | 0.34 | 7.95 | $0.0$ | 0.24 | 2.74 | 96.56 | $\begin{aligned} & 32.3 \\ & 6 \end{aligned}$ |
|  | $\begin{aligned} & \text { BH167 A2 57- } \\ & 59 \end{aligned}$ | average | 48.49 | 4.34 | $\begin{aligned} & 16.6 \\ & 4 \\ & \hline \end{aligned}$ | 0.60 | 0.05 | $\begin{aligned} & 13.2 \\ & 6 \\ & \hline \end{aligned}$ | 0.30 | $\begin{aligned} & 14.1 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.23 | 0.92 | 98.98 | $\begin{aligned} & 58.9 \\ & 3 \end{aligned}$ |
|  | BH167 A2 61 | core | 53.69 | 1.79 | 7.82 | 0.31 | 0.23 | $\begin{aligned} & 18.5 \\ & 0 \\ & \hline \end{aligned}$ | 0.23 | $\begin{aligned} & 18.0 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.17 | 0.00 | $\begin{aligned} & 100.8 \\ & 3 \end{aligned}$ | $\begin{aligned} & 80.8 \\ & 3 \end{aligned}$ |
|  | BH167 A2 62 | rim | 52.62 | 1.54 | $\begin{aligned} & 15.2 \\ & 7 \\ & \hline \end{aligned}$ | 0.35 | 0.02 | $\begin{aligned} & 18.1 \\ & 7 \\ & \hline \end{aligned}$ | 0.41 | $\begin{aligned} & 11.5 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.13 | 0.02 | $\begin{aligned} & 100.0 \\ & 8 \\ & \hline \end{aligned}$ | $67.9$ |
|  | $\begin{aligned} & \text { BH167 A2 61- } \\ & 62 \end{aligned}$ | average | 53.16 | 1.66 | $\begin{aligned} & 11.5 \\ & 5 \\ & \hline \end{aligned}$ | 0.33 | 0.13 | $\begin{aligned} & 18.3 \\ & 3 \end{aligned}$ | 0.32 | $\begin{aligned} & 14.7 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.15 | 0.01 | $\begin{aligned} & 100.4 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & 74.3 \\ & 9 \\ & \hline \end{aligned}$ |
|  | BH167 A4 8 | core | 51.62 | 1.77 | 8.93 | 0.50 | 0.00 | $\begin{aligned} & 16.6 \\ & 4 \end{aligned}$ | 0.22 | $\begin{aligned} & 18.6 \\ & 2 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 0.21 | 0.00 | 98.52 | $\begin{aligned} & 76.8 \\ & 6 \end{aligned}$ |
|  | BH167 A4 9 | rimwards | 52.56 | 1.60 | 8.44 | 0.43 | 0.00 | $\begin{aligned} & 17.3 \\ & 3 \\ & \hline \end{aligned}$ | 0.25 | $\begin{aligned} & 18.7 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 0.19 | 0.00 | 99.60 | $\begin{aligned} & 78.5 \\ & 4 \\ & \hline \end{aligned}$ |
|  | BH167 A4 10 | rim | 51.69 | 2.28 | 9.94 | 0.52 | 0.00 | $\begin{aligned} & 16.6 \\ & 1 \end{aligned}$ | 0.25 | $\begin{aligned} & 18.2 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.23 | 0.00 | 99.81 | $\begin{aligned} & 74.8 \\ & 5 \\ & \hline \end{aligned}$ |


| BH167 A4 8-10 | average | 51.96 | 1.88 | 9.10 | 0.48 | 0.00 | $\begin{aligned} & 16.8 \\ & 6 \\ & \hline \end{aligned}$ | 0.24 | $\begin{aligned} & 18.5 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.21 | 0.00 | 99.31 | $\begin{aligned} & 76.7 \\ & 5 \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| BH167 A4 11 | $\begin{aligned} & \text { rim (same } \\ & \text { x, next to } \\ & \text { ol) } \\ & \hline \end{aligned}$ | 51.42 | 1.67 | $\begin{aligned} & 11.3 \\ & 9 \end{aligned}$ | 0.64 | 0.06 | $\begin{aligned} & 15.8 \\ & 7 \end{aligned}$ | 0.29 | $\begin{aligned} & 17.8 \\ & 7 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.28 | 0.00 | 99.49 | $\begin{aligned} & 71.2 \\ & 8 \end{aligned}$ |
| BH167 A4 25 | rim | 49.86 | 2.65 | $\begin{aligned} & 12.5 \\ & 2 \\ & \hline \end{aligned}$ | 0.90 | 0.00 | $\begin{aligned} & 13.8 \\ & 7 \\ & \hline \end{aligned}$ | 0.30 | $\begin{aligned} & 18.7 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.27 | 0.01 | 99.10 | $\begin{aligned} & \hline 66.3 \\ & 8 \\ & \hline \end{aligned}$ |
| BH167 A4 26 | rimward | 50.25 | 2.75 | $\begin{aligned} & 12.9 \\ & 6 \end{aligned}$ | 0.72 | 0.09 | $15.0$ | 0.32 | $\begin{aligned} & 17.1 \\ & 2 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.25 | 0.01 | 99.48 | $\begin{aligned} & \hline 67.3 \\ & 5 \\ & \hline \end{aligned}$ |
| BH167 A4 27 | rimward | 50.57 | 2.66 | $\begin{aligned} & 12.9 \\ & 9 \end{aligned}$ | 1.02 | 0.00 | $\begin{aligned} & 14.9 \\ & 1 \\ & \hline \end{aligned}$ | 0.30 | $\begin{aligned} & 17.2 \\ & 3 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.26 | 0.01 | 99.95 | $\begin{aligned} & 67.1 \\ & 6 \\ & \hline \end{aligned}$ |
| BH167 A4 28 | rimward | 50.07 | 2.76 | $\begin{aligned} & 13.2 \\ & 6 \\ & \hline \end{aligned}$ | 0.92 | 0.00 | $\begin{aligned} & 14.8 \\ & 8 \\ & \hline \end{aligned}$ | 0.35 | $\begin{aligned} & 17.1 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.27 | 0.01 | 99.66 | $\begin{aligned} & 66.6 \\ & 6 \\ & \hline \end{aligned}$ |
| BH167 A4 29 | core | 50.06 | 2.67 | $\begin{aligned} & 12.8 \\ & 4 \\ & \hline \end{aligned}$ | 0.95 | 0.04 | $\begin{aligned} & 14.3 \\ & 9 \\ & \hline \end{aligned}$ | 0.30 | $\begin{aligned} & 17.6 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.27 | 0.00 | 99.18 | $\begin{aligned} & 66.6 \\ & 4 \\ & \hline \end{aligned}$ |
| BH167 A4 30 | rimward | 49.96 | 2.57 | $\begin{aligned} & 14.6 \\ & 8 \end{aligned}$ | 0.84 | 0.00 | $\begin{aligned} & 14.7 \\ & 1 \end{aligned}$ | 0.33 | $\begin{aligned} & 16.8 \\ & 0 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.29 | 0.00 | $\begin{aligned} & 100.2 \\ & 0 \end{aligned}$ | $\begin{aligned} & 64.1 \\ & 0 \\ & \hline \end{aligned}$ |
| BH167 A4 31 | rimward | 49.81 | 2.41 | $\begin{aligned} & 15.5 \\ & 7 \\ & \hline \end{aligned}$ | 0.97 | 0.00 | $14.0$ | 0.38 | $\begin{aligned} & 16.3 \\ & 7 \\ & \hline \end{aligned}$ | $0.0$ | 0.29 | 0.00 | 99.83 | $\begin{aligned} & \hline 61.6 \\ & 2 \\ & \hline \end{aligned}$ |
| BH167 A4 32 | rimward | 49.52 | 2.61 | $\begin{aligned} & 15.6 \\ & 6 \\ & \hline \end{aligned}$ | 0.87 | 0.00 | $\begin{aligned} & 13.2 \\ & 3 \\ & \hline \end{aligned}$ | 0.35 | $\begin{aligned} & 16.9 \\ & 0 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.27 | 0.00 | 99.42 | $\begin{aligned} & 60.0 \\ & 8 \end{aligned}$ |
| BH167 A4 33 | rim | 50.01 | 2.30 | $\begin{aligned} & 16.9 \\ & 6 \end{aligned}$ | 1.07 | 0.00 | $\begin{aligned} & 13.5 \\ & 7 \\ & \hline \end{aligned}$ | 0.41 | $\begin{aligned} & 15.2 \\ & 9 \end{aligned}$ | $0.0$ | 0.23 | 0.00 | 99.83 | $\begin{aligned} & \hline 58.7 \\ & 8 \\ & \hline \end{aligned}$ |
| BH167 A4 34 | core | 49.97 | 2.28 | $\begin{aligned} & 17.3 \\ & 8 \\ & \hline \end{aligned}$ | 0.96 | 0.00 | $\begin{aligned} & 13.1 \\ & 0 \\ & \hline \end{aligned}$ | 0.44 | $\begin{aligned} & 15.8 \\ & 0 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.24 | 0.00 | $\begin{aligned} & 100.1 \\ & 6 \end{aligned}$ | $\begin{aligned} & 57.3 \\ & 1 \\ & \hline \end{aligned}$ |
| BH167 A4 35 | rim | 49.77 | 1.65 | $\begin{aligned} & 20.4 \\ & 6 \\ & \hline \end{aligned}$ | 0.77 | 0.00 | $\begin{aligned} & 13.9 \\ & 0 \\ & \hline \end{aligned}$ | 0.50 | $\begin{aligned} & 11.8 \\ & 1 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | 0.16 | 0.01 | 99.07 | $\begin{aligned} & 54.7 \\ & 7 \\ & \hline \end{aligned}$ |
| $\begin{aligned} & \text { BH167 A4 25- } \\ & 35 \end{aligned}$ | average | 49.99 | 2.48 | $\begin{aligned} & 15.0 \\ & 3 \end{aligned}$ | 0.91 | 0.01 | $\begin{aligned} & 14.1 \\ & 4 \\ & \hline \end{aligned}$ | 0.36 | $\begin{aligned} & 16.4 \\ & 3 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.26 | 0.00 | 99.63 | $62.8$ |
| BH167 A4 41 | rim | 52.11 | 2.91 | 8.25 | 0.42 | 0.26 | $\begin{aligned} & 17.3 \\ & 6 \\ & \hline \end{aligned}$ | 0.20 | $18.9$ | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | 0.26 | 0.00 | $\begin{aligned} & 100.7 \\ & 5 \end{aligned}$ | $\begin{aligned} & 78.9 \\ & 4 \\ & \hline \end{aligned}$ |
| BH167 A4 42 | rimward | 52.13 | 2.83 | 7.86 | 0.54 | 0.17 | $\begin{aligned} & 17.3 \\ & 4 \end{aligned}$ | 0.19 | $\begin{aligned} & 19.2 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.27 | 0.01 | $\begin{aligned} & 100.6 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 79.7 \\ & 1 \\ & \hline \end{aligned}$ |
| BH167 A4 43 | rimward | 52.44 | 2.73 | 8.00 | 0.48 | 0.10 | $\begin{aligned} & 17.3 \\ & 2 \\ & \hline \end{aligned}$ | 0.24 | $\begin{aligned} & 19.2 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.24 | 0.01 | $\begin{aligned} & 100.7 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 79.4 \\ & 1 \\ & \hline \end{aligned}$ |
| BH167 A4 44 | rimward | 52.50 | 2.75 | 7.71 | 0.55 | 0.19 | $\begin{aligned} & 17.3 \\ & 5 \end{aligned}$ | 0.20 | $\begin{aligned} & 19.1 \\ & 5 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 8 \\ & \hline \end{aligned}$ | 0.26 | 0.01 | $\begin{aligned} & 100.7 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & 80.0 \\ & 5 \end{aligned}$ |
| BH167 A4 45 | core | 52.29 | 2.70 | 7.57 | 0.41 | 0.16 | $\begin{aligned} & 17.3 \\ & 7 \\ & \hline \end{aligned}$ | 0.21 | $\begin{aligned} & 19.0 \\ & 9 \end{aligned}$ | $0.0$ | 0.25 | 0.01 | $\begin{aligned} & 100.0 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 80.3 \\ & 5 \\ & \hline \end{aligned}$ |
| BH167 A4 46 | rimward | 53.10 | 1.66 | 7.60 | 0.26 | 0.10 | $\begin{aligned} & 17.6 \\ & 0 \end{aligned}$ | 0.22 | $\begin{aligned} & 19.3 \\ & 2 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.22 | 0.01 | $\begin{aligned} & 100.1 \\ & 0 \end{aligned}$ | $\begin{aligned} & 80.4 \\ & 9 \\ & \hline \end{aligned}$ |
| BH167 A4 47 | rimward | 53.00 | 1.53 | 8.41 | 0.43 | 0.05 | $\begin{aligned} & 17.4 \\ & 1 \\ & \hline \end{aligned}$ | 0.22 | $\begin{aligned} & 18.8 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.33 | 0.01 | $\begin{aligned} & 100.2 \\ & 1 \end{aligned}$ | $\begin{aligned} & 78.6 \\ & 7 \\ & \hline \end{aligned}$ |
| BH167 A4 48 | rimward | 52.97 | 1.53 | 9.94 | 0.33 | 0.03 | $\begin{aligned} & 17.7 \\ & 5 \\ & \hline \end{aligned}$ | 0.28 | $\begin{aligned} & 17.6 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.21 | 0.01 | $\begin{aligned} & 100.6 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 76.0 \\ & 9 \\ & \hline \end{aligned}$ |
| BH167 A4 49 | rim | 52.57 | 1.70 | $\begin{aligned} & 11.2 \\ & 1 \end{aligned}$ | 0.60 | 0.06 | $\begin{aligned} & 16.9 \\ & 1 \end{aligned}$ | 0.31 | $17.3$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.22 | 0.01 | $\begin{aligned} & 100.9 \\ & 0 \end{aligned}$ | $\begin{aligned} & 72.8 \\ & 8 \\ & \hline \end{aligned}$ |
| $\begin{aligned} & \text { BH167 A4 41- } \\ & 49 \\ & \hline \end{aligned}$ | average | 52.57 | 2.26 | 8.51 | 0.45 | 0.12 | $\begin{aligned} & 17.3 \\ & 8 \end{aligned}$ | 0.23 | $\begin{aligned} & 18.7 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.25 | 0.01 | $\begin{aligned} & 100.5 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & 78.5 \\ & 1 \\ & \hline \end{aligned}$ |
| BH167 A4 55 | rim | 50.27 | 1.33 | $\begin{aligned} & 19.6 \\ & 8 \\ & \hline \end{aligned}$ | 0.62 | 0.00 | $\begin{aligned} & 13.0 \\ & 8 \\ & \hline \end{aligned}$ | 0.48 | $\begin{aligned} & 14.7 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.20 | 0.00 | $\begin{aligned} & 100.4 \\ & 4 \end{aligned}$ | $\begin{aligned} & 54.2 \\ & 2 \\ & \hline \end{aligned}$ |
| BH167 A4 56 | core | 49.21 | 2.63 | $\begin{aligned} & 14.4 \\ & 0 \\ & \hline \end{aligned}$ | 0.93 | 0.00 | $\begin{aligned} & 13.1 \\ & 6 \\ & \hline \end{aligned}$ | 0.25 | $\begin{aligned} & 17.7 \\ & 7 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.32 | 0.01 | 98.67 | $\begin{aligned} & 61.9 \\ & 5 \end{aligned}$ |
| BH167 A4 57 | core | 49.71 | 2.45 | $\begin{aligned} & 15.4 \\ & 8 \\ & \hline \end{aligned}$ | 0.85 | 0.01 | $\begin{aligned} & 14.0 \\ & 4 \end{aligned}$ | 0.38 | $\begin{aligned} & 16.1 \\ & 9 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.24 | 0.01 | 99.36 | $\begin{aligned} & 61.7 \\ & 9 \\ & \hline \end{aligned}$ |
| BH167 A4 58 | rim | 50.01 | 1.73 | $\begin{aligned} & 18.5 \\ & 6 \\ & \hline \end{aligned}$ | 0.61 | 0.01 | $12.9$ | 0.42 | $\begin{aligned} & 15.1 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.24 | 0.00 | 99.66 | $\begin{aligned} & 55.3 \\ & 5 \\ & \hline \end{aligned}$ |
| $\begin{aligned} & \text { BH167 A4 55- } \\ & 58 \\ & \hline \end{aligned}$ | average | 49.80 | 2.04 | $\begin{aligned} & 17.0 \\ & 3 \end{aligned}$ | 0.75 | 0.01 | $\begin{aligned} & 13.3 \\ & 0 \\ & \hline \end{aligned}$ | 0.38 | $\begin{aligned} & 15.9 \\ & 7 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.25 | 0.01 | 99.53 | $\begin{aligned} & 58.3 \\ & 3 \end{aligned}$ |
| BH167 A4 59 | cpx overgrowth rep | 49.62 | 1.68 | $\begin{aligned} & 19.9 \\ & 2 \end{aligned}$ | 0.73 | 0.01 | $\begin{aligned} & 11.7 \\ & 0 \end{aligned}$ | 0.48 | $\begin{aligned} & 15.5 \\ & 9 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.22 | 0.00 | 99.93 | $\begin{aligned} & 51.1 \\ & 4 \end{aligned}$ |
| BH167 A4 69 | core | 50.59 | 2.06 | $\begin{aligned} & 16.5 \\ & 7 \end{aligned}$ | 0.63 | 0.00 | $\begin{aligned} & 16.8 \\ & 1 \\ & \hline \end{aligned}$ | 0.40 | $\begin{aligned} & 12.4 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | 0.19 | 0.00 | 99.76 | $\begin{aligned} & 64.3 \\ & 9 \\ & \hline \end{aligned}$ |
| BH167 A4 70 | rim | 50.97 | 1.37 | $\begin{aligned} & 14.4 \\ & 8 \\ & \hline \end{aligned}$ | 0.48 | 0.00 | $\begin{aligned} & 16.8 \\ & 0 \end{aligned}$ | 0.38 | $13.7$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.22 | 0.03 | 98.51 | $\begin{aligned} & 67.3 \\ & 9 \end{aligned}$ |
| $\begin{aligned} & \hline \text { BH167 A4 69- } \\ & 70 \\ & \hline \end{aligned}$ | average | 50.78 | 1.72 | $\begin{aligned} & 15.5 \\ & 2 \\ & \hline \end{aligned}$ | 0.56 | 0.00 | $\begin{aligned} & 16.8 \\ & 1 \\ & \hline \end{aligned}$ | 0.39 | $\begin{aligned} & 13.1 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.21 | 0.01 | 99.13 | $\begin{aligned} & \hline 65.8 \\ & 9 \\ & \hline \end{aligned}$ |
| BH167 A4 72 | rim | 54.86 | 3.38 | 7.69 | 0.51 | 0.42 | $\begin{aligned} & 17.6 \\ & 4 \end{aligned}$ | 0.22 | $18.2$ | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 0.20 | 0.02 | $\begin{aligned} & 103.2 \\ & 0 \end{aligned}$ | $80.3$ |
| BH167 A4 72 | rimward | 50.53 | 3.47 | 7.03 | 0.52 | 0.47 | $\begin{aligned} & 16.5 \\ & 8 \end{aligned}$ | 0.15 | $\begin{aligned} & 20.5 \\ & 1 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.23 | 0.00 | 99.52 | $\begin{aligned} & 80.7 \\ & 8 \\ & \hline \end{aligned}$ |
| BH167 A4 72 | rimward | 50.58 | 3.45 | 6.89 | 0.59 | 0.63 | $\begin{aligned} & 16.5 \\ & 1 \\ & \hline \end{aligned}$ | 0.16 | $\begin{aligned} & 20.4 \\ & 1 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.20 | 0.00 | 99.45 | $\begin{aligned} & 81.0 \\ & 3 \end{aligned}$ |
| BH167 A4 72 | rimward | 50.27 | 3.51 | 6.99 | 0.62 | 0.42 | $\begin{aligned} & 16.4 \\ & 9 \\ & \hline \end{aligned}$ | 0.20 | $\begin{aligned} & 19.9 \\ & 0 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 0.22 | 0.00 | 98.64 | $\begin{aligned} & 80.7 \\ & 9 \end{aligned}$ |


| BH167 A4 72 | core | 50.71 | 3.36 | 7.72 | 0.56 | 0.35 | $\begin{aligned} & 16.8 \\ & 0 \\ & \hline \end{aligned}$ | 0.17 | $\begin{aligned} & 19.5 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 7 \\ & \hline \end{aligned}$ | 0.21 | 0.00 | 99.49 | $\begin{aligned} & 79.4 \\ & 9 \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| BH167 A4 72 | core | 50.77 | 3.12 | 7.74 | 0.47 | 0.33 | $\begin{aligned} & 17.0 \\ & 9 \\ & \hline \end{aligned}$ | 0.17 | $\begin{aligned} & 19.5 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.19 | 0.01 | 99.50 | $\begin{aligned} & 79.7 \\ & 4 \end{aligned}$ |
| BH167 A4 72 | rimward | 50.28 | 2.73 | 8.74 | 0.51 | 0.22 | $\begin{aligned} & 15.8 \\ & 0 \\ & \hline \end{aligned}$ | 0.23 | $\begin{aligned} & 19.9 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.18 | 0.01 | 98.68 | $\begin{aligned} & 76.3 \\ & 1 \\ & \hline \end{aligned}$ |
| BH167 A4 72 | rimward | 51.14 | 2.67 | 7.79 | 0.50 | 0.19 | $\begin{aligned} & 16.7 \\ & 8 \\ & \hline \end{aligned}$ | 0.17 | $20.3$ | $\begin{aligned} & \hline 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.19 | 0.01 | 99.79 | $\begin{aligned} & 79.3 \\ & 4 \\ & \hline \end{aligned}$ |
| BH167 A4 72 | rimward | 51.19 | 2.50 | 8.69 | 0.51 | 0.04 | $\begin{aligned} & 17.4 \\ & 1 \\ & \hline \end{aligned}$ | 0.20 | $\begin{aligned} & 18.8 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.23 | 0.00 | 99.62 | $\begin{aligned} & 78.1 \\ & 2 \\ & \hline \end{aligned}$ |
| BH167 A4 72 | rim | 49.10 | 0.77 | $\begin{aligned} & 19.3 \\ & 2 \end{aligned}$ | 0.29 | 0.00 | 8.29 | 0.61 | $\begin{aligned} & 20.0 \\ & 1 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.15 | 0.05 | 98.59 | $\begin{aligned} & 43.3 \\ & 3 \\ & \hline \end{aligned}$ |
| BH167 A4 72 | x average | 50.94 | 2.90 | 8.86 | 0.51 | 0.31 | $15.9$ | 0.23 | $\begin{aligned} & 19.7 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.20 | 0.01 | 99.65 | $\begin{aligned} & 75.9 \\ & 3 \\ & \hline \end{aligned}$ |
| BH167 A4 73 | rim | 52.29 | 1.50 | $\begin{aligned} & 10.3 \\ & 2 \\ & \hline \end{aligned}$ | 0.44 | 0.00 | $\begin{aligned} & 17.0 \\ & 0 \end{aligned}$ | 0.27 | $\begin{aligned} & 18.2 \\ & 8 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.3 \\ & 1 \end{aligned}$ | $\begin{aligned} & \hline 74.5 \\ & 8 \\ & \hline \end{aligned}$ |
| BH167 A4 73 | rimward | 51.95 | 1.48 | $\begin{aligned} & 10.6 \\ & 6 \end{aligned}$ | 0.51 | 0.08 | $17.1$ | 0.30 | $\begin{aligned} & 17.4 \\ & 2 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.18 | 0.00 | 99.71 | $\begin{aligned} & 74.1 \\ & 4 \\ & \hline \end{aligned}$ |
| BH167 A4 73 | rimward | 51.79 | 1.48 | $\begin{aligned} & 11.2 \\ & 5 \\ & \hline \end{aligned}$ | 0.42 | 0.04 | $\begin{aligned} & 17.2 \\ & 0 \\ & \hline \end{aligned}$ | 0.30 | $\begin{aligned} & 17.0 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.21 | 0.00 | 99.70 | $\begin{aligned} & 73.1 \\ & 5 \\ & \hline \end{aligned}$ |
| BH167 A4 73 | rimward | 52.18 | 1.39 | $\begin{aligned} & 11.5 \\ & 2 \end{aligned}$ | 0.52 | 0.03 | $\begin{aligned} & 17.0 \\ & 0 \end{aligned}$ | 0.33 | $\begin{aligned} & 16.9 \\ & 4 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.23 | 0.01 | $\begin{aligned} & 100.1 \\ & 5 \end{aligned}$ | $\begin{aligned} & 72.4 \\ & 5 \\ & \hline \end{aligned}$ |
| BH167 A4 73 | rimward | 52.25 | 1.44 | $\begin{aligned} & 12.3 \\ & 9 \\ & \hline \end{aligned}$ | 0.37 | 0.00 | $\begin{aligned} & 17.3 \\ & 0 \\ & \hline \end{aligned}$ | 0.33 | $\begin{aligned} & 16.0 \\ & 4 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.22 | 0.00 | $\begin{aligned} & 100.3 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 71.3 \\ & 3 \\ & \hline \end{aligned}$ |
| BH167 A4 73 | rimward | 51.93 | 1.43 | $\begin{aligned} & 12.6 \\ & 3 \end{aligned}$ | 0.41 | 0.04 | $\begin{aligned} & 17.3 \\ & 1 \\ & \hline \end{aligned}$ | 0.31 | $\begin{aligned} & 15.7 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.19 | 0.01 | $\begin{aligned} & 100.0 \\ & 4 \end{aligned}$ | $70.9$ |
| BH167 A4 73 | rimward | 52.22 | 1.41 | $\begin{aligned} & 13.2 \\ & 6 \end{aligned}$ | 0.39 | 0.04 | $17.3$ | 0.33 | $15.3$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.21 | 0.01 | $\begin{aligned} & 100.6 \\ & 1 \end{aligned}$ | $\begin{aligned} & 70.0 \\ & 3 \end{aligned}$ |
| BH167 A4 73 | core | 52.00 | 1.43 | $13.4$ | 0.33 | 0.03 | $\begin{aligned} & 17.3 \\ & 0 \end{aligned}$ | 0.40 | $14.8$ | $\begin{aligned} & 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.18 | 0.00 | 99.97 | $\begin{aligned} & \hline 69.6 \\ & 8 \\ & \hline \end{aligned}$ |
| BH167 A4 73 | rimward | 52.36 | 1.41 | $\begin{aligned} & 14.2 \\ & 6 \\ & \hline \end{aligned}$ | 0.38 | 0.03 | $17.4$ | 0.38 | $\begin{aligned} & 14.0 \\ & 8 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.18 | 0.00 | $\begin{aligned} & 100.5 \\ & 3 \end{aligned}$ | $\begin{aligned} & 68.5 \\ & 5 \end{aligned}$ |
| BH167 A4 73 | rimward | 43.86 | 1.59 | $\begin{aligned} & 19.4 \\ & 5 \end{aligned}$ | 7.17 | 0.05 | $\begin{aligned} & 15.0 \\ & 5 \end{aligned}$ | 0.51 | 7.87 | $\begin{aligned} & 0.1 \\ & 6 \\ & \hline \end{aligned}$ | 0.29 | 0.04 | 96.04 | $\begin{aligned} & 57.9 \\ & 6 \\ & \hline \end{aligned}$ |
| BH167 A4 73 | rimward | 50.16 | 2.01 | $\begin{aligned} & 16.2 \\ & 9 \end{aligned}$ | 0.72 | 0.03 | $\begin{aligned} & 16.0 \\ & 0 \end{aligned}$ | 0.38 | $\begin{aligned} & 13.3 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.23 | 0.01 | 99.20 | $\begin{aligned} & 63.6 \\ & 5 \\ & \hline \end{aligned}$ |
| BH167 A4 73 | rimward | 49.75 | 2.01 | $\begin{aligned} & 18.2 \\ & 2 \\ & \hline \end{aligned}$ | 0.79 | 0.00 | $\begin{aligned} & 15.1 \\ & 4 \\ & \hline \end{aligned}$ | 0.42 | $\begin{aligned} & 13.0 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.20 | 0.00 | 99.61 | $\begin{aligned} & 59.6 \\ & 9 \\ & \hline \end{aligned}$ |
| BH167 A4 73 | rimward | 50.38 | 1.87 | $\begin{aligned} & 18.0 \\ & 2 \end{aligned}$ | 0.77 | 0.00 | $\begin{aligned} & 14.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.48 | $\begin{aligned} & 14.0 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 7 \end{aligned}$ | 0.21 | 0.02 | 99.86 | $\begin{aligned} & 58.1 \\ & 3 \\ & \hline \end{aligned}$ |
| BH167 A4 73 | rimward | 49.67 | 1.34 | $\begin{aligned} & 18.2 \\ & 3 \end{aligned}$ | 0.84 | 0.00 | $\begin{aligned} & 12.1 \\ & 9 \end{aligned}$ | 0.42 | $\begin{aligned} & 16.3 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.23 | 0.01 | 99.26 | $\begin{aligned} & 54.3 \\ & 8 \end{aligned}$ |
| BH167 A4 73 | rim | 50.48 | 0.27 | $\begin{aligned} & 20.5 \\ & 4 \end{aligned}$ | 0.23 | 0.07 | 8.89 | 0.54 | $\begin{aligned} & 18.2 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.16 | 0.00 | 99.44 | $\begin{aligned} & 43.5 \\ & 5 \\ & \hline \end{aligned}$ |
| BH167 A4 73 | x average | 50.88 | 1.47 | $\begin{aligned} & 14.7 \\ & 0 \\ & \hline \end{aligned}$ | 0.95 | 0.03 | $\begin{aligned} & 15.7 \\ & 6 \\ & \hline \end{aligned}$ | 0.38 | $\begin{aligned} & 15.2 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.21 | 0.01 | 99.65 | $\begin{aligned} & \hline 65.4 \\ & 8 \\ & \hline \end{aligned}$ |
| BH175 A1 87 | ground rep | 49.15 | 4.52 | 8.77 | 0.80 | 0.38 | $\begin{aligned} & 15.5 \\ & 7 \\ & \hline \end{aligned}$ | 0.16 | $19.5$ | $\begin{aligned} & 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.22 | 0.02 | 99.10 | $\begin{aligned} & 75.9 \\ & 9 \\ & \hline \end{aligned}$ |
| BH175 A1 88 | ground rep | 48.02 | 2.52 | $\begin{aligned} & 16.9 \\ & 7 \end{aligned}$ | 1.13 | 0.00 | $12.3$ | 0.36 | $\begin{aligned} & 16.9 \\ & 8 \end{aligned}$ | $0.0$ | 0.25 | 0.03 | 98.60 | $\begin{aligned} & 56.4 \\ & 3 \\ & \hline \end{aligned}$ |
| BH175 A1 101 | core | 51.47 | 2.52 | 6.12 | 0.31 | 0.67 | $17.3$ | 0.14 | $\begin{aligned} & 20.4 \\ & 5 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 7 \\ & \hline \end{aligned}$ | 0.18 | 0.01 | 99.28 | $\begin{aligned} & \hline 83.4 \\ & 9 \end{aligned}$ |
| BH175 A1 102 | rim | 50.79 | 2.94 | 6.38 | 0.37 | 0.61 | $\begin{aligned} & 17.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.17 | $\begin{aligned} & 20.6 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.23 | 0.03 | 99.23 | $\begin{aligned} & 82.5 \\ & 9 \\ & \hline \end{aligned}$ |
| $\begin{aligned} & \text { BH175 A1 101- } \\ & 102 \end{aligned}$ | average | 51.13 | 2.73 | 6.25 | 0.34 | 0.64 | $\begin{aligned} & 17.1 \\ & 8 \\ & \hline \end{aligned}$ | 0.15 | $\begin{aligned} & 20.5 \\ & 5 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 6 \\ & \hline \end{aligned}$ | 0.21 | 0.02 | 99.25 | $\begin{aligned} & 83.0 \\ & 4 \\ & \hline \end{aligned}$ |
| BH175 A1 107 | rim | 52.90 | 1.90 | 6.80 | 0.22 | 0.42 | $\begin{aligned} & 17.8 \\ & 1 \\ & \hline \end{aligned}$ | 0.20 | $\begin{aligned} & 19.8 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.19 | 0.02 | $\begin{aligned} & 100.3 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 82.3 \\ & \hline \end{aligned}$ |
| BH175 A1 107 | rimward | 52.11 | 2.38 | 5.74 | 0.36 | 0.50 | $\begin{aligned} & 17.0 \\ & 4 \end{aligned}$ | 0.14 | $21.0$ | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | 0.22 | 0.02 | 99.55 | $\begin{aligned} & 84.1 \\ & 0 \\ & \hline \end{aligned}$ |
| BH175 A1 107 | rimward | 52.16 | 2.42 | 6.06 | 0.42 | 0.66 | $\begin{aligned} & 17.0 \\ & 6 \end{aligned}$ | 0.14 | $\begin{aligned} & 21.0 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.21 | 0.00 | $\begin{aligned} & 100.1 \\ & 6 \end{aligned}$ | $\begin{aligned} & \hline 83.3 \\ & 8 \\ & \hline \end{aligned}$ |
| BH175 A1 107 | core | 52.30 | 2.29 | 6.18 | 0.37 | 0.53 | $\begin{aligned} & 17.4 \\ & 6 \\ & \hline \end{aligned}$ | 0.15 | $\begin{aligned} & 20.2 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 6 \\ & \hline \end{aligned}$ | 0.20 | 0.01 | 99.82 | $\begin{aligned} & \hline 83.4 \\ & 3 \\ & \hline \end{aligned}$ |
| BH175 A1 107 | core | 51.90 | 2.41 | 5.93 | 0.27 | 0.48 | $\begin{aligned} & 17.0 \\ & 7 \end{aligned}$ | 0.16 | $\begin{aligned} & 21.0 \\ & 0 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 6 \end{aligned}$ | 0.23 | 0.01 | 99.50 | $\begin{aligned} & 83.7 \\ & 0 \\ & \hline \end{aligned}$ |
| BH175 A1 107 | rimward | 51.68 | 2.81 | 6.33 | 0.27 | 0.77 | $\begin{aligned} & 16.9 \\ & 3 \\ & \hline \end{aligned}$ | 0.15 | $\begin{aligned} & 20.8 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 0.21 | 0.00 | 99.98 | $82.6$ |
| BH175 A1 107 | rimward | 50.17 | 3.78 | 7.06 | 0.51 | 0.67 | $\begin{aligned} & 16.2 \\ & 9 \end{aligned}$ | 0.11 | $\begin{aligned} & 20.1 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.21 | 0.05 | 99.06 | $\begin{aligned} & \hline 80.4 \\ & 4 \\ & \hline \end{aligned}$ |
| BH175 A1 107 | rim | 50.34 | 4.49 | 7.70 | 0.95 | 0.10 | $\begin{aligned} & 16.1 \\ & 7 \\ & \hline \end{aligned}$ | 0.19 | $\begin{aligned} & 19.7 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.22 | 0.01 | 99.91 | $\begin{aligned} & 78.9 \\ & 1 \end{aligned}$ |
| BH175 A1 107 | x average | 51.69 | 2.81 | 6.47 | 0.42 | 0.52 | $\begin{aligned} & 16.9 \\ & 8 \end{aligned}$ | 0.16 | $\begin{aligned} & 20.4 \\ & 9 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.21 | 0.01 | 99.79 | $\begin{aligned} & 82.3 \\ & 7 \\ & \hline \end{aligned}$ |
| BH175 A1 108 | rim | 51.85 | 2.78 | 5.91 | 0.47 | 0.71 | $\begin{aligned} & 16.9 \\ & 1 \\ & \hline \end{aligned}$ | 0.14 | $\begin{aligned} & 21.0 \\ & 5 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 5 \end{aligned}$ | 0.22 | 0.01 | $\begin{aligned} & 100.1 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 83.6 \\ & 1 \\ & \hline \end{aligned}$ |


| BH175 A1 108 | rimward | 50.59 | 3.94 | 6.77 | 0.46 | 0.57 | $\begin{aligned} & 16.4 \\ & 0 \\ & \hline \end{aligned}$ | 0.17 | $\begin{aligned} & 20.4 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.24 | 0.02 | 99.60 | $\begin{aligned} & \hline 81.1 \\ & 9 \\ & \hline \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| BH175 A1 108 | core | 50.93 | 3.56 | 5.95 | 0.50 | 0.65 | $\begin{aligned} & 16.4 \\ & 1 \end{aligned}$ | 0.16 | $\begin{aligned} & \hline 20.8 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.22 | 0.01 | 99.29 | $\begin{aligned} & 83.1 \\ & 0 \\ & \hline \end{aligned}$ |
| BH175 A1 108 | core | 50.55 | 3.49 | 6.74 | 0.49 | 0.51 | $\begin{aligned} & 16.2 \\ & 6 \\ & \hline \end{aligned}$ | 0.17 | $\begin{aligned} & \hline 20.8 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.23 | 0.01 | 99.26 | $\begin{aligned} & \hline 81.1 \\ & 3 \\ & \hline \end{aligned}$ |
| BH175 A1 108 | rimward | 50.97 | 2.88 | 7.32 | 0.47 | 0.24 | $\begin{aligned} & 16.0 \\ & 9 \end{aligned}$ | 0.16 | $20.6$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.25 | 0.02 | 99.04 | $\begin{aligned} & \hline 79.6 \\ & 6 \\ & \hline \end{aligned}$ |
| BH175 A1 108 | rim | 50.87 | 3.15 | 7.45 | 0.64 | 0.17 | $\begin{aligned} & \hline 16.4 \\ & 0 \\ & \hline \end{aligned}$ | 0.19 | $\begin{aligned} & \hline 20.3 \\ & 6 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.19 | 0.03 | 99.49 | $\begin{aligned} & 79.6 \\ & 9 \end{aligned}$ |
| BH175 A1 108 | x average | 50.96 | 3.30 | 6.69 | 0.50 | 0.47 | $\begin{aligned} & 16.4 \\ & 1 \\ & \hline \end{aligned}$ | 0.17 | $\begin{aligned} & 20.6 \\ & 9 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.22 | 0.02 | 99.46 | $\begin{aligned} & 81.4 \\ & 0 \end{aligned}$ |
| BH175 A1 109 | rim | 52.04 | 2.73 | 5.80 | 0.42 | 0.55 | $\begin{aligned} & 16.9 \\ & 6 \\ & \hline \end{aligned}$ | 0.15 | $\begin{aligned} & 21.0 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.23 | 0.01 | 99.97 | $\begin{aligned} & 83.8 \\ & 9 \end{aligned}$ |
| BH175 A1 109 | rimward | 50.41 | 8.34 | 8.31 | 0.37 | 0.62 | $\begin{aligned} & 15.0 \\ & 8 \end{aligned}$ | 0.13 | $\begin{aligned} & 15.5 \\ & 0 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 5 \end{aligned}$ | 0.23 | 1.03 | $\begin{aligned} & 100.0 \\ & 6 \end{aligned}$ | $\begin{aligned} & 76.3 \\ & 8 \end{aligned}$ |
| BH175 A1 109 | rimward | 50.78 | 3.73 | 6.15 | 0.40 | 0.88 | $\begin{aligned} & 16.3 \\ & 0 \end{aligned}$ | 0.11 | $20.7$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.24 | 0.01 | 99.42 | $\begin{aligned} & 82.6 \\ & 1 \end{aligned}$ |
| BH175 A1 109 | rimward | 54.50 | 3.45 | 6.50 | 0.44 | 0.69 | $\begin{aligned} & 16.6 \\ & 5 \\ & \hline \end{aligned}$ | 0.17 | $20.7$ | $\begin{aligned} & 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.26 | 0.01 | $\begin{aligned} & 103.4 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 82.0 \\ & 2 \\ & \hline \end{aligned}$ |
| BH175 A1 109 | rimward | 51.49 | 3.46 | 6.38 | 0.47 | 0.73 | $\begin{aligned} & 16.2 \\ & 6 \end{aligned}$ | 0.17 | $\begin{aligned} & 20.6 \\ & 0 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.32 | 0.16 | $\begin{aligned} & 100.0 \\ & 9 \end{aligned}$ | $\begin{aligned} & 81.9 \\ & 7 \\ & \hline \end{aligned}$ |
| BH175 A1 109 | core | 51.31 | 3.47 | 6.36 | 0.50 | 0.82 | $\begin{aligned} & 16.8 \\ & 1 \\ & \hline \end{aligned}$ | 0.14 | $\begin{aligned} & 20.2 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.23 | 0.01 | 99.96 | $\begin{aligned} & 82.4 \\ & 9 \\ & \hline \end{aligned}$ |
| BH175 A1 109 | rimward | 51.02 | 3.82 | 6.68 | 0.44 | 0.80 | $\begin{aligned} & 16.5 \\ & 5 \\ & \hline \end{aligned}$ | 0.15 | $20.3$ | $\begin{aligned} & 0.0 \\ & 6 \end{aligned}$ | 0.24 | 0.01 | $\begin{aligned} & 100.1 \\ & 0 \end{aligned}$ | $\begin{aligned} & 81.5 \\ & 3 \\ & \hline \end{aligned}$ |
| BH175 A1 109 | rimward | 51.14 | 3.44 | 6.35 | 0.42 | 0.55 | $\begin{aligned} & 16.7 \\ & 0 \\ & \hline \end{aligned}$ | 0.12 | $\begin{aligned} & 20.6 \\ & 2 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.26 | 0.03 | 99.62 | $82.4$ |
| BH175 A1 109 | rimward | 52.09 | 2.47 | 5.57 | 0.37 | 0.49 | $17.0$ | 0.15 | $\begin{aligned} & 20.9 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.22 | 0.05 | 99.34 | $84.4$ $8$ |
| BH175 A1 109 | rim | 51.55 | 3.28 | 6.24 | 0.52 | 0.68 | $\begin{aligned} & 16.4 \\ & 5 \\ & \hline \end{aligned}$ | 0.15 | $\begin{aligned} & 20.6 \\ & 4 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.25 | 0.02 | 99.76 | $\begin{aligned} & 82.4 \\ & 4 \end{aligned}$ |
| BH175 A1 109 | x average | 51.63 | 3.82 | 6.43 | 0.43 | 0.68 | $\begin{aligned} & 16.4 \\ & 0 \end{aligned}$ | 0.14 | $20.1$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.25 | 0.13 | $\begin{aligned} & 100.1 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 82.0 \\ & 2 \end{aligned}$ |
| BH175 A1 119 | ground rep | 50.04 | 3.33 | $\begin{aligned} & 10.4 \\ & 1 \\ & \hline \end{aligned}$ | 0.44 | 0.00 | $15.4$ | 0.22 | $\begin{aligned} & 18.1 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | 0.25 | 0.07 | 98.26 | $\begin{aligned} & 72.5 \\ & 0 \end{aligned}$ |
| BH175 A1 120 | core | 50.87 | 3.21 | 6.29 | 0.49 | 0.77 | $\begin{aligned} & 17.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.17 | $20.3$ | $0.0$ | 0.19 | 0.00 | 99.38 | $\begin{aligned} & 82.8 \\ & 4 \\ & \hline \end{aligned}$ |
| BH175 A1 120 | rimwards | 51.06 | 3.40 | 6.32 | 0.41 | 0.63 | $\begin{aligned} & 16.6 \\ & 7 \\ & \hline \end{aligned}$ | 0.14 | $\begin{aligned} & 20.6 \\ & 8 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.22 | 0.01 | 99.55 | $82.4$ |
| BH175 A1 120 | rimwards | 50.14 | 4.06 | 6.43 | 0.50 | 0.57 | $\begin{aligned} & 16.5 \\ & 3 \\ & \hline \end{aligned}$ | 0.16 | $\begin{aligned} & 20.3 \\ & 2 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 6 \\ & \hline \end{aligned}$ | 0.21 | 0.01 | 98.97 | $\begin{aligned} & 82.0 \\ & 9 \\ & \hline \end{aligned}$ |
| BH175 A1 120 | rimwards | 50.14 | 3.99 | 6.49 | 0.45 | 0.71 | $\begin{aligned} & 16.6 \\ & 1 \\ & \hline \end{aligned}$ | 0.16 | $\begin{aligned} & 20.1 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.25 | 0.02 | 98.99 | $\begin{aligned} & 82.0 \\ & 1 \end{aligned}$ |
| BH175 A1 120 | rimwards | 50.73 | 3.47 | 6.17 | 0.35 | 0.82 | $\begin{aligned} & 16.7 \\ & 4 \\ & \hline \end{aligned}$ | 0.14 | $\begin{aligned} & 20.4 \\ & 8 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.22 | 0.01 | 99.15 | $\begin{aligned} & 82.8 \\ & 7 \\ & \hline \end{aligned}$ |
| BH175 A1 120 | rimwards | 50.73 | 3.52 | 7.02 | 0.44 | 0.63 | $\begin{aligned} & 17.1 \\ & 3 \\ & \hline \end{aligned}$ | 0.16 | $\begin{aligned} & 19.9 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 6 \\ & \hline \end{aligned}$ | 0.22 | 0.01 | 99.88 | $\begin{aligned} & 81.3 \\ & 1 \\ & \hline \end{aligned}$ |
| BH175 A1 120 | rimwards | 50.63 | 3.69 | 6.93 | 0.49 | 0.41 | $16.8$ | 0.16 | $\begin{aligned} & 19.5 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 7 \\ & \hline \end{aligned}$ | 0.19 | 0.00 | 98.95 | $81.2$ $1$ |
| BH175 A1 120 | rim | 50.99 | 3.05 | 6.65 | 0.49 | 0.43 | $\begin{aligned} & 17.2 \\ & 7 \end{aligned}$ | 0.17 | $\begin{aligned} & 19.8 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.19 | 0.02 | 99.12 | $\begin{aligned} & 82.2 \\ & 3 \end{aligned}$ |
| BH175 A1 120 | x average | 50.66 | 3.55 | 6.54 | 0.45 | 0.62 | $\begin{aligned} & 16.8 \\ & 5 \\ & \hline \end{aligned}$ | 0.16 | $\begin{aligned} & \hline 20.1 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.21 | 0.01 | 99.25 | $\begin{aligned} & 82.1 \\ & 3 \\ & \hline \end{aligned}$ |
| BH175 A1 121 | core | 51.61 | 2.54 | 6.00 | 0.33 | 0.53 | $\begin{aligned} & 17.0 \\ & 6 \\ & \hline \end{aligned}$ | 0.15 | $\begin{aligned} & 20.8 \\ & 3 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | 0.21 | 0.03 | 99.33 | $\begin{aligned} & 83.5 \\ & 1 \\ & \hline \end{aligned}$ |
| BH175 A1 121 | rimwards | 53.90 | 2.24 | 6.41 | 0.42 | 0.27 | $\begin{aligned} & 17.7 \\ & 6 \end{aligned}$ | 0.15 | $\begin{aligned} & 19.7 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.19 | 0.01 | $\begin{aligned} & \hline 101.1 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 83.1 \\ & 5 \\ & \hline \end{aligned}$ |
| BH175 A1 121 | rimwards | 52.03 | 2.25 | 5.72 | 0.42 | 0.38 | $17.4$ | 0.18 | $20.7$ | $\begin{aligned} & 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.19 | 0.00 | 99.35 | $84.4$ <br> 3 |
| BH175 A1 121 | rimwards | 52.39 | 2.41 | 6.15 | 0.46 | 0.50 | $17.4$ | 0.14 | $\begin{aligned} & \hline 20.4 \\ & 2 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.18 | 0.01 | $\begin{aligned} & 100.1 \\ & 5 \end{aligned}$ | $\begin{aligned} & 83.5 \\ & 2 \end{aligned}$ |
| BH175 A1 121 | rimwards | 51.54 | 2.28 | 6.08 | 0.24 | 0.69 | $17.3$ | 0.21 | $\begin{aligned} & 20.7 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.19 | 0.01 | 99.33 | $\begin{aligned} & 83.5 \\ & 7 \\ & \hline \end{aligned}$ |
| BH175 A1 121 | rimwards | 52.21 | 2.07 | 6.12 | 0.24 | 0.53 | $\begin{aligned} & 17.6 \\ & 7 \\ & \hline \end{aligned}$ | 0.15 | $\begin{aligned} & 20.2 \\ & 4 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | 0.21 | 0.01 | 99.47 | $\begin{aligned} & 83.7 \\ & 3 \end{aligned}$ |
| BH175 A1 121 | rimwards | 50.89 | 2.37 | 5.98 | 0.32 | 0.67 | $\begin{aligned} & 17.1 \\ & 9 \\ & \hline \end{aligned}$ | 0.15 | $\begin{aligned} & 20.9 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 0.21 | 0.01 | 98.78 | $\begin{aligned} & 83.6 \\ & 6 \\ & \hline \end{aligned}$ |
| BH175 A1 121 | rim | 51.77 | 2.38 | 8.09 | 0.56 | 0.16 | $17.8$ | 0.20 | $\begin{aligned} & 18.4 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 0.15 | 0.02 | 99.67 | $79.7$ |
| BH175 A1 121 | x average | 52.04 | 2.32 | 6.32 | 0.37 | 0.47 | $\begin{aligned} & 17.4 \\ & 7 \\ & \hline \end{aligned}$ | 0.17 | $\begin{aligned} & 20.2 \\ & 6 \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | 0.19 | 0.01 | 99.66 | $\begin{aligned} & 83.1 \\ & 6 \end{aligned}$ |
| BH175 A1 123 | ground rep | 49.10 | 4.00 | $\begin{aligned} & \hline 14.1 \\ & 5 \\ & \hline \end{aligned}$ | 0.33 | 0.16 | $\begin{aligned} & 15.2 \\ & 3 \\ & \hline \end{aligned}$ | 0.28 | $\begin{aligned} & 15.0 \\ & 6 \end{aligned}$ | $\begin{aligned} & \hline 0.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.18 | 0.22 | 98.74 | $\begin{aligned} & 65.7 \\ & 4 \\ & \hline \end{aligned}$ |
| BH175 A1 125 | core | 52.03 | 2.83 | 6.28 | 0.39 | 0.72 | $\begin{aligned} & 17.3 \\ & 5 \end{aligned}$ | 0.20 | $\begin{aligned} & 20.1 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 0.22 | 0.02 | $\begin{aligned} & 100.2 \\ & 2 \end{aligned}$ | $\begin{aligned} & 83.1 \\ & 2 \end{aligned}$ |


|  | BH175 A1 125 | rimwards | 50.67 | 4.44 | 6.57 | 0.52 | 0.67 | 16.6 <br> 3 | 0.17 | 20.0 <br> 5 | 0.0 <br> 5 | 0.24 | 0.03 | 100.0 <br> 6 | 81.8 <br> 5 |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
|  | BH175 A1 125 | rimwards | 52.16 | 2.46 | 5.83 | 0.41 | 0.60 | 16.9 | 0.16 | 20.9 <br> 8 | 0.0 <br> 2 | 0.22 | 0.00 | 99.78 | 83.8 <br> 4 |  |
|  | BH175 A1 125 | rimwards | 51.56 | 3.13 | 6.64 | 0.44 | 0.56 | 16.6 <br> 8 | 0.18 | 20.3 <br> 6 | 0.0 <br> 4 | 0.21 | 0.02 | 99.83 | 81.7 <br> 4 |  |
|  | BH175 A1 125 | rimwards | 52.08 | 2.80 | 6.18 | 0.50 | 0.66 | 17.0 | 0.15 | 20.6 <br> 7 | 0.0 <br> 2 | 0.23 | 0.00 | 100.3 | 83.0 <br> 7 |  |
|  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |

## Appendix G4 - SFDC plagioclase mineral chemistry

| $\begin{aligned} & \text { ZON } \\ & \text { E } \\ & \hline \end{aligned}$ | SAMPLE ID | X Region | $\begin{aligned} & \mathrm{SiO} \\ & 2 \end{aligned}$ | $\begin{aligned} & \mathrm{TiO} \\ & 2 \end{aligned}$ | $\begin{aligned} & \text { Al2O } \\ & 3 \end{aligned}$ | $\begin{aligned} & \mathrm{Mg} \\ & \mathbf{O} \end{aligned}$ | CaO | $\begin{aligned} & \mathrm{Mn} \\ & \mathrm{O} \end{aligned}$ | $\begin{aligned} & \mathrm{Fe} \\ & \mathrm{O} \end{aligned}$ | $\begin{aligned} & \mathrm{Na} 2 \\ & \mathrm{O} \\ & \hline \end{aligned}$ | $\begin{aligned} & \mathrm{K} 2 \\ & \mathrm{O} \end{aligned}$ | $\begin{aligned} & \text { Ba } \\ & \mathbf{O} \end{aligned}$ | $\begin{aligned} & \mathrm{Sr} \\ & \mathbf{O} \end{aligned}$ | Total | An |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| D | BH268 A2 4 | chad rep | $\begin{aligned} & 48.5 \\ & 8 \end{aligned}$ | 0.03 | 30.31 | 0.24 | $\begin{aligned} & 14.5 \\ & 8 \\ & \hline \end{aligned}$ | 0.02 | $\begin{aligned} & \hline 0.8 \\ & 4 \end{aligned}$ | 3.36 | 0.08 | 0.04 | $\begin{aligned} & \hline 0.0 \\ & 6 \\ & \hline \end{aligned}$ | 98.12 | $\begin{aligned} & 70.1 \\ & 9 \end{aligned}$ |
|  | BH268 A2 5 | chad rep | $\begin{aligned} & 48.8 \\ & 3 \\ & \hline \end{aligned}$ | 0.05 | 30.42 | 0.22 | $\begin{aligned} & 14.5 \\ & 6 \\ & \hline \end{aligned}$ | 0.01 | $\begin{aligned} & 0.9 \\ & 0 \end{aligned}$ | 3.42 | 0.07 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 6 \end{aligned}$ | 98.54 | $\begin{aligned} & 69.9 \\ & 0 \\ & \hline \end{aligned}$ |
|  | BH268 A2 6 | core | $\begin{aligned} & 49.9 \\ & 3 \\ & \hline \end{aligned}$ | 0.07 | 30.20 | 0.24 | $\begin{aligned} & 14.6 \\ & 2 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & \hline 0.7 \\ & 4 \\ & \hline \end{aligned}$ | 3.33 | 0.09 | 0.03 | $\begin{aligned} & \hline 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 99.28 | $\begin{aligned} & 70.3 \\ & 9 \\ & \hline \end{aligned}$ |
|  | BH268 A2 7 | rim | $\begin{aligned} & 51.1 \\ & 4 \end{aligned}$ | 0.00 | 29.81 | 0.23 | $\begin{aligned} & 14.0 \\ & 7 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & 0.8 \\ & 0 \end{aligned}$ | 3.42 | 0.34 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 8 \end{aligned}$ | 99.88 | $\begin{aligned} & 68.0 \\ & 9 \\ & \hline \end{aligned}$ |
|  | BH268 A2 6-7 | average | $50.5$ | 0.04 | 30.00 | 0.24 | $\begin{aligned} & 14.3 \\ & 4 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & 0.7 \\ & 7 \end{aligned}$ | 3.38 | 0.21 | 0.01 | $\begin{aligned} & 0.0 \\ & 6 \end{aligned}$ | 99.58 | $\begin{aligned} & 69.2 \\ & 4 \\ & \hline \end{aligned}$ |
|  | BH268 A2 11 | chad rep | $\begin{aligned} & 50.3 \\ & 5 \end{aligned}$ | 0.04 | 30.40 | 0.22 | $\begin{aligned} & 14.6 \\ & 1 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & 0.8 \\ & 1 \end{aligned}$ | 3.39 | 0.07 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 8 \end{aligned}$ | 99.97 | $\begin{aligned} & 70.1 \\ & 6 \end{aligned}$ |
|  | BH268 A2 12 | core | $51.6$ | 0.07 | 30.59 | 0.25 | $14.6$ | 0.00 | $\begin{aligned} & \hline 0.7 \\ & 5 \\ & \hline \end{aligned}$ | 3.33 | 0.08 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | $\begin{aligned} & 101.3 \\ & 0 \end{aligned}$ | $\begin{aligned} & 70.5 \\ & 4 \\ & \hline \end{aligned}$ |
|  | BH268 A2 13 | rim | $\begin{aligned} & 51.1 \\ & 5 \end{aligned}$ | 0.02 | 30.89 | 0.23 | $\begin{aligned} & 14.8 \\ & 5 \end{aligned}$ | 0.03 | $\begin{aligned} & 0.7 \\ & 6 \end{aligned}$ | 3.18 | 0.07 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 8 \end{aligned}$ | $\begin{aligned} & 101.2 \\ & 5 \end{aligned}$ | $\begin{aligned} & 71.8 \\ & 0 \end{aligned}$ |
|  | $\begin{aligned} & \text { BH268 A2 12- } \\ & 13 \end{aligned}$ | average | $\begin{aligned} & \hline 51.4 \\ & 0 \end{aligned}$ | 0.04 | 30.74 | 0.24 | $\begin{aligned} & 14.7 \\ & 5 \end{aligned}$ | 0.02 | $\begin{aligned} & 0.7 \\ & 5 \end{aligned}$ | 3.25 | 0.07 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 5 \end{aligned}$ | $\begin{aligned} & 101.3 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & 71.1 \\ & 7 \\ & \hline \end{aligned}$ |
|  | BH268 A2 14 | ground (rep) | $\begin{aligned} & 57.6 \\ & 8 \end{aligned}$ | 0.10 | 26.66 | 0.05 | 9.40 | 0.00 | $\begin{aligned} & 0.8 \\ & 7 \end{aligned}$ | 6.21 | 0.38 | 0.00 | $\begin{aligned} & \hline 0.1 \\ & 3 \end{aligned}$ | $101.4$ | $\begin{aligned} & 44.5 \\ & 6 \\ & \hline \end{aligned}$ |
|  | BH268 A2 19 | chad core | $\begin{aligned} & \hline 51.1 \\ & 6 \end{aligned}$ | 0.09 | 30.27 | 0.25 | $\begin{aligned} & 14.5 \\ & 7 \end{aligned}$ | 0.05 | $\begin{aligned} & \hline 0.7 \\ & 3 \\ & \hline \end{aligned}$ | 3.49 | 0.07 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 8 \end{aligned}$ | $\begin{aligned} & 100.7 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 69.5 \\ & 0 \\ & \hline \end{aligned}$ |
|  | BH268 A2 20 | chad rim | $\begin{aligned} & \hline 52.1 \\ & 1 \\ & \hline \end{aligned}$ | 0.07 | 30.14 | 0.20 | $\begin{aligned} & 13.8 \\ & 5 \end{aligned}$ | 0.00 | $\begin{aligned} & \hline 0.8 \\ & 8 \end{aligned}$ | 3.82 | 0.10 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 6 \end{aligned}$ | $\begin{aligned} & 101.2 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 66.3 \\ & 5 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH268 A2 19- } \\ & 20 \end{aligned}$ | chad average | $\begin{aligned} & 51.6 \\ & 4 \\ & \hline \end{aligned}$ | 0.08 | 30.20 | 0.23 | $\begin{aligned} & 14.2 \\ & 1 \\ & \hline \end{aligned}$ | 0.03 | $\begin{aligned} & 0.8 \\ & 1 \end{aligned}$ | 3.65 | 0.08 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 101.0 \\ & 0 \end{aligned}$ | $\begin{aligned} & 67.9 \\ & 2 \\ & \hline \end{aligned}$ |
|  | BH268 A2 27 | ground (rep) | $\begin{aligned} & 52.8 \\ & 4 \end{aligned}$ | 0.16 | 27.63 | 0.33 | $\begin{aligned} & 11.4 \\ & 1 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & \hline 2.4 \\ & 2 \end{aligned}$ | 4.60 | 0.20 | 0.03 | $\begin{aligned} & \hline 0.0 \\ & 7 \\ & \hline \end{aligned}$ | 99.69 | $\begin{aligned} & \hline 57.1 \\ & 1 \\ & \hline \end{aligned}$ |
|  | BH268 A2 28 | plag rep (near cpx) | $52.8$ | 0.08 | 29.25 | 0.32 | $\begin{aligned} & 13.7 \\ & 1 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & \hline 1.0 \\ & 9 \end{aligned}$ | 3.91 | 0.09 | 0.02 | $\begin{aligned} & \hline 0.0 \\ & 6 \end{aligned}$ | $\begin{aligned} & 101.3 \\ & 2 \end{aligned}$ | $\begin{aligned} & 65.6 \\ & 2 \\ & \hline \end{aligned}$ |
|  | MH126 A2 44 | ground (rep) | $\begin{aligned} & 50.7 \\ & 0 \end{aligned}$ | 0.04 | 29.83 | 0.22 | $\begin{aligned} & 14.1 \\ & 0 \\ & \hline \end{aligned}$ | 0.02 | $\begin{aligned} & 1.1 \\ & 7 \\ & \hline \end{aligned}$ | 3.59 | 0.14 | 0.01 | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | 99.85 | $\begin{aligned} & 67.9 \\ & 0 \end{aligned}$ |
|  | BH296 A1 3 | core | $\begin{aligned} & 51.0 \\ & 4 \end{aligned}$ | 0.07 | 29.32 | 0.11 | $\begin{aligned} & 12.8 \\ & 6 \end{aligned}$ | 0.02 | $\begin{aligned} & \hline 0.9 \\ & 7 \end{aligned}$ | 4.32 | 0.14 | 0.01 | $\begin{aligned} & \hline 0.1 \\ & 0 \\ & \hline \end{aligned}$ | 98.94 | $\begin{aligned} & 61.6 \\ & 9 \\ & \hline \end{aligned}$ |
|  | BH296 A1 4 | rimwards | $52.3$ | 0.07 | 29.33 | 0.11 | $\begin{aligned} & 12.7 \\ & 8 \end{aligned}$ | 0.03 | $\begin{aligned} & \hline 0.8 \\ & 9 \\ & \hline \end{aligned}$ | 4.48 | 0.13 | 0.02 | $\begin{aligned} & \hline 0.0 \\ & 7 \end{aligned}$ | $\begin{aligned} & 100.2 \\ & 6 \end{aligned}$ | $\begin{aligned} & \hline 60.7 \\ & 3 \\ & \hline \end{aligned}$ |
|  | BH296 A1 5 | rim | $\begin{aligned} & \hline 52.3 \\ & 3 \\ & \hline \end{aligned}$ | 0.07 | 29.68 | 0.08 | $\begin{aligned} & 13.0 \\ & 4 \\ & \hline \end{aligned}$ | 0.08 | $\begin{aligned} & \hline 1.0 \\ & 0 \\ & \hline \end{aligned}$ | 4.28 | 0.13 | 0.03 | $\begin{aligned} & \hline 0.0 \\ & 8 \\ & \hline \end{aligned}$ | $\begin{aligned} & 100.7 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & 62.2 \\ & 2 \\ & \hline \end{aligned}$ |
|  | BH296 A1 3-5 | average | $\begin{aligned} & 51.9 \\ & 1 \end{aligned}$ | 0.07 | 29.44 | 0.10 | $\begin{aligned} & 12.8 \\ & 9 \end{aligned}$ | 0.04 | $\begin{aligned} & 0.9 \\ & 5 \end{aligned}$ | 4.36 | 0.13 | 0.02 | $\begin{aligned} & \hline 0.0 \\ & 8 \end{aligned}$ | $\begin{aligned} & 100.0 \\ & 0 \end{aligned}$ | $\begin{aligned} & 61.5 \\ & 4 \\ & \hline \end{aligned}$ |
|  | BH296 A1 6 | rep | $\begin{aligned} & 52.7 \\ & 3 \end{aligned}$ | 0.07 | 29.52 | 0.11 | $\begin{aligned} & 12.7 \\ & 7 \\ & \hline \end{aligned}$ | 0.05 | $\begin{aligned} & \hline 0.8 \\ & 7 \end{aligned}$ | 4.49 | 0.13 | 0.00 | $\begin{aligned} & 0.0 \\ & 8 \end{aligned}$ | $\begin{aligned} & 100.8 \\ & 3 \end{aligned}$ | $\begin{aligned} & 60.6 \\ & 5 \\ & \hline \end{aligned}$ |
|  | BH296 A1 7 | core | $53.2$ | 0.04 | 29.15 | 0.05 | $\begin{aligned} & 11.9 \\ & 9 \end{aligned}$ | 0.03 | $\begin{aligned} & \hline 0.5 \\ & 8 \\ & \hline \end{aligned}$ | 4.49 | 0.70 | 0.00 | $\begin{aligned} & 0.0 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & 100.3 \\ & 2 \end{aligned}$ | $\begin{aligned} & 57.2 \\ & 2 \\ & \hline \end{aligned}$ |
|  | BH296 A1 8 | rim | $\begin{aligned} & \hline 52.4 \\ & 4 \\ & \hline \end{aligned}$ | 0.11 | 29.19 | 0.06 | $\begin{aligned} & 12.7 \\ & 0 \\ & \hline \end{aligned}$ | 0.08 | $\begin{aligned} & \hline 1.0 \\ & 4 \end{aligned}$ | 4.45 | 0.13 | 0.00 | $0.0$ | $100.2$ | $60.7$ <br> 1 |
|  | BH296 A1 7-8 | average | $\begin{aligned} & 52.8 \\ & 5 \\ & \hline \end{aligned}$ | 0.07 | 29.17 | 0.06 | $\begin{aligned} & 12.3 \\ & 4 \end{aligned}$ | 0.05 | $\begin{aligned} & \hline 0.8 \\ & 1 \end{aligned}$ | 4.47 | 0.42 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 6 \end{aligned}$ | $\begin{aligned} & 100.2 \\ & 9 \\ & \hline \end{aligned}$ | $\begin{aligned} & 58.9 \\ & 7 \\ & \hline \end{aligned}$ |
|  | BH296 A1 9 | chad rep | $\begin{aligned} & 52.7 \\ & 8 \end{aligned}$ | 0.10 | 29.05 | 0.08 | $\begin{aligned} & 12.3 \\ & 8 \end{aligned}$ | 0.00 | $\begin{aligned} & 1.0 \\ & 7 \end{aligned}$ | 4.65 | 0.17 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | $\begin{aligned} & 100.2 \\ & 6 \end{aligned}$ | $\begin{aligned} & 58.9 \\ & 9 \end{aligned}$ |
|  | BH296 A1 10 | core | $\begin{aligned} & \hline 53.3 \\ & 1 \\ & \hline \end{aligned}$ | 0.09 | 28.09 | 0.06 | $\begin{aligned} & 11.2 \\ & 4 \end{aligned}$ | 0.03 | $\begin{aligned} & \hline 0.8 \\ & 2 \\ & \hline \end{aligned}$ | 5.30 | 0.25 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 99.19 | $\begin{aligned} & 53.2 \\ & 2 \end{aligned}$ |
|  | BH296 A1 11 | rimwards | $54.7$ | 0.09 | 27.82 | 0.05 | $\begin{aligned} & 10.6 \\ & 8 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & \hline 0.7 \\ & 8 \\ & \hline \end{aligned}$ | 5.43 | 0.21 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 8 \\ & \hline \end{aligned}$ | 99.86 | $\begin{aligned} & \hline 51.4 \\ & 5 \\ & \hline \end{aligned}$ |
|  | BH296 A1 12 | rim | $\begin{aligned} & 55.5 \\ & 8 \\ & \hline \end{aligned}$ | 0.03 | 26.90 | 0.05 | 9.81 | 0.00 | $\begin{aligned} & \hline 0.8 \\ & 4 \\ & \hline \end{aligned}$ | 5.88 | 0.31 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 7 \end{aligned}$ | 99.47 | $\begin{aligned} & 47.1 \\ & 3 \end{aligned}$ |
|  | $\begin{aligned} & \text { BH296 A1 10- } \\ & 12 \end{aligned}$ | average | $\begin{aligned} & 54.5 \\ & 4 \\ & \hline \end{aligned}$ | 0.07 | 27.61 | 0.05 | $\begin{aligned} & 10.5 \\ & 8 \end{aligned}$ | 0.01 | $\begin{aligned} & \hline 0.8 \\ & 1 \end{aligned}$ | 5.53 | 0.26 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 5 \end{aligned}$ | 99.51 | $\begin{aligned} & \hline 50.6 \\ & 0 \end{aligned}$ |
|  | BH296 A1 13 | chad core | $\begin{aligned} & 55.2 \\ & 9 \end{aligned}$ | 0.12 | 27.28 | 0.06 | 9.89 | 0.00 | $\begin{aligned} & 0.9 \\ & 2 \end{aligned}$ | 5.89 | 0.27 | 0.01 | $\begin{aligned} & \hline 0.0 \\ & 7 \end{aligned}$ | 99.80 | $47.3$ |
|  | BH296 A1 14 | chad rim | $\begin{aligned} & 55.8 \\ & 6 \end{aligned}$ | 0.09 | 27.07 | 0.05 | 9.59 | 0.03 | $\begin{aligned} & 0.9 \\ & 9 \\ & \hline \end{aligned}$ | 6.21 | 0.27 | 0.00 | $\begin{aligned} & \hline 0.1 \\ & 0 \end{aligned}$ | $\begin{aligned} & 100.2 \\ & 6 \end{aligned}$ | $\begin{aligned} & 45.3 \\ & 5 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH296 A1 13- } \\ & 14 \\ & \hline \end{aligned}$ | average | $\begin{aligned} & 55.5 \\ & 8 \\ & \hline \end{aligned}$ | 0.10 | 27.17 | 0.06 | 9.74 | 0.02 | $\begin{aligned} & 0.9 \\ & 6 \end{aligned}$ | 6.05 | 0.27 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 9 \end{aligned}$ | $\begin{aligned} & 100.0 \\ & 3 \end{aligned}$ | $\begin{aligned} & 46.3 \\ & 6 \end{aligned}$ |
|  | BH296 A1 15 | core | $\begin{aligned} & 52.9 \\ & 4 \end{aligned}$ | 0.03 | 28.74 | 0.06 | $\begin{aligned} & 11.9 \\ & 8 \end{aligned}$ | 0.01 | $\begin{aligned} & \hline 0.8 \\ & 4 \\ & \hline \end{aligned}$ | 4.79 | 0.19 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | 99.62 | $\begin{aligned} & 57.3 \\ & 7 \\ & \hline \end{aligned}$ |
|  | BH296 A1 16 | rimwards | $\begin{aligned} & 53.7 \\ & 9 \end{aligned}$ | 0.01 | 27.97 | 0.06 | $\begin{aligned} & 10.8 \\ & 7 \end{aligned}$ | 0.05 | $\begin{aligned} & 0.7 \\ & 3 \end{aligned}$ | 5.31 | 0.24 | 0.00 | $\begin{aligned} & \hline 0.1 \\ & 5 \end{aligned}$ | 99.16 | $\begin{aligned} & \hline 52.3 \\ & 7 \\ & \hline \end{aligned}$ |
|  | BH296 A1 17 | rimwards | $\begin{aligned} & \hline 54.1 \\ & 8 \\ & \hline \end{aligned}$ | 0.08 | 27.79 | 0.08 | $\begin{aligned} & 10.6 \\ & 2 \end{aligned}$ | 0.01 | $\begin{aligned} & \hline 0.7 \\ & 0 \\ & \hline \end{aligned}$ | 5.60 | 0.20 | 0.05 | $\begin{aligned} & \hline 0.0 \\ & 8 \end{aligned}$ | 99.39 | $\begin{aligned} & 50.5 \\ & 6 \end{aligned}$ |
|  | BH296 A1 18 | rim | 56.0 | 0.08 | 27.15 | 0.04 | 9.62 | 0.00 | 0.8 | 6.02 | 0.28 | 0.02 | 0.0 | 100.1 | 46.1 |




|  |  |  | 7 |  |  |  | 7 |  | 3 |  |  |  | 8 |  | 6 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | MH123 B2 89 | rim | $\begin{aligned} & \hline 50.4 \\ & 5 \\ & \hline \end{aligned}$ | 0.08 | 29.44 | 0.19 | $\begin{aligned} & 13.2 \\ & 1 \end{aligned}$ | 0.04 | $\begin{aligned} & \hline 0.7 \\ & 5 \\ & \hline \end{aligned}$ | 4.08 | 0.10 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 3 \\ & \hline \end{aligned}$ | 98.38 | $\begin{aligned} & \hline 63.7 \\ & 7 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { MH123 B2 88- } \\ & 89 \\ & \hline \end{aligned}$ | average | $\begin{aligned} & 50.2 \\ & 6 \\ & \hline \end{aligned}$ | 0.09 | 29.20 | 0.17 | $\begin{aligned} & 12.9 \\ & 4 \\ & \hline \end{aligned}$ | 0.02 | $\begin{aligned} & \hline 0.7 \\ & 9 \\ & \hline \end{aligned}$ | 4.22 | 0.12 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 97.87 | $\begin{aligned} & \hline 62.4 \\ & 2 \\ & \hline \end{aligned}$ |
|  | MH123 B2 90 | chad rep | $\begin{aligned} & 50.9 \\ & 0 \\ & \hline \end{aligned}$ | 0.07 | 29.72 | 0.19 | $\begin{aligned} & 13.7 \\ & 1 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & 0.8 \\ & 2 \end{aligned}$ | 3.87 | 0.11 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | 99.39 | $\begin{aligned} & 65.8 \\ & 1 \\ & \hline \end{aligned}$ |
|  | MH123 B2 95 | rim | $\begin{aligned} & 52.8 \\ & 4 \\ & \hline \end{aligned}$ | 0.07 | 28.18 | 0.23 | $\begin{aligned} & 11.8 \\ & 3 \\ & \hline \end{aligned}$ | 0.01 | $\begin{aligned} & 0.9 \\ & 6 \\ & \hline \end{aligned}$ | 4.78 | 0.16 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 99.08 | $\begin{aligned} & \hline 57.2 \\ & 5 \end{aligned}$ |
|  | MH123 B2 95 | rimward | $\begin{aligned} & 52.0 \\ & 0 \\ & \hline \end{aligned}$ | 0.02 | 28.53 | 0.18 | $\begin{aligned} & 12.2 \\ & 2 \end{aligned}$ | 0.04 | $\begin{aligned} & \hline 0.8 \\ & 6 \\ & \hline \end{aligned}$ | 4.60 | 0.17 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | 98.64 | $\begin{aligned} & 58.9 \\ & 4 \\ & \hline \end{aligned}$ |
|  | MH123 B2 95 | rimward | $\begin{aligned} & 52.0 \\ & 3 \\ & \hline \end{aligned}$ | 0.12 | 28.88 | 0.15 | $\begin{aligned} & 12.5 \\ & 9 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & 0.8 \\ & 6 \\ & \hline \end{aligned}$ | 4.32 | 0.13 | 0.00 | $\begin{aligned} & \hline 0.1 \\ & 1 \\ & \hline \end{aligned}$ | 99.19 | $\begin{aligned} & 61.2 \\ & 3 \\ & \hline \end{aligned}$ |
|  | MH123 B2 95 | rimward | $\begin{aligned} & 50.9 \\ & 4 \end{aligned}$ | 0.01 | 29.48 | 0.19 | $\begin{aligned} & 13.3 \\ & 0 \\ & \hline \end{aligned}$ | 0.02 | $\begin{aligned} & 0.7 \\ & 8 \end{aligned}$ | 4.10 | 0.13 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 9 \end{aligned}$ | 99.03 | $\begin{aligned} & 63.7 \\ & 5 \\ & \hline \end{aligned}$ |
|  | MH123 B2 95 | core | $\begin{aligned} & 49.4 \\ & 9 \\ & \hline \end{aligned}$ | 0.03 | 29.89 | 0.22 | $\begin{aligned} & 13.8 \\ & 5 \end{aligned}$ | 0.00 | $\begin{aligned} & 0.7 \\ & 7 \end{aligned}$ | 3.72 | 0.10 | 0.00 | $\begin{aligned} & 0.1 \\ & 2 \end{aligned}$ | 98.19 | $\begin{aligned} & 66.9 \\ & 1 \end{aligned}$ |
|  | MH123 B2 95 | core | $\begin{aligned} & \hline 49.8 \\ & 2 \\ & \hline \end{aligned}$ | 0.02 | 30.23 | 0.21 | $\begin{aligned} & 14.3 \\ & 1 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & \hline 0.7 \\ & 3 \\ & \hline \end{aligned}$ | 3.41 | 0.08 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 98.86 | $\begin{aligned} & \hline 69.5 \\ & 2 \\ & \hline \end{aligned}$ |
|  | MH123 B2 95 | rimward | $\begin{aligned} & 48.8 \\ & 5 \\ & \hline \end{aligned}$ | 0.00 | 30.99 | 0.15 | $\begin{aligned} & 14.7 \\ & 0 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & \hline 0.6 \\ & 9 \\ & \hline \end{aligned}$ | 3.26 | 0.07 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 9 \\ & \hline \end{aligned}$ | 98.81 | $\begin{aligned} & 71.0 \\ & 8 \\ & \hline \end{aligned}$ |
|  | MH123 B2 95 | rimward | $\begin{aligned} & 48.4 \\ & 5 \\ & \hline \end{aligned}$ | 0.09 | 31.18 | 0.20 | $15.1$ | 0.04 | $\begin{aligned} & 0.6 \\ & 5 \end{aligned}$ | 3.02 | 0.06 | 0.02 | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 98.92 | $\begin{aligned} & 73.2 \\ & 1 \\ & \hline \end{aligned}$ |
|  | MH123 B2 95 | rimward | $47.7$ | 0.06 | 32.30 | 0.19 | $\begin{aligned} & 15.7 \\ & 3 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & 0.7 \\ & 2 \end{aligned}$ | 2.58 | 0.06 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 6 \end{aligned}$ | 99.43 | $\begin{aligned} & 76.8 \\ & 1 \\ & \hline \end{aligned}$ |
|  | MH123 B2 95 | rim | $47.7$ | 0.08 | 30.65 | 0.46 | $\begin{aligned} & 14.5 \\ & 0 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & 1.5 \\ & 0 \\ & \hline \end{aligned}$ | 3.05 | 0.09 | 0.00 | $\begin{aligned} & 0.0 \\ & 6 \end{aligned}$ | 98.09 | $\begin{aligned} & 72.0 \\ & 5 \\ & \hline \end{aligned}$ |
|  | MH123 B2 95 | x average | $\begin{aligned} & 49.9 \\ & 0 \end{aligned}$ | 0.05 | 30.03 | 0.22 | $\begin{aligned} & 13.8 \\ & 2 \\ & \hline \end{aligned}$ | 0.01 | $\begin{aligned} & 0.8 \\ & 5 \\ & \hline \end{aligned}$ | 3.68 | 0.11 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 6 \\ & \hline \end{aligned}$ | 98.82 | $\begin{aligned} & 67.0 \\ & 7 \\ & \hline \end{aligned}$ |
|  | MH123 B2 97 | core | $\begin{aligned} & 51.4 \\ & 5 \\ & \hline \end{aligned}$ | 0.11 | 29.97 | 0.17 | $\begin{aligned} & 13.6 \\ & 4 \\ & \hline \end{aligned}$ | 0.08 | $\begin{aligned} & 0.8 \\ & 0 \\ & \hline \end{aligned}$ | 3.91 | 0.11 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & 100.2 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 65.4 \\ & 8 \\ & \hline \end{aligned}$ |
|  | MH123 B2 98 | rim | $51.7$ | 0.00 | 30.00 | 0.19 | $\begin{aligned} & 13.8 \\ & 3 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & 0.8 \\ & 4 \end{aligned}$ | 3.75 | 0.10 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 100.5 \\ & 0 \end{aligned}$ | $\begin{aligned} & 66.7 \\ & 0 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { MH123 B2 97- } \\ & 98 \end{aligned}$ | average | $\begin{aligned} & 51.5 \\ & 9 \end{aligned}$ | 0.05 | 29.99 | 0.18 | $\begin{aligned} & 13.7 \\ & 4 \\ & \hline \end{aligned}$ | 0.04 | $\begin{aligned} & \hline 0.8 \\ & 2 \\ & \hline \end{aligned}$ | 3.83 | 0.10 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 4 \end{aligned}$ | $\begin{aligned} & 100.3 \\ & 9 \end{aligned}$ | $\begin{aligned} & 66.0 \\ & 9 \end{aligned}$ |
|  | MH123 B2 100 | chad rep | $\begin{aligned} & 49.2 \\ & 9 \\ & \hline \end{aligned}$ | 0.03 | 30.28 | 0.11 | $\begin{aligned} & 13.9 \\ & 1 \end{aligned}$ | 0.03 | $\begin{aligned} & 0.9 \\ & 9 \\ & \hline \end{aligned}$ | 3.64 | 0.11 | 0.01 | $\begin{aligned} & 0.0 \\ & 6 \end{aligned}$ | 98.45 | $\begin{aligned} & \hline 67.3 \\ & 9 \\ & \hline \end{aligned}$ |
| C | MH129 A2 128 | chad rep | $\begin{aligned} & 50.0 \\ & 7 \\ & \hline \end{aligned}$ | 0.07 | 30.88 | 0.06 | $\begin{aligned} & 14.2 \\ & 2 \end{aligned}$ | 0.00 | $\begin{aligned} & \hline 0.6 \\ & 8 \\ & \hline \end{aligned}$ | 3.56 | 0.10 | 0.00 | $\begin{aligned} & \hline 0.1 \\ & 1 \\ & \hline \end{aligned}$ | 99.76 | $\begin{aligned} & 68.4 \\ & 0 \\ & \hline \end{aligned}$ |
|  | MH129 A2 129 | core | $\begin{aligned} & 48.7 \\ & 9 \end{aligned}$ | 0.00 | 31.05 | 0.07 | $\begin{aligned} & 14.7 \\ & 9 \\ & \hline \end{aligned}$ | 0.01 | $\begin{aligned} & 0.7 \\ & 1 \\ & \hline \end{aligned}$ | 3.23 | 0.09 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | 98.76 | $\begin{aligned} & 71.3 \\ & 1 \\ & \hline \end{aligned}$ |
|  | MH129 A2 130 | rim | $\begin{aligned} & 49.9 \\ & 7 \\ & \hline \end{aligned}$ | 0.04 | 30.82 | 0.09 | $\begin{aligned} & 14.6 \\ & 0 \\ & \hline \end{aligned}$ | 0.02 | $\begin{aligned} & \hline 0.8 \\ & 0 \end{aligned}$ | 3.33 | 0.09 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 99.77 | $\begin{aligned} & 70.4 \\ & 1 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { MH129 A2 } \\ & 129-130 \\ & \hline \end{aligned}$ | average | $49.3$ | 0.02 | 30.94 | 0.08 | $\begin{aligned} & 14.6 \\ & 0 \end{aligned}$ | 0.02 | $0.7$ | 3.28 | 0.09 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 2 \\ & \hline \end{aligned}$ | 99.26 | $\begin{aligned} & 70.8 \\ & 6 \end{aligned}$ |
|  | MH129 A2 131 | chad rep | $\begin{aligned} & 47.4 \\ & 9 \\ & \hline \end{aligned}$ | 0.00 | 30.78 | 0.14 | $\begin{aligned} & 13.9 \\ & 3 \end{aligned}$ | 0.06 | $\begin{aligned} & 0.8 \\ & 7 \end{aligned}$ | 3.60 | 0.12 | 0.00 | $\begin{aligned} & 0.0 \\ & 0 \end{aligned}$ | 96.99 | $\begin{aligned} & 67.6 \\ & 6 \end{aligned}$ |
|  | MH129 A2 135 | rim | $\begin{aligned} & 50.9 \\ & 3 \end{aligned}$ | 0.09 | 28.62 | 0.08 | $\begin{aligned} & 12.0 \\ & 0 \end{aligned}$ | 0.04 | $\begin{aligned} & 0.8 \\ & 5 \\ & \hline \end{aligned}$ | 4.70 | 0.18 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | 97.51 | $\begin{aligned} & 57.9 \\ & 2 \\ & \hline \end{aligned}$ |
|  | MH129 A2 135 | rimward | $\begin{aligned} & 50.5 \\ & 1 \\ & \hline \end{aligned}$ | 0.12 | 29.27 | 0.10 | $\begin{aligned} & 12.7 \\ & 8 \end{aligned}$ | 0.02 | $\begin{aligned} & \hline 0.8 \\ & 4 \end{aligned}$ | 4.24 | 0.14 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 5 \end{aligned}$ | 98.06 | $\begin{aligned} & 62.0 \\ & 0 \end{aligned}$ |
|  | MH129 A2 135 | rimward | $\begin{aligned} & 48.5 \\ & 0 \\ & \hline \end{aligned}$ | 0.06 | 30.23 | 0.10 | $\begin{aligned} & 13.7 \\ & 6 \\ & \hline \end{aligned}$ | 0.08 | $\begin{aligned} & \hline 0.8 \\ & 5 \\ & \hline \end{aligned}$ | 3.79 | 0.12 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 8 \\ & \hline \end{aligned}$ | 97.56 | $\begin{aligned} & \hline 66.2 \\ & 7 \\ & \hline \end{aligned}$ |
|  | MH129 A2 135 | rimward | $46.9$ | 0.00 | 30.73 | 0.09 | $\begin{aligned} & 14.5 \\ & 1 \end{aligned}$ | 0.05 | $\begin{aligned} & 0.8 \\ & 0 \end{aligned}$ | 3.24 | 0.09 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 6 \end{aligned}$ | 96.51 | $\begin{aligned} & 70.8 \\ & 4 \end{aligned}$ |
|  | MH129 A2 135 | core | $\begin{aligned} & 45.9 \\ & 8 \end{aligned}$ | 0.10 | 28.42 | 2.04 | $\begin{aligned} & 12.3 \\ & 9 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & 4.0 \\ & 7 \end{aligned}$ | 2.98 | 0.14 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | 96.12 | $\begin{aligned} & 69.0 \\ & 8 \end{aligned}$ |
|  | MH129 A2 135 | core | $48.1$ | 0.00 | 31.00 | 0.07 | $\begin{aligned} & 14.6 \\ & 8 \end{aligned}$ | 0.00 | $\begin{aligned} & 0.7 \\ & 1 \\ & \hline \end{aligned}$ | 3.18 | 0.08 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 7 \end{aligned}$ | 97.91 | $\begin{aligned} & \hline 71.4 \\ & 7 \\ & \hline \end{aligned}$ |
|  | MH129 A2 135 | rimward | $\begin{aligned} & 48.1 \\ & 9 \\ & \hline \end{aligned}$ | 0.00 | 30.58 | 0.07 | $\begin{aligned} & 13.9 \\ & 9 \end{aligned}$ | 0.00 | $\begin{aligned} & 0.5 \\ & 5 \end{aligned}$ | 3.55 | 0.10 | 0.00 | $\begin{aligned} & 0.0 \\ & 1 \\ & \hline \end{aligned}$ | 97.04 | $\begin{aligned} & 68.1 \\ & 7 \\ & \hline \end{aligned}$ |
|  | MH129 A2 135 | rimward | $\begin{aligned} & 49.1 \\ & 5 \\ & \hline \end{aligned}$ | 0.04 | 29.67 | 0.13 | $\begin{aligned} & 13.2 \\ & 0 \end{aligned}$ | 0.00 | $\begin{aligned} & 0.7 \\ & 2 \\ & \hline \end{aligned}$ | 3.87 | 0.16 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 7 \\ & \hline \end{aligned}$ | 97.00 | $\begin{aligned} & 64.7 \\ & 4 \\ & \hline \end{aligned}$ |
|  | MH129 A2 135 | rimward | $49.0$ $1$ | 0.11 | 30.44 | 0.06 | $\begin{aligned} & 13.6 \\ & 4 \\ & \hline \end{aligned}$ | 0.05 | $\begin{aligned} & 0.6 \\ & 4 \\ & \hline \end{aligned}$ | 3.71 | 0.12 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 9 \end{aligned}$ | 97.87 | $\begin{aligned} & 66.5 \\ & 4 \\ & \hline \end{aligned}$ |
|  | MH129 A2 135 | rim | $\begin{aligned} & \hline 49.5 \\ & 1 \end{aligned}$ | 0.11 | 29.99 | 0.07 | $\begin{aligned} & 13.0 \\ & 4 \end{aligned}$ | 0.00 | $\begin{aligned} & 0.6 \\ & 3 \end{aligned}$ | 4.16 | 0.12 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 8 \end{aligned}$ | 97.70 | $\begin{aligned} & 62.9 \\ & 8 \\ & \hline \end{aligned}$ |
|  | MH129 A2 135 | average | $\begin{aligned} & 48.6 \\ & 8 \\ & \hline \end{aligned}$ | 0.06 | 29.89 | 0.28 | $\begin{aligned} & 13.4 \\ & 0 \\ & \hline \end{aligned}$ | 0.02 | $\begin{aligned} & \hline 1.0 \\ & 7 \\ & \hline \end{aligned}$ | 3.74 | 0.12 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 5 \\ & \hline \end{aligned}$ | 97.33 | $\begin{aligned} & 66.0 \\ & 0 \\ & \hline \end{aligned}$ |
|  | MH129 A2 144 | rim | $\begin{aligned} & 51.1 \\ & 7 \\ & \hline \end{aligned}$ | 0.07 | 30.00 | 0.07 | $\begin{aligned} & 13.0 \\ & 8 \end{aligned}$ | 0.02 | $\begin{aligned} & 0.6 \\ & 2 \end{aligned}$ | 4.18 | 0.20 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 8 \end{aligned}$ | 99.48 | $\begin{aligned} & 62.6 \\ & 2 \\ & \hline \end{aligned}$ |
|  | MH129 A2 144 | rimward | $\begin{aligned} & 50.7 \\ & 3 \end{aligned}$ | 0.09 | 31.02 | 0.08 | $\begin{aligned} & 14.1 \\ & 2 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & 0.5 \\ & 4 \end{aligned}$ | 3.58 | 0.19 | 0.04 | $\begin{aligned} & \hline 0.0 \\ & 5 \end{aligned}$ | $\begin{aligned} & 100.4 \\ & 4 \end{aligned}$ | $\begin{aligned} & 67.7 \\ & 6 \\ & \hline \end{aligned}$ |
|  | MH129 A2 144 | rimward | $\begin{aligned} & 50.2 \\ & 3 \end{aligned}$ | 0.07 | 30.95 | 0.06 | $\begin{aligned} & 14.2 \\ & 1 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & \hline 0.4 \\ & 9 \\ & \hline \end{aligned}$ | 3.57 | 0.15 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 6 \\ & \hline \end{aligned}$ | 99.79 | $\begin{aligned} & \hline 68.1 \\ & 5 \\ & \hline \end{aligned}$ |
|  | MH129 A2 144 | rimward | 51.6 | 0.03 | 29.63 | 0.10 | 12.9 | 0.00 | 0.6 | 4.10 | 0.23 | 0.02 | 0.0 | 99.35 | 62.7 |



|  |  |  | 2 |  |  |  | 7 |  | 2 |  |  |  | 5 | 0 | 3 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | MH129 A2 171 | x average | $\begin{aligned} & 51.3 \\ & 9 \\ & \hline \end{aligned}$ | 0.07 | 30.16 | 0.11 | $\begin{aligned} & 13.2 \\ & 8 \end{aligned}$ | 0.01 | $\begin{aligned} & \hline 0.7 \\ & 7 \end{aligned}$ | 3.94 | 0.24 | 0.01 | $\begin{aligned} & \hline 0.0 \\ & 5 \end{aligned}$ | $\begin{aligned} & 100.0 \\ & 1 \end{aligned}$ | $\begin{aligned} & \hline 64.1 \\ & 2 \\ & \hline \end{aligned}$ |
| B | $\begin{aligned} & \hline \text { BH185 A2_1 } \\ & 41 \end{aligned}$ | rep | $\begin{aligned} & \hline 64.6 \\ & 1 \\ & \hline \end{aligned}$ | 0.00 | 19.10 | 0.03 | 0.41 | 0.00 | $\begin{aligned} & \hline 0.3 \\ & 0 \end{aligned}$ | 0.66 | $\begin{aligned} & \hline 14.7 \\ & 3 \\ & \hline \end{aligned}$ | 0.26 | $\begin{aligned} & \hline 0.1 \\ & 1 \end{aligned}$ | $\begin{aligned} & 100.2 \\ & 0 \\ & \hline \end{aligned}$ | 2.11 |
|  | $\begin{aligned} & \text { BH185 A2_1 } \\ & 44 \end{aligned}$ | rep | $\begin{aligned} & 58.6 \\ & 5 \\ & \hline \end{aligned}$ | 0.00 | 23.26 | 0.26 | 3.17 | 0.00 | $\begin{aligned} & \hline 0.7 \\ & 3 \end{aligned}$ | 1.77 | $\begin{aligned} & 10.8 \\ & 4 \end{aligned}$ | 0.05 | $\begin{aligned} & \hline 0.4 \\ & 8 \\ & \hline \end{aligned}$ | 99.20 | $16.4$ |
|  | $\begin{aligned} & \text { BH185 A2_1 } \\ & 45 \end{aligned}$ | core | $\begin{aligned} & 53.4 \\ & 0 \end{aligned}$ | 0.07 | 29.20 | 0.21 | $\begin{aligned} & 11.6 \\ & 1 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & \hline 0.9 \\ & 2 \end{aligned}$ | 4.25 | 1.06 | 0.00 | $\begin{aligned} & \hline 0.1 \\ & 0 \end{aligned}$ | $\begin{aligned} & 100.8 \\ & 2 \end{aligned}$ | $\begin{aligned} & 56.4 \\ & 9 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_1 } \\ & 46 \end{aligned}$ | rim | $\begin{aligned} & 55.8 \\ & 4 \\ & \hline \end{aligned}$ | 0.06 | 27.62 | 0.11 | $\begin{aligned} & 10.6 \\ & 7 \end{aligned}$ | 0.08 | $\begin{aligned} & \hline 0.8 \\ & 6 \end{aligned}$ | 5.40 | 0.46 | 0.03 | $\begin{aligned} & \hline 0.1 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 101.2 \\ & 4 \end{aligned}$ | $\begin{aligned} & 50.8 \\ & 3 \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_1 } \\ & 45-46 \end{aligned}$ | average | $\begin{aligned} & 54.6 \\ & 2 \\ & \hline \end{aligned}$ | 0.06 | 28.41 | 0.16 | $\begin{aligned} & 11.1 \\ & 4 \\ & \hline \end{aligned}$ | 0.04 | $\begin{aligned} & \hline 0.8 \\ & 9 \end{aligned}$ | 4.82 | 0.76 | 0.02 | $\begin{aligned} & 0.1 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 101.0 \\ & 3 \end{aligned}$ | $\begin{aligned} & 53.6 \\ & 6 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_1 } \\ & 47 \end{aligned}$ | core | $\begin{aligned} & 52.9 \\ & 2 \end{aligned}$ | 0.03 | 29.74 | 0.16 | $\begin{aligned} & 13.4 \\ & 9 \\ & \hline \end{aligned}$ | 0.05 | $\begin{aligned} & \hline 0.8 \\ & 8 \end{aligned}$ | 4.13 | 0.13 | 0.02 | $\begin{aligned} & 0.0 \\ & 6 \end{aligned}$ | $\begin{aligned} & 101.5 \\ & 9 \end{aligned}$ | $\begin{aligned} & 63.8 \\ & 6 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_1 } \\ & 48 \end{aligned}$ | rim | $\begin{aligned} & 55.8 \\ & 0 \\ & \hline \end{aligned}$ | 0.14 | 27.76 | 0.12 | 9.89 | 0.01 | $\begin{aligned} & 0.8 \\ & 9 \end{aligned}$ | 5.41 | 0.97 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 6 \end{aligned}$ | $\begin{aligned} & 101.0 \\ & 4 \\ & \hline \end{aligned}$ | $\begin{aligned} & 47.4 \\ & 6 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_1 } \\ & 47-48 \end{aligned}$ | average | $\begin{aligned} & 54.3 \\ & 6 \\ & \hline \end{aligned}$ | 0.09 | 28.75 | 0.14 | $\begin{aligned} & 11.6 \\ & 9 \end{aligned}$ | 0.03 | $\begin{aligned} & \hline 0.8 \\ & 8 \end{aligned}$ | 4.77 | 0.55 | 0.01 | $\begin{aligned} & \hline 0.0 \\ & 6 \end{aligned}$ | $\begin{aligned} & 101.3 \\ & 1 \end{aligned}$ | $\begin{aligned} & \hline 55.6 \\ & 6 \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_1 } \\ & 52 \end{aligned}$ | core | $\begin{aligned} & 52.1 \\ & 4 \\ & \hline \end{aligned}$ | 0.07 | 29.31 | 0.15 | $\begin{aligned} & 12.9 \\ & 5 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & \hline 0.9 \\ & 5 \end{aligned}$ | 4.44 | 0.13 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 3 \end{aligned}$ | $\begin{aligned} & 100.1 \\ & 7 \end{aligned}$ | $\begin{aligned} & 61.2 \\ & 4 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_1 } \\ & 53 \end{aligned}$ | rim | $\begin{aligned} & 53.4 \\ & 9 \end{aligned}$ | 0.05 | 28.69 | 0.15 | $\begin{aligned} & 12.6 \\ & 3 \end{aligned}$ | 0.00 | $\begin{aligned} & 1.0 \\ & 4 \end{aligned}$ | 4.59 | 0.16 | 0.00 | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | $\begin{aligned} & 100.8 \\ & 4 \end{aligned}$ | $\begin{aligned} & 59.8 \\ & 2 \end{aligned}$ |
|  | $\begin{aligned} & \hline \text { BH185 A2_1 } \\ & 52-53 \\ & \hline \end{aligned}$ | average | $\begin{aligned} & \hline 52.8 \\ & 2 \end{aligned}$ | 0.06 | 29.00 | 0.15 | $\begin{aligned} & 12.7 \\ & 9 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & \hline 0.9 \\ & 9 \end{aligned}$ | 4.51 | 0.14 | 0.00 | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | $\begin{aligned} & 100.5 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 60.5 \\ & 3 \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_2 } \\ & 51 \end{aligned}$ | rep | $54.5$ | 0.00 | 28.23 | 0.09 | $\begin{aligned} & 11.6 \\ & 6 \\ & \hline \end{aligned}$ | 0.02 | $\begin{aligned} & \hline 0.8 \\ & 1 \\ & \hline \end{aligned}$ | 5.00 | 0.19 | 0.01 | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | $\begin{aligned} & 100.5 \\ & 2 \end{aligned}$ | $\begin{aligned} & 55.6 \\ & 9 \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_2 } \\ & 54 \end{aligned}$ | core | $\begin{aligned} & 50.7 \\ & 8 \end{aligned}$ | 0.05 | 30.80 | 0.20 | $\begin{aligned} & 14.5 \\ & 1 \\ & \hline \end{aligned}$ | 0.00 | $\begin{aligned} & \hline 0.7 \\ & 3 \end{aligned}$ | 3.35 | 0.09 | 0.05 | $\begin{aligned} & 0.0 \\ & 2 \end{aligned}$ | $\begin{aligned} & 100.5 \\ & 7 \end{aligned}$ | $\begin{aligned} & 70.1 \\ & 2 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_2 } \\ & 55 \end{aligned}$ | rim | $\begin{aligned} & 51.7 \\ & 9 \\ & \hline \end{aligned}$ | 0.02 | 30.47 | 0.18 | $\begin{aligned} & 14.1 \\ & 0 \\ & \hline \end{aligned}$ | 0.07 | $\begin{aligned} & 0.8 \\ & 2 \end{aligned}$ | 3.68 | 0.14 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 0 \end{aligned}$ | $\begin{aligned} & 101.2 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 67.4 \\ & 0 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_2 } \\ & 54-55 \end{aligned}$ | average | $\begin{aligned} & 51.2 \\ & 8 \\ & \hline \end{aligned}$ | 0.04 | 30.64 | 0.19 | $\begin{aligned} & 14.3 \\ & 0 \\ & \hline \end{aligned}$ | 0.04 | $\begin{aligned} & \hline 0.7 \\ & 7 \\ & \hline \end{aligned}$ | 3.51 | 0.11 | 0.03 | $\begin{aligned} & \hline 0.0 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 100.9 \\ & 2 \\ & \hline \end{aligned}$ | $\begin{aligned} & 68.7 \\ & 6 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH185 A2_2 } \\ & 64 \end{aligned}$ | rep | $\begin{aligned} & \hline 49.0 \\ & 0 \end{aligned}$ | 0.00 | 31.28 | 0.92 | 0.07 | 0.00 | $\begin{aligned} & \hline 3.8 \\ & 0 \end{aligned}$ | 0.04 | 9.87 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 7 \end{aligned}$ | 95.05 | 0.60 |
|  | $\begin{aligned} & \text { BH185 A2_2 } \\ & 68 \end{aligned}$ | ground (rep) | $\begin{aligned} & 53.4 \\ & 9 \\ & \hline \end{aligned}$ | 0.13 | 28.27 | 0.28 | $\begin{aligned} & 11.6 \\ & 7 \\ & \hline \end{aligned}$ | 0.04 | $\begin{aligned} & \hline 1.4 \\ & 0 \end{aligned}$ | 4.92 | 0.20 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 100.3 \\ & 9 \end{aligned}$ | $\begin{aligned} & 56.0 \\ & 9 \end{aligned}$ |
| E | BH366 B1 82 | rep | $\begin{aligned} & 53.2 \\ & 0 \end{aligned}$ | 0.10 | 29.02 | 0.07 | $\begin{aligned} & 12.6 \\ & 5 \end{aligned}$ | 0.00 | $\begin{aligned} & 1.2 \\ & 3 \end{aligned}$ | 4.25 | 0.29 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 3 \\ & \hline \end{aligned}$ | $\begin{aligned} & 100.8 \\ & 2 \end{aligned}$ | $\begin{aligned} & 61.1 \\ & 9 \\ & \hline \end{aligned}$ |
|  | BH366 B1 86 | rep | $\begin{aligned} & \hline 51.1 \\ & 1 \\ & \hline \end{aligned}$ | 0.04 | 31.00 | 0.14 | $\begin{aligned} & 14.6 \\ & 6 \\ & \hline \end{aligned}$ | 0.02 | $\begin{aligned} & 1.0 \\ & 3 \\ & \hline \end{aligned}$ | 3.22 | 0.21 | 0.01 | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | $\begin{aligned} & 101.4 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 70.6 \\ & 7 \\ & \hline \end{aligned}$ |
|  | BH366 B1 87 | rep | $\begin{aligned} & 50.7 \\ & 0 \end{aligned}$ | 0.00 | 31.04 | 0.04 | $14.7$ | 0.01 | $\begin{aligned} & \hline 0.9 \\ & 8 \end{aligned}$ | 3.22 | 0.15 | 0.00 | $\begin{aligned} & 0.0 \\ & 4 \end{aligned}$ | $\begin{aligned} & 100.9 \\ & 5 \end{aligned}$ | $\begin{aligned} & 71.1 \\ & 3 \end{aligned}$ |
|  | BH366 B1 93 | rep | $\begin{aligned} & 50.9 \\ & 5 \end{aligned}$ | 0.09 | 30.82 | 0.11 | $\begin{aligned} & 14.7 \\ & 6 \end{aligned}$ | 0.00 | $\begin{aligned} & 0.9 \\ & 3 \end{aligned}$ | 3.22 | 0.18 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 2 \end{aligned}$ | $\begin{aligned} & 101.0 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 70.9 \\ & 7 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH366 B1 82- } \\ & 93 \end{aligned}$ | rep average | $\begin{aligned} & 51.4 \\ & 9 \\ & \hline \end{aligned}$ | 0.06 | 30.47 | 0.09 | $\begin{aligned} & 14.2 \\ & 1 \end{aligned}$ | 0.01 | $\begin{aligned} & 1.0 \\ & 4 \end{aligned}$ | 3.47 | 0.21 | 0.00 | $\begin{aligned} & 0.0 \\ & 3 \end{aligned}$ | $\begin{aligned} & 101.0 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 68.4 \\ & 9 \\ & \hline \end{aligned}$ |
|  | BH372 A1 111 | core | $\begin{aligned} & 51.3 \\ & 0 \\ & \hline \end{aligned}$ | 0.00 | 29.86 | 0.11 | $\begin{aligned} & 13.6 \\ & 5 \end{aligned}$ | 0.06 | $\begin{aligned} & 0.9 \\ & 3 \end{aligned}$ | 3.85 | 0.12 | 0.01 | $\begin{aligned} & 0.1 \\ & 1 \\ & \hline \end{aligned}$ | $\begin{aligned} & 100.0 \\ & 0 \\ & \hline \end{aligned}$ | $\begin{aligned} & 65.7 \\ & 4 \\ & \hline \end{aligned}$ |
|  | BH372 A1 112 | rimwards | $\begin{aligned} & 51.6 \\ & 2 \end{aligned}$ | 0.07 | 28.94 | 0.30 | 4.99 | 0.01 | $\begin{aligned} & 1.6 \\ & 2 \end{aligned}$ | 2.44 | 6.08 | 0.02 | $\begin{aligned} & \hline 0.1 \\ & 1 \\ & \hline \end{aligned}$ | 96.19 | $\begin{aligned} & 29.9 \\ & 7 \\ & \hline \end{aligned}$ |
|  | BH372 A1 113 | rim | $\begin{aligned} & 56.9 \\ & 6 \\ & \hline \end{aligned}$ | 0.13 | 26.64 | 0.05 | 9.29 | 0.03 | $\begin{aligned} & \hline 0.7 \\ & 8 \\ & \hline \end{aligned}$ | 6.37 | 0.24 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 7 \\ & \hline \end{aligned}$ | $\begin{aligned} & 100.5 \\ & 6 \\ & \hline \end{aligned}$ | $\begin{aligned} & 44.0 \\ & 4 \\ & \hline \end{aligned}$ |
|  | $\begin{aligned} & \text { BH372 A1 111- } \\ & 113 \\ & \hline \end{aligned}$ | average | $\begin{aligned} & 53.2 \\ & 9 \end{aligned}$ | 0.07 | 28.48 | 0.15 | 9.31 | 0.03 | $\begin{aligned} & \hline 1.1 \\ & 1 \end{aligned}$ | 4.22 | 2.14 | 0.01 | $\begin{aligned} & \hline 0.1 \\ & 0 \end{aligned}$ | 98.92 | $\begin{aligned} & 46.5 \\ & 8 \end{aligned}$ |
|  | BH372 A1 118 | rep | $55.2$ | 0.07 | 26.25 | 0.03 | 8.66 | 0.04 | $\begin{aligned} & \hline 0.6 \\ & 2 \end{aligned}$ | 6.60 | 0.25 | 0.00 | $\begin{aligned} & \hline 0.1 \\ & 1 \end{aligned}$ | 97.84 | $\begin{aligned} & 41.4 \\ & 5 \end{aligned}$ |
|  | BH372 A1 119 | core | $52.6$ | 0.10 | 28.53 | 0.23 | 9.46 | 0.00 | $\begin{aligned} & 1.0 \\ & 4 \end{aligned}$ | 4.31 | 1.61 | 0.01 | $\begin{aligned} & \hline 0.0 \\ & 8 \end{aligned}$ | 97.98 | $\begin{aligned} & 49.3 \\ & 4 \\ & \hline \end{aligned}$ |
|  | BH372 A1 120 | rim | $\begin{aligned} & 52.9 \\ & 9 \end{aligned}$ | 0.08 | 27.61 | 0.06 | $\begin{aligned} & 10.4 \\ & 6 \end{aligned}$ | 0.02 | $\begin{aligned} & \hline 0.7 \\ & 8 \end{aligned}$ | 5.56 | 0.20 | 0.00 | $\begin{aligned} & 0.0 \\ & 1 \end{aligned}$ | 97.76 | $\begin{aligned} & 50.3 \\ & 8 \end{aligned}$ |
|  | $\begin{aligned} & \text { BH372 A1 119- } \\ & 120 \end{aligned}$ | average | $\begin{aligned} & 52.8 \\ & 0 \end{aligned}$ | 0.09 | 28.07 | 0.14 | 9.96 | 0.01 | $0.9$ | 4.94 | 0.90 | 0.01 | $\begin{aligned} & 0.0 \\ & 5 \end{aligned}$ | 97.87 | $\begin{aligned} & 49.8 \\ & 6 \\ & \hline \end{aligned}$ |
| A | BH154 A2 68 | chad rep | $\begin{aligned} & 53.1 \\ & 5 \\ & \hline \end{aligned}$ | 0.08 | 31.12 | 0.48 | 0.78 | 0.01 | $\begin{aligned} & 1.0 \\ & 9 \end{aligned}$ | 2.85 | 8.14 | 0.00 | $\begin{aligned} & 0.1 \\ & 5 \end{aligned}$ | 97.86 | 4.99 |
|  | BH154 A2 69 | ground rep | $\begin{aligned} & 52.4 \\ & 5 \end{aligned}$ | 0.08 | 29.53 | 0.14 | $\begin{aligned} & 11.9 \\ & 3 \end{aligned}$ | 0.00 | $\begin{aligned} & 1.2 \\ & 4 \end{aligned}$ | 3.89 | 1.16 | 0.07 | $\begin{aligned} & \hline 0.0 \\ & 6 \end{aligned}$ | $\begin{aligned} & 100.5 \\ & 5 \end{aligned}$ | $\begin{aligned} & 58.5 \\ & 4 \end{aligned}$ |
|  | BH154 A2 79 | ground rep | $\begin{aligned} & \hline 51.5 \\ & 3 \\ & \hline \end{aligned}$ | 0.08 | 29.95 | 0.08 | $\begin{aligned} & 13.5 \\ & 3 \\ & \hline \end{aligned}$ | 0.07 | $\begin{aligned} & \hline 1.0 \\ & 9 \end{aligned}$ | 3.93 | 0.20 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 5 \end{aligned}$ | $\begin{aligned} & 100.4 \\ & 9 \end{aligned}$ | $\begin{aligned} & \hline 64.8 \\ & 3 \\ & \hline \end{aligned}$ |
|  | BH167 A2 54 | chad rep | $\begin{aligned} & 53.1 \\ & 9 \\ & \hline \end{aligned}$ | 0.04 | 29.68 | 0.16 | $\begin{aligned} & 13.2 \\ & 2 \end{aligned}$ | 0.00 | $\begin{aligned} & 1.0 \\ & 1 \end{aligned}$ | 4.29 | 0.14 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 1 \end{aligned}$ | $\begin{aligned} & 101.7 \\ & 3 \end{aligned}$ | $\begin{aligned} & 62.5 \\ & 5 \end{aligned}$ |
|  | BH167 A2 55 | core | $\begin{aligned} & 50.2 \\ & 0 \end{aligned}$ | 0.04 | 31.65 | 0.15 | 1.28 | 0.06 | $\begin{aligned} & 1.4 \\ & 2 \end{aligned}$ | 0.12 | $\begin{aligned} & 11.2 \\ & 5 \end{aligned}$ | 0.00 | $\begin{aligned} & 0.2 \\ & 5 \end{aligned}$ | 96.41 | 8.57 |
|  | BH167 A2 56 | rim | $\begin{aligned} & 55.1 \\ & 3 \\ & \hline \end{aligned}$ | 0.09 | 28.05 | 0.10 | $\begin{aligned} & 11.3 \\ & 0 \end{aligned}$ | 0.00 | $\begin{aligned} & \hline 1.1 \\ & 1 \end{aligned}$ | 5.21 | 0.24 | 0.00 | $\begin{aligned} & \hline 0.0 \\ & 5 \\ & \hline \end{aligned}$ | $\begin{aligned} & 101.2 \\ & 8 \end{aligned}$ | $\begin{aligned} & 53.7 \\ & 8 \\ & \hline \end{aligned}$ |
|  | BH167 A2 55- | average | 52.6 | 0.07 | 29.85 | 0.12 | 6.29 | 0.03 | 1.2 | 2.67 | 5.75 | 0.00 | 0.1 | 98.84 | 31.1 |





Appendix G5 - SFDC sulphur isotope data

| ZONE | $\begin{aligned} & \text { SAMPLE } \\ & \text { ID } \\ & \hline \end{aligned}$ | LITHOLOGY | DISTANCE FROM | $\begin{aligned} & \begin{array}{l} \boldsymbol{\delta}^{34} \mathrm{~S}(\mathrm{VCDT} \\ )^{2} \end{array} \\ & \hline \end{aligned}$ | $\begin{aligned} & \text { S (wt } \\ & \%) \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline \mathbf{N i} \\ & (\mathrm{ppm}) \end{aligned}$ | $\begin{aligned} & \hline \mathbf{C u} \\ & (\mathrm{ppm}) \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | INTRUSIVE CONTACT (m) |  |  |  |  |
| D | MH123A1 | Wynniatt Fm | NA | 7.41 | 0.17 | 16 | 12 |
|  | MH123B1 | Gabbro | NA | 2.98 | 0.077 | 132 | 189 |
|  | MH126A1 | Gabbro | NA | 4.54 | 0.078 | 110 | 166 |
|  | MH127A1 | Wynniatt Fm | NA | 11.47 | 0.30 | b.d.l. | b.d.l. |
|  | MH128A1 | Gabbro | NA | 2.74 | 0.091 | 125 | 161 |
| C |  |  |  |  |  |  |  |
|  |  |  |  |  |  |  |  |
| B | BH188A1 | Wynniatt Fm | NA | 19.55 | 0.18 | 9 | b.d.l. |
|  | BH193A1 | Gabbro | NA | 4.82 | 0.576 | 105 | 266 |
|  | BH195A1 | Gabbro | NA | 12.56 | 0.392 | 117 | 313 |
|  | BH196A1 | Wynniat Fm | NA | 12.74 | 0.27 | b.d.l. | b.d.l. |
|  | BH197A1 | Gabbro | NA | 8.24 | 0.244 | 78 | 178 |
| E | BH366B1 | Gabbro | -0.2 | 9.74 | 0.091 | 112 | 158 |
|  | BH366B1\#2 | Gabbro | -0.2 | 10.39 | 0.091 | 112 | 158 |
|  | BH366B1\#3 | Gabbro | -0.2 | 10.35 | 0.091 | 112 | 158 |
|  | BH371A1 | Kilian Fm | 1.8 | 16.96 | 0.13 | 15 | 8 |
|  | BH369A1 | Kilian Fm | 6.7 | 17.57 | 0.26 | 10 | 5 |
|  | BH368A1 | Kilian Fm | -5.06 | 10.95 | 0.28 | b.d.l. | b.d.l. |
|  | BH367A1 | Kilian Fm | 12.45 | 17.15 | 0.43 | 7 | 9 |
|  | BH365A1 | Kilian Fm | NA | 24.06 | 0.92 | 6 | 27 |
|  | BH371A2 | Kilian Fm | NA |  | 0.27 |  |  |
| A | BH167A4 | Gabbro | -22.7 | 8.86 | 0.076 | 88 | 204 |
|  | BH162A1 | Gabbro | -13.9 | 5.03 | 0.070 | 121 | 162 |
|  | BH162A2 | Gabbro | -4.9 | 8.00 | 0.074 | 96 | 179 |
|  | BH164A1 | Kilian Fm | 9.5 | 19.06 | 0.13 | b.d.l. | 7 |
|  | BH164A2 | Kilain Fm | 13.2 | 21.50 | 0.21 | 15 | 162 |
|  | BH166A1 | Kilian Fm (evaps) | 17 | 31.77 | 17.4 | b.d.l. | 11 |
|  | BH166A1\#2 | Kilian Fm (evaps) | 17 | 31.92 | 17.4 | b.d.l. | 11 |
|  | BH166A2 | Kilian Fm (evaps) | 35 | 32.59 | 17.4 | b.d.l. | b.d.l. |
|  |  |  |  |  |  |  |  |
| *b.d.l. $=$ below detection limit |  |  |  |  |  |  |  |


[^0]:    Advance online articles have been peer reviewed and accepted for publication but have not yet appeared in the paper journal (edited, typeset versions may be posted when available prior to final publication). Advance online articles are citable and establish publication priority; they are indexed by GeoRef from initial publication. Citations to Advance online articles must include the digital object identifier (DOIs) and date of initial publication.

[^1]:    ${ }^{\dagger}$ E-mail: jbedard@ nrcan.gc.ca

