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1	Petrogenesis of the ~2.77 Ga Monts de Cristal Complex, Gabon:					
2	evidence for direct precipitation of Pt- arsenides from basaltic magma					
3						
4	WD Maier, School of Earth and Ocean Sciences, Cardiff University, UK (corresponding					
5	author)					
6	B Rasmussen & I Fletcher, Curtin University, Australia					
7	B Godel, SJ Barnes & LA Fisher CSIRO, Perth, Australia					
8	Shenghong Yang, Department of Geosciences, University of Oulu, Finland					
9	H Huhma & Y Lahaye, GTK, Espoo, Finland					
10						
11	Running title: Petrogenesis of the Monts de Cristal Complex, Gabon					
12						

13 ABSTRACT

14 The Monts de Cristal Complex of Gabon consists of several igneous bodies interpreted to 15 be remnants of a tectonically dismembered, > 100km long and 1-3 km wide, ultramafic-16 mafic intrusion emplaced at 2765-2775 Ma. It is the most significant mafic-ultramafic 17 layered complex yet identified on the Congo craton. The Complex consists largely of 18 orthopyroxenite cumulates, with less abundant olivine-orthopyroxenites and norite, and rare harzburgites and dunites. Mineral compositions (Fool 84, Mg#Opx 85, Anplag 60-68, 19 20 Cr/Fe chromite 1-1.45) and whole rock data suggest that the magma was a low-Ti basalt 21 containing approximately 10% MgO and 0.5% TiO₂. Trace element and Rb-Sr and Sm-22 Nd isotope data indicate the presence of an enriched component, possibly derived from 23 crustal contamination of a magma generated in the sub-lithospheric mantle. Most rocks show a highly unusual pattern of strong Pt enrichment (10-150 ppb) at low concentrations of Pd (1-15 ppb), Au (1-2 ppb), Cu (1-20 ppm), and S (<500 ppm) suggesting that, unlike in most other PGE rich intrusions globally, platinum in the Monts de Cristal Complex is not hosted in magmatic sulfides. Synchrotron X-ray fluorescence mapping revealed the location of buried small Pt particles, most of which are associated with As. We propose that this constitutes some of the strongest evidence yet in support of magmatic crystallization of a Pt-As phase from S undersaturated magma.

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Keywords: Gabon; layered intrusion; platinum-group elements; Pt arsenide; synchrotron
 33

34 INTRODUCTION

35 The Monts de Cristal (MdC) Complex, locally also referred to as the Kinguele Complex, is located in NW Gabon, some 70-150 km to the NE of Libreville, and ~50-100 km north 36 37 of the Equator (Fig. 1a). It forms a tectonically dismembered intrusion, ~100 km in 38 length and mostly ~1-3 km wide (Fig. 1b), constituting the largest mafic-ultramafic 39 layered intrusion so far discovered on the Congo craton. In terms of its inferred shape and 40 the observed lithologies, it bears some resemblance to the PGE mineralized Great Dyke 41 of Zimbabwe. Based in part on these analogies, several companies have conducted 42 exploration for chromite seams and PGE reefs in the MdC during the last decades. After 43 initially obtaining promising exploration indicators, namely elevated Pt concentrations 44 across most of the strike of the Complex, it was ultimately realized that the Pt is not 45 associated with cumulus magmatic sulfide, in contrast to economic PGE reefs elsewhere. 46 However, the strong enrichment of Pt relative to Pd in the MdC is of considerable 47 petrological significance, in that it constitutes the most convincing case yet identified for 48 direct precipitation of Pt phases from basaltic magma. In the present paper, we provide a 49 description and compositional characterization of these unique Pt enriched rocks and 50 suggest a model for their formation.

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52 **REVIEW OF PREVIOUS WORK**

53 The first account of mafic rocks in the MdC area was published by Arsandaux (1912) 54 who reported the presence of ophitic gabbros associated with charnockite to the south of 55 Kinguele. Noritic massifs were first indicated in the Libreville Est geological map of 56 1959 (Aubague and Hausknecht, 1959), but reconnaissance mapping of the intrusion only 57 began in the 1980s when the Direction Générale des Mines et de la Géologie du Gabon, 58 in collaboration with the French Bureau de Recherches Géologiques et Minières, carried 59 out a regional mineral assessment comprising airborne geophysical suveys and stream geochemistry over an area of ~100.000 km². This program identified Archaean granite-60 61 gneiss basement rocks containing banded iron formation, amphibolites as well as 62 gabbronoritic intrusives. The presence of ultramafic intrusives was first indicated in the 63 1:200 000 Kango geological sheet (Kassa-Mombo et al., 1988). Mineral exploration 64 activities initially focused on chromite, but no massive seams were found and the detrital 65 chromite grains turned out to be too Fe- and Al-rich to be of economic interest 66 (Campiglio et al., 1983). Reconnaissance geochronological work indicated an age of ~ 67 2.78 Ga (2777±83 Ma, Rb-Sr on whole rocks, 2783±77 Ma, Pb-Pb on whole rocks, Caen-68 Vachette et al., 1988).

69 In 2002, Southern Era Resources Limited started an exploration program for PGE 70 in the MdC, comprising mapping, comprehensive stream sediment sampling, and soil 71 sampling. Significant Pt anomalies closely associated with all blocks of the intrusion 72 were delineated and a large number of outcrop samples were collected for high-precision 73 geochemistry and geochronology (Maier, 2004). The exploration licence was acquired by 74 Lonmin in 2005 which carried out extensive soil sampling at KG1, KG1 extension and 75 KG3, and remapped some of the bodies (Prendergast, 2009). The company also drilled 15 76 diamond drill holes (13 holes at KG1 and two holes at KG3). In addition, a detailed 77 magnetic-radiometric survey was conducted by Fugro Airborne Surveys in 2008.

78

79 ANALYTICAL METHODS

80 Whole rock and mineral chemistry

81 Seventy five outcrop samples were analysed for major and minor elements as well as a 82 range of lithophile trace elements including Sr, Rb, Nb, Y, Zr, Ni, Cu, Zn, Co, Cr, Sc, V 83 and REE using ICP-MS at Cardiff University, UK, after grinding to <75 µm in agate ring 84 and ball mills. The concentrations of the PGE were determined using ICP-MS after nickel 85 sulfide fire assay and tellurium co-precipitation, on 15 g aliquots. Full details of the 86 analytical procedures used are given in McDonald and Viljoen (2006). The analytical 87 procedure was validated by frequent analysis of reagent blank samples, replicate samples 88 and reference materials (Electronic Appendix 1). All whole rock data are listed in 89 Electronic Appendix 2, and average compositions of rocks types are given in Table 1. In 90 addition to the outcrop samples, Lonmin analysed ~2000 samples from 13 drillcores for 91 Pt, Pd and Au as well as selected major (Fe, Mg, Al, Ca), minor (Ti, Mn) and trace elements (Cr , Ni, Cu, Co, V, Zn, S, and Zr), at Genalysis, Johannesburg, using Pbcollection fire assay followed by mass spectrometry for Pt, Pd and Au, and ICP-OMS for
a range of other major and trace elements. The Pt contents and the Pt/Pd ratios from the
commercial lab show good overlap with the ICP-MS PGE data, suggesting the 2 methods
have broadly similar accuracy (Maier, 2009).

In situ Sr isotope analyses of plagioclase were performed by Laser ablation ICPMS using a Nu Plasma HR multi-collector inductively coupled plasma mass spectrometry
(MC-ICP-MS) and a Photon Machine Analyte G2 laser microprobe at the Geological
Survey of Finland in Espoo. Analytical details are given in Yang et al. (2013), and the
complete data are provided in Electronic Appendix 3.

Whole rock Nd isotopes were also determined at GTK in Espoo. Analytical details are given in Huhma et al. (2012), and data for 5 samples are provided in Electronic Appendix 4.

The compositions of orthopyroxene, olivine, plagioclase and chromite were determined in 5 samples of (olivine)orthopyroxenite, 1 sample of norite, and 2 samples of gabbro from the KG1 and KG3 blocks (Electronic Appendix 5-8). The instrument used was a JEOL JXA-8200 electron microprobe at the University of Oulu, operated at an accelerating voltage of 15 kV and a beam current of 30 nA. The accuracy of analyses was monitored using reference material of similar composition. The reproducibility varied by less than 2 %.

112

113 Synchrotron X-ray fluorescence microscopy

114 X-ray fluorescence microscopy using synchrotron radiation is a highly effective 115 technique for locating ultra-trace high-Z components, owing to the effective penetration 116 of the x-ray beam and the fluoresced target x-rays through sample volumes. A protocol 117 has been developed for searching for micron-scale precious metal (Au, PGE) enriched 118 grains within geological materials (Ryan et al. 2014). X-ray fluorescence microscopy 119 (XFM) was performed on the XFM beamline at the Australian Synchrotron in Melbourne 120 (Paterson et al., 2011) using the Kirkpatrick Baez mirror microprobe end-station. This 121 provides a monochromatic 2 μ m beam-spot size for energies in the range 4-20 keV. 122 Equipped with the Maia 384 detector array, the XFM beamline can acquire images at 2 123 µm resolution over areas of several square centimetres (Kirkham et al., 2010; Ryan et al., 124 2010a) with count rates of \sim 4-10 M/s and energy resolution of 300-400 eV. In this study, 125 large area element maps were collected on 100 micrometer thickness polished thin 126 sections backed by 1millimetre quartz glass slides with a beam energy of 18.5 keV, spot 127 size of 2µm and a dwell times per pixel of 0.97 milli-second to locate areas where Pt-rich 128 inclusions were observed. For each sub-area identified, additional maps were collected 129 with longer dwell time (15.26 millisecond per pixel) to enhance the signal to noise ratio 130 and improve further quantification. Each dataset was analysed with the GeoPIXE 131 software suite which uses a fundamental parameters approach, with spectral 132 deconvolution and imaging using the Dynamic Analysis method (Ryan, 2000; Ryan et 133 al., 2010c) and a detailed model of Maia detector array efficiency (Ryan et al., 2010b). 134 Data were fitted using a yield file which calculates the effects of X-ray absorption by the 135 given matrix or mineral phase (Ryan, 2000). Element maps were produced after spectra 136 were fitted and a dynamic analysis matrix file produced. Platinum abundance was

137 measured using the L α line. These maps were then interrogated to confirm the presence 138 of Pt and identify associated elements and phases. It must be noted that this method 139 samples potentially the entire 100 micrometer thickness of the section, and thus Pt may 140 be sitting on a buried grain boundary even if it appears within a grain. The true nature of 141 the grain location is difficult to resolve without milling down to the grain, but it is 142 possible to further interrogate the spectra to gain a rough quantification of probable depth 143 in the sample to assist with this. The correct identification of Pt has been confirmed by 144 matching XFM maps collected using this methodology with Pt phases identified at the 145 sample surface by optical microscopy in sulfide ore samples (Barnes et al., 2011), 146 although as yet we have not identified any optically visible Pt phases in the Monts de 147 Cristal samples. Identification of Pt phases in Monts de Cristal samples has been 148 confirmed by high resolution PIXE imaging (Barnes et al., in prep).

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150

151 Geochronology

Examination of orthopyroxenite samples Px12, Px39 and 1290 by scanning electron microscope revealed the presence of four U-Pb geochronometers: baddeleyite, zircon, zirconolite and monazite (Electronic Appendix 9). However, zircon grains appeared to have been affected by radiation damage and zirconolite was largely destroyed by secondary alteration, rendering both unsuitable for dating. Baddeleyite and monazite were viable for ion microprobe geochronology but the small size and limited number of grains constrain the data quality. Crystals of both minerals >10 μ m in size were drilled from polished thin sections in ~3 mm platelets and mounted in 25 mm resin disks for ion
microprobe (SHRIMP) analysis.

161 Baddeleyite was analysed using a ~0.15 nA O2- primary ion beam focused on a 162 <10 µm spot. A chip of Phalaborwa (2060 Ma) baddeleyite was mounted with the 163 samples for Pb/U calibration and as a monitor of Pb isotope data quality. Monazite 164 analyses used a $\sim 10 \,\mu\text{m}, 0.25 \,\text{nA}$ primary ion beam. Monazite reference samples (French, 165 PD-95, z2234 and QMa28-1) were in a separate mount that was cleaned and Au-coated with the sample mount. Baddeleyite data acquisition and reduction followed the 166 167 procedures of Wingate et al. (1998). Monazite procedures were those of Fletcher et al. 168 (2010). Primary data reduction was performed using Squid-2 software (Ludwig, 2009), 169 using "spot average" data for all ratios. Corrections to monazite data for matrix effects in 170 Pb/U and Pb/Th, and a small correction for instrumental mass fractionation were applied.

171

172 STRUCTURE AND LITHOLOGIES

173 The MdC is largely hosted by Archaean basement rocks comprising granite-gneiss and 174 greenstones. The basal contacts of the intrusions dip steeply towards the centres of the intrusive blocks (~80°) and are defined by aphanitic margins, grading inwards to 175 176 medium-grained norite, orthopyroxenite and, less abundantly, harzburgite and dunite 177 (Edou Minko et al., 2002). In addition, the Southern Era and Lonmin exploration 178 programs discovered isolated anorthosite layers and several occurrences of ophitic 179 gabbro, the latter located near the margins of the KG1 and KG3 blocks. All mafic-180 ultramafic rocks are generally poorly layered and lack distinct differentiation cycles. The 181 norites appear to form lenses within the orthopyroxenites rather than distinct, laterally

persistent, layers. In addition, norites appear to be relatively common towards the basal contact of the intrusions, e.g., at KG3 (drill core MPD 015) and at KG1 (drill cores MPO-03 and 04). Xenoliths of the country rocks, consisting mainly of feldspar-rich amphibolites as well as less abundant felsic fragments, have been intersected in several boreholes at the KG1 intrusive block.

187 The most comprehensive information on the geology of the Complex is available from KG1, where an extensive soil geochemical program was conducted (Prendergast, 188 189 2009). The location of the grab samples analysed by us, and some of the bore holes 190 drilled by Lonmin in 2007-2008 are indicated in Fig. 2. The KG1 mafic-ultramafic body 191 measures ~ 8 km by 2 km and appears to consist of two distinct blocks (Prendergast, 192 2009; Fig. 2). In the west and north, the intrusion is hosted by felsic gneisses and mafic 193 schists interpreted to represent a granite-gneiss-greenstone terrane. Tonalitic to 194 granodioritic granite gneiss containing intercalations of quartzite, amphibolite and 195 ultramafic slivers occurs to the south and east of the body. Proterozoic clastic and 196 chemical sedimentary rocks occur to the west of an inferred N-NE trending fault, 1-2 km 197 west of KG1. Most contacts of the intrusion with its host rocks are tectonic (Prendergast, 198 2009), but the intrusive basal contact of the body has been intersected by drill cores 199 MPD-03 and 04 (Fig. 2). This intrusive contact consists of a 1-2 m aphanitic rock interpreted as a chilled margin that grades to orthopyroxenite (Electronic Appendix 10). 200 201 Based on the concordance of the strike of layering with the NW flank of the intrusion, 202 Prendergast (2009) proposed that the intrusive contact strikes SW-NE and dips at $\sim 60^{\circ}$ to 203 the SE, and that the total stratigraphic thickness is ~ 1700 m. The dip of layering in most of the remainder of the Complex remains unclear, due to weathering and paucity oflayering.

206 The rocks of the KG1 block are mainly massive orthopyroxenites. Norites appear 207 to be relatively abundant along the south-central margin, whereas olivine pyroxenite and 208 occasional harzburgite predominantly occur in the east, consistent with elevated Ni 209 values in soil sediment surveys. In drill core MPD-03 there is a ~ 400 m basal reversal 210 expressed by a downward decrease in grain size together with an increase in modal 211 plagioclase and biotite, relatively prevalent orthocumulate textures, several noritic 212 intervals, occasional acicular textures, and enclaves of felsic melt rock. In another drill 213 core (MPD-01), relatively fine grained orthopyroxenite overlying the floor contains 214 xenoliths of amphibolite. All segments of the MdC intrusion are cut by abundant 215 amphibolized plagioclase-clinopyroxene-phyric dolerites of unknown age.

216

217 **PETROGRAPHY**

218 The ultramafic rocks are mostly heteradcumulates or mesocumulates with between ~ 5 219 and 15% intercumulus material (Fig. 3a-c). Orthocumulates are less abundant and mainly 220 confined to the stratigraphically lower portions of the intrusion. Considering the Archean 221 age of the Complex, most rocks are remarkably unaltered. Minor to moderate 222 serpentinization and chloritization is observed predominantly in the olivine-bearing 223 lithologies. Orthopyroxene is mostly up to 2 mm in size, commonly of euhedral or 224 subhedral habit (Fig. 3b-c), and aligned sub-parallel to the layering. Olivine forms 225 anhedral grains up to 4 mm in width that can be moderately altered to serpentine and 226 magnetite. Plagioclase occurs as either intercumulus, poikilitic or, in places, cumulus 227 grains. In the orthopyroxenites, chromite tends to be rare, forming occasional clusters and 228 chains of very small (< 0.01mm) euhedral and subhedral crystals included in 229 orthopyroxene (Fig. 3d-e). Chromite is slightly more abundant in olivine pyroxenites, 230 where grains tend to be larger (up to 0.1 mm) and may occur within, or interstitial to, 231 orthopyroxene. Clinopyroxene is relatively rare as an interstitial phase (mostly < 1-2) 232 modal%), but commonly forms exsolution lamellae and blebs in orthopyroxene. 233 Phlogopite is a common accessory phase and may constitute up to 2 modal % in some 234 samples. Hornblende, tremolite and actinolite are accessory phases, largely related to 235 alteration. Sulfides form small grain aggregates up to about 20 microns in size mostly 236 located on orthopyroxene grain boundaries and consist of pyrrhotite, pentlandite and 237 chalcopyrite. Notably, many orthopyroxenes are distinctly zoned showing dark rims (Fig. 238 3c). Synchrotron X-ray fluorescence mapping has shown the rims to be Ti enriched. The 239 absence of residual phases such as apatite, amphibole or K-feldspar implies that the last 240 few percent of trapped liquid was completely expelled from the rock during 241 heteradcumulate formation, at least at the scale of a thin section.

Norites consist mainly of subhedral orthopyroxene and intercumulus or cumulus plagioclase. The latter forms tabular subhedral crystals up to 4 mm in length (Fig. 3f). Clinopyroxene is intercumulus and phlogopite, quartz and alkali feldspar are accessory phases. Chromite is rare, occasionally occurring as clusters of very small grains included in orthopyroxene, similar to the orthopyroxenites. Based on the above observations, the crystallization sequence of the exposed portion of the Monts de Cristal intrusion is ol+cr - opx - opx+pl - opx+pl+cpx. 249 Gabbros are represented by just two samples, one from KG3 (sample Dy18), and 250 the other from KG1 west (sample Hz3). The rocks have a medium-grained ophitic 251 texture, containing tabular, elongated plagioclase (up to 1 mm long) and anhedral or 252 subhedral clinopyroxene (up to 2 mm) (Fig. 3g-h). Olivine is a minor phase, forming 253 mostly relatively small (<1mm) anhedral crystals, but occasionally they occur as larger 254 grains up to 1 cm across. Magnetite forms anhedral aggregates reaching up to 1-5 modal 255 %. Sulfides comprise mainly pyrrhotite and chalcopyrite that form very small trace 256 phases.

The amphibolites are fine to medium grained rocks. Plagioclase forms subhedral tabular crystals up to 2 mm long and < 1 mm wide. Amphibole grains tend to be anhedral or, in some cases, subhedral, forming pseudomorphs of pyroxene. Some samples have granular textures and are finer grained, suggesting metamorphic recrystallisation.

261

262 MINERAL CHEMISTRY

263 In general, the analysed MdC samples show relatively limited mineral compositional 264 variation compared to other layered intrusions. Olivine has mostly 75-84 Fo and 2400-265 3600 ppm Ni, with one sample having 4400 ppm Ni. These Ni contents are higher than 266 those of olivines with equivalent Fo contents in most other layered intrusions (Fig. 4). Orthopyroxene has Mg# 0.78-0.85, 200-1000 ppm Ni, 0.4-0.7% Cr₂O₃, and up to 2% 267 268 Al₂O₃. Cumulus plagioclase has An 67, whereas intercumulus plagioclase has An 60-68. 269 Chromite contains around 40% Cr_2O_3 , and has $Cr/Fe \sim 1$ to 1.4. Norites have 270 approximately similar mineral compositions as the orthopyroxenites.

Two samples of fine-grained ophitic gabbro from the margins of the intrusion have slightly less magnesian clinopyroxene (Mg#0.77-0.8) than the orthopyroxenites and norites. Plagioclase (An63-67) is of similar composition as in the norites, but olivine in the gabbros is significantly more iron-rich (Fo57-58) than in the other lithologies.

275

276 WHOLE ROCK CHEMISTRY

277 Lithophile elements

278 Concentrations of Al₂O₃, TiO₂, Cr, and Ni are plotted vs MgO in Figure 5a-d. The MdC 279 orthopyroxenites and norites have between 15 and 32% MgO, whereas the ophitic 280 gabbros have between 8.5-9% MgO. The concentrations of the major and minor elements 281 are mainly controlled by the modal proportions of orthopyroxene, olivine, and 282 plagioclase. Based on the lever rule, it is estimated that the proportion of olivine to 283 orthopyroxene may reach up to $\sim 20\%$, consistent with petrographic data. The composition of the trapped melt cannot be well constrained, except for TiO₂ where a best 284 285 fit line through the ultramafic rocks indicates TiO₂ levels in the trapped melt of around 286 0.5 wt%. REE patterns of most samples are fractionated (Electronic Appendix 11), 287 showing relative enrichment in LREE over HREE, but $Gd/Lu_N < 1$. Only a few samples 288 show slight positive or negative Eu anomalies. Incompatible trace element patterns show 289 marked enrichment in LILE, positive Th anomalies, and negative anomalies in Nb-Ta and 290 P (Fig. 6), indicating the presence of an enriched component in the rocks.

Trace element contents in the two gabbro samples are more than 5 times those in the orthopyroxenites and norites. Together with their ophitic texture, this suggests that the samples approach liquid compositions, either frozen against the margin of the intrusion or 294 emplaced as dykes or sills. The rocks have 47-48% SiO₂, 8.5% MgO, 12 % Fe₂O₃, 15% 295 Al₂O₃, 0.7-0.8% TiO₂, 2% Na₂O and 0.1-0.2% K₂O (Table 1). They can thus be 296 classified as Al-tholeiites. The multi-element patterns of the gabbros are much flatter and 297 less fractionated than those of the orthopyroxenites and norites (Fig. 6). The two samples 298 analysed have remarkably similar trace element patterns, considering that they were 299 collected from two different intrusive blocks located > 50 km apart. The patterns show 300 only subtle negative Nb-Ta anomalies, reminiscent of many Archean basalts (Barnes et 301 al., 2012). Zirconium contents of the gabbros are ~50ppm, in the range of Yilgarn low-Ti 302 basalts (Barnes et al., 2012), as are Cr and Ni contents (~400 and 200 ppm, respectively).

The amphibolite dykes also have relatively flat trace element patterns (Fig. 6h). The concentrations of most incompatible trace elements are significantly elevated relative to the orthopyroxenites and norites, by an order of magnitude. This is consistent with the amphibolites representing liquids. The shape of the multi-element patterns is broadly similar to those of the gabbros, except that the amphibolites have strong negative Sr anomalies and are relatively more enriched in LILE. Furthermore, the amphibolites have markedly higher total incompatible element concentrations than the gabbros.

310

311 Sulphur and chalcophile elements

The sulphur contents of the MdC cumulates are mostly < 100 ppm, consistent with the paucity of sulphides in the thin sections. Sulphur shows a weak positive correlation with TiO₂ (Fig. 7) suggesting that S behaved as an incompatible element during crystallization. We can place some constraints on the S content of the parent magma by means of the lever rule. The composition of the trapped melt can be delineated by a best fit line through the whole rock compositions of the least evolved orthopyroxenites. Assuming a TiO_2 content of ~0.4-0.5% for the liquid, as indicated by TiO_2 -MgO relationships (Fig. 6b), the S content of the magma is between 100 and 300 ppm. As the S solubility of basaltic magmas at crustal pressure is on the order of 1000-2000 ppm (Mavrogenes and O'Neill, 1999), the MdC magma is likely strongly S undersaturated.

Copper contents of the MdC orthopyroxenites and ol-pyroxenites are mostly between 5 and 40 ppm (Electronic Appendix 12) showing a positive correlation with TiO₂. This indicates that Cu behaved as an incompatible element during fractionation, that the initial liquid (at 0.4-0.5% TiO₂) contained ~50-60 ppm Cu, and that the magma did not equilibrate with sulphide melt during fractionation.

327 For the noble metals, two datasets are available, one being based on 75 outcrop 328 samples collected during the 2004 field season (Fig. 8), whereas the other is based on 329 several 1000 drill core samples analysed during 2007-2008 (Fig. 9-10 and Electronic 330 Appendix 13). The outcrop samples have up to 70 ppb Pt, but mostly <20 ppb Pd, <2ppb 331 Au, <6 ppb Rh, <9 ppb Ru, and <2 ppb Ir. As expected, the PGE contents in the drill cores 332 cover a wider range extending to higher concentrations. The orthopyroxenites contain up 333 to ~ 150 ppb Pt, 25 ppb Pd, 4 ppb Ir, 9 ppb Ru, 6 ppb Rh, and 3 ppb Au. With the 334 exception of Pd and Au, the norites tend to contain lower PGE contents than the 335 orthopyroxenites, i.e., up to ~ 20 ppb Pt, 15 ppb Pd, 1 ppb Ir, 3 ppb Ru, 2 ppb Rh, and 3 336 ppb Au. Pd and Au show weakly developed negative correlations with MgO (Fig. 8a-b), 337 consistent with an incompatible behavior of the two noble metals. In contrast to Pd and 338 Au, the other PGE (Pt, Rh and the IPGE) show positive correlations with MgO (Fig. 8c-339 f), indicating compatible behavior. Compatible behavior of the IPGE is not uncommon in

mafic-ultramafic rocks (Barnes et al., 1985), but compatible behavior of Pt is rare (Park
et al., 2013).

The drill core data show that the highest Pt values in unaltered rock are around 342 343 150 ppb (Fig. 9). These values occur in a specific layer that can be correlated laterally for 344 several 100m. Soil geochemistry (Prendergast, 2009) suggests that beyond that, the layer 345 either pinches out or is buried. Some strongly elevated Pt values are also observed within 346 the 2-20m lateritic cover where Cu and Cr also show anomalously high values (Fig. 10). 347 Notably, the highest Pt stream anomalies, in the centre of the KG1 body, do not correlate 348 with elevated Pt in the soils. Possibly, the streams locally record residual Pt enrichment 349 related to advanced weathering in a pronounced topographic gradient.

350 Of particular note are the high Pt/Pd ratios in the MdC (Pt/Pd mostly > 5), and the 351 lack of correlation between Pt and Pd (Fig. 11b). This pattern is in marked variance to that observed in most other layered intrusions and mafic-ultramafic systems in general 352 353 (Fiorentini et al., 2012; Maier et al., 2013). In contrast, Pt shows well defined positive 354 correlations with the IPGE and Rh (Fig. 11a,c,d). Different blocks and sub-blocks of the 355 MdC define distinct correlations, with KG1 having a relatively Rh and a relatively Pt 356 enriched population, whereas KG1 extension, KG3 and KG4 each define distinct 357 populations.

Highly anomalous Pd values have been found in a single example of a sulfidebearing quartz vein associated with an amphibolite xenolith in drill core MPD 001 (Electronic Appendix 14). The rock contains up to 13.92 ppm Pd, 0.98% Cu, minor Au, 32 ppb Pt and < 10 ppb Rh and Ru. The mantle normalized chalcophile element patterns for the orthopyroxenites and norites from all intrusive blocks (Electronic Appendix 15) show a progressive increase from Ir to Pt, followed by a decreasing trend through Pd and Au to Cu. The ophitic gabbros show very different patterns, with a progressive increase from Ir to Cu, and a small negative Au anomaly. The amphibolites have bowl-shaped patterns with strong PGE depletion relative to Cu and Ni.

368 Ratio plots of Cu/Pd and Cu/Zr (Fig. 12) allow an estimate of the fertility of the 369 magmas. Cu/Pd of the orthopyroxenites and olivine pyroxenites is mostly around or 370 below the level of the primitive mantle (PM), with outliers possibly explained by Pd 371 mobility or localized and temporary sulfide saturation. Cu/Pd of the norites and gabbros 372 tends to be slightly above mantle (\sim 11000), reflecting moderate Pd depletion of the 373 magma. The amphibolites have Cu/Pd significantly above mantle and thus crystallised 374 from a PGE depleted magma that equilibrated with sulfide prior to solidification. Cu/Zr 375 of most samples is 0.5-2, overlapping with, e.g., South African continental flood basalts 376 (Maier et al., 2003), consistent with crystallization of the MdC rocks from a S 377 undersaturated parent magma. Some samples have higher Cu/Zr. They could contain 378 small amounts of cumulus sulfide, consistent with their low Cu/Pd ratios. The PGE 379 depleted amphibolites have undepleted Cu/Zr suggesting that any sulfide that they 380 equilibrated with was of low volume.

381

382 Sr and Nd isotopes

383 Strontium isotope data of plagioclase from two orthopyroxenites, one olivine pyroxenite 384 and one gabbronorite from KG1 and KG3 show remarkable homogeneity between samples as well as between intrusive blocks (Electronic Appendix 3 and 16). The samples have 87 Sr/ 86 Sr_i 0.70255 - 0.70261 and positive ϵ Sr_T around +20, indicating an enriched component.

The four samples analysed for whole rock Nd isotope ratios (Electronic Appendix 4) show ϵ Nd_T -0.6 to 0.3, i.e. little variation between samples and intrusive blocks. As in the case of Sr isotopes, the Nd isotope data indicate the presence of an enriched component. The ophitic gabbro (DY18) has ϵ Nd +1.8, suggesting magma derivation from a depleted mantle source.

393

394 X-RAY FLUORESCENCE MICROSCOPY

X-ray fluorescence microscopy has provided added insight into the crystallization history
of the MdC rocks, allowing location and identification of trace Pt rich phases. Figures 13
and 14 show two orthopyroxenite heteradcumulates containing cumulus orthopyroxene,
approximately 5% plagioclase oikocrysts, minor intercumulus clinopyroxene, and trace
chromite and olivine inclusions within orthopyroxene.

More than a hundred small Pt-rich phases have been found in the five samples mapped, with at least 12 in each sample, suggesting that such phases are quite common. The elevated Pt content across significant core intervals, e.g., 200m with 50-100 ppb Pt in drill core MPD 001 (Electronic Appendix 13) suggests a relatively homogenous distribution of these phases, which in turn suggests that there are few large grains. Such grains would be systematically under-sampled during petrographic analysis (Godel, 2013) and are unlikely to be detected by conventional methods. 407 The Pt phases located in the two samples reported in detail in this study are all 408 less than 10 micrometres in size. The precise location in relation to the host grain 409 boundary is difficult to determine precisely, owing to the unknown shape of the grain 410 boundary in the third dimension, but it appears that all the Pt phases occur at or within a 411 few tens of microns of the margins of orthopyroxene grains. Although the sample size is 412 currently too small to be sure, none of the Pt phases was located within the Cr-zoned, 413 chromite-bearing pyroxene grains, and their occurrence is apparently regardless of 414 whether the host grain has a Ti-enriched rim. The great majority of grains so far 415 identified fall at or close to the margins of orthopyroxene grains enclosed within 416 plagioclase oikocrysts. This implies that the Pt phase or phases crystallised at the 417 cumulus stage, but there is no association with chromite inclusions, as might have been 418 predicted from the model of Finnigan et al. (2008).

419 The nature of the Pt-rich phases remains uncertain; all of the grains located so far 420 lie beneath the surface of the host sample, and none have yet been observed optically or 421 by SEM. Identification is therefore based on the element association obtained from the 422 XFM data. About 50% of Pt grains show an association with As. For example, in sample 423 Px21a there were 2 clusters of Pt grains, one associated with As, the other associated (but 424 not co-located) with Ni and Cu. Approximate estimation of the depth of the grain in the 425 sample in each case, based on variability of the apparent Pt peak height across the Maia 426 detector array (Barnes et al., in preparation) indicates that in most cases Pt and As are co-427 located while the Ni-Cu peaks, attributed to sulfide phases, are slightly offset from Pt and 428 As in 3D. The semi-quantified data indicate that in some, but not all, cases the visible 429 coincident As and Pt spots on the map correspond approximately to sperrylite stoichiometry, PtAs₂. It is likely that at least some of the Pt grains are sperrylite.
However, Pd is not detected by this technique, and neither is Te, although Bi is detectable
and none of the observed grains show any Bi association. Further investigation is ongoing
to increase the sample size of detected grains, and to locate these phases.

The small sulphide grains associated in some cases with the Pt rich phases could have formed in response to reaction of the PGM with S-saturated trapped liquid, or Sbearing hydrous fluid, although there is no evidence for magmatic hydrous phases within these particular samples. The great majority of the 100+ grains so far identified are therefore attributable to direct magmatic precipitation of sperrylite and/or other Ptdominant phases, thought most likely to be Pt-Fe alloys.

440

441 **GEOCHRONOLOGY**

442 Twelve analyses were taken from six baddeleyite grains (Electronic Appendix 17 and 18). The data scatter in ²⁰⁶Pb/U, and most are apparently discordant but this is presumed 443 to be due to grain orientation effects. They are closely grouped in ²⁰⁷Pb/²⁰⁶Pb, giving a 444 445 weighted mean age of 2775 ± 11 Ma. However, there is some excess scatter 446 (MSWD = 2.6). Low-side culling makes little difference to MSWD and high-side culling 447 reduces it to <0.7, implying overculling. Given the modest size of the data set, the variable precision of the ²⁰⁷Pb/²⁰⁶Pb data and the lack of clear justification for any data 448 449 rejection, we use the most conservative option, retaining all data and hence taking the age 450 of baddelevite crystallization to be 2775 ± 11 Ma.

451 Ten analyses were obtained from four monazite grains (Electronic Appendix 19 452 and 20). The majority of the results are apparently 5-6% discordant, however the 453 consistency of the offset from Concordia implies that this is due to a bias in Pb/U,
454 possibly due to an unidentified matrix effect. The variability of minor element contents
455 and lack of Y data make this impossible to assess, particularly for the one strongly
456 discordant point, which is from a very high-Th sample.

457 The data appear to fall into three categories. There are three points that are close 458 to Concordia and may be from concordant samples which have essentially identical ²⁰⁷Pb/²⁰⁶Pb. One other analysis matches this ²⁰⁷Pb/²⁰⁶Pb but is >10% discordant. This 459 analysis had low $CePO_2^+$ counts and very high Th contents, suggesting that the spot 460 461 inadvertently incorporated some huttonite. If this is true, the different (mixed) matrix 462 might explain the apparent discordance. It is also possible that huttonite is metamict and 463 has suffered "recent" Pb loss, and therefore that the analysed volume is genuinely 464 discordant. These four analyses come from four distinct sample grains. The weighted mean ${}^{207}\text{Pb}/{}^{206}\text{Pb}$ dates for all four analyses is 2767 \pm 7 Ma (including uncertainty 465 propagated from the ²⁰⁷Pb/²⁰⁶Pb reference sample PD-95 that was used for mass 466 467 fractionation correction), or 2766 ± 9 Ma if only the three most concordant analyses are 468 used. A fifth analysis is $\sim 2\sigma$ from the main cluster; incorporating it into the weighted 469 mean gives 2765 ± 11 Ma, with only minor excess scatter (MSWD = 1.5).

The other data scatter along a trend that might be a mixing Discordia. The trend is too short and Pb/U too susceptible to matrix-induced errors for a lower intercept to be defined. However, it is almost inescapable to conclude that there has been at least one younger episode of monazite growth or recrystallization. Therefore the monazite gives a minimum age of 2765 ± 11 Ma for the orthopyroxenite, consistent with the baddeleyite date. 476

477 **DISCUSSION**

478 **Parent magma composition**

479 The composition of the MdC parent magma remains difficult to constrain. We have used 480 two semi-quantitative approaches. First, we searched the literature to identify other 481 examples of orthopyroxenite cumulates of broadly similar composition to the MdC, 482 particularly those for which parent magma estimates are available. A relatively good 483 compositional overlap exists with Bushveld orthopyroxenites (Fig. 6, and Electronic 484 Appendix 11). We then simulated the crystallization of Bushveld B1 magma using PELE 485 (Boudreau, 1999). A particularly good fit with the MdC rocks is provided by sample 486 ECBV106 (Barnes et al., 2010: 10.07% MgO) at QFM and 100 bar. The simulation 487 yields spinel as the first mineral to crystallize followed by olivine (Fo85), orthopyroxene 488 (En86, after 10% crystallization) and then plagioclase (An69). Secondly, for one of our 489 best characterized and least evolved samples (Px21) we determined modal proportions by 490 point counting, and then subtracted the amount of MgO hosted by orthopyroxene from 491 the whole rock composition. This method gave 8% MgO for the intercumulus 492 component, but it needs to be borne in mind that the value is potentially affected by 493 residual liquid percolation.

To get a rough idea about the concentration of incompatible minor and trace elements in the trapped melt, we have plotted best fit lines through the ultramafic rocks (Electronic Appendix 21) and recorded the trace element concentration at 10% MgO (TiO₂ 0.4-0.5%, La ~8ppm, Ce ~10 ppm, Sm ~1-1.5 ppm, Th 2 ppm, Zr 40-60ppm, Ta 0.15 ppm, and Y 10-12ppm). Our data thus suggest that the MdC magma was a 499 magnesian basalt with approximately 10% MgO, relatively low TiO₂ contents around 0.5 500 wt%, yet relatively high concentrations of other incompatible trace elements. TiO₂/La ratios are comparable to continental flood-type basalts with similar MgO contents, e.g., 501 502 Ventersdorp and Dominion (Marsh et al., 1989, 1992). The whole rock trace element data 503 can be modeled by up to 20% combined assimilation-fractionation-contamination (AFC) 504 of magnesian basalt (having trace element contents assumed to be 4x primitive mantle, 505 i.e. equivalent to $\sim 25\%$ partial mantle melting) with a 17\% partial melt of MdC country 506 rock granite (average of samples GRA 2W and GRA 2X, Electronic Appendix 2). The 507 composition of the granitic partial melt was calculated by assuming restite modal 508 proportions determined in melting experiments of biotite gneiss at 875° and 3kbar (Patino 509 Douce and Beard, 1995), and D values summarized in Rollinson (2013) (Fig. 15). 510 However, as the model contains a large number of variables, other interpretations are 511 possible, including a magma source in the metasomatised mantle.

512 The Nd and Sr isotope data are broadly consistent with the interpretations from 513 the trace element data. Assuming that the host granites had Sr_i of 0.7013 at 3120±67Ma 514 (Rb/Sr whole rock method, Caen Vachette et al., 1988), and ENd 0.2 (i.e., approximately 515 CHUR_T), and that the MdC magma was a magnesian basalt with 4x PM levels of Rb, Sr, 516 Sm and Nd (Sun and McDonough, 1989), Sr_i 0.701, and εNd 2,1 (DM, De Paolo 1981), 517 the MdC samples ($Sr_i 0.70255-0.70261$, ϵ Nd 0.3 to -0.6) can be modeled by 10-15% bulk 518 contamination with host granite. Unfortunately, Caen Vachette et al. (1988) do not 519 provide detailed information on sampling sites or trace element geochemistry, and their 520 Rb-Sr dates have large uncertainties, which renders the above modeling semi-quantitative at best. To improve the reliability of the model, high-precision geochronology, isotopegeochemistry and trace element data of the host granites are required.

The two ophitic gabbroic samples cannot represent the parent magma to the orthopyroxenites and norites, as the gabbros have distinctly less fractionated trace element patterns and less enriched Nd isotope signatures, as well as depleted PGE contents (Cu/Pd >7000). Whether the gabbros do indeed belong to the same magmatic event as the remainder of the MdC cumulates needs to be confirmed by dating.

528

529 Nature of PGE enrichment and PGE host phases

The Pd levels of most MdC ultramafic samples can be modeled by assuming that the rocks contain 10-30% trapped liquid (broadly consistent with petrography) with 20-25 ppb Pd. Such levels are at the upper end of Pd contents in global basalts (Fiorentini et al., 2010; Maier et al., 2013). Other examples of Pd rich basalts include Finnish 2.45 Ga tholeiitic dykes having up to 30-35 ppb (Guo et al., 2012), and Svecofennian basalts reaching 25 ppb (Barnes et al., 2009).

In contrast to Pd, the measured Pt contents of the MdC cumulates are far too high to be explained by a trapped liquid component. Our global database indicates that basaltic magmas may contain up to ~20 ppb Pt (Maier et al., 2013). A 30% trapped liquid component could thus account for no more than 6-7 ppb Pt. Hydrothermal introduction of Pt or removal of Pd and S after crystallization is considered unlikely, in view of the relatively homogenous nature of the PGE contents in the MdC along ~100km strike, and the generally unaltered nature of the rocks. Localized mobility of Pd is indicated by the development of vein-style mineralization, but this is extremely rare and closelyassociated with amphibolite xenoliths.

A more likely scenario is that the MdC magma had Pt contents in the range of other basalts (15-20 ppb), but reached Pt (or PtAs₂) saturation prior to sulfide saturation. This model is consistent with the synchrotron XFM data reported above and the low S contents of the MdC rocks (Fig. 7). Compatible behavior of Pt has previously been reported from other basaltic suites, including the Pual ridge (Park et al., 2012), where it was assigned to precipitation of Pt alloys.

551 As platinum shows well defined positive correlations with Ir, and to a lesser 552 extent Ru and Rh (Fig. 11), these metals also appear to have been compatible during 553 fractionation. Compatibility of IPGE in mafic-ultramafic rocks has been described in 554 many igneous suites (Barnes et al., 1985, Puchtel and Humayun, 2001). From a crystalchemical point of view, the IPGEs could be hosted by orthopyroxene and olivine, 555 556 substituted for Mg in octahedral coordination, which has been shown to be the case for 557 Ru, Rh and Ir in olivine (Brenan et al., 2003; 2005). Several studies have proposed that 558 in mafic-ultramafic rocks the IPGE are mainly hosted by Os-Ir-Ru alloys and PGM (e.g., 559 Peck and Keays, 1990), but as yet no IPGE enriched PGM could be identified in the 560 MdC. Another possible host for the IPGE is chromite (Locmelis et al., 2011; Pagé et al., 561 2012), but chromite is rare in the MdC and thus unlikely to control IPGE distributions, at 562 least in the form of solid solution. Some of the IPGE and Rh could be hosted in Pt alloys. 563 For example, Pt alloys described by Garuti et al. (2002) have approximately 80% Pt, and 564 1-2% Ir, Ru, and Rh. However, in the MdC we do not see any detectable IPGE associated 565 with the As phases, or anywhere else.

566

567 Factors controlling Pt saturation

PGE levels in basaltic magmas depend largely on the solubility of sulfide liquid, 568 569 platinum-group minerals and PGE alloy during mantle melting and fractional 570 crystallization in the crust. Small to medium degree (<10-20%) melts of asthenospheric 571 mantle are saturated in sulfide (Barnes et al., 1985; Rehkämper et al., 1999; Mungall and 572 Brenan, 2014) and thus depleted in all PGE. Upon dissolution of the mantle sulfides at 573 larger degrees of melting, the concentrations of the individual PGE in the magma depend 574 on whether sulfide is the main PGE host phase or whether, in addition, any PGM or PGE 575 alloy is stable. Palladium is only controlled by sulfide and thus large degree mantle melts 576 such as komatiites have Pd levels 2-3 times that of the primitive mantle (Fiorentini et al., 577 2013). Platinum levels additionally depend on whether Pt alloy is stable, which in turn is 578 controlled by T, fO_2 and depth of melting (Mungall and Brenan, 2014). At relatively low 579 pressure, Pt alloy is stable even at large degree melting, resulting in sub-chondritic Pt/Pd 580 ratios in most asthenospheric mantle magmas (e.g., MORB and komatiites). Increased 581 temperature, fO_2 and melting depth enhance Pt alloy solubility (Borisov and Palme, 1997; 582 Fortenfant et al., 2003; Mungall and Brenan, 2014) leading to a decrease in the bulk 583 partition coefficient of Pt. This should result in relatively high Pt/Pd in the melt, perhaps 584 approaching the chondritic ratio as observed in Bushveld magmas. However, it is notable 585 that global mantle melts mostly show Pt contents between 10-15 ppb (Maier et al., 2009; 586 Fiorentini et al., 2010), with very few examples of higher values that might be expected 587 in large degree melts generated at high pressure. This is likely due to the fact that the Pt levels of the magmas are capped by decreasing Pt solubility during magma ascentresulting in the precipitation of Pt alloys or PGM.

590 Enigmatically, the empirical data indicate much higher Pt solubilities in mafic 591 magmas (10-15 ppb) than the available experimental data. Thus, Ertel et al. (1999) show 592 that Pt solubility in basalt is as low as 3 ppb. Borisov and Palme (1997) proposed higher 593 values, around 15 ppb, but their experiments probably overestimate the Pt solubility, as 594 that work involved bulk analyses in which samples may have been contaminated with 595 undissolved Pt particles. The mismatch between the natural magmas and the experimental 596 data suggests that either the solubility model is wrong, or that other factors play a role. 597 The experiments of Ertel et al. (1999) and Borisov and Palme (1997) were conducted on 598 synthetic basalts which contained no Fe or S. Fe addition should result in the formation of 599 Pt alloys and thus relatively lower Pt solubility (Borisov and Palme, 2000). However, 600 Laurenz et al. (2013) have shown that the presence of sulfur in magmas significantly 601 increases Pd and Ru solubility, and Mungall and Brenan (2014) have suggested a similar 602 effect on Pt.

603 Of particular interest is the origin of the Pt enriched layer of the MdC (Fig. 10, 604 and Electronic Appendix 13). The Pt rich samples are heteradcumulates that largely lack 605 chromite and olivine. They have the highest Mg#, i.e., are the least evolved, but in terms 606 of petrography, isotopes and incompatible trace element ratios they are compositionally 607 indistinguishable from the other rocks. Due to the small size of the Pt phases located, 608 gravitational concentration from convecting magma seems unlikely, unless the Pt phases 609 nucleated on the surface of cumulus orthopyroxene grains and were mechanically 610 concentrated with them. The Pt solubility could have been temporarily lowered due to 611 reduction of the magma in response to contamination, but not only would this potentially 612 have triggered sulfide saturation (Tomkins et al., 2012), but it is also inconsistent with the 613 V contents of the rocks; The experimental data of Toplis and Corgne (1992) have shown 614 that partitioning of V into pyroxene increases with falling O fugacity. Thus, one would 615 expect an increase in V to accompany the Pt enrichment, contrary to the observations 616 from the drillcore data that show inflections towards low V contents in the Pt rich layer. 617 Furthermore, there is no lithological evidence such as country rock xenoliths that could 618 suggest local magma contamination in the Pt enriched horizon. Localised reduction due 619 to chromite crystallization (Finnigan et al., 2008) can be ruled out because the most Pt 620 rich samples are chromite poor, and no Pt phases were observed by XFM mapping in 621 contact with chromite grains. The available evidence suggests that the formation of the Pt 622 enriched layer is related to magma replenishment with relatively unevolved magma, 623 consistent with a subtle increase in Mg# and a decrease in V and Ti of the rocks.

An additional factor suggested by our observations is that arsenic plays a role in the stability of magmatic Pt phases. The evidence for primary precipitation of Pt-As rich phases, thought to be sperrylite, suggest that the key factor may be the magma exceeding the solubility product for sperrylite. If Pt is dissolved in the magma as PtS, as implied by the experimental data of Mungall and Brenan (2014), then the following reaction may be operating:

630 $PtS(melt) + FeAs_2(melt) = PtAs_2(solid) + FeS(melt)$

631 with the reaction being driven by the increasing activity coefficient of PtS with 632 falling temperature along a given fO_2 buffer implied by the Mungall and Brenan 633 experiments. This conjecture requires testing by investigation of solution properties of Pt in the presence of As in silicate melts, which are currently unknown. The divergence of
Pt/Pd ratio noted in a number of layered intrusions and mafic lava suites (e.g., Stillwater,
Keays et al., 2012; Deccan Trap lavas, Keays and Lightfoot, 2010) may be explained by
preferential fractionation of magmatic Pt arsenides.

638 Unfortunately, no arsenic data are available to evaluate whether the elevated Pt 639 contents in the MdC rocks could be due to unusually high As concentrations in the 640 magma. However, assuming that all of the Pt occurs as $PtAs_2$, with overall Pt levels at 641 ~20-150 ppb, this would require ~15-100 ppb As for charge balance. This is in the range 642 of As contents reported by Jenner and O'Neill (2012) for MORB in the 8-10 wt% MgO 643 range (~125 ppb As). Hence, the required As contents implied by the Pt contents of the 644 rocks are not exceptional, adding support to the notion that Pt-arsenide might be more 645 significant in S-poor rocks than previously supposed (J Brenan, written comment, Feb. 2015). 646

647

648 CONCLUSIONS

649 The Monts de Cristal Complex of Gabon forms a large layered intrusion, approximately

650 100km long and 1-3 km wide. It consists predominantly of orthopyroxenite that

651 crystallised from a low-Ti basalt with approximately 10% MgO. The Complex is

characterised by strong enrichment in Pt (up to 150 ppb) relative to Pd (<20 ppb) and

other chalcophile elements and sulphur. X-Ray fluorescence microscopy has identified

numerous small (<10 micrometers) Pt rich phases, mostly arsenides. This is interpreted to

655 indicate precipitation of Pt-arsenides from S undersaturated magma, constituting some of

the best evidence yet for direct crystallization of platinum- group minerals from basalticmagma.

658

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673 **REFERENCES**

- Arsandaux, H. (1912). Sur la presence au Gabon de roches appartenant a la serie de la
- 675 charnockite. *Comptes Rendus Academie Sciences*, Paris, **154**, 896-898.
- 676
- 677 Aubague. M. & Hausknecht, J.J. (1959). Carte geologique de reconnaissance au l/500
- 678 000, feuille Libreville Est. Carte et notice explicative. Direction des Mines et de la
- 679 Geologie de l'AEF, Brazzaville, Congo, 34p.
- 680

681	Ballhaus, C., Berry, R.F. & Green, D.H. (1991). High pressure experimental calibration
682	of the olivine-orthopyroxene-spinel oxygen geobarometer: implications for the oxidation
683	state of the upper mantle. Contributions to Mineralogy and Petrology 107, pp 27-40
684	
685	Barnes, Sarah-Jane & Maier, W.D. (1999). The fractionation of Ni, Cu and the noble
686	metals in silicate and sulphide liquids, in Keays, R.R., Lesher, C.M., Lightfoot, P.C. and
687	Farrow, C.E.G., eds., Dynamic processes in magmatic ore deposits and their application
688	to mineral exploration: Geological Association Canada, Short Course Notes, 13, 69-106.
689	
690	Barnes, Sarah-Jane, Naldrett, A.J. & Gorton, M.P. (1985). The origin of the fractionation
691	of platinum-group elements in terrestrial magmas. Chemical Geology 53, 303-323.
692	
693	Barnes, Sarah-Jane, Maier, W.D. & Curl, E.A. (2010). Composition of the Marginal
694	rocks aand sills of the Rustenburg layered suite, Bushveld Complex, South Africa:
695	Implications for the formation of platinum-group element deposits. Economic Geology
696	105 , 1491-1511.
697	
698	Barnes, S.J. (2007). Cotectic precipitation of olivine and sulfide liquid from komatiite
699	magma and the origin of komatiite-hosted disseminated nickel sulfide mineralization at
700	Mount Keith and Yakabindie, Western Australia. Economic Geology 106, 298-304.
701	
702	Barnes, S.J. & Naldrett A.J. (1984). Geochemistry of the J-M Reef of the Stillwater
703	Complex, Minneapolis Adit area II. Slicate Mineral Chemistry and Petrogenesis. Journal
704	of Petrology, 27 , 791-825.
705	
706	Barnes, S.J., Makkonen, H.V., Dowling, S.E., Hill, R.E.T. & Peltonen, P. (2009). The
707	1.88 Ga Kotalahti and Vammala nickel belts, Finland: geochemistry of the mafic and
708	ultramafic metavolcanic rocks. Bulletin of the Geological Society of Finland 81, 103-
709	141.
710	

711	Barnes, S.J., Osborne, G.A., Cook, D., Barnes, L., Maier, W.D. & Godel, B. (2011), The				
712	Santa Rita Nickel sulfide deposit in the Fazenda Mirabela Intrusion, Bahia, Brazil:				
713	Geology, sulfide geochemistry and genesis. Economic Geology 106, 1083-1110.				
714					
715	Barnes, S. J., Van Kranendonk, M. J., and Sonntag, I., 2012, Geochemistry and tectonic				
716	setting of basalts from the Eastern Goldfields Superterrane: Australian Journal of Earth				
717	Sciences 59 , 707-735.				
718					
719	Borisov, A. & Palme, H. (1997), Experimental determination of the solubility of				
720	platinum in silicate melts. Geochimica et Cosmochimica Acta 20, 4349-4357.				
721					
722	Borisov A., Palme H. (2000) Solubilities of noble metals in Fe-containing silicate melts				
723	as derived from experiments in Fe-free system. American Mineralogist. 85 (11-12), pp.				
724	1665-1673				
725					
726	Boudreau, A.E. (1999), PELE-a version of the MELTS software programme for the PC				
727	platform. Computers and Geosciences 25, 201-203.				
728					
729	Bowles, J.F.W., Prichard, H.M., Suarez, S. & Fisher, P.C. (2013). The first report of				
730	PGM in magnetite-bearing gabbro, Freetown layered complex, Sierra Leone:				
731	Occurrences and genesis. Canadian Mineralogist 51, 455-473.				
732					
733	Brenan J. M., McDonough W. F. and Dalpe' C. (2003) Experimental constraints on the				
734	partitioning of rhenium and some platinum-group elements between olivine and silicate				
735	melt. Earth Planet. Sci. Lett. 212, 135–150.				
736					
737	Brenan J. M., McDonough W. F. and Ash R. (2005) An experimental study of the				
738	solubility and partitioning of iridium, osmium and gold between olivine and silicate melt.				
739	Earth Planet. Sci. Lett. 237, 855–872.				
740					

741	Caen-Vachette, M., Vialette, Y., Basset, J.P. & Vidal, P. (1988). Apport de la				
742	geochronologie isotopique a la connaissance de la geologic gabonaise. Chronique				
743	Recherche Miniere 491, 35-53.				
744					
745	Campbell, I. H. & Griffiths, R. W. (1992). The changing nature of mantle hotspots				
746	through time: implications for the chemical evolution of the mantle. Journal of Geology				
747	92 , 497-523.				
748					
749	Campiglio, C. (1987). Perspectives sur la recherche de platinoides dans le complexe des				
750	Monts de Cristal, Gabon. In: Rapport d'Execution de la Convention Concernant la				
751	Cellule d'orientation de l'Inventaire Minier Gabonais, 1986-1987. Armines-CGGM,				
752	France, 31-36.				
753					
754	Campiglio, C., Marion, C. & Vannier, M. (1983). Le complexe mafique-ultramafique de				
755	M'Bilan. Etude des premiers echantillons de roches et de concentres alluvionnaires. In:				
756	Annexes, Rapport Cellule d'orientation de l'Inventaire Minier, CGGM. Direction des				
757	Mines et de la Geologie Miniere, Libreville, Gabon, 22p.				
758					
759	Chevalier, L., Makanga, J.F. & Thomas, R.J. (2002). Carte geologique de la Republique				
760	Gabonaise. Notice explicative. Council for Geoscience, Pretoria, 195p.				
761					
762	Creighton, S., Stachel, T., Eichenberg, D. & Luth, R.W. (2010). Oxidation state of the				
763	lithospheric mantle beneath Diavik diamond mine, central Slave craton, NWT, Canada.				
764	Contributions to Mineralogy and Petrology 159, 645-657.				
765					
766	DePaolo, D.J. (1981). Neodymium isotopes in the Colorado Front Range and				
767	crust-mantle evolution in the Proterozoic. Nature 291, 684-687.				
768					
769	Edou-Minko, A., Grandin, G. & Campiglio, C. (2002). Petrologie et geomorphologie				
770	dans la region de Kango, Gabon: un grand dyke ultramafique-mafique archeen. Journal				

771 of African Earth Sciences **32**, 899-918.

772						
773	Ertel W., O'Neill H., St C., Sylvester P. J. and Dingwell D. B. (1999) Solubilities of Pt					
774	and Rh in a haplobasaltic melt at 1300 °C. Geochim. Cosmochim. Acta 63, 2439–2449.					
775						
776	Finnigan, C.S., Brenan, J.M., Mungall, J.E. & McDonough, W.F. (2008). Experiments					
777	and Models Bearing on the Role of Chromite as a Collector of Platinum Group Minerals					
778	by Local Reduction. Journal of Petrology 49, 1647-1665.					
779						
780	Fiorentini ML, Barnes SJ, Lesher CM, Heggie, GJ, Keays RR, Burnham OM (2010)					
781	Platinum Group Element Geochemistry of Mineralized and nonmineralized Komatiites					
782	and Basalts. <i>Economic Geology</i> , 105 , 795-823.					
783						
784	Fiorentini, M.L., Barnes, S.J., Maier, W.D. & Heggie, G.J. (2012). Global variability in					
785	the PGE contents of komatiites, Journal of Petrology 52, 82-112.					
786						
787	Fletcher, I.R., McNaughton, N.J., Davis, W.J., Rasmussen, B., 2010. Matrix effects and					
788	calibration limitations in ion probe U-Pb and Th-Pb dating of monazite. Chemical					
789	<i>Geology</i> 270 , 31–44.					
790						
791	Fortenfant, S.S., Gunther, D., Dingwell, D.B. & Rubie, D.C. (2003). Temperature					
792	dependence of Pt and Rh solubilities in a haplobasaltic melt. Geochimica et					
793	Cosmochimica Acta 67, 123-131.					
794						
795	Garuti, G., Pushkarev, E.V. & Zaccarini, F. (2002). Composition and paragenesis of Pt					
796	alloys from chromitites of the Uralian-Alaskan-type Kytlym and Uktus complexes,					
797	northern and central Urals, Russia. Canadian Mineralogist 40, 1127-1146.					
798						
799	Godel, B. (2013). High-Resolution X-Ray Computed Tomography and Its Application to					
800	Ore Deposits: From Data Acquisition to Quantitative Three-Dimensional Measurements					
801	with Case Studies from Ni-Cu-PGE Deposits. Economic Geology 108, 2005-2019.					
802						

803	Harris, C.	, Pronost, J.J.M.,	, Ashwal, L.D. &	z Cawthorn, R.G.	. (2004).	Oxygen and
		, ,	, ,			- 10

- 804 hydrogen isotope stratigraphy of the Rustenburg layered suite, Bushveld Complex:
- 805 constraints on crustal contamination. *Journal of Petrology* **46**, 579–601.
- 806
- 807 Huhma, H., Kontinen, A., Mikkola, P., halkoaho, T., Hokkanen, T., Hölttä, P., Juopperi,
- 808 H., Konnunaho, J., Luukkonen, E., Mutanen, T., Peltonen, P., Pietikäinen, K.,
- 809 Pulkkinen, A., 2012. Sm–Nd results on the Archean crust in Finland. In: Hölttä,
- 810 P. (Ed.), The Archean of the Karelia Province in Finland. Geological Survey of
- 811 Finland, Special Paper 54, pp. 176–213.
- 812
- 813 Irvine, T.N. (1970). Crystallization sequences in magmas of the Muskox intrusion and
- 814 some other layered intrusions. Geological Society of South Africa, Special Publication 1,
- 815 441-476.
- 816
- 817 Irvine, T.N. (1980). Magmatic infiltration metasomatism, double-diffusive fractional
- 818 crystallisation, and adcumulus growth in the Muskox Intrusion and other layered
- 819 intrusions. In: R.B. Hargraves (Editor), Physics of Magmatic Processes. Princeton
- 820 University Press, Princeton, N.J., 325-383.
- 821
- S22 Jacobsen, S.B. & Wasserburg, G.J. (1980). Sm-Nd isotopic evolution of chondrites.
- 823 Earth and Planetary Science Letters **50**, 139–155.
- 824
- 825 Kassa-Mombo, C., Minko-Bidza, V., Mikhailoff, N., Greenish, T., Devos, W., Boldrachi,
- 826 R., Degroote, V., Bonnot, H., Adlof, L., Mabiala, R., Melchior, A. & Abeme, M.J.
- 827 (1988). Apport de l'inventaire aux connaissances geologiques et minieres des feuilles a
- 828 l/200.000 de Libreville, Kango et Lambarene (Gabon). Chronique Recherche Miniere
- **491**, 105-124.
- 830
- 831 Keays, R.R., Lightfoot, 2010. Crustal sulfur is required to form magmatic Ni-Cu sulfide
- 832 deposits; evidence from chalcophile element signatures of Siberian and Deccan Trap
- 833 basalts. *Mineralium Deposita*, **45**, 241.
| 834 | |
|-----|---|
| 835 | Keays, R.R., Lightfoot, P.C., Hamlyn, P.R., 2012. Sulfide saturation history of the |
| 836 | Stillwater Complex, Montana: chemostratigraphic variation in platinum group elements. |
| 837 | Mineralium Deposita, 47, 151-173. |
| 838 | |
| 839 | Kirkham, R., Dunn, P. A., Kucziewski, A., Siddons, D. P., Dodanwela, R., Moorhead, G., |
| 840 | Ryan, C. G., De Geronimo, G., Beuttenmuller, R., Pinelli, D., Pfeffer, M., Davey, P., |
| 841 | Jensen, M., Paterson, D., de Jonge, M. D., Kusel, M. & McKinlay, J. (2010). The Maia |
| 842 | Spectroscopy Detector System: Engineering for Integrated Pulse Capture, Low-Latency |
| 843 | Scanning and Real-Time Processing. AIP Conference Series, 1234, 240-243. |
| 844 | |
| 845 | |
| 846 | Laurenz V, Fonseca ROC, Ballhaus C, Jochum KP, Heuser A, Sylvester PJ (2013) The |
| 847 | solubility of palladium and ruthenium in picritic melts: 2. The effect of sulfur. Geochim. |
| 848 | Cosmochim. Acta, 108, 172-183 |
| 849 | |
| 850 | Li, C. & Ripley, E.M. (2009). Sulphur contents at sulfide-liquid or anhydrite saturation in |
| 851 | silicate melts: empirical equations and example applications. <i>Economic Geology</i> 104, |
| 852 | 405-412. |
| 853 | |
| 854 | Lightfoot, P.C. & Naldrett, A.J. (1983). The geology of the Tabankulu section |
| 855 | of the Insizwa Complex, Transkei, southern Africa, with reference to the nickel sulphide |
| 856 | potential. Transactions of the Geological Society of South Africa, 86, 69–187. |
| 857 | |
| 858 | Lightfoot, P.C., Naldrett, A.J. & Hawkesworth, C.J. (1984). The geology and |
| 859 | geochemistry of the Waterfall Gorge Section of the Insizwa Complex with particular |
| 860 | reference to the origin of the nickel sulphide deposits. Economic Geology, 79, 1857- |
| 861 | 1879. |
| 862 | |
| 863 | Locmelis, M., Pearson, N.J., Barnes, S.J. & Fiorentini, M.I. (2011). Ruthenium in |
| 864 | komatiitic chromite. Geochimica et Cosmochimica Acta 75, 3645-3661. |

865	
866	Ludwig, K.R. (2003). User's Manual for Isoplot/Ex version 3.00-A Geochronology
867	Toolkit for Microsoft Excel, No. 4 - Berkeley Geochronological Center, Special
868	Publication
869	
870	Maier WD (2004). Monts de Cristal PGE project. Internal Report 2, SouthernEra
871	Resources Ltd.
872	
873	Maier WD (2009). Report on PGE reef potential of the Monts de Cristal PGE project,
874	Gabon, Lonmin Plc
875	
876	Maier, W.D. & Eales, H.V. (1997). Correlation within the UG2–Merensky Reef interval
877	of the Western Bushveld Complex, based on geochemical, mineralogical and petrological
878	data. Geological Survey of South Africa Bulletin 120, 56 p.
879	
880	Maier, W.D., Groves, D.I. (2012). Temporal and spatial controls on the formation of
881	magmatic PGE and Ni-Cu deposits. <i>Mineralium Deposita</i> 46, 841-857.
882	
883	Maier, W.D., Barnes, SJ. & Marsh, J.S. (2003). The concentrations of the noble metals
884	in Southern African flood-type basalts and MORB: implications for petrogenesis and
885	sulphide exploration. Contributions to Mineralogy and Petrology 146, 44-61.
886	
887	Maier, W.D., Gomwe, T., Barnes, SJ., Li, C. & Theart, H. (2004). Platinum-group
888	elements in the Uitkomst Complex, South Africa. Economic Geology 99, 499-516.
889	
890	Maier, W.D., Barnes, SJ. & Groves, D.I. (2013). The Bushveld Complex, South Africa:
891	formation of platinum- palladium, chrome- and vanadium-rich layers via hydrodynamic
892	sorting of a mobilized cumulate slurry in a large, relatively slowly cooling, subsiding
893	magma chamber. Mineralium Deposita 48, 1–56.
894	

895	Marsh JS, Bowen MP, Rogers NW, Bowen TB (1989) Volcanic rocks of the
896	Witwatersrand Triad, South Africa, I: Petrogenesis of mafic and felsic rocks of the
897	Dominion Group. Precambrian Research 44, 39-65
898	
899	Marsh JS, Bowen MP, Rogers NW, Bowen TB (1992) Petrogenesis of Late Archean
900	flood-type basic lavas from the Klipriviersberg group, Ventersdorp Supergroup, South
901	Africa. Journal of Petrology 33, 817–847
902	
903	Martin, H., Smithies, R.H., Rapp, R., Moyen, JF. & Champion, O. (2005). An overview
904	of adakite, tonalite-trondhjemite-granodiorite (TTG), and sanukitoid: relationships and
905	some implications for crustal evolution. Lithos 79, 1–24.
906	
907	Mavrogenes, J. A. & O'Neill, H.St.C. (1999). The relative effects of pressure,
908	temperature and oxygen fugacity on the solubility of sulfide in mafic magmas.
909	Geochimica et Cosmochimica Acta 63, 1173–1180.
910	
911	McDonald, I. & Viljoen, K.S. (2006). Platinum-group elemeent geochemistry of mantle
912	eclogites: a reconnaissance study of xenoliths from the Orapa kimberlite, Botswana.
913	Transactions Institution Mining and Metallurgy 115, 81-93.
914	
915	McDonough, W.F. & Sun, Ss. (1995). The composition of the Earth. Chemical Geology
916	120 , 223-253.
917	
918	McInnes, B.I.A., McBride, J.S., Evans, N.J., Lambert, D.D. & Andrew, A.S. (1999).
919	Osmium isotope constraints on metal recycling in subduction zones. Science 286, 512-
920	516.
921	
922	Merkle, R.K.W. (1992). Platinum-group minerals in the middle group of chromitite
923	layers at Marikana, western Bushveld Complex: indications for collection mechanisms
924	and postmagmatic modification. Canadian Journal of Earth Sciences 29, 209-221.
925	

- 926 Müller, W., Shelley, M., Miller, P. & Broude, S. (2009). Initial performance metrics of a
- 927 new custom-designed ArF excimer LA-ICPMS system coupled to a two-volume laser-
- ablation cell. *Journal Anal Atom Spectrom* **24**, 209–214.
- 929
- 930 Mungall, J.E. & Brenan, J.M. (2014). Partitioning of platinum-group elements and Au
- between sulfide liquid and basalt and the origins of mantle-crust fractionation of the
- 932 chalcophile elements. *Geochimica et Cosmochimica Acta* **125**, 265–289.
- 933
- 934 Nebel, O., Scherer, E.E. & Mezger, K. (2011). Evaluation of the ⁸⁷Rb decay constant by
- age comparison against the U–Pb system. *Earth and Planetary Science Letters* 301, 1–8.
- 937 Pagé, P., Barnes, S.-J., Bédard, J.H. & Zientek, M.L. (2012). In situ determination of Os,
- 938 Ir, and Ru in chromites formed from komatiite, tholeiite and boninite magmas:
- 939 implications for chromite control of Os, Ir and Ru during partial melting and crystal
- 940 fractionation. *Chemical Geology* **302–303**, 3–15.
- 941
- 942 Park, J.-W., Campbell, I.H. & Arculus, R.J. (2013). Platinum-alloy and sulfur saturation
- 943 in an arc-related basalt to rhyolite suite: Evidence from the Pual Ridge, eastern Manus
- 944 Basin. *Geochimica et Cosmochimica Acta* **101**, 76–95.
- 945
- 946 Paterson, D., de Jonge, M. D., Howard, D. L., Lewis, W., McKinlay, J., Starritt, A.,
- 947 Kusel, M., Ryan, C. G., Kirkham, R., Moorhead, G. & Siddons, D. P. (2011). The X-ray
- 948 *Flurorescence Microscopy Beamline at the Australian Synchroton.* AIP Conference
- 949 Proceedings, **1365**, 219-222.
- 950
- 951 Patino Douce, A.E. & Beard, J.S. (1995). Dehydration-melting of Biotite Gneiss and
- 952 Quartz Amphibolite from 3 to 15 kbar. *Journal of Petrology* **36**, 707-738.
- 953
- 954 Peck D. C., Keays R. R. & Ford R. J. (1992). Direct crystallization of refractory
- 955 platinum-group element alloys from boninitic magmas: evidence from western Tasmania.
- 956 Australian Journal of Earth Sciences **39**, 373–387.

957	
958	Prendergast, M. (2009). Monts de Cristal Platinum Project, NW Gabon. Geological
959	report on exploration activities. Lonmin plc.
960	
961	Puchtel, I.S. and Humayun, M. (2001) Platinum group element fractionation in a
962	komatiitic basalt lava lake. Geochim. Cosmochim. Acta 65: 2979-2994.
963	Raedeke, L.D. (1982). Petrogenesis of the Stillwater Complex. Unpublished PhD thesis,
964	University of Washington, Seattle, 212p.
965	
966	Rankenburg, K., Lassiter, J.C. & Brey, G. (2004). Origin of megacrysts in volcanic rocks
967	of the Cameroon volcanic chain – constraints on magma genesis and crustal
968	contamination. Contributions to Mineralogy and Petrology 147, 129-144.
969	
970	Rehkamper, M., Halliday, A.N., Fitton, J.G., Lee, DC., Wieneke, M., & Arndt, N.T.
971	(1999). Ir, Ru, Pt and Pd in basalts and komatiites: New constraints for the geochemical
972	behavior of the platinum group elements in the mantle. Geochimica et Cosmochimica
973	Acta 63, 3915-3934.
974	
975	Richard, P., Shimizu, N. & Allègre, C.J. (1976). 143Nd/146Nd, a natural tracer: an
976	application to oceanic basalts, Earth and Planetary Science Letters 31, 269–278.
977	
978	Rollinson HR (2013) Using geochemical data: evaluation, presentation, interpretation,
979	Routledge, Oxon, UK.
980	
981	Rudnick, R.L. & Fountain, D.M. (1995). Nature and composition of the continental crust:
982	A lower crustal perspective. Reviews in Geophysics 33, 267–309.
983	
984	Ryan, C., Kirkham, R., Hough, R., Moorhead, G., Siddons, D., de Jonge, M., Paterson,
985	D., De Geronimo, G., Howard, D. & Cleverley, J. (2010a), Elemental X-ray imaging
986	using the Maia detector array: The benefits and challenges of large solid-angle: Nuclear

- 987 Instruments & Methods in Physics Research, Section A–Accelerators, Spectrometers,
- 988 Detectors and Associated Equipment **619**, 37-43.
- 989
- 990 Ryan, C., Kirkham, R., Siddons, D., Dunn, P., Laird, J., Kuczewski, A., Moorhead, G.,
- 991 De Geronimo, G., Davey, P., Jensen, M., Paterson, D., de Jonge, M., Howard, D.
- 892 & Hough, R. (2010b). The Maia 384 detector array in a nuclear microprobe: A platform
- 993 for high definition PIXE elemental imaging: Nuclear Instruments & Methods in Physics
- Research Section B-Beam Interactions with Materials and Atoms 268, 1899-1902.
- 995
- 996 Ryan, C. G. (2000). Quantitative Trace Element Imaging using PIXE and the Nuclear
- 997 Microprobe. International Journal of Imaging Systems and Technology **11**, 219-230.
- 998
- 999 Ryan, C.G., Siddons, D.P., Kirkham, R., Li, Z.Y., de Jonge, M.D., Paterson, D.J.,
- 1000 Cleverley, J.S., Kuczewski, A., Dunn, P.A., Jensen, M., De Geronimo, G., Howard, D.L.,
- 1001 Godel, B., Dyl, K.A., Fisher, L.A., Hough, R.M., Barnes, S.J., Bland, P.A., Moorhead,
- 1002 G.F., James, S.A., Spiers, K.M., Falkenberg, G., U, Boesenberg, U., Wellenreuther, G.,
- 1003 (2014). The Maia detector array and x-ray fluorescence imaging system: locating rare
- 1004 precious metal phases in complex samples Proc. SPIE 8851, X-Ray Nanoimaging:
- 1005 Instruments and Methods 8851, 88510Q.
- 1006
- 1007 Ryan, C. G., Siddons, D. P., Kirkham, R., Dunn, P. A., Kuczewski, A., Moorhead, G., De
- 1008 Geronimo, G., Paterson, D. J., de Jonge, M. D., Hough, R. M., Lintern, M. J., Hoard, D.
- 1009 L., Kappen, P. & Cleverley, J. (2010c). The New Maia Detector System: Methods for
- 1010 High Definition Trace Element Imaging of Natural Material, in Denecke, M. A., and
- 1011 Walker, C. T., eds., X-Ray Optics and Microanalysis, **1221**, 9-17.
- 1012
- 1013 Sobolev S.V., Sobolev, A.V., Kuzmin D.V., Krivolutskaya, N.A., Petrunin, A.G., Arndt,
- 1014 N.T., Radko, V.A., & Vasiliev, Y.R. (2011). Linking mantle plumes, large igneous
- 1015 provinces and environmental catastrophes. *Nature* **477**, 312-316.
- 1016

1017	Teigler, B. & Eales, H.V. (1996). The lower and critical zones of the western limb of the
1018	Bushveld Complex, as indicated by the Nooitgedacht boreholes. Geological Survey of
1019	South Africa Bulletin 111, 126
1020	
1021	Thompson, R.N. Morison, M.A., Dicken, A.P. & Hendry, G.L. (1983). Continental flood
1022	basalts Archahnids rule OK? In: Hawkesworth, C.J. and Norry, M.J. (eds.) Continental
1023	basalts and mantle xenoliths. Shiva Nantwich, 158-185.
1024	
1025	Tomkins, A.G., Rebryna, K.C., Weinberg, R.F. & Schaefer, B.F. (2012). Magmatic
1026	Sulfide Formation by Reduction of Oxidized Arc Basalt. J Petrol., 53, 1537-1567.
1027	
1028	Toplis, M.J. & Corgne, A. (2002). An experimental study of element partitioning
1029	between magnetite, clinopyroxene and iron-bearing silicate liquids with particular
1030	emphasis on vanadium. Contributions to Mineralogy and Petrology 144, 22-37.
1031	
1032	Wille, M., Kramers, J.D., Nägler, T.F., Beukes, N.J., Schröder, S., Meisel, T., Lacassie,
1033	J.B. & Voegelin, A.R. (2007). Evidence for a gradual rise of oxygen between 2.6 and 2.5
1034	Gafrom Mo isotopes and Re-PGE signatures in shales. Geochimica et Cosmochimica
1035	<i>Acta</i> 71 , 2417–2435.
1036	
1037	Wilson, A.H. (2012). A chill sequence to the Bushveld Complex: insight into the first
1038	stage of emplacement and implications for the parental magmas. Journal of Petrology
1039	53 , 1123–1168
1040	
1041	Wingate, M.T.D., Campbell, I.H., Compston, W., and Gibson, G.M., 1998, Ion
1042	microprobe U-Pb ages for Neoproterozoic basaltic magmatism in southcentral Australia
1043	and implications for the breakup of Rodinia: Precambrian Research, v. 87, p. 135–159.
1044	
1045	Yang, S., Maier, W.D., Lahaye, Y. & O'Brien, H. (2013). Strontium isotope
1046	disequilibrium of plagioclase in the Upper Critical Zone of the Bushveld Complex:

1047 evidence for mixing of crystal slurries. Contributions to Mineralogy and Petrology, DOI

1048 10.1007/s00410-013-0903-4

- 1049
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- 1051 **FIGURES**





Fig. 1(a): Geological map of Gabon (modified after Chevalier et al. 2002). Insert shows
segment enlarged in Fig. 1b. (b): Geological map of the Monts de Cristal Complex and its
host rocks. Red line denotes outline of Monts de Cristal National Park. KG1-4 are
individual blocks of Complex, and shaded areas with pink outline represent exploration
focus areas. See insert of Fig. 1a for location.



- 1061 Fig. 2: Geological map of the KG1 body, showing sampling localities and drill core sites
- 1062 (denoted by stars, modified after Prendergast, 2009). Stippled line indicates a road.
- 1063 Coordinates are given in UTM system.
- 1064



1066 Fig. 3: Photomicrographs of Monts de Cristal rocks. (a) Olivine pyroxenite showing 1067 olivine inclusions in orthopyroxene. Sample Olpx21. (b) Orthopyroxene adcumulate, 1068 sample Px21. (c) Orthopyroxenite mesocumulate. Zoning is indicated by dark coronas 1069 along rims of grains. Sample Px18. (d) Chromite clusters within orthopyroxene of 1070 orthopyroxenite adcumulate. Sample Px21. (e) Chromite clusters within altered inclusion 1071 in orthopyroxene. Sample Px21. (f) Norite, showing euhedral and subhedral 1072 orthopyroxene as well as cumulus and intercumulus plagioclase. Sample Px16. (g) 1073 Olivine gabbro, showing intercumulus olivine and clinopyroxene as well as acicular 1074 plagioclase crystals. Sample Dy18. (h) Olivine gabbro, showing large euhedral 1075 clinopyroxene oikocryst containing plagioclase chadacrysts. Sample Dy18. Length of 1076 scale bar in each panel is 1mm.







1079 Fig. 4: Binary plot showing the variation of Fo and Ni contents of olivine in a number of 1080 layered intrusions. Note that the Monts de Cristal olivines have relatively high Ni 1081 contents compared to other intrusions with comparable Fo contents. Bushveld data are 1082 from Teigler and Eales (1996), Maier and Eales (1997) and Wilson (2012), Great Dyke 1083 data are from Wilson (pers. com., 2012), Koitelainen data are from Hanski (unpublished, 1084 pers. com., 2013), Stillwater data are from Raedeke (1982) and Barnes and Naldrett 1085 (1986), Mt Ayliff data are from Lightfoot and Naldrett (1983) and Lightfoot et al. (1984), 1086 and Finnish dykes are unpublished data of Fangfang Guo. Data for global komatiites and 1087 basalts are from Sobolev et al. (2011). 1088





1090 Fig. 5: Binary variation diagrams of (a) Al₂O₃, (b) TiO₂, (c) Cr, and (d) Ni, plotted vs

1091 MgO. Stippled arrows represent best fit lines through orthopyroxenites. Boxes linked by

1092 tielines indicate compositional ranges in analysed minerals.



Fig. 6: Multi-element variation diagrams of lithophile trace elements. Data for Bushveld
Upper Critical Zone are from Maier et al. (2013), and average Upper Crust is from





1098

1099 Fig. 7: Binary variation diagram of TiO₂ vs S. Regression line drawn through primitive

1100 orthopyroxenites (grey field, with exception of 2 samples) passes through origin and

1101 indicates approximately 100-150 ppm S in the putative trapped liquid at 0.4-0.5% TiO₂.



Fig. 8: Binary variation diagrams of chalcophile metals in various blocks of the Monts de
Cristal Complex, plotted vs MgO. (a) Pd, (b) Au, (c) Pt, (d) Ru, (e) Ir, (f) Rh. Note
negative correlations with MgO for Pd and Au, but positive correlations for Pt, Ru, Ir and
Rh. See text for discussion.



1110 Fig. 9: Analyses of drill core samples plotted in binary variation diagrams vs MgO of (a)





1116 Fig. 10: Chemical data for drill core MPD-005



Fig. 11: (a-d) Binary variation diagrams for PGE. Note poor correlation between Pt and
Pd, and high Pt/Pd ratios of most rocks. Outlined fields mark samples from Pt rich layer
at KG1 west.



1123

1124 Fig. 12: Cu/Pd and Cu/Zr of Monts de Cristal rocks. Primitive mantle (PM) estimate is

1125 from Barnes and Maier (1999). Shaded bar in (b) represents range of South African

- 1126 continental flood basalts (CFB, from Maier et al., 2003).
- 1127



1129 Fig. 13: a) and b) – three element false-colour RGB images of sample Px21. Each 1130 element is scaled to maximum values; Cr and Ti values are log transformed to enhance 1131 the compositonal range within the silicate phases. Note heterdacumulate texture of 1132 plagioclase oikocrysts (blue on both images) interstitial to orthopyroxene with minor 1133 intercumulus clinopyroxene (green on both images). Also note subtle complex internal 1134 zoning in Cr in some grains (grain 1 in image a) and fine rims of elevated Ti in image b 1135 in areas outside the plagioclase oikocrysts (point 1 in image b). Light green highlights on both images are small chromite inclusions, preferentially located in grains showing Cr 1136 1137 zoning. (c) Three element image for As (red) – Cu (green) and Ni (blue) – light green 1138 areas indicate probable presence of Cu-Ni rich sulphide. One of these sulfide areas 1139 occupies the grain boundary between opx host and an olivine inclusion. Pt peaks in (d) 1140 correspond to coincident highs for As, within areas enriched in Cu and Ni, but not exactly 1141 matching Cu-Ni peaks. Pt corresponds exactly to As peak in each case. Note also 1142 presence of isolated small Cu-rich phases, probably sulphide, with no associated Pt or As, 1143 at opx margin.



1145

Fig. 14: Sample Px17A, with same combinations of elements as in Fig. 13a. Note 1146 1147 abundant inclusions of olivine (red) and chromite (green) in Cr-zoned opx. No Pt grains 1148 were found in this grain, however. Enlarged areas show As-Cu-Ni three element maps as 1149 above, and corresponding single-element concentration maps for Pt. Note in all areas that 1150 Pt hits are associated with areas containing all three elements As, Cu and Ni, but the Pt 1151 peaks are very specifically associated with As, and do not necessarily occur directly 1152 within what appear to be sulphide aggregates. These aggregates are all occurring at or 1153 indistinguishably close to opx grain boundaries, but with no systematic relationship to 1154 plag oikocrysts.



1156

1157 Fig. 15: Plot of La/Sm vs La/Nb for MdC rocks. Also plotted is average upper continental

1158 crust (Rudnick and Fountain, 1995) and average 3.0-3.5 Ga TTG (Martin et al., 2005).

1159 Solid line represents mixing line between magnesian basalt and a contaminated magma

1160 produced by AFC (r=0.8, f=0.8) of magnesian basalt with a 17% partial melt of MdC

1161 country rock granite. Trace element contents of magnesian basalt are assumed to be 4x

1162 that of primitive mantle (PM), with PM data taken from McDonough and Sun (1995). See

1163 text for further discussion.