Cardiff School of Pharmacy and Pharmaceutical Sciences
Cardiff University

# Synthesis and Biological Evaluation of Novel Nucleosides and Nucleotides as <br> Potential Therapeutic Agents 

A thesis submitted in accordance with the conditions governing candidates for the degree of Philosophiae Doctor in Cardiff University

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#### Abstract

Nucleoside analogues are an important class of antimetabolites, used both as anticancer and antiviral agents for their resemblance to endogenous nucleosides, and their capacity to inhibit metabolic pathways in which these substrates are involved, leading to therapeutic potential. The need for both new anticancer and antiviral cures is significant, and often caused by the emergence of resistance to the available therapies. In the case of therapies with nucleoside analogues, the delivery of a nucleoside monophosphate prodrug inside the cell has potential to overcome resistance to treatment, and proved successful especially with the application of the ProTide approach. This strategy led to the progression into clinical trials of numerous antiviral and anticancer agents.

This work focused on the synthesis of novel ProTide prodrugs to different nucleoside analogues that are involved in clinical trials, such as cordycepin, 8-chloroadenosine, CNDAC and the recently approved agent trifluorothymidine. This strategy was also applied to 2'-mercapto-2'-deoxyuridine and 2'-trifluoromethyl-2'-deoxyuridine, whose in vitro evaluation has never been reported. The synthetic strategies to prepare each nucleoside analogue are also reported.

The in vitro evaluation of novel nucleotide prodrugs as anticancer and antiviral agents is described and discussed. Moreover the mechanism of activation of the ProTides is supported by studying their bioactivation to the corresponding monophosphate forms, through enzymatic NMR studies and molecular modeling simulations.

Modifications on the scaffold of the two most promising families of ProTides of cordycepin and 8 -chloroadenosine were also performed, leading to the introduction of groups on the nucleobase or alteration of the sugar moiety. Moreover, the phosphate group was also modified, with the application of alternative phosphorodiamidate and phosphonoamidate prodrug approaches, with the aim of improving the biological profile.

Cordycepin and 8 -chloroadenosine families of compounds emerged as the most potent prodrugs. Selected analogues from these two families were evaluated on resistant cancer cell lines, retaining better potency compared to the parent nucleoside. Moreover these analogues are stable in human plasma, serum and liver microsomes. Furthermore, the most promising ProTides showed selective targeting of the stem cell population of the KG1a leukaemic cell line.

Finally, two candidates in the cordycepin family of ProTides were tested in vivo on a mouse model of human non-Hodgkin lymphoma, leading to promising results.

Further investigations on these analogues are currently ongoing.


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## Abbreviations

- 1-Naph
- $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}$
- 2'-C-Me-C
- 2'-C-Me-G
- 2'-SCF ${ }_{3}-2^{\prime}-\mathrm{dU}$
- 2'-SH-2'dU
- 2F3'dA
- 3'dA
- 3'dAMP
- 3'dATP
- 3'dIno
- 3TC
- 5'-NT
- 5 FU
- 6-MP
- 8ClA
- 8CldA
- 8ClIno
- A-134974
aminocyclopentyl]-4-amino-5hydrate
- ABC
- ACV
- ADA
- Ade
- AMPK
- Ado
- ADP
- AK
- AMPK
- Ara-C
- Ara-G

1-Naphthyl
2-Methoxy-3'-deoxyadenosine
2'-C-methylcytidine
2'-C-methylgua ine
(2'-trifluoromethylthio-2'-deoxyuridine)
2'-mercapto-2'-deoxyuridine
2-Fluoro-3'-deoxyadenosine
3'-deoxyadenosine, cordycepin
3'-deoxyadenosine-5'-monophosphate
3'-deoxyadenosine-5'-triphosphate
3'-deoxyinosine
Lamivudine
5'-nucleotidase
5-fluorouracil
6-mercaptopurine
8-Chloroadenosine
8-Chloro-2'-deoxyadenosine
8-Chloroinosine
$\mathrm{N}^{7}-\left[\left(1^{\prime} \mathrm{R}, 2^{\prime} \mathrm{S}, 3^{\prime} \mathrm{R}, 4^{\prime} \mathrm{S}\right)-2^{\prime}, 3^{\prime}\right.$-dihydroxy-4'iodopyrrolopyrimidine dihydrochloride Abacavir

Acyclovir
Adenosine deaminase
Adenine
Adenylate kinase
Adenosine
Adenosine diphosphate
Adenosine kinase
Adenosine monophosphate kinase
Cytarabine
9- $\beta$-D-Arabinofuranosyl guanine

- ATP
- AZT
- Bn
- BNF
- BSA
- cAMP
- catA
- CDA
- cGMP
- cHex
- CMV
- CNDAC
pentofuranosylcytosine
- CNDAU
pentofuranosyluridine
- CPE
- CPY
- CSC
- Cyd
- Cyt
- CV
- d4T
- dAdo
- dCF
- dCK
- DCM
- dCyt
- ddAMP
- ddATP
- ddI
- DDQ
benzoquinone)
- ddU
- dGK

Adenosine triphosphate
zidovudine
Benzyl
British national formulary
$\mathrm{N}, \mathrm{O}$-bis(trimethylsilyl)acetamide
Cyclic adenosine monophosphate
Cathepsin-A
Cytidine deaminase
Cyclic guanosine monophosphate
Cyclohexyl
Cytomegalovirus
$2^{\prime}$ - $C$-cyano-2'-deoxy-1- $\beta$-D-arabino-

2'-C-cyano-2'-deoxy-1- $\beta$-D-arabino-

Cytophatic effect
Carboxypeptidase Y
Cancer Stem Cells
Cytidine
Cytosine
Column volumes
Stavudine
2'-Deoxyadenosine
Deoxycoformycin, pentostatin
Deoxycytidine kinase
Dichloromethane
2'-Deoxycytidine
Dideoxyadenosine monophosphate
Dideoxyadenosine triphosphate
Didanosine
2,3-Dichloro-5,6-dicyano-1,4-

Dideoxyuridine
Deoxyguanosine kinase

- dGuo
- DIPEA
- DLP
- DMA
- DMAP
- DMF
- DMG
- DMP
- DNA
- dNTP
- DTT
- dTTP
- dUTP
- $E D_{50}$
- EHNA
- ES/MS
- Et
- EtOAc
- EtSCSCl
- ETV
- FAD
- FCV
- FDA
- FdT
- FE
- FTC
- FUdR
- GCV
- Gly
- Gua
- Guo
- HBV
- HCV

2'-Deoxyguanosine
Diisopropyl ethyl amine
Dilauroyl peroxide
Dimethylacetamide
Dimethylamino pyridine
Dimethylformamide
Dimethylglycine
Dess-Martin periodinane
Deoxyribonucleic acid
Deoxynucleoside triphosphate
Dithiothreitol
Thymidine triphosphate
Deoxyuridine triphosphate
Median effective dose
Erythro-9-(2-hydroxy-3-nonyl) adenine
Electrospray mass spectrometry
Ethyl
Ethyl acetate
Ethyl chlorodithioformate
Entecavir
Flavin adenine dinucleotide
Famciclovir
Food and Drug Administration
3'-Fluoro-3'-deoxythymidine
Fast eluting
Emtricitabine
Floxuridine
Ganciclovir
Glycine
Guanine
Guanosine
Hepatitis B virus
Hepatitis C virus

- hENT

Human equilibrative nucleoside transporter 1

- Hint
- HSV
- $\quad i \mathrm{Bu}$
- IBX
- IPA
- L-Ala
- $L^{2} D_{50}$ viability
- LDA
- L-Leu
- L-Phe
- LSC
- MDS
- Me
- mCPBA
- MHz
- NAD
- NBTI
- NCS
- NDPK
- NDPK
- NDPK
- Neop
- $n$-Hex
- NMI
- NMPK
- NMR
- $n$-Pen
- $P$
- PARP
- PCV
- Ph

Hystidine triad nucleoside-binding protein
Herpes simplex virus
Isobutyl
2-Iodoxybenzoic acid
Isopropanol
L-Alanine
50\% Inhibitory concentration of cell

Lithium diisopropyl amide
L-leucine
L-phenylalanine
Leukemic Stem Cells
Myelodisplastic syndrome
Methyl
meta-chloroperbenzoic acid
Mega Hertz
Nicotinamide adenine dinucleotide
S-(4-Nitrobenzyl)-6-thioinosine
N -chlorosuccinimmide
Nucleoside diphosphate kinase
Nucleoside diphosphate kinase
Nucleoside diphosphate kinase
Neopentyl
Normal-hexyl
N -methylimidazole
Nucleoside monophosphate kinase
Nuclear magnetic resonance
Normal-pentyl
Phosphate group
Poly (ADP) ribose polymerase
Penciclovir
Phenyl

- PLC
- POC
- POM
- PTC-Cl
- $p$-TSA
- Pyr
- RBV
- RNA
- RNR
- ROS
- RSV
- S-8ClAMP
- SATE
- SDTE
- SE
- Ser
- TBAF
- TBAN
- TBDMSCl
- $t \mathrm{Bu}$
- TEA
- TFA
- TFAA
- TFT
- TG
- Thd
- THF
- Thy
- TIPSCl
tetraisopropyldisiloxane
- TK1/2
- TLC
- TMSCN
- TMSOMe

Phospholipase C
(isopropyloxycarbonyloxymethyl)
(pivaloyloxymethyl)
Phenoxythiocarbonyl chloride
para-toluene sulfonic acid
Pyridine
Ribavirin
Ribonucleic acid
Ribonucleotide reductase
Reactive oxygen species
Respiratory sincitial virus
Succinyl-8ClA monophosphate
$S$-acyl-2-thioethyl
$S$-[2-hydroxyethylsulfidyl]-2-thioethyl
Slow eluting
Serine
Tetrabutylammonium fluoride
Tetrabutylammonium nitrate
Tert-butyl dimethylsilyl chloride
Tert-butyl
Triethylamine
Trifluoroacetic acid
Trifluoroacetyc anhydride
Trifluorothymidine
Thioguanine
Thymidine
Tetrahydrofuran
Thymine
1,3-Dichloro-1,1,4,4-

Thymidine kinase $1 / 2$
Thin layer chromatography
Trimethylsilyl cyanide
Trimethylmethoxysilane

- TMSOTf
- TP
- TsCl
- UCK1/2
- Ura
- Urd
- UV
- VACV
- VGCV
- $\alpha$ - AIBBr

Trimethylsilyl trifluoromethanesulfonate
Thymidine phosphorylase
Tosyl chloride
Uridine-cytidine kinase $1 / 2$
Uracil
Uridine
Ultraviolet
Valaciclovir
Valganciclovir
$\alpha$-Acetoxyisobutyryl bromide

## 1 Nucleoside analogues as anticancer agents

### 1.1 The structure of endogenous nucleosides

Nucleosides are endogenous metabolites formed by a heterocyclic purine or pyrimidine ring called "base" linked to a pentose carbohydrate through a $\beta$ - N -glycosidic bond. ${ }^{1}$


Figure 1.1: Purine and pyrimidine deoxyribonucleoside and ribonucleoside structures.

The five major bases found in nature are the purines adenine (A) and guanine (G) and the pyrimidines cytosine (C), uracil (U) and thymine (T) (Figure 1.1).

The sugar can be either a D-ribose or a 2-deoxy-D-ribose. The nucleosides containing Dribose are called ribonucleosides and are represented by adenosine (Ado), guanosine (Guo), cytidine (Cyd), and uridine (Urd). The presence of 2-deoxy-D-ribose generates deoxynucleosides such as deoxyadenosine (dAdo), deoxyguanosine (dGuo), deoxycytidine (dCyt) and thymidine (Thd) (Figure 1.1).

Nucleotides are the 5 '-mono-, di- or triphosphorylated forms of nucleosides and are primarily found in the organism as units of the major nucleic acids of the cell, DNA and RNA. ${ }^{2}$ DNA is a polymer formed by deoxyribonucleoside monophosphate monomers linked together via a phosphodiester bond between the $3^{\prime}-\mathrm{OH}$ of one nucleotide and the $5^{\prime}$ 'OH of the second nucleotide (Figure 1.2).


Figure 1.2: Structural components of nucleic acids.

RNA has a similar structure but the nucleotide thymidine $5^{\prime}$-monophosphate is replaced by a uridine 5 '-monophosphate and the deoxyribose sugar by ribose (Figure 1.2).
Nucleosides and nucleotides are involved in other crucial processes such as neurotransmission, ${ }^{3}$ regulation of cardiovascular activity, ${ }^{4}$ and cell signaling as cyclic phosphates, such as cyclic adenosine monophosphate (cAMP) and cyclic GMP (cGMP). ${ }^{5-7}$ Furthermore, ATP involved in energy storage and liberation via hydrolysis of the phosphodiester bonds. ${ }^{8}$ Nucleoside analogues, linked to other moieties, especially in the form of adenosine diphosphate (ADP), are also involved in enzyme regulation as enzyme cofactors, such as nicotinamide adenine dinucleotide (NAD), flavin adenine dinucleotide (FAD) and coenzyme A. ${ }^{2,9}$

### 1.2 Modified nucleoside analogues as therapeutic agents

Nucleoside and nucleotide analogues are structurally related to the endogenous counterparts, hence can exploit cellular metabolism and subsequently be incorporated into DNA and RNA to inhibit cellular division ${ }^{10}$ and viral replication. ${ }^{11}$ Therapeutic properties of these molecules could also derive from the interaction with both human and host polymerase enzymes, responsible for the synthesis of nucleic acids (DNA-dependent DNA polymerases, ${ }^{12}$ RNA-dependent DNA polymerases or RNA-dependent RNA polymerases). ${ }^{13}$ Furthermore, ribonucleotide reductase (RNR), responsible for deoxyribonucleoside pool homeostasis, is recognised as a target of antitumour agents. ${ }^{14}$ Additional targets for treatment with anticancer nucleoside analogues are DNA methyltransferase, ${ }^{15}$ purine nucleoside phosphorylase ${ }^{16}$ and thymidylate synthase. ${ }^{17}$
Nucleoside and nucleotide analogues used in antiviral treatment are structurally more diverse than anticancer nucleosides (such as the acyclic sugar moiety in acyclovir, an antiviral agent used as a treatment against herpes infections); this difference causes antiviral agents to be less active on mammalian enzymes, leading to better tolerance profile than anticancer analogues. ${ }^{9}$

### 1.2.1 Mechanism of action of nucleoside analogues

Presently used nucleoside analogues exploit the same metabolic pathways as endogenous nucleosides and nucleotides, and compete with them acting as antimetabolites. ${ }^{9,18}$
Nucleosides depend on transporter proteins on the cell membrane to enter the cell, because their polarity inhibits passive diffusion through the lipophilic cell membrane. ${ }^{19,20}$

Once inside the cell, nucleosides are phosphorylated to their corresponding active $5^{\prime}$ ' $O$ triphosphate forms in three consecutive enzymatic steps. ${ }^{21}$ The nucleoside kinase (NK) enzymes involved in the first phosphorylating step, active on deoxyribonucleosides are: deoxycytidine kinase (dCK), deoxyguanosine kinase (dGK) and thymidine kinase 1 and 2 (TK1 and TK2). The known NK responsible for the phosphorylation of ribonucleosides are adenosine kinase (AK) and uridine-cytidine kinase 1 and 2 (UCK1, 2). ${ }^{22,23}$


Scheme 1.1: Nucleoside analogues intracellular mode of action. NK: nucleoside kinase, NA: nucleoside analogue, $5^{\prime}$-NT: 5'-nucleotidase, NDA: nucleoside deaminase, NtDA: nucleotide deaminase, P: phosphate group, TS: thymidylate synthase, NMPK: nucleoside monophosphate kinase, RR: ribonucleotide reductase, NDPK: nucleoside diphosphate kinase.

These enzymes are also involved in the first phosphorylation of most nucleoside drugs that are currently used. ${ }^{24}$ This step is often rate-limiting. ${ }^{25}$ The other two steps are catalysed by nucleoside monophosphate kinases (NMPKs), ${ }^{24}$ and nucleoside diphosphate kinases (NDPKs). ${ }^{26}$

There are also nucleoside and nucleotide deaminase enzymes (respectively NDA and NtDA) that deactivate nucleoside drugs. Cytidine deaminase (CDA), ${ }^{27}$ and adenosine deaminase (ADA), ${ }^{28}$ perform hydrolytic deamination of cytidine and adenosine derivatives, to the corresponding uridine or inosine derivatives. On the monophosphate level, adenosine monophosphate deaminase carries out similar conversions. ${ }^{29}$ Furthermore, the enzyme $5^{\prime}$-nucleotidase ( $5^{\prime}$-NT) catalyses the conversion of nucleoside monophosphates to nucleosides. ${ }^{30}$

As mono, di and triphosphorylated forms, nucleoside analogues can accumulate inside the cell and inhibit intracellular enzymes, while as the triphosphate form they can be incorporated into DNA and/or RNA, leading to inhibition of chain growth, ${ }^{11}$ accumulation of mutations in the viral progeny, ${ }^{31}$ and induction of apoptosis. ${ }^{32}$

### 1.2.2 Anticancer nucleoside analogues

Nucleoside analogues represent some of the oldest treatments in the anticancer and antiviral fields. Despite being old treatments, they still represent a cornerstone in these diseases, and research is still ongoing to improve the therapeutic profile of such agents. The British National Formulary (BNF) lists 14 anticancer agents with a nucleobase or nucleoside structure that are currently used in the clinic (Figure 1.3). ${ }^{33}$


Figure 1.3: Anticancer nucleobase and nucleoside analogues currently used in the clinic (BNF 2014). 6Mercaptopurine (1a), thioguanine (1b), 5-fluorouracil (1c), floxuridine (1d), capecitabine (1e), tegafur (1f), cytarabine ( $\mathbf{1 g}$ ), gemcitabine ( $\mathbf{1 h}$ ), fludarabine ( $\mathbf{1 i}$ ), cladribine ( $\mathbf{1 j}$ ), clofarabine ( $\mathbf{1 k}$ ), nelarabine ( $\mathbf{1 l}$ ), azacitidine ( $\mathbf{1 m}$ ), decitabine ( $\mathbf{1 n}$ ).

The nucleobases 6 -mercaptopurine ( $6-\mathrm{MP}, \mathbf{1 a}$ ) and thioguanine (TG, 1b) are thiopurines and some of the oldest cytotoxic agents. ${ }^{34}$ These structures require the activity of the salvage enzyme hypoxanthine-guanine phosphoribosyl transferase for the formation of respectively thioinosine monophosphate and thioguanosine monophosphate. These molecules are then converted to their nucleoside triphosphate forms and integrated into nucleic acids, leading to inhibition of DNA polymerases, ligases and endonucleases. Other
effects on de novo enzymes were documented, such as inhibition of 5-phosphoribosyl-1pyrophosphate amidotransferase, inosine monophosphate dehydrogenase and ribonucleotide reductase. Thiopurines are used against acute leukaemias, alone or more often in combination with other antineoplastic agents. ${ }^{35}$
5-Fluorouracil ( $5 \mathrm{FU}, \mathbf{1 c}$ ) is a derivative of uracil and initially synthesised in the 1950s and used in the treatment of several solid tumours including colorectal cancer. ${ }^{17}$ Floxuridine ( $\mathrm{FUdR}, \mathbf{1 d}$ ) is the deoxynucleoside derivative of 5 FU . Both drugs are in equilibrium in cancer cells due to the activity of the enzyme thymidine phosphorylase, which catalyses the coupling of 5 FU with deoxyribose and the cleavage of the glycosidic bond in floxuridine. ${ }^{36}$

Floxuridine monophosphate irreversibly inhibits thymidylate synthase, which catalyses the reductive methylation of deoxyuridine monophosphate into thymidine monophosphate. This causes an imbalance in the thymidine triphosphate/deoxyuridine triphosphate ratio (TTP/dUTP), leading to an increase of dUTP which is integrated into DNA and excised by the enzyme uracil-DNA glycosylase, with consequent DNA strand breaks and apoptosis. Moreover, after conversion into floxuridine triphosphate, this drug is integrated into RNA and DNA affecting intracellular metabolism. ${ }^{36}$
Capecitabine (1e) and tegafur (1f) are both 5-fluorouracil prodrugs to be administered orally. ${ }^{37}$ Tegafur is part of a combination with uracil called UFT. Both drugs share structural similarities and activation processes. After oral administration, both compounds cross the intestinal mucosa. Capecitabine undergoes metabolism with first cleavage of the amidic bond, followed by conversion into the uridine derivative by cytidine deaminase. The product is then finally converted into 5FU by thymidine phosphorylase activity, which catalyses the same conversion on tegafur.
Cytarabine ( $\mathbf{1 g}$ ) is a deoxycytidine analogue with activity on haematologic malignancies, expecially against acute myeloid leukaemia and no effect on solid tumours. ${ }^{34}$ This nucleoside is phosphorylated by dCK and inactivated by conversion to the uracil derivative by CDA. In the triphosphate form, it is integrated into DNA leading to chain termination and it directly inhibits DNA polymerase. ${ }^{34}$ Unlike cytarabine, the deoxycytidine derivative gemcitabine (1h) has activity on solid tumours such as pancreas, breast, lung and ovaries. ${ }^{38}$ The mechanism of action of this nucleoside analogue differs from cytarabine as after incorporation into DNA, an endogenous nucleotide is added, thus preventing DNA repair by base-pair excision. Gemcitabine diphosphate also has inhibitory activity on ribonucleotide reductase, blocking the de novo DNA synthesis pathway and determining a
self-potentiating effect of the drug, as dCTP levels decrease. This leads to an increase of gemcitabine intracellular activation and integration into both DNA and RNA. ${ }^{38}$
Fludarabine ( $\mathbf{1 i}$ ) and cladribine ( $\mathbf{(} \mathbf{j}$ ) are deoxyadenosine analogues. ${ }^{39}$ The former is orally administered as the $5^{\prime}$-monophosphate to improve solubility, and its approved application is against chronic lymphocytic leukaemia. ${ }^{39}$ Cladribine was approved against hairy cell leukaemia, and similarly to fludarabine, its triphosphate form is integrated in DNA stopping the elongation by DNA polymerases in the S phase of the cell cycle causing apoptosis. ${ }^{40}$ Both compounds also inhibit RNR and are toxic on resting cells, a feature that was suggested to derive from inhibition of DNA repair. Both nucleosides are integrated into RNA, causing premature termination of RNA transcription. ${ }^{40}$

More recently approved nucleoside analogues are the deoxyadenosine derivative clofarabine ( $\mathbf{1 k}$ ) and the deoxyguanosine derivative nelarabine (11). Clofarabine was approved in 2004 for the treatment of acute myeloid leukaemia. ${ }^{41}$ The triphosphate form inhibits DNA polymerase enzymes and RNR. These effects cause a self-potentiating profile with improved integration of clofarabine triphosphate into DNA and induction of apoptosis.

Nelarabine (11) was approved in 2005 for the treatment of T-cell acute lymphoblastic leukaemia and T-cell lymphoblastic lymphoma. ${ }^{42}$ This guanosine analogue was initially developed to improve the poor pharmacokinetic profile of 9- $\beta$-D-arabinofuranosyl guanine (Ara-G), as the introduction of a methoxy moiety in position 6 on the guanine portion improved the water solubility approximately 10 -fold. Both drugs were found to be integrated into DNA as their triphosphate form, inhibiting DNA synthesis and leading to apoptosis. ${ }^{42}$

Azacitidine (1m) and decitabine (1n) are respectively cytidine and deoxycytidine analogues. Both drugs were approved respectively in 2004 and 2006 for the treatment of myelodysplastic syndrome (MDS). The development of these drugs was initially due to their cytotoxic profile, and only several years later the epigenetic properties of both compounds were confirmed. ${ }^{15}$ Azacytidine and decitabine are inhibitors of the enzyme DNA methyltransferase and have shown substantial potency in reactivating epigenetically silenced tumour suppressor genes. In fact, DNA methyltransferase inhibition reverts hypermethylation induced gene silencing, which in turn causes silencing of tumour suppressor genes, which are responsible for human tumourigenesis. Azacytidine is integrated into RNA, although $10-20 \%$ of it is converted by RNR into decitabine, which is, instead, integrated into DNA. ${ }^{43}$

### 1.2.3 Antiviral nucleoside analogues

Nucleoside analogues were broadly explored in the antiviral field, and Figure 1.4 shows the compounds still used in the clinic in the UK (2a-2p). ${ }^{44}$ The nucleoside analogues zidovudine (2a), didanosine (2b), lamivudine (2c), emtricitabine (2d), stavudine (2e) and abacavir (2f) are active as anti-HIV reverse transcriptase inhibitors and HIV-DNA chain terminators. ${ }^{45}$


2a


2b


2c


2d

$2 e$

$2 f$


2g


2h

$2 i$

$2 j$


2k


21

$2 m$


2n


20


2p

Figure 1.4: Antiviral nucleoside analogues. Zidovudine (2a), didanosine (2b), lamivudine (2c), emtricitabine (2d), stavudine (2e), abacavir (2f), penciclovir ( $\mathbf{2 g}$ ), famciclovir ( $\mathbf{2 h}$ ), ribavirin ( $\mathbf{2} \mathbf{i}$ ), acyclovir ( $\mathbf{2} \mathbf{j}$ ), valaciclovir ( $\mathbf{2 k}$ ), ganciclovir (2I), valganciclovir (2m), cidofovir (2n), entecavir (20), telbivudine (2p).

Zidovudine (AZT, 2a), first approved as an anti-HIV nucleoside analogue in 1987, is active as the nucleoside triphosphate by inhibiting viral reverse transcriptase and HIVDNA chain growth. ${ }^{46}$

Didanosine (ddI, 2b), approved as an anti-HIV agent in 1991, is converted to dideoxyadenosine monophosphate (ddAMP) by the action of first a phosphotransferase enzyme and secondly by adenylosucinnate lyase and adenylosucinnate synthetase. After two additional phosphorylation steps, ddATP represents the final active metabolite. ${ }^{47}$
Lamivudine ( $3 \mathrm{TC}, \mathbf{2 c}$ ) and emtricitabine (FTC, 2d) are anti-HIV agents approved respectively in 1995 and 2003, as L-isomers. ${ }^{48}$ Lamivudine is also used as an anti-hepatitis $B$ virus (HBV) agent, at a lower dosage.
Stavudine (d4T, 2e) is a thymidine derivative approved in 1995 as an anti-HIV agent. ${ }^{45}$
Abacavir (ABC, 2f) was approved in 1998 as an anti-HIV agent and acts after conversion into the guanosine analogue carbovir triphosphate, which was initially synthesised and
biologically evaluated as an anti-HIV agent but poor pharmacokinetics halted its progression into the clinics. ${ }^{48}$

Penciclovir (PCV, 2g), is an acyclic nucleoside analogue approved in 1996 for the topical treatment of herpes simplex virus (HSV), followed in 1997 by its prodrug famciclovir (FCV, 2h), used also against herpes zoster virus. ${ }^{49}$ Both drugs act as chain terminators after intracellular phosphorylation and integration into the DNA growing chain. Famciclovir additionally requires an initial step of conversion into penciclovir after first-pass metabolism in intestine and liver. ${ }^{49}$
Ribavirin (RBV, 2i) is a guanosine analogue inhibitor of RNA synthesis and metabolism, ${ }^{50}$ which was approved in 1986 for the treatment of respiratory syncytial virus (RSV), and in 2001 against hepatitis C virus (HCV) in combination with interferon. ${ }^{51}$
Acyclovir (ACV, $\mathbf{2 j}$ ) is an acyclic adenosine analogue approved in 1982 for the treatment of HSV, ${ }^{52}$ and later also as an anti-varicella zoster virus agent. ${ }^{46}$ The valine prodrug valaciclovir (VACV, 2k), was approved in 1995 with the same purpose, due to increased bioavailability. ${ }^{52}$

Ganciclovir (GCV, 21), an acyclic guanosine analogue, was approved as an anti cytomegalovirus agent (CMV) in 1989, followed by its valine prodrug valganciclovir (VGCV, 2m), approved in 2001 for the same application due to increase in bioavailability compared to the parent nucleoside, as in the case of valaciclovir. ${ }^{52}$
Cidofovir (CDV, 2n) is an acyclic cytidine derivative bearing a phosphonate group, which allows bypassing of the first monophosphorylation step as this group is recognised as a phosphate group although endowed with increased stability. ${ }^{53}$ This nucleotide analogue was approved in 1996 for the treatment of CMV.
Entecavir (ETV, 20) is a RT and DNA replication inhibitor, approved in 2010 for the treatment of HBV, similar to the thymidine L-isomer telbivudine ( $\mathbf{2 p}$ ) approved in $2006 .{ }^{54,55}$

### 1.3 Resistance and limitations to nucleoside analogue treatment

Resistance to the anticancer activity of NAs can arise after modification in the tumour cell functions and genome mutations. ${ }^{40}$ Resistance to antiviral NAs could similarly derive from mutations in the host cells or from specific mutations in the viral genome. ${ }^{9}$ The rate of mutations in the viral genome is higher than in mammalian genomes due to limited proof reading capacities of viral polymerases, and this provides avenues to overcome resistance to such drugs. Such mutations could affect the interaction between the nucleoside analogue and viral polymerase enzymes. ${ }^{56}$

Cell mutations affect the insurgence of metabolic resistance that might cause insufficient intracellular concentration of the active triphosphate. ${ }^{57}$ This was reported to be due to deficiency in nucleoside transporters, such as the human equilibrative nucleoside transporter 1 (hENT1), which affects the sensitivity to cytarabine (ara-C), fludarabine and cladribine. ${ }^{19}$ Moreover, decreased activity of intracellular nucleoside kinase enzymes has been associated with resistance to NA treatment. ${ }^{25}$ In the case of dCK, an enzyme responsible for the phosphorylation of many NAs, such as cladribine, gemcitabine, fludarabine and ara-C, reduced enzyme activity could derive from either decreased expression of dCK gene or from inactivating mutations in the gene, such as hypermethylation of the promoter region of the gene. ${ }^{40}$ An alternative factor affecting the intracellular concentration of the active NA triphosphates is the increased expression of deactivating enzymes. High levels of $5^{\prime}$ '-nucleotidase enzymes have consistently been associated with poor response to anticancer NAs. ${ }^{30}$ This enzyme hydrolyses the phosphate bond of NA monophosphates, thus counteracting the activity of kinase enzymes. Furthermore, cytidine and adenosine derivatives can be substrates of cytidine or adenosine deaminase enzymes, which catalyse the conversion respectively into inactive uridine or inosine derivatives, as documented after treatment with gemcitabine, Ara-C or Ara-A. ${ }^{40}$ Once in their active triphosphate form, NAs could cause insufficient alterations in nucleic acid strands or to the nucleoside triphosphate pools, i.e. by altered interaction with DNA or RNA polymerases, ${ }^{58}$ or by ineffective inhibition of the RNR enzyme, ${ }^{59}$ responsible for the dNTP pool. Finally, NA treatment failure might derive from defective cell death pathways and p53 exonuclease activity. ${ }^{60}$ The p53 protein possesses exonuclease activity, and is able to excise DNA-incorporated NA. Alterations in the expression of such protein has been related in vitro to resistance to NAs. DNA damage caused by NA can induce the expression of p53, causing induction of pro-apoptotic molecules such as Bax and down regulation of anti-apoptotic proteins such a Bcl-2. ${ }^{40}$

### 1.4 Strategies to overcome NA resistance

Overcoming the dependency of nucleoside analogues on cell mediated activation pathways could lead to a reduction in resistance to therapy. ${ }^{61}$ One of the main documented factors affecting the activity of NA is the dependency on nucleoside kinase enzymes for the first, often rate-limiting, phosphorylation step. ${ }^{40,62}$ Nucleotides could be administered instead of nucleosides, hence bypassing this limiting stage. ${ }^{61}$ However, the much higher hydrophilic nature of the nucleotide (related to the phosphate group, which is acidic and thus
negatively charged at physiological pH ) would decrease its bioavailability and inhibit cell penetration. Moreover, the phosphate group is metabolically labile and could rapidly be cleaved by ecto- 5 '-NT enzymes, such as in the case of fludarabine monophosphate. One strategy to increase the stability of nucleotides is the use of nucleoside phosphonates, where the replacement of the phosphate O-P bond with a phosphonate C-P bond greatly increases the chemical and enzymatic stability of this moiety, still representing a moiety generally recognised by the nucleotide kinase enzymes. ${ }^{63-65}$ However, these molecules are often poorly bioavailable, due to the negative charges on the phosphonate.

A better strategy may be the use of liphophilic substituents to mask the charges on the phosphate and phosphonate groups, and different approaches are detailed in several reviews. ${ }^{61,66-69}$ This could additionally bypass the dependency on nucleoside transporters for cell penetration, due to increased lipophilicity, which would allow passive diffusion through cell membranes. Moreover, some of these strategies were found to inhibit NA inactivation by deaminase enzymes. The attached moieties should be then either chemically or enzymatically cleaved inside the cell, representing prodrugs of the corresponding nucleoside monophosphate analogue.

### 1.5 Monophosphate prodrugs

To overcome the previously mentioned issues correlated with the use of nucleosides as anticancer therapeutics, different approaches have been developed, and some of the most established ones will be described here.

### 1.5.1 $\operatorname{Bis}(\mathrm{POM})$ and $\operatorname{Bis}(\mathrm{POC})$ approach

Farquhar et al. were the first to use the bis(pivaloyloxymethyl) (POM) approach for the preparation of pro-nucleotides. ${ }^{70}$ The intracellular activation of the prodrug moiety depends on carboxyesterase enzymes yielding the free nucleotide.

This approach was used as a phosphate prodrug motif in the antiviral setting, on nucleosides such as dideoxyuridine (ddU) and AZT, ${ }^{71,72}$ although it proved more efficient on phosphonate substrates with bis(POM) adefovir dipivoxil (3a) approved as an anti-HBV agent, ${ }^{73,74}$ and LB80380 ( $\mathbf{3 b}$ ) in clinical trials as an anti-HBV agent (Figure 1.5). ${ }^{75,76}$ The methodology was also applied to FUdR (3c, Figure 1.5) with the aim of improving anticancer activity, but was not progressed in the clinic. ${ }^{77}$


Figure 1.5: Structures of the anti-HBV agents bis(POM) adefovir dipivoxil (3a), LB80380 (3b) and the anticancer agent bis(POM) FUdR (3c).

A modification of this strategy consisted of the introduction of bis(isopropyloxycarbonyloxymethyl) esters (POC). Similar intracellular conversions as for the bis(POM) prodrugs affords the desired nucleotide inside the cell. The application of this prodrug approach yielded the approved anti-HBV agent bis(POC) tenofovir (3d, Figure 1.6)..$^{78,79}$


3d
Figure 1.6: Structure of the anti-HBV agent $b i s(\mathrm{POC})$ tenofovir (3d).

### 1.5.2 $\operatorname{Bis}(S D T E)$ - and $\operatorname{Bis}(S A T E)$ Approach

Imbach and Gosselin explored the application of bis(S-[2-hydroxyethylsulfidyl]-2-thioethyl)- [bis(SDTE)] and the bis(S-acyl-2-thioethyl)- [bis(SATE)] nucleotide prodrug approaches. ${ }^{80-83}$ Both prodrug approaches require intracellular enzymatic cleavage of the pronucleotide moieties, catalysed by a reductase enzyme in the case of the bis(SDTE) and a carboxyesterase enzyme for the $b i s(\mathrm{SATE})$, similar to the $b i s(\mathrm{POM})$ and $b i s(\mathrm{POC})$ approaches. The bis(SATE) approach was applied to the anti-HIV nucleoside analogue AZT (4a, Figure 1.7), retaining good activity in a TK-deficient strain. The bis(SDTE) approach was applied to AZT (4b), yielding lower in vitro anti-HIV activity compared to the $b i s(S A T E)$ approach. ${ }^{84}$


4a


4b


4c

Figure 1.7: Structures of the anti-HIV agents bis(SATE) AZT (4a), bis(SDTE) AZT (4b), and the anticancer agent UA911 (4c).

Furthermore, the application of the bis(SATE) approach to Ara-C generated the agent UA911 (4c), currently involved in anticancer preclinical studies due to better activity on deoxycytidine kinase (dCK)-deficient cells than cytarabine. ${ }^{85}$

This prodrug approach was modified with the introduction of an aromatic or aliphatic amine or an amino acid ester in place of one of the two SATE chains. ${ }^{86}$ The activation of such prodrugs requires one esterase step followed by a putative phosphoramidase-mediated cleavage of the phosphoramidate bond.

One of the most important examples of such a prodrug approach was IDX184, a ${ }^{\prime}$ '-Cmethylguanosine prodrug that entered phase IIb clinical evaluation as an anti-HCV agent. ${ }^{87}$


4 d
Figure 1.8: Structure of the anti-HCV agent IDX184.

Although the progression in the clinic was halted, IDX184 is the only example of a SATE derivative reaching clinical investigations. ${ }^{88}$

### 1.5.3 CycloSal approach

Meier et al. developed this approach and it consists of masking the phosphate group on nucleotides with salicylic alcohols. The prodrug then requires a pH (acid) sensitive chemical hydrolysis mechanism of activation. ${ }^{66,89}$ This method was applied to different NAs, such as d4T (5a, Figure 1.9), ${ }^{90,91}$ AZT, ${ }^{92}$ etc. ${ }^{93}$ Second and third generations of cycloSal analogues have also been developed. The second generation of prodrugs has an esterase cleavable moiety on the aromatic ring that is a substrate of an esterase enzyme inside the cell (5b). This enzyme releases the more polar product, and hence traps it within
the cell. The third generation cycloSal prodrugs contain a geminal carboxylate attached to the aromatic ring, para to the phenoxy phosphate ester (5c). Upon enzymatic ester hydrolysis, this breaks down to an aldehyde, which is electron withdrawing and thus speeds up the chemical hydrolysis to release the nucleotide. The three prodrug approaches are described on the anti-HIV agent d 4 T in Figure 1.9.


Figure 1.9: Structures of first (5a), second (5b) and third (5c) generation cycloSal prodrugs on the anti-HIV agent d4T.

Moreover, the approach was also applied to the anticancer NA cladribine, yielding a compound more active than the parent nucleoside in dCK deficient cells. ${ }^{94}$ So far, none of these agents have progressed to clinical trial.

### 1.5.4 Phospholipid conjugates approach

This prodrug strategy has been developed by Hostetler et al. ${ }^{34}$ and applied to some acyclic nucleoside phosphonates, enhancing their antiviral potency. In this approach, one of the phosphonate negative charges is masked with a phospholipid moiety, such as hexadecyloxypropyl or octadecyloxyethyl groups. This phospholipid moiety mimics lysophosphatidylcholine, a naturally occurring phospholipid, and is hence recognised and cleaved intracellularly by phospholipase C (PLC). In the case of cidofovir, this approach increased the uptake inside the cell by 100 -fold, ${ }^{95}$ and this prodrug is currently involved in a phase III clinical trial study under the name of brincidofovir (6a, CMX001, Figure 1.10) on CMV infected patients. The application of this prodrug approach to tenofovir afforded derivative CMX157 (6b), which was terminated after phase I clinical study having been developed as an anti-HIV agent. ${ }^{96,97}$


6a


6b


6c

Figure 1.10: Phospholipid conjugate approach applied to cidofovir ( $\mathbf{6 a}$ ), tenofovir ( $\mathbf{6 b}$ ) and AZT ( $\mathbf{6 c}$ ).

This structure was also modified yielding fozivudine tidoxil, a thioether lipid prodrug of AZT, which reached phase II clinical trial for the treatment of HIV. ${ }^{98}$

### 1.5.5 HepDirect approach

This method consists of the introduction of a 1-aryl-1,3-propanyl ester (HepDirect) on the monophosphate group of the nucleotide. This group allows the monophosphate nucleoside analogue to be delivered in the liver via cytochrome $\mathrm{P}_{450}$ (CYP450) oxidation of the prodrug moiety. ${ }^{99}$ An important application of such a prodrug approach is pradefovir (7a, Figure 1.11), a derivative of adefovir that was successfully developed to overcome the renal toxicity derived from treatment of HBV affected patients with adefovir dipivoxil. ${ }^{100}$ Moreover, the approach was applied to cytarabine, yielding the phosphate prodrug MB07133 (7b), that has completed phase II clinical studies against primary liver cancer, ${ }^{101}$ and was granted orphan drug designation for the treatment of this malignancy. ${ }^{102}$


Figure 1.11: Structures of HepDirect prodrugs of adefovir (7a) and cytarabine (7b, MB07133).

### 1.5.6 Phosphoramidate diester approach

In this approach, developed by Wagner et al., only one of the two negative charges on the phosphate group of a nucleotide is masked by an aminoacyl methyl ester. The intracellular bioactivation of this prodrug involves the cleavage of the P-N bond by a phosphodiesterase or phosphoramidase enzyme. This approach has been applied to different nucleoside analogues, such as the anticancer agents FUdR (8a) and ara-C (8b), ${ }^{103}$ and the antiviral
analogue AZT (8c), ${ }^{104}$ which was endowed with a better pharmacokinetic profile compared to the parent nucleoside (Figure 1.12).


8a


8b


8c

Figure 1.12: Structures of phosphoramidate diesters derivatives of FUdR (8a), Ara-C (8b) and AZT (8c).

### 1.5.7 Aryloxy phosphoramidate approach

First developed by McGuigan and co-workers, the aryloxy phosphoramidate nucleotide prodrug (ProTide) approach ${ }^{105,106}$ involves masking of the negative charges on the monophosphate group by an aryloxy and an amino acid ester group.

The development of this prodrug strategy followed different steps, described in Figure 1.13 on AZT. At first, the prodrug structure was characterised by the presence of bis(alkyloxy) and haloalkyloxy substituents on the phosphate group (9a), which conferred stability but also prevented the bioactivation inside the cell, thus showing less activity than the parent nucleoside. ${ }^{107-113}$ The introduction of either an amino acid ester in place of one of the two alkyl chains ( $\mathbf{9 b}$ ), ${ }^{113}$ or two aryl moieties $(\mathbf{9 c}),{ }^{114-116}$ were reported to lead to release of free nucleoside inside the cell, and not the desired nucleotide.


R: Me, Et, Pr
$\mathrm{R}_{1}: \mathrm{CCl}_{3}, \mathrm{CH}_{2} \mathrm{~F}, \mathrm{CF}_{3} \mathrm{CH}_{2}$

9a

$\mathrm{R}_{2}$ : Me, Et, Pr, Bu, Hex, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CX}_{3}(\mathrm{X}=\mathrm{H}, \mathrm{F}, \mathrm{CI})$, $\mathrm{R}_{3}$ : $\mathrm{H}, \mathrm{Me}, \mathrm{iPr}, \mathrm{iBu}$ $\stackrel{R}{\text { secBu, }} \mathrm{Pr}, \mathrm{Bn}$
9b


Ar: Ph, p-X-Ph

9c


Ar: Ph, p-X-Ph
$\mathrm{R}_{4}: \mathrm{H}, \mathrm{Me}, \mathrm{iBu}, \mathrm{Bn}$

9d

Figure 1.13: Development of the aryl phosphoramidate (ProTide) approach using AZT.

The final step was the introduction of an aryloxy group and an amino acid ester moiety on the phosphate group (9d), which led to structures with improved in vitro antiviral activity in comparison to AZT. ${ }^{117}$ Application of the ProTide technology in the antiviral and anticancer settings, led to numerous examples of compounds endowed with better potency,
compared to the parent nucleoside. A pivotal example is represented by sofosbuvir ( $\mathbf{9} \mathbf{e}$, Figure 1.14), approved in 2013 for the treatment of HCV. ${ }^{118}$


Figure 1.14: Structures of ProTide analogues that have entered clinical evaluation: sofosbuvir (9e), stampidine (9f), PSI-353661 (9g), INX-08189 (9h), thymectacin (9i), NUC-1031(9j), NUC-3373 (9k), GS7340 (91), GS9131 (9m).

Many ProTides have entered clinical trials, such as the anti-HIV agent stampidine in phase I $(\mathbf{9 f}),{ }^{119}$ the anti-HCV agents PSI-353661 in phase I $(\mathbf{9 g}),{ }^{120}$ and INX-08189 (9h), ${ }^{121}$ in phase II. Despite progression into phase II clinical trials, the development of INX-08189 was halted in 2012 due to the development of severe cardiotoxicity in patients. ${ }^{122}$ The development of PSI-353661 was halted concomitantly, due to structural similarities to INX-08189 and the danger of development of similar toxicity. ${ }^{88}$
Within the anticancer setting, thymectacin has entered phase I/II (9i), ${ }^{123}$ and NUC-1031 is in phase III ( $\mathbf{9 j}$ ), ${ }^{124}$ along with FUdR ProTide NUC-3373 ( $\mathbf{9 k}$ ), which is currently in preclinical development and was announced to enter phase I in 2015. ${ }^{125}$

Moreover, tenofovir alafenamide fumarate (91, TAF, formerly known as GS-7340), ProTide of the antiviral acyclic nucleoside analogue tenofovir has also reached phase I/II development for the treatment of HIV and HBV viruses, ${ }^{126}$ and the phosphonate ProTide GS9131 ( $\mathbf{9 m}$ ) has reached phase II clinical development as an anti-HIV agent. ${ }^{127}$

The initial step of the ProTide putative activation consists of the cleavage of the ester moiety by an enzyme with esterase-type activity, such as lysosomal carboxypeptidase A (cathepsin A, catA), ${ }^{128-130}$ yielding the intermediate $\mathbf{B}$ (Scheme 1.2).


Scheme 1.2: ProTide activation pathway.

The initial ester cleavage is believed to be followed by an intramolecular attack of the newly formed carboxylate anion on the phosphorus atom, with formation of a putative five membered cyclic carboxylic phosphorus mixed anhydride, and resulting in elimination of the aryloxy group. ${ }^{131-133}$ Due to the likely instability of such an intermediate, hydrolysis of the five membered ring may be spontaneous, with water attack on the phosphorus centre, and opening of the ring. ${ }^{131}$ Finally, the cleavage of the P-N bond, mediated by an intracellular phosphoramidase-type enzyme, releases the monophosphate. ${ }^{103,104,134}$ This step may be performed by histidine triad nucleoside-binding protein 1 (Hint enzyme 1), which belongs to the HIT superfamily. ${ }^{135}$

The modifiable ProTide moieties are represented by the aryl group (Ar), the amino acid $\left(R_{1}, R_{2}\right)$ and the ester $(R)$, giving rise to vast sets of combinations.


Figure 1.15: Common structure of ProTide nucleotide prodrugs with highlighted key modifiable residues.

Concerning the ester moiety (moiety R, Figure 1.15), many different linear (such as methyl, ethyl, pentyl) and branched (such as isopropyl, tert-butyl, neopentyl) groups were used, along with the benzyl group. In most programs, the benzyl moiety proved to yield the most active prodrugs, whilst in the case of tert-butyl group, the prodrugs were found to be mostly inactive. ${ }^{62,106}$ Aryl moieties (Ar) such as differently substituted phenol groups were investigated and good results were obtained when using electron-withdrawing groups such as $p$ - Br . The naphthol functionality was used in many different programs, yielding promising activity as in the case of NUC-3373. The amino acids $\left(\mathbf{R}^{\mathbf{1}}, \mathbf{R}^{\mathbf{2}}\right)$ were extensively
explored, with a key moiety represented by L-alanine, used so far in all ProTides in clinical development. The use of unnatural D -alanine and other D -amino acids usually yielded inactive ProTides. ProTides bearing glycine were usually poorly active, while the alkylation to dimethylglycine afforded more potent analogues. ${ }^{136}$

### 1.5.8 Cyclic phosphate prodrugs

3',5'-Cyclic phosphoramidate prodrugs were developed with the intent of reducing the potential toxicity arising from the release of an aryloxy moiety via the ProTide approach. This method was applied to $2^{\prime}$-C-methylcytidine ( $\left.\mathbf{1 0 a}, 2^{\prime}-C-M e-C\right)$ and FUdR (10b) (Figure 1.16), but was not progressed further. ${ }^{137,138}$


Figure 1.16: Structures of $3^{\prime}, 5^{\prime}$-cylic phosphoramidate prodrugs of 2'-C-Me-C (10a) and FUdR (10b).

Interesting results were obtained with $3^{\prime}, 5$ '-cyclic ester prodrugs, reported to be activated by enzymatic P-O-dealkylation catalysed by CYP3A4 and then cleavage of the 3'-P-O bond by phosphodiesterase enzymes. ${ }^{139}$


Figure 1.17: Structure of the anti-HCV agent PSI-352938 (10c).

The application of this prodrug approach yielded PSI-352938 (10c, Figure 1.17), an antiHCV agent that entered phase I trials, ${ }^{139}$ although its progression into the clinic was halted due to hepatic side effects. ${ }^{140}$

### 1.5.9 Phosphorodiamidate approach

McGuigan et al. synthesised phosphorodiamidate derivatives of AZT and 3'-fluoro-3'deoxythymidine (FdT) by using two amino acid groups, ${ }^{141,142}$ which allowed the presence of an achiral phosphorus center, therefore a single isomer. The proposed activation of such prodrugs is similar to the ProTide approach, requiring initial step of cleavage of the ester moiety catalysed by a carboxypeptidase enzyme, inducing spontaneous cyclisation into a five membered ring (Scheme 1.3). ${ }^{143}$ Attack of the carboxylate ion to the phosphorus center triggers the release of one amino acid. A phosphoramidase enzyme finally catalyses the release of the nucleoside monophosphate. One additional positive aspect of this prodrug strategy is the release of two amino acid ester moieties during the activation process, avoiding the generation of potentially toxic aryloxy groups.


Scheme 1.3: Mechanism of activation of phosphorodiamidate nucleoside prodrugs.

Our group was involved in the synthesis of phosphorodiamidates of $6-O$-alkyl-2'-Cmethylguanosine ( $6-O$-alkyl-2'-C-Me-G) as anti HCV-agents, and different compounds such as 11a gave similar results to the phosphoramidate INX189 in in vitro assay against HCV replication (Figure 1.18). ${ }^{143}$


Figure 1.18: Structures of phosphorodiamidate prodrugs of $6-O$-alkyl-2'-C-Me-G (11a), ddA (11b), ABC (11c), ACV (11d).

A more recent study focused on the application of this approach on a broad selection of antiviral and anticancer nucleoside analogues, ${ }^{144}$ with improvement of the in vitro antiHIV activity of phosphorodiamidates of ddA (11b), ABC (11c) and ACV (11d), compared to the parent nucleosides.

### 1.6 Summary of the phosphate prodrug technologies

A summary of all the different monophosphate prodrug approaches described is reported in Table 1.1, along with the activation pathway required and whether any analogue has been taken to human trials.

| Prodrug class | Activation step required | Taken to human trials |
| :---: | :---: | :---: |
| Bis(POM)/(POC) | Esterase | $\checkmark$ |
| Bis(SATE)/(SDTE) | Esterase/(reductase + esterase) | $x$ |
| SATE phosphoramidate | Esterase + amidase | $\checkmark$ |
| CycloSal | Chemical | $x$ |
| Phospholipid conjugate | Phospholipase C | $\checkmark$ |
| HepDirect | CYP450 | $\checkmark$ |
| Phosphoramidate diester | phosphoramidase | $x$ |
| ProTide | Carboxypeptidase + phosphoramidase | $\checkmark$ |
| Cyclic phosphoramidate | Phosphoramidase + phosphodiesterase | $x$ |
| Cyclic phosphate ester | CYP3A4 + phosphodiesterase | $\checkmark$ |
| Phosphorodiamidate | Carboxypeptidase + phosphoramidase | $x$ |

Table 1.1: Summary of the phosphate prodrug approaches.

### 1.7 Aim of the work

Nucleoside analogues are therapeutics with a wide application in anticancer and antiviral therapies. Their use if often limited by the emergence of resistance to the treatment, due to different factors such as the down-regulation of activating pathways, or transporter proteins and up-regulation of inactivating enzymes. Many limiting steps can be overcome by the use of a lipophilic nucleotide prodrug, and our aim was to improve the biological profile of already known nucleoside analogues by applying the ProTide approach.
The synthesis of ProTides of different nucleoside analogues was planned, along with the evaluation of each analogue in vitro as anticancer agents. Selected compounds were also tested as antiviral agents.

NMR and molecular modelling studies on the enzymatic activation of selected compounds were planned in order to further validate the biological data, and respectively involved carboxypeptidase-Y and Hint enzymes.

Assays of the stability of promising derivatives were also considered, in mammalian plasma, serum and liver microsomes and against deactivating enzymes such as adenosine deaminase.

On the most promising ProTides families of known nucleoside analogues, modifications of the nucleotide structure at the nucleobase, sugar and phosphate level were also designed in order to gain an advantage over the already active analogues.

Anticancer evaluation of the most active derivatives was planned on the cancer stem cell population of the KG1a cell line.

The target nucleoside for the application of prodrug approaches were (Figure 1.19):

- 3'-Deoxyadenosine (chapter three).
- 2-Fluoro-3'-deoxyadenosine (chapter three).
- 8-Chloroadenosine (chapter four).
- 8 -Chloro-2'-deoxyadenosine (chapter four).
- 2'-C-cyano-2'-deoxy-1- $\beta$-D-arabino-pentofuranosylcytosine (chapter five).
- Trifluorothymidine (chapter six).
- 2'-Mercapto 2'-deoxyuridine (chapter seven).
- 2'-Trifluoromethylthio 2'-deoxyuridine (chapter seven).


Figure 1.19: Visual outline of the content of each chapter.

## 2 Synthesis of nucleotide prodrugs

### 2.1 ProTide synthesis

The ProTide approach was elected as our primary nucleotide prodrug method and applied to different nucleoside analogues. As already mentioned in Chapter 1, ProTides consist of a nucleoside monophosphate bearing different substituents on the phosphate group. One of these substituents is an aromatic moiety and the other is an amino acid ester, linked to the phosphorus through a phosphoramidate bond (12, Figure 2.1)..$^{62,105,106}$


12

Figure 2.1: ProTide general structure (12).

Different methodologies were reported for the synthesis of ProTides, although the most commonly used one is the coupling of the nucleoside analogue (NA) with a phosphorochloridate reagent, reported in Scheme 2.1. ${ }^{145}$


Scheme 2.1: Synthetic pathway to ProTide (12). Reagents and conditions: (a.) $\mathrm{POCl}_{3}$ (1 eq), $\mathrm{Et}_{3} \mathrm{~N}$ (1 eq), $\mathrm{Et}_{2} \mathrm{O},-78^{\circ} \mathrm{C}$ to $\mathrm{rt}, 3 \mathrm{~h}(90-91 \%)$; (b.) [i] $\mathrm{ROH}(15 \mathrm{eq}), \mathrm{SOCl}_{2}(2 \mathrm{eq})$, toluene, reflux, $16 \mathrm{~h},(88-98 \%)$; [ii] ROH (5 eq), pTSA (1.1 eq), toluene, reflux, 16h, (40-96\%); (c.) $\mathrm{Et}_{3} \mathrm{~N}(2 \mathrm{eq}), \mathrm{CH}_{2} \mathrm{Cl}_{2},-78{ }^{\circ} \mathrm{C}$ to rt, 2-4 h. (75-99\%) (d.) $t \mathrm{BuMgCl}(1 \mathrm{M}$ in THF, $1-3 \mathrm{eq}$ ), THF, rt, overnight or NMI ( 5 eq ), THF, rt, 16 h (see yields throughout the work). NA: nucleoside analogue.

The key moiety is the phosphorochloridate (17) that is coupled with the nucleoside analogue only in the last step (d.). The phosphorochloridate is synthesised by coupling aryl phosphorodichloridates (14) and amino acid ester salts (16). The aryl
phosphorodichloridate is obtained by reacting aryl alcohol (13) and phosphorus oxychloride $\left(\mathrm{POCl}_{3}\right)$, while the amino acid ester (16) is synthesised by esterification of the amino acid (15) with an alcohol.
All these different moieties can vary greatly, leading to a broad variability of ProTide structures and to the modulation of both biological activity and chemical stability of the structures. In this work, different modifications were attempted on the general ProTide structure, as reported in Figure 2.2.


Figure 2.2: Different modifications of ProTides developed in this work.

### 2.1.1 Phosphorodichloridate synthesis

Phosphorochloridates were synthesised in a coupling reaction between an amino acid ester salt (15) and an appropriate phosphorodichloridate (14) in the presence of triethylamine $\left(\mathrm{Et}_{3} \mathrm{~N}\right)$ as a base (Scheme 2.1, step c.). ${ }^{146}$

The phosphorodichloridate is the phosphorylating agent, which determines the properties of the aryl moiety in the final ProTide. Phenyl- (14a), 1-naphthyl- (14b) and the o-ethyl 3-(2-hydroxyphenyl)propanoylphenyl phosphorodichloridate (14c) were used in this work (Figure 2.3); unlike 14a, both $\mathbf{1 4 b}$ and $\mathbf{1 4 c}$ are not commercially available and were synthesised according to Scheme 2.1 (step a.), in approximately $90 \%$ yields (Table 2.1).


14a


14b


14c

Figure 2.3 Aryloxy phosphorochloridate moieties (14a-c) used within this work as ProTide building blocks.

In order to obtain compounds $\mathbf{1 4 b}$ and $\mathbf{1 4 c}$, phosphorus oxychloride $\left(\mathrm{POCl}_{3}\right)$ was added to a stirred solution of the aryl alcohol (respectively 1-naphthol 13b, and ethyl 3-(2hydroxyphenyl)propanoate 13c, Figure 2.4) in diethyl ether, in the presence of triethylamine at $-78{ }^{\circ} \mathrm{C}$ (Scheme 2.1, a.) and the reaction was monitored by ${ }^{31} \mathrm{P}$ NMR.


13b


13c

Figure 2.4: Structures of aryl alcohols used for the synthesis of phosphorodichloridates $\mathbf{1 3 b}$ and $\mathbf{1 3 c}$.

The reaction was complete with the disappearance of the ${ }^{31} \mathrm{P}$ NMR peak corresponding to $\mathrm{POCl}_{3}\left(\delta \mathrm{P}=4.27, \mathrm{CDCl}_{3}, 202 \mathrm{MHz}\right.$ ), paralleled by only the presence of the peak corresponding to the appropriate phosphorodichloridate (Table 2.1). ${ }^{105}$ Once the reaction was complete the triethylammonium chloride salt was filtered off and the solvent was removed in vacuo to give compounds $\mathbf{1 4 b}$ and $\mathbf{1 4 c}$ as clear oils, which were used in the next step without further purification.

| Compound | Ref. | Aryl | Yield \% | ${ }^{\mathbf{3 1}} \mathbf{P}$ P-NMR ס |
| :---: | :---: | :---: | :---: | :---: |
| $\mathbf{1 4 b}$ | 147 | 1-Naphthol | $91 \%$ | 3.69 |
| $\mathbf{1 4 c}$ |  | Ethyl 3-(2-hydroxyphenyl)propanoate | $90 \%$ | 3.22 |

Table 2.1 Reaction yields for the synthesis of phosphorodichloridates (14b and 14c).

Ethyl 3-(2-hydroxyphenyl)propanoate (13c) was synthesised in quantitative yield by treating 3,4-dihydrocoumarin (18) with a catalytic amount of 2,3-dichloro-5,6-dicyano-1,4benzoquinone (DDQ) in ethanol (Scheme 2.2). ${ }^{149}$


Scheme 2.2: Synthesis of ethyl 3-(2-hydroxyphenyl)propanoate (13c). Regents and conditions: DDQ (0.07 eq), $\mathrm{EtOH}, \mathrm{rt}, 8 \mathrm{~h}, 100 \%$.

This specific aryl group was chosen because ProTides bearing this moiety were shown to have good antiviral activity, suggesting efficient processing to the monophosphate. ${ }^{150}$ Moreover, the introduction of an ethyl propanoic ester side chain at the ortho position of the phenol portion is supposed to release during the ProTide activation hydroxyphenyl propanoic acid, which is a known non-toxic metabolite of dihydrocoumarin, a common flavouring agent widely used in food and cosmetics. ${ }^{150}$

### 2.1.2 Amino acid ester synthesis

The synthesis of amino acid esters that were not commercially available was performed according to two different synthetic routes, depending on the nature of the alcohol used for the esterification reaction.

In the first method, the amino acid and the appropriate alcohol were heated under reflux overnight in the presence of thionyl chloride. After evaporation of volatiles, pure amino acid esters were obtained as white hydrochloride salts (16a-c, Table 2.2, Scheme 2.1, b.[i]). This method is preferred when using low boiling point alcohols for the esterification, because it is easier to reach the reflux temperature required to promote the reaction.

| Common structure | Cpnd | Ref. | $\mathbf{R}^{1}$ | $\mathbf{R}^{2}$ | AA | R (ester) | Yield \% | X |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 16a | 152 | $i \mathrm{Bu}$ | H | L-Leu | $n$-Pen | 98 \% | $\mathrm{Cl}^{-\mathrm{a}}$ |
|  | 16b | - | $i \mathrm{Bu}$ | H | L-Leu | Et | 94 \% | $\mathrm{Cl}^{-\mathrm{a}}$ |
|  | 16c | - | Bn | H | L-Phe | Et | 88 \% | $\mathrm{Cl}^{-\mathrm{a}}$ |
|  | 16d | 151 | Me | H | L-Ala | $n$-Pen | 80 \% | $p$ TSA $^{\text {b }}$ |
|  | 16e | 151 | Me | H | L-Ala | cHex | 88\% | $p \mathrm{TSA}^{\mathrm{b}}$ |
|  | 16 f | - | H | H | Gly | $n$-Pen | 96 \% | $p \mathrm{TSA}^{\mathrm{b}}$ |
|  | 16 g | - | H | H | Gly | cHex | 88 \% | $p \mathrm{TSA}^{\text {b }}$ |
|  | 16h | - | $i \mathrm{Bu}$ | H | L-Leu | cHex | 40 \% | $p$ TSA $^{\text {b }}$ |
|  | 16i | 152 | Bn | H | L-Phe | $n$-Pen | 95 \% | $p$ TSA $^{\text {b }}$ |
|  | 16j | - | Bn | H | L-Phe | cHex | 83 \% | $p$ TSA $^{\text {b }}$ |

Table 2.2: Summary of amino acid esters synthesised (16a-j) using the two methods described above and isolated yields. ${ }^{\text {a }}$ amino acid ester synthesised through procedure [i]. ${ }^{\mathrm{b}}$ amino acid ester synthesised through procedure [ii].

In the second method, the amino acid and the appropriate alcohol were suspended in toluene and heated at reflux overnight, in the presence of para-toluenesulfonic acid monohydrate ( $p$-TSA, Scheme 2.1, b[ii]).

A Dean-Stark apparatus was fitted to the reaction flask to remove the water produced in the reaction. The corresponding pure amino acid esters were obtained through partial evaporation of the volatiles and diethyl ether-induced precipitation as white tosylate ( $p$ TSA) salts ( $\mathbf{1 6 d} \mathbf{- j}$, Table 2.2). A smaller volume of alcohol is used in comparison to the previous method and therefore this procedure is preferred when using high boiling point alcohols, in order to ease the evaporation work-up.
All newly synthesised amino acid esters, were used for the preparation of some of the synthesised phosphorochloridates ( $\mathbf{1 7} \mathbf{c}, \mathbf{d}, \mathbf{f}-\mathbf{h}, \mathbf{j}-\mathbf{l}, \mathbf{n}-\mathbf{p}$ ).

### 2.1.3 Phosphorochloridate synthesis

Phosphorodichloridates were coupled at low temperature $\left(-78{ }^{\circ} \mathrm{C}\right)$ in the presence of $\mathrm{Et}_{3} \mathrm{~N}$ in anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, with the different amino acid ester hydrochlorides or tosylate salts (Scheme 2.4, c.). After 1 hour the reaction mixture was allowed to reach room temperature. The formation of phosphorochloridate was monitored by ${ }^{31} \mathrm{P}$ NMR. The phosphorochloridates derived by tosylate amino acid salts were purified from the triethylamine tosylate salt by flash chromatography using a mixture of hexane/ethyl acetate as an eluent system. A simple filtration was sufficient to purify the phosphorochloridates derived by hydrochloric amino acid salts from the triethylamine chloridate salt.

Most of the phosphorochloridates obtained during the coupling method were represented by diasteromeric mixtures. ${ }^{31} \mathrm{P}$ NMR of synthesised compounds showed signal splitting except for the glycine and dimethylglycine species, which were obtained as a mixture of enantiomers and therefore showed only one single peak in the phosphorus spectra (Table 3.3).

| Common structure | Cpnd | Ref. | Ar | $\mathrm{R}^{1}$ | $\mathbf{R}^{2}$ | AA | R | Yield | 31P NMR $\delta$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 17a* | 147 | 1-Naph | Me | H | L-Ala | Bn | 87 \% | 8.2; 8.0 |
|  | 17b* | 147 | 1-Naph | Me | H | L-Ala | Et | 99\% | 8.3; 8.0 |
|  | 17c | 151 | 1-Naph | Me | H | L-Ala | cHex | 92 \% | 8.3; 7.9 |
|  | 17d | 151 | 1-Naph | Me | H | L-Ala | $n$-Pen | 85 \% | 8.6; 8.7 |
|  | 17e* | 152 | 1-Naph | $i \mathrm{Bu}$ | H | L-Leu | Bn | 89 \% | 8.3; 8.0 |
|  | 17f | 152 | 1-Naph | $i \mathrm{Bu}$ | H | L-Leu | $n$-Pen | 89 \% | 8.8; 8.5 |
|  | 17 g | - | 1-Naph | $i \mathrm{Bu}$ | H | L-Leu | cHex | $90 \%$ | 8.8; 8.5 |
|  | 17h | - | 1-Naph | $i \mathrm{Bu}$ | H | L-Leu | Et | 85 \% | 8.7; 8.5 |
|  | 17i* | - | 1-Naph | H | H | Gly | Et | $95 \%$ | $9.3{ }^{\text {i }}$ |
|  | 17j | 148 | 1-Naph | H | H | Gly | Bn | 80 \% | $8.9{ }^{\text {i }}$ |
|  | 17k | - | 1-Naph | H | H | Gly | $n$-Pen | 88 \% | $9.1{ }^{\text {i }}$ |
|  | 171 | - | 1-Naph | H | H | Gly | cHex | 77 \% | $9.0{ }^{\text {i }}$ |
|  | 17m* | 148 | 1-Naph | Bn | H | L-Phe | Bn | 87 \% | 8.29; 8.17 |
|  | 17n | 152 | 1-Naph | Bn | H | L-Phe | $n$-Pen | 82 \% | 8.35; 8.25 |
|  | 170 | - | 1-Naph | Bn | H | L-Phe | cHex | 86 \% | 8.45; 8.35 |
|  | 17p | - | 1-Naph | Bn | H | L-Phe | Et | $90 \%$ | 8.36; 8.23 |
|  | 17q** | 154 | 1-Naph | Me | Me | DMG | Me | $75 \%$ | $6.02{ }^{\text {i }}$ |
|  | 17\%* | 147 | Ph | H | H | Gly | Bn | 80 \% | $9.04{ }^{\text {i }}$ |
|  | 17s* | 147 | Ph | Me | H | L-Ala | Bn | 82\% | 7.5; 7.8 |
|  | 17t** | - | Ph | Me | H | L-Ala | Neop | 79\% | 7.5; 7.8 |
|  | 17u** | 151 | Ph | Me | H | L-Ala | $n$-Hex | 82\% | 7.6; 7.9 |
|  | 17v* | 150 | $o-\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{CO}_{2} \mathrm{EtPh}$ | Me | H | L-Ala | Bn | 92 \% | 7.70; 7.53 |

Table 2.3: Yields and ${ }^{31} \mathrm{P}$ NMR shifts of synthesised phosphorochloridates (17a-v). ${ }^{\text {a }}$ Spectra were performed in $\mathrm{CDCl}_{3}, 202 \mathrm{MHz}$. *Amino acid ester was purchased (SigmaAldrich/Fisher) ** Amino acid esters were provided by NuCana Biomed (as HCl salts). ${ }^{\mathrm{i}}$ No diasteromeric splitting ${ }^{31} \mathrm{P}$ NMR as compounds are a mixture of enantiomers.

### 2.2 Phosphoramidate synthesis

The phosphoramidates were obtained by coupling of nucleoside analogues with the appropriate phosphorochloridates in the presence of either tert-butyl magnesium chloride $(t \mathrm{BuMgCl}){ }^{6}$ or $N$-methylimidazole (NMI) ${ }^{17}$ in anhydrous THF, at room temperature, over 12-16 hours (Scheme 2.1). The use of $t \mathrm{BuMgCl}$ aimed at the deprotonation of the 5 'hydroxyl group on the sugar, which would then react with the phosphorochloridate. NMI, instead, functioned as an activator of the phosphorochloridate by replacing chlorine as a leaving group.


Scheme 2.3: General synthesis of phosphoramidates (12). Reagents and Conditions: a) $t-\mathrm{BuMgCl}(1-3 \mathrm{eq})$, THF, rt, overnight/ NMI (5 eq), THF, rt, overnight.

The Grignard reagent $t \mathrm{BuMgCl}$ was reported to act as a specific activator for $O$-selective phosphorylation. ${ }^{154}$ Despite the selectivity towards the alcohol groups, $t \mathrm{BuMgCl}$ mediated coupling reactions target both the primary and the secondary alcohols on unprotected nucleosides. NMI mediated coupling reactions favour the primary alcohol, although in some cases leading to poor reaction yields, caused either by poor solubility of the substrate in the reaction solvent, or by parallel formation of side products from the coupling on secondary hydroxyl functions. Strategies involving initial protection of the secondary hydroxyl groups on the sugar moiety, followed by coupling reaction and final deprotection step, often improved the reaction outcome, and will be described in the next chapters.

### 2.3 Phosphorodiamidate synthesis

The phosphorodiamidate approach is an alternative method to the already mentioned aryl phosphoramidate to mask the negative charges on the nucleotide molecule (chapter 1). The synthetic strategy applied in this work was adapted from work by Yoshikawa et al. ${ }^{144,155,}$ 156


19
20
Scheme 2.4: General synthesis of phosphorodiamidates. Reagents and Conditions: a) $\mathrm{POCl}_{3}(1 \mathrm{eq})$, trialkylphosphate (b) amino acid ester $p$-TSA or HCl salt ( 10 eq ), anhydrous DIPEA ( 5 eq ), anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}$.

According to the mentioned studies, in a one pot reaction the unprotected nucleosides were suspended in trialkyl phosphate, and phosphorus oxychloride is added in order to obtain
predominantly the $5^{\prime}$ 'phosphorodichloridate (19). The formation of this intermediate is monitored by ${ }^{31} \mathrm{P}-\mathrm{NMR}$ (appearance of a peak $\sim 7 \mathrm{ppm}$ corresponding to the 5 'dichlorophosphate, disappearance of the $\sim 4.27 \mathrm{ppm}$ peak of $\mathrm{POCl}_{3}$ ). Due to the relative instability of these intermediates, the isolation was not accomplished, and the compounds were immediately coupled with the appropriate amino acid ester salts in the presence of diisopropyl ethyl amine (DIPEA). The progression of the reaction was monitored again by ${ }^{31}$ P-NMR: the appearance of a single peak in the range of 13-17 ppm, depending on the amino acid used for the synthesis, indicated the formation of the product $\mathbf{2 0}$.

## 3 3'-Deoxyadenosine analogues

### 3.1 3'-Deoxyadenosine prodrugs

### 3.1.1 Background

3'-Deoxyadenosine ( 3 'dA, cordycepin, 21) is a nucleoside analogue of adenosine that lacks the 3 '-hydroxyl group on the ribose moiety (Figure 3.1).
$3^{\prime} \mathrm{dA}$ is one of the major bioactive substances produced by Cordyceps militaris, a parasitic fungus used for traditional Chinese medicine because of its activity on the immune system, and anti-aging and anti-tumour effects. ${ }^{157,158,159,160}$


Figure 3.1: Structure of 3'dA (21).

Because of its structure, $3^{\prime} \mathrm{dA}$ and its triphosphate form could potentially interfere with any process respectively requiring adenosine or adenosine triphosphate (ATP), 3'dA was suggested to enter the cell via the equilibrative nucleoside transporter 3 (hENT3), ${ }^{161}$ and to be phosphorylated by adenosine kinase (AK) to 3'-deoxyadenosine monophosphate (3'dAMP) and then to the active metabolite 3 '-adenosine triphosphate ( 3 '-dATP) by the combined actions of adenosine monophosphate kinase (AMPK) and nucleoside diphosphate kinase (NDPK) (Scheme 3.1, A). ${ }^{162}$
In addition, $3^{\prime} \mathrm{dA}$ is known to be quickly deaminated by adenosine deaminase (ADA), and rapidly metabolised to an inactive metabolite, 3'-deoxyinosine (3'-dIno, Scheme 3.1, В). ${ }^{163,164}$


Scheme 3.1: (A) Mechanism of intracellular activation of 3 ' dA. (B) Inactivation of 3' dA by ADA enzyme.

Following 3'dA intracellular phosphorylation, 3'-dATP was proposed to be a chain terminator for transcription, but it was reported that total nuclear mRNA precursors are less affected than cytoplasmic mRNA. ${ }^{165,166}$ In contrast, it has been demonstrated to be an efficient chain terminator for mRNA polyadenylation by poly (A) polymerases, thus causing mRNA instability. ${ }^{167,162,168,166}$ In addition, some studies indicate that 3 ' dA may act through adenosine receptors ${ }^{169-171,172}$ causing overproduction of reactive oxygen species (ROS) and apoptosis induction or cause inhibition of poly (ADP) ribose polymerase (PARP). ${ }^{173,167}$

The molecular effects generated by 3'dA translate into numerous biological applications of this nucleoside analogue reported to date, including anti-inflammatory activity. ${ }^{174,167,175}$ inhibition of platelet aggregation ${ }^{176}$ and anti-fungal ${ }^{177}$ and anti-trypanosomal activity, ${ }^{178,179,180}$ along with antiviral properties. ${ }^{181-183}$ However, this nucleoside has been studied most extensively as an anti-cancer agent. As such, 3'dA was reported to induce cytotoxic activity in various types of human cancer cells such as breast cancer, ${ }^{173}$ head and neck, ${ }^{184}$ colorectal, ${ }^{185,186}$ thyroid carcinoma cells, ${ }^{187}$ multiple myeloma cells, ${ }^{188}$ leukaemia ${ }^{171,189}$ and lymphoma cells. ${ }^{190}$ In addition, the inhibitory effect was demonstrated on haematogenic metastasis of mouse melanoma cells ${ }^{170}$ and lung carcinoma cells. ${ }^{169,191}$ Importantly, 3 ' dA has completed two phase I clinical studies in patients with refractory acute lymphocytic or chronic myelogenous leukaemia. ${ }^{192}$ Because of the very quick inactivation by ADA, during this study the administration of 3 ' dA was combined with the ADA-inhibitor pentostatin (deoxycoformycin, dCF, Figure 3.2) and the same combination
was involved in a phase $\mathrm{I} / \mathrm{II}$ clinical trial in patients with refractory terminal deoxynucleotidyl transferase (TdT)-positive leukaemia, although the status of the study is currently unknown. ${ }^{193}$


Figure 3.2: Adenosine deaminase inhibitor deoxycoformycin (dCF).

Although the combination of $3^{\prime} \mathrm{dA}$ and dCF was reported as effective both in vivo and in vitro, ${ }^{179,189}$ toxicity studies revealed that dogs treated with this combination of drugs presented toxicities to lymphoid tissue (lymphopenia and thymus lymphoid depletion), thrombocytopaenia, and decreased food consumption. ${ }^{194}$
The aim of this work is to apply a nucleotide prodrug approach to 3 ' dA in order to enhance the anticancer potential of this nucleoside. Some research groups have worked on the synthesis of 3 'dA prodrugs with the aim of bypassing the ADA-catalysed catabolism affecting the activity of this nucleoside analogue and enhance the pharmacokinetic properties. ${ }^{195,196}$ A patent disclosed the synthesis and anticancer evaluation of different $\mathrm{N}-6$ thioaminale prodrugs of 3 ' dA with the aim of bypassing ADA catabolism. ${ }^{197}$ In addition, some dinucleoside phosphate prodrugs of cordycepin and AZT were also synthesised as prodrugs of both $3^{\prime} \mathrm{dA}$ and AZT and proved to be weak HIV agents. ${ }^{198}$ This work focussed on the application of the ProTide ${ }^{105}$ and phosphorodiamidate ${ }^{67}$ phosphate prodrug approaches to 3'dA (Figure 3.3). Nucleosides bearing substitutions on the $5^{\prime}$ ' OH group of the sugar moiety were already reported to be immune to ADAcatabolism. ${ }^{199}$ Moreover ProTides were proved not to be substrates of nucleoside deaminase enzymes, ${ }^{151}$ and therefore this approach can be an interesting way to bypass ADA-deamination, along with the increasingly well established ability of ProTides to overcome the dependency on AK for the first phosphorylation step, whose downregulation was reported to induce resistance to $3^{\prime} \mathrm{dA} .{ }^{200}$



Figure 3.3: General structure of cordycepin prodrugs: ProTides and phosphorodiamidates.

In conclusion, the ProTide approach would make unnecessary the combination of dCF, therefore avoiding the toxicity risks connected with this treatment and the dosing and pharmacokinetic complications of dual drug administration.

### 3.1.2 Synthesis of 3'dA

Many different strategies for the synthesis of cordycepin are reported in the literature.
In some reports 3'dA is synthesised starting from non-nucleoside derivatives; ${ }^{201-204}$ however the synthesis requires multiple steps and is often low yielding. Other research groups developed a methodology requiring adenosine (Ado) as a starting material. The methodology by Meier et al. was initially applied in this work. This procedure involved a 4 step synthesis with an overall $55 \%$ yield, ${ }^{205}$ and was adapted from a similar procedure for the synthesis of 2'-deoxynucleosides. ${ }^{206}$ This procedure requires the selective protection of the 2'- and 5'-hydroxyl groups in the riboside moiety of adenosine (22) with tertbutyldimethylsilyl chloride (TBDMSCl) (Scheme 3.2). This intermediate (23) is then converted into the 3'-phenoxythiocarbonyl derivative (24) by treatment with phenoxythiocarbonyl chloride (PTC-Cl) and dimethylaminopyridine (DMAP) in acetonitrile $\left(\mathrm{CH}_{3} \mathrm{CN}\right)$. In the final step this product is transformed into protected 3'dAdo (25) by radical deoxygenation (Barton-McCombie reaction) with azobisisobutyronitrile (AIBN) and tributyltinstannane ( $n$ - $\mathrm{Bu}_{3} \mathrm{SnH}$ ). ${ }^{205,207,208}$


Scheme 3.2: First synthetic strategy for the synthesis of TBDMS-protected 3'dA (25). Reagents and conditions: (a.) see Table 3.1 (b.) PTC-Cl, DMAP, $\mathrm{CH}_{3} \mathrm{CN}$, rt, 16h. (c.) AIBN, $n$ - $\mathrm{Bu}_{3} \mathrm{SnH}$, toluene, $3 \mathrm{~h}, 75$ ${ }^{\circ} \mathrm{C}$.

For the first selective protection step, ${ }^{205,207}$ adenosine (22) was treated with 2.2 equivalents of TBDMSCl and an equimolar amount of $\mathrm{AgNO}_{3}$ in THF in the presence of 5 equivalents of pyridine (Table 3.1. entry 1). This procedure was reported by Ogilvie et al. to yield $2^{\prime}, 5^{\prime}$-bis- $O$-TBDMS-adenosine as the main product, ${ }^{209,210}$ but in our hands this attempt yielded only $5^{\prime}$-TBDMS-adenosine in quantitative yield. In order to improve the reaction outcome, the solubility of the reagents was enhanced by the use of a $1: 1 \mathrm{v} / \mathrm{v}$ mixture of THF and DMF and warming up the suspension to $30{ }^{\circ} \mathrm{C}$ until the formation of a clear solution.

| Entry | Reagents | Pyridine | Solvent | T | Yield |
| :---: | :---: | :---: | :---: | :---: | :---: |
| $\mathbf{1}$ | TBDMSCl (2.2 eq), $\mathrm{AgO}_{3}(2.2 \mathrm{eq})$ | 5 eq | THF | rt | - |
| $\mathbf{2}$ | $\mathrm{TBDMSCl}(2.2 \mathrm{eq}), \mathrm{AgO}_{3}(2.2 \mathrm{eq})$ | - | $\mathrm{THF} / \mathrm{DMF}(1: 1)$ | $30^{\circ} \mathrm{C}$ | $15 \%$ |

Table 3.1: Reagents and conditions for the selective $2^{\prime}, 5^{\prime}$-TBDMS-protection of adenosine.

Although allowing the formation of the desired product (23), the yield was still low (15\%), and the $3^{\prime}, 5^{\prime}$-regioisomer and $5^{\prime}$ 'silylated derivative could be recovered in lower yields (Table 3.1. entry 2). This step was not optimised further, and compound 23 was used for the second step (Scheme 3.2. step b.) of 3'dA synthesis. This intermediate was treated with 3 equivalents of phenoxythiocarbonyl chloride ( $\mathrm{PTC}-\mathrm{Cl}$ ) in the presence of 6 equivalents of DMAP. The reaction was attempted in different scales but never yielded sufficient amounts of the desired product 24 but only traces of compound whose purification was difficult due to similar retention factor (Rf) on silica gel to other impurities and the starting material. Because of the lack of success of this strategy, an alternative strategy reported by Robins et al. was followed, requiring first conversion of adenosine (22) into 9-(2,3-anhydro- $\beta$-D-ribofuranosyl)adenine (26), with final metal hydride mediated reductive opening of the epoxide ring, which was reported to afford 3'dA (21) in 91\% yield (Scheme 3.3). ${ }^{211-214}$ Following this procedure, adenosine (22) was reacted with $\alpha$-acetoxyisobutyryl bromide ( $\alpha-\mathrm{AIBBr}$ ) in acetonitrile at room temperature for 1 hour (Scheme 3.3, step a).


Scheme 3.3: Synthesis of 3 'dA. (a.) $\alpha$ - $\operatorname{AIBBr}$ (4 eq), $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$, rt, 1h; (b.) Amberlite (IRN78) (8 $\mathrm{mL} / \mathrm{mmol}$ ), $\mathrm{CH}_{3} \mathrm{OH}, \mathrm{rt}, 16 \mathrm{~h}$; (c.) conditions reported in Table 3.3.

The reaction described was first used by Moffatt and coworkers ${ }^{215}$ who applied the Mattocks reaction of diols with $\alpha$-AIB halides to nucleosides. ${ }^{216}$ The products of this reaction were reported to be a mixture of more liphophilic intermediates as shown in Figure $3.4 ;{ }^{212}$ these intermediates were not isolated in this work, but the following step was performed on the crude after neutralisation and extraction with ethyl acetate.


Figure 3.4: Liphophilic intermediates of reaction of Ado with $\alpha-\mathrm{AIBBr}$ in $\mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O} .^{212}$

The crude mixture thus obtained was treated with Amberlite (IRN78) resin in methanol for 1 hour, yielding intermediate $\mathbf{2 6}$ in $40 \%$ yield (Table 3.2, entry 1). This outcome differed from the reported $93 \%$ yield; however, the reaction yield was improved to $98 \%$ by adding Amberlite (IRN78) in separate portions, and allowing the reaction to stir for 16 hours (Table 3.2, entry 2).

| Entry | Reagents and conditions | yield |
| :---: | :---: | :---: |
| $\mathbf{1}$ | Amberlite (IRN78) $(4 \mathrm{~mL} / \mathrm{mmol}), \mathrm{CH}_{3} \mathrm{OH}, 1 \mathrm{~h}, \mathrm{rt}$ | $40 \%$ |
| $\boldsymbol{2}$ | Amberlite (IRN78) $(8 \mathrm{~mL} / \mathrm{mmol}$ in two portions $), \mathrm{CH}_{3} \mathrm{OH}, 16 \mathrm{~h}, \mathrm{rt}$ | $98 \%$ |
| $\mathbf{3}$ | $\mathrm{NaOMe}(4 \mathrm{eq}), \mathrm{CH}_{3} \mathrm{OH}$, up to $72 \mathrm{~h}, \mathrm{rt}$ | - |
| $\mathbf{4}$ | $\mathrm{K}_{2} \mathrm{CO}_{3}(4 \mathrm{eq}), \mathrm{CH}_{3} \mathrm{OH}, 48 \mathrm{~h}, \mathrm{rt}$ | $67 \%$ |

Table 3.2: Reagents and conditions for the synthesis of intermediate 26.

Very careful washing of the resin with methanol ensured the product was completely extracted. Since the use of basic Amberlite is required to cleave the acetyl group in position $2^{\prime}$ or $3^{\prime}$ of the starting material and thereafter trigger the intramolecular cyclisation with release of $\mathrm{Br}^{-}$, the use of alternative basic conditions was attempted, with the aim of reducing the reaction time and avoiding the tedious washing of the resin. When sodium methoxide ( NaOMe ) was used in $\mathrm{CH}_{3} \mathrm{OH}$, only traces of the desired product could be formed (Table 3.2, entry 3). As an alternative, $\mathrm{K}_{2} \mathrm{CO}_{3}$ in $\mathrm{CH}_{3} \mathrm{OH}$ was used and, along with the long reaction times required, filtration through celite pad yielded only $68 \%$ of the product 26, due to some degradation already noticed on TLC (Table 3.2, entry 4).

Therefore the method involving the use of basic resin was considered the most successful. The identity of $2^{\prime}, 3^{\prime}$-anhydro intermediate $\mathbf{2 6}$ was confirmed by NMR, and Figure 3.5 shows the ${ }^{1} \mathrm{H}$-NMR of this product, performed in DMSO- $d 6$, characterised by the presence of the 5 '-OH proton but lacking any other hydroxyl group. Moreover, from the distinctive coupling pattern, already described in the literature, ${ }^{212,217}$ it is evident that the H 1 ' is not coupled with the H 2 ', and the two signals appear respectively as a singlet and as a doublet, the latter due to the coupling with the H3'.


Figure 3.5: ${ }^{1} \mathrm{H}$ NMR (DMSO- $d 6,500 \mathrm{MHz}$ ) spectrum of $2^{\prime}, 3^{\prime}$-anhydroadenosine 26.

For the last step of 3'dA synthesis, Robins et al. reacted $\mathbf{2 6}$ with 12.5 equivalents of lithium triethylboron hydride (superhydride, $\mathrm{LiEt}_{3} \mathrm{BH}$ ) in anhydrous DMSO. In our hands, this procedure was initially attempted on a small scale ( 0.50 g of starting material), yielding 3 ' dA (21) in $23 \%$ yield due to incomplete conversion of the starting material (Table 3.3. entry 1), in contrast with the reported $98 \%$ yield.

| Entry |  | Reagents and conditions |  | Yield |
| :---: | :---: | :---: | :---: | :---: |
|  | LiEt $_{3} \mathbf{B H}^{*}$ | Solvent | Temperature. |  |
| $\mathbf{1}$ | 12.5 eq (one portion) | DMSO $(15 \mathrm{~mL} / \mathrm{mmoL})$ | $0{ }^{\circ} \mathrm{C}$ to rt. 16 h | $23 \%$ |
| $\mathbf{2}$ | 4.5 eq (two portions) | DMSO $(4 \mathrm{~mL} / \mathrm{mmoL}) . \mathrm{THF}(10 \mathrm{~mL} / \mathrm{mmoL})$ | $0{ }^{\circ} \mathrm{C}$ to rt. 16 h | $95 \%$ |
| $\mathbf{3}$ | 4.5 eq (two portions) | $\mathrm{THF}(15 \mathrm{~mL} / \mathrm{mmoL})$ | $0{ }^{\circ} \mathrm{C}$ to rt. 16 h | - |

Table 3.3: Reagents, conditions and yields of the different attempts to synthesise 3 ' $\mathrm{dA}(\mathbf{2 1}) .{ }^{*} \mathrm{LiEt}_{3} \mathrm{BH}$ was used as a 1 M solution in THF.

Following this procedure, no $2^{\prime}$-deoxyadenosine ( $2^{\prime} \mathrm{dA}, 27$ ) was synthesised, and the identity of the product was confirmed by both monodimensional and bidimensional NMR experiments, and by comparison with NMR characterisation reported in the literature (Figure 3.6, Figure 3.7). The ${ }^{1} \mathrm{H}$ NMR spectra of 3 ' dA yielding from this reaction method can be compared to commercially available $2^{\prime} \mathrm{dA}$, and the most evident difference lies in the chemical shifts of H2' and H3' groups of signals, which are the most upfield signals respectively in $2^{\prime} \mathrm{dA}$ and $3^{\prime} \mathrm{dA}$ (Figure 3.6).


Figure 3.6: ${ }^{1} \mathrm{H}$ NMR (DMSO-d6, 500 MHz ) spectra of overlayed 3'dA (21) and 2 'dA (27).

Moreover, analysis of the COSY 2D NMR experiment shows characteristic coupling patterns relative to 3 ' dA (Figure 3.7).


Figure 3.7: COSY NMR (DMSO-d6) spectrum of 3'dA (21).

When the reaction was performed on a bigger scale ( 1.5 g ), with the same reaction conditions, the yield was similar ( $28 \%$ ) and the removal of the large volume of DMSO proved tedious. In order to improve both the reaction yield and purification step, when the reaction was carried out on 9 g of intermediate 26, a reduced amount of anhydrous DMSO ( $4 \mathrm{~mL} / \mathrm{mmol}$ ) was used and anhydrous THF was added ( $10 \mathrm{~mL} / \mathrm{mmol}$ ). Furthermore, a reduced amount of $\mathrm{LiEt}_{3} \mathrm{BH}$ (3 eq.) was added, followed by an additional portion ( 1.5 eq .) after 16 hours.

This resulted in $95 \%$ final yield of 3 'dA (Table 3.3, entry 2 ). The mixture of a small amount of DMSO in THF does not allow the starting material to dissolve completely, but addition of the 1 M solution of $\mathrm{LiEt}_{3} \mathrm{BH}$ in THF allows the solid to dissolve and the solution to turn from clear to a dark orange colour. The attempt to completely avoid the use of DMSO as a solvent was unsuccessful: the reaction mixture of $\mathbf{2 6}$ in THF turned into a clear dark orange solution after addition of $\mathrm{LiEt}_{3} \mathrm{BH}$ but even after 16 hours no conversion to 3 'dA was noticed on TLC (Table 3.3, entry 3). A small amount of DMSO was added along with other 3 eq. of $\mathrm{LiEt}_{3} \mathrm{BH}$ and the conversion happened within 3 hours. This proves that although not crucial as a solvent, DMSO may participate in the reaction mechanism.

### 3.1.3 Prodrug synthesis

The phosphoroamidate (ProTide) and phosphorodiamidate phosphate prodrug technologies were applied to cordycepin, generating a family of nucleotide analogues for in vitro testing, primarily as anticancer agents. The following paragraphs will report synthetic methodologies and biological evaluation of 3'dA derivatives.

### 3.1.3.1 Synthesis of 3'dA ProTides

The first synthetic strategy for the synthesis of 5'-phosphoroamidates implied the treatment of 3 ' dA with 1.1 equivalents of $t \mathrm{BuMgCl}$ and 3 equivalents of the appropriate phosphorochloridate under argon atmosphere overnight (Scheme 3.4). ${ }^{154}$ The initially promising conversion of $3^{\prime} \mathrm{dA}$ into two other main species, as detected by TLC, resulted in the undesired synthesis of $2^{\prime}$-phosphoroamidate ( $\mathbf{2 8 a}, \mathbf{b}$ ) and $2^{\prime}, 5^{\prime}$-bis-phosphoroamidate (29a,b) products (Table 3.4).


Scheme 3.4: First attempt of the synthesis of 3 'dA ProTides using Grignard reagent. Reagents and conditions: a. $t \mathrm{BuMgCl}$ ( 1.1 eq ). appropriate phosphorochloridate ( 3 eq ). anhydrous THF, $\mathrm{rt}, 16 \mathrm{~h}$.

| Compound | Ar | $\mathbf{R}^{1}, \mathbf{R}^{\mathbf{2}}$ | Amino acid | Position | R | Yield |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 28a | Ph | $\mathrm{CH}_{3}, \mathrm{H}$ | $\mathrm{L}-\mathrm{Ala}$ | $2^{\prime}$ | Bn | $5 \%$ |
| 28b | Nap | $\mathrm{CH}_{3}, \mathrm{H}$ | $\mathrm{L}-\mathrm{Ala}$ | $2^{\prime}$ | Bn | $13 \%$ |
| 29a | Ph | $\mathrm{CH}_{3}, \mathrm{H}$ | $\mathrm{L}-\mathrm{Ala}$ | $2^{\prime}, 5^{\prime}$ | Bn | $35 \%$ |
| 29b | Nap | $\mathrm{CH}_{3}, \mathrm{H}$ | L-Ala | $2^{\prime}, 5^{\prime}$ | Bn | $11 \%$ |

Table 3.4: First family of cordycepin phosphoramidates synthesised through $t \mathrm{BuMgCl}-$ mediated coupling reaction.

Figure 3.8 and Figure 3.9 report the ${ }^{31} \mathrm{P}$ NMR spectra of the products of the reaction of 3'dA with 3 equivalents of L-alanine benzyloxy naphthyl phosphorochloridate in the presence of $t \mathrm{BuMgCl}$, and it is clear that this method yields only 2 '-phosphoramidate (28b) and $2^{\prime}, 5^{\prime}$-phosphoramidate (29b) products. In fact, the ${ }^{31} \mathrm{P}$ NMR spectrum of 29b
shows the presence of 8 peaks in a range between 4.5 and 2.5 ppm , corresponding to the 4 different isomers synthesised, each one containing two phosphoramidate moieties.


Figure 3.8: ${ }^{31} \mathrm{P}$ NMR ( $202 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) spectrum of 29b.

The ${ }^{31} \mathrm{P}$ NMR spectrum of the 2 '-phosphoramidate $\mathbf{2 8 b}$, on the other hand, is not confirmative of the presence of the phosphoramidate moiety in position $2^{\prime}$ or $5^{\prime}$, with two peaks corresponding to the two diasteroisomers with chemical shifts in the region where 5'-phosphoroamidates should also be expected.


Figure 3.9: ${ }^{31} \mathrm{P}$ NMR ( $202 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) spectrum of $\mathbf{2 8 b}$.

However, from the analysis of the ${ }^{13} \mathrm{C}$ NMR spectrum (Figure 3.10) it becomes evident that the structure of $\mathbf{2 8 b}$ corresponds to a 2'-phosphoramidate, because of the splitting of the two diastereomeric peaks of the C 1 ' into 4 peaks, due to a three bond coupling constant with phosphorus $\left({ }^{3} J_{C C O P}=7.5 \mathrm{~Hz}\right)$.


Figure 3.10: ${ }^{13} \mathrm{C}$ NMR spectrum of $\mathbf{2 8 b}$.

Another characteristic of the presence of a 2 '-phosphoroamidate moiety consists of the splitting of the carbon signals of the C2' due to the two bond coupling constant with phosphorus ( ${ }^{2} J_{C O P}=5.5 \mathrm{~Hz}$ ). On the other hand, the C 4 ' and C 5 ' signals are not split (Figure 3.10).

When the coupling reaction was performed with NMI and the appropriate phosphorochloridrate in THF, the 5 '-phosphoroamidates of 3'dA were synthesised as major products in moderate yields, although the $2^{\prime}, 5^{\prime}$-bis-phosphoroamidate side products were nevertheless formed (Figure 3.5, Table 3.5).


Scheme 3.5: Synthesis of cordycepin ProTides (30a-f) via the NMI method. Reagents and conditions: NMI ( 5 eq ). appropriate phosphorochloridate ( 3 eq ), anhydrous THF, rt, 16 h .

| Cpnd | Ar | $\mathbf{R}^{1} / \mathbf{R}^{2}$ | AA | R | Yield |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 30a | Ph | $\mathrm{Me} / \mathrm{H}$ | $\mathrm{L}-\mathrm{Ala}$ | Bn | $28 \%$ |
| 30b | 1-Naph | $\mathrm{Me} / \mathrm{H}$ | $\mathrm{L}-\mathrm{Ala}$ | Bn | $12 \%$ |
| 30c | Ph | $\mathrm{H} / \mathrm{H}$ | Gly | Bn | $19 \%$ |
| 30d | 1-Naph | $i \mathrm{Bu} / \mathrm{H}$ | $\mathrm{L}-\mathrm{Leu}$ | $n$-Pen | $25 \%$ |
| 30e | $\mathrm{O}_{\text {oet }}$ | $\mathrm{Me} / \mathrm{H}$ | $\mathrm{L}-\mathrm{Ala}$ | Bn | $22 \%$ |
| 30f | 1-Naph | $\mathrm{Me} / \mathrm{Me}$ | DMG | Me | $18 \%$ |

Table 3.5: 3'dA ProTides (30a-f) structures and yields.

Similar to the case of the $2^{\prime}$-phosphoroamidates, confirmation of the presence of the $5^{\prime}$ phosphoroamidate moiety is given primarily by the analysis of the signal splitting pattern of the ${ }^{13} \mathrm{C}$ NMR spectrum. In fact, the ${ }^{31} \mathrm{P}$ NMR spectrum of compound $\mathbf{3 0 b}$ presents two peaks corresponding to the two diasteroisomers in the same region as for the 2 'phosphoroamidate regioisomer 28b (Figure 3.11).


Figure 3.11: ${ }^{31} \mathrm{P}$ NMR ( $202 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) spectrum of $\mathbf{3 0 b}$.

On the other hand, the ${ }^{13} \mathrm{C}$ NMR spectrum presents crucial differences because of the splitting of the C 4 ' and $\mathrm{C} 5^{\prime}$ signals respectively due to three and two bond coupling constant ( ${ }^{3} \mathrm{~J}_{\mathrm{CCOP}}=2.7 \mathrm{~Hz}$ and ${ }^{2} \mathrm{~J}_{\mathrm{COP}}=5.2 \mathrm{~Hz}$ ), while in this case the C 1 ' and C 2 ' signals appear as two peaks due to the presence of two diasteroisomers (Figure 3.12).


Figure 3.12: ${ }^{13} \mathrm{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) spectrum of $\mathbf{3 0 b}$.

### 3.1.3.2 Synthesis of 3'dA phosphorodiamidates

The phosphorodiamidate approach was the second nucleotide prodrug system applied to $3^{\prime} \mathrm{dA} . .^{67,156,218}$ In the present work, the Yoshikawa ${ }^{156}$ procedure was adopted to obtain 5'phosphorylated nucleoside analogues using unprotected 3'dA. According to the cited study, in a one pot reaction the unprotected nucleoside is suspended in trimethyl phosphate (TMP), and phosphorus oxychloride is added in order to obtain the 5'phosphorodichloridate (31, Scheme 3.6) in situ. ${ }^{155}$ The formation of this intermediate was monitored by ${ }^{31} \mathrm{P}-\mathrm{NMR}$ (appearance of a peak $\sim 7 \mathrm{ppm}$ corresponding to the $5^{\prime}$ dichlorophosphate). Intermediate $\mathbf{3 1}$ was not isolated but immediately coupled with the appropriate amino acid ester salt in the presence of anhydrous diisopropyl ethyl amine (DIPEA). The progression of the reaction was monitored again by ${ }^{31} \mathrm{P}-\mathrm{NMR}$ : the appearance of a single peak in the range of 13-17 ppm, depending on the amino acid used for the synthesis, indicated the formation of the product. The synthesis of the benzyloxy-Lalaninyl phosphorodiamidate $\mathbf{3 2}$ was accomplished in $49 \%$ (Scheme 3.6).


Scheme 3.6: Synthesis of 3'dA phosphorodiamidate 32. Reagents and conditions: (a.) $\mathrm{POCl}_{3}$ (1 eq), TMP $0^{\circ} \mathrm{C}-\mathrm{rt}, 4 \mathrm{~h}$. (b.) L-AlaOBn $\cdot \mathrm{HCl}$ ( 5 eq ), DIPEA ( 10 eq ), $\mathrm{CH}_{2} \mathrm{Cl}_{2}, 0^{\circ} \mathrm{C}-\mathrm{rt}$, 16 h .

The product identity was confirmed by NMR characterisation and Figure 3.13 shows the characteristic ${ }^{31} \mathrm{P}$ NMR spectrum of compound $\mathbf{3 2}$ with only one peak due to the lack of chirality at the P centre.


Figure 3.13: ${ }^{31} \mathrm{P}$ NMR (202 MHz. $\mathrm{CD}_{3} \mathrm{OD}$ ) spectrum of $\mathbf{3 2}$.

### 3.1.4 Biological evaluation of $\mathbf{3}^{\prime}$ dA prodrugs

### 3.1.4.1 In vitro cytotoxic screening of 3'dA prodrugs

An initial cytotoxicity screening on a selection of cell lines was performed by WuXi AppTech on the synthesised ProTides (30a-f) and the $5^{\prime}$ 'phosphorodiamidate (32) prodrug of 3 'dA. Moreover, the $2^{\prime}$ '-phosphoroamidates ( $\mathbf{2 8 a , b}$ ), and one example of 2',5'bisphosphoroamidate ( $\mathbf{2 9 b}$ ) were included in this preliminary evaluation (Table 3.6).

| Comp. | Position | Ar | $\mathbf{R}^{1}, \mathbf{R}^{2}$ | AA | R | CLogP |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 3'-dA | - | - | - | - | - | -2.36 |
| 28a | 2 ' | Ph | $\mathrm{CH}_{3}, \mathrm{H}$ | L-Ala | Bn | 1.12 |
| 28b | 2 ' | Nap | $\mathrm{CH}_{3}$, H | L-Ala | Bn | 2.30 |
| 29a | 2',5' | Ph | $\mathrm{CH}_{3}, \mathrm{H}$ | L-Ala | Bn | 4.42 |
| 30a | 5 | Ph | $\mathrm{CH}_{3}$, H | L-Ala | Bn | 0.93 |
| 30b | 5 , | Nap | $\mathrm{CH}_{3}$, H | L-Ala | Bn | 2.10 |
| 30c | 5 | Ph | H, H | Gly | Bn | 0.81 |
| 30d | 5 , | Nap | $i \mathrm{Bu}, \mathrm{H}$ | L-Leu | $n$-Pent | 3.97 |
| 30e | $5 '$ |  | $\mathrm{CH}_{3}, \mathrm{H}$ | L-Ala | Bn | 1.60 |
| 30 f | 5 , | Nap | $\mathrm{CH}_{3}, \mathrm{CH}_{3}$ | DMG | $\mathrm{CH}_{3}$ | 0.70 |
| 32 | 5, | - (phosphorodiamidate) | $\mathrm{CH}_{3}$, H | L-Ala | Bn | 2.91 |

Table 3.6: Compounds tested on cancer cell lines by WuXi AppTech. CLogP calculated via ChemDraw program.

The selected cell lines derive mainly from haematologic malignancies (MOLT-4, KG-1, HL-60, CCRF-CEM, K562), since 3'dA has been involved in clinical trials against acute myeloid leukaemia (AML). Moreover, two solid tumour-derived cell lines (MCF-7, HepG2) were considered in light of the activity of 3 ' dA reported on breast cancer cell lines ${ }^{219,220}$ and hepatocellular carcinoma cell lines (Table 3.7). ${ }^{220}$

| Cell lines | Malignancy |
| :---: | :---: |
| MOLT-4 | Acute T lymphoblastic leukaemia |
| KG-1 | Acute myelogenous leukaemia |
| HL-60 | Acute promyelocytic leukaemia |
| CCRF-CEM | Acute lymphoblastic leukaemia |
| K562 | Chronic myelogenous leukaemia |
| MCF-7 | Breast adenocarcinoma |
| HepG2 | Hepatocellular carcinoma |

Table 3.7: Malignant cell lines selected for the cytotoxic screening of 3'dA prodrugs by WuXi AppTech.

This screening aimed to provide an initial understanding of whether 3' dA prodrugs would bring advantage over the parent nucleoside in terms of cytotoxic activity on the considered cell lines.

Table 3.8 reports the cytotoxic activities of 3 'dA prodrugs as $\mu \mathrm{M} \mathrm{IC}_{50}$ values, the concentration of compound leading to a $50 \%$ reduction in cell viability and percentage of maximum inhibition ( $\mathrm{MI}_{\%}$ ).

Across all range of cell line considered, $3^{\prime} \mathrm{dA}$ does not represent a potent anticancer drug, due both to the high $\mathrm{IC}_{50}$ values and the inability to reach $100 \%$ inhibition of viability. Particularly striking is the effect of cordycepin on the CCRF-CEM cell line, in which the maximum inhibition of viability reaches only $12 \%$ (Figure 3.14).


Figure 3.14: Cordycepin activity in CCRF-CEM cell line, reported as a set of data points corresponding to effect on cell viability of the different concentrations of drug. Screening performed by WuXi AppTech.

3'dA does not lead to a measurable $50 \%$ inhibition of cell growth in most of the cell lines, in fact the $\mathrm{IC}_{50}$ is reported as a predicted value ( $>100 \mu \mathrm{M}$ ) on most cell lines. On the cells
in which 3 'dA exerts inhibition of cell viability higher than the $50 \%$, the $\mathrm{IC}_{50}$ concentrations range from $34.23 \mu \mathrm{M}$ on the MCF-7 cell line and $151.92 \mu \mathrm{M}$ on the MOLT4 cell line.

3'dA derivatives, on the other hand, bring an advantage over the parent nucleoside in terms of cytotoxicity in almost all cell lines (Table 3.8).

| Cpnd | CCRF-CEM |  | HL-60 |  | KG-1 |  | MOLT-4 |  | K562 |  | HepG2 |  | MCF-7 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | IC ${ }_{50}$ | MI\% |
| 3'-dA | >198 | 12 | 76.84 | 88 | 70.82 | 78 | 151.92 | 52 | 59.31 | 88 | 142.47 | 66 | 34.23 | 78 |
| 28a | 64.83 | 65 | 143.47 | 57 | 111.95 | 63 | 62.86 | 75 | 26.3 | 78 | 85.5 | 667 | 10.35 | 97 |
| 28b | 27.14 | 99 | 78.59 | 99 | 66.28 | 93 | 14.78 | 100 | 25.35 | 87 | 30.16 | 90 | 5.05 | 99 |
| 29a | 17.63 | 99 | 31.09 | 80 | 79.44 | 69 | 18.62 | 100 | 26.87 | 89 | 83.06 | 55 | 5.06 | 94 |
| 30a | 2.36 | 100 | 17.78 | 97 | 17.36 | 92 | 0.51 | 98 | 6.1 | 92 | 18.73 | 76 | 3.07 | 94 |
| 30b | 1.17 | 100 | 6.37 | 100 | 11.03 | 100 | 0.32 | 100 | 4.93 | 97 | 11.53 | 95 | 1.39 | 99 |
| 30c | 14.93 | 94 | 45.23 | 82 | 136.56 | 65 | 3.55 | 93 | 15.42 | 92 | 142.35 | 59 | 15.46 | 87 |
| 30d | 4.72 | 100 | 8.9 | 100 | 12.02 | 99 | 1.46 | 100 | 11.7 | 99 | 9.53 | 99 | 2.48 | 100 |
| 30e | 40.09 | 90 | 78.51 | 74 | 102.14 | 69 | 12.62 | 97 | 174.71 | 61 | 96.11 | 59 | 15.78 | 78 |
| 30 f | 4.92 | 99 | 24.08 | 98 | >198 | 29 | 1.4 | 100 | 10.69 | 91 | 70.22 | 84 | 9.10 | 99 |
| 32 | 7.03 | 100 | 16.53 | 100 | 34.13 | 98 | 3.52 | 100 | 24.34 | 97 | 156.87 | 55 | 12.97 | 97 |
| PTX | 0.003 | 95 | 0.004 | 96 | 0.07 | 89 | 0.002 | 97 | 0.008 | 94 | \#Intersect | 54 | 0.003 | 79 |

Table 3.8: In vitro cell viability evaluations after treatment with 3'dA and 3'dA prodrugs. Cytotoxicity data reported as $\mu \mathrm{M} \mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ values (maximum inhibitory effect of the drug at the range of concentration considered). PTX: paclitaxel (control). Screening performed by WuXi AppTech.

The derivatives emerging as the most active on the majority of cell lines tested were $5^{\prime}$ 'phosphoramidates 30a (L-alanine- $O$-benzyl phenyl derivative), 30b (L-alanine- $O$-benzyl naphthyl derivative) and 30d (L-leucine-O-pentyl naphthyl derivative) (Table 3.8, Figure 3.15).


Figure 3.15: Cytotoxic activity of 3'dA and 5'-phosphoramidates 30a-f. * $\mathrm{IC}_{50}>198 \mu \mathrm{M}$.

However, replacing the amino acid portion with glycine (30c) or dimethylglycine (30f) caused a drop in anticancer activity, especially in the case of compound 30f. bearing a methyl ester instead of a benzyl ester. The introduction of a new moiety in place of the most frequently used phenyl or naphthyl group, represented by the ethyl 3-phenyl propanoate group in compound 30e, generated a compound with lower activity on KG-1 and HepG2 cell lines, compared to the naphthyloxy and phenyloxy-bearing compounds 30a, 30b and 30d. However, 30e had similar $\mathrm{IC}_{50}$ values to 30a and 30b when tested on the other cell lines.

The 2 '-phosphoroamidates 28a and 28b and the bis-phosphoroamidate 29b usually represent side products of the coupling reaction to synthesise the desired 5 'phosphoroamidate prodrugs.


Figure 3.16: Cytotoxic activity of 3'dA and 2'-phosphoroamidates 28a and b, and 2',5'-bisphosphoramidate 29a. ${ }^{*} \mathrm{IC}_{50}>198 \mu \mathrm{M}$.

However, some results generated in the McGuigan group ${ }^{221}$ reported that some examples of 3'-phosphoroamidate prodrugs of cladribine were more cytotoxic than the 5 'regioisomers, and this bizarre result stimulated the investigation on the cytotoxic potential of alternative phosphoroamidate derivatives (such as $2^{\prime}$ '-phosphoramidates 28a and 28b and $2^{\prime}, 5^{\prime}$-bis phosphoramidates 29a), especially in situations where as was the case of 3'dA ProTides synthesis, these side products could be generated in relatively large amounts. Although not as potent as the most active ProTides 30a, 30b and 30d, the naphthyl derivatives 28b and 29a were several-fold more potent than 3'dA, while 28a performed less efficiently, despite retaining higher potency than 3 ' dA on most cell lines (Table 3.8, Figure 3.16).

The $5^{\prime}$ '-phosphorodiamidate 32 was more active than $3^{\prime} \mathrm{dA}$ and the ProTides 30c, 30e and 30f but less active than the lead ProTides 30a, 30b and 30d (Figure 3.17 shows the comparison between 32 and $\mathbf{3 0 a}$, along with the parent nucleoside).


Figure 3.17: Cytotoxic activity of 3 ' dA and phosphorodiamidate 32, compared to the 5 '-phosphoramidate 30a. ${ }^{*} \mathrm{IC}_{50}>198 \mu \mathrm{M}$.

In conclusion, this initial cytotoxicity screening was aimed at understanding whether the ProTide and phosphorodiamidate prodrug approaches could generate potential anticancer agents by enhancing the potency of 3 ' dA . All the derivatives tested showed improved cytotoxicity on all cell lines considered, and some ProTides, such as the L-alanine derivatives 30a and 30b, reaching 2.6 log lower $\mathrm{IC}_{50}$ values (on MOLT-4 cell line), were considered as lead candidates for further preclinical development.

### 3.1.4.2 Antiviral screening on 3'dA derivatives

$3^{\prime} \mathrm{dA}$, the intermediate in 3 ' dA synthesis 9 -(2,3-anhydro- $\beta$-D-ribofuranosyl)adenine $\mathbf{2 7}$, the 2'-phosphoramidate derivatives 28a and 28b, the 2 ',5'-bis- $O$-phosphoramidate 29a and ProTides 30a and 30d were tested on yellow fever virus (YFV) infected cells by Dr. J. Bugert (School of Medicine, Cardiff University). The rationale behind this screening was that in the early literature reports, $3^{\prime} \mathrm{dA}$ was found to have some activity on both RNA and DNA viruses. ${ }^{182,183}$

| Compound | TOX $+/-$ | CPE $+/-$ |
| :---: | :---: | :---: |
| 3'dA | - | + |
| 27 | - | + |
| $28 a$ | - | + |
| 28b | - | + |
| 29b | - | + |
| 30a | - | + |
| 30d | - | + |

Table 3.9: Anti-YFV activity and toxicity profile of prodrugs 28a, 28b, 29b, 30a and 30d plus the parent nucleoside 3 '-dA and 27. Antiviral activity and cytotoxicity were determined at $10 \mu \mathrm{M}$ concentration. Assays performed by Dr. Joachim Bugert (School of Medicine, Cardiff University).

In an attempt to investigate the possible antiviral activity of 3'dA derivatives, they were sent for testing against YFV, a RNA virus of the Flaviviridae family that can cause devastating epidemics of potentially fatal, hemorragic disease. ${ }^{222}$ Although YFV vaccination exists, there is no approved treatment for infected patients. ${ }^{223}$
The compounds were first screened at a fixed concentration of $10 \mu \mathrm{M}$ to detect activity on YFV infected Vero and BSC-1 cells (both kidney epithelial cells extracted from an African green monkey) and the toxicity on these cells.
The antiviral activity was measured by analysis of the cytopathic effect (CPE) on the infected cells: this effect refers to the damage to the host cells during virus invasion that can be studied in tissue culture. A positive CPE value means the cells are infected and the tested compounds are not showing any effects, whereas a negative CPE value means the compounds are effective at the given concentration. ${ }^{224}$
As evident from Table 3.9, none of the compounds showed any interesting activity on yellow fever viruses infected cells; on the other hand the compounds are not toxic at 10 $\mu \mathrm{M}$ concentration on non malignant Vero and BSC-1 cell lines, and this could be positive data in terms of selective toxicity of compounds $\mathbf{3 0 a}$ and $\mathbf{3 0 d}$ ( 3 'dA ProTides) on cancer cells at that same concentration (compounds 30a and 30d showed activity at lower concentrations on almost all cancer cell lines shown in Table 3.8).

### 3.1.4.3 Additional cytotoxic screening of 3'dA prodrugs

The parent nucleoside 3 ' dA , ProTide 30b and the intermediate 27 were tested for their activity on L1210 (murine lymphocytic leukaemia), CEM (acute lymphoblastic leukaemia) and HeLa (cervical adenocarcinoma) cell lines by Prof. Balzarini group at the Rega Institute (Leuven, Belgium). The results are shown in Table 3.10.

| Compound | L1210 | CEM | HeLa |
| :---: | :---: | :---: | :---: |
| 3'dA | $39 \pm 9$ | $50 \pm 14$ | $85 \pm 15$ |
| 30b | $5.2 \pm 1.6$ | $1.7 \pm 0.6$ | $7.4 \pm 4.4$ |
| $\mathbf{2 7}$ | $>250$ | $>250$ | $>250$ |

Table 3.10: Compounds tested on cancer cell lines by Prof. Balzarini group at the Rega Institute (Leuven, Belgium), ( $\mathrm{IC}_{50} \mu \mathrm{M}$ activity).

The 5 '-phosphoramidate $\mathbf{3 0 b}$ retains better activity than 3 ' dA on these cell lines, confirming the data previously generated (Table 3.8). Compound 27 was previously tested as an anti-trypanosome agent, ${ }^{225}$ and anti-HIV agent, ${ }^{226}$ but was inactive. However, tested
as the triphosphate on different viral DNA-polymerase enzymes, it proved an efficient inhibitor. ${ }^{227}$ Compound 27 was never reported as potentially active as anticancer agent, and from this cytotoxic screening the compound was inactive.

### 3.1.4.4 In vitro screening of 3'dA ProTides on additional cell lines

The favourable activity profile of ProTides 30a, 30b, 30d, 30e prompted further testing of these compounds on additional cell lines (Table 3.11) by WuXi AppTech, in order to confirm the potency of the compounds and select a candidate for the preclinical studies. The parent nucleoside 3'dA was tested in parallel, along with ProTide 30f, selected as a negative control, due to the lower cytotoxic potential demonstrated in the previous assays.

|  | Cell line | Malignancy |
| :---: | :---: | :---: |
|  | RL | non-Hodgkin's lymphoma |
|  | HS445 | Hodgkin lymphoma |
| Haematologic | RPMI-8226 | human multiple myeloma |
| malignancies | Z-138 | acute monocytic leukaemia |
|  | MV4-11 | mantle cell lymphoma |
|  | NCI-H929 | biphenotypic B myelomonocytic leukaemia |
|  | HEL92.1.7 | plasmacytoma |
|  | Jurkat | erythroleukaemia |
|  | BxPC-3 | acute T cell leukaemia |
| Solid | HT29 | Pancreas carcinoma |
| tumours | MIA PaCa-2 | Colon adenocarcinoma |
|  | SW620 | Pancreas adenocarcinoma |
|  | Colon adenocarcinoma |  |

Table 3.11: Cell lines selection for the second anticancer evaluation of 3'dA ProTides by WuXi AppTech.

As shown in Table 3.12, the lack of potency of 3 'dA was confirmed in this second screening on different cell lines by the inability to reach $100 \%$ inhibition of cell viability up to the highest concentration considered $(198 \mu \mathrm{M})$.

| Cpnd | MV4-11 |  | THP-1 |  | RL |  | HS445 |  | HEL92.1.7 |  | NCI-H929 |  | RPMI-8226 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | IC 50 | MI\% | $\mathrm{IC}_{50}$ | $\mathbf{M I}_{\%}$ | IC 50 | MI\% | IC 50 | MI\% | IC50 | MI\% | IC 50 | MI\% | IC 50 | MI\% |
| 3'dA | >198 | 1 | $>198$ | -3 | >198 | 17 | >198 | 2 | 68.9 | 88 | $>198$ | 24 | >198 | 1 |
| 30a | 2.1 | 99 | 65.45 | 74 | 3 | 93 | 30.53 | 98 | 8.07 | 100 | 6.07 | 100 | 14.72 | 96 |
| 30b | 1.48 | 99 | 46.47 | 99 | 1.68 | 96 | 10.18 | 96 | 2.94 | 98 | 3.38 | 99 | 9.48 | 102 |
| 30d | 8.78 | 106 | 68.91 | 99 | 11.61 | 100 | 39.53 | 102 | 4.23 | 99 | 7.21 | 104 | 24.06 | 103 |
| 30 e | 3.27 | 100 | 36.66 | 100 | 7.26 | 90 | 42.73 | 92 | 14.49 | 99 | 7.57 | 100 | 31.05 | 106 |
| 30 f | 8.71 | 101 | $>198$ | 43 | 13.54 | 88 | 54.98 | 85 | 15.42 | 101 | 16.11 | 98 | 46.7 | 89 |
| PTX | 0.01 | 99 | 0.03 | 72 | 0.003 | 83 | 0.01 | 74 | 0.02 | 83 | 0.003 | 82 | 0.003 | 91 |

Table 3.12: In vitro cell viability evaluation after treatment with 3'dA and selected ProTides. Cytotoxicity data reported as $\mu \mathrm{M} \mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentration considered). PTX: paclitaxel (control). Screening performed by WuXi AppTech.

3'dA exerted measurable cell viability inhibition $\left(\mathrm{MI}_{\%}=88 \%\right)$ at high concentration $\left(\mathrm{IC}_{50}\right.$ $=68.9 \mu \mathrm{M})$ only on the erythroleukaemia cell line HEL92.1.7, and performed distinctively well on the mantle cell lymphoma cell line Z 138 , ${ }^{228}$ a unique example where $3^{\prime}$ 'dA could reach $95 \%$ of cell viability inhibition with an $\mathrm{IC}_{50}$ value of $12.15 \mu \mathrm{M}$. The cell line Z138 also represented an exception because ProTides 30b and 30d, usually many times more active than the parent nucleoside, were in this case only two-fold more active than $3^{\prime} \mathrm{dA}$ and more strangely $\mathbf{3 0 a}$ was two-fold less active than $3^{\prime} \mathrm{dA}$ (Table 3.13).

| Cpnd | Jurkat |  | Z138 |  | BxPC-3 |  | HT29 |  | MIA PaCa-2 |  | SW620 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% |
| 3'dA | >198 | 20 | 12.15 | 95 | >198 | 22 | >198 | 44 | >198 | 35 | >198 | 10 |
| 30a | 1.44 | 100 | 26.97 | 95 | 23.59 | 81 | 13.42 | 93 | 6.85 | 96 | 24.94 | 85 |
| 30b | 0.9 | 100 | 6.32 | 100 | 13.36 | 90 | 7.52 | 98 | 3.73 | 98 | 11.41 | 93 |
| 30d | 7.15 | 100 | 5.65 | 100 | 65.46 | 99 | 16.06 | 99 | 14.69 | 106 | 35.48 | 100 |
| 30e | 4.75 | 100 | 68.18 | 76 | 60.92 | 78 | 25.24 | 89 | 12.09 | 101 | 43.4 | 90 |
| 30 f | 7.01 | 95 | 43.84 | 93 | 76.18 | 71 | 37.94 | 76 | 17.55 | 91 | 56.48 | 74 |
| PTX | 0.005 | 97 | 0.002 | 99 | >0.5 | 42 | 0.004 | 75 | 0.002 | 87 | 0.02 | 96 |

Table 3.13: In vitro cell viability evaluation after treatment with 3'dA and selected ProTides. Cytotoxicity data reported as $\mu \mathrm{M} \mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentration considered). PTX: paclitaxel (control). Screening performed by WuXi AppTech.

No explanation for this peculiar result could be found in the literature, therefore one could only speculate that this cell line may down-regulate ADA, therefore there may be a reduction in deamination of 3 ' dA that could prolong and enhance the activity of this nucleoside. Alternatively, enzymes responsible for the ProTide activation
(carboxypeptidase/phosphoramidase enzymes) could be down-regulated, leading to a reduced activity of these molecules. However, this was an isolated event, as ProTides 30a, 30b, 30d, 30e and 30f were consistently more active than $3^{\prime} \mathrm{dA}$ on all the other cell lines considered, the only exception being the negative control ProTide 30f that did not lead to $50 \%$ of inhibition of cell viability on THP-1 cell line.


Figure 3.18: Cytotoxic activity of 3 'dA and selected ProTides, on haematologic malignancies. ${ }^{*} \mathrm{IC}_{50}>198$ $\mu \mathrm{M}$.

ProTides 30a and 30b were the most active compounds in most of the considered cell lines, 30b being usually 1 or 2 fold more active than 30a. The best results were achieved on Jurkat cell line (acute T-cell leukaemia), where 30b reached $100 \%$ of cell viability inhibition at a concentration of $0.9 \mu \mathrm{M}$, and 30a performed similarly at a concentration of $1.44 \mu \mathrm{M}$ (Table 3.13, Figure 3.18).


Figure 3.19: Cytotoxic activity of 3 ' dA and selected ProTides, on solid tumours ${ }^{*} \mathrm{IC}_{50}>198 \mu \mathrm{M}$.

Compounds 30e and 30d had similar potencies in most cell lines, although with higher $\mathrm{IC}_{50}$ values than 30a and 30b. On solid tumours (Table 3.13. Figure 3.19) the compounds were generally less active than in other settings, however with activity still in the $\mu \mathrm{M}$ range.

In conclusion, this second set of cytotoxic data demonstrated again the superior activity of 3'dA ProTides compared to the parent nucleoside, and the L-alanine benzyloxy compounds $\mathbf{3 0 a}$ and 30b proved the most potent derivatives in the series.

### 3.1.4.5 Evaluation of intracellular 3'dA triphosphate levels after treatment with 3'dA ProTides

The anticancer activity of $3^{\prime} \mathrm{dA}$ is primarily generated by the intracellular triphosphate metabolite ( $3^{\prime}$-dATP). In order to investigate whether the superior cytotoxic activity of 3'dA ProTides could be a consequence of increased intracellular 3'dATP levels, a selection of cell lines were treated with 3 'dA and the ProTides which emerged as most active in the previous screenings performed by Barts Institute (London). ProTide $\mathbf{3 0 f}$ was selected as a poorly performing analogue in order to have a negative control (Table 3.14).

| General Structure | Cpnd | Ar | $\mathbf{R}$ | $\mathbf{R}_{\mathbf{1}} / \mathbf{R}_{\mathbf{2}}$ | AA |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |

Table 3.14: Compound selected for in vitro evaluation of 3'dATP intracellular levels on CEM, K562 and HL-60 cell lines. Screening performed by Barts Institute (London).

The cell line selection includes the acute lymphoblastic leukaemia cell line CEM, the chronic myelogenous leukaemia cell line K562 and the acute promyelocytic leukaemia cell line HL-60. Cells were treated with concentrations of the selected compounds ranging from 10 nM to $20 \mu \mathrm{M}$ and were incubated for 2 hours at $37{ }^{\circ} \mathrm{C}$. The results reported in Table 3.15 as $\mathrm{IC}_{50}$ values confirm the better activity profile of 3 ' dA ProTides compared to the parent nucleoside, with the exception of compound 30f, which was choosen as a negative control, and emerged again as a poorly active compound.

| Compound | CEM | K562 | HL-60 |
| :---: | :---: | :---: | :---: |
| 3'dA | 19.5 | 10.9 | 11.4 |
| 30a | 0.87 | 2.4 | 4.6 |
| 30b | 0.13 | 0.21 | 2.6 |
| 30d | 6.1 | 4.2 | 5.4 |
| 30e | 4.3 | 6.3 | 8.3 |
| 30f | 10 | 13.9 | 10.2 |

Table 3.15: Evaluation of concentration inhibiting $50 \%$ of cell viability ( $\mathrm{IC}_{50}, \mu \mathrm{M}$ ) on CEM, K562 and HL60 cell lines. Assays performed by Barts Institute (London).

With $\mathrm{IC}_{50}$ activity ranging from 4 - to 15 -fold lower than 3 'dA, naphthyl L-alaninylbenzyloxy ProTide 30b emerged as the most potent compound, followed by the phenylbearing counterpart, compound 30a.

The analysis of 3'dATP levels generated in the intracellular environment after treatment with the selected compounds at a fixed concentration was performed at Barts Institute (London) via LC-MS. Table 3.16 shows these results as mean concentrations of intracellular 3'dATP expressed in $\mu \mathrm{g} / \mathrm{mL}$.

| Compound | CEM | K562 | HL-60 |
| :---: | :---: | :---: | :---: |
| 3'dA | 0.2 | 1.7 | 1.7 |
| 30a | 3.7 | 2.6 | 3.2 |
| 30b | 11.5 | 6.2 | 5.1 |
| 30d | 2.9 | 0.9 | 0.7 |
| 30e | 1.1 | 1.3 | 1.2 |
| 30f | 0.2 | 0.2 | 0.2 |

Table 3.16: 3'dATP intracellular levels generated after treatment of malignant cell lines with selected 3'dA ProTides. Values expressed as $\mu \mathrm{g} / \mathrm{mL}$ concentrations. Assays performed by Barts Institute (London).

The low levels of intracellular 3'dATP generated after treatment with 3'dA, ranging from 0.2 to $1.7 \mu \mathrm{~g} / \mathrm{mL}$, suggest that the low cytotoxicity of this compound derives from a poor intracellular processing to the active metabolite, which could depend on either an inefficient phosphorylation by nucleoside kinase enzymes or the catabolism into inactive intracellular species such as 3 'dIno. The 3 '-dATP levels generated after treatment with the least active ProTide $\mathbf{3 0 f}$ was comparable, potentially because of an inefficient intracellular processing to 3'dAMP. On the other hand, the levels of 3'dATP produced by the intracellular processing of the most active compound 30b are significantly higher, and range from 5.1 to $11.5 \mu \mathrm{~g} / \mathrm{mL}$. High levels of intracellular active metabolite are similarly generated by treatment with compound 30a. Cell treatment with ProTides 30e and 30f, on the other hand, led to intracellular 3'dATP levels 5 to 15 -fold higher than the parent
nucleoside on CEM cell line, and lower levels of HL-60 and K562 cells, despite the similar $\mathrm{IC}_{50}$ values of these compounds on all cell lines.


Figure 3.20 Correlation between cytotoxic $\mathrm{IC}_{50}$ values ( $\mu \mathrm{M}$ ) of selected compounds and intracellular 3'dATP concentrations ( $\mu \mathrm{g} / \mathrm{mL}$ ) after treatment with 3'dA and ProTides 30a,b,d-f).

Figure 3.20 shows a correlation between the cytotoxic activity ( $\mathrm{IC}_{50}$ values) of drugs ( $3^{\prime} \mathrm{dA}$ and selected ProTides $\mathbf{3 0 a}, \mathbf{b}, \mathbf{d} \mathbf{- f}$ ), and the intracellular levels of 3 'dATP measured. A link between activity of drug and level of putative intracellular metabolite ( 3 'dATP) emerged from these studies.

### 3.1.4.6 Selective cytotoxicity of 3'dA analogues on cancer stem cells

### 3.1.4.6.1 The concept of cancer stem cells (CSCs)

Cells with stem-cells characteristics have been identified in haematological malignancies and in the vast majority of solid tumours, such as breast cancer, ${ }^{229}$ lung cancer, ${ }^{230}$ colon cancer, ${ }^{231}$ prostate cancer, ${ }^{232}$ ovarian cancer, ${ }^{233}$ melanoma ${ }^{234}$ etc. These cells are considered responsible for cancer initiation, progression, metastasis, recurrence and drug resistance ${ }^{235-238}$ These cells share normal stem cells features such as the ability to selfrenew, proliferate and differentiate in a pluripotent manner. ${ }^{236,238}$ Two different models have been proposed in order to explain the heterogeneous cell phenotypes that constitute a tumour. ${ }^{237,239}$ According to the cancer stem cell model, rare cancer stem cells are responsible for the initiation and development of cancer, along with features such as recurrence and metastasis. ${ }^{239,240}$ Moreover the heterogeneity and hierarchy between cells in the tumour depend on CSCs asymmetric division, generating an identical cell clone and a differentiated daughter cell. A second theory, named the clonal evolution model considers all tumour cells responsible for the maintenance of the tumour, and the heterogeneity
derived from genetic and/or epigenetic modifications during cancer development. ${ }^{239,240}$ Understanding important features of these cells can help identify new anti-cancer therapeutics able to eradicate the tumour by targeting not only the bulk cells but also CSCs. ${ }^{235,236,241}$ The identification of CSCs from other cancer cells is based on the presence of specific CSCs biomarker phenotypes that can be used in fluorescence-activated cell sorting (FACS) to identify these populations and separate them from other cells. As an example, the leukaemia stem cells (LCS) were the first to be identified, due to the presence of a CD34 $/$ CD38 ${ }^{-}$surface marker phenotype. ${ }^{237}$ The lower expression of CD38 distinguished LSCs from normal hematopoietic stem cells, while the presence of CD34 is a feature of both cell types. Aside from extracellular and intracellular markers, CSCs appear to express high levels of ATP-binding cassette (ABC) transporters such as P-glycoprotein (P-gp), involved in the extracellular efflux of small molecules such as anti-tumour drugs, and contributing to the multidrug-resistance (MDR). ${ }^{239,242.243}$ Moreover, the up-regulation of self-renewal pathways and down-regulation of apoptotic pathways are features of CSCs. ${ }^{239}$ Examples of this are the up-regulation of PI3K/AKT signalling pathway causing cellular transformation and tumourigenesis in acute myeloid leukaemia (AML) and chronic myeloid leukaemia (CML). ${ }^{239}$ Moreover, aberration of the JAK/STAT pathway involved in tumour initiation through activation of several oncogenes was described in many tumours, including leukaemias. ${ }^{239}$ Aberrant activation of the transcription factor nuclear factor kappa B (NF-kB), affects the expression of several apoptosis-related proteins and causes cancer development, progression and chemo-resistance. ${ }^{239}$ Other important affected pathways are the Notch, Hedgehog ${ }^{238}$ and Wnt pathways, responsible for the CSCs selfrenewal capacity and differentiation. ${ }^{239}$
Resistance to chemotherapy and radiotherapy causes recurrence of malignancies, and this has been linked with the inability to eradicate CSCs through conventional therapy. ${ }^{237,244}$ For this reason, new therapeutic strategies are evolving in order to target the features of CSCs, exemplified by ligands or antibodies to target extracellular markers and agents capable of down-regulating the expression or inhibiting the activity of ABCtransporters. ${ }^{235,242}$ In addition, many agents were developed with the purpose of affecting the altered key signalling pathways in CSCs or to alter the tumour microenvironment responsible for promoting the CSCs development. ${ }^{239}$ Given that cells with a stem cell phenotype persist and are enriched in patients who have been treated with conventional therapies, ${ }^{245-247}$ there is a strong rationale for using stem cell targeting therapies in frontline therapy. This has the potential to induce more durable remissions and even cures.

### 3.1.4.6.2 Characterisation of KG1 a stem cell phenotype

The effect of 3 ' dA and some of the most active prodrugs was assessed on the CSC compartment of the acute myelogenous leukaemia cell line KGla by Prof. C. Pepper (School of Medicine, Cardiff University). A minor stem cell-like compartment is present in this cell line, with a distinct immunophenotype (CD34 $/$ /CD38 $/$ CD123 ${ }^{+}$). ${ }^{248,249}$


Figure 3.21: Gating strategy to define the LSC sub-population in the KG1a cell line. Analysis performed by Prof. C. Pepper (School of Medicine, Cardiff University).

This immunophenotype is evident through fluorescence-activated cell sorting (FACS) analysis of the cells after labelling with a cocktail of anti-lineage antibodies (PE-cy7), antiCD34 (FITC), anti-CD38 (PE) and anti-CD123 (PERCP cy5). The subpopulation expressing a leukaemic stem cell (LCS) phenotype can be identified and expressed as a percentage of all viable cells in the culture (Figure 3.21).

Furthermore, in the KG1a cell line, both the ATP-binding cassette sub-family G member 1 and 2 (ABCB1 and ABCG2) appear to be upregulated in the LSC portion of cells and the comparison between LSC and bulk tumour cells expression is given in Figure 3.22.


Figure 3.22: Increased expression of $A B C$ transporters in the stem cell population of KG1a. Analysis performed by Prof. C. Pepper (Cardiff University).

Hypoxic conditions, such as the tumour environment, are well known to induce stem cell phenotype. ${ }^{250-253}$ Analysis of the effect of these conditions on the KG1a cell line clearly showed that the percentage of cells expressing the distinct $\mathrm{CD} 34^{+} / \mathrm{CD} 38^{\circ} / \mathrm{CD} 123^{+}$ phenotype was increased (Figure 3.23).


Figure 3.23: Induction of cancer stem cell phenotype in hypoxic condition in the KG1a cell line. Analysis performed by Prof. C. Pepper (Cardiff University).

### 3.1.4.6.3 Analysis of LSC targeting after treatment with 3'dA and relative ProTides at the $L D_{50}$ concentration

The KG1a cell line was treated with 3'dA and some of the most active ProTides (30a-d) (Table 3.17).

| Common structure | Cpnd | Ar | R | $\mathrm{R}_{1}$ | AA |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  | 3'dA | - | - | - | - |
|  | 30a | Ph | Bn | Me | L-Ala |
|  | 30b | Nap | Bn | Me | L-Ala |
|  | 30c | Ph | Bn | H | Gly |
|  | 30d | Nap | $n$-Pen | $i \mathrm{Bu}$ | L-Leu |

[^0]Cell apoptosis was measured after 72 hours via the Annexin V/propidium iodide assay ${ }^{254}$ by dual-colour immunofluorescent flow cytometry. The phenyl L-alanine benzyl ester ProTide 30a emerged as the most active compound, with an $\mathrm{LD}_{50}$ value of $4.3 \mu \mathrm{M}, 100$ fold lower than the value reported for the parent nucleoside (Table 3.18). The napthyl Lleucine pentyl ester derivative 30d performed similarly to 30a, while on this cell line the naphthyl derivative 30b performed poorly, in contrast with the data reported on the other cell lines. The glycine analogue $\mathbf{3 0 c}$, on the other hand, was already reported as not very potent, and the poor activity was confirmed with the KG1a cell line (Table 3.18).

| Compound | KG1a |  |
| :---: | :---: | :---: |
|  | $\mathbf{L D}_{\mathbf{5 0}} \boldsymbol{\mu} \mathbf{M}$ | \% stem cells |
| 3'dA | 500 | 3.1 |
| 30a | 4.3 | 3.6 |
| 30b | 140 | 3.5 |
| 30c | 220 | 3.3 |
| 30d | 7.8 | 2.9 |
| Control | - | 3.3 |

Table 3.18: Cytotoxicity values and percentage of LSC in KG1a after treatment. Analysis performed under the supervision of Prof. C. Pepper (Cardiff University).

The $\mathrm{LD}_{50}$ values were calculated and the same concentrations were used to treat the cell lines and determine the effect of the selected compounds on the LSC compartment (CD34 ${ }^{+} / \mathrm{CD} 38^{-} / \mathrm{CD} 123^{+}$) after 72 hours of incubation.

This effect was analysed via FACS after labelling of the mentioned cell surface markers with specific luminescent antibodies: CD34-FITC, CD38-PE and CD123 PERCP-cy5. The effect of each compound on the stem cell compartment was compared with untreated cultures labelled with the same antibodies and was expressed as a proportion of the total cells analysed (Table 3.18). 3.3\% of cells express the LCS phenotype, and is reported as the control percentage. The parent nucleoside appears to selectively target this population by decreasing the percentage to $3.1 \%$, and this seems to be contrasted by ProTides 30a and 30b, which led to an increase in LCS population up to respectively 3.6 and $3.5 \%$. ProTide 30c leaves the percentage unaffected, while the leucine ProTide 30d decreases this percentage to $2.9 \%$.

These results could derive from an actual selective killing of the LSC population performed moderately by 3' dA and more significantly by ProTide 30d. Alternatively, these agents could lead to a differentiation of LSC cells into bulk tumour cells, which would not
express the typical extracellular phenotype and therefore this effect would lead to a decrease in the detected LSC population via FACS analysis.

### 3.1.4.6.4 Analysis of LSC targeting after treatment with 3'dA and relative ProTides at a range of concentrations

The evaluation of the preferential LSC targeting of 3 ' dA and relative ProTides on the KGla cell line was improved by the generation of a more complete dose-response curve for each compound and the evaluation of the effect on the LSC compartment at each concentration used, performed by Prof. C. Pepper (School of Medicine, Cardiff University).

| $\mathbf{C p n d}$ | $\mathbf{L D}_{\mathbf{5 0}} \boldsymbol{\mu} \mathbf{M}$ |
| :---: | :---: |
| $\mathbf{3} \mathbf{\prime} \mathbf{d A}$ | 120 |
| $\mathbf{3 0 a}$ | 4.2 |
| $\mathbf{3 0 b}$ | 90 |
| $\mathbf{3 0 c}$ | 110 |
| $\mathbf{3 0 d}$ | 8.2 |

Table 3.19: $\mathrm{LD}_{50}$ values for 3'dA, and selected prodrugs. All assays were carried out using KG1a cells and data are presented as mean of five independent experiments. Analysis performed under the supervision of Prof. C. Pepper (Cardiff University).


Figure 3.24: Comparison of the $\mathrm{LD}_{50}$ values for 3 ' dA , and selected prodrugs. All assays were carried out using KG1a cells and data are presented as mean ( $\pm \mathrm{SD}$ ) of five independent experiments. Analysis performed under the supervision of Prof. C. Pepper (Cardiff University).

The $\mathrm{LD}_{50}$ values were newly measured, and did not differ significantly from the previous data, with compounds 30a and 30d still emerging as the most active in this series (Table 3.19, Figure 3.24). The analysis of the LSC population percentage after treatment with compounds at different concentrations is reported in Figure 3.25.

Once again, treatment with both compounds 30b and 30c did not lead to any significant alteration of the LSC compartment, and this is confirmed throughout all the range of
concentrations. On the other hand, 3'dA and ProTides 30a and 30d appear to reduce this population at concentrations ranging from high micromolar to millimolar. At very high concentrations $\left(10^{-3}-10^{-2} \mathrm{M}\right), 3^{\prime} \mathrm{dA}$ seems to revert the pattern and leave the percentage unaffected. In contrast, 30a and 30d appear to consistently reduce the population even at the highest doses. However, only compound 30a can be considered to significantly reduce the LSC population to a larger extent than the parent nucleoside at concentrations above 1 mM (Figure 3.25).



Figure 3.25: Analysis of the LSC targeting capacity of 3 ' dA and selected prodrugs. All data are the mean ( $\pm$ SD) of three independent experiments. Analysis performed under the supervision of Prof. C. Pepper (Cardiff University).

Although a clear correlation between selective targeting of the LSC population and a potential mechanism of action could not be identified, at a broad range of concentrations compound 30a appears to clearly affect this subset, which is in contrast with the apparent increase in LSC population noted in the previous studies at $\mathrm{LD}_{50}$ dose. However, the selective cytotoxic effect of compounds 30a and 30d on LSC population becomes significant at concentrations 1 to $2 \log$ higher than the $L_{50}$ (which is around $10^{-6} \mathrm{M}$, Figure 3.25).

### 3.1.4.6.5 Effect of 3'dA and ProTides on $\beta$-catenin levels of the KG1a cell line

The nucleoside 3 ' dA was reported to supress cell proliferation affecting GSK- $3 \beta / \beta$-catenin signalling through down-regulation of the $\beta$-catenin intracellular levels in leukaemic cells. ${ }^{220} \beta$-Catenin is involved in the Wnt pathway, involved in LSCs self-renewal capacity. ${ }^{255,256.257}$ Glycogen synthase kinase $3 \beta$ (GSK-3 $)$ regulates the stability of $\beta$ catenin by phosphorylation. ${ }^{258}$ At the cellular basal level, $\beta$-catenin is part of a complex of proteins and undergoes phosphorylation by GSK- $3 \beta$, which activates $\beta$-catenin degradation via the proteasome-dependent pathway. ${ }^{220,259}$ Signalling through the Wnt pathway causes inactivation of GSK-3 $\beta$, resulting in $\beta$-catenin dissociation from the complex and
translocation to the nucleus, where it regulates the expression of self-renewal genes. ${ }^{220}$ The expression of $\beta$-catenin in AML cells was associated with clonogenic capacities and poor prognosis. ${ }^{260}$ Treatment of leukaemic cell lines K562, THP1 and the lymphoma cell line U937 with 3'dA was reported to cause a dramatic reduction in $\beta$-catenin intracellular levels, that was paralleled by a significant reduction of the colony forming capacity of these cell lines. ${ }^{220}$ These effects were reported to derive from increase in proteasomedependent degradation of $\beta$-catenin, which was in turn dependent on modulation of the Wnt/GSK-3 3 signalling.

These intriguing results prompted us to study the effects of 3 ' dA and relative ProTides on the intracellular levels of $\beta$-catenin in the KG1a cell line. First, the expression of $\beta$-catenin was assessed both in bulk tumour cells and in the stem-cell population within the KG1a cell line. Secondly, our goal was to understand whether these levels were modulated by the treatment with 3'dA and/or 3'dA ProTides and if the effects were similar on all cells or if the alteration was only visible in one cell population. The final aim was to understand whether the potential LSC targeting of 3 ' dA and ProTide 30d could derive from a perturbation of $\beta$-catenin levels.

Cells were treated with $\mathrm{LD}_{50}$ concentrations of both 3 ' $\mathrm{dA}\left(\mathrm{LD}_{50}=500 \mu \mathrm{M}\right)$ and ProTides 30a $\left(\mathrm{LD}_{50}=4.3 \mu \mathrm{M}\right)$ and $\mathbf{3 0 d}\left(\mathrm{LD}_{50}=7.8 \mu \mathrm{M}\right)$. Subsequently, cells were stained with both the already mentioned cocktail of antibodies useful to detect the presence of the distinctive extracellular markers (CD34 $/$ CD38 ${ }^{-} / \mathrm{CD} 123^{+}$) and with Anti- $\beta$-catenin-FITC. Before adding the last $\beta$-catenin specific antibody, cell permeability was increased by incubation with PBS-Tween. This step was necessary for the intracellular binding of the selective antibody to intracellular $\beta$-catenin.

Figure 3.26 reports the analysis of the $\beta$-catenin levels in the bulk subset of KGla cells and in the (CD34 ${ }^{+} / \mathrm{CD} 38^{-} / \mathrm{CD} 123^{+}$) population, measured as fluorescent intensity related to $\beta$ catenin intracellular levels. Increased levels of $\beta$-catenin appear to be present in the LSC population, which is in accordance with the characteristics of other leukaemic cells, ${ }^{220,249}$ confirming its interest as a target of investigation.


Figure 3.26: Effect of treatment with 3'dA and ProTides 30a and 30d on the $\beta$-catenin intracellular levels of the KG1a cells leukaemic stem cell compartment. Analysis performed under the supervision of Prof. C. Pepper (Cardiff University).

Treatment of leukaemic cell lines such as U937, K562 and THP1 was reported to lead to a decrease in the levels of intracellular $\beta$-catenin. ${ }^{220}$ Our results are in contrast to the previous reports, as we observed an increase in such levels both in the LSC compartment and in the bulk tumour cells. This may suggest that, although at the $\mathrm{LD}_{50}$ value 3 ' dA seems to reduce the amount of cancer stem like cells within the cell line, the treatment may select for cells with a higher amount of intracellular $\beta$-catenin, hence the higher levels detected via FACS analysis (Figure 3.26). On the other hand, at $\mathrm{LD}_{50}$ concentrations, both ProTides 30a and 30d were found not to significantly reduce the LSC population, although it should be highlighted that this population was not enriched either. Therefore, the activity of both compounds at $\mathrm{LD}_{50}$ concentrations was not selective for the LSC compartment over the bulk tumour cells. The effect of both compounds on the $\beta$-catenin levels in the bulk tumour cells was not leading to a significant decrease compared to the untreated cells. On the other hand, a difference could be noticed on the effect of ProTides 30a and 30d, on the $\beta$-catenin levels in the LSC compartment, which seemed to be increased in comparison to the control after treatment with compound 30d, while compound 30a did not significantly alter such levels. This might suggest that ProTide 30d, similarly to 3 'dA, enriched the LSC population with cells expressing higher levels of $\beta$-catenin, in contrast to the effects of $\mathbf{3 0 a}$. Although still preliminary, the reported data show that unlike other nucleoside based treatment, which lead to an enrichment in the cancer stem cell population within treated tumour, 3 ' dA and ProTides seem to avoid the enrichment of the cell line with cells with potentially more aggressive LSC characteristics. Moreover, at high concentrations both parent nucleoside and ProTides 30a and 30d appear to selectively target the LSC population in the KG1a cell line, leading to its reduction in percentage within all the cells. Analysis of the $\beta$-catenin levels after treatment with 3 ' dA and ProTides 30a and 30d did
not show a reduction in the levels of the intracellular mediator, however, seemed to indicate that only 30a over the other compounds does not lead to an increase in $\beta$-catenin levels, highlighting the intriguing potential of such compound.

### 3.1.5 Mechanistic investigations on the metabolic activation of 3'dA derivatives

### 3.1.5.1 Carboxypeptidase Y (CPY) enzymatic NMR assay

ProTides are activated inside the body in 4 steps, as discussed in Chapter 1. The process of intracellular activation involves a first step catalysed by a carboxyesterase enzyme (Cathepsin A), ${ }^{261}$ which is believed to be responsible for the cleavage of the amino acid ester moiety. ${ }^{129,130,261}$ Ester hydrolysis studies can be performed in order to gather information on these first steps in the activation of ProTides, and carboxypeptidase Y (CPY) can be used as a surrogate of cathepsin A enzyme, since it belongs to the same family of C-type carboxypeptidases and was reported to share similarities in the active site. ${ }^{262}$ The enzymatic reaction can be monitored by ${ }^{31} \mathrm{P}$ NMR, according to a procedure developed in the McGuigan group: the ProTide is dissolved in acetone- $d 6$ and trizma buffer ( pH 7.6 ) is added to the solution. ${ }^{147}$ A blank spectrum is recorded. The enzyme CPY, previously dissolved in trizma buffer ( pH 7.6 ), is added to the ProTide mixture (1 U per mmol of ProTide), and ${ }^{31} \mathrm{P}$ NMR experiments are recorded.

CPY experiments were carried out on different 3'dA derivatives, namely 5'phosphoroamidates 30a, 30b and 30f, in order to understand whether the better cytotoxic profile of 32a and 32b in comparison to 30f could derive from better intracellular processing, relative to the first activation step. Comparison between the processing rates of 2'-phosphoroamidate 28b and 2,5'-bisphosphoroamidate 29b, against the 5'phosphoroamidate counterpart 30b was also assessed. Moreover, the 5'phosphorodiamidate $\mathbf{3 2}$ was processed by CPY and the rate compared to the 5 '-ProTide counterparts 30a and 30b.

### 3.1.5.1.1 Enzymatic CPY processing of 3'dA 5'-phosphoroamidates

Simulation of the intracellular processing of $5^{\prime}$-phosphoramidates 30a, 30b and 30f was performed via CPY NMR assay, in order to monitor the conversion into final aminoacyl metabolites $\mathbf{3 0 a}{ }^{\mathrm{C}}, \mathbf{3 0 b}^{\mathrm{C}}$ and $\mathbf{3 0 f}{ }^{\mathrm{C}}$.



Scheme 3.7: Conversion of ProTides 30a, 30b and 30f into amino acyl phosphate monoester intermediates $\mathbf{3 0 a}^{\mathrm{C}}, \mathbf{3 0 b}^{\mathrm{C}}$ and $\mathbf{3 0 f}{ }^{\mathrm{C}}$.

The final conversion into $3^{\prime}$ dAMP (metabolites $\mathbf{3 0 a}{ }^{\mathbf{D}}, \mathbf{3 0 b}^{\mathbf{D}}$ and $\mathbf{3 0 f}{ }^{\mathbf{D}}$ ) can be simulated by docking of intermediates $\mathbf{3 0 a}{ }^{\mathbf{C}}, \mathbf{3 0 b}{ }^{\mathbf{C}}$ and $\mathbf{3 0 f}{ }^{\mathbf{C}}$ into the active pocket of the Hint enzyme, which was recognised as the enzyme responsible for the phosphoramidate bond cleavage. ${ }^{128}$ These assays will be described below.

Phenyl L-alanine benzyl ester ProTide 30a was treated with CPY enzyme and Figure 3.27 shows the stacked ${ }^{31} \mathrm{P}$ NMR spectra that were recorded overnight. Two peaks resonating at 3.71 and 3.53 ppm , corresponding to the two diasteroisomers, were processed into two more downfield peaks corresponding to intermediate $\mathbf{3 0 a}^{\mathbf{A}}$ (Scheme 3.7). Complete conversion of $\mathbf{3 0 a}$ into $\mathbf{3 0 a}{ }^{\mathbf{C}}$ happened within 1 hour from enzyme addition and was confirmed both by the presence of a ${ }^{31}$ P NMR peak at 7.30 ppm , and by ES/MS analysis of the crude mixture, showing a predominant peak corresponding to the structure of $\mathbf{3 0 a}{ }^{\mathbf{C}}$ (calculated: $m / z=402.11[M]$, found: $m / z=401.08\left[\mathrm{M} \mathrm{-} \mathrm{H}^{+}\right]$).


Figure 3.27 (A) ${ }^{31} \mathrm{P}$ NMR spectra ( 202 MHz , acetone- $d \sigma$ ) showing conversion of ProTide $\mathbf{3 0 a}$ to intermediate $\mathbf{3 0 a}{ }^{\mathrm{C}}$. (B) \% variation of ${ }^{31} \mathrm{P}$ NMR integrals of each species over the time.

The same assay was performed on ProTide 30b, the naphthyl counterpart of 30a. Two peaks corresponding to $\mathbf{3 0 b}$ are visible in the blank experiment at 4.02 and 3.84 ppm (Figure 3.28). After 7 minutes from the enzyme addition, the two diasteroisomers are converted almost entirely into two peaks corresponding to intermediate $\mathbf{3 0 b}^{\mathbf{A}}$ (4.92, 4.96 $\mathrm{ppm})$ and the simultaneous appearance of a peak at 7.30 ppm indicates the immediate formation of the metabolite $\mathbf{3 0 b}^{\mathbf{C}}$ (Scheme 3.7). The complete conversion to intermediate $\mathbf{3 0 b}^{\mathrm{C}}$ takes approximately 56 minutes, and is confirmed by mass analysis on the centrifuged mixture, showing a prevalent peak corresponding to compound $\mathbf{3 0 b}{ }^{\mathbf{C}}$ with the same $m / z$ as for $\mathbf{3 0 b}{ }^{\mathbf{C}}$. These results suggest that the anticancer activity of both compounds 30a and 30b in the in vitro assay could be a consequence of rapid intracellular metabolism to the active species.


Figure 3.28: (A) ${ }^{31} \mathrm{P}$ NMR spectra ( 202 MHz , acetone- $d 6$ ) showing conversion of ProTide 30b to intermediate $\mathbf{3 0 b}{ }^{\mathbf{C}}$ in 56 minutes. (B) Variation of ${ }^{31} \mathrm{P}$ NMR integrals of each species over the time.

Notably, the 30b diasteroisomers are processed at a similar speed, with no apparent difference in the metabolic stability of the two species, in contrast with the behaviour of the two diasteroisomers of 30a, which are converted more slowly (Figure 3.27).

The CPY assay, applied to the poorly active 3'dA ProTide 30f (Scheme 3.7), bearing naphthyl dimethylglycine methyl ester, showed that conversion into intermediate $\mathbf{3 0 f}{ }^{\mathbf{C}},\left({ }^{31} \mathrm{P}\right.$ NMR signal at 5.08 ppm ; calculated $m / z: 416.12$ [M], found $m / z=415.13\left[\mathrm{M}-\mathrm{H}^{+}\right]$) was very slow and not complete after 14 hours (Figure 3.29).


Figure $3.29{ }^{31} \mathrm{P}$ NMR spectra ( 202 MHz , acetone-d6) showing slow processing of ProTide $30 f$ to intermediate $\mathbf{3 0 f}{ }^{\text {C }}$.

Moreover, one of the two diasteroisomers (at 2.59 ppm ) appeared to be more stable than the other (at 2.56 ppm ). These results strongly suggested that the low cytotoxic profile of this compound might derive from inefficient intracellular processing of the ProTide moiety.

### 3.1.5.1.2 Enzymatic experiment on 2'-phosphoramidate

To the best of our knowledge, the synthesis and biological evaluation of phosphoramidate prodrugs of 3'-deoxynucleosides has never been reported before, nor the evaluation of the 2'-phosphoroamidate regioisomers. Although demonstrating lower anticancer activity compared to their 5'-regioisomers, compounds 28a and 28b still proved more potent than the parent nucleoside in most of the cell lines. These results could be due to their lipophilicity, that would allow cell entering by passive diffusion, and once within the cell these derivatives could be metabolically activated to $3^{\prime}$ dA-2'-monophosphate ( $\mathbf{2 8 b}^{\mathbf{D}}$, Scheme 3.8), and exert their activity with an unknown mechanism of action. ${ }^{263}$


Scheme 3.8: Putative bioactivation of 2'-phosphoroamidate 28b.

Alternatively, once inside the cell, these compounds could undergo cleavage of the phosphate moiety catalysed by nucleotidase enzymes, with release of the parent nucleoside 3'dA. These enzymes have already been reported to be responsible for resistance to the treatment with anticancer nucleoside analogues such as gemcitabine and cytarabine. ${ }^{264,265}$ A third more complex mechanism could be the isomerisation of the released $2^{\prime}$ monophosphate to the 5 '-position, generating 3'dAMP.

Presently, no clear mechanism of action could be confirmed. However, in an attempt to explain the activity of compound 28b, the 2 '-regioisomer of ProTide 30b, a CPY-mediated enzymatic assay was performed (Figure 3.30).


Figure 3.30: (A) ${ }^{31} \mathrm{P}$ NMR spectra ( 202 MHz , acetone-d $\sigma$ ) showing complete conversion of compound $\mathbf{2 8 b}$ to intermediate 28b ${ }^{\mathbf{C}}$. (B) Variation of ${ }^{31} \mathrm{P}$ NMR integrals of each species over the time.

Compound 28b peaks appear in the blank experiment at 2.63 and 2.54 ppm. After 7 minutes from the addition of CPY, the conversion to intermediate $\mathbf{2 8 b}^{\mathbf{A}}$ is complete, with the comparison of two peaks at 3.39 and 2.89 ppm . In 20 minutes, the peak corresponding to compound $\mathbf{2 8 b}{ }^{\mathbf{C}}$ could be seen at 6.28 ppm , and the complete conversion happens in 70 minutes. As for compound $\mathbf{3 0} \mathbf{b}^{\mathbf{C}}$, a mass experiment of the final solution from enzymatic reaction on compound $\mathbf{3 0 b}$ confirmed the final peak at 6.28 ppm corresponded to intermediate $\mathbf{2 8 b}{ }^{\text {C }}$ (calculated $m / z=400.09[M]$, found $m / z=402.12\left[M+2 \mathrm{H}^{+}\right]$). This result suggests that ProTide 2 '-regioisomes could be processed by similar enzymes involved in the metabolism of the $5^{\prime}$ 'counterparts, and release the 2 '-monophosphate inside the cell.

In addition to the reported data, the metabolism of $2^{\prime}, 5^{\prime}$-bisphosphoroamidate 29a by CPY was assessed, in order to gather information on the putative mechanism of intracellular activation. A bulk molecule such as 29a was hypothesised to be processed less efficiently by CPY.


Figure 3.31: Putative conversion of bis-phosphoroamidate 29b into amino acyl phosphate monoester intermediates $\mathbf{2 9 b}{ }^{\text {C }}$.


Figure 3.32: (A) ${ }^{31} \mathrm{P}$ NMR spectra ( 202 MHz , acetone-d $d$ ) showing processing of 29b by CPY.

However, the result of the CPY assay, shown in Figure 3.32 through the stacked ${ }^{31} \mathrm{P}$ NMR spectra relative to the overnight assay, suggest that $\mathbf{2 9 b}$ is potentially processed by the enzyme into the putative metabolites represented in Figure 3.31. The peaks corresponding to 29b are not clearly visible in the blank spectrum, due to low solubility of the molecule in acetone- $d 6$ and trizma buffer ( pH 7.6 ), in fact the spectra appears as a broad peak in the area between 2 and 4 ppm . However, once the processing by CPY begins, some more downfield peaks appear, potentially corresponding to the structures $\mathbf{2 9 b}^{\mathbf{A}}$ (Figure 3.31), as a mixture of isomers either totally or partially lacking the ester moieties. Concomitantly, peaks in the area between 6 and 7 ppm also develop, potentially corresponding to the different isomers $\mathbf{2 9 b}{ }^{\text {C }}$. These results suggest that 29b is processed by CPY to some extent, although at a slower rate that the 2 '-phosphoramidate or 5 '-phosphoramidate counterpart, and this could explain the activity of such prodrug inside the cell.

### 3.1.5.1.3 Enzymatic experiment on diamidate

The 5'-phosphorodiamidate of 3'dA $\mathbf{3 2}$ did not show similar anticancer potency as seen for 3'dA ProTides, although was still more potent than 3 ' dA in the cytotoxicity screenings. In order to get an insight into the possible reason why this phosphate prodrug did not show the same anticancer potential as other prodrugs, such as the ProTides 30a and 30b, a CPY assay was performed.


Scheme 3.9: Conversion of diamidate $\mathbf{3 2}$ into intermediate $\mathbf{3 2}^{\text {C }}$.

After addition of CPY, $\mathbf{3 2}\left({ }^{31} \mathrm{P}\right.$ NMR peak at 13.77 ppm$)$ was converted to intermediate $\mathbf{3 2}^{\mathbf{A}}$ (Scheme 3.9, Figure 3.33), lacking the ester moiety therefore presenting two diastereomeric peaks at 14.36 and 14.59 ppm . Finally the formation of the metabolite lacking one of the amino acids $\left(\mathbf{3 2}{ }^{\mathbf{C}}\right)$ was confirmed by comparison of the peak at 7.04 ppm and by mass analysis of the centrifuged crude (calculated $m / z=400.09$ [M], found $\left.m / z=402.1\left[\mathrm{M}+2 \mathrm{H}^{+}\right]\right)$.


Figure 3.33: (A) ${ }^{31} \mathrm{P}$ NMR spectra ( 202 MHz , acetone-d $\sigma$ ) showing complete conversion of compound $\mathbf{3 2}$ to intermediate $\mathbf{3 2}^{\text {C }}$ in 120 minutes. (B) Variation of ${ }^{31} \mathrm{P}$ NMR integrals of each species over the time.

The activation of $\mathbf{3 2}$ into intermediate $\mathbf{3 2}^{\text {C }}$ is slower than the conversion of $\mathbf{3 0 b}$ into $\mathbf{3 0 b}^{\mathbf{C}}$, as it happens within 2 hours (Figure 3.33), instead of 56 minutes. The slower activation of
the phosphorodiamidate derivative could thus be the reason for the lower anticancer potency of this derivative, compared to $\mathbf{3 0 b}$.

### 3.1.5.2 Docking studies

### 3.1.5.2.1 Docking of compounds in the active site of Hint-1

The second enzymatic step required for the activation of ProTides is the phosphoramidate bond cleavage, and the enzyme responsible for this step was reported to belong to the human histidine triad nucleotide-binding protein (Hint). ${ }^{135,261}$ In order to predict the likelihood of processing of 3 ' dA prodrugs by Hint, docking studies within the enzyme active site were carried out using PLANTS. The crystal structure of Hint was downloaded from the protein data bank (PDB 1RZY), ${ }^{266}$ and the interaction between each selected aminoacyl phosphate monoester intermediate and the crucial residues within the active pocket (Ser107, His 112 and His114), implicated in the P-N bond cleavage, were analysed with MOE software.



C Docking of $\mathbf{3 0 f}{ }^{\text {C }}$ within Hint
Figure 3.34 Docking of compounds (A) $\mathbf{3 0 a} \mathbf{a}^{\mathbf{C}}, \mathbf{3 0 b}^{\mathbf{C}}, \mathbf{3 2}^{\mathbf{C}}, \mathbf{( B )} \mathbf{2 8 b}^{\mathbf{C}}$ and (C) $\mathbf{3 0 f}^{\mathrm{C}}$ within the active pocket of Hint enzyme.

Comparison between the intracellular metabolic activation processes of the $5^{\prime}$ 'phosphoroamidate $\mathbf{3 0 a}{ }^{\mathbf{C}}$ (corresponding to the same structure as $\mathbf{3 0 b}^{\mathbf{C}}$ and $\mathbf{3 2}^{\mathbf{C}}$ ), and the $2^{\prime}$ regioisomer $\mathbf{2 8 b}{ }^{\mathrm{C}}$ was carried out by docking the L-alaninyl phosphate monester metabolites of both compounds within Hint (Figure 3.34, A and B). Similar positioning of the phosphoroamidate moiety close to the residues responsible for the phosphoramidase activity could be noticed for both compounds. However the positioning of the $5^{\prime}$ 'regioisomer seemed to allow better interaction of the nucleobase with the active site, which suggests more efficient positioning of the moiety within the pocket. However, the two moieties interact very similarly with Hint, therefore the P-N bond cleavage step should not represent a limiting step for the activation of compound $\mathbf{2 8 b}{ }^{\mathrm{C}}$ over $\mathbf{3 0 a}{ }^{\mathrm{C}}$. Very similar positioning could be observed for compound $\mathbf{3 2}^{\mathbf{C}}$ (Figure 3.34, C), therefore suggesting that the lower activity of the dimethylglycinyl compound $\mathbf{3 2}$ should not result from a lower activation step catalysed by Hint, but more likely by a slower processing by CPY, as suggested by the NMR assay performed (Figure 3.29).

### 3.1.6 Stability studies

### 3.1.6.1 Stability to adenosine deaminase-mediated catabolism

### 3.1.6.1.1 UV assay

The poor activity of cordycepin in the in vitro assays could derive primarily from a quick intracellular metabolism to inactive inosine, catalysed by adenosine deaminase enzyme, as already extensively reported. ${ }^{189}$ In order to have a proof of the quick metabolism of the nucleoside, and confirm the stability of $3^{\prime} \mathrm{dA}$ ProTides in the same conditions, an enzymatic assay was designed. In the first assay, $3^{\prime} \mathrm{dA}$ was treated with a solution of adenosine deaminase from calf and the UV absorbance was recorded over time.


Figure 3.35: Change in UV absorbance spectra of 3'dA after addition of a solution of ADA (in phosphate buffer pH 7.4).

Figure 3.35 reports the UV spectrum of 3 ' dA (red line) and the shift of the maximum wavelength of absorption from 259 nm to lower wavelengths of the other curves describes the appearance of the metabolite $3^{\prime}$ '-deoxyinosine ( 3 '-dIno), characterised by a maximum absorbance wavelength of $249 \mathrm{~nm} .{ }^{225}$ The metabolic deamination is complete within 2 minutes from the enzyme addition.


Figure 3.36: UV absorbance spectra of $\mathbf{3 0 a}$ after addition of a solution of ADA.

On the other hand, ProTide 30a, treated under the same conditions, was stable to deamination up to 2 hours from the addition of the enzyme solution, confirming the stability of 3'dA ProTides to deamination.

### 3.1.6.1.2 Docking within the ADA active site

In order to have an additional indication of the stability of $3^{\prime} \mathrm{dA}$ prodrugs to ADAmediated catabolism, we docked the parent nucleoside, $5^{\prime}$ '-phosphoroamidate $\mathbf{3 0 a}$ and the 2'-regioisomer 28a within the ADA active site (ADA crystal structure was downloaded from the protein data bank, PDB 3IAR).



Figure 3.37 A. Interaction of adenosine with catalytic residues within the ADA enzyme (image from Vodnala et al.). ${ }^{225} \mathbf{B}$. Docking interaction of 3 'dA within the ADA active site.

Figure 3.37 A shows the key interactions of the natural substrate adenosine within ADA, involving the crucial residues Glu217, Asp295, His238, Gly184 and Asp296. Figure 3.37B shows the interaction of 3 'dA in ADA, involving directly Glu217, Asp295, Gly184 and the close proximity to His 238 . The correct positioning resulting from docking confirms the very quick processing of this substrate by ADA, noticed in the UV assay. On the other hand, when both ProTide 30a and 2'-phosphoraminate 28a were docked within the ADA active pocket, these structures could not properly fit in the enzyme, due to the much larger dimensions compared to the natural substrates (Figure 3.38).


Figure 3.38 Overlap of 3'dA (white) and ProTide 30a (blue, A) or 2'-phosphoramidate 28a (blue, B) within the active site of ADA.

From the overlap of 3 ' dA and both compounds in ADA, it is clear that the 3 ' dA derivatives need to assume bent conformations in order to fit within the enzyme and in the case of 28a this does not allow correct positioning of the adenine ring close to the catalytic residues. These results, along with the stability of ProTide 30a noticed with the ADA UV assay, suggest that the improved activity of 3 ' dA prodrugs over the parent nucleoside might derive from increased stability to ADA-mediated catabolism.

### 3.1.6.2 Stability of 3'dA ProTides in Mice and Human plasma

The stability of 3 ' dA and ProTides 30a, 30b and 30f was assayed both in untreated human and mouse plasma, with or without the presence of ENHA, ADA inhibitor, by Dr. E. Ghazaly (Barts Institute, London). The study aimed at understanding whether 3'dA prodrugs could offer a stability advantage to plasmatic ADA-mediated deamination, which negatively affects the cytotoxic profile of the parent nucleoside.

Compounds were incubated in plasma at $10 \mu \mathrm{~g} / \mathrm{mL}$ concentrations and mixed for 30 seconds, after which the first sample was taken (time: 0 hr ). Afterwards, the compound was extracted and the concentration measured.


Figure 3.39: Stability assay of 3'dA in human plasma (A) and mouse plasma (B). Assays performed by Essam Ghazaly (Barts Institute, London).

When 3'dA was incubated in human plasma (Figure 3.39, A), without the addition of EHNA the concentration of compound decreased immediately and gradually, until completely metabolised after 4 hours. When EHNA was added, the stability of 3 'dA was significantly increased, and the compound could still be detected in plasma after 24 hours. The metabolism of 3 'dA in mouse plasma was similar although the nucleoside was completely metabolised after 1 hour of plasma incubation without the presence of ENHA.


Figure 3.40: Stability assay of ProTide 30a in human plasma (A) and mouse plasma (B). Assays performed by Essam Ghazaly (Barts Institute, London).

The incubation of ProTide 30a in human plasma showed increased stability compared to the parent nucleoside, due to stable concentration up to 4 hours from the beginning of the assay. Moreover, the addition of EHNA does not significantly affect the compound stability. Conversely, the stability of ProTide 30a in mouse plasma was poor; in fact the concentration of compound dropped immediately down to $\mathrm{ng} / \mathrm{mL}$ concentrations after 30 seconds from the incubation. Moreover the presence of EHNA did not significantly affect these results, suggesting that the compound was not metabolised by ADA (as expected) but through other pathways, potentially involving esterase enzymes. In fact, higher levels of carboxylesterase enzymes were reported in rodents, ${ }^{267}$ compared to human, monkey and
dogs, and were suggested to induce reduced stability of ProTides in in vivo mouse xenografts. ${ }^{151,268}$

Similar outcomes resulted from ProTide 30b incubation, with noticeable stability in human plasma, which was unaffected by addition of EHNA; on the other hand, the compound was rapidly metabolised in mouse plasma, down to $\mathrm{ng} / \mathrm{mL}$ concentration immediately after incubation.



Figure 3.41: Stability assay of ProTide 30b in human plasma (A) and mouse plasma (B). Assays performed by Essam Ghazaly (Barts Institute, London).

ProTide 30f emerged as very stable in human plasma, and more stable than the other ProTides in mouse plasma, with concentrations in the $\mu \mathrm{g} / \mathrm{mL}$ range still after 1 hour after the beginning of the assay (Figure 3.42).


Figure 3.42: Stability assay of ProTide $\mathbf{3 0 f}$ in human plasma (A) and mouse plasma (B). Assays performed by Essam Ghazaly (Barts Institute, London).

These results, along with the poor activity noticed in in vitro assays and the slow activation in the CPY assay, suggest that the lack of anticancer activity of this derivative may derive from poor processing of the ProTide moiety.

### 3.1.6.3 3'dA ProTides stability in human hepatocytes

The optimal metabolism of ProTides depends on their enzymatic and chemical stability. Upon administration, the circulating ProTides are exposed to challenging conditions, such as catalytic enzymes in the liver and should resist long enough to enter tumour cells before being degraded. Conversely, ProTides unaffected by enzymatic digestion might not be efficiently converted into active compounds inside the cancer cells. To further select the lead candidates with suitable chemical stability, five ProTides were incubated with human hepatocytes, performed by Eurofins CEREP USA laboratories. In the human hepatocyte extracts the half-lives of the six compounds varied from 24 minutes for compound 30e up to 99 min for compound 30f. The selection criteria included identifying a compound with a mid-range half-life in human hepatocytes. The short half-life of $3^{\prime} \mathrm{dA}(48 \mathrm{~min})$ may be an indication of quick conversion into 3 'dIno due to the activity of ADA. The very quick metabolism of $\mathbf{3 0 e}$, resulting in the highest clearance, may be due to the presence of a second ester moiety attached to the aryl group, that, along with the ester attached to the amino acid group as in each ProTide, would increase the susceptibility to esterasemediated processing.

| Common structure | Cpnd | $\mathbf{A r}$ | $\mathbf{R}$ | $\mathbf{R}_{\mathbf{1}} / \mathbf{R}_{\mathbf{2}}$ | $\mathbf{A A}$ | $\mathbf{t}_{\mathbf{1 \| 2}}$ | $\mathbf{C l}_{\text {int }}$ |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |

Table 3.20 Incubation of 3 ' dA and selected ProTides in human cryopreserved hepatocytes. Half life ( $\mathrm{t}_{1 / 2}$ ) expressed in minutes, intrinsic clearance ( $\mathrm{Cl}_{\text {int }}$ ) expressed as $\mu \mathrm{L} / \mathrm{min} / 10^{6}$ cells. Assays performed by Eurofins CEREP USA laboratories.

In contrast, compound $\mathbf{3 0 f}$ shows higher $\mathrm{t}_{1 / 2}$ value, and this, along with the poor cytotoxicity profile of this compound, could indicate slow processing. Compounds 30a and 30b (44-48 min) show very similar stability profiles, with favourable $t_{1 / 2}$ values, indicating a good balance between appropriate speed of processing and extracellular stability. Compound $\mathbf{3 0 e}$ has an intermediate stability profile ( 63 min ). From these stability studies, compounds 30a and 30b result again as promising 3'dA derivatives, with stability in human hepatocytes comparable to ProTides already progressing in the clinic. ${ }^{151}$

### 3.1.7 Conclusions

The synthesis of 3 'dA after optimisation of the literature reports was accomplished, and allowed the generation of a family of nucleoside derivatives through the ProTide and phosphorodiamidate prodrug approaches. 5'-Monophosphate prodrugs were produced along with $2^{\prime}$-monophosphate and $2^{\prime}, 5$ '-bis- $O$-phosphate prodrugs. All the synthesised derivatives proved to be more effective cytotoxic agents than the parent nucleoside on a broad selection of malignant cells. In particular 5'-phosphoramidates of 3'dA resulted as the most potent anticancer agents, with $\mathrm{IC}_{50}$ values in the low $\mu \mathrm{M}$ range especially on cells derived from haematologic malignancies and the ProTides emerging as the most active were the L-alanine benzyloxy derivatives $\mathbf{3 0 a}$ and $\mathbf{3 0 b}$. The $\mathrm{IC}_{50}$ values of some of the most active compounds and $3^{\prime} \mathrm{dA}$ were assayed on a small selection of leukaemic cell lines (CEM, K562 and HL60), and the 3'dATP levels were subsequently measured on these cells. The intracellular levels of 3'dATP, the active intracellular metabolite, were found to be higher after treatment with compounds 30a and 30b, correlating with the better cytotoxicity profile of these drugs. The most active compounds were also assayed, along with the parent nucleoside, on the KG1a cell lines, possessing a distinct sub-population endowed with cancer stem cells characteristics. At high concentrations, ProTide 30a was found to selectively target this population with higher efficiency compared to the other derivatives.

The metabolism of 3'dA prodrugs was assayed first with a NMR CPY assay, predicting quick intracellular processing for the active ProTides 30a and 30b and diamidate $\mathbf{3 2}$ and slow for the poorly active derivative 30f, therefore confirming that quick intracellular processing is a determinant for the intracellular activity of $3^{\prime} \mathrm{dA}$ prodrugs. CPY was also found to potentially process the $2^{\prime}$-phosphoromidate derivative 28b at a similar speed as the ProTide counterpart 30b, and to process the bis-phosphoroamidate derivative 29b to some extent. Docking into the Hint enzyme, responsible for the last step in 3'dA prodrugs activation, appeared to be similar with different intermediate structures, suggesting that this is not a limiting step in $3^{\prime} \mathrm{dA}$ prodrug activation. Metabolism by ADA was evaluated, and was found to be very quick in the case of $3^{\prime} \mathrm{dA}$, while not significant in the case of one example of ProTide (30a). Docking studies within the ADA active site correlated these results to the bulkier structure of 3 'dA prodrugs, impeding a correct positioning of these compounds inside the pocket. The stability of 3 'dA was assayed in human and mouse plasma, and was found to be similar in both assays and strongly dependent on ADA activity, as evaluated after addition of EHNA, an ADA inhibitor. On the other hand, for

ProTides 30a and 30b, the stability was found not to be significantly affected by the presence of EHNA, although considerably lower in mouse compared to human plasma, potentially due to higher expression of carboxyl esterase enzymes in rodent plasma. Stability evaluation in human hepatocytes confirmed a suitable pharmacokinetic profile for Protides 30a and 30b, comparable to ProTides in clinical development such as NUC1031. ${ }^{151}$ In conclusion, combining all stability and cytotoxicity data, ProTides 30a and 30b emerged as the most intriguing 3'dA prodrugs, deserving further evaluation. ProTide 30a, although endowed with lower cytotoxicity compared to $\mathbf{3 0 b}$ on some cell lines, was capable of selectively targeting the leukaemic stem cell compartment of KG1a cell line. This result was considered potentially useful for the in vivo application of such drug, as it could potentially overcome relapse to anticancer treatment. For this reason, ProTide 30a was included in further preclinical investigations.

### 3.2 2-Modified cordycepin analogues and their prodrugs

### 3.2.1 Background

The synthesis and biological evaluation of prodrugs of 3 ' dA has never been thoroughly investigated, and there is no literature report on the application of monophosphate prodrug approaches to any 3 '-deoxynucleoside. Considering the intriguing results of the application of the ProTide approach to $3^{\prime} \mathrm{dA}$, investigations on other $3^{\prime} \mathrm{dA}$ derivatives were carried out, with the aim of improving the activity of this family of ProTides. In this sense, the work by Vodnala et al. was of interest. This group recently published the synthesis and antitrypanosomal evaluation of different 3 ' dA analogues. ${ }^{225}$ The anti-trypanosomal activity of 3'dA had already been described both in vitro and in vivo. ${ }^{178-180,269}$ The most promising modification was represented by the introduction of a fluorine atom in position 2 on the nucleobase and resulted in increased metabolic stability of this derivative. In fact, the resistance of 2-fluorocordycepin (2F3'dA, 33, Figure 3.43) to ADA-mediated catabolism was already reported in 1967. ${ }^{270}$


Figure 3.43: Structure of 2-fluorocordycepin (2-fluoro-3'-deoxyadenosine, 2F3'dA, 33).

Similarly, the presence of such modification on 2-fluoroadenosine (2FA, 34) ${ }^{271}$ and fludarabine ( $2 \mathrm{FAraA}, \mathbf{1 i}$ ), ${ }^{272}$ is responsible for increased metabolic stability to ADAdeamination (Figure 3.44).


2FA
34

$\underset{1 i}{2 F A r a A}$

Figure 3.44: Structure of 2-fluoroadenosine (2FA, 34) and fludarabine (2FAraA, 1i).

Base modifications in position C 2 and C 8 have been reported to cause ADA-inhibition or, as in the case of small C 2 substitution, simply lack in recognition by the enzyme. ${ }^{273}$ Introduction of a halogen substitution in C 2 was reported to reduce the partial negative charges on N 1 and N 3 and therefore the capability of behaving as H -bond acceptors and the overall pKa , crucial features for interaction with ADA. ${ }^{225}$
Despite the increased metabolic stability of 2 F 3 ' dA and the more favourable activity profile on trypanosome-infected models, the activity of this nucleoside analogue as anticancer agent was not extensively investigated. In 1969 the cytotoxic screening of 2 F 3 ' dA resulted in an $\mathrm{ED}_{50}$ (median effective dose) concentration of $2 \mu \mathrm{M}$ on the hepatocellular carcinoma Hep-G2 cell line, compared to $\mathrm{ED}_{50} 80 \mu \mathrm{M}$ concentration of $3^{\prime} \mathrm{dA} .{ }^{274}$ Considering the interest in these reported results, investigation of the anticancer activity of both 2 F3'dA and a family of ProTides was carried out on a broader selection of cell lines. In fact, although 2 F3'dA represents a 3 'dA derivative less prone to ADAmediated inactivation, this nucleoside analogue might still benefit from the application of a monosphosphate prodrug approach in order to bypass potential limiting steps such as the dependency on a kinase enzyme for the first phosphorylation step, ${ }^{275}$ or inefficient transport-mediated cell penetration.

### 3.2.2 Synthesis of 2-fluorocordycepin

One reported procedure for the synthesis of 2 F 3 ' dA requires a four-step route starting from 3'dA (21, Scheme 3.10), which is first protected on the amino and hydroxyl groups via benzoyl chloride (35), to allow introduction of a nitro group in position 2 on the adenine moiety (36) by means of tetrabutylammonium nitrate (TBAN) and trifluoroacetic anhydride (TFAA). The $\mathrm{NO}_{2}$ group is finally replaced by fluorine by treatment with tetrabutylammonium fluoride (TBAF) in $\mathrm{CH}_{3} \mathrm{CN}$ (37) and de-benzoylation in methanolic ammonia to give the desired 2 F 3 ' $\mathrm{dA}(\mathbf{3 3}) .{ }^{225}$


Scheme 3.10: Synthesis of 2F3'dA (33). ${ }^{225}$ Reagents and conditions: (a.) BzCl, pyr; (b.) TBAN, TFAA; (c.) TBAF, $\mathrm{CH}_{3} \mathrm{CN}$; (d.) $\mathrm{NH}_{3}, \mathrm{CH}_{3} \mathrm{OH}$. (Overall yield of 4 steps from 3 'dA $=7 \%$ )

In an alternative methodology, the desired 2 F 3 ' dA is synthesised via Vorbrüggen-type glycosylation between 2-fluoroadenine (2FAde, 38) and 5-O-benzoyl-1,2-di-O-acetyl-3-deoxy-D-ribose in the presence of $\mathrm{N}, \mathrm{O}$-bis(trimethylsilyl)acetamide (BSA) and trimethylsilyl trifluoromethanesulfonate (TMSOTf) (Scheme 3.11) with a final deprotective step in methanolic ammonia yielding the desired 2 F3' $\mathrm{dA}(\mathbf{3 3}) .{ }^{276}$


Scheme 3.11: Synthesis of 2F3'dA (33). ${ }^{276}$ Reagents and conditions: (a.) BSA, 5-O-benzoyl-1,2-di- $O$-acetyl-3-deoxy-D-ribose, TMSOTf; (b.) $\mathrm{NH}_{3}, \mathrm{CH}_{3} \mathrm{OH}$. (Overall yield 2 steps from 2FAde $=35 \%$ )

Considering both the numerous steps required and the low yield reported for the first strategy ( $7 \%, 4$ steps), and both high cost of intermediates and moderate yield ( $35 \%$, 2 steps) needed for the second methodology, a different procedure was applied. The synthetic procedure successfully used for the synthesis of 3 ' dA ( $93 \%$ yield, 2 steps) was adapted, using 2FA as a starting material. Commercially available 2FA (34) was reacted with 4 equivalents of $\alpha-\mathrm{AIBBr}$ in a mixture of $\mathrm{CH}_{3} \mathrm{CN}$ and water. TLC monitoring of the crude confirmed the formation of more lipophilic species similarly to the reaction of adenosine in the same conditions. These intermediates were not isolated as the crude could be directly subjected to the next step after extraction of the mixture with EtOAc, washing of the organics with $\mathrm{NaHCO}_{3}$ (saturated solution) and evaporation of the organic phase. However, similarly to the intermediates isolated by Robins et al. ${ }^{212}$ during the synthesis of $3^{\prime} \mathrm{dA}$, the putative structures of the intermediates of this reaction were drawn as products of step a in Scheme 3.12.


40
Scheme 3.12: Attempted synthesis of $2^{\prime}, 3^{\prime}$-dehydro-2FA (41a-e). Reagents and conditions: (a.) $\alpha$ - $\operatorname{AIBBr}$ (4 eq), $\mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O}$; (b.) see Table 3.21. X: see Table 3.21.


41a


41b


41c


41d


Product 2
(yield)
(yield)

41b (54\%)
1h 30min 41a (11\%)
41c (83\%)
41d (*)
41b (traces)

41b (31\%) 41e (17\%)

Table 3.21: Reaction conditions and relative products and yields corresponding to step b. in Scheme 3.12. (*) Conversion confirmed via TLC and ES/MS. ${ }^{\text {a }}$ All volumes ( $\mathrm{mL} / \mathrm{mmol}$ ) of Amberlite were added in 2-3 portions over the time. ${ }^{\text {b }}$ Reactions were performed at rt .

The crude from step a (40, Scheme 3.12) was treated with Amberlite (IRN78, $8 \mathrm{~mL} / \mathrm{mmol}$ ) in two portions in $\mathrm{CH}_{3} \mathrm{OH}$ over 16 hours (Table 3.21, entry 1), following the procedure adopted previously in this work for the synthesis of 3 ' dA . The starting material 40 was completely converted into a less lipophilic spot on TLC that after filtration of the resin and evaporation of the solution was confirmed to be 2-methoxy- $2^{\prime}, 3^{\prime}$ '-anhydroadenosine 41a, by analysis of ${ }^{1} \mathrm{H}$ NMR (Figure 3.45), ${ }^{13} \mathrm{C}$ NMR, ES/MS and lack of peaks in the ${ }^{19} \mathrm{~F}$ NMR spectrum.


Figure 3.45: ${ }^{1} \mathrm{H}$ NMR of $\mathbf{4 1} \mathrm{a}$ in $\mathrm{CD}_{3} \mathrm{OD}(500 \mathrm{MHz})$.

Nucleophilic substitution of the fluorine in position 2 on the nucleobase was caused by reaction with methanol in basic conditions. In an attempt to limit this side reaction, the crude 40 was dissolved in $\mathrm{CH}_{3} \mathrm{OH}$ and $4 \mathrm{~mL} / \mathrm{mmol}$ of Amberlite ( $2 \times \mathrm{OH}^{-}$) were added (Table 3.21 , entry 2 ).

The reaction was carefully monitored and after only 1 hour a less liphophilic spot appeared on TLC, with a different Rf when compared to 41a. The conversion was not complete, however only traces corresponding to undesired 41a were noticed. An additional portion of Amberlite ( $2 \mathrm{x} \mathrm{OH}^{-}, 1 \mathrm{~mL} / \mathrm{mmol}$ ) was added, and the reaction was stirred for an additional 30 minutes, until the starting material 40 was present only in traces and the spot corresponding to undesired 41a became more intense on TLC. The resin was immediately filtered and washed with $\mathrm{CH}_{3} \mathrm{OH}$ and isolation of products by flash column chromatography yielded the desired 2-fluoro intermediate 41b in 54\% yield, along with $11 \%$ of 41a and traces of starting material (40), as confirmed by NMR analysis of the products ( ${ }^{1} \mathrm{H}$ NMR and ${ }^{19} \mathrm{~F}$ NMR of desired 41b are reported in Figure 3.46).


Figure 3.46: (A) ${ }^{1} \mathrm{H}$ NMR $(500 \mathrm{MHz})$ and $(\mathbf{B}){ }^{19} \mathrm{~F}$ NMR ( 470 MHz ) spectra of compound 41b in DMSO-d6.

In an attempt to improve the yield of the reaction, identification of a solvent for the reaction that could avoid risk of the aromatic nucleophilic substitution was necessary. At the same time, the solvent should have allowed the resin to behave as an ion exchanger. In fact, ion-exchange with conventional resins is frequently unsuccessful if anhydrous solvents of low polarity are used, ${ }^{277}$ due to the inability of the resin matrix to swell sufficiently in the solvent. If the resin is not swollen or is only partially swollen, the ionexchange rate-process or the penetration of the non-electrolyte or electrolyte into the resin will be limited. ${ }^{278-280}$

Our first attempt in replacing $\mathrm{CH}_{3} \mathrm{OH}$ as a solvent involved the use of EtOH (Table 3.21, entry 3), and $6 \mathrm{~mL} / \mathrm{mmol}$ of Amberlite ( $2 \mathrm{x} \mathrm{OH}^{-}$), reasoning that the increase in bulk of the alcohol could avoid or reduce the nucleophilic substitution.

However, 2-ethoxy-2', $3^{\prime}$-anhydroadenosine (41c) was recovered from resin filtration after 16 hours. The presence of an ethyl chain is clear from ${ }^{1} \mathrm{H}$ NMR spectrum (Figure 3.46) and the splitting pattern of the sugar moiety confirms the presence of the epoxide ring.


Figure 3.47: ${ }^{1} \mathrm{H}$ NMR of 2-O-ethyl-2', $3^{\prime}$-anhydroadenosine 41c in DMSO-d6.

Increasing the bulk of the alcohol with isopropanol ( $i \mathrm{PrOH}$ ), (Table 3.21, entry 4) as a solvent led to conversion into 2-isopropyloxy- $2^{\prime}, 3^{\prime}$-anhydroadenosine 41d as confirmed by TLC and ES/MS. Replacing the use of an alcohol with THF (Table 3.21, entry 5) caused a considerable reduction of the reaction rate, with only a small extent of conversion into desired 41b on TLC after 16 hours. The reaction attempted in water was unsuccessful due to the insolubility of the starting material in the solvent (Table 3.21, entry 6). Consequently, the use of a mixture of water in THF was attempted (Table 3.21, entry 7). Although the conversion into $\mathbf{4 1 b}$ appeared to be faster after 16 hours compared to the use of pure THF as a solvent, the reaction was pushed by stirring for 96 hours. Finally, complete conversion into 41b was obtained and a more lipophilic spot that was isolated and characterised as compound 41e, which was confirmed by ${ }^{1} \mathrm{H}$ NMR (Figure 3.48), ${ }^{19} \mathrm{~F}$ NMR and ES/MS analysis. Intermediates with the same 2,4,4-trimethyl-5-oxo-1,3dioxolane ring in position $5^{\prime}$ on the sugar moiety were already reported by Robins et al. ${ }^{212}$ during the synthesis of 3 ' dA (possible structures of such intermediates, which were not isolated, were reported in Scheme 3.12), however 41e can be considered as a partial product, because the formation of the epoxide ring on the sugar has happened.


Figure 3.48: ${ }^{1} \mathrm{HNMR}(500 \mathrm{MHz})$ spectra of 41e in DMSO-d6.

Therefore, these latter reaction conditions are too mild to complete the reaction with the total cleavage of this 5 '-ring, although favourably not leading to nucleophilic substitution of the fluorine on the adenine ring. However, considering the results of these attempts, the best yields achieved are represented by entry 2 , with $54 \%$ yield of desired product and $11 \%$ yield of side product 41a in 1 hour 30 minutes of reaction time.

The final step for the synthesis of 2 F 3 ' dA was performed on $\mathbf{4 1 b}$, which was subject to reductive opening of the epoxide ring via treatment with 4 equivalents of $\mathrm{LiEt}_{3} \mathrm{BH}(1 \mathrm{M}$ solution in THF) portioned in 2 additions: the first addition of 3 equivalents of the reagent was performed at $0^{\circ} \mathrm{C}$ and after overnight stirring at room temperature the mixture was brought to $0{ }^{\circ} \mathrm{C}$ a second time to allow the addition of the final equivalent of reagent (Scheme 3.13).


Scheme 3.13 Final step to the synthesis of 2 F 3 ' dA (33). Reagents and conditions. (a.) $\mathrm{LiEt}_{3} \mathrm{BH}$ (1M sol. in THF. 4 eq ), DMSO ( $4 \mathrm{~mL} / \mathrm{mmoL}$ ), THF ( $10 \mathrm{~mL} / \mathrm{mmoL}$ ), $0^{\circ} \mathrm{C}$ to $\mathrm{rt}, 16 \mathrm{~h}(82 \%)$.

Stirring of the mixture for an additional 5 hours at room temperature led to complete conversion of the starting material into desired 2 F 3 ' dA (33) (as confirmed by ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and ${ }^{19} \mathrm{~F}$-NMR analysis (Figure 3.49), which was isolated in $82 \%$ yield, reaching a 3 -step yield of $45 \%$.


Figure 3.49: (A) ${ }^{1} \mathrm{H}$ NMR ( 500 MHz ) and (B) ${ }^{19} \mathrm{~F}$ NMR ( 470 MHz ) spectra of 2 F 3 ' $\mathrm{dA}(\mathbf{3 3})$ in DMSO-d6.

Not completely satisfied with the yield of synthesis of 2 F3'dA, a different synthetic strategy was attempted. Hartwig et al. described the fluorination of multi-substituted pyridines and diazines at the position $\alpha$ to nitrogen accomplished by the use of silver fluoride $\left(\mathrm{AgF}_{2}\right)$ in $\mathrm{CH}_{3} \mathrm{CN}$, at rt for 3 hours. ${ }^{281}$ This mild methodology was applied to a variety of pyrimidine rings. Moreover, in a more recent publication by the same group, the introduction of a fluorine group $\alpha$ to nitrogen containing heterocycles was described as valuable for late-stage functionalisation of such substrates with oxygen, nitrogen and carbon-bearing nucleophiles, confirming the easy nucleophilic substitution on the 2fluoroadenine ring noticed in this work. ${ }^{282}$ The fluorination methodology required protection of both free hydroxyl and amino groups of the substrate.

When the reaction on $2^{\prime}, 5^{\prime}-O$-bistrityl-3'-deoxyadenosine (42) was carried out, the result was unsuccessful, with no trace of desired fluorinated product (43) detected (Scheme 3.14). The starting material (42) was recovered.


Scheme 3.14: Attempted fluorination of 43. Reagents and conditions: $\mathrm{AgF}_{2}(3 \mathrm{eq}), \mathrm{CH}_{3} \mathrm{CN}, \mathrm{rt}, 3 \mathrm{~h}$.

The same reaction procedure was performed on fully protected $N, N-5^{\prime}-O$-triacetyl- $2^{\prime}, 3^{\prime}$ isopropylidene adenosine (44) (Scheme 3.15), in an attempt to avoid any potential side reactions caused by the free amino group in substrate 42.


Scheme 3.15: Attempted fluorination of 44. Reagents and conditions: $\mathrm{AgF}_{2}(3 \mathrm{eq}), \mathrm{CH}_{3} \mathrm{CN}, \mathrm{rt}, 3 \mathrm{~h}$.

TLC analysis of the mixture showed the presence of a less lipophilic spot that was then confirmed by mass to be a side product lacking one acetyl group.

Therefore, the reaction on a simpler substrate such as fully protected adenine via di-tertbutyl dicarbonate (46) was attempted (Scheme 3.16).


Scheme 3.16: Attempted fluorination of 46. Reagents and conditions: $\mathrm{AgF}_{2}$ ( 3 eq ) $\mathrm{CH}_{3} \mathrm{CN}, \mathrm{rt}, 3 \mathrm{~h}$.

The reaction mixture visibly changed colour to yellow suggesting conversion of $\mathrm{AgF}_{2}$, although analysis of the crude via TLC revealed complete degradation into more hydrophilic products that could not be identified, with no trace of fluorinated product 47.

Despite not being able to improve the reaction yield for the synthesis of 2 F3'dA over $45 \%$, this still represented a significant improvement over the yields that were previously reported through different methodologies $\left(7 \%{ }^{225}\right.$ and $\left.35 \%{ }^{276}\right)$.

Due to the amount of 2-O-methyl-2', $3^{\prime}$-anhydroadenosine 41a synthesised in the attempts for the synthesis of 41b, conversion of this side-product into an alternative 3'deoxyadenosine analogue by treatment with $\mathrm{LiEt}_{3} \mathrm{BH}$ (4 equivalents in two portions) was carried out in a mixture of DMSO and THF.


Scheme 3.17: Final step of the synthesis of $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3$ 'dA (48). Reagents and conditions: $\mathrm{LiEt}_{3} \mathrm{BH}$ ( 1 M sol. in THF, 4 eq$)$, $\mathrm{DMSO}(4 \mathrm{~mL} / \mathrm{mmol})$, THF ( $10 \mathrm{~mL} / \mathrm{mmol}$ ), $0^{\circ} \mathrm{C}$ to $\mathrm{rt}, 16 \mathrm{~h}(85 \%)$.

This reaction led to the synthesis of 3'-deoxy-2-O-methyladenosine $\left(2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}, 48\right)$ in 85\% yield (Scheme 3.17), as confirmed in Figure 3.50.


Figure 3.50: ${ }^{1} \mathrm{H}$ NMR ( 500 MHz ) spectrum of $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}(\mathbf{4 8})$ in DMSO-d6.

The biological activity of this nucleoside analogue has never been reported, therefore the synthesis of a small family of ProTides of $\mathbf{4 8}$ was accomplished in order to investigate the potential anticancer activity of both the nucleoside and its ProTides.

### 3.2.3 Synthesis of 2-modified cordycepin ProTides

The synthesis of ProTides of both 2 F 3 ' dA and $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}$ was accomplished via treatment of the nucleoside analogue with 5 equivalents of N -methylimidazole (NMI) and

3 equivalents of the appropriate phosphorochloridates in anhydrous THF over 16 hours, at rt (Scheme 3.18).

Similarly to the synthesis of 3 ' dA ProTides, the NMI method applied to 2 F 3 ' dA and $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3$ 'dA also yielded 2'-phosphoramidate and 2'-5'-bis-O-phosphoramidate side products. However, these compound were not isolated, and the desired 5'phosphoramidates were obtained in yields ranging from $7 \%$ to $28 \%$ and 9 to $60 \%$ respectively for the 2 F 3 ' dA and $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}$ families of ProTides.


Scheme 3.18: Synthesis of ProTides of $2 \mathrm{~F} 3^{\prime} \mathrm{dA}(\mathbf{4 9 a - d})$ and $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}(\mathbf{5 0 a - d})$. Reagents and conditions: NMI ( 5 eq), appropriate phosphorochloridate (3 eq), THF, rt, 16 h .

| Cpnd | Nucleoside | Ar | R | $\mathbf{R}^{1}$ | AA | yield |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 49a | 2F3'dA (X = F) | Nap | Bn | Me | L-Ala | 15 \% |
| 49b | " | Ph | Bn | Me | L-Ala | 7\% |
| 49c | " | Nap | $n$-Pen | $i \mathrm{Bu}$ | L-Leu | 16 \% |
| 49d | " | Ph | $n$-Hex | Me | L-Ala | 28 \% |
| 50a | $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}\left(\mathrm{X}=\mathrm{OCH}_{3}\right)$ | Nap | Bn | Me | L-Ala | $9 \%$ |
| 50b | " | Ph | Bn | Me | L-Ala | 60\% |
| 50c | " | Nap | $n$-Pen | $i \mathrm{Bu}$ | L-Leu | 15\% |
| 50d | " | Ph | $n$-Hex | Me | L-Ala | 18\% |

[^1]
### 3.2.4 Cytotoxic in vitro screening

### 3.2.4.1 In vitro evaluation of 2F3'dA ProTides

The cytotoxic activity of $2 \mathrm{~F} 3^{\prime} \mathrm{dA}$ and $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}$ and relative prodrugs was tested on a panel of blood tumours and solid cancers (Table 3.23), by WuXi AppTech.

|  | Cell line | Malignancy | Cell line | Malignancy |
| :---: | :---: | :---: | :---: | :---: |
|  | $\begin{gathered} \text { MOLT-4 } \\ \text { CCRF- } \\ \text { RL } \\ \text { K562 } \\ \text { KG-1 } \\ \text { THP-1 } \\ \text { NCI-H929 } \end{gathered}$ | acute lymphoblastic acute lymphoblastic non-Hodgkin's lymphoma chronic myelogenous acute myelogenous leukaemia acute monocytic leukaemia plasmacytoma | $\begin{gathered} \text { HEL92.1.7 } \\ \text { HL-60 } \\ \text { Z-138 } \\ \text { Jurkat } \\ \text { HS445 } \\ \text { RPMI-8226 } \\ \text { MV4-11 } \end{gathered}$ | erythroleukaemia promyelocytic leukaemia mantle cell lymphoma acute T cell leukaemia hodgkin lymphoma human multiple myeloma biphenotypic B myelomonocytic |
|  | $\begin{gathered} \hline \text { BxPC-3 } \\ \text { SW620 } \\ \text { Cal } 27 \\ \text { HepG2 } \end{gathered}$ | pancreas adenocarcinoma colon adenocarcinoma oral adenosquamous hepatocellular carcinoma | HT29 <br> MCF-7 <br> MIA PaCa- $2$ | colon adenocarcinoma breast adenocarcinoma Pancreas adenocarcinoma |

Table 3.23: Malignant cell lines selected for the cytotoxic screening of ProTides of 2F3'dA and $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}$ (performed by WuXi AppTech).

The selection of cell lines used for the cytotoxic evaluation of 2 F3'dA (reported only on HepG2 cells) was broadened. ${ }^{270}$ Secondly, fewer cells were selected to test $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}$ and derivatives, since there was no report in the literature of any potential anticancer activity of such compound.

| Cpnd | CCRF-CEM |  | MOLT-4 |  | K562 |  | HEL92.1.7 |  | KG-1 |  | MV4-11 |  | HL-60 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{IC}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% |
| 2F3'dA | 15.23 | 101 | 8.82 | 99 | 16.83 | 97 | 10.9 | 99 | 10.85 | 99 | 2.76 | 99.5 | 10.64 | 96 |
| 49a | 0.62 | 99 | 0.5 | 98 | 3.86 | 99 | 3.41 | 100 | 6.48 | 99 | 0.43 | 99.8 | 1.78 | 101 |
| 49b | 0.41 | 99 | 0.32 | 96 | 2.63 | 97 | 2.91 | 99 | 2.6 | 97 | 0.61 | 99.9 | 2.04 | 100 |
| 49c | 0.35 | 101 | 0.25 | 97 | 1.66 | 100 | 0.7 | 99 | 6.07 | 103 | 0.32 | 99.7 | 2.74 | 99 |
| 49d | 0.37 | 100 | 0.25 | 97 | 14.7 | 96 | 2.52 | 99 | 2.15 | 96 | 0.53 | 99.4 | 1.99 | 100 |
| PTX | 0.003 | 98 | 0.003 | 95 | 0.01 | 91 | 0.02 | 83 | 0.08 | 88 | 0.003 | 98.2 | 0.003 | 85 |

[^2]Treatment of blood cancer cells with 2 F3'dA shows good values of $\mathrm{IC}_{50}$, consisting in low micromolar concentrations of $3.49 \mu \mathrm{M}$ on the plasmacytoma cell line NCI-H929, and 8.82 $\mu \mathrm{M}$ on the leukaemic cell line MOLT-4.

| Cpnd | Thp-1 |  | Z-138 |  | RL |  | Jurkat |  | Hs-445 |  | RPMI-8226 |  | NCI-H929 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% |
| 2F3'dA | 140.15 | 65 | 12 | 100 | 29.18 | 100 | 59 | 96 | 25 | 102 | 34 | 97 | 3.49 | 102 |
| 49a | 17.72 | 100 | 1.07 | 99 | 3.95 | 102 | 2.83 | 96 | 11 | 100 | 7.98 | 100 | 1.47 | 101 |
| 49b | 17.49 | 99 | 1.41 | 99 | 1.25 | 101 | 3.84 | 93 | 31 | 106 | 9.83 | 95 | 1.12 | 102 |
| 49c | 17.2 | 104 | 2.84 | 103 | 2.23 | 105 | 8.02 | 103 | 14 | 102 | 10 | 101 | 0.27 | 99 |
| 49d | 35.37 | 99 | 1.59 | 98 | 2.64 | 99 | 7.87 | 94 | 72 | 87 | 38 | 98 | 1.25 | 100 |
| PTX | 0.03 | 72 | 0.003 | 99 | 0.003 | 83 | 0.003 | 82 | 0.004 | 84 | 0.003 | 96 | 0.003 | 82 |

Table 3.25: In vitro cytotoxicity screening of parent nucleoside 2F3'dA and ProTides 49a-d on selected cell lines. Cytotoxicity data reported as $\mu \mathrm{M} \mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentrations considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

The treatment of the same cells with 3 ' dA gave $\mathrm{IC}_{50}$ values of $151.92 \mu \mathrm{M}$ and $>198 \mu \mathrm{M}$ respectively on MOLT-4 and NCI-H929 cells.

| Cpnd | MIA PaCa-2 |  | BxPC-3-Luc |  | HT29 |  | SW620 |  | HepG2 |  | MCF-7 |  | Cal 27 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | IC ${ }_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% |
| 2F3'dA | 32.44 | 100 | 38 | 70 | 54.19 | 73 | 23 | 95 | 17 | 82 | 51.48 | 97 | 35 | 99 |
| 49a | 3.31 | 100 | 23 | 95 | 18.21 | 84 | 7.73 | 99 | 13 | 83 | 3.24 | 95 | 9.09 | 101 |
| 49b | 2.42 | 98 | 32 | 78 | 14.83 | 87 | 15 | 97 | 5.88 | 74 | 1.53 | 95 | 11 | 95 |
| 49c | 4.88 | 100 | 55 | 90 | 16.54 | 88 | 12 | 104 | 4.02 | 94 | 1.32 | 99 | 35 | 101 |
| 49d | 5.82 | 98 | 34 | 67 | 35.36 | 92 | 37 | 89 | 5.3 | 82 | 1.17 | 87 | 32 | 86 |
| PTX | 0.002 | 87 | 0.016 | 51 | 0.004 | 75 | 0.011 | 84 | 0.04 | 50 | 0.01 | 65 | 0.002 | 92 |

Table 3.26: In vitro cytotoxicity screening of parent nucleoside 2F3'dA and ProTides 49a-d on selected cell lines. Cytotoxicity data reported as $\mu \mathrm{M} \mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and MI\% (maximum inhibitory effect of the drug at the range of concentrations considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

This improvement in cytotoxic activity is similar on all the other cell lines, including those from solid tumours: although in some cases generating $\mathrm{IC}_{50}$ values as high as $54.19 \mu \mathrm{M}$ (on the colon adenocarcinoma cell line HT29). 2F3'dA had a significant cytotoxic effect on all cell lines, leading to inhibition of cell viability higher than $73 \%$.


Figure 3.51: Activity of 2F3'dA and ProTides 49a-d on blood cancer cell lines. ${ }^{\mathrm{a}} \mathrm{IC}_{50}=29.18 \mu \mathrm{M} ;{ }^{\mathrm{b}} \mathrm{IC}_{50}=$ $140.15 \mu \mathrm{M}$; ${ }^{\mathrm{c}} \mathrm{IC}_{50}=35.37 \mu \mathrm{M}$.

ProTides of 2 F3'dA are more active than the parent nucleoside, the pentyloxy-L-leucine napthyl phosphate derivative 49c being consistently the most potent derivative, with $\mathrm{IC}_{50}$ values in the submicromolar range $(0.25-0.7 \mu \mathrm{M})$ on the blood cancer cell lines MOLT-4, NCI-H929, CCRFCEM and HEL92.1.7. The L-alanine phenyl benzyloxy derivative 49b was similarly potent on most cell lines, and this trend was generally maintained by all 2F3'dA prodrugs.


Figure 3.52: Activity of 2F3'dAdo and ProTides 49a-d on solid cancer cell lines.

Comparable results were gathered after treatment of solid tumour cell lines with 2F3'dA and relative prodrugs, with the parent nucleoside being moderately active. The activity of 2F3'dA ProTides was, in this case, overall very similar, with $\mathrm{IC}_{50}$ values rarely higher than $15 \mu \mathrm{M}$ and ranging between $1.17 \mu \mathrm{M}$ ( $\mathrm{IC}_{50}$ value of compound 49d on MCF-7 cell line) and $35.36 \mu \mathrm{M}$ ( $\mathrm{IC}_{50}$ value of compound 49 d on HT29 cell line).

### 3.2.4.2 In vitro screening of $2\left(\mathrm{CH}_{3} \mathrm{O}\right) \mathbf{3}^{\prime}$ 'dA ProTides

The introduction of a 2-methoxy group in place of fluorine on the adenine ring generated an inactive nucleoside analogue, as shown in Table 3.27, Table 3.28 and Figure 3.53.

| Cpnd | CCRFCEM |  | KG-1 |  | K562 |  | MOLT-4 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | IC $\mathbf{5 0}$ | MI\% | IC ${ }_{\mathbf{5} 0}$ | MI\% | IC50 | MI\% | $\mathbf{I C}_{50}$ | MI\% |
| 2( $\mathrm{CH}_{3} \mathrm{O}$ ) 3' ${ }^{\text {d }}$ | >198 | -4 | >198 | 1 | >198 | 40 | >198 | 1 |
| 50a | 22.2 | 100 | 106.4 | 92 | 42.1 | 104 | 29 | 101 |
| 50b | 12.9 | 102 | 86.6 | 99 | 29.6 | 100 | 17.9 | 104 |
| 50c | 10.6 | 100 | 20.6 | 102 | 13.7 | 100 | 6.5 | 101 |
| 50d | 75.5 | 93 | 82.9 | 96 | 84.4 | 78 | 58.1 | 97 |
| PTX | 0.003 | 98 | 0.08 | 88 | 0.01 | 91 | 0.003 | 95 |

Table 3.27 In vitro cytotoxicity screening of parent nucleoside $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}$ and ProTides 50a-d. Cytotoxicity data reported as $\mu \mathrm{M} \mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and MI\% (maximum inhibitory effect of the drug at the range of concentrations considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

2( $\left.\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime}$ 'dA did not significantly affect the growth of the considered cell lines, up to the highest concentration assayed ( $198 \mu \mathrm{M}$ ). Conversely, ProTides of $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}$ reached approximately $100 \%$ of inhibitory effect on almost all treated cell lines, with $\mathrm{IC}_{50}$ values ranging within the high micro-molar range.

| Cpnd | RL |  | NCI-H929 |  | HT29 |  | MCF7 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathbf{I C}_{\mathbf{5 0}}$ | $\mathbf{M I}_{\%}$ | $\mathbf{I C}_{\mathbf{5 0}}$ | MI $_{\%}$ | $\mathbf{I C}_{\mathbf{5 0}}$ | $\mathbf{M I}_{\%}$ | $\mathbf{I C}_{\mathbf{5 0}}$ | MI $_{\mathbf{\%}}$ |
| $\mathbf{2 ( \mathbf { C H } _ { \mathbf { 3 } } \mathbf { O } ) \mathbf { 3 } \mathbf { \prime } \mathbf { d A }}$ | $>198$ | 0 | $>198$ | 51 | $>198$ | 19 | $>198$ | 8 |
| $\mathbf{5 0 a}$ | 35.8 | 97 | 22 | 104 | 94.1 | 81 | 44.1 | 90 |
| $\mathbf{5 0 b}$ | 24.5 | 104 | 12.4 | 102 | 75.8 | 98 | 38.1 | 99 |
| $\mathbf{5 0 c}$ | 11.6 | 100 | 10.5 | 99 | 24.4 | 100 | 13.9 | 100 |
| $\mathbf{5 0 d}$ | 60.2 | 88 | 41.6 | 99 | 98.3 | 69 | 96.6 | 73 |
| PTX | 0.003 | 83 | 0.003 | 82 | 0.004 | 75 | 0.01 | 65 |

Table 3.28: In vitro cytotoxicity screening of parent nucleoside $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3$ 'dA and ProTides 50a-d. Cytotoxicity data reported as $\mu \mathrm{M} \mathrm{IC} 50$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and MI\% (maximum inhibitory effect of the drug at the range of concentrations considered). PTX: paclitaxel (control). Assays performed by WuXi app. Tech.


Figure 3.53: Cytotoxic activity of $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3$ ' dA and ProTides 50a-d. ${ }^{*} \mathrm{IC}_{50}>198 \mu \mathrm{M}$.

The L-leucine derivative 50c was the most active derivative in this series, with an $\mathrm{IC}_{50}$ of $6.51 \mu \mathrm{M}$ on MOLT-4 cell line and similar activity profile on colorectal and breast solid tumour cell lines (HT29 and MCF7).

In conclusion, the in vitro evaluation of 2-modified 3'dA analogues and relative ProTides confirmed the superior cytotoxic activity of 2 F 3 ' dA over 3 ' dA . Comparison between the cytotoxicity data relative to $3^{\prime} \mathrm{dA}, 2 \mathrm{~F} 3^{\prime} \mathrm{dA}$ and $2\left(\mathrm{OCH}_{3}\right) 3^{\prime} \mathrm{dA}$ is given in Figure 3.54. On this selection of cell lines, $3^{\prime} \mathrm{dA}$ was more active than $2\left(\mathrm{OCH}_{3}\right) 3^{\prime} \mathrm{dA}$, although $2 \mathrm{~F} 3^{\prime} \mathrm{dA}$ emerges as consistently the most active nucleoside.


Figure 3.54 Comparison between cytotoxicity ( $\mu \mathrm{M} \mathrm{IC}_{50}$ ) values of 3 ' $\mathrm{dA}, 2 \mathrm{~F} 3^{\prime} \mathrm{dA}$ and $2\left(\mathrm{OCH}_{3}\right) 3^{\prime} \mathrm{dA} .{ }^{*} \mathrm{IC}_{50}>$ $198 \mu \mathrm{M}$.

Figure 3.55 shows a comparison between the most active ProTides of the three families of 3'dA derivatives, namely the phenyl L-alanine benzyl ester ProTides of 3'dA (32a), 2F3'dA (51a) and 2( $\mathrm{OCH}_{3}$ ) 3'dA (50a), naphthyl L-alanine benzyl ester ProTides of 3'dA (30b), 2F3'dA (49a) and $2\left(\mathrm{OCH}_{3}\right) 3^{\prime} \mathrm{dA}(50 b)$ and naphthyl L-leucine $n$-pentyl ester ProTides of 3 'dA (30d), 2F3'dA (49c) and $2\left(\mathrm{OCH}_{3}\right) 3^{\prime} \mathrm{dA}(\mathbf{5 0 c})$. Despite the significant improvement in activity of 2 F3'dA in comparison to $3^{\prime} \mathrm{dA}$, ProTides of $2 \mathrm{~F} 3^{\prime} \mathrm{dA}$ are endowed with similar cytotoxicity as the 3 ' dA counterparts. The same moieties applied to $2\left(\mathrm{OCH}_{3}\right) 3^{\prime} \mathrm{dA}$, on the other hand, generate less active compounds, apart from the naphthyl

L-leucine $n$-pentyl ester moiety, which generate three compounds with very similar cytotoxicity (30d, 49c, 50c).


Figure 3.55: Comparison between cytotoxicity ( $\mu \mathrm{M} \mathrm{IC}_{50}$ ) values of phenyl L-alanine benzyl ester ProTides of 3'dA (30a), 2F3'dA (49a) and $2\left(\mathrm{OCH}_{3}\right) 3^{\prime} \mathrm{dA}(\mathbf{5 0 a})$, naphthyl L-alanine benzyl ester ProTides of $3^{\prime} \mathrm{dA}(\mathbf{3 0 b})$, 2F3'dA (49b) and $2\left(\mathrm{OCH}_{3}\right) 3^{\prime} \mathrm{dA}(\mathbf{5 0 b})$ and naphthyl L-leucine n-pentyl ester ProTides of 3'dA (30d), 2 F 3 ' $\mathrm{dA}(\mathbf{4 9 c})$ and $2\left(\mathrm{OCH}_{3}\right) 3^{\prime} \mathrm{dA}(\mathbf{5 0 c}) .{ }^{\mathrm{a}} \mathrm{IC}_{50}=106.41 \mu \mathrm{M},{ }^{\mathrm{b}} \mathrm{IC}_{50}=94.09 \mu \mathrm{M},{ }^{\mathrm{c}} \mathrm{IC}_{50}=86.58 \mu \mathrm{M},{ }^{\mathrm{d}} \mathrm{IC}_{50}=$ $75.81 \mu \mathrm{M}$.

The cytotoxic evaluation of new 2-modified 3'dA and related ProTides led to the identification of promising new derivatives within the 2 F3'dA family of prodrugs, which were further investigated.

### 3.2.4.3 Leukaemic stem cells selectivity

The intriguing anticancer activities of both 2 F 3 ' dA and relative ProTides prompted us to investigate the potential selective targeting of the LSC population present within the KG1a cell line. Initially, the $\mathrm{LD}_{50}$ values of these compounds were evaluated on KG1a cell line (Table 3.29) by Prof. C. Pepper (School of Medicine, Cardiff University).

| Common structure | Cpnd | Ar | $\mathbf{R}$ | $\mathbf{R}_{\mathbf{1}}$ | AA | $\mathbf{L D}_{\mathbf{5 0}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |

[^3]Consistently with the data previously reported, 2 F3'dA results indicated a good anticancer agent, with $\mathrm{LD}_{50}$ value in the micromolar range. ProTides 49a-d were more active than the
parent nucleoside, although by only 1.5 to 3.3 -fold. The most active compound on the KG1a cell line was the napthyl L-leucine pentyl derivative 49c (Table 3.29).

The KG1a cell line was then treated with a range of concentrations of $2 \mathrm{~F} 3^{\prime} \mathrm{dA}$ and ProTides 49a-d, and the effect on the LSC population within this cell line was evaluated via FACS analysis. Figure 3.56 reports a comparison between the $\mathrm{LD}_{50}$ values of 2 F 3 'dA and ProTides 49a-d, also in relation to the $\mathrm{LD}_{50}$ values of 3 ' dA and $3^{\prime} \mathrm{dA}$ ProTide 30a on the same cell line.


Figure 3.56: $\mathrm{LD}_{50}(\mathrm{M})$ values of 3 ' dA and 2 F 3 ' dA and ProTides 30a and 49a-d. Assays performed by Prof. C. Pepper, Cardiff University.

As noted from the previous in vitro screening, 2F3'dA appears to be a much more cytotoxic agent than 3 'dA. However, the $L_{50}$ values of 2 F3'dA ProTides were lower than the parent nucleoside, but not significantly different from ProTide 30a.


Figure 3.57 Effect on LCS population after treatment with 2 F3'dA and ProTides at a range of concentrations. Assays performed by Prof. C. Pepper, Cardiff University.

When the LSC targeting was analysed after treatment with 2 F3'dA derivatives at a broad dose range, both compound 49a and the parent nucleoside seemed to leave the percentage of this population unaffected (Figure 3.57, Figure 3.58). On the other hand, ProTides 49b,

49c and 49d led to selective targeting of the stem cell population at concentrations higher than $10^{-4} \mathrm{M}$.


Figure 3.58: Effect on LCS population after treatment with 2F3'dA and ProTides at a range of concentrations. Assays performed by Prof. C. Pepper, Cardiff University.

### 3.2.5 Adenosine deaminase (ADA) stability of 2-modified 3'dA analogues

### 3.2.5.1 ADA UV assay

The presence of a fluorine substituent in position 2 on the 3 ' dA structure was already reported to induce resistance to ADA-mediated catabolism. ${ }^{225,270}$


Figure 3.59 Stability of $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}(\mathbf{A})$ and $2 \mathrm{~F} 3^{\prime} \mathrm{dA}(\mathbf{B})$ to ADA mediated deamination. Spectra recorded every 0.1 minute for 10 minutes. No significant transformation into the respective 2 -modified inosine derivative was noticed for either substrates.

Therefore, the stability of $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3$ ' dA to ADA was assayed, due to the presence of a 2 methoxy functionality. The introduction of a methoxy group in position 2 on the adenine ring would be expected to lead to an opposite effect compared with the 2 -fluoro
substituent, owing to electron donation of the methoxy group. In fact, the introduction of a halogen in C2 was reported to reduce the partial negative charges on N1 and N3 and therefore the capability of behaving as H -bond acceptors and the overall pKa , crucial features for interaction with ADA. ${ }^{225}$ On the other hand, the $\mathrm{CH}_{3} \mathrm{O}$ - group should have an opposite electron donating effect, hence enhancing the partial negative charges on N 1 and N3.


Figure 3.60: Stability of 3 'dA to ADA mediated deamination. Spectra recorded every 0.1 minute for 10 minutes. 3'dA (UVmax $=259 \mathrm{~nm}$ ) was converted into 3 'dIno $(U V \max =249 \mathrm{~nm})$ in the presence of ADA over 10 minutes (A).

However, ADA UV assay performed by treating both 2 F 3 'dA and $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}$ nucleosides with bovine adenosine deaminase in phosphate buffer ( pH 7.6 ) showed that both compounds are resistant to ADA-mediated catabolism over 10 minutes (Figure 3.59), while 3 'dA was completely converted into the inosine counterpart within 5 minutes (Figure 3.60), as noticeable from the shift in $\lambda_{\max }$ from 259 nm (corresponding to $3^{\prime} \mathrm{dA}$ ) to 249 nm (corresponding to $3^{\prime}$ dIno). ${ }^{225}$
Since the electronic effect would not explain the stability of $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}$ resistace to ADA, other factors may contribute to stability of compounds to deamination such as the steric hindrance, as alreay noted for 3 ' dA prodrugs.

### 3.2.5.2 Docking of 2-modified 3'dA analogues within the ADA active site

In order to understand whether steric factors could affect the recognition of 2-modified 3 ' dA analogues by ADA, hence the deamination of the considered compounds, docking studies were performed within the active site of ADA (crystal structure PDB 3IAR).


Figure 3.61 Overlap of 2F3'dA (blue), $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}$ (yellow) and 3 ' dA (white) within the ADA active site.

Although overlapping almost entirely with $3^{\prime} \mathrm{dA}$, both $2 \mathrm{~F} 3^{\prime} \mathrm{dA}$ and $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}$ seemed to interact differently with the active site. This was suggested by the lack of interaction with the Asp295 residue. Moreover, the space around position 2 on the adenine ring is close to the substrate, and seems not to allow any bulky group in this position. In contrast there is a larger area around the amino group and position 8 . This may suggest that the lack of processing of $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3$ ' dA by ADA derives from a clash within the active site residues, leading to incorrect positioning of the molecule.

### 3.2.6 Mechanistic investigations

### 3.2.6.1 Enzymatic NMR experiments

A NMR enzymatic assay involving CPY enzyme was carried out on the phenyl L-alanine benzyl ester ProTide of 2 F 3 ' $\mathrm{dA}(\mathbf{4 9 b})$, and on the $2\left(\mathrm{CH}_{3} \mathrm{O}\right) \mathrm{O} 3$ ' dA ProTide (50b), which is endowed with significantly lower activity, as evident from Figure 3.55.

The aim was to compare the metabolism by CPY of these two prodrugs and predict whether the lower cytotoxic activity of $\mathbf{5 0 b}$ could derive from a less efficient processing by this enzyme.


Figure 3.62 Putative metabolic processing by CPY/Hint of ProTides 49b and 50b to final release of $5^{\prime}$ 'monophosphate nucleosides $\mathbf{4 9 b}{ }^{\mathbf{D}}$ and $\mathbf{5 0 b}$.

Conversion of ProTide 49b into L-alaninyl phosphate monoester intermediate 49b ${ }^{\text {C }}$ was rapid and complete after 48 minutes, as described in Figure 3.63.


Figure 3.63: (A) Stacked ${ }^{31}$ P NMR spectra (acetone- $d 6$, 202 MHz ) monitoring the conversion of ProTide 49b into metabolite $\mathbf{4 9 b}{ }^{\text {C }}$. (B) Graph describing the conversion of $\mathbf{4 9 b}$ into metabolite $\mathbf{4 9 b}{ }^{\text {C }}$ over time, as percentage of ${ }^{31} \mathrm{P}$ NMR integral of each species.

Similarly, compound 50b was processed very quickly by CPY, with complete disappearance of the initial peaks within 7 minutes from the enzyme addition. The ${ }^{31} \mathrm{P}$ NMR peak of the final compound resulting from CPY metabolism (50b ${ }^{\mathbf{C}}$ ), appears as a broad peak. However the positioning of the peak in the appropriate range of ppm (6-8 ppm ), and ES/MS analysis of the final solution (calculated $\mathrm{m} / \mathrm{z}=432.12$ [M], found $431.09\left[\mathrm{M}-\mathrm{H}^{+}\right]$) confirmed the presence of the expected metabolite $\mathbf{5 0 b}{ }^{\mathrm{C}}$.


Figure 3.64 Stacked ${ }^{31} \mathrm{P}$ NMR spectra (acetone- $d 6,202 \mathrm{MHz}$ ) monitoring the conversion of ProTide 50b into metabolite $\mathbf{5 0 b}^{\text {C }}$

Therefore, the CPY assay suggests that the cleavage of the ester moiety of compound 50b with release of metabolite $\mathbf{5 0 b}{ }^{\mathbf{C}}$ after intracellular rearrangements might not be a limiting step affecting the low activity of compound $\mathbf{5 0 b}$. This assay also suggests that the promising anticancer activity of compound 49b might be a consequence of quick intracellular release of intermediate $\mathbf{4 9 b}$. Metabolite $\mathbf{4 9 b}^{\mathbf{C}}$, along with the L-alaninyl intermediate $\mathbf{5 0 b}{ }^{\mathbf{C}}$, were docked into the active site of Hint, in order to predict the likelihood of processing to respectively 2 F 3 ' $\mathrm{dAMP}\left(\mathbf{4 9 b}^{\mathbf{D}}\right)$ and $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime}$ dAMP ( $\left.\mathbf{5 0 b} \mathbf{b}^{\mathbf{D}}\right)$ by this enzyme.

### 3.2.6.2 Docking studies

Prediction of the intracellular metabolism of compounds $\mathbf{4 9 b}$ and $\mathbf{5 0 b}$, regarding the last step of ProTide activation, catalysed by Hint enzyme, was carried out through docking simulations of intermediates $\mathbf{4 9 b}{ }^{\text {C }}$ and $\mathbf{5 0 b}{ }^{\mathrm{C}}$ within the active pocket of the enzyme.


Figure 3.65 Docking of metabolic intermediates $\mathbf{5 0 b} \mathbf{b}^{\mathbf{C}}(\mathbf{A})$ and $\mathbf{4 9 b}{ }^{\mathbf{C}}(\mathbf{B})$ in the active site of the Hint enzyme.

Both intermediates appear to interact very similarly in the active site, with both phosphoramidate moieties closely positioned to the catalytic residues and the nucleobase interacting with additional residues and allowing a correct positioning of both structures. Consequently, Hint enzyme does not appear to discriminate between the two compounds; hence it should not be responsible for the differences in cytotoxic profiles of the two compounds.

### 3.2.7 Conclusions

The already optimised strategy for the synthesis of 3 ' dA was modified and applied to the preparation of 2 F 3 ' dA , yielding as side product $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}$. 2-Modified analogues of 3'dA were already reported as more stable to ADA-catabolism compared to 3'dA, although were not thoroughly investigated as anticancer agents. In this work both the already reported 2 F 3 ' dA and the newly synthesised $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}$ along with some ProTide examples of these two families were tested, in search of modified 3'dA ProTides endowed with potentially improved anticancer activity. The introduction of fluorine on the adenine ring generated a nucleoside analogue with significantly improved cytotoxicity compared to $3^{\prime} \mathrm{dA}$. On the other hand the presence of a $2-\mathrm{CH}_{3} \mathrm{O}$ group on the adenine moiety was not beneficial in terms of anticancer activity. This was potentially due to a lack of recognition from adenosine kinase, as suggested by the significant improvement in activity when the ProTide moiety was applied to $2\left(\mathrm{CH}_{3} \mathrm{O}\right) 3$ ' dA . ProTides of 2 F 3 'dA were potent derivatives, with $\mathrm{IC}_{50}$ values ranging within sub-micromolar and low-micromolar range. In particular, compound 49b emerged as both one of the most active 2 F3' dA ProTides on a vast selection of malignant cell lines and as a selective cytotoxic agent against the LSC population of the KG1a cell line. On the basis of these promising in vitro
results, a new set of preclinical investigations were performed, aimed at comparing the activities of the lead candidates $\mathbf{3 0 a}$ and $\mathbf{4 9 b}$, respectively within the $3^{\prime} \mathrm{dA}$ and $2 \mathrm{~F} 3^{\prime} \mathrm{dA}$ families of ProTides.

### 3.3 Further preclinical investigations on 3'dA and 2F3'dA prodrugs

The cytotoxic in vitro screening of 3'dA ProTides showed the anticancer potential of this family of prodrugs. In particular, the L-alanine benzyl ester phenyl ProTide 30a, emerged as one of the most active derivatives in the in vitro screening on a broad selection of cell lines. In addition, this derivative was capable of targeting the LSC population within the KGla cell line, which could potentially offer an advantage in vivo. Based on the intriguing activity of 3'dA ProTides, 2F3'dA derivatives were synthesised and proved to be endowed with potent cytotoxic activity.


Figure 3.66: Structure of compounds 30a and 49b.

Therefore, the separation of the two diasteroisomer of the most promising derivatives 30a and 49b (Figure 3.66) was carried out, with the aim of separately evaluating their cytotoxicity on a selection of cell lines. Furthermore, in vivo anticancer studies on mice inoculated with RL non-Hodgkin's lymphoma cells were performed by WuXi AppTech, along with additional stability assays in human and rat serum.

### 3.3.1 Separation of isomers of ProTides 30a and 49b

### 3.3.1.1 Rationale behind diastereoisomer separation

As mentioned in the previous chapters, the two commonly used procedures for the synthesis of nucleoside ProTides consist of the NMI or $t \mathrm{BuMgCl}-$ mediated coupling reaction between the nucleoside and a phosphorochloridate. These reactions result in the generation of roughly a 1:1 mixture of two diasteroisomers ( $S \mathrm{p}$ and Rp ), due to the nonstereoselective formation of a new chiral center at the phosphorus atom.

Difference in the stereochemistry of drugs has often led to compounds with dissimilar activity, toxicology and metabolism, hence the preference for the use of a single isomer in the clinics. ${ }^{283,284}$ In the case of ProTides, the separation is not always straightforward, nor is the preparation of one single diasteroisomer, due to the lack of control of the
stereochemistry at the phosphorous centre during the coupling reaction. ${ }^{285}$ However, when the separation was successful, the different isomers were often found to be endowed with differences in the biological profile. ${ }^{286-288}$ Crucial examples are the marketed anti-HCV drug sofosbuvir ( $\mathbf{9 e}$, Figure 3.67), whose $S$ p isomer was 18 fold more active than the $R$ p, ${ }^{287}$ or the anti-HIV acyclic phosphonate analogue GS7340 (91, Figure 3.67), ${ }^{286}$ currently in phase III clinical trials as the $S$ p isomer, which was found to be 10 -fold more active than the $R \mathrm{p}$ isomer.


Figure 3.67: Structures of sofosbuvir (9e) and GS7340 (91).

A similar pattern was observed also in the family of cytostatic [5-(E-bromovinyl)]-2'deoxyuridine ProTides, whose separated diasteroisomers showed different activity profiles, with $S$ p isomers 10 times more active and $R$ p isomers slightly less active than the mixture. ${ }^{153}$ Moreover, phosphoro(no)amidate diasteroisomers were found to be processed by carboxypeptidase enzyme, ${ }^{151,152,289}$ or be transported through intestinal absorption cell models (Caco-2 and MDCK cells), at different rates. ${ }^{290,291}$

Separation of nucleoside phosphoroamidates is generally considered as a difficult task, however was reported as feasible on some substrates by means of reverse phase HPLC, ${ }^{286,290-293}$ molecularly imprinted stationary phase, ${ }^{294}$ capillary electrophoresis with chiral sectors ${ }^{295,296}$ and normal phase chiral HPLC. ${ }^{286,292,296-298}$ Nonetheless, the separation of diasteroisomers via reported methods is not always successful.

### 3.3.1.2 Attempt to separate diastereoisomers via silylation/desilylation methods

The attempts to separate the diasteroisomers of ProTides 30a and 49b via direct phase silica gel column chromatography were unsuccessful, due to the very close retention factor of both isomers on silica.

A recently developed method within the McGuigan group involved direct phase silica gel chromatography separation of sugar TBDMS protected ProTides of the anticancer agent

FUdR, due to different affinity for the stationary phase. Initially this method was applied to 3'dA ProTide 30a, in order to obtain the desired separated isomers 30a ( $R \mathrm{p}$ ) and 30a $(\mathrm{Sp})$ and be able to individually evaluate their potentially different cytotoxicity profile.


Scheme 3.19: Attempt to separate 30a diasteroisomers via silylation/desilylation method. Reagents and conditions: (a.) [1] TBDMSCl (3 eq), pyridine, rt, 16h, 50\%; [2] TBDMSCl (3 eq), DMAP ( 0.6 eq), imidazole ( 6 eq), DMF, rt, 16h. 100\%; (b.) THF: $\mathrm{H}_{2} \mathrm{O}:$ THF ( $1: 1: 4$ ), $5 \mathrm{~h}, 0^{\circ} \mathrm{C}, 98 \%$ (c.), phosphorochloridate (3 eq), NMI ( 5 eq ), THF, $16 \mathrm{~h}, \mathrm{rt}, 62 \%$. (d.) chromatographic separation. (e.) TFA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{rt}$.

This step was performed initially by treatment of the nucleoside analogue with tert-butyl dimethylsilyl chloride (TBDMSCl) in pyridine at room temperature for 16 hours, yielding compound 25 in 50\% yield (Scheme 3.19). ${ }^{208}$ Dissolving unprotected 3' dA in DMF and adding imidazole and DMAP to the mixture, followed by addition of the silylating agent, generated a significant improvement of this reaction to quantitative yield. ${ }^{299}$ Derivative 25 was subject to selective $5^{\prime}$-deprotection of the silyl group after treatment with a 1:1:4 mixture of trifluoroacetic acid (TFA) in water and THF, ${ }^{300}$ yielding $98 \%$ of desired product 51.

Coupling of the intermediate $\mathbf{5 1}$ with phenyl L-alanine benzyl ester phosphorochloridate by means of NMI in THF yielded $62 \%$ of ${ }^{\prime}$ '-TBDMS protected ProTide 52. However, efforts to separate by silica gel TLC using different binary and ternary eluent systems did not result in positive outcome.
A more hindered silyl protecting group was believed to allow a separation via the same method, therefore 30a was treated with tert-butyldiphenylsilyl chloride (TBDPCl), imidazole and DMAP in DMF, to yield compound $\mathbf{5 3}$ in $\mathbf{7 6 \%}$ yield.


Figure 3.68 Synthesis of $2^{\prime}$-TBDPS protected ProTide 53. Reagents and conditions: (a.) TBDPSCl (3 eq), DMAP ( 0.6 eq), imidazole ( 6 eq ), DMF, rt, 16h, $76 \%$.

Attempted methods to separate the two diasteroisomers of compound $\mathbf{5 3}$ on TLC yielded only modest differences in Rf between the two isomers. The outcome was not positive enough to promise good separation through flash chromatography.

The separation of $\mathbf{3 0 a}$ isomers was therefore attempted via reverse phase chromatography.

### 3.3.1.3 Separation of ProTides $30 a$ and $49 b$ diastereoisomers via reverse phase chromatography

Only a few milligrams of each separated isomer were required for preliminary in vitro testing, therefore preparative HPLC was initially used as a separating tool. The optimal elution method was firstly investigated on analytical HPLC, and at the beginning a gradient of $\mathrm{CH}_{3} \mathrm{CN}$ in water from a mixture of $40: 60$ to $70: 30$ in 30 minutes was used as eluent system (Figure 3.69, A), then a gradient of $\mathrm{CH}_{3} \mathrm{CN}$ in water 30:70 to 70:30 (Figure 3.69, B) and finally a gradient of $\mathrm{CH}_{3} \mathrm{CN}$ in water 10:90 to 90:10 (Figure 3.69, C).


Figure 3.69: Overlayed HPLC traces of 30a diastereomeric mixture in different solvent gradients. (Blue, A) HPLC conditions: $40: 60 \mathrm{CH}_{3} \mathrm{CN}: \mathrm{H}_{2} \mathrm{O}$ to $70: 30 \mathrm{CH}_{3} \mathrm{CN}: \mathrm{H}_{2} \mathrm{O}$ in 30 min . Flow $=1 \mathrm{~mL} \mathrm{~min}^{-1} ; \lambda=254 \mathrm{~nm}$; (green, B) HPLC conditions: $30: 70 \mathrm{CH}_{3} \mathrm{CN}: \mathrm{H}_{2} \mathrm{O}$ to $70: 30 \mathrm{CH}_{3} \mathrm{CN}: \mathrm{H}_{2} \mathrm{O}$ in 25 min . Flow $=1 \mathrm{~mL} \mathrm{~min}^{-1} ; \lambda$ $=254 \mathrm{~nm}$; (black, C) HPLC conditions: $10: 90 \mathrm{CH}_{3} \mathrm{CN}: \mathrm{H}_{2} \mathrm{O}$ to $90: 10 \mathrm{CH}_{3} \mathrm{CN}: \mathrm{H}_{2} \mathrm{O}$ in 25 min . Flow $=1 \mathrm{~mL}$ $\min ^{-1} ; \lambda=254 \mathrm{~nm}$.

The second eluent system method was believed to be the most suitable for the separation of 30a diasteroisomers via preparative HPLC. The separation of 15 milligrams of mixture was carried out, and two fractions of partially separated fast eluting (FE) and slow eluting (SE) isomers were collected. The FE fraction contained a 93:7 mixture of FE:SE isomers (Figure 3.70, A and B), whereas the SE fraction contained a 22:77mixture of FE:SE (Figure 3.70, C). Thus, the separation was evidently inefficient in yielding isomers characterised by a purity higher than $95 \%$.


Figure $3.70(\mathbf{A})$ HPLC trace of 30a diastereomeric mixture enriched of FE isomer. HPLC conditions: 30:70 $\mathrm{CH}_{3} \mathrm{CN}: \mathrm{H}_{2} \mathrm{O}$ to $70: 30 \mathrm{CH}_{3} \mathrm{CN}: \mathrm{H}_{2} \mathrm{O}$ in 25 min . Flow $=1 \mathrm{~mL} \mathrm{~min}^{-1} ; \lambda=254 \mathrm{~nm} ;(\mathbf{B}){ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 202\right.$ MHz ) of diastereomeric mixture depicted in (A), $\mathrm{FE}: \mathrm{SE}=93: 7$. (C) HPLC trace of 30a diastereomeric mixture enriched of SE isomer ( $\mathrm{FE}: \mathrm{SE}=22.3: 75.7$ ), HPLC conditions: $30: 70 \mathrm{CH}_{3} \mathrm{CN}^{2}: \mathrm{H}_{2} \mathrm{O}$ to $70: 30$ $\mathrm{CH}_{3} \mathrm{CN}: \mathrm{H}_{2} \mathrm{O}$ in 25 min .

Consequently, the replacement of $\mathrm{CH}_{3} \mathrm{CN}$ with $\mathrm{CH}_{3} \mathrm{OH}$, which by analytical HPLC seemed to improve the separation via the same gradient method $\left(\mathrm{CH}_{3} \mathrm{OH}\right.$ in $\mathrm{H}_{2} \mathrm{O}$ from 30:70 to 70:30 in 30 minutes, Figure 3.71), was attempted.


Figure 3.71: Analytical HPLC trace of 30a diastereomeric mixture. HPLC conditions: 30:70 $\mathrm{CH}_{3} \mathrm{OH}: \mathrm{H}_{2} \mathrm{O}$ to $70: 30 \mathrm{CH}_{3} \mathrm{OH}: \mathrm{H}_{2} \mathrm{O}$ in 25 min . Flow $=1 \mathrm{~mL} \mathrm{~min}^{-1} ; \lambda=280 \mathrm{~nm}$.

Separation via preparative HPLC of 25 mg of $\mathbf{3 0 a}$ mixture yielded 8 mg of each isomer characterised by a purity higher than $98 \%$ (Figure 3.72). Some material was collected as a mixture.


Figure 3.72: ${ }^{31} \mathrm{P}$ NMR spectra $\left(\mathrm{CD}_{3} \mathrm{OD}, 202 \mathrm{MHz}\right)$ of fast eluting (FE) and slow eluting (SE) isomers of ProTide 30a.

With the aim of identifying a possible scalable process for isomer separation, the gradient reverse phase method was applied to Biotage Isolera One flash chromatography system. The same gradient system $\left(\mathrm{CH}_{3} \mathrm{OH}\right.$ in $\mathrm{H}_{2} \mathrm{O}$ from 30:70 to 70:30 in 30 minutes) that was successful in yielding pure isomers via preparative HPLC was applied to the separation of 50 mg of mixture using a Snap C18 12 g cartridge. Unfortunately, no separation was achieved via this method, due to a solvent mixing issue in the build up of the gradient system.


Figure 3.73: Biotage Isolera One chromatogram of the elution of 30a. Run conditions: cartridge SNAP-Utra 12 g . flow: $12 \mathrm{~mL} / \mathrm{min}$. gradient eluent system: water methanol $\mathrm{A} / \mathrm{B} 30 \%$ to $70 \%$ in 25 min .50 mg sample. $\lambda$ $=280 \mathrm{~nm}$ (black). 254 nm (red).

Consequently, an isocratic eluent system that could allow separation was thought to provide a solution to this issue, avoiding automatic mixing of the solvent by the instrument. In addition, cartridges characterised by smaller particle size were used.
SNAP ultra C18 cartridges (with particle size of $25 \mu \mathrm{~m}$ of spheric silica) were therefore used in place of SNAP C18 (with particle size of $50 \mu \mathrm{~m}$ of irregular silica).

Analytical HPLC was initially used to identify a suitable isocratic method. Different isocratic eluent mixtures of $\mathrm{CH}_{3} \mathrm{OH}$ in water ( $60: 40,55: 45$, and $40: 60$ ) were attempted (Figure 3.74).


Figure 3.74: analytical HPLC trace of 30a diastereomeric mixture. HPLC isocratic conditions: (A) 60:40 $\mathrm{CH}_{3} \mathrm{OH}: \mathrm{H}_{2} \mathrm{O}$ in 30 min . (B) 55:45 $\mathrm{CH}_{3} \mathrm{OH}: \mathrm{H}_{2} \mathrm{O}$ in 30 min . (C) $40: 60 \mathrm{CH}_{3} \mathrm{OH}: \mathrm{H}_{2} \mathrm{O}$ in 30 mim . Flow $=1 \mathrm{~mL}$ $\min ^{-1} ; \lambda=254 \mathrm{~nm}$.

The best separation was afforded by isocratic elution with $55 \%$ of $\mathrm{CH}_{3} \mathrm{OH}$ in water (Figure 3.75 , B), which was applied to the separation of the 30a mixture via Biotage Isolera One through a SNAP Ultra 12 g cartridge. Due to the higher loading capacity of this improved cartridge ( $24 \% \mathrm{w} / \mathrm{w}$ ), separation up to 150 mg of mixture with high efficiency was possible.


Figure 3.75 Biotage Isolera One chromatogram of the elution of 30a. Run conditions: cartridge SNAP-Ultra 12g. Flow: $12 \mathrm{ml} / \mathrm{min}$, isocratic eluent system: $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{H}_{2} \mathrm{O} 55 / 45,100 \mathrm{mg}$ sample. $\lambda=280 \mathrm{~nm}$ (black), 254 nm (red).

An amount of 76 mg of the FE isomer was collected with purity higher than $98 \%$, along with 61 mg of SE isomer with similar purity.
With this methodology in hand to separate Protide 30a isomers, separation of ProTide 49b diastereoisomeric mixture via reverse phase preparative HPLC method was carried out, in order to have few mg of each isomer for in vitro cytotoxicity screening.


Figure 3.76: Analytical HPLC trace of 49b diastereomeric mixture. HPLC isocratic conditions: (A) 60:40 $\mathrm{CH}_{3} \mathrm{OH}: \mathrm{H}_{2} \mathrm{O}$ in 30 min . (B) 55:45 $\mathrm{CH}_{3} \mathrm{OH}: \mathrm{H}_{2} \mathrm{O}$ in 30 min . (C) $40: 60 \mathrm{CH}_{3} \mathrm{OH}: \mathrm{H}_{2} \mathrm{O}$ in 30 mim . Flow $=1 \mathrm{~mL}$ $\min ^{-1} ; \lambda=254 \mathrm{~nm}$.

The different eluent system mixtures used for the separation of 30a mixture were evaluated, and the isocratic elution with $55 \%$ of $\mathrm{CH}_{3} \mathrm{OH}$ in water was selected (Figure 3.76, B). This method gave the best isomer separation, and allowed purification of 10 mg of mixture yielding pure fast eluting and slow eluting 49b separated isomers.


Figure 3.77: ${ }^{31} \mathrm{P}$ NMR spectra $\left(\mathrm{CD}_{3} \mathrm{OD}, 202 \mathrm{MHz}\right)$ of fast eluting (FE) and slow eluting (SE) isomers of ProTide 49b.

Attempts to form crystals of each isomer are ongoing, with the aim of identifying the stereochemistry at the phosphorus center.

### 3.3.2 In vitro testing of separated isomers

The evaluation of parent nucleosides $3^{\prime} \mathrm{dA}$ and 2 F 3 ' dA , along with the mixtures of ProTides 30a and 49b and the separated fast eluting (FE) and slow eluting (SE) isomers was carried out on hematologic and solid tumour cell lines, already selected for previous in vitro screening of 3 ' dA and 2 F 3 ' dA ProTides (Table 3.30, Table 3.31, Table 3.32) by WuXi AppTech.

| Cpnd | CCRF-CEM |  | MOLT-4 |  | K562 |  | HEL92.1.7 |  | KG-1 |  | MV4-11 |  | HL-60 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | IC50 | MI\% | IC50 | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% |
| 3'dA | >198 | 35 | 124.0 | 57 | 47.0 | 98 | 54.0 | 83 | 67.0 | 70 | >198 | 32 | 82.0 | 74 |
| 30a(FE) | 1.86 | 97 | 0.57 | 98 | 3.70 | 99 | 13.0 | 101 | 21.0 | 92 | 1.43 | 99 | 8.7 | 100 |
| 30a(SE) | 2.04 | 97 | 0.43 | 98 | 2.60 | 95 | 11.0 | 98 | 14.0 | 90 | 0.98 | 100 | 11.0 | 92 |
| 30a(mix) | 1.90 | 97 | 0.48 | 98 | 3.0 | 96 | 11.0 | 100 | 18.0 | 90 | 1.03 | 100 | 9.10 | 100 |
| 2F3'dA | 14.0 | 99 | 12.0 | 102 | 13.0 | 98 | 9.69 | 97 | 6.50 | 93 | 2.79 | 100 | 8.80 | 98 |
| 49b(FE) | 0.93 | 98 | 0.74 | 97 | 2.20 | 99 | 5.63 | 102 | 4.20 | 93 | 0.62 | 100 | 3.70 | 98 |
| 49b(SE) | 1.06 | 98 | 0.82 | 97 | 2.50 | 99 | 5.98 | 99 | 6.20 | 92 | 0.69 | 100 | 3.70 | 98 |
| 49b(mix) | 1.77 | 99 | 1.19 | 95 | 8.10 | 97 | 18.0 | 104 | 6.60 | 92 | 1.60 | 100 | 14.0 | 103 |
| PTX | 0.004 | 93 | 0.002 | 94 | 0.004 | 80 | 0.065 | 72 | 0.08 | 89 | 0.03 | 98 | 0.003 | 93 |

Table 3.30 In vitro cytotoxicity screening of 3 ' $\mathrm{dA}, 2 \mathrm{~F} 3$ ' dA and diastereoisomeric mixture and separated diastereoisomers of 30a and 49b. Cytotoxicity data reported as $\mu \mathrm{M} \mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentration considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

| Cpnd | Thp-1 |  | RL |  | Jurkat |  | Hs-445 |  | RPMI-8226 |  | NCI-H929 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | IC 50 | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% |
| 3'dA | >198 | 1 | >198 | 14.2 | >198 | 19 | >198 | $<50$ | >198 | 0.6 | >198 | -0.2 |
| 32a(FE) | 35.0 | 91 | 4.0 | 96.8 | 2.82 | 97 | 42.0 | 99 | 25.0 | 90.8 | 19.0 | 97 |
| 32a(SE) | 66.0 | 69 | 2.64 | 92.6 | 2.29 | 90 | 67.0 | 75 | 19.0 | 86.8 | 16.0 | 94 |
| 32a(mix) | 38.0 | 89 | 3.21 | 95.7 | 3.37 | 94 | 41.0 | 93 | 21.0 | 89.8 | 14.0 | 98 |
| 2F3'dA | 71.0 | 88 | 30.0 | 96.7 | 62.0 | 96 | 25.0 | 102 | 35.0 | 97.4 | 18.0 | 99 |
| 51b(FE) | 8.48 | 102 | 3.11 | 98.3 | 4.21 | 93 | 30.0 | 106 | 11.0 | 94.7 | 5.63 | 100 |
| 51b(SE) | 8.01 | 101 | 4.04 | 96.1 | 4.57 | 93 | 25.0 | 103 | 11.0 | 95.6 | 6.19 | 100 |
| 51b(mix) | 29.0 | 96 | 4.33 | 96.3 | 11.0 | 91 | 74.0 | 88 | 31.0 | 93.5 | 12.0 | 99 |
| PTX | 0.02 | 30 | 0.005 | 64.9 | 0.003 | 82 | 0.05 | 84 | 0.003 | 95.6 | 0.03 | 95 |

Table 3.31 In vitro cytotoxicity screening of 3 ' $\mathrm{dA}, 2 \mathrm{~F} 3$ ' dA and diastereoisomeric mixture and separated diastereoisomers of 30a and 49b. Cytotoxicity data reported as $\mu \mathrm{M} \mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentration considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

| Cpnd | MIA PaCa-2 |  | BxPC-3-Luc |  | HT29 |  | SW620 |  | HepG2 |  | MCF-7 |  | Cal 27 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathbf{I C}_{\mathbf{5 0}}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | IC50 | MI\% | $\mathbf{I C}_{50}$ | MI\% | IC50 | MI\% | $\mathbf{I C}_{50}$ | MI\% |
| 3'dA | >198 | 16 | >198 | 4 | 132.0 | 63 | >198 | 22 | >198 | 11 | 76 | 79 | >198 | 20 |
| 32a(FE) | 18.0 | 94 | 99.0 | 73 | 25.0 | 95 | 35.0 | 87 | 57.0 | 73 | 5.87 | 94 | 24.0 | 101 |
| 32a(SE) | 14.0 | 96 | 74.0 | 71 | 16.0 | 97 | 29.0 | 82 | 32.0 | 78 | 2.94 | 97 | 32.0 | 82 |
| 32a(mix) | 15.0 | 97 | 73.0 | 76 | 18.0 | 97 | 29.0 | 91 | 47.0 | 86 | 3.59 | 96 | 25.0 | 108 |
| 2F3'dA | 45.0 | 100 | 63.0 | 70 | 74.0 | 74 | 25.0 | 95 | 44.0 | 85 | 61.0 | 92 | 34.0 | 99 |
| 51b(FE) | 6.35 | 98 | 52.0 | 78 | 25.0 | 86 | 17.0 | 97 | 17.0 | 71 | 4.06 | 91 | 12.0 | 95 |
| 51b(SE) | 8.78 | 94 | 66.0 | 73 | 31.0 | 86 | 18.0 | 98 | 18.0 | 66 | 6.27 | 90 | 12.0 | 93 |
| 51b(mix) | 18.0 | 92 | 114.0 | 63 | 56.0 | 87 | 41.0 | 89 | 28.0 | 70 | 4.90 | 92 | 26.0 | 95 |
| PTX | 0.004 | 85 | $>0.5$ | 51 | 0.004 | 79 | 0.014 | 84 | 0.1 | 51 | 0.004 | 80 | 0.003 | 92 |

Table 3.32 In vitro cytotoxicity screening of 3 ' $\mathrm{dA}, 2 \mathrm{~F} 3^{\prime} \mathrm{dA}$ and diastereoisomeric mixture and separated diastereoisomers of $\mathbf{3 0 a}$ and $\mathbf{4 9 b}$. Cytotoxicity data reported as $\mu \mathrm{M} \mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentration considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

The data obtained are consistent with the previous results on the same cell lines and there was no significant difference between the activity of ProTide 30a mixture and separated isomers, the same was observed for ProTide 49b mixture and separated isomers.


Figure 3.78 Activity of 3 'dA, 2F3'dA and diastereoisomeric mixture and separated diastereoisomers of derivatives 30a and 49b on tumour cell lines. ${ }^{\mathrm{a}} \mathrm{IC}_{50}>198 \mu \mathrm{M}$; ${ }^{\mathrm{b}} \mathrm{IC}_{50}=124 \mu \mathrm{M} ;{ }^{\mathrm{C}} \mathrm{IC}_{50}=54 \mu \mathrm{M} .{ }^{\mathrm{d}} \mathrm{IC}_{50}=67 \mu \mathrm{M}$. ${ }^{\mathrm{e}} \mathrm{IC}_{50}=82 \mu \mathrm{M} .{ }^{\mathrm{f}} \mathrm{IC}_{50}=62 \mu \mathrm{M}$.


Figure 3.79: Activity of 3 ' $\mathrm{dA}, 2 \mathrm{~F} 3$ ' dA and diastereoisomeric mixture and separated diastereoisomers of derivatives 30a and 49b on tumour cell lines. ${ }^{\mathrm{a}} \mathrm{IC}_{50}>198 \mu \mathrm{M} ;{ }^{\mathrm{b}} \mathrm{IC}_{50}=114 \mu \mathrm{M} ;{ }^{\mathrm{c}} \mathrm{IC}_{50}=132 \mu \mathrm{M}$.

Considering the promising in vitro cytotoxicity data relative to $\mathbf{3 0 a}$ and 49b, and the lack of noticeable difference between the activities of each isomer of both ProTides, in vivo evaluation of the activity of both diasteromeric mixtures of ProTides along with 3' dA were performed by WuXi AppTech on a mouse model of non-Hodgkin B lymphoma.

### 3.3.3 In vivo evaluation of ProTides 30a and 49b in a lymphoma mouse xenograft

Considering that 3 ' dA has already been the object of study in different in vivo anticancer evaluation plans, it was of interest to test the activity of both 3 ' dA and the most promising ProTide 30a in a mouse tumour model. In this evaluation ProTide 49b from the family of 2 F3'dA derivatives was also included as it was endowed with intriguing in vitro anticancer activity (Figure 3.80).


3'dA


30a


49b

Figure 3.80: Structures of compounds assayed in vivo on mouse xenograft by WuXi AppTech.

The in vivo studies were performed on female nude mice (Balb/c). An initial tolerability study with ProTide 30a ( $45 \mathrm{mg} / \mathrm{kg}$ ) was carried out on 4 mice not inocuolated with the tumour and resulted in no change in mobility and alimentary consumption. These positive results allowed the continuation of the study.

Therefore mice were inoculated subcutaneously with human RL (non-Hodgkin's Blymphoma) cells. At two weeks from the cell inoculation, the enlarged tumour lump was weighed and afterwards the treatment was started. Three groups of 10 mice each were treated separately with equimolar amounts of drug (respectively $20 \mathrm{mg} / \mathrm{kg}$ of 3 ' $\mathrm{dA}, 45$ $\mathrm{mg} / \mathrm{kg}$ of ProTide 30a or $47 \mathrm{mg} / \mathrm{kg}$ of ProTide 49b) dissolved in a mixture of $20 \%$ dimethylacetamide (DMA), $15 \%$ Solutol and $65 \%$ water. One group was administered vehicle ( $20 \%$ DMA $+15 \%$ Solutol $+65 \%$ water, control group). Treatments were administered via intraperitoneal injection per 5 days a week, over 14 days. The body weight was also recorded to assess the animal's tolerance to the compound.

A treatment-free period of 7 days at the end of the two-weeks of treatment was further carried out and 2 tumour assessments were performed during this period.

Notably, when examining 30a, 49b and 3' dA anticancer activities, both 3 ' dA and ProTide 30a did not alter the tumour growth compared to mice injected with vehicle (Figure 3.81). On the other hand, mice treated with ProTide 49b showed a decrease in tumour growth compared to control (vehicle). This effect was visible after day 10 from starting of the treatment and continued after the dosing was stopped (days 14-21).


Figure 3.81: Antitumour effect of 3'dA, 30a and 49b against RL lymphoma tumours in nude mice, compared to administration of vehicle. 30a ( 45 mpk , I.P., $5 \mathrm{x} /$ week $* 2$ weeks), $\mathbf{4 9 b}$ ( 47 mpk , I.P., $5 \mathrm{x} /$ week $* 2$ weeks) and 3'dA ( 20 mpk , I.P., $5 \mathrm{x} /$ week $* 2$ weeks). (A) Percentage of tumour growth. (B) Variation of tumour size $\left(\mathrm{mm}^{3}\right)$. Assays performed byWuXi AppTech.

Over the course of the study. the observed change in body weight for the mice treated with $3^{\prime} \mathrm{dA}, \mathbf{3 0 a}$ and 49b was stable and similar to mice treated with vehicle (Figure 3.82), and this was consistent after the treatment was stopped (days 14-21).


Figure 3.82: Mean body weight change (\%) of RL bearing nude mice following administration of vehicle, 30a ( 45 mpk ), 49b ( 47 mpk ) and 3'dA ( 20 mpk ). Assays performed byWuXi AppTech.

The lack of effect of 3 ' dA on the growth of RL tumour could be due to an ineffective targeting of the mass after intraperitoneal injection, possibly due to quick inactivation of the nucleoside analogue by ADA. The in vitro activity of 3'dA on RL tumour cell lines was similarly poor, suggesting that the lack of activity of the drug may also derive from intrinsic resistance mechanisms within the cell, such as ADA catabolism, lack of
phosphorylation or inefficient cell entering via nucleoside transporters. On the other hand, ProTide 30a is immune to ADA mediated catabolism and should enhance the activity of the parent nucleoside at the intracellular level, as suggested by the more favourable activity profile. However, the complete lack in in vivo activity of such derivative could derive from the already reported instability of ProTides in rodent models, described as a limitation for the anticancer and anti-HCV activity of these prodrugs. ${ }^{151,268}$ In fact, the lack in potency exhibited by 3 'dA ProTide 30a could be the consequence of a quick breakdown in mouse plasma, with putative release of the parent nucleoside 3 ' dA , which could be in turn inactivated by ADA, hence the similar negative results for 30a as for the parent nucleoside (Scheme 3.20).
On the other hand, the positive results shown by treatment of mice with 2 F 3 ' dA derived ProTide 49b could either indicate an increased stability to esterase-mediated inactivation of this ProTide moiety due to the presence of a fluorine on the nucleobase, or more likely be due to the fact that even after breakdown of the ProTide moiety, 2F3'dA released could retain anticancer activity due to its stability towards ADA-mediated catabolism (Scheme 3.20).

Nevertheless, ProTides were reported to be more stable in other species, including humans, and this might translate into superior clinical efficacy of both ProTide analogues 30a and 49b.


Scheme 3.20: Putative in vivo metabolism of ProTides 30a and 49b: (a.) Conversion into 3'dAMP and 2F3'dAMP catalysed by the joint activity of carboxypeptidase and phosphoramidase enzymes; (b.) release of 3'dA and 2F3'dA due to activity of 5'-nucleotidase enzymes; (c.) conversion of 3'dA into 3'dIno catalysed by ADA.

### 3.3.4 Stability NMR assays

In order to investigate whether the stability of ProTides in rodent plasma could be responsible of the negative outcome of the mouse xenograft, the stability of ProTides 30a and 49b was compared in rodent serum and human serum.


Figure 3.83: Putative metabolic processing of ProTides 30a and 49b in (rodent) serum.

Incubation of ProTide 30a in rat plasma caused partial processing of the ProTide moiety with release of L-alaninyl phosphate monoester intermediate $\mathbf{3 0 a}{ }^{\mathrm{C}}$ and to a compound which could potentially be identified as nucleotide intermediate $\mathbf{3 0} \mathbf{a}^{\mathbf{D}}$, as suggested by ES/MS analysis of the crude mixture and by previous literature report.


Figure 3.84 (A) Stacked ${ }^{31}$ P NMR spectra (acetone-d6, 202 MHz ) monitoring the stability of ProTide 30a in rat serum. (B) Percentage of 30a metabolites over time in rat serum ( $\%$ determined from ${ }^{31} \mathrm{P}$ NMR integrals).

The presence of untouched compound 30a is still prevalent after 12 h of incubation (Figure 3.84). On the other hand, treatment of ProTide 49b in the same conditions led to complete conversion of the original structure into intermediate $\mathbf{4 9 b}{ }^{\mathbf{C}}$ within 2 hours, paralleled by the formation of what potentially could be recognised as nucleotide $\mathbf{4 9 b}^{\mathbf{D}}$ (Figure 3.85).


Figure 3.85 (A) Stacked ${ }^{31}$ P NMR spectra (acetone- $d 6,202 \mathrm{MHz}$ ) monitoring the stability of ProTide 49b in rat serum. (B) Percentage of 49b metabolites over time in rat serum ( $\%$ determined from ${ }^{31} \mathrm{P}$ NMR integrals).

A very different picture could be noticed after incubation of both ProTides in human serum, with no detectable processing of both drugs after 12 h . The very different stability of 30a in human and rat serum potentially gives an explanation for the lack of activity of ProTide 30a, which could be processed partially or entirely in mouse plasma. This could lead to release of the parent nucleoside 3 'dA, which could lose activity after ADAcatalysed deamination.


Figure 3.86: A. Stacked ${ }^{31}$ P NMR spectra (acetone-d6, 202 MHz ) monitoring the stability of ProTide 30a in human serum. B. Stacked ${ }^{31}$ P NMR spectra (acetone-d $6,202 \mathrm{MHz}$ ) monitoring the stability of ProTide 49b in human serum.

On the other hand, the very quick metabolism of 49b would not be associated with complete lack of activity, due to the stability of 2 F 3 ' dA to deamination, and this could explain the anticancer activity exerted by ProTide 49b. The stability of both drugs in human serum suggests these drugs may have a different activity profile when tested in other species.

### 3.3.5 Conclusions

From anticancer evaluation of 3'dA derivatives, ProTide 30a emerged as the one with the most intriguing activity, along with the 2-fluoro counterpart, ProTide 49b, both deserving further preclinical investigations. Existing as mixtures of two diasteroisomers each, separation of the different isomers of $\mathbf{3 0 a}$ and $\mathbf{4 9 b}$ was performed, with the aim of testing the cytotoxic activity of each isomer separately. This was performed in order to understand whether any additional preclinical investigation would be performed on a single more active isomer of each drug or on the mixture. Reverse phase chromatography was successfully applied for the separation of the isomers, and anticancer evaluation showed no significant difference in activity of each isomer and the mixtures. Hence the diastereomeric mixtures of ProTides 30a and 49b were evaluated for efficacy and toxicity in a in vivo mouse model of non-Hodgkin's B lymphoma, along with the parent nucleoside 3'dA. ProTide 49b succeeded in reducing the tumour growth over 14 days of treatment and up to 21 days of tumour observation. On the other hand, both ProTide 30a and parent nucleoside failed to show any anticancer activity in this evaluation. NMR stability assays in rat and human serum suggested that the lack of activity of 30a could derive from a quick catabolism in rodent plasma, with release of 3 ' dA , that could potentially be inactivated by ADA, leading to loss of anticancer potential. However, the noticed instability of ProTide 49b in rat serum was not paralleled by a lack in activity in the in vivo mouse model. This was envisaged to be due to the potential release of $2 \mathrm{~F} 3^{\prime} \mathrm{dA}$, an analogue of 3 ' dA endowed with improved stability towards ADA-mediated catabolism. Both ProTides 30a and 49b were moreover found to be stable in human serum up to 12 hours of incubation, promising a better anticancer outcome for the administration of both compounds in different tumour models.

### 3.4 5'-Modified 3'dA prodrugs

### 3.4.1 Introduction on rationale for $\mathbf{5}^{\mathbf{\prime}}$-modification

Introduction of a moiety in the 2-position on the adenine base led to an expected increased stability to ADA-mediated inactivation, and provided improved in vitro and in vivo anticancer activity (when using fluorine as a substituent). The exploration of alternative modifications in the structure of 3 ' dA ProTides was believed to offer an additional biological advantage, such as the alteration of position $5^{\prime}$ on the sugar moiety. This moiety has already attracted attention, as groups in this position have been introduced to increase the metabolic stability in vivo, improving the pharmacokinetic drug profile when applied to anticancer and antiviral analogues. ${ }^{301-304}$ Moreover, the $5^{\prime}$-modification on the nucleoside level was reported to increase the stability of the phosphodiester linkage once the nucleotide was incorporated into the DNA, reducing the recognition by nuclease enzyme. ${ }^{305}$ Introduction of a group in position $5^{\prime}$, on the sugar moiety may improve stability of nucleoside analogues by reducing the recognition of substrates by intracellular and extracellular phosphatase enzymes, also known as $5^{\prime}$-nucleotidase enzymes ( $5^{\prime} \mathrm{NT}$ ), which antagonise the activity of nucleoside kinase enzymes by catalysing the cleavage of the monosphosphate group on nucleotides with release of the nucleoside structure. In this way they determine the sizes and turnover rates of the deoxyribonucleotide pools. ${ }^{30,306,307}$ These enzymes have been described as one of the resistance mechanisms to the activity of anticancer and antiviral nucleoside analogues. ${ }^{30,308-311}$

One alternative method to improve the stability of nucleotide analogues is the replacement of the natural and labile phosphoester (phosphorus-oxygen) bond (5'-P-O-C) of the monophosphate group with a chemically and enzymatically stable phosphonate bond (5’-P-C-C). ${ }^{312}$ Phosphonates have been extensively explored especially as antiviral agents and on acyclic nucleoside analogues. ${ }^{53,65,313,314}$ Moreover, acyclic nucleoside phosphonates such as PMEA have shown potent anticancer activity. ${ }^{315-317}$ To explore the potential benefits of a modification in position $5^{\prime}$ on the sugar moiety of $3^{\prime} \mathrm{dA}$ that could improve stability of 3'dA ProTide structures, first the introduction of a 5'-alkyl moiety, and second the replacement of the phosphate with a phosphonate group were attempted (Figure 3.87).



Figure 3.87 Plan of structural modifications on the scaffold of 3'dA ProTides. (A) Introduction of a 5'-alkyl moiety. (B) Replacement of the 5'-phosphate group with a phosphonate moiety.

### 3.4.2 Synthesis of 5'-modified 3'dA ProTides

### 3.4.2.1 Attempts to introduce an alkyl group in the 5'-position.

Introduction of an alkyl moiety in position 5 ' of 3 ' dA , with the aim of synthesising $3^{\prime} \mathrm{dA}$ ProTides endowed with improved metabolic stability was attempted according to Scheme 3.21. The synthesis was reported to first go through oxidation of the 5 ' primary alcohol with subsequent alkylation via a Grignard reagent.

Alternatively, the modification is introduced first on the sugar moiety, which is only then coupled with the appropriate nucleobase. ${ }^{303,304,318-320}$

With the nucleoside analogue 3'dA already available, the route reported in Scheme 3.21 was pursued.


Scheme 3.21: Synthesis of protected 5'-modified cordycepin. Reagents and conditions: (a.) [1] BzCl (3 eq), pyridine, 1.5 h , rt, 54a (52\%), 54b (32\%); [2] $\mathrm{Ac}_{2} \mathrm{O}(6 \mathrm{eq}), \mathrm{Et}_{3} \mathrm{~N}(6 \mathrm{eq})$, DMAP ( 0.4 eq ), $\mathrm{CH}_{3} \mathrm{CN}, \mathrm{rt}, 16 \mathrm{~h}, \mathbf{5 4 c}$ (55\%); [3] $\mathrm{Boc}_{2} \mathrm{O}$ (4 eq), DMAP ( 0.4 eq ), THF, rt, 16h, 54d (100\%); (b.) THF: $\mathrm{H}_{2} \mathrm{O}: T H F(1: 1: 4 \mathrm{v} / \mathrm{v} / \mathrm{v}), 5 \mathrm{~h}, 0$ ${ }^{\circ} \mathrm{C}$, 55a $(98 \%)$, 55b $(62 \%)$, 55c (74\%), 55d (66\%); (c.) Table 3.33; (d.) Table 3.34.
$2^{\prime}, 5^{\prime}-$ TBDMS protected $3^{\prime} \mathrm{dA}(\mathbf{2 5})$, synthesised according to Scheme 3.19 , was protected on the amino group of the adenine base with benzoyl group(s) by treatment with benzoyl chloride ( BzCl ) in pyridine, and conversion into both $N$-benzoyl (54a) and bis $N$-benzoyl (54b) derivatives, respectively in $52 \%$ and $32 \%$ yields. Derivative 54a was subject to selective $5^{\prime}$-deprotection of the silyl group after treatment with a 1:1:4 mixture of trifluoroacetic acid (TFA) in water and THF, ${ }^{300}$ yielding $98 \%$ of desired product 55a.

| Entry | Scheme 3.21 (step c.) |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | Substrate | $\boldsymbol{N}$-protecting group | Oxidising agent | Solvent | T | Time | Result |  |
| $\mathbf{1}$ | $\mathbf{5 5 a}$ | NHBz | DMP $(1.5 \mathrm{eq})$ | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | $0{ }^{\circ} \mathrm{C}$ | 1 h | Traces $^{\mathrm{a}}$ |  |
| $\mathbf{2}$ | $\mathbf{5 5 b}$ | $\mathrm{NBz}_{2}$ | DMP $(2 \mathrm{eq})$ | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | $0{ }^{\circ} \mathrm{C}$ | 2 h | $-{ }^{\mathrm{b}}$ |  |
| $\mathbf{3}$ | $\mathbf{5 5 c}$ | $\mathrm{NAc}_{2}$ | IBX $(1 \mathrm{eq})$ | $\mathrm{CH}_{3} \mathrm{CN}$ | $80{ }^{\circ} \mathrm{C}$ | 1.5 h | $-{ }^{\mathrm{b}}$ |  |
| $\mathbf{4}$ | $\mathbf{5 5 d}$ | $\mathrm{NBoc}_{2}$ | IBX $(1 \mathrm{eq})$ | $\mathrm{CH}_{3} \mathrm{CN}$ | $80{ }^{\circ} \mathrm{C}$ | 1.5 h | $-^{\mathrm{a}}$ |  |

Table 3.33 Substrates involved, reagents, conditions and results of Scheme 3.21 (step c.). ${ }^{\text {a }}$ Crude used directly for step d. Scheme 3.21. ${ }^{\mathrm{b}}$ Mixture of species.

The first attempt of oxidation of the $5^{\prime}$-hydroxyl group was performed by treatment of intermediate 55a with 1.5 equivalents of 1,1,1-triacetoxy-1,1-dihydro-1,2-benziodoxol$3(1 \mathrm{H})$-one (Dess-Martin periodinane, DMP) at $0{ }^{\circ} \mathrm{C}$ for 1 hour (Scheme 3.21, Table 3.33, entry 1 ). ${ }^{208}$ The mixture was then subject to the next step without purification, and the crude 56a was treated with methyl magnesium bromide ( $\mathrm{MeMgBr}, 1 \mathrm{M}$ solution) in THF at - $20{ }^{\circ} \mathrm{C}$. After 6 hours, TLC analysis of the reaction showed a mixture of spots. Chromatography separation of the components yielded one fraction containing traces of product lacking the benzoyl group on the amino function (57a), which was detected by ES/MS (calculated mass: $\mathrm{m} / \mathrm{z}=483.23$ [M], found: $\mathrm{m} / \mathrm{z}=506.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], 989.5[2 \mathrm{M}+$ $\left.\mathrm{Na}^{+}\right]$). However the yield was too low and the reaction needed to be optimised.

These side products were believed to derive from side reaction caused by partially protected amino group, ${ }^{299}$ therefore the fully benzoyl protected intermediate 54b was used for the following attempt. Compound 54b was selectively deprotected in position 5', affording 55b in $62 \%$ yield. Oxidation was tried on product $\mathbf{5 5 b}$ with 2 equivalents of DMP in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ for 2 hours at $-20^{\circ} \mathrm{C}$ (Table 3.33, entry 2).

| Entry | Substrate | $N$-protecting group | Scheme 3.21 (step c.) |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | $\mathbf{M e M g B r}^{\mathbf{a}}$ | Solvent | T | Time | Result |
| $\mathbf{1}$ | $\mathbf{5 6 a}$ (crude) | NHBz | 1.5 eq | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | $-20^{\circ} \mathrm{C}$ | 1 h | $\mathrm{Traces}^{\mathrm{b}}$ |
| $\mathbf{2}$ | $\mathbf{5 6 d}$ (crude) | $\mathrm{NBoc}_{2}$ | 1.5 eq | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | $-78{ }^{\circ} \mathrm{C}$ | 1 h | $-{ }^{\mathrm{c}}$ |

[^4]TLC analysis of the mixture at this step, revealed the presence a mixture of inseparable products, therefore the oxidation step was thought to limit the outcome of the reaction.

Modification of the protecting group on the amino moiety was attempted in order to stabilise the structure of the starting material, and improve the outcome of the reaction. Hence, acetyl moieties were introduced on the amino group. Fully silylated 3'dA (25) was therefore treated with acetic anhydride, DMAP and triethylamine in $\mathrm{CH}_{3} \mathrm{CN}$, yielding $55 \%$ of bis- $N$-acetyl derivative 54c (Scheme 3.21). This intermediate was selectively deprotected in the $5^{\prime}$ position, yielding compound $\mathbf{5 5}$ c in $74 \%$ yield.

For the following step, the oxidising conditions were changed by using 2-iodoxybenzoic acid (IBX) in place of DMP, which was reported to yield better outcomes in this step. ${ }^{299}$ Therefore $\mathbf{5 5 c}$ was treated with IBX in $\mathrm{CH}_{3} \mathrm{CN}$ at $80^{\circ} \mathrm{C}$ for 1.5 hours. Unfortunately, after work up the crude was again a mixture of spots (Table 3.33. entry 3 ).
For the last attempt, the procedure reported by Schinazi et al. was followed, to yield 5'oxidised derivatives. ${ }^{299}$ Hence, fully silylated $3^{\prime} \mathrm{dA}(\mathbf{2 5})$ was protected with di-tert-butyl dicarbonate ( $\mathrm{Boc}_{2} \mathrm{O}$ ) and DMAP in THF for 16 hours (Scheme 3.21). ${ }^{321}$ Product 54d was then selectively deprotected on the $5^{\prime}$ position yielding $66 \%$ of intermediate 55d (Scheme 3.21. step c.). Oxidation of the free hydroxyl was attempted again by treatment with IBX (1.2 eq) at $80^{\circ} \mathrm{C}$ for 1.5 hours. The mixture was then cooled down to $0{ }^{\circ} \mathrm{C}$, insoluble IBX was filtered off and the solvent evaporated. This procedure does not involve isolation of the intermediate aldehyde, therefore the crude was directly dissolved in THF, cooled down to $-78^{\circ} \mathrm{C}$ and 1.5 equivalents of MeMgBr were added (Table 3.34. entry 2). After 1.5 hours stirring, the mixture appeared as a combination of spots on TLC. The mixture of spots potentially derived from degradation of the material, removal of protecting groups and possibly cleavage of the glycosidic bond.
Since the treatment with the Grignard reagents was not successful, the second route leading to modification in the $5^{\prime}$ position involving introduction of a phosphonate bond was pursued.

### 3.4.2.2 Synthesis of phosphonate bond containing analogue of ProTide 32a

The synthesis of 5'-methylene nucleoside phosphonates prodrugs was already reported as a low yielding process, due to the requirement of different steps: (1) introduction of the phosphonate moiety (2) trimethylsilyl bromide driven cleavage of the phosphonate esters (3), purification of highly polar phosphonic acids, and (4) variable yields when coupling phosphonic acids with the prodrug portion (Scheme 3.22). ${ }^{299}$

The method used in this work involved the procedure reported by Schinazi et al. for the synthesis of bis-POM prodrugs of D-2'-deoxy-2'- $\alpha$-fluoro-2'- $\beta$-C-methylphosphonate. ${ }^{299}$ However, for the introduction of the ProTide moiety, a methodology already established in the group for the synthesis of PMPA and PMEA ProTide and bis-diamidate prodrugs, ${ }^{289}$ was applied, adapted from already reported procedures by Holy et al. ${ }^{322}$

The synthesis of the 5 '-phosphonate intermediate $\mathbf{6 1}$ involved the reaction of WadsworthEmmons (HWE) olefination by addition of dialkyl bisphosphonate salt 60 (Figure 3.88) on the $5^{\prime}$ 'aldehyde group in 58d (Scheme 3.22).

Intermediate 58d was treated with 2 equivalents of IBX for 1 hour at $80^{\circ} \mathrm{C}$ in $\mathrm{CH}_{3} \mathrm{CN}$. Then, the crude was directly dissolved in THF and a mixture of tetraethyl methylenebisphosphonate (60, Figure 3.88) and NaH in THF was added at $0^{\circ} \mathrm{C}$ and then stirred for 12 hours at room temperature.


60
Figure 3.88 Structure of tetraethyl methylene-bisphosphonate (60).

The reaction yielded olefin 61 in $43 \%$ yield, and the recovered starting material was unreacted alcohol 55d. The recovery of un-oxidised starting material suggested that the reaction could benefit from the use of either more equivalents of the oxidising agent (IBX) or by stirring for a longer period.


Scheme 3.22: Synthesis of 5'-phosphonate 3'dA analogue of ProTide 30a. Reagents and conditions: (a.) IBX (2eq), $\mathrm{CH}_{3} \mathrm{CN}, 80^{\circ} \mathrm{C}$, 4 h ; (b.) 60 (1.6eq), NaH (2.5eq), THF, $0^{\circ} \mathrm{C}$ to rt, $12 \mathrm{~h}, 58 \%$ two steps; (c.) Pd/C ( $10 \%$ ), $\mathrm{H}_{2}$, $\mathrm{EtOH} / \mathrm{EtOAc}(1: 1)$, rt, $5 \mathrm{~h}, 97 \%$; (d.) $\mathrm{HCOOH} / \mathrm{H}_{2} \mathrm{O}$ (1/1), rt, $72 \mathrm{~h}, 75 \%$; (e.) TMSBr (3eq), 2,6-lutidine (3 eq), $\mathrm{CH}_{3} \mathrm{CN}, 5 \mathrm{~h}, 0{ }^{\circ} \mathrm{C}$; (f.) L-AlaOBn pTSA (1eq), phenol ( 6 eq ), aldrithiol-2 (6eq), $\mathrm{Ph}_{3} \mathrm{P}$ ( 6 eq ), pyridine, TEA, $50^{\circ} \mathrm{C}, 3 \mathrm{~h}$; (g.) $\mathrm{HCl}(0.5 \mathrm{~N}), 2 \%$ three steps.

However, the positive result confirmed that the oxidation process took place in these conditions, therefore the previous negative results for the 5 '-alkylation of the nucleoside analogue potentially resulted from attempts to isolate the aldehyde intermediate, leading to degradation of the product.

In order to push the phosphonate synthesis to higher yields, the equivalents of IBX used were raised to 2 and the mixture was stirred at $80^{\circ} \mathrm{C}$ for 4 hours with improvement in the yield to $58 \%$.


Figure $3.89 \mathbf{A}^{1}{ }^{1} \mathrm{H}$ NMR spectrum of intermediate $\mathbf{5 9}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$. B H5', H1' and H 6 ' signals of compound 59. $\mathbf{C}^{31} \mathrm{P}$ NMR spectrum of compound $\mathbf{5 9}\left(\mathrm{CDCl}_{3}, 202 \mathrm{MHz}\right)$.

The vinylidene phosphonate $\mathbf{5 9}$ was obtained as the $E$-isomer with no trace of the $Z$-isomer (Figure 3.89), as expected from a HWE reaction mechanism. The structure of compound 59 was confirmed by ${ }^{1} \mathrm{H}$ NMR analysis (Figure $3.89, \mathbf{A}$ ). Particularly informative is the presence of two signals corresponding to the two olefinic proton H5' and H6' integrating for one proton (H5' signals overlap with the H1' signal). The H6' signal resonates at 6.80 ppm and presents as a triplet of doublets with coupling with phosphorus defined by a ${ }^{1} J_{\mathrm{H}-\mathrm{P}}$ $=22.1 \mathrm{~Hz}$, coupling with H 5 ' with ${ }^{1} J_{\mathrm{HH}}=17.0 \mathrm{~Hz}$ and coupling with H 4 ' with ${ }^{2} J_{\mathrm{HH}}=4.6$ Hz (Figure 3.89, B). The presence of the phosphonate moiety is moreover confirmed by ${ }^{31} \mathrm{P}$ NMR analysis (Figure 3.89, C).

Intermediate 59 was then reduced by catalytic hydrogenation conditions by treatment with palladium activated on carbon ( $\mathrm{Pd} / \mathrm{C} 10 \%$ ) in $\mathrm{H}_{2}$ atmosphere in a solution $1: 1(\mathrm{v} / \mathrm{v})$ of EtOAc and EtOH, yielding compound 60 in $97 \%$ yield (Scheme 3.22, step c.).

The deprotection of intermediate $\mathbf{6 0}$ was initially attempted in TFA, water and THF (1:1:4 $\mathrm{v} / \mathrm{v} / \mathrm{v}$ ) at room temperature. However, only the TBDMS groups were cleaved with this
method, as confirmed by ES/MS. The mixture was therefore dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and TFA was added (to yield a final mixture $2: 3 \mathrm{TFA}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ). This method proved to be too harsh for this substrate, leading to total glycosidic bond cleavage. The deprotection was finally achieved using a mixture of formic acid in water ( $2: 3 \mathrm{v} / \mathrm{v}$ ), yielding desired product 61 in $75 \%$ yield (Scheme 3.22), after 72 hours.

The first attempt for the synthesis of ProTide 64 involved treatment of deprotected intermediate $\mathbf{6 1}$ with TMSBr ( 5 equivalents) in $\mathrm{CH}_{3} \mathrm{CN}$ at room temperature over 16 hours in order to obtain bis(trimethylsilyl) phosphonate ester 62. The volatiles were then carefully removed to avoid contact with air. The crude ester, under an argon atmosphere, was dissolved in pyridine and triethylamine (TEA), then 1 equivalent of L-alanine benzyl ester para-toluene sulfonic salt ( $p$ TSA) and 6 equivalents of phenol were added, which were previously mixed in pyridine. In a separate flask aldrithiol-2 and triphenylphosphine $\left(\mathrm{Ph}_{3} \mathrm{P}\right)$ were dissolved in pyridine and added to the initial mixture, which was stirred at 50 ${ }^{\circ} \mathrm{C}$ for 3 hours. The final mixture contained multiple products, which attempts to separate via column chromatography and analysis by NMR and ES/MS, did not identify.

Afterwards, 2,6-lutidine was used as a buffer against hydrobromic acid $(\mathrm{HBr})$ that could develop after addition of TMSBr, leading to undesired side reactions such as glycosidic bond cleavage. ${ }^{323}$ Thus, the equivalents of TMSBr were reduced to 3.5 and an equimolar amount of 2,6-lutidine was added. The reaction was then treated with the same procedure as before (Scheme 3.22, step e.).

The product obtained was recognised as the 2 '-trimethylsilyl derivative $\mathbf{6 3}$ by ES/MS analysis of the reaction mixture. The silyl ether bond was easily cleaved by stirring with a 0.5 N solution of HCl in water, and then extracted with EtOAc.

Tedious purification by column chromatography and reverse phase HPLC were further required, and yielded desired compound $\mathbf{6 4}$ in $2 \%$ yield.

Although very poor, the yield was not further optimised at this stage, but the synthesised compound was tested for anticancer activity, with the aim of improving the synthetic procedures and preparing an enlarged family of prodrugs at a later stage.

### 3.4.3 Biological evaluation

Compound 64 was evaluated for anticancer activity on a broad selection of cell lines, previously used for the evaluation of 3'dA prodrugs by WuXi AppTech.

A comparison between the activity of phosphonate $\mathbf{6 4}$ and the phosphate counterpart ProTide 30a, along with the activity of 3' dA is given in Table 3.35, Table 3.36 and Table 3.37.

The activity of $\mathbf{6 4}$ is consistently higher than the parent nucleoside, with $\mathrm{IC}_{50}$ reaching values as low as 4.28 and $6.83 \mu \mathrm{M}$ respectively on Z-138 and MOLT- 4 cell lines, and maximum inhibition percentage always close to $100 \%$. On the other hand, 3 ' dA rarely reached MI\% higher than $50 \%$, as already noticed in previous in vitro evaluations.

| Cpnd | CCRF-CEM |  | MOLT-4 |  | K562 |  | HEL92.1.7 |  | KG-1 |  | MV4-11 |  | HL-60 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | IC ${ }_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% |
| 3'dA | >198 | 35 | 124.0 | 57 | 47.0 | 98 | 54.0 | 83 | 67.0 | 70 | >198 | 32 | 82.0 | 74 |
| 30a | 1.90 | 97 | 0.48 | 98 | 3.0 | 96 | 11.0 | 100 | 18.0 | 90 | 1.03 | 100 | 9.10 | 100 |
| 64 | 17.0 | 99 | 6.83 | 102 | 9.50 | 103 | 28.0 | 104 | 72.0 | 94 | 4.73 | 100 | 27.0 | 99 |
| PTX | 0.004 | 93 | 0.002 | 94 | 0.003 | 80 | 0.03 | 72 | 0.08 | 89 | 0.003 | 98 | 0.003 | 93 |

Table 3.35 In vitro cytotoxicity screening of compounds 3'dA, 30a and 64. Cytotoxicity data reported $\mu \mathrm{M}$ $\mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentration considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

The activity of $\mathbf{6 4}$ was generally lower than ProTide 30a, as it could be expected due to a less effective recognition of the prodrug moiety by metabolising enzymes and of the free phosphonate by kinase enzymes resposible for the second and third phosphorylating steps, required for intracellular activity.

| Cpnd | Thp-1 |  | Z-138 |  | RL |  | Jurkat |  | Hs-445 |  | RPMI-8226 |  | NCI-H929 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% |
| 3'dA | >198 | 1 | >198 | 25 | >198 | 14 | >198 | 19 | >198 | $<50$ | >198 | 0.6 | >198 | -0.2 |
| 30a | 38.0 | 89 | 1.35 | 96 | 3.21 | 96 | 3.37 | 94 | 41.0 | 93 | 21.0 | 90 | 14.0 | 98 |
| 64 | 13.0 | 99 | 4.28 | 100 | 12.0 | 104 | 24.0 | 103 | 43.0 | 108 | 16.0 | 105 | 12.0 | 100 |
| PTX | 0.01 | 60 | 0.003 | 97 | 0.003 | 65 | 0.003 | 82 | 0.004 | 84 | 0.003 | 96 | 0.003 | 95 |

Table 3.36 In vitro cytotoxicity screening of compounds 3'dA, 30a and 64. Cytotoxicity data reported $\mu \mathrm{M}$ IC50 values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI} \%$ (maximum inhibitory effect of the drug at the range of concentration considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

However, on many cell lines the activity of the two prodrugs was very similar, as in Hs445 and SW620 cells ( $\mathrm{IC}_{50}$ ranging between 28 and $43 \mu \mathrm{M}$ ). In some examples, as on cells Thp-1 and HepG2, the activity of the phosphonate analogue (64) was better than ProTide 30a. These positive results potentially indicate that after intracellular activation and release of 3'dAMP from ProTides, this moiety might undergo catabolism by phosphatase
enzymes, with release of $3^{\prime} \mathrm{dA}$, which in turn could be converted to the inosine counterpart by ADA.

| Cpnd | MIA PaCa-2 |  | BxPC-3-Luc |  | HT29 |  | SW620 |  | HepG2 |  | MCF-7 |  | Cal 27 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{IC}_{50}$ | MI\% | IC ${ }_{50}$ | MI\% | IC 50 | MI\% | IC ${ }_{50}$ | MI\% | IC 50 | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% |
| 3'dA | >198 | 16 | >198 | 4 | 132 | 63 | >198 | 22 | >198 | 11 | 76 | 79 | >198 | 20.3 |
| 30a | 15 | 97 | 73 | 76 | 18 | 97 | 29 | 91 | 47 | 86 | 3.59 | 95.7 | 25 | 108.2 |
| 64 | 30 | 105 | 81 | 88 | 38 | 96. | 28 | 98 | 18 | 101 | 23 | 100.7 | 45 | 99.2 |
| PTX | 0.004 | 85 | 0.016 | 51 | 0.004 | 79 | 0.01 | 84 | 0.008 | 51 | 0.003 | 79.7 | 0.002 | 92.1 |

Table 3.37 In vitro cytotoxicity screening of compounds 3 ' $\mathrm{dA}, \mathbf{3 0 a}$ and $\mathbf{6 4}$. Cytotoxicity data reported $\mu \mathrm{M}$ $\mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentration considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

The phosphonate analogue $\mathbf{6 4}$, on the other hand, would not undergo the same process, therefore might retain activity due to the increased stability.


Figure 3.90 Bar chart representing the $\mathrm{IC}_{50}$ values for $3{ }^{\prime} \mathrm{dA}$, and prodrugs 30a and $\mathbf{6 4}$. ${ }^{*} \mathrm{IC}_{50}>198 \mu \mathrm{M}$. ${ }^{* *} \mathrm{IC}_{50}=124 \mu \mathrm{M} . \S \mathrm{IC}_{50}=132 \mu \mathrm{M}$

Therefore the activity of 3 ' dA prodrugs could be considered as deriving from a balance between catabolic (catalysed by nucleotidase and deaminase enzymes) and anabolic processes (catalysed by kinase enzymes), which can lead to an advantage of one or the other prodrug analogue depending on the enzymatic expression of each cell line.

### 3.4.4 Mechanistic studies

A potential drawback in the activity of phosphonate prodrugs of nucleosides such as compound 64 might be the lack in recognition by enzymes responsible for the cleavage of
the prodrug moieties, such as CPY and Hint enzyme, due to the replacement of a natural phosphate group with the alternative phosphonate moiety.

The intracellular activation of compound 64 (Figure 3.91) was investigated by simulating the interaction with CPY and Hint enzyme through docking studies.


Figure 3.91: Putative intracellular metabolic activation of prodrug 64.

Initially, docking of both diasteroisomers ( $S \mathrm{p}$ and $R \mathrm{p}$ ) was performed within the active pocket of CPY enzyme.


Figure 3.92 (A) Docking of $R$ p isomer of 64 in CPY active site. (B) Docking of $S \mathrm{p}$ isomer of $\mathbf{6 4}$ in CPY active site. (Distances: $[\mathbf{A}] \mathrm{H}^{\mathrm{Gly} 52}{ }_{-} \mathrm{O}=\mathrm{C}: 5.19 \AA ; \mathrm{O}^{\text {Ser } 146}-\mathrm{C}=\mathrm{O}: 5.79 \AA ; \mathrm{H}^{\text {Gly53 }}-\mathrm{O}: 3.07 \AA ;[\mathbf{B}] \mathrm{H}^{\mathrm{Gly} 52}-\mathrm{O}: 5.05 \AA ;$ $\left.\mathrm{O}^{\text {Ser } 146}-\mathrm{C}: 3.97 \AA ; \mathrm{H}^{\mathrm{Gly} 53}-\mathrm{O}=\mathrm{C}: 2.76 \AA\right)$

Both isomers seemed to interact similarly inside the enzyme, positioning the amino acid ester moiety close to the pocket containing catalytic residues, although no clear interactions of the adenine ring with additional residues inside the pocket could be noticed.

This might suggest a slower processing by CPY compared to the phosphate counterpart 30a.


Figure 3.93 Docking of $\mathbf{6 4}{ }^{\mathrm{C}}$ in Hint active site. (Distances: $\mathrm{O}^{\text {Ser107 }}-\mathrm{N}: 7.59 \AA ; \mathrm{N}^{\mathrm{His} 112}-\mathrm{C} 6$ ': $6.42 \AA ; \mathrm{N}^{\text {His } 114}-\mathrm{P}$ : $5.24 \AA$ ).

L-alaninyl phosphonate monoester intermediate $\mathbf{6 4}{ }^{\mathrm{C}}$ was further docked into the Hint enzyme, and the results are shown in Figure 3.93. The positioning of the phosphoroamidate group is close to the key residues responsible for the cleavage of the P N bond, although at a bigger distance (5.24-7.59 $\AA$ ) compared to the phosphate analogue $\mathbf{3 0 a}^{\mathrm{C}}$ (3.99-6 $\AA$ ) (Figure 3.34, A).

### 3.4.5 Conclusions

During the explorations of different structural variations to apply to 3 ' dA prodrugs in order to enhance the biological properties, a 5 ' modification was introduced in the form of replacement of the phosphate group ( $\mathrm{P}-\mathrm{O}$ bond) with a phosphonate ( $\mathrm{P}-\mathrm{C}$ ) bond and insertion of an alkyl moiety. The latter modification was attempted by oxidation of the $5^{\prime}$ 'primary alcohol of 3 'dA followed by alkylation by different Grignard reagents, however with no success. Therefore the application of the phosphonate technology was carried out, and the analogue of ProTide 30a was synthesised via a 7 step synthetic procedure. Evaluation of phosphonate 64 in comparison with ProTide 30a and 3'dA showed that phosphonate $\mathbf{6 4}$ had a superior anticancer profile compared to 3 ' dA , with $\mathrm{IC}_{50}$ values in the same range as 30a although generally lower. These results may derive from lower recognition of the molecule by prodrug-activating enzymes such as CPY and Hint by kinase enzymes responsible for the second and third phosphorylation steps. Due to the potential of this kind of modification in terms of significant improvement of the activity of
the parent nucleoside, further investigations should be aimed at altering the phosphonate structure in order to mimic more efficiently the phosphate counterpart, in order to improve the recognition by key enzymes. This could be accomplished by introducing a double bond in the portion linking sugar and phosphonate (Figure 3.94, a. and b.), or by introducing a fluorine in the same position (Figure 3.94, c. and d.), which was already reported to mimic the hydrogen bonding capacity of the phosphate group. ${ }^{324}$


Figure 3.94 Possible modification of the phosphonate moiety in 64 aimed at improving the biological properties.

## 4 8-Chloroadenine nucleoside derivatives

### 4.1 8-Chloroadenosine

### 4.1.1 Background

Cyclic adenosine monophosphate (cAMP, 65, Figure 4.1) is a second messenger that plays an important role in cell growth and differentiation. ${ }^{325}$


Figure 4.1 Structure of cAMP (65).

As such, the cAMP signalling pathway has been used as a target in the development of new anticancer drugs. ${ }^{326}$ The most promising cAMP analogue, tested for its anticancer activity, is 8 -chloro cyclic adenosine monosphosphate (8ClcAMP, 66), also known as tocladesine (Scheme 4.1). Tocladesine was involved in a Phase II clinical trial study in patients with recurrent or refractory multiple myeloma (MM), a fatal B-cell malignancy with a median three years survival. ${ }^{327}$ Although reported as completed in 1999, the full clinical results of this study have not been published to date. Several studies confirmed that tocladesine acts after extracellular activation to its metabolite 8 -chloroadenosine ( 8 ClA , 67), via two enzymatic steps, involving the first conversion to 8 -chloroadenosine-5'monophosphate (8CIAMP) by a serum phosphodiesterase (PDE), followed by dephosphorylation to 8ClA by an extracellular $5^{\prime}$ '-nucleotidase ( $5^{\prime}-\mathrm{NT}$, Scheme 4.1). ${ }^{328}$


Scheme 4.1 Extracellular enzymatic conversion of 8ClcAMP (66) to 8ClA (67).

8 ClA is currently involved in a Phase I clinical trial on patients with chronic lymphocytic leukaemia which is planned to end in 2016. ${ }^{329,330}$ As described in Scheme 4.2, 8ClA enters the cell exploiting a nucleoside transporter belonging to the ENT (equilibrative nucleoside transporter) family, also involved in the active transport of adenosine. ${ }^{331}$ Once inside the cell, the intracellular concentration of 8 ClA depends on the activity of adenosine kinase (AK), responsible for the phosphorylation to 8ClAMP (Scheme 4.2, B). Moreover, 8ClA was reported to be a poor substrate of the ADA enzyme, and the metabolite 8chloroinosine ( 8 ClInno ) generated after deamination of 8 ClA was recognised as an inactive metabolite. ${ }^{322,333}$ The phosphorylation is considered more efficient than deamination in many cells, ${ }^{332}$ 8 ClAMP is subsequently converted into the di- (8ClADP) and triphosphate (8CIATP) metabolites respectively by adenylate kinase (AMPK) and nucleoside diphosphate kinase (NDPK). ${ }^{5,6}$ The accumulation of 8ClATP inside the neoplastic cell increases the incorporation of the analogue in the RNA growing chain, leading to premature transcript termination. ${ }^{329}$ Moreover, 8 ClA was found to inhibit mRNA processing inside neoplastic cells, by inhibiting polyadenylation of the mRNA by poly(A) polymerase, which is one of the key steps in the post-transcriptional processing of mRNA precursors. ${ }^{334}$


Scheme 4.2 Intracellular mechanism of action of 8ClA.

The poly(A) tail length has been shown to have a crucial role in mRNA stability, regulation of mRNA degradation and transport of mRNA from the nucleus to the cytoplasm where it promotes efficient translation. The enzyme poly(A) polymerase
belongs to the superfamily of the nucleotidyl transferases and incorporates ATP onto the 3 ' ends of RNA substrates in a template-dependent manner. ${ }^{335}$

Cleavage of the RNA and polyadenylation inhibition may contribute to the decreased RNA levels observed upon treatment of cells with 8 ClA . In addition to increased RNA degradation, the consequences of polyadenylation inhibition included impaired mRNA transport from the nucleus and altered translation efficiency. ${ }^{335}$
Furthermore, intracellular 8ClADP was reported to be a potential competitor of adenosine diphosphate (ADP) as a substrate of the ATP-synthase enzyme. This enzyme was involved in the oxidative phosphorylation process and it catalyses the synthesis of ATP from recycling ADP. The putative inhibition of this metabolic step may consequently explain the decrease of the ATP cellular pool that was noticed in the cells treated with $8 \mathrm{ClA} .{ }^{336,337}$ 8 ClA was also reported to cause indirect DNA damage by inhibition of the topoisomerase II enzyme. ${ }^{338}$ This enzyme resolves topological problems in DNA replication, transcription, recombination and chromosome condensation and segregation, by means of DNA double strand breakage and religation. ${ }^{339}$ To perform its activity, topoisomerase II needs energy derived from the hydrolysis of ATP, therefore 8ClATP could compete with the natural substrate and inhibit the activity of this enzyme leading to DNA doublestranded breaks. ${ }^{338}$

The intracellular energetic metabolism is additionally altered by 8ClAMP through the reversible succinylation of the amino group at the nucleobase level, presumably catalysed by the adenylosuccinase enzyme (ASase), with the synthesis of succinyl-8ClA monophosphate (S-8ClAMP) (Scheme 4.2C, Scheme 4.3). ${ }^{329}$


Scheme 4.3 Succinylation 8C1AMP by ASase and dephosphorylation to S-8C1A.

S-8ClAMP is then quickly dephosphorylated to succinyl-8ClA (S-8ClA). This reaction consumes fumarate, involved in the citric acid cycle, therefore causes perturbation of cellular metabolism and energetic pools inside the cell. Accumulation of 8 ClA succinyl metabolites may also afford a pharmacokinetic advantage, as these metabolites can constitute deposit forms of the nucleoside inside the neoplastic cells, due to the much
longer half-life of S-8ClA compared to 8ClATP, the putative active metabolite ( $12 \mathrm{~h} v s 6$ h). Thus, S-8ClA may preserve intracellular 8ClAMP levels as 8ClA concentrations fall in the plasma, and consequently prolong 8 ClA therapeutic effects. ${ }^{329}$

### 4.1.2 Preliminary biological evaluation of 8CIA ProTides

The application of the ProTide technology to 8ClA was previously explored within the McGuigan group, ${ }^{340}$ with the aim of delivering the key metabolite, 8ClAMP, from which the two main causes of anticancer activity branch out: phosphorylation leading to accumulation of 8ClATP and succinylation leading to accumulation of S-8ClA. Moreover the application of this prodrug approach would reduce the already low intracellular conversion of 8ClA into inactive 8 CIIno by $\mathrm{ADA}^{341}$ and bypass other potential resistance mechanisms such as the dependency on nucleoside transporters to enter the cell and on AK for the first phosphorylation step. ${ }^{341-343}$ The structures of 8ClA ProTides previously synthesised are reported in Table 4.1.

| Common structure | Cpn | R | $\mathbf{R}^{1}$ | $\mathbf{R}^{2}$ | AA |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  | 68a | $i \mathrm{Pr}$ | Me | H | L-Ala |
|  | 68b | cHx | Me | H | L-Ala |
|  | 68c | Bn | Me | H | L-Ala |
|  | 68d | Neop | Me | Me | DMG |
|  | 68e | cHx | H | H | Gly |
|  | $68 f$ | Bn | H | H | Gly |
|  | 68g | Bn | H | Me | D-Ala |
|  | 68h | Me | $\left(\mathrm{CH}_{2}\right)_{2} \mathrm{SCH}_{3}$ | H | L-Met |
|  | $68 i$ | Bn | $i \operatorname{Pr}$ | H | L-Val |
|  | 68j | Bn | $i \mathrm{Bu}$ | H | L-Leu |
|  | 68k | Bn | Bn | H | L-Phe |

Table 4.1 Previously synthesised 8ClA ProTides (68a-k).

These compounds were tested $v s$ different human and murine cancer cell lines (Table 4.2) by Tenovus (Cardiff University) and the Rega Institute (Leuven, Belgium), mainly derived from solid tumours, with the exception of L1210 and CEM leukaemic cell lines.

| Cell line | Malignancy |
| :---: | :---: |
| MCF7 | Human Breast adenocarcinoma |
| Lovo | Human Colorectal adenocarcinoma |
| PC3 | Human Prostatic adenocarcinoma |
| A549 | Human Lung carcinoma |
| L1210 | Murine lymphocitic leukaemia |
| FM3A | Murine mammary carcinoma |
| CEM | Human acute lymphoblastic leukaemia |
| HeLa | Human cervical adenocarcinoma |

Table 4.2 Set of cell lines used for the biological investigation of the first family of 8ClA ProTides.

The cytotoxic activities of compounds 68a-k in relation to the parent nucleoside 8ClA are reported in Table 4.3. None of the ProTides in this series were more active than 8ClA in vitro, although several were equipotent.

| Cpd | Lovo | MCF7 | PC3 | A549 | L1210 | FM3A | CEM | HeLa |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 8CIA (67) | 0.5 | 0.5 | 0.5 | 5 | 7 | 3 | 4 | 2 |
| $\mathbf{6 8 a}$ | 5 | 40 | 5 | 100 | 73 | 59 | 41 | 17 |
| $\mathbf{6 8 b}$ | 0.5 | 5 | 5 | 50 | - | - | - | - |
| $\mathbf{6 8 c}$ | 5 | 50 | 5 | 50 | 33 | 28 | 16 | 7 |
| $\mathbf{6 8 d}$ | 5 | 50 | 5 | $>100$ | 53 | 43 | 50 | 23 |
| $\mathbf{6 8 e}$ | 5 | 50 | 50 | 50 | 58 | $>100$ | 64 | 27 |
| $\mathbf{6 8 f}$ | 0.5 | 50 | 0.5 | 10 | 6 | 4 | 8 | 2 |
| $\mathbf{6 8 g}$ | 5 | 50 | 5 | 50 | 23 | 24 | 21 | 6 |
| $\mathbf{6 8 h}$ | 50 | 100 | 50 | $>100$ | 94 | $>100$ | 84 | 36 |
| $\mathbf{6 8 i}$ | 0.5 | 5 | 50 | 50 | 46 | 58 | 47 | 14 |
| $\mathbf{6 8 j}$ | 0.5 | 0.5 | 0.5 | 0.5 | 28 | 22 | 14 | 3 |
| $\mathbf{6 8 k}$ | 5 | 5 | 5 | 5 | 19 | 23 | 9 | 3 |

Table 4.3 Cytotoxic evaluation of previously synthesised 8ClA ProTides 68a-k. Activity reported as $\mu \mathrm{M} \mathrm{IC} 50$ values (compound concentration inhibiting the growth of $50 \%$ of cells). Data generated Cardiff University, Tenovus (Lovo, MCF7, PC3, A549 cell lines) and Rega Institute (L1210, FM3A, CEM, HeLa cell lines).

Glycine benzyl ester derivative ( $\mathbf{6 8 f}$ ) showed submicromolar activity $\left(\mathrm{IC}_{50}=0.5 \mu \mathrm{M}\right)$ in PC3, MCF7 and Lovo cells, and it also stood out with low micromolar activity in the other cell lines tested. This derivative was therefore included among the lead compounds in this series, along with 68j (L-leucine benzyl ester ProTide) and 68k (L-phenylalanine). In fact these compounds, although less potent than 8 ClA , still show the best overall activities in this series of prodrugs.

Considering this first set of data, the aim of the present work was to broaden the 8ClA ProTides structures by synthesising additional L-leucine, L-phenylalanine and glycine containing ProTides (Figure 4.2).


Figure 4.2 Structure of the new family of 8ClA ProTides.

Amino acid ester moieties varied among ethyl, $n$-pentyl, benzyl and cyclohexyl esters. LAlanine was also considered, as a generally promising moiety in terms of biological activity. ${ }^{152}$ The aryl moiety was modified to 1-naphthyl, a moiety that was already reported to yield compounds with good activity, ${ }^{152}$ and was not used in previously synthesised 8ClA ProTides. Moreover modification of the monophosphate prodrug structure were performed, including the diamidate symmetrical approach, which was reported as promising both in the antiviral and anticancer fields. ${ }^{67,143,144,322}$

### 4.1.3 Synthesis of 8CIA

8ClA was first synthesised in 1972 by chlorination of adenosine (Ado) using tetrabutylammonium iodotetrachloride, although in low yield. ${ }^{344}$ Since then different methods have been developed. ${ }^{345}$ The methodology developed by Ryu and McCoss required the treatment of Ado with HCl followed by meta-chloroperbenzoic acid ( $m \mathrm{CPBA}$ ) oxidation, ${ }^{346}$ and was more recently improved by replacing HCl with benzoyl chloride. ${ }^{347}$ The latter methodology was already used in the McGuigan group and was initially reproduced in this work, leading to a low $20 \%$ yield of the desired 8ClA (Scheme 4.4, [1]). In an attempt to improve both the yield and the tedious purification of this first method, an alternative procedure was applied, involving the treatment of Ado (22) with an excess of $N$-chlorosuccinimide (NCS) in a mixture of DMF and acetic acid (AcOH) as solvents for the reaction, affording the desired product in $50 \%$ yield (Scheme 4.4, [2]). ${ }^{348}$


Scheme 4.4 Synthesis of 8ClA (67). Reagents and Conditions: [1] BzCl (1 eq), mCPBA (1.1 eq), DMF, rt,
$20 \mathrm{~min}, 20 \%$. [2] NCS ( 3.5 eq ), AcOH, DMF, $\mathrm{rt}, 48 \mathrm{~h}, 50 \%$.

A final optimisation of 8ClA synthesis was investigated at a later stage, and involved a 3step process starting from Ado (22), firstly protected in quantitative yield on the hydroxyl groups of the sugar by treatment with tert-butyl dimethylsilyl chloride (TBDMSCl) and imidazole in DMF. ${ }^{349}$ Protected Ado (69) was then selectively lithiated in position 8 on the nucleobase with lithium diisopropyl amide (LDA) and subsequently chlorinated on the C-8 via tosyl chloride ( TsCl ) as an electrophilic chlorinating agent in a one pot reaction, yielding protected $8 \mathrm{ClA}(\mathbf{7 0})$ in $74 \%$ yield. ${ }^{350,351}$


Scheme 4.5 Synthesis of 8ClA (67). Reagents and Conditions: (a.) TBDMSCl (4 eq), imidazole (8 eq), DMF, 16h, rt; (b.) LDA (5 eq), TsCl (4 eq), THF, $-78{ }^{\circ} \mathrm{C}$ to $\mathrm{rt}, 2 \mathrm{~h}, 74 \%$; (c.) TFA: $\mathrm{H}_{2} \mathrm{O}: \mathrm{THF}(1: 1: 4 \mathrm{v} / \mathrm{v} / \mathrm{v})$, rt, 36 h (67\%).

Compound 70 was deprotected by treatment with TFA: $\mathrm{H}_{2} \mathrm{O}:$ THF (1:1:4) at room temperature over 36 hours, yielding 8ClA (67) in $64 \%$ yield (Scheme 4.5, step c.). Although requiring 3 steps, and with an overall yield of $50 \%$, this latter synthetic strategy was preferred due to the easier purification steps required.

NMR analysis confirmed the structure of the product, and Figure 4.3 shows the ${ }^{1} \mathrm{H}$ NMR spectrum of the final product, which lacks the H 8 , confirming the 8 ClA structure. Moreover, characteristic splitting pattern due to chlorine isotopes could be observed via ES-MS analysis.


Figure $4.3{ }^{1} \mathrm{H}$ NMR spectrum of 8ClA 67 (DMSO-d $6,500 \mathrm{MHz}$ ).

### 4.1.4 Synthesis of 8ClA prodrugs

### 4.1.4.1 8ClA phosphoroamidates

Two strategies for the synthesis of 8C1A ProTides were previously developed. ${ }^{105,152,154,340}$ Direct coupling between the nucleoside (67) and phosphorochloridates bearing non-bulky amino acids (such as glycine and L-alanine) was preformed using 1.1 equivalents of $t \mathrm{BuMgCl}$ in THF. Alternatively, for the synthesis of ProTides bearing bulky amino acids (such as L-phenylalanine or L-leucine), the nucleoside was first protected on the 2' and 3'hydroxyl groups and then coupled with the appropriate phosphorochloridate in the presence of up to 3 equivalents of $t \mathrm{BuMgCl}$. This method allowed the use of more equivalents of the Grignard reagent, hence to facilitate the reaction between the 5'hydroxyl group and the phosphorochloridate without regioisomerism.

For the first series of L-alanine and glycine bearing ProTides, the synthetic procedure is reported in Scheme 4.6.


Scheme 4.6 Synthesis of 8ClA ProTides 71a-h from unprotected 8ClA (67). Reagents and Conditions: $t \mathrm{BuMgCl}$ (1.1 eq), appropriate phosphorochloridate (3 eq), anhydrous THF, rt, 16h, 4-17\% (see Table 4.4).

The second series of napthyl-bearing 8ClA ProTides was synthesised according to Scheme 4.7. The isopropylidene protection of 8 ClA was performed using acetone in the presence of perchloric acid with $79 \%$ yield of intermediate $\mathbf{7 2}$. ${ }^{352}$ This intermediate was then reacted with the appropriate phosphorochloridate and 2-3 equivalents of $t \mathrm{BuMgCl}$, yielding protected ProTides 73a-h.


Scheme 4.7 Synthesis of 8ClA ProTides (74a-h) in a 3 -step process requiring protection of $8 \mathrm{ClA}(67)$, coupling and final deprotection procedures. Reagents and Conditions: (a.) acetone, perchloric acid 70\%, rt, $30 \mathrm{~min}, 79 \%$; (b.) $t \mathrm{BuMgCl}$, appropriate phosphorochloridate, anhydrous THF, rt, overnight; (c.) formic acid, water, rt , 16h, 6-35\% (two steps, see Table 4.4).

Final deprotection was performed by treatment of protected ProTides (73a-h) with a solution of aqueous formic acid in water overnight, ${ }^{353}$ affording ProTides 74a-h in low to moderate yields, as reported in Table 1.4.

| Common structure | Compound | R | $\mathbf{R}^{1}$ | AA | Yield \% |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  | 71a | Bn | Me | L-Ala | 17 \% |
|  | 71b | Et | " | " | 7 \% |
|  | 71c | cHex | " | " | 9 \% |
|  | 71d | $n-\mathrm{Pe}$ | " | " | 4 \% |
|  | 71e | Et | H | Gly | 4 \% |
|  | 71f | Bn | " | " | 4 \% |
|  | 71 g | $n-\mathrm{Pe}$ | " | " | $15 \%$ |
|  | 71h | cHex | " | " | 4 \% |
|  | 74 a | Bn | $i \mathrm{Bu}$ | L-Leu | $26 \%$ |
|  | 74b | $n-\mathrm{Pe}$ | " | " | $36 \%$ |
|  | 74 c | cHex | " | " | 8 \% |
|  | 74d | Et | " | " | $35 \%$ |
|  | 74e | Bn | Bn | L-Phe | 6 \% |
|  | 74f | $n-\mathrm{Pe}$ | " | " | $10 \%$ |
|  | 74 g | cHex | " | " | 21 \% |
|  | 74h | Et | " | " | 26 \% |

Table 4.4 Structures and \% yields of 8ClA ProTides synthesised via one step (71a-h), or three steps (74a-h) method.

### 4.1.4.2 8ClA phosphorodiamidates

The phosphorodiamidate approach was the second nucleotide prodrug system applied to 8 ClA . In the present work, the Yoshikawa ${ }^{156}$ procedure was adopted to obtain $5^{5}$ 'phosphorylated nucleoside analogues using unprotected 8ClA. According to the cited study, in a one pot reaction the unprotected nucleoside is suspended in trialkyl phosphate, and phosphorus oxychloride is added in order to obtain the $5^{\prime}$ 'phosphorodichloridate ( $\mathbf{7 5}$,

Scheme 4.8) in situ. ${ }^{155}$ In the first attempt, triethyl phosphate (TEP) was used, but no product could be observed and the starting material was recovered. For this reason trimethyl phosphate (TMP) was successfully used as a solvent in the second attempt, possibly due to better solubility of the 8 ClA . Analysis of the reaction mixture by ${ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CDCl}_{3}, 202 \mathrm{MHz}\right)$ revealed the formation of a species with a signal at $\sim 7 \mathrm{ppm}$, which corresponded to 75. This intermediate was not isolated but immediately coupled with the appropriate amino acid ester salt in the presence of diisopropyl ethyl amine (DIPEA). The progression of the reaction was monitored again by ${ }^{31} \mathrm{P}$-NMR: the appearance of a single peak in the range of 13-17 ppm, depending on the amino acid used for the synthesis, indicated the formation of the product. The overnight reaction, followed by purification on silica gel, resulted in achieving the final phosphorodiamidates 76a,b (Scheme 4.8, Table 4.5).


Scheme 4.8 General synthesis of 8 ClA phosphorodiamidates. Reagents and Conditions: (a.) $\mathrm{POCl}_{3}(1 \mathrm{eq})$, TMP, $-78{ }^{\circ} \mathrm{C}, 4 \mathrm{~h}$ (b.) amino acid ester $p$-TSA or HCl salt ( 5 eq ), DIPEA ( 10 eq ), $\mathrm{CH}_{2} \mathrm{Cl}_{2}, 27-45 \%$, (see Table 4.5).

| Cpnd | Common structure | $\mathbf{R}^{1}$ | R | AA | Yield |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $76 \mathbf{n}$ | 76b | Bn | Me | L-Ala | $45 \%$ |

Table 4.5: Structures and yields (\%) of 8ClA phosphorodiamidates 76a,b.

### 4.1.5 Biological evaluation of 8CIA prodrugs

The new 8C1A ProTides (71a-h, 74a-h) and the two diamidates (76a,b) were screened first on a small panel of cell lines including two leukemic cell lines (L1210 and CEM) and a solid tumour cell line (HeLa) by the Rega Institute (Leuven, Belgium) in order to compare the anticancer activity to the parent nucleoside 8ClA. The activity of some of the naphthylbearing new compounds could be directly compared to their phenyl-bearing counterparts, and the diamidates synthesised were compared to the corresponding ProTide analogues.

The compounds that emerged as the most promising in this initial screening and some interesting phenyl bearing analogues belonging to the first family of ProTides were further tested by WuXi AppTech on a broader number of cell lines in order to confirm the activity and potentially select candidates for additional biological evaluations.

### 4.1.5.1 Cytotoxic screening of the newly synthesised 8ClA prodrugs

The new 8ClA ProTides 71a-h, 74a-h and phosphorodiamidates 76a,b were tested for cytotoxicity and Table 4.6 shows the activities as $\mathrm{IC}_{50} \mu \mathrm{M}$ values (compound concentration required to inhibit tumour cell proliferation by $50 \%$ ).
None of the synthesised ProTides or phosphorodiamidates of 8ClA was found to be more active than the parent nucleoside. Most of the compounds, however, were reported to have similar activities compared to 8 ClA . ProTides bearing benzyl as an ester moiety (71a, 71f, 74a and 74e) had the lowest $\mathrm{IC}_{50}$ values, typically in the low micromolar range. The $n$ pentyl group is an alternative ester moiety found in some of the most active compounds of all the families (71d, 71g, 74b and 74f). On the other hand, ethyl and cyclohexyl ester groups were found to be the least active compounds of each series, with activities 10 to 20 fold lower than the parent nucleoside (respectively compounds 71b, 71e, 74d, 74h and compounds 71c, 71h, 74c, 74g). The two phosphorodiamidates (76a and 76b) showed similar activities, with $\mathrm{IC}_{50}$ values in the low micromolar range although higher than the parent nucleoside.

|  | Cpd | Ar | AA | Ester | L1210 | CEM | HeLa | cLogP |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| ProTides | 8ClA (67) | - | - | - | $3.2 \pm 0.7$ | $1.1 \pm 0.6$ | $0.89 \pm 0.50$ | -1.44 |
|  | 71a | 1 - | L-Ala | Bn | $10 \pm 1$ | $2.2 \pm 0.3$ | $1.6 \pm 0.8$ | 2.37 |
|  | 71b | 1 - | " | Et | $44 \pm 8$ | $8.4 \pm 2.4$ | $4.0 \pm 2.2$ | 1.19 |
|  | 71c | 1 - |  | cHex | $24 \pm 3$ | $10 \pm 2$ | $2.5 \pm 1.8$ | 2.69 |
|  | 71d | 1 - | " | $n$-Pen | $6.5 \pm 1.1$ | $2.5 \pm 1.5$ | $1.4 \pm 1.1$ | 2.77 |
|  | 71e | 1 - | Gly | Et | $48 \pm 20$ | $22 \pm 13$ | $4.2 \pm 3.8$ | 1.07 |
|  | 71f | 1 - | " | Bn | $5.3 \pm 1.8$ | $2.5 \pm 1.7$ | $1.4 \pm 0.6$ | 2.25 |
|  | 71i | 1 - | " | $n$-Pen | $4.6 \pm 1.5$ | $1.8 \pm 1.2$ | $1.1 \pm 0.4$ | 2.66 |
|  | 71h | 1 - | " | cHex | $39 \pm 12$ | $26 \pm 1$ | $19 \pm 11$ | 2.57 |
|  | 74a | 1- | L-Leu | Bn | $7.8 \pm 0.3$ | $3.5 \pm 1.4$ | $1.7 \pm 0.7$ | 3.82 |
|  | 74b | 1 - | " | $n$-Pen | $9.4 \pm 0.6$ | $6.2 \pm 1.6$ | $2.4 \pm 0.6$ | 4.23 |
|  | 74 c | 1 - | " | cHex | $12 \pm 5$ | $9.0 \pm 1.1$ | $5.7 \pm 1.4$ | 4.15 |
|  | 74d | 1 - | " | Et | $37 \pm 5$ | $12 \pm 4$ | $3.8 \pm 2.4$ | 2.65 |
|  | 74e | 1- | L-Phe | Bn | $9.2 \pm 2.6$ | $3.2 \pm 0.7$ | $1.8 \pm 0.5$ | 3.79 |
|  | 74f | $1-$ | " | $n$-Pen | $9.1 \pm 1.0$ | $3.6 \pm 1.2$ | $1.5 \pm 0.7$ | 4.19 |
|  | 74 g | $1-$ | " | cHex | $12 \pm 5$ | $5.5 \pm 1.1$ | $2.5 \pm 0.5$ | 4.11 |
|  | 74h | 1 - | " | Et | $29 \pm 3$ | $5.5 \pm 3.6$ | $2.0 \pm 1.1$ | 2.61 |
| Diamidates | 76 a | - | L-Ala | Bn | $10 \pm 4$ | $3.8 \pm 0.2$ | $2.3 \pm 1.0$ | 3.17 |
|  | 76b | - | Gly | Bn | $4.6 \pm 0.0$ | $3.4 \pm 0.4$ | $1.8 \pm 0.4$ | 2.55 |

Table 4.6 Biological evaluation of 8ClA, naphthyl-bearing ProTides (71a-h, 74a-h) and phosphorodiamidates ( $\mathbf{7 6 a} \mathbf{a}, \mathbf{b}$ ) of 8 ClA by the Rega Institute in Leuven, Belgium. Cytotoxicity is reported as $\mathrm{IC}_{50} \mu \mathrm{M}$ values ( $50 \%$ inhibitory concentration of cell viability).


Figure 4.4: Activities of 8ClA and selected 8ClA prodrugs (phenyl ProTides 68c,e,f,j,k; napthyl Protides 71a,h, 74a,e; phosphorodiamidates $\mathbf{7 6 a}, \mathbf{b}$ ). $\mathrm{IC}_{50} \mu \mathrm{M}$ values are reported on the Y axis.

Figure 4.4 shows a comparison between the activities of some of the new naphthyl-bearing ProTides with the phenyl containing counterparts, which were previously tested on a number of cell lines including the ones considered in this screening. Similarly, the two phosphorodiamidates were compared to their ProTide counterparts.
The phosphorodiamidate 76a and the naphthyl containing 5'-ProTide analogue 71a emerged as similarly active on the three cell lines considered and 3 to 7 -fold more potent
than the initially synthesised phenyloxy L-alanine benzyl ester ProTide 68c. A similar pattern resulted from the comparison between the two glycinyl cyclohexyl-containing ProTides, with the naphthyl derivative 71h being more potent than the phenyl analogue 68e. Within the series of glycinyl benzyloxy compounds, a similar low micromolar activity profile emerged both for the phenyl (68f) and naphthyl (71f) ProTides and for the phosphorodiamidate (76b). The naphthyl analogues of ProTides containing both L-leucine and L-phenylalanine benzyl ester emerged as more active than the phenyl counterparts, confirming the superior activity profile of naphthyl ProTides compared to the phenyl analogues. This could be due to an increase in lipophilicity that may be responsible for a more efficient cell entering via passive diffusion. As an example, the phenyl L-alanine benzyloxy ProTide 68c has a cLogP of 1.20 , while the naphthyl counterpart has a cLogP of 2.37, and this difference could affect cell penetration. Moreover, the recognition by enzymes responsible for the processing of the ProTide moiety (catA and Hint) could also differ, and was further explored. In conclusion, the synthesis of ProTides bearing a naphthyl moiety and phosphorodiamidates as new 8ClA phosphate prodrugs generated a series of compounds endowed with intriguing cytotoxic activity and this prompted further biological evaluation of the most active compounds on a broader selection of cell lines.

### 4.1.5.2 Cytotoxic screening of 8ClA prodrugs on a broader panel of malignant cell lines.

The new set of cell lines selected for in vitro testing of 8ClA prodrugs derived from both haematological malignancies and solid tumours (Table 4.7).

| Cell line | Malignancy | Cell line | Malignancy |
| :---: | :---: | :---: | :---: |
| MOLT-4 | Acute lymphoblastic leukaemia | HEL92.1.7 | Erythroleukaemia |
| CCRFCE | Acute lymphoblastic leukaemia | HL-60 | Promyelocytic leukaemia |
| RL | Non-Hodgkin's lymphoma | MV4-11 | Biphenotypic B myelomonocytic |
| HS445 | Hodgkin lymphoma | HepG2 | Hepatocellular carcinoma |
| RPMI8226 | Human multiple myeloma | HT29 | Colon adenocarcinoma |
| K562 | Chronic myelogenous | BxPC-3 | Pancreatic cancer |
| KG-1 | Acute myelogenous leukaemia | MCF-7 | Breast adenocarcinoma |
| THP-1 | Acute monocytic leukaemia | MIA PaCa-2 | Pancreatic adenocarcinoma |
| Z-138 | Mantle cell lymphoma | MDAMB23 | Human breast adenocarcinoma |
| NCI-H929 | Plasmacytoma | SW620 | Colon adenocarcinoma |
| Jurkat | Acute T cell leukaemia |  |  |
| Table 4.7: New set of cell lines for biological evaluation of 8C1A most active ProTides by WuXi AppTech. |  |  |  |

As previously mentioned, this second screening of 8 ClA prodrugs aimed at broadening the spectra of cell lines used in order to have a more complete evaluation of the most promising candidates from the initial phenyl family, and the new naphthyl and phosphorodiamidate families (Table 4.8).

| Structure | Cpnd | Ar | R | $\mathbf{R}^{1}$ | AA | cLogP |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |

Table 4.8: Selected compounds from first and second families of 8ClA prodrugs (phenyl-bearing ProTides $\mathbf{6 8 c}, \mathbf{e}, \mathbf{f}, \mathbf{j}$, naphthyl-bearing ProTides 71a,d,f,g, 74a,e,f and phosphorodiamidate 76b), screened for cytotoxic activity of cell lines reported in Table 4.7. cLogP values calculated with ChemDraw 2010 software.

| Cpnd | CCRFCEM |  | HEL92.1.7 |  | HS445 |  | KG-1 |  | Jurkat |  | K562 |  | THP-1 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% |
| 8CIA | 0.92 | 83 | 1.63 | 95 | 5.79 | 74 | 1.05 | 99 | 0.99 | 91 | 1.76 | 85 | 2.91 | 94 |
| 68c | 5.22 | 91 | 32.43 | 90 | >198 | 49 | >198 | 38 | 24.0 | 91 | 15.16 | 92 | 59.67 | 77 |
| 68e | 11.03 | 109 | 14.23 | 86 | >198 | 55 | >198 | 43 | 43.0 | 87 | 10.1 | 92 | 40.29 | 85 |
| 688 | 0.59 | 85 | 0.87 | 99 | 10.99 | 78 | 1.14 | 99 | 1.35 | 94 | 0.85 | 97 | 4.51 | 86 |
| 68j | 5.43 | 100 | 3.73 | 100 | 17.62 | 99 | 10.83 | 99 | 7.69 | 108 | 2.05 | 100 | 4.4 | 100 |
| 71a | 2.84 | 100 | 4.35 | 100 | 11.36 | 99 | 6.7 | 99 | 1.36 | 98 | 2.47 | 100 | 3.11 | 99 |
| 71d | 2.24 | 100 | 2.59 | 100 | 15.73 | 99 | 4.17 | 99 | 2.43 | 97 | 1.92 | 99 | 3.69 | 97 |
| 71f | 6.32 | 90 | 4.81 | 92 | ND | ND | 13.07 | 88 | ND | ND | 5.41 | 87 | ND | ND |
| 71 g | 26.03 | 100 | 21.12 | 100 | ND | ND | 41.77 | 99 | ND | ND | 13.5 | 100 | ND | ND |
| 74a | 6.53 | 100 | 3.07 | 100 | ND | ND | 13.88 | 99 | ND | ND | 1.84 | 100 | ND | ND |
| 74e | 5.77 | 100 | 3.70 | 100 | ND | ND | 14.92 | 99 | ND | ND | 2.28 | 100 | ND | ND |
| 74 f | 5.14 | 100 | 2.48 | 100 | ND | ND | 13.66 | 99 | ND | ND | 1.70 | 100 | ND | ND |
| 76b | 0.41 | 93 | 0.35 | 96 | 9.31 | 78 | 3.36 | 102 | 1.25 | 95 | 0.85 | 96 | 3.22 | 94 |
| PTX | 0.003 | 98 | 0.02 | 83 | 0.01 | 74 | 0.08 | 88 | 0.005 | 97 | 0.01 | 91 | 0.03 | 72 |

Table 4.9: Biological evaluation of 8ClA and relative prodrugs (phenyl-bearing ProTides 68c,e,f,j $\mathbf{j}$, naphthylbearing ProTides 71a,d,f,g, 74a,e,f and phosphorodiamidate 76b) by WuXi app. Tech. Cytotoxicity is reported as $\mathrm{IC}_{50} \mu \mathrm{M}$ values ( $50 \%$ inhibitory concentration of cell viability). $\mathrm{MI}_{\%}$ is the maximum percentage of inhibition exerted by compounds up to the top dose $(198 \mu \mathrm{M})$. PTX: paclitaxel, positive control. Assays performed by WuXi AppTech.

The anticancer activity of 8 ClA was confirmed in this set of data, with $\mathrm{IC}_{50}$ values in the low micromolar range on all cell lines and reaching sub-micromolar values in the leukaemic cell lines CCRFCEM, Jurkat, MV4-11, MOLT4, on the myeloma cell line RPMI-8226 and on the mantle cell lymphoma Z138 cell line (Table 4.9, Table 4.10).

Similar ranges of values were also gathered on the pancreatic tumour cell line MIA PaCa2, on the colorectal cancer cell lines HT29 and SW620 and on the hepatocellular carcinoma cell line HepG2 (Table 4.11).

| Cpnd | NCI-H929 |  | MV4-11 |  | RL |  | RPMI-8226 |  | MOLT-4 |  | HL-60 |  | Z138 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{IC}_{50}$ | $\mathbf{M I}_{\%}$ | IC50 | MI\% | $\mathbf{I C}_{50}$ | $\mathbf{M I}_{\%}$ | $\mathbf{I C}_{50}$ | $\mathbf{M I}_{\%}$ | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% |
| 8ClA | 4.35 | 98 | 0.25 | 100 | 6.64 | 85 | 0.43 | 98 | 0.53 | 99 | 1.62 | 99 | 0.17 | 100 |
| 68c | 8.82 | 101 | 2.18 | 92 | 31.95 | 94 | 7.7 | 91 | 4.15 | 95 | 8.67 | 84 | 4.32 | 86 |
| 68e | 38.92 | 103 | 4.14 | 100 | 14.94 | 98 | 10.29 | 96 | 15.72 | 110 | 11.77 | 94 | 0.41 | 99 |
| 68 f | 10.58 | 98 | 0.24 | 99 | 7.15 | 73 | 0.47 | 99 | 0.82 | 99 | 1.29 | 99 | 0.34 | 100 |
| 68j | 4.04 | 99 | 0.64 | 99 | >19.8 | 51 | 1.65 | 100 | 3.12 | 100 | 6.16 | 100 | 1.5 | 102 |
| 71a | 0.51 | 98 | 0.76 | 100 | 11.27 | 74 | 1.81 | 100 | 0.99 | 99 | 5.07 | 100 | 0.19 | 100 |
| 71d | 1.2 | 98 | 0.76 | 100 | 10.14 | 77 | 1.42 | 100 | 1.24 | 99 | 3.66 | 100 | 0.33 | 100 |
| 71f | 53.62 | 98 | 0.79 | 97 | 6.57 | 74 | 3.54 | 98 | ND | ND | 10.0 | 94 | ND | ND |
| 71 g | 38.05 | 99 | 3.64 | 96 | $>19.8$ | 27 | 7.52 | 100 | ND | ND | 18.21 | 100 | ND | ND |
| 74a | 4.28 | 98 | 0.85 | 100 | >19.8 | 28 | 1.09 | 100 | ND | ND | 6.36 | 100 | ND | ND |
| 74 e | 4.8 | 99 | 1.47 | 100 | 10.86 | 66 | 1.57 | 100 | ND | ND | 4.79 | 100 | ND | ND |
| 74 f | 5.9 | 99 | 0.72 | 100 | 7.47 | 79 | 2.78 | 100 | ND | ND | 5.92 | 100 | ND | ND |
| 76b | 4.97 | 103 | 0.39 | 100 | 1.74 | 94 | 0.70 | 98 | 0.50 | 100 | 0.39 | 100 | 0.20 | 98 |
| PTX | 0.003 | 82 | 0.01 | 99 | 0.003 | 83 | 0.003 | 91 | 0.003 | 95 | 0.003 | 85 | 0.002 | 99 |

Table 4.10 Biological evaluation of 8ClA and relative prodrugs (phenyl-bearing ProTides 68c,e,f,j, naphthylbearing ProTides 71a,d,f,g, 74a,e,f and phosphorodiamidate 76b) by WuXi app. tech. Cytotoxicity is reported as $\mathrm{IC}_{50} \mu \mathrm{M}$ values ( $50 \%$ inhibitory concentration of cell viability). $\mathrm{MI} \%$ is the maximum percentage of inhibition exerted by compounds up to the top dose $(198 \mu \mathrm{M})$. PTX: paclitaxel, positive control. Assays performed by WuXi AppTech.

Some of the screened compounds were consistently less active than 8C1A, as in the case of the phenyl derivatives 68c, 68e and the naphthyl-containing derivatives $71 \mathrm{~g}, 74 \mathrm{e}$ and 74 f . Some compound gave generally poor results apart from some sporadic cases, such as the phenyl L-leucine benzyl ester derivative $\mathbf{6 8 j}$, more active than 8ClA only on the leukaemic cell line NCI-H929 cell line, and on the breast and pancreatic cell lines MCF7 and BcPC-3. The naphthyl counterpart 74a, was generally less active than 8 ClA , although in some cases generated submicromolar $\mathrm{IC}_{50}$ values. Similarly to compound $\mathbf{6 8 j}$, the glycinyl derivative 71f had activity in the same range as 8 ClA on the RL cell line, although with an $\mathrm{IC}_{50}$ value of $6.57 \mu \mathrm{M}$.

| Cpnd | MiaPaCa-2 |  | SW620 |  | HepG2 |  | HT29 |  | MCF7 |  | BxPC-3 |  | MDAMB231 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{IC}_{50}$ | MI | IC ${ }_{50}$ | MI | $\mathbf{I C}_{50}$ | MI | $\mathrm{IC}_{50}$ | MI | IC $\mathbf{5 0}$ | MI | $\mathrm{IC}_{50}$ | MI | $\mathrm{IC}_{50}$ | MI\% |
| 8CI | 0.50 | 93 | 0.83 | 88 | 0.79 | 82 | 0.87 | 79 | 2.15 | 75 | 12.8 | 67 | 3.08 | 74 |
| 68c | 7.77 | 98 | 23.5 | 97 | 20.6 | 84 | 14.6 | 98 | 16.1 | 100 | 26.4 | 80 | ND | ND |
| 68e | 9.53 | 106 | 35.4 | 85 | 13.8 | 85 | 24.3 | 86 | 30.8 | 94 | 39.0 | 66 | ND | ND |
| 68 f | 0.78 | 96 | 1.44 | 87 | 1.03 | 86 | 2.05 | 86 | 3.51 | 92 | 13.8 | 68 | 7.78 | 76 |
| 68j | 1.77 | 99 | 2.47 | 100 | 1.51 | 100 | 2.53 | 92 | 1.48 | 97 | 9.72 | 99 | 15.31 | 99 |
| 71a | 0.95 | 99 | 2.01 | 99 | 1.53 | 102 | 1.6 | 92 | 2.63 | 99 | 6.92 | 99 | ND | ND |
| 71d | 1.28 | 100 | 2.47 | 99 | 2.57 | 101 | 3.28 | 91 | 1.91 | 98 | 11.8 | 92 | ND | ND |
| 71f | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND |
| 719 | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND |
| 74a | ND | ND | ND | ND | ND | ND | ND | ND | 1.09 | 99 | ND | ND | 13.45 | 99 |
| 74e | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND |
| 74f | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND | ND |
| 76b | 0.52 | 94 | 1.23 | 94 | 1.24 | 87 | 1.64 | 87 | 1.08 | 87 | 8.59 | 68 | ND | ND |
| PTX | 0.00 | 87 | 0.02 | 96 | 0.04 | 50 | 0.00 | 75 | 0.01 | 65 | $>0.5$ | 42 | 0.008 | 69 |

Table 4.11 Biological evaluation of 8ClA and relative prodrugs (phenyl-bearing ProTides 68c,e,f,j, naphthylbearing ProTides 71a,d,f,g, 74a,e,f and phosphorodiamidate 76b) by WuXi app. tech. Cytotoxicity is reported as $\mathrm{IC}_{50} \mu \mathrm{M}$ values ( $50 \%$ inhibitory concentration of cell viability). MI\% is the maximum percentage of inhibition exerted by compounds up to the top dose $(198 \mu \mathrm{M})$. Assays performed by WuXi AppTech.

On the other hand, compound phenyl glycine benzyloxy ProTide 68f, already mentioned as a promising derivative, confirmed its potency with generally low $\mathrm{IC}_{50}$ values and better potency than the parent nucleoside on CCRFCEM, HEL92.1.7, K562 and HL60 cell lines, belonging to hematologic malignancies. The naphthyl L-alanine benzyloxy ProTide 71a was 1 to 6 -fold less active than the parent nucleoside, with the exception of cell lines NCIH929 where it demonstrated an improvement in activity by ca. ten-fold, with an $\mathrm{IC}_{50}$ value of $0.51 \mu \mathrm{M}$, in contrast with the poor results obtained with the phenyl analogue $\mathbf{6 8 c}$, on all cell lines. On the pancreatic cell line BxPC-3, compound 71a was slightly more active than 8 ClA , although at relatively high concentration $(6.92 \mu \mathrm{M})$.
The phosphorodiamidate 76b was one of the most active 8ClA phosphate prodrugs, with better potency compared to the parent nucleoside in the leukaemic cell lines CCRFCEM, HEL92.1.7, K562, MOLT-4 and HL-60 and on the lymphoma cell line RL. This compound was also more active than 8 ClA on the pancreatic and breast cancer cell lines BxPC-3 and MCF7, at low $\mu \mathrm{M}$ concentrations.
In conclusion, this second set of data helped identify compounds with improved activity compared to 8 ClA on many of the considered cell lines, such as the phosphorodiamidate 76b, the glycine derivative 68f, belonging to the initial family of 8ClA prodrugs, and the L-alanine-bearing naphthyl analogue 71a, all containing the benzyl ester moiety.

### 4.1.5.3 Biological evaluation of selected ProTides on resistant cell lines

The cytotoxic activity of 8ClA and the ProTides 68e, 68f and 71a was evaluated on a set of leukaemic cell lines by Barts Institute (London), both under normal conditions and under induced resistance conditions, by inhibiting adenosine kinase and equilibrative nucleoside transporter (hENT1), essential for the activity of 8ClA. Moreover adenosine deaminase deficiency was induced in cells, in order to evaluate the extent of ADA-mediated inactivation of 8 ClA and the putative stability of ProTides in these conditions. Compounds $68 f$ and 71a were selected due to the favourable cytotoxicity profile in the previous screenings, while 68e was chosen as a weakly performing ProTide in order to have a negative control that could validate the data (Table 4.12).

| Common structure | Cpnd | Ar | $\mathbf{R}$ | $\mathbf{R}^{1}$ | AA | cLogP |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |

Table 4.12 Selected compounds for in vitro cytotoxicity evaluation on resistant cell lines. cLogP calculated with Chem Draw 2010 software. Assays performed by Barts Institute.

These assays were conducted on the leukaemic cell lines HL60 (human acute promyelocytic leukaemia), CTS (human acute myeloid leukaemia) and K562 (chronic myelogenous leukaemia). Resistance conditions were generated by the pre-treatment of the same cells with one of the following inhibitors: the ADA inhibitor erythro-9-(2-hydroxy-3nonyl) adenine (EHNA) or the AK inhibitor $\mathrm{N}^{7}-\left[\left(1^{\prime} R, 2^{\prime} S, 3^{\prime} R, 4^{\prime} S\right)-2^{\prime}, 3^{\prime}\right.$-dihydroxy-4'-aminocyclopentyl]-4-amino-5- iodopyrrolopyrimidine dihydrochloride hydrate (A-134974) or the hENT1 inhibitor S-(4-nitrobenzyl)-6-thioinosine (NBTI), reported in Figure 4.5.


EHNA


A-134974


NBTI

Figure 4.5 Structures of the ADA inhibitor EHNA, AK inhibitor A-134974 and hENT1 inhibitor NBTI.

When tested on cell lines not previously treated with inhibitor (Figure 4.6, Table 4.13column $\mathbf{A}$ ), 8 ClA had $\mathrm{EC}_{50}$ values in the submicromolar range, ranging from 0.32 to 0.64 $\mu \mathrm{M}$ on the three cell lines considered. When the cells were pre-treated with hENT1 nucleoside transporter inhibitor (Figure 4.6, Table 4.13-column B), the activity of 8ClA was roughly retained in the HL60 cell line, while it decreased considerably in CTS and K562 cells, respectively by 8 and 7 -fold. This data confirmed the dependency of the nucleoside on the hENT-1 nucleoside transporter in CTS and K562 cell lines, and may suggest that it uses a different transport mechanism in HL60 cells. On the other hand, when cells were treated with an adenosine kinase inhibitor (Figure 4.6, Table 4.13, column-C), 8 ClA was inactive up to the highest concentration considered $(20 \mu \mathrm{M})$ in all cell lines, confirming that the activity of 8 ClA is highly dependent on the first phosphorylation step catalysed by AK. The inhibition of adenosine deaminase (Figure 4.6, Table 4.13, columnsD), did not visibly affect 8 ClA activity, in agreement with the already reported low substrate recognition of this nucleoside by ADA. ${ }^{332}$

| Comp. | HL60 |  |  |  | CTS |  |  |  | K562 |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | A | B | C | D | A | B | C | D | A | B | C | D |
| 8CIA | 0.64 | 0.96 | >20 | 0.04 | 0.32 | 5.48 | >20 | 0.54 | 0.64 | 4.69 | $>20$ | 0.32 |
| 68 e | 12.4 | $>20$ | $>20$ | $>20$ | >20 | $>20$ | $>20$ | >20 | >20 | $>20$ | $>20$ | $>20$ |
| 68 f | 0.49 | 8.76 | 14.62 | 0.93 | 16.3 | $>20$ | $>20$ | 1.54 | 2.86 | 12.8 | $>20$ | 7.94 |
| 71a | 0.33 | 1.23 | 21.3 | 1.07 | 2.64 | 1.46 | 10.3 | 2.63 | 6.63 | 1.73 | 19.7 | 12 |

Table 4.13 Biological evaluation of 8ClA and ProTides 68e, 68f and 71a by Barts Institute. Cytotoxicity reported as $\mathrm{EC}_{50} \mu \mathrm{M}$ values ( $50 \%$ effective concentration on inhibition of cell viability). A: control. B: cells pretreated with NBTI (ENT inhibitor). C: cells were pretreated with A-134974 (AK inhibitor). D: cells were pretreated with EHNA (ADA inhibitor). Assays performed by Barts Institute, London.

ProTide 68e (GlyOcHex phenyl), chosen as a poorly cytotoxic ProTide in order to have a negative control, confirmed its cytotoxic profile as it was found to be inactive in all cell lines considered, up to the top concentration used ( $20 \mu \mathrm{M}$ ) apart from a $12.4 \mu \mathrm{M} \mathrm{EC}_{50}$ value on the HL60 cell line. Likewise, the $\mathrm{EC}_{50}$ values of $\mathbf{6 8}$ e were consistently higher than $20 \mu \mathrm{M}$ also in resistance conditions, and when ADA was inhibited. These results reflect the previously reported $\mathrm{IC}_{50}$ data regarding compound $\mathbf{6 8 e}$, which were generally higher than $15 \mu \mathrm{M}$.


Figure 4.6: Biological evaluation of 8 ClA and ProTides 68e, $\mathbf{6 8 f}$ and 71a. Cytotoxicity reported as $\mathrm{EC}_{50} \mu \mathrm{M}$ values ( $50 \%$ effective concentration on inhibition of cell viability). A: control. B: cells pretreated with NBTI (ENT inhibitor). C: cells were pretreated with A-134974 (AK inhibitor). D: cells were pretreated with EHNA (ADA inhibitor). Assays performed by Barts Institute, London.

ProTide $\mathbf{6 8 f}$ was approximately as active as 8 ClA on the HL60 cell line but respectively 51 and 5 -fold less active in the CTS and K562 cell lines. When cells were pre-treated with a hENT-1 inhibitor, the activity of this compound dropped by 18 and 4.5-fold in HL60 and K562 and was completely lost in CTS cell line. This surprising data may suggest a relation between the activity of this compound and this nucleoside transporter. In fact, $\mathbf{6 8 f}$ could penetrate the cell through passive diffusion and once within the cell it could be enzymatically and/or chemically processed to release 8ClA (Figure 4.7). The nucleoside could be a substrate of an efflux pump. ${ }^{354,355}$ In the eventuality of 8 ClA being expelled from the cell, and hENT1 being inhibited, the penetration inside the cell would be halted, hence potentially the reduced activity of $\mathbf{6 8 f}$ is due to hENT1 inhibition.



Figure 4.7: Putative explanation for compound $\mathbf{6 8 f}$ reduced activity after hENT1 inhibition.

Similarly, 68 activity decreased when the cells were treated with an AK inhibitor, becoming completely inactive up to $20 \mu \mathrm{M}$ concentration in CTS and K562 cell lines, and
losing 30-fold potency in HL60 cell line. This result, compared to the drop in potency of 8 ClA in the same conditions, may again confirm the conversion of $\mathbf{6 8 f}$ into 8 ClA within the cell, therefore the dependency on AK for intracellular activation to 8CIAMP. The prodrug $\mathbf{6 8 f}$ was also affected by ADA inhibition on CTS cell line, in fact pre-treatment with ENHA caused an increase in the activity by 10 -fold. On the other hand, the activity on the other cell lines treated in the same conditions did not differ from the untreated settings.

Compound 71a was 8 to 10 -fold less active that 8ClA on CTS and K562 cells, and approximately as active as 8 ClA on the HL60 cell line. When the hENT1 inhibitor NBTI was used, the activity of this prodrug was generally retained in all the cell lines tested. Although the activity of 71a decreased in parallel to the inhibition of AK, the compound still maintained a considerable cytotoxic activity in the micromolar range $\left(\mathrm{EC}_{50}\right.$ values ranging from 10.3 to $21.3 \mu \mathrm{M}$ ). On the other hand, the cytotoxicity of 71a remained constant after ADA inhibition. Considering these data, compound 71a emerged as the only ProTide able to generally overcome the mechanisms of resistance to 8 ClA , by maintaining the cytotoxic activity even after inhibition of hENT1 and AK and being unaffected by ADA inhibition.

### 4.1.5.4 ATP assay on 8ClA ProTides

The cytotoxic activity of 8 ClA and the prodrugs $\mathbf{6 8 e}, \mathbf{6 8 f}$ and 71a was also monitored through ATP assay by Dr. E. Ghazaly at the Barts Institute (London), by measuring the intracellular ratio between 8ClATP and ATP after cell treatment. In fact, one of the mechanisms of action of 8 ClA is the decrease in this ratio, due to the competition between 8 ClADP and ADP for the ATP-synthase enzyme, which leads to an increase in intracellular 8ClATP at the expense of ATP. ${ }^{336}$ As visible from the results reported in Figure 4.8, the ratio of 8CIATP/ATP was close to 1 when HL60, CTS and K562 cells were treated with 8 ClA . The inhibition of hENT1 halved this ratio to 0.5 , while AK inhibition seemed to totally abolish the level of 8ClATP inside the cells, lowering the ratio down to 0 . On the other hand, treatment of cells with the ADA inhibitor EHNA did not significantly affect 8ClATP levels after treatment with 8ClA, generating a 8ClATP/ATP ratio close to unity.


Figure 4.8: 8ClATP/ATP ratio after treatment of the cells with 8ClA (control), 8ClA and hENT1 inhibitor (NBTI), 8ClA and AK inhibitor (A-134974), 8ClA and ADA inhibitor (EHNA). Assays performed at Barts Institute, London.

The treatment of the cells with compound 68e generated 8ClATP levels much lower than after treatment with 8 ClA , inducing a ratio of $8 \mathrm{ClATP} / \mathrm{ATP}$ ranging from 0.02 and 0.05 in all cell lines considered (Figure 4.8). Both pre-treatments of the cells with hENT1 and AK inhibitors did not significantly affect the ratio of 8CIATP/ATP generated by treatment with compound 68e.


Figure 4.9: 8ClATP/ATP ratio after treatment of the cells with 68e (control), 68e and hENT1 inhibitor (NBTI), 68e and AK inhibitor (A-134974), 68e and ADA inhibitor (EHNA). Assays performed at Barts Institute, London.

However, ADA inhibition increased the ratio of 8ClATP/ATP, an explanation could be that the levels of 8ClATP released upon treatment with $\mathbf{6 8 e}$ would be low and therefore ADA inhibition would mainly affect the levels of ATP. When these level decrease the ratio inevitably increases. When cells are treated with compound 68f, as expected from the poor cytotoxic profile in the previously described assays, the levels of 8ClATP recorded are low, therefore the ratio of $8 \mathrm{ClATP} / \mathrm{ATP}$ is close to 0.1 , one-tenth of the levels generated by treatment with 8ClA (Figure 4.10).


Figure 4.10: 8ClATP/ATP ratio after treatment of the cells with $\mathbf{6 8 f}$ (control), $\mathbf{6 8 f}$ and hENT1 inhibitor (NBTI), $68 f$ and AK inhibitor (A-134974), $68 f$ and ADA inhibitor (EHNA). Assays performed at Barts Institute, London.

The 8 ClATP released by treatment with compound $\mathbf{6 8 f}$ are generally unaffected by treatment with hENT1, AK or ADA inhibitors. In contrast, upon treatment with compound 71a, the intracellular levels of 8ClATP were raised compared to the treatment with other ProTides, reaching a ratio of 8 CIATP/ATP ranging from 0.6 and 0.8 , similar to the 8 ClA related ratios (Figure 4.10).


Figure 4.11: 8ClATP/ATP ratio after treatment of the cells with 71a (control), 71a and hENT1 inhibitor (NBTI), 71a and AK inhibitor (A-134974), 71a and ADA inhibitor (EHNA). Assays performed at Barts Institute, London.

The 8ClATP/ATP ratio was not significantly affected by hENT1 and ADA inhibition and only marginally by AK inhibition. These data suggested that compound 71a, although not as active as 8 ClA in normal conditions, is many-fold more potent than compounds $\mathbf{6 8 e}$ and $68 f$ both in terms of cytotoxic activity and considering the 8ClATP levels generated in the intracellular compartment. Moreover, when tested in conditions of AK-deficiency, both cytotoxicity and release of 8ClATP after treatment with 71a were retained, conversely to the complete drop in potency of 8 ClA under the same conditions. This suggests that ProTide 71a could potentially behave as an efficient prodrug of 8ClAMP especially in resistance conditions, and that it may represent a significant new lead.

### 4.1.6 8CIA and prodrug stability

### 4.1.6.1 8ClA ProTides stability in plasma

The optimal metabolism of ProTides depends on their chemical and enzymatic stability. Upon administration, the circulating ProTides are exposed to challenging conditions, such as catabolic enzymes in the liver, and they should resist long enough to enter tumour cells before being degraded. Conversely, ProTides unaffected by enzymatic digestion might not be efficiently converted into active compounds inside the cancer cells. Drug stability in human plasma was assayed by Barts Institute (London), both in the presence and in absence of the ADA inhibitor EHNA. Plasma contains ADA, therefore the concentration of substrates of this catabolic enzyme should decrease in plasma over time. $10 \mu \mathrm{~g} / \mathrm{mL}$ concentrations of 8 ClA and ProTides 68e, 68f and 71a were incubated in untreated plasma at $37^{\circ} \mathrm{C}$ for 2 hours and their concentration was later analysed in the sample to monitor the stability of compounds to ADA-mediated catabolism. The same assay was repeated after treatment of plasma with EHNA (Figure 4.12).


Figure 4.12 Concentration of compounds left after incubation in untreated plasma (control) and plasma treated with ADA-inhibitor (EHNA). Assays performed at Barts Institute, London.

8 ClA is less stable than ProTides 68e, $\mathbf{6 8 f}$ and 71a in plasma, in fact after 2 hours incubation the concentration of the parent nucleoside is significantly lower compared to ProTide concentrations, indicating a higher susceptibility to ADA-mediated catabolism. Moreover, inhibiting ADA by EHNA the stability of 8 ClA increases, while the residual concentrations of ProTides 68e, 68f and 71a are not significantly affected by ADA inhibition. These results suggest that 8 ClA is a substrate of ADA, as previously reported, ${ }^{341}$ while this enzyme does not affect ProTides stability, in human plasma.

### 4.1.6.2 8ClA ProTide stability in human hepatocytes

The stability of 8ClA and ProTides 68e, 68f and 71a was further assayed by Eurofins CEREP laboratories (USA) in cryopreserved human hepatocytes.

| Compound | $\mathbf{t}_{\mathbf{1 / 2}}$ | $\mathbf{C l}_{\mathbf{i n t}}$ |
| :---: | :---: | :---: |
| $\mathbf{8 C 1 A}$ | $>120$ | $<8.2$ |
| $\mathbf{6 8 e}$ | 39 | 25.70 |
| $\mathbf{6 8 f}$ | 57 | 17.60 |
| 71a | 53 | 18.70 |

Table 4.14 Stability of 8C1A (1) and ProTides 68e, 68f and 71a in human hepatocytes, defined as $\mathrm{t}_{1 / 2}$ (half life reported in minutes) and intrinsic clearance ( $\mathrm{Cl}_{\text {int }}$ expressed in $\mu \mathrm{L} / \mathrm{min} / 10^{6}$ cells). Assays performed by Eurofins CEREP laboratories, USA.

The parent nucleoside is very stable in the assay conditions, with a half-life higher than 120 minutes. This would suggest that the molecule is not a substrate of catabolic enzymes although this would be in contrast with the previously reported data, which underline increased stability of the nucleoside in the presence of the ADA inbibitor EHNA. On the other hand, the half-lives of 8 ClA ProTides $\mathbf{6 8 f}$ and 71a are very similar and just under 1 hour, representing an appropriate stability compared to other ProTides in the clinic. ${ }^{151}$ Compound 68e is slightly less stable, with a half-life under 40 minutes.

### 4.1.7 Evaluation of 8CIA ProTides on the CSC compartment of the KG1a cell line

A pilot unpublished study of the gemcitabine ProTide NUC-1031 currently in Phase III clinical trial and developed by the McGuigan group indicated some selectivity for the leukaemic stem cell compartment when compared to gemcitabine. This interesting result prompted the McGuigan group to study the effect of the most promising nucleoside prodrugs on the stem-like cell compartment of KG-1a cell line.
Similarly to the studies performed on 3'dA and relative ProTides, this study was performed on the KG-1a cell line, which is an acute myeloid leukaemia cell line manifesting a minor stem cell-like compartment with a distinct immunophenotype (CD34 ${ }^{+} / \mathrm{CD} 38^{\circ} / \mathrm{CD} 123^{+}$). 8C1A and ProTides 71a, 68f and 71d (Table 4.15) were tested on the KG1a cell line and $\mathrm{LD}_{50}$ values (the concentration required to kill $50 \%$ of the cells) were determined by Prof. C. Pepper (School of Medicine, Cardiff University).

| Cpnd | Common structure | Ar | R | $\mathbf{R}^{1}$ | AA |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 8 ClA | $\mathrm{NH}_{2}$ | - | - | - |  |
| 68f | $\bigcirc{ }^{\circ}$ | Ph | Bn | H | Gly |
| 71a | -p-0 | Nap | Bn | $\mathrm{CH}_{3}$ | L-Ala |
| 71d | ${ }^{\text {R0 }}$ ) $\mathrm{R}^{1} \mathrm{OH} \mathrm{OH}$ | Nap | $n$-Pen | $\mathrm{CH}_{3}$ | L-Ala |

Table 4.15: 8ClA ProTides selected for evaluation of potential CSC selective cytotoxicity.

The cells were then treated with the calculated $\mathrm{LD}_{50}$ concentration and the effect generated on the KG1a sub-population expressing the immunophenotype (CD34 ${ }^{+} / \mathrm{CD} 38^{-} / \mathrm{CD} 123^{+}$) was subsequently identified via immunophenotypic identification of the leukaemic stem cell compartment, as percentage of all viable cells left in the culture. The percentages of stem cells remaining were then plotted on a dose-response graph and the effects of the ProTides were compared with the parent nucleosides.


Figure 4.13: Cytotoxic evaluation of 8ClA and relative ProTides 71a,d and $\mathbf{6 8 f}$ on $\mathrm{KG1a}$ cell line. $\mathrm{LD}_{50}$ : concentration required to kill $50 \%$ of the cells. Data are presented as mean $( \pm \mathrm{SD})$ of five independent experiments Assays performed by Prof. C. Pepper, Cardiff University.

| Cpnd | $\mathbf{L D}_{50}$ values ( $\mu \mathrm{M}$ ) | KG-1a (CD34 ${ }^{+} / \mathrm{CD38}^{-} / \mathrm{CD} 123^{+}$) \% |
| :---: | :---: | :---: |
| Control | - | 3.3 |
| 8CIA | 9.7 | 3.7 |
| 68 f | 7.1 | 2.7 |
| 71a | 3.8 | 2.5 |
| 71d | 20 | 3.5 |

Table 4.16: $\mathrm{LD}_{50}$ cytotoxicity ( $\mu \mathrm{M}$ ) values for 8 ClA and relative ProTides 71a,d and 68f. Percentage of CSCs left after treatment with the $\mathrm{LD}_{50}$ dose of each compound. All assays were carried out using KG1a cells. Assays performed by Prof. C. pepper, Cardiff University.

Table 4.16 shows that ProTides $\mathbf{6 8 f}$ and 71a are more potent than 8 ClA on the KG1a cell line, while 71d has a much higher $\mathrm{LD}_{50}$ value. The percentage of CSCs in the untreated sample (control) was $3.3 \%$. When cells were treated at the $\mathrm{LD}_{50}$ concentration of 8 ClA and
compound 71d, the percentage of stem cell population increased to respectively 3.7 and $3.5 \%$. On the other hand this population decreased to $2.7 \%$ and $2.5 \%$ respectively after treatment with $\mathbf{6 8 f}$ and 71a at their respective $\mathrm{LD}_{50} \mathrm{~S}$. These results suggested that some ProTide structures may offer an advantage over the parent nucleoside in terms of CSCs selectivity which could be linked to an increased cytotoxicity (as in the case of compound 71a).

A further study was designed to extend these investigations and generate a more complete dose-response curve for ProTides 68f, 71a and 71d and 8ClA, in order to evaluate the effect on the LSC compartment not only at the $\mathrm{LD}_{50}$ values of each compound, but at a range of concentrations reaching 1 mM . Compounds 68f and 71a demonstrated again a preferential LSC targeting when compared to 8ClA. These effects were observed at concentrations up to $1 \mu \mathrm{M}$. There was no significant difference in the ability of $\mathbf{6 8 f}$ and 71a to deplete LSCs at this range of concentrations.


Figure 4.14: Analysis of the LSC targeting capacity of 8ClA and proTides 71a, $\mathbf{6 8 f}$ and 71d. All data are the mean $( \pm$ SD $)$ of three independent experiments. Assays performed by Prof. C. pepper, Cardiff University.

The effect of ProTide 71a to decrease the CSCs percentage in a KG1a cells sample might derive from a selective killing of these cells. Alternatively, the drug might induce a differentiation of these stem-like cells into bulk tumour cells. This would similarly lead to a redution in \% of CSCs in the sample. Further studies are ongoing to understand these results.

### 4.1.8 Mechanistic investigations

In order to suggest an explanation for the better in vitro activity of some of the 8ClA ProTides over others, enzymatic studies of the ester hydrolysis and molecular modelling studies of the Hint-1 processing of the phosphoramidate monoesters were carried out.

### 4.1.8.1 CPY NMR assays

CPY enzymatic experiments were carried out on three 8 ClA prodrugs, bearing the same Lalanine benzyl ester motif: the ProTides 71a and 68c, respectively bearing naphthyl and phenyl aryloxy groups, and the diamidate 76a. The aim of these experiments was to understand whether the higher activity of both 71a and 76a over 68c on the cell lines L1210, CEM and HeLa could derive from a better intracellular processing to the same intermediate $5^{\prime}$-L-alaninyl-8-chloroadenosine. Moreover, ProTide 71a showed consistently better activity on a broader selection of cell lines, compared to the phenyl-containing counterpart (68c).


Figure 4.15: Putative enzymatic activation of ProTides 68c, 71a and diamidate 76a by CPY. (a.) carboxypeptidase-catalysed ester cleavage. (b.) intramolecular cyclisation with release of the aryloxy/amino acid ester moiety. (c.) ring opening via hydrolysis.

The conversion of 71a into intermediate 71a ${ }^{\mathbf{A}}$ (Figure 4.15) after benzyl ester cleavage by CPY enzyme was complete in 35 minutes as suggested by the formation of two additional signals at a more downfield chemical shift (Figure 4.16). The conversion into derivative $\mathbf{7 1 a}^{\mathrm{C}}$ was confirmed by the appearance of a peak at 7 ppm and by mass analysis of the crude sample after 12 hours.


Figure 4.16: ${ }^{31} \mathrm{P}$ NMR monitoring of 71a metabolism by CPY enzyme (Acetone- $d 6,202 \mathrm{MHz}$ ).

A similar picture appeared after treatment of ProTide 68c in the same conditions, yielding complete conversion to intermediate $\mathbf{6 8 c}{ }^{\mathrm{C}}$ within 37 minutes.


Figure 4.17: ${ }^{31} \mathrm{P}$ NMR monitoring of $\mathbf{6 8 c}$ metabolism by CPY enzyme. (Acetone- $d 6,202 \mathrm{MHz}$ ).

The same assay was performed on phosphorodiamidate 76a and the conversion of the initial signal belonging to the prodrug at 13.6 ppm into a more downfield peak corresponding to intermediate $\mathbf{7 6 a}^{\mathbf{A}}$ was complete along with the formation of the final metabolite $\mathbf{7 6 a} \mathbf{a}^{\mathbf{C}}$, within 57 minutes. Notably all of the compounds assayed gave an identical final peak in the ${ }^{31} \mathrm{P}$ NMR, consistent with the proposed metabolic route.


Figure 4.18: ${ }^{31} \mathrm{P}$ NMR monitoring of 76a metabolism by CPY enzyme. (Acetone-d6, 202 MHz ).

ProTides 71a and 68c were efficiently converted into the final common intermediate 71a ${ }^{\text {C }}$ $=\mathbf{6 8} \mathbf{a}^{\mathrm{C}}$ within 37 minutes, while the phophorodiamidate 76a required twenty more minutes for complete conversion into $\mathbf{7 1 a}{ }^{\mathbf{C}}=\mathbf{6 8 a}{ }^{\mathbf{C}}$. This result might explain the relatively lower activity of 76a compared to 71a but not the improved activity of 76a over 68c. Moreover the very similar processing of 68c and 71a cannot explain the significantly improved activity of the naphthyl analogue (71a) over the phenyl derivative (68c). However, considering the differences in lipophilicity of the three considered compounds, in terms of cLogP corresponding to $1.19,2.37$ and 3.17 respectively for compounds 68c, 71a and 76a, it could be suggested that a higher cLogP could cause an increase in activity of the compound. This could be a consequence of more efficient cell entering via passive diffusion. Therefore, this could explain the low activity of 68c due to low lipophilicity compared to 71a, and the better activity of 76a over 68c, in spite of the slower processing by CPY.

### 4.1.8.2 Molecular modelling studies

### 4.1.8.2.1 Docking of ProTides in the active site of CPY

As noticeable from in vitro cytotoxicity data, some ProTide moieties showed significantly better activity than others on most of the cell line considered. Some clear examples within the naphthyl-bearing family of ProTides are the superior activity of the L-alanine benzyl ester derivative 71a, compared to compounds bearing glycine cyclohexyl ester (71h) and the L-leucine ethyl ester 74d. An explanation for the substantial loss in activity of the latter compounds could be an inefficient intracellular processing. In order to understand whether this could be a consequence of inadequate recognition by the carboxypeptidase enzyme
responsible for the cleavage of the ester moiety, docking studies within the active site of CPY were carried out. The crystal structure of CPY was downloaded from the protein data bank (PNB 1YSC); ${ }^{356}$ interactions between each selected ProTide diasteroisomer and the crucial residues within the active pocket (Ser146, Gly53, Gly54) were analysed with MOE software. The residue Ser146 is reponsible for the nucleophilic attack at the carbonyl group of the ester, and the glycine residues facilitate the right positioning of the ester moiety by forming H-bonds with the carbonyl group. The likelyhood of correct interaction of each structure with the crucial residues was predicted by measuring the distances between the carbonyl group and the hydroxyl (Ser146) and amido (Gly53 and 54) groups.

As shown in Figure 4.19, the close positioning (a distance within $4.68 \AA$ ) of the carbonyl moiety of both isomers of compound 71a from the specific residues could potentially predict a more efficient ester cleavage by CPY, while the increased distance of compounds 71h and 74d (respectively within 6.88 and $6.02 \AA$ ) could suggest that the low activity derives from a less efficient processing. This could be caused by the hindrance and rigidity of the cyclohexyl ester for compound 71h and the presence of the bulky L-leucine amino acid for $\mathbf{7 4 d}$, causing a mispositioning of the carbonyl portion of the relative ProTides.


71a ( Sp diasteroisomer)


71h (Sp diasteroisomer)


74d ( Sp diasteroisomer)


71a ( Rp diasteroisomer)


71h (Rp diasteroisomer)


74d ( Rp diasteroisomer)

Figure 4.19: Docking of ProTides 71a, 71h and 74d (each diasteroisomer) within the active site of CPY.

### 4.1.8.2.2 Docking of compounds in the active site of Hint-1

The second enzymatic step required for the activation of ProTides is the phosphoramidate bond cleavage, and the enzyme responsible for this step was reported to belong to the human histidine triad nucleotide-binding proteins (Hint). ${ }^{135,261}$ Adenosine monophosphatebound amines are natural substrates of Hint, and the residues of Ser107, His 112 and

His114 are implicated in the P-N bond cleavage, therefore the distances between these residues and the phosphoramidate portion of substrates is crucially involved in the processing.

Dockings of the glycinyl and L-alaninyl monoester 8C1A intermediates (respectively $\mathbf{6 8 c}{ }^{\mathbf{C}}$, $\mathbf{7 1 a}^{\mathrm{C}}, \mathbf{7 6 a}^{\mathrm{C}}$ and 77) were carried out within the binding site of Hint enzyme.


Figure 4.20: Structures of L-alaninyl (68c ${ }^{\mathbf{C}}, \mathbf{7 1 a}^{\mathbf{C}}, \mathbf{7 6 a}^{\mathbf{C}}$ ) and glycinyl (77) phosphate monoester derivatives of 8 ClA docked within the active site of Hint enzyme.

These structures represent intermediates in the activation process of the families of glycine and L-alanine-bearing ProTides and phosphorodiamidates, which yielded some of the most active derivatives such as 71a, 68f, 76a and 76b. The very similar and close distances between both structures and the crucial residues of Hint suggest that a correct positioning within the binding site is the cause of the good activity of these analogues, as a consequence of efficient processing to 8ClAMP by Hint.

$68 \mathrm{c}^{\mathrm{C}}, 71 \mathrm{a}^{\mathrm{C}}, 76 \mathrm{a}^{\mathrm{C}}$


77

Figure 4.21: Docking of L-alaninyl (68c ${ }^{\mathbf{C}}, \mathbf{7 1 a}^{\mathbf{C}}, \mathbf{7 6 a}^{\mathbf{C}}$ ) and glycinyl (77) phosphate monoester derivatives of 8ClA within the active site of Hint.

### 4.1.9 Conclusions

The nucleoside analogue 8ClA, currently involved in phase III clinical studies against CML, was already considered by the McGuigan group as an interesting target for the application of a prodrug approach. In a previous programme, phenyl-bearing 8 8 ClA ProTides were synthesised and tested in vitro, showing an interesting cytotoxic profile, although no improvement over the already active parent nucleoside. In this work, expansion of the 8 ClA prodrug family was carried by synthesising naphthyl-bearing ProTides and phosphorodiamidates of 8ClA. Cytotoxic in vitro evaluation of these prodrugs on an extensive selection of cancer cell lines helped the identification of prodrugs with improved anticancer activity over the parent nucleoside, such as the naphthyl-bearing derivative 71a, the phenyl-bearing compound $\mathbf{6 8 f}$ and the phosphorodiamidate 76b. A selection of derivatives was selected for cytotoxic evaluation in resistance conditions caused by inhibition of AK and hENT1 and ProTide 71a proved to be the only derivative able to consistently retain cytotoxicity. Moreover, the activity of this compound correlated with intracellular levels of 8C1ATP in the cell, which were retained in resistance conditions, unlike the parent nucleoside. In addition, 71a was the most stable analogue to ADA-catalysed breakdown, both in cell culture and in plasma. The same analogue demonstrated stability to hepatocellular metabolism, showing a half-life comparable to ProTides currently in the clinic. A selection of compounds was also evaluated on the KG1a cell line and the effect on the CSC compartment was assayed. ProTides 71a and 68f showed the most significant reduction in the CSC population, suggesting a preferential targeting for this compartment.

Finally, mechanistic investigations performed both via CPY enzymatic metabolism and by docking within the active pocket of both CPY and Hint helped to reach the conclusion that the superior activity of L-alanine and glycine benzyl ester derivatives could be a consequence of efficient ester cleavage due to correct positioning in the active pocket of the carboxypeptidase enzyme. Moreover both glycinyl and L-alaninyl monoester intermediates fit correctly in the active site of Hint, suggesting an additional quick P-N bond cleavage with release of 8CIAMP.

The identification of 71a as an active 8ClA derivative, capable of bypassing resistance mechanisms related to 8 ClA activity is promising for pursuing this prodrug in further preclinical investigations.
Moreover, the promising in vitro activity of phosphorodiamidate 76b should be investigated more thoroughly, also in resistance conditions, and a broader family of 8ClA
phosphorodiamidates should be prepared, carefully considering a balance between good enzymatic processing and lipophilicity of the prodrug.

### 4.2 5'-Modified 8-chloroadenosine analogues

### 4.2.1 Rationale behind the $\mathbf{5}^{\prime}$-modification on 8CIA ProTides

In an attempt to improve the activity of 8ClA ProTides, the modification of the phosphate structure into a phosphonate was carried out.


Figure 4.22: Modification of the phosphate moiety of 8 ClA ProTides into a phosphonate group.

After a thorough investigation of the cytotoxic activity of a broad selection of differently modified 8ClA ProTides, most of the derivatives did not improve the biological potential of the parent nucleoside. Some analogues, such as the ProTide 71a and phosphorodiamidate $\mathbf{7 6 b}$, were found to be more potent than the parent nucleoside by 1 to 3 fold. However, evaluation of some 8ClA ProTides in resistance conditions revealed that the inhibition of both the nucleoside transporter hENT1 and adenosine kinase enzyme led to a decrease in cytotoxicity of some of the tested derivatives. Hence, this could be a consequence of the lability of the phosphate bond of 8ClAMP, released after intracellular processing of 8 ClA ProTides. In order to investigate the validity of this hypothesis, and the potential benefit of stabilising the phosphate bond in terms of cytotoxic activity, ${ }^{312}$ phosphonate derivatives of 8 ClA ProTides were synthesised.

This methodology has already been applied in this work to the family of 3'dA ProTides, with the synthesis and biological evaluation of phosphonate prodrug $\mathbf{6 4}$, which was found to be endowed with a 10 -fold improved cytotoxic activity compared to the parent nucleoside. Although this derivative was not as potent as the phosphate ProTide counterpart 30a, the promising anticancer activity of this compound prompted the application of this method to other promising families of Protides, such as 8 ClA derivatives.

### 4.2.2 Synthesis of 8ClA phosphonate ProTides

The synthetic strategy applied for the synthesis of 8CIA phosphonate ProTides was similar to the one used for the synthesis of phosphonate prodrugs of 3 ' dA . The starting material was the silyl protected derivative 70, synthesised according to Scheme 4.5 . As previously mentioned, the synthetic pathway to prepare phosphonoamidate prodrugs requires initial protection of amino and hydroxyl groups in the starting material. ${ }^{299}$ The adenine amino group was therefore protected with tert-butoxycarbonyl after treatment with $\mathrm{Boc}_{2} \mathrm{O}$ and DMAP in THF (Scheme 4.9), yielding intermediate 78 in $87 \%$ yield.


Scheme 4.9 Synthesis of 5'-phosphonate 8ClA ProTide 86. Reagents and conditions: (a.) $\mathrm{Boc}_{2} \mathrm{O}$ (4 eq), DMAP ( 0.4 eq ), THF, rt, $16 \mathrm{~h}, 87 \%$ (b.) TFA, $\mathrm{H}_{2} \mathrm{O}$, THF ( $1: 1: 4$ ), $0^{\circ} \mathrm{C}, 5 \mathrm{~h}, 68 \%$ (c.) IBX ( 2 eq ), $\mathrm{CH}_{3} \mathrm{CN}, 80^{\circ} \mathrm{C}$, 4 h ; (d.) tetraethyl methylenediphosphonate (1.6eq), $\mathrm{NaH}\left(2.5 \mathrm{eq}\right.$ ), THF, $0^{\circ} \mathrm{C}$ to $\mathrm{rt}, 12 \mathrm{~h}, 68 \%$ (two steps); (e.) $\mathrm{Pd} / \mathrm{C}(10 \%), \mathrm{H}_{2}$, $\mathrm{EtOH} / \mathrm{EtOAc}(1 / 1)$, rt, $5 \mathrm{~h}, 81 \%$; (f.) $\mathrm{HCOOH} / \mathrm{H}_{2} \mathrm{O}$ (1/1), rt, $72 \mathrm{~h}, 75 \%$; (g.) TMSBr (3eq), 2,6-lutidine ( 3 eq ), $\mathrm{CH}_{3} \mathrm{CN}, 5 \mathrm{~h}, 0{ }^{\circ} \mathrm{C}$; (h.) L-AlaOBn pTSA (1eq), phenol ( 6 eq ), aldrithiol-2 ( 6 eq ), $\mathrm{Ph}_{3} \mathrm{P}$ (6eq), pyridine, $\mathrm{Et}_{3} \mathrm{~N}, 5{ }^{\circ} \mathrm{C}, 3 \mathrm{~h}$; (i.) $\mathrm{HCl} 0.5 \mathrm{~N}, 5 \%$ (three steps).

Selective deprotection of the primary hydroxyl function of intermediate 78 was performed by treatment with a mixture of TFA in water and THF (1:1:4 $\mathrm{v} / \mathrm{v} / \mathrm{v})$ at $0{ }^{\circ} \mathrm{C}$, affording intermediate 79 in $68 \%$ yield. Refluxing this intermediate with 2 equivalents of IBX in $\mathrm{CH}_{3} \mathrm{CN}$ for 4 hours then resulted in oxidation of $\mathbf{7 9}$ into $\mathbf{8 0}$, which was not isolated but immediately subject to Wadsworth-Emmons (HWE) olefination by addition of tetraethyl methylenediphosphonate sodium salt. These two steps yielded olefinic intermediate $\mathbf{8 1}$ in $68 \%$ yield. Saturation of the double bond in intermediate 81 was carried out by treatment
with $\mathrm{Pd} / \mathrm{C}$ in a mixture of EtOH and EtOAc under hydrogen atmosphere for 5 hours, yielding $\mathbf{8 2}$ in $81 \%$ yield. Deprotection of the amino and hydroxyl groups of intermediate $\mathbf{8 2}$ was afforded in $75 \%$ yield by the use of a mixture of formic acid in water for 72 hours, similar to the procedure set up for the synthesis of 3'dA phosphonate ProTide 66. Despite the long reaction time required for this deprotection step, the use of any harsher conditions was avoided. This was decided in order to prevent the cleavage of the glycosidic bond that had happened before after treatment of the 3 ' dA intermediate with TFA in water at room temperature. The final step for the synthesis of 8ClA phosphonate ProTide required at first the deprotection of the phosphonate bond by treatment with 3 equivalents of trimethylsilyl bromide (TMSBr) in $\mathrm{CH}_{3} \mathrm{CN}$ for 5 hours at $0{ }^{\circ} \mathrm{C}$. The presence of 3 equivalents of 2,6lutidine was necessary to prevent the complete degradation caused by the release of hydrobromic acid during the reaction, as was already noticed for the synthesis of the 3 ' dA phosphonate ProTide 66. ${ }^{127}$ The reaction conditions were carefully kept anhydrous and after evaporation of the volatiles, the crude containing intermediate $\mathbf{8 5}$ was dissolved in pyridine and triethylamine (TEA), then 1 equivalent of L-alanine benzyl ester para-toluene sulfonic salt ( $p$ TSA) and 6 equivalents of phenol were added, which were previously mixed in pyridine. In a separate flask aldrithiol-2 and triphenylphosphine $\left(\mathrm{Ph}_{3} \mathrm{P}\right)$ were dissolved in pyridine and added to the initial mixture, which was stirred at $50{ }^{\circ} \mathrm{C}$ for 3 hours. Evaporation of the mixture and extraction with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ afforded a mixture of compounds containing the $2^{\prime}, 3^{\prime}$-bis- $O$-trimethyl silyl protected intermediate $\mathbf{8 5}$, which was converted into desired $\mathbf{8 6}$ by quick treatment with a solution of HCl 0.5 N that cleaved the silyl ether bonds. Purification by column chromatography and reverse phase HPLC were required, and yielded desired compound 86 in $5 \%$ yield. Although very poor, the yield was not further optimised at this stage, but the synthesised compound was tested for anticancer activity, with the aim of improving the synthetic procedures and preparing an enlarged family of prodrugs at a later stage.

### 4.2.3 Biological evaluation

Compound $\mathbf{8 6}$ was evaluated for anticancer activity on a broad selection of malignant cell lines by WuXi AppTech, previously used for the evaluation of 3'dA prodrugs.
A comparison between the activity of phosphonate 86 and the phosphate counterpart ProTide 68c, along with the activity of 8ClA is given in Table 4.17, Table 4.18, Table 4.19 and Figure 4.23.

| Cpnd | CCRF-CEM |  | MOLT-4 |  | K562 |  | Hel92.1.7 |  | KG1 |  | MV4-11 |  | HL-60 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% |
| 8CIA | 1.47 | 91 | 1.08 | 100 | 0.64 | 95 | 0.66 | 94 | 1.72 | 100 | 0.19 | 100 | 1.29 | 100 |
| 68c | 9.48 | 96 | 1.85 | 95 | 1.41 | 96 | 4.46 | 96 | 6.3 | 98 | 0.52 | 100 | 2.83 | 98 |
| 86 | 28.0 | 98 | 25.0 | 104.2 | 47.0 | 103 | 74.0 | 91 | 156.0 | 29 | 31.0 | 100 | 73.0 | 93 |
| PTX | 0.004 | 93 | 0.002 | 9 | 0.003 | 80 | 0.03 | 72 | 0.08 | 89 | 0.003 | 99 | 0.003 | 93 |

Table 4.17 In vitro cytotoxicity screening of 8ClA, ProTide 68c and phosphonate ProTide 86. Cytotoxicity data reported $\mu \mathrm{M} \mathrm{IC} 50$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentration considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

| Cpnd | THP-1 |  | Z138 |  | RL |  | Jurkat |  | Hs-445 |  | RPMI-8226 |  | NCI-H929 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | IC 50 | MI\% | $\mathbf{I C}_{50}$ | MI\% | IC 50 | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathrm{IC}_{50}$ | MI\% |
| 8ClA | 1.86 | 96 | 1.06 | 94 | 1.83 | 92 | 2.85 | 79 | 3.42 | 83 | 2.86 | 96 | 5.55 | 100 |
| 68c | 2.58 | 99 | 0.11 | 95 | 9.66 | 94 | 3.46 | 86 | 0.11 | 79 | 3.40 | 97 | 5.89 | 98 |
| 86 | 48.0 | 86 | 28.0 | 99 | 73.0 | 81 | 40.16 | 99 | 76.0 | 78 | 52.0 | 79 | 52.0 | 86 |
| PTX | 0.01 | 60 | 0.003 | 99 | 0.003 | 65 | 0.003 | 82 | 0.004 | 84 | 0.003 | 96 | 0.003 | 95 |

Table 4.18 In vitro cytotoxicity screening of 8ClA, ProTide 68c and phosphonate ProTide 86. Cytotoxicity data reported $\mu \mathrm{M} \mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentration considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

| Cpnd | MiaPaCa-2 |  | BxPC-3-Luc |  | HT29 |  | SW620 |  | HepG2 |  | MCF-7 |  | Cal 27 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | IC 50 | MI\% | IC 50 | MI\% | IC 50 | MI\% | IC 50 | MI\% | $\mathbf{I C}_{50}$ | MI\% | IC 50 | MI\% | $\mathrm{IC}_{50}$ | MI\% |
| 8ClA | 0.79 | 81 | 12 | 62 | 2.17 | 70 | 2.32 | 94 | 1.18 | 69 | 1.67 | 83 | 2.12 | 75 |
| 68c | 2.28 | 93 | 15 | 74 | 3.62 | 88 | 3.06 | 87 | 3.09 | 82 | 4.29 | 100 | 13 | 88 |
| 86 | 86.0 | 91 | 73 | 11 | 81 | 43 | 88 | 58 | 91 | 43 | 103 | 59 | 116 | 28 |
| PTX | 0.004 | 85 | 0.016 | 51 | 0.004 | 79 | 0.01 | 84 | 0.008 | 51 | 0.003 | 80 | 0.002 | 92 |

Table 4.19 In vitro cytotoxicity screening of 8C1A, ProTide 68c and phosphonate ProTide 86. Cytotoxicity data reported $\mu \mathrm{M} \mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentration considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.


Figure 4.23: Comparison of the cytotoxic activity of 8 ClA , ProTide 68c and phosphonate ProTide 86.

The phosphonate derivative $\mathbf{8 6}$ is consistently less active than both the parent nucleoside and the phosphate analogue 68c. This might derive from an inefficient processing of the ProTide moiety with release of 8ClA phosphonate, which should in turn mimic 8ClAMP. Alternatively, the released 8ClA phosphonic acid might be poorly recognised by the kinase enzymes that would catalyse the next phosphorylation steps, required for intracellular activity. Despite being less active than 8 ClA , compound $\mathbf{8 6}$ is still endowed with poor anticancer activity in the range between 30 and $80 \mu \mathrm{M}$ on most cell lines, with a preference for cells deriving from haematologic malignancies, such as MOLT-4 and CCRF-CEM.

### 4.2.4 Mechanistic studies

As previously mentioned in the case of phosphonamidate ProTide of 3 ' $\mathrm{dA} \mathbf{6 4}$, a potential cause of the lack in activity of such prodrugs might be due to inefficient processing of the prodrug moiety by carboxypeptidase and phosphoramidase enzymes. This would negatively affect the release of nucleoside phosphonic acid, which should mimic the nucleoside monophosphate.

The investigation of the intracellular activation of 8ClA phosphonate prodrug 86 (Scheme 4.10) was carried out through docking studies, by simulation of the interaction with CPY and Hint enzymes.


Scheme 4.10: Putative intracellular metabolic activation of prodrug $\mathbf{8 6}$ by CPY/Hint enzymes.

Initially, docking of both diasteroisomers ( $S$ p and $R \mathrm{p}$ ) was performed within the active pocket of the CPY enzyme. Both isomers seemed to interact similarly inside the enzyme, positioning the amino acid ester moiety close to the pocket containing catalytic residues.


Figure 4.24 (A) Docking of $R$ p isomer of $\mathbf{8 6}$ in CPY active site. (B) Docking of $S$ p isomer of $\mathbf{8 6}$ in CPY active site.

These results suggest that the cleavage of the ester moiety might not be a limiting step in the intracellular activation of prodrug 86.

The metabolite that is potentially formed after cleavage of the benzyl moiety and intramolecular rearrangements is $\mathbf{8 6}^{\mathbf{C}}$ (Scheme 4.10). This L-alaninyl phosphonate monoester intermediate was then docked inside the active site of Hint enzyme, which is thought to be responsible for the cleavage of the phosphoramidate bond. The results are shown in Figure 3.93. The positioning of the phosphoroamidate group is close to the key residues responsible for the cleavage of the P-N bond, although the phosphonate moiety might lack crucial interactions owing to the missing oxygen from the phosphate group.


Figure 4.25 Docking of $\mathbf{8 6}^{\mathrm{C}}$ in Hint active site.

However, the docking studies suggest that compound $\mathbf{8 6}$ would interact efficiently with carboxypeptidase and phosphoramidase processing enzymes. Hence the lack in improved activity of this prodrug compared to the phosphate counterpart 68c might derive from an inefficient processing to diphosphonate and triphosphonate metabolites, by respectively nucleoside monophosphate and diphosphate kinase enzymes.

### 4.2.5 Conclusions

In an attempt to improve the activity of 8ClA ProTides, the modification of the phosphate group into more stable moieties such as phosphonamidate prodrugs was carried out. The synthetic route applied had already been determined for the synthesis of 3'dA phosphonamidate prodrugs, and yielded one example of an 8 ClA derivative via a 7 step synthetic procedure, in low yield. Evaluation of this compound showed a poor anticancer profile, both in comparison with ProTide 68c and 8ClA. Dockings of derivate $\mathbf{8 6}$ within the active pocked of CPY and Hint enzymes predict a correct interaction with the active pockets. This does not provide an explanation for the lack in activity of derivative $\mathbf{8 6}$. However, provided compound $\mathbf{8 6}$ is effectively processed in the intracellular environment, with release of the metabolite 8 ClA phosphonic acid, this compound might not mimic efficiently 8ClAMP. Therefore lack of recognition by kinase enzymes responsible for the conversion into active nucleoside triphosphate could be responsible for the low anticancer profile of this molecule.

### 4.3 8-Chloro-2'-deoxyadenosine

### 4.3.1 Background

8-Chloro-2'-deoxyadenosine ( $8 \mathrm{CldA}, 87$ ) is the $2^{\prime}$-deoxyribose analogue of 8 ClA . This deoxyadenosine analogue was synthesised by Chen et al. ${ }^{351}$ in order to investigate the effect of its incorporation into DNA growing chains. In fact, 8 CldA was suggested to assume a preferential syn conformation instead of the anti conformation of natural nucleosides.


Figure 4.26: Syn and anti conformations of 8CldA caused by rotation around the glycosidic bond.

This preference was confirmed for purine nucleoside analogues modified at the C8 position with hindered groups, which frequently adopt syn glycosidic torsion angles instead of the preferred anti conformation found in unmodified nucleosides ${ }^{357}$ and was proven for 8-bromoadenosine and 8-methoxy-2'-deoxyadenosine, ${ }^{358-360}$ and suggested for $8 \mathrm{ClA} .{ }^{351,361}$ The presence of such modification within the DNA growing chain was suggested to lead to base mispairing effects.

8CldA was inserted into a DNA template and although pairing with thymidine, the nucleotide exhibited decreased nucleotide incorporation efficiencies compared to the natural substrate 2'-deoxyadenosine $5^{\prime}$ '-triphosphate and induced DNA polymerase pausing during DNA synthesis, possibly caused by its altered conformation. ${ }^{351}$

A study to explore the potential correlation between the reported data and the anticancer activity of this 8 ClA derivative was of interest, especially concerning the application of the ProTide approach to this nucleoside analogue with the aim of potentially bypassing limiting steps in the phosphorylation and cell penetration, that could decrease the efficiency of intracellular release of the potential active metabolite 8 -chloro-2'-deoxyadenosine-5'-triphosphate (8CldATP).

### 4.3.2 Synthesis of 8-chloro-2'-deoxyadenosine

The first attempt to synthesise 8 -chloro- 2 '-deoxyadenosine involved the treatment of commercially available $2^{\prime}$-deoxyadenosine (29) with $N$-chlorosuccinimmide (NCS) in a solution of DMF and acetic acid, as successfully adopted for the synthesis of 8ClA (87). ${ }^{348}$


Scheme 4.11: Attempted synthesis of 8CldAdo using the NCS method. Reagents and Conditions: a) NCS (3.5 eq), acetic acid, DMF, rt, 48 hours.

Unfortunately, due to the increased instability and acid lability of the glycosidic bond in the 2'-deoxyadenosine nucleoside structure, ${ }^{362,363}$ this methodology caused degradation of the starting material with adenine detection via ${ }^{1} \mathrm{H}$ NMR. ${ }^{27}$

This unsuccessful result prompted the use of a methodology involving first a tertbutyldimethyl protection of 2'-deoxyadenosine (29), followed by chlorination via one-pot treatment with LDA and tosyl chloride ( TsCl ) and final deprotection to yield desired 8CldA (87). This methodology was already applied to the synthesis of 8 ClA and was used by Chen et al. for the synthesis of $8 \mathrm{CldA} .{ }^{351}$


Scheme 4.12 Synthesis of 8CldAdo. Reagents and Conditions: (a.) TBDMSCl (4 eq), imidazole (8 eq), DMF, rt, 15 hours ( $94 \%$ ); (b.) LDA ( 5 eq ), TsCl ( 10 eq ), THF, $-78{ }^{\circ} \mathrm{C}, 2$ hours ( $75 \%$, two steps); (c.) TBAF (3.5 eq), THF, rt, 1 hour (33\%).

The initial protection of $\mathbf{2 9}$ was performed with TBDMSCl and imidazole in DMF, yielding desired $\mathbf{8 8}$ in $94 \%$ yield. ${ }^{349}$ This intermediate was then lithiated on the C-8 position of adenine using LDA. This makes the C-8 position available for an electrophilic substitution using tosyl chloride as a chlorine cation donor, yielding protected 8 CldA (89) in $75 \%$ yield. ${ }^{350}$ The final deprotection step was accomplished using a solution of TBAF $(1 \mathrm{M})$ in THF. ${ }^{26}$ This procedure, has the drawback of a challenging purification of the
product from hydrolysed TBAF residues, that were still present after column chromatography. In an attempt to remove the TBAF residues, a small portion of the impure compound was stirred with DOWEX sulfonic acid resin, and $\mathrm{CaCO}_{3}$ in anhydrous $\mathrm{CH}_{3} \mathrm{OH} .{ }^{364}$ Unfortunately, through this procedure, it was impossible to recover the compound that appeared to have been degraded, likely due to the strong acidic nature of the resin. However, washing the residue with abundant $\mathrm{Et}_{2} \mathrm{O}$ finally yielded clean product, as reported in the ${ }^{1} \mathrm{H}$ NMR spectrum in Figure 4.27.


Figure 4.27: ${ }^{1} \mathrm{H}$ NMR spectrum of 8 CldA (DMSO- $d 6,500 \mathrm{MHz}$ ).

### 4.3.3 Synthesis of 8CIdA prodrugs

In the first attempt of the synthesis of 8 CldA ProTides, 8 CldA was treated with tBuMgCl (2.5 eq) and 1-naphthyl-(pentyloxy-L-leucinyl) phosphorochloridate (3 eq) in THF (Table 4.20 , entry 1). This procedure failed to yield the desired 5 '-phosphoramidate derivative but afforded the $3^{\prime}, 5^{\prime}$-difunctionalised analogue 90a (Scheme 4.13) in $8 \%$ yield.
A similar result was obtained when lowering the amount of Grignard reagent used and modifying ( 1.5 eq ) the phosphorochloridate structure (Table 4.20, entry 2), yielding compound 90b in $28 \%$ yield. Reducing the phosphorochloridate and $t \mathrm{BuMgCl}$ equivalents used (respectively 2 and 1.1 eq ) (Table 4.20, entry 2), and further lowering the amount of phosphorochloridate used (to 1.1 eq ) (Table 4.20, entry 3), yielded similar results.


Scheme 4.13 Synthesis of 8CldA phosphoramidates (90a-d) from unprotected 8CldA. Reagents and conditions: $t \mathrm{BuMgCl}$, appropriate phosphorochloridate, THF, rt, 16 h (Table 4.20).

| Entry | Phosphorochloridate |  |  |  |  | tBuMgCl* ${ }^{\text {eq }}$. | 3',5'-bisProTide (yield) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | Ar | R | $\mathbf{R}^{1}$ | AA | Eq. |  |  |
| 1 | Nap | $n$ Pen | $i \mathrm{Bu}$ | L-Leu | 3 | 2.5 | 90a (8\%) |
| 2 | Ph | $n \mathrm{Hex}$ | Me | L-Ala | 3 | 1.5 | 90b (28\%) |
| 3 | Ph | Bn | Me | L-Ala | 2 | 1.1 | 90c (25\%) |
| 4 | Nap | Bn | H | Gly | 1.1 | 1.1 | 90d (16\%) |

Table 4.20: Structures and number of equivalent of phosphorochloridate used, amount of $t \mathrm{BuMgCl}$ used and relative yields for the synthesis of derivatives $\mathbf{9 0 a} \mathbf{- d}$. ${ }^{*} 1 \mathrm{M}$ solution in THF.

Replacing the Grignard reagent with NMI (5 equivalents), yielded the desired 5'phosphoramidates, in low yield (2\%) without trace of $3^{\prime}, 5^{\prime}$-bisProTide (Scheme 4.14, Table 4.21, entry 1 ).


91a,b
Scheme 4.14: Synthesis of 8CldA phosphoramidates (91a-b) from unprotected 8CldA. Reagents and conditions: NMI, appropriate phosphorochloridate, THF, rt, 16 h (Table 4.21).

A similar yield resulted using this same method and an alternative phosphorochloridate (Table 4.21, entry 2).

| Entry | Phosphorochloridate |  |  |  |  |  | NMI eq. |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |

Table 4.21: Structures and number of equivalent of phosphorochloridate used, amount of NMI used and relative yields for the synthesis of derivatives $91 \mathbf{a}, \mathbf{b}$.

In an attempt to improve the yield of 8 -chloro- $2^{\prime}$-deoxyadenosine ProTides, an alternative method was envisaged, involving selective deprotection of the 5 '-position of the fully protected nucleoside (89) followed by $t \mathrm{BuMgCl}$-mediated coupling reaction with the appropriate phosphorochloridate and final 3'-deprotection via TFA in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ to yield the desired 5 '-ProTide. Compound $\mathbf{8 9}$, was selectively deprotected on the 5 '-hydroxyl position by treatment with a 1:1:4 mixture of TFA and water in THF in an ice-cold bath for 5 hours, affording 92 in $51 \%$ yield. ${ }^{300}$


Scheme 4.15: Attempted synthesis of 8CldA ProTides from 5'-protected 8CldA. Reagents and conditions: (a.) TFA, water, THF, (1:1:4 v/v/v), $0{ }^{\circ} \mathrm{C}, 5 \mathrm{~h}, 51 \%$; (b.) conditions reported in Table; (c.) TFA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$.

The 3 '-silyl protected nucleoside ( $\mathbf{9 2}$ ) was then reacted with 3 equivalents of 1-phenyl-(benzyloxy-glycinyl)] phosphorochloridate and 3 equivalents of $t \mathrm{BuMgCl}$ ( 1 M in THF) in anhydrous THF (Table 4.22, entry 1). The mixture turned cloudy after 5 hours and after TLC analysis of the crude it was clear that the starting material had decomposed with no trace of product.

| Entry | Phosphorochloridate |  |  |  |  | time | T | $\begin{gathered} \text { tBuMgCl } \\ \text { eq. } \end{gathered}$ | $\begin{gathered} \hline \text { NMI } \\ \text { eq. } \end{gathered}$ | 5'-ProTide <br> yield |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | Ar | R | $\mathbf{R}^{1}$ | AA | Eq. |  |  |  |  |  |
| 1 | Ph | Bn | H | Gly | 3 | 5h | rt | 3 | - | - |
| 2 | Nap | nPen | $i \mathrm{Bu}$ | L- | 3 | 16h | $-10{ }^{\circ} \mathrm{C}$ to | 1 | - | - |
| 3 | Nap | nPen | $i \mathrm{Bu}$ | L- | 3 | 16h | rt | - | 5 | - |

Table 4.22: Structures and number of equivalent of phosphorochloridates, number of equivalents of $t \mathrm{BuMgCl}$ or NMI used in the attempt of synthesising derivatives 93a-c.

The same reaction was attempted again using 3 equivalents of 1-naphthyl-(pentyloxy-Lleucinyl)] phosphorochloridate and only one equivalent of ${ }^{t} \mathrm{BuMgCl}$ and adding the reagents to the $3^{\prime}$-protected starting material in THF at $-10^{\circ} \mathrm{C}$ in order to slow down any decomposition process, possibly due to the harsh alkaline conditions caused by the Grignard base. TLC analysis of the mixture revealed no conversion of the starting material after 2 hours at low temperature, therefore the reaction temperature was allowed to reach
room temperature. After overnight stirring, however, starting material was decomposed and no product could be detected (Table 4.22, entry 2).

When the 3 '-protected starting material was reacted with 3 equivalents of 1 -naphthyl-(pentyloxy-L-leucinyl)] phosphorochloridate and 5 equivalent of $N$-methylimidazole in THF overnight, TLC analysis of the crude mixture revealed complete lack of reactivity of the starting material, which was recovered (Table 4.22, entry 3).
In conclusion, attempts to synthesise $5^{\prime}$-ProTides of 8 CldA using the $t \mathrm{BuMgCl}$ methodology failed on both the unprotected nucleoside and the 3 '-silyl protected one. Although in low yields, the NMI method afforded two desired products.

### 4.3.3.1 Schwartz's reagent as dephosphorylating agent

The Schwartz's reagent is an organometallic compound corresponding to the structure of chlorobis(cyclopentadienyl)hydrido-zirconium also known as zirconocene hydrochloride.


Figure 4.28: Structure of the Schwartz's reagent.

Among the many applications of this reagent there is the chemoselective reduction of tertiary and secondary amidic bonds, ${ }^{365-367}$ which was investigated on a broad selection of differently protected $N$-acetylated nucleosides, ${ }^{368}$ confirming that this reagent is selective for the cleavage of $N$-acetyl groups with no effect on other protecting moieties. During these investigations, this methodology was also applied to one example of $N$-acetyl- $3^{\prime}, 5^{\prime}$ -bis-ProTide ( $N$-acetyl-2'-deoxycitidine-3', ${ }^{\prime}$ '-[1-naphthyl-(cyclohexyloxy-L-alaninyl)] diphosphate) and the main product of this reaction was the N-deacetylated 2'-deoxycitidine-5'-[1-naphthyl-(cyclohexyloxy-L-alaninyl)] phosphate. This serendipitous discovery suggested that the Schwartz's reagent could cleave preferentially the phosphoramidate group on the secondary hydroxyl moiety of 2'-deoxycytidine and not the primary one.
Intrigued by these results and considering the previously synthesised $3^{\prime}, 5^{\prime}$-bisProTides, obtained as undesired products via application of the $t \mathrm{BuMgCl}$ method on unprotected 8 CldA , attempt to use of the Schwartz's reagent on such substrates as selective dephosphorylating agent was carried out.


Scheme 4.16: Synthesis of 8CldA 5'-phosphoramidates (95a-c) and 8CldA 3'-phosphoramidate (96) by dephosphorylation of $3^{\prime}, 5^{\prime}$-bisProTides 95a,b,d. Reagents and conditions: Schwartz reagent (3 eq), THF, 16h, rt.

|  | Starting material |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Entry | $\mathbf{3}^{\prime}, \mathbf{5}^{\prime}$-BisProTide |  |  |  |  |  |  |  |  | Product(s) |  |
|  | Cpnd | Ar | $\mathbf{R}$ | $\mathbf{R}^{1}$ | AA | (yield) | (yield) |  |  |  |  |
| $\mathbf{1}$ | $\mathbf{9 0 a}$ | Nap | $n$ Pen | $i \mathrm{Bu}$ | L-Leu | $\mathbf{9 5 a}(3 \%)$ | traces |  |  |  |  |
| $\mathbf{2}$ | $\mathbf{9 0 b}$ | Ph | $n \mathrm{Hex}$ | Me | L-Ala | $\mathbf{9 5 c}(15 \%)$ | traces |  |  |  |  |
| $\mathbf{3}$ | $\mathbf{9 0 c}$ | Ph | Bn | Me | L-Ala | $\mathbf{9 5 b}(15 \%)$ | $\mathbf{9 6}(5 \%)$ |  |  |  |  |

Table 4.23: Structures of starting material $3^{\prime}, 5^{\prime}$ 'bisProTides 90a-c and yields of 5' and/or 3' ProTides obtained.

For this purpose, previously synthesised compound 90a was treated with 3 equivalents of Schwartz's reagent in THF for 3 hours at room temperature, yielding only $3 \%$ of 5 'ProTide 95a and traces of 3'-phosphoramidate, due to the low amount of starting material used (Table 4.23, entry 1). A similar yield was obtained when repeating the strategy on substrate 90b, although due to the low amount of material available, only 5'phosphoramidate 95b could be isolated (Table 4.23, entry 2).
On the other hand, when a larger amount of starting material $90 \mathbf{c}$ was treated in the same conditions, the reaction yielded $15 \%$ of 3 '-deophosphorylated derivative $\mathbf{9 5 c}$ and $5 \%$ of $5^{\prime}$ dephosphorylated derivative 96 (Table 4.23 , entry 2 ).
In conclusion, the Schwartz's reagent proved to be a useful reagent for the dephosphorylation of bis-ProTides of both 2'-deoxycytidine and 8CldA and investigations to understand the mechanism of such reaction and to potentially improve this methodology are ongoing.

### 4.3.4 Cytotoxic in vitro screening of 8CldA prodrugs

The 8CldA derivatives selected for cytotoxic evaluation by WuXi AppTech were the 3',5'bisProTides 90a and 90b, the 5'-ProTides 95a-c and the 3 '-regioisomer 96 (Table 4.24).


| Cpnd | regioisomer | Ar | R | $\mathrm{R}_{1}$ | AA | cLogP |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 95a | 5 | Nap | $n$ Pen | $i \mathrm{Bu}$ | L-Leu | 4.76 |
| 95b | 5 | Ph | $n \mathrm{Hex}$ | Me | L-Ala | 2.13 |
| 95 c | 5 | Ph | Bn | Me | L-Ala | 1.73 |
| 90 a | 3',5' | Nap | $n$ Pen | $i \mathrm{Bu}$ | L-Leu | 11.09 |
| 90c | 3',5' | Ph | Bn | Me | L-Ala | 6.09 |
| 96 | 3 ' | Ph | Bn | Me | L-Ala | 1.91 |

Table 4.24: Structures and cLogP of 8 CldA derivatives evaluated for their cytotoxic activity. cLogP calculated with ChemDraw 2010 software.

These derivatives were tested on a small selection of cell lines deriving from haematological malignancies and breast adenocarcinoma (Table 4.25).

| Cell line | Malignancy |
| :---: | :---: |
| HEL92.1.7 | Erythroleukaemia |
| HL-60 | Promyelocytic leukaemia |
| RPMI8226 | Human multiple myeloma |
| Z-138 | Mantle cell lymphoma |
| MCF-7 | Breast adenocarcinoma |
| MIA PaCa-2 | Pancreatic adenocarcinoma |

Table 4.25: Selection of cell lines used for cytotoxic evaluation of 8CldA derivatives.

8CldA was poorly active on most of the cell lines, reaching maximum inhibitions of cell viability ( $\mathrm{MI} \%$ ) ranging between 11.6 and $22.6 \%$ at the highest concentration considered $(198 \mu \mathrm{M})$. However, on the pancreatic adenocarcinoma cell line MIA PaCa-2, it reached a maximum inhibitory effect of $86.1 \%$ and the $\mathrm{IC}_{50}$ was $1.16 \mu \mathrm{M}$ (Table 4.26, Figure 4.29). This result suggests that the cytotoxic activity of this nucleoside analogue should be evaluated on a broader selection of solid tumour cell lines.

| Cpnd | HEL92.1.7 |  | HL-60 |  | RPMI-8226 |  | Z138 |  | MCF-7 |  | MIA PaCa-2 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathbf{I C}_{\mathbf{5 0}}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{\mathbf{5 0}}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{\mathbf{5 0}}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% |
| 8CldA | >198 | 16 | >198 | 23 | >198 | 12 | >198 | 22 | >198 | 12 | 1.16 | 86 |
| 95 c | >198 | 34 | >198 | 20 | 105 | 63 | >198 | -8 | >198 | 53 | 5.22 | 90 |
| 96 | 78 | 99 | 63 | 100 | 49 | 107 | >198 | 33 | 46 | 106 | 14 | 100 |
| 90c | >198 | 19 | >198 | 53 | 129 | 54 | >198 | 48 | >198 | 45 | 14 | 88 |
| 95a | 9.08 | 99 | 6.85 | 99 | 7.36 | 100 | 22 | 104 | 9.67 | 99 | 2.6 | 95 |
| 90 a | >198 | 47 | 29 | 51 | 11 | 79 | >198 | 61 | 18 | 66 | 4.65 | 100 |
| 95b | 109 | 70 | 92 | 86 | 80 | 91 | >198 | 33 | 102 | 68 | 16 | 84 |
| PTX | 0.02 | 83 | 0.003 | 85 | 0.003 | 91 | 0.002 | 99 | 0.01 | 65 | 0.002 | 87 |

Table 4.26: Biological evaluation of 8 CldA and relative prodrugs by WuXi app. tech. Cytotoxicity is reported as $\mathrm{IC}_{50} \mu \mathrm{M}$ values ( $50 \%$ inhibitory concentration of cell viability). MI $\%$ is the maximum percentage of inhibition exerted by compounds up to the top dose ( $198 \mu \mathrm{M}$ ).

Within the family of L-alanine benzyl ester phenyl phosphoroamidates, the 3 '-regioisomer (96) was generally the most active derivative, although not a very potent agent (Table 4.26, Figure 4.29). Although reaching complete inhibition of almost all the cell lines considered, the $\mathrm{IC}_{50}$ values ranged between 14 and $>198 \mu \mathrm{M}$. On the other hand, the 5 '-analogue ( $\mathbf{9 5 c}$ ) was less active on all cell lines apart from MIA PaCa-2 cell line, where it reached $\mathrm{IC}_{50}$ value of $5.22 \mu \mathrm{M}$. The $3^{\prime}, 5^{\prime}$-bisProTide 90 c was generally poorly active, although reaching better cell viability inhibition percentages compared to the parent nucleoside. The $3^{\prime}, 5^{\prime}$ 'and 5'-phosphoroamidates bearing L-leucine pentyl naphthyl moieties (respectively compounds 90a and 95a), were more active than the previously considered ones. ProTide 95a reached complete inhibition of cell viability on all cells and $\mathrm{IC}_{50}$ values ranging between 2.6 and $22 \mu \mathrm{M}$.


Figure 4.29: Comparison between cytotoxic data of 8 ClA and relative prodrugs (phenyl L-alanine benzyl ester derivatives $95 \mathbf{5}, 96$ and 90 c, napthyl L-leucine $n$-pentyl derivatives 95 a and 90a and phenyl L-alanine $n$-hexyl analogue 95b).

The $3^{\prime}, 5^{\prime}$ '-bisProTide derivative $\mathbf{9 0 a}$ was less active than $\mathbf{9 5 a}$ but still more active than the parent nucleoside, reaching $\mathrm{IC}_{50}$ values as low as $11 \mu \mathrm{M}$ on cell lines where the parent nucleoside was only partially active. The L-alanine pentyl phenyl derivative $\mathbf{9 5 c}$ was not as active as the other 5'-ProTides but still reached better results than the parent nucleoside on all cell lines but MIA PaCa-2 (Table 4.26, Figure 4.29).
Rationalising these data was easier when considering the calculated LogP, describing the lipophilicity of these compounds. Within the 5'-ProTides (95a-c), the most lipophilic derivative $\mathbf{9 5 a}$ was the most active, followed by $\mathbf{9 5 b}$ and $\mathbf{9 5}$ c, whose activity decreased with the lipophilic character. Similar results were also obtained in the $3^{\prime}, 5^{\prime}$-series, with the very lipophilic analogue 90 a being more active than $90 \mathbf{b}$. The superior activity of the 3 'phosphoramidate (96) over the $5^{\prime}$, analogue ( $\mathbf{( 9 5 b}$ ) could also be linked to a slightly higher lipophilicity. These results suggest that a main reason for improved activity of the 8CldA prodrug over the parent nucleoside may be the potential to cross the membrane by passive diffusion due to increased lipophilicity, therefore bypassing transport through nucleoside transporter. On the other hand, the higher activity of the 3 '-ProTide (96) over the 5 'Protide (95b) suggest that these compounds act as a prodrug of 8CldA more than 8CldAMP, therefore they all tend to release 8 ClA in the intracellular compartment and the better results are obtained only due to improved penetration in the cell.

### 4.3.5 Mechanistic investigations

### 4.3.5.1 Docking in CPY active site

The three most active 8CldA derivatives, 3'-phosphoramidate 96, and the two ProTides 95a and 95b were docked within the binding site of CPY in order to predict the likelihood of intracellular ester cleavage. The puzzling higher cytotoxicity of compound $\mathbf{9 6}$ compared to the 5 '-regioisomer $\mathbf{9 5 b}$ on most of the cell lines prompted us to investigate whether it could be related to an enzymatic mechanism. Compound 96 could release 3'-8CldAMP which could act with an unknown mechanism of action. Otherwise, thus released 3'8 CldAMP could be finally hydrolysed to 8 CldA , and be more active than the parent nucleoside because of an easier intracellular uptake by passive diffusion.



$$
95 \mathrm{a} \mathrm{Ar}=\mathrm{Nap}, \mathrm{R}=n \text {-Pen, } \mathrm{R}^{1}=\mathrm{Bu}, \mathrm{O}^{5}
$$

$$
95 \mathrm{a}^{\mathrm{A}} \mathrm{Ar}=\mathrm{Nap}, \mathrm{R}^{1}=\mathrm{Bu}, \mathrm{O}^{5^{\prime}}
$$

$$
95 c^{A} \mathrm{Ar}=\mathrm{Ph}, \mathrm{R}^{1}=\mathrm{Me}^{1} \mathrm{O}^{5^{\prime}}
$$

$95 a^{B} R^{1}=1 B u, O^{5}$
95c $\mathrm{Ar}=\mathrm{Ph}, \mathrm{R}=\mathrm{Bn}, \mathrm{R}^{1}=\mathrm{Me} \mathrm{O}^{5}$ $96^{A} \mathrm{Ar}=\mathrm{Ph}, \mathrm{R}^{1}=\mathrm{Me} \mathrm{O}^{3^{\prime}}$
$96 \mathrm{Ar}=\mathrm{Ph}, \mathrm{R}=\mathrm{Bn}, \mathrm{R}^{1}=\mathrm{Me} \mathrm{O}^{3}$
$95 \mathrm{c}^{\mathrm{B}} \mathrm{R}^{1}=\mathrm{Me}, \mathrm{O}^{5}$
$96^{B} R^{1}=\mathrm{Me}, \mathrm{O}^{3}$


$$
\begin{array}{cc}
95 a^{C} R^{1}=1 B u, O^{5} & 95 a^{D}=95 c^{C}=8 C I d A-5^{\prime} M P \\
95 c^{C} R^{1}=M e, O^{5} & 96^{C} 8 \mathrm{CldA}-3^{\prime} M P \\
96^{C} R^{1}=\mathrm{Me}, \mathrm{O}^{3} &
\end{array}
$$

Figure 4.30: Putative intracellular metabolic activation of 8 CldA prodrugs $\mathbf{9 5 a}, \mathbf{9 5 c}$ and $\mathbf{9 6}$ by CPY/Hint enzymes.

In comparison to both diasteroisomers of $\mathbf{9 5 c}$ (Figure 4.30), the positioning of the regioisomer 96 in the binding site of CPY (Table 4.27) is significantly closer to the crucial residues represented by Gly52, Gly53 and Ser146, for both diasteroisomer. This suggests that $\mathbf{9 6}$ may be processed quicker to the 8 CldA -3'-L-alaninyl monoester phosphate $\left(\mathbf{9 6}^{\mathbf{A}}\right.$, Figure 4.30 ), and this could potentially be due to a greater flexibility of the molecule, responsible for the better accomodation of the amino acid ester moiety inside the pocket.

|  |  |
| :---: | :---: |
| 96 (Sp diasteroisomer) | 96 (Rp diasteroisomer) |
|  |  |
| 95c (Sp diasteroisomer) | 95c (Rp diasteroisomer) |
|  |  |
| 95a (Sp diasteroisomer) | 95a (Rp diasteroisomer) |

Table 4.27: Docking interactions of compounds $\mathbf{9 5 a}, \mathbf{9 5 b}$ and $\mathbf{9 6}$ in the active site of CPY enzyme.

Similarly, compound 95a, seems to position the amino acid ester moiety close to the crucial residues, and this suggests a very quick processing, despite the very hindered structure.

### 4.3.5.2 Docking in Hint active site

The amino acid phosphate monoester intermediates resulting from the first enzymatic processing of compounds $\mathbf{9 5 a}, \mathbf{9 5 c}$ and 96 (shown in Figure 4.30) were docked in the active pocket of the Hint enzyme (Table 4.28). Intermediates $\mathbf{9 5 c}{ }^{\mathbf{C}}$ and $\mathbf{9 6}^{\mathbf{C}}$ were characterised by a very similar arrangement within the enzyme, $\mathbf{9 6}^{\mathrm{C}}$ accommodating the phosphoramidate moiety slightly further from the catalytic residues. Very close positioning to the key residues was also observed for intermediate $\mathbf{9 5 a}{ }^{\mathbf{C}}$.


Table 4.28: Docking interactions of compounds $\mathbf{9 5 a}{ }^{\mathbf{C}}, \mathbf{9 5 c}{ }^{\mathbf{C}}$ and $\mathbf{9 6}^{\mathbf{C}}$ in the active site of Hint enzyme.

Molecular modelling studies suggest that the higher activity of the 3 '-regioisomer $\mathbf{9 6}^{\mathbf{C}}$ over $\mathbf{9 5 c}{ }^{\mathbf{C}}$ could derive from a better processing by CPY. Moreover, slightly higher lipophilicity of $\mathbf{9 6}^{\mathbf{C}}$ (Table 4.28) could account for better cell penetration by passive diffusion, in comparison to $\mathbf{9 5 c}{ }^{\mathbf{C}}$, hence improved activity. On the other hand, the better cytotoxic profile of compound $\mathbf{9 5 a}{ }^{\mathbf{C}}$ over the other 8 CldA derivatives could derive from both high lipophilicity and good intracellular processing.

### 4.3.6 Conclusions

8CldA was synthesised in order to explore in vitro the reported distrupting potential on DNA chain growth, potentially leading to cell death. ProTides of 8CldA were synthesised in low yields either by the NMI coupling method or by first synthesis of $3^{\prime}, 5^{\prime}$ '-bisProTides followed by 3'-dephosphorylation by the Schwartz's reagent. Investigations are currently underway to explain the mechanism of this previously unreported reaction. Biological evaluation of 8 CldA derivatives and parent nucleoside on a small selection of cancer cell lines showed a generally poor cytotoxic profile in relation to 8 CldA . The nucleoside analogue was active at low micromolar concentration only on the MIA $\mathrm{PaCa}-2$ cell line. Tested $3^{\prime}, 5$ '-bisProTides were generally poorly active, while the L-leucine 5 '-ProTide 95a was the most potent compound within the family, with consistently low micromolar activity on all cell lines, bringing a substantial advantage over the parent nucleoside in terms of anticancer potency. Surprisingly, L-alanine ProTide 95c was poorly active, while the 3 '-regioisomer 96 showed improved activity. Molecular modelling studies suggested that this could derive from a quicker intracellular processing by carboxypeptidase enzyme, coupled with an efficient phosphoramidase step.
More derivatives bearing linear amino acid esters and lipophilic amino acids could be synthesised in order to broaden the structure activity relationships (SAR) relative to this family of compounds, with the aim of identifying more active compounds. Moreover, 3'regioisomers should be synthesised and investigated further, based on the improvement in activity yielded by compound $\mathbf{9 6}$ over $\mathbf{9 5 c}$. The enhanced activity of these unusual 3 'regioisomers merits further study.

## 5 CNDAC

### 5.1 Background

2'-C-cyano-2'-deoxy-1- $\beta$-D-arabino-pentofuranosylcytosine (CNDAC, 97, Figure 5.1) is an analogue of the endogenous nucleoside $2^{\prime}$-deoxycytidine. The deoxyriboside moiety has been replaced by an arabinoside group, bearing a cyano ( CN ) group in the $2^{\prime}$ position. ${ }^{369}$


CNDAC
97


Ara-C
1 g

Figure 5.1 Structures of CNDAC (97) and Ara-C (1g).

CNDAC is therefore the $2^{\prime}$-cyano derivative of the anti-leukaemic agent Ara-C (1g, Figure 5.1), and it was developed in 1991 by Matsuda et. al. as an analogue of Ara-C that could, in principle, offer advantage over the parent nucleoside. ${ }^{369}$ Ara-C was the first anticancer nucleoside approved by the FDA, in 1969, for the treatment of acute myelogenous leukaemia. ${ }^{18}$ Despite its potent activity on haematological malignancies, Ara-C lacks activity towards solid cancers. Moreover, it is a good substrate for cytidine deaminase (CDA), an enzyme that converts it to its inactive uracil derivative Ara-U, ${ }^{370}$ decreasing the plasma half-life of Ara-C and its activity (Scheme 5.1). ${ }^{371}$


Scheme 5.1: Deactivation of Ara-C to Ara-U catalysed by CDA.

CNDAC was synthesised by Matsuda et al. along with other 2'-modified Ara-C analogues, in an attempt to induce resistance to CDA-mediated inactivation and to broaden the spectra of anticancer activity to solid tumours. ${ }^{369,372}$

This nucleoside analogue resulted as one of the most promising analogues, improving both in vitro and in vivo activity compared to Ara-C.
After entering the cell through a nucleoside transporter recognised as the human equilibrative nucleoside transporter 1 (hENT1), ${ }^{373}$ this nucleoside is phosphorylated by deoxycytidine kinase (dCK) to CNDAC-monophosphate (CNDAC-MP) and finally to CNDAC-triphosphate (CNDAC-TP) by two sequential phosphorylation steps (Scheme 5.2). ${ }^{374}$ Although being a poorer substrate for cytidine deaminase, compared to Ara-C, CNDAC activity is still partially affected by CDA levels, causing the conversion into $2^{\prime}$ - ${ }^{-}$-cyano-2'-deoxy-1- $\beta$-D-arabino-pentofuranosyluracil (CNDAU). ${ }^{371}$


Scheme 5.2: CNDAC intracellular metabolism and mode of action. P: phosphate group.

Once in the triphosphate form, CNDAC is incorporated into the DNA growing chain by DNA polymerase $\alpha$, which catalyses the addition of an additional endogenous nucleotide (Scheme 5.2) to the DNA growing chain. ${ }^{375,376}$ This process causes chemical instability of DNA. In fact, once CNDAC is integrated into the DNA in the $3^{\prime}-5$ ' phosphodiester linkage, upon the addition of a subsequent nucleotide (dNTP), the electron-withdrawing effect of the CN group at the arabinose $2^{\prime}-\beta$ position increases the acidity of the $2^{\prime}-\alpha$
proton. This in turn facilitates a $\beta$-elimination reaction, leading to single strand breakage. ${ }^{375,376}$

Sapacitabine (CYC-682, Figure 5.2) is a CNDAC derivative, bearing a $N^{4}$-palmitoyl chain on the amino group of the base. ${ }^{377}$ This compound was developed as a prodrug that could bypass plasma deamination by CDA due to the presence of a bulky alkyl chain on the amino group of its structure.


Figure 5.2: Structure of sapacitabine (CYC682, 98).

Moreover, the increased lipophilicity of the molecule enhances intestinal permeability of the drug, allowing oral administration. Once in the blood stream, sapacitabine is activated by plasma and liver amidase enzymes, releasing CNDAC. Oral administration of sapacitabine showed increased activity and lower toxicity compared to CNDAC in a mouse model implanted with reticulum cell sarcoma. ${ }^{376}$ Several Phase I and II clinical trials of sapacitabine alone or in combination with other anticancer drugs are underway in patients with chronic lymphocytic leukaemia, acute myeloid leukaemia (AML), non-smallcell lung cancer and other advanced solid tumours, and a Phase III study of sapacitabine has been initiated in elderly patients with newly diagnosed AML. ${ }^{378}$
In this work, the application of the ProTide approach to CNDAC was carried out in order to analyse a different prodrug moiety that could potentially increase the activity of this nucleoside, and offer advantage over sapacitabine. In fact, sapacitabine is a prodrug that increases liphophilicity and deaminase stability of CNDAC, but does not bypass other important limiting steps, such as the first phosphorylation step by deoxycytidine kinase. In fact, CNDAC was reported to be a worse substrate of dCK compared to the natural substrate dCyd. ${ }^{374,376}$ On the other hand, ProTides of CNDAC could potentially bypass the dependency on dCK for the first phosphorylation step and release CNDAC-MP inside the cell, along with increasing the lipophilicity of CNDAC and the stability towards deamination, ${ }^{151}$ therefore representing a potential advantage over sapacitabine. The synthesis of a family of CNDAC ProTides was therefore performed, along with the
required 8 -step synthesis of the parent nucleoside, which is not commercially available. Moreover, to have a comparison with the prodrug in clinical development, sapacitabine was synthesised and tested in vitro along with CNDAC and CNDAC ProTides.

### 5.2 Synthesis of CNDAC

The CNDAC synthetic strategy was performed as reported in the literature, ${ }^{369,370,379}$ with some optimisation steps (Scheme 5.3).


Scheme 5.3: Synthesis of CNDAC. Reagents and conditions: (a.) TIPDSCl (1.1 eq), pyridine, rt, 16h, 57\%; (b.) $\mathrm{Ac}_{2} \mathrm{O}$ (1.2 eq), EtOH , reflux, $3.5 \mathrm{~h}, 100 \%$; (c.) DMP (1 eq), $\mathrm{CH}_{2} \mathrm{Cl}_{2}, 150{ }^{\circ} \mathrm{C}$ to rt, 16 h ; (d.) TMSCN (3 eq), $\mathrm{AlCl}_{3}(1.5 \mathrm{eq}), \mathrm{CH}_{2} \mathrm{Cl}_{2}, 0{ }^{\circ} \mathrm{C}$ to rt, 1.5 h ; (e.) $\mathrm{EtS}_{2} \mathrm{CCl}$ (1.2 eq), $\operatorname{DMAP}(0.4 \mathrm{eq}), \mathrm{Et}_{3} \mathrm{~N}(1.5 \mathrm{eq}), \mathrm{CH}_{2} \mathrm{Cl}_{2}, 0$ ${ }^{\circ} \mathrm{C}$ to rt, 2 h ; (f.) $2,4,6$-collidine ( 1.2 eq ), IPA, DLP ( 0.7 eq ), $100{ }^{\circ} \mathrm{C}, 2 \mathrm{~h}, 60 \%$ ( 4 steps); (g.) Schwartz reagent (3 eq), THF, rt, 3h, $75 \%$; (h.) [1] HF-pyr ( $70 \%$, 4 eq), pyr, EtOAc, $0^{\circ} \mathrm{C}$, 2 hours, $45 \%$; [2] TFA: $\mathrm{H}_{2} \mathrm{O}:$ THF ( $1: 1: 4 \mathrm{v} / \mathrm{v} / \mathrm{v}$ ), $0^{\circ} \mathrm{C}$ to $\mathrm{rt}, 48 \mathrm{~h}, 75 \%$.

Commercially available cytidine (99) was protected in the first step by treatment with 1,3-dichloro-1,1,4,4-tetraisopropyldisiloxane (TIPSCl, Markiewicz reagent) in pyridine (pyr) at room temperature overnight, to afford intermediate $\mathbf{1 0 0}$ in $57 \%$ yield. ${ }^{380-382}$ Compound $\mathbf{1 0 0}$ was then protected by $N$-acetylation of the amino group on the base by treatment with
acetic anhydride $\left(\mathrm{Ac}_{2} \mathrm{O}\right)$ in ethanol $(\mathrm{EtOH})$ under reflux over a period of 3.5 hours. $\mathrm{Ac}_{2} \mathrm{O}$ was added in two portions to push the reaction to completion, affording compound $\mathbf{1 0 1}$ in quantitative yield. ${ }^{379}$
The free 2'-hydroxyl group in intermediate $\mathbf{1 0 1}$ was oxidised to a ketone by treatment with Dess-Martin periodinane (DMP) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The reagent (2 eq) was added at $10{ }^{\circ} \mathrm{C}$ and the mixture was stirred at room temperature overnight and finally quenched with a solution of sodium bicarbonate $\left(\mathrm{NaHCO}_{3}\right)$ and sodium thiosulfate $\left(\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3} \cdot 5 \mathrm{H}_{2} \mathrm{O}\right)$; extraction of the mixture with diethyl ether yielded a crude mixture containing only desired compound $\mathbf{1 0 2}$ as confirmed by ES-MS and ${ }^{1} \mathrm{H}$-NMR analysis of the crude, without any trace of starting material 101. This crude was not purified further but used directly for the next step. ${ }^{379,383,384}$ The ketone group in $\mathbf{1 0 2}$ was converted into a cyanohydrin moiety by treatment with aluminium chloride $\left(\mathrm{AlCl}_{3}\right)$ and trimethylsilyl cyanide (TMSCN) that were added carefully to a solution of $\mathbf{1 0 2}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ in an ice-cold bath under anhydrous conditions. The mixture was stirred in ice for 45 minutes and then at room temperature for another 45 minutes, and was then quenched with ammonium chloride $\left(\mathrm{NH}_{4} \mathrm{Cl}\right)$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. All publications describing the product of this reaction state that it yields a mixture of 2'-cyanohydrins ( $\mathbf{1 0 3 a}$ and $\mathbf{1 0 3 b}$ ) as reported in Figure 9.3. ${ }^{369,370,379}$


103a


103b

Figure 5.3 Structure of the two 2'-cyanohydrins 103a and $\mathbf{b}$.

However the ratio between the two diasteroisomers is never clearly specified. In our hands, the product of this reaction was not purified by chromatography, due to the well known instability of the 2 '-cyanohydrin moiety, that easily reverts into the original ketone functionality (starting material 102). ${ }^{385}$ Analysis of the crude mixture by ${ }^{1} \mathrm{H}-\mathrm{NMR}$ suggested the presence of only one species, which could potentially be represented by isomer 103b. In fact the steric hindrance on the $\beta$-face would interfere with the attack of the cyano anion from the $\beta$-face, leading to the synthesis of $\mathbf{1 0 3 b}$ as a main product. This
would be in accordance with the reported synthesis of 3'-cyanohydrins of different pyrimidine and purine nucleosides via similar oxidation and cyano addition reaction. ${ }^{385}$

Different substituents could be used to protect the $2^{\prime}$-hydroxyl group of the 2 '-cyanohydrin 103a,b in order to generate a good leaving group for the following elimination reaction. ${ }^{379}$ The initial attempt involved the use of commercially available phenyl chlorothionocarbonate $(\mathrm{PhOCSCl})^{206}$ in acetonitrile $\left(\mathrm{CH}_{3} \mathrm{CN}\right)$ in the presence of DMAP as a catalyst and triethylamine (TEA) as a base. ${ }^{369,379}$ This reaction led to decomposition of the starting material without isolation of the desired product 104 (Scheme 9.4).


Scheme 5.4: Attempted synthesis of thiocarbonate derivatives 104. Reagents and conditions: PhOCSCl ( 1.5 eq), $\mathrm{Et}_{3} \mathrm{~N}(1.5 \mathrm{eq})$, DMAP ( 0.5 eq ), $\mathrm{CH}_{3} \mathrm{CN}, 0^{\circ} \mathrm{C}, 2 \mathrm{~h}$.

A second strategy was attempted, involving the use of ethyl chlorodithioformate (EtSCSCl, 109), ${ }^{379}$ a less hindered reagent which is not commercially available and was synthesised according to Godt et al. as reported in Scheme 5.5. ${ }^{386}$


Scheme 5.5: Synthesis of ethyl chlorodithioformate (EtSCSCl, 109). Reagents and conditions: thiophosgene ( 0.0015 eq ), toluene, $10^{\circ} \mathrm{C}, 16 \mathrm{~h}$.

Ethanethiol (EtSH) was dissolved in toluene (a less toxic alternative solvent to the reported benzene) and added to a solution of thiophosgene $\left(\mathrm{CSCl}_{2}\right)$ in toluene at $10^{\circ} \mathrm{C}$. the mixture was stirred overnight at low temperature, then the solvent and side-products were removed under reduced pressure and desired EtSCSCl was isolated by distillaton as a dark orange oil. ${ }^{386}$

Freshly prepared 109 was then reacted with 103 to yield the thiocarbonyl intermediate 104. The crude was not purified further but used directly in the next step, involving deoxygenation with 2,4,6-collidine and dilauroyl peroxide (DLP) in isopropanol (IPA) at $100{ }^{\circ} \mathrm{C}$ for 2 hours. ${ }^{387,388}$ This radical step afforded 2'-cyano arabinosyl derivative $\mathbf{1 0 6}$ as
the only product. According to previous reports on similar reactions, the 2 '-tert radical formed from the deoxygenation reaction are in equilibrium between $\alpha(\mathbf{A})$ and $\beta(\mathbf{B})$ face, as shown in Scheme 5.6. ${ }^{369,370,372,389-391}$


Scheme 5.6: Equilibrium between position $\alpha(\mathbf{A})$ and $\beta(\mathbf{B})$ for the tert-radical formed after deoxygenation of 105 and reaction of $\alpha$-radical with isopropanol with formation of $\beta$-cyano derivative 106.

Since the $\beta$ face is more hindered than the $\alpha$-face, the radical $\mathbf{B}$ would react more favorably with isopropanol, affording $\mathbf{1 0 5}$ as 2 '-cyano arabinonucleoside. (Scheme 5.6). ${ }^{369,370,372,389-391}$


Figure 5.4: ${ }^{1} \mathrm{H}$ NMR spectrum of deoxygenated intermediate $\mathbf{1 0 6}$.

The ${ }^{1} \mathrm{H}$ NMR spectra of intermediate $\mathbf{1 0 6}$ reported in Figure 5.4 confirms the success of the deoxygenation, due to the presence of a doublet signal corresponding to the H 1 ' in the sugar moiety, due to coupling with $\mathrm{H} 2{ }^{\prime}\left({ }^{1} J=7.0 \mathrm{~Hz}\right)$. The signal corresponding to this same proton is a singlet in the previous intermediates $\mathbf{1 0 2}, \mathbf{1 0 3}$ and $\mathbf{1 0 5}$ due to the lack of a proton in position 2 '. The presence of only one product and the extensively documented
selectivity of the radical deoxygenation reaction strongly support that this product corresponds to the $2^{\prime}$-cyano arabino nucleoside 106.

The selective removal of the acetyl group on the base in compound $\mathbf{1 0 6}$ was performed in $75 \%$ yield by treatment with zirconocene hydrochloride (Schwartz's reagent) in tetrahydrofuran (THF) for 3 hours, according to methodology by Bhat et al. ${ }^{392}$ that had been previously applied with good results by our team on a broad selection of N -acetylated nucleoside analogues. ${ }^{368}$ This procedure was applied as an innovative and better yielding alternative to the reported treatment of intermediate $\mathbf{1 0 6}$ with a $3 \%$ solution of HCl in $\mathrm{CH}_{3} \mathrm{OH}$, which was described to afford desired 107 in $53 \%$ yield. ${ }^{369}$
The final step of the synthesis of CNDAC was the removal of the TIPDS protecting group. The reported procedure described the use of a 1 N solution of TBAF in THF in the presence of acetic acid, yielding deprotected 107 in $84 \%$ yield. In order to avoid possible problems related to the removal of hydrolysed TBAF residues, and to the known instability of CNDAC in basic conditions, ${ }^{369,393}$ different deprotective strategies were attempted. Consequently, the deprotection was tested first on a small amount of the substrate (106), by treatment with HF-pyridine (HF-pyr, 70\%) in pyridine and ethyl acetate (EtOAc) at 0 ${ }^{\circ} \mathrm{C}$ for 2 hours (Scheme 5.3). ${ }^{394}$ This method yielded free CNDAC in $45 \%$ yield.


Figure 5.5: ${ }^{1} \mathrm{H}$ NMR ( $\mathbf{A}, 500 \mathrm{MHz}$ ) and ${ }^{13} \mathrm{C}$ NMR (B, 125 MHz ) spectra of CNDAC (97) in $\mathrm{CD}_{3} \mathrm{OD}$.

The characteristic upfield signal corresponding to H 2 ' in the ${ }^{1} \mathrm{H}$ NMR (Figure 9.5) and the presence of a CN signal ( 142.73 ppm ) in the ${ }^{13} \mathrm{C}$ NMR (Figure 9.6) confirm the structure of the synthesised nucleoside. A different procedure that afforded CNDAC in better yield was carried out by treating protected 106 in a 1:1:4 mixture of trifluoroacetic acid (TFA),
water and tetrahydrofuran (THF) at $0^{\circ} \mathrm{C}$ and then stirring the mixture at rt (Scheme 5.3). The reaction was very slow but yielded $75 \%$ of fully deprotected CNDAC after 48 hours. ${ }^{300}$

A similar method was used to synthesise selectively deprotected intermediate $\mathbf{1 1 0}$ in position 5' in order to have a suitable precursor for the synthesis of CNDAC ProTides. This intermediate was allowed to react with the appropriate phosphorochloridate and $t \mathrm{BuMgCl}$ to yield $5^{\prime}$ '-phosphoramidate silylated intermediates. These structures finally underwent a deprotection step in order to afford the desired CNDAC ProTides and the steps to the synthesis of CNDAC ProTides briefly outlined here will be thoroughly explained in the next paragraphs. The selective deprotection of 107 was carried out by treatment with a 1:1:4 mixture of TFA: $\mathrm{H}_{2} \mathrm{O}: \mathrm{THF}$ at $0{ }^{\circ} \mathrm{C}$. The mixture was stirred for 2 hours and then stopped through evaporation of the volatiles to avoid complete deprotection of the nucleoside, yielding intermediate $\mathbf{1 1 0}$ in $62 \%$ yield, as confirmed by ${ }^{1} \mathrm{H}-\mathrm{NMR}$ and bidimensional COSY experiment. ${ }^{300}$


Scheme 5.7: Synthesis of 5'-deprotected intermediate 110. Reagents and conditions: TFA: $\mathrm{H}_{2} \mathrm{O}:$ THF (1:1:4 $\mathrm{v} / \mathrm{v} / \mathrm{v}$ ), $0^{\circ} \mathrm{C}, 2 \mathrm{~h}, 62 \%$.

### 5.3 Synthesis of sapacitabine (CYC682)

CNDAC was used as a starting material for the synthesis of sapacitabine, $N^{4}$-pamitoyl derivative involved in clinical studies.


Scheme 5.8: Synthesis of sapacitabine. Reagents and conditions: palmitic anhydride (1.7 eq), 1,4dioxane: $\mathrm{H}_{2} \mathrm{O}(20: 1), 8{ }^{\circ} \mathrm{C}, 4 \mathrm{~h}, 45 \%$.

The nucleoside was refluxed in a mixture of 1,4-dioxane and water (20:1) at $85^{\circ} \mathrm{C}$ in the presence of palmitic anhydride heating the mixture for 4 hours, and was converted into the desired $N^{4}$-protected nucleoside in $45 \%$ yield. ${ }^{379}$

### 5.4 Synthesis of CNDAC ProTides

As anticipated, $3^{\prime}$-TIPDS protected CNDAC (110) was selectively functionalised of the 5'hydroxyl group via treatment with an appropriate phosphorochloridate and $t \mathrm{BuMgCl}$ in THF, ${ }^{105}$ to afford the desired CNDAC ProTides (Scheme 5.9).


Scheme 5.9: Synthesis of CNDAC 3'-protected ProTides. Reagents and conditions: appropriate phosphorochloridate (3 eq), $\mathrm{tBuMgCl}(1.5 \mathrm{eq})$, anh. THF, rt, 16h.

The coupling step between 3'-protected nucleoside $\mathbf{1 1 0}$ and the appropriate phosphorochloridate afforded moderate to good yields (38-70\%) of the 3'-protected ProTides (111a-e, Table 5.1).

| Cpnd | Ar | $\mathbf{R}$ | $\mathbf{R}^{1}$ | AA | Yield |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 111a | Nap | $n$-Pen | $i \mathrm{Bu}$ | L-Leu | $60 \%$ |
| 111b | Ph | Bn | $\mathrm{CH}_{3}$ | $\mathrm{~L}-\mathrm{Ala}$ | $63 \%$ |
| 111c |  | Bn | $\mathrm{CH}_{3}$ | (L-Ala) | $70 \%$ |
| 111d | Nap | Bn | $\mathrm{CH}_{3}$ | L-Ala | $38 \%$ |
| 111e | Ph | $n$-Hex | $\mathrm{CH}_{3}$ | L-Ala | $44 \%$ |

Table 5.1: Structures and synthetic yield of 3'-TIPDS protected CNDAC ProTides (111a-e).

The first attempt of deprotection was carried out by treatment with TFA: $\mathrm{H}_{2} \mathrm{O}:$ THF (1:1:4) at room temperature (Scheme 5.10, [1]). The reaction was monitored over 6 hours and then stopped because only small traces of potential product 112a presented on TLC, along with degradation of the material into a baseline spot (Table 5.2, entry 1). This method was
therefore not considered suitable to our aims. Due to the small amount of starting material on which the reaction was attempted, the isolation of pure product was not successful.


Scheme 5.10: [1] TFA: $\mathrm{H}_{2} \mathrm{O}: T H F(1: 1: 4)$, rt, 6h. [2] TFA: $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ (1:1) $0{ }^{\circ} \mathrm{C}$ to rt, 8 h . [3] KF (6 eq), 18-crown-6 ( 0.3 eq ), THF, rt, 48h. [4] HF-pyr (9.2 eq), pyr (9.2 eq), EtOAc, $0^{\circ} \mathrm{C}, 16 \mathrm{~h}$.

| Entry | Starting material | Procedure | Yield |
| :---: | :---: | :---: | :---: |
| $\mathbf{1}$ | $\mathbf{1 1 1 a}$ | $[1]$ | Traces |
| $\mathbf{2}$ | $\mathbf{1 1 1 a}$ | $[2]$ | Traces |
| $\mathbf{3}$ | $\mathbf{1 1 1 b}$ | $[3]$ | $17 \%$ |
| $\mathbf{4}$ | $\mathbf{1 1 1 b}$ | $[4]$ | $30 \%$ |
| $\mathbf{5}$ | $\mathbf{1 1 1 \mathbf { c }}$ | $[4]$ | $11 \%$ |
| $\mathbf{6}$ | $\mathbf{1 1 1 d}$ | $[4]$ | $7 \%$ |
| $\mathbf{7}$ | $\mathbf{1 1 1 e}$ | $[4]$ | $5 \%$ |

Table 5.2: Structures of products from Scheme 5.10 and relative yields.

A different attempt of deprotection of 111a was tried by stirring the material in a 1:1 mixture of TFA in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at $0{ }^{\circ} \mathrm{C}$ for 2 hours and then allowing the reaction mixture to reach room temperature (Scheme 5.10, [2]). ${ }^{395}$ After 8 hours the mixture was evaporated and the unreacted material recovered (Table 5.2, entry 2); however due to the small amount it could not be subjected to any other deprotecting step.

Alternative deprotecting steps were further applied on the phenoxy L-alanine benzyl ester 111b. One portion was treated with potassium fluoride (KF) and a catalytic amount of 18-crown-6 ether in THF, at room temperature (Scheme 5.10, [3]). ${ }^{396}$ The reaction proved to be very slow with only traces of product detected by TLC and HPLC after 16 hours. However the reaction was stirred for an additional 32 hours, until HPLC analysis of the crude showed $59 \%$ conversion. Purification of the crude was tedious, due to the close retention factor ( Rf ) on silica gel chromatography between starting material 111b and product 112b and especially because of the low amount of starting material used for the
reaction, and finally the desired deprotected product was obtained in only $17 \%$ yield (Table 5.2, entry 3 ).
The other portion of $\mathbf{1 1 1 b}$ was submitted to an alternative deprotecting strategy, involving the treatment with HF-pyridine (HF-pyr) in ethyl acetate (EtOAc) at $0{ }^{\circ} \mathrm{C}$, in the presence of an additional amount of pyridine (Scheme 5.10, [4]). ${ }^{394}$ The reaction was monitored over time by HPLC and after 24 hours the conversion to desired product 112b was $63 \%$. The mixture was quenched by the addition of trimethylmethoxysilane (TMSOMe), evaporated and purified by column chromatography affording the desired product (112b) in $30 \%$ isolated yield (Table 5.2, entry 4).
Due to the higher yield obtained, this latter methodology was considered the most suitable to afford deprotected CNDAC ProTides, and was adopted in all other examples shown in Table 5.2 (entries 5-7).
Despite the moderate to good yields deriving from the coupling reaction between 3'protected CNDAC (110) and the appropriate phosphorochloridates, final yield of desired compounds (112a-e) was greatly affected by the deprotection step that was unusually difficult. This was believed to be due to the steric hindrance of the cyano group in the $\alpha$ position, as similar deprotecting attempts on different Ara-nucleosides were reported to lead to poor results. ${ }^{397}$ However an additional factor affecting the yield was the amount of starting material used, as the purification of small amounts of product was very low yielding when starting with amounts of $\mathbf{1 1 0}$ lower than 50 mg .
One attempt for the synthesis of phenyl (benzyloxy-L-alaninyl) CNDAC ProTide (112b) was also performed by treatment of the unprotected parent nucleoside (97) with phenyl (benzyloxy-L-alaninyl) phosphorochloridate and NMI in anhydrous THF. ${ }^{105}$ However after 16 hours, analysis of the crude reaction by TLC did not show any trace of product but only evident decomposition of the nucleoside, possibly due to the already known instability of CNDAC in basic conditions. ${ }^{369,393}$


Scheme 5.11: Attempted synthesis of CNDAC ProTide 112b via the NMI method on unprotected CNDAC (97). Reagents and conditions: phenyl (benzyloxy-L-alaninyl) phosphorochloridate (3 eq), NMI (5 eq), anh. THF, rt, 16h.

### 5.5 In vitro anticancer screening of CNDAC ProTides

CNDAC ProTides (112b-e) were tested by WuXi AppTech against a panel of cell lines and their cytotoxic activity was compared to the activity of the prepared parent nucleoside CNDAC (97) and synthesised sapacitabine (CYC682, 98), a CNDAC prodrug currently in clinical development. The selection of cell lines (Table 5.3) depended on the clinical development of sapacitabine against haematologic malignancies. ${ }^{398}$

| Cell line | Malignancy |
| :---: | :---: |
| CCRF-CEM | Acute lymphoblastic leukaemia |
| HEL92.1.7 | Erythroleukaemia |
| HS445 | Hodgkin's lymphoma |
| MOLT-4 | Acute T lymphoblastic leukaemia |
| RL | Non Hodgkin's B-lymphoma |
| Z-138 | Mantle cell lymphoma |
| HL-60 | Acute promyelocytic leukaemia |
| K562 | Chronic myelogenous leukaemia |
| MCF7 | Breast adenocarcinoma |
| KG-1 | Acute myelogenous leukaemia |

Table 5.3: Selected cell lines for in vitro cytotoxic screening of CNDAC derivatives by WuXi AppTech.

In these in vitro evaluations, CNDAC was very potent, especially on leukaemic cells CCRF-CEM, HEL92.1.7 and on the lymphoma cell lines Z138, with $\mathrm{IC}_{50}$ values often lower than the minimum concentration considered. CNDAC was also active at the submicromolar range on the solid cell line MCF7. On the other cells, the parent nucleoside did not perform as effectively, but with $\mathrm{IC}_{50}$ still in the low micromolar range. However, a complete lack in activity was observed on the acute myeloid leukaemia cell line MOLT-4, which was paralleled by similar results for sapacitabine (CYC682) and CNDAC proTides 112b-e.
Common structure

Table 5.4: CNDAC ProTides tested for in vitro cytotoxicity on selected cell lines by WuXi AppTech.

Despite being 2 to 10 fold more active than CNDAC ProTides, sapacitabine (CYC686) was less active than the parent nucleoside on all considered cell lines. CNDAC ProTides 112b-e were often more than ten fold less active than CNDAC.

| Cpnd | CCRFCEM |  | HEL92.1.7 |  | HS445 |  | MOLT-4 |  | RL |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathbf{I C}_{50}$ | MI\% | IC50 | MI\% | $\mathrm{IC}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | IC $\mathbf{5 0}$ | MI\% |
| CNDAC | $<0.02$ | 100 | 0.05 | 91 | 4.59 | 88 | >198 | -1 | 0.97 | 72 |
| CYC686 | 0.11 | 98 | 0.43 | 93 | 3.83 | 98 | >198 | -5 | 6.9 | 81 |
| 112b | 1.47 | 98 | 1.36 | 89 | 25.95 | 90 | >198 | 0 | 30.89 | 85 |
| 112c | 3.26 | 102 | 6.05 | 93 | 81.52 | 83 | >198 | -1 | 55.34 | 82 |
| 112d | 3.45 | 103 | 4.37 | 93 | 49.95 | 75 | >198 | -4 | 189.52 | 62 |
| 112e | 0.63 | 99 | 0.64 | 92 | 9.84 | 93 | >198 | -2 | 17.81 | 81 |
| PTX | 0.003 | 98 | 0.02 | 83 | 0.01 | 74 | 0.003 | 95 | 0.003 | 83 |

Table 5.5: In vitro cell viability screening of CNDAC, sapacitabine (CYC686) and CNDAC ProTides (112ae). Cytotoxicity data reported as $\mu \mathrm{M} \mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentration considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

The phenyl (hexyloxy-L-alaninyl) CNDAC ProTide 112e, although never as potent as the parent nucleoside, showed the best activity among this family of prodrugs, with $\mathrm{IC}_{50}$ values close to the clinically developed prodrug sapacitabine on cell lines CCRF-CEM, HEL92.1.7, Z138, and in the submicromolar range ( $0.63-0.78 \mu \mathrm{M}$ ).

| Cpnd | K562 |  | KG-1 |  | MCF7 |  | Z138 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | IC $\mathbf{5 0}$ | $\mathbf{M I}_{\%}$ | $\mathbf{I C}_{50}$ | MI\% |
| CNDAC | 3.59 | 63 | 6.84 | 90 | 0.15 | 92 | $<0.02$ | 100 |
| CYC686 | 16.71 | 66 | 35.45 | 97 | 0.72 | 84 | 0.64 | 100 |
| 112b | 39.8 | 82 | 86.95 | 87 | 14.29 | 94 | 2.39 | 100 |
| 112c | 78.3 | 66 | 70.02 | 78 | 17.45 | 95 | 3.36 | 101 |
| 112d | 83.28 | 59 | 67.65 | 76 | 8.9 | 70 | 2.01 | 100 |
| 112e | 26.82 | 91 | 92.51 | 68 | 7.1 | 93 | 0.78 | 100 |
| PTX | 0.01 | 91 | 0.08 | 88 | 0.01 | 65 | 0.002 | 99 |

Table 5.6 In vitro cell viability screening of CNDAC, sapacitabine (CYC686) and CNDAC ProTides (112ae). Cytotoxicity data reported as $\mu \mathrm{M} \mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentration considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

This result suggests that modification in the ProTides scaffold could be introduced, enlarging the number of ProTide bearing linear ester chains and also modifying the amino acid moiety, with the aim of identifying a suitable active alternative to sapacitabine.


Figure 5.6: Cytotoxic evaluation of CNDAC proTides 112b-e, compared to parent nucleoside and sapacitabine (CYC682). Molt4 cell line was not included as all compounds had $\mathrm{IC}_{50}>198 \mu \mathrm{M}$. ${ }^{*} \mathrm{IC}_{50}>198$ $\mu \mathrm{M}$.

Moreover, investigations on the lack of activity of some of the least active prodrugs could explain whether this is due to an ineffective processing of the ProTide moiety and therefore a lack of release of CNDAC-MP inside the cell.

### 5.6 Mechanistic investigations

Investigations on the enzymatic steps required for ProTides activation were performed in order to evaluate whether the reduced activity of CNDAC ProTides in comparison to the parent nucleoside could derive from inefficient intracellular processing. An enzymatic NMR assay was performed treating CNDAC ProTide 112b with carboxypeptidase Y (CPY). Moreover, the final product of CPY-catalysed activation, $\mathbf{1 1 2 b}^{\mathbf{C}}$, was docked into the active site of Hint enzyme, responsible for the final phosphoramidate bond cleavage, with release of CNDAC-monophosphate (112b ${ }^{\mathbf{D}}$ ). ${ }^{128}$


Scheme 5.12 Putative intracellular activation of CNDAC ProTide 112b catalysed by CPY/Hint enzymes.

### 5.6.1 CPY enzymatic assay

CNDAC ProTide 112b was treated with CPY, and Figure 5.7 shows the spectra obtained after the overnight enzymatic processing. After 7 minutes from the enzyme addition, one more downfield peak corresponding to intermediate $\mathbf{1 1 2 b}^{\mathbf{A}}$ appears, and after 14 minutes one broad peak at 7.3 ppm shows the conversion to intermediate $\mathbf{1 1 2 b}^{\mathbf{C}}$, which is complete after 42 minutes from addition of the enzyme, and is confirmed by ES-MS analysis of the crude after 12 hours (calculated m/z: $401.09[\mathrm{M}]$, found $\mathrm{m} / \mathrm{z}: 403.1\left[\mathrm{M}+2 \mathrm{H}^{+}\right]$).


Figure 5.7: ${ }^{31} \mathrm{P}$ NMR spectra showing conversion of ProTide $\mathbf{1 1 2 b}$ to intermediate $\mathbf{1 1 2 b}^{\mathbf{C}}$ by CPY enzyme.

These results suggest that lower cytotoxic activity of CNDAC ProTide 112b is not a consequence of inefficient processing to intermediate $\mathbf{1 1 2 b}^{\mathbf{C}}$, therefore might derive from low recognition by the phosphoramidase enzyme Hint.

### 5.6.2 Docking in the active site of Hint1

In order to investigate whether the low activity of CNDAC ProTides such as 112b could derive from inefficient intracellular processing to CNDAC-MP, docking of the L-alaninylphosphate intermediate $\left(\mathbf{1 1 2 b}^{\mathbf{C}}\right.$ ) was performed within the active site of Hint. Figure 5.8 shows an overlap between the co-crystallised substrate AMS-Lys and $\mathbf{1 1 2 b}^{\mathbf{C}}$ in the active pocket of Hint. ${ }^{399}$ Although fitting in the binding pocket, intermediate $\mathbf{1 1 2 b}^{\mathrm{C}}$ interacts differently from AMS-Lys. The nucleobase moiety is positioned outside the pocket,
potentially due to the presence of the cyano group in position $2^{\prime}$, which would otherwise clash with the pocket residues. This leads to an incorrect positioning of the phosphoramidate moiety, which does not interact with the crucial residues involved in the phosphoramidase activity (His-51, His-112 and His-114). ${ }^{400}$


Figure 5.8: (A) Docking of the L-alaninyl phosphoramidate monoester of CNDAC (112b ${ }^{\mathbf{C}}$ ) in the active site of Hint1. (B) 2D interactions between $\mathbf{1 1 2 b}^{\mathrm{C}}$ and Hintl binding pocket.

In conclusion the docking simulation in Hint1 active pocket suggests that the poor activity of CNDAC ProTides could be a consequence of an inefficient processing by this enzyme.

### 5.7 Conclusions

The promising anticancer potential of sapacitabine (CYC686), a CNDAC prodrug, prompted the investigation of the cytotoxic activity of CNDAC ProTides, as an alternative prodrug strategy that could overcome potentially limiting steps in CNDAC activity, such as the cytidine deaminase-mediated catabolism or the dependency on dCK for the first phosphorylation step or on nucleoside transporters for the cell penetration. Synthesis of CNDAC, sapacitabine and a family of CNDAC ProTides was performed. Cytotoxic screening of these compounds showed that CNDAC ProTides represent less potent derivatives of the parent nucleoside in vitro and do not bring an advantage in terms of cytotoxic activity over sapacitabine. Investigations on the intracellular processing of CNDAC ProTides suggested that these results could be due to an inefficient recognition by the phosphoramidase enzyme Hint-1, involved in the last step of ProTides activation. However, interest should be given to the evaluation of CNDAC ProTides in resistance conditions, such as dCK deficiency, because as CNDAC-MP prodrugs, ProTides could
afford an advantage over both CNDAC and sapacitabine. This could lead to a clinical advantage of ProTides over sapacitabine.

## 6 Trifluorothymidine ProTides

### 6.1 Background

Trifluorothymidine (trifluridine, trifluoromethyl-2'-deoxyuridine, TFT, 113), is a nucleoside analogue closely related to thymidine, bearing a trifluoromethyl $\left(\mathrm{CF}_{3}\right)$ functionality in position 5 on the pyrimidine ring (Figure 6.1).


Figure 6.1: Structure of trifluorothymidine (TFT, 113).

TFT was first synthesised by Heidelberger et al. in 1962 in the quest for new fluoropyrimidine analogues of 5 -fluorouracil ( $5-\mathrm{FU}$ ) that could avoid the insurgence of resistance to the anticancer treatment. ${ }^{401-404}$ TFT enters the cell via a nucleoside transporter identified as equilibrative transporters 1 and/or 2 (hENT1, hENT2) A recent study reports the involvement of the concentrative nucleoside transporter 1 (hCNT1) in the intestinal absorption of TFT. ${ }^{405}$ TFT is phosphorylated to TFT-monophosphate (TFT-MP) within the cell by the thymidine kinase 1 (TK1) enzyme (Scheme 6.1), ${ }^{403}$ involved in the salvage pathway. ${ }^{406}$ The efficacy of this nucleoside analogue was reported to derive from inhibition of thymidylate synthase (TS) by TFT-MP, ${ }^{404,407,408}$ similar to 5 -fluoro- ${ }^{2}$-deoxyuridine monophosphate (FUdR-MP). TS is a rate-limiting enzyme in the pyrimidine de novo deoxynucleotide synthesis which catalyses the methylation of $2^{\prime}$-deoxyuridine monophosphate (dUMP) to thymidine monophosphate (TMP) using 5,10methylenetetrahydrofolate $\left(5,10-\mathrm{CH}_{2} \mathrm{THF}\right)$ as the methyl-donor. An important difference between the activity of TFT and the other fluorinated pyrimidine analogues on TS is that TFT forms a non-covalent complex between with the enzyme and the cofactor 5,10$\mathrm{CH}_{2}$ THF, in contrast with the covalent bond formed by FUdR-MP. TFT-MP is then phosphorylated to TFT-triphosphate (TFT-TP), which can be incorporated into the DNA of the cell with subsequent induction of fragmentation (Scheme 6.1). ${ }^{409-412}$


Scheme 6.1: Intracellular metabolism of TFT. $\mathrm{P}=$ monophosphate group.

Studies on the administration of TFT to cancer patients led to promising initial results, with responses in 8 of 23 patients with breast cancer, and in 1 out of 6 patients with colon cancer. ${ }^{413,414}$ Unfortunately, rapid recurrence seen after initial regression was evident. Moreover, attempts at using maintenance therapy were hindered by an unacceptable level of bone marrow toxicity. Therefore, the progression of TFT in the clinics as a single agent was halted. ${ }^{15,16}$
TFT was also investigated as an antiviral agent against herpes simplex virus (HSV) ${ }^{414}$ infections and was approved in 1980 by the Food and Drug Administration (FDA) for use in the topical treatment of primary keratoconjunctivitis and epithelial keratitis (Viroptic). ${ }^{46,415,416}$
Metabolic studies after administrations of TFT reveal that the major metabolites found in the urine are 5 -carboxy- ${ }^{\prime}$-deoxyuridine $(5 \mathrm{COOH}-\mathrm{dU})$, 5 -trifluorouracil $\left(5 \mathrm{CF}_{3}-\mathrm{U}\right)$ and 5 carboxyuracil ( $5 \mathrm{COOH}-\mathrm{U})^{403,413,414}$.


Scheme 6.2 Principal metabolites of TFT in urine, generated by TP de-phosphorylation and oxidation mechanisms.
$5 \mathrm{COOH}-\mathrm{dU}$ is formed in the liver due to the lability of the $\mathrm{CF}_{3}$ group, which is oxidised in alkaline conditions. ${ }^{406,407} 5 \mathrm{CF}_{3}-\mathrm{U}$ is formed after cleavage of the glycosidic moiety by thymidine phosphorylase (TP) and is then oxidised into $5 \mathrm{COOH}-\mathrm{U}$. A thorough
description of all the metabolites of TFT was documented by Wyrwicz et al. ${ }^{406}$ When given orally, TFT is well absorbed, although quickly degraded by thymidine phosphorylase (TP) in the liver. ${ }^{411}$ Recently, the interest in TFT as an anticancer agent has arisen as a combined therapy as TAS-102 (molar ratio TFT:TPI $=1: 0.5$ ) with the TP-inhibitor tipiracil hydrochloride (TPI, Figure 6.2). ${ }^{417,418}$


Figure 6.2 Structure of Tipiracil hydrochloride (TPI).

The increase stability of TAS-102 has led to an increase in antitumour activity, ${ }^{416}$ therefore this combination has been involved in Phase I, II and III clinical studies showing prolonged survival in patients with metastatic colorectal cancer that were refractory or intolerant to standard chemotherapy often including 5FU. ${ }^{419,420}$ Based on the results of the phase II clinical study, TAS-102 was approved in 2012 in Japan for the treatment of unresectable advanced and recurrent colon cancers. ${ }^{416}$ A phase III clinical study confirmed the efficacy of TAS102 with median overall survival improved from 5.3 months with placebo to 7.1 months with treatment. ${ }^{421}$ Temmink et al. have conducted a study aimed at understanding the potential resistance mechanisms that can arise from TFT treatment. ${ }^{422}$ In this study, resistance to TFT was induced in H630 colorectal cell line by intermittent short exposure to high concentrations of the drug. The results indicate the principal resistance mechanisms are the down-regulation of TK and a reduced expression of hENT transporters. ${ }^{422}$ These findings prompted us to investigate the potential of applying the ProTide approach to TFT, with the aim of bypassing these two resistance mechanisms. Moreover, ProTides were already reported to be stable to TP deactivation, ${ }^{152,423}$ and this advantage could avoid the co-administration of TPI. A family of TFT ProTides was therefore synthesised and the cytotoxic potential of these prodrugs was analysed on colorectal, breast and pancreatic cell lines. Moreover, additional screening of these prodrugs was also performed on HSV infected cells, due to the well-known activity of TFT on DNA-viruses.

### 6.2 Synthesis of TFT ProTides

ProTides of TFT were synthesised from commercially available TFT that was treated with N -methylimidazole (NMI) and the appropriate phosphorochloridate in THF, ${ }^{105}$ affording ProTides 114a-f in 11-45 \% yields.


Scheme 6.3: Synthesis of TFT ProTides 114a-f. Reagents and conditions: NMI (5eq), appropriate phosphochloridate ( 3 eq ), anhydrous THF, 16h, rt.

| Common structure | Cpnd | Aryl | $\mathbf{R}$ | $\mathbf{R}_{\mathbf{1}}$ | $\mathbf{A A}$ | ${ }^{19} \mathbf{F}(\mathbf{p p m})$ | ${ }^{31} \mathbf{P}(\mathbf{p p m})$ | Yield \% |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
|  | $\mathbf{1 1 4 a}$ |  |  |  |  |  |  |  |

Table 6.1: Structures, ${ }^{19} \mathrm{~F}-\mathrm{NMR}$ and ${ }^{31} \mathrm{P}$-NMR peaks corresponding to the two diastereoisoemrs of each Protide, and yields of synthesised compounds (114a-f), ${ }^{19} \mathrm{~F}$-NMR spectra preformed in $\mathrm{CD}_{3} \mathrm{OD}(470 \mathrm{MHz})$, ${ }^{31} \mathrm{P}-\mathrm{NMR}$ spectra preformed in $\mathrm{CD}_{3} \mathrm{OD}(202 \mathrm{MHz})$.

The structures of synthesised ProTides included mainly L-alanine as the amino acid moiety, with the exception of compound $\mathbf{1 1 4 f}$ bearing L-Leucine. The aryl moiety varied between phenyl ( $\mathbf{1 1 4} \mathbf{c}$ and $\mathbf{1 1 4 e}$ ) and naphthyl ( $\mathbf{1 1 4 b}, \mathbf{1 1 4} \mathbf{d}, \mathbf{1 1 4 f}$ ) with one example of a compound bearing the ethyl-3-phenyl propanoate group (114a). The ester group on the synthesised ProTides varied between the benzyl, cyclohexyl, the linear $n$-pentyl and the branched neopentyl (Table 6.1).

The identity of compounds 114a-f was confirmed by NMR analyses. The distinctive splitting pattern in ${ }^{31} \mathrm{P}$ NMR and ${ }^{19} \mathrm{~F}$ NMR spectra due to the presence of an approximately 1:1 ratio of two diastereoisomers is reported in Table 6.1. ${ }^{19} \mathrm{~F}$ NMR and ${ }^{31} \mathrm{P}$ NMR spectra of compound $\mathbf{1 1 4} \mathbf{c}$ are reported in Figure 6.3.


Figure 6.3: ${ }^{31} \mathrm{P}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 202 \mathrm{MHz}, \mathbf{A}\right)$ and ${ }^{19} \mathrm{~F}$ NMR $\left(\mathrm{CD}_{3} \mathrm{OD}, 470 \mathrm{MHz}, \mathbf{B}\right)$ spectra of compound 114 c.

The presence of a $\mathrm{CF}_{3}$ group in position 5 on the pyrimidine ring is confirmed by the presence of a quadruplet at 123.87 ppm corresponding to the $\mathrm{CF}_{3}$ carbon, whose coupling constant with fluorine is ${ }^{1} J_{\mathrm{FC}}=267.5 \mathrm{~Hz}$ (Figure 6.4, A). Moreover, two overalapping quadruplets at 105.5 ppm correspond to the C 5 carbon of the two diastereoisomers, with ${ }^{2} J_{\text {FCC }}$ coupling constant of 32.5 Hz (Figure 6.4, B).


Figure $6 .{ }^{13} \mathrm{C}$-NMR spectrum $\left(\mathrm{CD}_{3} \mathrm{OD}, 125 \mathrm{MHz}\right)$ of compound $\mathbf{1 1 4} \mathbf{c}$, and expansions of selected regions of ${ }^{13} \mathrm{C}$-NMR. A: $\mathrm{CF}_{3}\left(\mathrm{q}, 123.87 \mathrm{ppm},{ }^{1} J_{\mathrm{FC}}=267.5 \mathrm{~Hz}\right)$, B: C5 ( $2 \mathrm{xq}, 105.5 \mathrm{ppm},{ }^{1} J_{\mathrm{FC}}=32.5 \mathrm{~Hz}$ ).

### 6.3 Biological evaluation

TFT ProTides 114a-f were screened for their anticancer potential by WuXi AppTech on the cell lines reported, belonging to colon adenocarcinoma, pancreas and breast tumours (Table 6.2, Figure 6.5). The activity of TFT was, in fact, extensively documented in vitro and in vivo on colon carcinoma models, prior to the clinical trials in patients affected by this malignancy. ${ }^{416,424,425}$ However, TFT was also tested on breast ${ }^{426,427}$ and pancreatic cancer models, ${ }^{428}$ hence the decision to broaden the set of solid tumours cell lines for this screening.

| Cell line | Malignancy |
| :---: | :---: |
| SW620 | colon adenocarcinoma |
| BxPC-3-Luc | pancreatic cancer |
| HT29 | colon adenocarcinoma |
| MIA PaCa-2 | pancreatic carcinoma |
| MCF-7 | breast carcinoma |

Table 6.2: Selection of cell lines for the in vitro screening of TFT ProTides by WuXi AppTech.

The activity of TFT on the two colorectal cell lines was dissimilar, with submicromolar $\mathrm{IC}_{50}$ on SW620 cell line $(0.7 \mu \mathrm{M})$ and lower cytotoxicity on HT29 cells $\left(\mathrm{IC}_{50}=31.7 \mu \mathrm{M}\right)$ (Table 6.3). However the $\mathrm{MI}_{\%}$ was similarly low, and corresponded to approximately $60 \%$, meaning that TFT exerts cytotoxic activity only on this percentage of cells.

| Cpnd | SW620 |  | HT29 |  | BxPC-3 |  | MiaPaCa-2 |  | MCF7 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{IC}_{50}$ | MI\% | $\mathbf{I C}_{50}$ | MI\% | $\mathbf{A b I C}_{50}$ | MI\% | $\mathbf{A b I C}_{50}$ | MI\% | $\mathrm{AbIC}_{50}$ | MI\% |
| TFT | 0.7 | 62 | 31.7 | 65 | >198 | 45 | 1.84 | 86 | 23.22 | 57 |
| 114a | 23.05 | 100 | 16.72 | 99 | 49.65 | 98 | 5.6 | 90 | 14.74 | 100 |
| 114b | 8.77 | 100 | 9.86 | 110 | 22.18 | 98 | 2.92 | 95 | 7.44 | 100 |
| 114c | 25.18 | 60 | 49.71 | 92 | >198 | 31 | 18.74 | 84 | 45.37 | 83 |
| 114d | 39.59 | 69 | 31.75 | 92 | >198 | 44 | 16.66 | 89 | 33.52 | 88 |
| 114e | 11.79 | 100 | 10.01 | 110 | 32.75 | 99 | 4.68 | 100 | 8.78 | 100 |
| 114f | 17 | 100 | 15.33 | 103 | 32.69 | 99 | 14.03 | 100 | 17.81 | 100 |
| PTX | 0.02 | 96 | 0.004 | 75 | $>0.5$ | 42 | 0.002 | 87 | 0.01 | 65 |

Table 6.3: Cytotoxic evaluation of TFT (113) and ProTides 114a-c. Cytotoxicity data reported as $\mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentrations considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

This could be due to resistance to TFT on the other $40 \%$ of cells. Treatment of the two pancreatic cell lines BxPC-3 and MIA PaCa-2 with TFT led again to contrasting results: the nucleoside reached $45 \%$ of maximum inhibition of viability on the former cell line therefore an $\mathrm{IC}_{50}$ value could not be calculated (reported as $>198 \mu \mathrm{M}$ ).
Although TFT ProTides 114a-f were characterised by higher $\mathrm{IC}_{50}$ values on these two cell lines, 4 out of 6 of the compounds reached a complete inhibition of cell viaibility at moderate $\mu \mathrm{M}$ values. On HT29 cell line, some of the compounds (114a, 114b, 114e and 114f) were also 1 to 2 -fold more active than TFT, in terms of $\mathrm{IC}_{50}$ values. These data may suggest that TFT ProTides could overcome resistance mechanisms associated with the treatment with TFT, due to the ability to exert cytotoxic activity on the total cell population. On this cell line, similarly to the results obtained on SW620, ProTides 114a, $\mathbf{1 1 4 b}, 114 \mathrm{e}$ and 114 f reached, instead, the complete inhibition of cell viability and moderate micromolar $\mathrm{IC}_{50}$ values. On the other hand, TFT had good cytotoxicity on the latter cell line, with $\mathrm{IC}_{50}$ of $1.84 \mu \mathrm{M}$ and $\mathrm{MI}_{\%}$ of $86 \%$, while ProTides were all less active than the parent nucleoside in terms of $\mathrm{IC}_{50}$ values. TFT was a weak cytotoxic agent on the breast cancer cell line MCF7, while some of the ProTides (114a and 114e) were approximately 3-fold more potent than TFT with complete inhibition of cell viability.


Figure 6.5: Cytotoxic activity of TFT and ProTides $\mathbf{1 1 4 a}-\mathbf{f}$ on selected cell lines. ${ }^{*} \mathrm{IC}_{50}>198 \mu \mathrm{M}$.

These data show that some TFT ProTides appear to be more active than TFT itself on some of the cell lines considered, both in terms of $\mathrm{IC}_{50}$ values and $\mathrm{MI}_{\%}$. However high concentrations of compound are required to exert the cell viability inhibition. This could be due to a slow intracellular processing of the ProTide moiety which affects the rate of release of TFT-MP, the active species within the cell. In order to investigate whether the ProTide processing could be the limiting step in the activity of TFT ProTides, mechanistic investigations were performed.

### 6.4 Mechanistic investigations on TFT ProTides

As previously described, an NMR enzymatic assay with CPY and docking within the active site of Hint were performed.

### 6.4.1 CPY enzymatic assay

CPY is putatively involved in the cleavage of the amino acid ester, which triggers the metabolic activation of ProTide 114c with spontaneous cyclisation and release of the phenyloxy moiety $\left(\mathbf{1 1 4} \mathbf{c}^{\mathbf{B}}\right)$, followed by hydrolysis to L-alaninyl intermediate $\mathbf{1 1 4 \mathbf { c } ^ { \mathbf { C } }}$ (Scheme 6.4).


Scheme 6.4 Putative intracellular conversion of ProTide $\mathbf{1 1 4 c}$ into TFT-MP (114c ${ }^{\mathbf{D}}$ ) catalysed by CPY/Hint enzymes.
${ }^{31} \mathrm{P}$ NMR spectra were recorded over 12 hours to monitor the processing by CPY and Figure 6.6 shows that the peaks corresponding to the diastereomeric mixture of $\mathbf{1 1 4 c}$ were quickly converted into two more downfield peaks corresponding to intermediate $\mathbf{1 1 4} \mathbf{c}^{\mathbf{A}}$, lacking the ester group on the amino acid moiety (Scheme 6.4).


Figure 6.6: ${ }^{31} \mathrm{P}$ NMR spectra showing conversion of ProTide $\mathbf{1 . 1 2}$ to intermediate $\mathbf{1 . 1 2}^{\mathbf{C}}$ in 1 h 10 minutes.

Subsequent, the conversion into a peak at 6.80 ppm potentially corresponding to final compound $\mathbf{1 1 4} \mathbf{c}^{\mathrm{C}}$, was confimed by ES/MS analysis of the solution.

The activation catalysed by CPY is complete in 1 hour and 10 minutes, therefore the first enzymatic step does not appear to be rate-limiting for the initial activation towards TFTMP.

The $\mathrm{CF}_{3}$ group in position 5 on the pyrimidine ring of TFT also represents a probe that can be used to monitor the metabolism of TFT ProTides through ${ }^{19} \mathrm{~F}$ NMR, in parallel to ${ }^{31} \mathrm{P}$ NMR. In fact comparison between the ${ }^{19} \mathrm{~F}$ NMR spectrum of $\mathbf{1 1 4 c}$ at time 0 and after treatment with CPY shows a shift from -63.07 and -63.09 ppm (peaks corresponding to the diastereomeric mixture of $\mathbf{1 1 4 c}$ ) to one peak at -62.89 ppm , after 12 hours, corresponding to the final structure $\mathbf{1 1 4} \mathbf{c}^{\mathrm{C}}$.


Figure 6.7: ${ }^{19} \mathrm{~F}$ NMR spectra showing conversion of ProTide $\mathbf{1 1 4} \mathbf{c}$ to intermediate $\mathbf{1 1 4 c}$.

### 6.5 Docking in the Hint enzyme

The efficient processing by CPY of the ProTide 114c suggests the first step in the ProTide activation is not a limiting one, and therefore a reason for lack in potent activity of ProTides could depend on the effectiveness of the phosphoramidase-mediated second enzymatic step, responsible for the cleavage of the phosphoramidate moiety with release of the active metabolite TFT-MP. Docking of the L-alaninyl-phosphate intermediate (114c ${ }^{\mathbf{C}}$ ) was performed within the active site of Hint.


Figure 6.8 (A) Docking of the L-alaninyl phosphoramidate monoester of TFT (114c $\left.{ }^{\mathbf{C}}\right)$ in the active site of Hint1. (B) Predicted interactions between $\mathbf{1 1 4} \mathbf{c}^{\mathrm{C}}$ and Hintl binding pocket.

The important interactions for the correct positioning of the phosphoramidate group of the substrate into the binding site are potentially those between the carboxylic acid portion and a residue of asparagine (AsnD99), while the nitrogen involved in the phosphoramidate bond positively interacts with two serine residues (SerB107 and SerD107). Problems encountered in the binding of the substrate depended on the positioning of the uracil group outside the pocket, which apparently is a common feature of pyrimidine nucleoside amino acyl phosphates but could be heightened by the presence of the bulky trifluoromethyl $\left(\mathrm{CF}_{3}\right)$ group. ${ }^{429}$

In conclusion, the docking simulation shows some good aspects, with some likely crucial interactions observed in the binding of the selected substrate within the binding site of Hint1 but the presence of a bulky $\mathrm{CF}_{3}$ group could be the reason for the incorrect positioning of the nucleoside portion within the pocket, hence potentially slow processing by Hint1.

### 6.6 Biological evaluation on HSV infected cells.

TFT is a marketed antiviral agent for the topical treatment of epithelial keratitis caused by herpes simplex virus (HSV). ${ }^{414}$ When tested in vitro, TFT was shown to be endowed with $\mathrm{IC}_{50}$ values of activity ranging from 3 to $10 \mu \mathrm{M}$ on different strains of HSV. When tested on TK negative mutant strains of HSV, the activity was two-fold lower. ${ }^{430}$ TFT ProTides activity was screened by Dr. J. Bugert (School of Medicine, Cardiff Unievrsity) on HSV infected cells and the cytopathic effect (CPE, the damage to the host cells during virus
invasion) measured. ${ }^{224}$ The activity of TFT was compared both to TFT ProTides and to acyclovir (ACV), a well-known anti-HSV agent. ${ }^{431}$

| $\mathbf{C P F}$ | \% CPE inhibition | $\mathbf{C C}_{\mathbf{5 0}}$ |
| :---: | :---: | :---: |
| $\mathbf{A C V}$ | 50 | $>10$ |
| $\mathbf{T F T}$ | - | $>10$ |
| $\mathbf{1 1 4 a}$ | - | $>10$ |
| $\mathbf{1 1 4 b}$ | - | $>10$ |
| $\mathbf{1 1 4 c}$ | - | $>10$ |
| $\mathbf{1 1 4 d}$ | - | $>10$ |
| $\mathbf{1 1 4 e}$ | - | $>10$ |
| $\mathbf{1 1 4 f}$ | 50 | $>10$ |

Table 6.4: Anti-HSV activity of TFT and ProTide 114a-f. Activity expressed as \% reduction of viral CPE at $10 \mu \mathrm{M}$ concentration. Assays performed by Dr. J. Bugert, Cardiff University.

Compounds were tested at $10 \mu \mathrm{M}$ concentrations. ACV could reach $50 \%$ of protection at this dose, while both TFT and ProTides 114a-e were not effective. Only 114f, bearing Lleucine pentyloxy naphthyl phosphoramidate moiety, could reach $50 \%$ effect at the given dose. This could be due to bypass of resistance mechanisms and could deserve further investigations.

### 6.7 Conclusions

TFT is a very promising nucleoside analogue used as an antiviral in the topical treatment of HSV infections and has recently also shown intriguing anticancer activity both in vitro and in vivo. The combination of TFT and TPI, a TP inhibitor, is currently undergoing phase III clinical studies in the USA and has already been approved for the treatment of unresectable advanced and recurrent colon cancers in Japan. Due to the interesting anticancer profile of this agent and the documented resistance mechanisms due to TP inactivation and TK1 and hENT1 down regulation, causing a decrease in intracellular concentrations of the two active metabolites TFT-MP and TFT-TP, the ProTide prodrug approach was applied to TFT with the aim of bypassing all the aforementioned limiting steps in its anticancer activity. A series of TFT ProTides was synthesised in moderate yields and was evaluated in vitro on solid cancer cells. These studies revealed that although being a moderate to potent anticancer agent, TFT does not lead to complete inhibition of cell viability at the highest concentration considered. On the other hand, although often reported as less potent than the parent nucleoside, TFT ProTides tend to reach the complete
inhibition of cell viability. This finding could be linked to the bypassing by ProTides of the resistance mechanisms that potentially lead to lack in activity of TFT on some cells, and deserve some further investigations. Moreover, when tested on HSV infected cells, TFT resulted in a weak antiviral agent while one of the pro-drugs (the L-leucinyl derivative 114f) was effective at a concentration of $10 \mu \mathrm{M}$. In light of these results, the family of TFT ProTides could be expanded in order to include more lipophilic prodrugs such as the LLeucinyl derivative $\mathbf{1 1 4 f}$.

## 7 2'- $\mathrm{SCF}_{3}$ deoxyuridine and 2'-SH deoxyuridine

### 7.1 Background

The introduction of a trifluoromethylthio $\left(\mathrm{SCF}_{3}\right)$ functionality on the sugar moiety of nucleosides was reported to serve as a probe for RNA-ligand interactions and to monitor RNA structure at secondary and tertiary level, through ${ }^{19} \mathrm{~F}$ NMR spectroscopy. ${ }^{432-437}$ Moreover, the insertion of 2'- $\mathrm{SCF}_{3}$-bearing nucleoside analogues in double helical regions of the RNA was related to disrupting effects on base-pairing interactions. ${ }^{435,436,438}$ To date, $2^{\prime}$-deoxy- $2^{\prime}-$ SCF $_{3}$ modified uridine (115), ${ }^{437}$ cytidine (116) ${ }^{436}$ and, more recently, adenosine (117) and guanosine (118) ${ }^{435}$ (Figure 7.1) were synthesised and incorporated into oligonucleotides to study their physicochemical properties, however no cell-based in vitro data were reported.


115


116


117


118

Figure 7.1: Structures of 2'- $\mathrm{SCF}_{3}-2^{\prime}$ '-deoxynucleosides (115-118) synthesised to date.

The destabilising properties of these nucleoside analogues on RNA structure were considered as an interesting starting point for the investigation of the potential anticancer properties of $2^{\prime}-\mathrm{SCF}_{3}-2^{\prime}$-deoxyuridine ( $2^{\prime}-\mathrm{SCF}_{3}-2^{\prime}-\mathrm{dU}, 115$ ), an easily chemically accessible nucleoside analogue within the family of structurally related nucleosides depicted in Figure 7.1. No data regarding the recognition of this nucleoside by kinase enzymes required for the first phosphorylation step or nucleoside transporters responsible for cell penetration was reported. However, a bulky modification such as the 2 ' $-\mathrm{SCF}_{3}$ group may cause inefficient recognition of the structure by nucleoside transporters or kinase enzymes. ${ }^{439}$ ProTide $5^{\prime}$ 'monophosphate prodrugs could therefore offer an advantage in terms of bypassing the first monophosphorylation step or the dependency on a nucleoside transporter for cell entry, due to increased lipophilicity of the ProTide, enabling passive diffusion. Therefore, the ProTide approach was applied to this nucleoside analogue, for the synthesis of a family of $2^{\prime}-\mathrm{SCF}_{3}-2^{\prime}$-dU ProTides (Figure 7.2) to be tested
on a selection of malignant cell lines along with the parent nucleoside and investigate the potential anticancer activity of these derivatives.


Figure 7.2: Common structure of ${ }^{\prime}$ ' $-\mathrm{SCF}_{3}$ - 2 ' -dU ProTides.

Since the synthetic strategy described by Fauster et al. ${ }^{437}$ for the preparation of compound 115 consisted of four steps (Scheme 7.1) including the synthesis of the 2'-mercapto-2'deoxyuridine intermediate ( $2^{\prime}$-SH-2'-dU, 119, Figure 7.3), the anticancer activity of this nucleoside analogue was also investigated.


120

121

Figure 7.3: Structure of 2'-SH-2'-dU (119), relative ProTides (120) and intracellular active metabolite 2'-SH-2'-dUDP (121). PP: diphosphate group.

In fact, 2'-mercapto-2'-deoxyuridine $5^{\prime}$ '-diphosphate (2'-SH-2'-dUDP, 121, Figure 7.3) was found to be a potent inhibitor of ribonucleotide reductase (RNR). ${ }^{440}$ This enzyme has a crucial function in DNA biosynthesis and repair, as it is involved in the synthesis of deoxynucleotides. ${ }^{441}$ As such, this enzyme has already been a target for the design of antitumour and antiviral agents such as cytarabine. ${ }^{442,443}$ The in vitro activity of the mercapto nucleoside 119 has not been reported to date. However, Roy et al. investigated both the cytotoxic activity of the disulphide dimer form of this nucleoside (122) and some disulphide prodrugs of $\mathbf{1 1 9}$ (123, Figure 7.4), assuming that these disulphide-containing molecules would release the $2^{\prime}$-mercapto analogue $\mathbf{1 1 9}$ after intracellular reduction. ${ }^{444}$


122


123

Figure 7.4: Structures of 2'-deoxyuridine 2'-thio derivatives, which were previously reported in the literature.

However, they observed neither intracellular inhibition of the RNR enzyme nor cytotoxic activity on malignant cell lines. They hypothesised that these negative results could be a result of poor intracellular phosphorylation to the active 5'-diphosphate metabolite. ${ }^{444}$ As a consequence of these results, the ProTide approach was applied on this nucleoside analogue, in order to potentially release the 5 '-monophosphate analogue in the intracellular environment bypassing the potentially limiting step of first phosphorylation. The in vitro cytotoxic screening of this family of prodrugs and the parent nucleoside 119 was then performed, in order to investigate the potential anticancer properties of these structures.

### 7.2 Synthesis of nucleosides 2'-SH-2'-dU and 2'-SCF ${ }_{3} \mathbf{- 2}^{\prime}$ '-dU

Parent nucleosides $2^{\prime}$ - $\mathrm{SH}-2^{\prime}$-dU (119) and $2^{\prime}-\mathrm{SCF}_{3}-2^{\prime}$-dU (115) were synthesised following Scheme 7.1. ${ }^{437,445,446}$


Scheme 7.1: Synthesis of 2'-SH-2'-dU (119) and $2^{\prime}-\mathrm{SCF}_{3}-2^{\prime}$ ' dU (115). Reagents and conditions: (a.) $\mathrm{Ac}_{2} \mathrm{O}$ (4 eq), DMAP (0.4 eq), TEA (4 eq) $\mathrm{CH}_{3} \mathrm{CN}, \mathrm{rt}, 16 \mathrm{~h}, 97 \%$; (b.) [1] AcSH, DMF, $110{ }^{\circ} \mathrm{C}, 6$ hours, $25 \%$ [2] AcSH ( 25 eq ), 1,4-dioxane, $110^{\circ} \mathrm{C}$, 6 hours, $73 \%$; (c.) $1 \mathrm{~N} \mathrm{KOH}, \mathrm{H}_{2} \mathrm{O}: \mathrm{EtOH}(1: 1, \mathrm{v} / \mathrm{v}), 0^{\circ} \mathrm{C}, 30 \mathrm{~min}, 87 \%$; (d.) Togni's reagent $(\mathbf{1 2 3}, 2.5 \mathrm{eq}), \mathrm{CH}_{3} \mathrm{OH},-78^{\circ} \mathrm{C}, 1 \mathrm{~h}, 76 \%$.

Commercially available $2^{\prime}$ '-deoxy- $2^{\prime}$-anhydrouridine (124) was first acetylated in position 3' and 5' on the sugar moiety by treatment with acetic anhydride, triethylamine and DMAP in acetonitrile at room temperature for 16 hours, affording the desired protected compound (125) in $97 \%$ yield (Scheme 7.1, a.). ${ }^{477,448}$

Intermediate $\mathbf{1 2 5}$ was treated with thioacetic acid (AcSH) in DMF at $110^{\circ} \mathrm{C}$ for 6 hours to afford the 2'-thioacetylated product (126) (Scheme 7.1b.[1]). ${ }^{45}$ The cleavage of the 2,2'-
anhydro linkage with the introduction of the thioacetic moiety in 2'-position proceeded with only $25 \%$ yield, similar to the yield reported in the literature ( $30 \%$ ) due to unreacted starting material and possibly the already reported reactivity of DMF with thioacetic acid. ${ }^{446}$ However, the product formation was confirmed by the presence of a new singlet integrating for 3 protons corresponding to the thioacetyl group (SAc), at 2.37 ppm in the ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}, 500 \mathrm{MHz}\right.$ ), slightly downfield by comparison to the OAc groups ( $\delta 2.14$ and 2.19 ppm ) and comparable to the reported spectroscopic data. ${ }^{446}$ An improvement in the reaction yield up to $73 \%$ was reached by using 1,4-dioxane instead of DMF as a solvent (Scheme 7.1, b.[2]). ${ }^{446}$ Complete deprotection of the acetylated precursor 126 was accomplished by treatment with KOH in a mixture of water and methanol at $0^{\circ} \mathrm{C}$ (Scheme 7.1, c.), ${ }^{446}$ affording deprotected $2^{\prime}$-mercapto-2'-deoxyuridine (119) in $87 \%$ yield.


Figure 7.5: ${ }^{1} \mathrm{H}$ NMR spectrum of $2{ }^{\prime}-\mathrm{SH}-2{ }^{\prime}-\mathrm{dU}(\mathbf{1 1 9})$ in $\mathrm{CD}_{3} \mathrm{OD}(500 \mathrm{MHz})$.


Figure 7.6: ${ }^{13} \mathrm{C}$ NMR (pendant) spectrum of 2'-SH-2'-dU (119) in $\mathrm{CD}_{3} \mathrm{OD}$ ( 125 MHz ).

The structure of $\mathbf{1 1 9}$ was confirmed by ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR and ES/MS analysis, and by comparison with the reported spectroscopic data. ${ }^{446}$ The most characteristic features of these spectra is the presence of very upfield $\mathrm{H} 2^{\prime}$ and $\mathrm{C}^{\prime}$, signals due to the presence of the $2^{\prime}-\mathrm{SH}$ group. This group has a lower electronegativity compared to the $2^{\prime}$ ' OH group,
therefore has a lower de-shielding effect, hence the resonance of both H2' and C2' at lower ppm values.

The final step for the synthesis of $\mathbf{1 1 5}$ consisted of the trifluoromethylation of the 2'-SH group of derivative 119. This step was accomplished by treatment with 3,3-dimethyl-1-(trifluoromethyl)-1,2-benzodioxole (Togni's reagent, 127, Figure 7.7) ${ }^{449,450}$ in methanol at $-78{ }^{\circ} \mathrm{C}$, yielding the desired product $\mathbf{1 1 5}$ in $\mathbf{7 6 \%}$ yield (Scheme 7.1, d.). ${ }^{437}$


Figure 7.7: Structure of 3,3-dimethyl-1-(trifluoromethyl)-1,2-benzodioxole (Togni's reagent, 127).

This electrophilic trifluoromethylation is totally selective for the thiol group over the hydroxyl group, because the soft nucleophiles (with high lying HOMO) such as the SH group are preferred over harder nucleophiles such as OH groups at low temperature. ${ }^{450}$


Figure 7.8: ${ }^{1} \mathrm{H}$ NMR spectrum of 2' $-\mathrm{SCF}_{3}-2$ '- $\mathrm{dU}(\mathbf{1 1 5})$ in $\mathrm{CD}_{3} \mathrm{OD}(500 \mathrm{MHz})$.

The presence of the $\mathrm{CF}_{3}$ moiety exclusively on the thiol group was suggested by the shift in the ${ }^{1} \mathrm{H}$ NMR signal of the proton in 2 '-position higher ppm values, while all the other signals were in a similar range as precursor $\mathbf{1 1 9}$, in $\mathrm{CD}_{3} \mathrm{OD}$ (Figure 7.8). Moreover, the ${ }^{19} \mathrm{~F}$ NMR spectrum showed one signal corresponding to one $\mathrm{SCF}_{3}$ group at -41.60 ppm (in $\mathrm{CD}_{3} \mathrm{OD}$ ), in accordance with the value reported in the literature ( $\delta-38.94 \mathrm{ppm}$ in DMSO$d \sigma$ ).


Figure 7.9: ${ }^{19}$ F-NMR spectrum of $2^{\prime}-\mathrm{SCF}_{3}-2$ '- $\mathrm{dU}(\mathbf{1 1 5})$ in $\mathrm{CD}_{3} \mathrm{OD}(470 \mathrm{MHz})$.

Very characteristic is also the splitting pattern related to the $\mathrm{CF}_{3}$ group in the ${ }^{13} \mathrm{C}$ NMR spectrum (Figure 7.10), showing a quadruplet with a ${ }^{1} J_{\mathrm{C} \text { - F }}$ value of 303.5 Hz . Moreover, the $\mathrm{SCF}_{3}$ group represents an isolated spin system, ${ }^{437}$ hence the lack of coupling between fluorine and the carbon in the 2 '-position.


Figure 7.10: ${ }^{13} \mathrm{C}$ NMR spectrum of $2^{\prime}-\mathrm{SCF}_{3}-2^{\prime}-\mathrm{dU}(\mathbf{1 1 5})$ in $\mathrm{CD}_{3} \mathrm{OD}(125 \mathrm{MHz})$.

### 7.3 Synthesis of 2'-SCF3-2'-dU and 2'-SH-2'-dU ProTides

### 7.3.1 Synthesis of 2'-SCF $\mathbf{3}^{-2}$ '-dU ProTides

The synthetic strategy applied to the synthesis of $2^{\prime}-\mathrm{SCF}_{3}-2^{\prime}$-dU ProTides involved the treatment of the nucleoside with NMI and the appropriate phosphorochloridate in THF at room temperature overnight (Scheme 7.2). ${ }^{105}$


Scheme 7.2: Synthesis of $2^{\prime}-\mathrm{SCF}_{3}-2$ '-dU ProTides 128a-c. Reagents and conditions: appropriate phosphorochloridate (3 eq), NMI ( 5 eq ), THF, rt, 16h.

| Common structure | Compound | Ar | R | $\mathrm{R}_{1}$ | AA | Yield \% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 128a | Ph | Bn | Me | L-Ala | 87 |
|  | 128b | Ph | Hex | " | " | 56 |
|  | 128c | Nap | Bn | " | " | 42 |

Table 7.1: Structures and synthetic yields for ProTides 128a-c.

Three examples of ProTides were synthesised in moderate to good yields (Table 7.1), with no traces of 3'-phosphoramidate regioisomer or 3',5'-bisphosphoramidate. These analogues were then evaluated for anticancer activity as approximate $1 / 1$ mixtures of two diastereoisomers.

### 7.3.2 Synthesis of 2'-SH-2'-dU prodrugs

The attempted of synthesis of 2'-mercapto-2'-deoxyuridine ProTides (129) by treatment of the parent nucleoside $\mathbf{1 2 0}$ with the appropriate phosphorochloridate and NMI in THF yielded prodrugs characterised by dimeric disulphide structure (129a-c, Scheme 7.3).

This result might be connected to the reaction environment created by NMI, as a consequence of the tendency of the thiol groups to form disulphide bonds, as was specifically reported also on nucleosides bearing the thiol functionality on the sugar moiety. ${ }^{451-453}$


Scheme 7.3: Synthesis of of ${ }^{\prime}$ '-SH-2'-dU derivatives 129a-c. Reagents and conditions: appropriate phosphorochloridate (3 eq), NMI (5 eq), THF, rt, 16h.

| Common structure | Cpnd | Ar | ( $\mathrm{R}_{1}$ ) | R | AA | Yield \% |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 129a | Ph | Me | Bn | L-Ala | 21\% |
| , | 129b | Nap | " | Bn | " | 11\% |
| ${ }^{\mathrm{RO}} \mathrm{O}_{\left.\mathrm{R}^{1} \mathrm{OH} \mathrm{S}-\right]_{2}}$ | 129c | Ph | " | Hex | " | 5\% |

Table 7.2: Structures and synthetic yields of disulfide dimers 129a-c.

An example of the symmetric disulphide structures yielding from the coupling reaction reported in Scheme 7.3 is given in Figure 7.13, with compound 129a. The dimeric structure of the product was firstly suggested by mass, $\left(\mathrm{m} / \mathrm{z}\left[\mathrm{M}+\mathrm{Na}^{+}\right]=1175.2\right.$, $\mathrm{C}_{50} \mathrm{H}_{54} \mathrm{~N}_{6} \mathrm{O}_{18} \mathrm{P}_{2} \mathrm{~S}_{2}$, calculated $\mathrm{m} / \mathrm{z}[\mathrm{M}]=1152.24$ ). The ${ }^{31} \mathrm{P}$ NMR spectrum (Figure 7.11), nevertheless, shows only three peaks, presumably due to overlapping of two species.


Figure 7.11: ${ }^{31} \mathrm{P}$ NMR spectrum of compound 129a. $\left(\mathrm{CD}_{3} \mathrm{OD}, 202 \mathrm{MHz}\right)$

However, the ${ }^{1} \mathrm{H}$ NMR analysis of the product is shown in Figure 7.12 and the enlarged area of the spectra corresponding to protons H 6 and $\mathrm{H}^{\prime}$ ' showed characteristic signal pattern of the structures depicted in Figure 7.13. The presence of 4 doublets for the H6 signal and 8 doublets corresponding to H 2 ' confirmed the presence of a disulphide bond, giving rise to three different species depending on the phosphorus chirality.


Figure 7.12: ${ }^{1} \mathrm{H}$ NMR spectrum of compound $\mathbf{1 2 9 a}$ in $\mathrm{CD}_{3} \mathrm{OD}(500 \mathrm{MHz})$, enlarged portions of the spectra with signals corresponding to H 6 and H 2 '.

As shown in Figure 7.13, in fact, the coupling of 5'-hydroxyl groups with either a $S_{\mathrm{p}}$ or a $R_{\mathrm{p}}$ phosphorochloridrate (respectively yielding structures $\mathbf{A}$ and $\mathbf{B}$ ), would generate two different isomers, while the coupling of the two $5^{\prime}-\mathrm{OH}$ groups with two different phosphorochloridate diastereoisomers ( $S_{\mathrm{p}}$ and $R_{\mathrm{p}}$ ) would generate a single meso structure (C).




Figure 7.13: Structures (A,B and C) corresponding to compound 129a.

Therefore, the ${ }^{1} \mathrm{H}$ NMR spectrum of compound 129a shows the presence of 4 doublets ( $\delta$ $7.58^{2} J=8.0 \mathrm{~Hz} ; \delta 7.57^{2} J=8.5 \mathrm{~Hz} ; 7.56^{2} J=8.0 \mathrm{~Hz} ; 7.55^{2} J=8.0 \mathrm{~Hz}$ ), corresponding to the H6 signal. In fact, the symmetric species A and $\mathbf{B}$ give one doublet each respectively
corresponding to the two overlapping H 6 protons of each species (Figure 7.12). In addition, the isomer $\mathbf{C}$ will give two doublets of the two non-overlapping signals of the H6 protons, due to the different chemical environments around the two protons generated by the different chirality of the two phosphoramidate groups.
The synthesised derivatives 129a-c were tested on malignant cell lines. In fact, the disulphide bond was thought to be cleaved in the reducing intracellular environment, similarly to both disulphide containing bioconjugates ${ }^{454}$ and other prodrugs in clinical development. ${ }^{455-457}$ These compounds (129a-c) would therefore be double prodrugs, requiring both intracellular reduction of the disulphide bridge and the enzymatic activation of the ProTide moiety.

### 7.4 Biological evaluation of 2'- ${ }^{\prime} \mathrm{SCF}_{3}$-2'-dU ProTides

### 7.4.1 In vitro cytotoxicity assay

The synthesised prodrugs of $2^{\prime}-$ SCF $_{3}-2^{\prime}$-dU (128a-c) were tested by WuXi AppTech on a small selection of malignant cell lines, described in Table 7.3.

| Cell line | Malignancy |
| :---: | :---: |
| MOLT-4 | Acute T lymphoblastic leukaemia |
| K562 | chronic myelogenous leukaemia |
| HL-60 | Acute promyelocytic leukaemia |
| MCF-7 | breast adenocarcinoma |
| Mia-Pa-Ca | breast adenocarcinoma |
| Table 7.3: Malignant cell lines selected for testing of prodrugs of 2'-SCF $\mathbf{3}_{3}$-2'-dU (128a-c) and 2'-SH-2'-dU |  |

Table 7.3: Malignant cell lines selected for testing of prodrugs of 2'- SCF $_{3}$ - $^{\prime}$ '-dU (128a-c) and $2^{\prime}-\mathrm{SH}-2^{\prime}-\mathrm{dU}$ (129a-c) by WuXi AppTech.

The cytotoxic screening of the parent nucleoside (115) on both leukaemic cell lines and breast cancer cell lines revealed very poor to negligible anticancer activity on all cell lines, with cell viability inhibition reaching a maximum of $16.5 \%$ on HL60 cell line at the highest concentration considered ( $198 \mu \mathrm{M}$ ).

| Cpnd | MOLT4 |  | K562 |  | HL60 |  | Mia-Pa-Ca |  | MCF7 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathbf{I C}_{50}$ | MI\% | IC ${ }_{\mathbf{5 0}}$ | MI\% | IC $\mathbf{5 0}$ | MI\% | IC $\mathbf{5 0}^{0}$ | MI\% | IC $\mathbf{5 0}$ | MI\% |
| 2'SCF ${ }^{\prime}{ }^{\prime}$ 'dU (115) | >198 | -2 | >198 | 4.2 | >198 | 16.5 | >198 | 4.6 | >198 | 11 |
| 128a | 33.0 | 100 | 52.0 | 106 | 45.0 | 101 | 61.0 | 99 | 80.0 | 94 |
| 128b | 104.0 | 66 | 139.0 | 50 | 86.0 | 71 | >198 | 35 | >198 | 34 |
| 128c | 18.0 | 102 | 23.0 | 100 | 23.0 | 102 | 38.0 | 99 | 58.0 | 99 |

Table 7.4: In vitro cytotoxicity screening of parent nucleoside $\mathbf{1 1 5}$ and ProTides (128a-c). Cytotoxicity data reported as $\mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentration considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

ProTides 128a-c were generally more active than the parent nucleoside, especially compounds 128a and 128c, respectively the phenyl and naphthyl-L-alanine benzyl ester ProTides, which reached complete inhibition of cell viability within the range of concentrations considered ( $19 \mu \mathrm{M}$ to $198 \mu \mathrm{M}$ ). Moreover, the $\mathrm{IC}_{50}$ value of compound 128c was as low as 18 and $23 \mu \mathrm{M}$ respectively on MOLT4 and HL60 cell lines. The significantly improved anticancer activities of ProTides 128a-c over the parent nucleoside suggested that the lack of activity of $\mathbf{1 1 5}$ could potentially derive from a poor phosphorylation to $2^{\prime}-\mathrm{SCF}_{3}-2^{\prime}$-dUMP, causing inefficient intracellular activation to the supposedly active 5 '-triphosphate species, or improved cell permeation. On the other hand, ProTides 128a-c should release $2^{\prime}$ '- $\mathrm{SCF}_{3}-2^{\prime}$ '-dUMP efficiently and therefore facilitate the intracellular metabolism to the active metabolite 5 '-triphosphate.

### 7.4.2 Mechanistic investigations

### 7.4.2.1 Enzymatic CPY experiment

Enzymatic carboxypeptidase-Y (CPY) experiment was performed on compound 128a in order to predict the likelihood of intracellular processing of this ProTide to L-alaninyl phosphate intermediate $\mathbf{1 2 8 a}^{\mathbf{C}}$ (Scheme 7.4).


Scheme 7.4: Intracellular conversion of $\mathbf{1 2 8 a}$ to $\mathbf{1 2 8 a}^{\mathbf{D}}$ catalysed by CPY/Hint enzymes.

From Figure 7.14, showing the ${ }^{31} \mathrm{P}$ NMR monitoring of this conversion, it is clear that ProTide 128a was quickly and completely converted to intermediate $\mathbf{1 2 8 a}^{\mathbf{A}}$ within 4 minutes from enzyme addition, and the total conversion into 5'-L-alaninyl phosphate monoester 128a ${ }^{\mathbf{C}}$ was complete after 32 minutes.


Figure 7.14: ${ }^{31} \mathrm{P}$ NMR spectra monitoring the conversion of $\mathbf{1 2 8 a}$ to $\mathbf{1 2 8 a}{ }^{\mathbf{C}}$ recorded in acetone-d6 (202 MHz ).

The presence of the $\mathrm{SCF}_{3}$ functionality allows the additional use of ${ }^{19} \mathrm{~F}$ NMR spectroscopy to evaluate the enzymatic conversion catalysed by CPY, and Figure 7.15 depicts the signals corresponding to starting material 128a ( $\delta_{\mathrm{F}}-39.43 \mathrm{ppm},-40.01 \mathrm{ppm}$ ) being converted into a more downfield single signal ( $\delta_{\mathrm{F}}-39.60 \mathrm{ppm}$ ), corresponding to $\mathbf{1 2 8 a}{ }^{\mathbf{C}}$.


Figure 7.15: ${ }^{19} \mathrm{~F}$ NMR spectra corresponding to $\mathbf{1 2 8 a}$ (at time 0 ) and $\mathbf{1 2 8}^{\text {C }}$ (at time $=13 \mathrm{~h}$ ) recorded in acetone-d6 ( 407 MHz ).

### 7.4.2.2 Docking into Hint active site

Prediction of the effective conversion of 2' $-\mathrm{SCF}_{3}-2$ ' -dU ProTides into the desired $2^{\prime}-\mathrm{SCF}_{3}{ }^{-}$ 2'-dUMP intracellular metabolite was performed by docking the L-alaninyl monophosphate intermediate $\mathbf{1 2 8 a}^{\mathbf{C}}$ into the active site of Hint. As shown in Figure 7.16 the intermediate appears to correctly position the phosphoramidate moiety close to the residues recognised to have a crucial role in the cleavage of such portion (Ser107, His112 and His114), and this suggests the intracellular release of the desired metabolite $2^{\prime}-\mathrm{SCF}_{3}{ }^{-}$ 2'-dUMP.


Figure 7.16: Docking of intermediate $\mathbf{1 2 8 a}^{\mathbf{C}}$ into the active site of Hint enzyme.

However, as noted also for the docking of other pyrimidine nucleoside derivatives within the same pocket (e.g. TFT), the rest of the molecule is exposed outside the active site, and this could potentially cause a less efficient interaction with the enzyme that may cause a slower processing of such prodrugs.

### 7.5 Biological evaluation of 2'-SH-2'-dU ProTides

### 7.5.1 In vitro cytotoxicity evaluation

The biological evaluation of $2^{\prime}-\mathrm{SH}-2^{\prime}$-dU (119) and relative disulphide-bond bearing ProTide dimers (129a-c) was performed by WuXi AppTech on the cell lines reported in Table 7.3. Parent nucleoside $\mathbf{1 1 9}$ was completely inactive in the assay conditions, causing no significant cell viability inhibition up to the highest concentration used ( $198 \mu \mathrm{M}$ ). ProTides 129a-c had very poor activity, despite causing a more effective inhibition of cell viability compared to the parent nucleoside, up to $49 \%$ at the highest concentrations used (such as $\mathrm{MI} \%=49.5$ recorded for compound $\mathbf{1 2 9}$ con HL60 cell line).

| Cpnd | MOLT4 |  | K562 |  | HL60 |  | Mia-Pa-Ca |  | MCF7 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathbf{I C}_{50}$ | MI\% | IC $\mathbf{5 0}$ | MI\% | IC $\mathbf{5 0}$ | MI\% | IC $\mathbf{5 0}^{\mathbf{0}}$ | MI\% | IC $\mathbf{5 0}^{0}$ | MI\% |
| 2'-SH-2'-dU (119) | >198 | 1 | >198 | 3 | >198 | -0.5 | >198 | 1 | >198 | 0.1 |
| 129a | $>198$ | 40 | >198 | 45 | >198 | 48 | $>198$ | 22 | >198 | 7 |
| 129b | >198 | 49 | >198 | 34 | >198 | 35 | >198 | 16 | >198 | 3 |
| 129c | >198 | 38 | >198 | 49 | >198 | 49 | >198 | 4 | >198 | 5 |

Table 7.5: In vitro cytotoxicity screening of parent nucleoside $\mathbf{1 1 9}$ and derivatives 129a-c. Cytotoxicity data reported as $\mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentration considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

The low activity of parent $\mathbf{1 1 9}$ potentially derives from different factors, such as inefficient cell penetration due to lack of recognition by a nucleoside transporter or poor phosphorylation by nucleoside and nucleotide kinases to active diphosphate 121. Moreover, $\mathbf{1 1 9}$ was already reported to be unstable at a range of $\mathrm{pH},{ }^{458}$ including $\mathrm{pH}=7$, close to physiological conditions, and this would decrease the percentage of nucleoside being phosphorylated to active species once inside the cell. At this pH , the prevalent species detected were dimer $\mathbf{1 2 2}$ and uracil, respectively in 44 and $27 \%{ }^{458}$
Regarding compounds $\mathbf{1 2 9 a} \mathbf{- c}$, it is important to specify that these potential 'bis-prodrugs' should undergo a double activation process inside the cell: disulphide bridge reduction to 2'-thiol functionality and ProTide activation to $2^{\prime}$ '-mercapto-2'-deoxyuridine 5'monophosphate ( $2^{\prime}$-SH-2'-dUMP). However, $2^{\prime}$ '-SH-2'-dUMP must also be recognised by a nucleotide kinase enzyme and be phosphorylated to diphosphate 121, the active species that could interact and inhibit RNR enzyme.

Cleavage of disulphide bonds as a way of activating prodrugs inside cancer cells has been extensively exploited, ${ }^{454-456,459,460}$ considering the higher amount of reducing agent
glutathione inside neoplastic cells compared to normal cells, ${ }^{461}$ which cleaves these bonds by disulphide exchange. ${ }^{459,462}$ Chemical cleavage of the disulphide bond in different nucleoside based compounds was performed by treatment with a reducing agent such as dithiothreitol (DTT). ${ }^{440,463-466}$ An NMR assay was set up by Gerland et al., ${ }^{465}$ in order to monitor the reduction of alkyl and aryl disulphide bonds in position $3^{\prime}$ on the uridine sugar ring by treatment with an excess ( 5 equivalents) of DTT at $20^{\circ} \mathrm{C} .{ }^{465}$ This group noted that the reduction was complete within 10 minutes from the DTT addition in the case of alkyl disulfides, while the presence of an aromatic moiety linked to the disulphide group caused increase in the half-life of these compounds up to 8 hours. The susceptibility of the disulphide bond cleavage in compound 129a was assayed in the same conditions.


Scheme 7.5: Disulfide bond cleavage of compound 129a yielding ProTide 120a. Reagents and conditions: DTT ( 5 eq), $\mathrm{CD}_{3} \mathrm{OD}$, rt, $12 \mathrm{~h}, 50 \%$ (analysed by ${ }^{1} \mathrm{H}$ NMR).
${ }^{1} H$ NMR evaluation of this reaction was performed over 12 hours, and key spectra are reported in Figure 7.17 (this conversion was also monitored by ${ }^{31} \mathrm{P}$ NMR, although not reported because overlapping of the signals corresponding to $\mathbf{1 2 9 a}\left(\delta_{\mathrm{p}} 3.97 \mathrm{ppm}, 3.63 \mathrm{ppm}\right.$, 3.59 ppm ) and 120a ( $\delta_{\mathrm{p}} 4.14 \mathrm{ppm}, 3.73 \mathrm{ppm}$ ) caused unclear spectra.


Figure 7.17: Stacked ${ }^{1} \mathrm{H}$ NMR spectra $\left(\mathrm{CD}_{3} \mathrm{OD}, 500 \mathrm{MHz}\right)$ showing the partial conversion of dimer 129a into ProTide 120a.

The disulfide bond cleavage reaches approximately $50 \%$ within 12 hours as calculated from the ratio of starting material over product, whose structure was confimed by analysis of the ${ }^{1} \mathrm{H}$ NMR new peaks that were formed (Figure 7.17), which matched with the structure of ProTide 120a. As an example, the enlarged portion (between 6.35-5.9 ppm) of Figure 7.17, shows the H1' doublets corresponding to compound 120a ( $\delta_{\mathrm{H}} 6.31 \mathrm{ppm}^{2} J=$ $8.5 \mathrm{~Hz} ; 6.30 \mathrm{ppm},{ }^{2} J=8.2 \mathrm{~Hz} ; 6.28 \mathrm{ppm},{ }^{2} J=8.2 \mathrm{~Hz}, 6.27 \mathrm{ppm},{ }^{2} J=8.5 \mathrm{~Hz}$ ) slowly converting into new doublets corresponding to the H1’ signals of the two diastereoisomers of ProTide 120a ( $\delta_{\mathrm{H}} 5.97 \mathrm{ppm},{ }^{2} J=9.0 \mathrm{~Hz} ; \delta 5.95 \mathrm{ppm},{ }^{2} J=9.5 \mathrm{~Hz}$ ).
Comparison between this reaction and the half-life of compounds reported by Gerland et $a l .{ }^{464}$ seems to suggest a relatively slow activation process for compound 129a. The low activity of compounds $\mathbf{1 2 9 a} \mathbf{- c}$ could therefore derive from either an ineffective reduction of the disulphide bond or by a slow processing of the ProTide moiety to $2^{\prime}$-SH-2'-dUMP.

In addition, there could be a lack of recognition of $2^{\prime}$-SH- ${ }^{\prime}$ '-dUMP by a nucleotide kinase, with poor conversion to active metabolite $2^{\prime}-$ SH-2'-dUDP.

### 7.5.2 Mechanistic investigations on the processing of 2'-SH-2'-dU derivatives

As previously mentioned, $2^{\prime}$ '-SH-2'-dU derivatives, prior to having an effect on cell viability, need to be processed to $2^{\prime}-\mathrm{SH}-2^{\prime}$-dUMP, which should then in turn undergo phosphorylation to $2^{\prime}$-SH-2'-dUDP. The intracellular prodrug processing requires both cleavage of the disulphide bond, which was simulated by treatment of derivative 129a with DTT (Figure 7.17), and cleavage of the ProTide moieties with release of the monophosphate group(s). The order in which these two processes would happen in the intracellular environment is not clear, therefore we decided to simulate the interaction of derivative 129a within the active site of CPY in order to understand whether the phosphoramidate moiety processing could happen first. Figure 7.18 shows the results obtained for the structure with $R_{\mathrm{p}} / R_{\mathrm{p}}$ stereochemistry (Figure 7.13, B), as an example of what resulted from the docking also of the other two isomers $\mathrm{S}_{\mathrm{p}} / S_{\mathrm{p}}$ and $R_{\mathrm{p}} / S_{\mathrm{p}}$ (Figure 7.13, A and $\mathbf{C}$ ), which gave similar results.


3D interactions


2D interactions

Figure 7.18: Docking of dimer 129a ( $\mathrm{Rp} / \mathrm{Rp}$ stereochemistry), within the active pocket of CPY enzyme.

Dimer 129a does not appear to fit appropriately within the CPY active site, as a result of the very bulky structure, as evident from the 2D ligand interactions, showing clashing of a portion of the molecule with amino acid residues in the pocket (red portion, Figure 7.18). These results suggest that a first ProTide activation would be unlikely to occur and dimer 129a should first undergo disulphide bond cleavage.

### 7.5.2.1 CPY docking of ProTide 120a in CPY

One possible mechanism of intracellular activation of $2^{\prime}$-SH-2'-dU derivatives 129a-c involves first cleavage of the disulphide bond, followed by processing of the ProTide moiety with final release of $2^{\prime}$-SH-2'-dUMP.


Scheme 7.6: Intracellular conversion of $\mathbf{1 2 0 a}$ to $\mathbf{1 2 0} \mathbf{a}^{\mathbf{D}}$ catalysed by CPY/Hint enzymes.

In order to predict the likelihood of processing of $2^{\prime}$-SH- $2^{\prime}$-dU ProTides by carboxypeptidase and phosphoramidase enzymes, docking of compound 120a was performed within the active pocket of CPY enzyme, which should cleave the benzyl ester moiety and trigger the conversion into metabolite $\mathbf{1 2 0 a}^{\mathbf{C}}$ (Scheme 7.6, Figure 7.19). The activation process was then finally investigated with the docking of the L-alaninyl intermediate $120 \mathrm{a}^{\mathrm{C}}$ within the Hint active pocket, in order to predict the likelihood of P-N bond cleavage with release of $2^{\prime}-$ SH- ${ }^{\prime}$ '-dUMP 120a ${ }^{\text {D }}$.


Figure 7.19: Docking of isomers of ProTide 120a within the active site of CPY.

Both isomers interact with the pocket with the amino acid ester moiety close to the key residues responsible for ester cleavage (Ser146, Gly52 and Gly53), however, the $S$ p isomer seems to position such moiety in closer proximity.

### 7.5.2.2 CPY docking of 2'-SH-2'-dU intermediate $120 \mathrm{a}^{\mathrm{C}}$ in Hint

Docking of intermediate $\mathbf{1 2 0 a}^{\mathbf{C}}$ in the active site of Hint (Figure 7.20) are predictive of a relatively efficient processing of the phosphoramidate group with release of $2^{\prime}$-SH-2'dUMP. In fact, such groups are positioned in proximity with the catalytic residues, and would thus be processed by the enzyme.


Figure 7.20 Docking of intermediate $\mathbf{1 2 0 a}{ }^{\mathbf{c}}$ in the active site of Hint enzyme.

The outcome of docking of ProTide 120a within CPY and intermediate $\mathbf{1 2 0 a}{ }^{\text {C }}$ within Hint active sites, in parallel with the docking results obtained for dimer 129a, suggest that the ProTide moiety is more likely to be processed after cleavage of the disulphide bond.

### 7.6 Conclusions

In the present work the anticancer potential of $2^{\prime}-\mathrm{SCF}_{3}-2^{\prime}-\mathrm{dU}$, already reported to be endowed with RNA-base pairing disrupting effects, was assessed in vitro for the first time. Moreover the cytotoxic properties of the $2^{\prime}-\mathrm{SH}-2^{\prime}-\mathrm{dU}$ were evaluated, as the diphosphate analogue of such molecule was reported to be an efficient inhibitor of the RNR enzyme, with potential anticancer activity. In addition, the ProTide approach was applied to both nucleoside analogues, with the aim of enhancing their properties by potentially bypassing limiting steps in the intracellular activation. ProTides of $2^{\prime}-\mathrm{SCF}_{3}-2$ '-dU were synthesised in moderate to good yields, while the NMI-mediated coupling reaction, applied to $2^{\prime}$ ' SH -$2^{\prime}$-dU, led to the synthesis of dimeric ProTide analogues, linked via a $2^{\prime}, 2^{\prime}$-SS disulphide
bond, as the $2^{\prime}$-thiol moiety is prone to oxidation. All derivatives were tested and $2^{\prime}-\mathrm{SCF}_{3}{ }^{-}$ 2 '-dU ProTides were found to enhance the very poor anticancer profile of the parent nucleoside, leading to $100 \%$ cell viability inhibition at concentrations as low as $18 \mu \mathrm{M}$, potentially a consequence of the very quick processing noted by CPY and the correct positioning within the active site of Hint. With regard to $2^{\prime}$ 'SH-2'-dU disulphide dimers, these compounds were poor anticancer compounds, leading to inhibition of cell viability up to $50 \%$ at the highest concentration considered $(198 \mu \mathrm{M})$. However, these derivatives improved the activity of the parent nucleoside, which was completely inactive up to 198 $\mu \mathrm{M}$. The poor activity of $2^{\prime}$-SH-2'-dU disulphide derivatives was proposed to derive from slow disulphide bond cleavage, as simulated via the DTT assay, leading to a delay in the processing of the ProTide moiety.
The promising results obtained after cytotoxic evaluation of proTides of $2^{\prime}-\mathrm{SCF}_{3}-2^{\prime}$ - dU should direct future investigations at applying the ProTide approach to other ${ }^{\prime}-\mathrm{SCF}_{3}$ bearing nucleosides such as compounds 128a-c.

Moreover, the synthesis and biological evaluation of prodrugs of 2'-SH-2'dUDP should be included in a future project, since the low activity of monophosphate prodrugs could also be a consequence of defective processing to diphosphate by nucleotide kinase. In addition, any $2^{\prime}$-SH nucleoside prodrug should be treated prior to cytotoxic screening with DTT, as disulphide bond cleavage might potentially be an issue for such hindered molecules. Alternatively, exploration of structures such as $\mathbf{1 3 2}$ (Figure 7.21) could lead to potential double prodrugs of $2^{\prime}-\mathrm{SH}-2^{\prime} \mathrm{dU}$.


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Figure 7.21: Potential new prodrugs of 2'-SH-2'-dU.

The small disulphide moiety in the $2^{\prime}$ 'position should be easily cleaved intracellularly, as already suggested by previous studies. ${ }^{464}$ This would allow the synthesis of prodrugs of 2'-SH-2'dU endowed with improved chemical stability and potentially better cytotoxicity.

## 8 Conclusions

Herein is reported the design, synthesis and biological evaluation of different families of ProTides as potential therapeutic agents. All synthesised compounds were tested on a broad selection of cell lines, and in some cases as antiviral agents.

The mechanism of activation of the ProTides was supported by studying their bioactivation to the corresponding monophosphate forms, through enzymatic NMR studies and molecular modeling simulations.

The most active analogues were tested on resistant cancer cell lines, and their stability was assayed in serum, plasma and liver microsomes. Furthermore, the investigation of the selective targeting of the most promising ProTides on the stem cell population of the KG1a leukaemic cell line was carried out.

8-Chloroadenosine prodrugs were found to be endowed with higher cytotoxic activity compared to the parent nucleoside in conditions of AK of hENT1 deficiency, such as the 1-napthyl-L-alanine benzyl ester ProTide 71a. The 8-ClA phosphorodiamidate 76b also emerged as one of the most potent derivatives.

Regarding the families of ProTides of TFT, CNDAC, 2'-SCF3-2'-dU and 2'-SH-2'-dU, many compounds were found to be more potent than the parent nucleosides, validating the potential of applying the ProTide approach as a way to improve the activity of nucleoside analogues.

Moreover, within the 3 '-deoxyadenosine family of prodrugs, compounds with nearly 300 fold improved activity compared to their parent nucleoside were synthesised, with the two most potent derivatives, the phenyl L-alanine benzyl ester ProTide 30a and the 2-fluoro counterpart 49b, with low to sub micromolar activity. These analogues were evaluated in a mouse model of human non-Hodgkin's lymphoma, leading to promising results. These compounds represent potential new anticancer agents and will be further investigated in alternative tumour in vivo models.

Alteration of the phosphate structure of 30a using the phosphonoamidate prodrug approach generated a compound, $\mathbf{6 4}$, with improved anticancer activity in comparison to 3 ' dA .

## 9 Experimental part

### 9.1 General experimental details

All solvents used were anhydrous and used as supplied by Sigma-Aldrich. All commercially available reagents were supplied by Sigma-Aldrich, Fisher, Carbosynth or TCI chemicals and used without further purification. All nucleosides and solid reagents were dried for several hours under high vacuum prior to use.

All glassware was oven-dried at $130^{\circ} \mathrm{C}$ for several hours or overnight and allowed to cool in a desiccator or under a stream of dry nitrogen.

For analytical thin-layer chromatography (TLC), precoated aluminium-backed plates (60 F-54, 0.2 mm thickness; supplied by E. Merck AG, Darmstadt, Germany) were used and developed by an ascending elution method. For preparative thin-layer chromatography (prepTLC), preparative TLC plates ( $20 \mathrm{~cm} \times 20 \mathrm{~cm}, 500-2000 \mu \mathrm{~m}$ ) were purchased from Merck. After solvent evaporation, compounds were detected by quenching of the fluorescence, at 254 nm upon irradiation with a UV lamp.

Column chromatography purifications were carried out by means of normal or reverse phase silica gel flash chromatography or automatic Biotage Isolera One. Fractions containing the product were identified by TLC and pooled, and the solvent was removed under vacuum.
${ }^{1} \mathrm{H},{ }^{31} \mathrm{P},{ }^{13} \mathrm{C}$ and ${ }^{19} \mathrm{~F}$ NMR spectra were recorded on a Bruker Avance 500 spectrometer operating at $500 \mathrm{MHz}, 202 \mathrm{MHz}$ and 125 MHz and 407 MHz respectively and autocalibrated to the deuterated solvent reference peak. All ${ }^{31} \mathrm{P}$ and ${ }^{13} \mathrm{C}$ NMR spectra were proton-decoupled.
Chemical shifts are given in parts per million (ppm) and coupling constants (J) are measured in Hertz. The following abbreviations are used in the assignment of NMR signals: s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet), br s (broad singlet), dd (doublet of doublet), ddd (doublet of doublet of doublet), dt (doublet of triplet). The assignment of the signals in ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR was done based on the analysis of coupling constants and additional two-dimensional experiments (COSY, HSQC).

All analytical high-performance liquid chromatography (HPLC) experiments were done on a Thermo Fisher Scientific Spectra System SCM1000 provided with a System Controller SN4000, a pump Spectra System P4000 and a Spectra UV2000 detector set or a Varian Prostar (LC Workstation-Varian Prostar 335 LC detector) using a C18-Varian Pursuit (150 $\times 4.6 \mathrm{~mm}, 5 \mu \mathrm{~m})$ reverse phase column. All final compounds were isolated with purity $\geq$ 95\%

Low resolution mass spectrometry was perfomed on a Bruker Daltonics microTof-LC system (atmospheric pressure ionization, electron spray mass spectroscopy) in positive mode.

UV experiments were conducted using a Varian 50 Bio UV-visible spectrophotometer.

### 9.2 Molecular modelling studies

All molecular modelling studies were performed on a MacPro dual 2.66 GHz Xeon running Ubuntu 9.10, using molecular operating environment (MOE) 2008.10. ${ }^{467}$

The structures of adenosine deaminase (ADA, PDB 3IAR), carboxypeptidase Y (CPY, PDB 1YSC), ${ }^{468}$ human Hint1 (Hint, PDB 1RZY) ${ }^{266}$ enzymes were downloaded from the PDB website (http://www.rcsb.org/).

Hydrogen atoms were added to the protein and all minimisations were performed with MOE keeping all heavy atoms fixed until RMSD gradient of $0.05 \mathrm{kcal} \mathrm{mol}^{-1}$ was reached with MMFF94x force field and partial charges were automatically calculated. Ligands in mol2 format, prepared using MOE, were minimised using the default settings and used as input for the docking. Ligand docking was performed using the default values configured, using PLANTS 1.1.469 The binding site center and radius corresponding to each enzyme considered were as follows:

- [ADA] binding site center: $9.386,-1.572,0.772$; binding site radius: $9 \AA$.
- [Hint] binding site center: 30.48, 4.591, 32.494; binding site radius: 10 Å.
- [CPY] binding site center: 54.903, -1.594, 31.704; binding site radius: $9 \AA$.

Docking parameters: aco_ants 20; aco_evap 1; aco_sigma 1; flip_amide_bonds 0; flip_planar_n 0; force_flipped_bonds_planarity $0 ;$ force_planar_bond_rotation 1 ; rescore_mode simplex
scoring_function chemplp; outside_binding_site_penalty 50; enable_sulphur_acceptors 0 ;
ligand_intra_score clash2; chemplp_clash_include_14 0.25; chemplp_clash_include_HH 0 ; chemplp_weak_cho 1 ; chemplp_charged_hb_weight 2 ; chemplp_charged_metal_weight 2; chemplp_hbond_weight -3; chemplp_hbond_cho_weight -3; chemplp_metal_weight -3; chemplp_plp_weight 1 ; chemplp_plp_steric_e -0.4 ; chemplp_plp_burpolar_e -0.1; chemplp_plp_hbond_e -1 ; chemplp_plp_metal_e -1 ; chemplp_plp_repulsive_weight 1 ; chemplp_tors_weight 2 ; chemplp_lipo_weight $0 ;$ chemplp_intercept_weight -20 ; write_protein_conformations 0; write_protein_bindingsite 0; write_protein_splitted 0; write_rescored_structures $0 ; \quad$ write_multi_mol2 $\quad 1 ;$ write_ranking_links 0 ; write_ranking_multi_mol2 0 ; write_per_atom_scores 0 ; write_merged_ligand 0 ; write_merged_protein 0 ; write_merged_water 0 ; keep_original_mol2_description 1 ; merge_multi_conf_output $0 ;$ merge_multi_conf_character ; merge_multi_conf_after_characters 1 ; rigid_ligand 0 ; rigid_all 0 .
Twenty-five output configurations were obtained from each input compound and a visual inspection, in MOE, was used to identify interaction types between ligand and protein (discussed in each chapter).

### 9.3 Enzymatic and stability assays

### 9.3.1 Carboxypeptidase $Y$ assay

The appropriate substrate ( $5 \mathrm{mg}, \pm 0.008 \mathrm{mmol}$ ) was dissolved in $150 \mu \mathrm{~L}$ of acetone- $d 6$ and $300 \mu \mathrm{~L}$ of TRIZMA buffer ( pH 7.6 ) was added. $\mathrm{A}^{31} \mathrm{P}$ NMR ( $202 \mathrm{MHz}, 64-128$ scans) was conducted at this stage as a reference (blank, $t=0$ ). To this mixture, $130 \mu \mathrm{~L}$ of a stock solution of carboxypeptidase enzyme (Purchased from Sigma Aldrich, > 50 unit $/ \mathrm{mg}$, dissolved in TRIZMA buffer 7.6 pH to a concentration of 50 units $/ \mathrm{mL}$, EC 3.4.16.1) was added. ${ }^{31} \mathrm{P}$ NMR ( 128 scans) were carried out with 1 minute of delay between experiments for 14 hours at $25^{\circ} \mathrm{C}$.

### 9.3.2 Adenosine deaminase UV assay

Bovine recombinant adenosine deaminase (ADA, EC:3.5.4.4) was purchased from Sigma (Sigma-Aldrich, product number 59722). The ADA suspension (containing $>2 \mathrm{U} / \mathrm{mL}$ ), was aliquoted into $55 \mu \mathrm{~L}$ portions and kept at $-20^{\circ} \mathrm{C}$. Before use, a $55 \mu \mathrm{~L}$ sample was thawed and 1 mL of phosphate buffer ( $0.1 \mathrm{~N}, \mathrm{pH} 7.6$ ) was added. The suspension was filtered through a $0.2 \mu \mathrm{~m}$ filter to remove precipitates that could interfere with the absorbance measurements. The clear supernatant was used for the degradation
experiments. The experiments were conducted in UV transparent cuvettes (Sigma Aldrich, code: Z637092) using an Envision microplate reader (PerkinElmer). Samples contained $62.5 \mu \mathrm{M}$ of substrate, which were added to $17 \mu \mathrm{~L}$ of the ADA solution. Scans of the absorption spectrum were taken between 220 and 350 nm with 2 nm resolution every 0.1 min over a period of 10 minutes to 12 hours. One sample containing only ADA was also prepared, and this was subtracted to obtain the spectrum over time for substrates.

### 9.3.3 Serum stability assay

The appropriate substrate ( $5 \mathrm{mg}, \pm 0.008 \mathrm{mmol}$ ) was dissolved in $100 \mu \mathrm{~L}$ of DMSO- $d 6$ and $300 \mu \mathrm{~L}$ of $\mathrm{D}_{2} \mathrm{O} . \mathrm{A}{ }^{31} \mathrm{P}$ NMR ( $202 \mathrm{MHz}, 256$ scans) was conducted at this stage as a reference (blank, $t=0$ ). To this mixture, $300 \mu \mathrm{~L}$ of a stock solution of rat/human serum (Purchased from Sigma Aldrich, codes R9759 and H4522) was added. ${ }^{31}$ P NMR (256 scans) were carried out with 1 minute of delay between experiments for 14 hours at $37^{\circ} \mathrm{C}$.

### 9.3.4 KG1a cell culture conditions

The acute myeloid leukaemia (AML) KG1a cell line was maintained in RPMI medium (Invitrogen, Paisley, UK) supplemented with 100 units $/ \mathrm{mL}$ penicillin, $100 \mu \mathrm{~g} / \mathrm{mL}$ streptomycin and $20 \%$ foetal calf serum. Cells were subsequently aliquoted $\left(10^{5}\right.$ cells $/ 100 \mu \mathrm{~L}$ ) into 96 -well plates and were incubated at $37^{\circ} \mathrm{C}$ in a humidified $5 \%$ carbon dioxide atmosphere for 72 hours in the presence of nucleoside analogues and their respective ProTides at concentrations that were experimentally determined for each series of compounds. In addition, control cultures were carried out to which no drug was added. Cells were subsequently harvested by centrifugation and were analysed by flow cytometry using the Annexin V assay.

### 9.3.5 Measurement of in vitro apoptosis

Cultured cells were harvested by centrifugation and then resuspended in $195 \mu \mathrm{~L}$ of calcium-rich buffer. Subsequently, $5 \mu \mathrm{~L}$ of Annexin V (Caltag Medsystems, Botolph Claydon, UK) was added to the cell suspension and cells were incubated in the dark for 10 min prior to washing. Cells were finally resuspended in $190 \mu \mathrm{~L}$ of calcium-rich buffer together with $10 \mu \mathrm{~L}$ of propidium iodide. Apoptosis was assessed by dual-colour immunofluorescent flow cytometry. Subsequently $\mathrm{LD}_{50}$ values (the dose required to kill $50 \%$ of the cells in a culture) were calculated for each nucleoside analogue and ProTide.

### 9.3.6 Immunophenotypic identification of the leukaemic stem cell compartment

Once the $\mathrm{LD}_{50}$ value for each compound was established, KGla cells were cultured with the $\mathrm{LD}_{50}$ concentration of each compound for 72 h . Cells were then harvested and labelled with $5 \mu \mathrm{~L}$ each of CD34-FITC, CD38-PE and CD123 PERCP-cy5 antibodies. The $\mathrm{CD} 34^{+} / \mathrm{CD} 38^{-} / \mathrm{CD} 123^{+}$sub-population in each culture was quantified and expressed as a proportion of the total culture. The effect of each compound on the stem cell compartment was compared with untreated cultures labelled with the same antibodies.

### 9.3.7 Identification of $\boldsymbol{\beta}$-catenin levels in the leukaemic stem cell compartment

KG1a cells were cultured with the $\mathrm{LD}_{50}$ concentration of each compound for 72 h . Cells were then harvested and labelled with $5 \mu \mathrm{~L}$ each of CD34-FITC, CD38-PE and CD123 PERCP-cy5 antibodies and with anti- $\beta$-catenin-FITC. Before adding the last $\beta$-catenin specific antibody, cell permeability was increased by incubation with PBS-Tween. The levels of $\beta$-catenin were quantified and compared in $\mathrm{CD} 34^{+} / \mathrm{CD} 38^{-} / \mathrm{CD} 123^{+}$sub-population in each culture and in the total culture.

### 9.4 Standard synthetic procedures

### 9.4.1 Standard procedure $A_{1}$ for the synthesis of amino acid ester hydrochloride salt

 The appropriate alcohol ( 15 eq ) was cooled to $0{ }^{\circ} \mathrm{C}$ and $\mathrm{SOCl}_{2}(2 \mathrm{eq})$ was added dropwise. After stirring for 30 minutes at $0{ }^{\circ} \mathrm{C}$ the solution was allowed to reach rt . The appropriate amino acid (1 eq) was added and the reaction mixture was stirred at reflux overnight. The solvent was then removed under vacuum and co-evaporation with increasingly volatile solvents yielded the hydrochloride salt as a white solid. ${ }^{105}$
### 9.4.2 Standard procedure $A_{2}$ for the synthesis of amino acid ester $p$-toluene sulfonate salts

A mixture of the appropriate amino acid (1 eq), the appropriate alcohol (5eq), and $p$ toluene sulfonic acid ( $p$-TSA) monohydrate ( 1.1 eq ), in toluene ( $20 \mathrm{~mL} / \mathrm{mmol}$ ) was heated at reflux overnight, using a Dean-Stark apparatus. The solvent was removed under reduced pressure and co-evaporation with increasingly volatile solvents yielded the solid $p$-toluene sulfonate salt as a white solid. ${ }^{105}$

### 9.4.3 Standard procedure $B$ for the synthesis of aryl phosphorodichloridate

Phosphorus oxychloride ( 1 eq ) and the appropriate substituted aryl alcohol ( 1 eq ) were stirred in $\mathrm{Et}_{2} \mathrm{O}\left(5 \mathrm{~mL}\right.$ per 1 mmol of $\left.\mathrm{POCl}_{3}\right)$. The solution was cooled to $-78{ }^{\circ} \mathrm{C}$ and $\mathrm{Et}_{3} \mathrm{~N}$ ( 1 eq ) was added. After 30 minutes the solution was allowed to reach rt and stirred for 1.5 hours. The triethylamine hydrochloride salt was filtered off through a sintered funnel and the solvent removed under vacuum to afford the crude product as a clear oil. ${ }^{105}$

### 9.4.4 Standard procedure $C$ for the synthesis of aryl amino acid ester phosphorochloridates from amino acid hydrochloride or $\boldsymbol{p}$-toluene sulfonate salts

The appropriate amino acid ester salt (1 eq) was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \mathrm{~mL}$ per mmol of amino acid ester) under Argon atmosphere. A solution of the appropriate phosphorodichloridate ( 1 eq ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL}$ per mmol of amino acid ester) was added and the mixture cooled to $-78{ }^{\circ} \mathrm{C} . \mathrm{Et}_{3} \mathrm{~N}(2 \mathrm{eq})$ was added dropwise and the reaction mixture was stirred at $-78^{\circ} \mathrm{C}$ for 20 minutes and thereafter at rt for 2.5 hours.

In the case of amino acid ester hydrochloride salt as starting material, the solvent was evaporated under reduced pressure and the resulting oil was triturated with anh. $\mathrm{Et}_{2} \mathrm{O}$ and the filtrate reduced to give the crude product as an oil.
In the case of amino acid ester $p$-toluene sulfonate salt as starting material, the solvent was evaporated under vacuum and the crude was purified by silica gel column chromatography ( $50 \% \mathrm{EtOAc} /$ hexane), to yield the title compound as a white solid. ${ }^{105}$

### 9.4.5 Standard procedure $D$ for the synthesis of $\mathbf{2}^{\prime}, 3$ '-anhydrous adenosine analogues

To a stirring suspension of the appropriate adenosine analogue (1 eq) in $\mathrm{CH}_{3} \mathrm{CN}(7 \mathrm{~mL}$ per mmol of adenosine analogue), $\alpha-\operatorname{AIBBr}(4 \mathrm{eq})$ and $\mathrm{H}_{2} \mathrm{O}(0.001 \mathrm{eq})$ were added, and stirring was continued at room temperature. After 1 hour, the mixture was neutralised by addition of a saturated solution of $\mathrm{NaHCO}_{3}$ (a change in colour from dark orange to clear-white could be noticed) and the solution was extracted with EtOAc ( $2 \times 5 \mathrm{~mL}$ per mmol of adenosine analogue). The combined organic phase was washed with brine ( $1 \mathrm{~mL} x \mathrm{mmol}$ of adenosine analogue). The aqueous phase was extracted with EtOAc ( $1 \times 5 \mathrm{~mL}$ per mmol of adenosine analogue) and the combined organic phase was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to give a white gum. The crude mixture was dissolved in the appropriate
solvent $\left(\mathrm{CH}_{3} \mathrm{OH}\right.$, EtOH , or $\mathrm{THF} / \mathrm{H}_{2} \mathrm{O}(4 / 1), 7 \mathrm{~mL}$ per mmol of adenosine analogue initially used) and stirred for 1 to 16 hours with Amberlite $\left(2 \mathrm{x} \mathrm{OH}^{-}\right)$resin ( 4 mL per mmol of initial adenosine analogue used, $\geq 1.1 \mathrm{meq} / \mathrm{mL}$ by wetted bed volume), previously washed well with the appropriate solvent $\left(\mathrm{CH}_{3} \mathrm{OH}, \mathrm{EtOH}\right.$ or THF). The solution was then filtered and the resin carefully washed with $\mathrm{CH}_{3} \mathrm{OH}$ until no spot of the product by TLC could be detected in the filtrate. Evaporation of the combined filtrate and crystallisation of the residue from EtOH gave the appropriate $2^{\prime}, 3^{\prime}$-anhydrous adenosine analogue as a white powder.

### 9.4.6 Standard procedure $\mathbf{E}$ for the synthesis of 3'-deoxyadenosine analogues

The appropriate $2^{\prime}, 3^{\prime}$ '-anhydrous adenosine analogue ( 1 eq ) was dissolved in DMSO (1.5 mL per mmol of starting material) and diluted with THF ( 15 mL per mmol of starting material), under Argon atmosphere. The solution was cooled down in an ice-cold bath and a solution of $\mathrm{LiEt}_{3} \mathrm{BH}$ ( 1 M in THF, 3 eq ) was added dropwise.
Stirring was continued at $\sim 4{ }^{\circ} \mathrm{C}$ for 1 h and at rt for 16 h . The mixture was cooled down to $0{ }^{\circ} \mathrm{C}$ and an additional portion of $\mathrm{LiEt}_{3} \mathrm{BH}(1 \mathrm{M}$ in THF, 3 eq ) was added dropwise. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 1 hour and then at rt for one additional hour. The reaction mixture was carefully acidified $\left(5 \% \mathrm{AcOH} / \mathrm{H}_{2} \mathrm{O}\right)$, purged with $\mathrm{N}_{2}$ for 1 h (under the fume hood) to remove pyrophoric triethylborane, and evaporated. The residue was chromatographed on silica gel (see specifications for each nucleoside) to give the title compound as a white powder.

### 9.4.7 Standard procedure $\mathbf{F}_{\mathbf{1}}$ for the synthesis of ProTides

Nucleoside ( 1 eq ) was dissolved in THF ( 20 mL per 1 mmol of nucleoside) under Argon atmosphere. $t \mathrm{BuMgCl}(1.0 \mathrm{M}$ in THF, 1.1-3 eq) was added dropwise. The appropriate phosphorochloridate (3-4 eq) was dissolved in THF ( 4 mL per 1 mmol of phosphorochloridate) and added to the initial mixture. The mixture was stirred overnight and the solvent evaporated under vacuum. The obtained crude was purified by silica gel CC or Biotage Isolera One. In some cases further purification by prep. TLC and prep. HPLC was necessary. The title compound was obtained as a white solid.

### 9.4.8 Standard procedure $\mathbf{F}_{\mathbf{2}}$ for the synthesis of ProTides

Nucleoside ( 1 eq ) was dissolved in THF ( 7 ml per 0.1 mmol of nucleoside) under Argon atmosphere. The appropriate phosphorochloridate (3-4 eq) was dissolved in THF ( 4 mL per 1 mmol of phosphorochloridate) and added to the initial mixture, followed by NMI ( 5 eq). The mixture was stirred for 12-16 hours and the solvent evaporated under vacuum. The obtained crude was purified by silica gel CC or Biotage Isolera One. In some cases further purification by prep. TLC and/or prep. HPLC was necessary. The title compound was obtained as a white solid.

### 9.4.9 Standard procedure $G$ for the synthesis of phosphorodiamidates.

To a solution of nucleoside ( 1 eq ) in trimethyl phosphate ( 20 mL per 1 mmol of nucleoside), phosphoryl chloride ( 1 eq ) was added dropwise at $-5^{\circ} \mathrm{C}$ and the reaction mixture was left stirring for 4 hours. The formation of the intermediate was monitored by ${ }^{31}$ P-NMR. A suspension of the appropriate amino acid ester ( 5 eq ) in dichloromethane (4 mL per 1 mmol of amino acid ester) was added followed by diisopropyl ethyl amine (10 eq) at $-78{ }^{\circ} \mathrm{C}$. After stirring at room temperature for 20 hours, water ( 10 mL per mmol of nucleoside) was added and the layers were separated. The aqueous phase was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ (10 mL per mmol of nucleoside) and the organic phase washed with brine (10 mL per mmol of nucleoside). The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated. The residue was purified by column chromatography or Biotage Isolera One to give the title compound as a white solid.

### 9.5 Experimental details

### 9.5.1 Synthesis of aryl alcohol

## [13c] Ethyl 3-(2-hydroxyphenyl)propanoate



2,3-Dichloro-5,6-dicyano-1,4-benzoquinone ( $771 \mathrm{mg}, 3.4 \mathrm{mmol}$ ) was added to a solution of 3,4-dihydrocoumarin $(7.63 \mathrm{~g}, 51.5 \mathrm{mmol})$ in EtOH ( 40 mL ) and the resulting orange solution was stirred at room temperature for 8 hours. The crude was purified by flash column chromatography (eluent: $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to afford $\mathbf{1 3 c}$ as a yellow solid ( $9.6 \mathrm{~g}, 96 \%$ ). Lit. $\mathrm{mp}=36-36.5^{\circ} \mathrm{C} .{ }^{150}$ ${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 7.29-7.20 (m, 1H, Ar), 7.19-7.16 (m, 1H, Ar), 7.10-7.05 ( $\mathrm{m}, 2 \mathrm{H}, \mathrm{Ar}$ ), $4.12\left(\mathrm{q}, ~ J=7.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 3.00(\mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), $2.61\left(\mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ ), $1.22(\mathrm{t}, J=7.2 \mathrm{~Hz}$, $3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ).

### 9.5.2 Synthesis of aryl phosphorodichloridates

## [14b] Naphth-1-yl dichlorophosphate



Prepared according to the general procedure $\mathbf{B}$ using $\mathrm{POCl}_{3}(3.9 \mathrm{~mL}, 41.8$ mmol), 1-naphthol ( $6.06 \mathrm{~g}, 42.03 \mathrm{mmol}$ ) and $\mathrm{Et}_{3} \mathrm{~N}(5.9 \mathrm{~mL}, 42.3 \mathrm{mmol})$. The product was obtained as a clear oil $(9.98 \mathrm{~g}, 91 \%) .{ }^{105}$
${ }^{\mathbf{3 1}} \mathbf{P} \mathbf{N M R}\left(\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{P}} 3.69\left[\mathrm{lit} .{ }^{31} \mathrm{P}\right.$ NMR ( $202 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\boldsymbol{\delta}_{\mathrm{P}}$ 3.72]. ${ }^{147}$
${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.14(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 8), 7.90(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}$, H5), 7.81 (d, $J=8.3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4), 7.68-7.54$ (m, 3H, H2, H6, H7), 7.46 (d, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}$, H3).

## [14c] Ethyl 3-(2-hydroxyphenyl)propanoyl dichlorophosphate

Prepared according to the general procedure $\mathbf{B}$ using ethyl 3-(2-
 hydroxyphenyl)propanoate [13c] ( $5.17 \mathrm{~g}, 26.62 \mathrm{mmol}$ ), $\mathrm{POCl}_{3}$ $(2.48 \mathrm{~mL} \mathrm{~g}, 26.62 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(3.71 \mathrm{~mL}, 26.62 \mathrm{mmol})$. The product was obtained as a clear oil $(7.45 \mathrm{~g}, 90 \%) .{ }^{150}$
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathrm{CDCl}_{3}$ ) $\boldsymbol{\delta}_{\mathrm{P}} 3.22$.
${ }^{1} \mathbf{H}$ NMR ( $\left.\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{H}} 7.43-7.40(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ar}), 7.36-7.33(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ar}), 7.31-7.23$
( $\mathrm{m}, 2 \mathrm{H}, \mathrm{Ar}$ ), $4.15\left(\mathrm{q}, ~ J=7.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 3.06(\mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}$,
$\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 2.67 ( $\mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), $1.25(\mathrm{t}, J=7.2 \mathrm{~Hz}$, $3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ).

### 9.5.3 Synthesis of amino acid ester salts

[16a] L-Leucine pent-1-yl ester hydrochloride salt
Prepared according to the general procedure $\mathbf{A}_{\mathbf{1}}$ using $\mathrm{SOCl}_{2}$

( $6.10 \mathrm{~mL}, 84.12 \mathrm{mmol}$ ), $n$-pentanol ( 68.57 mL ) and L-leucine $(5.52 \mathrm{~g}, 42.06 \mathrm{mmol})$. The product was obtained as a white solid $(9.80 \mathrm{~g}, 98 \%) .{ }^{105}$
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{M H z}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.72$ (br s, $3 \mathrm{H}, \mathrm{NH}_{3}$ ), 4.18-4.09 (m, 2 H , $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), $3.88\left(\mathrm{t}, J=7.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 1.82-1.73 $\left(\mathrm{m}, 1 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 1.74-1.64 (m, 2H, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 1.64-1.57 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 1.35-1.27 (m, $4 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 0.93-0.85 (m, $9 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu).
[16b] L-Leucine ethyl ester hydrochloride salt


Prepared according to the general procedure $\mathbf{A}_{\mathbf{1}}$ using $\mathrm{SOCl}_{2}(5.53 \mathrm{~mL}$, $76.24 \mathrm{mmol})$, ethanol ( 33.39 mL ) and L-leucine ( $5.0 \mathrm{~g}, 38.12 \mathrm{mmol}$ ). The product was obtained as a white solid ( $7.01 \mathrm{~g}, 94 \%$ ).
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.85\left(\mathrm{br} \mathrm{s}, 3 \mathrm{H}, \mathrm{NH}_{3}\right), 4.32-4.21(\mathrm{~m}$, $\left.2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right), 4.06\left(\mathrm{t}, J=7.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 2.05-1.94 (m, 2 H , $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu) 1.89-1.80 (m, $1 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 1.32 ( $\mathrm{t}, \mathrm{J}=7.12$ $\left.\mathrm{Hz}, 3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right), 0.99\left(\mathrm{~d}, J=5.4 \mathrm{~Hz}, 6 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu).
[16c] L-Phenylalanine ethyl ester hydrochloride salt


Prepared according to the general procedure $\mathbf{A}_{\mathbf{1}}$ using $\mathrm{SOCl}_{2}(3.16 \mathrm{~mL}$, 43.54 mmol ), ethanol ( 19.07 mL ) and L-phenylalanine ( $3.60 \mathrm{~g}, 21.77$ $\mathrm{mmol})$. The product was obtained as a white solid $(4.20 \mathrm{~g}, 84 \%)$.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.72\left(\mathrm{br} \mathrm{s}, 3 \mathrm{H}, \mathrm{NH}_{3}\right), 7.34-7.25(\mathrm{~m}$, 5H, Ar L-Phe), 4.43-4.36 (m, 1H, CHCH2Ph L-Phe), 4.18-4.10 (m, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}$ ), 3.413.31 (m, 2H, CHCH ${ }_{2} \mathrm{Ph}$ L-Phe), $1.20-1.12\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right) .{ }^{467}$
[16d] L-Alanine pent-1-yl ester tosylate salt


Prepared according to the general procedure $\mathbf{A}_{2}$ using $n$-pentanol ( 16.4 mL ), L-alanine ( 2.7 g , $30.3 \mathrm{mmol})$ and $p-$ TSA ( $6.3 \mathrm{~g}, 33.3 \mathrm{mmol}$ ) and toluene $(109 \mathrm{~mL})$. The product was obtained as a white solid $(8.03 \mathrm{~g}, 80 \%) .{ }^{151}$
${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.17\left(\mathrm{br} \mathrm{s}, 3 \mathrm{H}, \mathrm{NH}_{3}\right), 7.77(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}, \operatorname{Ar} p$-TSA), 7.15 (d, $J=8.0 \mathrm{~Hz}, 2 \mathrm{H}, \operatorname{Ar} p$-TSA), 4.10-3.97 (m, $3 \mathrm{H}, \mathrm{CHCH}_{3}$ L-Ala, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n$-Pen), 2.36 (s, $3 \mathrm{H}, \quad \mathrm{CH}_{3} \quad p$-TSA), $1.59-1.51 \quad(\mathrm{~m}, \quad 2 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 1.46 (d, $J=7.4 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CHCH}_{3}$ L-Ala), 1.32-1.20 (m, 4H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 0.88 ( $\mathrm{t}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen).
[16e] L-Alanine cyclohex-1-yl ester tosylate salt


Prepared according to the general procedure $\mathbf{A}_{2}$ using cyclohexanol ( 23.4 mL ), L-alanine ( 4.0 g , 44.9 mmol ), p-TSA ( $9.39 \mathrm{~g}, 49.39 \mathrm{mmol}$ ) and toluene ( 102 mL ). The product was obtained as a
white solid ( $13.57 \mathrm{~g}, 88 \%$ ). ${ }^{151}$
${ }^{1} \mathbf{H} \operatorname{NMR}\left(500 \mathbf{M H z}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.52\left(\mathrm{br} \mathrm{s}, 3 \mathrm{H}, \mathrm{NH}_{3}\right), 7.79(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}, \operatorname{Ar} p$-TSA), $7.25\left(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar} p\right.$-TSA), 4.90-3.87 (m, 1H, $\left.\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 4.24-4.10(\mathrm{~m}, 1 \mathrm{H}$, $\left.\left.\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 2.31\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3} p \text {-TSA), 1.89-1.61 (m, 2H, CH(CH2 }\right)_{5} \mathrm{cHex}\right), 1.66-1.52$ (m, $3 \mathrm{H}, \mathrm{CHCH}_{3}$ L-Ala), 1.59-1.11 (m, $\left.8 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$.

## [16f] Glycine pent-1-yl ester tosylate salt



Prepared according to the general procedure $\mathbf{A}_{2}$, using 1-pentanol ( 11.87 mL ), glycine ( 1.64 g , 21.85 mmol ), p-TSA ( $4.57 \mathrm{~g}, 24.03 \mathrm{mmol}$ ) and toluene ( 113 mL ). The product was obtained as a white solid ( $6.66 \mathrm{~g}, 96 \%$ ).
${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.01\left(\mathrm{br} \mathrm{s}, 3 \mathrm{H}, \mathrm{NH}_{3}\right), 7.73(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}, \operatorname{Ar} p-\mathrm{TSA})$, $7.12\left(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}, \operatorname{Ar} p\right.$-TSA), $4.01\left(\mathrm{t}, J=6.87 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), 3.72-3.67 (m, 2H, CH Gly), 2.35 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3} p$-TSA), 1.55-1.48 (m, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 1.32-1.19 (m, 4H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 0.90-0.86 (m, $3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen).

## [16g] Glycine cyclohexyl ester tosylate salt



Prepared according to the general procedure $\mathbf{A}_{\mathbf{2}}$ using cyclohexanol ( 11.06 mL ), glycine ( $1.60 \mathrm{~g}, 21.25$ $\mathrm{mmol})$ and $p$-TSA ( $4.45 \mathrm{~g}, 23.37 \mathrm{mmol}$ ) and toluene $(110 \mathrm{~mL})$. The product was obtained as a white solid $(6.15 \mathrm{~g}, 88 \%)$.
${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.01\left(\mathrm{br} \mathrm{s}, 3 \mathrm{H}, \mathrm{NH}_{3}\right), 7.74(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}, \operatorname{Ar} p-\mathrm{TSA})$, 7.12 (d, $J=8.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar} p$-TSA), 4.77-4.69 (m, 1H, $\left.\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 3.70-3.64(\mathrm{~m}, 2 \mathrm{H}$, $\left.\mathrm{CH}_{2} \mathrm{Gly}\right), 2.35\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3} p\right.$-TSA $), 1.76-1.20\left(\mathrm{~m}, 10 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$.
[16h] L-Leucine cyclohexyl ester tosylate salt
Prepared according to the general procedure $\mathbf{A}_{\mathbf{2}}$ using
 cyclohexanol ( 20.82 mL ), L-leucine ( $6.61 \mathrm{~g}, 40.0$ mmol ), p-TSA ( $8.37 \mathrm{~g}, 44.0 \mathrm{mmol}$ ) and toluene (104 mL ). The product was obtained as a white solid (6.17 $\mathrm{g}, 40 \%)$.
${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(500 \mathbf{M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.08\left(\mathrm{br} \mathrm{s}, 3 \mathrm{H}, \mathrm{NH}_{3}\right), 7.78(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}, \operatorname{Ar} p-\mathrm{TSA})$, 7.15 (d, $J=8.0 \mathrm{~Hz}, 2 \mathrm{H}, \operatorname{Ar} p$-TSA), 4.77-4.68 (m, 1H, CH( $\left.\left.\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 3.96-3.88(\mathrm{~m}, 1 \mathrm{H}$, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 2.37 (s, $3 \mathrm{H}, \mathrm{CH}_{3} p$-TSA), 1.81-1.14 (m, 13H, $\mathrm{CHCH} \mathrm{CH}_{2}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu, $\left.\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 0.81\left(\mathrm{t}, J=7.2 \mathrm{~Hz}, 6 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}-\mathrm{Leu}\right)$.

## [16i] L-Phenylalanine pent-1-yl ester tosylate salt



Prepared according to the general procedure $\mathbf{A}_{2}$ using 1-pentanol ( 10.83 mL ), L-phenylalanine ( 3.30 $\mathrm{g}, 20.0 \mathrm{mmol}), p$-TSA ( $4.18 \mathrm{~g}, 22 \mathrm{mmol}$ ) and toluene ( 114 mL ). The product was obtained as a white solid ( $7.74 \mathrm{~g}, 95 \%$ ). ${ }^{152}$
${ }^{\mathbf{1}} \mathbf{H} \operatorname{NMR}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.25\left(\mathrm{br} \mathrm{s}, 3 \mathrm{H}, \mathrm{NH}_{3}\right), 7.77(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}, \operatorname{Ar} p-\mathrm{TSA})$, 7.21-7.09 (m, 7H, Ar $p$-TSA, $\mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe), 4.31-4.24 (m, 1H, $\mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe), 3.96-3.86 (m, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 3.32-3.23 (m, 1H, CHCH ${ }_{2} \mathrm{Ph}$ L-Phe), 3.13-3.05 (m, $1 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe), $2.35\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3} p\right.$-TSA), 1.42-1.34 (m, 2 H , $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 1.25-1.16 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 1.11-1.02 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), $0.87-0.81$ (m, $3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen).
[16j] L-Phenylalanine cyclohexyl ester tosylate salt


Prepared according to the general procedure $\mathbf{A}_{\mathbf{2}}$ using cyclohexanol ( 10.41 mL ), L-phenylalanine ( 3.30 g , 20.0 mmol ), $p$-TSA ( $4.18 \mathrm{~g}, 22 \mathrm{mmol}$ ) and toluene $(100 \mathrm{~mL})$. The product was obtained as a white solid ( $6.76 \mathrm{~g}, 83 \%$.).
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.22\left(\mathrm{br} \mathrm{s}, 3 \mathrm{H}, \mathrm{NH}_{3}\right), 7.76,7.12(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar} p-$ TSA), 7.21-7.11 (m, 7H, Ar p-TSA, $\mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe), 4.67-4.59 (m, $1 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5}$ cHex), 4.30-4.22 (m, 1H, CHCH2 Ph L-Phe), 3.32-3.24 (m, 1H, CHCH $\mathrm{CH}_{2} \mathrm{Ph}$ L-Phe), 3.143.05 ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe), 2.35 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3} p$-TSA), 1.64-1.49 (m, $4 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5}$ cHex), 1.47-1.39 (m, 1H, CH(CH2 $\left.)_{5} \mathrm{cHex}\right), 1.25-1.08\left(\mathrm{~m}, 5 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$.

### 9.5.4 Synthesis of aryl amino acid phosphorochloridates

[17a] Naphth-1-yl-(benzoxy-L-alaninyl) dichlorophosphate


Prepared according to the general procedure $\mathbf{C}$ using L-alanine benzyl ester hydrochloride salt ( $430 \mathrm{mg}, 1.99 \mathrm{mmol}$ ), naphth-1-yl dichlorophosphate [14b] $(0.52 \mathrm{~g}, 1.99 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(0.55 \mathrm{~mL}$, $3.98 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(8 \mathrm{ml})$. The product was obtained as a clear oil ( $87 \%, 0.70 \mathrm{~g}$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 8.23,8.02$ [ ${ }^{31} \mathrm{P}$ NMR ( $202 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\left.\boldsymbol{\delta}_{\mathrm{P}} 8.14,7.88\right] .{ }^{147}$
${ }^{1} \mathbf{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.10-7.30(\mathrm{~m}, 12 \mathrm{H}, \mathrm{Ar}), 5.12\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}\right), 5.19(\mathrm{~s}, 1 \mathrm{H}$, $\left.\mathrm{CH}_{2} \mathrm{Ph}\right), 4.70-4.64(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NH}), 4.42-4.33\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CHCH}_{3}\right.$ L-Ala), 1.59-1.57 (m, 3H, $\mathrm{CHCH}_{3}$ L-Ala).

## [17b] Naphth-1-yl-(ethoxy-L-alaninyl)dichlorophosphate



Prepared according to the general procedure $\mathbf{C}$ using L-alanine ethyl ester hydrochloride salt ( $0.77 \mathrm{~g}, 3.57 \mathrm{mmol}$ ), naphth-1-yl dichlorophosphate [14b] $(0.93 \mathrm{~g}, 3.57 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(0.99 \mathrm{~mL}, 7.14 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(15$ $\mathrm{mL})$. The product was obtained as a clear oil ( $1.20 \mathrm{~g}, 99 \%$ ).
${ }^{31} \mathbf{P}$ NMR (202 MHz, $\mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 8.30,8.05$. [ ${ }^{31} \mathrm{P}$ NMR (202 MHz, $\left.\left.\mathrm{CDCl}_{3}\right) \delta_{\mathrm{P}} 8.23,7.91\right] .{ }^{147}$
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.13-8.08 (m, 1H, Nap, H8), 7.92-7.87 (m, 1H, Nap, H5), 7.78-7.72 (m, 1H, H4 Nap), 7.63-7.56 (m, 3H, H2, H6, H7, Nap) 7.46 (t, $J=7.78 \mathrm{~Hz}, 1 \mathrm{H}$,

H3, Nap), 4.61-4.37 (m, 1H, NH), 4.32-4.22 (m, 2H, CH2 $\mathrm{CH}_{3} \mathrm{Et}$ ), 4.18-3.96 (m, 1 H , $\mathrm{CHCH}_{3}$ L-Ala), 1.62-1.52 (m, 3H, CH2 $\left.\mathrm{CH}_{3} \mathrm{Et}\right), 1.36-1.27$ (m, 3H, CHCH $\mathrm{C}_{3}$ L-Ala).
[17c] Naphth-1-yl-(cyclohexyloxy-L-alaninyl)dichlorophosphate


Prepared according to the general procedure $\mathbf{C}$ using L-alanine cyclohexyl ester tosylate salt [16e] ( $8.24 \mathrm{~g}, 24.0 \mathrm{mmol}$ ), naphth-1-yl dichlorophosphate [14b] ( $6.26 \mathrm{~g}, 24.0 \mathrm{mmol}$ ) and $\mathrm{Et}_{3} \mathrm{~N}(6.69 \mathrm{ml}, 48$ $\mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(100 \mathrm{~mL})$. The product was obtained as clear oil (7.58 g, $92 \%) .{ }^{151}$
${ }^{31} \mathbf{P}$ NMR (202 MHz, $\mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 8.33,7.99$.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(500 \mathbf{M H z}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{H}}$ 8.24-7.40(m, $\left.7 \mathrm{H}, \mathrm{Ar}\right), 4.92-4.84\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5}\right.$ cHex) 4.55-4.39 (m, 1H, NH), 4.34-4.25 (m, 1H, CHCH ${ }_{3}$ L-Ala), 1.88-1.75 (m, 4H, $\mathrm{CHCH}_{3}$ L-Ala, $\left.\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 1.58-1.55\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 1.54-1.35(\mathrm{~m}, 6 \mathrm{H}$, $\left.\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$.
[17d] Naphth-1-yl-(pent-1-yloxy-L-alaninyl)dichlorophosphate


Prepared according to the general procedure $\mathbf{C}$ using L-alanine pent-1-yl ester tosylate salt [16d] ( $7.95 \mathrm{~g}, 24.0 \mathrm{mmol}$ ), naphth-1-yl dichlorophosphate [14b] $(6.26 \mathrm{~g}, 24.0 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(6.69 \mathrm{ml}$, $48.0 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(100 \mathrm{~mL})$. The product was obtained as a clear oil $(7.83 \mathrm{~g}, 85 \%) .{ }^{151}$
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z , ~} \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 8.74,8.63$.
${ }^{1} \mathbf{H}$ NMR ( $500 \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.16(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Nap}, \mathrm{H} 8)$, $7.98-7.89(\mathrm{~m}, 1 \mathrm{H}$, Nap, H5), 7.72, 7.67 (m, 1H, Nap, H4), 7.65-7.52 (m, 3H, H2, H6, H7) 7.49-7.43 (m, 1H, H3), 4.36-4.18 (m, 1H, NH), 4.25-4.18 (m, 3H, CH $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen, $\mathrm{CHCH}_{3}$ LAla), 1.92-1.76 (m, $3 \mathrm{H}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ), 1.77-1.58 ( $\mathrm{m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 1.41-1.30 (m, 4H, CH2 CH2 $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), $0.99-0.82$ (m, $3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$ Pen).
[17e] Naphth-1-yl-(benzyloxy-L-leucinyl)dichlorophosphate


Prepared according to the general procedure $\mathbf{C}$ using L-leucine benzyl ester tosylate salt $(2.00 \mathrm{~g}, 5.08 \mathrm{mmol})$, naphth-1-yl dichlorophosphate [14b] $(1.33 \mathrm{~g}, 5.08 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(1.42 \mathrm{~mL}$, $10.16 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$. The product was obtained as a clear oil $(2.01 \mathrm{~g}, 89 \%) .{ }^{152}$
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 8.32,8.07$.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.12-8.04 (m, 1H, Nap, H8), 7.91-7.86 (m, 1H, Nap, H5), 7.78-7.71 (m, 1H, H4 Nap), 7.63-7.53 (m, 3H, H2, H6, H7, Nap) 7.44 (t, $J=7.97 \mathrm{~Hz}, 1 \mathrm{H}$, H3, Nap), 7.40-7.31 (m, 5H, Ph), 5.27-4.93 (m, 2H, CH ${ }_{2} \mathrm{Ph}$ ), 4.31-4.21 (m, 2H, NH, $\mathrm{CHCH} 2 \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 1.74-1.61 (m, 2H, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 1.36-1.31 (m, 1 H , $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 0.98-0.92 (m, $6 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu).

## [17f] Naphth-1-yl-(pentyl-1-oxy-L-leucinyl)dichlorophosphate



Prepared according to the general procedure $\mathbf{C}$ using L-leucine pent-1-yl ester hydrochloride salt [16a] ( $2.00 \mathrm{~g}, 8.41 \mathrm{mmol}$ ), naphth-1-yl dichlorophosphate [14b] ( $2.19 \mathrm{~g}, 8.41 \mathrm{mmol}$ ) and $\mathrm{Et}_{3} \mathrm{~N}(2.34 \mathrm{~mL}, 16.82 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(35 \mathrm{~mL})$. The product was obtained as a clear oil $(3.19 \mathrm{~g}, 89 \%) .{ }^{152}$
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 8.78,8.52$.
${ }^{1} \mathbf{H}$ NMR ( $500 \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.12(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Nap}, \mathrm{H} 8), 7.91-7.87(\mathrm{~m}, 1 \mathrm{H}$, Nap, H5), 7.77, 7.73 (m, 1H, Nap, H4), 7.65-7.54 (m, 3H, H2, H6, H7) 7.47-7.42 (m, 1H, H3), 4.48-4.36 (m, 1H, NH), 4.28-4.21 (m, 1H, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 4.21-4.12 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 1.91-1.76 (m, $2 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}$-Leu), 1.76-1.60 (m, $3 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 1.41-1.30 (m, 4 H , $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 1.01-0.94 (m, $6 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 0.94-0.86 (m, $3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen).
[17g] Naphth-1-yl-(cyclohexyloxy-L-leucinyl)dichlorophosphate


Prepared according to the general procedure $\mathbf{C}$ using L-leucine cyclohexyl ester tosylate salt [16h] ( $1.80 \mathrm{~g}, 4.67 \mathrm{mmol}$ ), naphth-1-yl dichlorophosphate [14b] ( $1.22 \mathrm{~g}, 4.67 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(1.30 \mathrm{~mL}, 9.34$ $\mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$. The product was obtained as a clear oil ( $1.84 \mathrm{~g}, 90 \%$ ).
${ }^{31} \mathbf{P} \mathbf{N M R}\left(\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{P}} 8.82,8.55$.
${ }^{1} \mathbf{H}$ NMR ( $500 \mathbf{M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.12(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Nap}, \mathrm{H} 8), 7.91-7.86(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 5$ Nap), 7.74 (d, $J=8.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4 \mathrm{Nap}$ ), 7.64-7.54 (m, 3H, H2, H6, H7, Nap) 7.47-7.42 (m, 1H, H3, Nap), 4.92-4.82 (m, 1H, CH $\left.\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), ~ 4.42-4.31(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NH})$, 4.25-4.16 (m, $1 \mathrm{H}, \mathrm{CHCH} \mathrm{CH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 1.93-1.26 (m, $13 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu, $\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5}$ cHex), 1.00-0.94 (m, $6 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu).
[17h] Naphth-1-yl-(ethyloxy-L-leucinyl)dichlorophosphate


Prepared according to the general procedure $\mathbf{C}$ using L-leucine ethyl ester hydrochloride salt [16b] ( $1.06 \mathrm{~g}, 5.42 \mathrm{mmol})$, naphth-1-yl dichlorophosphate [14b] $(1.41 \mathrm{~g}, 5.42 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(1.51 \mathrm{~mL}, 10.84$ $\mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$. The product was obtained as a clear oil (1.77 g, $85 \%$.).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 8.72,8.49$.
${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.12(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Nap}, \mathrm{H} 8), 7.89(\mathrm{~d}, J=7.6 \mathrm{~Hz}$, 1H, Nap, H5), 7.75 (d, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4 \mathrm{Nap}$ ), 7.65-7.53 (m, 3H, H2, H6, H7, Nap) 7.45 ( $\mathrm{t}, \mathrm{J}=7.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3, \mathrm{Nap}$ ), 4.52-3.91 (m, 4H, NH, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu, $\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}$ ), 1.95-1.75 (m, 2H, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 1.74-1.59 (m, 1H, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), $1.35-1.27\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right), 1.00-0.93\left(\mathrm{~m}, 6 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu).

## [17i] Naphth-1-yl-(ethoxy-glycinyl)dichlorophosphate



Prepared according to the general procedure $\mathbf{C}$ using glycine ethyl ester hydrochloric salt ( $0.85 \mathrm{~g}, 6.10 \mathrm{mmol}$ ), naphth-1-yl dichlorophosphate [14b] $(1.59 \mathrm{~g}, 6.10 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(1.70 \mathrm{~mL}, 12.21 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(24 \mathrm{~mL})$. The product was obtained as a clear oil $(1.90 \mathrm{~g}, 95 \%)$.
${ }^{31} \mathbf{P} \mathbf{N M R}\left(\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{P}} 9.29$.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.12-8.08 (m, 1H, Nap, H8), 7.98-7.87 (m, 1H, Nap, H5), 7.77-7.72 (m, 1H, H4 Nap), 7.64-7.55 (m, 3H, H2, H6, H7, Nap) 7.45 (t, J = 8.0 Hz, 1H, H3, Nap), 4.28 (q, $J=7.1 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}$ ), 4.03-3.98 (m, 2H, CH2 Gly), 1.32 (t, $J=$ $\left.7.1 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right)$.
[17j] Naphth-1-yl(benzyloxy-glycinyl)dichlorophosphate


Prepared according to the general procedure $\mathbf{C}$ using glycine benzyl ester hydrochloric salt $(1.03 \mathrm{~g}, 5.13 \mathrm{mmol})$, naphth-1-yl dichlorophosphate [14b] $(1.34 \mathrm{~g}, 5.13 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(1.43 \mathrm{~mL}$, $10.26 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$. The product was obtained as a clear oil ( $1.60 \mathrm{~g}, 80 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 8.87$. [Lit. ${ }^{31} \mathrm{P}$ NMR ( $202 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{P}} 8.90$ ]. ${ }^{148}$
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.10-8.06 (m, 1H, Nap, H8), 7.92-7.88 (m, 1H, Nap, H5), 7.76 (d, $J=8.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4 \mathrm{Nap}$ ), 7.63-7.56 (m, 3H, H2, H6, H7, Nap) 7.46 (t, $J=8.0 \mathrm{~Hz}$, 1H, H3, Nap), 7.42-7.38 (m, 5H, Ph), 5.27 (s, 2H, CH $\mathrm{H}_{2} \mathrm{Ph}$ ), 4.41-4.26 (m, 1H, NH), 4.104.05 (m, 2H, CH2 Gly).

## [17k] Naphth-1-yl(pentyl-1-oxy-glycinyl)dichlorophosphate



Prepared according to the general procedure $\mathbf{C}$ using glycine pent1 -yl ester tosylate salt $[\mathbf{1 6 f}](2.57 \mathrm{~g}, 8.11 \mathrm{mmol})$, naphth-1-yl dichlorophosphate [14b] $(2.12 \mathrm{~g}, 8.11 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(2.26 \mathrm{~mL}$, $16.22 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(32 \mathrm{~mL})$. The product was obtained as a clear oil ( $2.64 \mathrm{~g}, 88 \%$ ).
${ }^{31} \mathbf{P}$ NMR (202 MHz, $\mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 9.10$.
${ }^{1} \mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.10(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Nap}, \mathrm{H} 8), 7.92-7.88$ (m, 1H, Nap, H5), 7.75 (d, $J=8.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4 \mathrm{Nap}$ ), $7.64-7.54$ (m, 3H, H2, H6, H7, Nap) 7.46 (t, $J=8.0$ $\mathrm{Hz}, 1 \mathrm{H}, \mathrm{H} 3, \mathrm{Nap}), 4.48-4.41(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NH}), 4.23\left(\mathrm{t}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n-\right.$ Pen), 4.05-3.99 (m, 2H, CH2 Gly), 1.75-1.65 (m, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 1.39$1.34\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $0.93\left(\mathrm{t}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$ Pen).

## [171] Naphth-1-yl(cyclohexyloxy-glycinyl)dichlorophosphate



Prepared according to the general procedure $\mathbf{C}$ using glycine cyclohexyl ester tosylate salt $[\mathbf{1 6 g}](2.59 \mathrm{~g}, 7.86 \mathrm{mmol})$, naphth-1-yl dichlorophosphate [14b] ( $2.05 \mathrm{~g}, 7.86 \mathrm{mmol}$ ) and $\mathrm{Et}_{3} \mathrm{~N}(2.19 \mathrm{~mL}, 15.72$ $\mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$. The product was obtained as a clear oil (2.31 $\mathrm{g}, 77 \%)$.
${ }^{31} \mathbf{P}$ NMR ( $\left.\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{P}} 9.04$.
${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.10(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 8 \mathrm{Nap}), 7.90(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}$, H5 Nap), 7.76 (d, $J=8.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4 \mathrm{Nap}$ ), $7.64-7.57$ ( $\mathrm{m}, 3 \mathrm{H}, \mathrm{H} 2, \mathrm{H} 6, \mathrm{H} 7 \mathrm{Nap}$ ), 7.46 (t, $J$ $=8.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3 \mathrm{Nap}), 4.95-4.88\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 4.43-4.30(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NH}), 4.04-$ $3.98\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Gly}\right), 1.96-1.37\left(\mathrm{~m}, 10 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$.

## [17m] Naphth-1-yl-(benzoxy-L-phenylalaninyl)dichlorophosphate



Prepared according to the general procedure $\mathbf{C}$ using L-phenylalanine benzyl ester tosylate salt ( $3.42 \mathrm{~g}, 8.00 \mathrm{mmol}$ ), naphth-1-yl dichlorophosphate [14b] $(2.09 \mathrm{~g}, 8.00 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(2.23 \mathrm{~mL}$, $16.00 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$. The product was obtained as clear oil ( $3.34 \mathrm{~g}, 87 \%$ ).
${ }^{\mathbf{3 1}} \mathbf{P}$ NMR (202 MHz, $\mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 8.29,8.17$. [Lit. ${ }^{31} \mathrm{P}$ NMR ( 202 MHz , $\left.\left.\mathrm{CDCl}_{3}\right) \delta_{\mathrm{P}} 8.32,8.19\right] .{ }^{148}$
${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 7.93-6.96 (m, 17H, Nap, Ph L-Phe, $\mathrm{CH}_{2} \mathrm{Ph}$ ), 5.20-4.95 ( $\mathrm{m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}$ ), 4.66-4.51 (m, 1H, $\mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe), 4.32-4.11 (m, 1H, NH), 3.27-3.07 (m, 2H, $\mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe).
[17n] Naphth-1-yl-(pentyloxy-L-phenylalaninyl)dichlorophosphate


Prepared according to the general procedure $\mathbf{C}$ using Lphenylalanine pent-1-yl ester tosylate salt [16i] (3.26 g, 8.00 mmol), naphth-1-yl dichlorophosphate [14b] ( $2.09 \mathrm{~g}, 8.00 \mathrm{mmol}$ ) and $\mathrm{Et}_{3} \mathrm{~N}(2.23 \mathrm{~mL}, 16.00 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$. The product was obtained as a clear oil ( $3.02 \mathrm{~g}, 82 \%$ ). ${ }^{152}$
${ }^{31} \mathbf{P}$ NMR (202 MHz, $\left.\mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{P}} 8.35,8.25$.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.13-7.07 (m, $\mathbf{1 2 H}$, Nap, $\mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe), 4.56-4.46 $\left(\mathrm{m}, \quad 1 \mathrm{H}, \quad \mathrm{CHCH}_{2} \mathrm{Ph}\right.$ L-Phe), 4.36-4.21 (m, $\left.1 \mathrm{H}, \quad \mathrm{NH}\right), \quad 4.13-4.07(\mathrm{~m}, \quad 1 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 4.03-3.94 (m, $1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 3.25-3.10 ( $\mathrm{m}, 2 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe), 1.66-1.56 (m, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 1.37-1.31 (m, $4 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 0.94-0.87 (m, $3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen).
[170] Naphth-1-yl-(cyclohexyloxy-L-phenylalaninyl)dichlorophosphate


Prepared according to the general procedure $\mathbf{C}$ using L-phenylalanine cyclohexyl ester tosylate salt [16j] ( $3.36 \mathrm{~g}, 8.00 \mathrm{mmol}$ ), naphth-1-yl dichlorophosphate [14b] $(2.09 \mathrm{~g}, 8.00 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(2.23 \mathrm{~mL}, 16.00$ $\mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(32 \mathrm{~mL})$. The product was obtained as a clear oil (3.25 g, 86\%).
${ }^{31} \mathbf{P} \mathbf{N M R}\left(\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{P}} 8.45,8.35$.
${ }^{1} \mathbf{H}$ NMR ( $500 \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.12-7.10 (m, 12 H , Nap, $\mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe), 4.86-4.78 $\left(\mathrm{m}, 1 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 4.57-4.42\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{Ph}\right.$ L-Phe), 4.39-4.33 (m, 0.5H, NH), 4.29-4.23 (m, 0.5H, NH), 3.24-3.06 (m, 2H, CHCH ${ }_{2} \mathrm{Ph}$ L-Phe), 1.85-1.31 (m, 10H, $\left.\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$.
[17p] Naphth-1-yl-(ethyloxy-L-phenylalaninyl)dichlorophosphate


Prepared according to the general procedure $\mathbf{C}$ using L-phenylalanine ethyl ester hydrochloride salt [16c] ( $1.84 \mathrm{~g}, 8.00 \mathrm{mmol})$, naphth-1-yl dichlorophosphate [14b] $(2.09 \mathrm{~g}, 8.00 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(2.23 \mathrm{~mL}, 16.00$ $\mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(32 \mathrm{~mL})$. The product was obtained as a clear oil (3.01 g, $90 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 8.36, ~ 8.23$.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(500 \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{H}}$ 8.12-7.10 (m, $\mathbf{1 2 H}$, Nap, $\mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe), 4.84-4.71 ( $\mathrm{m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}$ ), 4.57-4.42 (m, $1 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe), 4.35-4.31 (m, 0.5H, NH), 4.284.19 (m, 0.5H, NH), 3.29-3.16 (m, 2H, CHCH $\mathrm{CH}_{2} \mathrm{Ph}$ L-Phe), 1.95-1.73 (m, 3H, CH2CH3 Et).

## [17q] Naphth-1-yl-(methoxy-dimethylglycinyl)dichlorophosphate



Prepared according to the general procedure $\mathbf{C}$ using dimethyl glycine methyl ester hydrochloride salt ( $1.0 \mathrm{~g}, 6.50 \mathrm{mmol}$ ), naphth-1-yl dichlorophosphate $[\mathbf{1 4 b}](1.70 \mathrm{~g}, 6.50 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(1.81 \mathrm{~mL}, 13.00$ $\mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(26 \mathrm{~mL})$. The product was obtained as a clear oil (1.66 g, $75 \%$ ). ${ }^{153}$
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathrm{P}} 6.02$.
${ }^{1} \mathbf{H} \operatorname{NMR}\left(500 \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.15(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 8 \mathrm{Nap}), 7.87(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}$, H5 Nap), 7.73 (d, $J=8.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4 \mathrm{Nap}$ ), 7.64-7.61 (m, 1H, H2 Nap), 7.60-7.52 (m, 2H, H6, H7 Nap), 7.44 (t, $J=7.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3 \mathrm{Nap}$ ), 4.97 (br s, 1H, NH), 3.81 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3}$ ), 1.77 (s, 3H, CH ${ }_{3}$ DMG), 1.73 (s, 3H, CH ${ }_{3}$ DMG).
[17r] Phenyl-(benzyloxy-glycinyl)dichlorophosphate


Prepared according to the general procedure $\mathbf{C}$ using glycine benzyl ester hydrochloride salt $(1.78 \mathrm{~g}, 8.83 \mathrm{mmol})$, phenyl dichlorophosphate $(1.32 \mathrm{~mL}, 8.83 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(2.50 \mathrm{~mL}, 17.66$ $\mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(35 \mathrm{~mL})$. The product was obtained as a clear oil ( $2.40 \mathrm{~g}, 80 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 9.04$. [Lit. ${ }^{31} \mathrm{P}$ NMR (202 MHz, $\mathrm{CDCl}_{3}$ ) $\delta_{\mathrm{P}} 8.75$ ]. ${ }^{147}$
${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 7.42-7.34(\mathrm{~m}, 7 \mathrm{H}, \mathrm{Ar}), 7.31-7.22(\mathrm{~m}, 3 \mathrm{H}, \mathrm{Ar}), 5.25(\mathrm{~s}$, $\left.2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.62-4.48(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NH}), 4.01-3.94\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Gly}\right)$.

## [17s] Phenyl-(benzyloxy-L-alaninyl)dichlorophosphate



Prepared according to the general procedure $\mathbf{C}$ using L-alanine benzyl ester hydrochloride salt ( $1.12 \mathrm{~g}, 5.20 \mathrm{mmol}$ ), phenyl dichlorophosphate $(1.10 \mathrm{~g}, 5.20 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(1.45 \mathrm{ml}, 10.40$ $\mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{ml})$. The product was obtained as a clear oil $(1.51 \mathrm{~g}, 82 \%)$.
${ }^{31} \mathbf{P}$ NMR (202 MHz, $\mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 7.51,7.85$. [Lit. ${ }^{31} \mathrm{P}$ NMR (202 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta_{\mathrm{P}} 7.86$, 7.52]. ${ }^{147}$
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 7.36-7.25(\mathrm{~m}, 10 \mathrm{H}, \mathrm{Ar}), 5.15\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}\right), 5.11(\mathrm{~s}$, $1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}$ ), 4.25-4.14 (m, 2H, NH, $\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ), 1.59-1.57 (m, 3H, $\mathrm{CHCH}_{3}$ L-Ala).
[17t] Phenyl-(neopentyloxy-L-alaninyl)dichlorophosphate
Prepared according to the general procedure $\mathbf{C}$ using L-alanine neopentyl ester hydrochloride salt ( $1.20 \mathrm{~g}, 6.12 \mathrm{mmol}$ ), phenyl dichlorophosphate ( $1.29 \mathrm{~mL}, 6.12 \mathrm{mmol}$ ) and $\mathrm{Et}_{3} \mathrm{~N}(1.71 \mathrm{ml}, 12.24$ $\mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25 \mathrm{ml})$. The product was obtained as a clear oil ( $1.61 \mathrm{~g}, 79 \%$ ).
${ }^{31} \mathbf{P}$ NMR (202 MHz, $\left.\mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{P}} 7.51,7.85$.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{H}} 7.16-7.22(\mathrm{~m}, 5 \mathrm{H}, \mathrm{Ar}), 4.26-4.21(\mathrm{~m}, 1 \mathrm{H}, \mathrm{NH}), 4.20-4.15$ ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ), 3.90-3.93 (m, 2H, $\left.\mathrm{CH}_{2} \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3} \mathrm{Neop}\right), 1.59-1.52\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{CHCH}_{3}\right.$ L-Ala), 1.34-1.01 (m, 9H, CH2 $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3} \mathrm{Neop}\right)$.

## [17u] Phenyl-(hexyl-1-oxy-L-alaninyl)dichlorophosphate

Prepared according to the general procedure $\mathbf{C}$ using L-alanine hex-1-yl ester hydrochloride salt ( $1.55 \mathrm{~g}, 7.40 \mathrm{mmol}$ ), phenyl dichlorophosphate $(1.93 \mathrm{~g}, 7.40 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(2.06 \mathrm{~mL}$, $14.80 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$. The product was obtained as a clear oil ( $2.11 \mathrm{~g}, 82 \%$ ). ${ }^{151}$
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 7.96,7.64$.
${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{H}}$ 7.24-7.18(m,5H, Ar), 4.34-4.20(m, 1H, NH), 4.20-4.05 (m, 3H, $\mathrm{CHCH}_{3}$ L-Ala, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex), 4.03-3.94 (m, 2 H , $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ n-Hex), 1.66-1.56 (m, $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ n-Hex), 1.59-1.53 (m, 3H, $\mathrm{CHCH}_{3}$ L-Ala), 1.37-1.31 (m, $4 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ n-Hex), $0.94-0.87\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{m}, 1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Hex).
[17v] Ethyl 3-(2-hydroxyphenyl)propanoyl-(benzyloxy-L-alaninyl)dichlorophosphate
 Prepared according to the general procedure $\mathbf{C}$ using L-alanine- $O$-benzyl ester hydrochloric salt $(1.70 \mathrm{~g}, 7.88$ mmol), ethyl 3-(2-hydroxyphenyl)propanoyl dichlorophosphate $[\mathbf{1 4 c}](2.45 \mathrm{~g}, 7.88 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}$ ( $2.20 \mathrm{~mL}, 15.76 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(35 \mathrm{~mL})$. The product was obtained as a clear oil $(3.29 \mathrm{~g}, 92 \%) .{ }^{150}$
${ }^{31} \mathbf{P} \mathbf{N M R}\left(\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{P}} 7.53,7.70$.
${ }^{1} \mathbf{H}$ NMR ( $\left.\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{H}} 7.50-7.44(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ar}), 7.41-7.32(\mathrm{~m}, 5 \mathrm{H}, \mathrm{Ar}), 7.28-7.14$ (m, 3H, Ar), 5.26-5.17 (m, 2H, CH2Ph), 4.67-4.56 (m, 1H, NH), 4.34-4.22 (m, 1H, CHCH L-Ala), $\quad 4.15-4.09 \quad\left(\mathrm{~m}, \quad 2 \mathrm{H}, \quad \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), \quad 3.15-2.93 \quad(\mathrm{~m}, \quad 2 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 2.66-2.59 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.58-1.52 (m, 3 H , $\mathrm{CHCH}_{3}$ L-Ala), 1.24-1.20 (m, 3H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ).

### 9.5.5 Synthesis of 3'-deoxyadenosine

## [21] 3'-Deoxyadenosine



3'-Deoxyadenosine was synthesised according to standard procedure $\mathbf{E}$ using 9-(2,3-anhydro- $\beta$-D-ribofuranosyl) adenine [26] ( $9.12 \mathrm{~g}, 36.59 \mathrm{mmol}$ ), $\mathrm{LiEt}_{3} \mathrm{BH}$ ( 1 M in THF, 109.77 mmol followed by a second addition of 36.59 mmol ) in DMSO ( 55 mL ) and THF $(550 \mathrm{~mL})$. The title compound was obtained as a white solid $(9.01 \mathrm{~g}$, 98\%).
Melting point: $188-190{ }^{\circ} \mathrm{C}\left(\right.$ Lit. mp: 191-192 $\left.{ }^{\circ} \mathrm{C}\right) .{ }^{212}$
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}$, DMSO-d6) $\boldsymbol{\delta}_{\mathbf{H}} 8.37$ (s, 1H, H8), 8.17 (s, 1H, H2), 7.29 (br s, 2H, $\mathrm{NH}_{2}$ ), $5.89\left(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H1}{ }^{\prime}\right), 5.68\left(\mathrm{~d}, J=4.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{OH}-2^{\prime}\right), 5.19(\mathrm{t}, J=6.0 \mathrm{~Hz}$, 1H, OH-5'), 4.63-4.58 (m, 1H, H2'), 4.40-4.34 (m, 1H, H4'), 3.71 (ddd, $J=12.0,6.0,3.0$ Hz, 1H, H5'), 3.53-3.49 (ddd, $\left.J=12.0,6.0,4.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5^{\prime}\right)$.
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}$, DMSO-d6) $\boldsymbol{\delta}_{\mathbf{C}} 156.00$ (C6), 152.41 (C2), 148.82 (C4), 139.09 (C8), 119.06 (C5), 90.79 (C1'), 80.66 (C4'), 74.56 (C2'), 62.61 (C5'), 34.02 (C3').

MS (ES+) m/z found $258.12\left[M+\mathrm{Li}^{+}\right], 274.09\left[\mathrm{M}+\mathrm{Na}^{+}\right], 252.11\left[\mathrm{M}+\mathrm{H}^{+}\right] \mathrm{C}_{10} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{O}_{3}$ required m/z 251.24 [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 0$ to $75 / 25$ in 30 minutes, $1 \mathrm{ml} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed one peak with tR 11.22 min .
[23] 2', 5'-(Bis-O-tert-butyldimethylsilyl)adenosine
Adenosine ( $1 \mathrm{~g}, 3.74 \mathrm{mmol}$ ) was dissolved in a mixture of THF and
 DMF ( $1: 1,100 \mathrm{~mL}$ ) and pyridine was added ( $1.51 \mathrm{~mL}, 18.71$ $\mathrm{mmol})$, followed by $\mathrm{AgNO}_{3}(1.90 \mathrm{~g}, 11.22 \mathrm{mmol})$. After 15 minutes, TBDMSCl ( $1.24 \mathrm{~g}, 8.23 \mathrm{mmol}$ ) was finally added to the reaction mixture and the solution was stirred at $30{ }^{\circ} \mathrm{C}$ for 2 hours. The solvent was evaporated in vacuo and the residue was dissolved in
$\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL})$ and washed with brine ( 50 mL ). The organic phase was dried over $\mathrm{NaSO}_{4}$, filtered and evaporated. The crude was purified by column chromatography $\mathrm{nHex} / \mathrm{EtOac}=5 / 5$ to $2 / 8$ to give the title compound as a white foam $(0.28 \mathrm{~g}, 15 \%) .{ }^{208}$
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}$, DMSO-d6) $\boldsymbol{\delta}_{\mathbf{H}} 8.28$ (s, 1H, H8), 8.14 ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H} 2$ ), 7.30 (br s, 2H, $\mathrm{NH}_{2}$ ), $5.94\left(\mathrm{~d}, J=5.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 5.15\left(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{OH}-3^{\prime}\right), 4.62\left(\mathrm{t}, J=5.0 \mathrm{~Hz}, \mathrm{H} 2^{\prime}\right), 4.17-$ 4.13 (m, 1H, H4'), $3.92\left(\mathrm{dd}, J=11.5,4.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}^{\prime}\right), 3.79(\mathrm{dd}, J=11.5,4.0 \mathrm{~Hz}, 1 \mathrm{H}$, $\left.\mathrm{H} 5^{\prime}\right), 0.90\left(\mathrm{~s}, 9 \mathrm{H},\left(\mathrm{CH}_{3}\right)_{3} t \mathrm{Bu}\right), 0.74\left(\mathrm{~s}, 9 \mathrm{H},\left(\mathrm{CH}_{3}\right)_{3} t \mathrm{Bu}\right), 0.09\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.08(\mathrm{~s}, 3 \mathrm{H}$, $\mathrm{CH}_{3}$ ), $0.06\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.15\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.
(ES+) m/z found $518.27\left[\mathrm{M}+\mathrm{Na}^{+}\right], 496.29\left[\mathrm{M}+\mathrm{H}^{+}\right] \mathrm{C}_{22} \mathrm{H}_{41} \mathrm{~N}_{5} \mathrm{NaO}_{4} \mathrm{Si}_{2}$ required $\mathrm{m} / \mathrm{z} 495.76$ [M].


3',5'-O-tert-butyldimethylsilyladenosine was isolated through the same reaction procedure $(0.06 \mathrm{~g}, 3 \%)$, as a more hydrophilic regioisomer.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}$, DMSO-d6) $\boldsymbol{\delta}_{\mathbf{H}} 8.31$ ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H} 8$ ), 8.14 ( s , $1 \mathrm{H}, \mathrm{H} 2$ ), 7.27 (br s, 2H, NH2), 5.90 (d, $J=6.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), 5.41 (d, $\left.J=6.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{OH}-2^{\prime}\right), 4.80-4.74\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}{ }^{\prime}\right), 4.36(\mathrm{dd}, J=5.0 \mathrm{~Hz}, 3.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3$ '), 3.98-3.94 (m, 1H, H4'), 3.88 (dd, $J=11.5,6.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H})^{\prime}$ ), 3.69 (dd, $J=11.0,4.0 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{H} 5$ '), $0.92(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.87(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.134\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.131\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.05$ ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3}$ ), $0.04\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.
[25] 2',5'-(Bis-O-tert-butyldimethylsilyl)3'-deoxyadenosine


3'-Deoxyadenosine ( $1.00 \mathrm{~g}, 3.98 \mathrm{mmol}$ ) was dissolved in anhydrous pyridine ( 10 mL ) and $\operatorname{TBDMSCl}(1.2 \mathrm{~g}, 7.96 \mathrm{mmol})$ was added to the flask. The reaction was stirred under argon atmosphere at room temperature for 16 hours. The reaction mixture was diluted with $\mathrm{CHCl}_{3}(20 \mathrm{~mL})$ and the organics were washed with $\mathrm{NH}_{4} \mathrm{Cl}$ (saturated solution, $20 \mathrm{~mL} \times 3$ ). The organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, evaporated and the crude was purified by flash column chromatography to afford the product as a sticky solid ( $0.89 \mathrm{~g}, 50 \%$ ). ${ }^{208}$
[Alternative procedure] 3'-Deoxyadenosine ( $1.13 \mathrm{~g}, 4.5 \mathrm{mmol}$ ), was dissolved in DMF ( 50 $\mathrm{mL})$ and TBDMSCl ( $2.03 \mathrm{~g}, 13.5 \mathrm{mmol}$ ) and imidazole ( $1.84 \mathrm{~g}, 27 \mathrm{mmol}$ ) were added to the flask. The mixture was stirred for 16 hours at rt . The reaction mixture was diluted with $\mathrm{CHCl}_{3}(20 \mathrm{~mL})$ and the organics were washed with $\mathrm{NH}_{4} \mathrm{Cl}$ (saturated solution, $20 \mathrm{~mL} \times 3$ ). The organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to afford the product as a sticky solid ( $1.56 \mathrm{~g}, 72 \%$ ).
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.36(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 8.34(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 6.03(\mathrm{~d}, J=1.2 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{H} 1$ '), 5.54 (br s, 2H, NH2 $), 4.66-4.63$ (m, 1H, H2'), 4.61-4.55 (m, 1H, H4'), 4.14 (dd, $J=11.7,2.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ '), $3.80(\mathrm{dd}, J=11.7,2.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ '), 2.32-2.25 (m, 1H, H3'), 1.91-1.85 (m, 1H, H3'), $0.97(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.92(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.17\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.16(\mathrm{~s}, 3 \mathrm{H}$, $\mathrm{CH}_{3}$ ), $0.15\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.10\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.
[26] 9-(2,3-Anhydro- $\beta$-D-ribofuranosyl) adenine
The title compound was synthesised according to standard procedure


D using adenosine ( $10.00 \mathrm{~g}, 37.42 \mathrm{mmol}$ ) , $\alpha-\mathrm{AIBBr}(22.03 \mathrm{~mL}$, $149.68 \mathrm{mmol})$ and $\mathrm{H}_{2} \mathrm{O}(0.67 \mathrm{~mL}, 0.037 \mathrm{mmol})$, in $\mathrm{CH}_{3} \mathrm{CN}(250 \mathrm{~mL})$. The crude was then dissolved in $\mathrm{CH}_{3} \mathrm{OH}(260 \mathrm{~mL})$ and stirred with Amberlite ( $2 \times \mathrm{OH}^{-}, 150 \mathrm{~mL}$ ) for 16 hours. The title compound was obtained as a white powder $(8.86 \mathrm{~g}, 95 \%)$.
Melting point $178-180{ }^{\circ} \mathrm{C}$ (Lit. mp: 180-181 ${ }^{\circ} \mathrm{C}$ )..$^{212}$
 $\mathrm{NH}_{2}$ ), 6.21 ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H} 1^{\prime}$ ), 5.05 (br s, $\left.1 \mathrm{H}, \mathrm{OH}-5^{\prime}\right), 4.46\left(\mathrm{~d}, J=2.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2^{\prime}\right), 4.22$ (d, $J=$ $\left.2.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3^{\prime}\right), 4.18\left(\mathrm{t}, J=5.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4^{\prime}\right), 3.60-3.55\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 5^{\prime}\right)$, 3.53-3.49(m, 1 H , H5').
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}$, DMSO-d6) $\boldsymbol{\delta}_{\mathbf{C}} 156.01$ (C6), 152.61 (C2), 149.11 (C4), 139.55 (C8), 119.52 (C5), 91.25 ( $\mathrm{C}^{\prime}$ ), 81.15 ( C 4 '), 75.06 ( $\mathrm{C}^{\prime}$ '), 58.75 ( C 3 '), 57.70 ( C 5 ').

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 0$ to $75 / 25$ in 30 minutes, $\mathrm{F}=1 \mathrm{ml} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed one peak with tR 13.20 min .
(ES+) m/z found $272.09\left[\mathrm{M}+\mathrm{Na}^{+}\right], 250.09\left[\mathrm{M}+\mathrm{H}^{+}\right] \mathrm{C}_{10} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{O}_{3}$ required $\mathrm{m} / \mathrm{z} 249.23$ [M].

### 9.5.6 Synthesis of 3'-deoxyadenosine prodrugs

## [28a] 3'-Deoxyadenosine-2'-O-phenyl-(benzyloxy-L-alaninyl)-phosphate

 Prepared according to general procedure $\mathbf{F}_{1}$ using 3'deoxyadenosine ( $0.05 \mathrm{~g}, 0.20 \mathrm{mmol}$ ) in THF ( 4 mL ), $t \mathrm{BuMgCl}(1.0 \mathrm{M}$ solution in THF, $0.22 \mathrm{~mL}, 0.22 \mathrm{mmol}$ ), phenyl(benzyloxy-L-alaninyl) phosphorochloridate [17s] $(0.21 \mathrm{~g}, 0.60 \mathrm{mmol})$ in THF ( 2.4 mL ). Purification by column chromatography (eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$
$0 / 100$ to $8 / 92$ ) and preparative TLC ( $500 \mu \mathrm{M}$, eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}=5 / 95$ ) afforded the compound as a white solid $(0.006 \mathrm{~g}, 5 \%)$.
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 2.44,2.92$.
${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ): $\boldsymbol{\delta}_{\mathbf{H}} 8.41$ ( $\left.\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 8\right), 8.28$ ( $\left.\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 8\right), 8.19$ ( $\mathrm{s}, 0.5 \mathrm{H}$, $\mathrm{H} 2), 8.18(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 2), 7.39-7.30(\mathrm{~m}, 4 \mathrm{H}, \mathrm{Ar}), 7.28-7.18(\mathrm{~m}, 4 \mathrm{H}, \mathrm{Ar}), 7.17-7.11(\mathrm{~m}, 1 \mathrm{H}$, Ar), 7.08-7.03 (m, 1H, Ar), $6.23\left(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 6.08(\mathrm{~d}, J=3.4 \mathrm{~Hz}, 0.5 \mathrm{H}$, H1'), 5.52-5.43 (m, 1H, H2'), 5.19-5.12 (m, 1H, CH ${ }_{2} \mathrm{Ph}$ ), 5.07-4.95 (m, 1H, CH $\mathrm{H}_{2} \mathrm{Ph}$ ), 4.484.42 ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{H} 4$ '), 4.05-3.97 (m, 1H, $\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ), 3.95-3.87 (m, 1H, H5'), 3.69-3.61 (m, 1H, H5'), 2.59-2.45 (m, 1H, H3'), 2.31-2.23 (m, 1H, H3'), 1.36-1.27 (m, 3H, CHCH L-Ala).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O H}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 174.76\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.52\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 157.44 (C6), 153.76 (C2), 151.93 (C4), 150.06 (C-Ar), 149.93 (C-Ar), 141.38 (C8), 141.18 (C8), 137.33 (C-Ar), 137.10 (C-Ar), 130.69 (CH-Ar), 130.79 (CH-Ar), 129.61 (CH-Ar), 129.51 (CH-Ar), 129.40 (CH-Ar), 129.30 (CH-Ar), 129.23 (CH-Ar), 126.33 (CH-Ar), 126.16 (CH-Ar), 121.53 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 121.20\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.5\right.$ H, CH-Ar), 120.76 (C5), 91.56 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.7 \mathrm{~Hz}, \mathrm{C} 1{ }^{\prime}\right), 91.45\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.7 \mathrm{~Hz}, \mathrm{C} 1{ }^{\prime}\right)$, $82.78\left(\mathrm{C}^{\prime}\right), 82.28\left(\mathrm{C} 4\right.$ '), $80.98\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=4.7 \mathrm{~Hz}, \mathrm{C} 2\right.$ '), $80.95\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=4.7 \mathrm{~Hz}, \mathrm{C} 2{ }^{\prime}\right)$, $67.95\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.92\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 64.13\left(\mathrm{C}^{\prime}\right), 63.59\left(\mathrm{C}^{\prime}\right), 51.88\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.75$ $\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 33.75\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.0 \mathrm{~Hz}, \mathrm{C} 3\right.$ '), $33.59\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.0 \mathrm{~Hz}, \mathrm{C} 3\right.$ '), $20.33(\mathrm{~d}$, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.1, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 20.18\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.1, \mathrm{CHCH} 3\right.$ L-Ala $)$.
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{OH}$ from $90 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed two peaks of the diastereoisomers with tR 22.16 min and tR 22.43 min .
(ES+) $\mathbf{m} / \mathbf{z}$ found: $569.2\left[\mathrm{M}+\mathrm{H}^{+}\right], 591.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], 1159.4\left[2 \mathrm{M}+\mathrm{Na}^{+}\right] \mathrm{C}_{26} \mathrm{H}_{29} \mathrm{~N}_{6} \mathrm{O}_{7} \mathrm{P}$ required m/z $568.2[\mathrm{M}]$.
[28b] 3'-Deoxyadenosine 2'-O-naphth-1-yl-(benzyloxy-L-alaninyl)-phosphate


Prepared according to general procedure $\mathbf{F}_{\mathbf{1}}$ using 3'deoxyadenosine ( $0.05 \mathrm{~g}, 0.20 \mathrm{mmol}$ ) in THF ( 4 mL ), $t \mathrm{BuMgCl}(1.0 \mathrm{M}$ solution in THF, $0.22 \mathrm{~mL}, 0.22 \mathrm{mmol}), 1-$ naphth-1-yl(benzyloxy-L-alaninyl) phosphorochloridate [17a] ( $0.32 \mathrm{~g}, 0.80 \mathrm{mmol}$ ) in THF ( 3.2 mL ). Purification by column chromatography (eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ $0 / 100$ to $6 / 94)$ and preparative TLC $(500 \mu \mathrm{M}$, eluent system $\left.\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 5 / 95\right)$ afforded the compound as a white solid ( $0.014 \mathrm{~g}, 11 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\left.\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 3.27, 2.75.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.37$ ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 8$ ), 8.18 ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 8$ ), 8.14 ( $\mathrm{s}, 0.5 \mathrm{H}$, H2), 8.13-8.11 (m, 0.5H, Ar) 8.11 ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 2$ ), $7.94-7.90$ (m, 0.5H, Ar), 7.90-7.87 (m, $0.5 \mathrm{H}, \mathrm{Ar}), 7.86-7.82(\mathrm{~m}, 0.5 \mathrm{H}, \mathrm{Ar}), 7.74-7.70(\mathrm{~m}, 0.5 \mathrm{H}, \mathrm{Ar}), 7.66-7.61(\mathrm{~m}, 0.5 \mathrm{H}, \mathrm{Ar})$, 7.57-7.47 (m, 1.5H, Ar), 7.46-7.37 (m, 2.5H, Ar), 7.34-7.27 (m, 4H, Ar), 7.25-7.17 (m, 1H, Ar), $6.19\left(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H1}{ }^{\prime}\right), 6.04\left(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{Hl}{ }^{\prime}\right), 5.60-5.54(\mathrm{~m}, 0.5 \mathrm{H}$, H2'), 5.50-5.42 (m, 0.5H, H2'), 5.16-4.99 (m, 2H, CH ${ }_{2} \mathrm{Ph}$ ), 4.46-4.40 (m, 0.5H, H4'), 4.36-4.30 (m, 0.5H, H4'), 4.13-4.04 (m, 1H, CHCH 3 L-Ala), 3.90-3.83 (m, 1H, H5'), 3.643.56 ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{H} 5$ '), 2.61-2.54 (m, 0.5H, H3'), 2.49-2.41 (m, 0.5H, H3'), 2.35-2.27 (m, $0.5 \mathrm{H}, \mathrm{H} 3$ '), 2.22-2.16 (m, 0.5H, H3'), 1.35-1.24 (m, 3H, $\mathrm{CHCH}_{3}$ L-Ala).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 174.52$ ( $\mathrm{C}=\mathrm{O}$ ), 174.49 ( $\mathrm{C}=\mathrm{O}$ ), 157.27 (C6), 153.58 (C2), 149.97 (C4), 149.93 (C-4), $147.70\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 147.48\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5\right.$ $\mathrm{Hz}, \mathrm{C}-\mathrm{Ar}), 141.36$ (C8), 141.19 (C8), 137.25 (C-Ar), 137.05 (C-Ar), 136.31 (C-Ar), 136.20 (C-Ar), 129.58 (CH-Ar), 129.48 (CH-Ar), 129.37 (CH-Ar), 129.26 (CH-Ar), 129.22 (CH-Ar), 128.88 (CH-Ar), 127.84 (CH-Ar), 127.75 (CH-Ar), 127.49 (CH-Ar), 127.44 (CH-Ar), 126.48 (CH-Ar), 126.39 (CH-Ar), 126.26 (CH-Ar), 126.05 (CH-Ar), 122.76 (CH-Ar), 122.38 (CH-Ar), 120.68 (C5), 120.61 (C5), 116.64 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.75 \mathrm{~Hz}$, CH-Ar), $116.13\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.75, \mathrm{CH}-\mathrm{Ar}\right), 91.60\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{C} 1{ }^{\prime}\right), 91.43\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=\right.$ $\left.7.5 \mathrm{~Hz}, \mathrm{C} 1{ }^{\prime}\right), 82.74\left(\mathrm{C} 4\right.$ '), $82.27\left(\mathrm{C}{ }^{\prime}\right), 81.99\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.5 \mathrm{~Hz}, \mathrm{C} 2\right.$ '), $81.12\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.5\right.$ $\mathrm{Hz}, \mathrm{C} 2$ ') , $67.97\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.94\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 64.16\left(\mathrm{C}^{\prime}\right), 63.51\left(\mathrm{C}^{\prime}\right), 51.96\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$, $51.89\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $33.89\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{CHCH}_{3}\right.$ L-Ala), $33.63\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}\right.$, $\mathrm{CHCH}_{3}$ L-Ala).
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{OH}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{ml} / \mathrm{min}, \lambda=200 \mathrm{~nm}$, showed two peaks of the diastereoisomers with tR 24.84 min and tR 25.43 min .
$\left(\mathbf{E S}^{+}\right) \mathbf{m} / \mathbf{z}$, found $619.2\left[\mathrm{M}^{+} \mathrm{H}^{+}\right], 641.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], 1259.4\left[2 \mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{30} \mathrm{H}_{31} \mathrm{~N}_{6} \mathrm{O}_{7} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 618.20[\mathrm{M}]$.

## [29a] 3'-Deoxyadenosine $\mathbf{2}^{\prime}$, ' $^{\prime}$ '-bis- $O$-phenyl-(benzyloxy-L-alaninyl)-phosphate



Isolated from reaction used for the synthesis of 3'-deoxyadenosine-2'-O-phenyl-(benzyloxy-L-alaninyl)-phosphate [28a] ( $0.019 \mathrm{~g}, 11 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 3.98,3.88$, 3.59, 3.12, 3.05, 2.45, 2.32.
${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.24-8.13 (m,

2H, H8, H2), 7.39-7.08 (m, 20H, Ph), 6.27-6.23 (m, 0.5H, H1'), 6.16-6.13 (m, 0.5H, H1'), 5.61-5.48 (m, 1H, H2'), 5.17-4.91 (m, 4H, CH $\mathrm{C}_{2} \mathrm{Ph} \times 2$ ), 4.57-4.49 (m, 1H, H4'), 4.41-4.29 (m, 1H, H5'), 4.25-4.15 (m, 1H, H5'), 4.10-4.01 (m, 1H, CHCH ${ }_{3}$ L-Ala), 3.99-3.89 (m, $1 \mathrm{H}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ), 2.57-2.41 (m, 1H, H3'), 2.28-2.17 (m, 1H, H3'), 1.38-1.23 (m, 6H, $\mathrm{CHCH}_{3}$ L-Alax 2).
${ }^{13} \mathbf{C}$ NMR ( $125 \mathrm{MHz}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 174.88(\mathrm{C}=\mathrm{O}), 174.83(\mathrm{C}=\mathrm{O}), 174.79(\mathrm{C}=\mathrm{O}), 174.73$ ( $\mathrm{C}=\mathrm{O}$ ), 174.61 ( $\mathrm{C}=\mathrm{O}), 174.57(\mathrm{C}=\mathrm{O}), 174.53$ ( $\mathrm{C}=\mathrm{O}$ ), 157.36 (C6), 157.34 (C6), 157.32(C6), 157.29 (C6), 154.04 (C2), 154.01 (C2), 153.97 (C2), 153.94 (C2), 152.09 (C4), 152.04 (C4), 152.02 (C4), 151.97 (C4), 150.31 (C-Ar), 150.29 (C-Ar), 150.16 (CAr), 140.98 (C8), 140.91 (C8), 140.81 (C8), 137.31 (C-Ar), 137.28 (C-Ar), 137.22 (C-Ar), 137.09 (C-Ar), 130.86 (CH-Ar), 130.78 (CH-Ar), 130.77 (CH-Ar), 129.65 (CH-Ar), 129.61 (CH-Ar), 129.58 (CH-Ar), 129.55 (CH-Ar), 129.44 (CH-Ar), 129.42 (CH-Ar), 129.38 (CH-Ar), 129.34 (CH-Ar), 129.32 (CH-Ar), 129.30 (CH-Ar), 129.28 (CH-Ar), 129.23 (CH-Ar), 129.21 (CH-Ar), 12.42 (CH-Ar), 126.23 (CH-Ar), 126.20 (CH-Ar), 126.17 (CH-Ar), 121.65 (CH-Ar), 121.63 (CH-Ar), 121.61 (CH-Ar), 121.59 (CH-Ar), 121.52 (CH-Ar), 121.50 (CH-Ar), 121.47 (CH-Ar), 121.46 (CH-Ar), 121.40 (CH-Ar), 121.39 (CH-Ar), 121.36 (CH-Ar), 121.35 (CH-Ar), 121.30 (CH-Ar), 121.28 (CH-Ar), 121.26 (CH-Ar), 121.24 (CH-Ar), 120.61 (C5), 120.57 (C5), 120.56 (C5), 120.54 (C5),
 81.79 (C2'), 81.27 ( $\mathrm{C}^{\prime}$ ), 81.22 ( $\mathrm{C}^{\prime}$ ), 81.18 ( $\mathrm{C}^{\prime}$ ), 80.49 ( $\mathrm{C}^{\prime}$ ), 80.43 ( C 4 '), 80.06 ( C 4 '), 79.99 (C4'), 68.29 ( $\left.\mathrm{C}^{\prime}, C H_{2} \mathrm{Ph}\right), 68.25$ ( $\left.\mathrm{C}^{\prime}, \mathrm{CH}_{2} \mathrm{Ph}\right), 68.00\left(\mathrm{C}^{\prime}, \mathrm{CH}_{2} \mathrm{Ph}\right), 67.96$ (C5’, $\left.\mathrm{CH}_{2} \mathrm{Ph}\right), 67.94\left(\mathrm{C}^{\prime}, \mathrm{CH}_{2} \mathrm{Ph}\right), 67.90\left(\mathrm{C}^{\prime}, \mathrm{CH}_{2} \mathrm{Ph}\right), 67.71$ (C5', CH ${ }_{2} \mathrm{Ph}$ ), 67.67 (C5', $\left.\mathrm{CH}_{2} \mathrm{Ph}\right), 51.91\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.74\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.70\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.59$ $\left(\mathrm{CHCH}_{3}\right.$ L-Ala), 34.22 ( C 3 '), 34.20 ( C 3 '), 34.16 ( $\mathrm{C}^{\prime}$ ), 33.97 ( $\mathrm{C}^{\prime}$ ), 33.94 ( C 3 '), 33.91 (C3'), $20.44\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $20.43\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $20.39\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 20.29$ $\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $20.27\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $20.24\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $20.21\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $20.19\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$.

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{ml} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed one broad peak with tR 15.97 min .
(ES+) $\mathbf{m} / \mathbf{z}$, found $886.3\left[\mathrm{M}+\mathrm{H}^{+}\right], 1771.6\left[2 \mathrm{M}+\mathrm{H}^{+}\right] \mathrm{C}_{42} \mathrm{H}_{45} \mathrm{~N}_{7} \mathrm{O}_{11} \mathrm{P}_{2}$ required $\mathrm{m} / \mathrm{z} 885.3[\mathrm{M}]$.
[29b] 3'-Deoxyadenosine 2',5'-bis-O-naphth-1-yl-(benzyloxy-L-alaninyl)-phosphate


Isolated from reaction used for the synthesis of 3'-deoxyadenosine-2'-O-naphth-1-yl-(benzyloxy-L-alaninyl)-phosphate [28b] as a white solid ( $0.069 \mathrm{~g}, 35 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\left.\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}} 4.23,4.18$, 3.97, 3.94, 3.48, 3.39, 2.78, 2.77.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.18-8.05 (m, $3 \mathrm{H}, \mathrm{H} 8, \mathrm{H} 2, \mathrm{Ar}), 8.03-7.81$ (m, 3H, Ar), 7.74-7.62 (m, 2H, Ar), 7.54-7.39 (m, 7H, Ar), 7.37-7.16 (m, 11H, Ar), 6.22-6.18 (m, 0.5H, H1'), 6.05-6.02 (m, 0.5H, H1'), 5.67-5.54 (m, $1 \mathrm{H}, \mathrm{H} 2$ '), 5.15-4.96 (m, 4H, CH2 $\mathrm{Ph} \times 2$ ), 4.50-4.42 (m, 1H, H4'), 4.39-4.31 (m, 1H, H5'), 4.24-4.16 (m, 1H, H5'), 4.15-4.08 (m, 1H, $\mathrm{CHCH}_{3}$ L-Ala), 4.04-3.91 (m, 1H, $\mathrm{CHCH}_{3}$ LAla), 3.60-2.41 (m, 1H, H3'), 2.29-2.09 (m, 1H, H3'), 1.37-1.27 (m, 6H, $\mathrm{CHCH}_{3}$ L-Ala x $2)$.
[30a] 3'-Deoxyadenosine 5'-O-phenyl-(benzyloxy-L-alaninyl)-phosphate


Prepared according to general procedure $\mathbf{F}_{2}$ using 3'-
 deoxyadenosine ( $0.05 \mathrm{~g}, 0.20 \mathrm{mmol}$ ) in anhydrous THF (4 $\mathrm{mL}), \quad N$-methylimidazole $\quad(80 \quad \mu \mathrm{~L}, \quad 1.0 \quad \mathrm{mmol})$, phenyl(benzyloxy-L-alaninyl) phosphorochloridate [17s] ( $0.021 \mathrm{~g}, 0.6 \mathrm{mmol}$ ) in THF ( 2.4 mL ) Purification by Biotage Isolera One (cartridge SNAP $25 \mathrm{~g}, 25 \mathrm{~mL} / \mathrm{min}$, $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 1-8 \% 10 \mathrm{CV}, 8 \% 5 \mathrm{CV}$ ) and preparative TLC $\left(1000 \mu \mathrm{M}\right.$, eluent system $\left.\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 5 / 95\right)$ afforded the title compound as a white solid ( $0.032 \mathrm{~g}, 28 \%$ ).
${ }^{31} \mathbf{P} \mathbf{N M R}\left(\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 3.91, 3.73.
${ }^{1} \mathbf{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.26(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.24(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.22(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 2)$, $8.21(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 2), 7.34-7.25(\mathrm{~m}, 7 \mathrm{H}, \mathrm{Ar}), 7.21-7.13(\mathrm{~m}, 3 \mathrm{H}, \mathrm{Ar}), 6.01(\mathrm{~d}, J=1.5 \mathrm{~Hz}, 0.5 \mathrm{H}$, H1'), $6.00\left(\mathrm{~d}, J=1.5 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1\right.$ ') , $5.15-5.04\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.73-4.63$ (m, 2H, H2', H4'), 4.43-4.35 (m, 1H, H5'), 4.27-4.20 (m, 1H, H5'), 4.03-3.91 (m, 1H, CHCH ${ }_{3}$ L-Ala), 2.35-2.28 (m, 1H, H3'), 2.09-2.02 (m, 1H, H3'), 1.32 (d, $J=7.4 \mathrm{~Hz}, 1.5 \mathrm{H}, \mathrm{CHCH}_{3} \mathrm{~L}-$ Ala), 1.28 (d, $\left.J=7.4 \mathrm{~Hz}, 1.5 \mathrm{H}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$.
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 174.84\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.5 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.63\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.5\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 157.32 (C6), 157.31 (C6), 153.86 (C2), 153.84 (C2), 152.13 (C4), 152.07 (C4), 150.20 (C-Ar), 150.18 (C-Ar), 140.47 (C8), 137.26 (C-Ar), 137.19 (C-Ar), 130.76 (CH-

Ar), 130.74 (CH-Ar), 129.57 (CH-Ar), 129.32 (CH-Ar), 129.31 (CH-Ar), 129.29 (CH-Ar), 129.26 (CH-Ar), 126.16 (CH-Ar), 126.14 (CH-Ar), 121.46 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), 121.38 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right) 120.54$ (C5), 120.53 (C5), 93.24 (C1'), 93.18 (C1’), $80.43\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.6 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $80.36\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.6 \mathrm{~Hz}, \mathrm{C} 4{ }^{\prime}\right), 76.62\left(\mathrm{C} 2^{\prime}\right), 68.62\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=\right.$ $5.3 \mathrm{~Hz}, \mathrm{C} 5$ '), $68.30\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.3 \mathrm{~Hz}, \mathrm{C} 5{ }^{\prime}\right), 67.95\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.92\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 51.74\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $51.60\left(\mathrm{CHCH}_{3}\right.$ L-Ala), 34.91 ( $\left.\mathrm{C} 3^{\prime}\right), 34.70\left(\mathrm{C} 3\right.$ '), 20.45 (d, ${ }^{3} \mathrm{~J}_{\mathrm{C}-\mathrm{P}}=7.0 \mathrm{~Hz}, \mathrm{CHCH}_{3}$ L-Ala), 20.28 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.0 \mathrm{~Hz}, \mathrm{CHCH}_{3}$ L-Ala).

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=200 \mathrm{~nm}$, showed two peaks of the diastereoisomers with tR 14.02 min . and tR 14.26 min .
$\mathbf{M S}\left(\mathbf{E S}^{+}\right) \mathbf{m} / \mathbf{z}$ found $569.2\left[\mathrm{M}+\mathrm{H}^{+}\right]$, $591.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], 1159.4\left[2 \mathrm{M}+\mathrm{Na}^{+}\right] \mathrm{C}_{26} \mathrm{H}_{29} \mathrm{~N}_{6} \mathrm{O}_{7} \mathrm{P}$ required m/z 568.2 [M].

The two diasteroisomers were separated via Biotage Isolera One (cartridge SNAP-Ultra C18 $12 \mathrm{~g}, \mathrm{~F}: 12 \mathrm{~mL} / \mathrm{min}$, isocratic eluent system: $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{OH} 45 \%$ in $30 \mathrm{~min}, 50 \mathrm{mg}$ sample).

## Fast eluting isomer:

${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 3.91$.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.26(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 8.22(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 7.37-7.25(\mathrm{~m}, 7 \mathrm{H}, \mathrm{Ar})$, 7.22-7.12 (m, 3H, Ar), $6.01\left(\mathrm{~d}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 5.12\left(\mathrm{AB} \mathrm{q}, J_{\mathrm{AB}}=12.0 \mathrm{~Hz}, \Delta \delta_{\mathrm{AB}}=\right.$ 0.04, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}$ ), 4.74-4.70 (m, 1H, H2'), 4.69-4.62 (m, 1H, H4'), 4.44-4.38 (m, 1H, H5'), 4.28-4.21 (m, 1H, H5'), 3.99-3.90 (m, 1H, CHCH ${ }_{3}$ L-Ala), 2.35-2.27 (m, 1H, H3'), 2.09-2.02 (m, 1H, H3'), 1.29 (d, J=7.0 Hz, 3H, CHCH $H_{3}$ L-Ala).

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $F: 1 \mathrm{~mL} / \mathrm{min}, \lambda=200 \mathrm{~nm}$, showed one peak with tR 14.02 min .

## Slow eluting isomer:

${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 3.73$.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.24(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 8.22(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 7.36-7.26(\mathrm{~m}, 7 \mathrm{H}, \mathrm{Ar})$, 7.22-7.13 (m, 3H, Ar), $6.01\left(\mathrm{~d}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 5.08\left(\mathrm{AB} \mathrm{q}, J_{\mathrm{AB}}=12.0 \mathrm{~Hz}, \Delta \delta_{\mathrm{AB}}=\right.$ 0.01, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}$ ), 4.70-4.67 (m, 1H, H2'), 4.66-4.60 (m, 1H, H4'), 4.41-4.35 (m, 1H, H5'), 4.26-4.19 (m, 1H, H5'), 4.02-3.94 (m, 1H, CHCH ${ }_{3}$ L-Ala), 2.36-2.27 (m, 1H, H3'), 2.08-2.01 (m, 1H, H3'), 1.34-1.30 (m, 3H, $\mathrm{CHCH}_{3}$ L-Ala).

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}: 1 \mathrm{~mL} / \mathrm{min}, \lambda=200 \mathrm{~nm}$, showed one peak with tR 14.26 min .
[30b] 3'-Deoxyadenosine 5'-O-naphth-1-yl(benzyloxy-L-alaninyl)-phosphate


Prepared according to general procedure $\mathbf{F}_{2}$ using 3'deoxyadenosine ( $0.15 \mathrm{~g}, 0.6 \mathrm{mmol}$ ) in THF ( 12 mL ), N methylimidazole ( $240 \mu \mathrm{~L}, 3.0 \mathrm{mmol}$ ), naphth-1-yl-(benzyloxy-L-alaninyl) phosphorochloridate [17a] (727 mg, 1.8 mmol ) in THF ( 7 mL ). Purification by Biotage Isolera One (cartridge SNAP KP-SIL $50 \mathrm{~g}, 100 \mathrm{~mL} / \mathrm{min}$, $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 1-10 \% 10 \mathrm{CV}, 10 \% 5 \mathrm{CV}$ ) and preparative TLC ( $2000 \mu \mathrm{M}$, eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 5 / 95$ ) afforded compound as white solid ( $0.045 \mathrm{~g}, 12 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 4.32,4.14$.
${ }^{1} \mathbf{H}$ NMR ( $500 \mathbf{M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.24(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.22(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.20(\mathrm{~s}, 0.5 \mathrm{H}$, H2), 8.19 (s, 0.5H, H2), 8.14-8.09 (m, 1H, Ar), 7.89-7.85 (m, 1H, Ar), 7.70-7.67 (m, 1H, Ar), 7.53-7.42 (m, 3H, Ar), 7.39-7.34 (m, 1H, Ar), 7.31-7.25 (m, 5H, Ar), 5.99 (d, $J=2.0$ $\left.\mathrm{Hz}, 0.5 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 5.98\left(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H1}{ }^{\prime}\right), 5.10-5.01\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.72-4.61(\mathrm{~m}$, 2H, H2', H4'), 4.47-4.40 (m, 1H, H5'), 4.33-4.24 (m, 1H, H5'), 4.09-3.98 (m, 1H, CHCH L-Ala) 2.35-2.26 (m, 1H, H3'), 2.07-1.98 (m, 1H, H3'), 1.30-1.24 (m, 3H, CHCH $H_{3}$ L-Ala). ${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 174.85\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.56\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 157.33 (C6), 157.31 (C6), 153.87 (C2), 153.85 (C2), 150.24 (C4), 150.23 (C4), $147.91\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 147.95$, (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 140.56$ (C8), 140.50 (C8), 137.22 (C-Ar), 137.17 (C-Ar), 136.28 (C-Ar), 129.55 (CH-Ar), 129.53 (CH-Ar), 129.30 (CH-Ar), 129.25 (CH-Ar), 128.88 (CH-Ar), 128.82 (CH-Ar), $127.91\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=6.2\right.$ $\mathrm{Hz}, \mathrm{C}-\mathrm{Ar}), 127.83\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 127.77(\mathrm{CH}-\mathrm{Ar}), 127.75$ (CH-Ar), 127.49 (CH-Ar), 127.45 ( $\mathrm{CH}-\mathrm{Ar}$ ), 126.48 ( $\mathrm{CH}-\mathrm{Ar}$ ), 126.47 (CH-Ar), 126.02 (CH-Ar), 125.97 (CH-Ar), 122.77 (CH-Ar), 122.63 (CH-Ar), 120.58 (C5), 120.53 (C5), 116.35 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=$ $3.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}), 116.15\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 93.22\left(\mathrm{C} 1{ }^{\prime}\right), 93.20\left(\mathrm{C} 1^{\prime}\right), 80.30\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}}-\right.$ $\left.{ }_{\mathrm{P}}=2.7 \mathrm{~Hz}, \mathrm{C} 4{ }^{\prime}\right), 80.24\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.7 \mathrm{~Hz}, \mathrm{C} 4\right.$ ') , $76.51\left(\mathrm{C} 2^{\prime}\right), 76.44\left(\mathrm{C} 2^{\prime}\right), 68.87\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=\right.$ $5.2 \mathrm{~Hz}, \mathrm{C} 5$ '), $68.64\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.2 \mathrm{~Hz}, \mathrm{C} 5{ }^{\prime}\right), 67.93\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 51.82\left(\mathrm{CHCH}_{3}\right.$ L-Ala), 51.73 $\left(\mathrm{CHCH}_{3}\right.$ L-Ala), 35.01 ( C 3 '), 34.76 ( $\mathrm{C}^{\prime}$ ), 20.41 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.7 \mathrm{~Hz}, \mathrm{CHCH}_{3}$ L-Ala), 20.22 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.7 \mathrm{~Hz}, \mathrm{CHCH}_{3}$ L-Ala).
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $F=1 \mathrm{ml} / \mathrm{min}, \lambda=200 \mathrm{~nm}$, showed two peaks of the diastereoisomers with tR 16.36 min . and tR 16.60 min .
(ES+) m/z found $619.2\left[\mathrm{M}+\mathrm{H}^{+}\right], 641.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], 1259.4\left[2 \mathrm{M}+\mathrm{Na}^{+}\right] \mathrm{C}_{30} \mathrm{H}_{31} \mathrm{~N}_{6} \mathrm{O}_{7} \mathrm{P}$ required m/z 618.58 [M].

## [30c] 3'-Deoxyadenosine 5'-O-(benzyloxy-glycinyl)-phosphate



Prepared according to general procedure $\mathbf{F}_{\mathbf{2}}$ using 3'-deoxyadenosine ( $0.05 \mathrm{~g}, 0.20 \mathrm{mmol}$ ) in THF ( 4 mL ), $N$-methylimidazole ( $80 \mu \mathrm{~L}, 1.0 \mathrm{mmol}$ ), phenyl (benzyloxy-glycinyl) phosphorochloridate [17r] ( $204 \mathrm{mg}, 0.6 \mathrm{mmol}$ ) in THF ( 2.4 mL ). Purification by column chromatography (eluent system
$\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} \quad 0 / 100$ to $6 / 94$ ) and preparative TLC (500 $\mu \mathrm{M}$, eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 5 / 95$ ) afforded the compound as a white solid ( $0.021 \mathrm{~g}, 19 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 5.12, 4.91.
${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.27(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.24(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.22(\mathrm{~s}, 0.5 \mathrm{H}$, H2), 8.21 (s, 0.5H, H2), 7.37-7.26 (m, 7H, Ph), 7.22-7.13 (m, 3H, Ph), 6.02 (d, $J=1.8 \mathrm{~Hz}$, $0.5 \mathrm{H}, \mathrm{H} 1$ '), $6.00\left(\mathrm{~d}, J=1.8 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H1}{ }^{\prime}\right), 5.15-5.12\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.73-4.64(\mathrm{~m}, 2 \mathrm{H}$, H2', H4'), 4.50-4.39 (m, 1H, H5'), 4.36-4.24 (m, 1H, H5'), 3.53-3.71 (m, 2H, CH2 Gly), 2.39-2.25 (m, 1H, H3'), 2.13-2.02 (m, 1H, H3').
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 172.30\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 172.27\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 157.34 (C6), 157.32 (C6), 153.88 (C2), 153.87 (C2), 152.08 ( $\mathrm{d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}$, C-Ar), 152.05 ( $\mathrm{d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}$ ), 150.20 (C4), 150.19 (C4), 140.52 (C8), 140.42 (C8), 137.15 (C-Ar), 130.79 (CH-Ar), 129.57 (CH-Ar), 129.55 (CH-Ar), 129.35 (CH-Ar), 129.34 (CH-Ar), 129.33 (CH-Ar), 126.22 (CH-Ar), 121.44 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), 121.40 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), 120.51 (C5), 120.49 (C5), 93.19, 93.14 (C1'), 80.46 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.6 \mathrm{~Hz}, \mathrm{C} 4$ '), $80.39\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.6 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $76.66\left(\mathrm{C} 2{ }^{\prime}\right), 68.68\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.4 \mathrm{~Hz}\right.$, C 5 '), $68.24\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.95\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.93\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 43.90\left(\mathrm{CH}_{2}\right.$ Gly $)$, 43.83 ( $\mathrm{CH}_{2}$ Gly), 34.83 ( C 3 '), 34.54 ( $\mathrm{C}^{\prime}$ ).

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=200 \mathrm{~nm}$, showed two peaks of the diastereoisomers with tR 13.63 min . and tR 13.41 min .
(ES+) $\mathbf{m} / \mathbf{z}$ found: $555.2\left[\mathrm{M}+\mathrm{H}^{+}\right]$, $577.2\left[\mathrm{M}+\mathrm{Na}^{+}\right]$, $1131.4\left[2 \mathrm{M}+\mathrm{Na}^{+}\right] \mathrm{C}_{25} \mathrm{H}_{27} \mathrm{~N}_{6} \mathrm{O}_{7} \mathrm{P}$ required m/z 554.2 [M].
[30d] 3'-Deoxyadenosine 5'-O-naphth-1-yl-(pentyl-1-oxy-L-leucinyl)-phosphate


Prepared according to general procedure $\mathbf{F}_{2}$ using 3'deoxyadenosine ( $0.048 \mathrm{~g}, 0.19 \mathrm{mmol}$ ) in THF ( 4 mL ), $N$ methylimidazole ( $76 \mu \mathrm{~L}, 0.95 \mathrm{mmol}$ ), naphth-1-yl(pentyl-1-oxy-L-leucinyl) phosphorochloridate [17f] ( $0.25 \mathrm{~g}, 0.6$ mmol ) in THF ( 2.4 mL ).

Purification by Biotage Isolera One, cartridge SNAP $25 \mathrm{~g}, 25 \mathrm{~mL} / \mathrm{min}, \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 2$ $20 \%(15 \mathrm{CV}), 20 \%(5 \mathrm{CV})$ and preparative TLC ( $1000 \mu \mathrm{M}$, eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ $4 / 96$ ) afforded the compound as white solid $(0.027 \mathrm{~g}, 22 \%)$.
${ }^{\mathbf{3 1}} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 4.64,4.37$.
${ }^{1} \mathbf{H}$ NMR ( $500 \mathbf{M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.28(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.25(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.21(\mathrm{~s}, 0.5 \mathrm{H}$, H2), 8.20 ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 2$ ), 8.17-8.12 (m, 1H, H8 Nap), 7.88-7.83 (m, 1H, H5 Nap), 7.69-7.66 (m, 1H, H4 Nap), 7.54-7.42 (m, 3H, H2, H6, H7 Nap), 7.40-7.35 (m, 1H, H3 Nap), 6.01 (d, $\left.J=2.1 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 6.00(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1$ '), 4.47-4.67 (m, 2H, H2', H4'), 4.55-4.44 (m, 1H, H5'), 4.43-4.31 (m, 1H, H5'), 4.00-3.87 (m, 3H, CHCH2 $\mathrm{CH}_{2}\left(\mathrm{CH}_{3}\right)_{2}$ LLeu, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 2.44-2.30 (m, 1H, H3'), 2.14-2.04 (m, 1H, H3'), 1.661.39 (m, 5H, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 1.1.28-1.21 (m, 4H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), $0.86-0.81\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $0.81-0.68$ (m, $6 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 175.42\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 175.04\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 157.32 (C6), 153.87 (C2), 153.86 (C2), 150.23 (C4), 147.97 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}$, C-Ar), 140.55 (C8), 136.30 (C-Ar), 136.29 (C-Ar), 128.89 (CH-Ar), 128.84 (CH-Ar), 127.95 (C-Ar), 127.91 (C-Ar), 127.84 (C-Ar), 127.78 (CH-Ar), 127.76 (CH-Ar), 127.46 (CH-Ar), 126.50 (C-Ar), 126.48 (C-Ar), 126.46 (C-Ar), 126.01 (CH-Ar), 125.91 (CH-Ar), 122.80 (CH-Ar), 122.70 (CH-Ar), 120.58 (C5), 120.56 (C5), 116.40 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}$, CH-Ar), 116.01 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), $93.31\left(\mathrm{C} 1^{\prime}\right), 93.27\left(\mathrm{C} 1^{\prime}\right), 80.35\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.5\right.$ $\mathrm{Hz}, \mathrm{C} 4$ '), $80.29\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.5 \mathrm{~Hz}, \mathrm{C} 4^{\prime}\right), 76.54\left(\mathrm{C} 2^{\prime}\right), 76.50\left(\mathrm{C} 2^{\prime}\right), 69.07\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.5 \mathrm{~Hz}\right.$, C 5 '), $68.85\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.5 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $66.33\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), 66.32 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen $), 54.81\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $54.71\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $44.22\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.6 \mathrm{~Hz}, \mathrm{CHCH} \mathrm{H}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $43.93\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.6 \mathrm{~Hz}\right.$, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 35.15 ( $\left.\mathrm{C} 3^{\prime}\right), 34.86$ ( $\left.\mathrm{C} 3 '\right), 29.32\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $29.30 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen $), \quad 29.11\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen $), \quad 25.67$ $\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $25.45\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $23.30\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ $n$-Pen), $23.12\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $23.02\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 22.04
$\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $21.78\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $14.28\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ $n$-Pen).
(ES+) m/z found $641.3\left[\mathrm{M}+\mathrm{H}^{+}\right], 663.3\left[\mathrm{M}+\mathrm{Na}^{+}\right], 1303.6\left[2 \mathrm{M}+\mathrm{Na}^{+}\right] \mathrm{C}_{31} \mathrm{H}_{41} \mathrm{~N}_{6} \mathrm{O}_{7} \mathrm{P}$ required m/z 640.3 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $F=1 \mathrm{~mL} / \mathrm{min}, \lambda=200 \mathrm{~nm}$, showed one peak of the two overlapping diastereoisomers with tR 20.84 min .
[30e] 3'-Deoxyadenosine 5'-O-naphth-1-yl(methoxy-2,2-dimethylglycinyl)-phosphate


Prepared according to general procedure $\mathbf{F}_{\mathbf{2}}$ using 3'deoxyadenosine ( $0.15 \mathrm{~g}, 0.6 \mathrm{mmol}$ ) in THF ( 4 mL ), N methylimidazole $(240 \quad \mu \mathrm{~L}, 3.0 \mathrm{mmol})$, naphth-1-yl(methoxy-2,2-dimethylglycinyl) phosphorochloridate [ $\mathbf{1 7 q} \mathbf{q}](0.61 \mathrm{~g}, 1.8 \mathrm{mmol})$ in THF ( 2.4 mL ). Purification by Biotage Isolera One, cartridge SNAP KP-SIL $50 \mathrm{~g}, 100 \mathrm{~mL} / \mathrm{min}, \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 2-20 \%$ $(10 \mathrm{CV}), 20 \%(5 \mathrm{CV})$ and preparative TLC ( $1000 \mu \mathrm{M}$, eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 4 / 96$ ) afforded the compound as a white solid $(0.02 \mathrm{~g}, 6 \%)$.
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 2.73$.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.28$ ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 8$ ), 8.25 ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 8$ ), 8.21 ( $\mathrm{s}, 0.5 \mathrm{H}$, H2), 8.19 ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 2$ ), 8.18-8.14 (m, 1H, H8 Nap), 7.90-7.84 (m, 1H, H5 Nap), 7.71-7.66 (m, 1H, H4 Nap), 7.53-7.47 (m, 3H, H2, H6, H7 Nap), 7.41-7.35 (m, 1H, H3 Nap), 6.03 (d, $J=2.1 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1$ '), 5.99 (d, $J=2.1 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1$ '), 4.76-4.67 (m, 2H, H2', H4'), 4.52-4.44 (m, 1H, H5'), 4.42-4.33 (m, 1H, H5'), $3.65\left(\mathrm{~s}, 1.5 \mathrm{H}, \mathrm{OCH}_{3}\right), 3.64(\mathrm{~s}, 1.5 \mathrm{H}$, $\mathrm{OCH}_{3}$ ), 2.48-2.41 (m, 0.5H, H3'), 2.37-2.30 (m, 0.5H, H3'), 2.15-2.09 (m, 0.5H, H3'), 2.08-2.02 (m, 0.5H, H3'), 1.47-1.44 (m, 6H, ( $\left.\mathrm{CH}_{3}\right)_{2}$ DMG).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 177.25$ ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}$ ), 157.53 (C6), 157.51 (C6), 153.86 (C2), 150.28 (C4), 150.25 (C4), 148.06 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}$ ), 148.04 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 140.67$ (C8), 140.60 (C8), 136.28 (C-Ar), 136.27 (C-Ar), 128.82 (CH-Ar), 128.80 (CH-Ar), 127.93 (d, $\left.{ }^{2} J_{\mathrm{C}-\mathrm{P}}=6.25 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 127.92\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=6.25 \mathrm{~Hz}\right.$, C-Ar), 127.71 (CH-Ar), 127.69 (CH-Ar), 127.32 (CH-Ar), 126.44 (CH-Ar), 125.84 (CH$\mathrm{Ar}), 122.93$ (CH-Ar), 120.56 (C5), 120.50 (C5), 116.38 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.75 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), $116.36\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 93.25\left(\mathrm{Cl}^{\prime}\right), 80.40\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.0 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $80.33(\mathrm{~d}$, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.0 \mathrm{~Hz}, \mathrm{C} 4$ '), 76.57 (C2'), $76.43\left(\mathrm{C} 2\right.$ '), $68.99\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.5 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $68.84\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}\right.$ $\left.=5.5 \mathrm{~Hz}, \mathrm{C} 5^{\prime}\right), 53.01\left(\mathrm{OCH}_{3}\right), 35.22\left(\mathrm{C} 3^{\prime}\right), 34.90\left(\mathrm{C} 3^{\prime}\right), 27.85\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.0 \mathrm{~Hz},\left(\mathrm{CH}_{3}\right)_{2}\right.$

DMG), $27.80\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.0 \mathrm{~Hz},\left(\mathrm{CH}_{3}\right)_{2} \mathrm{DMG}\right), 27.60\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.0 \mathrm{~Hz},\left(\mathrm{CH}_{3}\right)_{2} \mathrm{DMG}\right)$, $27.56\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.0 \mathrm{~Hz},\left(\mathrm{CH}_{3}\right)_{2} \mathrm{DMG}\right)$.

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $F=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed two peaks with $\mathrm{tR} 16.51 \mathrm{~min}, \mathrm{tR} 16.75 \mathrm{~min}$. (ES+) m/z found $557.2\left[\mathrm{M}+\mathrm{H}^{+}\right], 579.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], 1135.4\left[2 \mathrm{M}+\mathrm{Na}^{+}\right] \mathrm{C}_{25} \mathrm{H}_{29} \mathrm{~N}_{6} \mathrm{O}_{7} \mathrm{P}$ required $\mathrm{m} / \mathrm{z}$ [M] 556.51.
[30f] 3'-Deoxyadenosine 5'-O-ethyl-3-(2-hydroxyphenyl)propanoyl-(benzyloxy-L-alaninyl)-phosphate


Prepared according to general procedure $\mathbf{F}_{2}$ using 3'deoxyadenosine ( $0.021 \mathrm{~g}, 0.84 \mathrm{mmol}$ ) in THF ( 4 mL ), $N$-methylimidazole ( $320 \mu \mathrm{~L}, 4.2 \mathrm{mmol}$ ), ethyl 3-(2-hydroxyphenyl)propanoyl-L-alanine benzyl ester phosphorochloridate [17v] ( $1.14 \mathrm{~g}, 2.5 \mathrm{mmol}$ ) in THF $(2.4 \mathrm{~mL})$. Purification by Biotage Isolera One, cartridge ZIP Sphere $80 \mathrm{~g}, 100 \mathrm{~mL} / \mathrm{min}, \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 1-8 \%(10 \mathrm{CV}), 8 \%(5 \mathrm{CV})$ and preparative TLC ( $1000 \mu \mathrm{M}$, eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 5 / 95$ ) afforded the title compound as a white solid ( $0.032 \mathrm{~g}, 18 \%$ ).
${ }^{31} \mathbf{P}$ NMR (202 MHz, CD $\left.\mathbf{H}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}} 3.95,3.65$.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.25(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.21(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8, \mathrm{H} 2), 8.20(\mathrm{~s}, 0.5 \mathrm{H}$, H2), 7.35-7.29 (m, 6H, Ar), 7.25-7.21 (m, 1H, Ar), 7.16-7.07 (m, 2H, Ar), 6.00 (d, $J=1.9$ $\left.\mathrm{Hz}, 0.5 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 5.98$ (d, $\left.J=1.9 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 5.17-5.05\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.76-4.73$ ( m, $\left.0.5 \mathrm{H}, \mathrm{H} 2^{\prime}\right)$, 4.70-4.59 (m, 1.5H, H2', H4'), 4.45-4.34 (m, 1H, H5'), 4.30-4.22 (m, 1H, H5'), 4.08-3.96 (m, 3H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OCH}_{2} \mathrm{CH}_{3}, \mathrm{CHCH}_{3}$ L-Ala), 2.98-2.92 (m, 2 H , $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OCH}_{2} \mathrm{CH}_{3}$ ), 2.62-2.56 (m, 2H, CH2 $\mathrm{CH}_{2} \mathrm{OCH}_{2} \mathrm{CH}_{3}$ ), 2.40-2.29 (m, 1H, H3'), 2.112.03 ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{H}{ }^{\prime}$ ) , 1.36 (d, $J=6.9 \mathrm{~Hz}, 1.5 \mathrm{H}, \mathrm{CHCH}_{3}$ L-Ala), 1.33 (d, $J=6.9 \mathrm{~Hz}, 1.5 \mathrm{H}$, $\mathrm{CHCH}_{3}$ L-Ala), 1.17 (t, $J=7.0 \mathrm{~Hz}, 1.5 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OCH}_{2} \mathrm{CH}_{3}$ ), $1.16(\mathrm{t}, J=7.0 \mathrm{~Hz}, 1.5 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OCH}_{2} \mathrm{CH}_{3}$ ).
${ }^{13} \mathbf{C}$ NMR ( $\left.\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{C}} 174.82\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.62(\mathrm{C}=\mathrm{O}), 174.58$ (C=O), 174.55 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}$ ), 157.34 (C6), 157.32 (C6), 153.86 (C2), 153.84 (C2), $150.48\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 150.44$ (C4), 150.22 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right)$, 140.49 (C8), 137.29 (C-Ar), 137.21 (C-Ar), 133.09 (d, ${ }^{2} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}$ ), 132.94 (d, $\left.{ }^{2} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 131.62(\mathrm{CH}-\mathrm{Ar}), 131.59$ (CH-Ar), 129.58 (CH-Ar), 129.34 (CH-Ar), 129.31 (CH-Ar), 129.28 (CH-Ar), 128.70 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 128.69\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0\right.$ $\mathrm{Hz}, \mathrm{CH}-\mathrm{Ar}), 126.18$ (CH-Ar), 121.02 (d, ${ }^{3} J=2.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), 120.49 (d, ${ }^{3} J=2.5 \mathrm{~Hz}, \mathrm{CH}-$

Ar), 120.58 (C5), $93.28\left(\mathrm{C}^{\prime}\right)$, $93.24\left(\mathrm{C}^{\prime}\right)$, $80.32\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.7 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $76.57(\mathrm{C} 2$ '), $68.86\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{C} 5\right.$ ') , $68.53\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.98\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.95$ $\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 61.57\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OCH}_{2} \mathrm{CH}_{3}\right), 51.76\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.65\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 35.37$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OCH}_{2} \mathrm{CH}_{3}\right)$, $35.30\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OCH}_{2} \mathrm{CH}_{3}\right)$, $35.08\left(\mathrm{C}^{\prime}\right), 34.85\left(\mathrm{C}{ }^{\prime}\right), 26.77$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OCH}_{2} \mathrm{CH}_{3}\right), 26.72\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OCH}_{2} \mathrm{CH}_{3}\right), 20.55\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$, $20.33\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}, \mathrm{CHCH}_{3}\right.$ L-Ala), $14.53\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{OCH}_{2} \mathrm{CH}_{3}\right)$.

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $F=1 \mathrm{~mL} / \mathrm{min}, \lambda=245 \mathrm{~nm}$, showed one peak with tR 15.99 min .
(ES+) $\mathbf{m} / \mathbf{z}$ found $669.3\left[\mathrm{M}+\mathrm{H}^{+}\right], 691.3\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{31} \mathrm{H}_{37} \mathrm{~N}_{6} \mathrm{O} 9 \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 668.63[\mathrm{M}]$.

## [32] 3'-Deoxyadenosine 5'-O-bis(benzyloxy-L-alaninyl) phosphate



Prepared according to general procedure $\mathbf{G}$ using 3'deoxyadenosine $(0.20 \mathrm{~g}, 0.80 \mathrm{mmol}),\left(\mathrm{OCH}_{3}\right)_{3} \mathrm{PO}(5$ mL ), $\mathrm{POCl}_{3}(75 \mu \mathrm{~L}, 0.80 \mathrm{mmol}), \mathrm{L}$-alanine O-benzyl ester tosylate salt ( $1.4 \mathrm{~g}, 4.0 \mathrm{mmol}$ ), $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and diisopropyl ethyl amine ( $1.4 \mathrm{~mL}, 8.0 \mathrm{mmol}$ ). The residue was purified by column chromatography (gradient elution of $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{CH}_{3} \mathrm{OH}=100 / 0$ to $93 / 7$ ) to give white foam ( $0.26 \mathrm{~g}, 49 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{P}} 13.9$.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.28(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 8.22(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 7.37-7.26(\mathrm{~m}, 10 \mathrm{H}$, Ph), 6.00 (d, $J=1.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), 5.15-5.05 (m, 4H, CH ${ }_{2} \mathrm{Ph}$ ), 4.74-4.70 (m, 1H, H2'), 4.63-4.56 (m, 1H, H4'), 4.24-4.18 (m, 1H, H5'), 4.11-4.05 (m, 1H, H5'), 3.97-3.87 (m, 2H, $\mathrm{CHCH}_{3}$ L-Ala x 2), 2.35-2.27 (m, 1H, H3'), 2.07-2.01 (m, 1H, H3'), 1.34-1.27 (m, 6H, $\mathrm{CHCH}_{3}$ L-Alax 2).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 175.40\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 175.36\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 157.36 (C6), 153.91 (C2), 150.25 (C4), 140.64 (C8), 137.33 (C-Ar), 137.29 (CAr), 129.58 (CH-Ar), 129.57 (CH-Ar), 129.33 (CH-Ar), 129.31 (CH-Ar), 129.29 (CH-Ar), $120.55(\mathrm{C} 5), 93.18\left(\mathrm{C} 1{ }^{\prime}\right), 80.67\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.4 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $76.59\left(\mathrm{C} 2^{\prime}\right), 67.90\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.47$ (d, ${ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.2 \mathrm{~Hz}, \mathrm{C} 5$ '), $51.14\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=1.7 \mathrm{~Hz}, C \mathrm{HCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.11\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=1.7 \mathrm{~Hz}\right.$, $\mathrm{CHCH}_{3}$ L-Ala), 35.08 ( C 3 '), 20.77 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.5 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ), 20.59 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.5$ $\mathrm{Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ).
(ES+) $\mathbf{m} / \mathbf{z}$ found $654.2\left[\mathrm{M}+\mathrm{H}^{+}\right], 676.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], 1329.5\left[2 \mathrm{M}+\mathrm{Na}^{+}\right] \mathrm{C}_{30} \mathrm{H}_{36} \mathrm{~N}_{7} \mathrm{O}_{8} \mathrm{P}$ required m/z 653.62 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed one peak with tR 13.87 min .

### 9.5.7 Synthesis of 2-modified 3'-deoxyadenosine analogues

[33] 2-Fluoro-3'-deoxyadenosine


2-Fluoro-3'-deoxyadenosine was synthesised according to standard procedure $\mathbf{E}$ using 2-fluoro-9-(2,3-anhydro- $\beta$-Dribofuranosyl) adenine [41b] ( $1.13 \mathrm{~g}, 4.18 \mathrm{mmol}$ ), $\mathrm{LiEt}_{3} \mathrm{BH}$ ( 1 M in THF, $12.54 \mathrm{mmol}, 12.54 \mathrm{~mL}$ followed by a second addition of $4.18 \mathrm{mmol}, 4.18 \mathrm{~mL}$ ) in DMSO ( 6.5 mL ) and THF ( 64 mL ). The title compound was obtained as a white solid $(0.87 \mathrm{~g}, 77 \%) .{ }^{225}\left(\right.$ Lit. mp: 259-260 $\left.{ }^{\circ} \mathrm{C}\right) .{ }^{270}$
${ }^{1} \mathbf{H}$ NMR ( 500 MHz, DMSO-d6) $\boldsymbol{\delta}_{\mathbf{H}} 8.34$ ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H} 8$ ), 7.80 (br s, 2H, NH2), 5.78 (d, $J=$ 2.2 Hz, 1H, H1'), 5.68 (br s, 1H, OH-2'), 5.01 (br s, 1H, OH-5'), 4.55-4.51 (m, 1H, H2'), 4.39-4.32 (m, 1H, H4'), 3.73-3.76 (m, 1H, H5'), 3.56-3.50 (m, 1H, H5'), 2.26-2.18 (m, 1H, H3'), 1.94-1.85 (m, 1H, H3').
${ }^{13}$ C NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}$, DMSO-d6) $\boldsymbol{\delta}_{\mathbf{C}} 158.51$ (d, ${ }^{1} J_{C-F}=202.7 \mathrm{~Hz}, \mathrm{C} 2$ ), 157.55 (d, ${ }^{3} J_{C-F}=$ $21.2 \mathrm{~Hz}, \mathrm{C} 6), 150.11\left(\mathrm{~d},{ }^{3} J_{C-F}=20.3 \mathrm{~Hz}, \mathrm{C} 4\right), 139.22\left(\mathrm{~d},{ }^{6} J_{C-F}=2.2 \mathrm{~Hz}, \mathrm{C} 8\right), 117.37(\mathrm{~d}$, ${ }^{4} J_{C-F}=4.1 \mathrm{~Hz}, \mathrm{C} 5$ ), 90.67 ( $\mathrm{C}^{\prime}$ '), $80.90(\mathrm{C} 4$ '), 74.73 (C2'), 62.35 ( C 5 '), 33.89 ( C 3 ').
${ }^{19}$ F NMR ( $\mathbf{4 7 0} \mathbf{~ M H z}$, DMSO-d6) $\boldsymbol{\delta}_{\mathrm{F}}-52.19$.
(ES+) m/z found $276.11\left[\mathrm{M}+\mathrm{Li}^{+}\right], 292.08\left[\mathrm{M}+\mathrm{Na}^{+}\right], 270.10\left[\mathrm{M}+\mathrm{H}^{+}\right] \mathrm{C}_{10} \mathrm{H}_{13} \mathrm{~N}_{5} \mathrm{O}_{3}$ required m/z 269.23 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 0$ to $75 / 25$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=280 \mathrm{~nm}$, showed one peak with tR 11.86 min .

## [41a] 2-Methoxy-9-(2,3-anhydro- $\boldsymbol{\beta}$-D-ribofuranosyl) adenine



The title compound was synthesised according to standard procedure $\mathbf{D}$ using 2-fluoroadenosine ( $2.0 \mathrm{~g}, 7.01 \mathrm{mmol}$ ), $\alpha-\mathrm{AIBBr}$ ( $4.10 \mathrm{~mL}, 28.05 \mathrm{mmol}$ ) and $\mathrm{H}_{2} \mathrm{O}(0.14 \mathrm{~mL}, 0.008 \mathrm{mmol})$ in $\mathrm{CH}_{3} \mathrm{CN}(50 \mathrm{~mL})$. The crude was then dissolved in $\mathrm{CH}_{3} \mathrm{OH}(50$ $\mathrm{mL})$ and stirred with Amberlite ( $2 \mathrm{x} \mathrm{OH}^{-}, 28 \mathrm{~mL}$ ) for 16 hours.

The title compound was obtained as a white powder ( $1.57 \mathrm{~g}, 84 \%$ ).
Melting point $175-176^{\circ} \mathrm{C}$.
${ }^{1} \mathbf{H}$ NMR ( 500 MHz, DMSO-d6) $\boldsymbol{\delta}_{\mathbf{H}} 8.12$ (s, 1H, H8), 7.28 (br s, 2H, NH2), 6.12 ( $\mathrm{s}, 1 \mathrm{H}$, H1'), 4.99 (br s, 1H, OH-5'), 4.43 (d, $J=3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2$ '), 4.21 (d, $J=2.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3^{\prime}$ ), $4.16\left(\mathrm{t}, J=6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4\right.$ '), $3.82\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3} \mathrm{O}\right), 3.60\left(\mathrm{dd}, J=11.5,6.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5^{\prime}\right), 3.50$ (dd, $J=11.5,5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}{ }^{\prime}$ ).
[41b] 2-Fluoro-9-(2,3-anhydro- $\boldsymbol{\beta}$-D-ribofuranosyl) adenine


The title compound was synthesised according to standard procedure $\mathbf{D}$ using 2 -fluoroadenosine ( $2.0 \mathrm{~g}, 7.01 \mathrm{mmol}$ ), $\alpha$ - AIBBr $(4.10 \mathrm{~mL}, 28.05 \mathrm{mmol})$ and $\mathrm{H}_{2} \mathrm{O}(0.14 \mathrm{~mL}, 0.008 \mathrm{mmol})$ in $\mathrm{CH}_{3} \mathrm{CN}$ $(50 \mathrm{~mL})$. The crude was then dissolved in $\mathrm{CH}_{3} \mathrm{OH}(50 \mathrm{~mL})$ and stirred with Amberlite ( $2 \times \mathrm{OH}^{-}, 28 \mathrm{~mL}$ ) for 1.5 hours. Purification by Biotage Isolera One, cartridge SNAP Ultra $50 \mathrm{~g}, 100 \mathrm{~mL} / \mathrm{min}, \mathrm{CH}_{3} \mathrm{OH} / E t O A c 1-10 \%$ $(10 \mathrm{CV}), 10 \%(2 \mathrm{CV})$ afforded the title compound as a white solid ( $1.01 \mathrm{~g}, 54 \%)$.
${ }^{1} \mathbf{H}$ NMR ( $\left.\mathbf{5 0 0} \mathbf{~ M H z}, ~ D M S O-d \boldsymbol{d}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.30$ ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H} 8$ ), 7.84 (br s, 2H, NH2), 6.11 ( $\mathrm{s}, 1 \mathrm{H}$, $\mathrm{H}^{\prime}$ ), $5.02\left(\mathrm{t}, J=5.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{OH}-5^{\prime}\right), 4.44(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2$ '), $4.20(\mathrm{~d}, J=3.0 \mathrm{~Hz}$, $\left.1 \mathrm{H}, \mathrm{H} 3^{\prime}\right), 4.18$ (t, $J=5.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4$ '), 3.59-3.48 (m, 2H, H5').
${ }^{19}$ F NMR ( $\mathbf{4 7 0} \mathbf{~ M H z}$, DMSO-d6) $\boldsymbol{\delta}_{\mathrm{F}}$-51.57.
[41c] 2-Ethoxy-9-(2,3-anhydro- $\beta$-D-ribofuranosyl) adenine


The title compound was synthesised according to standard procedure $\mathbf{D}$ using 2-fluoroadenosine ( $0.50 \mathrm{~g}, 1.75 \mathrm{mmol}$ ), $\alpha$ $\operatorname{AIBBr}(1.0 \mathrm{~mL}, 7.01 \mathrm{mmol})$ and $\mathrm{H}_{2} \mathrm{O}(0.03 \mathrm{~mL}, 0.002 \mathrm{mmol})$ in $\mathrm{CH}_{3} \mathrm{CN}(12 \mathrm{~mL})$. The crude was then dissolved in $\mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{OH}$ (12 mL ) and stirred with Amberlite ( $2 \times \mathrm{OH}^{-}, 7 \mathrm{~mL}$ ) for 16 hours. The title compound was obtained as a white solid ( $0.43 \mathrm{~g}, 83 \%$ ).
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}$, DMSO-d6) $\boldsymbol{\delta}_{\mathbf{H}} 8.16$ (s, 1H, H8), 7.28 (br s, 2H, NH2), 6.16 (s, 1H, H1'), 5.06 (br s, 1H, OH-5'), 4.48 (d, $J=2.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2^{\prime}$ ), 4.35-4.28 (m, 2H, OCH ${ }_{2} \mathrm{CH}_{3}$ ), 4.25 (d, $J=2.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3$ '), $4.21\left(\mathrm{t}, J=5.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}^{\prime}\right), 3.63$ (dd, $J=11.5,6 \mathrm{~Hz}, 1 \mathrm{H}$, H5'), 3.54 (dd, $J=11.5,6.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ '), $1.35\left(\mathrm{t}, J=7.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{3}\right)$.

## [41e] <br> 2-Fluoro-9-[2,3-anhydro-5-(2,5,5-trimethyl-1,3-dioxolan-4-on-2-yl)- $\beta$-Dribofuranosyl) adenine



Compound 40e was was synthesised according to standard procedure $\mathbf{D}$ using 2-fluoroadenosine ( $2.0 \mathrm{~g}, 7.01 \mathrm{mmol}$ ), $\alpha$ $\operatorname{AIBBr}(4.10 \mathrm{~mL}, 28.05 \mathrm{mmol})$ and $\mathrm{H}_{2} \mathrm{O}(0.14 \mathrm{~mL}, 0.008$ $\mathrm{mmol})$ in $\mathrm{CH}_{3} \mathrm{CN}(50 \mathrm{~mL})$. The crude was then dissolved in THF ( 40 mL ) and water ( 10 mL ) was added and stirred with Amberlite ( $2 \mathrm{x} \mathrm{OH}^{-}, 28 \mathrm{~mL}$ ) for 96 h , affording 2-fluoro-2', $3^{\prime}$-anhydroadenosine $\mathbf{4 0 b}$ ( 0.58 $\mathrm{g}, 31 \%)$ and the title compound $40 \mathrm{e}(0.47 \mathrm{~g}, 17 \%)$ as a white foam.
${ }^{1} \mathbf{H}$ NMR ( 500 MHz, DMSO-d6) $\boldsymbol{\delta}_{\mathbf{H}} 8.26$ (s, 1H, H8), 7.89 (br s, 2H, NH2), 6.18 ( $\mathrm{s}, 1 \mathrm{H}$, H1'), 4.52 (d, $\left.J=2.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2^{\prime}\right), 4.36$ (dd, $J=7.0,5.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}^{\prime}$ ), 4.30 (dd, $J=12.0$, $\left.5.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5^{\prime}\right), 4.21\left(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3^{\prime}\right), 4.15\left(\mathrm{dd}, J=11.5,7.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}{ }^{\prime}\right), 1.95$ (s, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), $1.40\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.38\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.
${ }^{19}$ F NMR ( $\mathbf{4 7 0} \mathbf{~ M H z}$, DMSO-d6) $\boldsymbol{\delta}_{\mathrm{F}}-51.57$.

## [42] 2', $\mathbf{5}^{\prime}$-Bis- $O$-trityloxy- $\mathbf{3}^{\prime}$-deoxyadenosine



3'-Deoxyadenosine ( $0.20 \mathrm{~g}, 0.80 \mathrm{mmol}$ ) was coevaporated twice with dry pyridine and dissolved in dry pyridine (3.6 $\mathrm{mL})$. Triphenylmethyl chloride ( $0.62 \mathrm{~g}, 2.24 \mathrm{mmol}$ ) was added and the solution was heated to $80^{\circ} \mathrm{C}$ overnight under argon. The reaction was quenched by adding $\mathrm{CH}_{3} \mathrm{OH}$ (3 mL ) at room temperature, and stirred for 30 min . The solution was then co-evaporated with toluene and diluted in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$. The organic solution was washed with a solution of saturated $\mathrm{NaHCO}_{3}(3$ x 15 mL ) and the combined aqueous layers were extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL})$. Combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under vacuum. The residue was purified by flash column chromatography on silica (eluent system $\mathrm{nHex} / \mathrm{EtOAc}=1 / 1$ containing $0.5 \%$ triethylamine), and afforded the title compound as a white foam ( $0.22 \mathrm{~g}, 38 \%$ ).
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.06(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 8.00(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 7.44-7.37(\mathrm{~m}, 12 \mathrm{H}$, Ar), 7.32-7.25 (m, 18H, Ar), 5.88 (d, $J=2.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Hl}$ '), 5.73 (br s, $1 \mathrm{H}, \mathrm{NH}_{2}$ ), 4.80-4.76 (m, 1H, H2'), 4.72-4.66 (m, 1H, H4'), 3.44 (dd, $J=10.5,3.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ '), 3.28 (dd, $J=$ $10.5,4.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ '), 2.26-2.12 (m, 2H, H3').
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 154.21$ (C6), 151.75 (C2), 147.64 (C4), 144.95 (C-Ar), 143.71 (C-Ar), 137.97 (C8), 129.13 (CH-Ar), 128.75 (CH-Ar), 128.01 (CH-Ar), 127.95 (CH-Ar), 127.28 (CH-Ar), 127.00 (CH-Ar), 121.39 (C5), 93.24 ( C 1 '), 87.06 (C-Tr), 80.55 (C4'), 76.05 ( C 2 '), 71.60 (C-Tr), 65.17 ( $\mathrm{C}^{\prime}$ '), 33.92 ( C 3 ').

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $80 / 20$ to $0 / 100$ in $50 \mathrm{~min}, \mathrm{~F}=$ $1 \mathrm{~mL} / \mathrm{min}, \lambda=245 \mathrm{~nm}$, showed one peak with tR 30.27 min .
(ES+ + ) $\mathbf{m} / \mathbf{z}$ found $736.3\left[\mathrm{M}+\mathrm{H}^{+}\right], 758.3\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{48} \mathrm{H}_{41} \mathrm{~N}_{5} \mathrm{O}_{3}$ required $\mathrm{m} / \mathrm{z} 735.87[\mathrm{M}]$.
[44] 5'-O-Acetyl-2', 3'-O-isopropylidene-adenosine


2',3'-O-Isopropylidene-adenosine ( $0.20 \mathrm{~g}, 0.65 \mathrm{mmol}$ ) was dissolved in $\mathrm{CH}_{3} \mathrm{CN}(15 \mathrm{~mL})$. To this stirring solution, DMAP $(0.032 \mathrm{~g}, 0.26 \mathrm{mmol})$, acetic anhydride $(0.36 \mathrm{~mL}, 3.9 \mathrm{mmol})$ and triethylamine $(0.54 \mathrm{~mL}, 3.9 \mathrm{mmol})$ were added, and the reaction was stirred at $70{ }^{\circ} \mathrm{C}$ for 4 hours. The mixture was evaporated, redissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$ and washed with aqueous $\mathrm{HCl}(0.5$ $\mathrm{N}, 15 \mathrm{~mL}$ ), aqueous $\mathrm{NaHCO}_{3}$ saturated solution ( 15 mL ) and brine ( 15 mL ). The organic layers were combined, dried over $\mathrm{MgSO}_{4}$ and evaporated. The crude was purified by silica gel column chromatography (eluent system $\mathrm{EtOAc} / \mathrm{nHex}=50-100 \%$ ) to yield the title compound as a sticky solid ( $0.038 \mathrm{~g}, 15 \%$ ). ${ }^{368}$
${ }^{1} \mathbf{H}-\mathbf{N M R}\left(\mathbf{C D C l}_{3}, \mathbf{5 0 0} \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.91(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 8.16(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 6.14(\mathrm{~d}, J=2.3 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{H} 1^{\prime}$ ), 5.41 (dd, $J=2.3 \mathrm{~Hz}, 6.3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2$ '), 4.96 (dd, $J=3.4 \mathrm{~Hz}, 6.3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3$ '), 4.50-4.46 (m, 1H, H4'), 4.29 (dd, $J=4.8 \mathrm{~Hz}, 12.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ '), 4.22 (dd, $J=6.5 \mathrm{~Hz}, 12.1$ $\mathrm{Hz}, 1 \mathrm{H}, \mathrm{H} 5$ '), $2.29\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{N}\left(\mathrm{COCH}_{3}\right)_{2}\right), 1.91\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OCOCH}_{3}\right), 1.58\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.35(\mathrm{~s}$, $3 \mathrm{H}, \mathrm{CH}_{3}$ ).
[46] $N^{9}$-tert-butylcarbonate-[bis- $N^{4}, N^{4}$-(tert-butylcarbonate)]-adenine


Adenine ( $0.30 \mathrm{~g}, 2.22 \mathrm{mmol}$ ) was dissolved in ahydrous THF (10 $\mathrm{mL})$ and DMAP $(0.027 \mathrm{mg}, 0.22 \mathrm{mmol})$ and $\mathrm{Boc}_{2} \mathrm{O}(1.94 \mathrm{~g}, 8.88$ mmol) were added. The mixture was stirred at rt for 8 h and evaporated. The mixture was extracted with $n \mathrm{Hex} /$ water to yield the title compound as a clear oil $(0.97 \mathrm{~g}, 81 \%) .{ }^{470}$
${ }^{1} \mathbf{H}-\mathbf{N M R}\left(\mathbf{C D C l}_{\mathbf{3}}, \mathbf{5 0 0} \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.94$ ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H} 8$ ), 8.43 (s, 1H, H2), 1.64 ( $\mathrm{s}, 9 \mathrm{H}, t \mathrm{Bu}$ ), 1.37 (s, $18 \mathrm{~h}, t \mathrm{Bu} \times 2$ ).
[48] 2-Methoxy-3'-deoxyadenosine


The title compound was synthesised according to standard procedure $\mathbf{E}$ using 2-methoxy-9-(2,3-anhydro- $\beta$-D-ribofuranosyl) adenine [41a] ( $0.76 \mathrm{~g}, 2.84 \mathrm{mmol}$ ), $\mathrm{LiEt}_{3} \mathrm{BH}$ ( 1 M in THF, 8.53 $\mathrm{mL}, 8.53 \mathrm{mmol}$, followed by a second addition of $2.84 \mathrm{~mL}, 2.84$ mmol ) in DMSO ( 4.3 mL ) and THF ( 43 mL ). The residue was chromatographed on silica gel (3-17\% $\mathrm{CH}_{3} \mathrm{OH}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give 3'-deoxy-2-methoxyadenosine as a white powder $(0.65 \mathrm{~g}, 81 \%)$.

Melting point $179-180^{\circ} \mathrm{C}$.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.20(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 5.90(\mathrm{~d}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), 4.754.71 (m, 1H, H2'), 4.54-4.48 (m, 1H, H4'), 3.91 (dd, $\left.J=12.3,2.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5^{\prime}\right), 3.69$ (dd, $J=12.3,4.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ '), 3.37 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{OCH}_{3}$ ), 2.43-2.35 (m, 1H, H3'), 2.08-2.02 (m, 1H, H3').
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 163.68$ (C2), 158.12 (C6), 151.94 (C4), 139.71 (C8), 116.64 (C5), $93.36\left(\mathrm{C}^{\prime}\right), 82.53\left(\mathrm{C}^{\prime}\right), 76.59\left(\mathrm{C}^{\prime}\right), 64.24\left(\mathrm{C}^{\prime}\right), 55.29\left(\mathrm{OCH}_{3}\right), 34.81$ (C3').
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 0$ to $75 / 25$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=280 \mathrm{~nm}$, showed one peak with tR 8.84 min .
(ES+) $\mathbf{m} / \mathbf{z}$ found $282.3\left[\mathrm{M}+\mathrm{H}^{+}\right] \mathrm{C}_{11} \mathrm{H}_{15} \mathrm{~N}_{5} \mathrm{O}_{4}$ required $\mathrm{m} / \mathrm{z} 281.27$ [M].

### 9.5.8 Synthesis of 2-fluoro-3'-deoxyadenosine ProTides




Prepared according to general procedure $\mathbf{F}_{\mathbf{2}}$ using 2-fluoro-3'-deoxyadenosine [33] ( $0.05 \mathrm{~g}, 0.18 \mathrm{mmol}$ ), $N$-methylimidazole ( $74 \mu \mathrm{~L}, 0.93 \mathrm{mmol}$ ) and 1 -naphth-1-yl(benzyloxy-L-alaninyl)
phosphorochloridate [17a] ( $0.20 \mathrm{~g}, 0.56 \mathrm{mmol})$ in THF ( 2.4 mL ). Purification by Biotage Isolera One, cartridge SNAP KP-Sil 50 g , 100 $\mathrm{mL} / \mathrm{min}, \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 1-10 \%(10 \mathrm{CV}), 10 \%(5 \mathrm{CV})$ and preparative TLC $(1000 \mu \mathrm{M}$, eluent system $\left.\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 5 / 95\right)$ afforded the title compound as a white solid $(0.017 \mathrm{~g}$, 15\%).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 4.33,4.08$.
${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathbf{C D}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.17$ ( $\left.\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 8\right), 8.14(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.14-8.09(\mathrm{~m}, 1 \mathrm{H}$, Ar), 7.89-7.85 (m, 1H, Ar), 7.70-7.66 (m, 1H, Ar), 7.54-7.42 (m, 4H, Ar), 7.40-7.24 (m, $5 \mathrm{H}, \mathrm{Ar}), 5.89\left(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 5.88\left(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 5.08-5.01(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{Ph}$ ), 4.70-4.60 (m, 2H, H2', C4'), 4.46-4.39 (m, 1H, C5'), 4.32-4.24 (m, 1H, C5'), 4.09-3.97 (m, 1H, $\mathrm{CHCH}_{3}$ L-Ala), 2.36-2.25 (m, 1H, H3'), 2.06-1.98 (m, 1H, H3'), 1.321.25 (m, 3H, $\mathrm{CHCH}_{3}$ L-Ala).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 175.54(\mathrm{CO}), 175.22(\mathrm{CO}), 161.02\left(\mathrm{~d},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=207.3 \mathrm{~Hz}\right.$, C2), $160.89\left(\mathrm{~d},{ }^{1} J_{C-F}=207.3 \mathrm{~Hz}, \mathrm{C} 2\right), 158.45\left(\mathrm{~d},{ }^{3} J_{C-F}=18.2 \mathrm{~Hz}, \mathrm{C} 6\right), 158.23\left(\mathrm{~d},{ }^{3} J_{C-F}=\right.$ $18.2 \mathrm{~Hz}, \mathrm{C} 6), 150.63$ (d, $\left.{ }^{3} J_{C-F}=18.4 \mathrm{~Hz}, \mathrm{C} 4\right), 140.67$ (C8), 136.26 (C-Ar), 131.62, 131.54,
129.56 (CH-Ar), 129.52 (CH-Ar), 129.37 (CH-Ar), 129.31 (CH-Ar), 129.26 (CH-Ar), 128.87 (CH-Ar), 128.81 (CH-Ar), 128.29 (CH-Ar), 128.02 (CH-Ar), 127.79 (CH-Ar), 127.76 (CH-Ar), 127.51 (CH-Ar), 127.49 (CH-Ar), 127.47 (CH-Ar), 126.47 (CH-Ar), 126.33 (C-Ar), 126.27 (C-Ar), 125.97 (CH-Ar), 122.78 (CH-Ar), 122.74 (CH-Ar), 122.64 (CH-Ar), 122.62 (CH-Ar), 116.35 (d, $\left.{ }^{4} J_{C-F}=3.0 \mathrm{~Hz}, \mathrm{C} 5\right), 116.15$ ( $\left.\mathrm{d},{ }^{4} J_{C-F}=3.0 \mathrm{~Hz}, \mathrm{C} 5\right)$, $93.25\left(\mathrm{C}^{\prime}\right), 93.20\left(\mathrm{C} 1^{\prime}\right), 80.41\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $80.33\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), 76.43 (C2'), 76.35 (C2'), $68.84\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.5 \mathrm{~Hz}, \mathrm{C} 5^{\prime}\right), 68.45\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.5 \mathrm{~Hz}, \mathrm{C} 5^{\prime}\right)$, $67.92\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.92\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 51.75\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.52\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 34.97(\mathrm{C} 3)$ ), $34.74\left(\mathrm{C} 3\right.$ '), $20.42\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.7 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 20.20\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.7 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\right.$ Ala).
${ }^{19}$ F NMR (470 MHz, CD $\mathbf{D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{F}}-53.14,-53.22$.
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $F=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed two peaks of the diastereoisomers with tR 17.09 min . and tR 17.34 min .
(ES + ) $\mathbf{m} / \mathbf{z}$ found $637.2\left[\mathrm{M}+\mathrm{H}^{+}\right], \mathrm{C}_{30} \mathrm{H}_{30} \mathrm{FN}_{6} \mathrm{O}_{7} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 636.57$ [M].
[49b] 2-Fluoro-3'-deoxyadenosine-5'-O-phenyl-(benzyloxy-L-alaninyl) phosphate


Prepared according to general procedure $\mathbf{F}_{2}$ using 2-fluoro-3'-deoxyadenosine [33] ( $0.05 \mathrm{~g}, 0.18 \mathrm{mmol}$ ), N methylimidazole (74 $\mu \mathrm{L}, \quad 0.93 \mathrm{mmol})$ and phenyl(benzyloxy-L-alaninyl) phosphorochloridate [17s] $(0.20 \mathrm{~g}, 0.56 \mathrm{mmol})$ in THF ( 2.4 mL ). Purification by Biotage Isolera One (cartridge SNAP Ultra 50 g , 100 $\mathrm{mL} / \mathrm{min}, \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 1-10 \% 10 \mathrm{CV}, 10 \% 3 \mathrm{CV}$ ) and preparative TLC ( $1000 \mu \mathrm{M}$, eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 5 / 95$ ) afforded the title compound as a white solid ( $0.005 \mathrm{~g}, 7 \%$ ).
${ }^{31} \mathbf{P} \mathbf{N M R}\left(\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 3.95, 3.67.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.19$ ( $\left.\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 8\right), 8.16$ ( $\left.\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 8\right), 7.36-7.27$ (m, 7H, Ar), 7.22-7.13 (m, 3H, Ar), 5.91 (d, $J=1.5 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1$ '), $5.89(\mathrm{~d}, J=1.7 \mathrm{~Hz}, 0.5 \mathrm{H}$, H1'), 5.15-5.06 (m, 2H, CH ${ }_{2} \mathrm{Ph}$ ), 4.73-4.58 (m, 2H, H2', H4'), 4.42-4.34 (m, 1H, H5'), 4.02-3.90 (m, 1H, H5'), 3.89-3.77 (m, 1H, CHCH3 L-Ala), 3.27-3.24 (m, 1H, H3'), 2.08$2.00\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 3\right.$ ') , 1.33 (d, $\left.J=7.1 \mathrm{~Hz}, 1.5 \mathrm{H}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 1.29(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 1.5 \mathrm{H}$, $\mathrm{CHCH}_{3}$ L-Ala).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 175.85\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.63\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}), 160.58\left(\mathrm{~d},{ }^{l} J_{C-F}=207.5 \mathrm{~Hz}, \mathrm{C} 2\right), 160.53\left(\mathrm{~d},{ }^{l} J_{C-F}=207.5 \mathrm{~Hz}, \mathrm{C} 2\right), 159.06(\mathrm{~d}$,
$\left.{ }^{3} J_{C-F}=18.7 \mathrm{~Hz}, \mathrm{C} 6\right), 159.05\left(\mathrm{~d},{ }^{3} J_{C-F}=17.5 \mathrm{~Hz}, \mathrm{C} 6\right), 152.11\left(\mathrm{~d},{ }^{2} J_{C-P}=8.7 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right)$, $152.08\left(\mathrm{~d},{ }^{2} J_{C-P}=8.7 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 151.58\left(\mathrm{~d},{ }^{3} J_{C-F}=19.7 \mathrm{~Hz}, \mathrm{C} 4\right), 151.56\left(\mathrm{~d},{ }^{3} J_{C-F}=19.5\right.$ Hz, C4), 140.63 (C8), 137.28 (C-Ar), 137.21 (C-Ar), 130.78 (CH-Ar), 130.75 (CH-Ar), 129.58 (CH-Ar), 129.38 (CH-Ar), 129.34 (CH-Ar), 129.32 (CH-Ar), 129.28 (CH-Ar), 128.3 (CH-Ar), 128.02 (CH-Ar), 121.16 (CH-Ar), 121.18 (CH-Ar), 121.47 (CH-Ar), 121.51 ( $\mathrm{CH}-\mathrm{Ar}$ ), 121.42 ( $\mathrm{CH}-\mathrm{Ar}$ ), $121.39(\mathrm{CH}-\mathrm{Ar}), 121.36(\mathrm{CH}-\mathrm{Ar}), 118.75\left(\mathrm{~d},{ }^{4} J_{C-F}=3.7\right.$ $\mathrm{Hz}, \mathrm{C} 5), 118.72\left(\mathrm{~d},{ }^{4} J_{\mathrm{C}-\mathrm{F}}=3.7 \mathrm{~Hz}, \mathrm{C} 5\right), 93.25\left(\mathrm{C} 1^{\prime}\right), 93.18\left(\mathrm{C} 1^{\prime}\right), 80.48\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.3 \mathrm{~Hz}\right.$, C4'), 80.46 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.1 \mathrm{~Hz}, \mathrm{C} 4$ '), 76.51 (C2'), 76.49 ( C 2 '), $68.54\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.2 \mathrm{~Hz}\right.$, C 5 '), $68.18\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.6 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.94\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.91\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 51.71\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$, $51.56\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 34.85\left(\mathrm{C} 3\right.$ '), $34.64\left(\mathrm{C} 3\right.$ '), $20.42\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.1 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$, 20.25 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{CHCH}_{3}$ L-Ala).
${ }^{19}$ F NMR ( $\mathbf{4 7 0} \mathbf{M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{F}}-53.17,-53.23$.
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \mathrm{l}=280 \mathrm{~nm}$, showed two peaks of the diastereoisomers with $\mathrm{t}_{\mathrm{R}}$ 14.98 min . and $\mathrm{t}_{\mathrm{R}} 15.12 \mathrm{~min}$.
(ES + ) $\mathbf{m} / \mathbf{z}$ found $587.1\left[\mathrm{M}+\mathrm{H}^{+}\right], \mathrm{C}_{26} \mathrm{H}_{28} \mathrm{FN}_{6} \mathrm{O}_{7} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 586.17$ [M].
The two diastereoisomers were separated via preparative HPLC (isocratic eluent system: $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{OH} 45 \%$ in $30 \mathrm{~min}, 10 \mathrm{mg}$ sample).
Slow eluting isomer (SE)
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 3.67$.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.19$ ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H} 8$ ), 7.36-7.27 (m, 7H, Ar), 7.22-7.13 (m, $3 \mathrm{H}, \mathrm{Ar}), 5.91\left(\mathrm{~d}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 5.09\left(\mathrm{AB} \mathrm{q}, J_{\mathrm{AB}}=12.5 \mathrm{~Hz}, \Delta \delta_{\mathrm{AB}}=0.01,2 \mathrm{H}\right.$, $\mathrm{CH}_{2} \mathrm{Ph}$ ), 4.73-4.70 (m, 1H, H2'), 4.66-4.61 (m, 1H, H4'), 4.42-4.37 (m, 1H, H5'), 4.264.20 (m, 1H, H5'), 3.97-3.89 (m, 1H, $\mathrm{CHCH}_{3}$ L-Ala), 2.34-2.27 (m, 1H, H3'), 2.07-2.02 (m, 1H, H3'), 1.30-1.27 (m, 3H, $\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ).
${ }^{19}$ F NMR ( $\mathbf{4 7 0} \mathbf{~ M H z}$, CD $_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{F}}-53.23$.

## Fast eluting isomer ( FE )

${ }^{31} \mathbf{P}$ NMR ( $\left.\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathrm{P}} 3.95$.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.17$ ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H} 8$ ), 7.36-7.27 (m, 7H, Ar), 7.24-7.12 (m, $3 \mathrm{H}, \mathrm{Ar}), 5.89\left(\mathrm{~d}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H1}{ }^{\prime}\right), 4.95\left(\mathrm{AB} \mathrm{q}, J=12.0 \mathrm{~Hz}, \Delta \delta_{\mathrm{AB}}=0.02,2 \mathrm{H}, \mathrm{CH} H_{2} \mathrm{Ph}\right)$, 4.69-4.66 (m, 1H, H2'), 4.64-4.59 (m, 1H, H4'), 4.39-4.34 (m, 1H, H5'), 4.25-4.19 (m, 1H, H5'), 4.02-3.95 (m, 1H, CHCH ${ }_{3}$ L-Ala), 2.34-2.27 (m, 1H, H3'), 2.05-2.00 (m, 1H, H3'), 1.34-1.31 (m, 3H, $\mathrm{CHCH}_{3}$ L-Ala).
${ }^{19}$ F NMR ( $\mathbf{4 7 0} \mathbf{M H z}$, CD $_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{F}}-53.17$. phosphate


Prepared according to general procedure $\mathbf{F}_{2}$ using 2-fluoro-3'-deoxyadenosine [33] ( $0.05 \mathrm{~g}, 0.18 \mathrm{mmol}$ ), N methylimidazole ( $74 \mu \mathrm{~L}, 0.93 \mathrm{mmol}$ ) and naphth-1-yl(pentyl-1-oxy-L-leucinyl) phosphorochloridate [17f] ( $0.25 \mathrm{mg}, 0.56 \mathrm{mmol}$ ) in THF ( 2.2 mL ). Purification by Biotage Isolera One, cartridge SNAP Ultra $50 \mathrm{~g}, 100 \mathrm{~mL} / \mathrm{min}, \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 1-10 \%$ (10 $\mathrm{CV}), 10 \%(3 \mathrm{CV})$ and preparative TLC ( $1000 \mu \mathrm{M}$, eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 5 / 95$ ) afforded the title compound as a white solid $(0.018 \mathrm{~g}, 16 \%)$.
${ }^{31} \mathbf{P} \mathbf{N M R}\left(\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}} 4.60,4.35$.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.23(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.20(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.18-8.12(\mathrm{~m}, 1 \mathrm{H}$, H8 Nap), 7.92-7.86 (m, 1H, H5 Nap), 7.73-7.68 (m, 1H, H4 Nap), 7.57-7.46 (m, 3H, H2, H6, H7 Nap), 7.42-7.36 (m, 1H, H3 Nap), 5.93-5.91 (m, 1H, H1'), 4.74-4.62 (m, 2H, H2', H4'), 4.55-4.50 (m, 0.5H, H5'), 4.49-4.44 (m, 0.5H, H5'), 4.43-4.37 (m, 0.5H, H5'), 4.364.31 (m, 0.5H, H5'), 4.02-3.86 (m, 3H, CHCH ${ }_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$ Pen), 2.43-2.29 (m, 1H, H3'), 2.12-2.04 (m, 1H, H3'), 1.67-1.20 (m, 9H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n$-Pen, $\quad \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \quad$ L-Leu), $0.89-0.67 \quad(\mathrm{~m}, \quad 9 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 175.03\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.93\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}), 161.45\left(\mathrm{~d},{ }^{l} J_{\mathrm{C}-\mathrm{F}}=205.5 \mathrm{~Hz}, \mathrm{C} 2\right), 160.39\left(\mathrm{~d},{ }^{l} J_{\mathrm{C}-F}=205.5 \mathrm{~Hz}, \mathrm{C} 2\right), 158.33$ (C6), 151.60 (C4), 147.92 (C-Ar), 140.69 (C8), 136.30 (C-Ar), 128.88 (CH-Ar), 128.83 (CH$\mathrm{Ar}), 127.80$ (CH-Ar), 127.76 (CH-Ar), 127.49 (CH-Ar), 127.46 (CH-Ar), 126.48 (CH-Ar), 126.45 (CH-Ar), 126.02 (CH-Ar), 125.91 (CH-Ar), 123.03 (C-Ar), 122.81 (CH-Ar), 122.69 (CH-Ar), 116.39 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.9 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), 116.28 (C5), 116.26 (C5), 115.97 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.9 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 93.29\left(\mathrm{C} 1\right.$ '), $93.23\left(\mathrm{C} 1\right.$ '), $80.45\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.0 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $80.38(\mathrm{~d}$, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.0 \mathrm{~Hz}, \mathrm{C} 4$ '), $76.45(\mathrm{C} 2 '), 76.41(\mathrm{C} 2 '), 68.99\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $68.78\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}\right.$ $\left.=5.4 \mathrm{~Hz}, \mathrm{C}{ }^{\prime}\right)$, $66.31\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen $), 66.29\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen $)$, $54.78\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu $), 54.66\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $44.16\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.25\right.$ $\mathrm{Hz}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), $43.84\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.3 \mathrm{~Hz}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 35.09 ( $\mathrm{C} 3^{\prime}$ ), $34.79\left(\mathrm{C}^{\prime}\right)$, $29.31\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $29.12\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$ Pen), $25.65\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \text { L-Leu }\right)^{2} 25.41 \quad\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \quad\right.$ L-Leu), 23.33 $\left(\mathrm{H}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $23.11\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}-\mathrm{Leu}\right)$, $23.00\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}\right.$ Leu), $21.95\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu $), 21.68 \quad\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \quad\right.$ L-Leu), 14.29 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen).
${ }^{19}$ F NMR (470 MHz, CD $\left.\mathbf{3}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathrm{F}}-53.15,-53.20$.
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, 1=254 \mathrm{~nm}$, showed one peak of the overlapping diastereoisomers with tR 21.95 min .
(ES+ $+\mathbf{m} / \mathbf{z}$ found $659.3\left[\mathrm{M}+\mathrm{H}^{+}\right], \mathrm{C}_{31} \mathrm{H}_{40} \mathrm{FN}_{6} \mathrm{O}_{7} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 658.66[\mathrm{M}]$.
[49d] 2-Fluoro-3'-deoxyadenosine-5'-O-phenyl-(hexyl-1-oxy-L-alaninyl) phosphate


Prepared according to the general procedure $\mathbf{F} 2$ using 2-fluoro-3'-deoxyadenosine [33] ( $0.05 \mathrm{~g}, 0.18 \mathrm{mmol}$ ), $N$-methylimidazole (74 $\mu \mathrm{L}, \quad 0.93 \mathrm{mmol}$ ) and phenyl(hex-1-yloxy-L-alaninyl) phosphorochloridate [17u] ( $0.19 \mathrm{~g}, 0.56 \mathrm{mmol}$ ). Purification by column chromatography (eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CHCl}_{3} 0 / 100$ to $6 / 94$ ) and preparative TLC ( 500 $\mu \mathrm{M}$, eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 5 / 95$ ) afforded the title compound as a white solid ( 0.03 g, 28\%).
${ }^{\mathbf{1}} \mathbf{P}$ NMR ( $\left.\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 3.91, 3.73.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.21(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.20(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 7.37-7.29$ (m, $2 \mathrm{H}, \mathrm{Ar}), ~ 7.26-7.13$ (m, 3H, Ar), 5.94-5.91 (m, 1H, H1'), 4.76-4.64 (m, 2H, H2', H4'), 4.49-4.44 (m, 0.5H, H5'), 4.43-4.37 (m, 0.5H, H5'), 4.33-4.26 (m, 1H, H5'), 4.11-3.99 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex), 3.97-3.83 (m, 1H, CHCH ${ }_{3}$ L-Ala), 2.41-2.32 (m, 1H, H3'), 2.13-2.06 (m, 1H, H3'), 1.62-1.52 (m, 2H, CH2 $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex), 1.37-1.23 (m, 9H, $\mathrm{CHCH}_{3}$ L-Ala, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex), 0.92-0.85 (m, 3 H , $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$ - Hex ).
${ }^{13} \mathbf{C}$ NMR ( $\left.\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{C}} 175.15\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.96\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), $160.59\left(\mathrm{~d},{ }^{1} J_{C-F}=207.5 \mathrm{~Hz}, \mathrm{C} 2\right), 160.56\left(\mathrm{~d},{ }^{1} J_{C-F}=207.5 \mathrm{~Hz}, \mathrm{C} 2\right), 159.09(\mathrm{~d}$, $\left.{ }^{3} J_{C-F}=21.2 \mathrm{~Hz}, \mathrm{C} 6\right), 159.08\left(\mathrm{~d},{ }^{3} J_{C-F}=20.0 \mathrm{~Hz}, \mathrm{C} 6\right), 152.16\left(\mathrm{~d},{ }^{2} J_{C-P}=7.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right)$, $152.14\left(\mathrm{~d},{ }^{2} J_{C-P}=6.3 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 151.71\left(\mathrm{~d},{ }^{3} J_{C-F}=20.0 \mathrm{~Hz}, \mathrm{C} 4\right), 151.67\left(\mathrm{~d},{ }^{3} J_{C-F}=20.0\right.$ $\mathrm{Hz}, \mathrm{C} 4), 140.70\left(\mathrm{~d},{ }^{5} J_{C-F}=2.5 \mathrm{~Hz}, \mathrm{C} 8\right), 140.68\left(\mathrm{~d},{ }^{5} J_{C-F}=2.5 \mathrm{~Hz}, \mathrm{C} 8\right), 130.77(\mathrm{CH}-\mathrm{Ar})$, 130.74 (CH-Ar), 126.16 (CH-Ar), 126.24 (CH-Ar), 121.48 (CH-Ar), 121.44 (CH-Ar), 121.41 (CH-Ar), 121.37 (CH-Ar), $118.80\left(\mathrm{~d},{ }^{4} J_{C-F}=3.7 \mathrm{~Hz}, \mathrm{C} 5\right), 118.77\left(\mathrm{~d},{ }^{4} J_{C-F}=3.7 \mathrm{~Hz}\right.$, C5), 93.37 ( $\mathrm{Cl}^{\prime}$ ), $93.25\left(\mathrm{C} 1^{\prime}\right), 80.52\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $80.45\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.1 \mathrm{~Hz}, \mathrm{C} 4^{\prime}\right)$, 76.52 ( $\mathrm{C}^{\prime}$ ) , $76.49\left(\mathrm{C} 2^{\prime}\right), 68.69\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.4 \mathrm{~Hz}, \mathrm{C} 5^{\prime}\right), 68.30\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=4.9 \mathrm{~Hz}, \mathrm{C} 5^{\prime}\right)$, $66.46\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Hex $)$, $51.68\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$, $51.57\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$, 35.02 ( $\left.\mathrm{C} 3^{\prime}\right), \quad 34.80 \quad(\mathrm{C} 3 '), \quad 32.58 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex), $\quad 29.65$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex), $26.61\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex), 23.59
$\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Hex), $20.60\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.1 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}\right.$-Ala), $20.43\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}\right.$ $=7.5 \mathrm{~Hz}, \mathrm{CHCH} 3$ L-Ala $), 14.35\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Hex $)$.
${ }^{19}$ F NMR (470 MHz, CD3OD) $\boldsymbol{\delta}_{\mathrm{F}}-53.15,-53.20$.
(ES+ $+\mathbf{m} / \mathbf{z}$, found: $581.2\left(\mathrm{M}+\mathrm{H}^{+}\right), \mathrm{C}_{25} \mathrm{H}_{34} \mathrm{FN}_{6} \mathrm{O}_{7} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 580.55(\mathrm{M})$.
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \mathrm{l}=254 \mathrm{~nm}$, showed two peaks of the diastereoisomers with tR 17.83 min . and tR 18.02 min .

### 9.5.9 Synthesis of 2-methoxy-3'-deoxyadenosine ProTides

## [50a] 2-Methoxy-3'-deoxyadenosine-5'-O-naphth-1-yl(benzyloxy-L-alaninyl)phosphate



Prepared according to the general procedure A using 2-methoxy-3'-deoxyadenosine [48] ( $0.07 \mathrm{~g}, 0.25 \mathrm{mmol}$ ), $N$-methylimidazole (99 $\mu \mathrm{L}, \quad 1.24 \mathrm{mmol}) \quad$ and naphthyl(benzyloxy-L-alaninyl) phosphorochloridate [17a] ( $0.30 \mathrm{~g}, 0.75 \mathrm{mmol}$ ). Purification by column chromatography (eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CHCl}_{3} 0 / 100$ to 6/94) and preparative TLC ( $2000 \mu \mathrm{M}$, eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 5 / 95$ ) afforded the title compound as a white solid $(0.096 \mathrm{~g}, 60 \%)$.
${ }^{\mathbf{3 1}} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 4.38, 4.08.
${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.31-8.23(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ar}), 8.14-8.06(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ar}), 8.05(\mathrm{~s}$, $0.5 \mathrm{H}, \mathrm{H} 8$ ), 8.02 (s, 0.5H, H8), 7.90-7.78 (m, 3H, Ar), 7.70-7.20 (m, 6H, Ar), 5.91-5.88 (m, $1 \mathrm{H}, \mathrm{H} 1$ '), 5.10-4.99 (m, 2H, CH2 Ph ), 4.75-4.68 (m, 1H, H4'), 4.66-4.57 (m, 1H, H2'), 4.46-4.37 (m, 1H, H5'), 4.33-4.25 (m, 1H, H5'), 4.16-3.98 (m, 1H, CHCH ${ }_{3}$ L-Ala), 3.92 ( s , $1.5 \mathrm{H}, \mathrm{OCH}_{3}$ ), $3.91\left(\mathrm{~s}, 1.5 \mathrm{H}, \mathrm{OCH}_{3}\right), 2.43-2.34(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 3$ '), 2.06-1.99 (m, 1H, H3'), 1.301.21 (m, 3H, $\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C H}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 174.83\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.60\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 163.70 (C2), 158.10 (C6), 151.95 (C4), 147.95 ( $\mathrm{d},{ }^{3} \mathrm{~J}_{\mathrm{C}-\mathrm{P}}=7.5, \mathrm{C}-\mathrm{Ar}$ ), 147.91, ( $\left.\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5, \mathrm{C}-\mathrm{Ar}\right), 139.39$ (C8), 139.37 (C8), 137.12 (C-Ar), 137.17 (C-Ar), 136.22 (CAr), 129.57 (CH-Ar), 129.54 (CH-Ar), 129.48 (CH-Ar), 129.32 (CH-Ar), 129.27 (CH-Ar), 129.12 (CH-Ar), 129.24 (CH-Ar), 128.89 (CH-Ar), 128.83 (CH-Ar), 127.85 (d, ${ }^{2} J_{\mathrm{C}-\mathrm{P}}=6.2$ Hz, C-Ar), 127.86 (CH-Ar), 127.76 (CH-Ar), 127.51 (CH-Ar), 127.48 (CH-Ar), 126.49 (CH-Ar), 126.00 (CH-Ar), 125.97 (CH-Ar), 122.73 (CH-Ar), 122.63 (CH-Ar), 116.86
(C5), 116.72 (C5), 116.29 ( $\left.\mathrm{d}^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 116.22\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right)$,
 ${ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{C} 5$ ' $), 68.16\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=8.2 \mathrm{~Hz}, \mathrm{C} 5\right.$ ' $), 67.95\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 55.28\left(\mathrm{OCH}_{3}\right), 55.32$ $\left(\mathrm{OCH}_{3}\right), 51.79\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $51.71\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 35.40(\mathrm{C}-3$ '), $35.12(\mathrm{C} 3$ '), 20.49 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.7 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 20.35\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.7 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$.

MS (ES+ $\mathbf{~ m / z}$ found $649.2\left[\mathrm{M}+\mathrm{H}^{+}\right] \mathrm{C}_{31} \mathrm{H}_{33} \mathrm{~N}_{6} \mathrm{O}_{8} \mathrm{P}$ required m/z 648.21 [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $F=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed two peaks with $\mathrm{t}_{\mathrm{R}} 16.22 \mathrm{~min}$ and $\mathbf{t}_{\mathrm{R}} 16.48$ min.
[50b] 2-Methoxy-3'-deoxyadenosine-5'-O-phenyl-(benzyloxy-L-alaninyl)- phosphate


Prepared according to the general procedure F2 using 2-methoxy-3'-deoxyadenosine [48] (0.07 g, 0.25 mmol), $N$-methylimidazole ( $99 \mu \mathrm{~L}, 1.24 \mathrm{mmol}$ ) and phenyl(benzyloxy-L-alaninyl) phosphorochloridate [17s] ( $0.26 \mathrm{~g}, 0.75 \mathrm{mmol}$ ). Purification by column chromatography (eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CHCl}_{3} 0 / 100$ to $6 / 94$ ) and preparative TLC ( $2000 \mu \mathrm{M}$, eluent system
$\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 5 / 95$ ) afforded the title compound as a white solid ( $0.013 \mathrm{mg}, 10 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 3.97, 3.64.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.06(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.04(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 7.33-7.28(\mathrm{~m}, 7 \mathrm{H}$, $\mathrm{Ph}), 7.20-7.14(\mathrm{~m}, 3 \mathrm{H}, \mathrm{Ph}), 5.92$ (d, $\left.J=1.5 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H}{ }^{\prime}\right), 5.90(\mathrm{~d}, J=1.5 \mathrm{~Hz}, 0.5 \mathrm{H}$, H1'), 5.14-5.04 (m, 2H, OCH ${ }_{2} \mathrm{Ph}$ ), 4.78-4.76 (m, 0.5H, H4'), 4.74-4.72 (m, 0.5H, H4'), 4.63-4.59 (m, 1H, H2'), 4.10-4.34 (m, 1H, H5'), 4.25-4.20 (m, 1H, H5'), 3.94 (s, 1.5H, $\mathrm{OCH}_{3}$ ), $3.95\left(\mathrm{~s}, 1.5 \mathrm{H}, \mathrm{OCH}_{3}\right), 3.99-3.90\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 2.40-2.37(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 3$ '), 2.07-2.04 (m, 1H, H3'), 1.31 (d, $\left.J=7.0 \mathrm{~Hz}, 1.5 \mathrm{H}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 1.26(\mathrm{~d}, J=7.0 \mathrm{~Hz}$, $\left.1.5 \mathrm{H}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$.
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 174.82\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.62\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 163.80 (C2), 158.16, 158.13 (C6), 152.15 (C4), 152.05 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.8 \mathrm{~Hz}, \mathrm{C}-$ Ar), 152.00 ( $\left.\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.8 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 139.39$ (C8), 137.30 (C-Ar), 137.21 (C-Ar), 130.72 (CH-Ar), 129.57 (CH-Ar), 129.31 (CH-Ar), 129.27 (CH-Ar), 126.122 (CH-Ar), 121.42 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), 121.37 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), 116.72 (C5), 116.69 (C5), 93.33, $93.24\left(\mathrm{C} 1^{\prime}\right), 80.26\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.9, \mathrm{C} 4{ }^{\prime}\right), 80.19\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.9, \mathrm{C} 4{ }^{\prime}\right), 76.35\left(\mathrm{C} 2^{\prime}\right)$, $68.78\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $68.35\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.94\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.92$ $\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 55.25\left(\mathrm{OCH}_{3}\right), 55.28\left(\mathrm{OCH}_{3}\right), 51.69\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.57\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$,
$35.23\left(\mathrm{C} 3\right.$ '), $34.96\left(\mathrm{C} 3\right.$ '), $20.38\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.7 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 20.26\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.7 \mathrm{~Hz}\right.$, $\mathrm{CHCH}_{3}$ L-Ala).
(ES + ) $\mathbf{m} / \mathbf{z}$ found $599.2\left[\mathrm{M}^{+} \mathrm{H}^{+}\right], \mathrm{C}_{27} \mathrm{H}_{31} \mathrm{~N}_{6} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 598.19$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $F=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed two peaks of the diastereoisomers with $\mathbf{t}_{\mathrm{R}}$ 14.22 min and $\mathbf{t}_{\mathrm{R}} 14.51 \mathrm{~min}$.
[50c] 2-Methoxy-3'-deoxyadenosine-5'-O-naphth-1-yl-(pentyl-1-oxy-L-leucinyl)phosphate


Prepared according to the general procedure F2 using 2-methoxy-3'-deoxyadenosine [48] (0.07 g, 0.25 mmol ), $N$-methylimidazole ( $99 \mu \mathrm{~L}, 1.24 \mathrm{mmol}$ ) and naphthyl(pentyl-1-oxy-L-leucinyl) phosphorochloridate [17f] ( $0.33 \mathrm{~g}, 0.75 \mathrm{mmol})$. Purification by column chromatography (eluent system gradient $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 0 / 100$ to $6 / 94$ ) and preparative TLC ( $2000 \mu \mathrm{M}$, eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 7 / 93$ ) afforded the title compound as a white solid ( $0.05 \mathrm{mg}, 30 \%$ ).
${ }^{31} \mathbf{P} \mathbf{N M R}\left(\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}} 4.53,4.28$.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.04-7.96 (m, 1H, H8), 7.77-7.71 (m, 1H, Nap), 7.587.53 (m, 1H, Nap), 7.45-7.17 (m, 5H, Nap), 5.83-5.75 (m, 1H, H1'), 4.64-4.51 (m, 2H, H2', H4'), 4.40-4.16 (m, 2H, H5'), 3.88-3.75 (m, 6H, OCH ${ }_{3}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen, $\left.\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}-\mathrm{Leu}\right), 2.38-2.24$ (m, 1H, H3'), 2.00-1.91 (m, 1H, H3'), 1.53-1.05 (m, $9 \mathrm{H}, \quad \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n$-Pen, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 0.77-0.55 (m, 9 H , $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 175.02\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.78\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 163.76 (C2), 158.14 (C6), 151.03 (C4), 147.96 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.2 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 138.96$ (C8), 136.30 (C-Ar), 136.28 (C-Ar), 136.22 (C-Ar), 128.93 (CH-Ar), 128.88 (CH-Ar), 128.81 (CH-Ar), 128.48 (CH-Ar), 127.77 (CH-Ar), 127.73 (CH-Ar), 127.44 (CH-Ar), 127.42 (CH-Ar), 127.06 (CH-Ar), 126.86 (CH-Ar), 126.45 (CH-Ar), 126.44 (CH-Ar), 126.31 (CH-Ar), 125.98 (CH-Ar), 125.88 (CH-Ar), 123.83 (CH-Ar), 123.43 (CH-Ar), 123.24 (CH-Ar), 122.81 ( $\mathrm{CH}-\mathrm{Ar}$ ), 122.77 ( $\mathrm{CH}-\mathrm{Ar}$ ), 122.69 (CH-Ar), 116.34 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7$ $\mathrm{Hz}, \mathrm{CH}-\mathrm{Ar}), 116.02$ (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 115.71$ (C5), 93.42 ( $\mathrm{Cl}{ }^{\prime}$ ), 93.32 (C1’), $80.22\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.3 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $80.15\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.3 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), 76.29 ( $\mathrm{C}^{\prime}$ ), 76.27 (C2'), $69.22\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.2 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $69.028\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.2 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $66.31\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ $n$-Pen $), 66.30\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $55.29\left(\mathrm{OCH}_{3}\right), 55.24\left(\mathrm{OCH}_{3}\right), 54.79$
$\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $54.68\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $44.20\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{p}}=7.2 \mathrm{~Hz}\right.$, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 43.93 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.2 \mathrm{~Hz}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 35.49 ( C 3 '), 35.17 ( $\mathrm{C}^{\prime}$ '), $29.31\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $29.11\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $25.67 \quad\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \quad\right.$ L-Leu $), \quad 25.44 \quad\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \quad\right.$ L-Leu $), \quad 23.30$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $23.10\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $23.00\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $22.94\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \text { L-Leu }\right)^{2} 22.81\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 14.27 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen).
(ES+ $) \mathbf{m} / \mathbf{z}$, found: $671.3\left[\mathrm{M}^{+} \mathrm{H}^{+}\right], \mathrm{C}_{32} \mathrm{H}_{43} \mathrm{~N}_{6} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 670.69$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed two peaks with tR 20.83 min and tR 20.93 $\min$.
[50d] 2-Methoxy-3'-deoxyadenosine-5'-O-phenyl-(hex-1-yloxy-L-alaninyl)-phosphate


Prepared according to the general procedure F2 using 2-methoxy-3'-deoxyadenosine [48] (0.07 g, 0.25 mmol ), $N$-methylimidazole ( $99 \mu \mathrm{~L}, 1.24 \mathrm{mmol}$ ) and phenyl(hex-1-yloxy-L-alaninyl) phosphorochloridate [ $\mathbf{1 7 u}](0.26 \mathrm{~g}, 0.75 \mathrm{mmol})$. Purification by column chromatography (eluent system gradient $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 0 / 100$ to $6 / 94$ ) and preparative TLC $\left(1000 \mu \mathrm{M}\right.$, eluent system $\left.\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 7 / 93\right)$ afforded the title compound as a white solid ( $0.026 \mathrm{~g}, 18 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 3.87, 3.65.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(500 \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.08(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.07(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 7.36-7.29(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{Ph}), 7.24-7.14(\mathrm{~m}, 3 \mathrm{H}, \mathrm{Ph}), 5.94\left(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 5.92(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 0.5 \mathrm{H}$, H1'), 4.81-4.76 (m, 1H, H2'), 4.71-4.62 (m, 1H, H4'), 4.48-4.43 (m, 0.5H, H5'), 4.42-4.36 $\left(\mathrm{m}, \quad 0.5 \mathrm{H}, \quad \mathrm{H} 5^{\prime}\right), \quad 4.33-4.25 \quad\left(\mathrm{~m}, \quad 1 \mathrm{H}, \quad \mathrm{H} 5^{\prime}\right), \quad 4.10-3.83 \quad\left(\mathrm{~m}, \quad 6 \mathrm{H}, \quad \mathrm{OCH}_{3}\right.$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex, $\mathrm{CHCH}_{3} \mathrm{~L}$-Ala), 2.48-2.40 (m, 1H, H3'), 2.13-2.07 (m, $1 \mathrm{H}, \mathrm{H} 3$ '), 1.61-1.51 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex), 1.33-1.24 (m, 9 H , $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex, $\mathrm{CHCH}_{3} \mathrm{~L}$-Ala), $0.89\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ $n$-Hex).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 175.13\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.3 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.94\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.3\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 163.80 ( C 2 ), 163.78 (C2), 158.17 (C6), 158.15 (C6), 152.17 (d, ${ }^{2} J_{C-P}=6.3 \mathrm{~Hz}$, C-Ar), 152.15 ( $\mathrm{d},{ }^{2} J_{C-P}=6.3 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}$ ), 152.03 (C4), 151.99 (C4), 139.42 (C8), 139.39 (C8), 130.75 (CH-Ar), 130.74 (CH-Ar), 126.13 (CH-Ar), 121.43 (CH-Ar), 121.41 (CHAr), 121.39 (CH-Ar), 121.37 (CH-Ar), 116.74 (C5), 116.69 (C5), 93.40 (C1'), 93.27 (C1’),
$80.26\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.7 \mathrm{~Hz}, \mathrm{C} 4\right.$ ') , $76.40\left(\mathrm{C} 2^{\prime}\right), 68.85\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.2 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $68.42\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=\right.$ $5.2 \mathrm{~Hz}, \mathrm{C} 5$ '), $66.43\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Hex $)$, $55.30\left(\mathrm{OCH}_{3}\right), 55.26\left(\mathrm{OCH}_{3}\right)$, $51.64\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.54\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 35.30(\mathrm{C} 3$ '), 35.04 ( C 3 '), 32.58 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex), $29.67\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex), 29.64 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex), $26.61\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex), 23.59 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Hex), $20.56\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.4 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}\right.$-Ala), $20.41\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}\right.$ $\left.=6.4 \mathrm{~Hz}, \mathrm{CHCH}_{3}\right), 14.36\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Hex).
(ES+) m/z, found: $593.3\left[\mathrm{M}+\mathrm{H}^{+}\right], \mathrm{C}_{32} \mathrm{H}_{43} \mathrm{~N}_{6} \mathrm{O}_{8} \mathrm{P}$ required m/z 592.58 [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $F=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed two peaks with tR 17.02 min and tR 17.23 $\min$.

### 9.5.10 Synthesis of 5'-modified 3'-deoxyadenosine analogues

[51] 2'-O-tert-butyldimethylsilyl-3'-deoxyadenosine


2',5'-O-Bis-tert-butyldimethylsilyl-3'-deoxyadenosine [25] (1.56 g, 3.25 mmol ) was dissolved in THF ( 2 mL ) and cooled down to $0{ }^{\circ} \mathrm{C}$. A solution of TFA in water ( $1 \mathrm{~mL}, 1 / 1 \mathrm{v} / \mathrm{v}$ ) was added dropwise and the mixture was stirred for 5 hours at $0{ }^{\circ} \mathrm{C}$. The solution was evaporated under vacuum and the crude purified via Biotage Isolera One ( 30 g ZIP cartridge KP SIL, $60 \mathrm{~mL} / \mathrm{min}$, gradient eluent system 2-20\% $\left.\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 10 \mathrm{CV}, 20 \% 5 \mathrm{CV}\right)$ to yield the title compound as a white foam (1.16 g, $98 \%) .{ }^{208}$
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.45(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 7.97(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 6.29\left(\mathrm{br} \mathrm{s}, 2 \mathrm{H}, \mathrm{NH}_{2}\right)$, 5.74 (d, $\left.J=6.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 5.19-5.12$ (m, 1H, H2'), 4.65-4.60 (m, 1H, H4'),
4.11 (dd, $J=13.0,1.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}{ }^{\prime}$ ), 3.67 (dd, $\left.\left.J=12.5,1.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}\right)^{\prime}\right)$, 2.72-2.62 (m, $1 \mathrm{H}, \mathrm{H} 3$ '), 2.37-2.26 (m, 1H, H3'), 1.48 (d, $J=7.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{OH}-5$ '), 0.91 ( $\mathrm{s}, 9 \mathrm{H}, t \mathrm{Bu}$ ), 0.00 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3}$ ), $-0.12\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.


Prepared according to general procedure $\mathbf{F}_{\mathbf{2}}$ using $2^{\prime}$ - $O$ -tert-butyldimethylsilyl-3'-deoxyadenosine [51] ( 0.05 g , 0.14 mmol ) in THF ( 2 mL ), phenyl-(benzyloxy-Lalaninyl) phosphorochloridate [17s] $(0.14 \mathrm{~g}, 0.41 \mathrm{mmol})$ in THF $(0.5 \mathrm{~mL})$ and $N$-methylimidazole ( $56 \mu \mathrm{~L}, 0.7$ $\mathrm{mmol})$.The crude was purified via Biotage Isolera One ( 10 g SNAP cartridge KP SIL, $10 \mathrm{~mL} / \mathrm{min}$, gradient eluent system $1-10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 12 \mathrm{CV}, 10 \% 2 \mathrm{CV}$ ) to afford the title compound as a white solid ( $0.06 \mathrm{~g}, 62 \%$ ).
${ }^{\mathbf{3 1}} \mathbf{P} \mathbf{N M R}\left(\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C H}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 2.89, 2.98.
${ }^{1} \mathbf{H}$ NMR ( $500 \mathbf{M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.436$ ( $\left.\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 8\right), 8.431(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.20(\mathrm{~s}, 0.5 \mathrm{H}$, H2), 8.19 ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 2$ ), $7.49-7.20(\mathrm{~m}, 10 \mathrm{H}, \mathrm{Ar}), 6.23\left(\mathrm{br} \mathrm{s}, 2 \mathrm{H}, \mathrm{NH}_{2}\right), 6.05(\mathrm{~d}, J=9.0 \mathrm{~Hz}$, $0.5 \mathrm{H}, \mathrm{H} 1$ '), 6.03 (d, $\left.J=9.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 4.98-4.89(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 2$ '), 4.81-4.72 (m, 1H, H3'), 4.56-4.32 (m, 2H, H5'), 4.26-4.16 (m, 1H, CHCH ${ }_{3}$ L-Ala), 2.32-1.99 (m, 2H, H3'), $1.01(\mathrm{~s}, 4.5 \mathrm{H}, t \mathrm{Bu}), 1.00(\mathrm{~s}, 4.5 \mathrm{H}, t \mathrm{Bu}), 0.23\left(\mathrm{~s}, 1.5 \mathrm{H}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 0.21(\mathrm{~s}, 1.5 \mathrm{H}$, $\left.\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 0.20\left(\mathrm{~s}, 1.5 \mathrm{H}, \mathrm{CH}_{3}\right), 0.19\left(\mathrm{~s}, 1.5 \mathrm{H}, \mathrm{CH}_{3}\right)$.


3'-Deoxyadenosine 5'-O-phenyl-(benzyloxy-L-alaninyl)phosphate [30a] $(0.025 \mathrm{~g}, 0.04 \mathrm{mmol})$ was dissolved in DMF ( 0.5 mL ). $\mathrm{TBDPSCl}(0.036 \mathrm{~g}, 0.13 \mathrm{mmol})$, imidazole $(16 \mathrm{mg}, 0.24 \mathrm{mmol})$ and DMAP $(0.003 \mathrm{~g})$ were added to the mixture, which was stirred at rt for 16 hours. The mixture was evaporated, and the crude purified via Biotage Isolera One ( 10 g ZIP cartridge KP SIL $10 \mathrm{~mL} / \mathrm{min}$, gradient eluent system $1-10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 10 \mathrm{CV}, 10 \% 2 \mathrm{CV}$ ), to yield the title compound as a white solid ( $0.024 \mathrm{~g}, 76 \%$ ).
${ }^{\mathbf{3 1}} \mathbf{P} \mathbf{N M R}\left(\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C H}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 2.71, 2.80.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.146(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 8.142(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 8), 7.94-7.92(\mathrm{~m}$, 1H, H2), 7.51-7.43 (m, 4H, Ar), 7.34-6.95 (m, 16H, Ar), 5.94-5.81 (m, 3H, H1', NH2), 5.02-4.88 (m, 2H, CH ${ }_{2} \mathrm{Ph}$ ), 4.78-4.73 (m, 1H, H2'), 4.63-4.56 (m, 1H, H4'), 4.33-4.22 (m, 1H, H5'), 4.16-4.09 (m, 1H, H5'), 4.04-3.86 (m, 2H, NH, CHCH3 L-Ala), 2.20-2.11 (m,

1H, H3'), 1.95-1.86 (m, 1H, H3'), 1.21-1.14 (m, 3H, CHCH ${ }_{3}$ L-Ala), 1.00 (s, 3H, CH3 $t \mathrm{Bu}), 0.99\left(\mathrm{~s}, 6 \mathrm{H}, 2 \times \mathrm{CH}_{3} t \mathrm{Bu}\right)$.
(ES + ) $\mathbf{m} / \mathbf{z}$ found $807.3\left[\mathrm{M}^{+} \mathrm{H}^{+}\right], 829.3\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{42} \mathrm{H}_{47} \mathrm{~N}_{6} \mathrm{O}_{7} \mathrm{PSi}$ required $\mathrm{m} / \mathrm{z} 806.3$ [M].

## [54a] 2',5'-Bis-O-tert-butyldimethylsilyl- $\mathrm{N}^{6}$-benzoyl-3'-deoxyadenosine



2',5'-Bis-O-(tert-butyldimethylsilyl)-3'-deoxyadenosine
( $0.89 \mathrm{~g}, 1.85 \mathrm{mmol}$ ) was dissolved in anhydrous pyridine ( 15 mL ) and benzoyl chloride ( $0.65 \mathrm{~mL}, 5.56 \mathrm{mmol}$ ) was added dropwise. The reaction was stirred at room temperature for 1.5 hours, then the flask was lowered in an ice-cold bath and water ( 5 mL ) was added. After 10 minutes $28 \% \mathrm{NH}_{3}$ in water ( 10 mL ) was added and the mixture stirred for 30 minutes. The reaction was allowed to reach room temperature, the solvent evaporated in vacuo, the residue dissolved in EtOAc (150 mL ) and washed with water $(70 \mathrm{~mL})$ and $\mathrm{NaHCO}_{3}$ (saturated solution). The organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to yield the product as a clear oil ( 0.56 g , $52 \%$ ). ${ }^{208}$
${ }^{1} \mathbf{H} \operatorname{NMR}\left(500 \mathbf{~ M H z}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{H}} 10.11$ (br s, $1 \mathrm{H}, \mathrm{NH}$ ), $8.54(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 8.36(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2)$, $7.83(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.28(\mathrm{t}, J=7.3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ar}), 7.18(\mathrm{t}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 5.90$ ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H} 1^{\prime}$ ), 4.49-4.46 (m, 1H, H2'), 4.45-4.39 (m, 1H, H4'), 3.97-3.92 (m, 1H, H5'), 3.64-3.58 (m, 1H, H5'), 2.11-2.03 (m 1H, H3'), 1.75-1.68 (m, 1H, H3'), 0.74 ( $\mathrm{s}, 9 \mathrm{H},{ }^{t} \mathrm{Bu}$ ), $0.72\left(\mathrm{~s}, 9 \mathrm{H},{ }^{t} \mathrm{Bu}\right), 0.00\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.05\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.06\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.07(\mathrm{~s}, 3 \mathrm{H}$, $\mathrm{CH}_{3}$ ).
[54b] (2',5'-O-bis-tert-butyldimethylsilyl- $N$-benzoyl)cordycepin


Isolated from the same procedure to obtain [54a]. Product obtained as a clear oil ( $0.41 \mathrm{~g}, 32 \%$ ).
${ }^{1} \mathbf{H} \operatorname{NMR}\left(500 \mathbf{M H z}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.67$ ( $\left.\mathrm{s}, 1 \mathrm{H}, \mathrm{H} 8\right), 8.54(\mathrm{~s}, 1 \mathrm{H}$, H2), 7.88 (d, $J=8.1 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}$ ), 7.87 (d, $J=8.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}$ ), 7.50-7.45 (m, 2H, Ar), 7.38-7.32 (m, 4H, Ar), 6.07 (d, $J=1.4$ $\mathrm{Hz}, 1 \mathrm{H}, \mathrm{H1}$ '), 4.74-4.70 (m, 1H, H2'), 4.63-4.57 (m, 1H, H4'), 4.09 (dd, $\left.J=11.5 \mathrm{~Hz}, 2.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5^{\prime}\right), 3.80(\mathrm{dd}, J=11.5 \mathrm{~Hz}$,
$2.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ '), 2.28-2.20 (m 1H, H3'), 1.97-1.90 (m, 1H, H3'), 0.94 ( $\mathrm{s}, 9 \mathrm{H},{ }^{\text {t }} \mathrm{Bu}$ ), 0.92 (s, $\left.9 \mathrm{H},{ }^{t} \mathrm{Bu}\right), 0.13\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{CH}_{3}\right), 0.12\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.10\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.
[54c] 2',5'-Bis-O-tert-butyldimethylsilyl-bis- $N^{6}, N^{6}$-acetyl-3'-deoxyadenosine 2',5'-Bis-O-(tert-butyldimethylsilyl)-3'-deoxyadenosine [25] (0.38
 $\mathrm{g}, 0.79 \mathrm{mmol})$ was dissolved in $\mathrm{CH}_{3} \mathrm{CN}(15 \mathrm{~mL})$ and acetic anhydride was added dropwise ( $0.45 \mathrm{~mL}, 4.74 \mathrm{mmol}$ ), followed by DMAP ( $0.04 \mathrm{~g}, 0.31 \mathrm{mmol}$ ), and $\mathrm{Et}_{3} \mathrm{~N}(0.66 \mathrm{ml}, 4.74 \mathrm{mmol})$. The mixture was diluted with aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ (saturated solution, 10 mL ) and extracted with $n$-Hex ( 3 x 20 mL ). The collected organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated. The crude was purified by flash column chromatography (eluent system: EtOAc/n-Hex 5-15\%) to yield the title compound as a clear oil ( $0.24 \mathrm{mg} 55 \%$ ).
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.80(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 8.56(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 5.98(\mathrm{~d}, J=1.0 \mathrm{~Hz}$, $\left.1 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 5.62-4.58\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 2^{\prime}\right), 4.52-4.46\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 4^{\prime}\right), 4.03(\mathrm{dd}, J=11.7 \mathrm{~Hz}, 2.3 \mathrm{~Hz}$, H5'), 3.67 (dd, $J=11.7 \mathrm{~Hz}, 2.3 \mathrm{~Hz}, \mathrm{H} 5$ ), 2.21 ( $\mathrm{s}, 6 \mathrm{H}, 2 \times \mathrm{COCH}_{3}$ ), 2.18-2.11 (m, 1H, H3'), 1.77 (ddd, $\left.J=13.05 \mathrm{~Hz}, 5.55 \mathrm{~Hz}, 1.90 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}{ }^{\prime}\right)$, 0.81 (s, $9 \mathrm{H},{ }^{t} \mathrm{Bu}$ ), 0.79 ( $\mathrm{s}, 9 \mathrm{H},{ }^{t} \mathrm{Bu}$ ), $0.04\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.01\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.00\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{CH}_{3}\right)$.


2',5'-Bis-O-(tert-butyldimethylsilyl)-3'-deoxyadenosine [25] $(1.56 \mathrm{~g}, 3.25 \mathrm{mmol})$ was dissolved in THF ( 20 mL ) and $\mathrm{Boc}_{2} \mathrm{O}$ was added $(2.83 \mathrm{~g}, 13.0 \mathrm{mmol})$, followed by DMAP $(0.16 \mathrm{mg}, 1.3 \mathrm{mmol})$. The mixture was stirred at room temperature overnight and was then diluted with $n$-Hex ( 20 mL ) and washed with brine ( 3 x 15 mL ). The collected organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated. The title compound was obtained as a clear oil $(2.21 \mathrm{~g}, 100 \%)$.
${ }^{1} \mathbf{H}$ NMR ( $500 \mathbf{M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.68(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 8.49(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 5.95(\mathrm{~d}, J=1.4 \mathrm{~Hz}$, 1H, H1'), 4.49-4.42 (m, 2H, H2', H4'), 4.00 (dd, $J=11.6 \mathrm{~Hz}, 2.5 \mathrm{~Hz}, \mathrm{H} 5$ '), 3.65 (dd, $J=$ $11.6 \mathrm{~Hz}, 2.5 \mathrm{~Hz}, \mathrm{H}^{\prime}$ ), 2.15-2.08 (m, 1H, H3'), 1.74 (ddd, $J=13.0 \mathrm{~Hz}, 5.6 \mathrm{~Hz}, 2.1 \mathrm{~Hz}, 1 \mathrm{H}$, H 3 '), $1.30\left(\mathrm{~s}, 18 \mathrm{H}, \mathrm{Boc}_{2}\right), 0.80(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.76(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.00\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{CH}_{3}\right),-0.01(\mathrm{~s}$, $\left.3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.04\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.
[55a] 2'-O-tert-butyldimethylsilyl- $\boldsymbol{N}^{6}$-benzoyl-3'-deoxyadenosine

(2',5’-O-bis-tert-butyldimethylsilyl- N -benzoyl)cordycepin [54a] ( $0.56 \mathrm{~g}, 0.96 \mathrm{mmol}$ ) was stirred in a solution of trifluoroacetic acid, water and THF ( $0.5 \mathrm{~mL}: 0.5 \mathrm{~mL}: 2 \mathrm{~mL}$ ) for 5 hours in an ice-cold bath. The solution was carefully neutralised by dropwise addition of $\mathrm{NaHCO}_{3}$ (saturated aqueous solution), the mixture was diluted with $n$-Hex ( 20 mL ), washed with brine ( $2 \times 20 \mathrm{~mL}$ ); the organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to give the product as a sticky solid $(0.46 \mathrm{mg}, 100 \%) .{ }^{208}$
${ }^{1} \mathbf{H}$ NMR ( $500 \mathbf{M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 9.24$ (br s, 1H, NH), $8.83(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 8.13(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2)$, 8.05 (d, $J=7.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 7.63$ (t, $J=8.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Ar}), 7.54$ (t, $J=7.4 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{Ar}), 5.74$ (d, $J=5.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Hl}$ '), 5.42 (d, $\left.J=10.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{OH}-5^{\prime}\right), 5.03$ (dd, $J=14.3,6.5 \mathrm{~Hz}, 1 \mathrm{H}$, H2'), 4.59-4.54 (m, 1H, H4'), 4.06-4.03 (m, 1H, H5'), 3.64-3.59 (m, 1H, H5'), 2.60-2.53 (m 1H, H3'), 2.55-2.16 (m, 1H, H3'), $0.81(\mathrm{~s}, 9 \mathrm{H}, \mathrm{tBu}),-0.07\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.20(\mathrm{~s}, 3 \mathrm{H}$, $\mathrm{CH}_{3}$ ).
[55b] 2'-O-tert-butyldimethylsilyl-bis- $N^{6}, N^{6}$-benzoyl-3'-deoxyadenosine

(2',5'-O-bis-tert-butyldimethylsilyl-bis- $N^{6}, N^{6}$-benzoyl) $3^{\prime}$ deoxyadenosine [54b] ( $1.22 \mathrm{~g}, 1.77 \mathrm{mmol}$ ) was stirred in a solution of trifluoroacetic acid, water and THF ( $1 \mathrm{~mL}: 1 \mathrm{~mL}: 4$ mL ) for 5 hours in an ice-cold bath. The solution was carefully neutralised by dropwise addition of aqueous $\mathrm{NaHCO}_{3}$ (saturated solution), the mixture was diluted with $n$ - $\mathrm{Hex}(20 \mathrm{~mL})$, washed with brine ( $2 \times 20 \mathrm{~mL}$ ); the organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to give the product as a sticky solid ( $0.88 \mathrm{~g}, 87 \%$ ).
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.68(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 8.19$ (s, 1H, H2), 7.88-7.86 (m, 4H, Ar), 7.52-7.50 (m, 2H, Ar), 7.40-7.36 (m, 4H, Ar), 5.73 (d, $J=5.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Hl}$ '), 5.01-4.96 (m, $1 \mathrm{H}, \mathrm{H} 2$ '), 4.93-4.90 (m, 1H, OH-5'), 4.59-4.56 (m, 1H, H4'), 4.05-4.02 (m, 1H, H5'), 3.64-3.58 (m, 1H, H5'), 2.57-2.52 (m 1H, H3'), 2.24-2.18 (m, 1H, H3'), 0.82 ( $\mathrm{s}, 9 \mathrm{H},{ }^{t} \mathrm{Bu}$ ), $-0.10\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.23\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.
[55c] 2'-O-tert-butyldimethylsilyl-bis- $N^{6}, N^{6}$-acetyl-- $3^{\prime}$ '-deoxyadenosine
 (2', $5^{\prime}$-O-bis-tert-butyldimethylsilyl-bis- $N^{6}, N^{6}$-acetyl) 3'deoxyadenosine [54c] $(0.20 \mathrm{~g}, 0.35 \mathrm{mmol})$ was stirred in a solution of trifluoroacetic acid, water and THF ( $0.25 \mathrm{~mL}: 0.25 \mathrm{~mL}: 1 \mathrm{~mL}$ ) for 5 hours in an ice-cold bath. The solution was carefully neutralised by dropwise addition of $\mathrm{NaHCO}_{3}$ (saturated aqueous solution), the mixture was diluted with $n$-Hex ( 20 mL ), washed with brine ( $2 \times 20$ mL ); the organics were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to give the product as a sticky solid ( $0.12 \mathrm{~g}, 75 \%$ ).
${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.89(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 8.61(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 5.84(\mathrm{~d}, J=3.0 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{H} 1$ '), 4.80-4.76 (m, 1H, H2'), 4.58-4.49 (m, 1H, H4'), 4.06-4.02 (m, 1H, H5'), 3.773.68 ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{H} 5$ '), $2.38-2.31\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 3\right.$ '), $2.23\left(\mathrm{~s}, 6 \mathrm{H}, 2 \times \mathrm{COCH}_{3}\right), 2.08-2.01(\mathrm{~m}, 1 \mathrm{H}$, H3'), $0.81(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.00\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.09\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.13\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{CH}_{3}\right)$.
[55d] 2'-O-tert-butyldimethylsilyl-bis- $N^{6}, N^{6}$-tert-butylcarbonyl-3'-deoxyadenosine

$2^{\prime}$-O-tert-butyldimethylsilyl-(bis- $N^{6}, N^{6}$-tert-butylcarbonate)-3'-deoxyadenosine [54d] ( $1.26 \mathrm{~g}, 1.85 \mathrm{mmol})$ was stirred in a mixture of TFA, $\mathrm{H}_{2} \mathrm{O}$ and THF $(1: 1: 4)$ at $0{ }^{\circ} \mathrm{C}$ for 5 hours. The mixture was then neutralised with solid $\mathrm{NaHCO}_{3}$ and extracted with $n$-Hex ( $3 \times 40 \mathrm{~mL}$ ). The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and the volatiles were concentrated under vacuum. The residue was purified by silica gel column chromatography (EtOAc: $n \mathrm{Hex}=1: 1 \mathrm{v} / \mathrm{v}$ ) to give the title compound as a sticky solid ( $0.70 \mathrm{~g}, 66 \%$ ).
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.80(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 8.68(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 5.77(\mathrm{~d}, J=4.0 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{H} 1$ ') , 4.75-4.71 (m, 1H, OH-5'), 4.56-4.48 (m, 2H, H2', H4'), 4.01 (dd, $J=12.5 \mathrm{~Hz}$, $\left.2.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5^{\prime}\right), 3.61$ (dd, $\left.J=12.5 \mathrm{~Hz}, 2.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5^{\prime}\right), 2.04-1.97$ (m 1H, H3'), 1.801.72 (m, 1H, H3'), 1.36 ( $\mathrm{s}, 18 \mathrm{H}, 2 \times \mathrm{mu}$ Boc), $0.80(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}),-0.09\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.13$ ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3}$ ). hexofuranosyl] bis- $N^{6}, N^{6}$-tert-butoxycarbonyl 3-deoxyadenosine


To a solution of $2^{\prime}$-O-tert-butyldimethylsilyl-(bis- $N^{6}, N^{6}$ -tert-butylcarbonate)-3'-deoxyadenosine [55d] (0.50 g, $0.73 \mathrm{mmol})$ in $\mathrm{CH}_{3} \mathrm{CN}(5 \mathrm{~mL})$ was added 2iodoxybenzoic acid (IBX) ( $1.2 \mathrm{~g}, 4.28 \mathrm{mmol}$ ). After stirring for 4 h at $80^{\circ} \mathrm{C}$, the reaction mixture was concentrated under reduced pressure and co-evaporated with THF. The residue was then dissolved in THF ( 10 mL ) and filtered through a sintered funnel. The solution was concentrated under vacuum and dried under high vacuum for 1 h . Subsequently, the crude was dissolved in THF ( 10 mL ) and a solution of tetraethyl methylenediphosphonate sodium salt (freshly prepared by addition, at $0^{\circ} \mathrm{C}$, of tetraethyl methylenediphosphonate ( $0.8 \mathrm{~mL}, 3.21 \mathrm{mmol}$ ) to a suspension of $\mathrm{NaH}(128 \mathrm{mg}, 5.35$ $\mathrm{mmol})$ in THF ( 2 mL ) and stirring for 10 min ) was added dropwise at $0^{\circ} \mathrm{C}$. The reaction mixture was then stirred at room temperature for 12 h before addition of $\mathrm{NH}_{4} \mathrm{Cl}(15 \mathrm{~mL}$, saturated aqueous solution). The mixture was then extracted with EtOAc ( $3 \times 10 \mathrm{~mL}$ ). The combined organic layers were washed with water $(2 \times 10 \mathrm{~mL})$ and brine $(10 \mathrm{~mL})$ and the resulting solution was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under vacuum. The residue was purified by Biotage Isolera One ( 50 g SNAP cartridge KP SIL $50 \mathrm{~mL} / \mathrm{min}$, gradient eluent system $16-100 \% \mathrm{EtOAc} / \mathrm{nHex} 7 \mathrm{CV}, 100 \% \mathrm{5CV}$ ) to yield the title compound as a sticky solid (single $E$ isomer, $0.29 \mathrm{~g}, 0.51 \mathrm{mmol}$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 17.15.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.71(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 7.99(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 6.90\left(\mathrm{ddd}, J_{\mathrm{HP}}=22.1\right.$
$\left.\mathrm{Hz}, J_{\mathrm{HH}}=17.0,4.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6^{\prime}\right), 5.93\left(\mathrm{~d}, J=1.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H1}{ }^{\prime}\right), 5.91$ (ddd, $J_{\mathrm{HP}}=19.1 \mathrm{~Hz}$, $\left.J_{\mathrm{HH}}=17.0,1.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5^{\prime}\right), 4.96-4.89\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 2^{\prime}\right), 4.69-4.64\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 4{ }^{\prime}\right), 4.01-3.91$ ( $\mathrm{m}, 4 \mathrm{H}, 2 \times \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 2.09-2.03 (m, 1H, H3'), 2.02-1.95 (m, 1H, H3'), 1.33 ( $\mathrm{s}, 18 \mathrm{H}, 2 \times$ $t \mathrm{Bu} \mathrm{Boc}_{2}$ ), 1.22-1.17 (m, $6 \mathrm{H}, 2 \times \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 0.77 ( $\mathrm{s}, 9 \mathrm{H}, t \mathrm{Bu}$ ), $0.00\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.03(\mathrm{~s}$, $3 \mathrm{H}, \mathrm{CH}_{3}$ ).
[60] 9-[5',6'-Dideoxy-6'-(bis-ethylphosphinyl)-2'-O-tert-butyldimethylsilyl-D-ribohexofuranosyl] bis- $N^{6}, N^{6}$-tert-butoxycarbonyl $3^{\prime}$ '-deoxyadenosine

To a solution of 9-[5', $6^{\prime}$-vinyl-6'-(bis-ethylphosphinyl)-2'-
 $O$-tert-butyldimethylsilyl-D-ribo-hexofuranosyl] bis$N^{6}, N^{6}$-tert-butoxycarbonyl 3-deoxyadenosine [59] ( 0.14 g , 0.20 mmol ) was dissolved in a $1: 1$ mixture of EtOAc and EtOH ( 10 mL ) under Argon atmosphere. To this solution, $\mathrm{Pd} / \mathrm{C}(0.02 \mathrm{~g}, 10 \% \mathrm{Pd}$ on activated carbon) was added, and the suspension was stirred under $\mathrm{H}_{2}$ atmosphere for 1 hour at room temperature. $\mathrm{NaCl}(0.02 \mathrm{~g})$ was added and the mixture stirred for additional 15 minutes, and then filtered through a short pad of celite and washed with EtOH ( $10 \mathrm{~mL} \times 3$ ). The collected solution was concentrated under vacuum to yield the title compound as a sticky solid ( $107 \mathrm{mg}, 93 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathrm{P}} 31.03$.
${ }^{1} \mathbf{H} \operatorname{NMR}\left(500 \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.75(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 8.08(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 5.88(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 1$ '), 4.76 (d, $J=4.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2$ '), 4.47-4.37 (m, 1H, H4'), 4.09-3.97 (m, 4H, $2 \times \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 2.031.71 ( $\mathrm{m}, 6 \mathrm{H}, \mathrm{H} 3^{\prime}, \mathrm{H} 5^{\prime}, \mathrm{H}^{\prime}$ ), 1.38 ( $\mathrm{s}, 18 \mathrm{H}, \mathrm{Boc}_{2}$ ), $1.25\left(\mathrm{t}, J=7.0 \mathrm{~Hz}, 6 \mathrm{H}, 2 \times \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ ), $0.81(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.02\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.00\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.
[61] 9-[5',6'-Dideoxy-6'-(bis-ethylphosphinyl)-3'-deoxy-D-ribo-hexofuranosyl] adenine


9-[5',6'-Dideoxy-6'-(bis-ethylphosphinyl)-2'-O-tert-butyldimethylsilyl-D-ribo-hexofuranosyl] bis- $N^{6}, N^{6}$-tertbutoxycarbonyl 3'-deoxyadenosine [60] ( $0.19 \mathrm{~g}, 0.27 \mathrm{mmol}$ ) was stirred in a solution of HCOOH and water ( $3 / 2 \mathrm{v} / \mathrm{v}, 5 \mathrm{~mL}$ ) for 96 hours. The solution was evaporated under vacuum and the residue purified via Biotage Isolera One ( 10 g ZIP cartridge KP-SIL, $36 \mathrm{~mL} / \mathrm{min}$, gradient eluent system 2-20\% $\mathrm{CH}_{3} \mathrm{OH}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2} 10 \mathrm{CV}, 20 \% 6 \mathrm{CV}$ ) to yield the title compound as a white solid ( $0.10 \mathrm{~g}, 96 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 32.82$.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.23(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 8.20(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 5.98(\mathrm{~d}, J=2.0 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{H} 1$ '), 4.82-4.79 (m, 1H, H2'), 4.52-4.46 (m, 1H, H4'), 4.14-4.04 (m, 4H, 2 x $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 2.24-2.15 (m, 2H, H3'), 2.06-1.83 (m, 4H, H5', H6'), 1.31 (t, J = 7 Hz, 6H, 6H, $2 \mathrm{x} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ).
[64] 9-[5',6'-Dideoxy-6'-(phenyloxy-(benzyloxy-L-alanin- $N$-yl-phosphinyl-3'-deoxy)-D-ribo-hexofuranosyl] adenine


9-[5',6'-Dideoxy-6'-(bis-ethylphosphinyl)-3'-deoxy-D-ribohexofuranosyl] adenine [61] ( $0.06 \mathrm{~g}, 0.15 \mathrm{mmol}$ ) was dissolved in $\mathrm{CH}_{3} \mathrm{CN}(6 \mathrm{~mL})$ under Argon atmosphere, and 2,6-lutidine ( $55 \mu \mathrm{~L}, 0.47 \mathrm{mmol}$ ) was added to the solution, followed by TMSBr ( $62 \mu \mathrm{~L}, 0.47 \mathrm{mmol}$ ), and the mixture was stirred at $0^{\circ} \mathrm{C}$ for 3 hours. The solvents were removed under reduced pressure without any contact with air. The residue was dissolved in anhydrous $\mathrm{Et}_{3} \mathrm{~N}(0.31 \mathrm{~mL})$ and pyridine $(1.25 \mathrm{~mL})$ and L-alanine benzyl ester HCl salt was added ( $0.05 \mathrm{~g}, 0.16 \mathrm{mmol}$ ) along with phenol $(0.09 \mathrm{mg}, 0.94 \mathrm{mmol})$. In a separated flask, aldrithiol-2 $(0.21 \mathrm{~g}, 0.93 \mathrm{mmol})$ and $\mathrm{Ph}_{3} \mathrm{P}(0.24 \mathrm{~g}, 0.93 \mathrm{mmol})$ were dissolved in pyridine $(1.56 \mathrm{ml})$ and this solution was immediately added to the reaction. The resulting mixture was stirred at $50^{\circ} \mathrm{C}$ for 16 h , then cooled to room temperature and the solvent evaporated. The residue was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$ and washed with HCl (aqueous solution, 0.5 N , $5 \mathrm{~mL} \times 3$ ), the organics collected, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated. The crude was purified by Biotage Isolera One ( 10 g SNAP cartridge KP-SIL, $36 \mathrm{~mL} / \mathrm{min}$, gradient eluent system $2-16 \% \mathrm{CH}_{3} \mathrm{OH}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ in 25 CV ), and preparative HPLC (20 $\mathrm{mL} / \mathrm{min}$, isocratic eluent system $55 \% \mathrm{CH}_{3} \mathrm{OH}$ in water, 30 minutes), to afford the title compound as a white solid ( $0.002 \mathrm{~g}, 2 \%$ ).
${ }^{31} \mathbf{P} \mathbf{N M R}\left(\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 34.21, 33.56.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.13-8.09 (m, 2H, H8, H2), 7.26-7.17 (m, 7H, Ar), 7.11$7.03(\mathrm{~m}, 3 \mathrm{H}, \mathrm{Ar}), 5.89-5.87\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 5.02\left(\mathrm{AB} \mathrm{q}, J_{\mathrm{AB}}=12.3 \mathrm{~Hz}, \Delta \delta_{\mathrm{AB}}=0.03,1 \mathrm{H}\right.$, $\mathrm{CH}_{2} \mathrm{Ph}$ ), 4.94 ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}$ ), 4.68-4.63 (m, 1H, H2'), 4.40-4.33 (m, 1H, H4'), 3.95-3.87 (m, 1H, $\mathrm{CHCH}_{3}$ L-Ala), 2.11-1.84 (m, 6H, H3', H5', H6'), 1.19-1.08 (m, 3H, CHCH ${ }_{3}$ LAla).

MS (ES+) m/z found $589.2\left[\mathrm{M}+\mathrm{Na}^{+}\right]$, $566.2\left[\mathrm{M}+\mathrm{H}^{+}\right], \mathrm{C}_{30} \mathrm{H}_{30} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required m/z 566.5 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, two peaks with tR 13.37 min , tR 13.53 min .

### 9.5.11 Synthesis of 8-chloroadenosine

## [67] 8-Chloroadenosine



Adenosine ( $1.97 \mathrm{~g}, 7.37 \mathrm{mmol}$ ) was suspended in DMF ( 20 mL ) and $\mathrm{BzCl}(0.9 \mathrm{~mL}, 7.7 \mathrm{mmol})$ was added. $m \mathrm{CPBA}(70-75 \%, 2.03 \mathrm{~g}, 8.6$ $\mathrm{mmol})$ was dissolved in DMF ( 10 mL ) and added dropwise. The reaction mixture was stirred at rt for 20 min before it was poured into ice-cold water ( 100 mL ) and filtered. The filtrate was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 100 \mathrm{~mL})$. The combined aqueous phases were evaporated under vacuum to give an oil. The oil was absorbed on silica and purified by silica gel CC (packed in $5 \% \mathrm{CH}_{3} \mathrm{OH}$ $/ \mathrm{CHCl}_{3}$, eluted with $7 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CHCl}_{3}$ ) to yield the title compound as a white solid ( 0.53 g, 24\%).
[Alterative procedure-1] Adenosine ( $1.09 \mathrm{~g}, 4.08 \mathrm{mmol}$ ) was suspended in DMF ( 50 mL ) and glacial acetic acid ( 10 mL ) was added. A solution of $N$-chlorosuccinimmide ( $2 \mathrm{~g}, 15$ mmol ) in DMF ( 15 mL ) was added dropwise. The reaction mixture was stirred at rt for 48 hours and the volatiles were evaporated in vacuo to give a yellow gum. The crude was absorbed on silica and purified by silica gel CC (packed in $5 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CHCl}_{3}$, eluted with $7 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CHCl}_{3}$ ) to yield the title compound as a white powder $(0.65 \mathrm{~g}, 52 \%) .{ }^{344}$
[Alterative procedure-2] 2'-3',5'-tri-O-tert-butyldimethylsilyl 8-chloroadenosine [70] $(1.0 \mathrm{~g}, 1.55 \mathrm{mmol})$ was dissolved in THF $(10 \mathrm{~mL})$ and the solution was cooled down to 0 ${ }^{\circ} \mathrm{C}$. TFA $(2.5 \mathrm{~mL})$ was diluted with water $(2.5 \mathrm{~mL})$ and the solution was added dropwise to the initial mixture. The reaction was stirred at rt for 36 hours. The solvent was evaporated and the crude was absorbed on silica and purified by silica gel CC (packed in $5 \% \mathrm{CH}_{3} \mathrm{OH}$ $/ \mathrm{CHCl}_{3}$, eluted with $7 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CHCl}_{3}$ ) to yield the title compound as a white powder ( $0.31 \mathrm{~g}, 64 \%$ ).
Melting point: $188-189{ }^{\circ} \mathrm{C}\left(\right.$ Lit. $\left.\mathrm{mp}: 189-191{ }^{\circ} \mathrm{C}\right) .{ }^{471}$
${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}$, DMSO-d6) $\boldsymbol{\delta}_{\mathbf{H}} 8.16$ (s, 1H, H2), 7.55 (br s, 2H, NH2), 5.86 (d, $J=$ $\left.6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 5.48\left(\mathrm{~d}, J=6.2 \mathrm{~Hz}, 1 \mathrm{H}, 2^{\prime} \mathrm{OH}\right), 5.45\left(\mathrm{~d}, J=4.0 \mathrm{~Hz}, 1 \mathrm{H}, 5^{\prime}-\mathrm{OH}\right), 5.23(\mathrm{~d}$, $\left.J=4.6 \mathrm{~Hz}, 1 \mathrm{H}, 3^{\prime} \mathrm{OH}\right), 5.08-5.06$ (m, 1H, H2'), 4.23-4.18 (m, 1H, H3'), 4.01-3.96 (m, 1H, H4'), 3.72-3.65 (m, 1H, H5'), 3.57-3.50 (m, 1H, H5').
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 6} \mathbf{~ M H z}$, DMSO-d6) $\boldsymbol{\delta}_{\mathrm{C}} 156.2$ (C6), 152.3 (C2), 149.6 (C4), 137.0 (C8) 117.9 (C5), 89.2 (C1'), 86.7 (C4'), 71.2 (C2'), 70.77 ( $\mathrm{C}^{\prime}$ ), 62.0 ( C 5 ').
[69] 2', 3',5'-Tri-O-tert-butyldimethylsilyl-D-ribo-hexofuranosyl-adenosine


Adenosine ( $10.0 \mathrm{~g}, 37.42 \mathrm{mmol}$ ) was dissolved in DMF ( 60 mL ) and TBDMSCl ( $24.8 \mathrm{~g}, 149.7 \mathrm{mmol}$ ), imidazole ( $22.4 \mathrm{~g}, 329.3$ mmol) and DMAP ( $1.83 \mathrm{~g}, 14.97 \mathrm{mmol}$ ) were added. The mixture was stirred at rt for 16 hours, then the solution was evaporated under vacuum and the mixture was suspended in $n$ hexane ( 250 mL ) and washed with aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ (saturated solution, $2 \times 150 \mathrm{~mL}$ ) and brine ( $1 \times 150 \mathrm{~mL}$ ). The organic were collected, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered off and evaporated to yield the title compound as a sticky solid ( 22.83 g , $100 \%) .{ }^{471}$
${ }^{1} \mathbf{H}$ NMR ( $\left.\mathbf{C D C l}_{3}, \mathbf{5 0 0} \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.23(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 8.03(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 5.90(\mathrm{~d}, J=5.0 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{H} 1^{\prime}$ ), 5.46 (br s, 2H, NH 2 ), $4.59(\mathrm{t}, J=4.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2$ '), $4.22(\mathrm{t}, J=3.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3$ '), 4.03-3.99 (m, 1H, H4'), 3.92 (dd, $J=11.5,4.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5^{\prime}$ ), 3.68 (dd, $J=11.0,3.0 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{H} 5$ '), $0.846(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.826(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.68(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.02\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.02$ ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3}$ ), $0.00\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.05\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.15\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.33\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.
[70] 2',3',5'-tri-O-tert-butyldimethylsilyl-D-ribo-hexofuranosyl-8-chloroadenosine To a stirring solution of LDA ( 2 M in THF, $93.55 \mathrm{~mL}, 187.1$ mmol ) in a $-78{ }^{\circ} \mathrm{C}$ bath was added a solution of $2^{\prime}, 3^{\prime}, 5^{\prime}$-tri- $O$ -tert-butyldimethylsilyl-D-ribo-hexofuranosyl-adenosine
[69] ( $22.83 \mathrm{~g}, 37.42 \mathrm{mmol}$ ) in THF ( 15 mL ) dropwise via a dropping funnel. The funnel was washed with THF ( 5 mL ) and the reaction was stirred for 1 hour at $-7{ }^{\circ} \mathrm{C}$. A solution of $\mathrm{TsCl}(28.54 \mathrm{~g}, 149.68 \mathrm{mmol})$ in THF ( 20 mL ) was added dropwise, then the funnel rinsed with THF ( 5 mL ). The reaction was stirred for 1 hour and then quenched by dropwise addition of glacial $\mathrm{AcOH}(10 \mathrm{~mL})$. The mixture was allowed to reach room temperature and evaporated. The crude was dissolved in $n$-hexane ( 50 mL ) and washed with brine ( $3 \times 50 \mathrm{~mL}$ ). The organic phase was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated. The crude was purified by Biotage Isolera flash chromatography (100g SNAP cartridge KP SIL, eluent system 3-30\% EtOAc/ $n \mathrm{Hex}$ $10 \mathrm{CV}, 30 \% \mathrm{AcOEt} / n \mathrm{Hex} 4 \mathrm{CV})$ to yield the title compound as a clear oil ( 17.84 g , $74 \%$ ). ${ }^{350}$
${ }^{1} \mathbf{H}$ NMR ( $\left.\mathbf{C D C l}_{3}, \mathbf{5 0 0} \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.11(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 5.79(\mathrm{~d}, \mathrm{~J}=5.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), 5.34-5.28 ( $\mathrm{m}, 2 \mathrm{H}$,), 4.42-4.40 (dd, J = 4, 2.4 Hz, 1H, H3'), 3.94-3.87 (m, 2H, H4', H5'), 3.59-3.54 (m, 1H, H5'), $0.80(\mathrm{~s}, 9 \mathrm{H}, \mathrm{tBu}), 0.68(\mathrm{~s}, 9 \mathrm{H}, \mathrm{tBu}), 0.63(\mathrm{~s}, 9 \mathrm{H}, \mathrm{tBu}), 0.003\left(\mathrm{~s}, 3 \mathrm{H} . \mathrm{CH}_{3}\right)$,
$0.00\left(\mathrm{~s}, 3 \mathrm{H} . \mathrm{CH}_{3}\right),-0.13\left(\mathrm{~s}, 3 \mathrm{H} . \mathrm{CH}_{3}\right),-0.16\left(\mathrm{~s}, 3 \mathrm{H} . \mathrm{CH}_{3}\right),-0.21\left(\mathrm{~s}, 3 \mathrm{H} . \mathrm{CH}_{3}\right),-0.484(\mathrm{~s}, 3 \mathrm{H}$. $\mathrm{CH}_{3}$ ).

### 9.5.12 Synthesis of 8-chloroadenosine ProTides

## [71a] 8-chloroadenosine-5'-naphth-1-yl(benzyloxy-L-alaninyl) phosphate



Prepared according to the general procedure $\mathbf{F}_{1}$ using 8-chloroadenosine [67] ( $0.10 \mathrm{~g}, 0.33 \mathrm{mmol}$ ) in THF (7 $\mathrm{mL}), t \mathrm{BuMgCl}(1.1 \mathrm{M}$ in THF, $0.36 \mathrm{~mL}, 0.36 \mathrm{mmol})$, naphth-1-yl-(benzyloxy-L-alaninyl)phosphorochloridate $[\mathbf{1 7 a}](0.42 \mathrm{~g}, 1.04 \mathrm{mmol})$ in THF ( 3 mL ). The crude was purified first by silica gel $\mathrm{CC}\left(0-6 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ and secondly by prep. TLC $(5 \%$ $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to yield the title compound as a white solid ( $0.037 \mathrm{~g}, 17 \%$ ).
${ }^{31} \mathbf{P} \mathbf{N M R}\left(\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 3.82, 3.94.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.12-8.03(\mathrm{~m}, 3 \mathrm{H}, \mathrm{H} 2, \mathrm{Ar}), 7.81-7.83(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ar}), 7.66-$ 7.61 (m, 1H, Ar), 7.52-7.18 (m, 8H, Ar) 6.05-6.01 (m, 1H, H1'), 5.35-5.31 (m, 0.5H, H2'), 5.30-5.26 (m, 0.5 H, H-2'), 5.05-4.91 (m, 2H, CH ${ }_{2} \mathrm{Ph}$ ), 4.69-4.63 (m, 1H, H3'), 4.53-4.45 (m, 1H, H5') 4.43-4.34 (m, 1H, H5'), 4.26-4.20 (m, 1H, H4'), 4.08-4.00 (m, 0.5H, CHCH L-Ala), 3.97-3.89 (m, 0.5H, $\mathrm{CHCH}_{3}$ L-Ala), 1.28-1.14 (m, 3H, $\mathrm{CHCH}_{3}$ L-Ala).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 174.70\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.60\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 156.24 (C6), 156.22 (C6), 154.11 (C2), 154.01 (C-2), 151.62 (C4), 151.58 (C4), 147.94 (C8), 147.88 (C8) 139.82 (C-Ar), 139.76 (C-Ar), 137.17 (C-Ar), 137.12 (C-Ar), 136.19 (C-Ar), 129.54 (CH-Ar), 129.46 (CH-Ar), 129.27 (CH-Ar), 129.22 (CH-Ar), 129.17 (CH-Ar), 129.14 (CH-Ar), 128.82 (CH-Ar), 128.78 (CH-Ar), $127.84\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=6.2\right.$ $\mathrm{Hz}, \mathrm{C}-\mathrm{Ar}), 127.78$ (d, $\left.{ }^{2} J_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 127.75(\mathrm{CH}-\mathrm{Ar}), 127.71$ (CH-Ar), 127.39 (CH-Ar), 127.36 (CH-Ar), 126.42 (CH-Ar), 125.88 (CH-Ar), 122.79 (CH-Ar), 122.62 (CH-Ar), 119.53 (C5), 116.23 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), $116.08\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{CH}-\right.$ Ar), $91.40\left(\mathrm{C}^{\prime}\right), 91.31\left(\mathrm{C}^{\prime}\right), 84.34\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.0 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $84.22\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.0 \mathrm{~Hz}, \mathrm{C} 4^{\prime}\right)$, 72.78 ( $\mathrm{C}^{\prime}$ '), 72.65 ( $\mathrm{C}^{\prime}$ ), 71.66 ( $\mathrm{C} 3^{\prime}$ ), 71.14 ( C 3 '), 67.96 ( $\mathrm{C}^{\prime}$ ', $\mathrm{CH}_{2} \mathrm{Ph}$ ), 67.87 ( $\mathrm{C}^{\prime}$ ', $\left.\mathrm{CH}_{2} \mathrm{Ph}\right), 67.80\left(\mathrm{C}^{\prime}, \mathrm{CH}_{2} \mathrm{Ph}\right), 67.15\left(\mathrm{C}^{\prime}, \mathrm{CH}_{2} \mathrm{Ph}\right), 67.11\left(\mathrm{C}^{\prime}, \mathrm{CH}_{2} \mathrm{Ph}\right), 51.63\left(\mathrm{CHCH}_{3} \mathrm{~L}-\right.$ Ala), $51.56\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $20.48\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}, \mathrm{CHCH}_{3}\right.$ L-Ala), $20.28\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.2\right.$ $\mathrm{Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ).

MS (ES+) $\mathbf{m} / \mathbf{z}$ found $691.16\left[\mathrm{M}+\mathrm{Na}^{+}\right], 707.13\left[\mathrm{M}+\mathrm{K}^{+}\right], \mathrm{C}_{30} \mathrm{H}_{30} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathbf{z}$ 669.02 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{CH}_{3} \mathrm{CN} / \mathrm{H}_{2} \mathrm{O}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, two peaks with tR 16.48 min , tR 17.07 min .
[71b] 8-chloroadenosine-5'-naphth-1-yl(ethyloxy-L-alaninyl) phosphate


Prepared according to the general procedure $\mathrm{F}_{1}$ using 8chloroadenosine [67] ( $0.10 \mathrm{~g}, 0.33 \mathrm{mmol}$ ) in THF ( 7 mL ), $t \mathrm{BuMgCl}(1.1 \mathrm{M}$ in THF, $0.36 \mathrm{~mL}, 0.36 \mathrm{mmol})$, naphth-1-yl-(ethoxy-L-alaninyl) phosphorochloridate [17b] ( 0.35 g , $1.04 \mathrm{mmol})$ in THF ( 3 mL ). The crude was purified first by silica gel $\mathrm{CC}\left(0-5 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ and secondly by prep. TLC $\left(10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to yield the title compound as a white solid $(0.015 \mathrm{~g}, 7 \%)$.
${ }^{31}$
P NMR ( $\mathbf{2 0 2 ~ M ~ M z , ~}_{\mathbf{~ M H}}^{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{P}}$ 3.98, 3.80.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.12$ ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 2$ ) 8.11-8.07 (m, 1H, Naph) 8.05 (s, 0.5 H, H2), 7.87-7.84 (m, 1H, Nap), 7.67 (d, $J=8.1$ 1H, Nap), 7.54-7.31 (m, 6H, Nap), 6.056.02 (m, 1H, H1'), 5.36-5.32 (m, 0.5H, H2'), 5.31-5.28 (m, 0.5H, H2'), 4.72-4.66 (m, 1H, H-3'), 4.55-4.49 (m, 1H, H5') 4.46-4.37 (m, 1H, H5'), 4.28-4.22 (m, 1H, H4'), 4.06-3.91 ( $\mathrm{m}, 2.5 \mathrm{H}, \mathrm{CHCH}_{3}$ L-Ala, $\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}$ ), 3.87-3.79 (m, $0.5 \mathrm{H}, \mathrm{CHCH}_{3}$ L-Ala), 1.28-1.25 (m, $1 \mathrm{H}, \mathrm{CHCH}_{3}$ L-Ala), 1.19-1.08 (m, 5H, $\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}, \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}$ ).
${ }^{13} \mathbf{C} \mathbf{N M R}\left(\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{C}} 174.97\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.9 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.80\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.9\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 156.28 (C-6), 154.09 (C2), 154.01 (C2), 151.63 (C4), 151.60 (C4), 147.98 (C8), 147.92 (C8), 139.82 (C-Ar), 139.79 (C-Ar), 136.22 (C-Ar), 128.81 (C-Ar), 128.79 (C-Ar), 127.83 (C-Ar), 127.78 (C-Ar), 127.72 (CH-Ar), 127.36 (CH-Ar), 127.34 (CH-Ar), 126.43 (CH-Ar), 126.39 (CH-Ar), 125.82 (CH-Ar), 122.73 (CH-Ar), 122.62 (CH-Ar), 119.53 (C5), 116.06 (CH-Ar), 116.04 (CH-Ar), 91.42 ( $\mathrm{Cl}^{\prime}$ ), 91.35 ( C 1 '), $84.34\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.1 \mathrm{~Hz}\right.$,
 $67.80\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.2 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.09\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.2 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $62.36\left(\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right), 62.23$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right), 51.55\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.45\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $20.54\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.0 \mathrm{~Hz}\right.$, $\mathrm{CHCH}_{3}$ L-Ala), 20.33 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.0 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ), $14.39\left(\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right)$.
$\mathbf{M S}\left(\mathbf{E S}+\right.$ ) $\mathbf{m} / \mathbf{z}$ found $629.14\left[\mathrm{M}+\mathrm{Na}^{+}\right], 645.11\left[\mathrm{M}+\mathrm{K}^{+}\right], \mathrm{C}_{25} \mathrm{H}_{28} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathrm{z}$ 606.95 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, two peaks with $\mathrm{tR} 14.43 \mathrm{~min}, \mathrm{tR} 13.73 \mathrm{~min}$.

## [71c] 8-chloroadenosine-5'-naphth-1-yl(cyclohexyloxy-L-alaninyl) phosphate



Prepared according to the general procedure $\mathbf{F}_{1}$ using 8chloroadenosine [67] ( $0.10 \mathrm{~g}, 0.33 \mathrm{mmol}$ ) in THF ( 7 mL ), $t \mathrm{BuMgCl}(1.1 \mathrm{M}$ in THF, $0.36 \mathrm{~mL}, 0.36 \mathrm{mmol})$, naphth-1-yl-(cyclohexyloxy-L-alaninyl) phosphorochloridate [17c] ( $0.41 \mathrm{~g}, 1.04 \mathrm{mmol}$ ) in THF ( 3 mL ). The crude was purified first by silica gel $\mathrm{CC}\left(0-5 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ and secondly by prep. TLC $(10 \%$ $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to yield the title compound as a white solid ( $0.02 \mathrm{~g}, 9 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 3.93, 3.90.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.13(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 2)$ 8.11-8.07 (m, 1H, Nap), 8.06 ( $\mathrm{s}, 0.5$ H, H2), 7.83 (d, $J=8.3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Nap}), 7.64$ (d, $J=8.3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Nap}), 7.52-7.39(\mathrm{~m}, 3 \mathrm{H}$, Nap), 7.37-7.30 (m, 1H, Nap), 6.06-6.02 (m, 1H, H1'), 5.35-5.31 (m, 0.5H, H2'), 5.305.27 ( m, 0.5 H, H2'), 4.70-4.65 (m, 1H, H3'), 4.65-4.38 (m, 3H, H5', CH( $\left.\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}$ ), 4.29-4.23 (m, 1H, H4'), 4.00-3.92 (m, 0.5H, $\mathrm{CHCH}_{3}$ L-Ala), 3.91-3.83 (m, 0.5H, $\mathrm{CHCH}_{3}$ L-Ala), 1.79-1.13 (m, 13H, $\left.\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$.
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 174.44$ (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.1 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}$ ), $174.28\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.1\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 156.24, (C6), 154.10 (C2), 154.03 (C2), 151.64 (C4), 151.59 (C4), 147.98 (C8), 147.93 (C8), 139.80 (C-Ar), 139.76 (C-Ar), 136.21 (C-Ar), 128.81 (C-Ar), 127.82 (CH$\mathrm{Ar}), 127.77$ (CH-Ar), 127.72 (CH-Ar), 127.36 (CH-Ar), 127.34 (CH-Ar), 126.44 (CH-Ar), 125.83 (CH-Ar), 122.76 (CH-Ar), 122.64 (CH-Ar), 119.51 (C5), 116.07 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}$, CH-Ar), $116.03\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 91.42\left(\mathrm{C} 1{ }^{\prime}\right), 91.28\left(\mathrm{C} 1{ }^{\prime}\right), 84.36\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.0\right.$ $\mathrm{Hz}, \mathrm{C} 4$ ' $), 84.30\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.0 \mathrm{~Hz}, \mathrm{C} 4^{\prime}\right), 74.96\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 74.93\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$,
 $\left(\mathrm{d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.6 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $51.71\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.69\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 32.44\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5}\right.$ cHex), $32.41\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 32.37\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 32.32\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 32.30$ $\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 26.37\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 26.34\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 24.63\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5}\right.$ cHex), $24.59\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right) 20.73\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.3 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 20.53\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=\right.$ $6.3 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ).

MS (ES+ $\mathbf{)} \mathbf{m} / \mathbf{z}$ found $683.19\left[\mathrm{M}+\mathrm{Na}^{+}\right], 699.17\left[\mathrm{M}+\mathrm{K}^{+}\right], \mathrm{C}_{29} \mathrm{H}_{34} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathbf{z}$ 660.19 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, two peaks with tR 17.36 min , tR 17.93 min .

## [71d] 8-chloroadenosine-5'-naphth-1-yl(pentyl-1-oxy-L-alaninyl) phosphate



Prepared according to the general procedure $\mathbf{F}_{1}$ using 8chloroadenosine [67] ( $0.10 \mathrm{~g}, 0.33 \mathrm{mmol}$ ) in THF ( 7 mL ), $t \mathrm{BuMgCl}(1.1 \mathrm{M}$ in THF, $0.36 \mathrm{~mL}, 0.36 \mathrm{mmol})$, naphth-1-yl-(pentyl-1-oxy-L-alaninyl) phosphorochloridate [17d] $(0.40 \mathrm{~g}, 1.04 \mathrm{mmol})$ in THF ( 3 mL ). The crude was purified first by silica gel $\mathrm{CC}\left(0-5 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ and secondly by prep. TLC $\left(10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to yield the title compound as a white solid ( $0.009 \mathrm{~g}, 4 \%$ ).
${ }^{31} \mathbf{P}-\mathrm{NMR}\left(\mathbf{2 0 2} \mathbf{~ M H z}, \mathrm{CD}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathrm{P}}$ 3.88, 3.89.
${ }^{\mathbf{1}} \mathbf{H}-\mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.13$ (s, $0.5 \mathrm{H}, \mathrm{H} 2$ ) 8.12-8.09 (m, 1H, Nap), 8.06 (s, 0.5 H, H2), 7.88-7.85 (m, 1H, Nap), 7.68 (d, $J=8.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Nap}), 7.55-7.48$ (m, 2H, Nap), 7.45-7.33 (m, 2H, Nap), 6.04 (d, $J=4.7 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1$ '), 6.03 (d, $J=4.7 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1$ '), 5.36-5.32 (m, 0.5H, H2'), 5.30-5.27 (m, 0.5 H, H2'), 4.69-4.64 (m, 1H, H3'), 4.55-4.49 (m, 1H, H5'), 4.45-4.36 (m, 1H, H5'), 4.27-4.22 (m, 1H, H4'), 4.02-3.83 (m, 3H, CHCH L-Ala, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 1.55-1.46 (m, 2H, $\mathrm{CHCH}_{3}$ L-Ala), 1.32-1.16 (m, 7H, $\mathrm{CHCH}_{3}$ L-Ala, $\quad \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n$-Pen), $0.86 \quad(\mathrm{t}, \quad J=6.9 \mathrm{~Hz}, \quad 1.5 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), $0.82\left(\mathrm{t}, J=6.9 \mathrm{~Hz}, 1.5 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen).
${ }^{13} \mathbf{C - N M R}\left(125 \mathrm{MHz}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathrm{C}} 175.04\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.86\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 156.29 (C6), 154.09 (C2), 154.01 (C2), 151.65 (C4), 151.61 (C4), 147.98 (C8), 147.92 (C8), 139.82 (C-Ar), 139.77 (C-Ar), 136.23 (C-Ar), 128.82 (CH-Ar), 128.79 (CHAr), 127.85 (C-Ar), 127.80 (C-Ar), 127.71 (C-Ar), 127.36 (CH-Ar), 126.33 (CH-Ar), 126.43 (CH-Ar), 126.42 (CH-Ar), 126.40 (CH-Ar), 125.83 (CH-Ar), 122.77 (CH-Ar), 122.64 (CH-Ar), 119.54, (C5), 116.11 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 116.03\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}\right.$, CH-Ar), $91.42\left(\mathrm{C} 1^{\prime}\right), 91.32\left(\mathrm{C} 1^{\prime}\right), 84.36\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.2 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $84.27\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.2 \mathrm{~Hz}\right.$, C4'), 72.75 ( $\mathrm{C}^{\prime}$ '), 72.62 ( $\mathrm{C}^{\prime}$ ), 71.66 ( $\left.\mathrm{C} 3^{\prime}\right), 71.16$ ( $\mathrm{C} 3^{\prime}$ ), 67.81 ( $\mathrm{d}^{2}{ }^{2} J_{\mathrm{C}-\mathrm{P}}=4.8 \mathrm{~Hz}, \mathrm{C} 5^{\prime}$ ), 67.11, (d, ${ }^{2} J_{\mathrm{C}-\mathrm{P}}=4.8 \mathrm{~Hz}, \quad \mathrm{C} 5$ '), $66.43 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen $), \quad 66.31$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen $), 51.58\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $51.52\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 30.76$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen $), \quad 30.47 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen), $\quad 29.29$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen), $29.07 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen), 23.73 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $23.31\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), 20.56 (d, ${ }^{3} J_{\mathrm{C} \text { - }}=7.2$ $\mathrm{Hz}, \mathrm{CHCH}_{3}$ L-Ala), 20.37 (d, ${ }^{3} J_{\text {C-P }}=7.2 \mathrm{~Hz}, \mathrm{CHCH}_{3}$ L-Ala), $14.25\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ $n$-Pen), $14.22\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen $)$.

MS (ES+) m/z found $671.19\left[\mathrm{M}+\mathrm{Na}^{+}\right], 687.16\left[\mathrm{M}+\mathrm{K}^{+}\right], \mathrm{C}_{28} \mathrm{H}_{34} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathrm{z}$ 649.03 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, two peaks with $\mathrm{tR} 18.44 \mathrm{~min}, \mathrm{tR} 17.79 \mathrm{~min}$.

## [71e] 8-chloroadenosine-5'-naphth-1-yl(ethyloxy-glycinyl) phosphate



Prepared according to the general procedure $\mathbf{F}_{\mathbf{1}}$ using 8chloroadenosine [67] ( $0.10 \mathrm{mg}, 0.33 \mathrm{mmol}$ ) in THF ( 7 mL ), $t \mathrm{BuMgCl}(1.1 \mathrm{M}$ in THF, $0.36 \mathrm{~mL}, 0.36 \mathrm{mmol})$, naphth-1-yl-(ethyloxy-glycinyl) phosphorochloridate [17i] ( $0.43 \mathrm{~g}, 1.33$ $\mathrm{mmol})$ in THF ( 3 mL ). The crude was purified first by silica gel CC $\left(0-6 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ and secondly by prep. TLC $\left(5 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to yield a white solid ( $0.007 \mathrm{~g}, 4 \%$ ).
${ }^{\mathbf{3 1}} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 4.96, 4.94.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.15-8.07(\mathrm{~m}, 1.5 \mathrm{H}, \mathrm{H} 2, \mathrm{H} 8 \mathrm{Nap}) 8.05(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 2)$, 7.89-7.85 (m, 1H, H5 Nap), 7.69-7.66 (m, 1H, H4 Nap), 7.55-7.33 (m, 4H, H2, H6, H7, H3 Nap), 6.04 (d, $\left.J=5.1 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{Hl}^{\prime}\right), 6.02\left(\mathrm{~d}, J=5.1 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H}{ }^{\prime}\right.$ ), $5.35-5.29$ (m, 1H, H2'), 4.69-4.66 (m, 0.5H, H3'), 4.64-4.61 (m, 0.5H, H3'), 4.56-4.50 (m, 1H, H5'), 4.49-4.43 (m, 1H, H5'), 4.27-4.22 (m, 1H, H4'), 4.10-4.00 (m, 2H, CH ${ }_{2} \mathrm{CH}_{3} \mathrm{Et}$ ), 3.73-3.62 (m, 2H, CH2 Gly), 1.21-1.13 (m, 3H, CH2 $\mathrm{CH}_{3} \mathrm{Et}$ ).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 175.02\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.92\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 156.99 (C6), 154.02 (C2), 153.99 (C2), 151.30 (C4), 151.22 (C4), 147.42 (C8), 147.36 (C8), 139.82 (C-Ar), 139.75 (C-Ar), 136.19 (C-Ar), 128.44 (C-Ar), 128.36 (C-Ar), 127.80 (C-Ar), 127.75 (C-Ar), 127.67 (CH-Ar), 127.36 (CH-Ar), 127.32 (CH-Ar), 126.45 (CH-Ar), 126.32 (CH-Ar), 125.79 (CH-Ar), 122.71 (CH-Ar), 122.62 (CH-Ar), 119.51 (C5), 116.05 (CH-Ar), 116.02 (CH-Ar), $91.40\left(\mathrm{Cl}^{\prime}\right), 91.35$ (C1'), $84.25\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.6 \mathrm{~Hz}\right.$, C 4 '), 84.22 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.6 \mathrm{~Hz}, \mathrm{C} 4$ '), 72.72 (C2'), 72.67 ( C 2 '), 71.64 ( $\mathrm{C}^{\prime}$ ), 71.39 ( $\mathrm{C}^{\prime}$ ), $67.78\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.09\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $62.35\left(\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right), 62.29$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right), 44.02\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $43.88\left(\mathrm{CH}_{2} \mathrm{Gly}\right), 14.22\left(\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right)$.

MS (ES-) m/z found $591.10\left[\mathrm{M}-\mathrm{H}^{+}\right], 627.08\left[\mathrm{M}+\mathrm{Cl}^{-}\right], \mathrm{C}_{24} \mathrm{H}_{26} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required m/z 592.93 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{ml} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, two peaks with $\mathrm{tR} 13.88, \mathrm{tR} 13.92$.

## [71f] 8-chloroadenosine-5'-naphth-1-yl(benzyloxy-glycinyl) phosphate



Prepared according to the general procedure $\mathbf{F}_{\mathbf{1}}$ using 8chloroadenosine [67] ( $0.15 \mathrm{~g}, 0.50 \mathrm{mmol}$ ) in THF ( 10 $\mathrm{mL}), t \mathrm{BuMgCl}(1.0 \mathrm{M}$ in THF, $0.55 \mathrm{~mL}, 0.55 \mathrm{mmol})$, naphth-1-yl-(benyloxy-glycinyl) phosphorochloridate [17j] ( $0.78 \mathrm{~g}, 2 \mathrm{mmol}$ ) in THF ( 5 mL ). The crude was purified first by silica gel $\mathrm{CC}\left(0-6 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ and secondly by prep. TLC $\left(5 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to yield the title compound as a white powder ( $0.023 \mathrm{~g}, 7 \%$ ).

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P NMR (202 MHz, CD $\mathbf{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 4.93$.
${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.13-8.06(\mathrm{~m}, 1.5 \mathrm{H}, \mathrm{H} 2, \mathrm{H} 8 \mathrm{Nap}), 8.05$ (s, $0.5 \mathrm{H}, \mathrm{H} 2$ ), 7.87-7.84 (m, 1H, H5 Nap), 7.68-7.64 (m, 1H, H4 Nap), 7.53-7.26 (m, 9H, H2, H6, H7, H3 Nap, Ph), 6.04 (d, $J=5.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), 6.02 (d, $J=5.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), 5.33-5.28 (m, $1 \mathrm{H}, \mathrm{H} 2$ '), $5.10-5.01$ ( $\mathrm{m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}$ ), 4.67-4.63 (m, 0.5H, H3'), 4.63-4.59 (m, 0.5H, H3'), 4.55-4.48 (m, 1H, H5'), 4.48-4.40 (m, 1H, H5'), 4.26-4.20 (m, 1H, H4'), 3.79-3.71 (m, 2H, $\mathrm{CH}_{2}$ Gly).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 172.09\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 156.25$ (C6), 154.01 (C2), 151.66 (C4), 151.60 (C4), 147.91 (C8), 148.85 (C8), 139.79 (C-Ar), 137.12 (C-Ar), 137.11 (C-Ar), 136.22 (C-Ar), 129.53 (CH-Ar), 129.30 (CH-Ar), 129.28 (CH-Ar), 129.26 (CH-Ar), 128.80 (CH-Ar), 128.79 (CH-Ar), 127.85 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=1.8 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), 127.79 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=1.8 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 127.69(\mathrm{CH}-\mathrm{Ar}), 127.38(\mathrm{CH}-\mathrm{Ar}), 127.36(\mathrm{CH}-\mathrm{Ar}), 126.44(\mathrm{CH}-$ $\mathrm{Ar})$, 126.39 ( $\mathrm{CH}-\mathrm{Ar}$ ), 125.89 (CH-Ar), 122.69 (CH-Ar), 122.65 (CH-Ar), 119.53 (C5), 119.52, (C5), 116.18 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 116.12\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 91.37$ (C1'), 91.17 ( $\mathrm{Cl}^{\prime}$ ), 84.45 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}, \mathrm{C} 4$ '), 84.39 ( $\mathrm{d},{ }^{3} \mathrm{~J}_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}, \mathrm{C} 4$ '), 2.75 ( C 3 '), 72.57 ( C 3 '), 71.64 ( C 2 '), 71.28 ( C 2 '), $67.90\left(\mathrm{C}^{\prime}\right), 67.82$ ( $\mathrm{C}^{\prime}$ ), 67.41 ( $\mathrm{CH}_{2} \mathrm{Ph}$ ), $67.37\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 43.91\left(\mathrm{CH}_{2} \mathrm{Gly}\right), 43.85\left(\mathrm{CH}_{2} \mathrm{Gly}\right)$.

MS (ES-) m/z found $653.19\left[\mathrm{M}-\mathrm{H}^{+}\right], 688.64\left[\mathrm{M}+\mathrm{Cl}^{-}\right], \mathrm{C}_{29} \mathrm{H}_{28} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required m/z 654.99 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, two peaks with $\mathrm{tR} 15.84 \mathrm{~min}, \mathrm{tR} 15.35 \mathrm{~min}$.

## [71g] 8-chloroadenosine-5'-naphth-1-yl(pentyl-1-oxy-glycinyl) phosphate



Prepared according to the general procedure $\mathbf{F}_{\mathbf{1}}$ using 8chloroadenosine [67] ( $0.13 \mathrm{~g}, 0.43 \mathrm{mmol}$ ) in THF ( 10 $\mathrm{mL}), t \mathrm{BuMgCl}(1.0 \mathrm{M}$ in THF, $0.47 \mathrm{~mL}, 0.47 \mathrm{mmol})$, naphth-1-yl-(pentyl-1-oxy-glycinyl) phosphorochloridate [ $\mathbf{1 7 k} \mathbf{k}](0.64 \mathrm{~g}, 1.72 \mathrm{mmol})$ in THF ( 5 mL ). The crude was purified first by silica gel $\mathrm{CC}\left(0-6 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ and secondly by prep. TLC $(7 \%$ $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ), to yield the product as a white solid ( $0.039 \mathrm{~g}, 15 \%$ ).
${ }^{\mathbf{3 1}} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 4.93$.
${ }^{1} \mathbf{H}$ NMR ( $\left.\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.15-8.07(\mathrm{~m}, 1.5 \mathrm{H}, \mathrm{H} 2, \mathrm{H} 8 \mathrm{Nap}), 8.05(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 2)$, 7.87-7.82 (m, 1H, H5 Nap), 7.68-7.64 (m, 1H, H4 Nap), 7.54-7.42 (m, 3H, H2, H6, H7 Nap), 7.35 (t, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3 \mathrm{Nap}), 6.04(\mathrm{~d}, J=4.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), $6.02(\mathrm{~d}, J=4.7 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{H} 1$ '), 5.37-5.25 (m, 1H, H2'), 4.69-4.65 (m, 0.5H, H3'), 4.64-4.60 (m, 0.5H, H3'), 4.58-4.50 (m, 1H, H5'), 4.50-4.41 (m, 1H, H5'), 4.29-4.22 (m, 1H, H4'), 4.03-9.94 (m, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n$-Pen), 3.75-3.64 (m, 2H, $\mathrm{CH}_{2}$ Gly), 1.57-1.47 (m, 2 H , $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 1.31-1.22 (m, 4H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 0.86 (t, $J=$ $6.9 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 172.29\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.4 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 172.30\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.4\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 156.26 (C6), 154.02 (C2), 151.67 (C4), 151.60 (C4), 147.95 (C8), 147.89 (C8), 139.77 (C-Ar), 136.22 (C-Ar), 128.82 (CH-Ar), 128.80 (CH-Ar), 127.85 (C-Ar), 127.80 (C-Ar), 127.72 (C-Ar), 127.70 (C-Ar), 127.37 (CH-Ar), 127.36 (CH-Ar), 126.45 (CH-Ar), 126.41 (CH-Ar), 125.89 (CH-Ar), 122.70 (CH-Ar), 122.65 (CH-Ar), 116.15 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.0$ $\mathrm{Hz}, \mathrm{CH}-\mathrm{Ar}$ ), 116.17 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.0 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), $91.38\left(\mathrm{C} 1^{\prime}\right), 91.17\left(\mathrm{C} 1^{\prime}\right), 84.47\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=\right.$ $7.5 \mathrm{~Hz}, \mathrm{C} 4$ '), $84.40\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), 72.78 ( $\mathrm{C}^{\prime}$ ), 72.57 ( $\left.\mathrm{C} 2^{\prime}\right), 71.66\left(\mathrm{C} 3^{\prime}\right), 71.27$ (C3'), $67.81\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.1 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.37\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.1 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), 67.41 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $67.35\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $43.83\left(\mathrm{CH}_{2} \mathrm{Gly}\right), 43.75$ ( $\mathrm{CH}_{2}$ Gly), $29.32\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $29.29\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), 29.07 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen), $29.04 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen $), \quad 23.33$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $14.28\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen).

MS (ES+) $\mathbf{m} / \mathbf{z}$ found $657.19\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{27} \mathrm{H}_{32} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 635.01$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \boldsymbol{\lambda}=254 \mathrm{~nm}$, two peaks with $\mathrm{tR} 17.91 \mathrm{~min}, \mathrm{tR} 17.39 \mathrm{~min}$.
[71h] 8-chloroadenosine-5'-naphth-1-yl(cyclohexyloxy-L-alaninyl) phosphate


Prepared according to the general procedure $\mathbf{F}_{\mathbf{1}}$ using 8chloroadenosine [67] ( $0.13 \mathrm{~g}, 0.43 \mathrm{mmol}$ ) in THF ( 10 $\mathrm{mL}), t \mathrm{BuMgCl}(1.0 \mathrm{M}$ in THF, $0.47 \mathrm{~mL}, 0.47 \mathrm{mmol}$ ), naphth-1-yl-(cyclohexyloxy-glycinyl)dichlorophosphate [171] ( $0.50 \mathrm{~g}, 1.29 \mathrm{mmol})$ in THF ( 5 mL ). The crude was purified first by silica gel $\mathrm{CC}\left(0-6 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ and secondly by prep. TLC $\left(7 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$, to yield the product as a white solid ( 0.012 g, 4\%).

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P NMR (202 MHz, CD $\mathbf{3}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 4.96, 4.94.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.15-8.04 (m, $2 \mathrm{H}, \mathrm{H} 2, \mathrm{H} 8 \mathrm{Nap}$ ), 7.89-7.85 (m, 1H, H5 Nap), 7.70-7.65 (m, 1H, H4 Nap), 7.55-7.44 (m, 3H, H2, H6, H7 Nap), 7.42-7.33 (m, 1H, H3 Nap), $6.04(\mathrm{~d}, J=6.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1$ '), $6.02(\mathrm{~d}, J=6.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1$ '), $5.32-5.28(\mathrm{~m}$, $1 \mathrm{H}, \mathrm{H} 2$ '), 4.70-4.63 (m, 1.5H, H3', $\left.\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), ~ 4.61-4.58$ (m, 0.5H, H3'), 4.57-4.50 (m, 1H, H5'), 4.49-4.42 (m, 1H, H5'), 4.27-4.23 (m, 1H, H4'), 3.72-3.62 (m, 2H, CH2 Gly), 1.81-1.64 (m, 4H, $\left.\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 1.56-1.49\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 1.39-1.22$ ( $\mathrm{m}, 5 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}$ ).
${ }^{13} \mathbf{C}$ NMR ( $\left.\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{C}} 172.67\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 156.14$ (C6), 153.82 (C2), 151.66 (C4), 151.59 (C4), 147.96 (C8), 147.87 (C8), 139.84 (C-Ar), 128.82 (C-Ar), 128.80 (C-Ar), 127.81 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 127.71\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 126.45$ (CH-Ar), 126.41 (CH-Ar), 125.89 (CH-Ar), 122.70 (CH-Ar), 122.65 (CH-Ar), 116.16 (CH-Ar), 116.14 (CH-Ar), 91.38 ( $\mathrm{Cl}^{\prime}$ ), 91.14 ( $\mathrm{Cl}^{\prime}$ ), 84.53 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.2 \mathrm{~Hz}, \mathrm{C} 4$ '), 84.43 $\left(\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.2 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $75.08\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 75.02\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 72.76(\mathrm{C} 2$ '), $72.55\left(\mathrm{C} 2^{\prime}\right), 71.64\left(\mathrm{C} 3^{\prime}\right), 71.27\left(\mathrm{C} 3^{\prime}\right), 67.79\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.39\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.0\right.$ Hz, C5'), $44.05\left(\mathrm{CH}_{2}\right.$ Gly $), 43.99\left(\mathrm{CH}_{2}\right.$ Gly $), 32.48\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 32.45\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5}\right.$ cHex), $26.36\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 26.35\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 24.69\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$.
$\mathbf{M S}\left(\mathbf{E S}+\right.$ ) $\mathbf{m} / \mathbf{z}$ found $669.01\left[\mathrm{M}+\mathrm{Na}^{+}\right], 685.12\left[\mathrm{M}+\mathrm{K}^{+}\right], \mathrm{C}_{28} \mathrm{H}_{32} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathbf{z}$ 647.02 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, two peaks with tR 17.52 min , tR 17.00 min .

## [72] 2',3'-Isopropylidene-8-chloroadenosine



8-Chloroadenosine [67] ( $0.40 \mathrm{~g}, 1.33 \mathrm{mmol}$ ) was suspended in acetone $(30 \mathrm{~mL})$ and $\mathrm{HClO}_{4}(70 \%, 0.2 \mathrm{~mL})$ was added dropwise. The mixture was stirred at rt for 30 minutes and then neutralised by addition of $\mathrm{NaHCO}_{3}$ (saturated aqueous solution). The mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( $3 \times 30 \mathrm{~mL}$ ); the organics were collected and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated. The crude was purified by flash silica gel chromatography (gradient eluent system $0-3 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ), to afford the title compound as a white foam $(0.36 \mathrm{~g}, 79 \%)$.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}$, DMSO-d6) $\boldsymbol{\delta}_{\mathbf{H}} 8.18$ (s, 1H, H2) 7.53 (br s, 2H, NH ${ }_{2}$ ), 6.05 (d, $J=2.5$ $\mathrm{Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), 5.68 (dd, $J=6.5 \mathrm{~Hz}, 2.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2$ '), 5.07 (t, $\left.J=6.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{OH} 5^{\prime}\right), 5.04$ (dd, $\left.J=6.5 \mathrm{~Hz}, 3.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3^{\prime}\right)$, 4.17-4.14 (m, 1H, H4'), 3.53-3.48 (m, 1H, H5'), 3.453.41 (m, 1H, H5'), $1.55\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.34\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.

## [74a] 8-chloroadenosine-5'-naphth-1-yl(benzyloxy-L-leucinyl) phosphate



Prepared according to the general procedure $\mathbf{F}_{\mathbf{1}}$ using 2',3'-O-isopropylidene-8-chloroadenosine [72] ( 0.20 g , 0.59 mmol ), in THF ( 5 mL ), $t \mathrm{BuMgCl}$ ( 1.0 M in THF, $3 \mathrm{eq}, 1.76 \mathrm{~mL}, 1.76 \mathrm{eq})$ and naphth-1-yl-(benzyloxy-Lleucinyl)dichlorophosphate [17e] ( $0.80 \mathrm{~g}, 1.77 \mathrm{mmol}$ ) in THF ( 3 mL ). Purification by silica gel CC $(0-4 \%$ $\left.\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ gave a residue that was stirred at rt in $\mathrm{HCOOH} / \mathrm{H}_{2} \mathrm{O}(3 / 2,10 \mathrm{~mL})$ overnight. The solvents were evaporated in vacuo and the crude purified first by silica gel $\mathrm{CC}\left(3-6 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ and secondly by prep. TLC $\left(10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to yield the product as a white foam ( $0.118 \mathrm{~g}, 26 \%$ over two steps).
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P NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 4.13,3.95$.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.10(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 2)$ 8.09-8.05 (m, 1H, H8 Nap), 8.04 (s, $0.5 \mathrm{H}, \mathrm{H} 2$ ), 7.83-7.79 (m, 1H, H5 Nap), 7.62 (d, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4 \mathrm{Nap}$ ), 7.49- 7.38 (m, 3H, H2, H6, H7 Nap), 7.32-7.19 (m, 6H, H3 Nap, Ph), 6.07-6.02 (m, 1H, H1'), 5.36-5.32 ( $\mathrm{m}, 0.5 \mathrm{H}, \mathrm{H} 2$ '), $5.29-5.25\left(\mathrm{~m}, 0.5 \mathrm{H}, \mathrm{H} 2\right.$ '), $5.02-4.91\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.67-4.60(\mathrm{~m}, 1 \mathrm{H}$, H3'), 4.55-4.41 (m, 2.5H, H5'), 4.40-4.33 (m, 0.5H, H5'), 4.29-4.22 (m, 1H, H4'), 3.99$3.86\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 1.65-1.54 (m, $2.5 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 1.51-1.31 (m, $0.5 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), $0.80-0.62\left(\mathrm{~m}, 6 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}-\right.$ Leu).
$\left.{ }^{13} \mathbf{C ~ N M R ~ ( 1 2 5 ~ M H z , ~ C D} \mathbf{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{C}} 174.96\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.5 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.94\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.5\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 156.19 (C6), 156.17 (C6), 154.05 (C2), 154.02 (C2), 151.64 (C4), 151.57 (C4), 147.95 (C8), 147.89 (C8), 139.79 (C-Ar), 139.71 (C-Ar), 137.07 (C-Ar), 137.03 (C-Ar), 136.20 (C-Ar), 136.18 (C-Ar), 129.52 (CH-Ar), 129.50 (CH-Ar), 129.35 (CH-Ar), 129.33 (CH-Ar), 129.32 (CH-Ar), 129.28 (CH-Ar), 128.83 (CH-Ar), 128.79 (CH-Ar), 127.81 (d, $\left.{ }^{2} J_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 127.77\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 127.71(\mathrm{CH}-\mathrm{Ar}), 127.40(\mathrm{CH}-\mathrm{Ar})$, 127.35 (CH-Ar), 126.43 (CH-Ar), 125.88 (CH-Ar), 125.82 (CH-Ar), 122.79 (CH-Ar), 122.66 (CH-Ar), 119.51 (C5), 119.49 (C5), 116.24 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), 115.92 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 91.40(\mathrm{C} 1 ’), 91.23\left(\mathrm{C} 1\right.$ '), $84.53\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.2 \mathrm{~Hz}, \mathrm{C} 4\right.$ ) $), 84.33(\mathrm{~d}$, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.2 \mathrm{~Hz}, \mathrm{C} 4$ '), $72.83\left(\mathrm{C} 2^{\prime}\right), 72.65\left(\mathrm{C} 2^{\prime}\right), 71.75\left(\mathrm{C} 3^{\prime}\right), 71.48\left(\mathrm{C} 3\right.$ '), $68.05\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=\right.$ $5.3 \mathrm{~Hz}, \mathrm{C} 5$ '), $68.01\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.3 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.91\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.83\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.62\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}\right.$ $=5.5 \mathrm{~Hz}, \mathrm{C} 5 '), 67.57\left(\mathrm{~d}^{2}{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.5 \mathrm{~Hz}, \mathrm{C} 5\right.$ ' $), 54.68\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu $), 54.64$ $\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $44.09\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.9 \mathrm{~Hz}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 43.09 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.9 \mathrm{~Hz}, \quad \mathrm{CHCH} 2 \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu $), 25.57\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 25.37 $\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}-\mathrm{Leu}\right), 23.09\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}-\mathrm{Leu}\right), 23.05\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}\right.$ Leu), $22.04\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $21.92\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}-\mathrm{Leu}\right)$.
$\mathbf{M S}\left(\mathbf{E S}+\right.$ ) $\mathbf{m} / \mathbf{z}$ found $733.20\left[\mathrm{M}+\mathrm{Na}^{+}\right], 749.19\left[\mathrm{M}+\mathrm{K}^{+}\right], \mathrm{C}_{33} \mathrm{H}_{36} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathbf{z}$ 711.10 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \boldsymbol{\lambda}=254 \mathrm{~nm}$, two peaks with $\mathrm{tR} 19.33 \mathrm{~min}, \mathrm{tR} 18.93 \mathrm{~min}$.
[74b] 8-chloroadenosine-5'-O-naphth-1-yl-(pentyl-1-oxy-L-leucinyl) phosphate
Prepared according to the general procedure $\mathbf{F}_{\mathbf{1}}$ using $2^{\prime}, 3$ '-
 $O$-isopropylidene-8-chloroadenosine [72] (0.18 g, 0.53 mmol ), in THF ( 5 mL ), $t \mathrm{BuMgCl}(1.0 \mathrm{M}$ in THF, 1.59 mL , $1.59 \mathrm{eq})$ and naphth-1-yl-(pentyl-1-oxy-L-leucinyl) phosphorochloridate [17f] ( $0.68 \mathrm{~g}, 1.59 \mathrm{mmol}$ ) in THF (3 $\mathrm{mL})$. Purification by silica gel $\mathrm{CC}\left(0-4 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ gave a residue that was stirred at rt in $\mathrm{HCOOH} / \mathrm{H}_{2} \mathrm{O}(3 / 2,10 \mathrm{~mL})$ overnight. The solvents were evaporated in vacuo and the crude purified first by silica gel CC (3-6\% $\left.\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ and secondly by prep. TLC $\left(10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to yield the product as a white foam ( $0.13 \mathrm{~g}, 36 \%$ over two steps).
${ }^{31} \mathbf{P}$ NMR (202 MHz, CD $\left.\mathbf{D}_{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 4.06, 4.02.
${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.12$ (s, $0.5 \mathrm{H}, \mathrm{H} 2$ ) 8.11-8.06 (m, $\left.1 \mathrm{H}, \mathrm{H} 8 \mathrm{Nap}\right), 8.06$ (s, 0.5H, H2), 7.84-7.79 (m, 1H, H5 Nap), 7.65-7.60 (m, 1H, H4 Nap), 7.51-. 40 (m, 3H, H2, H6, H7 Nap), 7.32 (t, $J=8.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3 \mathrm{Nap}$ ), 6.07-6.03 (m, 1H, H1'), 5.37-5.33 (m, 0.5H, H2'), 5.30-5.26 (m, 0.5H, H2'), 4.69-4.62 (m, 1H, H-3'), 4.57-4.46 (m, 1.5H, H5'), 4.42-4.36 (m, 0.5H, H5'), 4.32-4.24 (m, 1H, H4'), 3.95-3.89 (m, 2H, CH ${ }_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ $n$-Pen), $\quad 3.89-3.83\left(\mathrm{~m}, \quad 0.5 \mathrm{H}, \quad \mathrm{CHCH} \mathrm{CH}_{2}\left(\mathrm{CH}_{3}\right)_{2} \quad\right.$ L-Leu), $1.70-1.57 \quad(\mathrm{~m}, \quad 0.5 \mathrm{H}$, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 1.52-1.35 (m, $5 \mathrm{H}, \quad \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \quad$ L-Leu, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n$-Pen, $\quad \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \quad$ L-Leu), 1.26-1.15 (m, 4 H , $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), $0.85-0.80\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $0.80-0.68$ (m, $6 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu).
${ }^{13} \mathbf{C} \mathbf{N M R}\left(\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{C}} 175.27\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.0 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 175.08\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.0\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 156.21 (C6), 156.18 (C6), 154.04 (C2), 151.64 (C4), 151.58 (C4), 148.00 (C8), 147.94 (C8) 139.79 (C-Ar), 139.73 (C-Ar), 136.22 (C-Ar), 136.19 (C-Ar), 129.52 (CHAr), 129.35 (CH-Ar), 129.31 (CH-Ar), 128.84 (CH-Ar), 128.80 (CH-Ar), 127.84 (CHAr), 127.79 ( $\mathrm{CH}-\mathrm{Ar}$ ), 127.73 ( $\mathrm{CH}-\mathrm{Ar}$ ), 127.71 (CH-Ar), 127.36 (CH-Ar), 127.32 (CH-Ar), 125.84 (CH-Ar), 125.80 (CH-Ar), 122.77 (CH-Ar), 122.68 (CH-Ar), 119.50, (C5), 116.13 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 115.89\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 91.42$ (C1'), 91.25 (C1'), 84.58 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.2 \mathrm{~Hz}, \mathrm{C} 4$ '), 84.34 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.2 \mathrm{~Hz}, \mathrm{C} 4$ '), 72.83 (C2'), 72.67 (C2'), 71.79 (C3'), 71.56 (C3'), $68.01\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.5 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), 67.63 , ( $\mathrm{d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.5 \mathrm{~Hz}, \mathrm{C} 5$ '), $66.35\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen $), \quad 66.29\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen $), \quad 54.64$ $\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \quad \mathrm{~L}\right.$-Leu), $44.30 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen $), 44.24$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen), $44.15 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen $), \quad 44.09$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $29.27\left(\mathrm{~d},{ }^{3} J_{\text {C-P }}=7.0 \mathrm{~Hz}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 29.11 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.0 \mathrm{~Hz}, \quad \mathrm{CHCH} 2 \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu $), 25.65\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 25.47 $\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \quad \mathrm{~L}-\mathrm{Leu}\right), \quad 23.31 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen $), \quad 23.28$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $23.11\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $23.09\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $22.13\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \text { L-Leu }\right)^{2} 22.03\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 14.33 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen).
MS (ES+) $\mathbf{m} / \mathbf{z}$ found $713.24\left[\mathrm{M}+\mathrm{Na}^{+}\right], 729.20\left[\mathrm{M}+\mathrm{K}^{+}\right], \mathrm{C}_{31} \mathrm{H}_{40} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathbf{z}$ 691.11 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, two peaks with tR 21.33 min , tR 20.87 min .

## [74c] 8-chloroadenosine 5'-O-naphth-1yl-(cyclohexyloxy-L-leucyn) phosphate



Prepared according to the general procedure $\mathbf{F}_{\mathbf{1}}$ using 2',3'-$O$-isopropylidene-8-chloroadenosine [72] ( $0.064 \mathrm{~g}, 0.19$ $\mathrm{mmol})$, in THF ( 2 mL ), $t \mathrm{BuMgCl}(1.0 \mathrm{M}$ in THF, 0.57 mL , $0.57 \mathrm{mmol})$ and naphth-1-yl-(cyclohexyloxy-L-leucinyl) phosphorochloridate $[\mathbf{1 7 g}](0.25 \mathrm{~g}, 0.57 \mathrm{mmol})$ in THF ( 1 $\mathrm{mL})$. Purification by silica gel $\mathrm{CC}\left(0-3 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ gave a residue that was stirred at rt in $\mathrm{HCOOH} / \mathrm{H}_{2} \mathrm{O}(3 / 2,10 \mathrm{~mL})$ overnight. The solvents were evaporated in vacuo and the crude purified first by silica gel CC $\left(3-6 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ and secondly by prep. TLC $\left(10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to yield the product as a white foam $(0.01 \mathrm{~g}, 8 \%$ over two steps $)$.

P NMR (202 MHz, CD $\mathbf{3}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 4.12, 4.02.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.13$ ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 2$ ) 8.12-8.07 (m, 1H, H8 Nap), 8.07 (s, 0.5 H, H2), 7.89-7.84 (m, 1H, H5 Nap), 7.69-7.64 (m, 1H, H4 Nap), 7.54-7.40 (m, 3H, H2, H6, H7 Nap), 7.37-7.32 (m, 1H, H3 Nap), 6.06-6.02 (m, 1H, H1'), 5.36-5.32 (m, 0.5H, H2'), 5.29-5.25 (m, 0.5H, H2'), 4.67-4.45 (m, 3.5H, H3', H5', $\left.\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), ~ 4.41-4.34$ ( $\mathrm{m}, 0.5 \mathrm{H}, \mathrm{H} 5$ '), 4.30-4.21 (m, 1H, H4'), 3.91-3.79 (m, 1H, CHCH $\mathrm{CH}^{2}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 1.741.22 ( $\mathrm{m}, ~ 13 \mathrm{H}, \quad \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu, $\left.\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$, $0.85-0.70(\mathrm{~m}, 6 \mathrm{H}$, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu).
$\left.{ }^{13} \mathbf{C ~ N M R ~ ( 1 2 5 ~ M H z , ~ C D} \mathbf{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathrm{C}} 174.61\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.9 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.43\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.9\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 156.27 (C6), 156.25 (C6), 154.03 (C2), 151.67 (C4), 151.60 (C4), 147.95 (C8) 139.79 (C-Ar), 139.72 (C-Ar), 136.25 (C-Ar), 136.23 (C-Ar), 128.82 (C-Ar), 128.79 (CAr), $127.84\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 127.70\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 127.33$ (CH-Ar), 127.29 (CH-Ar), 126.40 (CH-Ar), 125.81 (CH-Ar), 125.76 (CH-Ar), 122.79 (CH-Ar), 122.69 (CH-Ar), 119.53 (C5), 116.12 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 115.85\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}\right.$, CH-Ar), 91.43 (C1'), $91.24\left(\mathrm{Cl}^{\prime}\right), 84.60\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.0 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $84.37\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.0 \mathrm{~Hz}\right.$, C4'), 72.78 ( $\mathrm{C}^{\prime}$ ), 72.61 ( $\mathrm{C}^{\prime}$ ), 71.76 ( $\left.\mathrm{C} 3^{\prime}\right), 71.53\left(\mathrm{C} 3^{\prime}\right), 68.00\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.61,\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.4 \mathrm{~Hz}, \mathrm{C}{ }^{\prime}\right), 54.76\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 54.71\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu $)$, $44.34\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.3 \mathrm{~Hz}, \quad \mathrm{CHCH} \mathrm{H}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $44.15\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.3 \mathrm{~Hz}\right.$, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), $32.40\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \text { cHex }\right)^{2} 32.35\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 32.32$ $\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$, $26.37\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$, $26.35\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 25.66\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5}\right.$ cHex), $25.48 \quad\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \quad\right.$ cHex $), \quad 24.62 \quad\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \quad\right.$ L-Leu), 24.57 $\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}-\mathrm{Leu}\right), 23.02\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}-\mathrm{Leu}\right), 23.01\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}\right.$ Leu), $22.11\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $21.99\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu $)$.

MS (ES-VE) $\mathbf{m} / \mathbf{z}$ found $701.21\left[\mathrm{M}-\mathrm{H}^{+}\right], 737.19\left[\mathrm{M}+\mathrm{Cl}^{-}\right], \mathrm{C}_{32} \mathrm{H}_{40} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathbf{z}$ 702.2 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, two peaks with $\mathrm{tR} 20.53 \mathrm{~min}, \mathrm{tR} 20.00 \mathrm{~min}$.

## [74d] 8-chloroadenosine-5'-O-naphth-1-yl-(ethyloxy-L-leucinyl) phosphate



Prepared according to the general procedure $\mathbf{F}_{1}$ using $2^{\prime}, 3^{\prime}$ -$O$-isopropylidene-8-chloroadenosine [72] ( $0.08 \mathrm{~g}, 0.23$ $\mathrm{mmol})$ in THF $(15 \mathrm{~mL}), t \mathrm{BuMgCl}(1.0 \mathrm{M}$ in THF, 0.69 $\mathrm{mL}, 0.69 \mathrm{mmol}$ ) and naphth-1-yl-(ethyloxy-L-leucinyl) phosphorochloridate [17h] ( $0.25 \mathrm{~g}, 0.69 \mathrm{mmol})$ in THF $(1.5 \mathrm{~mL})$. Purification by silica gel $\mathrm{CC}\left(0-3 \% \mathrm{CH}_{3} \mathrm{OH} /\right.$ $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ gave a residue that was stirred at rt in $\mathrm{HCOOH} / \mathrm{H}_{2} \mathrm{O}(3 / 2,10 \mathrm{~mL})$ overnight. The solvents were evaporated in vacuo and the crude purified by prep. TLC ( $10 \%$ $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to yield a white solid ( $0.03 \mathrm{~g}, 35 \%$ over two steps).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 4.11, 4.04.
${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.12(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 2)$ 8.10-8.06 (m, 1H, H8 Nap), 8.05 ( s , 0.5H, H2), 7.86-7.82 (m, 1H, H5 Nap), 7.67-7.63 (m, 1H, H4 Nap), 7.53-7.39 (m, 3H, H2, H6, H7 Nap), 7.36-7.31 (m, 1H, H3 Nap), 6.06-6.03 (m, 1H, H1'), 5.36-5.33 (m, 0.5H, H2'), 5.30-5.27 (m, 0.5H, H2'), 4.68-4.62 (m, 1H, H3'), 4.57-4.46 (m, 1.5H, H5'), 4.414.35 (m, 0.5H, H5'), 4.30-4.23 (m, 1H, H4'), 4.02-3.92 (m, 2H, CH ${ }_{2} \mathrm{CH}_{3} \mathrm{Et}$ ), 3.91-3.81 (m, $1 \mathrm{H}, \mathrm{CHCH} 2 \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 1.70-1.61 (m, $0.5 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 1.50-1.35 (m, 2.5H, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 1.16-1.08 (m, $3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}$ ), 0.84-0.68 (m, 6 H , $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 175.18\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.7 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.97\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.7\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 156.26 (C6), 156.22 (C6), 154.02 (C2), 151.66 (C4), 151.60 (C4), 147.98 (C8), 148.93 (C8), 139.77 (d, $\left.{ }^{2} J_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 136.22\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 128.81$ (CAr), 128.78 (C-Ar), 127.84 (C-Ar), 127.79 (C-Ar), 127.72 (C-Ar), 127.69 (CH-Ar), 127.34 (CH-Ar), 127.30 (CH-Ar), 126.40 (CH-Ar), 125.81 (CH-Ar), 125.78 (CH-Ar), 122.75 (CH-Ar), 122.66 (CH-Ar), 119.52 (C5), 116.08 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), 115.88 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}$ $=3.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}), 91.42\left(\mathrm{C} 1\right.$ '), $91.27\left(\mathrm{C} 1\right.$ '), $84.58\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.9 \mathrm{~Hz}, \mathrm{C} 4{ }^{\prime}\right), 84.33\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=\right.$ $8.9 \mathrm{~Hz}, \mathrm{C} 4$ '), $72.77\left(\mathrm{C} 2^{\prime}\right), 72.64\left(\mathrm{C}^{\prime}\right)$, $71.73\left(\mathrm{C} 3^{\prime}\right), 71.52\left(\mathrm{C} 3\right.$ '), $67.96\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.4 \mathrm{~Hz}\right.$, $\mathrm{C}^{\prime}$ ) , 67.58, (d, ${ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.4 \mathrm{~Hz}, \mathrm{C} 5$ '), $62.23\left(\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right), 62.17\left(\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right), 54.58$
$\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}-\mathrm{Leu}\right), 44.17\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.3 \mathrm{~Hz}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 43.97 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.3 \mathrm{~Hz}, \quad \mathrm{CHCH} 2 \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), $25.60\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 25.41 $\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}-\mathrm{Leu}\right), 23.09\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}-\mathrm{Leu}\right), 21.98\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}\right.$ Leu), $21.87\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $14.40\left(\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right)$.

MS (ES -VE) m/z found $647.22\left[\mathrm{M}-\mathrm{H}^{+}\right], 683.14\left[\mathrm{M}+\mathrm{Cl}^{-}\right], \mathrm{C}_{28} \mathrm{H}_{34} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathrm{z}$ 649.03 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, two peaks with $\mathrm{tR} 17.88 \mathrm{~min}, \mathrm{tR} 17.43 \mathrm{~min}$.
[74e] 8-chloroadenosine-5'-O-naphth-1-yl-(benzyloxy-L-phenylalaninyl)phosphate


Prepared according to the general procedure $\mathbf{F}_{\mathbf{1}}$ using 2',3'-O-isopropylidene-8-chloroadenosine [72] (0.10 $\mathrm{g}, 0.29 \mathrm{mmol}$ ) in THF ( 3 mL ), $t \mathrm{BuMgCl}(1.0 \mathrm{M}$ in THF, $0.88 \mathrm{~mL}, \quad 0.88 \mathrm{mmol}$ ) and naphth-1-yl-(benzyloxy-L-phenylalaninyl)phosphorochloridate [ $\mathbf{1 7 m}](0.42 \mathrm{~g}, 0.87 \mathrm{mmol})$ in THF $(1.5 \mathrm{~mL})$. Purification by silica gel CC $\left(0-3 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ gave a residue that was stirred at rt in $\mathrm{HCOOH} / \mathrm{H}_{2} \mathrm{O}(3 / 2,10 \mathrm{~mL})$ overnight. The solvents were evaporated in vacuo and the crude purified by prep. TLC $\left(10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to yield the title compound as a white solid ( $0.012 \mathrm{~g}, 6 \%$ over two steps).
${ }^{31}$
P NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{P}}$ 3.66, 3.46.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.08$ ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 2$ ), 8.03 ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 2$ ), $8.00-7.97(\mathrm{~m}, 1 \mathrm{H}$, H8 Naph), 7.85-7.82 (m, 1H, H5 Nap), 7.64-7.61 (m, 1H, H4 Nap), 7.52-7.47 (m, 1H, H2 Nap), 7.44-7.38 (m, 1H, H6 Nap), 7.28-7.20 (m, 5H, Ph), 7.19-7.11 (m, 5H, Ph), 7.05-7.02 (m, 1H, H7 Nap), 7.00-6.96 (m, 1H, H3 Nap), 6.02-6.00 (m, 1 H, H1'), 5.32-5.25 (m, 1H, H2'), 4.98-4.88 (m, 2H, CH2Ph), 4.58-4.54 (m, 0.5H, H3'), 4.52-4.49 (m, 0.5H, H3'), 4.34-4.10 (m, 4H, H4', H5', $\mathrm{CHCH}_{2}$ Ph L-Phe), 3.01-2.90 (m, 1H, CHCH ${ }_{2} \mathrm{Ph}$ L-Phe), 2.832.76 (m, 1H, $\mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 173.73\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.6 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 173.65\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.6\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 156.23 (C6), 154.08 (C2), 154.02 (C2), 151.68 (C4), 151.60 (C4), 147.90 (C8), 147.78 (C8), 139.79 (C-Ar), 139.79 (C-Ar), 137.75 (C-Ar), 136.90 (C-Ar), 136.87 (C-Ar), 136.18 (C-Ar), 136.15 (C-Ar), 130.48 (C-Ar), 129.49 (CH-Ar), 129.44 (CH-Ar), 129.42 (CH-Ar), 129.39 (CH-Ar), 129.28 (CH-Ar), 129.23 (CH-Ar), 128.77 (C-Ar), 128.73 (CAr), 127.91 (C-Ar), 127.87 (C-Ar), 127.78 (CH-Ar), 127.76 (CH-Ar), 127.73 (CH-Ar),
127.71 (CH-Ar), 128.77 (C-Ar), 128.73 (CH-Ar), 126.40 (CH-Ar), 125.80 (CH-Ar), 125.74 (CH-Ar), 122.78 (CH-Ar), 122.71 (CH-Ar), 119.54 (C5), 119.50 (C5), 115.98 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 115.95\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 91.33\left(\mathrm{Cl}\right.$ '), $91.13\left(\mathrm{Cl}{ }^{\prime}\right), 84.38$ (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.1 \mathrm{~Hz}, \mathrm{C} 4$ '), $84.29\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.1 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $72.63\left(\mathrm{C} 2\right.$ '), 72.51 ( $\mathrm{C}^{\prime}$ ), 71.66 ( C 3 '), $71.33(\mathrm{C} 3 '), 67.96\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.88\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.71\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.5 \mathrm{~Hz}, \mathrm{C}{ }^{\prime}\right), 67.22$ (d, ${ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.5 \mathrm{~Hz}, \mathrm{C} 5$ '), $57.77\left(\mathrm{CHCH}_{2} \mathrm{Ph}\right.$ L-Phe $), 57.70\left(\mathrm{CHCH}_{2} \mathrm{Ph}\right.$ L-Phe $), 41.05\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}\right.$ $=7.3 \mathrm{~Hz}, \mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe), 40.96 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.3 \mathrm{~Hz}, \mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe).

MS (ES+) $\mathbf{m} / \mathbf{z}$ found $767.21\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{36} \mathrm{H}_{34} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 745.12$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, two peaks with $\mathrm{tR} 20.48 \mathrm{~min}, \mathrm{tR} 20.03 \mathrm{~min}$.
[74f] 8-chloroadenosine-5'-O-naphth-1-yl-(pentyl-1-oxy-L-phenylalaninyl)phosphate


Prepared according to the general procedure $\mathbf{F}_{1}$ using $2^{\prime}, 3^{\prime}$ -$O$-isopropylidene-8-chloroadenosine [72] (0.10 g, 0.29 mmol ) in THF ( 3 mL ), $t \mathrm{BuMgCl}(1.0 \mathrm{M}$ in THF, 0.88 mL , 0.88 mmol ) and naphth-1-yl-(pentyl-1-oxy-Lphenylalaninyl)phosphorochloridate $[\mathbf{1 7 n}](0.40 \mathrm{~g}, 0.87$ mmol ) in THF ( 1.5 mL ). Purification by silica gel CC ( $0-$ $3 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) gave a residue that was stirred at rt in $\mathrm{HCOOH} / \mathrm{H}_{2} \mathrm{O}(3 / 2,10 \mathrm{~mL})$ overnight. The solvents were evaporated in vacuo and the crude purified by prep. TLC $\left(10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to yield the title compound as a white solid ( $0.02 \mathrm{~g}, 10 \%$ over two steps).
31
P NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 3.72, 3.39.
${ }^{1} \mathbf{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.11(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 2), 8.05(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 2), 8.04-8.00(\mathrm{~m}, 1 \mathrm{H}$, H8 Nap), 7.87-7.83 (m, 1H, H5 Nap), 7.67-7.61 (m, 1H, H4 Nap), 7.54-7.43 (m, 2H, H2, H6 Nap), 7.32-7.26 (m, 2H, Ph), 7.22-7.12 (m, 3H, Ph), 7.11-7.08 (m, 1H, H7 Nap), 7.077.04 (m, 1H, H3 Nap), 6.04-6.00 (m, 1H, H1'), 5.32-5.26 (m, 1H, H2'), 4.58-4.54 (m, $0.5 \mathrm{H}, \mathrm{H} 3$ '), 4.53-4.50 (m, 0.5H, H3'), 4.37-4.31 (m, 0.5H, H5'), 4.29-4.24 (m, 0.5H, H5'), 4.18-4.05 (m, 3H, H5', H4', $\mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe), 3.90-3.79 (m, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n-$ Pen), 3.00-2.90 (m, 1H, CHCH ${ }_{2} \mathrm{Ph}$ L-Phe), 2.85-2.78 (m, 1H, CHCH ${ }_{2} \mathrm{Ph}$ L-Phe), 1.45-1.35 (m, 2H, CH ${ }_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 1.25-1.09 (m, 4H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 0.83 ( $\mathrm{t}, J=7.2 \mathrm{~Hz}, 1.5 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 0.80 (t, $J=7.2 \mathrm{~Hz}, 1.5 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen).
${ }^{13} \mathbf{C}$ NMR ( $\left.\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathrm{C}} 174.06\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.2 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 173.95\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.2\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 156.27 (C6), 156.25 (C6), 154.08 (C2), 154.02 (C2), 151.70 (C4), 151.62 (C4), 147.91 (C8), 147.85 (C8), 139.79 (C-Ar), 139.78 (C-Ar), 137.94 (C-Ar), 136.21 (C-Ar), 136.18 (C-Ar), 130.46 (C-Ar), 129.48 (C-Ar), 129.44 (C-Ar), 128.78 (C-Ar), 128.75 (CAr), 127.92 (C-Ar), 127.89 (C-Ar), 127.80 (CH-Ar), 127.75 (CH-Ar), 127.67 (CH-Ar), 127.32 (CH-Ar), 127.28 (CH-Ar), 126.40 (CH-Ar), 125.79 (CH-Ar), 125.70 (CH-Ar), 122.77 (CH-Ar), 122.73 (CH-Ar), 119.55 (C5), 119.52 (C5), 115.96 ( $\mathrm{d}^{3}{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.0 \mathrm{~Hz}, \mathrm{CH}-$ Ar ), 115.84 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.0 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), 91.34 ( $\mathrm{Cl}^{\prime}$ ), 91.14 ( C 1 '), $84.43\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.4 \mathrm{~Hz}\right.$, C 4 '), 84.31 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.4 \mathrm{~Hz}, \mathrm{C} 4$ '), 72.63 ( C 2 '), 72.54 ( $\mathrm{C}^{\prime}$ ), 71.69 ( $\mathrm{C}^{\prime}$ ), 71.39 ( $\mathrm{C}^{\prime}$ ), $67.72\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.23\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $66.38\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ $n$-Pen), $66.33\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen $), 57.74\left(\mathrm{CHCH}_{2} \mathrm{Ph}\right.$ L-Phe $), 57.70\left(\mathrm{CHCH}_{2} \mathrm{Ph}\right.$ L-Phe), $41.21\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}, \mathrm{CHCH}_{2} \mathrm{Ph}\right.$ L-Phe), $41.16\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}, \mathrm{CHCH}_{2} \mathrm{Ph}\right.$ LPhe), $29.20\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $29.03\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), 28.99 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen), $23.30 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen $), \quad 23.26$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen), $14.24 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen $), \quad 14.21$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen).

MS (ES+ $) \mathbf{m} / \mathbf{z}$ found $747.23\left[\mathrm{M}^{2} \mathrm{Na}^{+}\right], \mathrm{C}_{34} \mathrm{H}_{38} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 725.13$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, two peaks with tR 22.04 min , tR 21.48 min .
[74g] 8-chloroadenosine-5'-O-naphth-1-yl-(cyclohexyloxy-L-alaninyl)phosphate


Prepared according to the general procedure $\mathbf{F}_{1}$ using $2^{\prime}, 3^{\prime}$ - $O$-isopropylidene-8-chloroadenosine [72] ( 0.12 mg , 0.36 mmol ) in THF ( 3 mL ), $t \mathrm{BuMgCl}$ ( 1.0 M in THF, 1.1 $\mathrm{mL}, 1.1 \mathrm{mmol})$ and naphth-1-yl-(cyclohex-1-yloxy-Lphenylalaninyl)phosphorochloridate [170] $(0.50 \mathrm{~g}, 1.1$ $\mathrm{mmol})$ in THF $(1.5 \mathrm{~mL})$. Purification by silica gel CC ( $0-$ $3 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) gave a residue that was stirred at rt in $\mathrm{HCOOH} / \mathrm{H}_{2} \mathrm{O}(3 / 2,10 \mathrm{~mL})$ overnight. The solvents were evaporated in vacuo and the crude purified by prep. TLC $\left(10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to yield the title compound as a white solid $(0.053 \mathrm{~g}, 21 \%$ over two steps).
${ }^{31} \mathbf{P}$ NMR (202 MHz, CD $\mathbf{3}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 3.73, 3.74.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.10$ ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 2$ ), 8.05 ( $\left.\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 2\right), 8.04-8.00(\mathrm{~m}, 1 \mathrm{H}$, H8 Nap), 7.86-7.82 (m, 1H, H5 Nap), 7.65-7.61 (m, 1H, H4 Nap), 7.53-7.48 (m, 1H, H2

Nap), 7.47-7.42 (m, 1H, H6 Nap), 7.32-7.22 (m, 2H, Ph), 7.21-7.12 (m, 3H, Ph), 7.11-7.08 (m, 1H, H7 Nap), 7.07-7.04 (m, 1H, H3 Nap), 6.04-6.01 (m, 1H, H1'), 5.32-5.27 (m, 1H, H2'), 4.59-4.50 (m, 2H, H3', $\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}$ ), 4.39-4.26 (m, 1H, H5'), 4.20-4.04 (m, 3H, H5', H4', $\mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe), 2.99-2.90 (m, 1H, $\mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe), 2.86-2.78 (m, 1 H , $\mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe $), 1.69-1.52\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 1.50-1.41\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5}\right.$ $\mathrm{cHex}), 1.32-1.12\left(\mathrm{~m}, 5 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$.
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 6} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 173.45\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.6 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 173.34\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.6\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 156.25 (C6), 154.09 (C2), 154.04 (C2), 151.70 (C4), 151.61 (C4), 147.91 (C8), 147.86 (C8), 139.79 (C-Ar), 139.77 (C-Ar), 137.94 (C-Ar), 137.89 (C-Ar), 136.21 (C-Ar), 136.19 (C-Ar), 130.52 (C-Ar), 130.51 (C-Ar), 129.46 (C-Ar), 129.42 (C-Ar), 128.79 (CHAr), 128.76 ( $\mathrm{CH}-\mathrm{Ar}$ ), 127.90 (C-Ar), 127.87 (C-Ar), 127.80 (CH-Ar), 127.74 (CH-Ar), 127.67 (CH-Ar), 127.32 (CH-Ar), 127.27 (CH-Ar), 126.41 (CH-Ar), 125.78 (CH-Ar), 125.70 (CH-Ar), 122.78 (CH-Ar), 122.73 (CH-Ar), 119.54 , (C5), 115.94 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.5 \mathrm{~Hz}$, CH-Ar), $115.83\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 91.35\left(\mathrm{Cl}{ }^{\prime}\right), 91.14\left(\mathrm{Cl}{ }^{\prime}\right), 84.46\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.2\right.$ $\mathrm{Hz}, \mathrm{C} 4$ '), 84.33 ( $\mathrm{d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.2 \mathrm{~Hz}, \mathrm{C} 4$ '), 75.03 ( $\mathrm{C}^{\prime}$ ), 75.01 ( $\mathrm{C}^{\prime}$ ), 72.64 ( $\left.\mathrm{C} 3^{\prime}\right), 72.55$ ( $\left.\mathrm{C} 3^{\prime}\right)$, $71.72\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 71.40\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 67.78\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.28(\mathrm{~d}$, ${ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.4 \mathrm{~Hz}, \mathrm{C} 5$ '), $57.78\left(\mathrm{CHCH}_{2} \mathrm{Ph}\right.$ L-Phe $), 41.33\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{CHCH}_{2} \mathrm{Ph}\right.$ L-Phe $)$, $41.25\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{CHCH} 2 \mathrm{Ph}\right.$ L-Phe $), 32.36\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 32.32\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5}\right.$ cHex), $26.32\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right.$ ex $), 26.29\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 24.56\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$.
MS (ES+) m/z found $759.19\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{35} \mathrm{H}_{38} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required m/z 737.14 [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, two peaks with $\mathrm{tR} 21.81 \mathrm{~min} ., \mathrm{tR} 21.27 \mathrm{~min}$.

## [74h] 8-chloroadenosine-5'-O-naphth-1-yl-(ethoxy-L-phenylalaninyl) phosphate



Prepared according to the general procedure $\mathbf{F}_{\mathbf{1}}$ using $2^{\prime}, 3^{\prime}-O$-isopropylidene-8-chloroadenosine [72] $(0.17 \mathrm{~g}$, 0.50 mmol ) in THF ( 3 mL ), $t \mathrm{BuMgCl}$ ( 1.0 M in THF, 1.5 $\mathrm{mL}, 1.5 \mathrm{mmol}$ ) and naphth-1-yl-(ethoxy-L-phenylalaninyl) phosphorochloridate [ $\mathbf{1 7 p}$ ] ( $1.05 \mathrm{~g}, 2.5 \mathrm{mmol}$ ) in THF ( 1.5 $\mathrm{mL})$. Purification by silica gel $\mathrm{CC}\left(0-2 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ gave a residue that was stirred at rt in $\mathrm{HCOOH} / \mathrm{H}_{2} \mathrm{O}(3 / 2$, 10 mL ) overnight. The solvents were evaporated in vacuo and the crude purified by prep. TLC $\left(7 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to yield the title compound as a white solid $(0.09 \mathrm{~g}, 26 \%$ over two steps).

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P NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{P}}$ 3.73, 3.42.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.09(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 2), 8.04$ ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 2$ ), 8.00 (d, $J=8.4$ Hz, 1H, H8 Nap), 7.80 (d, $J=8.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5 \mathrm{Nap}$ ), 7.62-7.58 (m, 1H, H4 Nap), 7.50-7.40 (m, 2H, H2, H6 Nap), 7.32-7.23 (m, 2H, Ph), 7.20-7.07 (m, 4H, Ar), 7.05-7.01 (m, 1H, H3 Nap), 6.05-6.01 (m, 1H, H1'), 5.33-5.28 (m, 1H, H2'), 4.60-4.53 (m, 1H, H3'), 4.38-4.31 ( $\mathrm{m}, 0.5 \mathrm{H}, \mathrm{H} 5$ '), 4.30-4.24 (m, 0.5H, H5'), 4.20-4.10 (m, 4.5H, H5', H4', $\mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe, $\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}$ ), 4.09-4.03 (m, 0.5H, H5'), 3.97-3.85 (m, 2H, CHCH ${ }_{2} \mathrm{Ph}$ L-Phe), 1.07-0.98 (m, $\left.3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right)$.
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 174.00\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.6 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 173.90\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.6\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 156.20 (C6), 154.09 (C2), 154.04 (C2), 151.67 (C4), 151.59 (C4), 147.92, 147.86 (C8), 147.80 (C8), 139.82 (C-Ar), 139.79 (C-Ar), 137.90 (C-Ar), 136.17 (C-Ar), 136.15 (C-Ar), 130.51 (C-Ar), 129.48 (C-Ar), 129.43 (C-Ar), 128.76 (CH-Ar), 127.92 (CH-Ar), 127.89 (C-Ar), 127.79 (CH-Ar), 127.73 (CH-Ar), 127.69 (CH-Ar), 127.33 (CHAr), 127.30 (CH-Ar), 126.43 (CH-Ar), 125.80 (CH-Ar), 125.71 (CH-Ar), 122.73 (CH-Ar), 122.70 (CH-Ar), 119.52 , (C5), 115.97 (d, $\left.{ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.6 \mathrm{~Hz} \mathrm{CH}-\mathrm{Ar}\right), 115.81$ (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=3.6 \mathrm{~Hz}$ CH-Ar), $91.33\left(\mathrm{Cl}^{\prime}\right), 91.16\left(\mathrm{Cl}^{\prime}\right), 84.35\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.7 \mathrm{~Hz}, \mathrm{C}-4{ }^{\prime}\right), 84.27\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=8.7 \mathrm{~Hz}\right.$, C-4'), 72.66 (C2'), $72.60\left(\mathrm{C} 2\right.$ '), $71.70\left(\mathrm{C} 3\right.$ '), 71.38 ( $\left.\mathrm{C}^{\prime}\right), 67.74$ ( $\mathrm{d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.4 \mathrm{~Hz}, \mathrm{C} 5$ '), $67.23\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $62.33\left(\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right), 62.27\left(\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right), 57.69\left(\mathrm{CHCH}_{2} \mathrm{Ph}\right.$ L-Phe), $57.62\left(\mathrm{CHCH}_{2} \mathrm{Ph}\right.$ L-Phe $), 41.20\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.2 \mathrm{~Hz}, \mathrm{CHCH}_{2} \mathrm{Ph}\right.$ L-Phe), 41.13 (d, ${ }^{3} J_{\mathrm{C}}$ $\mathrm{P}=6.2 \mathrm{~Hz}, \mathrm{CHCH}_{2} \mathrm{Ph}$ L-Phe $), 14.35\left(\mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Et}\right)$.

MS (ES+ $) \mathbf{m} / \mathbf{z}$ found $705.18\left[\mathrm{M}^{+} \mathrm{Na}^{+}\right], \mathrm{C}_{31} \mathrm{H}_{32} \mathrm{ClN}_{6} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 683.05$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, two peaks with tR 17.00 min , tR 16.53 min .
[76a] 8-chloroadenosine 5'-O-bis(benzyloxy-L-alaninyl)phosphate


Prepared according to the general procedure $\mathbf{G}$, using 8-chloroadenosine [67] ( $0.25 \mathrm{~g}, 0.83 \mathrm{mmol})$, in $\left(\mathrm{CH}_{3} \mathrm{O}\right)_{3} \mathrm{PO}(5 \mathrm{~mL}), \mathrm{POCl}_{3}(77 \mu \mathrm{~L}, 0.83 \mathrm{mmol}), \mathrm{L}-$ alanine- $O$-benzyl ester tosylate salt $(1.46 \mathrm{~g}, 4.15$ mmol ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and DIPEA ( $1.45 \mathrm{~mL}, 8.3$ mmol ). The crude was purified by column chromatography (gradient elution of $4-7 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give the title compound as a white solid ( $0.24 \mathrm{~g}, 42 \%$ ).

31
P NMR ( $\left.\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}} 13.54$.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.21(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 7.36-7.26(\mathrm{~m}, 10 \mathrm{H}, \mathrm{Ph}), 6.02(\mathrm{~d}, J=$ $4.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), $5.30\left(\mathrm{t}, J=5.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2^{\prime}\right)$, $5.14-5.02\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.62(\mathrm{dd}, J=$ 7.5, 5.0 Hz, 1H, H3'), 4.31-4.24 (m, 1H, H5'), 4.18-4.11 (m, 2H, H5', H4'), 3.95-3.82 (m, $2 \mathrm{H}, \mathrm{CHCH}_{3}$ L-Ala), 1.28 (d, $J=7.2 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ), 1.23 (d, $J=7.2 \mathrm{~Hz}, 3 \mathrm{H}$, $\mathrm{CHCH}_{3}$ L-Ala).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 175.34\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 175.30\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=5.0\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 156.35 (C6), 154.22 (C2), 151.71 (C4), 139.83 (C-Ar), 137.32 (C-Ar), 137.27 (C-Ar), 129.58 (CH-Ar), 129.56 (CH-Ar), 129.32 (CH-Ar), 129.28 (CH-Ar), 119.55 (C5), $91.28\left(\mathrm{Cl}^{\prime}\right), 84.50\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.3 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $72.65\left(\mathrm{C} 2^{\prime}\right), 71.47\left(\mathrm{C} 3\right.$ '), $67.91\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.86$ $\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 66.17\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=4.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ ' $), 51.01\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 50.98\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$, $20.81\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.0 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 20.53\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=6.0 \mathrm{~Hz}, \mathrm{CHCH}_{3}\right.$ L-Ala).
MS (ES+ $) \mathbf{m} / \mathbf{z}$ found $704.1\left[\mathrm{M}^{+} \mathrm{H}^{+}\right], 726.1\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{30} \mathrm{H}_{35} \mathrm{ClN}_{7} \mathrm{O}_{9} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 704.07$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254$, one peak with tR 24.98 min .
[74b] 8-chloroadenosine 5'-O-bis(benzyloxy-glycinyl) phosphate


Prepared according to the general procedure G, using 8-chloroadenosine [67] ( $0.10 \mathrm{~g}, 0.33 \mathrm{mmol})$, in $\left(\mathrm{CH}_{3} \mathrm{O}\right)_{3} \mathrm{PO}(2.5 \mathrm{~mL}), \mathrm{POCl}_{3}(31 \mu \mathrm{~L}, 0.33 \mathrm{mmol})$, glycine- $O$-benzyl ester hydrochloride salt $(0.33 \mathrm{~g}, 1.65$ mmol ), in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and DIPEA ( $0.57 \mathrm{~mL}, 3.3$ mmol ). The crude was purified by column chromatography (gradient elution of 4-7\% $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give the title compound as a white solid ( $0.06 \mathrm{~g}, 27 \%$ ).
${ }^{\mathbf{3 1}} \mathbf{P}$ NMR ( $\left.\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}} 15.96$.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.23(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 7.35-7.26(\mathrm{~m}, 10 \mathrm{H}, \mathrm{Ph}), 6.03(\mathrm{~d}, \mathrm{~J}=$ $5.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1^{\prime}$ ), 5.31 (dd, $J=5.5,5.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2$ '), $5.11-5.08$ (m, 4H, CH ${ }_{2} \mathrm{Ph}$ ), 4.59 (dd, $5.5 \mathrm{~Hz}, 5.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3^{\prime}$ ), 4.34-4.28 (m, 1H, H5'), 4.27-4.16 (m, 2H, H5', H4'), 3.70-3.64 (m, 4H, CH2 Gly).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 172.91\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.7 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 172.89\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=4.7\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 156.24 (C6), 154.08 (C2), 151.69 (C4), 139.88 (C-Ar), 137.22 (C-Ar), 129.57 (CH-Ar), 129.56 (CH-Ar), 129.33 (CH-Ar), 129.31 (CH-Ar), 119.56 (C5), 91.16 ( C 1 '),
$84.70\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.2 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $72.62\left(\mathrm{C} 2\right.$ '), $71.58\left(\mathrm{C} 3^{\prime}\right), 67.86\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.84\left(\mathrm{CH}_{2} \mathrm{Ph}\right)$, $66.17\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=4.6 \mathrm{~Hz}, \mathrm{C} 5\right.$ ) $), 43.56\left(\mathrm{CH}_{2} \mathrm{Gly}\right), 43.47\left(\mathrm{CH}_{2} \mathrm{Gly}\right)$.

MS (ES+ $) \mathbf{m} / \mathbf{z}$ found $704.1\left[\mathrm{M}+\mathrm{H}^{+}\right], 726.1\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{28} \mathrm{H}_{31} \mathrm{ClN}_{7} \mathrm{O}_{9} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 676.01$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254$, one peak with tR 23.23 min .

### 9.5.13 Synthesis of 5'-modified 8-chloroadenosine analogues

[78] $2^{\prime}, 3^{\prime}, 5$ '-tri- $O$-tert-butyldimethylsilyl-(bis- $N^{6}, N^{6}$-tert-butylcarbonyl)-8chloroadenosine


Procedure
2',3',5'-tri-O-tert-butyldimethylsilyl-8-
chloroadenosine [70] ( $4.0 \mathrm{~g}, 6.2 \mathrm{mmol}$ ) was dissolved in anhydrous THF ( 50 mL ) and DMAP ( $0.30 \mathrm{~g}, 2.48 \mathrm{mmol}$ ) was added in one portion to the stirring solution, followed by $\mathrm{Boc}_{2} \mathrm{O}(5.41 \mathrm{~g}, 24.8 \mathrm{mmol})$. The reaction was stirred at rt for 18 h . The solvent was evaporated and the crude dissolved in $n$-Hex ( 30 mL ) and washed with brine ( 20 mL $x$ 3). The organic phase was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated. The crude was purified by Biotage Isolera One ( 100 g SNAP cartridge KP SIL, $100 \mathrm{~mL} / \mathrm{min}$, eluent system $1-20 \% \mathrm{EtOAc} / n \mathrm{Hex} 10 \mathrm{CV}, 20 \% \mathrm{CV}$ ) to yield the desired compound as syrup ( 5.24 g , 87\%).
${ }^{1} \mathbf{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.76(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 6.02(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), 5.42 (dd, $\left.J=6.5 \mathrm{~Hz}, 4.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2^{\prime}\right), 4.51(\mathrm{dd}, J=4.5 \mathrm{~Hz}, 2.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3$ '), 4.08 (ddd, $J=7.5 \mathrm{~Hz}$, $4.5 \mathrm{~Hz}, 2.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4$ '), 4.02 (dd, $J=11.0 \mathrm{~Hz}, 7.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ '), 3.74 (dd, $J=10.5 \mathrm{~Hz}$, $\left.4.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5^{\prime}\right), 1.50(\mathrm{~s}, 18 \mathrm{H}, \mathrm{Boc}), 0.953(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.864(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.761(\mathrm{~s}, 9 \mathrm{H}$, $t \mathrm{Bu}), 0.150\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.14\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.02\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.00\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.09(\mathrm{~s}$, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), $-0.42\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.

$2^{\prime}, 3^{\prime}, 5^{\prime}$-tri-O-tert-butyldimethylsilyl-(bis- $N^{6}, N^{6}$-tert-butylcarbonate)-8-chloroadenosine [78] ( $4.18 \mathrm{~g}, 5.04 \mathrm{mmol})$ was dissolved in a mixture of THF/water $(5 \mathrm{~mL}, 4 / 1)$ and the solution was cooled to $0^{\circ} \mathrm{C}$ in an ice bath. TFA ( 1 mL ) was added dropwise and the reaction was stirred for 5 h at $0^{\circ} \mathrm{C}$. The solution was then evaporated and the crude dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$ and washed with $\mathrm{NaHCO}_{3}(20 \mathrm{~mL})$ and brine ( $20 \mathrm{~mL} \times 2$ ). The organic phase was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated. The crude was purified via Biotage Isolera One ( 120 g ZIP cartridge KP SIL, $100 \mathrm{~mL} / \mathrm{min}$, eluent system $12-23 \% \mathrm{EtOAc} / n \mathrm{Hex} 5 \mathrm{CV}, 23 \% 5 \mathrm{CV}$ ) to afford the desired product as a white foam ( $2.50 \mathrm{~g}, 68 \%$ ).
1
H NMR ( $\mathbf{5 0 0} \mathbf{~ M H z , ~ C D C l} \mathbf{3}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.67(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 5.96(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), 5.66 (dd, $\left.J=12.1 \mathrm{~Hz}, 1.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{OH} 5^{\prime}\right), 4.93$ (dd, $J=8.0 \mathrm{~Hz}, 4.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2$ '), 4.20 (d, $J=4.5 \mathrm{~Hz}$, 1H, H3'), 4.01-4.03 (m, 1H, H4'), $3.80\left(\mathrm{~d}, J=13.1,1 \mathrm{H}, \mathrm{H} 5^{\prime}\right), 3.58\left(\mathrm{t}, J=12.7,1 \mathrm{H}, \mathrm{H} 5^{\prime}\right)$, $1.30(\mathrm{~s}, 18 \mathrm{H}, 2 \times \mathrm{Boc}), 0.81(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.63(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.00\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.01(\mathrm{~s}, 3 \mathrm{H}$, $\mathrm{CH}_{3}$ ), $-0.26\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.78\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$. ribo-hexofuranosyl] bis- $N^{6}, N^{6}$-tert-butoxycarbonyl-8-chloroadenine


To a solution of $2^{\prime}, 3^{\prime}$-bis-O-tert-butyldimethylsilyl-(bis$N^{6}, N^{6}$-tert-butylcarbonate)-8-chloroadenosine [79] (1.25 g, $1.80 \mathrm{mmol})$ in $\mathrm{CH}_{3} \mathrm{CN}(10 \mathrm{~mL})$ was added 2-iodoxybenzoic acid (IBX) $(1.00 \mathrm{~g}, 3.60 \mathrm{mmol})$. After stirring for 2 hours at $80^{\circ} \mathrm{C}$, the reaction mixture was cooled to $0^{\circ} \mathrm{C}$ and filtered. The filtrate was washed with anhydrous THF ( 10 mL ) and the solution was concentrated and co-evaporatd with THF. Subsequently, to a solution of the resulting aldehyde [80] in THF ( 2 mL ) was added dropwise, at $0{ }^{\circ} \mathrm{C}$, a freshly prepared solution of tetraethyl-bisphophonate sodium salt [prepared by addition, at $0{ }^{\circ} \mathrm{C}$, of tetraethyl-bisphophonate ( $0.71 \mathrm{~mL}, 2.88 \mathrm{mmol}$ ) to a suspension of $\mathrm{NaH}(0.11 \mathrm{~g}, 2.60 \mathrm{mmol})$ in THF ( 1.5 mL ) and stirring for 10 min$]$. The reaction mixture was then stirred at room temperature for 16 h before addition of $\mathrm{NH}_{4} \mathrm{Cl}$ ( 10 mL , saturated aqueous solution). The mixture was then extracted with EtOAc (3 x 5 $\mathrm{mL})$. The combined organic layers were washed with water ( $2 \times 5 \mathrm{~mL}$ ) and brine ( 5 mL )
and the resulting solution was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The collected solution was concentrated under reduced pressure and the residue was purified on Biotage Isolera One (50g SNAP cartridge KP SIL, $50 \mathrm{~mL} / \mathrm{min}$, eluent system $12-100 \% \mathrm{EtOAc} / n \mathrm{Hex} 10 \mathrm{CV}, 100 \% 5 \mathrm{CV}$ ) to give the title compound as a white foam (single $E$ isomer, $1.05 \mathrm{~g}, 68 \%$ ).
${ }^{31}$ PNMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l 3}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 17.55$.
${ }^{1} \mathbf{H N M R}(500 \mathrm{MHz}, \mathbf{C D C l} 3) \boldsymbol{\delta}_{\mathbf{H}} 8.64(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 6.83\left(\mathrm{ddd},{ }^{1} J_{\mathrm{HP}}=22.4 \mathrm{~Hz},{ }^{1} J_{\mathrm{HH}} 17.0 \mathrm{~Hz}\right.$, $\left.{ }^{2} J_{\mathrm{HH}} 5.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6^{\prime}\right), 5.99-5.89\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H} 5^{\prime}, \mathrm{H} 1^{\prime}\right), 5.16$ (dd, $J=6.0 \mathrm{~Hz}, 4.5 \mathrm{~Hz}, 1 \mathrm{H}$, H2'), 4.47-4.43 (m, 1H, H4'), 4.33 (dd, $J=4.5 \mathrm{~Hz}, 3.5 \mathrm{~Hz}, \mathrm{H} 3$ '), 3.98-3.85 (m, 4H, $2 \times$ $C H_{2} \mathrm{CH}_{3}$ ), $1.29\left(\mathrm{~s}, 18 \mathrm{H}, \mathrm{Boc}_{2}\right), 1.18\left(\mathrm{t}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.13(\mathrm{t}, \mathrm{J}=7.0 \mathrm{~Hz}, 3 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ), $0.81(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.62(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.01\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.00\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.21$ ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3}$ ), $-0.55\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.
MS (ES+) $\mathbf{m} / \mathbf{z}$ found $862.3\left[\mathrm{M}+\mathrm{H}^{+}\right], 884.3\left[\mathrm{M}+\mathrm{Na}^{+}\right], 762.3\left[\mathrm{M}(-\mathrm{Boc})+\mathrm{H}^{+}\right], 784.3[\mathrm{M}(-$ Boc) $+\mathrm{Na}^{+}$], required $\mathrm{m} / \mathrm{z} \mathrm{C}_{37} \mathrm{H}_{65} \mathrm{ClN}_{5} \mathrm{O}_{10} \mathrm{PSi}_{2} 861.37$ [M].
[82] 9-[5',6'-dideoxy-6'-(bis-ethylphosphinyl)-2',3'-bis-O-tert-butyldimethylsilyl-D-ribo-hexofuranosyl] bis- $N^{6}, N^{6}$-tert-butoxycarbonyl 8-chloroadenine


9-[5',6'-vinyl-6'-(bis-ethylphosphinyl)-2', $3^{\prime}$ 'bis-O-tert-butyldimethylsilyl-D-ribo-hexofuranosyl]bis- $N^{6}, N^{6}$-tert-butoxycarbonyl-8-chloroadenine [81] ( $0.30 \mathrm{~g}, 0.35 \mathrm{mmol}$ ) was dissolved in EtOH/EtOAc ( $15 \mathrm{~mL}, 1 / 1$ ) and $\mathrm{Pd} / \mathrm{C}(0.03 \mathrm{~g}, 10 \%$ Pd on activated carbon) was added. The mixture was stirred for 5 hours under an atmosphere of $\mathrm{H}_{2} . \mathrm{NaCl}(0.02 \mathrm{~g})$ was added and the mixture stirred for 15 minutes and then filtered through a short pad of celite and washed with EtOH ( $10 \mathrm{~mL} \times 3$ ). The collected solution was concentrated under vacuum and the residue was purified via Biotage Isolera One ( 25 g SNAP cartridge Ultra, $75 \mathrm{~mL} / \mathrm{min}$, eluent system $0-6 \%$ $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 10 \mathrm{CV}, 6 \% 2 \mathrm{CV}$ ) to give the title compound as a white foam $(0.24 \mathrm{~g}, 81 \%)$ ${ }^{31}$ PNMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 31.81$.
${ }^{1} \mathbf{H N M R}(500 \mathbf{M H z}, \mathbf{C D C l} 3) \boldsymbol{\delta}_{\mathbf{H}} 8.67$ (s, 1H, H2), 5.88 (d, $J=6.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), 5.17 (d, $J$ $\left.=4.0 \mathrm{~Hz}, 5.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2^{\prime}\right), 4.13$ (d, $J=2.5 \mathrm{~Hz}, 4.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3$ '), 4.00-3.86 (m, 5H, H4', 2 x $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 2.12-2.00 (m, 1H, H5'), 1.92-1.80 (m, 1H, H5'), 1.78-1.68 (m, 1H, H6'), 1.63$1.49\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 6^{\prime}\right), 1.30\left(\mathrm{~s}, 18 \mathrm{H}, t \mathrm{Bu} \mathrm{Boc}_{2}\right), 1.18-1.12\left(\mathrm{~m}, 6 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3} \times 2\right), 0.81(\mathrm{~s}, 9 \mathrm{H}$, $t \mathrm{Bu}), 0.64(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.002\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.00\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.22\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right),-0.56(\mathrm{~s}$, $3 \mathrm{H}, \mathrm{CH}_{3}$ ).
[83] 9-[5',6'-dideoxy-6'-(bis-ethylphosphinyl)-D-ribo-hexofuranosyl]-8-chloroadenine
A solution of 9-[5',6'-dideoxy-6'-(bis-ethylphosphinyl)-2', ${ }^{\prime}$ '-bis- $O$ -
tert-butyldimethylsilyl-D-ribo-hexofuranosyl]bis- $N^{6}, N^{6}$-tertbutoxycarbonyl 8 -chloroadenine [82] ( $0.24 \mathrm{~g}, 0.28 \mathrm{mmol}$ ) was stirred at room temperature for 72 hours in $\mathrm{HCOOH} / \mathrm{H}_{2} \mathrm{O}(10 \mathrm{~mL}$, $1 / 1$ ). The volatiles were evaporated and the crude purified via Biotage Isolera One ( 25 g SNAP cartridge Ultra, $75 \mathrm{~mL} / \mathrm{min}$, eluent system 2-20\% $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 10 \mathrm{CV}, 20 \% 7 \mathrm{CV}$ ) to afford the title compound as a white solid ( 0.09 g , 75\%).
${ }^{31}$ PNMR ( $202 \mathrm{MHz}, \mathbf{C D C l 3}$ ) $\boldsymbol{\delta}_{\mathrm{P}} 33.04$.
${ }^{1} \mathbf{H N M R}\left(500 \mathrm{MHz}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.20(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 6.01(\mathrm{~d}, J=4.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), $5.25(\mathrm{t}, J$ $\left.=5.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2^{\prime}\right), 4.43\left(\mathrm{~d}, J=5.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3^{\prime}\right), 4.13-4.01\left(\mathrm{~m}, 5 \mathrm{H}, \mathrm{H} 4{ }^{\prime}, \mathrm{CH}_{2} \mathrm{CH}_{3} \times 2\right)$, 2.14-1.84 (m, 4H, H5', H6'), $1.32\left(\mathrm{t}, J=7.5 \mathrm{~Hz}, 1.5 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.30(\mathrm{t}, J=7.5 \mathrm{~Hz}$, $\left.1.5 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$.
[86] 9-[5',6'-dideoxy-6'-(phenyloxy-(benzyloxy-L-alanin- $N$-yl-phosphinyl)-D-ribo-hexofuranosyl]-8-chloroadenine


A solution of 9-[5', $6^{\prime}$-dideoxy-6'-(bis-ethylphosphinyl)-D-ribo-hexofuranosyl]-8-chloroadenine [83] ( $0.05 \mathrm{~g}, 0.12$ mmol) in dry $\mathrm{CH}_{3} \mathrm{CN}(2 \mathrm{~mL})$ was cooled to $0{ }^{\circ} \mathrm{C}$ and 2,6 lutidine ( $0.04 \mathrm{~mL}, 0.36 \mathrm{mmol}$ ) was added, followed by $\mathrm{TMSBr}(0.05 \mathrm{~mL}, 0.36 \mathrm{mmol})$. The mixture was stirred under argon at $0{ }^{\circ} \mathrm{C}$ for 5 hours. Then the solvents were removed under reduced pressure without any contact with air. The residue was dissolved in anhydrous $\mathrm{Et}_{3} \mathrm{~N}(0.25 \mathrm{~mL})$ and pyridine ( 1 mL ) and L-alanine benzyl ester $p$ TSA salt $(0.042 \mathrm{~g}, 0.12 \mathrm{mmol})$ and phenol $(0.068 \mathrm{~g}, 0.72 \mathrm{mmol})$ were added. In a separated flask, aldrithiol-2 ( $0.06 \mathrm{~g}, 0.72 \mathrm{mmol}$ ) and $\mathrm{Ph}_{3} \mathrm{P}(0.188 \mathrm{~g}, 0.72 \mathrm{mmol})$ were dissolved in anhydrous pyridine $(1.2 \mathrm{~mL})$ and this solution was immediately added to the reaction. The resulting mixture (green solution) was stirred at $50^{\circ} \mathrm{C}$ for 16 hours. The residue was taken up in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(15 \mathrm{~mL})$ and washed with $\mathrm{HCl}(0.5 \mathrm{~N}$ in water) $(2 \times 15 \mathrm{~mL})$ and brine ( 15 mL ). The organics were combined and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to dryness. The crude was purified via Biotage Isolera One (10g SNAP cartridge Ultra, 12 $\mathrm{mL} / \mathrm{min}$, eluent system $2-100 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 10 \mathrm{CV}, 100 \% 7 \mathrm{CV}$ ), prep. HPLC (isocratic eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{H} 2 \mathrm{O} 55 \%$, in 30 minutes, $20 \mathrm{~mL} / \mathrm{min}, \lambda=254$ ) to afford the two separated diastereoisomers as white solids ( $0.004 \mathrm{~g}, 5 \%$ ).

More polar isomer (fast eluting-FE)
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 34.42.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.00(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 7.20-7.10(\mathrm{~m}, 7 \mathrm{H}, \mathrm{Ph}), 7.02-6.92(\mathrm{~m}$, $3 \mathrm{H}, \mathrm{Ph}), 5.84(\mathrm{~d}, J=5.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), $5.09(\mathrm{t}, J=5.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2$ '), 4.93-4.84 (m, 2H, $\mathrm{CH}_{2} \mathrm{Ph}$ ), $4.24\left(\mathrm{t}, J=5.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3\right.$ '), 3.86-3.76 (m, 2H, H4', $\mathrm{CHCH}_{3}$ L-Ala), 2.02-1.79 (m, 4H, H5', H6'), 0.99 (d, J=7.0 Hz, 3H, $\mathrm{CHCH}_{3}$ L-Ala).

## Least polar isomer (slow eluting-SE)

${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 33.67$.
${ }^{1}{ }^{\mathbf{H}} \mathbf{N M R}\left(500 \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.18(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 7.34-7.26(\mathrm{~m}, 7 \mathrm{H}, \mathrm{Ph}), 7.18-7.11(\mathrm{~m}$, $3 \mathrm{H}, \mathrm{Ph}), 6.00(\mathrm{~d}, J=4.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ ') , 5.19 (dd, $J=4.5,5.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2$ '), $5.02-4.98$ (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}$ ), 4.47 (t, $J=5.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3$ '), 4.04-3.94 (m, 2H, H4', $\left.\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$, 2.191.95 (m, 4H, H5', H6'), 1.26 (d, $J=7.0 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CHCH}_{3}$ L-Ala).

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254$, one peak of the overlapping diastereoisomers with tR 23.23 min .

### 9.5.14 Synthesis of 8-chloro-2'-deoxyadenosine

## [87] 8-Chloro-2'-deoxyadenosine



3',5'-(Bis-O-tert-butyldimethylsilyl) 8-chloroadenosine [89] (0.81 g, 1.57 mmol ) was dissolved in THF ( 10 mL ) and the solution was cooled to $0^{\circ} \mathrm{C}$. TBAF ( 1 M in THF, $5.51 \mathrm{~mL}, 5.51 \mathrm{mmol}$ ) was added dropwise and the mixture was stirred at room temperature for 1 hour. The solvents were evaporated and the crude washed with $n$-hex ( $4 \times 5$ mL ), The crude was packed as a silica slurry and purified via flash column chromatography (gradient eluent system $3-10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CHCl}_{3}$ ) to afford the title compound as a white solid $(0.094 \mathrm{~g}, 33 \%) .{ }^{351}$
${ }^{1} \mathbf{H}$ NMR ( $500 \mathbf{M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.16(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 6.39(\mathrm{dd}, J=8.5,6.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), 5.67 (br s, 2H, NH2), 4.67-4.65 (m, 1H, H3'), 4.12-4.10 (m, 1H, H4'), 3.89 (dd, $J=12.5$, $2.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5^{\prime}$ ), 3.75 (dd, $J=12.5,3.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}{ }^{\prime}$ ), 3.16-3.10 (m, 1H, H2'), 2.34-2.30 (m, 1H, H2').
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 6} \mathbf{~ M H z}$, DMSO-d6) $\boldsymbol{\delta}_{\mathbf{C}} 156.15$ (C6), 152.03 (C2), 149.52 (C4), 136.92 (C8), 116.54 (C5), 88.95 ( $\mathrm{C}^{\prime}$ ), 86.59 (C4'), 72.05 ( $\mathrm{C}^{\prime}$ ), 62.69 ( C 5 '), 38.39 (C2').
[88] 3',5'-Bis-O-tert-butyldimethylsilyl-2'-deoxyadenosine


2'-Deoxyadenosine ( $6.0 \mathrm{~g}, 23.88 \mathrm{mmol}$ ) was dissolved in DMF $(30 \mathrm{~mL})$ and imidazole ( $7.15 \mathrm{~g}, 105.07 \mathrm{mmol}$ ) was added, followed by TBDMSCl ( $7.98 \mathrm{~g}, 52.53 \mathrm{mmol})$.
The mixture was stirred at rt for 16 hours, then the volatiles were evaporated and the crude taken up with $n$-hex ( 40 mL ) and washed with aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ (saturated solution, 20 mL x 2) and brine ( 20 mL ). The organics were collected, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated, to yield the title compound as a sticky solid $(10.77 \mathrm{~g}, 94 \%) .{ }^{351}$
${ }^{1} \mathbf{H}$ NMR ( $500 \mathbf{M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.37(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 8), 8.15(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 6.47(\mathrm{t}, J=6.5 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{H} 1$ '), 5.78 (br s, $2 \mathrm{H}, \mathrm{NH}_{2}$ ), 4.63 (dt, $J=6.0,3.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3$ '), 4.05-4.00 (m, 1H, H4'), 3.89 (dd, $\left.J=11.0,4.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H})^{\prime}\right), 3.79$ (dd, $J=11.5,3.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ '), 2.69-2.62 (m, $1 \mathrm{H}, \mathrm{H} 2$ '), 2.49-2.42 (m, 1H, H2'), 0.937 ( $\mathrm{s}, 9 \mathrm{H}, t \mathrm{Bu}$ ), 0.935 ( $\mathrm{s}, 9 \mathrm{H}, t \mathrm{Bu}$ ), $0.12\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{CH}_{3}\right.$ x 2), $0.11\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.10\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.

## [89] 3',5'-Bis-O-tert-butyldimethylsilyl-8-chloro-2'-deoxyadenosine

 mmol) in a $-78{ }^{\circ} \mathrm{C}$ bath was added a solution of $3^{\prime}, 5^{\prime}$-O-tert-butyldimethylsilyl-2'-deoxyadenosine [88] ( $10.0 \mathrm{~g}, 52.10 \mathrm{mmol}$ ) in THF ( 25 mL ) dropwise via a dropping funnel. The funnel was washed with THF ( 5 mL ) and the reaction was stirred for 1 hour at $-78^{\circ} \mathrm{C}$. A solution of $\mathrm{TsCl}(39.73 \mathrm{~g}, 208.4 \mathrm{mmol})$ in THF $(30 \mathrm{~mL})$ was added dropwise, and the funnel rinsed with THF ( 5 mL ). The reaction was stirred for 1 h and then quenched by dropwise addition of glacial $\mathrm{AcOH}(15 \mathrm{~mL})$. The mixture was allowed to reach room temperature and evaporated. The crude was dissolved in $n \mathrm{Hex}(50 \mathrm{~mL})$ and washed with brine ( $3 \times 50 \mathrm{~mL}$ ). The organic phase was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated. The crude was purified by Biotage Isolera flash chromatography (100g SNAP cartridge KP SIL, eluent system 3-30\% EtOAc/nHex $10 \mathrm{CV}, 30 \% \mathrm{EtOAc} / n \mathrm{Hex} 4 \mathrm{CV}$ ) to yield the title compound as a white foam $(20.09 \mathrm{~g}, 75 \%) .{ }^{351}$
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.30(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 2), 6.47(\mathrm{t}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), 5.67 (br s, $2 \mathrm{H}, \mathrm{NH}_{2}$ ), 4.89-4.86 (m, 1H, H3'), 3.99-3.96 (m, 1H, H4'), $3.91(\mathrm{dd}, \mathrm{J}=11.0,6.5 \mathrm{~Hz}, 1 \mathrm{H}$, H5'), 3.69 (dd, $J=11.0,5.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ '), 3.65-3.60 (m, 1H, H2'), 2.28-2.23 (m, 1H, $\left.\mathrm{H} 2^{\prime}\right), 0.96(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.85(\mathrm{~s}, 9 \mathrm{H}, t \mathrm{Bu}), 0.17\left(\mathrm{~s}, 6 \mathrm{H}, \mathrm{CH}_{3} \times 2\right), 0.02\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 0.01(\mathrm{~s}$, $3 \mathrm{H}, \mathrm{CH}_{3}$ ).

### 9.5.15 Synthesis of 8-chloro-2'-deoxyadenosine prodrugs

## [90a] 8-Chloro-2'-deoxyadenosine-3',5'-bis-O-naphth-1-yl(pentyl-1-oxy-L-leucinyl) phosphate



Prepared according to general procedure $\mathbf{F}_{\mathbf{1}}$ using 8-chloro-2'-deoxyadenosine [87], ( $0.10 \mathrm{~g}, 0.35 \mathrm{mmol}$ ) in THF ( 7 mL ), $t \mathrm{BuMgCl}(1.0 \mathrm{M}$ in THF, $0.87 \mathrm{~mL}, 0.87$ mmol ) and naphth-1-yl-(pentyl-1-oxy-L-leucinyl) phosphorochloridate [17f] ( $0.45 \mathrm{~g}, 1.05 \mathrm{mmol}$ ) in THF (3 mL ). Purification by Biotage Isolera One ( 50 g SNAP cartridge, $50 \mathrm{~mL} / \mathrm{min}$, gradient eluent system $1-10 \%$ $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 10 \mathrm{CV}, 10 \% 5 \mathrm{CV}$ ) and prep TLC (2000 $\mu \mathrm{m}, 5 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded the title compound as a white solid ( $0.03 \mathrm{~g}, 8 \%$ )
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 4.05,3.97,3.92,3.84,3.49,2.46$.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.26-8.22 (m, 1H, Nap), 8.10-8.01 (m, 2H, H2, Nap), 7.93-7.87 (m, 1H, Nap), 7.85-7.89 (m, 1H, Nap), 7.76-7.69 (m, 1H, Nap), 7.65-7.60 (m, 1H, Nap), 7.60-7.35 (m, 7H, Nap), 7.33-7.23 (m, 1H, Nap), 6.49-6.45 (m, 0.5H, H1'), 6.35-6.29 (m, 0.5H, H1'), 5.78-5.60 (m, 1H, H3'), 4.62-4.35 (m, 3H, H5', H4'), 4.06-3.79 ( $\mathrm{m}, 6 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pent, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 3.79-3.71 (m, $0.5 \mathrm{H}, \mathrm{H} 2$ '), 3.68-3.53 (m, 0.5H, H2'), 2.83-2.75 (m, 0.5H, H2'), 2.62-2.52 (m, 0.5H, H2'), 1.77-1.13 ( $\mathrm{m}, 16 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pent, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 0.89-0.63 (m, 20 H , $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pent, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 175.44\left(\mathrm{~d},{ }^{3} J_{C P}=2.5 \mathrm{~Hz}, \mathrm{CO}\right), 175.37\left(\mathrm{~d},{ }^{3} J_{C P}=2.5 \mathrm{~Hz}\right.$, CO ), $175.23\left(\mathrm{~d},{ }^{3} J_{C P}=2.5 \mathrm{~Hz}, \mathrm{CO}\right), 175.03\left(\mathrm{~d},{ }^{3} J_{C P}=2.5 \mathrm{~Hz}, \mathrm{CO}\right), 175.00\left(\mathrm{~d},{ }^{3} J_{C P}=2.5\right.$ $\mathrm{Hz}, \mathrm{CO}), 174.89\left(\mathrm{~d},{ }^{3} J_{C P}=2.5 \mathrm{~Hz}, \mathrm{CO}\right), 174.87\left(\mathrm{~d},{ }^{3} J_{C P}=2.5 \mathrm{~Hz}, \mathrm{CO}\right), 156.22$ (C6), 156.17 (C6), 156.12 (C6), 153.90 (C2), 153.88 (C2), 153.85 (C2), 151.52 (C4), 151.45 (C4), 151.43 (C4), 151.33 (C4), 148.93 (C8), 148.89 (C8), 148.88 (C8), 148.84 (C8), 139.43 (C-Ar), 139.42 (C-Ar), 136.40 (C-Ar), 136.23 (C-Ar), 136.21 (C-Ar), 136.14 (CAr), 128.99 (CH-Ar), 128.95 (CH-Ar), 128.93 (CH-Ar), 128.80 (CH-Ar), 128.74 (CH-Ar), 128.02 (C-Ar), 128.01 (C-Ar), 127.98 (C-Ar), 127.96 (C-Ar), 127.93 (C-Ar), 127.88 (CHAr), 127.84 (CH-Ar), 127.82 (CH-Ar), 127.80 (C-Ar), 127.78 (C-Ar), 127.75 (C-Ar), 127.73 (C-Ar), 127.68 (CH-Ar), 127.63 (CH-Ar), 127.61 (CH-Ar), 127.60 (CH-Ar), 127.34 (CH-Ar), 127.29 (CH-Ar), 126.61 (CH-Ar), 126.57 (CH-Ar), 126.56 (CH-Ar), 126.43 (CH-Ar), 126.39 (CH-Ar), 126.24 (CH-Ar), 126.21 (CH-Ar), 126.15 (CH-Ar), 126.13 (CH-Ar), 125.78 (CH-Ar), 125.72 (CH-Ar), 122.91 (CH-Ar), 122.80 (CH-Ar),
122.75 (CH-Ar), 119.69 (C5), 119.56 (C5), 116.73 (CH-Ar), 116.71 (CH-Ar), 116.47 (CHAr), 116.44 (CH-Ar), 116.41 (CH-Ar), 116.39 (CH-Ar), 116.19 (CH-Ar), 116.17 (CH-Ar), 115.91 ( $\mathrm{CH}-\mathrm{Ar}$ ), 115.88 ( $\mathrm{CH}-\mathrm{Ar}$ ), 86.32 ( C 1 '), 86.20 ( $\mathrm{Cl}^{\prime}$ ), 86.17 ( $\left.\mathrm{Cl}^{\prime}\right), 86.03$ ( $\left.\mathrm{Cl}^{\prime}\right)$,
 85.18 (C4'), $85.13\left(\mathrm{C} 4\right.$ '), $85.07\left(\mathrm{C} 4\right.$ '), $78.55\left(\mathrm{~d},{ }^{2} J_{C P}=5.4 \mathrm{~Hz}, \mathrm{C} 3\right.$ '), $78.52\left(\mathrm{~d},{ }^{2} J_{C P}=5.2\right.$ $\mathrm{Hz}, \mathrm{C} 3$ '), $78.20\left(\mathrm{~d},{ }^{2} J_{C P}=5.3 \mathrm{~Hz}, \mathrm{C} 3\right.$ '), $78.16\left(\mathrm{~d},{ }^{2} J_{C P}=5.2 \mathrm{~Hz}, \mathrm{C} 3\right.$ '), $67.34\left(\mathrm{~d},{ }^{2} J_{C P}=5.4\right.$ $\mathrm{Hz}, \mathrm{C} 5$ '), $67.23\left(\mathrm{~d},{ }^{2} J_{C P}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.18\left(\mathrm{~d},{ }^{2} J_{C P}=5.2 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.08\left(\mathrm{~d},{ }^{2} J_{C P}=5.2\right.$ $\mathrm{Hz}, \mathrm{C} 5$ '), $66.43\left(\mathrm{CH}_{2}\right.$ Gly $), 66.42\left(\mathrm{CH}_{2}\right.$ Gly $), 66.41\left(\mathrm{CH}_{2} \mathrm{Gly}\right), 66.39\left(\mathrm{CH}_{2} \mathrm{Gly}\right), 66.25$ $\left(\mathrm{CH}_{2}\right.$ Gly), $66.23\left(\mathrm{CH}_{2}\right.$ Gly ), $54.94\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $54.90\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $54.66\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu $)$, $54.64\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 54.62 $\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $44.28\left(\mathrm{CH}_{2}\right), 44.26\left(\mathrm{CH}_{2}\right), 44.22\left(\mathrm{CH}_{2}\right), 44.20\left(\mathrm{CH}_{2}\right), 44.12$ $\left(\mathrm{CH}_{2}\right), 44.09\left(\mathrm{CH}_{2}\right), 44.06\left(\mathrm{CH}_{2}\right), 44.04\left(\mathrm{CH}_{2}\right), 43.98\left(\mathrm{CH}_{2}\right), 43.86\left(\mathrm{CH}_{2}\right), 43.93\left(\mathrm{CH}_{2}\right)$, $43.90\left(\mathrm{CH}_{2}\right), 36.62\left(\mathrm{C}^{\prime}\right), 36.59\left(\mathrm{C}^{\prime}\right), 36.54\left(\mathrm{C}^{\prime}\right), 36.51\left(\mathrm{C}^{\prime}\right), 29.39\left(\mathrm{CH}_{2}\right), 29.34\left(\mathrm{CH}_{2}\right)$, $29.29\left(\mathrm{CH}_{2}\right), 29.25\left(\mathrm{CH}_{2}\right), 29.16\left(\mathrm{CH}_{2}\right), 29.10\left(\mathrm{CH}_{3} / \mathrm{CH}\right), 29.08\left(\mathrm{CH}_{2}\right), 29.06\left(\mathrm{CH}_{2}\right), 25.76$ $\left(\mathrm{CH}_{3}\right), 25.65\left(\mathrm{CH}_{3}\right), 25.62\left(\mathrm{CH}_{3}\right), 25.47\left(\mathrm{CH}_{3}\right), 25.38\left(\mathrm{CH}_{3}\right), 23.34\left(\mathrm{CH}_{2}\right), 23.29\left(\mathrm{CH}_{2}\right)$, $23.27\left(\mathrm{CH}_{2}\right), 23.23\left(\mathrm{CH}_{3} / \mathrm{CH}\right), 23.22\left(\mathrm{CH}_{3} / \mathrm{CH}\right), 23.15\left(\mathrm{CH}_{3} / \mathrm{CH}\right), 23.11\left(\mathrm{CH}_{3} / \mathrm{CH}\right), 23.08$ $\left(\mathrm{CH}_{3} / \mathrm{CH}\right)$, $23.05\left(\mathrm{CH}_{3} / \mathrm{CH}\right)$, $23.03\left(\mathrm{CH}_{3} / \mathrm{CH}\right)$, $22.13\left(\mathrm{CH}_{3} / \mathrm{CH}\right), 22.06\left(\mathrm{CH}_{3} / \mathrm{CH}\right), 21.87$ $\left(\mathrm{CH}_{3} / \mathrm{CH}\right), 21.85\left(\mathrm{CH}_{3} / \mathrm{CH}\right), 21.79\left(\mathrm{CH}_{3} / \mathrm{CH}\right), 21.75\left(\mathrm{CH}_{3} / \mathrm{CH}\right), 14.30\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $14.28\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $14.26\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu).
$\mathbf{M S}\left(\mathbf{E S}+\right.$ ) $\mathbf{m} / \mathbf{z}$ found $1064.21\left[\mathrm{M}+\mathrm{H}^{+}\right], \mathrm{C}_{52} \mathrm{H}_{68} \mathrm{ClN}_{7} \mathrm{O}_{11} \mathrm{P}_{2}$ required $\mathrm{m} / \mathrm{z} 1064.54$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $80 / 20$ to $1 / 90$ in 15 minutes, then to $0 / 100$ in $35 \mathrm{~min}, \mathrm{~F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=280 \mathrm{~nm}$, broad peak with tR 21.98 min .

## [90b] 8-Chloro-2'-deoxyadenosine-3',5'-bis-O-phenyl-(hex-1-yloxy-Lalaninyl)phosphate



Prepared according to general procedure $\mathbf{F}_{\mathbf{1}}$ using 8-chloro-2'-deoxyadenosine [87], $(0.10 \mathrm{~g}, 0.35 \mathrm{mmol})$ in THF ( 10 mL ), $t \mathrm{BuMgCl}(1.0 \mathrm{M}$ in THF, 0.52 mL , 0.52 mmol ) and phenyl-(hexyl-1-oxy-Lalaninyl)phosphorochloridate [17u] (0.36 g, 1.05 mmol ) in THF ( 2 mL ). Purification by Biotage Isolera One ( 50 g SNAP cartridge, $50 \mathrm{~mL} / \mathrm{min}$, gradient eluent system $1-10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 10 \mathrm{CV}, 10 \% 5 \mathrm{CV}$ ) and prep TLC ( $2000 \mu \mathrm{~m}$, $5 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded the title compound as a white solid ( $0.089 \mathrm{~g}, 28 \%$ ).
${ }^{\mathbf{3 1}} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 3.55,3.52,3.49,3.26,3.14,2.89,2.84$.
${ }^{1} \mathbf{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathbf{C D}_{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}}$ 8.21-8.12 (m, 1H, H2), 7.42-7.07 (m, 10H, Ph), 6.536.45 (m, 0.5H, H1'), 6.42-6.36 (m, 0.5H, H1'), 5.69-5.53 (m, 1H, H3'), 4.53-4.43 (m, 1H, H5'), 4.41-4.26 (m, 2H, H4', H5'), 4.15-3.93 (m, 6H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex, $\mathrm{CHCH}_{3}$ L-Ala), 3.91-3.83 (m, 0.5H, H2'), 3.81-3.61 (m, 0.5H, H2'), 2.82-2.72 (m, 0.5H, H2'), 2.66-2.57 (m, 0.5H, H2'), 1.65-1.48 (m, 4H, $2 \times \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex), 1.43-1.18 (m, $18 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex, $\mathrm{CHCH}_{3}$ L-Ala), 0.91-0.82 (m, 6 H , $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex).

## [90c] <br> 8-Chloro-2'-deoxyadenosine-3',5'-bis-O-phenyl-(benzyloxy-L-alaninyl) phosphate



Prepared according to general procedure $\mathbf{F}_{\mathbf{1}}$ using 8-chloro-2'-deoxyadenosine [87], ( $0.10 \mathrm{~g}, 0.35 \mathrm{mmol})$ in THF ( 10 mL ), $t \mathrm{BuMgCl}(1.0 \mathrm{M}$ in THF, $0.38 \mathrm{~mL}, 0.38$ mmol) and phenyl-(benzyloxy-L-alaninyl) phosphorochloridate [17s] ( $0.25 \mathrm{~g}, 0.7 \mathrm{mmol}$ ) in THF $(1.5 \mathrm{~mL})$. Purification by Biotage Isolera One $(50 \mathrm{~g}$ SNAP cartridge, $50 \mathrm{~mL} / \mathrm{min}$, gradient eluent system 1$10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 10 \mathrm{CV}, 10 \% 5 \mathrm{CV}$ ) and prep TLC ( $2000 \mu \mathrm{~m}, 5 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded the title compound as a white solid ( $0.08 \mathrm{~g}, 25 \%$ ). ${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 3.49,3.43,3.42,3.39,3.30,3.18$.
${ }^{1} \mathbf{H}$ NMR ( $500 \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{H}} 8.17(\mathrm{~s}, 0.25 \mathrm{H}, \mathrm{H} 2), 8.158(\mathrm{~s}, 0.25 \mathrm{H}, \mathrm{H} 2), 8.152(\mathrm{~s}$, $0.25 \mathrm{H}, \mathrm{H} 2$ ), 8.13 (s, $0.25 \mathrm{H}, \mathrm{H} 2$ ), 7.39-7.06 (m, 20H, Ar), 6.44-6.40 (m, 0.5H, H1'), 6.366.32 (m, 0.5H, H1'), 5.63-5.53 (m, 1H, H3'), 5.15-5.00 (m, 4H, CH2 Ph, 4.47-4.38 (m, 1H, H5'), 4.34-4.22 (m, 2H, H5', H4'), 4.14-4.04 (m, 1H, CHCH ${ }_{3}$ L-Ala), 3.98-3.80 (m, 1H, $\mathrm{CHCH}_{3}$ L-Ala), 3.69-3.53 (m, 1H, H2'), 2.68-2.60 (m, 0.5H, H2'), 2.57-2.50 (m, 0.5H, H2'), 1.44-1.17 (m, 6H, $\mathrm{CHCH}_{3}$ L-Ala).
${ }^{13} \mathbf{C}$ NMR ( $\left.\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathrm{C}} 174.85(\mathrm{CO}), 174.84(\mathrm{CO}), 174.81$ (CO), 174.72 (CO), 174.69 (CO), 174.67 (CO), 174.57 (CO), 174.53 (CO), 156.30 (C6), 156.28 (C6), 156.27 (C6), 154.00 (C2), 153.94 (C2), 153.92 (C2), 152.16 (C-Ar), 152.15 (C-Ar), 152.12 (CAr), 152.11 (C-Ar), 152.07 (C-Ar), 152.04 (C-Ar), 152.99 (C-Ar), 151.51 (C4), 151.49 (C4), 151.44 (C4), 151.41 (C4), 139.54(C-Ar), 139.53(C-Ar), 139.48(C-Ar), 139.45(CAr), $137.32(\mathrm{C}-\mathrm{Ar}), 137.26(\mathrm{C}-\mathrm{Ar}), 137.24(\mathrm{C}-\mathrm{Ar}), 137.22(\mathrm{C}-\mathrm{Ar}), 137.20(\mathrm{C}-\mathrm{Ar}), 130.91$ (CH-Ar), 130.68 (CH-Ar), 129.65 (CH-Ar), 129.57 (CH-Ar), 129.39 (CH-Ar), 129.38 (CH-Ar), 129.32 (CH-Ar), 129.31 (CH-Ar), 129.26 (CH-Ar), 129.24 (CH-Ar), 126.37 (CH-Ar), 126.34 (CH-Ar), 126.07 (CH-Ar), 126.02 (CH-Ar), 121.67 (CH-Ar), 121.64
(CH-Ar), 121.60 (CH-Ar), 121.57 (CH-Ar), 121.56 (CH-Ar), 121.49 (CH-Ar), 121.46 (CH-Ar), 121.40 (CH-Ar), 121.37 (CH-Ar), 121.34 (CH-Ar), 119.68 (C5), 119.65 (C5),
 85.07 (C4'), 85.00 (C4'), 84.92 (C4'), 84.85 (C4'), 84.78 (C4'), 84.72 (C4'), 78.01 (C3'), 77.96 ( C 3 '), 77.94 ( C 3 '), $77.90\left(\mathrm{C} 3\right.$ '), 77.86 ( $\left.\mathrm{C}^{\prime}\right)$ ), $67.94\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.93\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.88$ $\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.86\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 66.99\left(\mathrm{~d},{ }^{2} J_{C P}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $66.94\left(\mathrm{~d},{ }^{2} J_{C P}=5.4 \mathrm{~Hz}, \mathrm{C} 5^{\prime}\right)$, $66.57\left(\mathrm{~d},{ }^{2} J_{C P}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ ) $), 66.53\left(\mathrm{~d},{ }^{2} J_{C P}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $51.92\left(\mathrm{CHCH}_{3}\right.$ L-Ala), 51.91 $\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $51.83\left(\mathrm{CHCH}_{3}\right.$ L-Ala), 51.81 ( $\left.\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 36.41$ (C2’), 36.39 (C2’), 36.37 ( $\mathrm{C}^{\prime}$ '), 36.35 ( $\mathrm{C}^{\prime}$ ), 36.33 ( $\mathrm{C}^{\prime}$ ), 36.30 ( $\mathrm{C}^{\prime}$ ), 36.27 ( $\mathrm{C}^{\prime}$ ), 20.58 ( $\mathrm{CHCH}_{3}$ L-Ala), $20.56\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 20.52\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 20.47\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 20.45\left(\mathrm{CHCH}_{3} \mathrm{~L}-\right.$ Ala), $20.43\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 20.39\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 20.38\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$.
$\mathbf{M S}\left(\mathbf{E S}+\right.$ ) $\mathbf{m} / \mathbf{z}$ found $920.2\left[\mathrm{M}+\mathrm{H}^{+}\right], \mathrm{C}_{42} \mathrm{H}_{44} \mathrm{ClN}_{7} \mathrm{O}_{11} \mathrm{P}_{2}$ required $\mathrm{m} / \mathrm{z} 920.24$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \boldsymbol{\lambda}=280 \mathrm{~nm}$, three broad peaks with $\mathrm{tR} 21.97 \mathrm{~min}, 22.26 \mathrm{~min}, 22.59 \mathrm{~min}$.


Prepared according to general procedure $\mathbf{F}_{\mathbf{1}}$ using 8-chloro-2'-deoxyadenosine [87], ( $0.10 \mathrm{~g}, 0.35 \mathrm{mmol}$ ) in THF ( 10 mL ), $t \mathrm{BuMgCl}(1.0 \mathrm{M}$ in THF, 0.38 mL , 0.38 mmol) and naphth-1-yl-(benzyloxy-Lalaninyl)phosphorochloridate [17a] (0.15 g, 0.38 mmol ) in THF ( 1.5 mL ). Purification by Biotage Isolera One ( 50 g SNAP cartridge, $50 \mathrm{~mL} / \mathrm{min}$, gradient eluent system $1-10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 10 \mathrm{CV}$,
$10 \% 5 \mathrm{CV})$ and prep TLC $\left(2000 \mu \mathrm{~m}, 5 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ yielded the title compound as a white solid ( $0.056 \mathrm{~g}, 16 \%$ ).
${ }^{\mathbf{3 1}} \mathbf{P}$ NMR (202 MHz, CD $\mathbf{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 4.96,4.89,4.80,4.71,4.68,4.61,4.59,4.46$.
${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.24-8.19$ (m, 1H, Nap), 8.10-8.02 (m, 2H, H2, Nap), 7.95-7.77 (m, 2H, Nap, Ph), 7.73-7.66 (m, 1H, Nap), 7.62-7.15 (m, 19H, Nap, Ph), 6.336.13 (m, 1H, H1'), 5.75-5.60 (m, 1H, H3'), 5.14-4.96 (m, 4H, CH ${ }_{2} \mathrm{Ph}$ ), 4.59-4.26 (m, 3H, H5', H4'), 3.95-3.83 (m, 2H, CH2 Gly), 3.78-3.51 (m, 3H, CH2 Gly, H2'), 2.60-2.41 (m, 1H, H2').
[91a] 8-Chloro-2'-deoxyadenosine-5'-O-naphth-1-yl-(cyclohexyloxy-Lleucinyl)phosphate


Prepared according to generale procedure $\mathbf{F}_{2}$ using 8-chloro-2'-deoxyadenosine [87], $(0.05 \mathrm{~g}, 0.17 \mathrm{mmol})$ in THF ( 1.5 mL ), $N$-methylimidazole ( $0.06 \mathrm{ml}, 0.87$ mmol ) and naphth-1-yl-(cyclohexyloxy-Lleucinyl)phosphorochloridate ( $230 \mathrm{mg}, 0.52 \mathrm{mmol}$ ) in THF ( 1 mL ). Purification by Biotage Isolera One ( 10 g SNAP cartridge, $10 \mathrm{~mL} / \mathrm{min}$, gradient eluent system $1-10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 10 \mathrm{CV}, 10 \%$ 5 CV ) and prep TLC ( $500 \mu \mathrm{~m}, 5 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded the title compound as a white solid ( $0.002 \mathrm{~g}, 2 \%$ ).
${ }^{31} \mathbf{P} \mathbf{N M R}\left(\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 4.11, 4.04.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.14-8.06 (m, 2H, H2, Nap), 7.89-7.84 (m, 1H, Nap), 7.70-7.65 (m, 1H, Nap), 7.55-7.45 (m, 2H, Nap), 7.44-7.40 (m, 1H, Nap), 7.39-7.31 (m, $1 \mathrm{H}, \mathrm{Nap}), 6.499\left(\mathrm{t}, J=7.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{Hl}\right.$ '), $6.49\left(\mathrm{t}, J=6.8 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H}{ }^{\prime}\right)$ ), 4.81-4.75 (m, 1H, H3'), 4.65-4.56 (m, 1H, CH( $\left.\left.\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), ~ 4.53-4.42$ (m, 1.5H, H5'), 4.36-4.29 (m, $0.5 \mathrm{H}, \mathrm{H} 5$ '), 4.23-4.18 (m, 0.5H, H4'), 4.18-4.12 (m, 0.5H, H4'), 3.90-3.81 (m, 1H, $\left.\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}-\mathrm{Leu}\right), 3.54-3.43$ ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{H} 2$ '), 2.24-2.32 (m, 1H, H2'), 1.76-1.59 (m, $\left.4 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 1.53-1.22\left(\mathrm{~m}, 9 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 0.85$0.62\left(\mathrm{~m}, 6 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu).
[91b] 8-Chloro-2'-deoxyadenosine-5'-O-phenyl-(benzyloxy-glycinyl)phosphate


Prepared according to generale procedure $\mathbf{F}_{2}$ using 8-chloro-2'-deoxyadenosine [87], ( $0.025 \mathrm{~g}, 0.08 \mathrm{mmol}$ ) in THF ( 1.0 mL ), $N$-methylimidazole ( $0.03 \mathrm{ml}, 0.46$ mmol ) and phenyl-(benzyloxyglycinyl)phosphorochloridate [17r] (0.085 g, 0.26 mmol ) in THF ( 1 mL ). Purification by Biotage Isolera One ( 10 g SNAP cartridge, $10 \mathrm{~mL} / \mathrm{min}$, gradient eluent system $1-10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ $10 \mathrm{CV}, 10 \% 5 \mathrm{CV}$ ) and prep TLC ( $500 \mu \mathrm{~m}, 5 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) yielded the title compound as a white solid ( $0.002 \mathrm{~g}, 4 \%$ )
${ }^{\mathbf{3 1}} \mathbf{P} \mathbf{N M R}\left(\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 4.61, 4.52.
${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.19$ (s, $\left.0.5 \mathrm{H}, \mathrm{H} 2\right), 8.16(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 2), 7.37-7.26(\mathrm{~m}, 7 \mathrm{H}$, Ph), 7.18-7.10 (m, 3H, Ph), 6.50-6.45 (m, 1H, Hl'), 5.15-5.08 (m, 2H, CH ${ }_{2} \mathrm{Ph}$ ), 4.82-4.78
(m, 1H, H3'), 4.46-4.39 (m, 1H, H5'), 4.37-4.28 (m, 1H, H5'), 4.15-4.10 (m, 1H, H4'), 3.76-3.67 (m, 2H, CH2 Gly), 3.56-3.46 (m, 1H, H2'), 2.41-2.33 (m, 1H, H2').
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 156.32$ (C6), 153.94 (C2), 151.60 (C4), 149.01 (C8), 139.62 (C-Ar), 137.22 (C-Ar), 134.64 (CH-Ar), 132.88 (CH-Ar), 130.68 (CH-Ar), 129.56 (CH-Ar), 129.36 (CH-Ar), 129.34 (CH-Ar), 129.31 (CH-Ar), 128.77 (C-Ar), 126.07 (CHAr), 121.42 ( $\mathrm{CH}-\mathrm{Ar}$ ), 121.38 ( $\mathrm{CH}-\mathrm{Ar}$ ), 121.36 ( $\mathrm{CH}-\mathrm{Ar}$ ), 119.48 (C5), 86.81 (d, ${ }^{3} J_{C P}=8.1$ $\mathrm{Hz}, \mathrm{C} 4$ '), 86.67 ( $\mathrm{d},{ }^{3} J_{\mathrm{CP}}=8.4 \mathrm{~Hz}, \mathrm{C} 4$ '), 86.53 ( $\mathrm{C}^{\prime}$ ), 86.50 ( $\mathrm{C}^{\prime}$ ), 72.51 ( $\left.\mathrm{C} 3^{\prime}\right), 72.33$ ( $\left.\mathrm{C} 3^{\prime}\right)$ ), $67.90\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.87\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.77\left(\mathrm{~d},{ }^{2} J_{C P}=5.6 \mathrm{~Hz}, \mathrm{C} 5\right.$ ' $), 67.35\left(\mathrm{~d},{ }^{2} J_{C P}=5.0 \mathrm{~Hz}\right.$, C5'), 43.78 ( $\mathrm{CH}_{2}$ Gly), 43.75 ( $\mathrm{CH}_{2}$ Gly), $37.80\left(\mathrm{C} 2^{\prime}\right), 37.77$ ( $\mathrm{C}^{\prime}$ ).

## [95a] 8-Chloro-2'-deoxyadenosine-5'-O-naphth-1-yl-(pentyl-1-oxy-Lleucinyl)phosphate



8-Chloro-2'-deoxyadenosine-3',5'-bis-O-naphth-1-yl(pentyl-1-oxy-L-leucinyl)phosphate [90a] ( $0.03 \mathrm{~g}, 0.03$ mmol ) was dissolved in THF ( 2.5 mL ), and zirconocene hydrochloride (Schwartz reagent) was added ( 0.023 g , 0.09 mmol ). The mixture was stirred at room temperature for 3 hours, quenched with water ( 0.2 mL ) and evaporated. The crude was purified via Biotage Isolera One (10g ZIP cartridge, 10 $\mathrm{mL} / \mathrm{min}$, gradient eluent system $1-12 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 10 \mathrm{CV}, 12 \% 2 \mathrm{CV}$ ) to afford the title compound as a white solid ( $0.003 \mathrm{~g}, 14 \%$ ).
${ }^{31} \mathbf{P}$ NMR (202 MHz, $\left.\mathbf{C D}_{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}} 4.03,4.01$.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.03-7.93 (m, 2H, Nap, H2), 7.77-7.72 (m, 1H, Nap), 7.58-7.53 (m, 1H, Nap), 7.43-7.19 (m, 4H, Nap), 6.39-6.34 (m, 1H, H1'), 4.70-4.64 (m, $1 \mathrm{H}, \mathrm{H} 3$ '), 4.40-4.30 (m, 1.5H, H5'), 4.23-4.16 (m, 0.5H, H5'), 4.10-4.06 (m, 0.5H, H4'), 4.05-4.01 (m, $\left.0.5 \mathrm{H}, \quad \mathrm{H}{ }^{\prime}\right), \quad 3.84-3.68 \quad\left(\mathrm{~m}, \quad 3 \mathrm{H}, \quad \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 3.42-3.30 (m, 1H, H2'), 2.29-2.20 (m, 1H, H2'), 1.55-1.06 (m, $9 \mathrm{H}, \quad \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n$-Pen, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \quad$ L-Leu), $0.77-0.56 \quad(\mathrm{~m}, ~ 9 \mathrm{H}$, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu).
${ }^{13} \mathbf{C}$ NMR ( $\left.\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{C}} 175.27$ (CO), 175.17 (CO), 175.03 (CO), 174.87 (CO), 156.02 (C6), 155.88 (C6), 153.86 (C2), 153.85 (C2), 151.61 (C4), 151.50 (C4), 149.02 (CAr), 148.97 (C8), 148.88 (C8), 147.97 (C-Ar), 147.80 (C-Ar), 147.75 (C-Ar), 144.47 (CAr), 144.44 (C-Ar), 144.29 (C-Ar), 144.25 (C-Ar), 139.53 (C-Ar), 139.51 (C-Ar), 136.29 (C-Ar), 136.23 (C-Ar), 128.84 (CH-Ar), 128.79 (CH-Ar), 127.87 (C-Ar), 127.78 (C-Ar), 127.37 (CH-Ar), 127.31 (CH-Ar), 126.42 (d, $\left.{ }^{2} J_{C P}=8.5 \mathrm{~Hz}, 5.0 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 126.40\left(\mathrm{~d},{ }^{2} J_{C P}\right.$
$=8.5 \mathrm{~Hz}, 5.0 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}), 125.81$ (CH-Ar), $125.80(\mathrm{CH}-\mathrm{Ar}), 122.76$ (CH-Ar), 122.71 (CHAr), 119.59 (C5), 119.46 (C5), $116.09\left(\mathrm{~d},{ }^{3} J_{C P}=3.75 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 115.02\left(\mathrm{~d},{ }^{3} J_{C P}=3.7\right.$ $\mathrm{Hz}, \mathrm{CH}-\mathrm{Ar}), 86.91$ (d, ${ }^{3} J_{C P}=8.4 \mathrm{~Hz}, \mathrm{C} 4$ '), 86.77 (d, ${ }^{3} J_{C P}=8.5 \mathrm{~Hz}, \mathrm{C} 4$ '), 86.57 (C1'), $86.44\left(\mathrm{Cl}^{\prime}\right), 72.57\left(\mathrm{C} 3\right.$ '), 72.46 ( C 3 '), $68.11\left(\mathrm{~d},{ }^{2} J_{C P}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.77\left(\mathrm{~d},{ }^{2} J_{C P}=5.6\right.$ Hz, C5'), $66.29\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 66.25\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 54.65\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), 54.63 $\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $44.23\left(\mathrm{~d},{ }^{3} J_{C P}=7.4 \mathrm{~Hz}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $44.06(\mathrm{~d}$, ${ }^{3} J_{C P}=8.2 \mathrm{~Hz}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), $37.81\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen $), 37.76$ $\left(C_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen), $29.29 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen), 29.28 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen), $\quad 29.12 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen), 29.08 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $25.56\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu) $25.45\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $23.31\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen $), 23.28\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), 23.06 $\left(\mathrm{CH}_{3} \mathrm{~L}-\mathrm{Leu} / n\right.$-Pen), $23.01\left(\mathrm{CH}_{3} \mathrm{~L}-\mathrm{Leu} / n\right.$-Pen $), 22.03\left(\mathrm{CH}_{3} \mathrm{~L}-\mathrm{Leu} / n\right.$-Pen $), 21.90\left(\mathrm{CH}_{3} \mathrm{~L}-\right.$ $\mathrm{Leu} / n$-Pen), $14.25\left(\mathrm{CH}_{3} \mathrm{~L}-\mathrm{Leu} / n\right.$-Pen $)$.

MS (ES+ $) \mathbf{m} / \mathbf{z}$ found $676.2\left[\mathrm{M}+\mathrm{H}^{+}\right], \mathrm{C}_{31} \mathrm{H}_{40} \mathrm{ClN}_{6} \mathrm{O}_{7} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 675.11$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=280 \mathrm{~nm}$, two peaks with tR 21.50 , tR 21.02 min .
[95b] 8-Chloro-2'-deoxyadenosine-5'-O-phenyl-(benzyloxy-L-alaninyl)phosphate


8-Chloro-2'-deoxyadenosine-3',5'-bis- $O$ -phenyl(benzyloxy-L-alaninyl)phosphate [90c] (0.08 $\mathrm{g}, 0.09 \mathrm{mmol}$ ) was dissolved in THF ( 5 mL ), and zirconocene hydrochloride (Schwartz reagent) was added $(0.067 \mathrm{~g}, 0.26 \mathrm{mmol})$. The mixture was stirred at room temperature for 3 hours, quenched with water ( 0.5 mL ) and evaporated. The crude was purified via Biotage Isolera One ( 80 g ZIP cartridge, $50 \mathrm{~mL} / \mathrm{min}$, gradient eluent system $1-12 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 10 \mathrm{CV}, 12 \% 2 \mathrm{CV}$ ) to afford the title compound as a white solid $(0.008 \mathrm{~g}, 15 \%)$.
${ }^{\mathbf{3 1}} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 3.53, 3.43.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O}: \boldsymbol{\delta}_{\mathbf{H}} 8.18$ ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 2$ ), 8.15 ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 2$ ), 7.37-7.25 (m, 7H, Ph), 7.18-7.07 (m, 3H, Ph), 6.50-6045 (m, 1H, H1'), 5.13-5.03 (m, 2H, CH $\mathrm{H}_{2} \mathrm{Ph}$ ), 4.94-4.86 (m, 1H, H3'), 4.42-4.36 (m, 1H, H5'), 4.31-4.22 (m, 1H, H5'), 4.14-4.08 (m, 1H, H4'), 3.98-3.91 (m, $0.5 \mathrm{H}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ), $3.90-3.81$ (m, $0.5 \mathrm{H}, \mathrm{CHCH}_{3}$ L-Ala), 3.57-3.44 (m, 1H, H2'), 2.41-2.34 (m, 1H, H2'), 1.32-1.22 (m, 3H, CHCH ${ }_{3}$ L-Ala).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 174.70$ (C=O), 174.54 ( $\mathrm{C}=\mathrm{O}$ ), 156.31 (C6), 156.26 (C6), 153.92 (C2), 153.88 (C2), 152.16 (C4), 152.10 (C4), 151.57 (C8), 151.52 (C8),
139.18 (C-Ar), 137.46 (C-Ar), 137.30 (C-Ar), 137.21 (C-Ar), 130.66 (CH-Ar), 130.64 (CH-Ar), 129.54 (CH-Ar), 129.29 (CH-Ar), 129.27 (CH-Ar), 129.25 (CH-Ar), 126.02 (CH-Ar), $121.42\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=4.9 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 121.34\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=4.9 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 86.65(\mathrm{~d}$, ${ }^{3} J_{\mathrm{CP}}=8.4 \mathrm{~Hz}, \mathrm{C} 4$ '), $86.60\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=8.4 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $86.45\left(\mathrm{C1}{ }^{\prime}\right), 72.50\left(\mathrm{C} 3^{\prime}\right), 72.19\left(\mathrm{C} 3^{\prime}\right)$, $67.93\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.86\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.78\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.20\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=5.4 \mathrm{~Hz}\right.$, C5'), $51.55\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $51.53\left(\mathrm{CHCH}_{3}\right.$ L-Ala), 37.80 (C2'), 37.71 ( $\left.\mathrm{C} 2^{\prime}\right), 20.39(\mathrm{~d}$, ${ }^{3} J_{\mathrm{CP}}=6.4 \mathrm{~Hz}, \mathrm{CHCH}_{3}$ L-Ala), $20.30\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=6.4 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$.
MS (ES+ $) \mathbf{m} / \mathbf{z}$ found $625.1\left[\mathrm{M}+\mathrm{Na}^{+}\right], 603.1\left[\mathrm{M}+\mathrm{H}^{+}\right], \mathrm{C}_{24} \mathrm{H}_{24} \mathrm{ClN}_{6} \mathrm{O}_{7} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 588.13$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=280 \mathrm{~nm}$, two peaks with $\mathrm{tR} 11.44, \mathrm{tR} 11.10 \mathrm{~min}$.

## [95c] 8-Chloro-2'-deoxyadenosine-5'-O-phenyl-(hex-1-yloxy-L-alaninyl)phosphate



8-Chloro-2'-deoxyadenosine-3',5'-bis-O-phenyl(hex-1-yloxy-L-alaninyl)phosphate [90b] $(0.089 \mathrm{~g}, 0.10$ mmol ) was dissolved in THF ( 5 mL ), and zirconocene hydrochloride (Schwartz reagent) was added ( 0.077 g , $0.30 \mathrm{mmol})$. The mixture was stirred at room temperature for 3 hours, quenched with water ( 0.5 mL ) and evaporated. The crude was purified via Biotage Isolera One (10g ZIP cartridge, $10 \mathrm{~mL} / \mathrm{min}$, gradient eluent system $1-12 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 10 \mathrm{CV}, 12 \% 2 \mathrm{CV}$ ) to afford the title compound as a white solid ( $0.003 \mathrm{~g}, 5 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 3.54, 3.49.
${ }^{1} \mathbf{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.07(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 2), 8.04(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 2), 7.22-7.15(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{Ph})$, 7.07-6.99 (m, 3H, Ph), 6.39-6.34 (m, 1H, H1'), 4.70-4.66 (m, 1H, H3'), 4.34-4.26 (m, $1 \mathrm{H}, \mathrm{H}{ }^{\prime}$ ), 4.21-4.12 (m, 1H, H5'), 4.04-3.98 (m, 1H, H4'), 3.95-3.86 (m, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex), 3.81-3.65 (m, $1 \mathrm{H}, \mathrm{CHCH}_{3}$ L-Ala), 3.53-3.33 (m, 1 H , H2'), 2.30-2.23 (m, 1H, H2'), 1.50-1.41 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex), 1.261.10 (m, 9H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex, $\mathrm{CHCH}_{3} \mathrm{~L}$-Ala), $0.83-0.74$ (m, 3 H , $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 156.32$ (C6), 153.92 (C2), 153.89 (C2), 151.37 (C4), 150.99 (C-Ar), 147.54 (C8), 130.67 (CH-Ar), 126.04 (CH-Ar), 121.44 (CH-Ar), 121.37 (CH-Ar), 121.33 (CH-Ar), 119.63 (C5), 86.77 ( $\mathrm{d}^{3}{ }^{3} J_{C P}=8.25 \mathrm{~Hz}, \mathrm{C} 4$ '), 86.49 ( C 1 '), 72.53 (C3'), 72.25 ( $\mathrm{C}^{\prime}$ '), 67.75 ( $\mathrm{d}^{2} J_{C P}=5.5 \mathrm{~Hz}, \mathrm{C} 5$ '), $67.22\left(\mathrm{~d},{ }^{2} J_{C P}=5.5 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), 66.43 $\left(C_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex), $66.28 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex $), 51.54$
$\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $51.49 \quad\left(\mathrm{CHCH}_{3} \quad\right.$ L-Ala), $37.79 \quad$ (C2’), $37.70 \quad$ (C2'), 29.67 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex), $29.64\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex), 26.63 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex $), \quad 26.60 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex $), 23.57$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Hex), $20.49\left(\mathrm{~d},{ }^{3} J_{C P}=7.2 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}\right.$-Ala), $20.40\left(\mathrm{~d},{ }^{3} J_{C P}\right.$ $=7.2 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}$-Ala), $14.35\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Hex $)$.
$\mathbf{M S}\left(\mathbf{E S}+\right.$ ) $\mathbf{~ m} / \mathbf{z}$ found $597.2\left[\mathrm{M}+\mathrm{H}^{+}\right], \mathrm{C}_{25} \mathrm{H}_{34} \mathrm{ClN}_{6} \mathrm{O}_{7} \mathrm{P}$ required m/z 597.00 [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=280 \mathrm{~nm}$, two peaks with tR 12.32 , tR 12.39 min .
[96] 8-Chloro-2'-deoxyadenosine-3'-O-phenyl-(benzyloxy-L-alaninyl)phosphate


Isolated from reaction to afford 8-chloro-2'-deoxyadenosine-5'- $O$ -phenyl-(benzyloxy-L-alaninyl)phosphate [95b] as a white solid (3 $\mathrm{mg}, 5 \%)$.
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 3.29,2.58$.
${ }^{1} \mathbf{H}$ NMR ( $500 \mathbf{M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.05$ ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 2$ ), 8.03 ( $\mathrm{s}, 0.5 \mathrm{H}$, H2), 7.31-7.21 (m, 5H, Ph), 7.20-7.08 (m, 5H, Ph), 6.29 (dd, $J=6.0$ $\mathrm{Hz}, 8.5 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1$ '), 6.17 (dd, $J=6.0 \mathrm{~Hz}, 8.5 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1$ '), 5.31-5.26 (m, 0.5H, H3'), 5.23-5.19 (m, 0.5H, H3'), 5.10-5.01 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}$ ), 4.16-4.13 (m, 0.5H, H4'), 4.11-4.07 (m, 0.5H, H4'), 4.00-3.92 (m, 1 H , $\mathrm{CHCH}_{3}$ L-Ala), 3.74-3.68 (m, 1H, H5'), 3.67-3.60 (m, 1H, H5'), 3.13-3.02 (m, 1H, H2'), 2.49-2.43 (m, 0.5H, H2'), 2.39-2.36 (m, 0.5H, H2'), 1.32-1.29 (m, 1.5H, CHCH ${ }_{3}$ L-Ala), 1.26-1.23 (m, 1.5H, $\mathrm{CHCH}_{3}$ L-Ala).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 174.92\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=4.5 \mathrm{~Hz}, \mathrm{CO}\right), 174.69\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=4.4 \mathrm{~Hz}\right.$, CO ), 156.77 (C6), 156.76 (C6), 153.53 (C2), 153.51 (C2), 152.11 (C4), 152.06 (C4), 150.79 (C8), 150.76 (C8), 138.91 (C-Ar), 138.86 (C-Ar), 137.37 (C-Ar), 137.3 (C-Ar), 130.92 (CH-Ar), 130.88 (CH-Ar), 129.65 (CH-Ar), 129.54 (CH-Ar), 129.39 (CH-Ar), 129.28 (CH-Ar), 129.26 (CH-Ar), 126.37 (CH-Ar), 121.63 (d, ${ }^{3} J_{\mathrm{CP}}=4.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), $121.55\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=4.7 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 88.64\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=5.3 \mathrm{~Hz}, \mathrm{C} 4\right.$ ) $), 88.45\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=5.3 \mathrm{~Hz}\right.$, $\mathrm{C} 4 '), 87.63(\mathrm{C} 1 '), 87.54\left(\mathrm{Cl}^{\prime}\right), 79.95\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=5.2 \mathrm{~Hz}, \mathrm{C} 3 '\right), 68.01\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.97$ $\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 63.62\left(\mathrm{C} 5\right.$ '), $51.91\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=1.9 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.72\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$, 38.43 (d, ${ }^{3} J_{\mathrm{CP}}=4.4 \mathrm{~Hz}, \mathrm{C} 2$ '), $38.27\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=4.4 \mathrm{~Hz}, \mathrm{C} 2{ }^{\prime}\right), 20.37\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.4 \mathrm{~Hz}\right.$, $\mathrm{CHCH}_{3}$ L-Ala), 20.28 (d, ${ }^{3} J_{\mathrm{CP}}=7.4 \mathrm{~Hz}, \mathrm{CHCH}_{3}$ L-Ala).
MS (ES+ $\mathbf{~ m / z}$ found $611.1\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{24} \mathrm{H}_{24} \mathrm{ClN}_{6} \mathrm{O}_{7} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 588.13$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 35 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=280 \mathrm{~nm}$, two peaks with tR 16.38 , tR 16.80 min .

### 9.5.16 Synthesis of CNDAC and sapacitabine

[97] 2'-C-cyano-2'-deoxy-1- $\beta$-D-arabino-pentofuranosylcytosine (CNDAC)
$N^{4}$-Acetyl-1-[3,5-O-(tetraisopropyldisiloxanyl)- $\beta$-D-erythro- $N$ -
 pentofuran-2- $\beta$-cyano]cytosine [106] ( $0.30 \mathrm{~g}, 0.6 \mathrm{mmol}$ ) was dissolved in THF ( 1 mL ), and cooled to $0{ }^{\circ} \mathrm{C}$ in an ice-cold bath. Trifluoroacetic acid $(0.25 \mathrm{~mL})$ and water $(0.25 \mathrm{~mL})$ were added dropwise, and the mixture was stirred at rt for 48 hours. The volatiles were evaporated and the mixture was purified by column chromatography (eluent $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 5-25 \%$ ), affording the title compound as a white solid ( $0.11 \mathrm{~g}, 75 \%$ ).
(Lit. $\mathrm{mp}=175-176{ }^{\circ} \mathrm{C}$ )..$^{369}$
H NMR ( $500 \mathrm{MHz}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.13(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6), 6.28(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}$, H1'), 6.05 (d, $J=7.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ ), 4.62-4.56 (m, 1H, H3'), 3.97-3.90 (m, 2H, H5', H4'), 3.84-3.77 (m, 2H, H5', H2').
$\left.{ }^{13} \mathbf{C} \mathbf{( 1 2 5 ~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{C}} 166.58(\mathrm{C} 4), 156.28(\mathrm{C} 2), 142.82(\mathrm{C} 6), 117.86(\mathrm{CN}), 96.30$ (C5), 86.96 (C4'), 85.37 ( $\mathrm{C}^{\prime}$ ), 73.81 ( C 3 '), 60.61 ( C 5 '), 44.62 (C2').
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{OH}$ from $100 / 0$ to $75 / 25$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed one peak with tR 6.79 min .
[98] 2'-C-cyano-2'-deoxy-1- $\beta$-D-arabino-pentofuranosyl- $N^{4}$-palmitoyl-cytosine (sapacitabine)


CNDAC [97] ( $0.04 \mathrm{~g}, 0.16 \mathrm{mmol}$ ) was suspended in 1,4-dioxane ( 1 $\mathrm{mL})$. Water ( 0.04 mL ) and palmitic anhydride $(0.14 \mathrm{~g}, 0.28 \mathrm{mmol})$ were added and the mixture was brought to $90{ }^{\circ} \mathrm{C}$ for 3 hours. The solvent was evaporated, the mixture diluted with $\mathrm{EtOAc}(5 \mathrm{~mL})$ and washed with brine ( $3 \times 5 \mathrm{~mL}$ ). The organic phase was collected, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to yield the title compound as a white solid ( $0.035 \mathrm{mg}, 45 \%$ ). ${ }^{379}$
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{H}} 9.73(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{NH}), 8.27-8.03(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 6), 7.62-7.43$ (m, 1H, H5), 6.26-6.06 (m, 1H, H1'), 4.79-4.62 (m, 1H, H2'), 4.13-3.66 (m, 4H, H3', H4', H5'), 2.32-2.20 (m, 2H, CH2), 1.63-1.49 (m, 2H, CH 2 ), 1.28-1.07 (m, 24H, CH2), 0.84$0.76\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{CH}_{3}\right)$.

MS (ES+) m/z found $491.3\left[\mathrm{M}^{+}+\mathrm{H}^{+}\right], \mathrm{C}_{26} \mathrm{H}_{42} \mathrm{~N}_{4} \mathrm{O}_{5}$ required m/z 490.64 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 45 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed one peak with tR 30.34 min .
[100] 3',5'-O-(tetraisopropyldisiloxanyl)-cytidine


Cytidine ( $5.0 \mathrm{~g}, 20.56 \mathrm{mmol}$ ) was pre-dried by azeotroping with pyridine ( $2 \times 10 \mathrm{~mL}$ ), and suspended in anhydrous pyridine ( 15 mL ) and the flask purged with Argon. 1,3-Dichloro-1,1,3,3tetraisopropyldisiloxane ( $7.1 \mathrm{~mL}, 22.20 \mathrm{mmol}$ ) was added dropwise at room temperature. A heavy white precipitate gradually settled at the bottom of the flask, and it was broken up to a suspension, by vigorous stirring overnight. The mixture was poured into water ( 100 mL ) and extracted with AcOEt ( $3 \times 100 \mathrm{~mL}$ ). The combined organics were washed with brine, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and finally evaporated to give the title compound as a white solid ( $5.70 \mathrm{~g}, 57 \%$ ). ${ }^{369}$ (Lit. mp: $177{ }^{\circ} \mathrm{C}$ ). ${ }^{471}$
${ }^{1} \mathbf{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.14(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6), 6.03(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}$, H5), 5.70 ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H} 1^{\prime}$ ), 4.33-4.16 (m, 4H, H3', H4', H2', H5'), 4.08 (dd, $J=13.6 \mathrm{~Hz}, 2.4$ Hz, 1H, H5'), 1.25-0.94 (m, 28H, iPr).

## [101] $N^{4}$-Acetyl-3',5'-O-(tetraisopropyldisiloxanyl)-cytidine

 $3^{\prime}, 5^{\prime}-O$-(tetraisopropyldisiloxanyl)-cytidine [100] (5.70 g, 11.73 $\mathrm{mmol})$ was dissolved in $\mathrm{EtOH}(40 \mathrm{~mL})$ and $\mathrm{Ac}_{2} \mathrm{O}(1.4 \mathrm{~mL}, 14.77$ mmol ) was added dropwise. The mixture was heated to reflux ( $\mathrm{T}=$ $65{ }^{\circ} \mathrm{C}$ ) for 3 hours and then cooled to room temperature, evaporated, neutralised with $5 \% \mathrm{NaHCO}_{3}$ solution. The mixture was then extracted with $\mathrm{Et}_{2} \mathrm{O} / n \mathrm{Hex}(1: 1,100 \mathrm{~mL})$ and the organic phase washed with brine and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, and evaporated to give the title compound as a white solid $(6.19 \mathrm{~g}, 100 \%) .{ }^{369}$ (Lit. mp: 157-159 $\left.{ }^{\circ} \mathrm{C}\right) .{ }^{471}$
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{H}} 10.07(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{NH}), 8.21(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6), 7.46$ (d, $J=7.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ ), 5.84 ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{H1}$ '), 4.32-4.26 (m, 2H, H3', H5'), 4.24-4.20 (m, 2H, H2', H4'), 4.03 (dd, $J=13.4,2.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ '), 3.02 (br s, 1H OH2'), 2.31 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{Ac}$ ), 1.15-0.97 (m, 28H, iPr).
$N^{4}$-Acetyl-3', $5^{\prime}$ - $O$-(tetraisopropyldisiloxanyl)-cytidine [101] ( 6.19 g ,
 $11.73 \mathrm{mmol})$ was dissolved in anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL})$ and cooled to $0{ }^{\circ} \mathrm{C}$ in an ice-cold bath. Dess-Martin periodinane (10.20 $\mathrm{g}, 24.05 \mathrm{mmol}$ ) was added in small portions, the resulting cloudy solution was stirred with cooling for 10 minutes and then at room temperature overnight. The mixture was cooled to $0{ }^{\circ} \mathrm{C}$ and a second portion of Dess-Martin periodinane ( $4.97 \mathrm{~g}, 11.73 \mathrm{mmol}$ ) was added. The reaction was stirred in the ice cold bath for 10 minutes and then at room temperature for 5 hours. The mixture was diluted with $\mathrm{Et}_{2} \mathrm{O}(100 \mathrm{~mL})$ and washed with $\mathrm{NaHCO}_{3}$ saturated aqueous solution $(100 \mathrm{~mL})$ in which $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{4} \cdot 5 \mathrm{H}_{2} \mathrm{O}(5.70 \mathrm{~g})$ had been dissolved. The aqueous phase was extracted with $\mathrm{Et}_{2} \mathrm{O}(100 \mathrm{~mL})$. The combined organics were washed with $\mathrm{NaHCO}_{3}$ saturated aqueous solution ( 100 mL ), brine $(100 \mathrm{~mL})$, dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated to white foam, which was used without purification for the next step (crude $5.49 \mathrm{~g}, 10.44 \mathrm{mmol}$ ). ${ }^{379}$
[103] $\quad N^{4}$-Acetyl-1-[3,5-O-(tetraisopropyldisiloxanyl)- $\beta$-D-erythro-n-pentofuran-2-hydroxyl-2-cyano]cytosine


Crude [102] from previous synthetic step ( $5.49 \mathrm{~g}, 10.44 \mathrm{~mol}$ ) was dissolved in anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}(60 \mathrm{~mL})$ and cooled to $0{ }^{\circ} \mathrm{C}$ in an ice-cold bath. $\mathrm{AlCl}_{3}(2.09 \mathrm{~g}, 15.66 \mathrm{mmol})$ was added in small portions, and the flask flushed with Argon. TMSCN ( 3.92 mL , 31.32 mmol ) was added dropwise at $0^{\circ} \mathrm{C}$. The mixture was stirred with cooling for 45 minutes and then for further 45 minutes at room temperature. The mixture was cooled in an ice-cold bath and quenched by the addition of $\mathrm{NH}_{4} \mathrm{Cl}$ saturated aqueous solution ( 30 mL ) in a steady stream. The reaction was then transferred into a separating funnel, the organic phase separated from the aqueous phase, which was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 30 \mathrm{~mL})$. The combined organics were washed with brine $(50 \mathrm{~mL})$ and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to give a crude $(5.48 \mathrm{~g}, 9.92$ mmol ) that was used without further purification for the following step. ${ }^{379}$

## [105] O-((6aR,8R,9R,9aR)-8-(4-acetamido-2-oxopyrimidin-1(2H)-yl)-9-cyano-2,2,4,4-

 tetraisopropyltetrahydro-6H-furo[3,2-f][1,3,5,2,4]trioxadisilocin-9-yl) carbonodithioate

Crude [103] ( $5.48 \mathrm{~g}, 9.92 \mathrm{mmol}$ ) and DMAP ( $0.05 \mathrm{~g}, 0.45 \mathrm{mmol}$ ) were dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(60 \mathrm{~mL})$ and the vessel was flushed with Argon. The yellowish solution was cooled in an ice-cold bath. Ethyl dithiochloroformate [109] ( $1.81 \mathrm{~g}, 12.90 \mathrm{mmol}$ ) was diluted with anhydrous dichloromethane ( 4 mL ) and added to the reaction mixture in a steady stream, followed by the dropwise addition of triethylamine ( $2.07 \mathrm{~mL}, 14.85 \mathrm{mmol}$ ). The mixture was stirred 1 hour with cooling and a further 1 hour at room temperature. Water was added $(33 \mathrm{~mL})$ and the organic phase was separated. The aqueous phase was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(33 \mathrm{~mL})$, the combined organics were washed with brine ( 50 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to yield a crude $(5.92 \mathrm{~g}, 9.03 \mathrm{mmol})$ that was used in the following reaction without purification. ${ }^{379}$
[106] $\quad N^{4}$-Acetyl-1-[3,5-O-(tetraisopropyldisiloxanyl)- $\beta$-D-erythro-n-pentofuran-2- $\beta$ cyano]cytosine


Crude [104] ( $5.92 \mathrm{~g}, 9.03 \mathrm{mmol}$ ), was dissolved in isopropyl alcohol ( 65 mL ), 2,4,6-collidine ( $1.55 \mathrm{~mL}, 11.74 \mathrm{mmol}$ ) was added, and the mixture was degassed under vacuum. Lauroyl peroxide $(2.39 \mathrm{~g}, 6.00 \mathrm{mmol})$ was added in three portions over 20 minutes and the vessel flushed with Argon. The reaction was refluxed at 100 ${ }^{\circ} \mathrm{C}$. After 1 hour a second portion of lauroyl peroxide $(0.78 \mathrm{~g}, 2.00$ mmol ) was added and the reaction mixture was stirred at $100{ }^{\circ} \mathrm{C}$ for another hour. The mixture was cooled to room temperature, the solvent was removed in vacuo and the residue was triturated with cold $n \mathrm{Hex}$ and filtered to afford the product as a white solid (3.78 g, 78\%). Overall yield (4 steps) $60 \%$. (Lit. $\mathrm{mp}=209-211{ }^{\circ} \mathrm{C}$ )..$^{369}$

H NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{\mathbf{3}}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 10.50(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{NH}), 8.08(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6), 7.46$ (d, $J=7.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ ), 6.36 (d, $J=6.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1^{\prime}$ ), 4.64 (dd, $J=8.7 \mathrm{~Hz}, 8.2 \mathrm{~Hz}, 1 \mathrm{H}$, H3'), 4.18 (dd, $\left.J=13.1 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5^{\prime}\right), 4.09$ (dd, $J=13.1 \mathrm{~Hz}, 3.3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}^{\prime}$ ), 3.91 (ddd, $\left.J=8.0 \mathrm{~Hz}, 3.3 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}^{\prime}\right), 3.75(\mathrm{dd}, J=8.7 \mathrm{~Hz}, 6.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2$ '), 2.29 (s, 3H, Ac), 1.15-1.03 (m, 28H, iPr). pentofuranosylcytosine

$N^{4}$-Acetyl-1-[3,5-O-(tetraisopropyldisiloxanyl)- $\beta$-D-erythro- $N$ -pentofuran-2- $\beta$-cyano]cytosine [106] ( $3.78 \mathrm{~g}, 7.04 \mathrm{mmol}$ ) was dissolved in anhydrous THF ( 20 mL ) under argon atmosphere, zirconocene hydrochloride (Schwartz reagent) ( $3.63 \mathrm{~g}, 14.08 \mathrm{mmol}$ ) was added and the yellow mixture was stirred for 1 hour at room temperature. A second portion of the Schwartz reagent (1.82 g, 7.04 mmol ) was added and the reaction was stirred for a further 2 hours. The reaction was quenched by addition of water ( 5 mL ) and the aqueous solution was extracted with EtOAc $(2 \times 25 \mathrm{~mL})$ to afford the product as a white solid $(2.61 \mathrm{~g}, 75 \%) .{ }^{369,379}$
${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 7.64(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6), 7.59\left(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{NH}_{2}\right), 6.30$ (d, $\left.J=6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1^{\prime}\right), 6.00\left(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{NH}_{2}\right), 5.85(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5), 4.64$ (dd, $J=$ $\left.8.8,7.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3^{\prime}\right), 4.15-4.03\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}{ }^{\prime}\right.$ ), 3.86-3.81 (m, 1H, H4'), 3.68 (dd, $J=7.9$, $6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2$ '), 1.15-1.01 (m, 28H, iPr).
[109] Ethyl chlorodithioformate


A solution of the ethanethiol $(25.5 \mathrm{~mL}, 353.48 \mathrm{mmol})$ in toluene $(75 \mathrm{~mL})$ at
$10{ }^{\circ} \mathrm{C}$ was added dropwise to a stirred solution of thiophosgene ( $62.1 \mathrm{~g}, 0.54$ $\mathrm{mol})$ in toluene $(50 \mathrm{~mL})$ maintaining the temperature at $10^{\circ} \mathrm{C}$. The reaction mixture was stirred for 16 hours in an ice bath. The solvent was removed and the product fractionated. The title compound was isolated at a temperature of $63^{\circ} \mathrm{C}$, and a pressure of 5.8 mmHg , as a dark orange liquid $(4.89 \mathrm{~g}, 65 \%) .{ }^{386}$
${ }^{1} \mathbf{H} \mathbf{N M R}\left(500 \mathrm{MHz}, \mathbf{C D C l}_{\mathbf{3}}\right) \boldsymbol{\delta}_{\mathbf{H}} 3.23\left(\mathrm{q}, J=7.5 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.40(\mathrm{t}, J=7.5 \mathrm{~Hz}$, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ).
${ }^{13} \mathbf{C}$ NMR ( $\left.\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{C}} 197.30(\mathrm{C}=\mathrm{S}), 35.06\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 11.82\left(\mathrm{CH}_{2} \mathrm{CH}_{3}\right)$.
[110] 3'-O-(tetraisopropyldisiloxanyl)-2'-C-cyano-2'-deoxy-1- $\beta$-D-arabinopentofuranosylcytosine

$3^{\prime}, 5^{\prime}-O$-(tetraisopropyldisiloxanyl)-2'- $C$-cyano-2'-deoxy-1- $\beta$-D-arabino-pentofuranosylcytosine [107] ( $3.03 \mathrm{~g}, 6.12 \mathrm{mmol}$ ) was dissolved in anhydrous THF ( 4 mL ), and cooled to $0^{\circ} \mathrm{C}$ in an icecold bath. Trifluoroacetic acid ( 1 mL ) and water $(1 \mathrm{~mL})$ were added dropwise, and the mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 5 hours.

The volatiles were evaporated and the mixture was purified by column chromatography (eluent EtOAc/nHex 5-25\%), affording the title compound as a white solid ( $1.94 \mathrm{~g}, 62 \%$ ).
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 7.64(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6), 7.59\left(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{NH}_{2}\right), 6.30$ (d, $J=6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Hl}$ '), $6.00\left(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{NH}_{2}\right), 5.85(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5), 4.64$ (dd, $J=$ 8.8, 7.9 Hz, 1H, H3'), 4.15-4.03 (m, 2H, H5'), 3.86-3.81 (m, 1H, H4'), 3.68 (dd, $J=7.9$, $6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2$ '), 1.15-1.01 (m, 28H, iPr).

### 9.5.17 Synthesis of CNDAC ProTides

 pentofuranosylcytosine-5'-O-naphth-1-yl-(pentyl-1-oxy-L-leucinyl)] phosphate

Prepared according to procedure $\mathbf{F}_{\mathbf{1}}$ using $3^{\prime}-O$ -(tetraisopropyldisiloxanyl)-2'-C-cyano-2'-deoxy-1- $\beta$-D-arabino-pentofuranosylcytosine [110] $(0.10 \mathrm{~g}, 0.19 \mathrm{mmol})$, in THF ( 4 mL ), $t \mathrm{BuMgCl}(0.50 \mathrm{~mL}, 0.50 \mathrm{mmol}$ ), naphth-1-yl (pentyl-1-oxy-L-leucinyl) phosphorochloridate [17f] ( 0.24 g , 0.57 mmol ), in THF ( 2 mL ). After evaporation, the crude was purified by Biotage Isolera One ( 10 g SNAP cartridge KP-SIL, $12 \mathrm{~mL} / \mathrm{min}$, using a gradient elution of $1-12 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over $10 \mathrm{CV}, 12 \%$ $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over 5 CV ), yielding the title compound as a white solid ( $0.103 \mathrm{~g}, 60 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 4.32, 4.27.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.24-8.19 (m, 1H, Nap), 7.94-7.89 (m, 1H, Nap), 7.86 (d, $0.5 \mathrm{H}, J=7.3 \mathrm{~Hz}, \mathrm{H} 6$ ), 7.83 (d, $0.5 \mathrm{H}, J=7.3 \mathrm{~Hz}, \mathrm{H} 6$ ), 7.75-7.71 (m, 1H, Nap), 7.597.50 (m, 3H, Nap), 7.47-7.41 (m, 1H, Nap), 6.22 (d, $0.5 \mathrm{H}, J=5.9 \mathrm{~Hz}, \mathrm{H} 1$ '), 6.21 (d, 0.5H, $J=6.1 \mathrm{~Hz}, \mathrm{H} 1$ '), 5.89 (d, $0.5 \mathrm{H}, J=7.6 \mathrm{~Hz}, \mathrm{H} 5$ ), 5.84 (d, $0.5 \mathrm{H}, J=7.6 \mathrm{~Hz}, \mathrm{H} 5), 4.95-4.92$ ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{H} 3^{\mathrm{\prime}}$ ), 4.62-4.48 (m, 1.5H, H5'), 4.46-4.40 (m, 0.5H, H5'), 4.32-4.24 (m, 1H, H4'), 4.01-3.97 (m, 2H, CH2 $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 3.95-3.91 (m, 2H, H2', $\mathrm{CHCH} \mathrm{CH}_{2}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 1.71-0.71 (m, 46H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen, $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu, $\left.\mathrm{OSi}(i \operatorname{Pr})_{2} \mathrm{OSi}(i \operatorname{Pr})_{2} \mathrm{OH}\right)$. pentofuranosylcytosine-5'-O-phenyl-(benzyloxy-L-alaninyl)] phosphate


Prepared according to procedure $\mathbf{F}_{1}$ using $3^{\prime}-O$ -(tetraisopropyldisiloxanyl)-2'- $C$-cyano-2'-deoxy-1- $\beta$-D-arabino-pentofuranosylcytosine [110] ( $0.10 \mathrm{~g}, 0.19 \mathrm{mmol}$ ), in THF ( 4 mL ), $t \mathrm{BuMgCl}(0.50 \mathrm{~mL}, 0.50 \mathrm{mmol})$, naphth-1yl (benzyloxy-L-alaninyl) phosphorochloridate [17a] (0.20 $\mathrm{g}, 0.57 \mathrm{mmol})$, in THF ( 2 mL ). After evaporation, the crude was purified by Biotage Isolera One (10g SNAP cartridge KP-SIL, $12 \mathrm{~mL} / \mathrm{min}$, using a gradient elution of $2-20 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over $10 \mathrm{CV}, 20 \%$ $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over 5 CV ), yielding the title compound as a white solid ( $0.10 \mathrm{~g}, 63 \%$ ).

31P NMR ( $202 \mathrm{MHz}, \mathrm{CD}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{P}}$ 3.64, 3.59.
${ }^{1} \mathbf{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathbf{C D}_{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 7.85(\mathrm{~d}, 0.5 \mathrm{H}, J=7.5 \mathrm{~Hz}, \mathrm{H} 6), 7.80(\mathrm{~d}, 0.5 \mathrm{H}, J=7.6$ $\mathrm{Hz}, \mathrm{H} 6$ ), 7.39-7.30 (m, 7H, Ph), 7.27-7.18 (m, 3H, Ph), 6.22 (d, $0.5 \mathrm{H}, J=6.1 \mathrm{~Hz}, \mathrm{Hl}$ '), $6.19\left(\mathrm{~d}, 0.5 \mathrm{H}, J=6.1 \mathrm{~Hz}, \mathrm{H} l^{\prime}\right), 5.92(\mathrm{~d}, 0.5 \mathrm{H}, J=7.65 \mathrm{~Hz}, \mathrm{H} 5), 5.88(\mathrm{~d}, 0.5 \mathrm{H}, J=7.45$ $\mathrm{Hz}, \mathrm{H} 5), 5.18-5.10$ (m, 2H, CH2 Ph ), 4.96-4.93 (m, 1H, H3'), 4.53-4.45 (m, 1H, H5'), 4.434.33 (m, 1H, H5'), 4.28-4.21 (m, 1H, H4'), 4.08-4.00 (m, 1H, CHCH ${ }_{3}$ L-Ala), 3.93 (dd, J $\left.=6.1,3.4 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 2{ }^{\prime}\right), 3.92$ (dd, $J=6.1,3.4 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 2$ '), 1.39-1.35 (m, 3H, $\mathrm{CHCH}_{3}$ L-Ala), 1.13-0.87 (m, 28H, $\left.\mathrm{OSi}(i P r)_{2} \mathrm{OSi}(i P r)_{2} \mathrm{OH}\right)$.
[111c] 3'-O-(tetraisopropyldisiloxanyl)-2'-C-cyano-2'-deoxy-1- $\beta$-D-arabino-pentofuranosylcytosine-[ethyl 3-(2-hydroxyphenyl)propanoyl (benzyloxy-L-alaninyl)] phosphate


Prepared according to procedure $\mathbf{F}_{1}$ using $3^{\prime}-O$ -(tetraisopropyldisiloxanyl)-2'-C-cyano-2'-deoxy-1- $\beta$-D-arabino-pentofuranosylcytosine [110] $\left(\begin{array}{llll}0.10 & \mathrm{~g}, & 0.19\end{array}\right.$ $\mathrm{mmol})$, in THF ( 4 mL ), $t \mathrm{BuMgCl}(0.50 \mathrm{~mL}, 0.50 \mathrm{mmol})$, ethyl 3-(2-hydroxyphenyl)propanoyl (benzyloxy-L-alanin- $N$-yl) phosphorochloridate $[\mathbf{1 7 v}](0.26 \mathrm{~g}, 0.57$ mmol ), in THF ( 2 mL ). After evaporation, the crude was purified by Biotage Isolera One ( 10 g SNAP cartridge $\mathrm{KP}-\mathrm{SIL}, 12 \mathrm{~mL} / \mathrm{min}$, using a gradient elution of $2-10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over $10 \mathrm{CV}, 10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over 5 CV ), yielding the title compound as a white solid $(0.11 \mathrm{~g}, 70 \%)$.
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 3.73, 3.56.
${ }^{1} \mathbf{H}$ NMR ( $\left.500 \mathbf{M H z}, \mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{H}} 7.83(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 6), 7.78(\mathrm{~d}, 0.5 \mathrm{H}, J=7.5 \mathrm{~Hz}$, H6), 7.38-7.28 (m, 7H, Ph), 7.16-7.07 (m, 2H, Ph), $6.21\left(\mathrm{~d}, 0.5 \mathrm{H}, J=6.0 \mathrm{~Hz}, \mathrm{H} 1^{\prime}\right), 6.18$ (d, $0.5 \mathrm{H}, J=6.0 \mathrm{~Hz}, \mathrm{H} 1$ '), $5.92(\mathrm{~d}, 0.5 \mathrm{H}, J=7.5 \mathrm{~Hz}, \mathrm{H} 5), 5.89(\mathrm{~d}, 0.5 \mathrm{H}, J=7.5 \mathrm{~Hz}, \mathrm{H} 5)$, 5.19-5.00 (m, 2H, CH ${ }_{2} \mathrm{Ph}$ ), 4.94-4.89 (m, 1H, H3'), 4.54-4.33 (m, 2H, H5'), 4.28-4.16 (m, $1 \mathrm{H}, \mathrm{H}^{\prime}$ ), 4.12-4.02 ( $\mathrm{m}, 3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}, \mathrm{CHCH}_{3}$ L-Ala), 3.94-3.89 (m, 1 H , H 2 '), 3.05-2.89 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 2.67-2.54 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.45-1.07 (m, $\left.34 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}, \mathrm{OSi}(i \operatorname{Pr})_{2} \mathrm{OSi}(i P r)_{2} \mathrm{OH}\right)$.
[112b] 2'-C-cyano-2'-deoxy-1- $\beta$-D-arabino-pentofuranosylcytosine-5'-O-phenyl-(benzoxy-L-alaninyl)] phosphate


Compound 111b ( $0.04 \mathrm{~g}, 0.05 \mathrm{mmol}$ ) was dissolved in THF ( 2 mL ), and treated with $\mathrm{KF}(0.028 \mathrm{~g}, 0.3 \mathrm{mmol})$ and 18 -crown-6 ( $0.004 \mathrm{~g}, 0.015 \mathrm{mmol}$ ), and stirred at room temperature. After 3 hours, the mixture was filtered and the solids washed with THF. The filtrate was concentrated, redissolved in ethyl acetate ( 15 mL ), washed with water ( 15 mL ) and brine ( 15 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated. The product was purified by flash chromatography (eluent system $0-7 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ), to afford the title compound as a white solid ( 5 mg , $17 \%)$.
[Alternative procedure] Compound $\mathbf{1 1 1 b}(0.04 \mathrm{~g}, 0.05 \mathrm{mmol})$ was dissolved in EtOAc $(1.5 \mathrm{~mL})$, pyridine ( $20 \mu \mathrm{~L}, 0.23 \mathrm{mmol}$ ) and the mixture was cooled to $0{ }^{\circ} \mathrm{C}$. Then HF-pyridine ( $70 \% \mathrm{HF}, 6 \mu \mathrm{~L}, 0.22 \mu \mathrm{~mol}$ ) was added. The progress of the reaction was monitored by HPLC. After stirring for six hours another portion of pyridine ( $20 \mu \mathrm{~L}, 0.23$ mmol ) and HF-pyridine ( $70 \% \mathrm{HF}, 6 \mu \mathrm{~L}, 0.22 \mu \mathrm{~mol}$ ) was added to the reaction mixture. After stirring for 23 hours an additional portion of pyridine ( $20 \mu \mathrm{~L}, 0.23 \mathrm{mmol}$ ) and HF-pyridine ( $70 \% \mathrm{HF}, 6 \mu \mathrm{~L}, 0.22 \mu \mathrm{~mol}$ ) was added and the stirring was continuing up to a total of 24 hours. After addition of TMSOMe ( $0.21 \mathrm{~mL}, 1.54 \mathrm{mmol}$ ) and stirring for one hour, the solution was evaporated and the crude purified by flash column chromatography (eluent $0-7 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ), and preparative TLC ( $500 \mu \mathrm{~m}, 5 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to yield the title compound as a white solid ( $8 \mathrm{mg}, 30 \%$ ). ${ }^{393}$
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{P}}$ 3.85, 3.69.
${ }^{1} \mathbf{H}$ NMR ( $\left.500 \mathbf{M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 7.76(\mathrm{~d}, 0.5 \mathrm{H}, J=7.5 \mathrm{~Hz}, \mathrm{H} 6), 7.70(\mathrm{~d}, 0.5 \mathrm{H}, J=7.6$
Hz, H6), 7.40-7.31 (m, 7H, Ph), 7.27-7.19 (m, 3H, Ph), 6.27 (d, $0.5 \mathrm{H}, J=7.0 \mathrm{~Hz}, \mathrm{Hl}$ '), $6.23\left(\mathrm{~d}, 0.5 \mathrm{H}, J=7.0 \mathrm{~Hz}, \mathrm{H} 1^{\prime}\right), 5.91(\mathrm{~d}, 0.5 \mathrm{H}, J=7.65 \mathrm{~Hz}, \mathrm{H} 5), 5.87(\mathrm{~d}, 0.5 \mathrm{H}, J=7.65$
$\mathrm{Hz}, \mathrm{H} 5$ ), 5.20-5.11 (m, 2H, CH2 Ph ), 4.58-4.51 (m, 1H, H3'), 4.49-4.26 (m, 2H, H5'), 4.094.01 ( $\mathrm{m}, 2 \mathrm{H}, \mathrm{CHCH}_{3}$ L-Ala, H4'), 3.81-3.75 (m, 1H, H2'), 1.40-1.35 (m, 3H, $\mathrm{CHCH}_{3} \mathrm{~L}-$ Ala).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 174.87\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=3.75 \mathrm{~Hz}, \mathrm{CO}\right), 174.64\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=3.75\right.$ $\mathrm{Hz}, \mathrm{CO}$ ), 167.83 (C4), 167.80 (C4), 152.16 (C2), 152.10 (C2), 141.88 (C6), 141.78 (C6), 137.43 (C-Ar), 137.25 (C-Ar), 130.90 (CH-Ar), 130.12 (CH-Ar), 129.39 (CH-Ar), 129.37 (CH-Ar), 129.34 (CH-Ar), 129.26 (CH-Ar), 129.24 (CH-Ar), 129.64 (CH-Ar), 129.55 (CH-Ar), 126.34 ( $\mathrm{CH}-\mathrm{Ar}$ ), 126.29 ( $\mathrm{CH}-\mathrm{Ar}$ ), 123.84 (CH-Ar), 121.61 (CH-Ar), 121.48 (CH-Ar), 121.44 (CH-Ar), 117.64 (CN), 117.62 (CN), 96.58 (C5), 96.51 (C5), 85.64 (C1'), $85.42\left(\mathrm{C} 1\right.$ ') , $84.47\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=8.1 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $84.37\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=3.7 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $74.45\left(\mathrm{C} 3^{\prime}\right), 74.07$ $\left(\mathrm{C} 3^{\prime}\right), 68.04\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.66\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 65.83\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=4.6 \mathrm{~Hz}, \mathrm{C} 5\right.$ ' $), 65.57\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=4.4\right.$ Hz, C5'), 51.86 ( $\mathrm{CHCH}_{3}$ L-Ala), 51.66 ( $\mathrm{CHCH}_{3}$ L-Ala), 44.50 ( C 2 '), 44.45 ( C 2 '), 20.45 (d, $\left.{ }^{3} J_{\mathrm{CP}}=7.1 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 20.30\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.2 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$.
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \boldsymbol{\lambda}=254 \mathrm{~nm}$, showed two peaks with $\mathrm{tR} 12.52 \mathrm{~min}, 12.75 \mathrm{~min}$.
$\mathbf{M S}\left(\mathbf{E S}+\right.$ ) $\mathbf{m} / \mathbf{z}$ found $570.2\left[\mathrm{M}+\mathrm{H}^{+}\right], 592.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{26} \mathrm{H}_{28} \mathrm{~N}_{5} \mathrm{O}_{8} \mathrm{P}$ required m/z 569.50 [M].
[112c] 2'-C-cyano-2'-deoxy-1- $\beta$-D-arabino-pentofuranosylcytosine-5'-O-phenyl-[ethyl 3-(2-hydroxyphenyl)propanoyl (benzyloxy-L-alaninyl)] phosphate

Compound 111c ( $0.11 \mathrm{~g}, 0.12 \mathrm{mmol}$ ) was dissolved in
 EtOAc ( 4.8 mL ), pyridine ( $48 \mu \mathrm{~L}, 0.55 \mathrm{mmol}$ ) and the mixture was cooled to $0{ }^{\circ} \mathrm{C}$. Then HF-pyridine ( $70 \% \mathrm{HF}$, $14.4 \mu \mathrm{~L}, 0.53 \mu \mathrm{~mol}$ ) was added. The progress of the reaction was monitored by HPLC. After stirring for six hours another portion of pyridine ( $48 \mu \mathrm{~L}, 0.55 \mathrm{mmol}$ ) and HF-pyridine ( $70 \% \mathrm{HF}, 14.4 \mu \mathrm{~L}, 0.53 \mu \mathrm{~mol}$ ) was added to the reaction mixture. After stirring for 23 hours an additional portion of pyridine ( $48 \mu \mathrm{~L}, 0.55 \mathrm{mmol}$ ) and HF-pyridine ( $70 \% \mathrm{HF}, 14.4 \mu \mathrm{~L}, 0.53 \mu \mathrm{~mol}$ ) was added and the stirring was continuing up to a total of 24 hours. After addition of TMSOMe ( $0.50 \mathrm{~mL}, 3.70 \mathrm{mmol}$ ) and stirring for one hour, the solution was evaporated purified by flash column chromatography (eluent 0-7\% $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ), and preparative TLC ( $500 \mu \mathrm{~m}, 5 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to yield the title compound as a white solid ( $0.009 \mathrm{~g}, 11 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\left.\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 3.73, 3.56.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 7.61(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 6), 7.53$ (d, $0.5 \mathrm{H}, J=7.5$ $\mathrm{Hz}, \mathrm{H} 6$ ), 7.32-6.99 (m, 7H, Ph), 7.16-7.07 (m, 2H, Ph), 6.13 (d, 0.5H, $J=7.3 \mathrm{~Hz}, \mathrm{H} 1$ '), $6.09\left(\mathrm{~d}, 0.5 \mathrm{H}, J=7.3 \mathrm{~Hz}, \mathrm{Hl}^{\prime}\right), 5.78(\mathrm{~d}, 0.5 \mathrm{H}, J=7.5 \mathrm{~Hz}, \mathrm{H} 5), 5.73(\mathrm{~d}, 0.5 \mathrm{H}, J=7.7 \mathrm{~Hz}$, H5), 5.09-4.96 (m, 2H, CH2Ph), 4.69-4.51 (m, 1H, H3'), 4.44-4.39 (m, 1H, H4'), 4.37-4.07 ( $\mathrm{m}, 2 \mathrm{H}, \mathrm{H} 5$ '), 4.00-3.89 (m, 3H, $\mathrm{CHCH}_{3}$ L-Ala, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 3.70-3.59 (m, 1 H , H2'), 2.91-2.80 (m, 2H, CH2 $\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 2.54-2.46 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{32}$ ), 1.33-1.26 (m, 3H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.12-1.00 (m, 3H, CHCH $\mathrm{C}_{3}$ L-Ala).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 174.59(\mathrm{C}=\mathrm{O}), 174.06(\mathrm{C}=\mathrm{O}), 167.78$ (C4), 150.50 (C2), 150.44 (C2), 141.85 (C6), 141.75 (C6), 137.26 (C-Ar), 133.02 (C-Ar), 132.96 (CAr), 131.72 (CH-Ar), 131.70 (CH-Ar), 129.72 (CH-Ar), 129.62 (CH-Ar), 129.38 (CH-Ar), 129.35 (CH-Ar), 129.32 (CH-Ar), 129.24 (CH-Ar), 129.21 (CH-Ar), 128.85 (CH-Ar), 128.83 (CH-Ar), 128.64 (CH-Ar), 126.30 (CH-Ar), 126.29 (CH-Ar), 121.00 (CH-Ar), 120.98 (CH-Ar), 117.61 (CN), 117.58 (CN), 96.54 (C5), 96.47 (C5), 85.68 ( Cl '), 85.46 ( $\mathrm{C} 1{ }^{\prime}$ ), $84.39\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.6 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $84.34\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.6 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), 74.57 ( C 3 '), 74.34 ( C 3 '), $68.06\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 62.26\left(\mathrm{C}^{\prime}\right), 62.22\left(\mathrm{C}^{\prime}\right), 62.21\left(\mathrm{C} 5^{\prime}\right), 61.65\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 51.86$ $\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $51.73\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 44.47\left(\mathrm{C} 2^{\prime}\right), 35.38\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 35.33$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 26.71\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 26.76\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 20.56$ $\left(\mathrm{d},{ }^{3} J_{\mathrm{CP}}=6.4 \mathrm{~Hz}, \mathrm{CHCH} 3\right.$ L-Ala), $20.37\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=6.4 \mathrm{~Hz}, \mathrm{CHCH}_{3}\right.$ L-Ala), 14.54 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$.

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed one broad peak with tR 15.41 min .
$\mathbf{M S}\left(\mathbf{E S}+\right.$ ) $\mathbf{m} / \mathbf{z}$ found $670.2\left[\mathrm{M}+\mathrm{H}^{+}\right], 692.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{31} \mathrm{H}_{36} \mathrm{~N}_{5} \mathrm{O}_{10} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 669.62$ [M].
[112d] 2'-C-cyano-2'-deoxy-1- $\beta$-D-arabino-pentofuranosylcytosine-5'-O-napht-1-yl-(benzyloxy-L-alaninyl)]phosphate


Compound $\mathbf{1 1 1 d}$ was prepared according to procedure $\mathbf{F}_{\mathbf{1}}$ using $110(0.10 \mathrm{~g}, 0.19 \mathrm{mmol})$, in THF ( 4 mL ), $t \mathrm{BuMgCl}$ ( $0.50 \mathrm{~mL}, 0.50 \mathrm{mmol}$ ), naphth-1-yl(benzyloxy-L-alanin-$N$-yl)]phosphorochloridate ( $0.23 \mathrm{~g}, 0.57 \mathrm{mmol}$ ), in THF ( 2 $\mathrm{mL})$. After evaporation, the crude was purified by Biotage Isolera One ( 50 g SNAP cartridge KP-SIL, $100 \mathrm{~mL} / \mathrm{min}$, using a gradient elution of 2-10\% $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over $10 \mathrm{CV}, 10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over 5 CV ). The crude ( 0.065 g ) was dissolved in EtOAc ( 2.8 mL ), pyridine ( $28 \mu \mathrm{~L}, 0.32 \mathrm{mmol}$ ) and the mixture was cooled to $0{ }^{\circ} \mathrm{C}$. Then HF-pyridine ( $70 \% \mathrm{HF}, 8.5 \mu \mathrm{~L}, 0.31 \mu \mathrm{~mol}$ ) was added. The progress of the
reaction was monitored by HPLC. After stirring for six hours another portion of pyridine ( $28 \mu \mathrm{~L}, 0.32 \mathrm{mmol}$ ) and HF-pyridine ( $70 \% \mathrm{HF}, 8.5 \mu \mathrm{~L}, 0.31 \mu \mathrm{~mol}$ ) was added to the reaction mixture. After stirring for 23 hours an additional portion of pyridine ( $28 \mu \mathrm{~L}, 0.32$ mmol ) and HF-pyridine ( $70 \% \mathrm{HF}, 8.5 \mu \mathrm{~L}, 0.31 \mu \mathrm{~mol}$ ) was added and the stirring was continuing up to a total of 24 hours. After addition of TMSOMe ( $0.29 \mathrm{~mL}, 2.18 \mathrm{mmol}$ ) and stirring for one hour, the solution was evaporated and the crude purified by flash column chromatography (eluent $0-7 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ), and preparative TLC ( $500 \mu \mathrm{~m}, 5 \%$ $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to yield the title compound as a white solid ( $0.004 \mathrm{mg}, 3 \%$ two steps).

31P NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 4.26, 4.22.
${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.11-8.04 (m, 1H, Ar), 7.81-7.78 (m, 1H, Ar), 7.64-7.60 ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{Ar}$ ), $7.56(\mathrm{~d}, 0.5 \mathrm{H}, J=7.5 \mathrm{~Hz}, \mathrm{H} 6), 7.53(\mathrm{~d}, 0.5 \mathrm{H}, J=7.6 \mathrm{~Hz}, \mathrm{H} 6), 7.46-7.36$ (m, $3 \mathrm{H}, \mathrm{Ar}), 7.33-7.28(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ar}), 7.22-7.17$ (m, $5 \mathrm{H}, \mathrm{Ar}), 6.11$ (d, $\left.0.5 \mathrm{H}, J=6.0 \mathrm{~Hz}, \mathrm{Hl}^{\prime}\right)$, $6.09\left(\mathrm{~d}, 0.5 \mathrm{H}, J=6.5 \mathrm{~Hz}, \mathrm{H1}{ }^{\prime}\right), 5.65(\mathrm{~d}, 0.5 \mathrm{H}, J=7.5 \mathrm{~Hz}, \mathrm{H} 5), 5.61(\mathrm{~d}, 0.5 \mathrm{H}, J=7.6 \mathrm{~Hz}$, H5), 5.01-4.97 (m, 2H, CH2 Ph), 4.46-4.42 (m, 1H, H3'), 4.42-4.36 (m, 1H, H5'), 4.29-4.24 (m, 1H, H5'), 4.02-3.96 (m, 1H, $\mathrm{CHCH}_{3}$ L-Ala), 3.95-3.91 (m, 1H, H4'), 3.67-3.61 (m, 1H, H2'), 1.26-1.19 (m, 3H, $\mathrm{CHCH}_{3}$ L-Ala).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 173.35$ ( $\mathrm{C}=\mathrm{O}$ ), 167.71 (C4), 167.64 (C4), 152.27 (C2), 141.79 (C6), 136.33 (C-Ar), 130.54 (C-Ar), 130.49 (C-Ar), 129.61 (CH-Ar), 129.60 (CH$\mathrm{Ar}), 129.36$ (CH-Ar), 129.32 (CH-Ar), 128.94 (CH-Ar), 127.93 (CH-Ar), 127.91 (CH-Ar), 127.68 (CH-Ar), 127.62 (CH-Ar), 126.60 (CH-Ar), 126.16 (CH-Ar), 126.13 (CH-Ar), 122.76 ( $\mathrm{CH}-\mathrm{Ar}$ ), 122.70 ( $\mathrm{CH}-\mathrm{Ar}$ ), 116.32 ( CN ), 116.29 (CN), 96.45 (C5), 96.43 (C5), 85.69 ( $\mathrm{Cl}^{\prime}$ ), 85.49 ( C 1 '), 84.39 ( $\mathrm{C}^{\prime}$ ) 84.19 ( $\mathrm{C}^{\prime}$ ), 74.39 ( C 3 '), 74.33 ( $\mathrm{C}^{\prime}$ ), 68.01 $\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 65.95\left(\mathrm{C}^{\prime}\right), 65.89\left(\mathrm{C}^{\prime}\right), 51.93\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.79\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 44.45$ (C2'), $20.45\left(\mathrm{CHCH}_{3}\right), 20.39\left(\mathrm{CHCH}_{3}\right)$.
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed one broad peak with tR 15.48 min .
MS (ES+) $\mathbf{m} / \mathbf{z}$ found $620.2\left[\mathrm{M}+\mathrm{H}^{+}\right], 642.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{30} \mathrm{H}_{30} \mathrm{~N}_{5} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 619.56$ [M].

## [112e] 2'-C-cyano-2'-deoxy-1- $\beta$-D-arabino-pentofuranosylcytosine-5'-O-phenyl-(hexyl-1-oxy-L-alaninyl)]phosphate



Compound 111e was prepared according to procedure $\mathbf{F}_{\mathbf{1}}$ using $110(0.10 \mathrm{~g}, 0.19 \mathrm{mmol})$, in THF ( 4 mL ), $t \mathrm{BuMgCl}$ ( $0.50 \mathrm{~mL}, 0.50 \mathrm{mmol}$ ), phenyl (hexyl-1-oxy-L-alanin- $N$ $\mathrm{yl})]$ phosphorochloridate ( $0.20 \mathrm{~g}, 0.57 \mathrm{mmol}$ ), in THF (2 mL ). After evaporation, the crude was purified by Biotage Isolera One ( 50 g SNAP cartridge KP-SIL, $100 \mathrm{~mL} / \mathrm{min}$, using a gradient elution of $1-10 \%$ $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over $10 \mathrm{CV}, 10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over 5 CV ). The crude ( 0.068 mg ) was dissolved in EtOAc ( 2.0 mL ), pyridine ( $28 \mu \mathrm{~L}, 0.32 \mathrm{mmol}$ ) and the mixture was cooled to $0{ }^{\circ} \mathrm{C}$. Then HF-pyridine ( $70 \% \mathrm{HF}, 8.5 \mu \mathrm{~L}, 0.31 \mu \mathrm{~mol}$ ) was added. The progress of the reaction was monitored by HPLC. After stirring for six hours another portion of pyridine ( $28 \mu \mathrm{~L}, 0.32 \mathrm{mmol}$ ) and HF-pyridine ( $70 \% \mathrm{HF}, 8.5 \mu \mathrm{~L}, 0.31 \mu \mathrm{~mol}$ ) was added to the reaction mixture. After stirring for 23 hours an additional portion of pyridine ( $28 \mu \mathrm{~L}, 0.32$ mmol ) and HF-pyridine ( $70 \% \mathrm{HF}, 8.5 \mu \mathrm{~L}, 0.31 \mu \mathrm{~mol}$ ) was added and the stirring was continuing up to a total of 24 hours. After addition of TMSOMe ( $0.29 \mathrm{~mL}, 2.18 \mathrm{mmol}$ ) and stirring for one hour, the solution was evaporated and the crude purified by flash column chromatography (eluent $0-7 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ), and preparative TLC ( $500 \mu \mathrm{~m}, 5 \%$ $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to yield the title compound as a white solid ( $0.002 \mathrm{~g}, 2 \%$ two steps).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}} 3.92,4.01$.
${ }^{1} \mathbf{H}$ NMR ( $\left.500 \mathbf{M H z}, \mathbf{C D}_{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 7.66(\mathrm{~d}, 0.5 \mathrm{H}, J=7.7 \mathrm{~Hz}, \mathrm{H} 6), 7.60(\mathrm{~d}, 0.5 \mathrm{H}, J=7.7$ Hz, H6), 7.31-7.24 (m, 2H, Ph), 7.19-7.08 (m, 3H, Ph), 6.16 (d, 0.5H, $J=7.0 \mathrm{~Hz}, \mathrm{H} 1$ '), $6.13\left(\mathrm{~d}, 0.5 \mathrm{H}, J=7.2 \mathrm{~Hz}, \mathrm{Hl}^{\prime}\right), 5.81(\mathrm{~d}, 0.5 \mathrm{H}, J=7.5 \mathrm{~Hz}, \mathrm{H} 5), 5.76(\mathrm{~d}, 0.5 \mathrm{H}, J=7.5 \mathrm{~Hz}$, H5), 4.47-4.22 (m, 3H, H3', H5'), 4.02-3.94 (m, 3H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex, H4'), 3.91-3.83 (m, 1H, $\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ), 3.69-3.64 (m, 1H, H2'), 1.55-1.46 (m, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex), 1.28-1.16 (m, $7 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex, $\mathrm{CHCH}_{3} \mathrm{~L}$-Ala), 1.02-0.95 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex), 0.81-0.76 (m, 3 H , $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$ - Hex ).
${ }^{13} \mathbf{C}$ NMR ( $\left.\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{C}} 175.18\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=5.0 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.97\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=5.0\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 167.84 (C4), 167.81 (C4), 157.67 (C-Ar), 157.03 (C-Ar), 152.44 (C2), 152.12 (C2), 141.85 (C6), 141.75 (C6), 130.90 (CH-Ar), 126.34 (CH-Ar), 126.29 (CH-Ar), 121.48 (CH-Ar), 121.46 (CH-Ar), 121.44 (CH-Ar), 121.43 (CH-Ar), 117.61 (CN), $117.59(\mathrm{CN})$, 96.55 (C5), 96.47 (C5), 85.59 (C1'), $85.40\left(\mathrm{C}^{\prime}\right), 84.48$ (d, ${ }^{3} J_{\mathrm{CP}}=8.1 \mathrm{~Hz}, \mathrm{C} 4$ '), 84.38 (d, ${ }^{3} J_{\mathrm{CP}}=8.1 \mathrm{~Hz}, \mathrm{C} 4$ '), $74.40\left(\mathrm{C} 3\right.$ '), $74.04(\mathrm{C} 3 '), 66.54\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Hex}\right), 66.52$
$\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Hex}\right), 65.78\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ ) $) 65.56\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=5.4 \mathrm{~Hz}\right.$, C 5 '), $51.80\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.63\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 44.49$ ( C 2 '), 44.46 ( $\mathrm{C}^{\prime}$ ), 32.62 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex $), \quad 32.61\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex $), \quad 29.68$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex), $26.66 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex), 23.63 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Hex), $20.57\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.12 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 20.42\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}\right.$ $=7.12 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}$-Ala), $14.39\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Hex $)$.
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $90 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed one broad peak with tR 16.09 min .
MS (ES+) $\mathbf{m} / \mathbf{z}$ found $564.3\left[\mathrm{M}+\mathrm{H}^{+}\right], 586.3\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{25} \mathrm{H}_{34} \mathrm{~N}_{5} \mathrm{O}_{8} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 563.54$ [M].

### 9.5.18 Synthesis of TFT ProTides

[114a] 5-Trifluoromethyl-2'-deoxyuridine-5'-[ethyl-3-(2-hydroxyphenyl)propanoyl (benzyloxy-L-alanin- $N$-yl)]phosphate


Prepared according to general procedure $\mathbf{F}_{2}$ using trifluorothymidine ( $0.05 \mathrm{~g}, 0.17 \mathrm{mmol}$ ) in THF ( 2 mL ), ethyl 3-(2-hydroxyphenyl)propanoyl (benzyloxy-L-alanin-$N$-yl)] phosphorochloridate [17v] ( $0.23 \mathrm{~g}, 0.51 \mathrm{mmol}$ ) in THF ( 1 mL ), and $N$-methylimidazole ( $67 \mu \mathrm{~L}, 0.84 \mathrm{mmol}$ ). The solvent was evaporated and the crude was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25 \mathrm{~mL})$ and washed with 0.5 M aqueous HCl solution $(5 \mathrm{~mL})$. The organic phase was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, evaporated and purified by silica gel column chromatography (eluent system $0-6 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) and by preparative TLC $(500 \mu \mathrm{M}$, eluent system $5 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give the title compound as a white solid ( 0.013 g , $11 \%$ ).
${ }^{\mathbf{3 1}} \mathbf{P}$ NMR ( $\left.\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 3.95, 3.54.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.20(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 6), 8.16(\mathrm{~s}, 0.5 \mathrm{H}, \mathrm{H} 6), 7.40-7.30(\mathrm{~m}, 6 \mathrm{H}$, Ar), 7.28-7.24 (m, 1H, Ar), 7.22-7.09 (m, 2H, Ar), 6.17 (dd, $J=7.3,6.2 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{Hl}$ '), 6.11 (dd, $J=7.3,6.2 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1$ '), $5.19-5.08\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}\right), 4.47-4.42(\mathrm{~m}, 0.5 \mathrm{H}$, H3'), 4.39-4.28 (m, 2H, H3',H5'), 4.26-4.20 (m, 0.5H, H5'), 4.17-3.97 (m, 4H, H4', $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}, \mathrm{CHCH}_{3}$ L-Ala), 3.00-2.89 (m, 2H, CH $\mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 2.65-2.53 ( $\mathrm{m}, 2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 2.40-2.33 (m, 1H, H2'), 2.24-2.17 (m, 0.5H, H2'), 2.14-
2.07 (m, 0.5H, H2'), 1.41-1.36 (m, 3H, CHCH3 L-Ala), 1.24-1.18 (m, 3H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH} 3$ ).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 174.79\left(\mathrm{~d},{ }^{3} J_{C-P}=4.5 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.56(\mathrm{C}=\mathrm{O}), 174.52$ (C=O), $174.47\left(\mathrm{~d},{ }^{3} J_{C-P}=5.2 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 161.13(\mathrm{C} 4), 151.12(\mathrm{C} 2), 151.02(\mathrm{C} 2), 150.39(\mathrm{~d}$, $\left.{ }^{2} J_{\mathrm{CP}}=7.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 150.37\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=7.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 143.04\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=6.2 \mathrm{~Hz}, \mathrm{C} 6\right)$, $142.91\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=6.2 \mathrm{~Hz}, \mathrm{C} 6\right), 137.24$ (C-Ar), 137.18 (C-Ar), 132.87 (d, ${ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{C}-$ $\mathrm{Ar}), 132.09\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=7.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 131.64(\mathrm{CH}-\mathrm{Ar}), 131.55(\mathrm{CH}-\mathrm{Ar}), 129.64$ (CH-Ar), 129.61 (CH-Ar), 129.43 (CH-Ar), 129.39 (CH-Ar), 129.37 (CH-Ar), 128.77 (CH-Ar), 128.71 (CH-Ar), 126.25 (CH-Ar), 123.81 ( $\mathrm{q},{ }^{1} J_{\mathrm{CF}}=267.5 \mathrm{~Hz}, \mathrm{CF}_{3}$ ), $120.91\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=2.5\right.$ $\mathrm{Hz}, \mathrm{CH}-\mathrm{Ar}), 120.89\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=2.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 105.20\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=32.5 \mathrm{~Hz}, \mathrm{C} 5\right), 105.19(\mathrm{q}$, $\left.{ }^{2} J_{\mathrm{CF}}=32.5 \mathrm{~Hz}, \mathrm{C} 5\right), 88.59\left(\mathrm{C} 1^{\prime}\right), 88.31\left(\mathrm{C} 1\right.$ '), $87.22\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.5 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $87.16\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}\right.$ $=7.5 \mathrm{~Hz}, \mathrm{C} 4$ '), $72.24\left(\mathrm{C} 3\right.$ '), $72.17\left(\mathrm{C} 3^{\prime}\right), 68.04\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.70\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=7.5 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.65\left(\mathrm{~d}^{2}{ }^{2} J_{\mathrm{CP}}=7.5 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $61.62\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH} 3\right), 61.61\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$, $51.77\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.68\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 41.47$ (C2'), 41.45 (C2'), 35.29 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 35.21\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 26.75\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 26.63$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 20.47\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=6.2 \mathrm{~Hz}, \mathrm{CHCH}_{3}\right.$ L-Ala), $20.35\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=6.2 \mathrm{~Hz}\right.$, $\mathrm{CHCH}_{3}$ L-Ala), $14.52\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CO}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right)$.

## ${ }^{19}$ F NMR (470 MHz, CD $\left.\mathbf{3}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathrm{F}}$ 64.34, 64.35 .

MS (ES+) m/z found $736.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], 714.2\left[\mathrm{M}+\mathrm{H}^{+}\right], \mathrm{C}_{31} \mathrm{H}_{35} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}_{11} \mathrm{P}$ required m/z 713.59 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $F=1 \mathrm{~mL} / \mathrm{min}, \lambda=254 \mathrm{~nm}$, showed one peak with tR 20.87 min .
[114b] alaninyl)phosphate


Prepared according to generale procedure $\mathbf{F}_{2}$ using trifluorothymidine ( $0.05 \mathrm{~g}, 0.17 \mathrm{mmol}$ ), in THF ( 2 mL ), naphth-1-yl(benzyloxy-L-alaninyl)phosphorochloridate [17a] ( $0.14 \mathrm{~g}, 0.34 \mathrm{mmol})$ in THF $(1 \mathrm{~mL})$, and $N$ methylimidazole ( $67 \mu \mathrm{~L}, 0.84 \mathrm{mmol}$ ). The solvent was evaporated and the crude was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25 \mathrm{~mL})$ and washed with 0.5 M aqueous HCl solution ( 5 mL ). The organic phase was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, evaporated and purified by silica gel column chromatography (eluent system 0-6\% $\mathrm{CH}_{3} \mathrm{OH}$ $/ \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) and by preparative TLC ( $500 \mu \mathrm{M}$, eluent system $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}=5 / 95$ ) to give the title compound as a white solid ( $0.035 \mathrm{~g}, 31 \%$ ).
${ }^{31} \mathbf{P}$ NMR (202 MHz, CD $\left.\mathbf{3}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 4.47, 4.03.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.16-8.08(m, 2H, H6, Ar), 7.81-7.87 (m, 1H, Ar), 7.73, $7.68(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ar}), 7.57-7.45(\mathrm{~m}, 2 \mathrm{H}, \mathrm{Ar}), 7.44-7.35(\mathrm{~m}, 1 \mathrm{H}, \mathrm{Ar}), 7.34-7.27(\mathrm{~m}, 6 \mathrm{H}, \mathrm{Ar})$, 6.07-6.01 (m, 1H, H1'), 5.13-5.03 (m, 2H, CH2 Ph), 4.42-4.23 (m, 3H, H3', H5'), 4.15-4.11 ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{H} 4$ '), 4.11-4.02 (m, 1H, $\mathrm{CHCH}_{3}$ L-Ala), 2.30-2.24 (m, 0.5H, H2'), 2.23-2.17 (m, $\left.0.5 \mathrm{H}, \mathrm{H} 2{ }^{\prime}\right), 1.93-1.87\left(\mathrm{~m}, 0.5 \mathrm{H}, \mathrm{H} 2{ }^{\prime}\right), 1.76-1.69\left(\mathrm{~m}, 0.5 \mathrm{H}, \mathrm{H} 2\right.$ '), 1.36-1.31 (m, 3H, $\mathrm{CHCH}_{3}$ L-Ala).
${ }^{13} \mathbf{C}$ NMR ( $\left.\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathrm{C}} 174.82\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=4.0 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.50\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=4.0\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 161.08 (C4), 161.07 (C4), 150.91 (C2), 150.88 (C2), $147.93\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.4 \mathrm{~Hz}\right.$, C-Ar), $147.83\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.4 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 142.91\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=6.0 \mathrm{~Hz}, \mathrm{C} 6\right), 142.63\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=6.0\right.$ Hz, C6), 137.15 (C-Ar), 137.13 (C-Ar), 136.27 (C-Ar), 129.60 (CH-Ar), 129.55 (CH-Ar), 129.39 (CH-Ar), 129.34 (CH-Ar), 129.32 (CH-Ar), 128.94 (CH-Ar), 128.90 (CH-Ar), 127.85 (CH-Ar), $127.55(\mathrm{CH}-\mathrm{Ar}), 123.80\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=268.7 \mathrm{~Hz}, \mathrm{CF}_{3}\right), 123.76\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=\right.$ $268.7 \mathrm{~Hz}, \mathrm{CF}_{3}$ ), $126.49\left(\mathrm{~d}, J_{\mathrm{CP}}=3.4 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 126.48\left(\mathrm{~d}, J_{\mathrm{CP}}=3.4 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 126.16$ (CH-Ar), 126.09 (CH-Ar), 116.28 (CH-Ar), 105.32 (q, $\left.{ }^{2} J_{\mathrm{CF}}=32.7 \mathrm{~Hz}, \mathrm{C} 5\right), 88.61$ (C1'), $88.44\left(\mathrm{C} 1^{\prime}\right), 87.46\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.8 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $87.22\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.8 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $72.33\left(\mathrm{C} 3^{\prime}\right), 72.22$ (C3'), $68.02\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.82\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $67.78\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), 51.84 $\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $51.76\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 41.42\left(\mathrm{C} 2^{\prime}\right), 41.40\left(\mathrm{C} 2^{\prime}\right), 20.37\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.6 \mathrm{~Hz}\right.$, $\mathrm{CHCH}_{3}$ L-Ala), 20.31 (d, ${ }^{3} J_{\mathrm{CP}}=7.6 \mathrm{~Hz}, \mathrm{CHCH}_{3}$ L-Ala).
${ }^{19}$ F NMR (470 MHz, CD $\left.\mathbf{D}_{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathrm{F}}$-64.30, -64.33.
MS (ES+) m/z: Found: $686.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], 664.2\left[\mathrm{M}+\mathrm{H}^{+}\right], \mathrm{C}_{30} \mathrm{H}_{29} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}_{9} \mathrm{P}$ required m/z 663.53 [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from 100/10 to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{ml} / \mathrm{min}, \lambda=245 \mathrm{~nm}$, showed two peaks with tR 17.25 min . and tR 17.51 min.
[114c]
alaninyl)]phosphate


Prepared according to general procedure $\mathbf{F}_{2}$ using trifluorothymidine ( $0.054 \mathrm{~g}, 0.18 \mathrm{mmol}$ ) in THF ( 3 mL ), phenyl(benzyloxy-L-alaninyl)phosphorochloridate [17s] $(0.13 \mathrm{~g}, \quad 0.36 \mathrm{mmol})$ in THF $(1 \mathrm{~mL})$ and N methylimidazole ( $72 \mu \mathrm{~L}, 0.90 \mathrm{mmol}$ ). The solvent was evaporated and the crude was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25 \mathrm{~mL})$ and washed with 0.5 M aqueous HCl solution ( 5 mL ). The organic phase was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered,
evaporated and purified by silica gel column chromatography (eluent system 0-6\% $\mathrm{CH}_{3} \mathrm{OH}$ $/ \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give the title compound as a white solid ( $0.05 \mathrm{~g}, 45 \%$ ).
${ }^{31} \mathbf{P}$ NMR (202 MHz, CD $\mathbf{3}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 4.12, 3.56.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.17$ ( $\left.\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 6\right), 8.15$ ( $\left.\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 6\right), 7.39-7.28(\mathrm{~m}, 7 \mathrm{H}$, Ar), 7.24-7.15 (m, 3H, Ar), 6.17 (dd, $J=7.6,6.2 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1$ '), 6.12 (dd, $J=7.6,6.2 \mathrm{~Hz}$, $0.5 \mathrm{H}, \mathrm{H} 1$ '), 5.18-5.08 (m, 2H, CH2 Ph ), 4.45-4.41 (m, 0.5H, H3'), 4.40-4.22 (m, 2.5H, H3', H5'), 4.17-4.12 (m, 1H, H4'), 4.05-3.94 (m, 1H, $\mathrm{CHCH}_{3}$ L-Ala), 2.42-2.29 (m, 1H, H2'), 2.14-2.07 (m, 0.5H, H2'), 2.03-1.94 (m, 0.5H, H2'), 1.37-1.31 (m, 3H, CHCH ${ }_{3}$ L-Ala). ${ }^{13} \mathbf{C}$ NMR ( $\left.\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathrm{C}} 174.87\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=5.0 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.75\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=5.0\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), $161.07(\mathrm{C} 4), 152.06\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=6.2 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 152.05\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=6.2 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right)$, $151.07(\mathrm{C} 2), 151.01(\mathrm{C} 2), 142.90\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=6.02 \mathrm{~Hz}, \mathrm{C} 6\right), 142.76\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=6.02 \mathrm{~Hz}, \mathrm{C} 6\right)$, 137.22 (C-Ar), 137.14 (C-Ar), 130.84 (CH-Ar), 129.66 (CH-Ar), 129.64 (CH-Ar), 129.47 (CH-Ar), 129.42 (CH-Ar), 129.38 (CH-Ar), 126.34 (CH-Ar), 126.27 (CH-Ar), 123.87 (q, $\left.{ }^{1} J_{\mathrm{CF}}=267.5 \mathrm{~Hz}, \mathrm{CF}_{3}\right), 121.42(\mathrm{CH}-\mathrm{Ar}), 121.38(\mathrm{CH}-\mathrm{Ar}), 105.50\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=32.5 \mathrm{~Hz}, \mathrm{C} 5\right)$, $105.39\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=32.5 \mathrm{~Hz}, \mathrm{C} 5\right), 88.50\left(\mathrm{Cl}^{\prime}\right), 88.17\left(\mathrm{C} 1^{\prime}\right), 87.40\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=8.2 \mathrm{~Hz}, \mathrm{C} 4{ }^{\prime}\right)$, $87.21\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=8.2 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $72.31\left(\mathrm{C} 3\right.$ '), $72.22\left(\mathrm{C} 3\right.$ '), $68.06\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.66\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=\right.$ $16.2 \mathrm{~Hz}, \mathrm{C} 5$ '), $67.54\left(\mathrm{~d}^{2}{ }^{2} J_{\mathrm{CP}}=16.2 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $51.79\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.63\left(\mathrm{CHCH}_{3} \mathrm{~L}-\right.$ Ala), 41.62 (C2'), 41.55 ( C 2 '), $20.52\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.0 \mathrm{~Hz}, \mathrm{CHCH}_{3}\right.$ L-Ala), $20.40\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.0\right.$ $\left.\mathrm{Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$.
${ }^{19}$ F NMR (470 MHz, CD $\mathbf{3}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{F}}-64.11,-64.15$.
MS (ES+ $) \mathbf{m} / \mathbf{z}$ found $636.1\left[\mathrm{M}+\mathrm{Na}^{+}\right], 614.1\left[\mathrm{M}+\mathrm{H}^{+}\right], \mathrm{C}_{26} \mathrm{H}_{27} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}_{9} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 613.48$ [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \mathrm{l}=254 \mathrm{~nm}$, showed one peak with tR 15.69 min .
[114d] 5-Trifluoromethyl-2'-deoxyuridine-5'-[naphth-1-yl-(cyclohexyloxy-Lalaninyl)] phosphate


Prepared according to general procedure $\mathbf{F}_{2}$ using trifluorothymidine ( $0.05 \mathrm{~g}, 0.178 \mathrm{mmol}$ ) in THF ( 3 mL ), naphth-1-yl(cyclohex-1-yloxy-L-alaninyl) phosphorochloridate [17c] ( $0.17 \mathrm{~g}, 0.51 \mathrm{mmol}$ ) in THF (1 mL ) and $N$-methylimidazole ( $70 \mu \mathrm{~L}, 0.85 \mathrm{mmol}$ ). The solvent was evaporated and the crude was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25 \mathrm{~mL})$ and washed with 0.5 M aqueous HCl solution ( 5 mL ). The organic phase was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered,
evaporated and purified by silica gel column chromatography (eluent system 0-6\% $\mathrm{CH}_{3} \mathrm{OH}$ $/ \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give the title compound as a white solid ( $0.048 \mathrm{~g}, 41 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 4.40, 4.10.
${ }^{1} \mathbf{H}$ NMR ( $500 \mathrm{MHz}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.17-8.10 (m, 2H, H6, Nap), 7.91-7.87 (m, 1H, Nap), 7.74-7.69 (m, 1H, Nap), 7.56-7.48 (m, 3H, Nap), 7.45-7.39 (m, 1H, Nap), 6.09-6.02 (m, $1 \mathrm{H}, \mathrm{H1}$ '), 4.73-4.66 (m, 1H, CH( $\left.\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}$ ), 4.47-4.30 (m, 3H, H3', H5'), 4.20-4.14 (m, 1H, H4'), 4.03-3.95 (m, 1H, CHCH L-Ala), 2.31 (ddd, $J=13.9 \mathrm{~Hz}, 6.2 \mathrm{~Hz}, 3.0 \mathrm{~Hz}, 0.5 \mathrm{H}$, H2'), 2.20 (ddd, $\left.J=13.9 \mathrm{~Hz}, 6.2 \mathrm{~Hz}, 3.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 2^{\prime}\right), 2.03-1.95$ (m, $0.5 \mathrm{H}, \mathrm{H} 2$ '), 1.81$1.65\left(\mathrm{~m}, 4.5 \mathrm{H}, \mathrm{H} 2{ }^{\prime}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 1.57-1.48\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$, 1.42-1.23 (m, 8H, CH( $\left.\left.\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$.
${ }^{13} \mathbf{C}$ NMR ( $\left.\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathrm{C}} 174.50\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=4.7 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.22\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=4.7\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), $161.05(\mathrm{C} 4), 161.02(\mathrm{C} 4), 150.95(\mathrm{C} 2), 150.87(\mathrm{C} 2), 147.99\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=7.1 \mathrm{~Hz}\right.$, C-Ar), $147.90\left(\mathrm{~d},{ }^{2} J_{\mathrm{C}-\mathrm{P}}=7.1 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 143.00\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=5.7 \mathrm{~Hz}, \mathrm{C} 6\right), 142.59\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=5.7\right.$ Hz, C6), 136.31 (C-Ar), 128.94 (CH-Ar), 127.87 (CH-Ar), 127.85 (CH-Ar), 127.54 (CHAr), $127.52(\mathrm{CH}-\mathrm{Ar}), 126.50(\mathrm{CH}-\mathrm{Ar}), 126.15(\mathrm{CH}-\mathrm{Ar}), 126.08(\mathrm{CH}-\mathrm{Ar}), 123.83\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=\right.$ $268.8 \mathrm{~Hz}, \mathrm{CF}_{3}$ ), $123.76\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=268.8 \mathrm{~Hz}, \mathrm{CF}_{3}\right), 122.58(\mathrm{CH}-\mathrm{Ar}), 122.55(\mathrm{CH}-\mathrm{Ar})$, $116.27\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.9 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 116.22\left(\mathrm{~d},{ }^{3} J_{\mathrm{C}-\mathrm{P}}=2.9 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 105.43\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=\right.$ $32.6 \mathrm{~Hz}, \mathrm{C} 5), 105.35\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=32.6 \mathrm{~Hz}, \mathrm{C} 6\right), 88.66\left(\mathrm{C} 1^{\prime}\right), 88.51\left(\mathrm{C} 1^{\prime}\right), 87.54\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=8.3\right.$ $\mathrm{Hz}, \mathrm{C} 4$ '), $87.22\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=8.3 \mathrm{~Hz}, \mathrm{C} 4\right.$ ' $), 75.02\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 74.99\left(\mathrm{C} 3^{\prime}\right), 67.87(\mathrm{~d}$, ${ }^{2} J_{\mathrm{CP}}=5.3 \mathrm{~Hz}, \mathrm{C} 5$ '), $67.80\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=5.3 \mathrm{~Hz}, \mathrm{C} 5^{\prime}\right), 51.88\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 41.43\left(\mathrm{C} 2^{\prime}\right)$, $41.37\left(\mathrm{C}^{\prime}\right)$, $32.41\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$, $32.38\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 32.36\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right)$, $32.33\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 24.88\left(\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{5} \mathrm{cHex}\right), 20.64\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=6.3 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$, $20.56\left(\mathrm{~d},{ }^{3} \mathrm{~J}_{\mathrm{CP}}=6.3 \mathrm{~Hz}, \mathrm{CHCH}_{3}\right.$ L-Ala).
${ }^{19}$ F NMR (470 MHz, CD $\mathbf{3}_{3}$ OD) $\boldsymbol{\delta}_{\mathrm{F}}-64.23,-64.30$.
MS (ES+ $) \mathbf{m} / \mathbf{z}$ found $678.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], 656.2\left[\mathrm{M}+\mathrm{H}^{+}\right], \mathrm{C}_{29} \mathrm{H}_{33} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}_{9} \mathrm{P}$ required m/z 655.56 [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, 1=254 \mathrm{~nm}$, showed two peaks with tR 18.38 min . and tR 18.63 min.

## 5-Trifluoromethyl-2'-deoxyuridine-5'-phenyl-(neopentyloxy-L-

 alaninyl)]phosphate

Prepared according to general procedure $\mathbf{F}_{2}$ using trifluorothymidine ( $0.05 \mathrm{~g}, 0.178 \mathrm{mmol}$ ) in THF ( 3 mL ), phenyl(neopentyloxy-L-alaninyl)phosphorochloridate [17t] ( $0.17 \mathrm{~g}, 0.51 \mathrm{mmol}$ ) in THF ( 1 mL ) and $N$-methylimidazole ( $70 \mu \mathrm{~L}, 0.85 \mathrm{mmol}$ ). The solvent was evaporated and the crude was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25 \mathrm{~mL})$ and washed with 0.5 M aqueous HCl solution (5 $\mathrm{mL})$. The organic phase was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, evaporated and purified by silica gel column chromatography (eluent system $0-6 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give the title compound as a white solid ( $0.029 \mathrm{~g}, 27 \%$ ).
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathbf{H}} 8.19$ ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 6$ ), 8.18 ( $\mathrm{s}, 0.5 \mathrm{H}, \mathrm{H} 6$ ), $7.39-7.33$ (m, 2H, Ph), 7.26-7.18 (m, 3H, Ph), 6.20-6.12 (m, 1H, H1'), 4.48-4.30 (m, 3H, H3', H5'), 4.204.14 ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{H} 4$ '), 4.04-3.95 (m, 1H, $\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ), 3.87-3.84 (m, 1H, CH2tBu Neop), 3.80-3.73 (m, 1H, CH2 CBu Neop ), 2.44-2.38 (m, 0.5H, H2'), 2.37-2.31 (m, 0.5H, H2'), 2.24-2.18 (m, 0.5H, H2'), 2.06-1.99 (m, 0.5H, H2'), 1.39-1.35 (m, 3H, CHCH ${ }_{3}$ L-Ala), 0.96 ( $\mathrm{s}, 4.5 \mathrm{H}, \mathrm{CH}_{2} t \mathrm{Bu}$ Neop), 0.95 ( $\mathrm{s}, 4.5 \mathrm{H}, \mathrm{CH}_{2}{ }^{t} \mathrm{Bu}$ ).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 175.06\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=4.9 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.72\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=4.9\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 161.07 (C4), 160.9 (C4), $152.06\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=6.2 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 152.05\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=6.2\right.$ $\mathrm{Hz}, \mathrm{C}-\mathrm{Ar}), 151.09(\mathrm{C} 2), 143.01\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=6.2 \mathrm{~Hz}, \mathrm{C} 6\right), 142.77\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=6.2 \mathrm{~Hz}, \mathrm{C} 6\right)$, 136.70 (C-Ar), $130.80(\mathrm{CH}-\mathrm{Ar}), 126.31(\mathrm{CH}-\mathrm{Ar}), 126.22(\mathrm{CH}-\mathrm{Ar}), 123.70\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=267.5\right.$ $\mathrm{Hz}, \mathrm{CF}_{3}$ ), $121.38\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=4.6 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 121.36\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=4.6 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 105.15(\mathrm{q}$, $\left.{ }^{2} J_{\mathrm{CF}}=32.3 \mathrm{~Hz}, \mathrm{C} 5\right), 88.59\left(\mathrm{C} 1^{\prime}\right), 88.27\left(\mathrm{C} 1^{\prime}\right), 87.48\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=8.1 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $87.24\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}\right.$ $=8.1 \mathrm{~Hz}, \mathrm{C} 4$ '), $75.46\left(\mathrm{CH}_{2} \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right.$ Neop $), 72.36\left(\mathrm{C}^{\prime}\right)$, $72.23\left(\mathrm{C} 3^{\prime}\right), 67.72\left(\mathrm{~d}^{2}{ }^{2} J_{\mathrm{CP}}=4.6\right.$ $\mathrm{Hz}, \mathrm{C} 5$ '), $67.59\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=4.6 \mathrm{~Hz}, \mathrm{C} 5 '\right), 51.73\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.65\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$, $41.58\left(\mathrm{C}^{\prime}\right)$, $41.46\left(\mathrm{C}^{\prime}\right)$, $28.05\left(\mathrm{CH}_{2} \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3} \mathrm{Neop}\right) 26.70\left(\mathrm{CH}_{2} \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3} \mathrm{Neop}\right), 20.70(\mathrm{~d}$, ${ }^{3} J_{\mathrm{CP}}=6.3 \mathrm{~Hz}, \mathrm{CHCH}_{3}$ L-Ala), $20.56\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=6.3 \mathrm{~Hz}, \mathrm{CHCH} 3\right.$ L-Ala $)$.
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 4.06, 3.71.
${ }^{19}$ F NMR (470 MHz, CD $\left.\mathbf{D}_{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathrm{F}}-64.32,-64.24$.
$\mathbf{M S}\left(\mathbf{E S}+\right.$ ) $\mathbf{m} / \mathbf{z}$ found $616.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], 594.2\left[\mathrm{M}+\mathrm{H}^{+}\right], \mathrm{C}_{24} \mathrm{H}_{31} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}_{9} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 593.49$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \mathrm{l}=254 \mathrm{~nm}$, showed one peak with tR 16.56 min .
[114f] 5-Trifluoromethyl-2'-deoxyuridine-5'-naphth-1-yl-(pentyl-1-oxy-Lalaninyl)]phosphate


Prepared according to general procedure $\mathbf{F}_{2}$ using trifluorothymidine ( $0.05 \mathrm{~g}, 0.17 \mathrm{mmol}$ ) in THF ( 3 mL ), naphth-1-yl(pentyl-1-oxy-L-leucinyl)phosphorochloridate [17f] ( $0.22 \mathrm{~g}, 0.51 \mathrm{mmol})$ in THF $(1 \mathrm{~mL})$ and N methylimidazole ( $70 \mu \mathrm{~L}, 0.85 \mathrm{mmol}$ ). The solvent was evaporated and the crude was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(25 \mathrm{~mL})$ and washed with 0.5 M HCl solution ( 5 mL ). The organic phase was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered, evaporated and purified by silica gel column chromatography (eluent system $0-6 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) and by prepTLC ( $1000 \mu \mathrm{M}, 5 \% \mathrm{CH}_{3} \mathrm{OH}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give a white solid ( $0.03 \mathrm{~g}, 26 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\mathbf{2 0 2} \mathbf{~ M H z}, \mathbf{C D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{P}}$ 4.79, 4.23.
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{5 0 0} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}}$ 8.19-8.11 (m, 2H, H6, Nap), 7.93-7.87 (m, 1H, Nap), 7.74-7.89 (m, 1H, Nap), 7.58-7.48 (m, 3H, Nap), 7.45-7.38 (m, 1H, Nap), 6.11-6.05 (m, 1H, H1'), 4.50-4.43 (m, 0.5H, H3'), 4.42-4.29 (m, 2.5H, H3', H5'), 4.20-4.10 (m, 3H, H4', $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 3.97-3.89 (m, $1 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), 2.36-2.29 (m, $\left.0.5 \mathrm{H}, \mathrm{H} 2^{\prime}\right), 2.27-2.19$ (m, 0.5H, H2'), 2.03-1.95 (m, 0.5H, H2'), 1.85-1.44 (m, 5.5H, H2', $\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \quad$ L-Leu, $\quad \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n$-Pen), 1.41-1.36 (m, 2 H , $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 1.32-1.25 (m, $2 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Pen), 0.99-0.91 ( $\mathrm{m}, 6 \mathrm{H}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}$ L-Leu), $0.88-0.83\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen).
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{1 2 5} \mathbf{~ M H z}, \mathbf{C D}_{3} \mathbf{O D}$ ), $\boldsymbol{\delta}_{\mathbf{C}} 175.42\left(\mathrm{~d},{ }^{3} J_{C-P}=2.5 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.95\left(\mathrm{~d},{ }^{3} J_{C-P}=2.5\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 161.04 (C4), 161.00 (C4), 150.96 (C2), 150.93 (C2), 147.97 (d, ${ }^{2} J_{\mathrm{CP}}=7.1 \mathrm{~Hz}$, C-Ar), $147.90\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=7.1 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 143.00\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=5.6 \mathrm{~Hz}, \mathrm{C} 6\right), 142.68\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=5.6\right.$ Hz, C6), 136.32 (C-Ar), 128.93 (CH-Ar), 128.91 (CH-Ar), 127.84 (CH-Ar), 127.82 (CHAr), 127.52 ( $\mathrm{CH}-\mathrm{Ar}$ ), 127.49 (CH-Ar), 126.48 (CH-Ar), 126.15 (CH-Ar), $126.02(\mathrm{CH}-\mathrm{Ar})$, $125.93\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=269.5 \mathrm{~Hz}, \mathrm{CF}_{3}\right), 125.88\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=269.5 \mathrm{~Hz}, \mathrm{CF}_{3}\right), 122.66(\mathrm{CH}-\mathrm{Ar})$, 122.61 (CH-Ar), 116.44 ( $\mathrm{d},{ }^{3} J_{\mathrm{CP}}=3.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), 116.10 (d, $\left.{ }^{3} J_{\mathrm{CP}}=3.5 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right)$, $105.44\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=33.5 \mathrm{~Hz}, \mathrm{C} 5\right), 88.61\left(\mathrm{C} 1^{\prime}\right), 88.31\left(\mathrm{C} 1^{\prime}\right), 87.45\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.9 \mathrm{~Hz}, \mathrm{C} 4^{\prime}\right)$, $87.17\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.9 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $72.30\left(\mathrm{C} 3\right.$ '), $72.24\left(\mathrm{C} 3^{\prime}\right), 67.89\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), 67.81 $\left(\mathrm{d},{ }^{2} J_{\mathrm{CP}}=5.4 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $66.37\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen $), 54.77\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}\right.$ Leu), $44.19\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.5 \mathrm{~Hz}, \mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $43.88\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=7.5 \mathrm{~Hz}\right.$, $\left.\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}-\mathrm{Leu}\right), 41.38$ ( $\mathrm{C}^{\prime}$ ), 41.32 ( $\mathrm{C}^{\prime}$ ), $29.33\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $29.31 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen), $25.70 \quad\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \quad\right.$ L-Leu), 25.43 $\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2} \mathrm{~L}-\mathrm{Leu}\right), 23.30\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), $23.16\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu), $22.97\left(\mathrm{CHCH}_{2} \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right.$ L-Leu $), 22.03\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen), 21.65
$\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen), $14.24 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Pen $), \quad 14.23$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Pen).
${ }^{19}$ F NMR (470 MHz, CD $\left.\mathbf{D}_{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathrm{F}}-64.32,-64.23$.
$\mathbf{M S}\left(\mathbf{E S}+\right.$ ) $\mathbf{m} / \mathbf{z}$ found $708.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], 686.2\left[\mathrm{M}+\mathrm{H}^{+}\right], \mathrm{C}_{31} \mathrm{H}_{39} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}_{9} \mathrm{P}$ required $\mathrm{m} / \mathrm{z} 685.63$ [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}$ from $100 / 10$ to $0 / 100$ in 30 minutes, $\mathrm{F}=1 \mathrm{~mL} / \mathrm{min}, \mathrm{l}=245 \mathrm{~nm}$, showed one broad peak with tR 21.58 min .

### 9.5.19 Synthesis of 2'-SCF ${ }_{3}$-2'-deoxyuridine and 2'-SH-2'-deoxyuridine

## [115] 2'-Trifluoromethylthio-2'-deoxyuridine



2'-Mercapto-2'-deoxyuridine [119] ( $0.54 \mathrm{~g}, 2.07 \mathrm{mmol}$ ) was dissolved in $\mathrm{CH}_{3} \mathrm{OH}(15 \mathrm{~mL})$ and the solution was cooled to $-78{ }^{\circ} \mathrm{C}$ under argon atmosphere. 3,3-Dimethyl-1-(trifluoromethyl)-1,2-benziodoxole (Togni Reagent) ( $0.76 \mathrm{~g}, 2.5 \mathrm{mmol}$ ) was dissolved in $\mathrm{CH}_{3} \mathrm{OH}(5 \mathrm{~mL})$ in a second flask and cooled to $-78^{\circ} \mathrm{C}$ under argon atmosphere. The solution was added via a syringe to the initial solution, the second flask was washed with $\mathrm{CH}_{3} \mathrm{OH}$ $(1 \mathrm{~mL})$ and the solution added to the first flask. The reaction mixture was allowed to stir at $-78{ }^{\circ} \mathrm{C}$ for 10 minutes and then at room temperature for 5 hours. The mixture was evaporated and the crude purified via Biotage Isolera One ( 50 g SNAP cartridge KP-SIL, $75 \mathrm{~mL} / \mathrm{min}$, using a gradient elution of $2-20 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over $10 \mathrm{CV}, 20 \%$ $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over 5 CV ) to yield the title compound as a white solid ( $0.52 \mathrm{~g}, 76 \%$ ). ${ }^{437}$ ${ }^{\mathbf{1}} \mathbf{H}$ NMR (CD $\mathbf{3}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 7.99(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6), 6.32(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H1}$ '), $5.78(\mathrm{~d}$, $J=8.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5), 4.43\left(\mathrm{dd}, J=5.1 \mathrm{~Hz}, 1.1 \mathrm{~Hz}, \mathrm{H}{ }^{\prime}\right), 4.11(\mathrm{tt}, J=4.6 \mathrm{~Hz}, 1.1 \mathrm{~Hz}, 1 \mathrm{H}$, H4'), 4.03 (dd, 9.1 Hz, 5.3 Hz, 1H, H2'), 3.789 (dd, $17.9 \mathrm{~Hz}, 12.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ '), 3.783 (dd, $\left.17.9 \mathrm{~Hz}, 12.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}{ }^{\prime}\right)$.
${ }^{13} \mathbf{C}$ NMR (CD $\mathbf{D}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 165.73(\mathrm{C} 4), 152.57(\mathrm{C} 2), 142.23(\mathrm{C} 6), 132.37\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=305.0\right.$ $\mathrm{Hz}, \mathrm{SCF}_{3}$ ), 103.83 (C5), 89.87 ( C 1 '), 88.63 (C4'), 74.62 (C3'), 63.11 (C5’), 52.06 (C2’).
${ }^{19}$ F NMR (CD $\left.{ }_{3} \mathbf{O D}\right) \boldsymbol{\delta}_{\mathrm{F}}-41.60$.
$\mathbf{M S}$ (ES+) $\mathbf{m} / \mathbf{z}$ found $351.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{10} \mathrm{H}_{11} \mathrm{~F}_{3} \mathrm{~N}_{2} \mathrm{O}_{5} \mathrm{~S}$ required $\mathrm{m} / \mathrm{z} 328.26$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{OH} 100 / 0$ to $75 / 25$ in $30 \mathrm{~min}, \lambda=254$ $\mathrm{nm}, \mathrm{F}=1 \mathrm{~mL} / \mathrm{min}$, showed one peak with tR 18.51 min .
[119] 2'-Mercapto-2'-deoxyuridine
2'-S-acetyl-3',5'-di- $O$-acetyluridine [126] ( $1.0 \mathrm{~g}, 2.59 \mathrm{mmol}$ ) was
 dissolved in EtOH $(20 \mathrm{~mL})$ and cooled to $0^{\circ} \mathrm{C}$ in a ice bath. An aqueous solution of $\mathrm{KOH}(2 \mathrm{~N}, 20 \mathrm{~mL})$ was added dropwise and the mixture was
 stirred for 15 minutes. Keeping the flask in the cool bath, the reaction solution was neutralised with HCl 2 N aqueous solution and the volatiles evaporated. The crude was packed in a silica slurry and purified by Biotage Isolera One (25g SNAP cartridge C-18 KP-SIL, $25 \mathrm{~mL} / \mathrm{min}$, using water as eluent solvent over 30 minutes) to yield the desired product as a white solid $(0.59 \mathrm{~g}, 87 \%)$.
(Lit. mp: 174-175 ${ }^{\circ} \mathrm{C}$ ). ${ }^{445}$
${ }^{1} \mathbf{H}$ NMR (CD $\mathbf{H}_{\mathbf{3}} \mathbf{O D}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 7.94(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6), 5.99(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 1$ '), 5.76 (d, $J=8.1 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5), 4.22(\mathrm{dd}, J=5.1 \mathrm{~Hz}, 1.9 \mathrm{~Hz}, \mathrm{H} 3$ '), $4.08(\mathrm{tt}, J=2.6 \mathrm{~Hz}, 1.9 \mathrm{~Hz}, 1 \mathrm{H}$, H4'), 3.779 (dd, 20.5 Hz, 11.9 Hz, 1H, H5'), 3.772 (dd, $20.5 \mathrm{~Hz}, 11.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H})^{\prime}$ ), 3.53 (dd, $J=8.6 \mathrm{~Hz}, 5.3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 2$ ').
${ }^{13} \mathbf{C}$ NMR (CD $\mathbf{N O D}_{\mathbf{3}} \mathbf{O D} \boldsymbol{\delta}_{\mathbf{C}} 164.64(\mathrm{C}=\mathrm{O}), 151.18$ ( $\mathrm{C}=\mathrm{O}$ ), 141.28 (C6), 102.00 (C5), 88.83 (C1'), 86.80 (C4'), 72.31 ( $\mathrm{C}^{\prime}$ ), 61.45 (C5'), 56.68 ( C 2 ').
MS (ES+) $\mathbf{m} / \mathbf{z}$ found $299.0\left(\mathrm{M}+\mathrm{K}^{+}\right), 283.0\left(\mathrm{M}+\mathrm{Na}^{+}\right), 557.1\left(2 \mathrm{M}+\mathrm{K}^{+}\right) \mathrm{C}_{9} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{5} \mathrm{~S}$ required $\mathrm{m} / \mathrm{z} 260.05$.
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{OH} 100 / 0$ to $75 / 25$ in $30 \mathrm{~min}, \lambda=280$ $\mathrm{nm}, \mathrm{F}=1 \mathrm{~mL} / \mathrm{min}$, showed one peak with tR 8.26 min .

## [125] 3',5'-Bis-O-acetyl-2,2'-anhydrouridine



2, ${ }^{\prime}$ '-Anhydrouridine ( $1.0 \mathrm{~g}, 4.42 \mathrm{mmol}$ ) was suspended in $\mathrm{CH}_{3} \mathrm{CN}$ (20 $\mathrm{mL})$ at room temperature and DMAP $(0.22 \mathrm{~g}, 1.77 \mathrm{mmol})$ was added in one portion, followed by the dropwise addition of $\mathrm{Ac}_{2} \mathrm{O}(1.67 \mathrm{~mL}, 17.68$ mmol ) and triethylamine ( $2.46 \mathrm{~mL}, 17.68 \mathrm{mmol}$ ). The solution was stirred at room temperature for 16 hours, the solvent evaporated and the crude was dissolved in water $(20 \mathrm{~mL})$ and washed once with $\mathrm{Et}_{2} \mathrm{O}(20 \mathrm{~mL})$ and the desired compound was extracted from the aqueous solution with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( $20 \mathrm{~mL} \times 3$ times). The $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ solution was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to yield the desired compound as a white solid ( $1.33 \mathrm{~g}, 97 \%$ ). (Lit. mp: 178-179 ${ }^{\circ} \mathrm{C}$ )..$^{448}$
${ }^{1} \mathbf{H}$ NMR ( $\mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 7.37(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6), 6.32\left(\mathrm{~d}, J=5.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H1}{ }^{\prime}\right), 6.07(\mathrm{~d}$, $J=7.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5), 5.47-5.44\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}{ }^{\prime}\right)$, $5.42-5.40\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 3^{\prime}\right), 4.52\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H} 4{ }^{\prime}\right)$, 4.33 (dd, $\left.J=3.8 \mathrm{~Hz}, 12.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5^{\prime}\right), 4.05$ (dd, $J=3.5 \mathrm{~Hz}, 12.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ '), 2.18 (s, $3 \mathrm{H}, \mathrm{CH}_{3} \mathrm{CO}$ ), $2.08\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3} \mathrm{CO}\right)$.
${ }^{13} \mathbf{C}$ NMR ( $\left.\mathbf{C D C l}_{3}\right) \boldsymbol{\delta}_{\mathbf{C}} 171.37(\mathrm{C} 2), 170.53\left(\mathrm{CH}_{3} \mathrm{CO}\right), 169.62\left(\mathrm{CH}_{3} \mathrm{CO}\right), 159.60(\mathrm{C} 4)$, 134.37 (C5), 110.66 (C6), 90.68 ( $\mathrm{C}^{\prime}$ ), 86.46 (C2'), 85.67 (C3'), 78.02 (C4'), 63.52 (C5'), $20.59\left(\mathrm{CH}_{3} \mathrm{CO}\right), 20.51\left(\mathrm{CH}_{3} \mathrm{CO}\right)$.
[126] 2'-S-Acetyl-3',5'-di-O-acetyluridine


3',5'-Bis- $O$-acetyl-2,2'-anhydrouridine [125] ( $3.5 \mathrm{~g}, 11.28 \mathrm{mmol}$ ) was suspended in anhydrous dioxane ( 30 mL ) at room temperature and thioacetic acid ( $19.25 \mathrm{~mL}, 270.7 \mathrm{mmol}$ ) was added dropwise. The reaction was refluxed for 12 hours, then allowed to reach room temperature and the solvent evaporated. The crude was suspended in water ( 50 mL ) and washed with $\mathrm{Et}_{2} \mathrm{O}(50 \mathrm{~mL})$; the compound was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL})$ from the aqueous solution, and the $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ solution was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated. The crude was purified by Biotage Isolera One ( 25 g SNAP cartridge KP-SIL, $75 \mathrm{~mL} / \mathrm{min}$, using a gradient elution of $15-100 \% \mathrm{EtOAc} / n \mathrm{Hex}$ over $8 \mathrm{CV}, 100 \% \mathrm{EtOAc} / n \mathrm{Hex}$ over 7 CV ) to yield the title compound as a white solid ( $2.68 \mathrm{~g}, 73 \%$ ). (Lit. mp: 124-126 ${ }^{\circ} \mathrm{C}$ )..$^{455,446}$
${ }^{\mathbf{1}} \mathbf{H}$ NMR ( $\mathbf{C D C l}_{3}$ ) $\boldsymbol{\delta}_{\mathbf{H}} 8.63(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{NH}), 7.32(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6), 6.29(\mathrm{~d}, J=9.8$ Hz, 1H, H1'), 5.84 (d, $J=8.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5$ ), 5.32-5.29 (m, 1H, H3'), 4.51 (dd, $J=13.4 \mathrm{~Hz}$, 4.6 Hz, 1H, H5'), 4.34-4.27 (m, 3H, H4', H2', H5'), 2.37 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH} H_{3} \mathrm{COS}$ ), 2.23 (s, 3H, $\mathrm{CH}_{3} \mathrm{CO}_{2}$ ), $2.17\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3} \mathrm{CO}_{2}\right)$.
${ }^{13} \mathbf{C}$ NMR $\left(\mathbf{C D C l}_{3}\right) \delta_{\mathbf{C}} 192.99\left(\mathrm{CH}_{3} \mathrm{COS}\right), 170.07\left(\mathrm{OCOCH}_{3}\right), 169.77\left(\mathrm{OCOCH}_{3}\right), 162.11$ (C2), 150.33 (C4), 138.68 (C6), 103.61 (C5), 86.51 (C1'), 81.98 (C4'), 74.38 (C3'), 63.83 ( $\mathrm{C}^{\prime}$ '), $46.75\left(\mathrm{C}^{\prime}\right), 30.40\left(\mathrm{CH}_{3} \mathrm{COS}\right), 20.78\left(\mathrm{OCOCH}_{3}\right), 20.63\left(\mathrm{OCOCH}_{3}\right)$.

### 9.5.20 Synthesis of 2'-SCF ${ }_{3}$-2'-dU ProTides

## [128a] 2'-Trifluoromethylthio-2'-deoxyuridine-5'-phenyl-(benzyloxy-Lalaninyl)phosphate



Prepared according to general procedure $\mathbf{F}_{\mathbf{2}}$ using 2'-trifluoromethylthio-2'-deoxyuridine [115] ( $0.025 \mathrm{~g}, 0.076$ mmol ) in THF (2 mL), phenyl(benzyloxy-Lalaninyl)phosphorochloridate [17s] ( $0.054 \mathrm{~g}, 0.152 \mathrm{mmol})$ in THF ( 0.5 mL ) and $N$-methylimidazole ( $30 \mu \mathrm{~L}, 0.38$ $\mathrm{mmol})$. The mixture was evaporated, dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and washed with HCl
aqueous 0.5 N ( $3 \times 1.5 \mathrm{~mL}$ ), the organic phase dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated. The crude was purified via Biotage Isolera One (10g SNAP cartridge KP-SIL, $36 \mathrm{~mL} / \mathrm{min}$, using a gradient elution of $1-14 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over $10 \mathrm{CV}, 14 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over $2 \mathrm{CV})$, to yield the title compound as a white solid ( $0.043 \mathrm{~g}, 87 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\left.\mathbf{C D}_{\mathbf{3}} \mathbf{O D}, \mathbf{2 0 2} \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 3.75, 4.04.
${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathbf{C D}_{3} \mathbf{O D}, 500 \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathbf{H}} 7.67(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 6), 7.59(\mathrm{~d}, J=8.0 \mathrm{~Hz}$, $0.5 \mathrm{H}, \mathrm{H} 6), 7.41-7.28(\mathrm{~m}, 7 \mathrm{H}, \mathrm{Ph}), 7.26-7.19(\mathrm{~m}, 3 \mathrm{H}, \mathrm{Ph}), 6.31(\mathrm{t}, J=9.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{Hl}$ '), 5.73 (d, $J=8.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 5$ ), 5.66 (d, $J=8.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 5$ ), $5.19-5.12$ (m, 2H, CH2 Ph ), 4.49 (dd, $J=5.6 \mathrm{~Hz}, 1.4 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 3$ '), 4.47 (dd, $J=5.6 \mathrm{~Hz}, 1.4 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 3$ '), 4.354.31 (m, 1H, H5'), 4.30-4.27 (m, 1H, H5'), 4.23-4.18 (m, 1H, H4'), 4.08-4.00 (m, 1H, $\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ), 3.94 (dd, $J=5.5 \mathrm{~Hz}, 9.15 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 2$ '), $3.91(\mathrm{dd}, J=5.5 \mathrm{~Hz}, 9.15 \mathrm{~Hz}$, $0.5 \mathrm{H}, \mathrm{H} 2$ '), 1.41-1.33 (m, 3H, $\left.\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$.
${ }^{13} \mathbf{C}$ NMR ( $\mathbf{C D}_{\mathbf{3}} \mathbf{O D}, \mathbf{1 2 5} \mathbf{~ M H z}$ ) $\boldsymbol{\delta}_{\mathrm{C}} 174.93\left(\mathrm{~d},{ }^{3} J_{C P}=3.7 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.57\left(\mathrm{~d},{ }^{3} J_{C P}=3.7\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 165.56 (C2), 165.50 (C2), 152.37 (C4), 152.33 (C4), 152.09 (d, $J_{C P}=2.5 \mathrm{~Hz}$, C-Ar), 152.04 (d, $\left.J_{C P}=2.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 141.87$ (C6), 141.75 (C6), 137.22 (C-Ar), 137.21 (C-Ar), $132.28\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=303.7 \mathrm{~Hz}, \mathrm{SCF}_{3}\right), 132.22\left(\mathrm{q},{ }^{1} J_{\mathrm{C}-\mathrm{F}}=303.7 \mathrm{~Hz}, \mathrm{SCF}_{3}\right), 130.98$ (CH-Ar), 130.92 (CH-Ar), 129.66 (CH-Ar), 129.44 (CH-Ar), 129.42 (CH-Ar), 129.38 (CH-Ar), 129.36 (CH-Ar), 126.38 (CH-Ar), 121.36 (CH-Ar), 121.33 (CH-Ar), 121.30 (CH-Ar), 104.13 (C5), 104.10 (C5), 89.97 (C1'), 89.83 (C1'), 86.06 (d, ${ }^{3} J_{C P}=8.1 \mathrm{~Hz}$, C 4 '), $85.90\left(\mathrm{~d},{ }^{3} J_{C P}=8.1 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $73.65\left(\mathrm{C} 3^{\prime}\right), 73.53\left(\mathrm{C} 3^{\prime}\right), 68.09\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 67.65(\mathrm{~d}$, ${ }^{2} J_{C P}=9.1 \mathrm{~Hz}, \mathrm{C} 5$ '), $67.60\left(\mathrm{~d}^{2}{ }^{2} J_{C P}=9.1 \mathrm{~Hz}, \mathrm{C} 5 '\right), 51.90\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 51.74\left(\mathrm{CHCH}_{3}\right.$ L-Ala), 51.54 (C2'), 51.45 (C2'), 20.43 (d, ${ }^{3} J_{C P}=6.1 \mathrm{~Hz}, \mathrm{CHCH}_{3}$ L-Ala), $20.18\left(\mathrm{~d},{ }^{3} J_{C P}=\right.$ $\left.6.1 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$.

## ${ }^{19} \mathbf{F}$ NMR ( $\left.\mathbf{C D}_{\mathbf{3}} \mathbf{O D}, 470 \mathbf{M H z}\right) \boldsymbol{\delta}_{\mathrm{F}}-41.37$-41.26.

MS (ES+ $\mathbf{)} \mathbf{m} / \mathbf{z}$ found $668.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{26} \mathrm{H}_{27} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}_{9}$ PS required $\mathrm{m} / \mathrm{z} 645.12$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN} 90 / 10$ to $0 / 100$ in $30 \mathrm{~min}, \lambda=254$ $\mathrm{nm}, \mathrm{F}=1 \mathrm{ml} / \mathrm{min}$, showed two peaks with $\mathrm{tR}=17.36,17.48 \mathrm{~min}$.

## [128b] 2'-Trifluoromethylthio-2'-deoxyuridine-5'-phenyl-(hex-1-yloxy-L-alaninyl) phosphate



Prepared according to general procedure $\mathbf{F}_{2}$ using 2'-trifluoromethylthio-2'-deoxyuridine [115] ( $0.025 \mathrm{~g}, 0.076$ $\mathrm{mmol})$, THF ( 2 mL ), phenyl(hex-1-yloxy-L-alaninyl) phosphorochloridate [ $\mathbf{1 7 u}$ ] $(0.05 \mathrm{~g}, 0.152 \mathrm{mmol})$ in 0.5 mL of THF and $N$-methylimidazole ( $30 \mu \mathrm{~L}, 0.38 \mathrm{mmol}$ ). The
compound was then dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and washed with $\mathrm{HCl} 0.5 \mathrm{~N}(3 \times 1.5 \mathrm{~mL})$, the organic phase dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to dryness. Purification was performed via Biotage Isolera One (10g SNAP cartridge KP-SIL using a gradient elution of 1-10\% $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over $10 \mathrm{CV}, 10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over 5 CV ), yielding the title compound as a white solid $(0.027 \mathrm{~g}, 56 \%)$.
${ }^{31} \mathbf{P} \mathbf{N M R}\left(\mathbf{C D}_{\mathbf{3}} \mathbf{O D}, \mathbf{2 0 2} \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 3.84, 4.07.
${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathbf{C D}_{\mathbf{3}} \mathbf{O D}, \mathbf{5 0 0} \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathbf{H}} 7.59(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 6), 7.48(\mathrm{~d}, J=8.0 \mathrm{~Hz}$, $0.5 \mathrm{H}, \mathrm{H} 6$ ), 7.31-7.19 (m, 2H, Ph), 7.18-7.07 (m, 3H, Ph), 6.22 (d, $J=9.1 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{Hl}$ '), 6.19 (d, $\left.J=9.3 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{Hl}^{\prime}\right), 5.64$ (d, $J=8.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 5$ ), 5.55 (d, $J=8.0 \mathrm{~Hz}, 0.5 \mathrm{H}$, H5), 4.40 (dd, $J=5.5 \mathrm{~Hz}, 1.2 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 3$ '), 4.36 (dd, $J=5.5,1.2 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 3$ '), 4.294.19 (m, 2H, H5'), 4.16-4.10 (m, 1H, H4'), 4.01-3.95 (m, 2H, CH2 $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n-$ Hex), 3.89-3.76 (m, 2H, CHCH $\mathrm{CH}_{3}$ L-Ala, H2'), 1.55-1.47 (m, 2H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ $n$-Hex), 1.29-1.14 (m, 9H, $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex, $\mathrm{CHCH}_{3}$ L-Ala), 0.82-0.74 (m, $3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n$-Hex).
${ }^{13} \mathbf{C} \mathbf{N M R}\left(\mathbf{C D}_{3} \mathbf{O D}, \mathbf{1 2 5} \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathbf{C}} 173.82\left(\mathrm{~d},{ }^{3} J_{C P}=3.8 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 174.17\left(\mathrm{~d},{ }^{3} J_{C P}=3.8\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 164.16 (C2), 164.10 (C2), 150.96 (C4), 150.91 (C4), 150.69 ( $\mathrm{d}, J_{C P}=12.7 \mathrm{~Hz}$, C-Ar), 150.55 ( $\mathrm{d}, J_{C P}=12.7 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}$ ), 140.51 (C6), 140.30 (C6), $130.89\left(\mathrm{q} \mathrm{x} \mathrm{2},{ }^{1} J_{\mathrm{CF}}=\right.$ $303.5 \mathrm{~Hz}, \mathrm{CF} 3$ ), 129.67 (CH-Ar), 129.59 (CH-Ar), 129.52 (CH-Ar), 129.21 (CH-Ar), 124.99 (CH-Ar), 124.27 (CH-Ar), 120.33 (CH-Ar), 120.29 (CH-Ar), 119.97 (CH-Ar), 119.93 (CH-Ar), 119.89 (CH-Ar), 102.72 (C5), 102.71 (C5), 88.47 (C1'), 88.28 (C1'), $84.68\left(\mathrm{~d},{ }^{3} J_{C P}=8.5 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $84.54\left(\mathrm{~d},{ }^{3} J_{C P}=8.5 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), 72.31 ( $\left.\mathrm{C}^{\prime}\right), 72.20$ ( $\mathrm{C}^{\prime}$ ), $66.28\left(\mathrm{~d},{ }^{2} J_{C P}=11.5 \mathrm{~Hz}, \mathrm{C}^{\prime}\right), 66.19\left(\mathrm{~d},{ }^{2} J_{C P}=11.5 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), 65.17 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex), $65.14\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \quad n\right.$-Hex), 50.43 $\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 50.26\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 50.12\left(\mathrm{C} 2\right.$ '), $50.06\left(\mathrm{C} 2{ }^{\prime}\right), 31.24\left(\mathrm{CH}_{2} n\right.$-Hex), $31.21\left(\mathrm{CH}_{2} n\right.$-Hex $), 28.30\left(\mathrm{CH}_{2} n\right.$-Hex $), 28.28\left(\mathrm{CH}_{2} n\right.$-Hex $), 25.30\left(\mathrm{CH}_{2} n\right.$-Hex $), 25.25$ $\left(\mathrm{CH}_{2} n\right.$-Hex), $22.25\left(\mathrm{CH}_{2} n\right.$-Hex), $22.22\left(\mathrm{CH}_{2} n\right.$-Hex $), 19.46\left(\mathrm{~d},{ }^{3} J_{C P}=6.0 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\right.$ Ala), 19.44 (d, ${ }^{3} J_{C P}=6.12 \mathrm{~Hz}, \mathrm{CHCH}_{3}$ L-Ala), $19.20\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$-Hex), $19.15\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} n\right.$ - Hex$)$.
${ }^{19}$ FNMR ( $\mathbf{C D}_{3} \mathbf{O D}, 470 \mathbf{M H z}$ ) $\boldsymbol{\delta}_{\mathbf{F}}-41.16,-41.26$.
MS (ES+ $\mathbf{~ m} / \mathbf{z}$ found $662.2\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{25} \mathrm{H}_{33} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}_{9} \mathrm{PS}$ required $\mathrm{m} / \mathrm{z} 639.16$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN} 90 / 10$ to $0 / 100$ in 30 min , $\lambda=254$ $\mathrm{nm}, \mathrm{F}=1 \mathrm{~mL} / \mathrm{min}$, showed two peaks with t ( $=20.08,20.26 \mathrm{~min}$.
[128c] 2'-Trifluoromethylthio-2'-deoxyuridine-5'-naphth-1-yl-(benzyloxy-L-alaninyl) phosphate


Prepared according to general procedure $\mathbf{F}_{\mathbf{2}}$ using 2'-trifluoromethylthio-2'-deoxyuridine [115] ( $0.015 \mathrm{~g}, 0.046$ $\mathrm{mmol})$, THF ( 2 mL ), naphth-1-yl(benzyloxy-L-alaninyl) phosphorochloridate [17a] ( $0.054 \mathrm{~g}, 0.137 \mathrm{mmol})$ in THF $(0.5 \mathrm{~mL})$ and $N$-methylimidazole ( $17 \mu \mathrm{~L}, 0.23 \mathrm{mmol}$ ). The compound was then dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and washed with $\mathrm{HCl} 0.5 \mathrm{~N}(3 \times 1.5 \mathrm{~mL})$, the organic phase dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to dryness to yield a white solid. The mixture was evaporated and the desired product isolated via Biotage Isolera One (SNAP Ultra $10 \mathrm{~g}, 36 \mathrm{~mL} / \mathrm{min}, 1-10 \%$ $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2} 10 \mathrm{CV}, 10 \% 5 \mathrm{CV}$ ) to yield the title compound as a white solid ( 0.013 g , $42 \%)$.
${ }^{31} \mathbf{P}$ NMR (CD $\mathbf{3}_{\mathbf{3}} \mathbf{O D}, \mathbf{2 0 2} \mathbf{~ M H z )} \boldsymbol{\delta}_{\mathbf{P}}$ 4.45, 4.22.
 (m, 1H, H6), 7.59-7.49 (m, 4H, Ar), 7.46-7.40 (m, 1H, Ar), 7.35-7.28 (m, 5H, Ar), 6.30 (d, $J=9.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1$ ') , 6.27 (d, $\left.J=9.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H}{ }^{\prime}\right), 5.58$ (d, $J=8.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 5$ ), $5.52(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 5), 5.11\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}\right), 5.10\left(\mathrm{ABq}, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{Ph}\right)$, 4.47 (dd, $J=6.0,1.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3$ '), 4.42-4.29 (m, 2H, H5'), 4.22-4.18 (m, 1H, H4'), 4.144.04 (m, 1H, CHCH ${ }_{3}$ L-Ala), 3.88-3.81 (m, 1H, H2'), 1.37-1.33 (m, 3H, $\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ).
${ }^{13} \mathbf{C}$ NMR ( $\left.\mathbf{C D}_{\mathbf{3}} \mathbf{O D}, \mathbf{1 2 5} \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathrm{C}} 173.49\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=3.6 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 173.10\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=3.6\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}$ ), 164.08 (C2), 164.05 (C2), 150.90 (C4), 150.87 (C4), 146.51 (C-Ar), 146.45 (CAr), 146.39 (C-Ar), 146.28 (CH-Ar), 140.25 (C6), 140.26 (C6), 135.73 (C-Ar), 134.95 (CAr), $130.76\left(2 \mathrm{x} \mathrm{q},{ }^{1} J_{\mathrm{CF}}=304.1 \mathrm{~Hz}, \mathrm{CF}_{3}\right), 128.19(\mathrm{CH}-\mathrm{Ar}), 127.97(\mathrm{CH}-\mathrm{Ar}), 127.92(\mathrm{CH}-$ Ar), 127.90 (CH-Ar), 127.63 (CH-Ar), 127.59 (CH-Ar), 126.55 (CH-Ar), 126.53 (CH-Ar), 126.26 (CH-Ar), 126.17 (CH-Ar), 125.19 (CH-Ar), 125.18 (CH-Ar), 125.12 (CH-Ar), 124.77 ( $\mathrm{CH}-\mathrm{Ar}$ ), 121.20 ( $\mathrm{CH}-\mathrm{Ar}$ ), 121.04 (CH-Ar), 114.70 (d, ${ }^{3} J_{\mathrm{CP}}=3.4 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}$ ), $114.66\left(\mathrm{~d},{ }^{3} \mathrm{~J}_{\mathrm{CP}}=3.4 \mathrm{~Hz}, \mathrm{CH}-\mathrm{Ar}\right), 102.64$ (C5), 102.60 (C5), 88.53 ( C 1 '), 88.43 ( $\mathrm{Cl}^{\prime}$ ), $84.59\left(\mathrm{~d}^{3} J_{\mathrm{CP}}=8.0 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $84.44\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=8.0 \mathrm{~Hz}, \mathrm{C} 4{ }^{\prime}\right), 72.03\left(\mathrm{C} 3^{\prime}\right), 71.96\left(\mathrm{C} 3^{\prime}\right), 66.65$ $\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 66.27\left(\mathrm{~d},{ }^{2} J_{\mathrm{CP}}=4.9 \mathrm{~Hz}, \mathrm{C} 5\right.$ '), $66.23\left(\mathrm{~d},{ }^{2} \mathrm{~J}_{\mathrm{CP}}=4.9 \mathrm{~Hz}, \mathrm{C} 5\right.$ ' $), 50.58\left(\mathrm{CHCH}_{3} \mathrm{~L}-\right.$ Ala), $50.44\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $50.08\left(\mathrm{C}^{\prime}\right), 50.02\left(\mathrm{C} 2^{\prime}\right), 18.95\left(\mathrm{~d},{ }^{3} \mathrm{~J}_{\mathrm{CP}}=6.5 \mathrm{~Hz}, \mathrm{CHCH}_{3} \mathrm{~L}-\right.$ Ala), $18.74\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=6.5 \mathrm{~Hz}, \mathrm{CHCH}_{3}\right.$ L-Ala).
${ }^{19} \mathrm{~F}$ NMR ( $\mathrm{CD}_{3} \mathbf{O D}, 470 \mathbf{M H z}$ ) $\boldsymbol{\delta}_{\mathrm{F}}-41.11$-41.15.
$\mathbf{M S}\left(\mathbf{E S}+\right.$ ) $\mathbf{m} / \mathbf{z}$ found $718.1\left[\mathrm{M}+\mathrm{Na}^{+}\right], \mathrm{C}_{30} \mathrm{H}_{29} \mathrm{~F}_{3} \mathrm{~N}_{3} \mathrm{O}_{9} \mathrm{PS}$ required $\mathrm{m} / \mathrm{z} 695.60$ [M].

HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN} 90 / 10$ to $0 / 100$ in $30 \mathrm{~min}, \lambda=254$ $\mathrm{nm}, \mathrm{F}=1 \mathrm{~mL} / \mathrm{min}$, showed two peaks with $\mathrm{tR}=22.15,22.23 \mathrm{~min}$.

### 9.5.21 Synthesis of 2'-SH-2'-dU ProTide dimers

[129a] Bis-(5'-[phenyl(benzyloxy-L-alaninyl)phosphate]-2'-deoxyuridine-2'yl)disulfide


Prepared according to general procedure $\mathbf{F}_{2}$ using 2'-mercapto-2'-deoxyuridine [119] ( $0.05 \mathrm{~g}, 0.19 \mathrm{mmol}$ ), anh. THF (5 mL), phenyl(benzyloxy-Lalaninyl)phosphorochloridate [17s] ( $0.14 \mathrm{~g}, 0.38 \mathrm{mmol}$ ) in THF ( 1 mL ) and $N$-methylimidazole ( $76 \mu \mathrm{~L}, 0.95$ mmol ). The compound was then dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 5 mL ) and washed with HCl aqueous $0.5 \mathrm{~N}(3 \times 1.5 \mathrm{~mL})$, the organic phase dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to dryness. Purification was performed via Biotage Isolera One ( 10 g SNAP cartridge KP-SIL using a gradient elution of $1-10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over $10 \mathrm{CV}, 10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over 5 CV ), yielding the title compound as a white solid ( $0.046 \mathrm{~g}, 21 \%$ ).
${ }^{31} \mathbf{P}$ NMR ( $\left.\mathbf{C D}_{\mathbf{3}} \mathbf{O D}, \mathbf{2 0 2} \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 3.94, 3.62, 3.58.
${ }^{1} \mathbf{H}$ NMR (CD $\left.\mathbf{3}_{\mathbf{3}} \mathbf{O D}, 500 \mathrm{MHz}\right) \boldsymbol{\delta}_{\mathbf{H}} 7.456(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 6), 7.453(\mathrm{~d}, J=8.0 \mathrm{~Hz}$, $0.25 \mathrm{H}, \mathrm{H} 6$ ), 7.43 (d, $J=8.0 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 6$ ), 7.42 (d, $J=8.0 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 6$ ), $7.26-7.16$ (m, $7 \mathrm{H}, \mathrm{Ph}), 7.13-7.03(\mathrm{~m}, 3 \mathrm{H}, \mathrm{Ph}), 6.18$ (d, $\left.J=8.0 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H}{ }^{\prime}\right), 6.17$ (d, $J=8.0 \mathrm{~Hz}$, $0.25 \mathrm{H}, \mathrm{Hl}$ '), 6.158 (d, $J=8.0 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 1$ '), 6.153 (d, $J=8.0 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 1$ '), $5.57-$ 5.51 (m, 1H, H5), 5.07-4.98 (m, 2H, CH 2 Ph ), 4.32-4.27 (m, 1H, H3'), 4.22-4.10 (m, 2H, H5'), 4.02-3.87 (m, 2H, H4', $\mathrm{CHCH}_{3}$ L-Ala), 3.56 (dd, $J=8.0 \mathrm{~Hz}, 6.1 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 2$ '), 3.53 (dd, $J=8.0 \mathrm{~Hz}, 6.1 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 2$ '), 3.41 (dd, $J=8.0 \mathrm{~Hz}, 6.1 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 2$ '), 3.37 (dd, $J=8.0 \mathrm{~Hz}, 6.1 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 2$ '), 1.26-1.20 (m, 3H, $\mathrm{CHCH}_{3}$ L-Ala).
${ }^{13} \mathbf{C}$ NMR (CD30D, $125 \mathbf{M H z}$ ) $\boldsymbol{\delta}_{\mathbf{C}} 175.05\left(\mathrm{~d},{ }^{3} J_{C P}=3.7 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 175.01\left(\mathrm{~d},{ }^{3} J_{C P}=3.7\right.$ $\mathrm{Hz}, \mathrm{C}=\mathrm{O}), 174.75\left(\mathrm{~d},{ }^{3} J_{C P}=4.8 \mathrm{~Hz}, \mathrm{C}=\mathrm{O}\right), 165.68(\mathrm{C} 2), 152.45\left(\mathrm{~d}, J_{C P}=2.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right)$, $152.40\left(\mathrm{~d}, J_{C P}=2.5 \mathrm{~Hz}, \mathrm{C}-\mathrm{Ar}\right), 152.13(\mathrm{C} 4), 152.08$ (C4), 152.07 (C4), 142.24 (C6), 142.19 (C6), 142.08 (C6), 142.03 (C6), 137.21 (C-Ar), 137.19 (C-Ar), 130.94 (CH-Ar), 129.66 (CH-Ar), 129.43 (CH-Ar), 129.40 (CH-Ar),129.34 (CH-Ar), 128.96 (CH$\mathrm{Ar}), 126.44$ (CH-Ar), 126.33 (CH-Ar), 121.43 (CH-Ar), 121.39 (CH-Ar), 103.86 (C5), 103.84 (C5), 103.81 (C5), 103.77 (C5), 89.99 (C1'), 89.62 (C1'), 85.85 (d, ${ }^{3} J_{C P}=7.5 \mathrm{~Hz}$, C 4 '), $85.79\left(\mathrm{~d},{ }^{3} J_{C P}=7.5 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $85.64\left(\mathrm{~d},{ }^{3} J_{C P}=7.5 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), $85.61\left(\mathrm{~d},{ }^{3} J_{C P}=7.5 \mathrm{~Hz}\right.$,

C4'), 73.48 ( $\mathrm{C}^{\prime}$ ), 73.37 ( C 3 '), 73.26 ( C 3 '), 73.21 ( $\mathrm{C}^{\prime}$ '), $68.16\left(\mathrm{CH}_{2} \mathrm{Ph}\right), 68.17\left(\mathrm{CH}_{2} \mathrm{Ph}\right)$, 68.12 ( $\mathrm{CH}_{2} \mathrm{Ph}$ ), 67.62 ( $\mathrm{C}^{\prime}$ ), 67.58 ( $\mathrm{C}^{\prime}$ ), 67.54 ( $\mathrm{C}^{\prime}$ ), 57.52 ( $\mathrm{C}^{\prime}$ ), 57.42 ( $\mathrm{C}^{\prime}$ ), 57.31 (C2'), 57.27 (C2'), $51.84\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $51.69\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 20.55\left(\mathrm{~d},{ }^{3} J_{C P}=6.2 \mathrm{~Hz}\right.$, $\mathrm{CHCH}_{3}$ L-Ala), 20.53 (d, ${ }^{3} J_{C P}=6.2 \mathrm{~Hz}, \mathrm{CHCH}_{3}$ L-Ala), $20.40\left(\mathrm{~d},{ }^{3} J_{C P}=6.2 \mathrm{~Hz}, \mathrm{CHCH}_{3}\right.$ L-Ala), 20.34 ( $\mathrm{d},{ }^{3} J_{C P}=6.2 \mathrm{~Hz}, \mathrm{CHCH}_{3}$ L-Ala).
MS (ES+ $) \mathbf{m} / \mathbf{z}$ : found: $1175.3\left[\mathrm{M}+\mathrm{Na}^{+}\right] . \mathrm{C}_{50} \mathrm{H}_{54} \mathrm{~N}_{6} \mathrm{O}_{18} \mathrm{P}_{2} \mathrm{~S}_{2}$ required $\mathrm{m} / \mathrm{z} 1152.24$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN} 90 / 10$ to $0 / 100$ in $30 \mathrm{~min}, \lambda=254$ $\mathrm{nm}, \mathrm{F}=1 \mathrm{~mL} / \mathrm{min}$, showed two broad peaks with $\mathrm{tR}=19.27,19.44 \mathrm{~min}$.
[129b] Bis-(5'-[naphth-1-yl(benzyloxy-L-alaninyl)phosphate]-2'-deoxyuridine-2'yl)disulfide


Prepared according to general procedure $\mathbf{F}_{\mathbf{2}}$ using 2'-mercapto-2'-deoxyuridine [119] ( $0.10 \mathrm{~g}, 0.38 \mathrm{mmol}$ ), anh. THF (10 mL), naphthyl(benzoxy-Lalaninyl)phosphorochloridate $[\mathbf{1 7 a}](0.46 \mathrm{~g}, 1.15 \mathrm{mmol})$ in 3 mL of anh. THF and $N$-methylimidazole ( $143 \mu \mathrm{~L}$, $1.9 \mathrm{mmol})$. The compound was then dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(5 \mathrm{~mL})$ and washed with $\mathrm{HCl} 0.5 \mathrm{~N}(3 \times 1.5 \mathrm{~mL})$, the organic phase dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to dryness. Purification was performed via Biotage Isolera One ( 10 g SNAP cartridge Ultra, $36 \mathrm{~mL} / \mathrm{min}$, gradient elution of $1-14 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over 10 $\mathrm{CV}, 14 \%$ over 5 CV ), yielding the title compound as a white solid ( $0.053 \mathrm{~g}, 11 \%$ ).
${ }^{31} \mathbf{P} \mathbf{N M R}\left(\mathbf{C D}_{\mathbf{3}} \mathbf{O D}, \mathbf{2 0 2} \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 4.42, 4.40, 4.09, 4.05.
${ }^{1} \mathbf{H}$ NMR (CD $\left.\mathbf{H}_{3} \mathbf{O D}, \mathbf{5 0 0} \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathbf{H}}$ 7.97-7.91(m, 1H, Ar), 7.69-7.62 (m, 1H, Ar), 7.51-7.44 (m, 1H, H6), 7.35-7.26 (m, 3H, Ar), 7.20-7.02 (m, 7H, Ar), 6.06-5.99 (m, 1H, H1'), 5.27 (d, $J=8.5 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 5), 5.25$ (d, $J=8.5 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 5$ ), 5.20 (d, $J=8.5 \mathrm{~Hz}, 0.25 \mathrm{H}$, H5), 5.19 (d, $J=8.5 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 5), 4.94-4.84$ (m, 2H, CH2Ph), 4.18-4.01 (m, 3H, H3', H5'), 3.95-3.83 (m, 2H, H4', $\mathrm{CHCH}_{3}$ L-Ala), 3.20 (dd, $J=8.0 \mathrm{~Hz}, 6.0 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 2$ '), $3.13\left(\mathrm{dd}, J=8.0 \mathrm{~Hz}, 6.0 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 2{ }^{\prime}\right), 3.05(\mathrm{dd}, J=8.0 \mathrm{~Hz}, 6.0 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 2$ '), 2.98 (dd, $J=8.0 \mathrm{~Hz}, 6.0 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 2$ ) $), 1.20-1.12$ ( $\mathrm{m}, 3 \mathrm{H}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ).
${ }^{13} \mathbf{C}$ NMR ( $\left.\mathbf{C D}_{\mathbf{3}} \mathbf{O D}, \mathbf{1 2 5} \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathbf{C}} 173.70(\mathrm{C}=\mathrm{O})$, $173.69(\mathrm{C}=\mathrm{O})$, $173.67(\mathrm{C}=\mathrm{O})$, 173.36 $(\mathrm{C}=\mathrm{O}), 173.33(\mathrm{C}=\mathrm{O}), 173.23(\mathrm{C}=\mathrm{O}), 173.20(\mathrm{C}=\mathrm{O}), 164.16(\mathrm{C} 2), 164.09(\mathrm{C} 2), 150.97$ (C4), 150.93 (C4), 146.56 (C-Ar), 146.51 (C-Ar), 146.45 (C-Ar), 146.39 (C-Ar), 140.51 (C6), 140.44 (C6), 140.20 (C6), 140.13 (C6), 135.70 (C-Ar), 134.89 (C-Ar), 128.22 (CHAr), 128.18 (CH-Ar), 127.97 (CH-Ar), 127.95 (CH-Ar), 127.93 (CH-Ar), 127.91 (CH-Ar), 127.83 (CH-Ar), 127.70 (CH-Ar), 127.67 (CH-Ar), 126.61 (CH-Ar), 126.59 (CH-Ar),
126.57 (CH-Ar), 126.30 (CH-Ar), 126.28 (CH-Ar), 125.25 (CH-Ar), 125.19 (CH-Ar), 124.86 (CH-Ar), 121.25 (CH-Ar), 121.21 (CH-Ar), 115.07 (CH-Ar), 115.04 (CH-Ar), 115.03 (CH-Ar), 115.00 (CH-Ar), 114.81 (CH-Ar), 114.79 (CH-Ar), 102.39 (C5), 88.24 $\left(\mathrm{C}^{\prime}\right), 88.21\left(\mathrm{Cl}^{\prime}\right), 87.89\left(\mathrm{Cl}^{\prime}\right), 87.86\left(\mathrm{C}^{\prime}\right), 84.46\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=8.9 \mathrm{~Hz}, \mathrm{C} 4{ }^{\prime}\right), 84.38\left(\mathrm{~d}^{3} J_{\mathrm{CP}}=\right.$ $8.9 \mathrm{~Hz}, \mathrm{C} 4$ '), $84.20\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=8.9 \mathrm{~Hz}, \mathrm{C} 4\right.$ '), 84.13 (d, ${ }^{3} J_{\mathrm{CP}}=8.9 \mathrm{~Hz}, \mathrm{C} 4$ '), 72.05 (C3'), 71.94 ( C 3 '), 71.87 ( C 3 '), 71.82 ( C 3 '), 66.82-66.89 (m, CH2 Ph ), 66.39-66.18 (m, C5'), 55.61 (C2'), 55.54 (C2'), 55.42 (C2'), 50.49 ( $\mathrm{CHCH}_{3}$ L-Ala), 50.44 ( $\mathrm{CHCH}_{3}$ L-Ala), 19.14 $\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $19.08\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $19.02\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $18.96\left(\mathrm{CHCH}_{3}\right.$ L-Ala), $18.95\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$.

MS (ES+) m/z found: $1305.33\left(\mathrm{M}+\mathrm{Na}^{+}\right) \mathrm{C}_{60} \mathrm{H}_{64} \mathrm{~N}_{6} \mathrm{O}_{18} \mathrm{P}_{2} \mathrm{~S}_{2}$ required $\mathrm{m} / \mathrm{z} 1282.34$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN} 90 / 10$ to $0 / 100$ in $30 \mathrm{~min}, \lambda=254$ $\mathrm{nm}, \mathrm{F}=1 \mathrm{~mL} / \mathrm{min}$, showed three broad peaks with $\mathrm{t} \mathrm{R}=22.18,22.31,22.58 \mathrm{~min}$.
[129c]
Bis-(5'-[phenyl(hexyl-1-oxy-L-alaninyl)phosphate]-2'-deoxyuridine-2'yl)disulfide


Prepared according to general procedure $\mathbf{F}_{2}$ using 2'-mercapto-2'-deoxyuridine [119] ( $0.05 \mathrm{~g}, 0.19 \mathrm{mmol}$ ), anh. THF (5 mL), phenyl(hex-1-yloxy-Lalaninyl)phosphorochloridate $[\mathbf{1 7 u}](0.013 \mathrm{~g}, 0.38$ mmol ) in THF ( 1 mL ) and $N$-methylimidazole ( $76 \mu \mathrm{~L}$, $0.95 \mathrm{mmol})$. The compound was then dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ and washed with HCl aqueous $0.5 \mathrm{~N}(3 \times 1.5 \mathrm{~mL})$, the organic phase dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and evaporated to dryness. Purification was performed via Biotage Isolera One (30g ZIP cartridge KP-SIL, gradient elution of $1-10 \% \mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ over 10 $\mathrm{CV}, 10 / 90$ over 5 CV ), yielding the title compound as a white solid ( $0.011 \mathrm{~g}, 5 \%$ ).
${ }^{\mathbf{3 1}} \mathbf{P} \mathbf{N M R}\left(\mathbf{C D}_{\mathbf{3}} \mathbf{O D}, \mathbf{2 0 2} \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathbf{P}}$ 3.97, 3.70, 3.66.
${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathbf{C D}_{\mathbf{3}} \mathbf{O D}, \mathbf{5 0 0} \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathbf{H}} 7.62(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 6), 7.61(\mathrm{~d}, J=8.0 \mathrm{~Hz}$, $0.25 \mathrm{H}, \mathrm{H} 6$ ), 7.59 (d, $J=8.0 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 6$ ), 7.58 (d, $J=8.0 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 6$ ), $7.41-7.35$ (m, $2 \mathrm{H}, \mathrm{Ph}), 7.30-7.19$ (m, 3H, Ph), 6.32 (d, $J=8.0 \mathrm{~Hz}, 0.5 \mathrm{H}, \mathrm{H} 1$ '), 6.30 (d, $J=8.0 \mathrm{~Hz}, 0.25 \mathrm{H}$, H1'), 6.28 (d, $J=8.0 \mathrm{~Hz}, 0.25 \mathrm{H}, \mathrm{H} 1$ '), 5.72-5.66 (m, 1H, H5), 4.46-4.41 (m, 1H, H3'), 4.38-4.27 (m, 2H, H5'), 4.19-4.07 (m, 3H, H4', $\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Hex}$ ), 4.04-3.95 ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}$ ), 3.75-3.64 (m, 0.5H, H2'), 3.57-3.48 (m, 0.5H, H2'), 1.68-1.59 (m, $4 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Hex}$ ), 1.41-1.27 (m, $7 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ Hex, $\mathrm{CHCH}_{3}$ L-Ala), $0.94-0.88\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Hex}\right.$ ).
${ }^{13} \mathbf{C}$ NMR ( $\left.\mathbf{C D}_{\mathbf{3}} \mathbf{O D}, \mathbf{1 2 5} \mathbf{~ M H z}\right) \boldsymbol{\delta}_{\mathrm{C}} 174.57(\mathrm{C}=\mathrm{O}), 174.15$ ( $\mathrm{C}=\mathrm{O}$ ), 151.01 (C2), 140.87 (C6), 140.83 (C6), 140.73 (C6), 140.69 (C6), 129.55 (CH-Ar), 125.53 (CH-Ar), 125.08 (CH-Ar), 125.02 (CH-Ar), 124.93 (CH-Ar), 124.61 (CH-Ar), 124.58 (CH-Ar), 120.02 (CHAr), 120.00 (CH-Ar), 119.98 (CH-Ar), 119.97 (CH-Ar), 102.49 (C5), 102.47 (C5), 102.43 (C5), 102.40 (C5), 102.38 (C5), 88.61 ( $\mathrm{Cl}^{\prime}$ ), 88.57 ( $\mathrm{Cl}^{\prime}$ ), 88.20 ( C 1 '), 88.19 ( C 1 '), $84.66-$ 84.06 (m, C4'), 72.15 ( $\mathrm{C} 3^{\prime}$ ), 72.06 ( $\mathrm{C}^{\prime}$ ), 72.04 ( $\left.\mathrm{C} 3^{\prime}\right), 71.94$ ( C 3 '), 66.27-66.00 (m, C5'), $65.20\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Hex}\right), 56.03\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right), 55.77\left(\mathrm{CHCH}_{3} \mathrm{~L}-\mathrm{Ala}\right)$, $31.19\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Hex}\right), 30.73\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Hex}\right), 28.28$ $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ Hex), $28.26 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ Hex), 25.26 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ Hex), $25.24 \quad\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ Hex), 22.20 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}\right.$ Hex), 19.34-18.38 (m, $\mathrm{CHCH}_{3} \quad$ L-Ala), 12.96 $\left(\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3} \mathrm{Hex}\right)$.
MS (ES+) m/z found: $1193.4\left[\mathrm{M}+\mathrm{Na}^{+}\right] \mathrm{C}_{50} \mathrm{H}_{72} \mathrm{~N}_{6} \mathrm{O}_{18} \mathrm{P}_{2} \mathrm{~S}_{2}$ required $\mathrm{m} / \mathrm{z} 1170.38$ [M].
HPLC Reverse-phase HPLC eluting with $\mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN} 90 / 10$ to $0 / 100$ in $30 \mathrm{~min}, \lambda=254$ $\mathrm{nm}, \mathrm{F}=1 \mathrm{Lmin}$, showed three broad peaks with $\mathrm{t}=23.96,24.06,24.28 \mathrm{~min}$.

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[^0]:    Table 3.17: Compound selected for evaluation on the KG1a cell line by Prof. C. Pepper (Cardiff University).

[^1]:    Table 3.22: Structures and yields of ProTides of 2F3'dA (49a-d) and 2( $\left.\mathrm{CH}_{3} \mathrm{O}\right) 3^{\prime} \mathrm{dA}(\mathbf{5 0 a} \mathbf{- d})$

[^2]:    Table 3.24: In vitro cytotoxicity screening of parent nucleoside 2F3'dA and ProTides 49a-d on selected cell lines. Cytotoxicity data reported as $\mu \mathrm{M} \mathrm{IC}_{50}$ values (concentration of drug causing $50 \%$ of cell viability inhibition) and $\mathrm{MI}_{\%}$ (maximum inhibitory effect of the drug at the range of concentrations considered). PTX: paclitaxel (control). Assays performed by WuXi AppTech.

[^3]:    Table 3.29 Structures of 2F3'dA ProTides assayed for cytotoxicity on KG1a cell line, reported as $\mathrm{LD}_{50}$ values $(\mu \mathrm{M})$. Assays performed by Prof. C. Pepper, Cardiff University.

[^4]:    Table 3.34 Substrates involved, reagents, conditions and results of Scheme 3.21 (step c.). ${ }^{\text {a }}$ Sol. 1 M in THF.
    ${ }^{\mathrm{b}}$ Detected by ES/MS. ${ }^{\text {c }}$ Complex mixture of products.

